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Electron-withdrawing group modified carbazophane donors for deep blue thermally activated delayed fluorescence OLEDs†

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We report two blue-emitting thermally activated delayed fluorescence (TADF) compounds employing a substituted carbazophane (Czp) donor (indolo[2.2]paracyclophane). The compounds **CNCzpPhTRZ** and **CF₃CzpPhTRZ** show emission maxima of 426 nm and 432 nm, respectively, with high photoluminescence quantum yields (Φ_{PL}) of 73% and 80%, respectively. The singlet–triplet energy gap (ΔE_{ST}) of both emitters is 0.22 eV, resulting in long-delayed lifetimes of 132 μs for **CNCzpPhTRZ** and 158 μs for **CF₃CzpPhTRZ** in PPT as the host matrix. Blue organic light-emitting diodes (OLEDs) showed maximum external quantum efficiencies (EQEs) of 7.4% for **CNCzpPhTRZ** and 11.6% for **CF₃CzpPhTRZ** with electroluminescence maxima of ca. 460 nm.

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Introduction

The development of purely organic thermally activated delayed fluorescence (TADF) emitters is considered as one of the most promising approaches to utilize singlet and triplet excitons to access highly efficient organic light-emitting diodes (OLEDs).^{1–4} The design principle of generating a structure with spatially separated donor and acceptor moieties to limit the exchange integral between the highest occupied molecular orbital (HOMO) and the lowest unoccupied molecular orbital (LUMO) thereby leading to a small singlet and triplet excited state energy difference, ΔE_{ST} , has been a reliable and fruitful strategy to construct TADF emitters. To obtain a small ΔE_{ST} , there are at present at least six design strategies, which are twisted donor–acceptor,¹ multi-resonance,⁵ spiro-conjugation,⁶ exciplex,⁷

excited-state intramolecular proton transfer (ESIPT)⁸ and through space-conjugation.⁹ The most popular of these is to electronically decouple the donor and acceptor by ensuring a highly twisted conformation.

The [2.2]paracyclophane (PCP) is a rigid molecule containing two slightly bent benzene rings connected by ethylene bridges. Due to the spatial stacked configuration and a short distance of 2.78–3.09 Å between the two benzene decks, which is smaller than the van der Waals distance between the layers of graphite (3.35 Å), a strong transannular electronic communication can take place, as demonstrated by Bazan *et al.*^{10–14} Substituted PCPs can possess planar chirity and emissive analogs can exhibit circularly polarized luminescence (CPL).¹⁵ The characteristics of rigid skeleton, chemical and photo stability, planar chirity, and through-space electronic communication possessed by the PCP core make it a potentially useful building block for organic semiconductor materials.

In 2018, we first demonstrated that the PCP moiety could be used to mediate weak electronic communication between donor and acceptor groups on different decks of the PCP.⁹ **Trans-Bz-PCP-TPA** and **cis-Bz-PCP-TPA** showed photoluminescence maxima, λ_{PL} , at 455 nm, photoluminescence quantum yields, Φ_{PL} , of 60% (in toluene), ΔE_{ST} = 0.19 eV and delayed lifetime, τ_{d} , of 3.6 μs for **trans-Bz-PCP-TPA** and λ_{PL} of 492 nm, Φ_{PL} of 45% (in toluene), ΔE_{ST} = 0.04 eV and τ_{d} of 1.8 μs for **cis-Bz-PCP-TPA** in 15 wt% mCP doped film (Fig. 1). Zhao *et al.* also reported a series of structurally similar chiral green TADF molecules **g-BNMe-Cp** and **m-BNMe-Cp** containing a PCP, with emission at 531 nm (Φ_{PL} = 72% in cyclohexane, ΔE_{ST} = 0.17 eV

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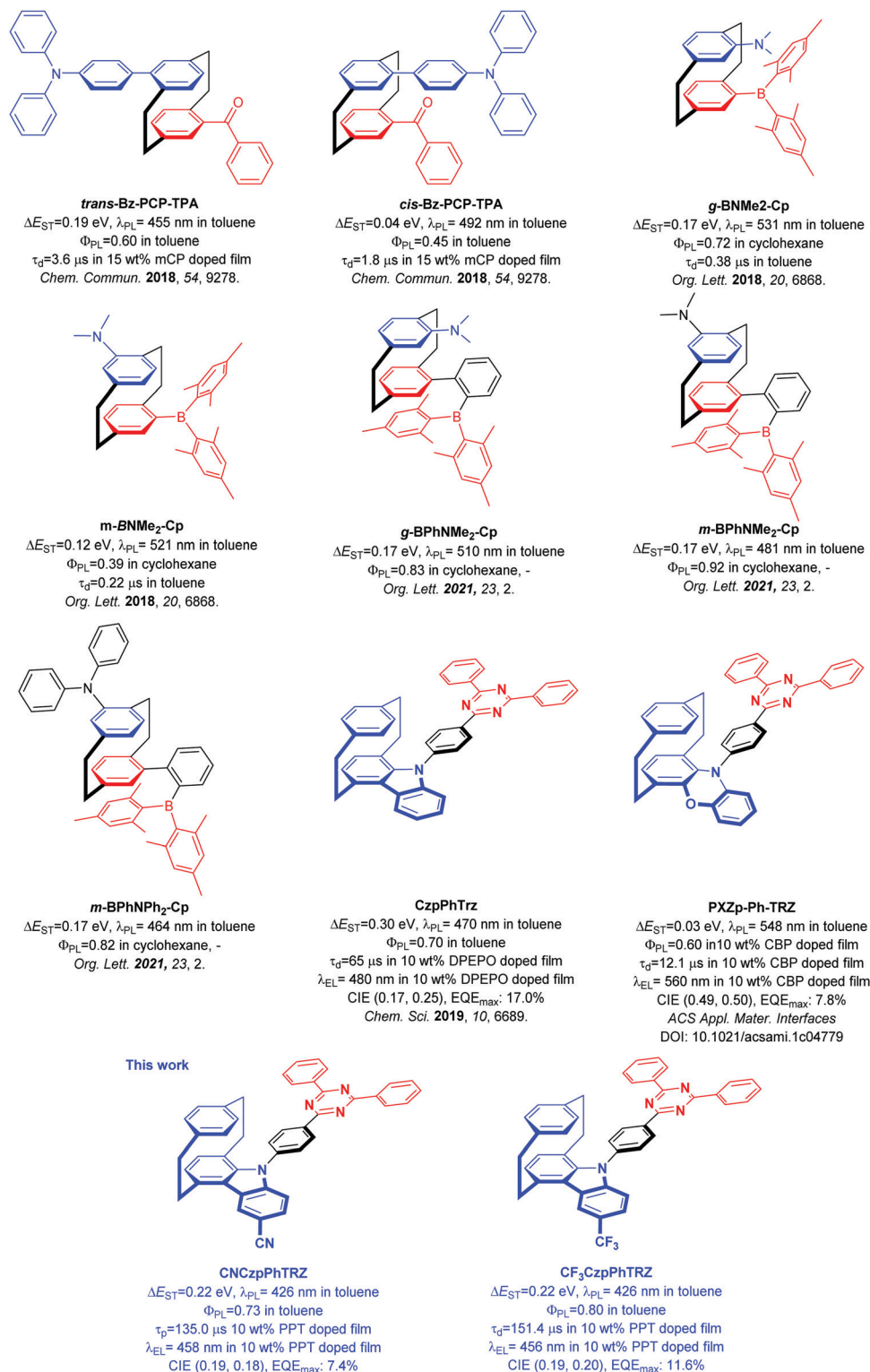
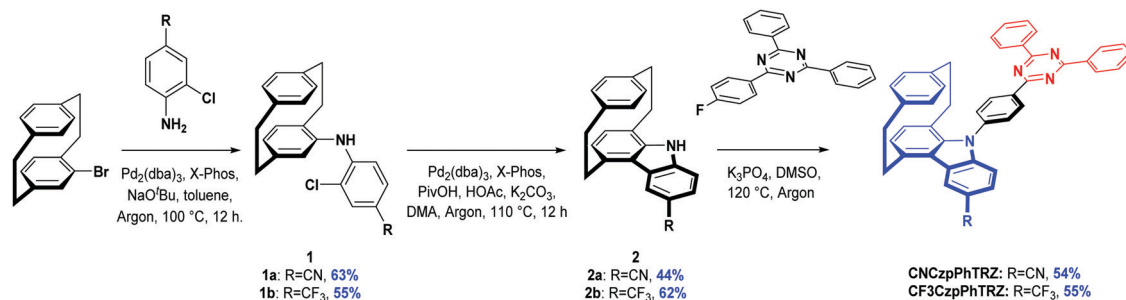


Fig. 1 Chemical structures and performance of PCP-based TADF molecules.

and $\tau_d = 0.38$ μ s in toluene) and 521 nm ($\Phi_{PL} = 39\%$ in cyclohexane, $\Delta E_{ST} = 0.12$ eV and $\tau_d = 0.22$ μ s in toluene), respectively (Fig. 1).¹⁶ Recently, the same group introduced a phenylene spacer between the PCP and the acceptor moiety to obtain sky-blue to green TADF emitters showing increased Φ_{PL}

(in cyclohexane) to 83% for **g-BPhNMe₂-Cp**, 93% for **m-BPhNMe₂-Cp**, and 82% for **g-BPhNPh₂-Cp** (Fig. 1).¹⁷ Besides using the PCP as a bridge motif to enable electronic coupling between donor and acceptor groups, we also demonstrated how the PCP skeleton could be incorporated within the carbazole donor unit. Carbazolophane

Scheme 1 Synthetic route to CNCzPhTRZ and CF₃CzPhTRZ.

(Czp, [2]paracyclo[2](1,4)carbazolophane) group is a stronger electron donor that adopts a more twisted conformation due to its greater steric bulk.⁹ We reported the chiral TADF emitter CzpPhTRZ, which showed a small ΔE_{ST} of 0.16 eV, and λ_{PL} at 482 nm with Φ_{PL} = 69% and a τ_d of 65 μ s in 10 wt% DPEPO doped film.¹⁸ Sky-blue OLEDs with CzpPhTRZ showed a maximum external quantum efficiency (EQE_{max}) of 17.0% and CIE coordinates of (0.17, 0.25). Zheng and co-workers reported an analog of CzpPhTRZ, PXZp-Ph-TRZ, where the Czp donor was replaced with a phenoxazinephane (PXZp). PXZp-Ph-TRZ possesses a ΔE_{ST} of 0.03 eV and shows yellow emission with λ_{PL} at 560 nm in 10 wt% CBP doped film. The OLED showed an EQE_{max} of 7.8% and CIE coordinates of (0.49, 0.50).¹⁹ As we can see from Fig. 1 since our first report, a handful of emitters have been reported containing the PCP moiety; however none to date show pure blue emission.

Building on our prior work and with the goal to obtain a blue-shifted emission, we designed two new TADF emitters CNCzPhTRZ and CF₃CzPhTRZ, by introducing electron-withdrawing groups on the Czp donor unit of the parent CzpPhTRZ emitter. We successfully blue-shifted the emission without adversely affecting the TADF efficiency. CNCzPhTRZ emits at 426 nm, and CF₃CzPhTRZ emits at 432 nm in toluene, representing a *ca.* 40 nm blue-shift compared to that of CzpPhTRZ. The Φ_{PL} remained high at 73% and 80%, respectively. Further, we fabricated bluer OLEDs with electroluminescence maxima at *ca.* 460 nm. However, we observed the evolution of a red emission band at around 600 nm in the OLEDs due to degradation of both blue emitters during the operation of the device.

Results and discussion

Synthesis and thermal stability

The synthesis is based on our previously reported route and shown in Scheme 1. Aniline derivatives were coupled to

bromo[2.2]paracyclophane with yields of 63% for **1a**, and 55% for **1b**. The generated intermediates were used to form the functionalized carbazolophane donor units with yields of 44% for **2a** and 62% for **2b** via a Pd-catalyzed intramolecular oxidative cyclization process where the *ortho*-chlorine atoms act as the directing group. Lastly, the prepared donor **2** and triazine-based acceptor were linked together via a nucleophilic aromatic substitution with yields above 50% using tripotassium phosphate as a base in DMSO at 120 °C. The targeted compounds, CNCzPhTRZ and CF₃CzPhTRZ, show strong blue luminescence under UV excitation. The identity and purity of CNCzPhTRZ and CF₃CzPhTRZ were determined by a combination of NMR, MS, HRMS, Mp, HPLC and IR.

Differential scanning calorimetry (DSC) and thermogravimetric analysis (TGA) were carried out to explore the thermal stability of the two emitters (Table 1 and Fig. S2, ESI[†]). CF₃CzPhTRZ shows a glass transition temperature (T_g) of 151 °C and 5% weight loss temperature (T_{d5}) of 360 °C. The T_g and T_{d5} of CNCzPhTRZ are higher than those of CF₃CzPhTRZ (T_g = 184 °C and T_{d5} = 412 °C).

Theoretical calculations

Density functional theory (DFT) calculations were performed to investigate the frontier orbital densities and energy states of model compounds of the two emitters. The geometries in the ground state of the two emitters were optimized in the gas phase at the PBE0/6-31G(d,p) level of theory. Time-dependent DFT calculations using the Tamm-Dancoff approximation (TDA-DFT) in the gas phase based on the optimized molecular structures provide insight into the energies and nature of the lowest-lying singlet and triplet excited states.²⁰ The frontier orbitals and energy levels of these molecules are shown in Fig. 2 and Table S1 (ESI[†]).

Table 1 Electrochemical data, HOMO–LUMO, ΔE_{ST} and thermal stability data of CF₃CzPhTRZ and CNCzPhTRZ

Emitters	E_{ox}^a/V	E_{red}^a/V	HOMO ^b /eV	LUMO ^b /eV	E_S/E_T^c /eV	ΔE_{ST}^d /eV	T_g ; $T_{d5}/^\circ C$
CF ₃ CzPhTRZ	1.30	−1.66	−5.64 (6.05)	−2.68	2.91/2.69	0.22	151; 360
CNCzPhTRZ	1.34	−1.67	−5.68 (5.93)	−2.67	2.92/2.69	0.23	184; 412

^a E_{ox} and E_{red} are anodic and cathodic peak potentials, respectively, obtained from DPV using F₀/F_c⁺ as the internal reference and referenced versus SCE (0.46 V vs. SCE) in DCM with 0.1 M [nBu₄N]PF₆ as the supporting electrolyte.²¹ ^b $E_{HOMO/LUMO} = -(E^{ox}/E^{red} + 4.8)$ eV,²² where E^{ox} is anodic peak potential and E^{red} is cathodic peak potential calculated from DPV relative to F₀/F_c⁺. Value inside parentheses determined by ambient pressure photoemission spectroscopy (APS). ^c Obtained from the onset of the prompt fluorescence (time window: 1 ns–100 ns) and phosphorescence spectra (time window: 1 ms–10 ms) of 10 wt% samples doped in PPT at 77 K. ^d $\Delta E_{ST} = E_S - E_T$.



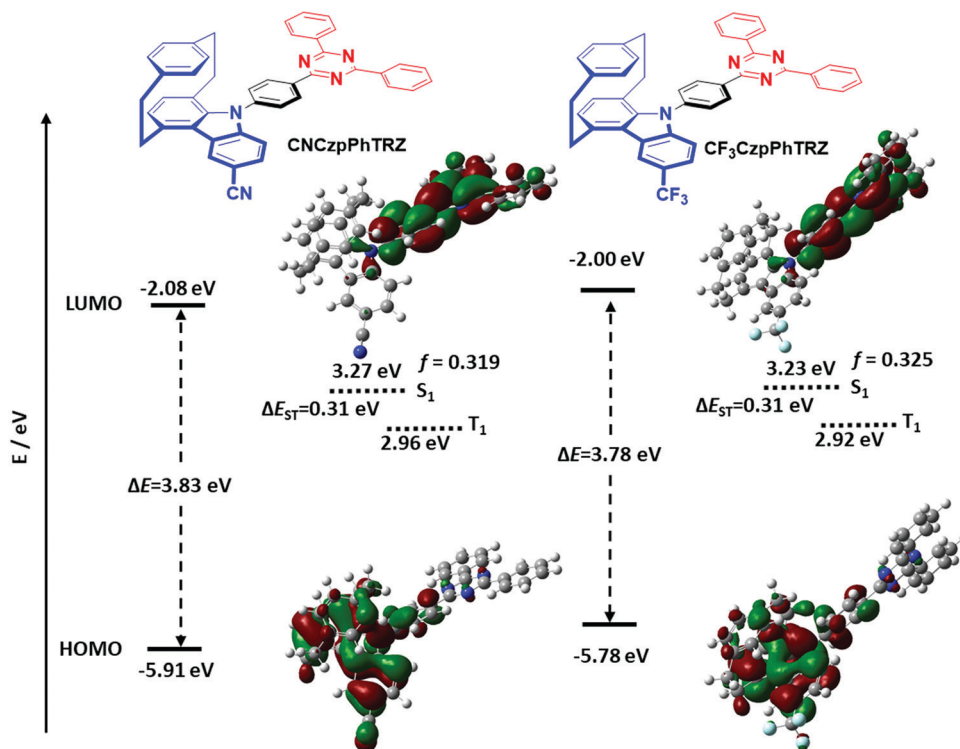


Fig. 2 DFT calculations of **CNCzpPhTRZ** and **CF₃CzpPhTRZ**.

The torsion angles between the **Czp** donors and the bridging phenylene are expected not unaffected by the substitution on the distal side of the donor. While for the previously reported **CzpPhTRZ**, the torsion angle obtained from DFT calculations was 58°, the torsion angles for **CF₃CzpPhTRZ** and **CNCzpPhTRZ** are 56° and 58°, respectively. The DFT calculations reveal that the frontier orbitals and states are effectively tuned *via* the introduction of electron-withdrawing groups. The HOMO is located primarily on the carbazolophane donor moiety, while the LUMO is distributed across the triazine acceptor and phenylene bridge. The HOMO levels of the two derivatives are expectedly significantly stabilized at −5.91 eV (**CNCzpPhTRZ**), and −5.78 eV (**CF₃CzpPhTRZ**) compared to that of the parent emitter **CzpPhTRZ** (−5.54 eV) due to the presence of the electron-withdrawing cyano and trifluoromethyl groups, respectively. The more positive Hammett σ_p value for cyano (0.66) *versus* trifluoromethyl (0.54) explains the 0.13 eV greater stabilization observed for the HOMO of **CNCzpPhTRZ**. The LUMO levels are somewhat less affected by substituting the carbazolophane at −2.08 and −2.00 eV, respectively, for **CNCzpPhTRZ** and **CF₃CzpPhTRZ** and deeper than that of **CzpPhTRZ** (−1.88 eV). There is a corresponding increase in the energy of the S₁ state from 3.12 eV (**CzpPhTRZ**) to 3.27 eV (**CNCzpPhTRZ**) and 3.23 eV (**CF₃CzpPhTRZ**). The triplet energies are unaffected by the substituents with 2.81 eV (**CzpPhTRZ**), 2.96 eV (**CNCzpPhTRZ**) and 2.92 eV (**CF₃CzpPhTRZ**). The ΔE_{ST} values of **CNCzpPhTRZ** (0.31 eV) and **CF₃CzpPhTRZ** (0.31 eV) are almost identical to that of **CzpPhTRZ** (0.30 eV), which suggests that these three molecules will be equally promising TADF emitters. The oscillator strengths, f , for the transition to the S₁ state of **CNCzpPhTRZ** ($f = 0.319$) and **CF₃CzpPhTRZ** ($f = 0.325$)

are slightly reduced compared to **CzpPhTRZ** ($f = 0.418$), indicating a slightly slower radiative decay rate.

Electrochemical studies

Cyclic voltammetry (CV) and differential pulse voltammetry (DPV) measurements in dichloromethane (DCM) with tetra-*n*-butylammonium hexafluorophosphate (TBAPF₆) as the supporting electrolyte were used to infer the HOMO/LUMO levels of **CF₃CzpPhTRZ** and **CNCzpPhTRZ** from the redox potentials. The resulting voltammograms are shown in Fig. S1 (ESI[†]), and the data are summarized in Table 1. Both emitters possess similar irreversible oxidation potentials. The oxidation potential of **CF₃CzpPhTRZ** is 1.30 V *vs.* SCE, and that of **CNCzpPhTRZ** is 1.34 V *vs.* SCE (Fig. S1, ESI[†]), which are each assigned to the oxidation of the carbazolophane (Czp) donor moieties, values that are anodically shifted relative to that of **CzpPhTRZ** (1.14 V).¹⁸ The estimated HOMO levels are −5.64 eV and −5.68 eV for **CF₃CzpPhTRZ** and **CNCzpPhTRZ**, respectively. These two HOMO levels are very similar, suggesting that the CN and CF₃ have a similar electron-withdrawing ability in these compounds.

Further, the ionization potential was directly measured by UV ambient pressure photoemission spectroscopy (APS) and showed deeper levels of 6.05 eV for **CF₃CzpPhTRZ** and 5.93 eV for **CNCzpPhTRZ** than those determined from the solution-state electrochemistry, this due in part to the different measurement environments (neat film for APS and DCM for DPV). These values align much more closely to the DFT calculated values (*vide supra*), although we note the calculations are *in vacuo* whereas the measurements are on solid films. The reduction



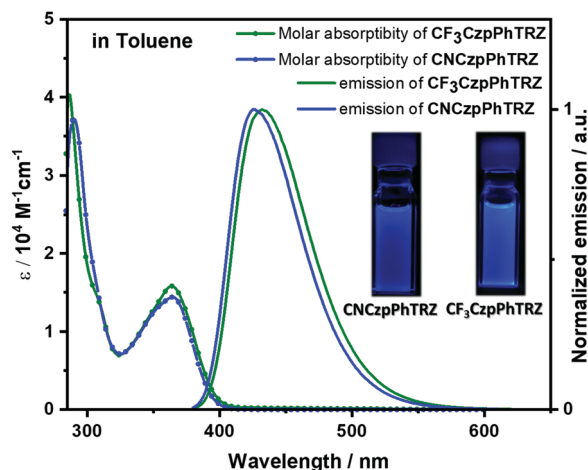


Fig. 3 Molar absorptivity and photoluminescence spectra of **CF₃CzpPhTRZ** and **CNCzpPhTRZ** in toluene at 298 K ($\lambda_{\text{exc}} = 360$ nm). Inset: photos of photoexcited solutions in toluene.

waves assigned to the reduction of the cyaphenine acceptor are quasi-reversible at -1.67 V vs. SCE. The reduction potentials are slightly shifted anodically compared to E_{red} of -1.78 V for CzpPhTRZ.¹⁸ The corresponding LUMO levels are -2.68 V and -2.67 V for **CF₃CzpPhTRZ** and **CNCzpPhTRZ**, respectively.

Solution-state photophysical studies

Both molecules show similar absorption and photoexcitation spectra in dilute toluene (Fig. 3 and Fig. S3, ESI†). The absorption bands of **CF₃CzpPhTRZ** at 363 nm and **CNCzpPhTRZ** at 365 nm are of similar intensities ($\epsilon = 1.6 \times 10^4$ vs. 1.4×10^4 $\text{M}^{-1} \text{cm}^{-1}$) and are assigned to an intramolecular charge transfer (ICT) from the carbazolophane donor to the triazine acceptor. These CT bands are *ca.* 10 nm hypsochromically shifted compared to the CT band of the reference compound **CzpPhTRZ**.¹⁸

The photoluminescence (PL) spectra of both emitters in toluene are shown in Fig. 3. The PL maxima of **CF₃CzpPhTRZ** at 432 nm and **CNCzpPhTRZ** at 426 nm are significantly blue-shifted compared to that of the reference **CzpPhTRZ** emitter ($\lambda_{\text{PL}} = 470$ nm). The emission spectra are unstructured, indicative of an excited state with a strong ICT character. The full-width-at-half maximum (FWHM) for **CNCzpPhTRZ** (0.36 eV) is slightly smaller than that of **CF₃CzpPhTRZ** (0.41 eV). The PL quantum yield (Φ_{PL}) values in degassed toluene are 80% for **CF₃CzpPhTRZ** and 73% for **CNCzpPhTRZ**, which decreased upon exposure to oxygen to 69% for **CF₃CzpPhTRZ** and 61% for **CNCzpPhTRZ**. The Φ_{PL} of both emitters in toluene is higher than that of the parent emitter (70% in degassed toluene and 55% after exposure to oxygen).¹⁸ Triplets are sensitive to quenching by oxygen and so the lower Φ_{PL} after exposure to oxygen shows that triplets play a role in the light emission process.^{23,24} The time-resolved PL decays of both compounds exhibited mono-exponential decay kinetics with PL lifetimes, τ , of 5.47 ns and 4.87 ns for **CF₃CzpPhTRZ** and **CNCzpPhTRZ**, respectively (Fig. S4, ESI†); no delayed emission in toluene was detected despite the oxygen sensitivity noted for the Φ_{PL} . Thus,

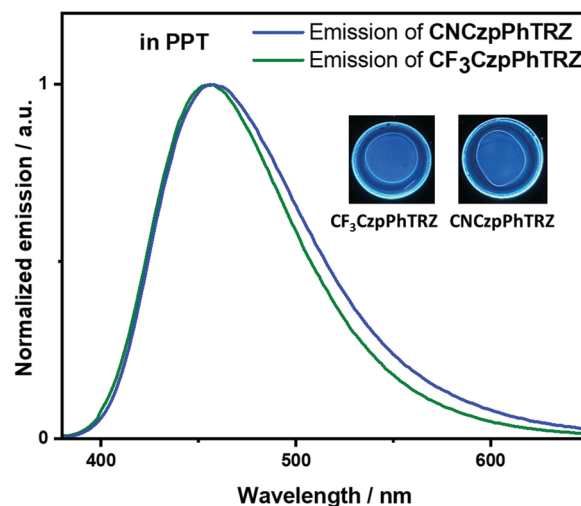


Fig. 4 Photoluminescence spectra of **CF₃CzpPhTRZ** and **CNCzpPhTRZ** dispersed at 10 wt% in PPT matrix at 295 K ($\lambda_{\text{exc}} = 360$ nm).

the delayed fluorescence in **CzpPhTRZ**, **CF₃CzpPhTRZ**, and **CNCzpPhTRZ** in toluene is smaller than we can reliably detect.

Photophysical properties in thin films

Recognizing examples where there is little evidence for TADF in solution but that TADF is prevalent in the solid-state for compounds with relatively large calculated ΔE_{ST} values,^{24,26} we next focused our attention on the photophysical properties of **CF₃CzpPhTRZ** and **CNCzpPhTRZ** in doped thin films. Initially, a photophysical investigation was conducted on spin-coated 10 wt% doped films in poly(methyl methacrylate) (PMMA) as the polarity of this host closely mimics that of toluene (PhMe). The absorption bands are of similar energy to those in PhMe, with the CT band observed at about 363 nm (Fig. S5, ESI†). Both compounds showed the unstructured ICT-based emission at around 435 nm, similar to that observed

Table 2 Photophysical properties of **CF₃CzpPhTRZ** and **CNCzpPhTRZ**^a

Emitters	$\lambda_{\text{PL}}/\text{nm}$	$\tau_{\text{p}}/\text{ns}$	$\tau_{\text{d}}/\mu\text{s}$	$\Phi_{\text{PL}}/\%$
In toluene ^b				
CF₃CzpPhTRZ	432	5.47	—	80 (69)
CNCzpPhTRZ	426	4.87	—	73 (61)
10 wt% doped in PMMA ^c				
CF₃CzpPhTRZ	436	6.84	151.4	63 (61)
CNCzpPhTRZ	435	7.32	132.1	62 (59)
10 wt% doped in PPT ^c				
CF₃CzpPhTRZ	456	6.63	158.3	70 (65)
CNCzpPhTRZ	458	6.75	135.0	65 (60)

^a At 298 K. ^b Quinine sulfate (0.5 M) in H_2SO_4 (aq.) was used as the reference (Φ_{PL} : 54.6%, $\lambda_{\text{exc}} = 360$ nm).²⁵ Values quoted are in degassed solutions, which were prepared by three freeze–pump–thaw cycles. The values in parentheses are in the presence of O_2 . ^c Thin films were prepared by spin-coating 10 wt% doped samples in PMMA and PPT. Steady-state and time-resolved emission spectra were recorded at 298 K under an O_2 -free atmosphere ($\lambda_{\text{exc}} = 360$ nm for steady-state and $\lambda_{\text{exc}} = 379$ nm for time-resolved emission). Photoluminescence quantum yields of thin films were determined using an integrating sphere ($\lambda_{\text{exc}} = 300$ nm or 360 nm) under N_2 atmosphere at 298 K. The values in parentheses are in the presence of O_2 .

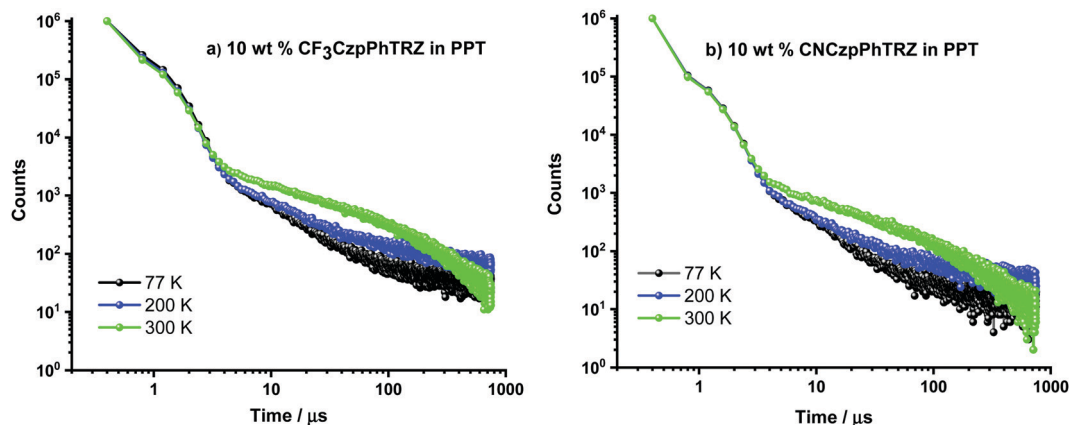


Fig. 5 Temperature-dependent time-resolved photoluminescence lifetime of 10 wt% doped films of (a) $\text{CF}_3\text{CzpPhTRZ}$ and (b) CNCzpPhTRZ in PPT ($\lambda_{\text{exc}} = 379 \text{ nm}$).

in PhMe (Fig. S5, ESI[†]). The PL spectra are blue-shifted by *ca.* 30 nm compared to that of the parent emitter **CzpPhTRZ**. The Φ_{PL} values are modestly decreased relative to those in solution at 63% and 62% for **CF₃CzpPhTRZ** and **CNCzpPhTRZ**, respectively, under N_2 atmosphere, likely reflective of some aggregation-caused quenching. Biexponential decay kinetics were observed in the time-resolved decays with τ_{p} values of 6.84 ns and 7.32 ns and τ_{d} values of 151.4 and 132.1 μs for **CF₃CzpPhTRZ** and **CNCzpPhTRZ**, respectively (Fig. S6, ESI[†]). We measured prompt fluorescence and phosphorescence spectra of both emitters in PMMA films at 77 K. The energy difference, taken from the onsets of these spectra, provides ΔE_{ST} values of 0.27 and 0.30 eV for **CF₃CzpPhTRZ** and **CNCzpPhTRZ**, respectively. These data are consistent with TADF being operative for the emitters in PMMA films. We then investigated the photophysical behavior of the both compounds in suitably high triplet energy and deep HOMO level OLED-relevant hosts such as bis[2-(diphenylphosphino)phenyl]ether oxide (DPEPO, $E_{\text{T}} = 3.0 \text{ eV}$, $E_{\text{HOMO}} = -6.3 \text{ eV}$)^{27,28} and 2,8-bis(diphenyl-phosphoryl)-dibenzo[*b,d*]thiophene (PPT, $E_{\text{T}} = 3.1 \text{ eV}$, $E_{\text{HOMO}} = -6.7 \text{ eV}$).²⁹ A concentration study of both emitters revealed that the Φ_{PL} was highest in 10 wt% doped films in PPT (Table S2, ESI[†]). The Φ_{PL} of 10 wt% **CF₃CzpPhTRZ** doped in PPT is 70%, while in DPEPO at the same doping concentration, it is 53%. Similarly, the Φ_{PL} of 10 wt% **CNCzpPhTRZ** doped in PPT is 65%, and in the corresponding DPEPO film, it is 52%. By contrast, the parent emitter **CzpPhTRZ** showed Φ_{PL} of 69% in the 10 wt% doped DPEPO film and only 55% in the 10 wt% doped PPT film.¹⁸ The absorption spectra in these host matrices are very similar to those in PMMA (Fig. S7, ESI[†]). However, λ_{PL} of both emitters are about 20 nm red-shifted at around 456 nm in these two host matrices, suggesting stronger aggregation^{30,31} (Fig. 4 and Fig. S8 (ESI[†]) and Table 2). The emission of both emitters is expectedly blue-shifted ($\sim 25 \text{ nm}$) in comparison to parent emitter **CzpPhTRZ** ($\lambda_{\text{PL}} = 482 \text{ nm}$ in 10 wt% doped films in DPEPO).¹⁸ The time-resolved PL decays in PPT show a nanosecond prompt emission and microsecond delayed emission at room temperature with comparable values to those measured in PMMA. The τ_{d} values of the 10 wt% doped PPT films are 158.3 μs for **CF₃CzpPhTRZ** and 135.0 μs for **CNCzpPhTRZ**, which is significantly

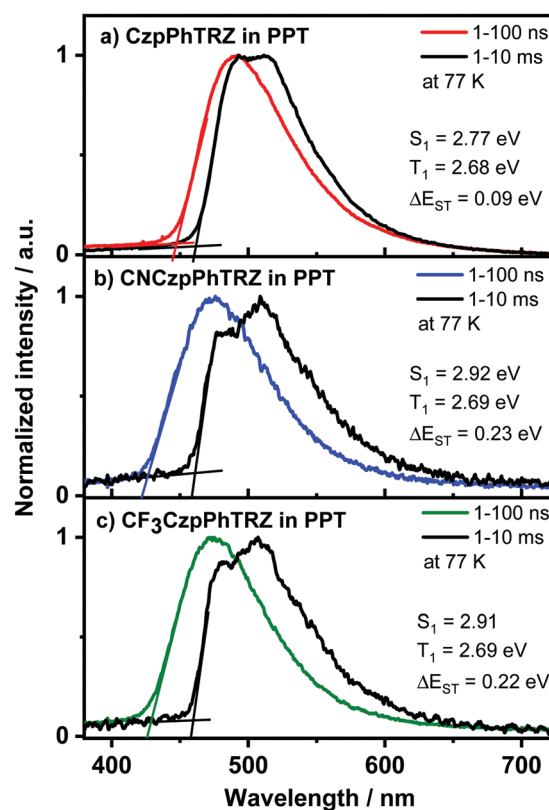


Fig. 6 77 K prompt PL and phosphorescence spectra measurement of 10 wt% (a) **CzpPhTRZ** (b) **CF₃CzpPhTRZ** and (c) **CNCzpPhTRZ** doped in PPT host ($\lambda_{\text{exc}} = 343 \text{ nm}$), the ΔE_{ST} value is taken from the onset value difference between the 77 K prompt fluorescence and phosphorescence spectra.

longer than that measured for parent emitter **CzpPhTRZ** in the 10 wt% DPEPO film ($\tau_{\text{d}} = 65 \mu\text{s}$). The temperature-dependent time-resolved PL decays in PPT are shown in Fig. 5. The prompt emission is insensitive to temperature, while the delayed emission is thermally activated, a behavior consistent with TADF.

We next prepared 10 wt% doped films of each emitter in PPT as a host matrix. As shown in Fig. 6, the S_1 levels of

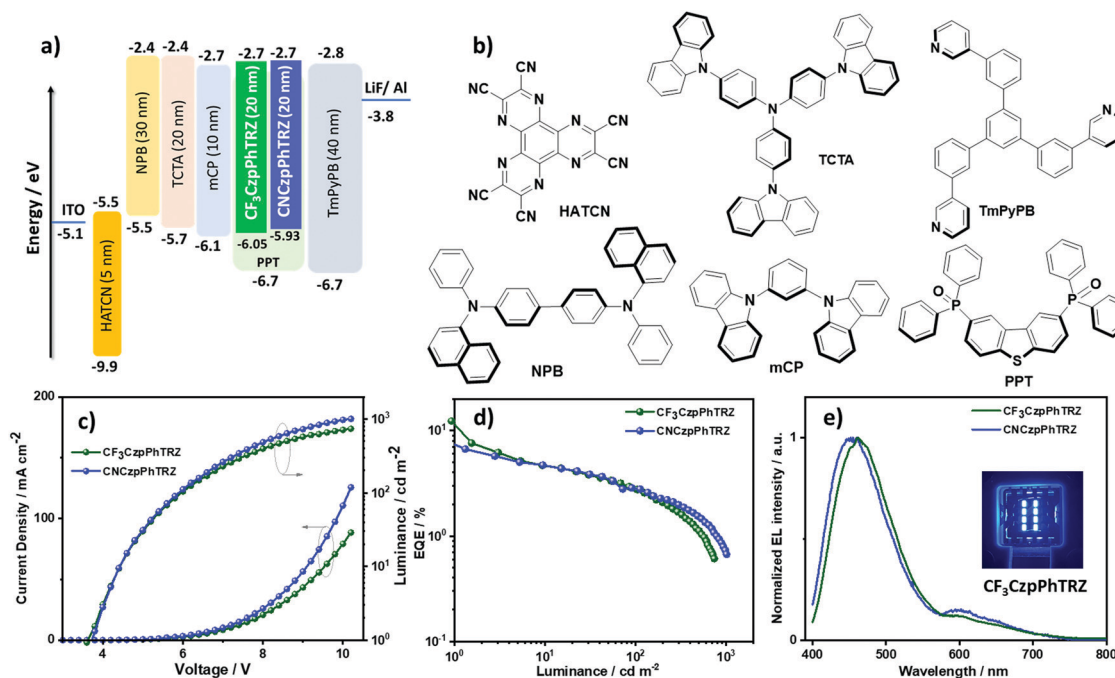


Fig. 7 (a) Energy level diagram of materials employed in the devices; (b) molecular structure of materials used in the devices; (c) current density and luminescence versus voltage characteristics for the devices; (d) external quantum efficiency versus luminescence curves for the devices; (e) electroluminescence spectra of the device of CF₃CzpPhTRZ and CNCzpPhTRZ, the inset is the electroluminescence of CF₃CzpPhTRZ.

CF₃CzpPhTRZ ($S_1 = 2.92$ eV) and CNCzpPhTRZ ($S_1 = 2.91$ eV) are very similar and at higher energy than that of the parent emitter CzpPhTRZ ($S_1 = 2.77$ eV); interestingly, the S_1 level of CzpPhTRZ in DPEPO is 2.89 eV. The S_1 energies in PPT are only slightly lower in energy than those determined in PMMA ($S_1 = 3.0$ eV, Fig. S9 (ESI[†])). The T_1 levels remain largely the same, regardless of the nature of the host medium or the structure of the emitter, and are of similar magnitude to their corresponding triphenyl triazine acceptor segments in toluene,³² indicating that the T_1 state is locally excited (³LE) character. The T_1 levels for CF₃CzpPhTRZ, CNCzpPhTRZ and CzpPhTRZ are all around 2.69 eV in PPT (Fig. 6) and closely resembles those measured in PMMA at 2.73 eV (Fig. S9, ESI[†]) and the reported triplet energy of CzpPhTRZ in DPEPO (2.73 eV).¹⁸ The corresponding ΔE_{ST} values are 0.22 eV for CF₃CzpPhTRZ, 0.23 eV for CNCzpPhTRZ, and 0.09 eV for CzpPhTRZ in PPT (Fig. 6 and Table 2); the reported ΔE_{ST} of 10 wt% CzpPhTRZ doped in DPEPO is 0.16 eV.¹⁸ The larger ΔE_{ST} values for CF₃CzpPhTRZ and CNCzpPhTRZ explain the longer delayed lifetimes and less efficient RISC, which can be rationalized due to the weaker electron-donor leading to a larger overlap integral.

Device characterization

Based on the promising photophysical properties of CF₃CzpPhTRZ and CNCzpPhTRZ in PPT films, we fabricated thermally-evaporated multilayer bottom-emitting OLEDs. The OLED stack architecture is shown in Fig. 7a and consisted of: indium tin oxide (ITO)/1,4,5,8,9,11-hexaazatriphenylene-heacarbonitrile (HATCN) (5 nm)/N,N'-di(1-naphthyl)-N,N'-diphenyl-(1,1'-biphenyl)-4,4'-diamine (NPB) (30 nm)/tris(4-carbazoyl-9-ylphenyl)amine (TCTA) (20 nm)/1,3-bis(N-carbazoyl)benzene (mCP) (10 nm)/emissive layer (20 nm)/1,3,5-tri[(3-pyridyl)-phen-3-yl]benzene (TmPyPB) (40 nm)/LiF (0.5 nm)/Al (100 nm), where HATCN plays the role of hole-injection layer (HIL), NPB and TCTA were employed as hole-transport layer (HTL), mCP acts as an electron/exciton-blocker layer (EBL) and TmPyPB plays the role of an electron-transport layer (ETL) and hole-blocking layer (HBL) due to its deep HOMO (−6.7 eV);³³ LiF acts as electron-injection layer (EIL) by modifying the work function of the aluminum cathode. The molecular structures of the materials used in these OLEDs are shown in Fig. 7b. The emission layer (EML) comprises 10 wt% of CF₃CzpPhTRZ or 10 wt% CNCzpPhTRZ doped into PPT. The electroluminescence properties of the OLEDs are summarized in Table 3, and current density–voltage–brightness

Table 3 Electroluminescence data for the devices

Emitter	Host	V_{on}^a/V	λ_{EL}^b/nm	$CE^c/cd A^{-1}$	$PE_{max}/lm W^{-1}$	$EQE^c/\%$	$CIE^d/x, y$
CF ₃ CzpPhTRZ (10 wt%)	PPT	3.7	460	16.8	13.7	11.6/2.7	0.186, 0.200
CNCzpPhTRZ (10 wt%)	PPT	3.7	456	9.1	7.5	7.4/2.7	0.194, 0.182

^a The turn-on voltage at a brightness 1 cd m⁻². ^b The electroluminescence maximum recorded at 6 V. ^c EQE_{max} at 1 cd m⁻² and EQE at 100 cd m⁻².

^d The CIE coordinates were recorded at 7 V.

There are no conflicts to declare.

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