

CrossMark  
click for updatesCite this: *Chem. Sci.*, 2017, 8, 411

# Thioester reduction and aldehyde transamination are universal steps in actinobacterial polyketide alkaloid biosynthesis†

U. R. Awodi,‡ J. L. Ronan,‡ J. Masschelein,‡ E. L. C. de los Santos and G. L. Challis\*

Actinobacteria produce a variety of polyketide alkaloids with unusual structures. Recently, it was shown that a type I modular polyketide synthase (PKS) is involved in the assembly of coelimycin P1, a polyketide alkaloid produced by *Streptomyces coelicolor* M145. However, the mechanisms for converting the product of the PKS to coelimycin P1 remain to be elucidated. Here we show that the C-terminal thioester reductase (TR) domain of the PKS and an  $\omega$ -transaminase are responsible for release of the polyketide chain as an aldehyde and its subsequent reductive amination. Bioinformatics analyses identified numerous gene clusters in actinobacterial genomes that encode modular PKSs with a C-terminal TR domain and a homolog of the  $\omega$ -transaminase. These are predicted to direct the biosynthesis of both known and novel polyketide alkaloids, suggesting that reductive chain release and transamination constitutes a conserved mechanism for the biosynthesis of such metabolites.

Received 24th June 2016  
Accepted 21st August 2016

DOI: 10.1039/c6sc02803a

www.rsc.org/chemicalscience

## Introduction

An astonishing array of structurally-diverse polyketide metabolites with numerous applications in medicine, animal health and agriculture are produced by actinobacteria.<sup>1</sup> Remarkable molecular machines known as type I modular polyketide synthases (PKSs) are responsible for the biosynthesis of the overwhelming majority of these natural products.<sup>2</sup> Modular PKSs employ several distinct strategies for the creation of structural diversity, including the utilization of a variety of starter and extender units for assembly of the polyketide chain,<sup>3</sup> diverse mechanisms for release of the fully-assembled chain from the PKS,<sup>4</sup> and a wide range of on- and post-PKS tailoring reactions. Among these, the mechanism of chain release is arguably of prime importance in defining which structural class a metabolite belongs to. Thus, in macrolide biosynthesis chain release proceeds *via* macrolactonization,<sup>4</sup> whereas in ansamycin biosynthesis the chain is released *via* macrolactamization;<sup>4</sup> polyether biosynthesis employs hydrolytic chain release;<sup>4</sup> (spiro)tetronates are assembled by condensation of the polyketide chain with a glyceryl thioester;<sup>5</sup> and chain release in prodiginine biosynthesis proceeds *via* nucleophilic attack of an amino acid  $\alpha$ -carbanion equivalent (see ESI† for further details).<sup>6</sup>

Incorporation experiments with isotope-labeled precursors have shown that several mono-, di-, tri- and tetracyclic alkaloids produced by actinobacteria, including latumcicin 1 (also known as abikoviromycin),<sup>7</sup> nigrifactin 2,<sup>8</sup> pyrindicin 3,<sup>9</sup> cyclizidine 4,<sup>10</sup> and streptazolin 5 (ref. 11) (Fig. 1) are derived from polyketide precursors. Several other actinobacterial alkaloids, such as streptazone E 6,<sup>12</sup> 4-hydroxy-(2-penta-1,3-dienyl)-piperidine 7,<sup>13</sup> and indolizomycin 8 (ref. 14) can similarly be hypothesized to have a polyketide origin. In 2012, we reported the identification of the unusual alkaloid coelimycin P1 9 (Fig. 1) as the metabolic product of a cryptic polyketide (*cpk*) biosynthetic gene cluster in *Streptomyces coelicolor* M145.<sup>15</sup> This

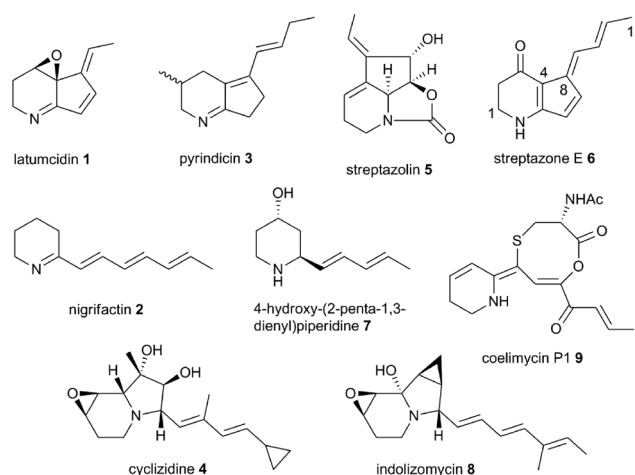


Fig. 1 Structures of polyketide alkaloids from actinobacteria.

Department of Chemistry, University of Warwick, Gibbet Hill Road, Coventry, CV4 7AL, UK. E-mail: g.l.challis@warwick.ac.uk

† Electronic supplementary information (ESI) available: Experimental procedures and additional tables, figures, chromatograms and results from biological assays. See DOI: 10.1039/c6sc02803a

‡ These authors contributed equally.



Fig. 2 The proposed roles played by the CpkABC modular PKS and the CpkG  $\omega$ -transaminase in coelimycin P1 **9** biosynthesis.

provided the first biochemical insights into the molecular mechanisms underlying actinobacterial polyketide alkaloid biosynthesis.

The *cpkA*, *cpkB* and *cpkC* genes within the coelimycin biosynthetic gene cluster encode a hexamodular PKS, which is proposed on the basis of the predicted specificities of its catalytic domains to assemble thioester **10** (Fig. 2).<sup>15</sup> This intermediate is thought to be released from the PKS by NAD(P)H-mediated reduction to the corresponding aldehyde **11**, catalyzed by the thioester reductase (TR) domain at the C-terminus of CpkC.<sup>15</sup> Reductive amination of aldehyde **11**, catalyzed by the putative  $\omega$ -transaminase encoded by *cpkG*, is proposed to give amine **12** (or its (Z)- $\Delta$ -3,4 isomer, depending on the timing of  $\Delta$ -2,3 isomerization).<sup>15</sup> Elaboration of **12** to **9** is believed to proceed *via* a series of oxidations, a cyclodehydration and bis-epoxide trapping by *N*-acetyl-L-cysteine, as illuminated by incorporation of labelled precursors.<sup>15</sup>

## Results and discussion

To experimentally investigate the role played by the CpkC TR domain in coelimycin **9** biosynthesis, it was overproduced in *E. coli* as an N-terminal His<sub>6</sub> fusion protein and purified to homogeneity using nickel affinity chromatography. UHPLC-ESI-Q-TOF-MS analysis confirmed that the purified protein had the expected molecular weight (see ESI†). The catalytic activity of the CpkC TR domain was investigated using octanoyl-CoA **13** as a mimic of the proposed natural substrate **10**. NADH and NADPH consumption was examined by monitoring the decrease in absorbance at 340 nm. A substantial decrease in absorbance with time was observed when NADH was used, whereas no decrease in absorbance was observed for NADPH (Fig. 3), showing that octanoyl-CoA is reduced by the former but not the latter. No NADH consumption was observed in control reactions from which the enzyme was omitted (Fig. 3).

In the absence of CpkG, which is proposed to intercept the octanal **14** formed by reduction of octanoyl-CoA **13** *via* transamination, CpkC-TR could catalyze further reduction of **14** to 1-

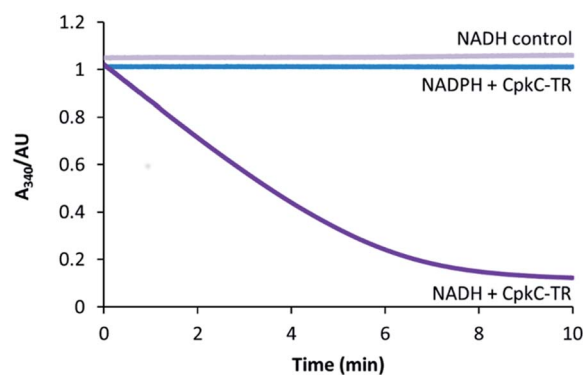
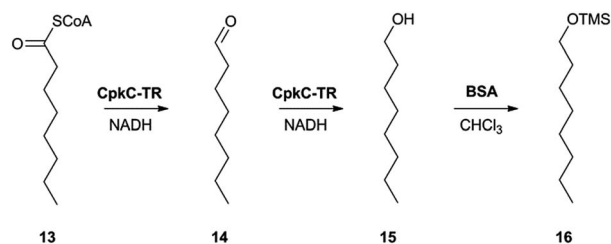


Fig. 3 Analysis of NAD(P)H consumption in the CpkC-TR-catalyzed reduction of octanoyl-CoA using a continuous spectrophotometric assay.

octanol **15** (Scheme 1). Thus, the reaction mixture containing CpkC-TR, octanoyl-CoA **13** and NADH was extracted with chloroform and derivatized with *N,O*-bis-(trimethylsilyl)acetamide (BSA) to convert any alcohols present to the corresponding trimethylsilyl (TMS) ether derivatives. GC-MS analysis of the derivatized extract identified an analyte, absent from a control reaction containing the boiled enzyme, with  $m/z = 187.2$ . This corresponds to the fragment ion of trimethylsilyl-1-octanol **16** resulting from neutral loss of a methyl group and was found to



Scheme 1 Reactions catalyzed by CpkC-TR using octanoyl-CoA **13** and octanal **14** as substrate analogues and conversion of 1-octanol **15** to its TMS ether **16** to facilitate GC-MS analysis.

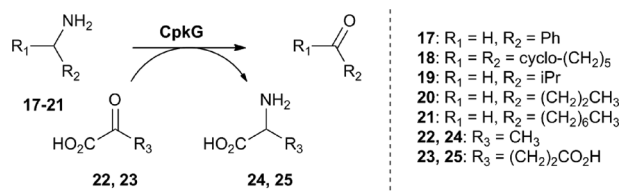




Fig. 4 Extracted ion chromatograms at  $m/z = 187.2$  (corresponding to the fragment ion of **16** resulting from loss of a methyl group) from GC-MS analyses of an authentic standard of **16** (top), the BSA-derivatized product of the CpkC-TR-catalyzed reduction of **13** with NADH (middle) and the BSA-derivatized product of a control reaction containing boiled enzyme (bottom).

have the same retention time as the  $m/z = 187.2$  ion resulting from GC-MS analysis of BSA-derivatised 1-octanol (Fig. 4). An analogous experiment in which octanal **14** was used in place of octanoyl-CoA **13** gave a similar result (see ESI†). Taken together, the data show that CpkC-TR catalyzes the NADH-mediated reduction of octanoyl-CoA **13** to octanal **14**, which is further reduced to 1-octanol **15** in the absence of a suitable transaminase to trap the transient aldehyde intermediate.

We next turned our attention to CpkG which shows sequence similarity to pyridoxal-5'-phosphate (PLP)-dependent  $\omega$ -transaminases. CpkG was overproduced in *E. coli* and purified as described for CpkC-TR. UHPLC-ESI-Q-TOF-MS analysis confirmed that the purified protein had the molecular weight expected for apo-CpkG and gel filtration chromatography showed that it is a homodimer (see ESI†). To investigate the ability of PLP to bind to CpkG, the UV-vis absorbance spectrum of the enzyme purified in the presence of the cofactor was compared to that of PLP alone. At pH 8, a characteristic  $\lambda_{\max}$  shift from 410 nm to 418 nm was observed, indicating that PLP forms an aldimine with CpkG (see ESI†).<sup>16</sup> We initially investigated whether *holo*-CpkG is able to catalyze the conversion of amines **17–21** to the corresponding aldehydes/ketones, along with concomitant conversion of either pyruvate **22** or  $\alpha$ -ketoglutarate **23** to alanine **24** or glutamic acid **25**, respectively (Scheme 2). A spectrophotometric assay that detects the production of amino acids *via* formation of a copper(II) complex



Scheme 2 Amines tested as substrates for CpkG, alongside pyruvate and  $\alpha$ -ketoglutarate as co-substrates.

was used.<sup>17</sup> All of the amines tested were found to stimulate conversion of the two  $\alpha$ -keto acids to the corresponding amino acids, demonstrating that CpkG has broad substrate tolerance (see ESI†). The highest level of activity was observed for octylamine **21** using pyruvate **22** as the co-substrate, indicating that CpkG prefers medium chain alkylamines, consistent with its proposed role in coelimycin biosynthesis. The production of octanal **15** in this reaction was confirmed by derivatization with D-cysteine and LC-MS comparison with an authentic standard of the resulting thiazolidine (see ESI†).<sup>18</sup>

The stereochemical outcome of the CpkG-catalyzed reaction of octylamine **21** with pyruvate **22** was investigated by derivatizing the alanine **24** produced with Marfey's reagent.<sup>19</sup> LC-MS comparisons with Marfey's derivatives of L- and D-alanine revealed that only L-alanine is formed (see ESI†), suggesting that CpkG has preference for L-amino acid substrates.

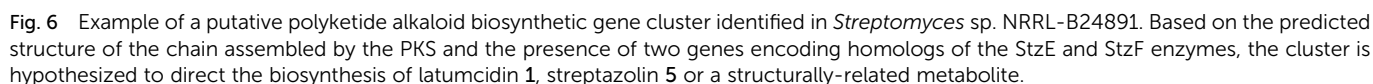
Having established that octylamine **21** and pyruvate **22** are good substrates for CpkG in the reverse of the proposed biosynthetic reaction, we investigated reductive amination of octanal **14** using alanine **24** as an amino donor. The pyruvate **22** produced was detected using a coupled assay with lactate dehydrogenase and NADH (see ESI†)<sup>20</sup> and octylamine **21** formation was confirmed by LC-MS comparisons with an authentic standard (Fig. 5). Despite the exclusive production of L-alanine in the reverse reaction (see above), CpkG was found to accept both L- and D-alanine as amino donors in the forward reaction.

Taken together, our data strongly support the key proposed roles played by CpkC-TR and CpkG in diverting the product of a modular PKS into alkaloid metabolism. We thus reasoned that homologous enzymes are likely to be ubiquitous in bacterial polyketide alkaloid biosynthesis. To investigate this hypothesis, we searched the GenBank database for gene clusters containing genes encoding a type I modular PKS with a TR domain and a CpkG homolog within 35 kb of each other (see ESI† for details). Functional annotation of the sequences obtained using antiSMASH resulted in the identification of 22



Fig. 5 Extracted ion chromatograms at  $m/z = 130.1590 \pm 0.005$  (corresponding to the  $[M + H]^+$  ion for **21**) from UHPLC-ESI-Q-TOF-MS analyses of an authentic standard of **21** (top), the CpkG-catalyzed reaction of **14** with L-alanine (middle) and a control reaction from which the enzyme was omitted (bottom).





aldehyde has been shown to be intercepted *via* transamination to form the corresponding amine.<sup>4,24–26</sup> TR-catalysed reductive chain release has also been proposed to occur in several fungal PKS systems, although direct biochemical evidence for this is currently lacking.<sup>27–31</sup> In contrast, relatively few bacterial modular PKSs have been reported, prior to this work, to contain a TR domain and to the best of our knowledge CpkC-TR is the first of these to be biochemically characterized. In common with a TR (Zmn14) that we recently showed to be responsible for chain release from a polyunsaturated fatty acid synthase-like assembly line in the biosynthesis of the zeamine antibiotics,<sup>32</sup> CpkC-TR has a strict preference for NADH. This contrasts with most other TR domains, which are able to utilize NADPH. The three residues that interact with the phosphate group of NADPH in the TR domain from the myxalamid NRPS are completely conserved in CpkC-TR and Zmn14.<sup>26</sup> Thus the structural basis for the cofactor preference of CpkC-TR and Zmn14 is unclear.

In conclusion, we have shown that reductive chain release and transamination are the key steps in the conversion of a modular PKS product to the unusual alkaloid coelimycin P1 9. Moreover, a further 22 gene clusters in actinobacteria that are known or predicted to direct the biosynthesis of diverse polyketide alkaloids have been identified. All of these encode a TR-terminated modular PKS and a CpkG homolog, suggesting a common enzymatic logic for actinobacterial polyketide alkaloid biosynthesis.

This research was supported by a Chancellor's International Scholarship (to U. R. A.), a Chancellor's Scholarship (to J. L. R.), a Marie Skłodowska-Curie Fellowship (to J. M., contract no.



656067), and grants from the BBSRC (BB/K002341/1, BB/L502017/1 and BB/M017982/1).

## Notes and references

- 1 L. Katz and R. H. Baltz, *J. Ind. Microbiol. Biotechnol.*, 2016, **43**, 155.
- 2 K. J. Weissman, *Nat. Prod. Rep.*, 2015, **32**, 436.
- 3 L. Ray and B. S. Moore, *Nat. Prod. Rep.*, 2016, **33**, 150.
- 4 L. Du and L. Lou, *Nat. Prod. Rep.*, 2010, **27**, 255.
- 5 L. Vieweg, S. Reichau, R. Schobert, P. F. Leadlay and R. D. Süssmuth, *Nat. Prod. Rep.*, 2014, **31**, 1554.
- 6 D. X. Hu, D. M. Withall, G. L. Challis and R. J. Thomson, *Chem. Rev.*, 2016, **116**, 7818–7853.
- 7 H. Seto, T. Sato, H. Yonehara and W. C. Jankowski, *J. Antibiot.*, 1973, **26**, 609.
- 8 T. Terashima, E. Idaka, Y. Kishi and T. Goto, *J. Chem. Soc., Chem. Commun.*, 1973, 75.
- 9 Y. Iwai, K. Kumano and S. Omura, *Chem. Pharm. Bull.*, 1978, **26**, 736.
- 10 F. J. Leeper, P. Padmanabhan, G. W. Kirby and G. N. Sheldrake, *J. Chem. Soc., Chem. Commun.*, 1987, 505.
- 11 M. Mayer and R. Thiericke, *J. Org. Chem.*, 1993, **58**, 3486.
- 12 Q. Liu, J. Wang, X. Wang, C. Liu, J. Zhang, Y. Pang, C. Yu and W. Xiang, *J. Asian Nat. Prod. Res.*, 2013, **15**, 221.
- 13 S. Grabley, P. Hammann, H. Kluge, J. Wink, P. Kricke and A. Zeeck, *J. Antibiot.*, 1991, **44**, 797.
- 14 S. Gomi, D. Ikeda, H. Nakamura, H. Naganawa, F. Yamashita, K. Hotta, S. Kondo, Y. Okami, H. Umezawa and Y. Iitaka, *J. Antibiot.*, 1984, **37**, 1491.
- 15 J. P. Gomez-Escribano, L. Song, D. J. Fox, V. Yeo, M. J. Bibb and G. L. Challis, *Chem. Sci.*, 2012, **3**, 2716.
- 16 R. A. Vacca, S. Giannattasio, G. Capitani, E. Marra and P. Christen, *BMC Biochem.*, 2008, **9**, 17.
- 17 B. Hwang and B. Kim, *Enzyme Microb. Technol.*, 2004, **34**, 429.
- 18 H. J. Kim and H. S. Shin, *Anal. Chim. Acta*, 2011, **702**, 225.
- 19 R. Bhushan and H. Brückner, *J. Chromatogr. B: Anal. Technol. Biomed. Life Sci.*, 2011, **879**, 3148.
- 20 M. Sova, G. Čadež, S. Turk, V. Majce, S. Polanc, S. Batson, A. J. Lloyd, D. I. Roper, C. W. G. Fishwick and S. Gobec, *Bioorg. Med. Chem. Lett.*, 2009, **19**, 1376.
- 21 T. Weber, K. Blin, S. Duddela, D. Krug, H. U. Kim, R. Bruccoleri, S. Y. Lee, M. A. Fischbach, R. Müller, W. Wohlleben, R. Breitling, E. Takano and M. H. Medema, *Nucleic Acids Res.*, 2015, **43**, W237.
- 22 S. Ohno, Y. Katsuyama, Y. Tajima, M. Izumikawa, M. Takagi, M. Fujie, N. Satoh, K. Shin-Ya and Y. Ohnishi, *ChemBioChem*, 2015, **16**, 2385.
- 23 W. Huang, S. J. Kim, J. Liu and W. Zhang, *Org. Lett.*, 2015, **17**, 5344.
- 24 Y. Li, K. J. Weissman and R. Müller, *J. Am. Chem. Soc.*, 2008, **130**, 7554.
- 25 A. Chhabra, A. S. Haque, R. K. Pal, A. Goyal, R. Rai, S. Joshi, S. Panjekar, S. Pasha, R. Sankaranarayanan and R. S. Gokhale, *Proc. Natl. Acad. Sci. U. S. A.*, 2012, **109**, 5681.
- 26 J. F. Barajas, R. M. Phelan, A. J. Schaub, J. T. Kliever, P. J. Kelly, D. R. Jackson, R. Luo, J. D. Keasling and S. Tsai, *Chem. Biol.*, 2015, **22**, 1018.
- 27 O. Salo, F. Guzman-Chavez, M. I. Ries, P. P. Lankhorst, R. A. L. Bovenberg, R. J. Vreeken and A. J. M. Driessen, *Appl. Environ. Microbiol.*, 2016, **82**, 3971.
- 28 J. Davison, A. al Fahad, M. Cai, Z. Song, S. Y. Yehia, C. M. Lazarus, A. M. Bailey, T. J. Simpson and R. J. Cox, *Proc. Natl. Acad. Sci. U. S. A.*, 2012, **109**, 7642.
- 29 T. Ugai, A. Minami, R. Fuiji, M. Tanaka, H. Oguri, K. Gomi and H. Oikawa, *Chem. Commun.*, 2015, **51**, 1878.
- 30 K. M. Fisch, E. Skellam, D. Ivison, R. J. Cox, A. M. Bailey, C. M. Lazarus and T. J. Simpson, *Chem. Commun.*, 2010, **46**, 5331.
- 31 A. M. Bailey, R. J. Cox, K. Harley, C. M. Lazarus, T. J. Simpson and E. Skellam, *Chem. Commun.*, 2007, 4053.
- 32 J. Masschelein, C. Clauwers, U. R. Awodi, K. Stalmans, W. Vermaelen, E. Lescrinier, A. Aertsen, C. Michiels, G. L. Challis and R. Lavigne, *Chem. Sci.*, 2015, **6**, 923.

