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Solvent-free confinement of ordered microparticle monolayers: effect of host substrate and pattern symmetry†

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The self-organisation of individual suspended colloids into ordered structures that can be mediated by confinement has garnered interest recently. Despite the push for solvent reduction for sustainability reasons, the comprehension and development of solvent-free assembly methods remain largely unaddressed. In this study, we explore the effect of confinement without rigid geometrical constraints, *i.e.*, wall-less confinement on the assembly of monodisperse PMMA powder microspheres (diameters of 3 μm and 10 μm) on fluorocarbon-patterned heterogeneous substrates using a solvent-free rubbing assembly approach. Our findings reveal that the PMMA microspheres self-align on the fluorocarbon patterns, adapting to various geometrical shapes of these patterns through symmetry matching. The assembly process is driven by triboelectric charging and elastic properties of the microspheres and substrates. Moreover, we observe that the host substrate and the particle and pattern size ratio significantly influence the ordering of the microparticles on the fluorocarbon patterns. Ultimately, we demonstrate the successful use of fluorocarbon patterns to assemble tunable crystal patterns on rigid substrates, which typically do not exhibit any ordering.

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1 Introduction

The self-organisation of individual building blocks into organized large collective structures is ubiquitous in nature at different length scales, *e.g.*, molecules, living cells, colloidal particles, granular matter, a flock of animals, pedestrians, robots, and even objects in space.^{1–4} In particular, the controlled assembly and crystallisation of colloids and granular matter into ordered structures is appealing from fundamental perspective,^{5–10} but can also be advantageous for a sustainable society as these structures unlock many industrial applications, *e.g.*, photonics, smart materials and sensors, surface coatings, robust light-weight materials, and miniaturized devices.^{11–14} As such, the literature on colloidal assembly is extensive.^{8,11,14}

Another approach gaining attention within the soft matter community is steering the self-organisation of individual

building blocks through the confinement of different natures, *e.g.*, inside microfluidic chips or simply using an evaporating suspension droplet.^{2,8,15,16} By spatially confining colloids, their translational and rotational degrees of freedom are restricted, altering possible configurations or system order.

Research on colloidal self-assembly phenomena into ordered structures has been primarily devoted to colloidal particles in suspension, using so-called “wet assembly” techniques. However, the impetus for reducing solvent waste towards more sustainable processes elucidates the necessity of comprehending solvent-free assembly, *i.e.*, “dry assembly” methods.¹⁷ To achieve this anticipated transition, a handful of dry assembly methods have been proposed on different length scales (size of building blocks) over the last decades as recently surveyed by Sotthewes and Jimidar,¹⁷ including shaking,^{18–20} acoustic-driven,²¹ vacuum-assisted techniques,^{22,23} and rubbing.^{24–27}

The rubbing method has recently emerged as an advanced solvent-free assembly approach to rapidly (<20 s) attain large areas of ordered crystals, spherical colloids and cubic particles,²⁸ using polydimethylsiloxane (PDMS). Recently, Sotthewes *et al.*²⁶ have reported the assembly of tunable hexagonal closely packed (HCP) arrays driven by triboelectrification and contact mechanics. Triboelectrification or contact electrification is the process of spontaneous electrical charge exchange when two bodies are in frictional contact.^{29,30} The exact mechanism at the onset of triboelectric charging has not been resolved yet, as many factors can play a

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role.^{29,30} A few of the mechanisms include electron transfer,³⁰ material transfer and bond scission in polymers,³¹ mechanochemistry,^{32–34} and water or humidity in ambient air.^{30,35–37} In particular, understanding the role of ambient humidity or water in contact electrification charging/discharging has garnered interest in the context of improving the performance of triboelectric nanogenerators (TENGs) and other green energy harvesting applications.^{34,36}

Besides the tribocharging-driven rubbing assembly proposed by Jimidar and co-workers,^{25,26} the Whiteside group explored the agitation concept to self-assemble millimetre-size polymer beads^{19,20} into closely packed crystals of different symmetry structures using the triboelectrification mechanism. On the other hand, Jimidar *et al.*¹⁸ reported that agitation was not commensurate with the scale to achieve a closely packed monolayer of silica and polystyrene microspherical particles (3–10 μm).

The solvent-free rubbing method offers significant control over the order and positioning of a broad spectrum of particles (size and material), enabling the assembly of ordered crystals on any desired pattern.²⁶ By utilizing a fluorocarbon-patterned (heterogeneous) substrate, microparticles are laterally confined to the patterns, despite the fluorocarbon pads being at least 50 nm higher than the host substrate. This confinement occurs without physical boundaries or rigid walls, referred to as wall-less confinement. The driving mechanism underpinning this confinement is the differences in triboelectric charging and contact mechanics forces between the fluorocarbon coating and the host substrate. However, the effect of the host substrate and size of the fluorocarbon patterns on the ordering of the particles remains unaddressed.

In this study, we leverage the solvent-free rubbing method to investigate the self-alignment of poly(methyl methacrylate) (PMMA) powder into hexagonal closely packed (HCP) monolayer arrays on fluorocarbon-patterned substrates. Using the Voronoi tessellation approach, we quantify the order of the assembled monolayers composed of monodisperse PMMA microspheres (3 and 10 μm) on the wall-less confined fluorocarbon patterns. Their effect on particle orderings is examined by varying the shape and size of the fluorocarbon pattern and the type of host substrate (glass and SiO_2). Our findings show that particles are confined to the CF_x and match the geometrical shape of the pattern, affecting the local ordering of the particles. Additionally, the edge between the fluorocarbon pattern and host substrate plays a crucial role in the order achieved using the rubbing assembly process. The insights gained in this study will impact our understanding of reaching and tuning structural ordering using solvent-free (dry) assembly methods, addressing the need for designing greener assembly processes.

2 Experimental and analysis section

2.1 Materials and methods

The rubbing experiments, illustrated in Fig. 1(A), were performed using monodisperse dry PMMA powder (15 ± 3 mg) with a diameter of (3.04 ± 0.11) μm , and (9.95 ± 0.22) μm that were purchased from microParticles GmbH. The manufacturer provided the standard deviations on the particle diameter. Note that the rubbing assembly (*cf.* Fig. 1(A)) is a dynamic process, and

concomitantly time-dependent. Since only the final result can be microscopically examined, we repeated the process at least 6 times to gain sufficient statistics. All experiments were performed under standard lab conditions ($T = 21 \pm 1$ °C and RH = 40–55%). The temperature and humidity are measured with a Digital Professional Thermo-Hygrometer KLIMA BEE, TFA, Germany.

The substrates are patterned using plasma polymerisation using a 50–75 nm thick fluorocarbon (CF_x with $2 \leq x \leq 3$) coating. The manufacturing process of these patterned substrates is described in detail in our previous studies,^{25,26} to which the interested reader is kindly referred.

Kelvin probe force microscopy (KPFM) measurements are performed using a Bruker icon atomic force microscope (AFM) under ambient conditions with RH = 45–50% (measured with a TFA Digital Professional Thermo-Hygrometer KLIMA BEE). A heavily boron-doped diamond cantilever with a resonance frequency of 100 kHz and a force constant of 8 Nm^{-1} (FM-LC, Adama Innovations) was used. The FM-KPFM mode is utilized to detect the electrostatic gradient through the frequency shift of the cantilever oscillation. During the KPFM measurement, the tip was grounded, allowing the determination of the contact potential difference (V_{CPD}) using the following equation:

$$V_{\text{CPD}} = \frac{\phi_s - \phi_{\text{tip}}}{|e|} \quad (1)$$

where e is the elementary charge and ϕ_s and ϕ_{tip} are the work functions of the sample and tip, respectively.

2.2 Structure analysis of particle monolayer arrays

After rubbing the particles onto the substrates, optical microscopy images were captured using a Leica DM2500 MH microscope connected to a ZWO ASI294MC Pro camera, while in some instances, scanning electron microscopy (SEM) images were taken using a ZEISS Merlin high-resolution scanning electron microscope.

Voronoi diagrams were constructed from the images using MATLAB routines to characterize the structure of the PMMA monolayers. A Voronoi diagram is constructed by using the centres of the microspheres to partition the space into polygonal cells comprising points closer to one microsphere centre than to all the others, as elaborated in Sections S1 and S2 of the ESI.[†]^{38,39} The shape of each obtained Voronoi cell i can be described by computing a dimensionless quantity $\mathcal{G}_i$, called shape factor:

$$\mathcal{G}_i = \frac{p_i^2}{4\pi A_i} \quad (2)$$

where p_i and A_i are the perimeter and area of Voronoi cell i , respectively. From the definition, it can be concluded that $\mathcal{G}_i = 1$ corresponds to a circular cell, whereas $\mathcal{G}_i > 1$ for regular polygons. For a perfect hexagonal crystal, the Voronoi diagrams would result in regular hexagons with $\mathcal{G}_{i,\text{hex}} = 1.1027$.

The orientational order of the obtained monolayers can be quantified by computing the local 6-fold bond-orientation, or hexatic, order parameter ψ for each particle, which is defined as:^{10,40}

$$\psi_i = \psi = \frac{1}{N_j} \sum_{j=1}^{N_j} e^{6i\theta(\mathbf{r}_{ij})} \quad (3)$$





Fig. 1 (A) Schematic representation of the manual rubbing method of dry PMMA powder on a fluorocarbon-patterned substrate using a PDMS stamp. SEM images of 10 μm PMMA microspheres assembled in a closely packed monolayer configuration on a (B) square and (C) circular fluorocarbon pattern on a SiO₂ substrate. (D) and (E) KPFM measurements performed on assembled 500 nm PMMA colloids and the CF_x-coated area. (D) Topographic image and (E) the simultaneously acquired surface potential map of the particles and CF_x layer, scale bar 700 nm. (F) and (G) The Voronoi diagrams correspond to the assembled monolayers on the fluorocarbon square in (B) and circle in (C), respectively. The Voronoi cells are coloured according to the normalized shape factor ϑ_{norm} . (H) The normalized average shape factor. (I) and (J) SEM images of 10 μm PMMA microspheres on an uncoated homogeneous glass wafer. Scale bar: 100 μm (green); 30 μm (red).

where $\theta(\mathbf{r}_{ij})$ is the angle between the vector connecting particles i and its nearest neighbour j and an arbitrary reference axis. The vector $\mathbf{r}_{ij}$ is computed for all its N_j nearest neighbours, identified from the Voronoi constructions. What follows from eqn (4) is that $|\psi| = 1$ for a particle oriented in a hexagonal closely packed crystal, and $|\psi| < 1$ for all particles with neighbours deviating from perfect hexagonal order, *i.e.*, $|\psi| \leq 1$.

One can also compute the spatially averaged orientational order parameter ψ_{av} of an ensemble of N particles as:

$$\psi_{\text{av}} = \frac{1}{N} \sum_{i=1}^N |\psi_i| \quad (4)$$

3 Results and discussion

3.1 Initial observations on rubbing-induced assembly

Dry monodisperse PMMA powder sandwiched between a PDMS stamp and another fluorocarbon-patterned substrate is manually rubbed for approximately 20 seconds in a circular motion across the substrates to assemble a monolayer array of hexagonal closely packed (HCP) crystal structures, as depicted in

Fig. 1(A). We vary the geometrical shape and size of the confined CF_x patterns (height of 50–75 nm) on the heterogeneous host substrates. During the rubbing motion, the operator applies a pressure P , which can be adjusted to assemble a monolayer array depending on two extreme cases: when the applied pressure is too high, the assembled monolayer of particles is disturbed, whereas, at low pressures, the powder is not spreading across the substrate, hindering the assembly process.²⁶ It has been reported that soft-elastomeric substrates, like PDMS, induce a rolling motion on the spherical particles confined between the PDMS stamp and the substrate, which promotes the formation of the closely packed monolayer, *i.e.*, HCP crystal structures.^{24,26}

Initial rubbing experiments were performed on a fluorocarbon-patterned SiO₂ substrate using 10 μm PMMA powder microspheres. It is striking to observe from Fig. 1(B) and (C) that despite the circular rubbing motion, the PMMA microspheres self-aligned into HCP crystal monolayers matching the geometrical shapes, square or circular, of the wall-less confined CF_x patterns. This result indicates that the spherical PMMA powder particles have a strong preference to remain and stick to the fluorocarbon coating rather than the SiO₂ substrate, highlighting that a key aspect for assembling HCP crystals on the substrates is the presence of

sufficient adhesion between the microspheres and underlying substrate. Our team recently reported that the main drivers for sufficient adhesion between the PMMA microspheres and the fluorocarbon coating are the contact mechanics force due to elastic deformations of the particles and the underlying surface and the triboelectric charging phenomenon.²⁶ The latter is induced as the microspheres and substrates get into frictional contact during the rubbing motion.

The triboelectric charging is supported by KPFM measurements performed on the PMMA particles and the fluorocarbon coating. The contact potential difference V_{CPD} on the PMMA particles and the CF_x coating is measured, and it can be observed from Fig. 1(D) and (E) that the particle mainly acquired a positive polarity, whereas the fluorocarbon coating was negatively charged. It should be noted that prior to the rubbing experiments, the PMMA particles were negatively charged.²⁵ This implies that as the particles roll across the substrate, charge is exchanged from the particles to the fluorocarbon coating, which aligns with the triboelectric series.²⁹ Thus, a tribocharging-induced electrostatic attraction exists between the PMMA particles and the coating. It is remarkable that although most of the PMMA particles have the same polarity, a closely packed monolayer is present, as one would expect that Coulombic repulsion between the particles would counter crystal formation. As such, it is elucidated that the electrostatic attraction between the PMMA microspheres and the fluorocarbon coating is sufficiently strong to overcome the repulsive force, enabling the assembly of HCP crystals. We want to point out that humidity effects may also influence the triboelectric charging between the PMMA particles and CF_x coating, particularly due to their difference in hydrophobicity.^{35,37} However, to exclude humidity effects, experiments were performed only under ambient conditions.

As noticed before, the area of the conspicuously present closely packed monolayers matches the geometrical shape of the confined coating patterns exhibiting symmetry structures and defects, e.g., a few excess particles on top, voids and grain boundaries, even though the rubbing direction is performed randomly. These symmetry structures and defects can be identified and quantified using Voronoi decomposition,^{39,41} as showcased in Fig. 1(F) and (G). Since our interest is in hexagonal symmetries, each cell in the Voronoi diagrams is coloured according to its normalized shape factor $\mathcal{S}$, such that a perfect hexagonal cell $\mathcal{S}_{\text{norm,hex}} = 1$. A feature that can be readily observed from Fig. 1(F) and (G) is that seemingly at the edge of the wall-less confined patterns, the symmetry structures in the monolayers were distorted, indicating that interactions of the particles on the edge (interface) between the coated and uncoated part of the host substrate were altered. Furthermore, defects associated with the misalignment of grains, e.g., square symmetry structures between two grains, indicate that grains with different orientations were assembled during the rubbing motion. This implies that several nucleation sites grow into larger HCP crystal structures as the rubbing process is performed. The assembled crystals then move as a single entity on the surface during the rubbing motion, as shown by Park *et al.*²⁴. Since no geometrical constraint exists in the system, the

attained crystals can leave the confined CF_x spaces during the rubbing assembly process, which changes the interaction forces acting on them when entering the uncoated areas on the heterogeneous substrate. When the particles slide instead of rolling, they also tend to disturb already assembled crystal structures due to their high kinetic energy, causing voids in the crystal domains.²⁴ Therefore, Section 3.2 will elaborately discuss the effects of the geometrical shape and size of the fluorocarbon patterns, as well as the properties of the host substrate on the defects and the symmetrical order present within the assembled monolayers.

Next to the HCP crystals on the CF_x coating, Fig. 1(B) and (C) show PMMA microspheres, mostly single particles, on the uncoated SiO_2 substrate. It is noteworthy that mostly the uncoated areas between the fluorocarbon patterns also contained HCP crystal structures, *i.e.*, the assembly of HCP crystals was not limited to the fluorocarbon layers. This can be corroborated by our earlier reported findings²⁶ in which it was shown that similar to the fluorocarbon coating, the adhesion on the SiO_2 substrate was also promoted due to the contact mechanics and tribocharging mechanism. However, on the SiO_2 substrate, it was observed that the assembled monolayers were less stable as these could be disturbed during the rubbing motion, and more powder supply was needed to form these monolayers.²⁶ This observation ascertains distinct interactions on the coated and uncoated areas, *i.e.*, the adhesion on CF_x was stronger than the SiO_2 substrate, which is ascribed to the lower Young's modulus²⁶ and the stronger triboelectrification property of the CF_x layer.³⁵ This explains why certain areas on the substrate were still visible where we could distinguish between the fluorocarbon patterns covered with closely packed monolayers and the uncoated substrate (*cf.* Fig. 1(B) and (C)).

The above discussion underscores that by tuning the interactions induced during the rubbing motion between the microspheres and surface (triboelectric charging and contact mechanics), the structural order can be laterally confined on the CF_x patterns. The confinement is achieved without the use of physical boundaries or rigid walls, *i.e.* wall-less confinement. The properties of the chosen materials (coating surface as well as the host surface) determine the efficiency of the lateral confinement. This underlines the necessity of a host substrate on which the formation of HCP crystals is not promoted such that we can examine the effect of the confined CF_x pattern geometry and size on the assembled symmetry structures. This implies that the PMMA powder particles should clearly discriminate between the fluorocarbon layer and the surrounding host substrate. To select such a substrate, we first compare two available substrates that we have, namely SiO_2 and borosilicate glass substrates, on which the fluorocarbon patterns can be applied. This comparison is performed by assembling closely packed monolayers on 8 smaller samples ($14 \times 14 \text{ mm}^2$) of these homogeneous substrates. Subsequently, the quality of these monolayers can be quantified by determining the average normalized shape factor ($\mathcal{S}_{\text{norm,av}} = 1$).

Fig. 1(H) displays the normalized shape factor ($\mathcal{S}_{\text{norm,av}}$) values obtained on the bare SiO_2 and glass substrate and a uniformly fluorocarbon-coated SiO_2 substrate. Given the discussion above, it is unsurprising that the best results with



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Fig. 2 (A) The normalized average shape factor ($\vartheta_{\text{norm,av}}$) for (blue) 3 and (yellow) 10 μm PMMA spheres arranged on CF_x squares on glass and (red) 3 μm PMMA spheres on CF_x squares on SiO₂. The squares are defined by the length of the sides (l_{sq}), which is related to the total square area (A_{sq}). The number of samples per bar are between $N = 5$ and $N = 15$. (B) The number of grains per square (N_G) as a function of the square dimension for (blue) glass and (black) SiO₂ as a host substrate. (C) The average grain coverage with respect to the total square area (θ_G) as a function of the square dimensions for glass and SiO₂ as the host substrate. The data in (B) and (C) represent the assembled grains obtained using 3 μm PMMA microspheres.

The CNT explains the roughly linear increase of N_G with increasing square size (Fig. 2(B)). As the available area of the confined CF_x squares doubles, the number of critical nuclei that can occupy this area also doubles. Hence, a linear dependence is expected between the number of grains and the area.



Fig. 3 A single crystal structure identified comprising 3 μm PMMA microspheres on a CF_x pattern of $25 \times 25 \mu\text{m}^2$ on (A) glass and (B) SiO₂ as a host substrate. Multiple identified crystals (each coloured differently) comprising 3 μm PMMA microspheres on a CF_x pattern of $200 \times 200 \mu\text{m}^2$ on (C) glass and (D) SiO₂ as a host substrate.

Eqn (5) also accounts for the observation with larger microspheres. For the 10 μm PMMA spheres, a larger critical nucleus size is expected (large Ω , thus a larger A_c is needed), resulting in fewer nuclei per square. Since the critical nucleus size scales with Ω^2 , the squares need to be approximately 10 times larger ($> 1 \times 1 \text{ mm}^2$) for 10 μm particles compared to 3 μm particles to achieve low $\vartheta_{\text{norm,av}}$, *i.e.*, highly ordered monolayers comprising HCP crystals. In a previous study, we demonstrated that 10 μm microspheres can be assembled in HCP crystals using larger pattern dimensions of fluorocarbon-coated areas.²⁶ This is further supported by the data presented in Fig. 1(H) compared to Fig. 2(A), as low $\vartheta_{\text{norm,av}}$ was attained on the uniformly coated fluorocarbon-coated substrates ($14 \times 14 \text{ mm}^2$) compared to significantly smaller confined squares in the latter case.

From the slopes of the curves in Fig. 2(B), the average areas (particles and surrounding spaces) per grain (A_G , not to be confused with the grain size) are extracted, which are approximately $30 \times 30 \mu\text{m}^2$ for glass and $45 \times 45 \mu\text{m}^2$ for SiO₂ (when projected on a square grain). Thus, the smallest squares in this study ($25 \times 25 \mu\text{m}^2$) are smaller than A_G and indeed, only one grain (and sometimes even no grain) is present on an average on these squares (Fig. 2(B)). However, the average normalized shape factor is large, indicating poor overall ordering.

The relatively high-shape factor for small square sizes is reflected in the total grain coverage (θ_G , relative to the square area), shown in Fig. 2(C). For the small confined CF_x squares, coverage of $< 50\%$ is obtained regardless of the host surface. This is due to the small grain sizes (much smaller than A_G) that result in ill-defined grain boundaries and poor ordering of the microspheres.

As the wall-less confined squares on the heterogeneous substrates become larger, the grain coverage also increases, decreasing the shape factor (Fig. 2). Larger grains imply that the proportion of grain boundaries and defects on the overall

ordering decreases, reducing the average shape factor. For large CF_x squares, the difference between the glass and SiO_2 host substrates becomes more pronounced. The coverage on the CF_x squares is twice as large on SiO_2 compared to glass, even though the number of grains is less (Fig. 2(B) and (C)). This is attributed to the much larger grain sizes obtained when a SiO_2 host substrate is involved (see Fig. 3(C) and (D) and Fig. S5 in ESI†). In addition, it is observed that on the $200\ \mu\text{m}$ CF_x squares on the SiO_2 host substrate, the normalized shape factor coincides with the value obtained on uniformly fluorocarbon-coated samples (*cf.* Fig. 1), implying the order enhancement effect that the SiO_2 substrate has on the formation of HCP monolayers.

The above observations underscore a combined effect of the confined CF_x squares and the host substrate on the order of the attained monolayer arrays using the solvent-free rubbing assembly process. As elaborated in the previous section (Section 3.1), rolling particles are needed during the rubbing process to attain ordered structures.^{17,24,26} Due to the wall-less confinement, the microspheres can enter, leave or stay on the CF_x patterns, affecting their assembly behaviour. Those that are located and stay on the CF_x pattern are arranged into ordered structures, as shown in Fig. 1(B) and (C) and previous studies.^{17,26}

Assembled grains attained on the CF_x layer, which cross the edge towards the host substrate may remain intact or be disturbed, depending on the host substrate's properties, as was noted in Section 3.1. Due to the higher contact mechanics force and the better triboelectric charging, particles tend to roll more easily on SiO_2 (resulting in a lower shape factor, see Fig. 1(H)) compared to glass,²⁶ making them less likely to be disturbed when crossing the edge. This promotes grain formation. Additionally, the formation of closely packed monolayers will be promoted on SiO_2 (see Fig. 1(H)), increasing the likelihood that an ordered monolayer can cross the edge to the CF_x layer as a grain, further enhancing grain formation and ordering on the CF_x layer. This is reflected in the values found for A_G , which are higher for SiO_2 compared to glass. Therefore, the grain size and the ordering heavily depend on the host substrate and pattern material. As noted before, glass is a poor performing host substrate, significantly affecting the grains on the CF_x squares. Thus, as the host substrate's properties improve, A_G and Θ_G will increase, while N_G will decrease as the host substrate aids in the formation of larger grains.

Altogether, the results elucidate that particle arrangement on the confined CF_x square patterns is significantly affected by the host substrate in two main ways: (i) the formation of grains on the host substrate that are transferred to the CF_x squares, and (ii) the disturbance of pre-existing grains that cross the wall-less edge towards the host substrate. The quality of particle ordering is determined by both the contact mechanics force and the triboelectric charging strength of the host surface.^{17,26} In addition, larger patterns exhibit better particle arrangement (for the same particle size) due to the reduced influence of the host substrate, leading to larger grains (*cf.* Fig. S5, ESI†) and relatively fewer grain boundaries or defects. However, the edge

and its geometry have not been considered in this analysis, which is investigated in the next section.

3.3 Fluorocarbon geometrical shapes: edge effect

As briefly pointed out in Section 3.1, the geometrical shape of the confined CF_x patterns seemingly affected the order, particularly at the edge. Therefore, the influence of the edge confinement on the particle ordering is explored in this section, focusing on the geometrical shape and symmetry. To make a fair comparison between the different patterns, a form factor is introduced, which is defined as $\text{FF} = \frac{A_p}{L_p}$ with A_p being the pattern area and L_p being the pattern perimeter. A higher form factor indicates a larger surface area of the confined CF_x pattern with a relatively small edge (perimeter) contribution.

Fig. 4(A) presents the normalized average shape factor as a function of the form factor for different geometries of the CF_x patterns on the glass substrate. Similar to the squares in Fig. 2(A), $\vartheta_{\text{norm,av}}$ decreases for circles and right-angle triangles as the form factor increases (or larger pattern sizes), and $\vartheta_{\text{norm,av}}$ saturates at higher FFs. For all three geometries on the heterogeneous glass substrate, the shape factor converges to approximately the same shape factor, suggesting that for large patterns, geometry has a minor to negligible effect on the order of the assembled monolayers. However, as soon as the different geometrical patterns become smaller, distinctive packing orders are attained on the various patterns.

The local ordering of particles at the edge is analysed to further examine the geometrical effect of the CF_x pattern on the quality of the assembled packing. As such, this analysis focuses solely on particles located at the edge of the pattern, *i.e.*, the interface between the CF_x pattern and host substrate. Using this analysis, the number of nearest neighbours (NNs) is quantified for each individual particle at the edge (*cf.* Section S4 in ESI†), which can range from zero (no neighbours) to four (the maximum number of neighbours). A distribution of the number of the nearest neighbours for different geometrical shapes is shown in Fig. 4(B) and Fig. S4 (ESI†) for various form factors. Across all geometries (and form factors), the contribution of particles at the edge with three neighbours is most common, but their relative distribution varies. Circular patterns host more edge particles with two neighbours than four neighbours, while this trend is reversed for right-angle triangular and hexagonal geometries.

From the neighbours' distribution shown in Fig. 4(B), the average number of the nearest neighbours is calculated by

$$\text{NN}_{\text{av}} = \sum_{n=0}^4 n k_n \text{ with } n \text{ being the number of the nearest neighbours and } k_n \text{ being the percentage of particles with the nearest neighbours } n.$$

In Fig. 4(C), the average number of the nearest neighbours is plotted as a function of the form factor. As expected based on Fig. 4(B), the right-angle triangular geometry exhibits the best particle ordering at the edge. For the triangular pattern NN_{av} remains constant for large form factors, unlike the square and circular geometries, where NN_{av} continues to increase with larger FF.





Fig. 4 (A) The normalized average shape factor ($g_{\text{norm,av}}$) value for 3 μm PMMA particles on CF_x patterns on glass for different pattern geometries: (○) circular, (□) square and (▽) triangular geometries. (B) Distribution of the number of the nearest neighbours (NNs) of 3 μm PMMA edge particles (particles located at the edge) on patterns with different geometrical shapes. The form factors (FFs) of the different shapes is approximately 37. (C) Distribution of the average number of the nearest neighbours (NN_{av}) of edge particles on patterns with different geometrical shapes. CF_x patterns on glass: (○) circular, (□) square and (▽) triangular geometries. (★) Square geometry pattern on SiO_2 . The edge ordering is improved when SiO_2 is employed as a host substrate. All data represent a sample size of $N = 5\text{--}15$.

The quality of particle arrangement at the edge follows the order: circle < square < right-angle triangle for all form factors. This sequence is also observed in $g_{\text{norm,av}}$ for small form factors ($\text{FF} < 30$, Fig. 4(A)). This observation can be directly attributed to a discrepancy in the symmetry of the CF_x pattern and the packing symmetry of the assembled PMMA

microspheres. When the microspheres are arranged in an HCP orientation, a three-fold symmetry is present, matching the three-fold symmetry of triangular patterns. On small right-angle triangular patterns, a single grain of HCP-assembled particles can align with two of the three edges. However, in the 90° corner of the right-angle triangle, the ordering is locally disrupted due to the mismatch in symmetry. On the other hand, for square patterns, only two of the four edges would match the grain symmetry. Especially, at the corners of the pattern, disordered particle assembly is obtained, resulting in fewer neighbouring particles (Fig. 4(C)). Regarding the circular patterns, it is obvious that straight edges are absent, preventing alignment with HCP grain symmetry, which leads to even lower local ordering compared to the other geometries. Thus, the results highlight that the edges influence the ordering, while perfect ordering is unequivocally also affected by the host substrate (Section 3.2). Perfect ordering is occasionally achieved on small patterns, particularly on equilateral triangles, where the 60° angles align perfectly with the three-fold symmetry of the HCP-arranged particles.

As the form factor increases, the impact of the geometrical shape of the pattern and its symmetry on the overall particle ordering diminishes, as shown in Fig. 4(A). This reduction is caused by two factors: (i) the larger pattern-to-particle ratio and (ii) the reduced edge influence on the ordering. With larger patterns, the pattern-to-particle ratio increases, reducing the influence of the CF_x pattern corners that define their symmetry. More particles experience straight edges, which simplifies the particle arrangement through grain rotation compared to alignment in corners. This is particularly true for larger circles, where the edge becomes locally straighter, enhancing edge ordering and increasing the average number of nearest neighbours (NN_{av}). Additionally, the influence of the edge is reduced on the total ordering, as shown in Fig. 2 and 4(A). For $\text{FF} > 40$, the ordering is primarily dictated by the presence and properties of the host substrate, even though the ordering at the edge of the triangular and hexagonal patterns remains superior compared to square and circular patterns (*cf.* Fig. 4(B)). While the local symmetry of the pattern still affects particle ordering near the edge, it no longer influences particles away from the edge where grain boundaries and defects become dominant factors consistent with the observations in Fig. 2. This is further confirmed by the average nearest neighbour data of square CF_x patterns on SiO_2 (stars in Fig. 4(C)). The obtained NN_{av} values on SiO_2 are higher than for any pattern on glass. Thus, these results emphasize the importance of the host substrate on the HCP monolayer ordering on the confined CF_x patterns.

3.4 Enhancing particle assembly using nucleation sites

From the previous sections, it is understood that poor ordering of HCP crystals on the confined CF_x patterns on the glass substrate is attained (*cf.* Fig. 2). To improve the ordering on this host substrate, we performed an experiment in which the surface density of confined CF_x squares was higher than the previous patterns on glass, *i.e.*, the separation between subsequent squares became smaller.





Fig. 5 SEM images of 3 μm PMMA spherical particles on the glass host substrates with (A) and (B) 200 μm CF_x confined squares and spacing of 50 μm , (C) 50 μm CF_x confined squares (indicated with dashed lines) and spacing of 10 μm , and (D) 50 μm CF_x confined squares and spacing of 500 μm . Scale bar: 100 μm (green); 20 μm (red).

Fig. 5 shows the result of the 3 μm PMMA microspheres on 200 μm squares (A) and (B) and 50 μm squares. It can be readily observed that assembled HCP crystals are predominantly present in Fig. 5(A)–(C). In particular, it is challenging in Fig. 5(C) to distinguish the confined coated areas from the uncoated glass regions on squares of only 50 μm . This implies that the assembled crystals do not discriminate between the two areas on the heterogeneous substrate. As the spacing on the uncoated areas is less (on the order of 10–50 μm), the crystal has limited time to experience the uncoated areas of the glass substrate. Consequently, the weaker interactions between the PMMA microspheres and the glass surface have an insufficient chance to disturb already assembled crystals, leading to stable crystals that can move across the heterogeneous glass substrate (almost 80% covered with CF_x pattern overall). Additionally, it is implied that the particle–particle interactions are sufficiently strong to keep the assembled crystals intact. On the other hand, Fig. 5(D) shows that when the distance between subsequent CF_x squares of 50 μm becomes larger (coverage <10%), poor ordering is achieved on the confined patterns, which aligns with preceding results in Sections 3.2 and 3.3. Thus, these results signify that a local high density of nucleation sites is needed to achieve long-range ordering on a host substrate unfavourable (low elasticity and tribocharging) for HCP crystal formation.

As inferred from the previous sections, SiO_2 is an ideal host to form HCP crystals on the wall-less confined fluorocarbon patterns. To demonstrate the merit of the solvent-free rubbing assembly technique, we performed an experiment that highlights the point that the HCP crystals self-align to match any arbitrary geometry of the CF_x confinement on a SiO_2 substrate, as showcased in Fig. 6(A). Fig. 6(B) corroborates the notion that several nucleation sites exist, which grow into grains, as different orientations of the particles can be identified.

As the assembled HCP crystals comprising PMMA microspheres match the geometry of the confined fluorocarbon patterns, Fig. 6 shows that the shape of the ordered monolayer



Fig. 6 (A) SEM images of 3 μm PMMA spherical particles on a CF_x pattern (shaped in the logo of the RSC) with SiO_2 as the host substrate, scale bar: 100 μm . (B) Same image as in (A) but now with the grain distribution highlighted. The color bar indicates the rotational orientation of the grains. A few of the excess particles remaining on the uncoated areas were removed by blowing pressurized air (4 bar) along the substrate.

array patterns can be tuned, which can open avenues in photonics,^{17,24,26} (bio-)analytics,^{17,46} anti-counterfeiting,⁴⁷ sensing,¹⁷ and (flexible) microelectronics.⁴⁸

4 Conclusions

This study demonstrates that PMMA microspheres (3 and 10 μm) can be rapidly (<20 s) assembled into highly ordered, closely packed monolayers on various substrates using the solvent-free rubbing method. These microspheres can be confined laterally into any desired shape without physical barriers, attributed to the different contact mechanics force and tribo-electrification of the materials. The ordering process of dry



powder into HCP crystals was examined by taking two distinct host substrates and various pattern geometries and sizes. In agreement with previous studies on homogeneous substrates, we found that the best ordering is achieved on CF_x patterns, whereas poor ordering was attained on glass substrates due to the lower contact mechanics force and triboelectrification. The tribocharging is corroborated through KPFM measurements, and elastic properties explain the contact mechanics contribution.²⁶

The critical nucleation theory effectively describes the ordering process using the rubbing method. The results show that HCP ordering on the wall-less confined geometrical patterns is primarily influenced by two factors: (i) the ratio between the pattern and particle size and (ii) the host substrate properties (the uncoated surface surrounding the pattern). In particular, larger patterns promote the formation of grains and reduce the amount of grain boundaries and defects, leading to better overall ordering. The ability of the host substrate to support grain formation, determined by the aforementioned contact mechanics force and triboelectrification, further enhances ordering.

For smaller patterns with a form factor $\text{FF} \leq 25$, pattern symmetry affects particle assembly. Geometrical shapes that match the symmetry of the particle grains (three-fold symmetry) result in better edge and overall ordering. For larger patterns, edge influence on ordering diminishes and is mainly determined by the host substrate, such that conspicuous HCP monolayer arrays are attained on the SiO_2 substrate. However, even for large patterns, those with matching symmetry, e.g., triangles, exhibit better edge ordering.

Ultimately, it is elucidated that with a high density of pattern distribution, highly ordered closely packed monolayers can be formed on hard and non-elastomeric surfaces. The patterns act as nucleation points, promoting the growth of ordered monolayers.

This solvent-free rubbing technique unlocks the potential of reducing solvent waste in a more sustainable assembly era to create fully ordered monolayers and patterned arrays of HCP crystals on wall-less confined areas. The assembly of such tunable patterned arrays can hold promising applications in fields such as (bio-)analytics, photonics and electronic/sensing technologies.⁴⁹ Additionally, from a fundamental perspective, the system can function as an assembly toolbox to study grain formation on various patterns, geometries, and substrates with different colloids, which can be supported by numerical modelling.⁴⁹ From recent experiments,²⁶ we already know that under the right environmental conditions, HCP monolayers of 500 nm colloids (silica and PMMA) can be assembled using the rubbing method. But, more research is needed to understand what the size limit is to obtain ordered arrays and how surface interaction forces influence the ordering and which surface forces become dominant.^{18,50} Furthermore, by changing the wettability properties of the particles, we will explore in the near future the effect of humidity conditions on the achieved order in the system (see e.g., our work^{25,26} on silica microspheres at different humidity levels), as ambient humidity levels may affect triboelectric charging and other surface force interactions, e.g., capillary forces, among the particles themselves and between the particles and substrates.^{25,26,35}

Author contributions

ISMJ and KS conceptualised and supervised the work. ISMJ and GR performed the experiments. MTJW was involved in sample fabrication and data analysis. ISMJ and KS wrote the initial draft of the manuscript. All authors reviewed the manuscript.

Data availability

The data and detection code are accessible via DOI: <https://doi.org/10.4121/01365bbe-72ba-47fb-8044-505e1ace0e29> after submitting a request to the corresponding authors.

Conflicts of interest

The authors have no conflicts to declare.

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Notes and references

- 1 G. M. Whitesides and B. Grzybowski, *Science*, 2002, **295**, 2418–2421.
- 2 N. A. Araújo, L. M. Janssen, T. Barois, G. Boffetta, I. Cohen, A. Corbetta, O. Dauchot, M. Dijkstra, W. M. Durham and A. Dussutour, *et al.*, *Soft Matter*, 2023, **19**, 1695–1704.
- 3 T. Vicsek and A. Zafeiris, *Phys. Rep.*, 2012, **517**, 71–140.
- 4 G. Volpe, C. Bechinger, F. Cichos, R. Golestanian, H. Löwen, M. Sperl and G. Volpe, *npj Microgravity*, 2022, **8**, 54.
- 5 B. Li, D. Zhou and Y. Han, *Nat. Rev. Mater.*, 2016, **1**, 1–13.
- 6 J. Schockmel, E. Mersch, N. Vandewalle and G. Lumay, *Phys. Rev. E: Stat., Nonlinear, Soft Matter Phys.*, 2013, **87**, 062201.
- 7 J. Menath, R. Mohammadi, J. C. Grauer, C. Deters, M. Böhm, B. Liebchen, L. M. Janssen, H. Löwen and N. Vogel, *Adv. Mater.*, 2023, **35**, 2206593.
- 8 C. P. Royall, P. Charbonneau, M. Dijkstra, J. Russo, F. Smallenburg, T. Speck and C. Valeriani, *arXiv*, 2023, preprint, arXiv:2305.02452, DOI: [10.48550/arXiv.2305.02452](https://doi.org/10.48550/arXiv.2305.02452).
- 9 E. Locatelli and E. Bianchi, *Soft Matter*, 2018, **14**, 8119–8136.
- 10 P. Digregorio, D. Levis, L. F. Cugliandolo, G. Gonnella and I. Pagonabarraga, *Soft Matter*, 2022, **18**, 566–591.
- 11 N. Vogel, M. Retsch, C.-A. Fustin, A. Del Campo and U. Jonas, *Chem. Rev.*, 2015, **115**, 6265–6311.
- 12 S. Jiang, A. Van Dyk, A. Maurice, J. Bohling, D. Fasano and S. Brownell, *Chem. Soc. Rev.*, 2017, **46**, 3792–3807.
- 13 Y. Hu, Z. Tian, D. Ma, C. Qi, D. Yang and S. Huang, *Adv. Colloid Interface Sci.*, 2024, 103089.
- 14 V. Lotito and T. Zambelli, *Adv. Colloid Interface Sci.*, 2017, **246**, 217–274.



- 15 M. Schelling, T. Verouden, T. Stevens and J.-M. Meijer, *Soft Matter*, 2024, **20**, 6343–6352.
- 16 C. O. Solano-Cabrera, P. Castro-Villarreal, R. E. Moctezuma, F. Donado, J. C. Conrad and R. Castañeda-Priego, *Annu. Rev. Condens. Matter Phys.*, 2024, **16**, DOI: [10.1146/annurev-conmatphys-041124-120513](https://doi.org/10.1146/annurev-conmatphys-041124-120513).
- 17 K. Sotthewes and I. S. Jimidar, *Small*, 2024, 2405410, DOI: [10.1002/smll.202405410](https://doi.org/10.1002/smll.202405410).
- 18 I. S. Jimidar, K. Sotthewes, H. Gardeniers, G. Desmet and D. van der Meer, *Soft Matter*, 2022, **18**, 3660–3677.
- 19 B. A. Grzybowski, A. Winkleman, J. A. Wiles, Y. Brumer and G. M. Whitesides, *Nat. Mater.*, 2003, **2**, 241–245.
- 20 S. Battat, D. A. Weitz and G. M. Whitesides, *Soft Matter*, 2023, **19**, 3190–3198.
- 21 M. X. Lim, A. Souslov, V. Vitelli and H. M. Jaeger, *Nat. Phys.*, 2019, **15**, 460–464.
- 22 W. Van Geite, I. S. Jimidar, K. Sotthewes, H. Gardeniers and G. Desmet, *Mater. Des.*, 2022, **216**, 110573.
- 23 W. Van Geite, I. S. Jimidar, H. Gardeniers and G. Desmet, *Powder Technol.*, 2023, **415**, 118177.
- 24 C. Park, T. Lee, Y. Xia, T. J. Shin, J. Myoung and U. Jeong, *Adv. Mater.*, 2014, **26**, 4633–4638.
- 25 I. S. Jimidar, K. Sotthewes, H. Gardeniers and G. Desmet, *Langmuir*, 2020, **36**, 6793–6800.
- 26 K. Sotthewes, G. Roozendaal, A. Sutka and I. S. Jimidar, *ACS Appl. Mater. Interfaces*, 2024, **16**, 12007–12017.
- 27 D. Park, H. Kwak, S. Kim, H. Choi, I. Lim, M. Kwak, I.-S. Kim, H. Park, I.-Y. Eom and J.-W. Lee, *et al.*, *Adv. Funct. Mater.*, 2024, 2408902.
- 28 D. N. t Napel, J.-M. Meijer and A. V. Petukhov, *Appl. Sci.*, 2021, **11**, 5117.
- 29 K. Sotthewes, H. J. Gardeniers, G. Desmet and I. S. Jimidar, *ACS Omega*, 2022, **7**, 41828–41839.
- 30 D. J. Lacks and T. Shinbrot, *Nat. Rev. Chem.*, 2019, **3**, 465–476.
- 31 A. Šutka, K. Mālnieks, L. Lapčinskis, P. Kaufelde, A. Linarts, A. Bērziņa, R. Zābels, V. Jurkāns, I. Gorņevs and J. Blūms, *et al.*, *Energy Environ. Sci.*, 2019, **12**, 2417–2421.
- 32 L. B. da Silveira Balestrin, D. Del Duque, D. S. da Silva and F. Galembeck, *Faraday Discuss.*, 2014, **170**, 369–383.
- 33 E.-C. Shin, J.-H. Mun, S. Baek, J. Jung and Y.-H. Kim, *Chem. Phys. Rev.*, 2023, **4**, 031305.
- 34 F. Galembeck, T. A. Burgo, L. B. Balestrin, R. F. Gouveia, C. A. Silva and A. Galembeck, *RSC Adv.*, 2014, **4**, 64280–64298.
- 35 I. S. Jimidar, W. Kwiecinski, G. Roozendaal, E. S. Kooij, H. J. Gardeniers, G. Desmet and K. Sotthewes, *ACS Appl. Mater. Interfaces*, 2023, **15**, 42004–42014.
- 36 S. Lin, X. Chen and Z. L. Wang, *Chem. Rev.*, 2021, **122**, 5209–5232.
- 37 F. Galembeck, L. P. Santos, T. A. Burgo and A. Galembeck, *Chem. Soc. Rev.*, 2024, **53**, 2578–2602.
- 38 V. Lotito and T. Zambelli, *Langmuir*, 2018, **34**, 7827–7843.
- 39 V. Lotito and T. Zambelli, *Adv. Colloid Interface Sci.*, 2020, **284**, 102252.
- 40 P. Dillmann, G. Maret and P. Keim, *Eur. Phys. J.: Spec. Top.*, 2013, **222**, 2941–2959.
- 41 A. T. Gray, E. Mould, C. P. Royall and I. Williams, *J. Phys.: Condens. Matter*, 2015, **27**, 194108.
- 42 R. Becker and W. Döring, *Ann. Phys.*, 1935, **416**, 719–752.
- 43 X. C. Zeng, *J. Chem. Phys.*, 1996, **104**, 2699–2704.
- 44 D. Schwarz, R. van Gastel, H. J. W. Zandvliet and B. Poelsema, *Phys. Rev. Lett.*, 2012, **109**, 016101.
- 45 S. Karthika, T. K. Radhakrishnan and P. Kalaichelvi, *Cryst. Growth Des.*, 2016, **16**, 6663–6681.
- 46 M. Serra, D. Ferraro, I. Pereiro, J.-L. Viovy and S. Descroix, *Lab Chip*, 2017, **17**, 3979–3999.
- 47 X. Li, L. Chen, Y. Ma, D. Weng, Z. Li, L. Song, X. Zhang, G. Yu and J. Wang, *Adv. Funct. Mater.*, 2022, **32**, 2205462.
- 48 R. Pan, W. Zhang, H. Cheng, J. Yang, Y. Gong, R. Hu, Y. Zhuo, J. Ding, X. Zhang and L. Chen, *et al.*, *Adv. Mater. Technol.*, 2023, **8**, 2300514.
- 49 E. B. Lindgren, H. Avis, A. Miller, B. Stamm, E. Besley and A. J. Stace, *J. Colloid Interface Sci.*, 2024, **663**, 458–466.
- 50 J. N. Israelachvili, *Intermolecular and surface forces*, Academic Press, 2011.

