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Introduction to Photocatalytic Materials for Clean Energy, Renewable Chemicals production, and Sustainable Catalysis

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Dr Rajendra Srivastava is a Professor at the Department of Chemistry, Indian Institute of Technology (IIT Ropar), Punjab, India. He joined IIT Ropar as an Assistant Professor, headed the Department from March 2015 to May 2018, and is currently serving as Dean of Postgraduate & Research. He obtained a PhD degree from CSIR-National Chemical Laboratory, Pune, India, where his doctoral work won him the Best Thesis Award of 2006, conferred by the Catalysis Society of India. Afterward, he worked as a postdoctoral research fellow at KAIST, South Korea (2005–2007) and as a JSPS fellow at Hokkaido University, Japan (2007–2009). He received the NASI-SCOPUS Young Scientist Award 2017, The Catalysis Society of India Young Scientist Award 2018, Mid-career Faculty Research and Innovation Award 2019 from IIT Ropar, and Shiv Nath Rai Kohli Memorial Mid-Career Best Scientist Award 2022 from Punjab University, Chandigarh, India. He is a Fellow of the Royal Society of Chemistry, and the Indian Chemical Society. His fields of interest include the design and synthesis of nanostructured functional materials for catalytic applications, including photocatalysis and electrocatalysis. Thermocatalytic and photocatalytic processes are being developed for biomass and plastic waste valorisation, and CO₂ adsorption and conversion. He has published more than 194 peer-reviewed research articles in reputed international journals (Google Scholar: citations = 10 808, h-index = 53, i10-index = 169; Scopus: citations = 9431, h-index = 51), and has been granted nine patents.



The growing population has increased the demand for energy sources and commodity chemicals, which are mostly fulfilled by fossil-fuel sources. The excessive use of fossil fuels pollutes the air and water. The atmospheric CO₂ concentration reached 417.5 ppm in 2022, which is ~50% higher than the CO₂ concentration during the pre-industrial era (mid-1700).¹ The last 40 years have produced approximately 50% of the cumulative CO₂ emitted since 1750, a direct consequence of the growing population and the increasing fossil-fuel consumption for producing energy and chemicals.² According to the Paris Climate Agreement, to limit the global temperature rise to 1.5 °C above that of the pre-industrial era and to avoid temperature-driven catastrophic climatic events, CO₂ emissions must drop to zero by 2050.³ Lowering of the atmospheric CO₂ concentration can be accomplished by adopting three main strategies, namely: (a) reducing atmospheric CO₂ emissions *via* CO₂ capture & storage, and converting CO₂ into value-added chemicals;⁴ (b) renewable chemicals and fuel production employing sustainable energy sources; and (c) environmental remediation employing sustainable energy sources.

Sunlight provides a thousand times more energy to the earth's surface each year than global energy consumption annually, offering a sustainable energy source.⁵ Harnessing the power of sunlight, photocatalytic materials need to be developed to catalyze chemical reactions that hold promise for clean energy production, synthesizing renewable chemicals, and mitigating environmental pollutants. The early research on solar-light-driven technologies, namely photocatalysis (PC) and photoelectrocatalysis (PEC), mainly focused on H₂ fuel generation *via* H₂O splitting and producing valuable organic compounds, such as pharmaceutical building blocks and fine chemicals. However, the recent focus has shifted towards producing hydrocarbon fuels and chemicals from CO₂, intending to decarbonize the earth's atmosphere.⁵ The photocatalytic conversion of CO₂ into methane, methanol, formaldehyde, acetic acid, and higher hydrocarbon fuels and alcohols is attracting growing

interest.^{6–8} The recent focus has shifted towards producing renewable chemicals, fuels and NH₃, and removing water and air pollutants employing light energy, as it offers an alternative strategy to produce value-added chemicals and hydrocarbon fuels currently obtained from non-renewable fossil fuels.

Air pollution remains a pressing global issue due to the continuous release of NO_x from fossil-fuel combustion, contributing to photochemical smog and acid rain, and adversely affecting human health and ecosystems. Semiconductor photocatalysis offers a sustainable solution by harnessing solar energy to convert NO_x into less-harmful compounds under mild conditions;^{9,10} efficient photocatalysts enhance NO removal and selectively convert NO to environmentally benign products like NO₃[−], minimizing secondary emissions. Developing advanced photocatalytic systems tailored for effective NO_x reduction holds promise for mitigating air pollution and advancing environmental sustainability. In addressing water pollution caused by industrial dyes, photocatalytic dye degradation plays a crucial role.^{11,12} When irradiated with light, the photocatalyst generates reactive oxygen species (ROS) such as hydroxyl radicals (•OH) and superoxide anions (O₂^{•−}). These highly reactive ROS oxidize dye molecules, breaking them down into less-harmful compounds like water and carbon dioxide.

Photoelectrochemical water splitting and the photocatalytic hydrogen evolution reaction (HER) are two advanced approaches for sustainable H₂ production from water using sunlight.¹³ PEC utilizes semiconducting materials to generate H₂ and O₂ through efficient optimization of material nanostructures, heterojunction engineering, and surface functionalities. The HER, on the other hand, focuses specifically on using semiconductor photocatalysts to drive the reduction of protons (H⁺) from water to produce hydrogen gas (H₂). This process harnesses the absorption of sunlight by photocatalysts to generate electron–hole pairs. PEC water splitting and photocatalytic HER support the transition to a hydrogen-based economy by offering a renewable, environmentally

friendly alternative to fossil fuels, thus advancing global sustainability efforts.

All of the above-discussed photocatalytic applications for environmental remediation can be realized by synthesizing efficient photocatalytic materials. This can be accomplished by adopting modern synthetic strategies and integrating advanced materials science with photocatalytic principles to unlock unprecedented opportunities to harness solar energy. Innovations in catalyst design, such as creating multi-component systems and incorporating plasmonic materials, have shown significant promise in enhancing photocatalytic performance. One of the primary strategies in photocatalytic synthesis is the development of semiconductor-based materials with tailored nanostructures. By engineering the size, shape, and composition of semiconductor nanoparticles, researchers can significantly improve light absorption, charge separation, and surface reactivity. Modifications such as doping with metal or non-metal elements, creating heterojunctions, and surface functionalization improve photocatalytic performance by extending the light absorption range into the visible spectrum and enhancing charge separation.

This collection of reviews, minireviews and research articles highlights the pivotal role of multifunctional nanomaterials in advancing the efficiency, selectivity, and scalability of photocatalytic processes in harnessing clean energy, producing renewable chemicals, and developing sustainable catalytic routes.

The collection begins with a comprehensive minireview by Wang *et al.* (<https://doi.org/10.1039/D3NA00837A>), on metal–organic framework (MOF)-based photocatalytic materials. MOFs have high crystallinity, large surface area, and variable metal nodes, linkers, and functional ability, which makes them ideally suited for the design and development of photocatalytic processes. The minireview elucidates the design rules and principles for engineering MOFs in manipulating the interfacial charge dynamics in MOFs for enhanced photocatalytic applications.



Enhancing photocatalytic efficiency for NO_x removal focuses on refining materials properties and integrating advanced nanostructures, significantly boosting the degradation rates of nitrogen oxides. These advancements offer more efficient and environmentally friendly solutions for air pollution control. Pham *et al.* (<https://doi.org/10.1039/D4NA00035H>) explore the efficient photocatalytic removal of NO_x pollutants using triangular Ag nanoparticles coupled with TiO₂. This study underscores significant enhancements in photocatalytic efficiency while minimizing toxic byproducts, highlighting the role of

Surface plasmon resonance (SPR)-based heterogeneous photocatalysts have demonstrated significantly enhanced photocatalytic efficiency under

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