

Cite this: *J. Mater. Chem. C*, 2023,
11, 8885

Models of orientational disorder in hybrid organic–inorganic piezoelectric materials

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Hybrid organic–inorganic materials show potential as flexible and Pb-free piezoelectric materials, but ferroelectric-to-paraelectric phase transitions are responsible for a loss of properties at increased temperatures. Here, we develop a model Hamiltonian from first-principles calculations for the archetypical hybrid organic–inorganic piezoelectric TMCMCdCl₃ (TMCM = trimethylchloromethyl ammonium, Me₃NCH₂Cl⁺) to model the order–disorder phase transition. We show that the configurational entropy from molecular orientational disorder is alone unable to quantitatively reproduce the phase transition temperature, but inclusion of vibrational entropy leads to good quantitative agreement with experiments. This highlights the importance of vibrational contributions for the phase stability of soft and flexible materials. We furthermore show that defective structures and intergrowths are readily formed at ambient temperature which is suggested to be responsible for the exceptional piezoelectric response of these materials, which cannot be explained from the conventional small-displacement limit.

Received 26th May 2023,
Accepted 5th June 2023

DOI: 10.1039/d3tc01835k

rsc.li/materials-c

1 Introduction

Piezoelectric materials allow for the interconversion between mechanical and electrical energy and find applications in diverse areas including as sensors, actuators, and high precision motors. Current state-of-the-art piezoelectric materials are based on inorganic ceramics such as lead zirconate titanate (PbZr_{1-x}Ti_xO₃, PZT) with large piezoelectric responses.^{1–3} However, these materials are brittle, require high processing temperatures, and may contain toxic and environmentally hazardous lead, which impose challenges for flexible and bio-compatible technologies.

In the last decade, hybrid organic–inorganic materials have shown promising applications as, *e.g.*, photovoltaics^{4,5} and light emitting diodes,⁶ and more recently, high performance hybrid organic–inorganic ferroelectric and piezoelectric materials have been reported.^{7–12} These materials are flexible, solution-processable, and can be made from non-toxic and environmentally friendly chemistries.

Hybrid organic–inorganic materials are commonly based on an inorganic framework structure hosting organic dipolar cations—often in perovskite or perovskite-inspired structures—and they commonly feature several temperature induced phase transitions with increasing degrees of orientational disorder of the organic cations at increasing temperature.¹³ Elevated temperature may lead to either stabilisation of desired phases such as phases with high dielectric constants for increased screening of charge-carriers in

photovoltaics,^{14,15} or to detrimental disordered paraelectric phases, in which the ferro- and piezoelectric performance is lost.^{9,13} Thus, it is of great importance to understand the influence of molecular orientational disorder and other entropic contributions in hybrid organic–inorganic materials on their stability, phase transitions and physical properties.^{16,17}

In this article, we focus on the hybrid ferroelectric material, TMCMCdCl₃ (TMCM = trimethylchloromethyl ammonium, Me₃NCH₂Cl⁺) with piezoelectric performance comparable to BaTiO₃ in its pure form,⁹ and to PZT for its solid-solution with the fluorine analogue of the molecular cation (TMFM = trimethylfluoromethyl ammonium).¹¹ TMCMCdCl₃ crystallises in the non-centrosymmetric monoclinic space group *Cc* at low temperature with ordered molecular cations, and transitions to a disordered paraelectric phase in the centrosymmetric hexagonal space group *P6₃/mmc* above 400 K (Fig. 1).^{9,11}

Besides the interest in its phase transitions, the mechanism behind the high piezoelectric performance cannot be described within the conventional small-displacement limit, and is thus still poorly understood. Originally, it was suggested that ferroelastic switching was responsible,⁹ but it has been argued that this is not possible when the piezoelectric response is measured with the Berlincourt method.¹⁸ Instead, it was suggested that phase switching between the ground state and a metastable phase is involved.^{18,19}

Through the last decade, several studies have proposed effective model Hamiltonians to study the order–disorder transitions in hybrid organic–inorganic materials, mainly those with cubic perovskite topology. These have been based on classical dipole–dipole interactions and strain models,^{20–22}

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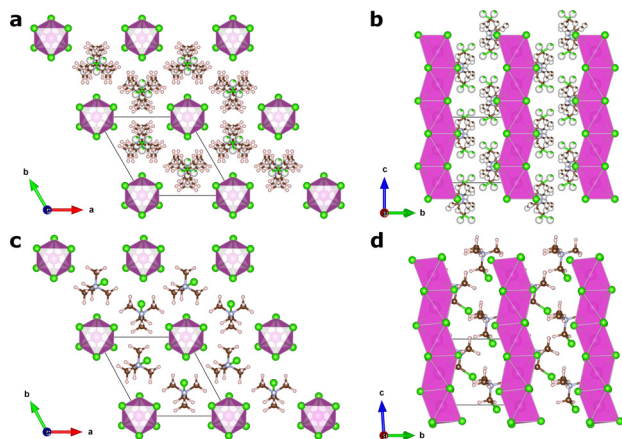


Fig. 1 Crystal structures of TCMCdCl₃. (a) and (b) show the high temperature disordered phase with $P6_3/mmc$ symmetry, and (c) and (d) show the low temperature monoclinic Cc phase, which is responsible for the ferroelectric and piezoelectric properties. Hydrogen atoms are omitted from b for clarity.

Potts- and Ising-like models,^{23,24} or pair interactions.²⁵ However, none of these have been developed solely from first principles and then applied to model a ferroelectric-to-paraelectric phase transition.

Here, we report the construction of a model Hamiltonian based on molecular pair interactions fitted to first principles calculated energies. Applying the model Hamiltonian, we perform Monte Carlo simulations to gain insight into the order–disorder phase transition as well as nanoscale effects and defective phases that are likely to be involved in the piezoelectric performance. Particularly, we show the importance of including contributions from vibrational entropy to model phase transition temperatures in quantitative agreement with experiment. Finally, we show that the easy development of disorder in the ab -plane results in formation of isolated and extended defects, which are likely to be responsible for the exceptional piezoelectric response.

2 Computational methods

2.1 Model construction

A model Hamiltonian for the orientational disorder of TCMC cations was built using a model inspired by the pair mode concept introduced by Li *et al.*²⁵ Using the hexagonal setting of the high temperature phase, the structure of TCMCdCl₃ consists of one-dimensional face-sharing CdCl₆-octahedra along the c -direction with TCMC molecules located around the $2c$ Wyckoff positions. Assuming that the molecular mirror plane is retained in the solid state and coincides with the mirror plane of the local $\bar{6}m2$ symmetry of the Wyckoff site, there are six different molecular orientations per site.

The low temperature phase is of monoclinic Cc symmetry, with its primitive cell coinciding with that of the high temperature phase. The conventional C-centered monoclinic unit cell can be constructed from the three primitive basis vectors $\mathbf{a}_C = \mathbf{a}_P$, $\mathbf{b}_C = \mathbf{a}_P + 2\mathbf{b}_P$ and $\mathbf{c}_C = \mathbf{c}_P$, followed by a monoclinic distortion. Here, we use the primitive representation of the cell throughout.

In order to construct our model Hamiltonian, we include symmetry unique pair interactions corresponding to three different symmetry unique intermolecular vectors, (i) the interactions with the nearest neighbour at the other $2c$ site in the cell, *i.e.* with an intermolecular vector of $\left[\frac{1}{3}, -\frac{1}{3}, \frac{1}{2}\right]$ (and symmetry equivalent directions), (ii) the nearest neighbour along the c -directions at an intermolecular vector of $[0,0,1]$ (and symmetry equivalent directions), and (iii) the next-nearest neighbour in the ab -plane with an intermolecular vector of $[1,0,0]$ (and symmetry equivalent directions). This gives rise to 13, 6 and 12 symmetry unique interactions along the first, second and third intermolecular separations. All three types of interactions are needed in order to distinguish the experimentally observed ground state from specific defective states that we will consider below.

The final energy expression for any supercell size can be written as

$$E = \sum_i n_i J_i, \quad (1)$$

where n_i is the number of interactions of a given symmetry, and J_i is their associated energy. It should be noted that in the final expression, the total number of parameters, J_i , to be determined from fitting to energies calculated from density functional theory (DFT), is not equal to 31, but rather 23, since the parameters do not form a linearly independent basis—the simplest reduction being (considering the up-down orientation along the c -direction) that an equal number of head-to-head and tail-to-tail interactions of the molecules must be present due to the periodic boundary conditions. In practice, 8 parameters are eliminated using QR-decomposition of the original design matrix. This also means that the parameters cannot be interpreted directly as interaction strengths, as they will contain indirect information about other interaction types.

Fig. 2 shows an illustration of how one of the training structures is represented as molecular vectors of the different sites. We represent the molecular cations as vectors with directions along the $[1\bar{1}3]$ -direction and its symmetry equivalents for the first $2c$ Wyckoff site, and the $[\bar{1}13]$ -direction at its equivalents for the second one. This gives directions in agreement with experiment⁹ and previous computations.¹⁸ Here, we will represent the dipoles as unit vectors in the given directions throughout.

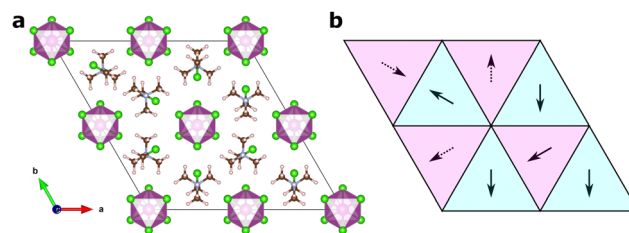


Fig. 2 Illustration of the structural representation in the pair mode concept. (a) Example of an input structure with randomly generated TCMC orientations, and (b) the corresponding representation of the structure with orientation vectors. Full (dotted) arrows note a vector pointing out of (into) the plane, and arrows in purple (turquoise) tiles are located at a z -coordinate of $\frac{1}{4}\left(\frac{3}{4}\right)$.



2.2 Density functional theory calculations

60 supercells of varying sizes are constructed based on random orientations (within the six allowed ones for each site) of TCMC molecules in the framework hexagonal structure. Of the 60 supercells, 20 $2 \times 2 \times 2$, 10 $1 \times 1 \times 2$, 10 $2 \times 2 \times 1$, 10 $3 \times 3 \times 1$, 6 $1 \times 1 \times 3$, and 4 $1 \times 1 \times 1$ cells are constructed, where the 4 $1 \times 1 \times 1$ cells are chosen as the 4 symmetry unique ones possible within that cell size, and the rest are chosen randomly.

All supercells—including cell parameters—are relaxed with DFT to a convergence threshold of 0.01 eV Å⁻¹. The Vienna *ab initio* simulation (VASP) package with the projector augmented wave approximation is used for all simulations.^{26,27} The PBEsol functional is used,²⁸ the plane wave cutoff is set to 700 eV, and a $2 \times 2 \times 2$ Γ -centered k -point grid is used for the base cell, and scaled accordingly for larger supercells.

The pair interaction parameters, J_i , are fitted to the energies extracted from the DFT calculations using linear least-squares.

2.3 Phonon calculations

As we will see below, fitting our model Hamiltonian based only on internal energies from DFT leads to inaccurate predictions of the phase transition temperature in TCMC₃. To get a better description of the energy as a function of temperature, we include the vibrational free energy calculated from harmonic phonon dispersions.

The phonon calculations are performed with PHONOPY²⁹ and forces are calculated with VASP. The structures are relaxed to a slightly stricter convergence criterion of 0.005 eV Å⁻¹, and $2 \times 2 \times 2$ supercells (of the primitive base cell) are used for all phonon calculations. Vibrational free energies given as

$$F_{\text{vib}} = k_B T \sum_{\nu, \mathbf{q}} \ln [2 \sinh(\hbar \omega_{\nu \mathbf{q}} / 2k_B T)], \quad (2)$$

where $\omega_{\nu \mathbf{q}}$ are the phonon frequencies at wave vector \mathbf{q} and band index ν , are then calculated with PHONOPY.

Due to the computational cost of calculating phonon dispersions for supercells, we opt to calculate them only for a subset of the structures. We then construct a surrogate model based only on the internal energy, E_0 , and equilibrium volume, V_0 relative to the ground state structure, to reproduce the relative free energy as

$$\Delta F = \alpha \Delta E_0 + \beta \Delta V_0 \quad (3)$$

where α and β are fitting parameters determined at each temperature of interest. The surrogate model reproduces the relative free energies quite well, especially at temperatures up to ~ 600 K as seen in Fig. 4b.

2.4 Monte Carlo simulations

Monte Carlo (MC) simulations were performed with the Metropolis algorithm³⁰ using the model Hamiltonians described above, *i.e.* both the one based only on internal energies calculated with DFT, and the one based on free energies. In the second case, the temperature of the Monte Carlo simulation was matched with the free energy from phonon calculations.

A simulation box of $8 \times 8 \times 8$ unit cells (with two molecules per unit cell) was used for all simulations, and 2×10^6 MC steps were performed with the first 1×10^5 steps discarded as equilibration. Each simulation was repeated 10 times to improve the statistical sampling. In all cases, the simulations were started from the ground state distribution of molecular orientations, as random (and high-temperature initialised) starting configurations always led to trapping in metastable states.

3 Results & discussions

3.1 Model quality

Fig. 3 shows the agreement between energies from the models based on the pair mode concept and the energies calculated from DFT. For the internal energies from DFT an RMSE of 4.76 meV f.u.⁻¹ is obtained. Correcting the DFT energies for vibrational entropy results in a lower spread in energies and a smaller RMSE of only 3.79 meV f.u.⁻¹ at 300 K. The models coarse grain over ionic relaxations *etc.*, and considering the simplicity and short range nature of the model, the agreement is good. Note however, that for both athermal and thermal energies, the ground state energy (lowest energy point in Fig. 3) is predicted to be larger using the model Hamiltonian than what is obtained with DFT. We may attribute this to lack of long-range terms in the model Hamiltonian, which tend to stabilise specific local environments, but less so long range interactions. This includes strain contributions, which are likely to be underestimated by the model.

3.2 Vibrational contributions

As we shall see below, simulations based on internal energies from DFT tend to overestimate the stability of the ground state structure. However, besides configurational entropy that we will model below with our MC model, vibrational entropy will also contribute to the energetics at elevated temperatures. Combining configurational and vibrational entropic contributions has also previously been shown to be important for accurate predictions of alloy stability.³¹

In Fig. 4a, the phonon dispersion and density-of-states of the ground state structure is shown. The vibrational degrees of freedom give rise to vibrational entropy, or more conveniently expressed as vibrational free energy as in eqn (2). Fig. 4b shows

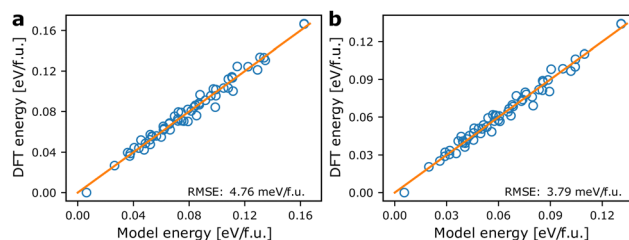


Fig. 3 Comparison of model energies based on eqn (1) and DFT energies. (a) Model energies and internal energies from DFT, (b) model energies and temperature dependent energies corrected using eqn (3) at 300 K. The orange lines are of unity slope and indicate perfect agreement between model and calculated energies.



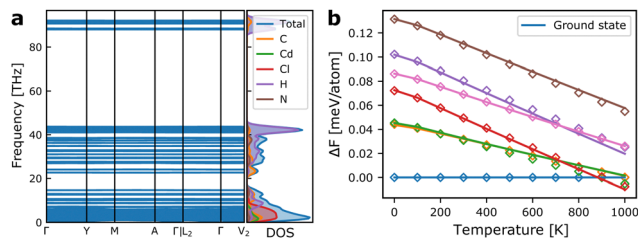


Fig. 4 (a) Phonon dispersion and vibrational density of states for the ground state monoclinic structure of TMCMCdCl₃. (b) Vibrational free energies of a set of metastable structures relative to that of the ground state. The solid lines indicate the calculated free energies from the phonon dispersions, and diamonds show the value from the correction model based only on the internal energy and the equilibrium volume of the metastable phases given in eqn (3).

the relative free energy between the ground state and a set of metastable structures that are used in the model construction. It is evident that all metastable structures are stabilised compared to the ground state at increasing temperatures, with one metastable state even becoming the thermodynamic ground state above ~ 900 K. This shows that vibrational entropy contributes significantly to phase transitions in hybrid organic–inorganic materials as it has been suggested in previous reports.^{32–34}

Fitting the model for the relative free energy given in eqn (3) at each temperature in 100 K intervals, gives rise to predicted energies shown in Fig. 4b. As it is computationally expensive to calculate the vibrational free energies for all training structures, we instead use this surrogate model to estimate the relative free energy for all other structures, which in turn is used to fit a new model Hamiltonian at each temperature, with the model quality indicated in Fig. 3b for 300 K.

3.3 Polarisation & local correlations

From the MC simulations, we can extract expectation values of relevant structural parameters as a function of temperature. Of main interest for probing the ferroelectric-to-paraelectric phase transition in TMCMCdCl₃ is the total polarisation of the simulation cell. In Fig. 5, the polarisation as a function of temperature is shown for model Hamiltonians based on DFT internal energies and free energies including vibrational contributions. The polarisation is calculated from the molecular orientations using the (normalised) $[1\bar{1}3]$ and $[\bar{1}13]$ family of directions as stated above for the orientation of the dipole of two different sites, and averaged over the number of molecules.

First, we note that the transition to a paraelectric phase, characterised by zero polarisation, is observed to occur in the 600–900 K range when neglecting effects of vibrational entropy, whereas it is predicted to occur in the range of 400–600 K based on free energies. The results including vibrational contributions are in good agreement with experiment, where the transition is observed close to 400 K.⁹ Thus, the interplay between vibrational and configurational entropy is of great importance for the relative phase stability in hybrid organic–inorganic materials. The gradual transition observed here compared to the sharper transition in experiment is most likely a result of finite size effects of the simulation cell.³⁵

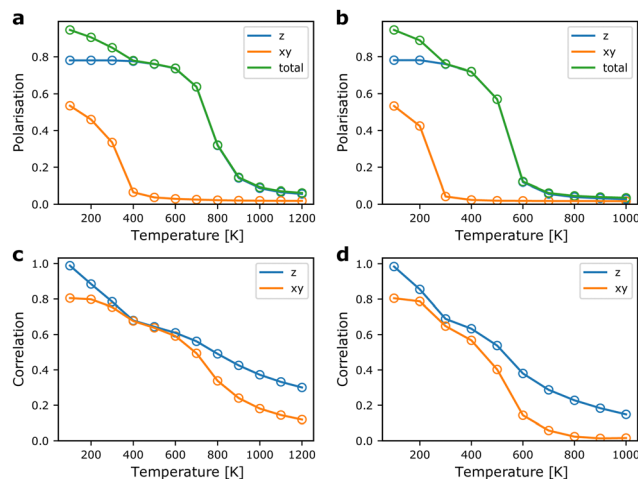


Fig. 5 Polarisation and local correlations as a function of temperature from MC simulations. (a) and (b) show the total polarisation, and (c) and (d) show the local correlations along the *z*-direction and in the *xy*-plane. Polarisation is given in units of the dipole moment of a single molecule and per molecule, while the correlation is given as the dot product between nearest neighbour unit vector dipoles along the *z*-direction and in the *xy*-plane. (a) and (c) show the results for the model based on internal energies from DFT, and (b) and (d) show the results based on the energies corrected for vibrational free energies.

The ferroelectric-to-paraelectric phase transition is observed when considering the *z*-component of the polarisation, whereas the polarisation in the *xy*-plane gradually disappears below room temperature. This is in disagreement with experiment, where only one phase transition is observed in which all components of the polarisation are lost simultaneously.^{9,11} We suggest that this is a result of neglecting long-range interactions in the *xy*-plane. Thus, in our model, it is possible to form local regions where the favourable relative local orientations are retained, but at relatively little energetic cost, defect-centers and grain boundaries can be formed. Put in another way, the strain energy arising from lattice mismatch of different domains is only included indirectly as an energetic penalty of a local relative orientation, but not from its overall effect on the strain field. The failure to capture the effect of strain and other long-range interactions is a well-known deficiency of local model Hamiltonians such as the cluster expansion technique often employed for alloys and solid solutions.³⁶

Interestingly, the observed behaviour of the model around room temperature is reminiscent of the experimentally observed room temperature phases of the related TMCMCdBr₃,³⁷ and the TMCM_{1-x}TMFM_xCdCl₃ solid-solution at concentrations, *x*, between 0.3 and 0.5,¹¹ as well as the intermediate temperature phase in the TMCMCdBrCl₂ solid solution.³⁸ In all three cases, a structure with *P6₃mc* symmetry is observed, which corresponds to disorder between the three sites with the same *z*-polarisation. This suggests a strong competition between a fully ordered *Cc*-phase stabilised by long-range strain contributions, and an entropically stabilised—but still ferroelectric—*P6₃mc*-phase. For TMCMCdBrCl₂ the *Cc*-phase is observed at lower temperature. Whether the two first mentioned systems also condense to the fully ordered *Cc*-phase at lower temperature is unknown.



To experimentally validate the presence of large concentrations of both isolated and extended defects, accurate X-ray diffraction experiments—possibly modelled with the maximum entropy method⁴²—should be able to reveal significant disorder on the molecular sites even in the low temperature phase,^{43,44} and diffuse scattering experiment may reveal local correlations among these orientations.³⁹ Furthermore, high resolution microscopy techniques such as transmission electron microscopy could lead to direct visualisation of (extended) defects, possibly through mapping of the associated strain fields.^{45,46}

Data availability

The data that supports the findings of this study is openly available at <https://doi.org/10.5281/zenodo.7669567>.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

K. T. is funded by the Independent Research Fund Denmark through the International Postdoctoral grant (0164-00015B). Via our membership of the UK's HEC Materials Chemistry Consortium, which is funded by EPSRC (EP/R029431 and EP/X035859), this work used the ARCHER2 UK National Supercomputing Service (<https://www.archer2.ac.uk>).

References

- 1 B. Jaffe, R. Roth and S. Marzullo, *J. Appl. Phys.*, 1954, **25**, 809–810.
- 2 D. Damjanovic, *Rep. Prog. Phys.*, 1998, **61**, 1267.
- 3 F. Li, D. Lin, Z. Chen, Z. Cheng, J. Wang, C. Li, Z. Xu, Q. Huang, X. Liao and L.-Q. Chen, *et al.*, *Nat. Mater.*, 2018, **17**, 349–354.
- 4 A. Kojima, K. Teshima, Y. Shirai and T. Miyasaka, *J. Am. Chem. Soc.*, 2009, **131**, 6050–6051.
- 5 J. M. Frost, K. T. Butler, F. Brivio, C. H. Hendon, M. Van Schilfgaarde and A. Walsh, *Nano Lett.*, 2014, **14**, 2584–2590.
- 6 H. Cho, S.-H. Jeong, M.-H. Park, Y.-H. Kim, C. Wolf, C.-L. Lee, J. H. Heo, A. Sadhanala, N. Myoung and S. Yoo, *et al.*, *Science*, 2015, **350**, 1222–1225.
- 7 P. Jain, V. Ramachandran, R. J. Clark, H. D. Zhou, B. H. Toby, N. S. Dalal, H. W. Kroto and A. K. Cheetham, *J. Am. Chem. Soc.*, 2009, **131**, 13625–13627.
- 8 W. Zhang and R.-G. Xiong, *Chem. Rev.*, 2012, **112**, 1163–1195.
- 9 Y.-M. You, W.-Q. Liao, D. Zhao, H.-Y. Ye, Y. Zhang, Q. Zhou, X. Niu, J. Wang, P.-F. Li and D.-W. Fu, *et al.*, *Science*, 2017, **357**, 306–309.
- 10 H.-Y. Ye, Y.-Y. Tang, P.-F. Li, W.-Q. Liao, J.-X. Gao, X.-N. Hua, H. Cai, P.-P. Shi, Y.-M. You and R.-G. Xiong, *Science*, 2018, **361**, 151–155.
- 11 W.-Q. Liao, D. Zhao, Y.-Y. Tang, Y. Zhang, P.-F. Li, P.-P. Shi, X.-G. Chen, Y.-M. You and R.-G. Xiong, *Science*, 2019, **363**, 1206–1210.
- 12 Y. Hu, L. You, B. Xu, T. Li, S. A. Morris, Y. Li, Y. Zhang, X. Wang, P. S. Lee and H. J. Fan, *et al.*, *Nat. Mater.*, 2021, **20**, 612–617.
- 13 W. Li, Z. Wang, F. Deschler, S. Gao, R. H. Friend and A. K. Cheetham, *Nat. Rev. Mater.*, 2017, **2**, 1–18.
- 14 I. Anusca, S. Balciunas, P. Gemeiner, Š. Svirskas, M. Sanlialp, G. Lackner, C. Fettkenhauer, J. Belovickis, V. Samulionis and M. Ivanov, *et al.*, *Adv. Energy Mater.*, 2017, **7**, 1700600.
- 15 J. N. Wilson, J. M. Frost, S. K. Wallace and A. Walsh, *APL Mater.*, 2019, **7**, 010901.
- 16 K. T. Butler, A. Walsh, A. K. Cheetham and G. Kieslich, *Chem. Sci.*, 2016, **7**, 6316–6324.
- 17 K. Tolborg, J. Klarbring, A. M. Ganose and A. Walsh, *Digital Discovery*, 2022, **1**, 586–595.
- 18 P. Ghosh, S. Lisenkov and I. Ponomareva, *Phys. Rev. Lett.*, 2020, **125**, 207601.
- 19 P. Ghosh, J. Doherty, S. Lisenkov and I. Ponomareva, *J. Phys. Chem. C*, 2021, **125**, 16296–16303.
- 20 J. M. Frost, K. T. Butler and A. Walsh, *APL Mater.*, 2014, **2**, 081506.
- 21 A. Leguy, J. M. Frost, A. P. McMahon, V. G. Sakai, W. Kockelmann, C. Law, X. Li, F. Foglia, A. Walsh and B. C. O'regan, *et al.*, *Nat. Commun.*, 2015, **6**, 7124.
- 22 D. J. Allen, N. C. Bristowe, A. L. Goodwin and H. H.-M. Yeung, *J. Mater. Chem. C*, 2021, **9**, 2706–2711.
- 23 M. Šimėnas, S. Balčiūnas, M. Maćzka, J. Banyš and E. E. Tornau, *Phys. Chem. Chem. Phys.*, 2016, **18**, 18528–18535.
- 24 M. Šimėnas, A. Ibenskas, A. Stroppa, A. Gagor, M. Maćzka, J. Banyš and E. E. Tornau, *J. Phys. Chem. C*, 2019, **123**, 19912–19919.
- 25 J. Li, J. Järvi and P. Rinke, *Phys. Rev. B*, 2018, **98**, 045201.
- 26 G. Kresse and J. Furthmüller, *Comput. Mater. Sci.*, 1996, **6**, 15–50.
- 27 G. Kresse and D. Joubert, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1999, **59**, 1758.
- 28 J. P. Perdew, A. Ruzsinszky, G. I. Csonka, O. A. Vydrov, G. E. Scuseria, L. A. Constantin, X. Zhou and K. Burke, *Phys. Rev. Lett.*, 2008, **100**, 136406.
- 29 A. Togo and I. Tanaka, *Scr. Mater.*, 2015, **108**, 1–5.
- 30 N. Metropolis, A. W. Rosenbluth, M. N. Rosenbluth, A. H. Teller and E. Teller, *J. Chem. Phys.*, 1953, **21**, 1087–1092.
- 31 A. Van De Walle, *Calphad*, 2009, **33**, 266–278.
- 32 K. T. Butler, K. Svane, G. Kieslich, A. K. Cheetham and A. Walsh, *Phys. Rev. B*, 2016, **94**, 180103.
- 33 W. Wei, W. Li, K. T. Butler, G. Feng, C. J. Howard, M. A. Carpenter, P. Lu, A. Walsh and A. K. Cheetham, *Angew. Chem.*, 2018, **130**, 9070–9074.
- 34 K. Li, Z.-G. Li, J. Xu, Y. Qin, W. Li, A. Stroppa, K. T. Butler, C. J. Howard, M. T. Dove and A. K. Cheetham, *et al.*, *J. Am. Chem. Soc.*, 2022, **144**, 816–823.
- 35 K. Binder, *Ferroelectrics*, 1987, **73**, 43–67.
- 36 D. B. Laks, L. Ferreira, S. Froyen and A. Zunger, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1992, **46**, 12587.



- 37 W.-Q. Liao, Y.-Y. Tang, P.-F. Li, Y.-M. You and R.-G. Xiong, *J. Am. Chem. Soc.*, 2018, **140**, 3975–3980.
- 38 X.-G. Chen, Y.-Y. Tang, H.-P. Lv, X.-J. Song, H. Peng, H. Yu, W.-Q. Liao, Y.-M. You and R.-G. Xiong, *J. Am. Chem. Soc.*, 2023, **145**, 1936–1944.
- 39 D. A. Keen and A. L. Goodwin, *Nature*, 2015, **521**, 303–309.
- 40 J.-S. Park, Z. Li, J. N. Wilson, W.-J. Yin and A. Walsh, *ACS Energy Lett.*, 2020, **5**, 2231–2233.
- 41 S. Macpherson, T. A. Doherty, A. J. Winchester, S. Kosar, D. N. Johnstone, Y.-H. Chiang, K. Galkowski, M. Anaya, K. Frohna and A. N. Iqbal, *et al.*, *Nature*, 2022, **607**, 294–300.
- 42 M. Sakata and M. Sato, *Acta Crystallogr., Sect. A: Found. Crystallogr.*, 1990, **46**, 263–270.
- 43 G. J. Snyder, M. Christensen, E. Nishibori, T. Caillat and B. B. Iversen, *Nat. Mater.*, 2004, **3**, 458–463.
- 44 J. Zhang, N. Roth, K. Tolborg, S. Takahashi, L. Song, M. Bondesgaard, E. Nishibori and B. B. Iversen, *Nat. Commun.*, 2021, **12**, 6709.
- 45 M. U. Rothmann, J. S. Kim, J. Borchert, K. B. Lohmann, C. M. O'Leary, A. A. Sheader, L. Clark, H. J. Snaith, M. B. Johnston and P. D. Nellist, *et al.*, *Science*, 2020, **370**, eabb5940.
- 46 S. D. Stranks, *Matter*, 2021, **4**, 3852–3866.

