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Introduction

Gold nanoparticles have a long history in biomedicine for drug delivery and imaging.^{1–15} This is due to their comparatively easy synthesis, the absence of toxic effects, and their chemical inertness. They can be surface-functionalized by thiol- or phosphane-ligands to carry drugs, dyes, or other molecules.^{2,5,7,9,11,13,14,16–20} The surface functionalization of gold nanoparticles with nucleic acids has been pioneered by Mirkin,

Silencing of proinflammatory NF-κB and inhibition of herpes simplex virus (HSV) replication by ultrasmall gold nanoparticles (2 nm) conjugated with small-interfering RNA[†]

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Azide-terminated ultrasmall gold nanoparticles (2 nm gold core) were covalently functionalized with alkyne-terminated small-interfering siRNA duplexes by copper-catalyzed azide–alkyne cycloaddition (CuAAC; click chemistry). The nanoparticle core was visualized by transmission electron microscopy. The number of attached siRNA molecules per nanoparticle was determined by a combination of atomic absorption spectroscopy (AAS; for gold) and UV-Vis spectroscopy (for siRNA). Each nanoparticle carried between 6 and 10 siRNA duplex molecules which corresponds to a weight ratio of siRNA to gold of about 2.2 : 1. Different kinds of siRNA were conjugated to the nanoparticles, depending on the gene to be silenced. In general, the nanoparticles were readily taken up by cells and highly efficient in gene silencing, in contrast to free siRNA. This was demonstrated in HeLa-eGFP cells (silencing of eGFP) and in LPS-stimulated macrophages (silencing of NF-κB). Furthermore, we demonstrated that nanoparticles carrying antiviral siRNA potently inhibited the replication of Herpes simplex virus 2 (HSV-2) *in vitro*. This highlights the strong potential of siRNA-functionalized ultrasmall gold nanoparticles in a broad spectrum of applications, including gene silencing and treatment of viral infections, combined with a minimal dose of gold.

Alivisatos, Parak and others.^{21–25} DNA on the surface of nanoparticles also permits the construction of three-dimensional networks of nanoparticles that carry complementary DNA strands.^{26–28} If small-interfering RNA (siRNA) is attached to 13 nm gold nanoparticles by ligand exchange, they can be used for gene silencing after cellular uptake.^{29–32} RNA interference (RNAi) is a potent approach in regulating gene expression during the treatment of diseases.³³ In particular, exogenous small interfering RNAs (siRNAs) delivered into cells initiate the degradation of complementary mRNA molecules by their own cellular machinery.³⁴ In contrast, dissolved ("naked") siRNA is rapidly degraded in the presence of RNAses and not taken up by cells due to its negative charge.³⁵

Ultrasmall nanoparticles with a diameter of 1 to 2 nm are on the borderline between molecular atom-sharp clusters and "classical" plasmonic nanoparticles (typically 10 to 15 nm in diameter).^{36–43} They have a much higher specific surface area than bigger nanoparticles which permits a higher dose of attached molecules per mass of administered gold. They also do not quench the fluorescence of attached molecules.⁴⁴ The common ligand exchange reactions for thiolated DNA make it difficult to control the number of DNA molecules.²⁴ However, gold nanoparticles functionalized with siRNAs were prepared by



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[†] Electronic supplementary information (ESI) available: ESI, Fig. 1 shows images of protein bands from inflamed murine macrophages after exposure with Au-siRNA-p65-NPs. See https://doi.org/10.1039/d2na00250g

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Yamankurt *et al.* by ligand exchange:²⁹ Citrate-stabilized gold nanoparticles with a diameter of 13 nm from a modified Turkevich synthesis⁴⁵ were coated by thiolated siRNAs *via* ligand exchange in dispersion and denoted as "siRNA spherical nucleic acids".²⁹

Here, we demonstrate how click chemistry can be used for a covalent attachment of alkyne-terminated siRNA duplex molecules to the surface of azide-terminated ultrasmall gold nanoparticles.46 This method permits the attachment of alkynylated compounds like fluorescent dyes,46,47 molecular tweezers,⁴⁸ or aggregation-induced emission molecules (AIE)⁴⁹ to the particle surface. Ultrasmall gold nanoparticles are taken up by endocytosis, but they appear to use different endocytosis pathways. If one is blocked, another one opens as we have recently shown.⁵⁰ This is consistent with earlier results on nanoparticles and quantum dots of similar size as reported by Yang et al.⁵¹ and Carrillo-Carrion et al. 52 and different from the behavior of larger nanoparticles.41 Ultrasmall nanoparticles have the advantage of even penetrating cell nuclei in an in vitro 3D brain spheroid model.53 Thus, ultrasmall nanoparticles represent an interesting agent for drugs that are to be transported across the blood-brain barrier.53,54 They are usually non-toxic to cells41 due to the inert gold-sulfur bond.55 This is in line with general studies on the absence of toxicity of conventional gold nanoparticles (10-15 nm)⁵⁶⁻⁵⁸ It shall be noted that cytotoxic effects have been discussed for very small gold nanoparticles and gold clusters in eukaryotic cells,59-61 bacteria,62 and zebrafish.63 However, there are other studies (including our previous results) where it has been shown that ultrasmall gold nanoparticles are not cytotoxic.2,43,50,54,64-67 The attached ligands appear to play an important role for the cytotoxicity, especially if they can leave the gold surface.⁵⁷ In any case, the particles prepared here were not cytotoxic. So far, there are no data on the fate of the nanoparticles after cellular uptake, but an eventual excretion is likely.^{52,57,68} Due to the noble character of gold, a dissolution inside the cell or the body can be excluded.⁵⁷

We show that gold nanoparticles are advantageous as siRNA delivery vehicles into cells because they promote siRNA uptake and silence the expression of specific genes. This is demonstrated in a model system for gene silencing (HeLa-eGFP cells) and in two examples of clinical relevance where they suppress inflammation and viral infection: the silencing of NF- κ B in macrophages and the inhibition of viral replication in Herpes Simplex Virus (HSV)-infected cells.

Results and discussion

For the copper-catalyzed click reaction with different siRNAalkyne sequences, Au-GSH nanoparticles were prepared with a modified Brust-Schiffrin synthesis.⁶⁹ Azide groups were introduced by substituting the amine group in the stabilizing ligand glutathione (GSH) to Au-N₃ nanoparticles with an azidetransfer reagent.⁴⁶ Au-GSH and Au-N₃ nanoparticles were analyzed with UV-Vis-spectroscopy and disc centrifugal sedimentation (DCS). Dynamic light scattering (DLS) was not possible because the particles were too small to scatter the light, in agreement with earlier reports on Au-GSH and Au-N₃ nanoparticles.⁴⁶ UV-Vis spectroscopy was used to verify the successful synthesis of ultrasmall nanoparticles and the absence of larger nanoparticles. There was no typical surface plasmon resonance (SPR) band between 500 and 520 nm that would have indicated larger gold nanoparticles.⁴⁷ The hydrodynamic diameter of Au-



Fig. 1 Reaction scheme for the attachment of alkyne-terminated siRNA (depicted as blue balls).

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GSH and Au-N₃ nanoparticles was determined by DCS with less than 2 nm. The ligand shell influenced the particle size by DCS, thereby underestimating the actual particle size.⁷⁰ NMR spectroscopy which can be applied to ultrasmall nanoparticles⁷¹⁻⁷⁶ was used to identify and quantify GSH and N₃ bound to the ultrasmall nanoparticles.⁴⁶ Alkyne-terminated duplex siRNA was then clicked to the Au-N₃ nanoparticles. Fig. 1 shows the full reaction scheme.

We have used functional duplex siRNA as well as Cy5labelled duplex siRNA. The labelling with Cy5 was important to quantify the number of siRNA molecules on each nanoparticle and to follow the nanoparticles inside a cell by fluorescence spectroscopy. The gold nanoparticles that carry only siRNA are only very weakly fluorescent and cannot be detected inside cells. The siRNA-carrying nanoparticles were characterized by UV-Vis-spectroscopy and high-resolution transmission electron microscopy (HRTEM). DCS turned out to be not applicable because the large organic shell reduced the effective density of the nanoparticle that no sedimentation occurred at the given rotational speed of the centrifuge within reasonable times (>12 h). Dynamic light scattering (DLS) did not show the ultrasmall nanoparticle (core too small) but again confirmed the absence of larger (plasmonic) or agglomerated gold nanoparticles. The nanoparticle gold core was analyzed by HRTEM. Fourier transformation analysis gave the *d*-spacings of elemental gold (*fcc*). Due to the low scattering contrast of the organic ligands, only the metallic core was visible in HRTEM. HRTEM showed a mostly uniform shape of the gold core without agglomerates. An average diameter of 2.2 ± 0.6 nm for Au-siRNA-eGFP nanoparticles and 2.2 ± 0.5 nm for Au-siRNAeGFP-Cy5 nanoparticles was found (Fig. 2). We have demonstrated earlier that the click reaction does not influence the size of the gold core,⁴⁶⁻⁴⁸ a result that is confirmed here.

A UV-Vis-spectrum of Au-siRNA-eGFP and Au-siRNA-eGFP-Cy5 nanoparticles showed the absorption band of the attached siRNA molecules (absorption band from 228 to 303 nm, Fig. 3) and confirmed the covalent binding of siRNA to the azide-terminated gold nanoparticles. Again, no absorption by larger (plasmonic) gold nanoparticles was found. For each synthesis, 129 nmol of alkyne-terminated siRNA and 11.9 nmol of azide-terminated gold nanoparticles were used (molar ratio 11 : 1), assuming a surface loading as with molecular tweezers.⁴⁸ The click efficiency of the reaction was determined by analysis of the siRNA-loaded gold nanoparticles. Between 6 and 10 siRNA molecules were attached to the surface of each nanoparticle, with good agreement of the data from the UV absorption of RNA and Cy5. Depending on the



Fig. 2 High-resolution transmission electron micrographs of Au-siRNA-eGFP (A) and Au-siRNA-eGFP-Cy5 (B) nanoparticles and corresponding particle size distribution diagrams.



Fig. 3 UV-Vis spectra of azide-terminated gold nanoparticles (Au-N₃, left) and of siRNA-functionalized gold nanoparticles (right). The absorption of RNA is visible by the band at 258 nm. The absorption of Cy5 is visible by the band at 647 nm.

siRNA used, at least 58% of the applied siRNA-alkyne was bound to the nanoparticle. This is of considerable importance due to the high price of siRNA. The recovery of gold was between 60 and 90%, i.e. also comparatively high. The number of clicked molecules determines the molecular footprint on the nanoparticle surface. For a spherical nanoparticle with an average diameter of 2 nm, the nominal surface area is 12.6 nm².⁴⁶ Thus, the siRNA ligands had a molecular footprint between 1.35 and 1.96 nm², in good agreement with clicked dye molecules like FAM (1.6 to 2.1 nm²),^{46,47} or AlexaFluor-647 (1.15 nm²).⁴⁶ Lach et al. found a footprint of 0.20 nm² for quaternary ammonium ligands on 3.4 nm gold nanoparticles.⁷⁷ This is also in line with Girard et al. who showed that a ssDNA molecule (18 bases) required a surface area of about 1 nm² on a 2 nm gold nanoparticle, *i.e.* geometrically, each 2 nm-nanoparticle can carry about 12 ssDNA molecules.²⁴ All analytical data of the nanoparticles are summarized in Table 1. The nanoparticles have a composition of about $Au_{\sim 250}$ - $GSH_{\sim 125}(siRNA)_8$. With molar masses of Au (197 g mol⁻¹), GSH (306 g mol^{-1}) and siRNA (~14 000 g mol⁻¹), we obtain a molar mass of one nanoparticle of about 199 500 g mol⁻¹ with weight fractions of Au = 25 wt%, GSH = 19 wt%, and siRNA = 56 wt%.

This is a considerably higher loading of siRNA than that reported for "standard" gold nanoparticles prepared by the Turkevich method (citrate reduction), followed by ligand exchange.⁴⁵ Yamankurt *et al.* prepared siRNA-coated gold nanoparticles (13 nm, "spherical nucleic acids") with a maximum of 98 siRNA duplex molecules per particle.²⁹ This corresponds to a molecular footprint of about 5.4 nm² per siRNA duplex. In comparison, the surface density of siRNA after clicking is about 3 to 4 times higher than after ligand exchange. In terms of the particle composition, the composition of a 13 nm gold nanoparticle is about $Au_{70\ 000}(siRNA)_{98}$. This leads to a molar mass of one nanoparticle of about $15.1 \cdot 10^6$ g mol⁻¹ with weight fractions of Au = 91 wt% and siRNA = 9 wt% (assuming 14 kDa per siNA duplex as in our case). In terms of siRNA loading, ultrasmall nanoparticles carry about 2.2 : 0.1 = 22 times more siRNA per mass of gold applied.

siRNA for silencing the eGFP gene is a useful model because the extent of gene silencing can be easily demonstrated by analyzing the eGFP fluorescence intensity by fluorescence microscopy. Moreover, we also focused on siRNA for silencing nuclear factor-kappaB (NF- κ B) dimer gene expression, since such gene products are important cellular players responsible for many regulatory processes including inflammatory and cancerous diseases.⁷⁸⁻⁸²

The uptake of fluorescently labelled Au-siRNA-eGFP-Cy5 nanoparticles was observed with confocal laser scanning microscopy (CLSM) HeLa-eGFP cells. All slices shown are taken

 Table 1
 Analytical data of all prepared gold nanoparticles. Quantification was done assuming an average particle diameter of 2 nm and spherical nanoparticles. The gold content in the final products was calculated and compared to the gold amount used in the synthesis

	Ligands per nanoparticle (¹ H-NMR and UV-Vis spectroscopy)	Recovery of gold/%	Yield of siRNA attachment/%	Gold core diameter (HRTEM)/nm	Ligand footprint/nm ²	Hydrodynamic diameter (DCS)/nm
Au-GSH ⁴⁶	125 (¹ H-NMR)	72	_	1.7 ± 0.9	0.10	1.5 ± 0.2
Au-N ₃ (ref. 46)	118 (¹ H-NMR)	86	94	2.0 ± 0.4	0.11	1.5 ± 0.2
Au-siRNA-eGFP	9.5 (UV-Vis)	89	86	2.2 ± 0.6	1.35	_
Au-siRNA-eGFP-Cy5	8.9 (UV-Vis, RNA)/8.7 (UV-Vis, Cy5)	79	81 (RNA)/ 79 (Cy5)	2.2 ± 0.5	1.41 (RNA)/ 1.45 (Cy5)	—
Au-siRNA-eGFP-r	8.3 (UV-Vis)	75	76	_	1.50	_
Au-siRNA-p65	8.7 (UV-Vis)	62	79	_	1.45	_
Au-siRNA-18.1	6.4 (UV-Vis)	76	58	_	1.96	_
Au-siRNA-29.2	6.8 (UV-Vis)	75	63	_	1.83	_

Fig. 4 Confocal laser scanning images showing the uptake of ultrasmall Au-siRNA-eGFP-Cy5 nanoparticles by HeLa-eGFP cells. The cells were incubated with 5 μ g mL⁻¹ gold nanoparticles per well (13.7 μ g siRNA mL⁻¹) for 24 h, washed, fixed, and stained. Control: untreated HeLa-eGFP cells (no nanoparticles). Nuclei: stained with DAPI; actin: stained with AlexaFluorTM 568 Phalloidin; Cy5: Au-siRNA-eGFP-Cy5 nanoparticles. Scale bars are 20 μ m. Data represent the mean of three individual experiments with the standard deviation (N = 3).

from the focal plane in the middle of the whole cell volume (*z*-stacks). Based on the *z*-stacks, the nanoparticles were well taken up by all cells but did not enter the cell nucleus. They were found in the cytoplasm or decorated the nucleus (Fig. 4).

The viability of the nanoparticle-treated HeLa-eGFP cells was determined with an MTT cell viability assay. The siRNA-loaded nanoparticles did not lead to a significant change in cell viability when compared to the untreated control. In contrast to the nanoparticles, Lipofectamine/siRNA-based formulations showed a significant decrease in cell viability. This is well known and due to the adverse nature of Lipofectamine (Fig. 5).⁸³

The nanoparticles were able to silence the production of eGFP in HeLa-eGFP cells as shown by fluorescence microscopy (Fig. 6). The mean fluorescence intensity (MFI) in these images was used to quantify the eGFP gene silencing efficiency. Only fully confluent parts of a well were monitored. Interestingly, AusiRNA-eGFP-Cy5 nanoparticles remained for a much longer time inside the cells than Lipofectamine-transported siRNAeGFP-Cy5 which was barely visible after the incubation time of 72 h. The fluorescent label Cy5 on Au-siRNA-eGFP-Cy5 nanoparticles decreased the gene silencing efficiency. This underscores the fact that labels on biomolecules may compromise their biological effect. A possible explanation could be that the functionalization of an alkyne and Cy5 at the 5'- and 3'-ends of the siRNA sequence made the siRNA inaccessible for the RISC. Thus, the separation of the sense and antisense strands could have been blocked by the design of the siRNA-eGFP-Cy5 (for sequence designs see Table 2) when coupled to the surface of the gold nanoparticles. It has been reported that the conjugation of a fluorophore to a bioactive molecules can change its subcellular localization and function.84

Based on the fluorescence microscopy images as shown in Fig. 6 the mean fluorescence intensity data were calculated and are presented in Fig. 7A. The nanoparticles were remarkably efficient in silencing the eGFP gene in HeLa-eGFP cells, but still less efficient than Lipofectamine (Fig. 7B). However, it is advantageous that they were considerably less cytotoxic than Lipofectamine. Higher nanoparticle concentrations did not substantially

increase the gene silencing efficiency. There was also no significant difference between the nanoparticles where the antisense strand was attached to the gold (siRNA-eGFP) compared to the case when the sense strand was attached to the gold (siRNA-eGFPr). This might have been expected because the antisense strand is considered to be the active agent, but is obviously not the case.⁸⁵

Yamankurt *et al.* have studied the mechanism of the intracellular processing of gold-conjugated siRNA in detail. They

Fig. 5 MTT viability assay with HeLa-eGFP cells, incubated with Lipofectamine/siRNA-eGFP or water-dispersed Au-siRNA-eGFP nanoparticles. Nanoparticle dispersions were tested at three concentrations, *i.e.* c_1 : 14.8 µg siRNA mL⁻¹ (13.7 µg siRNA mL⁻¹ for siRNA-eGFP-Cy5); c_2 : 74 µg siRNA mL⁻¹; c_3 : 148 µg siRNA mL⁻¹. The incubation time was 24 h, except for Lipofectamine where it was 5 h. Control: untreated HeLa-eGFP cells. One-way ANOVA was used, followed by Tukey's comparison test: n.s., not significant; *p < 0.05; **p < 0.01; ***p < 0.001. Data represent the mean of three individual experiments with the standard deviation (N = 3).

Cy5

Brightfield eGFP Control Lipofectamine siRNA-eGFP siRNA-eGFP-Cy5 Lipofectamine Au-siRNAeGFP Au-siRNAeGFP-Cy5

Fig. 6 Representative fluorescence images showing the silencing of eGFP expression in HeLa-eGFP cells by Lipofectamine/siRNA-eGFP polyplex ($3.2 \ \mu g \ mL^{-1} \ siRNA/Lipofectamine$) and by Au-siRNA-eGFP ($14.8 \ \mu g \ siRNA \ mL^{-1}$) or Au-siRNA-eGFP-Cy5 ($13.7 \ \mu g \ siRNA \ mL^{-1}$) nanoparticle treatment. Data represent the mean of three individual experiments with the standard deviation (N = 3).

reported that the enzyme Dicer was cleaving the duplex from the gold surface, thereby liberating the siRNA for intracellular gene silencing.²⁹ It is likely that a similar mechanism occurs with siRNA-functionalized ultrasmall nanoparticles as well. This could also explain the higher stability of nanoparticle-conjugated siRNA towards enzymatic degradation in comparison to dissolved siRNA. They reported gene silencing efficiencies of the protein HER2 in SKOV-3 cells to about 15% on the RNA level by nanoparticle treatment. However, they also reported that the nanoparticles had a significant cytotoxicity if applied at doses above about 1 nmol L^{-1} .²⁹

As medically relevant protein, we chose nuclear factor (NF)- κ B. This protein plays an important role in inflammationassociated diseases because numerous NF- κ B target genes encode proinflammatory proteins.⁸² The principal means to control the NF- κ B activity is by retention of inactive forms of heterogeneous transcription factor proteins in the cell's cytoplasm by inhibitors of κ B (I κ B) molecules.⁷⁸ Upon activation by a diverse range of stimuli (immunoreceptors, cytokines, growth factors, and other stress impulses),⁸⁰ the freely released NF- κ B proteins immediately form homo- and hetero-conjugations, which bind to specific DNA regions. The active NF- κ B protein

 Table 2
 Sequences of all siRNA duplexes used for gene silencing. The given molar masses refer to the duplexes, including alkyne and dye groups where applicable

siRNA code	siRNA target	Sequence 5'-3' of sense and antisense strand of duplex siRNA
siRNA-eGFP	eGFP	rGrCrArArGrCrUrGrArCrCrCrUrGrArArGrUrUrCrArU (sense)
$(M = 14 \ 414 \ \mathrm{g \ mol}^{-1})$	eGFP	5-Octadinyl dU/rArUrGrArArCrUrUrCrArGrGrGrUrCrArGrCrUrUrGrC (antisense)
siRNA-eGFP-Cy5	eGFP	5-Cy5/rGrCrArArGrCrUrGrArCrCrCrUrGrArArGrUrUrCrArU (sense)
$(M = 14 \ 946 \ \mathrm{g} \ \mathrm{mol}^{-1})$	eGFP	5-Octadinyl dU/rArUrGrArArCrUrUrCrArGrGrGrUrCrArGrCrUrUrGrC (antisense)
siRNA-eGFP-r	eGFP	5-Octadinyl dU/rGrCrArArGrCrUrGrArCrCrCrUrGrArArGrUrUrCrArU (sense)
$(M = 14 \ 414 \ \mathrm{g \ mol}^{-1})$	eGFP	rArUrGrArArCrUrUrCrArGrGrGrUrCrArGrCrUrUrGrC (antisense)
siRNA-p65	NF-κB	5-Octadinyl dU/rCrCrArUrGrGrArGrUrUrCrCrArGrUrArCrUrUTT (sense)
$(M = 13 694 \text{ g mol}^{-1})$	NF-ĸB	rArArGrUrArCrUrGrGrArArCrUrCrCrArUrGrGTT (antisense)
siRNA-18.1	HSV-UL18.1	5-Octadinyl dU/rGrCrArCrCrGrUrUrArArCrCrUrUrCrGrCrArATT (sense)
$(M = 13 694 \text{ g mol}^{-1})$	HSV-UL18.1	rUrUrGrCrGrArArGrGrUrUrArArCrGrGrUrGrCTT (antisense)
siRNA-29.2	HSV-UL29.2	5-Octadinyl dU/rCrUrUrUrCrGrCrArArUrCrArArUrUrCrCrArA (sense)
$(M = 12 \ 448 \ \mathrm{g \ mol}^{-1})$	HSV-UL29.2	rUrUrGrGrArArUrUrGrArUrUrGrCrGrArArArG (antisense)

monomers are termed p50, p52, p65 (RelA), c-Rel, and RelB. Especially the Rel proteins can form several different NF- κ B dimers. Of these, the p50/p65 heterodimer is physiologically the most abundant.⁷⁹ Importantly, the combination of subunits within the NF- κ B protein components results in distinct DNA-sequence-specific transactivation properties.⁸¹

It has been shown that a key regulator of macrophage behavior is NF- κ B signaling, *via* nuclear translocation and binding of p65/p50 transcription factors. In consequence, there is increasing evidence that increased NF- κ B activity in macrophages polarizes them towards an M1-like phenotype.^{86–88} In our study after incubation of inflamed (LPS-primed) macrophages with Au-siRNA-p65 nanoparticles carrying anti-p65 NF- κ B siRNA (1 µg siRNA mL⁻¹, 72 h), a considerable number of cells was shown to be nanoparticle-positive (Fig. 8A; see also ESI, Fig. 1†). We chose to target p65 NF- κ B siRNA over other NF- κ B dimers on the basis of our previous experience with calcium phosphate nanoparticles for gene silencing.^{89–91} Moreover, the presence of nanoparticles did not reduce the viability of the target cells (Fig. 8B). Protein expression analyses revealed a statistically significant depression not only of p65 NF-κB, but also of other NF-κB dimers like p50, p100, and p105 (the latter two not significant). In contrast, the impact on p52 was inconspicuous (Fig. 7C). Since the mentioned NK-κB proteins are responsible for encoding numerous pro-inflammatory proteins, cytokines, and growth factors,⁹² we expect that the nanoparticle-mediated anti-p65 siRNA inhibition in macrophages will lead to a decreased pro-inflammatory activity. In particular, the protein expression of the two most common NF-κB dimer components p65 and p50 was depressed. These are known to regulate many physiological activities.⁷⁹

Finally, we demonstrate that gold nanoparticles carrying siRNA can be used to combat viral infections by gene silencing. Herpes simplex viruses type 1 (HSV-1) and type 2 (HSV-2) belong

Fig. 7 (A) Normalized mean fluorescence intensity as calculated from cell culture images as shown in Fig. 6. (B) Normalized fold decrease of eGFP gene expression in HeLa-eGFP cells. Nanoparticle dispersions were tested in three concentrations, *i.e.* c_1 : 14.8 µg siRNA mL⁻¹; c_2 : 74 µg siRNA mL⁻¹; c_3 : 148 µg siRNA mL⁻¹. One-way ANOVA was used, followed by Tukey's comparison test: n.s., not significant; *p < 0.05; **p < 0.01; ***p < 0.001. Data represent the mean of three individual experiments with the standard deviation (N = 3).

Fig. 8 Impact of Au-siRNA-p65 nanoparticles in inflamed murine macrophages. (A) Light microscopy image of inflamed macrophages (J774.A1, LPS-primed) after nanoparticle exposure. (B) Cell viability after nanoparticle exposure. (C) Protein expression (western and immunoblot) relative to GAPDH of different NF- κ B members, T: inflamed cells treated with anti-p65 NK- κ B Au-siRNA-p65 nanoparticles (1 μ g siRNA mL⁻¹, 72 h), S: inflamed cells treated with Lipofectamine/siRNA-p65 (1 μ g siRNA mL⁻¹, 72 h), I: non-treated inflamed cells, C: non-treated, non-inflamed cells. *: statistically different compared to I with *p* < 0.05 (*) or *p* < 0.01 (**) (Student's *t*-test). Data represent the mean of three individual experiments with the standard deviation (*N* = 3).

to the most common viruses affecting humans.93,94 Upon primary infection, the viruses establish a life-long latent infection in the sensory neurons.95,96 Periodic reactivations of the viruses may lead to genital, orofacial, or ocular infections that are clinically indistinguishable between HSV-1 and HSV-2.97 Most recurrent episodes are mild and self-limiting.98 However, besides severe, life-threatening reactivations in immunocompromised patients,99 HSV-infections of the eye represent a serious risk to human health.¹⁰⁰ Frequent recurrent manifestations of the virus at the cornea may lead to the development of herpetic stromal keratitis and finally to blindness.¹⁰⁰ The increasing incidence of acyclovir (ACV)- and multidrug-resistant strains in patients with corneal HSV-infections makes the treatment increasingly challenging.101,102 Clearly, novel strategies for the topical treatment of HSV-infections of the cornea are urgently needed. Paradowska et al. have recently reported that citrate-coated 10-15 nm gold nanoparticles can interact with HSV-1 in an unspecific way.103

To investigate whether siRNA-functionalized gold nanoparticles can be used to treat viral infections like genital HSVinfections of the eye, we examined the antiviral efficacy of AusiRNA-18.1 or Au-siRNA-29.2 nanoparticles against HSV-2 in cell culture, respectively. siRNA-18.1 targets the capsid encoding gene UL18 ¹⁰⁴ and siRNA-29.2 targets the DNA-binding protein UL29.¹⁰⁵ We found that Au-siRNA-18.1 or Au-siRNA-29.2 both strongly inhibited HSV-2 replication in cell culture in a dose-dependent way (Fig. 9). A complete inhibition of the viral replication was observed at a concentration of 2 pmol μ L⁻¹ Au-siRNA-18.1 (26 μ g siRNA mL⁻¹) and 4 pmol μ L⁻¹ Au-siRNA-29.2 (52 μ g siRNA mL⁻¹). In contrast, dissolved siRNA-18.1 and siRNA-29.2 had no effect on the viral replication. These results clearly emphasize that siRNA-functionalized gold nanoparticles are a promising platform for the development of novel anti-invective drugs. Conventionally, siRNAs such as anti-HSV-siRNA 18.1 or 29.2 need to be formulated with transfection reagents to achieve uptake into target cells.^{104,105} siRNA-functionalized gold nanoparticles clearly avoid the need for an additional transfection agent to achieve the desired gene silencing.

The copper-catalyzed azide–alkyne click reaction (CuAAC) obviously does not lead to a structural damage of the siRNA as it is still able to silence the target genes, demonstrated here for a number of cases. Whereas a detailed investigation of the chemical integrity of the clicked siRNA is difficult to impossible (NMR does not work; detachment may lead to chemical damage as well), it is obvious that at least a significant fraction of siRNA has preserved its chemical nature. This is supported by literature data on click reactions of nucleic acids by CuAAC. It has been reported that CuAAC as a mild and orthogonal click reaction as defined by Sharpless¹⁰⁶ does not damage nucleic acids and is therefore a suitable means for their modification.¹⁰⁷⁻¹¹⁰

Fig. 9 Inhibition of HSV-2 replication by Au-siRNA-18.1 and Au-siRNA-29.2. Vero cells were pre-incubated for 4 h with various concentrations of Au-siRNA-18.1 or Au-siRNA-29.2 nanoparticles (0.25 pmol siRNA μ L⁻¹ to 4 pmol siRNA μ L⁻¹) and subsequently infected with 25 TCID₅₀ units of HSV-2 GFP reporter virus. siRNA-18.1 and siRNA-29.2 dissolved in cell culture medium ("naked") at a concentration of 4 pmol siRNA μ L⁻¹ were used as control. After 48 h, the cytopathic effects (CPE) were assessed by fluorescence microscopy. Representative immunofluorescence images acquired at a 100-fold magnification from three independent experiments (*N* = 3) are shown. Scale bar 200 μ M.

Conclusions

Ultrasmall gold nanoparticles (2 nm core diameter) can be functionalized with siRNA duplexes by clicking them to the particle surface with a surface density that is 3 to 4 times higher compared to ligand exchange reactions. Due to the small particle diameter, the weight ratio of siRNA to gold is about 22 times higher than with "classical" 13 nm gold nanoparticles. The particles were easily taken up by HeLa cells and led to a strong silencing of eGFP in HeLa-eGFP cells. This effect is therapeutically relevant to treat inflammations as demonstrated with LPS-stimulated macrophages. Their treatment with gold nanoparticles carrying siRNA led to a significant silencing of the two most common NF-κB dimer components p65 and p50. Furthermore, the treatment of Vero cells that were previously infected with herpes simplex virus 2 (HSV-2) with such nanoparticles fully inhibited HSV-2 replication in a dose-dependent way. The fact that the nanoparticle-attached siRNA was still active in gene silencing indicates that it was not significantly damaged by the click reaction. Thus, siRNA functionalized Aunanoparticles Au-siRNA-18.1 and Au-siRNA-29.2 represent an efficient tool for the topical treatment of HSV-infections such as ACV-resistant genital herpes or HSV-infections of the cornea. The cytotoxicity of the nanoparticles was generally low at the applied doses. We conclude first that surface-clicking is a superior method to attach siRNA to a gold nanoparticle surface. Second, ultrasmall gold nanoparticles are especially suitable to transport high amounts of siRNA into cells for gene silencing with a minimum gold dose.

In control studies, the dissolved siRNA did not lead to the desired gene silencing effect because it was not taken up by the cells. We ascribe the fact that nanoparticle-attached siRNA is much more active in gene silencing than dissolved siRNA to two reasons. First, it can enter a cell together with a nanoparticle much more easily than in dissolved form due to the nanoparticle-specific uptake mechanisms. Second, it appears to be better protected against enzymatic degradation by RNAses. The reason for this effect is still unknown, but steric effects associated with the dense packing of nucleic acids on the nanoparticle surface could play an important role.

Materials and methods

Chemicals and reagents

Elemental gold (>99%) was dissolved in *aqua regia* to prepare tetrachloroauric acid (HAuCl₄). Hydrochloric acid (HCl, 37%), nitric acid (HNO₃, 67%), and sodium hydroxide (NaOH, 1 M) were obtained from Bernd Kraft (Duisburg, Germany). Sodium borohydride (NaBH₄, >96%), sodium L-ascorbate (>99%), copper(II) sulfate pentahydrate (>99%), deuterium oxide (D₂O, 99%), sodium ascorbate (99%), and Spin-X[®] UF 30k MWCO PES spin filters (Corning[®]) were obtained from Sigma-Aldrich (Steinheim, Germany). Sucrose (99%) was obtained from VWR Chemicals (Langenfeld, Germany). Methanol (99.8%) was obtained from Fisher Scientific (Geel, Belgium). Aminoguanidine hydrogen carbonate (98%) was obtained from Alfa Aesar (Kandel, Germany). Tris(3-hydroxypropyltriazolylmethyl)amine (THPTA, >95%) was obtained from Jena Bioscience (Jena, Germany). Dodecane (99%)

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and 2-(4,5-dimethylthiazol-2-yl)-3,5-diphenyl-2*H*-tetrazol-3-ium bromide (MTT), and InvitrogenTM LipofectamineTM 2000 transfection reagent were obtained from Thermo Fisher Scientific (Schwerte, Germany). Dimethyl sulfoxide (DMSO, >99.5%) was obtained from Carl Roth (Karlsruhe, Germany). The PVC nanoparticle calibration dispersion (Lot#149, 1.385 g L⁻¹) for differential centrifugal sedimentation (DCS) was obtained from CPS Instruments Inc. (Oosterhout, The Netherlands).

All siRNA duplexes were obtained from Integrated DNA Technologies (Coralville, Iowa, USA) in purified form (standard desalting or HPLC purification for dye-modified sequences). They were obtained with an alkyne group for clicking and in one case with an additional Cy5 group for fluorescent labelling. All siRNA sequences are given in Table 2. Note that siRNA against eGFP was tested in two sequence modifications. siRNA was attached to the nanoparticle at the 5' ends of the sequences, *i.e.* the antisense strand (siRNA-eGFP) or the sense strand (siRNA-eGFP-r) (r stands for reverse).

For the synthesis and purifications of Au-GSH nanoparticles, ultrapure water (Purelab ultra instrument from ELGA, Celle, Germany) was used. All glassware was cleaned with boiling *aqua regia* and washed twice with ultrapure water before use. Ultra-PureTM water (distilled water, DNase/RNase free, Invitrogen) was used for the reconstitution of siRNA. For synthesis, purifications, and analyses of Au-N₃ and all Au-siRNA nanoparticles, Ultra-PureTM water was used. To create conditions that were as RNase-free as possible after the synthesis of the Au-GSH nanoparticles, all solutions were prepared with sterile pipette tips in sterile vessels and filtered using non-pyrogenic sterile syringe filters (0.2 μ m, Sarstedt). In addition, all glassware was sterilized before use at 200 °C for at least 2 h before use. All working surfaces were treated beforehand with RNase AWAYTM (Carl Roth).

Methods

Atomic absorption spectroscopy (AAS) was used to determine the gold concentration of the nanoparticle dispersion. Nanoparticle dispersions (5 μ L) were dissolved in *aqua regia* (450 μ L) and diluted with ultrapure water (3.5 mL) for each measurement. A Thermo Electron M-Series spectrometer (graphite tube furnace; operated according to DIN EN ISO/IEC 17025:2005) was used.

Analytical disc centrifugation (differential centrifugal sedimentation; DCS) was carried out with a CPS Instruments DC 24000 disc centrifuge (24 000 rpm, 29 000 relative centrifugal force; rcf). A density gradient of sucrose solutions with different concentrations (8 and 24 wt%) was applied. For stabilization, the solution was covered with 0.5 mL dodecane. The calibration was performed before each run with 100 μ L of a dispersion of poly(vinyl chloride) (PVC) latex in water with a defined hydrodynamic diameter of 483 nm. 100 μ L nanoparticle dispersion was used for each run. To compute the hydrodynamic diameter, the density of elemental gold (19 300 kg m⁻³) was used.

UV-Vis spectroscopy was performed with a Genesis 50 instrument (Thermo Scientific) in quartz glass cuvettes after background correction from 200 nm to 800 nm. Sample volumes of 600 μ L were analyzed.

High-resolution transmission electron microscopy (HRTEM) was performed with an aberration-corrected FEI Titan transmission electron microscope equipped with a Cs-probe corrector (CEOS Company), operating at 300 kV.¹¹¹ The nanoparticle dispersion was drop-cast on a copper grid, coated with an ultrathin amorphous carbon film.

Synthesis of gold nanoparticles

Ultrasmall gold nanoparticles (diameter 2 nm) were covalently functionalized with siRNA as described below, following the procedure for click reaction reported earlier in detail.46 First, glutathione-stabilized gold nanoparticles (Au-GSH) were synthesized by reduction of tetrachloroauric acid with sodium tetraborohydride. Next, they were then modified by converting the amino group of the ligand into an azide group.⁴⁶ Azideterminated nanoparticles (Au-N₃; 118 azide groups per 2 nm gold nanoparticle) were then conjugated with alkyneterminated double-stranded siRNA by copper-catalyzed azidealkyne cycloaddition (CuAAC).46 This ensured the formation of a stable covalent bond between siRNA and particle surface. Au-N₃ nanoparticles (1 eq., 1.4 µmol azide groups, 11.9 nmol Au-N₃, 0.58 mg Au) and (as example) alkyne-terminated siRNA-eGFP (0.09 eq. to N_3 , 129 nmol, 1.86 mg) were dissolved in 100 μ L of water. A solution of copper sulphate (CuSO₄, 7.5 µmol, 1.2 in mL water together with mg) 6 tris(3hydroxypropyltriazolylmethyl)amine (THPTA, 119 µmol, 63 mg) and aminoguanidine hydrogen carbonate (7.5 µmol, 1.0 mg) was prepared. 98 μ L of this solution (CuSO₄, 0.07 eq. to N₃, 15.7 µg; THPTA, 2.2 eq. to N₃, 1.65 mg; aminoguanidine hydrogencarbonate, 0.07 eq. to N_3 , 13.3 µg) were added to the dispersion to start the click reaction. Next, 28.1 µL of a 10 mM sodium ascorbate solution (0.2 eq. to N_3 , 55.6 µg) was added. The dispersion was stirred overnight for 16 h. Then, the nanoparticle dispersion was centrifuged five times by 30 kDa Spin-X® UF spin filters at 4000 rpm (2500 g) for 45 min to remove any unbound siRNA. The full characterization data of Au-GSH and Au-N₃ nanoparticles, including the calculation of the number of GSH groups per Au-NP (125) and of N₃ groups per gold nanoparticle (118), has been reported in ref. 46, including 1D and 2D NMR spectra.

The following alkyne-terminated siRNAs were clicked to Au- N_3 nanoparticles in this way: siRNA-eGFP-Cy5 (0.09 eq. to N_3 , 129 nmol, 1.93 mg), siRNA-eGFP-r (0.09 eq. to N_3 , 129 nmol, 1.86 mg), siRNA-p65 (0.09 eq. to N_3 , 129 nmol, 1.77 mg), siRNA-18.1 (0.09 eq. to N_3 , 129 nmol, 1.77 mg), and siRNA-29.2 (0.09 eq. to N_3 , 129 nmol, 1.61 mg).

For all types of nanoparticles, the gold content of the nanoparticle dispersions was determined by atomic absorption spectroscopy (AAS) and converted to the number of gold nanoparticles, using the density of gold and the average particle diameter (solid core) of 2 nm. As each gold nanoparticle has a mass of $8.08 \cdot 10^{-20}$ g, 1 g of gold nanoparticles corresponds to $1.24 \cdot 10^{19}$ particles.⁴⁶ The amount of siRNA was measured by UV-Vis spectroscopy with calibration curves of siRNA-alkyne solutions. The quantification of siRNA was performed by the absorbance of ribonucleic acids (integral from 228 to 303 nm),

and additionally with Cy5 for siRNA-eGFP-Cy5 (integral from 500 to 700 nm). To determine the concentration of the gold nanoparticles, atomic absorption spectroscopy (AAS) was used as described above. The ratio of siRNA molecules to gold nanoparticles (diameter 2 nm) in a given dispersion gave the number of siRNA molecules per nanoparticle. For a typical sample, the siRNA concentration per nanoparticle was calculated as follows:

Concentration of gold nanoparticles (2 nm) from the molar gold concentration by AAS:

$$N(\text{NP}) \ \left[\text{m}^{-3}\right] = \frac{3 \ c(\text{Au})}{4 \cdot \pi \cdot r(\text{NP})^3 \cdot \rho(\text{Au})} \tag{1}$$

$$\frac{3 \cdot 0.141 \cdot 10^{-3} \text{ kg m}^{-3}}{4 \cdot \pi \cdot (1 \cdot 10^{-9} \text{ m})^3 \cdot 19 \ 320 \text{ kg m}^{-3}} = 1.74 \cdot 10^{18} \text{ m}^{-3} \qquad (2)$$

$$c(\mathrm{NP})[\mathrm{mol}\ \mathrm{m}^{-3}] = \frac{N(\mathrm{NP})}{\mathrm{N}_{\mathrm{A}}} \tag{3}$$

$$\frac{1.74 \cdot 10^{18} \text{ m}^{-3}}{6.02 \cdot 10^{23} \text{ mol}^{-1}} = 2.90 \cdot 10^{-6} \text{ mol m}^{-3}$$
(4)

Concentration of Cy5 in an Au-siRNA-Cy5 dispersion (1 Cy5 per siRNA) by UV spectroscopy, using Lambert–Beer's law after recording previous calibration curves (500–700 nm):

$$c(\text{siRNA-Cy5})_{\text{Cy5 extinction}} [\text{mol m}^{-3}] = \frac{\int_{500 \text{ nm}}^{700 \text{ nm}} (\text{extinction}) \cdot V_{\text{f}} \cdot V_{\text{i}}^{-1}}{\text{calibration slope}}$$
(5)

$$\frac{28.57 \cdot 0.3 \cdot 10^{-6} \text{m}^{-3} \cdot 0.3 \cdot 10^{-7} \text{m}^3}{11.4 \cdot 10^6 \text{ mol }^{-1} \text{m}^3} = 25.1 \cdot 10^{-6} \text{ mol } \text{m}^{-3}$$
(6)

Concentration of siRNA in an Au-siRNA-Cy5 dispersion by UV spectroscopy, using Lambert–Beer's law after recording previous calibration curves (234–297 nm; absorption of nucleobases):

$$c(\text{siRNA-Cy5})_{\text{RNA extinction}} [\text{mol } \text{m}^{-3}] = \frac{\int_{234 \text{ nm}}^{297 \text{ nm}} (\text{extinction}) \cdot V_{\text{f}} \cdot V_{\text{i}}^{-1}}{\text{calibration slope}} - 0$$
(7)

$$\frac{24.72 \cdot 0.3 \cdot 10^{-6} \text{m}^{-3} \cdot 0.3 \cdot 10^{-7} \text{m}^3}{9.6 \cdot 10^6 \text{ mol }^{-1} \text{m}^3} = 25.8 \cdot 10^{-6} \text{ mol } \text{m}^{-3}$$
(8)

Number of siRNA molecules per nm from Cy5 absorption:

siRNA per NP (UV-Vis, Cy5) =
$$\frac{c(siRNA-Cy5)_{Cy5 \text{ extinction}}}{c(NP)}$$
 (9)

$$\frac{25.1 \cdot 10^{-6} \text{ mol m}^{-3}}{2.90 \cdot 10^{-6} \text{ mol m}^{-3}} = 8.7$$
(10)

Number of siRNA molecules per nm from nucleobase absorption:

siRNA per NP (UV-Vis, RNA) =
$$\frac{c(siRNA-Cy5)_{RNA \text{ extinction}}}{c(NP)}$$
(11)

$$\frac{25.8 \cdot 10^{-6} \text{ mol m}^{-3}}{2.90 \cdot 10^{-6} \text{ mol m}^{-3}} = 8.9$$
(12)

Cells and viruses

The eGFP-expressing human cervix carcinoma cell line HeLaeGFP was generated in our group. The cells were cultivated at 37 °C and 5% CO₂ in GibcoTM Dulbecco's modified Eagle's medium (DMEM, Thermo Fisher Scientific) supplemented with 10% fetal bovine serum (FBS, Thermo Fisher Scientific), 50 µg mL⁻¹ GibcoTM Geneticin (G418 sulfate, Thermo Fisher Scientific), 1 mM GibcoTM sodium pyruvate (Thermo Fisher Scientific), and 1 mM GibcoTM GlutaMAX (Thermo Fisher Scientific). Cells were passaged either at 70–90% confluency or every two to three days by trypsinization with 0.05% GibcoTM Trypsin-EDTA (Thermo Fisher Scientific). The cells were washed two to three times with GibcoTM Dulbecco's buffered saline (DPBS, Thermo Fisher Scientific) between the individual steps of each experiment.

Vero cells (American Type Culture Collection, ATCC, CCL81, Rockville, MD, USA) were cultivated in Dulbecco's modified Eagle medium (DMEM, Life Technologies Gibco, Darmstadt, Germany), supplemented with 100 U mL⁻¹ penicillin, 0.1 mg mL⁻¹ streptomycin and 10% (v/v) fetal calf serum (FCS; Life Technologies Gibco), at 37 °C and 5% CO₂. HSV-2-(333)-GFP (further referred to as HSV-2 GFP) was provided by B. Sodeik (Institute of Virology, Medical University of Hannover, Germany). Viral titers were determined by a standard endpoint dilution assay and calculated as 50% tissue culture infectious dose (TCID₅₀) mL⁻¹ as described earlier.¹¹²

The murine monocyte cell line J774A.1 (CLS Cell Lines Service GmbH, Eppelheim, Germany) was cultured in DMEM/F-12 (1 : 1) (+)-L-glutamine (Gibco, Darmstadt, Germany) supplemented with 10 vol% fetal calf serum (FCS; Gibco). To induce an inflammation status, the monocytes were stimulated with lipopolysaccharide (LPS, Sigma-Aldrich; 1 μ g mL⁻¹) 4 h prior to investigation (denoted as "LPS-primed" in the following).

Viability of HeLa cells

The cell viability was determined by an MTT assay. First, HeLaeGFP cells were seeded at a density of 50 000 cells per well in a 24-well plate and incubated with 0.5 mL DMEM overnight at 37 °C in 5% CO₂ atmosphere. Next, the cells were incubated either with water-dispersed siRNA-clicked ultrasmall gold nanoparticles or with Lipofectamine/siRNA. Lipofectamine/ siRNA was applied at a concentration of 3.2 µg siRNA (siRNAeGFP/siRNA-eGFP-Cy5/siRNA-eGFP-r) mL⁻¹ per well according to the manufacturer's instructions. Nanoparticle dispersions were tested at three concentrations, *i.e.* c_1 : 5 µg Au mL⁻¹ and 14.8 µg siRNA mL⁻¹ (13.7 µg siRNA mL⁻¹ for Au-siRNA-eGFP-Cy5), c_2 : 25 µg Au mL⁻¹ and 74 µg siRNA mL⁻¹, c_3 : 50 µg Au

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mL⁻¹ and 148 µg siRNA (siRNA-eGFP/-eGFP-Cy5/-eGFP-r) mL⁻¹. The control groups for cell viability were cells treated with 500 µL of a Lipofectamine/DMEM mixture (1:1000 Lipofectamine: DMEM) and HeLa-eGFP cells cultivated in medium alone. After the incubation for 5 h (only for Lipofectamine) or 24 h (all other cases), the cells were washed three times with DPBS to remove weakly adhering lipoplexes and nanoparticles. For the staining solution, 5 mg MTT were dissolved in 1 mL PBS and then diluted with 4 mL DMEM to a final concentration of 1 mg mL⁻¹. 0.3 mL of the staining solution was pipetted into each well, and the cells were incubated for 1 h. Next, the solution was replaced with 0.3 mL DMSO and incubated for another 30 min. The dissolved formazan was quantitatively determined in a 96-well plate with a Multiscan plate reader (Thermo Fisher Scientific GmbH) at 570 nm.

Nanoparticle uptake and gene silencing in HeLa-eGFP cells

The uptake of Au-siRNA-eGFP-Cy5 nanoparticles by HeLa-eGFP cells was analyzed by multiple focal plane (z-stacks, interval 0.5 µm) confocal laser scanning microscopy (CLSM) in an 8-well chamber polymer slide surface modified with ibiTreat for tissue culture applications (µ-Slide 8-well, ibidi). CLSM was performed with a TCS SP8X Falcon instrument (Leica Microsystems) with a $63 \times / 1.2$ water immersion objective. 20 000 cells were seeded per well and incubated with 0.2 mL DMEM overnight at 37 °C in 5% CO2 atmosphere. The cells were incubated with siRNAeGFP-Cy5-clicked ultrasmall gold nanoparticles (final concentration 25 μ g Au mL⁻¹ per well). The control group were HeLaeGFP cells cultivated in medium alone. After incubation for 24 h, the cells were washed three times with DPBS, and fixed with 4 vol% formaldehyde solution according to standard protocols. For post-fixation, the actin skeleton of the cells was stained with AlexaFluorTM 568 Phalloidin. Nuclei were stained with DAPI according to the manufacturer's protocol. The excitation wavelengths were 405 nm for DAPI (emission: 420 to 460 nm), 488 nm for eGFP (emission: 495 to 515 nm), 568 nm for AlexaFluor[™] 568 Phalloidin (emission: 590 to 610 nm), and 647 nm for Cy5 (emission: 660 to 700 nm).

Gene silencing in HeLa-eGFP cells was analyzed by fluorescence microscopy with a Biorevo BZ-9000 instrument (Keyence). Images were recorded with the filter cubes for FITC (excitation: 470/40 nm, emission: 525/50 nm) and Cy5 (excitation: 620/ 60 nm, emission: 700/75 nm) and 10 \times and 20 \times air objectives together with the BZ-II viewer software. Gene silencing of eGFP in HeLa-eGFP cells was quantified by calculating the mean fluorescence intensity (MFI) in the fluorescence images and the fold change in the decrease of signal intensity. 50 000 cells per well were seeded in a 24-well plate (Sarstedt). The cells were incubated with siRNA-functionalized ultrasmall nanoparticles for 24 h or with Lipofectamine/siRNA (siRNA-eGFP/-eGFP-Cy5/eGFP-r) for 5 h. The lipoplex formation was carried out according to the manufacturer's protocol with a final concentration of 3.2 $\mu g \; si RNA \; m L^{-1}$ per well. Nanoparticle dispersions were tested in three concentrations of applied siRNA, *i.e.* c₁: 6-8 μg Au mL^{-1} and 14.8 μg siRNA $mL^{-1},$ c_2 : 30–40 μg Au mL^{-1} and 74 µg siRNA mL⁻¹, c_3 : 60–80 µg Au mL⁻¹ and 148 µg siRNA

 mL^{-1} . After incubation, the cells were washed twice with DPBS and submerged in fresh DMEM. All cells were fixed 72 h after the initial treatment. Next, four images of cell-confluent areas were taken per well by fluorescence microscopy. The mean fluorescence intensity (MFI) of the images was analyzed with the program Image J version 1.53k.¹¹³ Then, the average eGFP MFI of the untreated control group was divided by the MFI of the siRNA-treated cells. The fold change in signal decrease (gene silencing) of eGFP and was expressed as follows:

fold change_{decrease} =
$$-\left(\frac{MFI_{control}}{MFI_{sample}}\right)$$
 (13)

Only fully confluent areas with cells were analyzed to compute the mean fluorescence intensity, therefore the eGFP expression could be compared between the samples.

Cell viability determination on inflamed murine monocytes

For cell viability determination, LPS-primed monocytes were seeded in untreated 96-well cell culture plates and pre-incubated for 24 h. Au-siRNA-p65 nanoparticles were added with concentration of 1 µg siRNA-p65 mL⁻¹ and incubated for 72 h. Nontreated LPS-primed and non-treated native cells were used as controls. Afterwards, the cells were incubated with the Alamar-Blue[®]Cell Proliferation Reagent (Life Technologies GmbH, Darmstadt, Germany) according to the manufacturer's instruction. After 1 h of incubation, the fluorescence emission ($\lambda_{exc}/\lambda_{em}$: 560/590 nm) was measured. The cell viability was calculated by normalizing fluorescence data to that of non-treated native cells.

Gene silencing in inflamed murine monocytes

To study the impact of gold nanoparticles carrying anti-p65 siRNA on J774A.1 monocytes, the following experimental routes were used: LPS-primed cells exposed to Au-siRNA-p65 nanoparticles (1 μ g siRNA mL⁻¹, 72 h); free siRNA-p65 (1 μ g mL⁻¹, 72 h) together with Lipofectamine® (Thermo Fisher); non-treated but LPS-primed monocytes (native inflamed monocytes); non-treated cells without LPS-priming (non-inflamed monocytes). Cells were lysed with RIPA buffer for subsequent analysis of protein expression. The total protein concentration was measured with the Bradford assay.

Proteins were separated in multiple 10% (w/v) SDS-PAGE assays (10 to 20 μ g total protein per lane) and transferred to an Immobilon-P membrane (Merck Millipore, Darmstadt, Germany). A multi-strip western blotting was performed.⁹¹ After blocking (PUREBlockTM Blocking Buffer, Vilber Lourmat Deutschland GmbH, Eberhardzell, Germany; overnight, 4 °C), the membranes were probed with the corresponding primary antibodies (p65, p105/p50, and p100/p52, all mouse anti-rabbit, 1 : 1000 from Cell Signaling Technology, Danvers, USA; or GAPDH 1 : 3000 from Santa Cruz Biotechnology, Texas, USA, all overnight at 4 °C), washed, and finally incubated with the corresponding HRP-coupled secondary mouse anti-rabbit antibody (1 : 10 000, 1 h at RT, Dianova GmbH, Hamburg, Germany). Chemiluminescence was detected by PURECL and Fusion FX7 Edge (Vilber Lourmat Deutschland GmbH, Eberhardzell,

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Germany). To quantify the expression of the proteins of the different experimental groups, a densitometric analysis of western blot bands was performed with the software Bio1D (Vilber Lourmat Deutschland GmbH, Eberhardzell, Germany). Data were normalized to β -actin (loading control).

Inhibition of HSV-2 replication by gene silencing

Small interfering RNA 18.1 (siRNA-18.1) targeting the capsid encoding gene UL18 (ref. 104) and siRNA-29.2 targeting the DNAbinding protein UL29 (ref. 105) show a potent antiviral activity against HSV-2.104,105 Ultrasmall gold nanoparticles were functionalized with siRNA-18.1 siRNA-29.2 as described above. The impact of Au-siRNA-18.1 and Au-siRNA-29.2 nanoparticles on HSV-2 replication was tested in Vero cell culture. Serial dilutions of Au-siRNA-18.1 and Au-siRNA-29.2 nanoparticle stock dispersions (0.25 to 4 pmol siRNA μ L⁻¹) of Au-bound siRNA were added to confluent Vero cell monolayers grown in 96-well microtiter plates. Dissolved siRNA-18.1 and siRNA-29.2 in cell culture medium at concentrations of 4 pmol siRNA μL^{-1} , respectively, were used as controls. After 4 h of incubation, the cells were infected with 25 TCID₅₀ HSV-2 GFP. Non-infected Vero cells and mock-treated but HSV-2 infected cells served as controls. After 48 h, the cytopathic effects (CPE) were assessed by fluorescence microscopy. Immunofluorescence images were acquired with a Zeiss Observer Z1 fluorescence microscope (Carl Zeiss, Oberkochen, Germany) at 100-fold magnification. The concentrations of Au-siRNA-18.1 and Au-siRNA-29.2 required for reducing virusinduced CPE by 100% were defined as the completely neutralizing titer. With a molar mass of siRNA of about 13 000 g mol⁻¹ (Table 2), a concentration of 1 pmol siRNA $\mu L^{-1} = 1$ nmol siRNA mL^{-1} corresponds to about 13 µg siRNA mL^{-1} .

Conflicts of interest

There are no conflicts to declare.

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References

1 M. M. Xie, Y. L. Xu, J. Huang, Y. C. Li, L. Y. Wang, L. L. Yang and H. Mao, *WIREs Nanomed. Nanobiotechnol.*, 2020, **12**, e1644.

- 2 M. Fan, Y. Han, S. T. Gao, H. Y. Yan, L. Z. Cao, Z. H. Li, X. J. Liang and J. C. Zhang, *Theranostics*, 2020, **10**, 4944– 4957.
- 3 K. M. Rice, G. K. Ginjupalli, N. Manne, C. B. Jones and E. R. Blough, *Nanotechnology*, 2019, **30**, 372001.
- 4 K. A. Willets, A. J. Wilson, V. Sundaresan and P. B. Joshi, *Chem. Rev.*, 2017, **117**, 7538–7582.
- 5 L. A. Dykman and N. G. Khlebtsov, *Biomaterials*, 2016, **108**, 13–34.
- 6 A. B. Chinen, C. M. Guan, J. R. Ferrer, S. N. Barnaby, T. J. Merkel and C. A. Mirkin, *Chem. Rev.*, 2015, 115, 10530–10574.
- 7 Y. Ding, Z. W. Jiang, K. Saha, C. S. Kim, S. T. Kim, R. F. Landis and V. M. Rotello, *Mol. Ther.*, 2014, 22, 1075– 1083.
- 8 L. C. Cheng, X. M. Jiang, J. Wang, C. Y. Chen and R. S. Liu, *Nanoscale*, 2013, **5**, 3547–3569.
- 9 K. Saha, S. S. Agasti, C. Kim, X. Li and V. M. Rotello, *Chem. Rev.*, 2012, **112**, 2739–2779.
- 10 L. Dykman and N. Khlebtsov, *Chem. Soc. Rev.*, 2012, **41**, 2256–2282.
- 11 W. J. Stark, Angew. Chem., Int. Ed., 2011, 50, 1242-1258.
- 12 C. M. Cobley, J. Y. Chen, E. C. Cho, L. V. Wang and Y. N. Xia, *Chem. Soc. Rev.*, 2011, **40**, 44–56.
- 13 M. Homberger and U. Simon, *Philos. Trans. R. Soc.*, A, 2010, 368, 1405–1453.
- 14 D. A. Giljohann, D. S. Seferos, W. L. Daniel, M. D. Massich, P. C. Patel and C. A. Mirkin, *Angew. Chem., Int. Ed.*, 2010, 49, 3280–3294.
- 15 R. Sardar, A. M. Funston, P. Mulvaney and R. W. Murray, *Langmuir*, 2009, **25**, 13840–13851.
- 16 Y. G. Srinivasulu, Q. F. Yao, N. Goswami and J. P. Xie, *Mater. Horiz.*, 2020, 7, 2596–2618.
- 17 C. J. Zeng, Pure Appl. Chem., 2018, 90, 1409-1427.
- 18 J. Langer, S. M. Novikov and L. M. Liz-Marzan, Nanotechnology, 2015, 26, 322001.
- 19 Y. C. Yeh, B. Creran and V. M. Rotello, *Nanoscale*, 2012, 4, 1871–1880.
- 20 E. C. Dreaden, A. M. Alkilany, X. Huang, C. J. Murphy and M. A. El-Sayed, *Chem. Soc. Rev.*, 2012, **41**, 2740–2779.
- 21 D. Zanchet, C. M. Micheel, W. J. Parak, D. Gerion and A. P. Alivisatos, *Nano Lett.*, 2001, 1, 32–35.
- 22 D. Zanchet, C. M. Micheel, W. J. Parak, D. Gerion,
 S. C. Williams and A. P. Alivisatos, *J. Phys. Chem. B*, 2002,
 106, 11758–11763.
- 23 J. R. McMillan, J. D. Brodin, J. A. Millan, B. Lee, M. O. de la Cruz and C. A. Mirkin, *J. Am. Chem. Soc.*, 2017, **139**, 1754– 1757.
- 24 M. Girard, S. Z. Wang, J. S. Du, A. Das, Z. Y. Huang,
 V. P. Dravid, B. Lee, C. A. Mirkin and M. O. de la Cruz, *Science*, 2019, 364, 1174.
- 25 S. D. Huo, N. Q. Gong, Y. Jiang, F. Chen, H. B. Guo, Y. L. Gan, Z. S. Wang, A. Herrmann and X. J. Liang, *Sci. Adv.*, 2019, 5, eaaw6264.
- 26 S. Z. Wang, J. S. S. Du, N. J. Diercks, W. J. Zhou, E. W. Roth, V. P. Dravid and C. A. Mirkin, *J. Am. Chem. Soc.*, 2019, 141, 20443–20450.

- 27 C. R. Laramy, M. N. O'Brien and C. A. Mirkin, *Nat. Rev. Mater.*, 2019, 4, 201–224.
- 28 C. A. Mirkin, Inorg. Chem., 2000, 39, 2258-2272.
- 29 G. Yamankurt, R. J. Stawicki, D. M. Posadas, J. Q. Nguyen, R. W. Carthew and C. A. Mirkin, *Proc. Natl. Acad. Sci. U. S. A.*, 2020, **117**, 1312.
- 30 K. L. Young, A. W. Scott, L. L. Hao, S. E. Mirkin, G. L. Liu and C. A. Mirkin, *Nano Lett.*, 2012, **12**, 3867–3871.
- 31 W. L. Zhang, B. Meckes and C. A. Mirkin, *ACS Cent. Sci.*, 2019, **5**, 1983–1990.
- 32 S. S. Zhu, H. Xing, P. Gordiichuk, J. Park and C. A. Mirkin, *Adv. Mater.*, 2018, **30**, 1707113.
- 33 A. M. Gorabi, N. Kiaie, S. Aslani, T. Jamialahmadi, T. P. Johnston and A. Sahebkar, *J. Autoimmun.*, 2020, 114, 102529.
- 34 B. Kim, J. H. Park and M. J. Sailor, *Adv. Mater.*, 2019, **31**, 1903637.
- 35 B. Hu, L. Zhong, Y. Weng, L. Peng, Y. Huang, Y. Zhao and X. J. Liang, *Signal Transduction Targeted Ther.*, 2020, **5**, 101.
- 36 A. A. Sousa, P. Schuck and S. A. Hassan, Nanoscale Adv., 2021, 3, 2995–3027.
- 37 Z. H. Liu, Z. N. Wu, Q. F. Yao, Y. T. Cao, O. J. H. Chai and J. P. Xie, *Nano Today*, 2021, **36**, 101053.
- 38 Y. X. Du, H. T. Sheng, D. Astruc and M. Z. Zhu, *Chem. Rev.*, 2020, **120**, 526–622.
- 39 R. A. Revia, Z. R. Stephen and M. Zhang, Acc. Chem. Res., 2019, 52, 1496–1506.
- 40 M. Kopp, S. Kollenda and M. Epple, *Acc. Chem. Res.*, 2017, 50, 1383–1390.
- 41 K. Zarschler, L. Rocks, N. Licciardello, L. Boselli, E. Polo,
 K. P. Garcia, L. De Cola, H. Stephan and K. A. Dawson, *Nanomedicine*, 2016, **12**, 1663–1701.
- 42 N. H. Kim, M. J. Hackett, J. Park and T. Hyeon, *Chem. Mater.*, 2014, 26, 59–71.
- 43 L. Shang, S. J. Dong and G. U. Nienhaus, *Nano Today*, 2011, 6, 401–418.
- 44 R. Yu, L. M. Liz-Marzan and F. J. Garcia de Abajo, *Chem. Soc. Rev.*, 2017, **46**, 6710–6724.
- 45 J. Turkevich, P. C. Stevenson and J. Hilliery, *Discuss. Faraday* Soc., 1951, **11**, 55.
- 46 K. Klein, K. Loza, M. Heggen and M. Epple, *ChemNanoMat*, 2021, 7, 1330–1339.
- 47 S. B. van der Meer, K. Loza, K. Wey, M. Heggen, C. Beuck, P. Bayer and M. Epple, *Langmuir*, 2019, 35, 7191–7204.
- 48 S. B. van der Meer, I. Hadrovic, A. Meiners, K. Loza, M. Heggen, S. K. Knauer, P. Bayer, T. Schrader, C. Beuck and M. Epple, *J. Phys. Chem. B*, 2021, **125**, 115–127.
- 49 K. Klein, M. Hayduk, S. Kollenda, M. Schmiedtchen, J. Voskuhl and M. Epple, *Molecules*, 2022, **27**, 1788.
- 50 N. Bialas, V. Sokolova, S. B. van der Meer, T. Knuschke, T. Ruks, K. Klein, A. M. Westendorf and M. Epple, *Nano Sel.*, 2022, DOI: 10.1002/nano.202200049.
- 51 L. Yang, L. Shang and G. U. Nienhaus, *Nanoscale*, 2013, 5, 1537–1543.
- 52 C. Carrillo-Carrion, A. I. Bocanegra, B. Arnaiz, N. Feliu, D. C. Zhu and W. J. Parak, *ACS Nano*, 2019, **13**, 4631–4639.

- 53 V. Sokolova, G. Mekky, S. B. van der Meer, M. C. Seeds,A. J. Atala and M. Epple, *Sci. Rep.*, 2020, **10**, 18033.
- 54 V. Sokolova, G. Nzou, S. B. van der Meer, T. Ruks, M. Heggen, K. Loza, N. Hagemann, F. Murke, B. Giebel, D. M. Hermann, A. J. Atala and M. Epple, *Acta Biomater.*, 2020, 111, 349–362.
- 55 H. Häkkinen, Nat. Chem., 2012, 4, 443-455.
- 56 D. Mahl, C. Greulich, W. Meyer-Zaika, M. Köller and M. Epple, *J. Mater. Chem.*, 2010, **20**, 6176–6181.
- 57 G. Schmid, W. G. Kreyling and U. Simon, Arch. Toxicol., 2017, 91, 3011–3037.
- 58 A. Leifert, Y. Pan-Bartnek, U. Simon and W. Jahnen-Dechent, *Nanoscale*, 2013, **5**, 6224–6242.
- 59 J. Broda, G. Schmid and U. Simon, *Struct. Bonding*, 2014, **161**, 189–242.
- 60 Y. Pan, S. Neuss, A. Leifert, M. Fischler, F. Wen, U. Simon, G. Schmid, W. Brandau and W. Jahnen-Dechent, *Small*, 2007, 3, 1941–1949.
- 61 L. Dong, M. Li, S. Zhang, J. Li, G. Shen, Y. Tu, J. Zhu and J. Tao, *Small*, 2015, **11**, 2571–2581.
- 62 S. K. Boda, J. Broda, F. Schiefer, J. Weber-Heynemann, M. Hoss, U. Simon, B. Basu and W. Jahnen-Dechent, *Small*, 2015, **11**, 3183–3193.
- 63 Y. Pan, A. Leifert, M. Graf, F. Schiefer, S. Thoröe-Boveleth, J. Broda, M. C. Halloran, H. Hollert, D. Laaf, U. Simon and W. Jahnen-Dechent, *Small*, 2013, 9, 863–869.
- 64 S. Son, V. G. Deepagan, S. Shin, H. Ko, J. Min, W. Um, J. Jeon, S. Kwon, E. S. Lee, M. Suh, D. S. Lee and J. H. Park, *Acta Biomater.*, 2018, **79**, 294–305.
- 65 A. A. Sousa, S. A. Hassan, L. L. Knittel, A. Balbo,
 M. A. Aronova, P. H. Brown, P. Schuck and
 R. D. Leapman, *Nanoscale*, 2016, 8, 6577–6588.
- 66 Z. Luo, K. Zheng and J. Xie, *Chem. Commun.*, 2014, **50**, 5143– 5155.
- 67 K. Huang, H. Ma, J. Liu, S. Huo, A. Kumar, T. Wei, X. Zhang, S. Jin, Y. Gan, P. C. Wang, S. He, X. Zhang and X. J. Liang, *ACS Nano*, 2012, 6, 4483–4493.
- 68 J. Bourquin, D. Septiadi, D. Vanhecke, S. Balog, L. Steinmetz, M. Spuch-Calvar, P. Taladriz-Blanco, A. Petri-Fink and B. Rothen-Rutishauser, *ACS Nano*, 2019, 13, 7759–7770.
- 69 L. M. Liz-Marzan, Chem. Commun., 2013, 49, 16-18.
- 70 H. Fissan, S. Ristig, H. Kaminski, C. Asbach and M. Epple, *Anal. Methods*, 2014, **6**, 7324–7334.
- 71 T. Ruks, C. Beuck, T. Schaller, F. Niemeyer, M. Zähres, K. Loza, M. Heggen, U. Hagemann, C. Mayer, P. Bayer and M. Epple, *Langmuir*, 2019, 35, 767–778.
- 72 G. Salassa and T. Burgi, Nanoscale Horiz., 2018, 3, 457-463.
- 73 C. Guo and J. L. Yarger, *Magn. Reson. Chem.*, 2018, 56, 1074– 1082.
- 74 L. E. Marbella and J. E. Millstone, *Chem. Mater.*, 2015, 27, 2721–2739.
- 75 O. Wetzel, O. Prymak, K. Loza, N. Gumbiowski, M. Heggen,
 P. Bayer, C. Beuck, C. Weidenthaler and M. Epple, *Inorg. Chem.*, 2022, 61, 5133–5147.
- 76 O. Wetzel, S. Hosseini, K. Loza, M. Heggen, O. Prymak,P. Bayer, C. Beuck, T. Schaller, F. Niemeyer,

C. Weidenthaler and M. Epple, *J. Phys. Chem. B*, 2021, **125**, 5645–5659.

- 77 M. Lach, C. Strelow, A. Meyer, A. Mews and T. Beck, *ACS Appl. Mater. Interfaces*, 2022, **14**, 10656–10668.
- 78 P. A. Baeuerle and D. Baltimore, *Science*, 1988, 242, 540-546.
- 79 T. Huxford and G. Ghosh, *Cold Spring Harbor Perspect. Biol.*, 2009, **1**, a000075.
- 80 A. Israel, Cold Spring Harbor Perspect. Biol., 2010, 2, a000158.
- 81 C. Kunsch, S. M. Ruben and C. A. Rosen, *Mol. Cell. Biol.*, 1992, **12**, 4412–4421.
- 82 T. Liu, L. Zhang, D. Joo and S. C. Sun, *Signal Transduction Targeted Ther.*, 2017, **2**, 17023.
- 83 S. Chernousova and M. Epple, *Gene Ther.*, 2017, 24, 282–289.
- 84 H. H. Szeto, P. W. Schiller, K. Zhao and G. Luo, *FASEB J.*, 2005, **19**, 118–120.
- 85 J. Kurreck, Angew. Chem., Int. Ed., 2009, 48, 1378-1398.
- 86 M. G. Dorrington and I. D. C. Fraser, *Front. Immunol.*, 2019, 10, 705.
- 87 E. B. Glass, S. Masjedi, S. O. Dudzinski, A. J. Wilson, C. L. Duvall, F. E. Yull and T. D. Giorgio, *ACS Omega*, 2019, 4, 16756–16767.
- 88 R. A. Ortega, W. Barham, K. Sharman, O. Tikhomirov, T. D. Giorgio and F. E. Yull, *Int. J. Nanomed.*, 2016, **11**, 2163–2177.
- 89 N. Bialas, E. K. Müller, M. Epple and I. Hilger, *Biomaterials*, 2021, 276, 121013.
- 90 E. K. Müller, N. Bialas, M. Epple and I. Hilger, *Biomedicines*, 2022, **10**, 1571.
- 91 E. K. Müller, N. Bialas, M. Epple and I. Hilger, *Pharmaceutics*, 2022, 14, 419.
- 92 M. Mussbacher, M. Salzmann, C. Brostjan, B. Hoesel, C. Schoergenhofer, H. Datler, P. Hohensinner, J. Basilio, P. Petzelbauer, A. Assinger and J. A. Schmid, *Front. Immunol.*, 2019, 10, 85.
- 93 D. I. Bernstein, A. R. Bellamy, E. W. Hook III, M. J. Levin,
 A. Wald, M. G. Ewell, P. A. Wolff, C. D. Deal,
 T. C. Heineman, G. Dubin and R. B. Belshe, *Clin. Infect. Dis.*, 2013, 56, 344–351.
- 94 K. J. Looker, G. P. Garnett and G. P. Schmid, *Bull. W. H. O.*, 2008, **86A**, 805–812.

- 95 J. T. Vrabec and R. L. Alford, *J. NeuroVirol.*, 2004, **10**, 216–222.
- 96 R. Whitley, D. W. Kimberlin and C. G. Prober, in *Human Herpesviruses: Biology, Therapy, and Immunoprophylaxis*, ed. A. Arvin, G. Campadelli-Fiume, E. Mocarski, P. S. Moore, B. Roizman, R. Whitley and K. Yamanishi, Cambridge, 2007.
- 97 D. Bauer, M. Alt, M. Dirks, M. Kasper, A. Buch, U. Dittmer, A. Goergens, B. Sodeik, A. Heiligenhaus, R. Michael and A. Krawczyk, *Invest. Ophthalmol. Visual Sci.*, 2016, 57, 3300.
- 98 P. Sen and S. E. Barton, BMJ, 2007, 334, 1048-1052.
- 99 U. Meyding-Lamade and C. Strank, *Ther. Adv. Neurol. Disord.*, 2012, 5, 279–296.
- 100 S. Burrel, D. Boutolleau, G. Azar, S. Doan, C. Deback,
 I. Cochereau, H. Agut and E. E. Gabison, *J. Clin. Virol.*, 2013, 58, 321–324.
- 101 A. V. Farooq and D. Shukla, *Surv. Ophthalmol.*, 2012, 57, 448–462.
- 102 R. Duan, R. D. de Vries, J. M. van Dun, F. B. van Loenen,
 A. D. M. E. Osterhaus, L. Remeijer and
 G. M. G. M. Verjans, *J. Infect. Dis.*, 2009, 200, 1402–1414.
- 103 E. Paradowska, M. Studzinska, A. Jablonska, V. Lozovski, N. Rusinchuk, I. Mukha, N. Vitiuk and Z. J. Lesnikowski, *Molecules*, 2021, 26, 5960.
- 104 F. Jin, S. Li, K. Zheng, C. Zhuo, K. Ma, M. Chen, Q. Wang, P. Zhang, J. Fan, Z. Ren and Y. Wang, *PLoS One*, 2014, 9, e96623.
- 105 D. Palliser, D. Chowdhury, Q. Y. Wang, S. J. Lee, R. T. Bronson, D. M. Knipe and J. Lieberman, *Nature*, 2006, **439**, 89–94.
- 106 V. V. Rostovtsev, L. G. Green, V. V. Fokin and K. B. Sharpless, *Angew. Chem., Int. Ed.*, 2002, **41**, 2596–2599.
- 107 T. Efthymiou, W. Gong and J. P. Desaulniers, *Molecules*, 2012, **17**, 12665–12703.
- 108 A. H. El-Sagheer and T. Brown, *Acc. Chem. Res.*, 2012, **45**, 1258–1267.
- 109 S. Kath-Schorr, Top. Curr. Chem., 2016, 374, 4.
- 110 L. Li and Z. Zhang, Molecules, 2016, 21, 1393.
- 111 A. Thust, J. Barthel and K. Tillmann, *J. Large-scale Res. Fac.*, 2016, 2, A41.
- 112 J. L. Reed and H. Muench, Am. J. Hyg., 1938, 27, 493-497.
- 113 C. T. Rueden, J. Schindelin, M. C. Hiner, B. E. DeZonia, A. E. Walter, E. T. Arena and K. W. Eliceiri, *BMC Bioinf.*, 2017, 18, 529.