

PAPER

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Ensuring site-selectivity in covalent chemical modification of proteins is one of the major challenges in chemical biology and related biomedical disciplines. Most current strategies either utilize the selectivity of proteases, or are based on reactions involving the thiol groups of cysteine residues. We have modified a pair of heterodimeric coiled-coil peptides to enable the selective covalent stabilization of the dimer without using enzymes or cysteine moieties. Fusion of one peptide to the protein of interest, in combination with linking the desired chemical modification to the complementary peptide, facilitates stable, regio-selective attachment of the chemical moiety to the protein, through the formation of the covalently stabilized coiled-coil. This ligation method, which is based on the formation of isopeptide and squaramide bonds, respectively, between the coiled-coil peptides, was successfully used to selectively modify the HIV-1 envelope glycoprotein. Covalent stabilization of the coiled-coil also facilitated truncation of the peptides by one heptad sequence. Furthermore, selective addressing of individual positions of the peptides enabled the generation of mutually selective coiled-coils. The established method, termed Bind&Bite, can be expected to be beneficial for a range of biotechnological and biomedical applications, in which chemical moieties need to be stably attached to proteins in a site-selective fashion.

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Introduction

The development of methods for the site-selective chemical modification of proteins represents one of the hallmark accomplishments in protein research and chemical biology.¹ Applications of such chemically modified proteins are highly diverse and include the *in vivo* tracking of fluorophore-labelled proteins,² PEGylation^{3–6} of therapeutic proteins, as well as the covalent immobilization of proteins on surfaces.⁷ Ideally, the modification reaction should proceed rapidly, completely, and in bio-compatible media. Furthermore, the starting material should be stable and easily accessible through chemical, biochemical or biosynthetic methods that are feasible in standard laboratories.

Most current strategies for chemical protein modification achieve site-selectivity by utilizing either the unique chemical properties of selectively introduced cysteine and lysine residues,

respectively, or the substrate-selectivity of proteases. These methods have been reviewed expertly and extensively.^{8–11} A complementary method of chemo-selective ligation is based on pairs of peptide adapter domains, which selectively recognize each other, forming heterodimers. Linking these peptide adapters to the protein and the desired chemical moiety, respectively, in conjunction with covalent stabilization of the peptide heterodimer, enables a perfectly site-selective chemical protein modification. Among these peptide adapter domains, heterodimeric α -helical coiled coils are particularly appropriate for chemical ligation reactions, as their sequences are fairly short, *i.e.* less than 50 amino acids, and they bind to each other with high affinity and selectivity.^{12,13} The sequences of coiled-coil peptides are typically presented as three to five heptad repeat fragments, *i.e.* they are 21 to 35 amino acids long.¹⁴ Their interaction is driven by the formation of an intermolecular hydrophobic core involving leucine and isoleucine residues, and stabilized by salt bridges between positively (arginine and lysine) and negatively (glutamate and aspartate) charged residues.^{15,16} Covalent stabilization of coiled-coils has thus far been achieved through sophisticated reactions involving additional cysteine residues flanking the coiled-coil sequences,¹⁷ in conjunction with the use of a hetero-bivalent cross-linker¹⁸ and native chemical ligation

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(NCL) using coiled-coils as tag-carriers,^{19–21} respectively. Here, we set out to explore the feasibility of covalently stabilizing coiled-coils through isopeptide bonds bridging the side chains of glutamate and lysine residues. This setup eliminates the need to introduce cysteine residues, which typically bear the risk of engaging in disulfide shuffling with cysteine residues in the protein, as well as other side reactions.^{22,23} The principal challenge here was to maintain the selectivity of the interaction, as glutamate and lysine residues are present not only in coiled-coil peptides, but also abundant in proteins, which may give rise to covalent peptide homodimerization, -cyclization or unselective cross-linking with the protein. In order to facilitate chemical synthesis and minimize potential immunogenicity, we selected for this study a pair of relatively short (21 amino acids) coiled-coil peptides, which had been reported to assemble into a parallel heterodimer with high affinity and selectivity.^{16,24}

Results and discussion

Bind&Bite: covalent stabilization of a coiled-coil heterodimer

The coiled-coil selected for this study is formed by interaction of an acidic peptide (PepE, net charge -3) with a complementary basic peptide (PepK, net charge $+3$) (Fig. 1A). As the covalent stabilization of the peptide heterodimer was intended to be accomplished through the formation of isopeptide bonds between glutamate residues in PepE and lysine residues in PepK, the lysine residues in PepE were replaced with arginine, in order to prevent intramolecular acylation by activated glutamates.^{18,25} The guanidine group of arginine is expected to be largely protonated at the relevant pH range (7–8) used for the reaction, impeding acylation by an activated carboxylic acid.

The interaction of the two coiled-coil peptides was assessed in a binding assay based on fluorescence polarization (Flupol) measurement, for which PepE was fluoresceinylated (Fluo-PepE), and PepK was biotinylated (Bio-PepK, see Table 1 for peptide sequences). Binding of Bio-PepK to streptavidin generated a far larger ligand molecule (Bio-PepK-StA),²⁶ which in initial experiments served as a surrogate for PepK-fusion proteins in later applications of the method. After validating the assay by reproducing the reported affinity of the peptide heterodimer (Fig. 1B), as well as the selectivity, *i.e.* the lack of homodimerization (Fig. 1C), the heterodimer was covalently stabilized by pseudopeptide bonds. The formation of this covalent crosslink was enabled *via* activation of the glutamate carboxylic acids by addition of *N*-(3-dimethylaminopropyl)-*N'*-ethylcarbodiimide hydrochloride (EDC) and *N*-hydroxysuccinimide (NHS) to the Fluo-PepE solution, prior to incubation with Bio-PepK-StA. The respective binding curves (with and without EDC/NHS) show a high similarity (Fig. 1D), indicating that pre-activation of Fluo-PepE does not adversely affect its interaction with Bio-PepK-StA. On the other hand, while the non-activated interaction was completely abrogated by addition of the denaturant guanidinium hydrochloride (GdnHCl), the interaction in the presence of EDC/NHS was largely stable at 2 M GdnHCl (Fig. 1E). This fact strongly indicates a covalent stabilization of the heterodimer through isopeptide bonds between the two peptides. Notably, the selectivity of heterodimerization was largely preserved in the presence of EDC/NHS, as homodimerization was detected only at peptide concentrations $> 5 \mu\text{M}$ (100-fold molar excess) (Fig. 2F).

In order to achieve complete activation of glutamates in PepE, the required EDC/NHS concentration resulted in a very large molar excess over the peptides, which, in ligation reactions

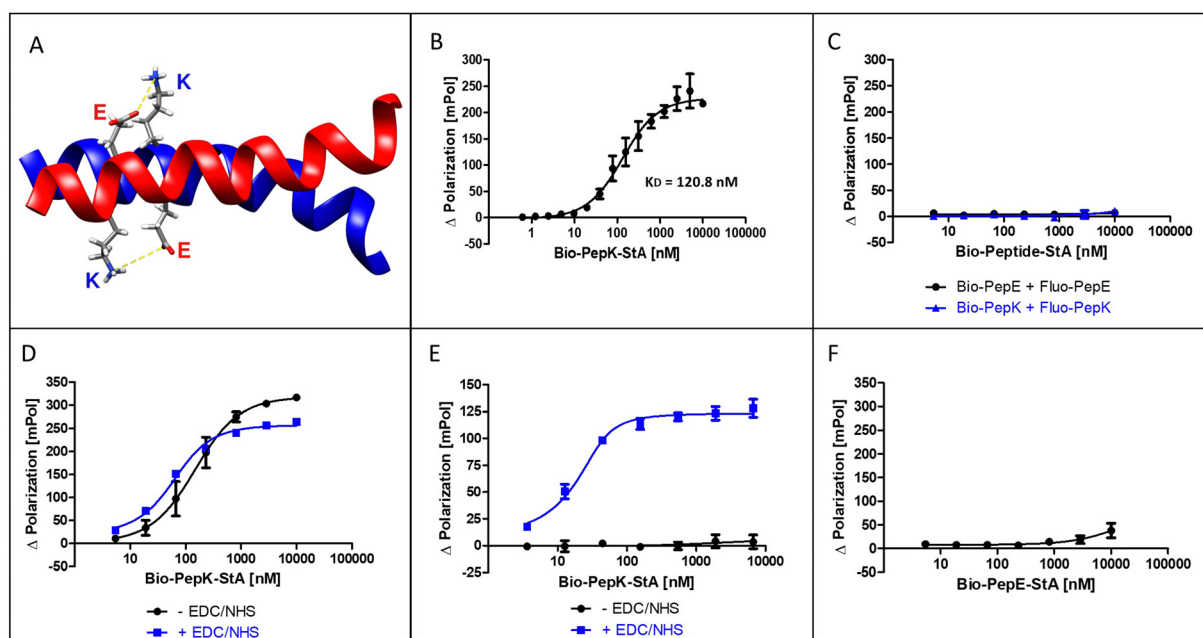


Fig. 1 The Bind&Bite reaction. (A) Structure of the parallel, heterodimeric coiled-coil (pdb code 1u0i). (B) Formation of the non-covalent heterodimer. (C) Lack of non-covalent homodimer formation. (D) Activation of Fluo-PepE does not affect the formation of the heterodimer. (E) Covalent heterodimer is stable in GdnHCl. (F) Lack of covalent homodimer formation. See Experimental section for details.



Table 1 Coiled-coil peptide variants

Peptide	Sequence
Bio-PepK	Bio ^a -Aoa ^b -KIAALKEKIAALKEKIAALKE-NH ₂
Bio-PepE	Bio-Aoa-EIAALEKEIAALEKEIAALEK-NH ₂
Fluo-PepE	Fluo ^c -Aoa-EIAALEREIAALEREIAALER-NH ₂
Fluo-PepK	Fluo-Aoa-RIAALRERIAALRERIAALRE-NH ₂
PepE-Bio	Ac-EIAALEREIAALEREIAALER-Aoa-Aoa-Lys(Bio)-NH ₂
PepK(K13)	Bio-Aoa-RIAALRERIAALKERIAALRE-NH ₂
PepK(K6)	Bio-Aoa-RIAALKERIAALRERIAALRE-NH ₂
PepE(-1C)	Fluo-Aoa-EIAALEREIAALEREIAALE-NH ₂
PepE(-2C)	Fluo-Aoa-EIAALEREIAALEREIAAL-NH ₂
PepE(-3C)	Fluo-Aoa-EIAALEREIAALEREIAA-NH ₂
PepE(-4C)	Fluo-Aoa-EIAALEREIAALEREIA-NH ₂
PepE(-5C)	Fluo-Aoa-EIAALEREIAALEREI-NH ₂
PepE(-6C)	Fluo-Aoa-EIAALEREIAALERE-NH ₂
PepE(-7C)	Fluo-Aoa-EIAALEREIAALER-NH ₂
PepE(ARS8)	NBD ^d -Sar ^e -Aoa-EIAALER-Dap ^f (EDCB ^g)-IAALEREIAALER-NH ₂
PepE(ARS15)	NBD-Sar-Aoa-EIAALEREIAALER-Dap(EDCB)-IAALER-NH ₂

^a Bio, biotin. ^b Aoa, 8-amino-3,6-dioxo-octanoic acid. ^c Fluo, fluorescein. ^d NBD, 7-nitrobenzo-2-oxa-1,3-diazol-4-yl. ^e Sar, sarcosine. ^f Dap, 2,3-diaminopropionic acid. ^g EDCB, 2-ethoxy-3,4-dioxocyclobut-1-en-1-yl.

involving PepK fused to a protein, may give rise to activation of glutamate residues in the protein, resulting in undesired intramolecular cross-linking of the protein. Therefore, we assessed the feasibility of purifying the pre-activated Fluo-PepE NHS ester by preparative HPLC. Interestingly, the Fluo-PepE NHS ester turned out to be sufficiently stable, as long as it was kept in acidic solution (pH < 4). The storage stability was demonstrated by the interaction with Bio-PepK-StA, which proceeded similar to the reaction of *in-situ* activated Fluo-PepE (Fig. 2A). It should be noted that the reaction of Fluo-PepE with EDC/NHS does not yield a completely activated peptide in which all six glutamates are present as NHS esters. In fact, the pre-activated Fluo-PepE is present as a mixture of peptides containing up to three NHS esters, or none at all, which could be separated and detected by LC-MS analysis (Fig. 2B). This heterogeneity, however, does not appear to affect the reactivity of the peptide (Fig. 2A), as even one NHS ester, yielding a single isopeptide bond with PepK, is

sufficient to covalently stabilize the coiled-coil. In the mass spectrum of the product of the reaction of pre-activated PepE-Bio with Bio-PepK, we identified the masses of the ligation products containing one and two isopeptide bonds, respectively (Fig. S3, ESI[†]). The remaining free glutamate carboxylates in the activated PepE may even facilitate the selective interaction with PepK, as they are still available for salt bridges with the complementary lysine residues in PepK. Furthermore, the pre-activated Fluo-PepE NHS ester was shown to be stable at acidic pH at -20 °C for at least one week (Fig. 2B), illustrating the robustness of the developed method.

Truncation of coiled-coil peptides

Although the sequences of the coiled-coil peptides used for this study are relatively short, *i.e.* 21 amino acids, they are still long enough to be potentially immunogenic, in particular when linked to a protein. Therefore, we set out to explore the scope

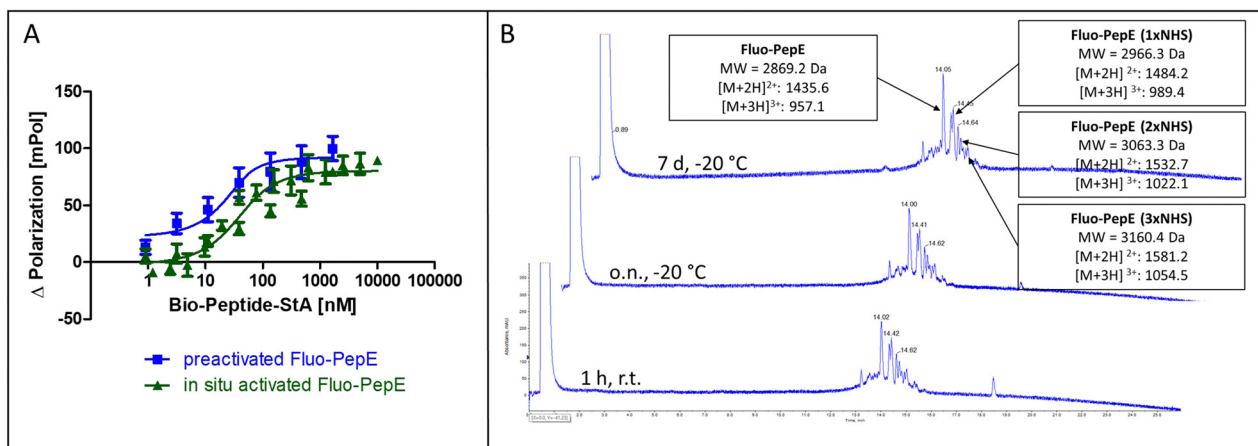


Fig. 2 Reactivity and stability of pre-activated, purified PepE. (A) Covalent interaction of *in situ* activated Fluo-PepE and pre-activated, purified Fluo-PepE with Bio-PepK-StA (Flupol binding assay, binding curves after treatment with GdnHCl). (B) LC-MS chromatogram of pre-activated, purified Fluo-PepE in acidic solution (0.1% TFA in 50% water/acetonitrile) after 1 hour at room temperature (bottom), as well as after overnight (center) and seven days (top) storage at -20 °C. See Experimental section for details.

of possible sequence truncation, by evaluating the behaviour of PepE variants upon stepwise, C-terminal truncation. These shortened variants were tested against full-length PepK. Interestingly, PepE tolerates C-terminal truncation by one or two amino acids (PepE(-1C) and PepE(-2C)) in the non-covalent interaction with full-length PepK, as the affinities of these interactions are very similar to full-length PepE (Fig. 3A). Deletion of up to three more amino acids (PepE(-3C), PepE(-4C) and PepE(-5C)), on the other hand, causes a drop in affinity by approximately 40-fold. Further truncation (PepE(-6C) and PepE(-7C)) resulted in complete loss of binding, which, however, could be counteracted by covalently stabilizing the coiled-coils using the Bind&Bite method (Fig. 3B). Using this method, PepE truncated by up to seven residues (PepE(-7C)), *i.e.* a complete heptad, still binds to full-length PepK at submicromolar concentrations. In conclusion, the Bind&Bite method to covalently stabilize the coiled-coils, facilitated truncation of the PepE sequence by one heptad, which was not possible for the non-covalent coiled-coil.

Chemical modification of HIV-1 envelope protein

The HIV-1 envelope protein (Env) presents a perfect target for chemical modification, as such modifications can be used to anchor the protein to liposome membranes or other carriers for the generation of vaccines.^{27–29} The limited scope of strategies for the covalent immobilization/modification of this viral protein poses a significant bottleneck in the generation of such vaccines. To circumvent this limitation in the context of HIV-1, trimeric BG505 NFL2P HIV-1 envelope protein (Env) with a C-terminal His tag³⁰ was used as a model protein to test the Bind&Bite method. The C-terminal His tag of Env was replaced by two additional G4S linkers and the PepK sequence, in order to ensure sufficient flexibility of the coiled-coil peptide, correct presentation, as well as to maintain correct folding of Env into a homotrimer. The fusion protein Env-PepK was expressed in HEK 293-FT cells. Using the Flupol binding assay, it could be shown that the Bind&Bite reaction was well preserved in the interaction of Fluo-PepE with the Env-PepK fusion

protein (Fig. 4A), demonstrating that PepK is sterically available for interaction with PepE also when fused to a protein.

Addressing the question whether addition of the PepK tag to the Env sequence, and the subsequent reaction with activated PepE, respectively, affects the functional integrity of the protein, the interaction of the modified proteins (Env-PepK and Env-PepK + PepE-Bio) with mAb PG9 was compared to wt protein (Env(wt)). PG9 is an antibody that selectively recognizes a conformational epitope formed by gp120 subunits of the functional Env trimer.³¹ As shown in Fig. 4B, all three Env variants were recognized by PG9, clearly demonstrating preservation of the conformation of Env-PepK prior to, as well as after reaction with pre-activated PepE-Bio. As a negative control, monomeric gp120 was not recognized by PG9.

The selectivity of this interaction of Env-PepK with activated PepE was then probed in a more complex system. Cell culture medium (DMEM) containing 10% fetal calf serum (FCS) was spiked with purified Env-PepK and Env(wt), respectively, at 71 nM. Upon addition of *in situ* activated (Fig. 4C) and pre-activated (Fig. 4D) PepE-Bio, respectively, no reaction of PepE-Bio, regardless of the mode of activation (*in situ* or pre-activated), with Env(wt) could be detected, demonstrating a selective reaction of activated PepE with lysine residues in the PepK tag of Env-PepK (green marks). Upon incubation of *in situ* activated PepE-Bio with Env(wt), though, considerable off-target reaction was evident by the presence of additional bands that were detected with streptavidin-HRP (Fig. 4C, blue marks), indicating reaction of activated glutamate residues in PepE-Bio with nucleophiles in other components of the highly complex medium (FCS/DMEM). Such off-target reactions, however, were largely eliminated when using pre-activated, purified PepE-Bio, (Fig. 4D). These results clearly highlight the benefit of using pre-activated, purified PepE for the ligation reaction, as compared to *in situ* activation, in particular for reactions in solution. For the interaction of PepK fusion proteins with PepE immobilized on particles or surfaces, on the other hand, *in situ* activation may still be useful. In that setup, which is potentially useful for the purification of proteins

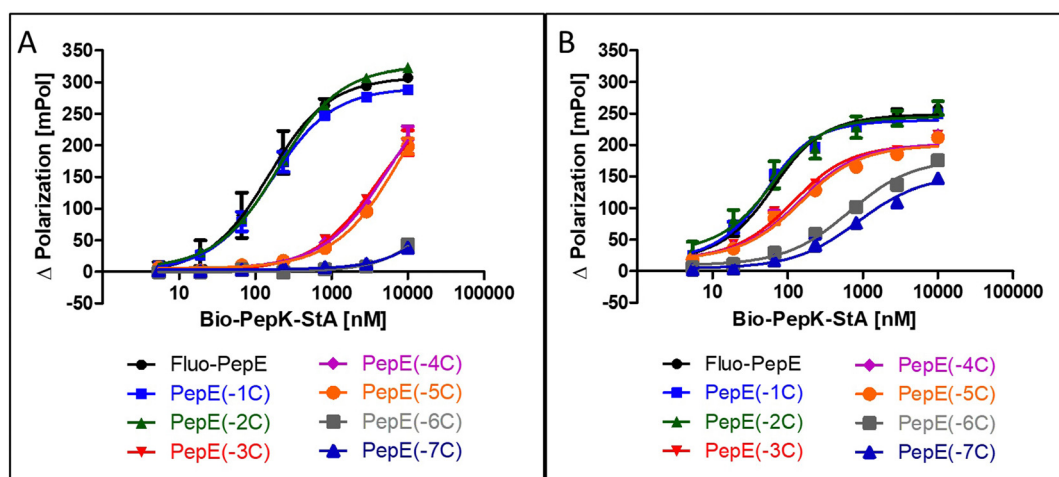
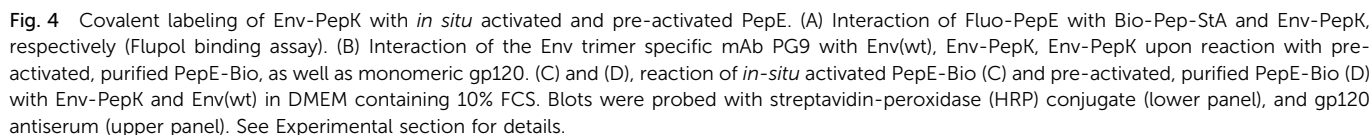


Fig. 3 Effect of C-terminal truncation of Fluo-PepE on its non-covalent (A) and covalent (B) interaction with Bio-PepK-StA (Flupol binding assay). See Experimental section for details.



only the complementary lysine in PepK. The remaining five lysines in PepK were replaced with arginine, which can still engage in ionic interactions with glutamates in PepE, but is not sufficiently reactive for a covalent reaction with the ARS in PepE (see Table 1 for peptide sequences). The ARS was introduced by first coupling β -Alloc-protected diaminopropionic acid (Dap(Alloc)), which, after selective removal of the Alloc group, was alkylated with squaric acid diethyl ester (SADE). The resulting 2-ethoxy-3,4-dioxocyclobut-1-en-1-yl moiety could then react with the single lysine in PepK, generating a squaramide (Fig. 5B). This reaction, however, was expected to proceed only when the single lysine in PepK was placed in the correct position, *i.e.* juxtapositioned to the ARS in PepE. In order to probe this selectivity, PepE with the ARS at position 8 and 15, respectively, was reacted with PepK presenting the single lysine at positions 6 and 13, respectively. Essentially no product could be detected for the reaction of PepE(ARS8) with PepK(K13), in which mismatched positions were modified (Fig. 5A, mismatch 1, Fig. 5D, top and Fig. S2, ESI⁺). In contrast, for the reaction of PepE(ARS8) with PepK(K6), the reactive positions were placed correctly, *i.e.* in close proximity within the coiled-coil. In this case the ligation provided >90% yield after overnight reaction (Fig. 5A, match 1, Fig. 5D, bottom and Fig. S2, ESI⁺). On the other hand, PepK(K13) was able to react with PepE(ARS15) (Fig. 5C, match 2, Fig. 5E, bottom and Fig. S2, ESI⁺), while no product was generated upon combination of PepK(K6) with PepE(ARS15) (Fig. 5C, mismatch 2, Fig. 5E, top and Fig. S2, ESI⁺). Thus, we have generated two pairs of mutually selective pairs of peptides, *i.e.* PepK(K6) and PepE(ARS8), as well as PepK(K13) and PepE(ARS15), based on the same original pair of

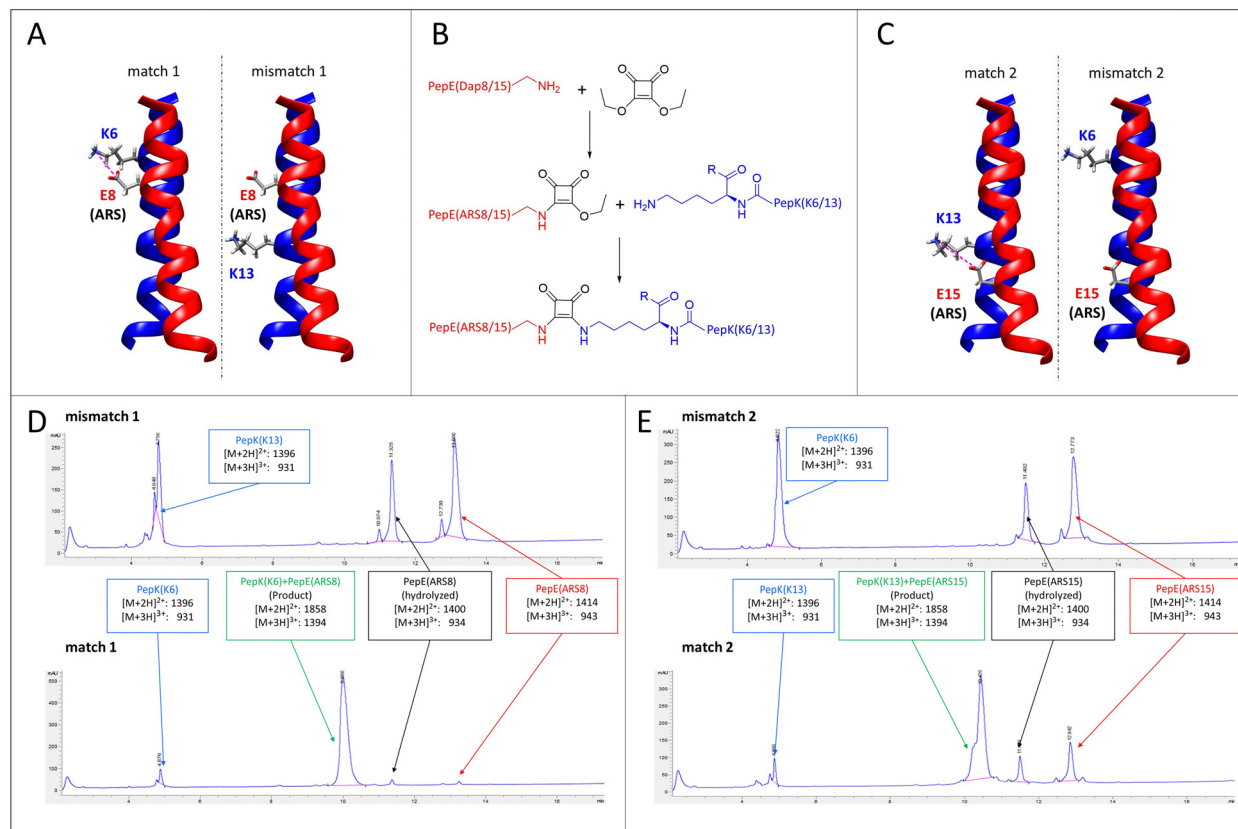


Fig. 5 Mutually selective coiled-coils. (A) and (C), illustration of the location of individual lysine residues in PepK and ARS in PepE in the pairs of peptides (structure: pdb code 1u0i). (B) Scheme of the ligation reaction. (D) and (E), HPLC chromatograms (UV absorbance at 220 nm) of the products of the reaction of PepE(ARS8) with PepK(K13) (mismatch 1, D top) and PepK(K6) (match 1, D bottom), as well as of PepE(ARS15) with PepK(K6) (mismatch 2, E top) and PepK(K13) (match 2, E bottom). The shoulder in the product peak of match 2 corresponds to the peptide containing biotin sulfoxide. See Experimental section for details.

coiled-coil peptides. This excellent selectivity can be expected to enable the simultaneous labelling of two or more proteins with different chemical entities.

It should also be noted that the chemical nature of the ARS is not limited to the squaramide used here, but can be extended to moieties such as isothiocyanates, fluorinated nitroaromates, and aza-Michael acceptors (e.g. sulfonylacrylates). The varying degree of reactivity of these groups enables the fine-tuning of the relationship between speed of reaction and stability (speed of hydrolysis). This flexibility may also enable application of the Bind&Bite method in cells, by using an ARS that is sufficiently stable in biological media, while still being reactive enough to react with the PepK tag attached to proteins.

Experimental

Peptide synthesis

Peptides were synthesized as C-terminal amides by Fmoc/tBu-based solid-phase synthesis using an automated multiple peptide synthesizer (ResPep from INTAVIS Bioanalytical Instruments AG, Köln, Germany), as previously described.³⁴ Protected amino acids were obtained from Iris Biotech GmbH, Marktredwitz, Germany).

After the final amino acid coupling, the N-terminal amino groups were either acetylated with a mixture of acetic anhydride/pyridine/DMF (1 : 2 : 3) for 30 min, biotinylated (3 eq. biotin, DIC and Oxyma pure (R) in NMP), or fluoresceinylated by coupling of 5(6)-carboxyfluorescein (3 eq. 5(6)-carboxyfluorescein, DIC and Oxyma pure (R) in DMF) over night. Peptides were cleaved from the resin by use of trifluoroacetic acid (TFA)/water/phenol/thioanisole/triisopropylsilane 80 : 5 : 5 : 5, precipitated in cold *tert*-butyl methyl ether, extracted with water, and lyophilized. Peptides were purified by preparative HPLC (Phenomenex Kinetex C18 column, 100×21.2 mm, flow rate 30 mL min^{-1} , gradient of acetonitrile in water (both containing 0.1% TFA, 25–60% over 10 min). Stock solutions of purified peptides were prepared at 2.5 mM in 50% acetonitrile in water. Analytical data (LC-MS) of purified peptides are shown in Fig. S1 (ESI[†]).

PepE(ARS8) and PepE(ARS15). At position 8 and 15, respectively, of PepE, Fmoc-Dap(Alloc)-OH was coupled instead of Fmoc-Glu(*t*Bu)-OH. The amino group of the N-terminal sarcosine was arylated by incubating the peptide resin twice for 20 min with NBD-Cl (4-chloro-7-nitrobenzo-2-oxa-1,3-diazol, 10 eq., 250 μmol , 49.9 mg) in 0.5 mL DMF, to which DIPEA (10 equiv. 250 μmol , 45.1 μL) was added. NBD was used as a fluorescent identification tag, which is inert to reaction with the ARS. After washing with DMF,



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