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Recycling hazardous and energy-demanding piezoelectric ceramics using an oxide–halide perovskite upside-down composite method†

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Piezoelectric ceramics are widely employed in electromechanical coupling components such as sensors, actuators, and transducers owing to their excellent performance, cost-effectiveness, versatile shapes and configurations, and generous chemical/functional tuneability. However, most piezoceramics contain a significant amount of the hazardous and toxic element Pb, which is considered unsustainable. Although more environmentally friendly Pb-free counterparts have been developed, production of both Pb-containing and Pb-free piezoceramics still demands high energy consumption. On the other hand, retired and rejected piezoceramics are not properly recycled, leaving a large ecological footprint for the piezoelectric industry. Thus, there is a need for proper recycling methods which give wasted piezoceramics a second life at only a marginal energy cost, so that the energy and environmental footprints are minimized before the piezoceramics are permanently disposed of. This work demonstrates the fabrication and use of oxide–halide perovskite composites to recycle piezoceramics. The composites consist of oxide perovskite piezoceramic particles as the fillers that mimic crushed/recycled piezoceramics and an organometal halide perovskite, PTMA–CdCl₃ (PTMA is C₆H₅N(CH₃)₃), as the matrix/binder. The composites consume only about 1% of the energy budget required for making new equivalent piezoceramics whilst retaining comparable capabilities for sensing applications. The results suggest that such a recycling method is applicable to a broad range of piezoceramics, providing an incentive to possibly extend the method for simultaneously recycling other oxide and halide perovskite materials and components in the future.

Received 28th September 2023
Accepted 4th March 2024

DOI: 10.1039/d3su00348e

rsc.li/rscsus

Sustainability spotlight

Piezoelectric materials are essential for modern electronics, devices and equipment but their manufacturing suffers from high energy consumption and hazardous source materials. For example, they need to be treated at ultrahigh temperatures and many products contain toxic Pb. Piezoelectric materials and products are not properly recycled nowadays, leading to a heavy carbon footprint for this polluting yet important industrial sector. In this work, we demonstrate a reliable method for recycling these energy-demanding and non-environmentally friendly materials that is transferrable to many compounds. Moreover, our method gives a second life to the recycled materials in the smart sensing and energy harvesting domains, which helps to eliminate or postpone permanent disposal of existing hazardous elements. Therefore, our work contributes to the UN sustainable development goals of affordable and clean energy (SDG 7), sustainable cities and communities (SDG 11), responsible consumption and production (SDG 12), and climate action (SDG 13).

1. Introduction

Piezoelectric materials benefit from their capability of converting a mechanical input in the form of stress or strain to an electrical potential, and *vice versa*. Such a versatile electromechanical coupling behavior has made piezoelectrics a forefront of electronics research since their discovery in 1880.¹ Essential applications of piezoelectric components can be found in

devices including impact/touch sensors, accelerometers, gyroscopes, ultrasonic transducers, SAW (surface acoustic wave) devices, micro-mass sensors, transformers, high-precision actuators, micromotors, and energy harvesters.^{2,3}

Among piezoelectric materials, bulk piezoceramics based on the oxide perovskite structure offer many industrial and economic benefits. For instance, there is a broad range of selections of compositions for different working conditions and functional requirements. Those compositions are highly tunable to achieve desired properties, and the materials can be fabricated into complex shapes *via* cost-effective processes for various device configurations.¹

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† Electronic supplementary information (ESI) available. See DOI: <https://doi.org/10.1039/d3su00348e>

However, a major concern regarding piezoceramics is the extensive use of the hazardous and toxic element Pb. This is not compliant with the EU's legislation, RoHS (Restriction of Hazardous Substances in Electrical and Electronic Equipment), which intends to ban the use of hazardous elements in electronic devices,^{4,5} whereas the widely used piezoceramic composition, the PZT ($\text{Pb}(\text{Zr,Ti})\text{O}_3$) family, contains a significant amount of Pb (up to 68 wt%). Environmentally friendly Pb-free piezoceramics have been developed to tackle this issue.⁶ Some Pb-free alternatives are able to perform similarly to their Pb-based counterparts.^{6,7}

Regardless of being environmentally friendly, manufacturing of Pb-free piezoceramics, as is the case also for Pb-based piezoceramics and most other ceramics in general, involves at least two high-temperature treatment steps: (i) calcination (usually $>600^\circ\text{C}$) for formation of desired phases, and (ii) sintering (usually $>1000^\circ\text{C}$) for necessary material densification and grain growth.¹ The sintering is particularly important because most signature functionalities of piezoceramics, such as excellent piezoelectric coefficients and strong electromechanical coupling, are obtained only after this step.

The need for high-temperature treatments inevitably leads to high energy consumption, making all industrial sectors which deploy piezoelectric components energy intensive and unsustainable. In addition, the high-temperature densification procedure causes unpredictable shrinkage as large as 15–25% and unfavored chemical diffusion when co-sintered with other materials, both of which significantly degrade the compatibility of piezoceramics for integration with other components.⁸

One method to reduce the energy footprint of piezoceramic manufacturing is to properly recycle retired or broken piezoelectric components after use as well as the low-quality ones that have already been rejected during fabrication. Ideally, the recycled materials should also be given a second life at service to the best possible extent before they are permanently disposed of. A prerequisite here is that the recycling procedure consumes only a marginal amount of energy compared to that needed for making correspondingly new materials. Meanwhile, the materials made *via* recycling should obtain comparable properties to those of newly sintered piezoceramics. Otherwise, the energy benefit provided by recycling will be outweighed by the better performance of newly made materials.

Among possible methods that can be utilized for recycling piezoceramics,^{7,9} the ultralow-temperature densification route is considered viable due to low energy consumption and simple fabrication procedures.^{10–12} In this method, a small amount of liquid acts as a transport phase during densification of a functional filler phase under pressure, with or without mildly elevated temperature ($<350^\circ\text{C}$). In particular, the transport phase coats the filler particles to form a core-shell microstructure,⁸ and upon applying high pressure, the transport phase redistributes itself due to surface tension and capillary action. Dissolution of the filler particles into the transport phase occurs at contacting points which experience a very high, GPa-level local pressure. The dissolved filler phase then, due to evaporation of the transport phase at mild temperature or at room temperature though with a slower pace, reprecipitates at

sites away from the contacted edges due to a favorable concentration gradient which drives densification and grain growth.¹³

The ultralow-temperature densification method has shown promise using dielectric ceramic particles as the filler phase. For instance, the first truly room-temperature densified ceramic has been reported where Li_2MoO_4 (LMO) particles are used as the filler phase with deionized water as the transport phase, which are then shaped and densified under 130–150 MPa.^{10,11} The subsequently obtained dielectric properties are measured to be better than those of the LMO ceramics conventionally sintered at 540°C . The advantage here is that the functional phase, LMO, is soluble in water. The LMO–water solution also acts as a functional binder and precipitates in the intergranular areas between the filler particles through redistribution and evaporation of the solution under pressure. This has eased transportation of the functional filler phase to a large extent in the form of a transport solution and hence, after being shaped under high pressure, better functionality is achieved. The resultant material is a ceramic–ceramic composite with much higher filler volume fraction compared to the case for conventional composites, and it is thus termed an ‘upside-down’ composite. Using LMO as the binder phase, TiO_2 -based,¹² SrTiO_3 -based,¹⁴ and $(\text{Ba}_{0.55}\text{Sr}_{0.45})\text{TiO}_3$ -based¹⁵ dielectric upside-down composites have been successfully made.

Recently, piezoelectric upside-down composites have also been made using PZT as the filler.¹⁶ Instead of LMO–aqueous solution, an organotitanate precursor gel has also been applied as the binder.¹⁷ Under a pressure of 250 MPa and at 300°C , the precursor gel forms amorphous TiO_2 (am- TiO_2) in the intergranular areas of the PZT particles and thus an upside-down PZT–(am- TiO_2) composite is obtained in addition to the PZT–LMO composite.

It is worth noting that the PZT fillers used in the above-mentioned composites are made by crushing high-temperature sintered ceramics. Although the method itself does not save energy in a new fabrication round, it adds merits to a possible recycling procedure because mechanically and/or electrically broken piezoceramic components, which need to be reshaped, are naturally appropriate sources of the filler materials. Moreover, although the PZT–LMO and PZT–(am- TiO_2) composites perform worse than the PZT ceramics in terms of generic piezoelectric response, they are particularly competitive regarding the piezoelectric voltage coefficient (g)^{16,17} which is an indicator for potential sensitivity of piezoelectric sensors made from the same material. The possible recycling is then highly likely to succeed in making sensors from a wide range of recycled piezoceramics. Therefore, this work reports the recycling of Pb-based and Pb-free piezoceramics using the upside-down composite fabrication method with an organometal halide perovskite compound, PTMA– CdCl_2 where PTMA is $\text{C}_6\text{H}_5\text{N}(\text{CH}_3)_3$ (ref. 9 and 18), as the binder phase.

The motivation here is attributed to the compound's capability of dissolving in an organic solvent (advantageous over LMO to reduce hygroscopicity) and its subsequent recrystallization at closer to room temperature (advantageous over TiO_2 to eliminate the need for further elevated temperature). In this



work, four types of filler materials are chosen, including two Pb-free piezoceramics, BaTiO₃-based¹⁹ and (K,Na)NbO₃-based,²⁰ and two Pb-based ones, soft-type²¹ and hard-type²² PZT, respectively. The four chosen compositions tend to represent a wide range of piezoceramics used in different categories of applications under various working conditions and with distinctive functional requirements. The results obtained from these four types of upside-down composites are compared to their pristine ceramic counterparts to evaluate the extent of gain or degradation in functional properties.

2. Materials and methods

2.1. Fabrication of piezoceramic pellets and fillers

Table 1 summarizes the materials fabricated in this work, including information on filler compositions, binder volume fractions (where applicable), and densities. Four types of oxide perovskite structured piezoceramics were fabricated *via* the solid-state route. The two Pb-free ceramic powders were synthesized. Reactants of BaCO₃ (99%, Alfa Aesar, Germany) and TiO₂ (99.9%, Aldrich Chemistry, Canada) for BaTiO₃ (BT), and K₂CO₃ (99%, J.T. Baker, Holland), Na₂CO₃ (99%, Sigma-Aldrich, Germany), BaCO₃ (99.98%, Sigma-Aldrich, USA), Nb₂O₅ (99.9%, Aldrich, China) and NiO (99.999%, Aldrich Chemistry, USA) for (K_{0.52}Na_{0.38}Ba_{0.03})(Nb_{0.985}Ni_{0.015})O_{2.9575} (KNBNNO) were first weighed according to stoichiometry.

Ball-milling was then performed with ZrO₂ beads (for BT) or without beads (for KNBNNO as the use of beads during mixing of raw powders caused substantial stoichiometric loss of alkali carbonates) in ethanol for 6 hours. The dried mixture was subjected to calcination at 1100 °C (BT) or 750 °C (KNBNNO) for 4 hours. K₂CO₃, Na₂CO₃ and NiO were stored at 220 °C prior to weighing owing to their hygroscopicity.²³

The two Pb-based ceramic powders, H-PZT (APC-841, a hard-type PZT) and S-PZT (PZ29, a soft-type PZT), were purchased from American Piezo International (APC) Ltd. (USA) and Meggitt A/S (Denmark), respectively. To obtain a suitable particle size

distribution, a ball milling procedure was performed for the as-purchased powders with ZrO₂ beads in ethanol for 12 hours. The same procedure was also performed for the calcined Pb-free powders. After drying, each milled powder was sieved through a 200 µm mesh and was then shaped into green bodies with diameter of 10 mm for BT, KNBNNO and H-PZT, 13 mm for S-PZT, and 14.5 mm for all ceramic powders under uniaxial pressure of approximately 90–110 MPa using polyvinyl alcohol (PVA) as the binder. The binder was subsequently burnt off at 550 °C for 10 hours and the green bodies were finally sintered at 1450 °C for 4 hours (BT, on a Pt substrate), at 1150 °C for 2 hours (KNBNNO, on a Pt substrate), at 1100 °C for 4 hours (H-PZT, in an alumina crucible), and at 1200 °C for 2 hours (S-PZT, in an alumina crucible), respectively. Note that the KNBNNO samples were covered under an alumina crucible during sintering to help suppress possible evaporation of the alkali elements.²³ Also, powder beds were made for the H-PZT and S-PZT using the same corresponding powders and the samples were covered under an alumina crucible to help prevent possible loss of Pb during the high-temperature sintering process.¹ Dimensions of the four types of sintered pellets are shown in Table S1 in the ESI.†

Smaller ceramic pellets were structurally and electrically characterized, whereas the larger pellets shaped with the 14.5 mm die were crushed in a mortar using a pestle. The crushed ceramics were sieved through a series of meshes of 425 µm, 180 µm, and 63 µm, consecutively. Particles in sizes between 63 and 180 µm, which mimicked filler materials recycled from worn-out devices, were collected for the following upside-down composite fabrication.

2.2. Synthesis of PTMA–CdCl₂

The starting reactants of PTMA–Cl (≥98%, Sigma-Aldrich, USA) and CdCl₂ (99.99%, Sigma-Aldrich, USA) were weighed stoichiometrically. A mixed solution of acetonitrile and deionized water (volume ratio of 3 : 2) was used to dissolve the reactants in a concentration of 13.3 w/v% and then the solution containing

Table 1 Summary of compositions and densities of the piezoceramics and upside-down composites fabricated in this work

Sample ID	Form of material	Filler composition	Binder volume fraction (%)	Measured density (g cm ⁻³)	Theoretical density (g cm ⁻³)	Relative density (%)
BT-P	Ba(Zr _x Ti _{1-x})O ₃ ceramic pellet	—	—	~5.44	5.90 ± 0.06	92.2 ± 1.0
KNBNNO-P	(K _{0.52} Na _{0.38} Ba _{0.03})(Nb _{0.985} Ni _{0.015})O _{2.9575} ceramic pellet	—	—	~4.34	4.53 ± 0.07	95.8 ± 1.5
H-PZT-P	Hard-type PZT ceramic pellet	—	—	~7.58	7.86 ± 0.14	96.5 ± 1.7
S-PZT-P	Soft-type PZT ceramic pellet	—	—	~7.17	7.54 ± 0.10	95.1 ± 1.3
BT-C	Composite	Ba(Zr _x Ti _{1-x})O ₃	~14.7	4.52 ± 0.12	~5.30	85.4 ± 2.2
KNBNNO-C	Composite	(K _{0.52} Na _{0.38} Ba _{0.03})(Nb _{0.985} Ni _{0.015})O _{2.9575}	~11.7	3.59 ± 0.03	~4.21	85.2 ± 0.6
H-PZT-C	Composite	Commercial hard-type PZT	~18.7	5.93 ± 0.05	~6.72	88.1 ± 0.8
S-PZT-C	Composite	Commercial soft-type PZT	~18.1	6.05 ± 0.05	~6.50	93.0 ± 0.8



$\text{C}_6\text{H}_5\text{N}(\text{CH}_3)_3^+$, Cd^{2+} , and Cl^+ was stirred for 24 hours.¹⁸ After passing through a 0.2 μm PTFE (polytetrafluoroethylene) filter, cm-sized PTMA- CdCl_3 crystals were grown *via* slow precipitation statically in 3–4 weeks.

2.3. Fabrication of upside-down composites

Using crushed and particle size-controlled (63–180 μm) ceramics as the filler and PTMA- CdCl_3 as the binder, four types of upside-down composites were made. 60 w/v% of ceramic filler particles were added in a 0.6 v/w% PTMA- CdCl_3 acetonitrile-deionized water solution. The suspension was kept being stir-mixed whilst evaporating the solvents on a hot plate. Filler particles coated with PTMA- CdCl_3 were obtained after the suspension was dried. Such a coating procedure followed the same mechanism as is reported in the literature.¹⁵ The coated filler particles were then mixed with 5 wt% crushed PTMA- CdCl_3 crystals in a mortar with a pestle. During mixing, 3–4 drops of saturated PTMA- CdCl_3 solution (prepared in Section 2.2) was also added to each gram of the filler material as the transport phase during shaping. The obtained mixture was then uniaxially hot-pressed in a cylindrical steel mold of 10 mm diameter under 250 MPa and at 150 $^\circ\text{C}$ for 30 minutes. Before releasing the pressure and being removed from the mold, the sample was cooled down to 50 $^\circ\text{C}$. The amount of filler used to prepare each upside-down composite sample and dimensions of the composite samples are listed in Table S1.†

2.4. Structural and electrical characterization

The density of the samples was measured using either Archimedes' method in ambient air and deionized water or the dimensional method (weight/volume). XRD (X-ray diffractometry) was used to analyze phases of crushed ceramics using a diffractometer (SmartLab 9 kW, Rigaku, Japan) under $\text{Co K}\alpha_1$ radiation ($\lambda = 1.7889 \text{ \AA}$) at a scanning rate of 4° min^{-1} for 2θ . Lattice parameters were determined using the Rietveld refinement method in PDXL2 software. Elemental analysis was carried out on the carbon-coated surface of ceramic pellets using an electron-probe microanalyzer (EPMA) (JXA-8530F Plus, JEOL, Japan) with 15 kV acceleration voltage, 15 nA probe current, and 1–5 μm beam spot size. Nominal chemical formulae of the ceramics were calculated from the EPMA results. Using the unit cell volume (V) calculated from XRD results and theoretical molecular weight (A) calculated from EPMA results, the theoretical density (ρ_f) of each filler material was calculated based on eqn (1), where N is Avogadro's number and n is number of atoms per unit cell. Eqn (2) calculates the binder volume fraction (ϕ_b) in the composite, where m_f is the mass fraction of the filler (0.95 for all samples) and ρ_b is the theoretical density of PTMA- CdCl_3 ($\sim 1.80 \text{ g cm}^{-3}$, provided by single crystal XRD measurement in previous work¹⁸). Eqn (3) determines the theoretical density of the composite (ρ_c) with volume fractions of the filler (ϕ_f) and ϕ_b .

$$\rho_f = \frac{nA}{vN} \quad (1)$$

$$\phi_b = 1 - \frac{\rho_b}{\rho_f \left(\frac{1}{m_f} - 1 \right) + \rho_b} \quad (2)$$

$$\rho_c = \rho_b \phi_b + \rho_f \phi_f \quad (3)$$

The cross-sectional surface of the composite samples was polished in a vacuum under a cooling cross-section ion polisher (JEOL IB-19520CCP, Japan) with a stainless-steel source. The polished, carbon-coated surface was characterized under a field-emission scanning electron microscope (FESEM, ULTRA Plus, Carl Zeiss SMT AG, Germany) equipped with an energy dispersive X-ray spectroscopy (EDS) system under an acceleration voltage of 15 kV.

The ceramic pellets were polished down to 0.95–1.05 mm thick from both surfaces on a P1200 abrasive paper. Polished H-PZT and KNN pellets were painted with an Ag-epoxy paste (H20E-4GM, Epoxy Technology, USA) which was cured at 150 $^\circ\text{C}$ for 30 minutes as the electrode. Polished S-PZT and BT pellets were screen-printed with an Ag conductive paste (735825, Sigma-Aldrich Co., South Korea) which was cured at 120 $^\circ\text{C}$ for 30 minutes. Another Ag conductor paste (5065, Dupont, UK) was painted on the composite samples and was cured at 130 $^\circ\text{C}$ for 10 minutes as the electrode.

Dielectric properties were measured at room temperature in the dark with an LCR meter (E4980AL, Keysight Technologies, USA). Poling was performed in the dark, in a silicone oil bath, and at room temperature for a period of 10 minutes. The poling electric fields for both ceramics and composites were 3 kV mm^{-1} , 4 kV mm^{-1} , 6 kV mm^{-1} , and 3 kV mm^{-1} for BT, KNN, H-PZT, and S-PZT, respectively. The poled samples were electrically shorted for 24 hours, and their respective longitudinal piezoelectric charge coefficient (d_{33}) was subsequently measured with a d_{33} meter (YE2730A, APC International Ltd., USA) under 0.25 N alternating force at 110 Hz.²⁴

3. Results and discussion

3.1. Phases and microstructure

Data associated with this work are openly available.²⁵

The rationale for choosing these four piezoceramic compositions as the fillers was that they represent a broad range of material features and piezoelectric applications. BT-based compositions are popular Pb-free options showing moderate to very high piezoelectric response with relatively low Curie temperature compared to other piezoelectric counterparts.⁶ Their structural and electrical properties have been well understood in a large number of previous studies.^{26,27} They are also widely used in MLCCs (multilayer ceramic capacitors), which are an essential component with each electronic device containing hundreds or thousands of units.²⁸ Therefore, BT was an ideal fundamental composition to study in this work, assuming the large MLCC market and other BT-based piezoelectric components to be potential resources for future recycling.

The KNN ((K,Na)NbO₃) based compositions are another popular family of Pb-free piezoelectrics featuring high Curie



mmm and tetragonal crystal structure) with reflections at 2θ of 31.82° , 34.07° , 45.61° , and 58.14° . A reflection at 2θ of 33.40° could not be identified, possibly due to minor impurities present in the starting reactants.

In Fig. 1b, the lattice parameters of the perovskite phase for KNNBNNO agreed with those in the literature.³⁴ In addition, a secondary phase was formed with evidence of minor reflections between 29° and 42°. They matched a tungsten bronze phase with a chemical formula of Ba₃Nb_{4.67}Ni_{0.33}O₁₅ (space group: 100: *P4bm* and tetragonal crystal structure) which is also typically seen in the literature.²⁰ According to Fig. 1c and d, both the sintered H-PZT and S-PZT ceramics formed expected perovskite phases which were identical to those existing in the as-purchased powders. However, the H-PZT had some reflections attributed to secondary or unknown phases at various 2θ across the measurement range. This might be because of possible Pb loss during sintering which caused an uneven distribution of dopants. Evidence could be found in the EDS maps for the hard-PZT composite (H-PZT-C) in which Sb and Mn were seen to be inhomogeneously spread across the filler particles, as shown in Fig. S7.† As expected, the secondary phase was matched to Pb(Zr_{0.5}Ti_{0.5})O₃ (space group: 227: *Fd3m*, choice-2 and cubic crystal structure). Similarly, the S-PZT had an unidentified reflection at 2θ of 49.7° which could be assigned to the segregated Sr detected in the fillers of the soft-PZT composite (S-PZT-C), as shown in Fig. S8.†

According to subsequently calculated data shown in Table 1, all the ceramic pellets achieved relative densities of 91–97%, indicating good microstructural quality of the sintered ceramic fillers.

The good filler quality, together with the proper selection of the binder, consequently induced high-quality upside-down composites. As demonstrated in Fig. 2, a high packing density between the BT filler and PTMA-CdCl₃ binder particles can be observed. Referring to the term upside-down composite, the FESEM micrograph and EDS maps clearly visualized the high volume-fraction of the BT filler. Meanwhile, no major microstructural defects, such as voids, pores, or cracks, were noticed at the boundaries between BT and PTMA-CdCl₃ particles. Similar situations shown in Fig. S6-S8† concluded a successful fabrication of the upside-down composites between oxide and halide perovskites in this work with a similar microstructure that has been reported in previous works.^{15,16} EDS maps for all the composites indicated that constitutive elements of the fillers and those of the binder were seen only in their respective particles. This implies that no reaction happened between the filler and binder, likely due to the process combining a lower fabrication temperature and a pressure-assisted densification procedure which did not provide enough activation energy for cross reaction, as opposed to the case for high-temperature sintering.

3.2. Dielectric properties

Fig. 3 shows the frequency-dependent dielectric properties of samples BT-P (ceramic) and BT-C (composite) in the frequency range of 20 Hz to 100 kHz. The frequency dependent dielectric

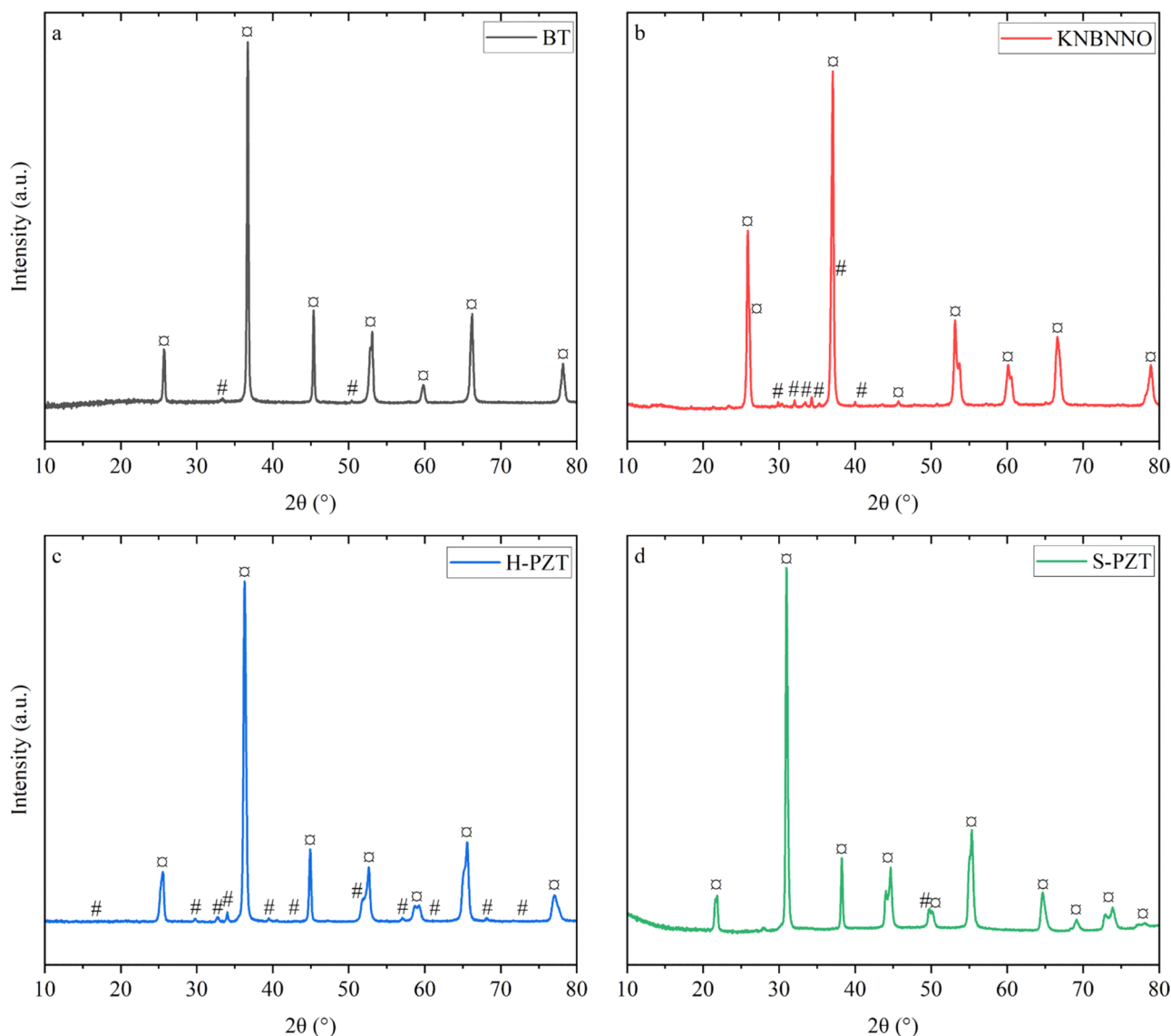


Fig. 1 XRD reflections of piezoceramic fillers: (a) BT, (b) KNNBNO, (c) H-PZT, and (d) S-PZT. Major perovskite phases are marked with 'ø' and secondary phases are marked with '#'.[†]

properties for the KNNBNO, H-PZT and S-PZT ceramic and composite samples in the same frequency range are shown in Fig. S9–S11 in the ESI.[†]

The relative permittivity (ϵ_r) and dielectric loss ($\tan(\delta)$) for both the unpoled and poled states are provided in the figures.

The distinctive values of unpoled and poled permittivity and dielectric loss indicated an effective poling procedure for all the ceramic and composite samples. From the figures, it is seen that the composite samples generally exhibited a decreasing trend of permittivity and dielectric loss with the increase of frequency,

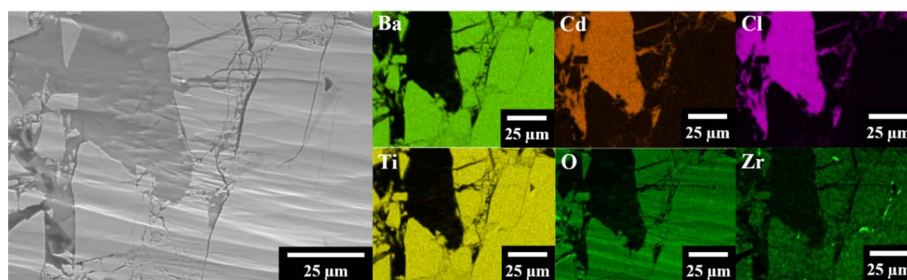


Fig. 2 FESEM micrograph and EDS maps of upside-down composite sample BT-C.



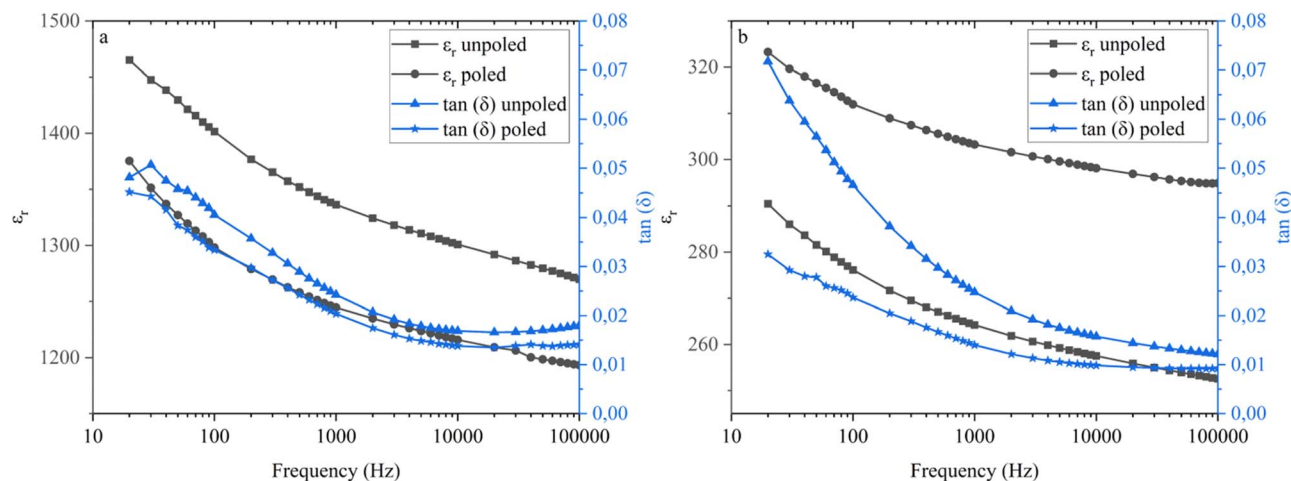


Fig. 3 Dependence of relative permittivity (ϵ_r) and dielectric loss ($\tan(\delta)$) on frequency for unpoled and poled (a) BT-P and (b) BT-C measured at room temperature in the dark.

whereas some ceramic samples such as KNBNNO-P and S-PZT-P tended to be more sensitive to frequency change. The behavior of frequency dependence was affected by different resonant frequencies of dipoles and ferroelectric domains.³⁵ All the composite samples had poled permittivity values greater than their respective unpoled permittivity values possibly owing to a ferroelastic contribution from the PTMA-CdCl₃ binder.¹⁸ During poling of the composites, the ferroelastic properties of the PTMA-CdCl₃ binder eased the domain pinning effect that would have otherwise resulted on the fillers with a non-ferroelastic binder. This type of polarization behaviour has already been observed in ferroelectric hysteresis loops conducted on upside-down composites fabricated with PTMA-CdCl₃ as a binder.⁹

In the ceramic samples, the trend of the unpoled and poled permittivity values was not straightforward compared to the composite samples. For instance, in the H-PZT-P and S-PZT-P ceramic samples, the unpoled permittivity values were smaller than their poled counterparts. This is attributed to the tetragonal crystal structures of their primary phases (Tables S7 and S9†). Increased permittivity after poling is often empirically observed due to the dominance of 180° domain reorientation.^{36,37}

The evolution of permittivity after poling for the BT-P and KNBNNO-P ceramics was analogous with orthorhombic phases identified (Tables S3 and S5†), giving decreased permittivity values in the poled states associated with non-180° domain reorientation.

Fig. 4 compares the dielectric properties between the ceramic and composite samples in a more straightforward way by extracting unpoled relative permittivity and dielectric loss values at 1 kHz. Percentages of ϵ_r and $\tan(\delta)$ for the composites with respect to those for the ceramic counterparts are also placed alongside. As can be seen in Fig. 4, all the ceramics achieved comparable dielectric properties to those reported in the literature correspondingly.^{20,26,38,39} More information can be found in data associated with this work.²⁵

As shown in Fig. 4a, the BT composite (BT-C) yielded the largest relative permittivity (≈ 245) compared to other composites (*i.e.*, ≈ 170 for KNBNNO-C and ≈ 210 for H-PZT-C and S-PZT-C). It can also be noticed that across different compositions, the distribution of ϵ_r values of the composites was relatively narrow (*i.e.*, within 30–45% variation) compared to that of the ceramics (*i.e.*, as large as 50–120% variation).

The reduced and narrowly distributed permittivity in the composites was due to the interplay between the inherent contribution of permittivity from the filler to the composite and thus the ‘composite effect’ that occurred between a high-permittivity filler and a low-permittivity binder ($\epsilon_r \approx 10$ for the PTMA-CdCl₃ crystal¹⁸). A larger difference between permittivity values of the filler and binder would increase the dominance of the binder in contribution to measured permittivity of the overall composite.^{40–42} This phenomenon appeared due to a lack of interconnectivity between the filler particles in 0–3 composites where fillers are randomly dispersed in the binder, which is the case in this work. During the LCR measurement, the input electric field was actually applied more on the low-permittivity binder than on the high-permittivity fillers.^{40,41,43}

Therefore, the largest ϵ_r value exhibited by BT-C among all the composites was attributed to a moderate permittivity value of the BT filler acting against its own composite effect (Fig. 3a). The composite effect could also be clearly seen when assessing the retainment of ϵ_r values in composites compared to those in corresponding ceramics. As shown in Fig. 4a, the percentage of ϵ_r values of the composites out of their ceramic counterparts was about 20% for the pairs of BT-C/BT-P, KNBNNO-C/KNBNNO-P and H-PZT-C/H-PZT-P. However, the percentage for the S-PZT-C/S-PZT-P pair was only about 10% owing to more disparate permittivity values between the S-PZT filler and the PTMA-CdCl₃ binder (Fig. S11a†).

It should be noted that the upside-down composites prepared in this study had filler volume fractions higher than that in conventional 0–3 dielectric and piezoelectric composites which enabled the filler's permittivity to have a more significant



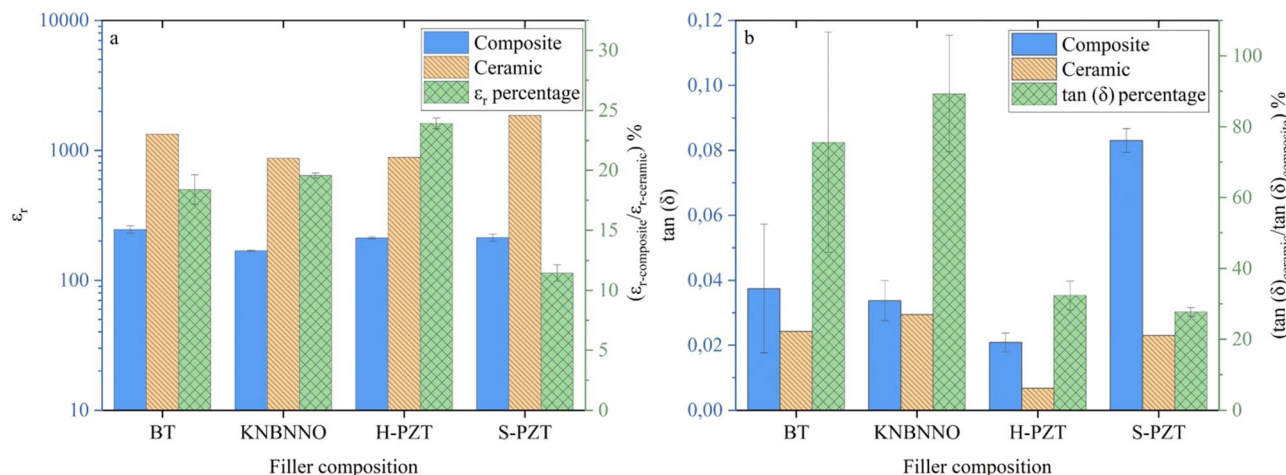


Fig. 4 (a) Relative permittivity (ϵ_r) and (b) dielectric loss ($\tan(\delta)$) of unpoled BT, KNBNNO, H-PZT, and S-PZT ceramic and composite samples measured at 1 kHz, and ratio of (a) the composite relative permittivity value ($\epsilon_{r-composite}$) to the corresponding ceramic relative permittivity value ($\epsilon_{r-ceramic}$) shown as a percentage ($(\epsilon_{r-composite}/\epsilon_{r-ceramic})\%$) and (b) the ceramic dielectric loss value ($\tan(\delta)_{ceramic}$) to the corresponding composite dielectric loss value ($\tan(\delta)_{composite}$) shown as a percentage ($(\tan(\delta)_{ceramic}/\tan(\delta)_{composite})\%$) for the four compounds.

influence in the whole composite sample and thus the composite effect in this work was less severe compared to the cases in previous works.^{44,45} This also provides an advantage of using upside-down composites over using conventional 0–3 composites for recycling.

As can be seen in Fig. 4b, $\tan(\delta)$ values of the composites were in the range of 0.02–0.08 and they were higher than those of their ceramic counterparts. This was due to the same composite effect discussed above, which forced the electric field during LCR measurement to concentrate on the highly lossy PTMA–CdCl₃ binder ($\tan(\delta) \approx 0.25$ (ref. 18)) rather than on the low-loss fillers ($\tan(\delta) \approx 0.01$ –0.03).

The composite effect phenomenon also explains the largest $\tan(\delta)$ value of ≈ 0.08 in S-PZT-C among all the composites because of exaggerated distortion of electric field distribution between the filler and binder, as discussed above. The S-PZT-P and H-PZT-P showed the lowest level of $\tan(\delta)$ values and, as a result, the ratio of the ceramic $\tan(\delta)$ value to the composite $\tan(\delta)$ value was significantly lower for these two compounds (Fig. 4b). However, the absolute $\tan(\delta)$ values for these two types of composites were distinctive where S-PZT-C was three times more lossy ($\tan(\delta) \approx 0.08$) than H-PZT-C ($\tan(\delta)$ as low as 0.02). The higher loss in S-PZT-C could also be augmented by the ‘soft’ properties of the S-PZT filler.^{46,47} The H-PZT-C implied a good retention of the ‘hard’ properties of its filler in the composite.

In addition to the influence caused by the difference between the filler and binder, the dielectric properties of the composites could also be affected by possible presence of microstructural voids such as intergranular pores, although no major pores were visualized in the microstructure of the fabricated upside-down composites (Fig. 2 and S6–S8†).

3.3. Piezoelectric properties and trade-off between energy consumption and performance

Fig. 5 and Table 2 compare the piezoelectric properties of the ceramic and composite samples fabricated in this work between

each other as well as with relevant piezoceramics and composites selected from the literature, including frequently studied excellent compositions and their variants based on BT, (Ba_{0.85}Ca_{0.15})(Ti_{0.9}Zr_{0.1})O₃ (BCZT), KNN, PZT, Pb(Mg_(1/3)Nb_(2/3))O₃–PbTiO₃ (PMN–PT), and commercial PZT-5A.^{16,17,32,48–58}

Fig. 5a shows the d_{33} values of the four types of ceramics and their corresponding composites studied in this work alongside the percentage of the composite d_{33} values with respect to those of their ceramic counterparts. The percentages provide information on the retainment of the piezoelectric properties of the fillers after being fabricated into upside-down composites. According to Fig. 5a, all the ceramics exhibited comparable d_{33} values to those reported in the literature.^{20,26,38,39} This proves that the presence of the identified and unidentified secondary or impurity phases detected by XRD (Fig. 1) did not degrade the piezoelectric properties to a noticeable extent and thus was not considered as an issue for the upside-down composites made with these ceramic fillers.

Fig. 5a also shows the fact that variation of the ceramic and composite d_{33} values followed the same trend, i.e., S-PZT-P (≈ 510 pC N^{−1}) > BT-P (≈ 295 pC N^{−1}) > H-PZT-P (≈ 200 pC N^{−1}) > KNBNNO-P (≈ 85 pC N^{−1}) for ceramics, and S-PZT-C (≈ 100 pC N^{−1}) > BT-C (≈ 45 pC N^{−1}) > H-PZT-C (≈ 40 pC N^{−1}) > KNBNNO-C (≈ 20 pC N^{−1}) for composites. This fact indicated that no net effect on piezoelectric properties was contributed by the binder. The decrease of d_{33} from the ceramics to the composites could be attributed to reasons as follows:

(1) As has been explained for dielectric properties, the electric field distribution inside the composite samples was dominated by the low-permittivity PTMA–CdCl₃ which was induced by the composite effect. As a result, the actual poling electric field in the composite was lower than the applied electric field,^{45,59} as illustrated by eqn (4), where E_{filler} is the electric field in the filler, $E_{applied}$ is the applied electric field, R is the ratio of the average filler particle size to the filler interparticle distance, and ϵ_{binder} and ϵ_{filler} are relative permittivities of the PTMA–CdCl₃ and filler, respectively.



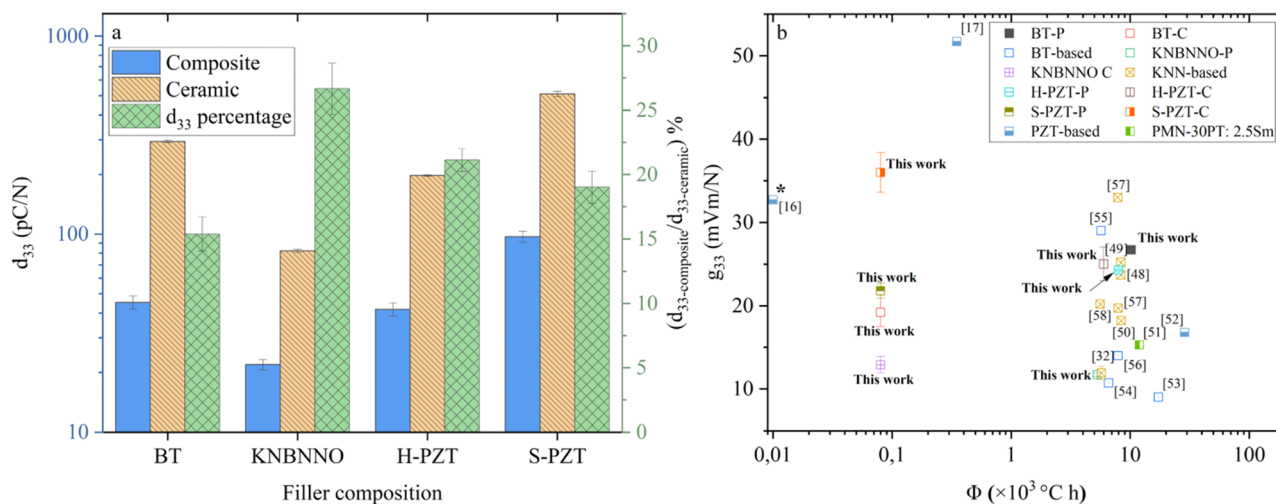


Fig. 5 (a) d_{33} of poled BT, KNBNNO, H-PZT and S-PZT ceramic and composite samples and ratio of the composite d_{33} value ($d_{33\text{-composite}}$) to the corresponding ceramic d_{33} value ($d_{33\text{-ceramic}}$) shown as a percentage ($(d_{33\text{-composite}}/d_{33\text{-ceramic}})\%$); and (b) dependence of longitudinal piezoelectric voltage coefficient (g_{33}) on the energy budget of fabrication (Φ) for the ceramic and composite samples studied in this work as well as for selected high-performance counterparts reported in the literature.^{16,17,32,48–58} In (b), the datapoint marked with * has an actual value of zero.

$$E_{\text{filler}} = E_{\text{applied}} \left(\frac{(1+R)\epsilon_{\text{binder}}}{R\epsilon_{\text{binder}} + \epsilon_{\text{filler}}} \right) \quad (4)$$

This reason was supported by the comparison between KNBNNO-C and other composite samples where the ratio of composite d_{33} to ceramic d_{33} was the highest for KNBNNO-C because of the least disparate permittivity values between the KNBNNO filler and PTMA-CdCl₃ binder (Fig. S9a†) and thus the smallest influence on d_{33} by the composite effect.

(2) Lack of piezoelectricity in the PTMA-CdCl₃ binder⁹ formed an inactive barrier between the filler particles, reduced the interconnectivity of short-range piezoelectric response, and consequently suppressed long-range piezoelectric response.

(3) As can be seen in Table 1, the composite samples had lower relative densities (85–93%) compared to their ceramic counterparts (91–97%), indicating a higher porosity in the composites. Higher porosity is a characteristic of pressure-assisted densification processes made between two elastically dissimilar materials.^{15,16}

Despite inferior d_{33} , the composite samples showed piezoelectric voltage coefficient (g_{33}) values comparable to, or even larger than, those of corresponding ceramic samples, as is listed in Table 2. Variation of the g_{33} followed a similar trend to that for the d_{33} among the composite samples, as g_{33} is dependent on d_{33} and ϵ_r (calculated as $g_{33} = d_{33}/\epsilon_r\epsilon_0$ where ϵ_0 is permittivity in a vacuum). In this regard, S-PZT-C showed the optimum g_{33} value of ≈ 36 mV m N⁻¹ as a result of its largest d_{33} and moderate permittivity among the composites. S-PZT-C also showed a larger g_{33} than S-PZT-P (≈ 25 mV m N⁻¹). Such an increase of g_{33} benefited from the composite effect which significantly lowered the material's permittivity. A similar mechanism has also been employed by research works on porous piezoelectric materials.⁴⁴

Furthermore, the true value of using the oxide-halide perovskite upside-down composite method to potentially

recycle piezoceramics comes when considering the energy budget of fabrication (designated as ' Φ ' in Fig. 5b and Table 2). The Φ value was calculated based on the product of peak temperature and hold time implemented in each type of fabrication process to produce the desired functionality, as has been reported in previous work.⁷ For instance, in the solid-state route, the product of calcination temperature and time and that of sintering temperature and time were summed to provide a measure of energy budget needed for producing a piezoceramic sample that could exhibit desired piezoelectric properties. This parameter, Φ , is assigned the unit $\times 10^3$ °C h.

It is obvious from Fig. 5b that, if sourcing the filler materials from waste piezoceramics, whilst being able to induce similar or larger g_{33} , the upside-down composite fabrication method enabled a considerably lower Φ value, which was only about 1% of that to be used for fabrication of new piezoceramics.

Therefore, both the lower Φ and larger g_{33} in S-PZT-C compared to its own piezoceramic counterpart as well as to other high-performance piezoceramics (Fig. 5b and Table 2) sufficiently demonstrated the feasibility of the oxide-halide upside-down composite and recyclability of piezoceramics *via* such a method. It is also noted that in addition to S-PZT-C, H-PZT-C also showed a comparable g_{33} (≈ 22 mV m N⁻¹) to those of S-PZT-P and H-PZT-P but with a negligible Φ . This implies that deployed piezoceramics like S-PZT and H-PZT in aged/worn-out devices could be potentially recycled and be made for a second life in applications demanding high g_{33} , such as sensors, but at a much lower energy price. This would eventually contribute to sustainable production and use of toxic and hazardous Pb-based piezoelectric materials by avoiding, or at least postponing, the disposal of old PZT-based components that have a negative environmental contribution.

Recyclability using the oxide-halide perovskite upside-down composite method could also be proven with Pb-free piezoelectric compositions, with BT-C and KNBNNO-C showing g_{33}

Table 2 Summary of piezoelectric properties and energy budget of fabrication for Pb-free and Pb-based piezoelectric ceramics and composites which were fabricated in this work and selected from the literature

Sample ID/material	Densification method	Maximum processing temperature (°C)	g_{33} (mV m N ⁻¹)	FOM (×10 ⁻¹² m ² N ⁻¹)	Φ (×10 ³ °C h)	Reference
BT-P	Solid-state sintering	1450	26.7 ± 0.2	7.9 ± 0.1	10.2	This work
BT-C	Upside-down composite	150	19.2 ± 1.7	0.9 ± 0.2	0.1	This work
BT	Reactive hydrothermal liquid-phase densification	240	9.0 ^a	0.8	17.3	53
BT	Sintering with LiF as sintering aid	1100	10.7 ^a	2.9	6.6 ^b	54
BCZT	Sol-gel followed by solid-state sintering	1450	29.0	18.5	5.7	55
BCZT	Sintering with CeO ₂ as sintering aid	1350	14.0 ^a	8.4	7.9	56
KNBNNO-P	Solid-state sintering	1150	11.7 ± 0.2	1.0 ± 0.0	5.3	This work
KNBNNO-C	Upside-down composite	150	12.9 ± 1.0	0.3 ± 0.0	0.1	This work
(K _{0.5} Na _{0.5})NbO ₃	Cold sintering-assisted sintering	1115	19.7 ^a	2.6	7.9	57
(K _{0.5} Na _{0.5})NbO ₃	Solid-state sintering	1115	33.0 ^a	3.9	7.9	57
(K _{0.5} Na _{0.5})NbO ₃	Cold sintering followed by annealing	850	20.2	3.0	5.6	58
0.98(K _{0.5} Na _{0.5})NbO ₃ –0.02Ba(Ni _{0.5} Nb _{0.5})O ₃ –Δ	Solid-state sintering	1165	11.9 ± 0.8	1.2 ± 0.2	5.7	32
0.965(K _{0.48} Na _{0.52})(Nb _{0.95} Sb _{0.05})O ₃ –	Solid-state sintering	1090	23.7 ^a	12.5	8.4	48
0.035Bi _{0.5} (Na _{0.82} K _{0.18}) _{0.5} HfO ₃	Solid-state sintering	1085	25.2 ^a	12.3	8.4	49
0.96(K _{0.48} Na _{0.52})(Nb _{0.95} Sb _{0.05})O ₃ –	Solid-state sintering	1085	25.2 ^a	12.3	8.4	49
0.04Bi _{0.5} (Na _{0.82} K _{0.18}) _{0.5} ZrO ₃	Sintering with ZnO as sintering aid	1200	18.2 ^a	6.4	8.4	50
0.3 wt% ZnO-doped (K _{0.5} Na _{0.5})NbO ₃ –0.02(Bi _{0.5} Li _{0.5})TiO ₃ –0.06BaZrO ₃	Sintering with ZnO as sintering aid	1200	18.2 ^a	6.4	8.4	50
H-PZT-P	Solid-state sintering	1100	24.3 ± 0.2	4.8 ± 0.1	8.0 ^b	This work
H-PZT-C	Upside-down composite	150	21.8 ± 0.9	0.9 ± 0.1	0.1	This work
S-PZT-P	Solid-state sintering	1200	24.3 ± 1.3	12.4 ± 1.4	6.0 ^b	This work
S-PZT-C	Upside-down composite	150	36.0 ± 2.4	3.5 ± 0.5	0.1	This work
S-PZT-LMO	Upside-down composite	22	32.7	2.7	0.0	16
S-PZT-(am-TiO ₂)	Upside-down composite	350	51.7	7.8	0.4 ^b	17
PZT-5A	Cold sintering followed by annealing	900	16.8 ^a	3.3	28.7	52
2.5 mol% Sm-doped 0.7PMN–0.3PT	Solid state sintering	1250	15.3 ^a	16.7	11.9	51

^a Details on whether measured permittivity was from an unpoled or poled sample were not provided in the reference for g_{33} calculation. ^b Details for calculation of Φ were not fully provided in the cited study and a close substitute was used instead.

values of ≈ 19 mV m N⁻¹ and ≈ 13 mV m N⁻¹, respectively. Although these g_{33} values were not as competitive as those of their Pb-based counterparts, the significantly low Φ still provides an incentive to recycle Pb-free piezoceramics which had already offered potential environmental benefits by replacing Pb. On top of the environmental benefits, recycling Pb-free piezoceramics would offer additional benefits of saving on energy and material resource for our planet.

Table 2 also compares the piezoelectric energy harvesting figure of merit ($\text{FOM} = d_{33} \times g_{33}$), which is a universal indicator of energy conversion capability in kinetic energy harvesting applications.³ A FOM value of $\approx 3.5 \times 10^{-12}$ m² N⁻¹ put the recycled sample, S-PZT-C, at the middle level among selected piezoceramics prepared *via* the solid-state high-temperature sintering route. However, if factors such as energy payback

time,⁷ scaling-up capability and integration compatibility with other low-temperature fabricated components are emphasized in practice, the upside-down composite method may make the moderate FOM still attractive in some lower-end energy harvesters that require cost-effectiveness which could tolerate the use of recycled materials with reduced FOM.

4. Critical outlook

In this work, commercial piezoelectric ceramics (purchased from Morgan Technical Ceramics Plc, UK), referred to as O.S-PZT-P, which were older than 20 years, were also truly recycled using the method reported above. Table 3 compares the properties of the O.S-PZT-P with those claimed by the manufacturer in freshly made samples as well as those of S-PZT-P, S-



Material	d_{33} (pC N ⁻¹)	ϵ_r after poling at 1 kHz (—)	g_{33} (mV m N ⁻¹)	Reference
Soft-type PZT ceramic	630	3300 ^a	21	38
S-PZT-P	509	2363	24	This work
O.S-PZT-P	588	2460	27	This work
S-PZT-C	97	304	36	This work
O.S-PZT-C	30	240	14	This work

$$k_{\text{eff}} \approx \frac{f_a^2 - f_r^2}{f_a^2} \quad (5)$$
$$Y_{11}^E = 4\rho_m(f_r^2)(d^2) \quad (6)$$

$$Q_M = \frac{1}{2\pi|Z_m|k_{\text{eff}}C_T f_r} \quad (7)$$

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administration, resources, supervision, validation, writing-review & editing.

Conflicts of interest

This provides a strong motivation for future recycling of piezoceramics from aged and retired piezoelectric devices *via* this method and then giving the recycled materials a second life in sensing applications. The moderate energy harvesting figure of merit obtained from soft-PZT composites may also be suitable for less-demanding energy harvesters made from recycled materials.

Acknowledgements

This research has been funded by Infotech Institute, University of Oulu as a spearhead project. The authors acknowledge the Centre for Material Analysis, University of Oulu for providing necessary equipment and facilities. The authors also acknowledge Dr Jaakko Palosaari and Mr Vasilii Balanov for research assistance.

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Data availability

Data associated with this work are available at <https://doi.org/10.23729/099a0c0b-c416-45c2-832e-27c6bfa1ae15>.

Author contributions

S. S. A. – data curation, formal analysis, investigation, methodology, visualization, writing-original draft, writing-review & editing. M. T. – data curation, methodology. M. N. – investigation, methodology, supervision, writing review & editing. J. P. – investigation, supervision, writing-review & editing. H. J. – conceptualization, funding acquisition, resources. J. J. – funding acquisition, resources. Y. B. – conceptualization, formal analysis, funding acquisition, methodology, project

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