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## **REVIEW**

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# Recent advances in the application of alkynes in multicomponent reactions

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Alkynes have two active positions to carry out chemical reactions:  $C \equiv C$  and C-H. These two positions are involved and activated in different reactions using different reagents. In this study, we investigated the reactions of alkynes that are involved in multi-component reactions through the C-C and C-H positions and examined the progress and gaps of each reaction by carefully studying the mechanism of the reactions. Firstly, we investigated and analyzed the reactions involving the  $C \equiv C$  position of alkynes, including the reactions between derivatives of alkynes with RN<sub>3</sub>, sulfur compounds (RSO<sub>2</sub>R', DMSO, S<sub>8</sub>, DABCO(SO<sub>2</sub>)<sub>2</sub> and DABSO), barbituric acids, aldehydes and amines, COOH,  $\alpha$ -diazoesters or ketones, and isocyanides. Then, we examined and analyzed the important reactions involving the C-H position of alkynes and the progress and gaps in these reactions, including the reaction between alkyne derivatives with amines and aldehydes for the synthesis of propargylamines, the reaction between alkynes with CO<sub>2</sub> and the reaction between alkynes with CO.

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### 1 Introduction

Alkynes are versatile molecules that can be widely applied for transformations in organic synthesis.<sup>1</sup> Also, alkynes and their derivatives are among the most valuable chemical motifs owing to their versatile reactivities and abundance.<sup>2</sup> These basic chemicals can serve as molecular building blocks for the



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user=XDWRu9gAAAAJ&hl=en.

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development of new organic reactions and the construction of functional materials.<sup>3</sup> Alkynes have at least one C–C triple bond in which the carbon atoms are sp hybridized. The C atoms are stage is irreversible.<sup>26</sup>

functional materials.<sup>3</sup> Alkynes have at least one C-C triple bond in which the carbon atoms are sp hybridized. The C atoms are connected by one  $\sigma$  bond and two  $\pi$  bonds, in which the two p orbitals in the C atoms overlap.4 In recent years, several novel organic transformations involving alkynes have been explored for the construction of valuable molecular scaffolds.5-13 However, because sp-hybridized carbon atoms are less reactive, activating the C≡C bonds in alkynes is key to their synthetic transformation. In addition, C≡C bond substituents in alkynes can be selectively activated by either both  $\pi$ - and  $\sigma$ -activation (terminal alkynes) or  $\pi$ -activation (internal alkynes) alone.<sup>14</sup> Alternatively, alkynes play a significant role as a coupling agent in transition metal-catalysed directed C-H activation reactions (Fig. 1). Also, because of the linear geometry and low steric requirements of alkynes, they can be effectively coordinated with metal d-orbitals. All these applications make alkynes useful coupling partners in terms of their numerous beneficial transformations.15

Creating molecular complexity and molecular diversity from available and simple compounds is considered one of the basic challenges in organic synthesis. Therefore, the use of methods that use simple compounds to form complex molecules while creating several links in chemical processes in a single process is progressing. In this regard, multicomponent reactions provide these conditions.16 The applications of multicomponent reactions include the synthesis of natural products, 17 polymer chemistry, 18 agrochemistry, 19 medicinal chemistry, and drug discovery.20 In addition, through multicomponent reactions, it is possible to synthesize different and important molecular skeletons, such as γ-lactams, quinoxalinones, benzodiazepinedione, benzimidazoles, and thiazoles.21,22 These molecular skeletons have been demonstrated to possess medicinal properties, for example, thiazole scaffolds have anticancer, anti-inflammatory, and antibacterial properties.<sup>23</sup> Among the advantages of multi-component reactions, the most notable are the purification and separation of products at low cost, the formation of low tracer waste, high atom economy, and high efficiency of products.24 However, many practical features of these methods and the role of multicomponent reactions in solving delicate chemical problems still have to be demonstrated.25 Alternatively, the mechanism of multi-component In this review, we focus on some C-H and C≡C activation reactions of alkynes. In the case of the activation of C-C alkynes, we review the gaps, developments, and mechanisms of reactions of alkynes with carboxylic acids, RN₃, sulfur-containing compounds, barbituric acids, aldehydes, amines, alpha-diazo esters or ketones, and isocyanides. Then, for the activation of C-CH alkynes, we investigate the gaps, advances and reaction mechanisms of alkynes with amines and aldehydes for the synthesis of propargylamines, reaction between alkynes with CO₂, and reaction between alkynes with CO₂.

### 2 C≡C activation

#### 2.1 Reaction of RN<sub>3</sub>

Click chemistry was discovered by Sharpless *et al.*<sup>27</sup> These reactions refer to reactions that are easy to perform, generate products with high efficiency and little or no side products, are not affected by the nature of different groups, and perform well under different conditions.<sup>28</sup> The features of click reactions have are presented in Fig. 2.

Among the known reactions, a small number has the characteristics of click reactions, including the azide-alkyne cycloaddition (AAC) reaction, which produces 1,2,3-triazoles (T) (Fig. 3). This is one of the most important click reactions.<sup>28</sup> It should be noted that 1,2,3-triazoles have wide biological activities such as anti-fungal and anti-bacterial, anti-virus, antituberculosis, anti-cancer.29 The AAC reaction has the following characteristics: (1) high chemical potential energy of azides and alkynes, (2) slow reaction rate (long-term heating for inactive alkynes), (3) in the presence of solvent, electrophiles and nucleophiles are stable (azide and alkyne groups) and (4) they are accelerated by catalysts such as Cu(1) (CuAAC).28 One of the most widely used click reactions is the CuAAC reaction; however, despite its widespread use, it has disadvantages because copper is cytotoxic and the excessive consumption of copper can cause Alzheimer's, neurological disorders, and hepatitis.30

Click chemistry is used in drug discovery, organic synthesis, pharmaceutical industry, and polymer chemistry, materials

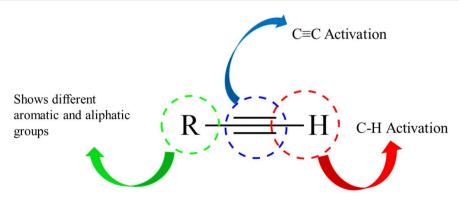


Fig. 1 Structure of an alkyne showing its two active positions: C−H and C≡C.

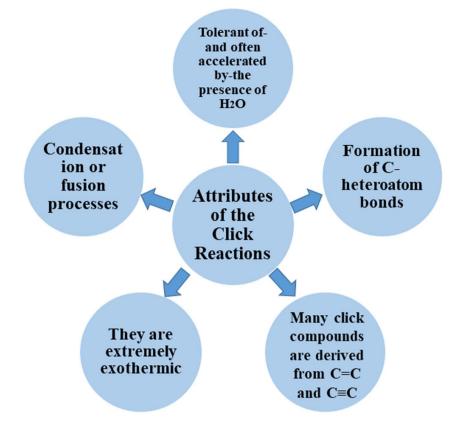


Fig. 2 General attributes of click reactions.

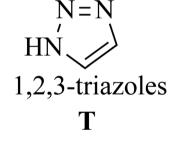


Fig. 3 Structure of 1,2,3-triazoles.

science, and bioconjugation.<sup>29</sup> Meanwhile, drug discovery based on click chemistry classified as shown in Fig. 4.

A series of  $N^1$ - and  $N^2$ -oxyalkylated 1,2,3-triazole derivatives 4 and 5 was synthesized *via* the reaction of ethers 3, alkynes 1, and TMSN<sub>3</sub> 2 in the presence of CuI and TBHP (as an oxidant) at 80 °C for 16 h by Bao and co-workers (Scheme 1).<sup>31</sup>

In the first stage for the formation of **4**, the radical species t-BuO' and t-BuOO' are produced through the Cu-catalyzed homolytic decomposition of TBHP. This process causes **A** to convert to **B**. Next, **B** is oxidized, producing  $\alpha$ -alkoxyalkyl cation

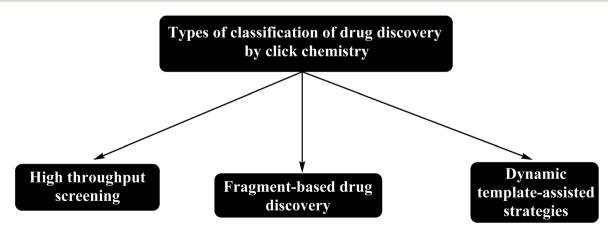


Fig. 4 Classification of drug discovery based on click chemistry.

Scheme 1 (A) Synthesis of  $N^1$ -oxyalkylated triazoles 4 and (B) synthesis of  $N^2$ -oxyalkylated triazoles 5.31

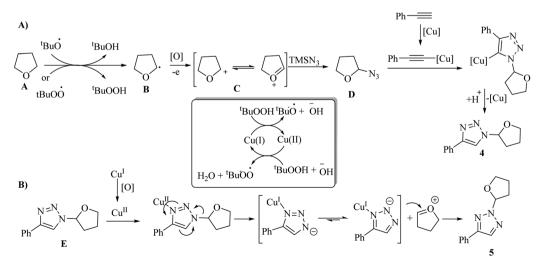
C. Then, C undergoes nucleophilic attack by TMSN<sub>3</sub> to produce oxyalkylated azide **D**, which in the presence of phenylacetylene, undergoes the click reaction and produces **4** (Scheme 2A). **E** produces 5 *via* the copper-promoted 1,2-shift of the ether group. This process leads to a sequential electronic transfer process (Scheme 2B).

In the case of 4 (Scheme 1A), the Me group at different positions of the benzene ring (*ortho*, *meta*, and *para*) in aromatic alkynes produced the desired triazole derivatives in 70%, 75%, and 66% yield, respectively. <sup>t</sup>Bu and OMe groups at the *para* position of phenylacetylene produced the desired products in 76% and 66% yield, respectively. Also, phenylacetylene with CF<sub>3</sub>, Cl, Br, and F substituents gave the corresponding triazole derivatives in 57–71% yield. Notably, aliphatic alkynes (hex-1-yne and ethynylcyclohexane) and heterocyclic aromatic alkynes (3-ethynylpyridine and 3-ethynylthiophene) produced the corresponding triazole derivatives in 65% and 60% yield and 67% and 72%, respectively. In the case of 5 (Scheme 1B), aromatic alkyne derivatives with various substituents produced

the corresponding triazole derivatives in 64–80% yield. In addition, aliphatic and heterocyclic aromatic alkynes (3-ethynylthiophene) produced the corresponding triazole derivatives in 66–78% yield.

In another study, Bao and co-workers<sup>32</sup> synthesized a selective assembly of  $N^1$ - and  $N^2$ -alkylated 1,2,3-triazole derivatives 9 and 10 via the reaction between ether 7, alkynyl carboxylic acid derivatives 6, and azidotrimethylsilane (TMSN<sub>3</sub>) 8 in the presence of CuCl<sub>2</sub> in TBHP at 80 °C for 12 h (Scheme 3).

Two pathways were proposed for the formation of **9**. In the first pathway, t-BuO' and t-BuOO' abstract the  $\alpha$ -H atom from **A**, producing alkoxyalkyl radical **B**. Then, **B** is oxidized to produce **C**. Next, **B** undergoes nucleophilic attack by TMSN<sub>3</sub>, generating **D**. Subsequently, an alkynyl copper intermediate is produced *via* the reaction between the alkynyl carboxylic acid and Cu salt and *via* the reaction with one molecule of HCl, releasing CO<sub>2</sub>. The alkynyl copper intermediate reacts with **D** and generates **E** *via* a click reaction. Finally, **D** is quenched by a proton to give **9** (Scheme 4A). In the second pathway, propiolic acids **6** react with



Scheme 2 Possible reaction mechanism for the synthesis of (A)  $N^1$ -oxyalkylated triazoles and (B)  $N^2$ -oxyalkylated triazoles.

Scheme 3 (A) Synthesis of  $N^1$ -alkylated 1,2,3-triazoles 9 via CuCl<sub>2</sub> and (B) synthesis of  $N^2$ -alkylated 1,2,3-triazoles 10 via CuCl<sub>2</sub>.<sup>32</sup>

8 via the copper-catalyzed decarboxylative cycloaddition, which generates **F**. Next, **F** undergoes alkylation to generate **9** (Scheme 4A). Then, the copper-mediated 1,2-shift of the ether group in **9** generates **10** (Scheme 4B).

The above-mentioned method is a selective and efficient approach for the synthesis of the desired triazole derivatives in 40-92% yield. In the case of 9 (Scheme 3A), arylpropiolic acids with electron-poor or -rich aryl rings react well and the desired triazole derivatives were obtained in 57-92% yield. Also, 2naphthalenyl propiolic acid, 3-(thiophen-3-yl)propiolic acid and biphenyl propiolic acid produced the corresponding triazole derivatives in 61%, 75%, and 57% yield, respectively. In addition, alkylpropiolic acid (oct-2-ynoic acid) also reacted and produced the desired 1,2,3-triazole derivative in 60% yield. Furthermore, the tolerance of ether derivatives was examined in the above-mentioned method, and besides THF, other cyclic ether derivatives such as tetrahydro-2*H*-pyran and 1,4-dioxane produced the corresponding triazole derivatives in 60% and yield, respectively. Also, acyclic

dimethoxyethane and methylphenylethyl ether) produced the desired triazole derivatives in 40% and 71% yield, respectively. In the case of **10** (Scheme 3B), similar to the preparation of  $N^1$ -alkylated 1,2,3- triazoles, substituted arylpropiolic acid derivatives containing electron-withdrawing and -donating groups produced the corresponding triazole derivatives in 70–85% yield. Specifically, 3-(thiophen-3-yl)propiolic acid produced the desired product in 73% yield, while oct-2-ynoic acid also produced the desired 1,2,3-triazole derivatives in 70% yield. Furthermore, cyclic and acyclic ether derivatives produced the corresponding triazole derivatives in 40–68% yield.

He and co-workers<sup>33</sup> synthesized 3-triazolylcoumarin derivatives  $14 \ via$  a one-pot, copper-catalyzed multicomponent reaction. The desired product was produced from the reaction of arylacetylenes 13, salicylaldehydes 11, ethyl 2-azidoacetate 12, and CuI in EtOH and  $K_3PO_4$  at  $50\ ^{\circ}C$  for  $5\ h$  (Scheme 5).

Initially, **12** and **13** generates **A** via the copper(i)-catalyzed azide–alkyne cycloaddition (CuAAC) reaction. Then, **A** reacts with an aldehyde group of **11** in the presence of  $K_3PO_4$  via Knoevenagel condensation, generating intermediate **B**. Next, in the presence of  $K_3PO_4$  the intermediate **C** is generated. Subsequently, **14** is generated via the intramolecular nucleophilic addition of **C**, followed by transesterification (Scheme 6).

The above-mentioned method has advantages such as ecofriendly nature, convenience and production of substituted 3trizolylcoumarin derivatives in moderate to excellent yields. According to this study, halide substituents in arylacetylenes were compatible to produce the desired triazolylcoumarin derivatives in good yield in the range of 67% to 82%. Also, ferrocenylacetylene 15 was investigated via the one-pot optimized condition, which produced the desired products 16 in a good yield (R = H, 85% and R = Cl, 87%).

Saikia and co-workers<sup>34</sup> synthesized 1,4-disubstituted 1,2,3-triazole derivatives **20** *via* the reaction of benzyl bromide derivatives **19**, NaN<sub>3</sub> **18**, and alkyne derivatives **17** in the presence of Cu(II) precatalyst in MeOH at 65 °C for 8–10 min under microwave irradiation (Scheme 7).

In the first stage, the reaction between **18** and organic halides in MeOH forms azides. Then, the Cu(II) catalyst is added and reduced by MeOH to Cu(I) species. Subsequently, **17** is added and forms **T**. Finally, **20** is generated *via* the protonolysis of **T** (Scheme 8).

Different types of benzyl bromide and alkyne derivatives were investigated, and results showed that the corresponding products were produced in 85–95% yield. Also, aromatic alkyne derivatives bearing an electron-withdrawing substitute such as F did not have an effect on the reaction result.

Zhang and co-workers<sup>35</sup> synthesized 5-trifluoromethylthio-1,2,3-triazole derivatives **23** using  $S_8$ , alkynes **21**, (trifluoromethyl)trimethylsilane (TMSCF<sub>3</sub>), and 4-methoxybenzyl azide (PMBN<sub>3</sub>) **22** (Scheme 9).

The desired products were obtained in two steps. In the first step, PMBN<sub>3</sub>, alkynes, CuI, and KF in DMF were mixed at 0 °C for 3 h. In step 2, S<sub>8</sub>, Ag<sub>2</sub>CO<sub>3</sub> as a silver salt, 1,10-phen as a ligand, and TMSCF<sub>3</sub> were mixed at room temperature for 12 h.

S<sub>8</sub> and TMSCF<sub>3</sub> convert to [(phen)Cu(SCF<sub>3</sub>)]<sub>2</sub> in the presence of CuI, 1,10-phen, and KF, which reacts with intermediate **S** and

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A) Pathway 1 <sup>t</sup>BuOO tBuOOH <sup>t</sup>BuO <sup>t</sup>BuOH THF D CH3 Pathway 2 [Cu]/TBHP OH + <sup>t</sup>BuO <sup>t</sup>BuOOH B) Cu(II) Cu(I) <sup>t</sup>BuOOH <sup>t</sup>BuOO + H Cu(II)

Possible reaction mechanism for the synthesis of (A)  $N^1$ -alkylated 1,2,3-triazoles and (B)  $N^2$ -alkylated 1,2,3-triazoles.

Scheme 5 (A) Synthesis of 3-triazolylcoumarin derivatives 14 and (B) synthesis of 3-(4-ferrocenyl)-trizaolycoumarins 16.33

generates key intermediate K. Finally, 23 is prepared via the reductive elimination of K (Scheme 10).

Alkyne derivatives with electron-donating substituents on the aryl gave the desired products in 74-75% yield, whereas electron-withdrawing groups led to lower yields with an isolated yield of 30-48%. Also, alkyl alkynes produced the corresponding triazole derivatives in moderate yields in the range of 36-56%.

Khandar et al.36 synthesized 1,4-disubstituted 1,2,3-triazoles 28 via the reaction of  $R'N_3$  26, alkyne derivatives 27 and benzyl halide derivatives (24 and 25) using a novel binuclear copper(II)oxalate complex as the catalyst in H<sub>2</sub>O at 70 °C (Scheme 11). Various benzyl chloride derivatives 24 with both electrondonating and -withdrawing groups (H, 2-Me, and 4-NO2) were reacted with phenylacetylene and produced the corresponding triazole derivatives in 99%, 77%, and 87% yields, respectively. It This article is licensed under a Creative Commons Attribution-NonCommercial 3.0 Unported Licence.

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Scheme 6 Possible reaction mechanism for the synthesis of 3-triazolylcoumarin derivatives.33

seems that 2-Me substituted benzyl chloride resulted in a much lower yield due to the steric hindrance effect. Also, substituting phenylacetylene with dimethyl propargyl alcohol and propargyl alcohol decreased the reaction yield (34% and 64%, respectively). In addition, the reaction of benzyl bromide 25 with dimethyl propargyl alcohol and propargyl alcohol led to the production of the corresponding products in 73% and 74% yield, respectively.

Liu and co-workers37 synthesized sulfuryl-substituted Nvinyl-1,2,3-triazole derivatives 32 via the reaction of vinyl azide surrogate 30, S- and O-nucleophiles 31 and alkynes 29 in the presence of CuI/Pd(PPh<sub>3</sub>)<sub>4</sub>, and Et<sub>3</sub>N in DMSO at 60 °C for 2 h (Scheme 12).

Active palladium species Q was prepared via the dissociation of Pd(PPh<sub>3</sub>)<sub>4</sub> catalyst, which underwent oxidative addition with 30 to generate complex R. Then, intermediate R reacted with 31 and generated functionalized vinyl azide S. Next, activated acetylide T and intermediate S reacted via a click reaction and

Scheme 8 Possible reaction mechanism for the synthesis of 1,4disubstituted 1,2,3-triazole derivatives.34

produced Cu(1) triazoles U. Finally, 32 was generated via the protolysis of U by trace H2O in solvent or air (Scheme 13).

In the case of 29, para-substituted aromatic alkyne derivatives bearing various groups gave the desired triazole derivatives in 82% to 88% yield. Also, substrates with meta-substituents such as halogen, Me, and amino gave the desired triazole derivatives in 78% to 88% yield. In addition, ortho-substituents on the benzene ring of the alkyne derivatives were reacted and produced the desired triazole derivatives in 80% to 83% yield. Also, S-containing aromatic alkynes in the reaction were tolerated and the corresponding triazole derivatives obtained in 75-77% yield. Furthermore, alkenyl and aliphatic alkynes were tested and the corresponding triazole derivatives obtained in 72% to 82% yield. A variety of aliphatic alkynes with heteroatoms was reacted and produced the corresponding triazole derivatives in 74% to 79% yield. Various sulfinates were reacted with phenylacetylene and the results showed that aryl sulfinates

Scheme 7 Synthesis of 1,4-disubstituted 1,2,3-triazole derivatives 20 using a ionic liquid-supported Cu(ii) catalyst.<sup>34</sup>

$$R = \frac{PMBN_3}{21} + \frac{PMBN_3}{22} = \frac{1)\text{CuI, KF, DMF, 0°C}}{2) S_8, \text{Ag}_2\text{CO}_3, 1, 10\text{-phen}}, \qquad PMBN_N \\ TMSCF_3, \text{ rt} \qquad F_3C - S \\ R' = -4\text{-H} \\ -4\text{-OMe} \\ -4\text{-n-pentyl} \\ -4\text{-NO}_2 \qquad \qquad R'' = \text{Me, OTs, OBz}$$

Scheme 9 Synthesis of 5-trifluoromethylthio-1,2,3-triazole derivatives 23.35

-2-C1

Scheme 10 Possible reaction mechanism for the synthesis of 5-trifluoromethylthio-1,2,3-triazole derivatives 23.35

R=H, 2-Me, 4-NO<sub>2</sub>

R'= octane, cyclohexane

Scheme 11 Synthesis of 1,4-disubstituted 1,2,3-triazoles 28 via the binuclear copper(II)-oxalate complex containing 2,9-dimethyl-1,10phenanthroline.36

R'= H, 4-OMe, 4-C<sub>2</sub>H<sub>5</sub>, 4-Me, 4-OC<sub>2</sub>H<sub>5</sub>, 4-C<sub>3</sub>H<sub>7</sub>, 4-F, 4-Cl 4-Br, 4-Bu, 4-CN, 3-Me, 3-NH<sub>2</sub>, 3-Cl, 3-F, 2-OMe, 2-NH<sub>2</sub>, 2-F, 2-Cl

R''= R'''= R''= R''= R''= R''= R''=

Scheme 12 Synthesis of N-vinyl-1,2,3-triazoles 32 via Cul.<sup>37</sup>

$$Pd^{0}(PPh_{3})_{4} \longrightarrow 2PPh_{3} + Pd^{0}(PPh_{3})_{2}$$

$$Q$$

$$Q$$

$$GF_{S}^{S} \longrightarrow Q$$

$$N_{3} \longrightarrow \begin{pmatrix} FG \\ Pd(PPh_{3})_{2} & Nu & 31 \\ Pd(PPh_{3})_{2} & FG \end{pmatrix}$$

$$R \longrightarrow Cu^{1}$$

$$R \longrightarrow Nu$$

$$N_{2} \longrightarrow Nu$$

$$N_{2} \longrightarrow Nu$$

$$R \longrightarrow Cu^{1}$$

$$R \longrightarrow Nu$$

Scheme 13 Possible reaction mechanism for the synthesis of sulfuryl-substituted N-vinyl-1,2,3-triazole derivatives 32. $^{37}$ 

with *para*-substituents produced the corresponding triazole derivatives in 78% to 89% yield. *Ortho*- and *meta*-substituted aryl sulfinates were also reacted and produced the desired triazole derivatives in 83% to 82% yield, respectively. In addition, 2-naphthyl sulfinate produced the corresponding product in 85% yield. Alkyl-substituted sulfonate derivatives were also evaluated in the above-mentioned method and produced the corresponding triazole derivatives in 74% to 77% yield. Finally, a variety of O-nucleophiles 31 was reacted and the corresponding triazole derivatives obtained in 69% to 83% yield.

Wang and co-workers<sup>38</sup> synthesized 5-aryl-1,4-disubstituted 1,2,3-triazole derivatives **36** *via* one-pot *in situ* oxidative-

coupling. The compounds were prepared using arylboronic acid derivatives 35, organic azide derivatives 33 and alkyne derivatives 34 in the presence of  $O_2$ /chloramine-T as an oxidant, CuCl as a catalyst, and TEA in toluene at 50 °C (Scheme 14).

Intermediate L was produced *via* the reaction between 34 and CuCl in the presence of TEA. Then, intermediate L reacted with 33 *via* the CuAAC reaction and generated triazolide copper Z. Next, Z reacted with 35 under oxidative conditions and generated X. It should be noted that transmetalation must take place, and then the protonation of Z (Scheme 15).

The benefits of this method include the use of a low reaction temperature, low-cost catalyst, and less-toxic additive. The reaction among aryl alkynes with various substituents, arylboronic acids, and benzyl azides produced 5-aryl-1,2,3-triazoles in 80–90% yield. The reaction among various alkyl azides with different substituted groups, benzyl azides, and arylboronic acids produced triazole derivatives in moderate to good yield (68–80%). Azide **M** with 2,5-difluorobenzyl motifs that exist in the clinic drug rufinamide reacted with arylboronic acids and terminal alkynes and produced triazole derivatives, showing that this method has potential applications for drug screening.

Camberlein and co-workers<sup>39</sup> synthesized 1,5-disubstituted 1,2,3-triazole derivatives **40** via the reaction between NaN<sub>3</sub> **39**, alkyne derivatives **38**, and bromides **37** in the presence of Cp<sub>2</sub>Ni (precatalyst), Cs<sub>2</sub>CO<sub>3</sub>, and xantphos ligand in DMF under microwave irradiation at 50 °C for 4 h (Scheme 16). Various alkynes (aromatic and aliphatic) **38** were evaluated, and the desired triazole derivatives were obtained in 27% to 84% yield. According to the regioselectivity of para-NO<sub>2</sub> phenylacetylene (ratio 66:34) with respect to phenylacetylene (ratio 89:11), and then para-F phenylacetylene (ratio 91:9) and para-OMe

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arylboronic acid R""= H, Me, F, Br R'"= H, Me, Cl, Br, F, OMe,

Scheme 14 Synthesis of 1,2,3-triazole derivatives 36 using CuCl.<sup>38</sup>

$$R' = \frac{35}{34} + \text{CuCl}$$

$$\text{base}$$

$$R' = \frac{35}{\text{L}} [\text{Cu}] \xrightarrow{\text{Ar-B(OH)}_2} R' = \frac{\text{CuCl/TEA}}{\text{R-N}_3} \xrightarrow{\text{NN-Ar}} R'$$

$$R' = \frac{1}{\text{R-N}_3} [\text{Cu}] \xrightarrow{\text{R-N}_3} R'$$

$$R' = \frac{1}{\text{R-N}_3} = \frac{\text{CuCl/TEA}}{\text{R-N}_3} \xrightarrow{\text{NN-Ar}} R'$$

$$R' = \frac{1}{\text{R-N}_3} = \frac{\text{CuCl/TEA}}{\text{R-N}_3} \xrightarrow{\text{NN-Ar}} R'$$

$$R' = \frac{1}{\text{R-N}_3} = \frac{1}{\text$$

Scheme 15 Possible reaction mechanism for the synthesis of 5-aryl-1,4-disubstituted 1,2,3-triazole derivatives 36.38

phenylacetylene, it had a positive mesomeric effect on the phenyl ring, but did not affect the yield. Interestingly, heteroaryl and alkyl chains can be employed in the above-mentioned reaction. In the case of the scope of bromide derivatives 37, aromatic benzyl bromides bearing electron-donating or

electron-withdrawing groups after flash chromatography purification (64-79%) and semi-preparative SFC (31-84%) purification produced the corresponding 1,5-regioisomers in high yields. The ortho isomer in this series was obtained in the lowest yield due to the steric effect. The regioisomerism of the

$$R = Br + R' = R' = H, 4^{-1}Bu, 4-OMe, 4Br, 4-NO_2,$$

$$R = 4-CF_3, 3-CF_3, 2-CF_3,$$

$$R'' = R'' = R''$$

Scheme 16 Synthesis of 1,2,3-triazole derivatives using the nickelocene (Cp<sub>2</sub>Ni) precatalyst.<sup>39</sup>

phenyl substitution did not influence the regioselectivity (ratio = 85:15 to 97:3) and only had a minor effect on the yield. Also, functional groups (ester and carbamate-protected amine) were tolerated in the reaction and produced the corresponding triazole derivatives in 34% and 90% yield, respectively.

R'''=H,  $NO_2$ , F, OMe

Yuan *et al.*<sup>40</sup> synthesized 5-iodo-1,4-substituted 1,2,3-triazoles 44 via the reaction among terminal alkyne derivatives 41, organic azide derivatives 42, and NaI 43 in the presence of CuI, DIPEA as a base, chloramine-T as an oxidant, and L as a ligand in THF under an N<sub>2</sub> atmosphere at 50 °C (Scheme 17).

$$R = + R' - N_3 + NaI$$

$$41$$

$$42$$

$$43$$

$$Cul/L/DIPEA$$

$$Chloramine-T/THF$$

$$N$$

$$R'' = H, Me, F, Cl, Br, OMe, I$$

$$CH_2CH_3$$

$$R'' = Cl, Br, NO_2, OMe$$

$$I = O$$

$$N_3$$

$$R''' = Cl, Br, NO_2, OMe$$

$$I = O$$

Scheme 17 Synthesis of 5-iodo-1,4-substituted 1,2,3-triazoles 44 by Cul.<sup>40</sup>

Scheme 18 Possible reaction mechanism for the synthesis of 5-iodo-1,4-substituted 1,2,3-triazoles 44.40

Scheme 19 Synthesis of triazole derivatives using CuCl catalyst. (A) Substrate scope of alkyne derivatives and azide derivatives. (B) Substrate scope of epoxide derivatives.41

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Two reaction pathways were proposed by the authors. It should be noted that **43** was oxidized to electrophilic "I<sup>+</sup>" by chloramine-T. In pathway B, iodoalkyne(III) **O** may be generated *via* iodination directly on **41**, whereas in pathway A, iodination may occur on intermediate copper triazolide(II) **P** (Scheme 18).

Also, 41 and 42 were investigated. Phenylacetylenes bearing various groups were reacted and produced the desired triazole derivatives in 71% to 78% yield. Also, benzyl azides with electron-withdrawing and -donating substituents were tested and the desired triazole derivatives obtained in 67% to 84% yield. In addition, alkyl azides were tolerated in the reaction and the desired triazole derivatives obtained in 73% to 74% yield.

In 2021, Wang and co-workers<sup>41</sup> synthesized triazoles bearing  $\beta$ -hydroxy chalcogenides (49 and 49') via the reaction among terminal alkyne derivatives 45, azide derivatives 46, epoxide derivatives (47 and 47'), and Se/K<sub>2</sub>S 48 in the presence of CuCl, Me<sub>4</sub>Phen as a ligand, KOH as a base, and TBAI as an additive in acetone/H<sub>2</sub>O [1/1(v/v)] at 25 °C for 6 h under an air atmosphere (Scheme 19).

Initially,  $Se^{2-}$  and  $SeO_3^{2-}$  were produced *via* the disproportionation reaction of Se in the presence of KOH. Then, **45** reacted with copper salt **A** and generated copper(I) acetylide **B**. Subsequently, **B** reacted with **46** and produced triazolyl-copper **D** by [3 + 2] cycloaddition. In this stage, cycloaddition included the coordination of **46** with **B** to afford a six-membered ring intermediate, and then generated 5-cuprated triazole *via* reductive elimination. Next, copper(III) species **E** was generated *via* the ligand exchange of **D** with  $Se^{2-}/S^{2-}$  and oxidation by  $SeO_3^{2-}/O_2$ . In the next step, **E** generated triazolylchalcogenide anion **F** *via* reductive elimination together with the release of the copper(I) catalyst. Finally, **49** or **49**′ was produced by nucleophilic attack of **F** to **47** (Scheme 20).

In the case of 45 (Scheme 11a), phenylacetylenes with both electron-withdrawing and -donating groups (Me, OMe, F, Cl, Br, CF<sub>3</sub>, Me<sub>3</sub>Si, <sup>t</sup>Bu, Ph, and NO<sub>2</sub>) on the benzene ring were reacted and produced the desired triazole derivatives in 39% to 81% yield and >20:1 dr. Importantly, the electronic effect of phenylacetylenes affected the yield. Phenylacetylene derivatives containing CF<sub>3</sub> and NO<sub>2</sub> gave the products in low yields (45% and 39%, respectively). In addition, thienylacetylene and naphthylacetylene reacted well and produced the corresponding products in 82% and 65% yield, respectively. Also, alkenylsubstituted alkyne derivatives showed high reactivity under the above-mentioned conditions and produced the corresponding products in 64-65% yield. Moreover, alkyl-substituted alkyne derivatives containing 3,3-dimethylbut-1-yne, ethynylcyclopropane, prop-2-vn-1-vlbenzene, and hex-1-vne were reacted with elemental selenium, epoxide, and azide and produced the corresponding products in 70%, 66%, 72%, and 70% yield, respectively. In the case of 46 (Scheme 19A), Cl- and Me-substituted benzyl azides were tested and produced the corresponding triazole derivatives in 68% and 76% yield, respectively. Also, phenethyl azide was tested and produced the corresponding triazole in 68% yield. In addition, phenyl azide was tested and produced the desired triazole in 40% yield. Especially, (trimethylsilyl)methyl azide also reacted well and produced the 1-Me substituted 1,2,3triazole in 54% yield and >20:1 dr. In the case of the scope of 47 (Scheme 19B), epoxide derivatives containing linear, aromatic, aliphatic, and cyclic epoxides reacted with elemental selenium 48, phenylacetylene 45', and benzyl azide 46' and produced the corresponding 5-selenotriazoles 49' in 47% to 79% yield.

In 2023, Sun *et al.*<sup>42</sup> synthesized ditriazolyl diselenides (53 and 53') *via* the reaction among azides (51 and 51'), terminal alkyne derivatives 50, and elemental selenium 52 in the

$$3Se + 6OH \longrightarrow 2Se^2 + SeO_3^{2-} + 3H_2O$$

$$O \qquad Ph \longrightarrow B$$

$$Ph \longrightarrow A$$

$$Ph \longrightarrow BnN_3$$

$$46$$

$$Ph \longrightarrow Cu(I)L_n$$

$$Ph \longrightarrow Cu(I)L_n$$

$$Ph \longrightarrow Cu(I)L_n$$

$$Ph \longrightarrow Cu(I)L_n$$

$$N \supset Bn$$

$$N \supset N$$

$$Ph \longrightarrow Cu(I)L_n$$

$$N \supset Bn$$

$$N \supset N$$

$$N \supset$$

Scheme 20 Possible reaction mechanism for the synthesis of triazoles bearing β-hydroxy chalcogenides. 41

presence of  $Cu(CH_3CN)_4PF_6$ ,  $Me_4phen$ , KOH, and tetrabuty-lammonium bromide in dioxane/ $H_2O$  at 25 °C for 12 h (Scheme 21).

 $SeO_3^{2-}$  and  $Se^{2-}$  were produced *via* the disproportionation of 52 in the presence of KOH. Then, Cu(i) reacted with 50 and generated alkynyl copper S. Next, S in the presence of 51 underwent cycloaddition and generated 5-cuprated triazole U. After that, copper(iii) species V was produced *via* the exchange of U with

 $Se_2^{2-}$  and oxidation by  $SeO_3^{2-}$ . Finally, intermediate **V** generated 53 or 53' and Cu(i) *via* reductive elimination (Scheme 22).

The above-mentioned reaction produced the corresponding products with high atom-economy. In the case of **50** (Scheme 21A), phenylacetylenes containing Me, <sup>t</sup>Bu, OMe, OCF<sub>3</sub>, and Cl on the phenyl ring were tested and produced the corresponding products in 40–75% yield. Also, ethynylthiophenes showed high reactivity and produced the corresponding ditriazolyl

$$R = \frac{1}{50} + \frac{1}{8n} - \frac{1}{51} + \frac{1}{52} \frac{\frac{\text{Cu}(\text{CH}_3\text{CN})_4\text{PF}_6(10 \text{ mol }\%)}{\text{Meaphen } (12 \text{ mol }\%)}}{\frac{1}{18AB}(1.0 \text{ equiv})} \frac{1}{18AB} \frac{$$

Scheme 21 Synthesis of ditriazolyl diselenides in the presence of  $Cu(CH_3CN)_4PF_6$ . (A) Substrate scope of alkynes and (B) substrate scope of azides.<sup>42</sup>

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Scheme 22 Possible reaction mechanism for the synthesis of ditriazolyl diselenides. 42

diselenides in 53-57% yield. Unfortunately, 2-ethynylpyridine did not produce the corresponding product. Alkenyl-substituted alkyne in the above-mentioned reaction was tested and produced the corresponding ditriazolyl diselenide in 62% yield. Also, alkyl-substituted alkynes in the above-mentioned method were tolerated in the present transformation. Moreover, phenoxy-, phenylthio-, and methoxy-incorporated alkyl alkynes reacted well and the corresponding ditriazolyl diselenides obtained in 51-73% yield. Also, butyldimethylsilyl (TBS)substituted alkyne was compatible and gave the desired product in 55% yield. It should be noted that ethisteronederived alkyne demonstrated high reactivity and afforded the desired diselenide in 51% yield. In the case of 51 (Scheme 21B), tBu-, OMe-, and phenyl-substituted benzyl azide derivatives 51 reacted with 50' and 52 and produced the corresponding ditriazolyl diselenides 53' in 47-59% yield. Also, 2-(azidomethyl)-thiophene reacted smoothly in the abovementioned reaction and afforded the desired ditriazolyl diselenide in 73% yield. Subsequently, (trimethylsilyl)-methyl azide reacted well with elemental selenium and phenylacetylene to obtain the desired ditriazolyl diselenide in 48% yield. 4-Azidobutyl benzoate was suitable in the abovementioned reaction and produced the desired ditriazolyl diselenide in 57% yield. Unfortunately, phenyl azide failed to give the corresponding product in this method.

A summary of the reaction of alkyne derivatives with various  $\mathrm{RN}_3$  derivatives, and also the limitation of some reactions are presented in Table 1.

#### 2.2 Reaction of sulfur-containing compounds

Ansari *et al.*<sup>49</sup> synthesized (Z)-β-phenoxy vinylsulfones **57** *via* the reaction between phenols **56**, 1-phenyl-1-propyne **54**, and sodium sulfonate derivatives **55** in the presence of  $I_2$  and  $K_2CO_3$  in EtOH at 60 °C (Scheme 23).

Initially, sulfonyl radical  $\mathbf{Q}$  was generated via the reaction between  $I_2$  and 55. Subsequently, radical  $\mathbf{Q}$  reacted with 54 and produced vinyl sulfone radicals  $\mathbf{X}$ . After that, the vinyl sulfone radical was attacked by R'SO<sub>2</sub>I,  $I_2$ , or I' and generated  $\beta$ -iodovinylsulfone intermediate  $\mathbf{Y}$ . Next, intermediate  $\mathbf{Z}$  was generated via the nucleophilic addition of PhO $^-$  to intermediate  $\mathbf{Y}$ . Finally, intermediate  $\mathbf{Z}$  produced the desired product via the elimination of iodide (Scheme 24).

The desired products were obtained in 73% to 87% yield. According to the results, the electron-deficient groups (Cl and  $CF_3$ ) on the phenyl ring gave lower reactivity than electron-rich groups (Me and t-Bu). In addition, the internal alkynes and aliphatic terminal alkynes produced 57 in 72–77% yield.

Chen *et al.*<sup>50</sup> constructed C–S and C–Se bonds and produced unexpected  $\beta$ -sulfonylvinylselane structures **60** *via* the reaction among alkyne derivatives **58**, SO<sub>2</sub>, and diselenides **59** in high selectivity for the E configuration (Scheme 25).

Initially, **59** under blue light irradiation generated selenium radical **O**. Then, **58** reacted with **O** and produced vinylic radical intermediate **P**, which after being captured by SO<sub>2</sub>, produced sulfonyl radical intermediate **Q**. After that, vinylic radical intermediate **R** was generated *via* the attack of radical intermediate **Q** to **59** and trapped by **59**. Finally, **60** was obtained (Scheme 26).

The starting materials were mixed in CH<sub>3</sub>CN under blue LEDs. The unexpected  $\beta$ -sulfonylvinylselanes were produced in good yield (up to 71%). According to the results, when various substrate alkynes 58 on the phenyl ring were used, the desired unexpected  $\beta$ -sulfonylvinylselanes were produced in 35–71% yield and when 4-ethynyl-1,1'-biphenyl was used, the unexpected  $\beta$ -sulfonylvinylselane was obtained in 54% yield. Also, 3-ethynylpyridine and its isomer were investigated and the corresponding unexpected  $\beta$ -sulfonylvinylselanes were obtained in moderate to low yield (42% and 3%, respectively). In contrast,

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Table 1 A summary of the reaction of alkynes with various RN<sub>3</sub> and the limitations of the reactions

Ref.	Wang et al. <sup>43</sup>	Xiong et al. <sup>44</sup>	Chen et al. <sup>45</sup>	Liu et al. <sup>46</sup>	He et al. <sup>47</sup>	Cheung et al. <sup>48</sup>
Limitations	Alkyl-substituted alkynes had poor reactivity and produced trace amount of corresponding product	Reaction of ethynylbenzene and 1-iododecane did not produce the desired product	Alkyne containing electron-donating group produced the products in lower yields than those with an electron-withdrawing group	<ol> <li>(1) C≡C triple bond cleavage</li> <li>(2) High product yield</li> <li>(3) High efficiency</li> <li>(4) An unprecedented 4-component reaction</li> <li>(5) Good functional group tolerance</li> <li>(1) Four new honds formation</li> </ol>		– 2 equiv. AgNO <sub>3</sub> as the oxidant – Only CF <sub>3</sub> - substitution – Only allyl azides as valid reactants
Yield	70%	30-	17-86%	52- 85%	Trace- 90%	Up to 87%
Reaction	(1) 8 h (2) 2–6 h	(1) 5–20 min (2) 10 min	8 8 9	4–10 h	رد ط	15 h
Method t employed	(1) RT, N <sub>2</sub> (2) <i>hv</i>	(1) DME (1) TBPB, RT (2) (2) 120 °C Toluene	K <sub>2</sub> CO <sub>3</sub> , H <sub>2</sub> O, 70 ° C	H <sub>2</sub> O, 70 °C	K <sub>3</sub> PO <sub>4</sub> , 50 °C	Phen, Ag <sub>2</sub> CO <sub>3</sub> , Et <sub>3</sub> N, Ar, RT
Catalyst Solvent	(1) MeOH (2) (7) Toluene	(1) DME ( 3 (2) ( Toluene	DMF	Ag <sub>3</sub> PO <sub>4</sub> DMSO	ЕгОН	DMF
Catalys	CuBr	$(1)$ $Fe(OTf)_3 (2)$ $Tolo$	Ag <sub>2</sub> SO <sub>4</sub> DMF	${ m Ag_3PO_4}$	Cul	Cul
Product	N CF3	Ar	Z Z Z	NH R NHTs	Z <sub>z</sub> Z	F <sub>3</sub> C NH <sub>2</sub>
Reactants	$ \begin{array}{c c} A_{\mathbf{L}} \\ + & CF_{3} \\ TMSN_{3} & OF_{3} \end{array} $	$Ar$ $==$ $+$ $RI$ $TMSN_3$	TMSN <sub>3</sub>	R—=== ToISO <sub>2</sub> Na + TMSN <sub>3</sub> TsN <sub>3</sub>	$ \begin{array}{c c} R & \downarrow \downarrow \\ \hline  & \downarrow \\  & \downarrow \\ R' & \longrightarrow \\ \hline  & COOEt \end{array} $	R'N <sub>3</sub> TMSCF <sub>3</sub>

Scheme 23 Synthesis of (Z) β-phenoxy vinylsulfone derivatives 57.49

$$R \stackrel{l_{2}}{=} \underbrace{\overset{l_{2}}{\circ}}_{55} \underbrace{\overset{l_{2}}{\circ}}_{0} \underbrace{\overset{l_{2}}{\circ}}_$$

Scheme 24 Possible reaction mechanism for the synthesis of (Z)-β-phenoxy vinylsulfones 57.49

alkyl acetylene was unsuitable and the corresponding product not obtained. In the case of alkyl diselenides, the tandem reactions of 1,2-dimethyldiselane with 3- ethynylthiophene, 4-methylphenylacetylene, phenylacetylene, and 4-bromophenylacetylene produced the corresponding unexpected  $\beta$ -sulfonylvinylselanes in good yield of 47% to 64%.

Zhu and co-workers<sup>51</sup> synthesized stereo-defined olefins **64** *via* single electron/triplet energy transfer and nickel catalysis. This strategy is a good method for producing tri-substituted alkenes. The reaction proceeded *via* the reaction among aryl halide derivatives **61**, alkyne derivatives **62**, and sodium sulfonate **63** in the presence of an Ru photocatalyst, NiCl<sub>2</sub>·dme, and 4,4′,4″-tri-*tert*-butyl-2,2':6′,2″-terpyridine as a ligand in DMF at room temperature for 24 h under blue LED irradiation (7.4 W) (Scheme 27).

Initially, the  $Ru^{II}$  or  $Ir^{III}$  complex absorbed visible light in the photocatalytic cycle and generated \* $Ir^{III}$  or \* $Ru^{II}$  triplet excited state. Then, the radical precursor was oxidized to the corresponding radical and formed  $Ru^{I}$  and  $Ir^{II}$  reductants. The desired radical was added to a  $Ni^{I}$  complex and entered the Ni catalytic cycle. Subsequently, it underwent reduction and 1,2 migratory insertion of alkyne, producing A. Then, intermediate B was produced by intermediate C was produced

Subsequently, **64** was generated from intermediate C *via* reductive elimination (Scheme 28).

According to the results, various alkynes with various functional groups were used (boronic ester, acetal, amine, bromide, ester, fluoride, chloride, alkyl, methoxy, and methylthio groups) and produced the desired olefins in 29–99% yields. Furthermore, 3-ethynylthiophene produced the corresponding olefin in 76% yield and cyclohexenyl alkyne and internal alkyne derivatives were suitable substrates for producing the desired products (29–60%).

Yi et al.  $^{52}$  synthesized arylamidines **69** via the four-component reaction of sulfonyl azide derivatives **65**, terminal alkyne derivatives **66**, water **67**, and oxime halide **68** by [3 + 2] cycloaddition in the presence of triethylamine and Cu(i) in DCM at room temperature for 12 h under open air (Scheme 29).

Initially, ketenimine intermediate **F** was generated *via* the reaction between **66** and TsN<sub>3</sub> in the presence of Et<sub>3</sub>N and CuI. Then, oxime halide **68** generated nitrile oxide **G** under Et<sub>3</sub>N. Subsequently, **G** reacted with intermediate **F** and produced intermediate **H** *via* the formal [3 + 2] cycloaddition reaction. Next, intermediate **I** was generated *via* the electronic ring opening of intermediate **H** with the help of OH<sup>-</sup> or H<sub>2</sub>O. Intermediate **I** reacted with OH<sup>-</sup> or H<sub>2</sub>O and produced intermediate **J**, accompanied by the formation of **66**. Finally, **69** was obtained from intermediate **J** in the presence of OH<sup>-</sup> or H<sub>2</sub>O,

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blue LEDs 58 CH<sub>2</sub>CN

**R**= Ph, 4-Br-Ph, 4-OMe-Ph, 4-NO<sub>2</sub>-Ph, 3-OMe-Ph,4-Ph-Ph 4-Cl-Ph, 4-Me-Ph, 4-COOMe-Ph, 3-Me-Ph, 3-NH<sub>2</sub>-Ph, pentane

R'= Ph, Me, CH<sub>2</sub>Ph,4-Me-Ph, 4-CF3-Ph, 4-Cl-Ph, thienyl

Scheme 25 Synthesis of β-sulfonylvinylselane 60 via the visible-light-mediated multicomponent reaction.<sup>50</sup>

$$R' \xrightarrow{Se} Se \xrightarrow{R'} O$$

$$R' \xrightarrow{Se} Se Se O$$

$$R' \xrightarrow{Se} Se O$$

$$R' \xrightarrow{Se}$$

Scheme 26 Possible reaction mechanism for the synthesis of unexpected β-sulfonylvinylselanes 60.50

accompanied by the release of O2. It should be noted that the role of H<sub>2</sub>O in the reaction is important, which provides the OH<sup>-</sup> anion and the source of proton, and also may promote the cleavage of the N-O and C-O bonds (Scheme 30).

According to the results, arylamidines were produced in 68-92% yield. Importantly, oxime chloride derivatives 68 as the carbon source and water 67 played a critical role in the reaction. In addition, terminal alkyne derivatives, which can lead to high yields of corresponding products using lower amounts of reactant, may play a catalytic role.

Hazarika et al.53 synthesized 2-[(ortho-methylthio)aryloxy] substituted dialkyl maleates 73 via the reaction among alkyne derivatives 70, DMSO 72, and aryne derivatives 71 in the presence of KF at 50 °C (Scheme 31).

Initially, four-membered ring intermediate E was formed via the reaction between benzyne, which was produced in situ, and DMSO. Then, intermediate E underwent ring opening due to the ring strain, and o-quinone intermediate F was generated. Zwitterion G was the resonance form of F. G underwent nucleophilic attack to form the activated alkyne, and also RSC Advances Review

Scheme 27 Synthesis of tri-substituted alkenes via nickel catalysis.51

demethylation, which led to the generation of anion **H**. Finally, anion **H** produced 73 in the presence of H<sub>2</sub>O (Scheme 32).

This strategy produces trisubstituted vinyl ethers with remarkable stereospecificity. According to the results, different types of activated alkyne derivatives treated with *o*-silyl aryl triflate produced the corresponding products in 63–81% yield. Notably, bulky alkyl group-substituted activated alkyne derivatives as a comparison to small alkyl group-substituted activated alkyne derivatives showed lower yields. Investigation of symmetrical aryne precursors showed that the desired products were produced in 59–77% yield. Interestingly, the symmetrical aryne precursors 3-(trimethylsilyl)-2-naphthyl trifluoromethanesulfonate, and 4,5-dimethoxy-2-(trimethylsilyl) phenyl trifluoromethanesulfonate produced a small amount of the corresponding products, which could be due to steric

hindrance. Also, unsymmetrical aryne precursors such as 4methoxy-2-(trimethylsilyl)phenyltrifluoromethanesulfonate and 2-methyl-6-(trimethylsilyl)phenyltrifluoromethanesulfonate were evaluated and produced the desired products in 66-69% yield. When the unsymmetrical aryne precursor 4-methyl-2-(trimethylsilyl)phenyltrifluoromethanesulfonate was treated activated derivatives diwith alkyne such as methylacetylenedicarboxylate and diethylacetylenedicarboxylate, regioisomer mixtures, they gave 69% and 63% yield, respectively.

In 2019, indolizine thione derivatives **76** were synthesized *via* the formation of C–N, C–C and C–S bonds *via* the reaction among ynals **75**, elemental sulfur, and 2-alkylpyridines **74** was reported by Chen and co-workers (Scheme 34).<sup>54</sup> Two conditions

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Possible reaction mechanism for the synthesis of stereo-defined olefins.<sup>47</sup>

Scheme 29 Synthesis of arylamidines 69 from oxime chlorides.<sup>52</sup>

were investigated for the synthesis of indolizine thione derivatives 76.

Firstly, 74 and 75 reacted and generated enyne intermediate U via Knoevenagel condensation. Intermediate V was produced from intermediate U via the intramolecular nucleophilic attack and activation of S. Then, the cleavage of the disulfide bond produced 76 (Scheme 33a). Also, W generated intermediate Z via the elimination process and electrophilic addition. Next, intermediate Z produced 76 via aminothiation with elemental sulfur (Scheme 33b).

The desired products were synthesized once in 50-88% yield in DMF under argon at 80 °C for 1 h, and again the reaction process for the synthesis of the corresponding products was carried out in the presence of iodine as a catalyst and under the same conditions for 4 h and products were obtained in 52-93%

yield. According to the results without the catalyst, ynals 75 with different groups (electron-donating and -withdrawing) produced the corresponding indolizine thiones in 50-86% yield. Me or OMe at the o-position of the benzene ring had a low impact on the efficiency of the reaction. In addition, the disubstituted substrates (3,5-(Me)2 and 3,5-(F)2) produced the desired indolizine thiones in 70% and 52% yield, respectively. Ynals with strong and weak electron-poor substitutes reacted smoothly. In contrast, aliphatic ynal did not produce the corresponding indolizing thione. According to the results from the I<sub>2</sub>-catalyzed aminothiation of 2-alkylpyridines, the ynal bearing a naphthalene moiety produced the desired indolizine thione in 93% yield. Interestingly, the heterocycle ynal produced the desired indolizine thione in 53% yield.

$$CI \longrightarrow H_{2}O \longrightarrow Et_{3}N \cdot H^{+} + OH$$

$$CI \longrightarrow NH_{2}O \longrightarrow G$$

$$CI \longrightarrow H_{2}O \longrightarrow G$$

$$G \longrightarrow$$

Scheme 30 Possible reaction mechanism for the synthesis of arylamidines 69.52

Scheme 31 Synthesis of dialkyl maleates 3 from activated alkyne. 53

Jin and co-workers<sup>55</sup> synthesized 2,3-dihydro-1H-imidazo[1,2-a]indole derivatives **80** via the reaction among allylamine derivatives **79**, sulfonyl azide derivatives **78**, and alkyne derivatives **77** (Scheme 35). In the first step, triethylamine (Et<sub>3</sub>N) as a base and CuI as a catalyst were added to the mixture in DMSO at room temperature for 1 h. After that, L1, K<sub>2</sub>CO<sub>3</sub>, and CuI were added to the mixture and heated at the temperature of 80  $^{\circ}$ C for 6 h.

Initially, 77 and 78 generated intermediate G *via* a Cucatalyzed azide–alkyne cycloaddition reaction. Then, intermediate G produced ketenimine H by the extrusion of  $N_2$ , and carboxamidine I and/or its tautomer I' was generated *via* the nucleophilic addition of 79 to intermediate H. After that, 80 was produced *via* the Cu-catalyzed C-N coupling reaction (Scheme 36).

The corresponding products were produced with in isolated yield of 16–77%. According to the results, 1-bromo-2-ethnylbenzene demonstrated higher reactivity compared to the iodo- and chloro-substituted substrates. In addition, aromatic alkyne derivatives with different groups reacted with *para*-toluenesulfonyl azide.

Xu *et al.*<sup>56</sup> synthesized  $\alpha$ , $\alpha$ -bis-sulfonyl arylketone derivatives **84** *via* the reaction among H<sub>2</sub>O **83**, bromoalkyne derivatives **81**, and sulfinic acid derivatives **82** at room temperature under visible-light irradiation in DMSO solvent and N<sub>2</sub> atmosphere for 12 h (Scheme 37).

Initially, sulfinyl sulfones **B** were produced *via* the condensation of **A**, which decomposed to radical **C**. Then, radical **C** was added to **81** and generated intermediate **D**, which removed the

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TMS 
$$\frac{70}{\text{ROOC}} = \frac{70}{\text{COOR}}$$
 $\frac{70}{\text{TMSO}} = \frac{70}{\text{COOR}}$ 
 $\frac{1}{\text{TMSO}} = \frac{70}{\text{COOR}}$ 
 $\frac{1}{\text{TMSO}} = \frac{70}{\text{COOR}}$ 
 $\frac{1}{\text{TMSO}} = \frac{70}{\text{COOR}}$ 
 $\frac{1}{\text{COOR}} = \frac{1}{\text{COOR}}$ 
 $\frac{1}{\text{$ 

Possible reaction mechanism for the synthesis of 2-[(o-methylthio)aryloxy]-substituted dialkyl maleate.<sup>53</sup>

Scheme 33 Possible reaction mechanism for the synthesis of (a) indolizine thione derivatives without I<sub>2</sub> and (b) indolizine thione derivatives with I<sub>2</sub>.<sup>54</sup>

bromide radical (Br'). After that, surplus sulfonyl C reacted with E to afford intermediate F via conjugate addition. Next, the hydroxyl radical ('OH) coupled with intermediate F and produced intermediate G, which was tautomerized to 84. It should be noted that 'OH may be generated via the reaction between H<sub>2</sub>O and possible radical intermediates (Scheme 38a). Also, E was afforded via a similar procedure, except intermediate H was produced via the 1,4-addition reaction of E and extra A', and also an initial hydrolysis. In addition, Br' and alkynyl radical were generated via the homolytic cleavage of 81. Next, intermediate I was produced via the coupling of intermediate H with an alkynyl radical. The important point about intermediate I was that it can be unstable. Finally, R'SO<sub>3</sub>H reacted with intermediate I via electrophilic addition and generated 86 (Scheme 38b).

In the case of 81, aromatic bromoalkynes bearing para-Me, meta-Me, para-Et, para-Pr, and para-<sup>t</sup>Bu were reacted with benzenesulfonic acid and produced the corresponding arylketone derivatives in 76% to 89% yield. A phenyl substituent at the para-position on the benzene ring of bromoalkyne produced the corresponding arylketone in 94% yield. Bromoalkyne derivatives with a halogen-substituent at the meta- and para-position

were tested and produced the corresponding arylketone derivatives in 72% to 82% yield. In contrast, substrates with Cl at the o-position did not react with benzenesulfonic acid to obtain the corresponding product. CF<sub>3</sub> and thiophene substituents on the alkynyl bromide were also reacted and produced the corresponding arylketone derivatives in 68% and 53% yield, respectively. Importantly, the aliphatic bromoalkyne of 1-bromohex-1yne was not transformed in this reaction, and no desired product was produced. Toluenesulfonic acid was tested and produced the corresponding arylketone in 60% yield. In contrast, when o-toluenesulfonic acid was employed, no reaction occurred. Furthermore, sulfonic acid derivatives with halogen substituents on the benzene ring produced the desired arylketone derivatives in 64% to 84% yield. An CF<sub>3</sub> group on the benzene ring was reacted and produced the desired product in 57% yield. In addition, benzenesulfonic acid with disubstituted groups of Me and F produced the corresponding product in 43% yield. Also, ethanesulfonic and methylsulfonic acid were tolerated and the desired arylketone derivatives obtained in 85% and 81% yield, respectively.

Bhat and Lee<sup>57</sup> synthesized 3-arylsulfonylated thioflavones 90 via the three-component reaction of aryl diazonium salts 88,

$$\begin{array}{c} S_8,I_2\ (10\ mol\%)\\ \hline DMF, 80^\circ C, 4h\\ \hline \\ R'\\ \hline \\ 75\\ \hline \\ DMF, 80^\circ C, 1h\\ \hline \end{array}$$

**R'**=H, Me

$$\mathbf{R}^{"}=\mathbf{Ph},$$

$$\mathbf{R}^{"}$$

$$\mathbf{R}^{"}$$

$$\mathbf{R}^{"}$$

R''= 3-Me, 2-Me, 3,5-(Me)<sub>2</sub>, 4-¹Bu, 4-OMe, 2-OMe,CN 4-F, 3-F, 4-Cl, 4-Br, 3,5-(F)<sub>2</sub>, 3-CF<sub>3</sub>, 4-Ac, CO<sub>2</sub>Me, NO<sub>2</sub>

Scheme 34 Synthesis of indolizine thiones 76 from ynals.54

$$R = -H$$

$$-Me$$

$$-OMe$$

$$-F$$

$$-Br$$

$$-OMe$$

$$-F$$

$$-OMe$$

$$-F$$

$$-CI$$

$$-CF_3$$

$$R = -H$$

$$-Me$$

$$-OMe$$

$$-F$$

$$-CI$$

$$-CF_3$$

$$R = -H$$

$$-Me$$

$$-OMe$$

$$-F$$

$$-R = -F$$

$$-CI$$

$$-CF_3$$

$$R = -H$$

$$-Me$$

$$-OMe$$

$$-F$$

$$-R = -F$$

$$-CI$$

$$-CF_3$$

$$R = -H$$

$$-Me$$

$$-OMe$$

$$-F$$

$$-R = -F$$

$$-CI$$

$$-CF_3$$

Scheme 35 Synthesis of 2,3-dihydro-1*H*-imidazo[1,2-*a*]indole derivatives 80.<sup>55</sup>

double C-N coupling ľ

Scheme 36 Possible reaction mechanism for the synthesis of 2,3-dihydro-1*H*-imidazo[1,2-a]indole derivatives 80.55

R= Ph, 4-OMePh, 4-MePh, 3-MePh, 4-EtPh, 4-PrPh, 4tBuPh, 4-PhPh, 4-FPh, 3-FPh, 4-ClPh 3-ClPh, 4-BrPh, 3-BrPh, 2-ClPh, 4-CF<sub>3</sub>Ph,thiophene

R' = Ph, Me, Et,  $C_6H_4p-Me$ ,  $C_6H_4o-Me$ ,  $C_6H_4p-F$ ,  $C_6H_4m$ -F,  $C_6H_4p$ -Cl,  $C_6H_4m$ -Cl,  $C_6H_4p$ -Br,  $C_6H_4m$ -Br,  $C_6H_4p$ - $CF_3$ ,  $C_6H_3m$ -F-p-Me,

MeO Br + 
$$R'$$
 S-ONa +  $H_2O$  blue LEDs (450-455 nm) DMSO,  $N_2$ , 16 h

 $R' = C_6H_5$ , p-Cl-C<sub>6</sub>H<sub>4</sub>,m-F-C<sub>6</sub>H<sub>4</sub>, m-Br-C<sub>6</sub>H<sub>4</sub>, p-Me-C<sub>6</sub>H<sub>4</sub>, p-<sup>t</sup>Bu-C<sub>6</sub>H<sub>4</sub>, P-F-C<sub>6</sub>H<sub>4</sub>

Synthesis of  $\alpha$ , $\alpha$ -bis-sulfonyl arylketones under visible-light irradiation.<sup>56</sup>

Scheme 38 Possible reaction mechanism for the synthesis of  $\alpha, \alpha$ -bis-sulfonyl arylketones. 56

DABSO 89, and alkyl-substituted alkynones 87 in CH<sub>3</sub>CN under N<sub>2</sub> at 23 °C for 4 h (Scheme 39).

Initially, complex B was produced via the electrostatic interaction between A and DABSO. Then, complex B produced C and D via homolytic cleavage of the N-S bond and concomitant single-electron transfer. After that, a radical intermediate was generated via the reaction between E and 87. Finally, via intramolecular radical cyclization, 90 was produced. It should be

$$Ar + Ar'N_2BF_4 + DABSO \xrightarrow{MeCN, N_2} SAr'$$
88 89  $39$   $3°C, 4 h$ 

90 Alkynes

R'= 2-Me, 3-Me, 4-Me, 4-OMe 2-F, 3-F, 4-F, 4-Cl, 4-Br, 4-CN,

#### Diazonium salts

Scheme 39 Synthesis of 3-arylsulfonylated thioflavone derivatives 90 by alkyl-substituted alkynone derivatives.<sup>57</sup>

noted that by-product S was observed in this reaction, which was produced via the reaction between 87 and the methylsulfonyl radical (Scheme 40).

In the case of 88, the corresponding products in the presence of different substituents produced in 40-97% yield. No obvious electronic effects of these substituents were observed.

Scheme 40 Possible reaction mechanism for the synthesis of 3-arylsulfonylated thioflavones.<sup>57</sup>

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Moreover, the p-, m-, and o-substituted aryl diazonium salt derivatives produced the desired -arylsulfonylated thioflavones in 87-97% yield. Heteroaromatic and naphthyl substituents were tested, although lower yields were shown with quinolinesubstituted diazonium salt derivatives. In the case of 87, alkyne derivatives with electron-donating and -withdrawing substituents produced the corresponding 3-arylsulfonylated thioflavones in 52-91% yield. No significant differences in reactivity were observed for the ortho, meta, and para substituents. Also, thienyl or pyridyl groups produced the correspond-3-arylsulfonylated thioflavones in 57-96% yield. Unfortunately, the use of 1-(2-(methylthio)phenyl)hept-2-yn-1one or 1-(2-(methylthio)phenyl)-4-phenylbut-2-yn-1-one as the alkyl-substituted alkynone was not successful and did not produce the corresponding products.

In 2022, Kalari and co-workers<sup>58</sup> synthesized β-acyl allyl sulfone 94 via the reaction among alkynes 91, DMSO 93, and

$$R = \begin{bmatrix} 92 \\ R'SO_2Na & (2.5 \text{ equiv}) \\ Selectfluor & (2 \text{ equiv}) \\ O \\ 93 \end{bmatrix}, 110^{\circ}C \\ 94 \\ R' = Ts, Ph$$

$$R = \begin{bmatrix} R'' = H, Me, OMe, Br \\ CH_2CH_3, tBu, Cl \end{bmatrix}$$

Scheme 41 Synthesis of β-acyl allyl sulfone 94 by Selectfluor.<sup>58</sup>

sodium sulfinate derivatives 92 in the presence of Selectfluor at 110 °C for 12 h (Scheme 41).

Initially, 93 was activated by Selectfluor to afford E (radical cation). Then, E reacted with F and generated key intermediate F. Intermediate G was produced via the reaction between 91 and intermediate F, which reacted with additional F to afford intermediate H. After that, I was generated via the demethylthiolation of intermediate H. Finally, 94 was obtained via the reaction between 92 and I (Scheme 42).

In the above-mentioned reaction, DMSO acts as a dualcarbon synthon. Phenyl acetylene without substrates and phenyl acetylenes bearing Me-, OMe-, tBu-, ethyl-, Br-, and Clafforded the desired products in 41-66% yield. It should be noted that the investigation of aliphatic alkyne ethynylcyclohexane or 1-ethynylcyclohex-1-ene showed that no desired product was synthesized.

A summary of the reaction of alkynes with some of sulfurcontaining structures, and also the limitations and progress of reactions are presented in Table 2.

#### 2.3 Reaction of acids

2.3.1 Reaction of barbituric acids. Jana and co-workers<sup>65</sup> synthesized 2-arylbenzo[d]imidazo[2,1-b]thiazole derivatives 98 via the I<sub>2</sub>/DMSO-mediated reaction among arylacetylene 95, 2aminobenzothiazole 96, and barbituric acid derivatives 97 at 110 °C for 12 h (Scheme 43).

Initially, 95 reacted with I2 in DMSO and generated B via intermediate A. Then, electron-deficient alkene C was produced via Knoevenagel condensation between B and 97. After that, D was produced via the aza-Michael addition of 96 to C. Finally, 98 was prepared via the intramolecular cyclization of D under acidic conditions together with H2O removal (Scheme 44).

Scheme 42 Possible reaction mechanism for the synthesis of β-acvl allvl sulfones 94.58

Table 2 Summary of the reaction of alkynes with sulfur-containing structures and limitations or progress of the reactions

Reactants	Product	Catalyst	Solvent	Method employed	Reaction time		Limitations or progress	Ref.
R O O O O O O O O O O O O O O O O O O O	R' N O=S=O R"	$ m Ag_3PO_4$	DMSO	H <sub>2</sub> O, 70 °C	6 h	Up to 74%	<ul><li>(1) New transformation of alkynes</li><li>(2) C-C, C-N, and C-S bonds formed.</li><li>(3) Excellent FG tolerance</li></ul>	Tang et al. <sup>59</sup>
HO Ar + TMSN <sub>3</sub> RSO <sub>2</sub> Na	O HN Ar SO <sub>2</sub> R	${ m Ag_3PO_4}$	DMSO	H <sub>2</sub> O, 70 °C	4 h		Efficient C to N aryl migration of homopropargylic alcohol	Ning et al. <sup>60</sup>
$R = $ $+$ $R'SO_2N_3  CS_2$	R NHSO <sub>2</sub> R'	$\text{CuBr} \cdot \text{SMe}_2$	Toluene	BDU, 100 °C	22 h	53- 97%	Electronic and steric variations had not remarkable effect on the reaction	Khalaj et al. <sup>61</sup>
$\begin{array}{c} R \\ & \longrightarrow H \\ + & \stackrel{\text{Nu}}{+} \\ R' \longrightarrow & \text{SO}_2\text{Na} \end{array}$	R H O	Eosin Y	EtOH	I <sub>2</sub> , K <sub>2</sub> CO <sub>3</sub> , visible light, RT	48 h	52- 80%	(1) Exclusive Z-selectivity (2) C–S and C–O bond formation	Sahoo et al. <sup>62</sup>
R NaO	Y-N O S O	_	DMF	I <sub>2</sub> , K <sub>2</sub> CO <sub>3</sub> , 70 ° C	5 h	Up to 85%	<ul><li>(1) C-S and C-N bond formation</li><li>(2) Excellent stereoselectivity</li><li>(3) Mild operation</li></ul>	Ansari et al. <sup>63</sup>
ROOC + DMSC	COOR	-	_	KF, 50 °C	5 h	Up to 81%	(1) Stereoselective (2) Transition-metal-free (3) Multiple bond cleavage and bond formation	Hazarika et al. <sup>64</sup>

**R'**= 6-Me, 6-OEt, 6-Cl, 6-NO<sub>2</sub>, 5,6-diMe

Scheme 43 Synthesis of 2-arylbenzo[d]imidazo[2,1-b]thiazole derivatives 98.65

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Scheme 44 Possible reaction mechanism for the synthesis of 2-arylbenzo[d]imidazo[2,1-b]thiazole derivatives 98.65

According to the results, 2-aminobenzothiazoles **96** with a variety of groups such as 6-NO<sub>2</sub>, 6-Me, 5,6-dimethyl, 6-OEt and 6-Cl produced the desired products in 74%,78%, 80%, 83%, and 76% yield, respectively. Moreover, phenyl acetylene derivatives **95** with either electron-donating or -withdrawing groups (Br, Me, OMe, CF<sub>3</sub>, and NO<sub>2</sub>) produced the desired products in 79%, 85%, 70%, 85%, and 84% yield, respectively.

Panday *et al.*<sup>66</sup> synthesized pyrimidine-linked naphthoquinone-fused pyrrole derivatives **102** *via* iodine-mediated one-pot multicomponent reactions (Scheme 45). The desired product was obtained *via* the reaction among 2-amino-1,4-naphthoquinone **101**, barbituric acid derivatives **100**, and terminal aryl alkynes **99** in DMSO and molecular iodine at 110  $^{\circ}$ C for 4–5 h.

2-Iodoacetophenone C was generated via the reaction between 99 and  $I_2$ . Next, C oxidized to phenylglyoxal T in the presence of DMSO. Then, alkene compound U was produced via

the reaction between **T** and **100**. After that, trisubstituted methane intermediate **V** was obtained via the 1,4 addition of reaction between **U** and **101**. Finally, **102** was obtained via intramolecular cyclization together with  $H_2O$  removal (Scheme 46).

Various alkynes (4-OMe, 4-Br, 4-NO<sub>2</sub>, and 4-CF<sub>3</sub> phenyl acetylenes) produced the desired products with an isolated yield of 60% to 78%.

In 2022, Bhuyan *et al.*<sup>67</sup> synthesized pyrano[2,3-d]pyrimidine-2,4(3H,5H)-dione derivatives **106** *via* the reaction among aromatic aldehydes **104**, N,N-dimethyl barbituric acid **103**, and terminal alkyne derivatives **105** in the presence of ZnCl<sub>2</sub> in HFIP under ultrasonic irradiation at 50 °C for 1 h (Scheme 47).

Firstly, **103** reacted with **104** *via* Knoevenagel condensation and generated **G**. Simultaneously,  $ZnCl_2$  reacted with **105** and produced zinc(II) alkynilide **H**. After that, **H** underwent conjugative alkynylation with **G** and produced the intermediate **I**.

$$\frac{1}{100}$$
 +  $\frac{1}{100}$   $\frac{1}{100}$ 

Scheme 45 Synthesis of naphthoguinone-fused pyrroles 102.66

 $NO_2, CF_3$ 

DMSO 
$$Me_2S$$
  $Me_2S$   $Me_2S$ 

Scheme 46 Possible reaction mechanism for the synthesis of pyrimidine-linked naphthoquinone-fused pyrrole derivatives 102.66

$$R' = H, 4-Me, 2-NH2, 4-F$$

$$R' = H, 4-Me, 2-NH2, 4-F$$

$$R' = \frac{ZnCl_2, HFIP}{US (40 KHz)}$$

$$O = \frac{R'}{US (40 KHz)}$$

Scheme 47 ZnCl<sub>2</sub>-catalyzed synthesis of pyrano[2,3-d]pyrimidine-2,4(3H,5H)-dione derivatives 106.67

Finally, the alkynylated product **106** was produced through 6-endo-dig cyclization (Scheme 48).

Various electronic functional groups such as -F,  $-NO_2$ ,  $-NH_2$ , -Me, and -Cl present on both aldehyde derivatives 104 and alkyne derivatives 105 were shown to tolerate the reaction conditions. However, although all terminal alkynes performed well, aromatic alkynes (yield: 79–95%) gave higher yields than

aliphatic alkynes (yield: 68–73%). Electronic factors play a role among aromatic alkynes, given that those with electron-donating groups such as –NH<sub>2</sub> and CH<sub>3</sub> were observed to produce slightly lower products than their unsubstituted counterparts. Also, the electron-withdrawing –F group has little impact on the product yield. Among the various groups on the aromatic ring of aldehydes, the electron-withdrawing groups -Cl

Scheme 48 Possible reaction mechanism for the synthesis of pyrano[2,3-d]pyrimidine-2,4(3H,5H)-dione derivatives 106.67

and  $-NO_2$  in the *para* position were found to support the reactions when aromatic alkyne derivatives were used as one of the reaction partners. However, when using aliphatic alkynes, this trend was not observed, but the effect was minimal. Interestingly, the -Cl group placed in the *ortho* position in aromatic aldehydes had the opposite effect, with lower yields being obtained in these cases. The aromatic bicyclic aldehyde also reacted well and gave a good yield of product (70%). *n*-Octanal provided the desired product in trace amounts, as identified by HRMS. All other aliphatic aldehydes did not give suitable products under the reaction conditions.

Asghari and co-workers<sup>68</sup> synthesized spiro 1,3-oxazines **110** *via* the reaction among N,N'-disubstituted parabanic acid derivatives **109**, imine derivatives **107**, and dialkyl

acetylenedicarboxylate derivatives (DAAD) **108** in dry dichloromethane without catalyst at room temperature for 1–6 h (Scheme 49).

Initially, **107** reacted with **108** and produced 1,4-dipolar intermediate **J**. After that, zwitterionic intermediate **K** was generated via attack of **J** to the carbonyl group of **109**. Finally, **110** was obtained via the intramolecular reaction between iminium and alkoxide group of intermediate **K** (Scheme 50).

The desired products in the above-mentioned method were produced in 44–98% yield. According to the results, the different groups (withdrawing and donating) at the *para* position of the *N*-arylaldimine derivatives as nucleophiles were effective and the desired 1,3-oxazines obtained in 84–98% yield. The reaction yields decreased when using electron-withdrawing

Scheme 49 Synthesis of spiro 1,3-oxazine derivatives 110.68

Scheme 50 Possible reaction mechanism for the synthesis of spiro 1,3-oxazines 110.68

groups at the *meta* position in the *N*-arylaldimine derivatives (65–84%) but the yield of the reaction in the presence of a methyl group increased, which could be due to the inductive effect at the *meta* position (96%). In addition, when 2-substituted *N*-arylaldimines were tested, the reaction yields decreased, which could be because of the hindrance effect at the *ortho* position (44–76%).

**2.3.2 Reaction of RCOOH.** Shinde *et al.*<sup>69</sup> reported a novel method for the synthesis of ethanopyrido[1,2-b]isoquinolines **114** *via* the reaction among ninhydrin **111**, proline **112**, and alkynes **113** in CH<sub>3</sub>CN at 50 °C for 1 h (Scheme 51).

Initially, intermediate **K** was generated via the reaction between ninhydrin and **112**. After that, aziridinium cation intermediate **L** was produced via the attack of amine on the ninhydrin carbonyl carbon. Next, intermediate **L** underwent ring expansion and produced intermediate **M**. After that, intermediate **N** was obtained via the nucleophilic attack of the enol carbon on the proline carbon. It should be noted that the authors could not isolate intermediate **N**. In the next step, intermediate **O** was produced from intermediate **N** by the loss of  $H_2O$ , followed by decarboxylation. Next, intermediate **O** was

converted into isoquinolinium ylide O', probably. Finally, **114** was produced *via* the reaction between O' and **113** by [3 + 2]-cycloaddition reaction (Scheme 52).

According to the results, various aromatic alkyne derivatives containing different types of groups at different positions on the aromatic ring produced the desired of ethanopyrido derivatives in 67-90% yield. Heteroaryl alkyne (thiophene) also produced the desired ethanopyrido derivatives in 75-80% yield. The above-mentioned method was extended with ethyl phenylpropiolate derivatives and produced the corresponding ethanopyrido derivatives in 70-90% yield. Furthermore, unactivated terminal alkynes produced the desired products in 70-88% yield. Importantly, terminal aromatic alkynes containing free amine group also produced the corresponding of ethanopyrido [1,2-b]isoquinoline in 88% yield. Moreover, TMS acetylene was also treated with isoquinolinium ylide and produced the corresponding ethanopyrido[1,2-b]isoquinoline in 73% yield. However, the reaction did not proceed with 1-butyne and cyclohexylacetylene.

Thakur *et al.*<sup>70</sup> synthesized 3-substituted 2-quinolone derivatives **118** *via* the reaction among terminal alkynes **116**, 2-

 $\mathbf{R'}= \mathbf{H}, \mathbf{CO}_2\mathbf{Me}, \mathbf{CO}_2\mathbf{Et}$ 

Scheme 51 Synthesis of pyrido [1,2-b] isoquinoline derivatives 114.69

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Scheme 52 Possible reaction mechanism for the synthesis of ethanopyrido[1,2-b]isoquinolines 114.69

iodoaniline derivatives 115, and oxalic acid 117 as the C1 source in the presence of polystyrene-supported palladium (Pd@PS) nanoparticle catalyst, and TBAI as a base in DMF at 135 °C under MW irradiations for 1 h (Scheme 53).

Initially, 115 and Pd@PS reacted and generated arylpalladium intermediate A via oxidative addition. Then, vinylpalladium intermediate B was obtained via the insertion of 116 in intermediate A. When intermediate B is produced, two pathways (A or B) may be followed. In the case of pathway A, intermediate C was produced via the interaction of CO with Pd metal. Next, acylpalladium intermediate D was obtained via the insertion reaction. After that, 118 was generated via the nucleophilic attack of the amine group on the carbonyl of intermediate D. In this pathway for synthesis of the 118, another minor pathway exists, where intermediate E was obtained via the formylation of intermediate B. Then, intermediate E formed 118. In the case of pathway B, intermediate F was produced via the

reaction between CO<sub>2</sub> and the amine group of intermediate B. Next, intermediate F produced intermediate G in the presence of base. Finally, intermediate G generated 118 via the nucleophilic attack of the vinyl group at the carbon of the isocyanate group together with the regeneration of Pd@PS (Scheme 54).

All the 4-CH<sub>3</sub>, 4-COOCH<sub>3</sub>, 4-CN, 4-Cl, 5-Cl, 4-Br, and 4-F substituted 2-iodoanilines reacted well with 4-ethynyltoluene and phenylacetylene to produce the desired 2-quinolone derivatives in 43-75% yield. The relatively lower yield of 2-ethynyltoluene compared to 3-ethynyltoluene could be due to the steric hindrance at the *ortho* position of the alkyne. The reaction of 2iodoaniline with 4-tert-butylphenylacetylene led to the formation of the corresponding 3-substituted 2-quinolone in 64% yield. Moreover, the C<sub>2</sub>H<sub>5</sub>/4-OCH<sub>3</sub>-substituted alkynes produced the desired 2-quinolone derivatives in 56% and 58% yield, respectively. The halogen-substituted aromatic alkynes also produced the corresponding products in 43-55%. The aromatic alkyne

Scheme 53 Synthesis of 3-substituted 2-quinolone derivatives 118 by oxalic acid. 70

Pathway B Pathway A base В R' ,CO Pd@PS Pd@PS base 118 115 d@PS -HX D Pd@PS base -HX

118

Scheme 54 Possible reaction mechanism for the synthesis of 3-substituted 2-quinolone derivatives 118.70

derivatives containing electron-withdrawing groups ( $-\text{CF}_3$  and -CN) produced the desired 2-quinolone derivatives in 42% and 55% yield, respectively. In addition, alicyclic alkynes (1-ethynylcyclohexene and 1-ethynylcyclohexane) also produced the desired 2-quinolone derivatives in 52% and 49% yield, respectively. The aliphatic alkynes were tested and produced the desired 2-quinolone derivatives in 54–57% yield.

Zeng and co-workers<sup>71</sup> synthesized  $\alpha$ -carbonyloxy ester derivatives **122** *via* a two-stage process. In the first stage, carboxylic acids **119**, and ynol ethers **120** reacted in the presence of Ag<sub>2</sub>O in dioxane at 100 °C for 5 h and produced alpha-alkoxy enol ester intermediate (1), and then 3-chloroperoxybenzoic acid **121** added to the mixture and stirred at room temperature for 12 h (Scheme 55).

Firstly, **119** and **120** reacted and produced  $\alpha$ -alkoxy enol ester **K** under catalytic Ag<sub>2</sub>O promotion. After the synthesis of **K**, two pathways may exist. In pathway A, intermediate **L** was generated *via* the reaction between **121** and the carbonyl group of **K**. Next, **122** was produced from intermediate **L** *via* fragment cascade and Baeyer–Villiger-type reaction. In pathway B, intermediate **M** was generated *via* the epoxidation of **K** by **121**. Finally, **122** was obtained from **M** (Scheme 56).

When various carboxylic acid derivatives **119** and ynol ether derivatives **120** were evaluated, the desired  $\alpha$ -carbonyloxy ester

derivatives were produced in 45-91% yields. In the case of 119, various groups produced the corresponding products in 69-83% yield. Also, heterocyclic carboxylic acid derivatives (thiophene, pyrrole, and furan) and naphthyl carboxylic acids produced the corresponding product in 91%, 57%, 53%, 86% and 70% yield, respectively. Moreover, aliphatic carboxylic acids also produced the corresponding products in 49-79% yield. Also, N-protected amino acid derivatives produced the corresponding products in 72-81% yield. Importantly, the products generated from L-proline and L-alanine showed dr values of 54: 46 and 53:47, respectively. In the case of 120, various substituted groups in the benzene ring such as para-methoxy, para-fluoro, para-chloro, ortho-chloro, meta-methyl, and paramethyl produced the corresponding products in 49-83% yield. In addition, the aliphatic and terminal ynol ether in the abovementioned method engaged and produced the desired α-carbonyloxy ester derivatives in 45% and 49% yield, respectively.

Wang *et al.*<sup>72</sup> synthesized β-amino amides **127** with high diastereoselectivity *via* the four-component reaction of triazenyl alkynes **123**, carboxylic acids **125**, aldehydes **124** and anilines **126** in the presence of  $Sc(OTf)_3$  and 4 Å MS in DCE at 80 °C (Scheme 57). In the above-mentioned method, the alkyne scaffold serves as a C2 fragment, which can lead to high diastereoselectivity.

Scheme 55 (A) Ag<sub>2</sub>O catalyzed synthesis of  $\alpha$ -carbonyloxy esters 122. (B) Gram-scale reaction for the synthesis of  $\alpha$ -carbonyloxy ester; 1:  $\alpha$ alkoxy enol ester intermediate synthesized via the reaction between benzoic acid and ynol ether.71

Initially, 124 and 126 reacted and produced imine R, probably, and also probably 123 reacted with 125 and generated Eenol ether Q. After that, a Zimmerman-Traxler chair-like transition state was formed. It should be noted that Sc(OTf)<sub>3</sub> could be coordinated with the nitrogen atom of the imine and triazene group. Next, intermediate T was obtained via intramolecular addition of the enol ether to R. Finally, 127 was produced via the Mumm rearrangement together with the regeneration of the catalyst. The important point about this reaction was that 123 acted as a bifunctional building block (Scheme 58).

In the case of 123, various substitutions in the aromatic ring such as F, Cl, Me, and OMe were tolerated and produced the corresponding products in 81-87% yield with excellent diastereoselectivity (d.r = up to >20:1). It should be noted that the substitution position did not have a remarkable effect on the reaction yields. The thiophene-2-yl-substituted triazenyl alkyne also produced the corresponding β-amino amide in 90% yield (d.r = >20:1). The alkyl-substituted triazenyl alkyne derivatives such as benzyloxymethyl and phenylethyl produced the corresponding  $\beta$ -amino amides in 87% and 86% yield (d.r = 5:3 and >20:1), respectively. The benzyloxymethyl-substituted triazenyl alkyne produced the corresponding beta-amino amide with moderated d.r, which may be due to the O atom of the benzyloxymethyl group coordinating with Sc(OTf)<sub>3</sub>. In addition, the trimethylsilyl-substituted triazenyl alkyne produced the desilylation beta-amino amide product in 70% yield, and when piperidine-substituted triazenyl alkyne was tested, the desired β-amino amide could be formed in 76% yield (d.r = >20:1). In the case of 125, aldehydes, and anilines, various substituted benzoic acids such as methoxy, [1,3]dioxole, and cyano were tested in the reaction and produced the corresponding β-amino amides in 89%, 90%, and 80% yield, respectively with excellent d.r (d.r = >20:1). The heterocyclic carboxylic acids bearing indole and thiophene produced the corresponding β-amino

Scheme 56 Possible reaction mechanism for the synthesis of  $\alpha$ -carbonyloxy ester derivatives 122.71

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Synthesis of  $\beta$ -amino amide derivatives 127 using Sc(OTf)<sub>3</sub>.72

amides in 82% and 90% yield, respectively. Unsaturated carboxylic acids such as 3-phenylpropiolic acid and cinnamic acid also produced the corresponding beta-amino amides in 80% and 90% yield, respectively. Aliphatic carboxylic acids (allyl, protected alcohol, amino, and cyclopropyl) were well engaged in the reaction and produced the corresponding βamino amide derivatives in 92%, 90%, 90%, and 87% yield, respectively. Also, N-Boc L-proline produced the desired product in 80% yield. Next, a variety of benzaldehydes with the substitution of OMe, Cl, F, Br, CF<sub>3</sub>, CN, and CO<sub>2</sub>Me was tolerated and

produced the corresponding beta-amino amide derivatives in 80% to 95% yield. In addition, anisaldehyde produced the desired product in 63% yield. The heterocyclic aldehyde derivatives were tested and the desired beta-amino amide derivatives obtained in 62% to 73% yield. Ethyl oxoacetate also produced the desired β-amino amide in 70% yield. Finally, various aniline derivatives 126 were tested and anilines with differential substituted positions and functionalities (F, Cl, CF<sub>3</sub>, CO<sub>2</sub>Me, NO<sub>2</sub>, CN, ethyne, and OCF<sub>3</sub>) were tolerated and the desired βamino amide derivatives obtained in 71-89% yield. Importantly, aliphatic aldehydes and alkylamines failed in the reaction and did not produce the desired  $\beta$ -amino amides.

#### Synthesis of heterocyclic structures *via* the reaction among alkynes, aldehydes and amines

Fernández and co-workers73 synthesized pyrano[2,3,4-ij]isoquinoline derivatives 131 (132 was generated in a molar ratio of 2.5:1:1) via the reaction among dimethyl acetylenedicarboxylate 133, ortho-alkynylsalicylaldehydes 135 and anilines 134 (molar ratio: 1:1:1) in the presence of AgOTf in THF under reflux (Scheme 59).

Initially, intermediate L was generated via the coordination of AgOTf to the triple bond of K, which was prepared by the reaction between 129 and 130. After that, intermediate M was obtained via the intramolecular addition of the nitrogen of the imine to the alkyne. Subsequently, 8-isoquinolinones N was produced via the intramolecular protodemetalation reaction together with the regeneration of the silver catalyst. Next, intermediate N via [4 + 2] cycloaddition reactions with 128 generated the first desired product 131. In addition, when an excess of 128 was used, derivatives 131 reacted with it and produced cyclobutenes O via [2 + 2] cycloaddition reactions. After that, the cyclobutene underwent electrocyclic ring opening and produced novel tricyclic intermediates P. Then, P may produce the bicyclic intermediates Q via another ring-opening

R 
$$=$$
 N  $=$  N  $=$ 

Scheme 58 Possible reaction mechanism for the synthesis of β-amino amides 127.72

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Scheme 59 (A) Synthesis of pyrano[2,3,4-ij]isoquinoline derivatives (molar ratio = 1:1:1) and (B) synthesis of benzo[de]chromene derivatives (molar ratio = 2.5:1:1).

process. Finally, 132 was obtained *via* the electrocyclic ringclosing process of intermediates **Q** (Scheme 60).

The desired products were produced in moderate yields. According to the results, when an excess of dimethyl acetylenedicarboxylate (molar ratio: 2.5) was used under the same condition, benzo[de]chromene derivatives 137 were synthesized instead of pyrano[2,3,4-ij]isoquinoline derivatives 136. The corresponding products were obtained in moderate yield (40–48%).

et al.<sup>74</sup> synthesized 1,3-disubstituted Das tetrahydroisoquinolines 141 and isoindolines 142 via the reaction among aldehyde derivatives 138, aniline derivatives 140 and alkyne derivatives 139 with high yield (up to 92%), diastereoselectivity (up to 9:1), and enantioselectivity (up to 99%) (Scheme 61). Imination-alkynylation occurred at room temperature in C<sub>6</sub>H<sub>5</sub>CH<sub>3</sub>, but the desired products were produced in the presence of an equimolar solution of lithium hexamethyldisilyl amide in THF at 0 °C. According to the results, due to higher yield and enantioselectivity using 4methoxyaniline, the above-mentioned reaction was extended to various terminal alkyne derivatives with this amine. When different substituents were used on the aromatic ring of the alkyne, no remarkable changes in selectivities and yields were shown and the corresponding products obtained in high yields (up to 86%), and ee (up to 96%). Trimethylsilyl acetylene in this reaction was unable to produce the desired product. In addition, aliphatic terminal alkynes were investigated and produced the corresponding products with enantioselectivities of 87-93% and high yields of 84-86%.

Chandra *et al.*<sup>75</sup> synthesized 4-arylated quinolines **146** *via* the reaction among paraformaldehyde **145**, anilines **143**, and alkynes **144** through [4 + 2] cycloaddition of alkyne derivatives and imine in the presence of  $(\pm)$  camphor-10-sulfonic acid

(CSA) in TFE under microwave irradiation at 90 °C for 20 to 30 min (Scheme 62).

Initially, the reaction between **145** and **143** may *in situ* produce intermediate **A**. After that, intermediate **D** was generated from [4 + 2] cycloaddition of **142** with intermediate **B** *via* a Povarov-type multicomponent reaction. Finally, **146** was produced *via* the spontaneous oxidation of intermediate **D** (Scheme 63).

According to the results, the corresponding products with different paraformaldehyde, aniline, and alkyne derivatives were produced in excellent yields (up to 90%). In the case of **144**, various alkyne derivatives were reacted with *ortho*-toluidine and different *para*-substituted phenyl acetylenes produced the corresponding quinolines in 31–76% yield. Interestingly, polyaromatic alkyne was tolerated and the desired quinoline was produced in 68% yield. In addition, 3-ethynylthiophene produced the desired quinoline in 48% yield. In contrast, internal alkyne did not react.

In 2019, 2,4-disubstituted quinolones **150** were synthesized *via* the reaction among alkynes **148**, aldehydes **147**, and amines **149** in toluene at 110  $^{\circ}$ C (Scheme 64). When different alkynes **148**, aldehydes **147**, and amine derivatives **149** were used, the desired quinolines were produced in 46–92% yield. Also, 3-thienylacetlyene, 1-hexyne, and 1-ethynylcyclohexene produced the desired quinolone derivatives in 91%, 55%, and 86% yield, respectively. <sup>76</sup>

Yu and co-workers<sup>77</sup> synthesized spirofuranhydropyrimidinone derivatives **154** *via* the three-component Biginelli-like tandem reaction of alkynol **153** as an enolizable carbonyl equivalent, aromatic aldehydes **151**, and (thio)urea **152** in the presence of co-catalyst of palladium chloride (transition metal) and trifluoroacetic acid (Brønsted acid) in dioxane at 50 ° C for 12 h (Scheme 65).

Scheme 60 Possible reaction mechanism for the synthesis of benzo[de]chromenes and pyrano[2,3,4-ii]isoquinolines.73

Initially, *N*-acyliminium intermediate **G** was generated *via* the condensation of **151** and **152** in the TFA. After that, vinylpalladium species **H** was obtained *via* intramolecular addition of the OH group to the electron-deficient alkyne, where it should be mentioned that the coordination of PdCl<sub>2</sub> with **153** increased the electrophilicity of the triple bond. Next, intermediate **I** was produced *via* the attack of **H** to intermediate **G**. Then, endocyclic enol ether **J** was generated *via* the isomerization of the exocyclic double bond in the presence of the Pd catalyst. Next, oxonium intermediate **K** was obtained *via* the nucleophilic attack of the enol ether to a second aldehyde. Finally, the catalyst was regenerated and **154** was obtained *via* intramolecular cyclization (Scheme 66).

0

Ε

The desired spirofuran-hydropyrimidinone derivatives were produced in 45–87% yield and diastereoselectivity. In the case

of **151**, some groups such as ester, carbonyl, nitro, halide, and hydroxyl were amenable in this catalytic system. Generally, benzaldehyde with electron-withdrawing substituents produced the corresponding spirofuran-hydropyrimidinone derivatives in worse yields compared to that with donating groups (71–78% and 80–87% yield). Notably, the reaction of benzaldehyde with a substituent at the *meta* position proceeded worse than that at the *para* position. Also, di- and tri-substituted benzaldehydes produced the corresponding spirofuran-hydropyrimidinone derivatives in 77–85% yield. Other aldehydes such as furfural and 1-naphthaldehyde produced the corresponding spirofuran-hydropyrimidinone derivatives in 73% and 81% yield, respectively but no desired product could be produced with 2-pyridine formaldehyde. When urea was replaced with thiourea, the corresponding spirofuran-hydropyrimidinone derivatives were

A)

COOR

$$R' = R''' = H$$
, OMe, F, Cl

 $R'' = Br$ , CF3

COOR

 $R' = Pr$ 
 $R'' = Pr$ 
 $R''$ 

R"=4-OMePh, 4-EtPh, 4-nPnPh, 4-ClPh, 4-BrPh, 4-FPh, SiMe<sub>3</sub>,nBu,

R'"= Ph, 4-OMePh, 4-MePh, 4-ClPh, 4-BrPh, 4-FPh 3-FPh, 4-NO<sub>2</sub>Ph,Ts, Cbz, nPn,PMP, COOMe 1.Cu<sup>I</sup>-L B) Toluene, rt 2.LHMDS, THF, 0°C CHO R=H, Me, F, R'= H, OMe, Et 142

Scheme 61 (A) Synthesis of 1,3-disubstituted tetrahydroisoquinolines and (B) isoindolines.<sup>74</sup>

Scheme 62 Synthesis of 4-arylated quinoline derivatives 146.75

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$$(HCHO)n$$

$$NH_{2}$$

$$143$$

$$H_{2}O$$

$$A$$

$$N$$

$$CSA$$

$$B$$

$$CSA$$

$$Ph$$

$$144$$

$$N$$

$$I_{4+2}$$

$$Cycloaddition$$

$$C$$

$$Ph$$

$$CSA$$

$$Ph$$

$$I_{4+2}$$

$$I_{4+2}$$

$$Cycloaddition$$

$$I_{2}O$$

$$I_{3}O$$

$$I_{46}$$

Scheme 63 Possible reaction mechanism for the synthesis of 4-arylated quinolones 146.75

**R**= Ph, 4-ClPh, 4-OMePh, 4-NO<sub>2</sub>Ph, 2-ClPh, 2-OMePh 2-NO<sub>2</sub>Ph, 3-ClPh, 3-OMePh, 3-NO<sub>2</sub>Ph,cyclohexane

$$\mathbf{R'}=\mathrm{Ph}, \quad \mathbf{R''}=\mathrm{H}, \mathrm{Br}, \mathrm{Me}, \mathrm{OMe}$$

$$\mathbf{R''}=\mathrm{H}, \mathrm{Br}, \mathrm{Me}, \mathrm{OMe}$$

$$\mathbf{R''}=\mathrm{H}, \mathrm{Br}, \mathrm{Me}, \mathrm{OMe}$$

Scheme 64 Synthesis of 2,4-disubstituted quinoline derivatives 150 via the triazole-gold catalyst.76

produced in 45–57% yield. The scope of **153** also was investigated and the desired products were produced in 62–77% yield.

Purohit and co-workers<sup>78</sup> *via* the reaction among aldehyde derivatives **156**, aminopyridines **155** and alkynes **157** synthesized substituted imidazopyridines **158** in the presence of PW-CIS500 nano-catalyst at 100 °C under neat condition (Scheme 67). The reaction proceeded smoothly when used various alkynes **157**, aldehydes **156**, and amines **155**. According to the study, the desired imidazopyridines were produced with an isolated yield of 82–95%. Also, various aryl-alkynes/aldehydes with electron-donating and electron-withdrawing groups produced the desired imidazopyridines with an isolated yield of 82% to 95%.

Xian *et al.*<sup>79</sup> synthesized 2,4-diphenylpyrimido[1,2-b]indazole derivatives **162** *via* the reaction among alkyne derivatives **161**, 1H-indazol-3-amine **159**, and aldehydes **160** under solvent-free conditions in the presence of FeCl<sub>3</sub> at 110 °C (Scheme 68).

Initially, the condensation reaction between **159** and **160** led to the formation of imine **A**, which released an  $H_2O$  molecule through the reaction. Then, iron complex intermediate **B** was obtained *via* the attack of iron acetylide complex to **A**, which isomerized to intermediate **C**. After that, intermediate **D** was

obtained after regioselective attack of the intramolecular N–H bond activated the triple bond through 6-endo-dig-cyclization. Finally, **162** was generated *via* the autoxidation and demetalation of intermediate **D** (Scheme 69).

The desired indazole derivatives were produced in 69% to 93% yield. Alkynes with different groups gave the desired indazole derivatives in 83–85% yield. Under the same conditions, substituted aromatic alkynes and substituted aromatic aldehydes were evaluated and produced the corresponding products with an isolate yield of 73% to 79%. This method has advantages such as the use of available materials and an environmentally friendly catalyst. Also, the abovementioned method is a good way to produce pyrimidine derivatives.

Yuan *et al.*<sup>80</sup> synthesized  $\gamma$ -carbonyl- $\alpha$ -amino acids **166** *via* the reaction among amines **165**,  $\alpha$ -carbonyl aldehyde derivatives **163**, and propargylic alcohols **164** in the presence of AgNO<sub>3</sub> in CH<sub>2</sub>Cl<sub>2</sub> at room temperature for 24 h (Scheme 70).

Initially, the reaction among **164**, **165**, and glyoxylate in the presence Ag catalyst generated product **A** *via* a Mannichtype reaction with intermediate **B**. In this section, intermediate **B** acted as a nucleophile (cycle Z). Subsequently,

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Scheme 65 Synthesis of spirofuran-hydropyrimidinones 154 through a Biginelli-like tandem reaction with co-catalyst of PdCl<sub>2</sub> and TFA.<sup>77</sup>

Scheme 66 Possible reaction mechanism for the synthesis of spirofuran-hydropyrimidinone derivatives 154.77

intermediate silver-alkyne complex C was produced via the coordination of intermediate A with Ag<sup>+</sup>. Next, intermediate **D** was obtained *via* the reaction between intermediate **C** and H<sub>2</sub>O. In the last step, enol E was generated via the protonation of intermediate D, and then E was tautomerized to 166 (cycle X) (Scheme 71).

According to the results, γ-carbonyl-α-amino acids were produced in 71% to 91% yield. Aromatic amine derivatives

-4-OMe

Scheme 67 Synthesis of imidazopyridine derivatives 158.78

Scheme 68 Synthesis of 2,4-diphenylpyrimido[1,2-b]indazole derivatives 162 under solvent-free conditions.<sup>79</sup>

Scheme 69 Possible reaction mechanism for the synthesis of 2,4-diphenylpyrimido[1,2-b]indazole derivatives 162.79

containing either electron-withdrawing or -donating groups on the aromatic ring produced the corresponding  $\gamma$ -carbonyl- $\alpha$ -amino acids in 78–90% yield. Aliphatic amine derivatives were also compatible and produced the corresponding  $\gamma$ -carbonyl- $\alpha$ -

amino acids in 71–87% yield. In the above-mentioned method, when morpholine was used, none of the desired  $\gamma$ -carbonyl- $\alpha$ -amino acids were detected. In addition, a variety of secondary and tertiary propargyl alcohol derivatives including acyclic and

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$$\mathbf{R} = \mathbf{C}_{2}\mathbf{H}_{5},$$

$$\mathbf{R} = \mathbf{H} \mathbf{O}$$

$$\mathbf{R} + \mathbf{R} \mathbf{E} \mathbf{O}$$

$$\mathbf{R} + \mathbf{R} \mathbf{E} \mathbf{O}$$

$$\mathbf{R} + \mathbf{R} \mathbf{E} \mathbf{O}$$

$$\mathbf{R} \mathbf{H} \mathbf{O}$$

$$\mathbf{R} \mathbf{O}$$

$$\mathbf{O}$$

$$\mathbf{R} \mathbf{O}$$

$$\mathbf{O}$$

$$\mathbf{R} \mathbf{O}$$

$$\mathbf{O}$$

$$\mathbf{$$

Scheme 70 AgNO<sub>3</sub>-catalyzed synthesis of  $\gamma$ -carbonyl- $\alpha$ -amino acid derivatives 166.80

$$R'''$$
 $R'''$ 
 $R''''$ 
 $R'''$ 
 $R'''$ 
 $R'''$ 
 $R'''$ 
 $R'''$ 
 $R'''$ 
 $R'''$ 
 $R'''$ 
 $R''''$ 
 $R'''$ 
 $R'''$ 
 $R'''$ 
 $R'''$ 
 $R'''$ 
 $R'''$ 
 $R'''$ 
 $R'''$ 
 $R'''$ 

Scheme 71 Possible reaction mechanism for the synthesis of  $\gamma$ -carbonyl- $\alpha$ -amino acids 166.80

cyclic substituted propargyl alcohol derivatives produced the desired products. However, general aliphatic and aromatic alkynes such as 1-hexyne and phenylacetylene were used, and

165

none of the corresponding products were detected (B and A, respectively). In this transformation, the alcohol OH group of propargylic alcohol played an essential role.

Mohammod and co-workers<sup>81</sup> synthesized 3-aminoindolysines **170** *via* the reaction among pyridine-2-carbaldehyde **167**, secondary amine derivatives **168**, and terminal alkyne derivatives **169** in the presence of the CuCN/[bmim][PF<sub>6</sub>] system under an  $N_2$  atmosphere at 70 °C (Scheme 72). Firstly, Cu-acetylide intermediate **G** was generated *via* the activation of **169** by CuCN. Then, *N*-propargylamine amine intermediate **H** was obtained *via* the reaction between intermediate **G** and iminium ion **F**. In this stage, the electrophilicity of the internal alkyne increased because the triple bond of **169** 

Scheme 72 Synthesis of 3-aminoindolysines 170 in the presence of the CuCN/[bmim][PF $_6$ ] system.<sup>81</sup>

Scheme 73 Possible reaction mechanism for the synthesis of 3-aminoindolysines 170.81

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was coordinated with CuCN. After that, cationic heterocyclic intermediate I was produced via the nucleophilic attack of the N lone pair to the activated triple bond of alkyne, which proceeded via intramolecular cyclization. Next, intermediate I was generated through the deprotonation of intermediate I. Finally, the desired product 170 was produced together with the regeneration of catalyst via the protonation and demetallation of intermediate J (Scheme 73).

It should be noted that the CuCN/[bmim][PF<sub>6</sub>] system was reused six times with no noticeable change in its performance. In the case of 169, phenylacetylene and phenylacetylene derivatives with various substituents such as 4-Me, 4-OMe, 4-CF<sub>3</sub>, 4-Br, 4-Cl, and 4-NO<sub>2</sub> provided the corresponding 3-aminoindolizines in 67-89% yield. However, only modest yields of the corresponding 3-aminoindolysines were isolated with aliphatic alkynes (70-71%), which is possibly due to the instability of the products, where they rapidly degraded during column chromatography. In addition, the use of heteroaryl alkynes was effective to obtain the corresponding 3-aminoindolizines in 82-84% yield. In the case of 168, moderate yields of the resulting 3-aminoindolizine derivatives were obtained with aliphatic acyclic amine derivatives (67-72%).

Cui et al.82 synthesized pyrrolo[2,1-a]isoquinolines 174 via the reaction among terminal alkyne derivatives 173, aldehydes 172, and tetrahydroisoguinolines 171 in the presence of CuCl<sub>2</sub> and PhCOOH (condensation promoter) in DMF at 130 °C (Scheme 74).

Initially, the Cu catalyst reacted with 173 and produced F. Meanwhile, two imine ions G and G' were generated via the reaction between 171 and 172. Then, amine H was obtained via the Mannich-type addition of activated alkyne F to ion G', which is more stable than the other ion. After that, H underwent oxidation in the presence of the Cu catalyst, and then its product was deprotonated to produce azomethine ylide. After that, intermediate I was generated via the 5-endo-dig cyclization of azomethine ylide. Lastly, 174 was obtained via the final proton transfer together with the release of the Cu catalyst (Scheme 75). It should be noted that the Cu catalyst may have played three roles in the reaction, as follows: (1) activation of 173 for Mannich-type addition, (2) producing an iminium ion via the oxidation of the tertiary amine, and (3) 5-endo-dig cyclization via the activation of propargyl amine.

According to the results, the corresponding isoquinoline derivatives were produced in 17-69% yield via a condensation/ Mannich-type addition/oxidation/cyclization cascade sequence. In the case of 171, with the exception of the NO<sub>2</sub> group, the usual electron-withdrawing and -donating groups can be compatible under the optimal conditions at the tetrahydroisoguinoline moiety such as MeO, OH, Br, and Cl (28–66% yield). The NO2 group dramatically deactivated tetrahydroisoquinoline and did not produce the corresponding product. In the case of 172, various substituted benzaldehydes were tested in the above-mentioned reaction and produced the desired isoquinoline derivatives in 17-69% yield. In general, derivatives with electron-withdrawing groups

Scheme 74 CuCl<sub>2</sub>-catalyzed synthesis of pyrrolo[2,1-a]isoquinolines 174.82

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Scheme 75 Possible reaction mechanism for the synthesis of pyrrolo[2,1-a]isoquinolines 174.82

performed better in terms of reaction yield, which is likely due to the easier condensation and better stability of the intermediates generated *in situ*. Also, when using 2,4-

dinitrobenzaldehyde, no corresponding product was detected, which is probably due to steric hindrance. All pyridyl, thienyl, and naphthyl substituents could be incorporated in the pyrrole

Scheme 76 (A) Synthesis of five-membered indolizine (scope for [3 + 2] cycloaddition) and (B) synthesis of seven-membered 1,4-oxazepine (scope for [5 + 2] cycloaddition).<sup>83</sup>

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[2,1-a]isoquinoline backbone in reasonable yields (31–33%). Moreover, a trace amount of the corresponding isoquinoline derivative was detected when using aliphatic aldehyde and the corresponding intermediate (37%). In the case of 173, the electron-rich and -deficient terminal alkynes were compatible in the present reaction (53-66%). Heterocyclic alkynes such as 2pyridyl, 3-pyridyl, and 2-thienyl also were tolerated and produced the corresponding isoquinoline derivatives in 31%, 46%, and 51% yield, respectively. In addition, alkyl-substituted terminal alkynes were tested and produced the corresponding products in 19-35% yield. Unfortunately, ethyl propiolate did not give the corresponding product, which may be related to the high reactivity of this alkyne.

#### 2.5 Reaction of α-diazoesters or ketones

He et al.83 synthesized seven-membered 1,4-oxazepine structures 180 and five-membered indolizine derivatives 178 in 34-93% yield via the rhodium-catalyzed regiodivergent [5 + 2] and [3 + 2] cycloaddition of quinolinium ylide derivatives 175 with alkyne derivatives (176 or 176), respectively, in  $CH_2Cl_2$  for [5+2]cycloaddition and xylenes for [3 + 2] cycloaddition at room temperature (Scheme 76).

Initially, rhodium carbene A was produced via the reaction between diazo compound 177 and Rh complex together with N<sub>2</sub> removal. Then, intermediate B was generated via the nucleophilic addition of 175 to A. After that, intermediates C and C' were produced via the dissociation of the rhodium salt. Next, intermediate D was obtained via the reaction between an alkyne and intermediate C, and this reaction was processed via 1,3dipolar [3 + 2] cycloaddition. Subsequently, 178 was produced via a metal-assisted 1,3-ester migration process. In addition, the

reaction between an alkyne and intermediate C' led to 180, which proceeded via 1,5-dipolar [5 + 2] cycloaddition (Scheme 77).

In the case of 180, various  $\alpha$ -diazoketone derivatives were investigated and produced the desired oxazepines in 51% to 96% yield. The types of groups (electron-donating and -withdrawing) on both aryl rings of the α-diazoketone derivatives produced the desired oxazepines in 51-96% yield. In addition, diazoketone derivatives containing a bulky 2-naphthyl group, thiophene, and furan gave the desired products in 61%, 56%, and 68%, yield, respectively. Diethyl acetylenedicarboxylate also produced the corresponding oxazepine in 87% yield. A variety of quinoline derivatives was evaluated, and the desired oxazepines were obtained in 34-93% yield, except for 8-fluoroquinoline, which could not produce the corresponding product. In the case of 178, various aryl diazoacetates were tested and produced the corresponding indolizines in 71–87% yield. Moreover, a diazo compound with thiophene was compatible in the transformation and produced the corresponding indolizine in 38% yield. Next, a variety of quinolines was tested and quinolone derivatives with various 6- and 7substituents produced the corresponding indolizines in 58% to 92% yield. Notably, 6-Br-7-fluoroquinoline produced the corresponding indolizine in 68% yield, but 8-fluoroquinoline did not produce the corresponding product in the abovementioned method.

Li et al.84 synthesized multi-substituted 1-arylaziridine-2carboxylates via the reaction among alkyne derivatives 181,  $\alpha$ diazoester derivatives 183 and nitrosoarenes 182 and the mixture was stirred in 1,4-dioxane at 50 °C for 24 h under catalyst-free conditions (Scheme 78).

Scheme 77 Possible reaction mechanism for the synthesis of seven-membered 1.4-oxazepines 180 and five-membered indolizines 178.83

$$R = \begin{array}{c} R' + Ph \\ 181 \end{array}$$

$$R = \begin{array}{c} R'' + Ph \\ 182 \end{array}$$

$$R''' = Cl, Me$$

$$R''' = Cl, Me$$

$$R''' = Cl, Me$$

$$R''' = Cl, Me$$

$$R'''' = Cl, Me$$

R'= H,  $CO_2Me$ ,  $CO_2Et$ ,  $CO_2Ph$ ,

Scheme 78 Synthesis of 1-arylaziridine-2-carboxylates 184.84

Firstly, intermediate **K** was formed from **183**. Then, intermediate **L** was generated via the reaction between **182** and intermediate **K**. After that, intermediate **M** was produced via the removal of  $N_2$  from intermediate **L**. Next, isoxazoline **N** was prepared from the reaction between **M** and **181** via [3 + 2] cycloaddition. Finally, the desired product cis-**184** or trans-**184** was generated via a Baldwin rearrangement. The important point

about this reaction is that the use of an ester group instead of R' and an aryl group instead of R led to the production of *cis'-184* due to steric hindrance in transition state **O**. Also, when using an H atom instead of R', two transition states (**P** and **Q**) were formed, where the *cis* form was dominant (Scheme 79).

The above-mentioned method provides mild conditions for the synthesis of different types of multi-functionalized

Scheme 79 Possible reaction mechanism for the synthesis of multi-substituted 1-arylaziridine-2-carboxylates 184.84

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aziridine derivatives in up to 91% yield. Also, this method has advantages such as the use of catalyst-free strategy and readily accessible and cheap starting materials. According to the screened internal and terminal alkynes, they all produced the desired products in 56-91% yield. Internal alkynes produced the corresponding products (as single diastereomeric) exclusively in 71-91% yield. The evaluation of ester moieties of phenylpropiolic acid ester (Me, Et, and Ph) demonstrated that steric hindrance affected the yield, where sterically hindered aryl group provided relatively lower yields (89%, 87%, and 73%, respectively). Also, ethyl 3-phenylpropiolate derivatives with an electron-donating group (Me at ortho, meta, and para position) resulted in higher yields than that having an electron-withdrawing group such as Cl (84-91% and 71-80% vield). In addition, the structure with two substituents (3,5dimethyl) on the phenyl ring gave the corresponding product produced in 75% yield, which showed that steric hindrance affected the yield. Furthermore, internal alkyne derivatives with a naphthyl group gave the corresponding products in 83% to 91% yield. Moreover, aliphatic terminal alkyne derivatives were evaluated and produced the corresponding cis isomers as the major products (56-82% yield) with moderate diastereoselectivity (up to 71:29 cis/trans). Unfortunately, ethynylcyclopropane and 2-ethynylthiophene gave the corresponding products in 78% and 83% yield with very low diastereoselectivity (50:50 cis/trans and 52:48 cis/trans), respectively.

Zhou et al. 85 synthesized tetrasubstituted pyrrole 188 via the reaction among alkynes 187, 1-substituted benzimidazoles 186, diazoacetates 185, and [Fe(TPP)Cl] in 1,4-dioxane for 12 h under an Ar atmosphere at room temperature (Scheme 80).

Initially, the reaction between 185 and 186 produced benzimidazolium N-vlide intermediate A in the presence of an iron catalyst. Then, intermediate B was produced via the reaction between 187 and intermediate. Finally, 188 was generated via fast isomerization of intermediate B, which was a thermodynamically favorable ring-opening aromatization process (Scheme 81).

The desired products were produced with an isolated yield of 70-90%. The above-mentioned method has advantages such as the use of mild reaction conditions, available starting materials, operational simplicity, and inexpensive catalyst.

Lv and co-workers86 synthesized (E)-beta-monofluoroalkyl- $\beta, \gamma$ -unsaturated esters (192 and 195) via the reaction among 1,1alkylmonofluoroalkylation of terminal alkyne derivatives 189, 2fluoro-1,3-dicarbonyl compounds (191 and 194), and diazo compounds (190 and 193) in the presence of CuI and K2CO3 in acetone at 80 °C for 2 h (Scheme 82).

Initially, Cu acetylide was produced via the reaction between 189 and CuI in the presence of K<sub>2</sub>CO<sub>3</sub>. Then, copper carbene K was generated via the reaction between 190 and Cu acetylide. Next, intermediate L was obtained from the migratory insertion of the alkynyl group into the K, followed by protonation of intermediate L to produce intermediate M. After that, intermediate N was produced by the rearrangement of intermediate M. Finally, the desired product 192 was produced via the reaction between intermediate N and dimethyl 2-fluoromalonate in the presence of K<sub>2</sub>CO<sub>3</sub>. It should be noted that the nucleophilic attack of the monofluoroalkyl group to the Ph from the opposite side was preferred to avoid steric interactions (Scheme 83).

In the case of 189, the reaction of dimethyl 2-fluoromalonate, 2-diazoacetate, and various terminal alkynes gave the

Scheme 80 Synthesis of multi-substituted pyrroles 188.85

Scheme 81 Possible reaction mechanism for the synthesis of tetra-substituted pyrrole 188.85

A)
$$R = \frac{189}{189} + \frac{N_2}{190} \underbrace{OEt}_{191} + \underbrace{MeO}_{F} \underbrace{OMe}_{191} \underbrace{Cul (10 \text{ mpl}\%)}_{Acctone, 80^{\circ}C, N_2} \underbrace{R + \underbrace{CF(CO_2Me)_2}_{CF(CO_2Me)_2}}_{CF(CO_2Me)_2}$$

$$R = Ph, 2-FPh, 2-CIPh, 3-FPh, 3-CIPh, 4-FPh, 4-CIPh, 4-BrPh}_{4-CO_2MePh, 2-OMePh, 4-MePh, 4-EtPh, 4-PrPh, 4-PuPh, 4-PuP$$

Scheme 82 Cul-catalyzed synthesis of (E)- $\beta$ -monofluoroalkyl- $\beta$ , $\gamma$ -unsaturated esters or ketones: (A) scope of alkynes in the 1,1-alkylmonofluoroalkylation and (B) scope of diazo compounds and monofluoroalkylation reagents in 1,1-alkylmonofluoroalkylation.<sup>86</sup>

.COOMe `COOMe COOMe

`COOMe

corresponding  $\beta$ , $\gamma$ -unsaturated products in 35–91% yield. Also, para, ortho, and *meta*-fluoro, chloro ethynylbenzenes under the 1,1-alkylmonofluoroalkylation reaction produced the desired products in the same yields, which showed that steric hindrance had no effect. 4-Br and 4-COOMe ethynylbenzenes were reacted and produced the corresponding products in 76% and 70% yield, respectively. In addition, alkyne derivatives with electron-donating groups on the aromatic rings gave the corresponding  $\beta$ , $\gamma$ -unsaturated products in 70% to 85% yield. Alkynes with a pyridine or thiophene moiety gave the desired  $\beta$ , $\gamma$ -unsaturated in 45–77% yield. *N*-propargyl amides, conjugated enynes, and alkyl propargyl ethers gave the desired products in 70–75% yield. Moreover, the non-activated aliphatic alkynes produced the corresponding  $\beta$ , $\gamma$ -unsaturated products

in 35% to 40% yield. Further, the citronellol-based alkyne derivative,  $_{D,L}$ - $\alpha$ -tocopherol-based alkyne derivative, and purine-based alkyne derivative produced the desired  $\beta$ , $\gamma$ -unsaturated products in 85%, 70%, and 81% yield, respectively. Diethyl 2-fluoromalonate produced the desired product in 78% yield. Also, 2-F-1-morpholinobutane-1,3-dione and N,N-diethyl-2-F-3-oxobutanamide were tested and produced the desired products in 70% and 56% yield, respectively. Interestingly, ethyl 2-fluoro-3-oxobutanoate reacted with ethynylbenzene and ethyl 2-diazoacetate and produced the desired  $\beta$ , $\gamma$ -unsaturated product in 70% yield. In addition, 2-F-acetoacetamides with different groups on the aromatic rings were converted to the desired  $\beta$ , $\gamma$ -unsaturated products in 68% to 81% yield. Importantly, when  $K_2$ CO<sub>3</sub> was replaced with  $K_3$ PO<sub>4</sub>, the corresponding products

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Scheme 83 Possible reaction mechanism for the synthesis of (E)-beta-monofluoroalkyl-β,γ-unsaturated esters.86

were produced in 52% to 78% yield. The reaction of ethyl 2diazoacetate compared to 2-diazo-1-phenylethanone and methyl aryldiazoacetates was higher under identical conditions. When acetone was replaced with THF, methyl aryldiazoacetates were suitable substrates and produced the desired  $\beta$ , $\gamma$ -unsaturated products in 39-40% yield. Finally, 2-diazo-1phenylethanone was transformed into the 1,1-alkylmonofluoroalkylation products (Scheme 82B(a) and (b)) in 1,4dioxane with a total yield of 35%.

# 2.6 Reaction of isocyanides

Pelliccia et al.87 synthesized iminofurans 199 via an isocyanidebased three-component reaction with the reactants of diethyl bromomalonate 196 and phenylacetylene derivatives 197 in the presence of fac-Ir(ppy)<sub>3</sub> as the photoredox catalyst and Na<sub>2</sub>CO<sub>3</sub> as the base in MeCN under irradiation with a 30 W blue LED at room temperature for 20 h (Scheme 84).

Initially, the Ir<sup>3+</sup>\* photocatalyst was generated from fac-Ir(ppy)3, which underwent metal-to-ligand charge transfer under visible-light irradiation. Then, alkyl radical A was formed via single-electron transfer (SET) between Ir<sup>3+\*</sup> and 196. Next, vinyl radical B was generated via the reaction between A and 197, followed by the reaction of 198 with B, which led to the formation of imidoyl radical intermediate C. After that, Ir<sup>4+</sup> oxidized this key species to nitrilium ion D. Next, amide E was prepared from **D** in the presence of  $H_20$ . In the following step, 199 was generated from E, which underwent oxidative cyclization (Scheme 85).

Investigation of isocyanides in the above-mentioned reaction demonstrated that a variety of groups on the phenyl ring could produce the corresponding products in 35-54% yield. In contrast, aliphatic isocyanides such as cyclohexylisocyanide and tert-butylisocyanide did not produce the corresponding products. Moreover, the evaluation of phenylacetylene derivatives showed that neither the position of the aromatic ring substituents nor their electronic nature significantly affected the reaction yield (23-60%) except when bromo phenylacetylene was tested, where the reaction yield dramatically decreased (23%). In the case of aliphatic derivatives of alkynes, the use of these derivatives was not possible due to them not being reactive at all.

Gao and co-workers<sup>88</sup> synthesized captodative olefins (203 or 207) via the palladium-catalyzed (Pd(OAc)<sub>2</sub>) three-component reaction of alkynes (200 or 204), carboxylic acids (206 or 202), and isocyanides (201 or 205) in the presence of Ag<sub>2</sub>O as a silver salt, and  $P(o-tol)_3$  as a ligand in PhCl under  $N_2$  at 30–50 °C for 5 h (or 70 °C for 5-15.5 h) (Scheme 86).

Initially, the reaction among 200, 206, and Ag<sub>2</sub>O produced silver acetylide K together with the formation of silver carboxylate. Then, complex L was obtained via the attack of the carboxylic anion on K, which subsequently caused the transmetalation process with Pd(II). Then, imidoyl palladium(II) intermediate N was generated via the isocyanide migratory insertion reaction to vinyl palladium complexes M. Finally, via the successive reductive elimination process and treatment with H<sub>2</sub>O 270 or 203 was produced (Scheme 87).

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Scheme 84 Fac-Ir(ppy)<sub>3</sub>-catalyzed synthesis of iminofuran derivatives 199.87

Scheme 85 Possible reaction mechanism for the synthesis of iminofurans 199.87

A variety of alkyne, carboxylic acid, and isocyanide derivatives was investigated. According to the results, aryl acetylenes bearing different substitutions (alkyl, alkoxy, halides, acyl, nitro, and amide) produced the corresponding olefins in 48% to 90% yield. Interestingly, when phenylpropiolic acid was used instead of phenylacetylene, the corresponding olefin was

produced in 87% yield. In addition, thienyl alkyne gave the corresponding olefin in 44% yield, and the enyne substrate also could be transformed into the corresponding olefin in 57% yield. Also, alkyl acetylenes (<sup>4</sup>Bu and cyclopropane) produced the corresponding olefins in 70% and 67% yields, respectively (Scheme 86A). A variety of acids and isocyanides was tested in the above-mentioned reaction. It was found that for other carboxylic acids, higher temperature was more favorable and produced the corresponding olefins in 47% to 86% yield (Scheme 86B). Moreover, 1-adamantyl and cyclohexyl isocyanide were tested and produced the corresponding olefins in 44% to 79% yield, respectively. Interestingly, when α-acidic isocyanides such as isocyanoacetates were used, no desired products could be observed.

Singh and co-workers\*9 reported a novel method for the synthesis of tetrahydroquinoline-fused tetracyclic derivatives 2a. This method was performed in two steps. In the first step, aniline derivatives 209, benzaldehyde derivatives 208, propiolic acid derivatives 211, and isocyanide derivatives 210 were stirred in MeOH at room temperature and produced derivatives 1a. Then derivatives 1a were reacted with molecular iodine (electrophile) and produced the corresponding derivatives 2a (Scheme 88).

Initially, intermediate **A** was generated *via* the activation of a triple bond of alkyne ( $\pi$  activation) by the iodonium ( $I^{\dagger}$ ) ion. After that, the regioselective spirocyclic carbocation intermediate **B** was obtained *via* the ipso nucleophilic attack of the aniline to the activated triple bond. In this step, it should be noted that this process was done *via* a 5-endo-dig method. Finally, 2a was produced *via* the immediate trapping of the carbocation by the intramolecular nucleophile. In this step, it should be noted that this process was done through the Friedel–Crafts-type cyclization (Scheme 89).

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$$\mathbf{R}$$
= tBu,  $\mathbf{R}'$ 

R'= H, 4-Me, 4-OMe, 4-Cl, 4-Br 4-Ac, 3-F, 3-NO<sub>2</sub>, 3-AcNH, 3-Me, 2-Cl, 2-Me

R= H, 4-Cl, 2-OMe, 4-OMe R'= Bu, Ad, cyclohexyl

R"= Me, cyclohexyl,

R"'= H, F, OMe

Scheme 86 Synthesis of captodative olefins: (A) substrate scope of alkyne derivatives and (B) substrate scope of isocyanides, alkynes, and carboxylic acids.88

$$2 \text{ R"COOH} + \text{Ag}_2\text{O} \longrightarrow 2 \text{AgO}_2\text{CR"} \longrightarrow R \text{"COOH}$$

$$R \text{"CO$$

Scheme 87 Possible reaction mechanism for the synthesis of captodative olefins.88

The Ugi adducts were produced in 67-91% yield and were compatible with substituted phenyl propiolic acid derivatives, which transformed to the corresponding tetrahydroquinolines in 64% to 87% yield. Also, in this transformation, substituent electronic factors had an influence on the phenyl propiolic acid. The presence of an electron-donating group on phenyl propiolic acid produced the required products in better yields (71-87%) compared to fluorine-substituted adducts (58-64%). The Ugi adducts derived from 2,5-dimethoxy and 3,4,5-trimethoxy aldehyde reacted with the various isocyanide derivatives and produced the corresponding products in 71% to 76% yield. Furthermore, functionalized phenyl propiolic acid, aldehyde and cyclohexylisocyanide-derived Ugi adduct produced the corresponding products in 58% to 77% yield. It should be noted that the aliphatic acid derivatives having Ugi adducts transformed the corresponding products in 69% to 71% yield. In addition, a variety of aromatic and aliphatic substituted isocyanides was tested and produced the corresponding cyclized products in 52% to 83% yield. Among the aliphatic isocyanide derivatives, only *n*-pentyl isonitrile showed the minimum yield. Moreover, the aromatic isonitrile-based Ugi adduct produced the corresponding products in 56% and 62% yield.

Yang and co-workers% synthesized perfluoroalkyl-containing pyrano[3,4-c]pyrroles 215 via the reaction among methyl perfluoroalk-2-ynoate derivatives 214, 3-aroyl (or heteroyl)

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**R**= 2,5-diOMe, 2,5-diMe, 3,4,5-triOMe

R'=H, 3-Me

R"= <sup>t</sup>Bu, EtOOC-CH<sub>2</sub>-, 1,1,3,3-tetramethylbutyl,cyclohexane, 2,6-dimethylphenyl

R""= H, Me, F, OMe

Scheme 88 Synthesis of tetrahydroquinoline-fused tetracyclic heterocycles 2a.89

Scheme 89 Possible reaction mechanism for the synthesis of tetrahydroquinoline-fused tetracyclic derivatives 2a.89

acrylic acid derivatives **212**, and isocyanide derivatives **213** in MeCN at room temperature for 12 h (Scheme 90).

Initially, zwitterion **J** was obtained *via* the Michael addition of **213** to **214**. Then, **J** was protonated by **212** to give aminium **K** and carboxyl anion **L**. After that, isoimide **M** was generated from the reaction between intermediate **K** and **L** *via* a Passerini-type reaction. Next, **M** produced conformers **N** and **N**' *via* the Mumm rearrangement. Finally, the exo isomer was generated conformer **N** as the major diastereoisomer *via* the intramolecular oxo-Diels–Alder reaction. It should be noted that the endo isomer was obtained as the minor product due to the steric hindrance between  $\mathrm{CO}_2\mathrm{Me}$  and the R group in conformer **N**' (Scheme 91).

Various R-substituted acrylic acid derivatives were evaluated and acrylic acid derivatives with different types of heteroaryl or aryl groups at the terminal position produced the desired perfluoroalkyls in 45% to 97% yield with high diastereoselectivity (up to >99:1 dr). Generally, an electron-withdrawing group on the acrylic acid phenyl ring afforded the corresponding perfluoroalkyls in higher yields compared to an electron-donating substituent at the same position. Substrates with  $n-C_3F_7$ , CF<sub>2</sub>H, and C<sub>2</sub>F<sub>5</sub> were tested and produced the corresponding perfluoroalkyls in 45% to 93% yield, which showed little detrimental effect on the yields and had no adverse influence on the diastereoselectivity. Next, various isocyanides were investigated and the corresponding perfluoroalkyls produced in 32% to 98% yield (up to >92:1 dr). A summary of the reaction of alkynes with isocyanide derivatives, and also the limitations and progress of the reactions are presented in Table 3.

The reaction among isocyanides, alkynes and fullerenes was reviewed by Li *et al.*<sup>91-93</sup> Investigating the types of alkynes in this type of multicomponent reaction requires a more detailed investigation to discover a way to carry out the reaction of alkynes that are not involved in the reaction. Even checking the position of C–H alkynes and activating this part to be involved in the reaction need to be investigated. The yield of the obtained products is also low, which is a limitation in this field and requires more extensive research in this type of reaction. Some recent developments in this field are shown in Table 4.

## 3 C-H activation

#### 3.1 Synthesis of propargylamines

Yan and co-workers<sup>96</sup> reported a green method for the synthesis of propargylamine derivatives **219** *via* the reaction among alkynes **218**, aldehyde derivatives or ketone derivatives **216**, and amines **217** in the presence of PS-PEG-BPy-CuBr<sub>2</sub> as a catalyst

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OH + R'-N=C + R" 
$$=$$
 CO<sub>2</sub>Me  $=$  MeCN, rt 12 h  $=$  CO<sub>2</sub>Me  $=$  R= C6H5, 4-MeOC<sub>6</sub>H<sub>4</sub>, 4-MeC<sub>6</sub>H<sub>4</sub>, 4-FC<sub>6</sub>H<sub>4</sub>, 4-ClC<sub>6</sub>H<sub>4</sub>, 4-BrC<sub>6</sub>H<sub>4</sub>,  $=$  215

4-NO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>, 3-MeC<sub>6</sub>H<sub>4</sub>, 3-CF<sub>3</sub>C<sub>6</sub>H<sub>4</sub>, 2-BrC<sub>6</sub>H<sub>4</sub>, 2-Thienyl, 2-puryl, R'= t-Butyl, Cyclohexyl, 4-NO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>, 4-MeOC<sub>6</sub>H<sub>4</sub>, 2,6-dimethylC<sub>6</sub>H<sub>4</sub>  $R''=CF_3$ ,  $CF_2H$ ,  $C_2F_5$ ,  $n-C_3F_7$ 

Scheme 90 Synthesis of perfluoroalkyl derivatives 215 without a catalyst. 90

Scheme 91 Possible reaction mechanism for the synthesis of perfluoroalkyl-containing pyrano[3,4-c]pyrroles.90

and N<sub>2</sub> at 110 °C under solvent-free conditions (Scheme 92). The desired propargylamines were produced in 66-97% yield. Several alkynes were investigated and the results showed that aromatic alkynes with p-trifluoromethyl or p-methyl substituents produced the corresponding products in 99% and 98% yield, respectively. When the aromatic alkyne has a methyl substitution on the ortho position, it gave the corresponding product in 94% yield. Furthermore, aliphatic acetylenes were used and the desired propargylamines were produced in yields of 79% and 74%.

Propargylamines 223 were synthesized by Dhanasekaran and co-workers97 via the reaction among 2-ethynyl anilines 222, amines 221, and substituted benzaldehydes 220 in the presence of a Cu<sup>I-i</sup>PrpyboxdiPh complex and N-Boc-(L)-proline as the catalyst system in DCE at room temperature (Scheme 93). According to the results, N-substituted 2-ethynyl aniline derivatives gave the desired propargylamines in high yields (up to 87%). N-Tosyl-2-ethynyl aniline produced the products in low enantioselectivities with respect to N-substituted 2-ethynylaniline derivatives. Also, 2-ethynyl anilines with various substituents on the aromatic ring produced the desired products with high levels of enantioselectivities (up to 95% ee) in good yields (up to 87%).

Milen et al. 98 synthesized propargylamines 227 via the reaction among acetylenes 225, aldehydes 224, and secondary amines 226, where they studied twelve readily available coppercontaining mineral derivatives. It should be noted that the **RSC Advances** Review

Table 3 Summary of the reaction of alkynes with isocyanides and limitations or progress of the reactions

	Reaction							
Reactants	Product	Catalyst	Solvent	Method employed	time	Yield	Limitations or progress	Ref.
$\begin{array}{c c} R & NH_2 \\ \hline & OH \\ Br & C & \\ \hline & \\ & \\ \end{array}$	R II N N N N N N N N N N N N N N N N N N	Pd(PPh <sub>3</sub> ) <sub>2</sub> Cl <sub>2</sub>	1,4- Dioxane	PPh <sub>3</sub> , Cs <sub>2</sub> CO <sub>3</sub> , 80 °	2 h	47- 94%	2-Amino-3-pyridinol and 2- aminothiophenol did not react	Liu et al. <sup>94</sup>
R" OH O'S An CN + R	R''	A —	DMSO/ H <sub>2</sub> O	Method 1: pyridine, LED, $O_2^a$ , RT Method 2: pyridine, UV, $O_2^a$ , RT	Method 1: 16 h Method 2: 24 h	Up to 73%	(1) High selectivity (2) Made new EDA complexes from arylsulfinate anions	Li et al. <sup>95</sup>

<sup>&</sup>lt;sup>a</sup> Investigation of scope of arylacetylenes under N<sub>2</sub> (UV and LED method = 24 h).

Table 4 A summary of the reaction of alkynes with isocyanides and fullerene and limitations or progress of the reactions

Reactants	Product	Catalyst Solvent	Method Reaction employed time		Limitations or progress	Ref.
R + F	R'HNOC R'	<i>o-</i> Dichlorobenzene	Air, 120° e C 1 h	27- 30%	(1) One-step reaction for synthesis of a static solid-state cluster in an NCF (2) This research proposed a vision about potential regioselectivity in further reactions and advances in quantum information materials	
R C <sub>60</sub> + tBuNO	tBu CO <sub>2</sub> Me N R  + NtBu C MeO <sub>2</sub> C R	<i>o-</i> Dichlorobenzene	$N_2$ , 80 ° $C^a$ or 120 1 h $C^b$	4- 36%	(1) Alkynes containing a substituent and an ester group on one side provided resonance and inductive effects on the other side of the triple bond (ester group has key role) (2) The 4-NO₂-Ph substituent at both ends of the C≡C triple bond of did not show similar results	•

<sup>&</sup>lt;sup>a</sup> Used in the reaction between fullerenes, alkyl isocyanides, and DMAD. <sup>b</sup> Used in the reaction between fullerenes, alkyl isocyanides, and unsymmetric alkyne substrates.

reaction was performed in the presence of chrysocolla, brochantite, azurite, and malachite in PhMe under reflux for 20 h and the corresponding propargylamines were obtained in 36% to 96% yield (Scheme 94). Firstly, various benzaldehyde derivatives with phenylacetylene and morpholine were tested and the results showed that aldehydes with either electron-donating or -withdrawing groups on the phenyl ring produced the desired

propargylamine derivatives in 21% to 92% yield. Also, the reaction of 9H-fluorene-2-carbaldehyde and naphthalene-2carbaldehyde produced the desired propargylamines in 87% and 91% yield, respectively. Furthermore, heteroaromatic carbaldehyde derivatives with a furanyl, thiazolyl, pyridinyl or thiophenyl moiety produced the corresponding propargylamines in 78% to 94% yield. When 1-methylpyrrol-2-carboxaldehyde and 5Review

**R**= H, Ph, 4-Me-Ph, 4-Cl-Ph, Cyclohexane. Pentan

R'''= Ph, 4-Me-Ph, 4-CF<sub>3</sub>-Ph, 2-Me-Ph, SiMe<sub>3</sub>, -5

Scheme 92 Green synthesis of propargylamine derivatives 219 via A3 coupling reactions. 96

Scheme 93 Synthesis of propargylamines 223 via a chiral Cul-box complex with 2-ethynyl anilines. 97

fluoro-1-benzothiophene-2-carbaldehyde were used as the starting materials, lower yields were obtained (36% and 51%, respectively). In addition, aliphatic aldehyde heptanal produced the corresponding propargylamine in 88% yield. Subsequently, the reaction of benzaldehyde and morpholine with 4-substituted phenylacetylenes was evaluated. 4-Methylphenylacetylene showed good reactivity and produced the corresponding propargylamine in 93% yield. However, 4-nitrophenylacetylene gave the corresponding product in 47% yield. Furthermore, the reaction could be performed with aliphatic acetylene derivatives (dec-1-yne and oct-1-yne) and produced the corresponding propargylamines in 77% and 83% yield, respectively. Next, secondary amine derivatives were also investigated and produced the corresponding propargylamines in 59% to 96%

yield. Dibutylamine was also reacted with phenylacetylene and benzaldehyde to produce the corresponding propargylamine in 71% yield. Importantly, an attempted extension of the reaction to aniline, with phenylacetylene and benzaldehyde, led to the formation of a Schiff base.

Thakur and Khare99 synthesized stereoselective chiral propargylamines 231 via the reaction among heterocyclic amine derivatives 229, aromatic aldehyde derivatives 228, deoxy sugar based alkyne 230 in the presence of CuI, and L-proline (L) in  $CH_3C_6H_5$  under reflux (Scheme 95).

Initially, complex A was generated via the reaction between CuI and 230. Then, 228 and 229 reacted and produced intermediate aminal B via the Aldol-type condensation. Next, A and B reacted together and produced the complex C. Then, the

**R**= Ph, 2-F<sub>3</sub>CC<sub>6</sub>H<sub>4</sub>, 3-BrC<sub>6</sub>H<sub>4</sub>, 4-F<sub>3</sub>CC<sub>6</sub>H<sub>4</sub>, 4-NCC<sub>6</sub>H<sub>4</sub>

4-i-PrC<sub>6</sub>H<sub>4</sub>, 4-MeOC<sub>6</sub>H<sub>4</sub>, 4-PhOC<sub>6</sub>H<sub>4</sub>, 2,3-diF-C<sub>6</sub>H<sub>3</sub>

naphthalen-2-yl, 9H-fluoren-2-yl, thiophen-2-yl, 5-methylfuran-2-yl

1,3-thiazol-4-yl, pyridin-3-yl, 5-F-benzothiophen-3-yl,

1-Me-1H-pyrrol-2-yl, hexyl

 $\mathbf{R'}=\text{Ph}$ ,  $4-\text{MeC}_6\text{H}_4$ ,  $4-\text{O}_2\text{NC}_6\text{H}_4$ , hexyl, octyl,

 $R''R''' = -(CH_2)_2O(CH_2)_2$ -,  $-(CH_2)_4$ -,  $-(CH_2)_5$ -,  $-(CH_2)_6$ -,  $-(CH_2)_2NMe(CH_2)_2$ -,

Buf

Scheme 94 Synthesis of propargylamine derivatives via Cu-containing minerals.98

HO 
$$\frac{1}{128}$$
 H +  $\frac{R}{H}$  +  $\frac{R}{AcO}$  OAc  $\frac{H}{H}$  Cul  $\frac{L-proline}{Toluene, reflux, 6-8 h}$  R  $\frac{R}{AcO}$  OAc  $\frac{H}{H}$  OAc  $\frac{R}{AcO}$  OAc  $\frac{H}{H}$  OAc  $\frac{R}{AcO}$  OAc  $\frac{R}{H}$  OAc  $\frac{R}$ 

R=H, OAc

**R'**= H, OAc

228= 2-hydroxybenzaldehyde, 3-hydroxybenzaldehyde, 4-hydroxybenzaldehyde

229= Morpholine, Pyrrolidine, Piperidine

Scheme 95 Cul-catalyzed synthesis of deoxy sugar-based chiral propargylamine derivatives.<sup>99</sup>

Scheme 96 Possible reaction mechanism for the synthesis of stereoselective chiral propargylamines.<sup>99</sup>

coordinated alkyne in the complex C was deprotonated together with the removal of  $H_2O$ . After that, complex **D** was produced *via* the ligation of L-proline (L) with complex C, which gave Cuacetylide with a coordinated iminium ion at the end. Finally, 231 was generated via the addition of Cu-acetylide with a coordinated iminium ion together with the regeneration of the catalyst and L (Scheme 96).

According to the results, when ortho/meta-substituted benzaldehyde derivatives reacted with 3,4,6-tri-O-acetyl-2-deoxy propargyl-O-α-D-glucopyranoside/3,4,6-tri-O-acetyl-2-deoxy propargyl-O-α-D-galactopyranoside and piperidine/morpholine/ pyrrolidine the desired propargylamine derivatives were produced in 68% to 86% yield. Under similar conditions, 4hydroxy benzaldehyde compounds decreased the percentage vield of the corresponding products in comparison with other synthesized compounds (30-65%).

Martinez-Amezaga<sup>100</sup> synthesized 1-substituted propargylic tertiary amines 235 via the three-component reaction of alkynes 234, amines 233, and aldehydes or 2-oxoacid derivatives 232 in the presence of Cu(OTf)<sub>2</sub> in toluene at 120 °C for 6 h (Scheme 97). When the reaction condition for the synthesis of 235 was done under microwave heating for 1 h, a mixture of 235 and 1,3disubstituted allene A was observed (Scheme 97).

The synthesis mechanism of A could have occurred as follows: initially, 235 was generated, followed by the coordination of Cu to the triple bond. After that, this process caused 1,5hydride transfer and  $\beta$ -elimination, which led to the synthesis of A (Scheme 98).

According to the study, propargylamines were obtained in 9% to 99% yield. In addition, propargylamines were obtained in 50% to 65% yield using phenylglyoxylic acid. Also, ethynyltrimethylsilane as the alkyne component in the abovementioned method gave the corresponding propargylamine in 49% yield. Also, Sonogashira coupling was performed using

Scheme 98 Possible reaction mechanism for the synthesis of 1substituted propargylic tertiary amines. 100

tetrakis(triphenylphosphine)palladium(0) as the catalyst and aryl iodide, giving the corresponding product in 79% yield.

Neofotistos and co-workers<sup>101</sup> synthesized tetrasubstituted propargylamines 239 via the reaction among amine derivatives 237, alkyne derivatives 238, and ketone derivatives 236 in the presence of MnBr<sub>2</sub> as a metal source under neat conditions at 130 for 20 h (Scheme 99). In the case of 236, the mixture of cyclopentanone, piperidine, and phenylacetylene produced the corresponding propargylamine in 67% yield. Cycloheptanone produced the corresponding product in 73% yield. In addition, when cyclohexanone or cyclopentanone was reacted with pyrrolidine produced, the desired propargylamines were obtained in 91% and 75% yield, respectively. In contrast, when

R'''=H, Ph, 4-MePh,  $C_5H_{11}$ ,  $C_8H_{17}$ ,  $C_7H_{15}$ , ethanol

Scheme 97 Cu(OTf)<sub>2</sub>-catalyzed synthesis of propargylamine derivatives. 100

R'= H, 3-Me, 4-Me, 4-OMe, 4-Cl, 2-Br, 4-CF<sub>3</sub>, 4-OMe-2-Me

Scheme 99 MnBr<sub>2</sub>-catalyzed synthesis of propargylamine derivatives.<sup>101</sup>

cyclohexane-1,2-dione was used, the corresponding propargylamine was not obtained. Linear ketones such as 3-pentanone, 2-pentanone, and 4-decanone decreased the yields (48%, 38%,

and 68%, respectively). Also, 4-methoxy-acetophenone and 4-chloro-acetetophenone produced the corresponding propargylamines in 34% and 42% yield, respectively. When

Scheme 100 Synthesis of propargylamine derivatives using glyoxylic acid monohydrate.<sup>102</sup>

benzophenone was used, the corresponding propargylamine was not observed, which may be due to the increased steric hindrance imparted from the two phenyl moieties in the alphaposition of the carbonyl group. In the case of 238, cyclohexanone with pyrrolidine or piperidine was coupled with a variety of terminal alkyne derivatives and the corresponding propargylamines obtained in 57% to 89% yield. 3-Ethynyl-toluene was coupled with piperidine and cyclohexanone and produced corresponding propargylamine in 81% yield. pMethoxyphenylacetylene reacted was with piperidine and cyclohexanone and produced the corresponding propargylamine in 87% yield. Interestingly, the reaction and nornicotine, 2-methylbut-3-yn-2-ol, and cyclohexanone gave the corresponding product in 81% yield, which was produced as a single diastereoisomer. In addition, 4-chloro-ethynylbenzene was reacted with cyclohexanone and pyrrolidine and produced the corresponding propargylamine in 77% yield, and when piperidine replaced pyrrolidine, the corresponding product was

Scheme 101 Possible reaction mechanism for the synthesis of propargylamines. 102

Table 5 Summary of the synthesis of propargylamine derivatives under metal-free conditions

Reactants	Product	Catalyst	Solvent	Method employed	Reaction time	Yield	Limitations or progress	Ref.
CHO  R  X  H  N  H	R <sub>U</sub> Ar	Nano-colloidal silica@APTPOSS	Toluene	Microwave irradiation, 50 °C	20–25 min	Up to 96%	<ol> <li>(1) Efficient catalyst</li> <li>(2) Recoverability of the catalyst.</li> <li>(3) Low catalyst loading</li> <li>(4) Short reaction times.</li> <li>(5) A simple procedure</li> <li>(6) High atom economy</li> </ol>	Safaei- Ghomi et al. <sup>103</sup>
R' $HN$ $X$ $R'$ $HN$ $X$	$R \xrightarrow{\text{II}} N \xrightarrow{N}_{n} X$	_	Solvent-free	80–90 °C	4–8 h	82- 90%	(1) Metal-catalyst- free (2) Solvent-free conditions	Ghosh et al. 104
$ \begin{array}{ccc} R & = & \\ O & + & N \\ Ar & H \end{array} $	N Ar	· _	Magnetized water	MW, 60 °C	110–160 min	65- 90%	(1) Simplicity (2) Low cost (3) Short reaction time (4) High reaction yield (5) Easy work-up (6) Absence of hazardous organic solvents	Bakherad et al. <sup>105</sup>
$ \begin{array}{ccc} R & & \\ X & & \\ K & & \\ K & & \\ K & & \\ X & $	N N	_	DCM (play solvent and reactant role)	70 °C	12 h		(1) Simplicity of operation (2) Mild conditions (3) High functional group tolerance	Rawat et al. <sup>106</sup>

produced in 64% yield. 2-Br-phenylacetylene was coupled with piperidine and cyclohexanone and produced the corresponding propargylamine in 57% yield. Furthermore, when ptrifluomethyl-phenylacetylene was used, the corresponding product was obtained in low yield. Finally, 1-octyne reacted well with piperidine and cyclohexanone and produced the corresponding propargylamine in 70% yield. In the case of 237, various amine derivatives were reacted with cyclohexanone and phenylacetylene. When morpholine was used, the corresponding product was obtained in 94% yield. Also, benzylamine produced the corresponding propargylamine in 46% yield. In addition, when p-methoxybenzylamine was used, it gave the corresponding propargylamine in 43% yield. Aliphatic amines were also investigated. N-Octylamine reacted with phenylacetylene and cyclohexanone and produced the desired propargylamine in 67% yield. Also, cyclohexylamine gave the corresponding propargylamine in 58% yield. In addition,

secondary amine di-n-propylamine produced the desired propargylamine in 61% yield. Finally, the corresponding product of p-methoxy-aniline decomposed when exposed to light.

Huang *et al.*<sup>102</sup> synthesized propargylamines **243** *via* the reaction among alkyne derivatives **242**, glyoxylic acid monohydrate **241**, and secondary amine derivatives **240** in DCE at 95  $^{\circ}$  C for 12 h (Scheme 100).

Scheme 104 Synthesis of alkynoates. 108

$$\begin{array}{c|c}
C & & & & & & \\
\hline
247 & & & & & \\
R & & & & & \\
\hline
R & & & & & \\
\hline
WO_4^2 & & & & \\
\hline
Q & & & & \\
\hline
O & & & \\
\hline
O & & & \\
\hline
O & & & \\
\hline
O &$$

Scheme 103 Possible reaction mechanism for the synthesis of alkyl 2-alkynoates. 107

Scheme 105 Possible reaction mechanism for the synthesis of alkynoates. 108

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Initially, iminium salt species **A** was generated via the reaction between **240** and **241** together with H<sub>2</sub>O removal. Then, the sp carbon of alkyne was activated by the carboxyl group of intermediate **A** and produced the coupled product **C** via the subsequent Michael addition. Finally, **243** was produced via the further conversion of **C** (Scheme 101).

In the case of 240 and 242, the yield increased when the steric bulk on the amine substrate increased, which may be due to intermediate A. All N-alkyl substituted benzylamine derivatives were tested and produced the desired propargylamines in 78-93% yield except for the product NH(Bn)<sub>2</sub> (21%). N-tert-Butyl benzylamines containing various groups on the phenyl moiety were reacted and produced the corresponding propargylamines in 78% and 93% yield, respectively. Moreover, secondary aliphatic amine derivatives were also reacted and produced the corresponding propargylamines in 54-79% yield. When the primary and cyclic amine derivatives were tested, only morpholine gave a low yield of the corresponding propargylamine (13%). Next, phenylacetylenes bearing electron-neutral groups such as phenyl and electrondonating substituents such as Me, tBu, and OMe produced the corresponding propargylamines in 89%, 72%, 78%, and 82% yield, respectively. Also, phenylacetylene derivatives with electronwithdrawing groups such as CF3, Cl, and F produced the corresponding propargylamines in 65-69% yield. In addition, aliphatic alkyne derivatives were investigated and produced the corresponding propargylamines in 72% to 83% yield. Importantly, methyl propiolate and prop-2-yn-1-ol, 3-chloroprop-1-yne were tested, but no desired products were produced.

A summary of the synthesis of propargylamine derivatives under metal-free conditions, and also the limitations and progress of the reactions are presented in Table 5.

#### 3.2 Reaction of CO<sub>2</sub>

Guo *et al.*<sup>107</sup> synthesized alkyl 2-alkynoates **247** (24–96%) *via* the reaction among terminal alkyne derivatives **245**,  $CO_2$  **244**, and alkyl halides **246** in the presence of  $Ag_2WO_4$  (cat.) and  $Cs_2CO_3$  (base) in DMF at room temperature for 24 h (Scheme 102).

Initially, Ag-acetylide **A** was generated *via* the coordination of Ag and **245** in the presence of the base. Then, simultaneously  $WO_4^{2-}$ - $CO_2$  adduct **B** was obtained *via*  $CO_2$  activation by tungstate. After that, Ag-phenylpropiolate **C** was produced *via* the insertion of **B** into the sp-hybridized Ag–C bond together with the release of  $WO_4^{2-}$ . Finally, **247** was generated from **C** using *n*-BuI (Scheme 103). It should be noted that the type of solvent in this transformation played an important role, where DMF showed the best result among the solvents.

Phenylacetylenes containing electron-withdrawing and electron-donating groups were carboxylated and produced the desired alkyl 2-alkynoate derivatives in 75–91% yield. 1,4-Diethynylbenzene generated the biscarboxylative and monocarboxylative products. Between them, the biscarboxylative product was the major product and the monocarboxylative product was the minor product (64% and 24% yield, respectively). In this transformation, heteroaromatic acetylenes and aliphatic alkyne derivatives also were tolerated and produced the desired alkyl 2-alkynoate derivatives. This method has

advantages such as use of a halogen-free catalyst and simple preparation of the catalyst.

Xie *et al.*<sup>108</sup> synthesized alkynoates **251** *via* the reaction among terminal alkyne derivatives **249**,  $CO_2$  **248**, and *n*-BuI **250** in the presence of L (cat.) and  $Cs_2CO_3$  in DMF at room temperature for 12 h (Scheme 104).

Initially, Cu-acetylide **B** was generated from activated alkyne **A**, in the presence of Cs<sub>2</sub>CO<sub>3</sub>. Then, propynoate intermediate **C** was produced from **B** *via* the insertion of CO<sub>2</sub> into the sphybridized Cu–C. Finally, **251** was obtained from intermediate **C** in the presence of **250**. According to this report, the ligand causes the reaction to proceed under mild conditions, and also advances the efficiency of the catalyst (Scheme 105).

Phenylacetylene derivatives **249** containing electro-rich groups at the *para* or *meta* positions generated the desired **251** in 88–96% yield. Also, the substrate with Br produced the desired **251** in 85% yield. In addition, aliphatic alkynes generated the desired **251** in up to 97% yield. Generally, the yield of the reaction decreased when using a strong electron-withdrawing group.

Zhang et al.  $^{109}$  synthesized 2-alkynoate derivatives 255 via the reaction among CO<sub>2</sub> 252, nBuI 254, and terminal alkyne

$$CO_2 + R = H + nBuI$$
252
 $R = H + nBuI$ 
254
 $R = O$ 
255
 $R = O$ 
255
 $R = O$ 
255

Scope of Alkynes

$$\mathbf{R} = \begin{array}{c} & \\ & \\ & \\ & \\ & \\ & \\ & \\ & \\ \end{array} , \text{ n-C}_{4} \\ \text{H}_{9} \\ \hline \end{array} + \text{ n-C}_{6} \\ \text{H}_{13} \\ \hline \end{array} - \text{H}$$

R'= H, 4-Me, 4-nPr, 4-Ph, 4-OMe, 4-F 3-F, 4-Cl, 4-CN, 4-CF<sub>3</sub>,

Scheme 106 Synthesis of 2-alkynoate derivatives. 109

255

Ag-NHC

AgNHC

AgNHC

AgNHC

AgNHC

AgNHC

AgNHC

CsO

OAg

Cs2CO<sub>3</sub>

CsHCO<sub>3</sub>

$$R = R$$
 $R = R$ 
 $R = R$ 

Scheme 107 Possible reaction mechanism for the synthesis of 2-alkynoate derivatives. 109

$$\begin{array}{c} \text{CO}_2 + R & \begin{array}{c} & \\ \hline \\ \textbf{256} \end{array} & \begin{array}{c} \text{X} \\ \textbf{257} \end{array} & \begin{array}{c} \text{AgI, Cs}_2\text{CO}_3 \\ \hline \text{CH}_3\text{CN}, 40^\circ\text{C}, 48 \text{ h} \end{array} & \begin{array}{c} \text{R} \\ \hline \\ \textbf{259} \end{array} & \begin{array}{c} \text{O} \\ \hline \\ \textbf{R'} \end{array} & \begin{array}{c} \text{R'} \\ \hline \\ \textbf{R''} \end{array} & \begin{array}{c} \text{R''} \\ \hline \\ \textbf{3-F, 4-Cl, 4-CN, 4-NO}_2 \end{array} & \begin{array}{c} \text{Br} \\ \hline \\ \textbf{Br} \end{array} & \begin{array}{c} \text{R''} \\ \\ \textbf{Scope of Benzyl Halide} \end{array}$$

R'''= H, 3-CN, 4-Me, 4-tBu, 4-OMe, 4-F 4-Cl, 4-Br, 2-I, 2,6-diCl, 2-CN, 4-CN

Scheme 108 Synthesis of benzyl 2-alkynoate derivatives. 110

derivatives 253 in the presence of Ag–NHC and Cs<sub>2</sub>CO<sub>3</sub> in DMF at 40 °C for 48 h (Scheme 106).

Initially, 253 coordinated with Ag-NHC and produced intermediate  $\bf A$ . Then, intermediate  $\bf B$  was produced from intermediate  $\bf A$  in the presence of  $Cs_2CO_3$ . After that, Agacetylide  $\bf C$  was obtained from intermediate  $\bf B$  together with the elimination of  $CsHCO_3$ . Next, Ag-propiolate intermediate  $\bf D$  was produced via the insertion of  $CO_2$  into the Ag-C bond of intermediate  $\bf C$ . Finally, 255 was generated via the reaction between nBuI and intermediate  $\bf D$  together with the regeneration the catalyst (Scheme 107).

Derivatives of 253 containing electron-donating groups gave the desired 2-alkynoate derivatives 255 in 77-94% yield. In addition, halo derivatives of 253 generated the desired 2-alkynoate derivatives 255 in 82-86% yield. Derivatives of 253

 $\begin{array}{c|cccc}
 & 259 \\
\hline
 & & 258 \\
\hline
 & & & & \\
 & & & & \\
\hline
 & & & & \\
 & & & & \\
\hline
 & & & & \\
 & & & & \\
\hline
 & & & & \\
 & & & & \\
\hline
 &$ 

**Scheme 109** Possible reaction mechanism for the synthesis of benzyl 2-alkynoate derivatives. <sup>110</sup>

containing a CN or  $\mathrm{CF}_3$  group produced the desired 2-alkynoate derivatives 255 in 84% and 81% yield, respectively. Furthermore, linear alkyl terminal alkyne derivatives 253 were tolerated and produced the desired 2-alkynoate derivatives 255 in 80–82% yield. It should be noted that the selectivity of the reaction was good due to the production of only one isolated product.

Br

Guo *et al.*<sup>110</sup> synthesized benzyl 2-alkynoate derivatives **259** *via* the reaction among  $CO_2$  **256**, aryl/alkyl terminal alkyne derivatives **257**, and benzyl halide derivatives **258** in the presence of  $Cs_2CO_3$  and AgI in  $CH_3CN$  under ligand-free conditions at 40 °C for 48 h (Scheme 108).

Initially, intermediate **K** was obtained *via* the coordination of 257 and Ag(1). Then, Ag(1)-acetylide **L** was generated *via* the deprotonation of **K** by  $Cs_2CO_3$  together with the release of  $CsHCO_3$ . After that, Ag-propiolate intermediate **M** was produced *via* the insertion of  $CO_2$  in the C-Ag bond. Finally, 259 was obtained *via* the reaction between **M** and 258 together with the release of Ag(1) (Scheme 109).

The electronic effect of the groups on the phenyl ring of aryl alkyne derivatives were demonstrated to have an important effect on the reactivity. In the case of 257, derivatives of 257 containing electron-donating groups produced the desired 259 in 66–73% yield. Also, derivatives of 257 containing F and Cl groups were produced the desired 259 in 65–75% yield. In contrast, aryl alkyne derivatives containing CN or NO<sub>2</sub> group produced the desired 259 in 45% and 33% yield, respectively. Also, aliphatic alkyne derivatives 257 generated desired 2-alkynoate derivatives 259 in 38–50% yield. The low product yields by aliphatic alkynes may be due to the low activity of the activation of their sp C–H bond. In the case of the benzyl halide derivatives 258, aryl chlorides or bromides 258

Scheme 110 Synthesis of propargylic esters. 111

containing electron-donating groups generated the desired 2alkynoate derivatives 259 in 71-85% yield. Also, halo derivatives of 258 gave the desired 2-alkynoate derivatives 259 in 62-91% yield. Aryl halide derivatives containing CF<sub>3</sub>, NO<sub>2</sub>, and CN gave the desired 2-alkynoate derivatives 259 in 63-87% yield. In addition, allylic/propargylic halide derivatives produced the desired 2-alkynoate derivatives 259 in 42-47% yield. It should be noted that secondary benzyl halide-(bromomethylene) dibenzene was tested and produced the desired 2-alkynoate 259 in 8% yield.

Papastavrou et al. 111 synthesized propargylic esters 263 via the reaction among organochlorides 262, CO<sub>2</sub> 260, and terminal alkyne derivatives 261 in the presence of K<sub>2</sub>CO<sub>3</sub> and NHC in DMF at 60 °C for 20 h (Scheme 110). In the case of 262, 2-chloroacetate esters generated the desired 263 in 25-28% yield. Organochlorides containing a Cl atom on the carbonyl carbon did not give the desired 263. In contrast, allylic chlorides showed good activity and produced the desired 263 in 42-97%. The low yield of chloropropene and

chlorobutene may be related to their low boiling points, which were lower than the reaction temperature. Similarly, the yield of crotyl chloride was high, which has a boiling point of 85 °C. In addition, benzyl chloride derivatives produced the desired ester 263 in 59-84% yields. In contrast, picolyl chloride did not generate the desired ester 263, which may be due to the presence of a pyridine moiety in its structure. It should be noted that organobromide derivatives did not give the desired esters 263. Furthermore, benzyl chlorides containing electron-poor and electron-rich groups produced the desired esters 263 in 59-84% yield. In the case of 261, alkyl-substituted terminal alkyne derivatives and a hydroxylbearing alkyne did not generate the desired esters 263. Phenyl acetylene bearing p-NO2 and p-CF3 produced the desired esters in trace and 91% yield, respectively. The result showed that electron-poor groups could produce the desired esters 263 but the NO2 group was not very consistent. Phenylacetylene derivatives containing an Me or OMe group on

 $R = Ph, 4-MeC_6H_4, 4-MeOC_6H_4, 4-tBuC_6H_4,$ 4-CF<sub>3</sub>C<sub>6</sub>H<sub>4</sub>, PhCH<sub>2</sub>CH<sub>2</sub>, n-hexyl

Scheme 111 Synthesis of 4-pentenyl 2-alkynoate derivatives. 112

X

# Scope of terminal alkynes

 $\mathbf{R}$ = Ph, 4-MeOC<sub>6</sub>H<sub>4</sub>, 2-MeOC<sub>6</sub>H<sub>4</sub>, 4-FC<sub>6</sub>H<sub>4</sub>, 4-ClC<sub>6</sub>H<sub>4</sub>, 4-BrC<sub>6</sub>H<sub>4</sub>, 4-MeC<sub>6</sub>H<sub>4</sub>, 4-CF<sub>3</sub>C<sub>6</sub>H<sub>4</sub>, 3-chloropropyl, 2-thiophenyl, 3-thiophenyl, cinnamyl, TMS

Scope of enallenes

 $\mathbf{R}$ = CO<sub>2</sub>Et, CH<sub>2</sub>CO<sub>2</sub>Et, CH<sub>2</sub>CH<sub>2</sub>OTs, CH<sub>2</sub>CH<sub>2</sub>OBn  $\mathbf{R}$ '= H. Me

Scheme 112 Synthesis of a-chirality ketones 272.113

the aromatic ring, and also containing no aromatic substituents produced the desired esters 263 in 64–97% yield.

Zhang *et al.*<sup>112</sup> synthesized 4-pentenyl 2-alkynoate derivatives **267** and 5-hexenyl 2-alkynoate derivatives **268** *via* the reaction among  $CO_2$  **264**, terminal alkyne derivatives **265**, and 5-bromopentene or 6-bromohexene derivatives **266** in the presence of AgI and  $Cs_2CO_3$  in DMF at 60 °C for 24 h (Scheme 111). The selectivity of the reaction was high and no direct coupling by-products were observed. Alkyne derivatives containing electron-withdrawing and -donating groups on the phenyl ring produced the desired carboxylative products (**267** and **268**) in 69–91% yield. Alkyl-substituted terminal alkynes in comparison to aryl-substituted substrates showed low yields (59–66%). These results may be due to the higher stability of the aryl-substituted Ag(1) acetylide with respect to the alkyl-substituted Ag(1) acetylide intermediate.

#### 3.3 Reaction of CO

Yang *et al.*<sup>113</sup> synthesized a-chirality ketones 272 *via* the reaction among terminal alkyne derivatives (271 and 271′), CO 269, and enallene derivatives (270 and 270′) in the presence of BQ, Pd(OAc)<sub>2</sub>, and chiral phosphoric acid (CPA) **X** in PhCl at 0 °C for 24 h (Scheme 112).

Initially, intermediate **A** was generated *via* the coordination of 270′ to Pd<sup>II</sup>. Then, carbonyl Pd<sup>II</sup> intermediate **B** was obtained *via* the insertion of CO and the subsequent attack of 270 on intermediate **A**. After that, olefin inserted into the C–Pd bond *via* enantioselective migration and produced intermediate **C**, which caused chirality at the alpha-position of the ketone. Lastly, 272 was generated *via* the carbonylative alkynylation of intermediate **C** together with the release of Pd<sup>0</sup>. To close the catalytic cycle, BQ reoxidized the released Pd<sup>0</sup> to Pd<sup>II</sup> (Scheme 113).

Scheme 113 Possible reaction mechanism for the synthesis of a-chirality ketones 272. 113

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#### Scope of terminal alkynes

$$EtO_{2}C$$

$$CO + 273$$

$$+ R = \frac{Pd(TFA)_{2}}{275} = \frac{BQ}{toluene, rt, time}$$

$$O = 276$$

 $R = Ph, 3-MeOC_6H_4, 2-MeOC_6H_4, 4-FC_6H_4, 4-MeC_6H_4, 4-CF_3C_6H_4, TMS$ n-hexyl, 2-thiophenyl, 3-thiophenyl, cyclopentyl, cinnamyl, 2-phenylethyl, 3-chloropropyl

# Scope of enallenes

R=CO<sub>2</sub>Et, CH<sub>2</sub>CO<sub>2</sub>Et, CH<sub>2</sub>CH<sub>2</sub>OTs, CH<sub>2</sub>CH<sub>2</sub>OBn **R'**= H, Et, Ph

Synthesis of ynone derivatives 276.114

In the case of 271, phenylacetylenes containing electrondonating and -withdrawing groups generated the desired 272 in 48-81% yield and high er (up to 95.5: 4.5). Aliphatic acetylene derivatives, as well as heteroaryl acetylene derivatives, were tolerated and produced the desired 272 in 64-73% yield. Importantly, alkyne containing TMS generated the desired 270 in 5% yield and poor er (56.5:43.5), and the reaction time for the preparation of this desired compound was 60 h. Poor enantioselectivity of the desired 272 was obtained when slow alkyne-quenching of the key Pd species existed. In the case of 270', when two Me were replaced with a terminal cycloalkyl group, the desired 272 was generated with a slight decrease in the enantiomeric ratio. The size of the ring was tested and the results showed that all rings could generate the desired 272 but had slightly lower er values compared to the six-membered rings. When an internal Me group on the olefin moiety was used, the desired 272 was generated in a lower er. Substituents on the allene moiety had a remarkable effect on the reactivity, as well as enantioselectivity. 2,3-Allenoate had lower er and yield with respect to 3,4-allenoate. Functional groups of sulfonyl ester and ether were suitable in the reaction and produced the desired 272 in 66% and 70% yield (93.5:6.5 and 95:5 er), respectively.

Zhu et al.114 synthesized ynone derivatives 276 via the reaction among CO 273, terminal alkyne derivatives 275, and enallene derivatives 274 in the presence of Pd(TFA)2 and BQ in  $CH_3Ph$  at room temperature (Scheme 114).

Initially, intermediate A was produced via the coordination of 247' to Pd(II). Then, intermediate B was generated from intermediate A via allene attack, where the additional coordination of the olefin to Pd(II) was vital. After that, intermediate C was obtained via the selective insertion of CO in the C-Pd bond of intermediate B. In this step, olefin insertion could occur, which led to the formation of intermediate F. Next, intermediate C produced intermediate E via olefin and carbon monoxide insertion reactions. Lastly, intermediate G was generated via the reaction between 275' and Pd in intermediate E, which after reductive elimination, gave the desired product together with the release of Pd(0). The released product was reoxidized to  $Pd(\Pi)$  by p-benzoquinone (Scheme 115).

In the case of 275, phenylacetylene derivatives containing electron-withdrawing and -donating groups generated the desired 276 in 70-81% yield. Also, heteroaryl acetylenes produced the desired 276 in 66-75% yield. It should be noted that aliphatic cyclic and acyclic terminal alkynes were tolerated well and produced the desired 276 in 74-83% yield. In addition,

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Scheme 115 Possible reaction mechanism for the synthesis of ynone derivatives 276.114

functional groups of TMS and Cl were tested and produced the desired 276 in 73% and 72% yield, respectively. In the case of 274′, cyclohexylidene enallene produced the desired 276 in 73% yield but cyclopentylidene enallene gave the desired 276 in a lower yield (58% yield). Tosyl and benzyl 3,4-dienol were tested and produced the desired 276 in 76% and 75% yield, respectively. It should be noted that the use of Ph or Et substituent rather than R″ substituent on the olefin had a great effect on the reaction (41% and 40% yield, respectively). Also, the dissymmetric allene containing i-Pr and Me gave the desired 276a and 276a′ in 40% and 30% yield, respectively. It should be noted that selective allenic C–H cleavage gave 276a and 276a′.

Qiu *et al.*<sup>115</sup> synthesized spiro[3.4]octene derivatives **280** and spiro[4.4]nonene derivatives **281** via the reaction among dienallene derivatives **278**, CO **277**, and alkyne derivatives **279** in the presence of Pd(TFA)<sub>2</sub> and BQ. For the preparation of **280**, CH<sub>3</sub>CN was used as the solvent at 80 °C, whereas DCE was used as the solvent for the preparation of **281** at room temperature (Scheme 116).

Initially, intermediate **A** was generated from the reaction between  $Pd(TFA)_2$  and 278 via allene attack involving allenic C–H bond cleavage. Then, intermediate **B** was produced from intermediate **A** via olefin insertion. After that, intermediate **D** was obtained from intermediate **C** via olefin and CO insertions. Lastly, intermediate **E** was generated via the reaction between intermediate **D** and 279, which gave the desired 280 after reductive elimination. Alternatively, intermediate **F** was

obtained from intermediate **A** *via* carbonylation. Then, intermediate **I** was generated from intermediate **G** and **H** *via* olefinolefin–CO insertion. Finally, the desired **281** was generated from the reaction between intermediate **I** and **279** *via* intermediate **J** (Scheme 117). It should be noted that CH<sub>3</sub>CN choose the pathway that was favorable for the production of intermediate **B** instead of intermediate **F**. This solvent suppressed CO coordination, and hence insertion.

In the case of 279 for the formation of 280, arylalkyne derivatives with electron-donating and -withdrawing groups generated the desired 280 in 64-79% yield. Also, heteroaryl acetylene derivatives were tolerated well and produced the desired 280 in 71-83% yield. Aliphatic terminal alkyne derivatives were tested and produced the desired 280 in 66-70% yield. Importantly, the TMS group reacted well and generated the desired 280 in 79% yield, which could be utilized for further functionalization after desilylation. In the case of 278 for the formation of 280, cyclooctylidene, cyclopentylidene, and cyclohexylidene produced the desired 280 in 80%, 71%, and 66% yield, respectively. In addition, the unsymmetric allene containing methyl and phenyl groups generated the desired 280 in 62% yield. It should be noted that when using n-butyl, cyclohexyl, and benzyl groups instead of R, the desired 280 was obtained in 55%, 60%, and 63% yield, respectively. In the case of the formation of 281, substrates of Me, cyclopentylidene, cyclohexylidene, or cyclooctylidene on the dienallene moiety were tolerated well and generated the desired 281 in 90%, 82%,

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A)

$$R$$
 $CO + R' = \frac{Pd(TFA)_2}{279} = \frac{BQ}{CH_3CN, 80^{\circ}C, time}$ 

280

R= CH<sub>2</sub>CO<sub>2</sub>Et, CH<sub>2</sub>CH<sub>2</sub>OAc, CH<sub>2</sub>Ph, cyclohexane, CH2CH2CH2CH3

 $\mathbf{R'} = \text{Ph}, 3 - \text{MeOC}_6 \text{H}_4, 2 - \text{MeOC}_6 \text{H}_4, 4 - \text{FC}_6 \text{H}_4, 4 - \text{BrC}_6 \text{H}_4, 4 - \text{MeC}_6 \text{H}_4, 4 - \text{CF}_3 \text{C}_6 \text{H}_4, 4 - \text{CF}_3 \text{C}_6 \text{H}_4, 4 - \text{MeC}_6 \text{H}_4, 4 - \text{CF}_3 \text{C}_6 \text{H}_4, 4 - \text{CF}_3 \text{C}_6 \text{H}_4, 4 - \text{MeC}_6 \text{H}_4, 4 - \text{CF}_3 \text{C}_6 \text{C}_6 \text{H}_4, 4 - \text{CF}_3 \text{C}_6 \text$ Cy, 2-thiophenyl, 3-thiophenyl, cinnamyl, TMS

Scheme 116 Synthesis of (A) spiro[3.4] octene derivatives 280 and (B) spiro[4.4] nonene derivatives 281.115

93%, and 89% yield, respectively. Arylacetylenes containing o-MeO and p-Me groups produced the desired product in 80% and 87% yield, respectively. Furthermore, heteroaryl acetylene derivatives were tested and the desired 281 obtained in 72-78% vield.

Volla and Backvall116 reported an oxidative carbocyclizationcarbonylation-alkynylation reaction for the synthesis of ynone derivatives 285 via the reaction among alkyne derivatives 283, CO 282, and enallene derivatives 284 in the presence of Pd(TFA)<sub>2</sub> and 1,4-BQ in ClCH<sub>2</sub>CH<sub>2</sub>Cl at room temperature for 12 h (Scheme 118).

Initially,  $\pi$ -complex **A** was generated with alkene and allene units by Pd. Then, intermediate B was obtained from intermediate A via the attack of the allene moiety on  $Pd(\Pi)$ , which was followed by the production of intermediate C via carbocyclization. After that, acylpalladium(II) species **D** was obtained from intermediate C via the insertion of CO in the Pd-C bond of intermediate C. Next, intermediate E was generated via the subsequent transfer of an alkynyl moiety. Finally, 285 was produced from E via reductive elimination together with the release of Pd(0), and Pd(0) was converted to Pd( $\pi$ ) via oxidization (Scheme 119).

In the case of 283, phenylacetylene derivatives possessing electron-donating and -withdrawing groups produced the desired 285 in 72-87% yield. It should be noted that steric and electronic factors on the aromatic ring did not show sensitivity. 1,3-Diethynylbenzene, which has two terminal alkyne units, produced monoalkynylation 285 without reaction of the second alkyne group during the reaction. Also, aliphatic cyclic and acyclic terminal alkyne derivatives generated the desired 285 in 54-94% yield. Importantly, TMS was tolerated well in the reaction and produced the desired TMS-product 285 in 81% yield, which could be utilized for further functionalization after desilylation. In the case of 284, when the dimethyl groups of the allene unit were replaced with methylethyl or pentamethylene units, the temperature was increased to 40 °C, and the desired 285 was produced in 88% and 76% yield, respectively. In addition, enallene, which contains a methyl group on the olefin unit, gave the desired 285 with an all-carbon quaternary center via the cascade carbocyclization-carbonylation-alkynylation in 73% yield at 80 °C. It should be noted that 1,6-enallene X did not generate the desired 258 even at high temperatures.

Chandrasekhar et al. 117 synthesized 3-aryl-4-(arylethynyl)-2H-chromen-2-one derivatives 289 via the reaction among CO

BQ + X280 278 281  $BQ + X^{-}$ Pd(0) PdX-ΗQ 278 E 279 allene attack slow step CO PdX insertion CO insertion olefin olefin insertion insertion olefin 'n CO insertion olefin insertion insertion В

Scheme 117 Possible reaction mechanism for the synthesis of spiro[3.4] octene derivatives 280 and spiro[4.4] nonene derivatives 281.115

**286**, terminal alkyne derivatives **287**, and 2-iodoaryl 2-arylacetate derivatives **288** in the presence of  $Pd(PPh_3)_2(Cl)_2$  and  $Et_3N$  in toluene at 80 °C for 3–10 h (Scheme 120A). According to the results, various substituents were evaluated on the three aryl rings, and all groups such as Cl,  $NO_2$ , Br, Me, OMe, and tBu were tolerated well in the reaction and produced the

desired **289** in 70–95% yield (Scheme 120a). In addition, trimethylsilylacetylene **291** generated the desired **292** in 78% yield, and the reaction time for the preparation was increased to 12 h (Scheme 120B). For more evaluations, 2-iodo-3,5-dimethylphenyl 2-phenylacetate **293** was investigated under the standard reaction conditions and produced just

**R**= H, Me

**R'**= H, Me,

**R"**= Ph, 4-Me- $C_6H_4$ , 4-(n-pent)- $C_6H_4$ , 4-MeO- $C_6H_4$ , 2-MeO- $C_6H_4$ , 4-F- $C_6H_4$ , 4-CF<sub>3</sub>- $C_6H_4$ , 3-ethynyl- $C_6H_4$ , n-butyl, 2-bromoethyl, 3-chloropropyl, cyclopentyl, 2-propenyl, 1-cyclohexenyl, TMS

Scheme 118 Synthesis of ynone derivatives 285. 116

Possible reaction mechanism for the synthesis of ynone derivatives 285.116

**R**= H, 4-Me, 2-Me, 2.4.6-Me<sub>3</sub>, 4-OMe

R'= H, 4-tBu, 4-Cl

**R''**= H, 4-OMe, 4-Br, 4-NO<sub>2</sub>

C)
$$H_{3}C \longrightarrow O \longrightarrow O \longrightarrow + Ph \longrightarrow + CO \xrightarrow{Pd(PPh_{3})_{2}(Cl)_{2}} \longrightarrow Ph \longrightarrow + CO \xrightarrow{Et_{3}N, \text{ toluene,}} O \longrightarrow O \longrightarrow CH_{3}$$

$$294 \longrightarrow 286 \xrightarrow{Et_{3}N, \text{ toluene,}} O \longrightarrow CH_{3}$$

$$295 \longrightarrow CH_{3}$$

Scheme 120 (A) Synthesis of 3-aryl-4-(arylethynyl)-2H-chromen-2-one derivatives 289, (B) 3-phenyl-4-((trimethylsilyl)ethynyl)-2H-chromen-2-one 292, (C) 3,5-dimethyl-2-(phenylethynyl)phenyl 2-phenylacetate 295, (D) 1-(phenylethynyl)naphthalen-2-yl acetate 297.<sup>117</sup>

Table 6 Summary of the reaction of alkynes with CO and limitations or progress of the reactions

Reactants	Product	Catalyst	Solvent	Method employed	Reaction time		Limitations or progress	Ref.
H R'\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	O R'	PdCl <sub>2</sub> (PPh <sub>3</sub> ) <sub>2</sub> - cataCXium A		CsOH·H <sub>2</sub> O, 80 °C, microwave	50 min	35- 70%	(1) CHCl <sub>3</sub> as <i>in situ</i> CO surrogate (2) Short reaction time (3) No prefunctionlization needed (4) Bioactive alk-2-ynamides (5) Catalytic system was new (6) Substrate scope was broad (7) 2-Ethynylpyridine could not produce the desired product (8) Trace amount of product was produced by trimethylsilylacetylene (9) Di- <i>n</i> -propylamine did not produce the	Rathod and Jain <sup>118</sup>
CO  R''-N' H	O N R' R"	IPr-Pd- Peppsi- $\mathrm{Cl}_2$	Dioxane	KI, K₃PO₄, 100 °C	18 h	50- 95%	desired product (1) Mild conditions (2) High atom efficiency (3) Alkyl and aryl of the alkynes could be efficiently	Zhang et al. <sup>119</sup>
R II I	R	$NiCl_2 \cdot 6H_2O$	DMF	dppb, K₂CO₃ 130 °C	7-12 h	20- 76%	(1) First time (CO <sub>2</sub> H) <sub>2</sub> was used as <i>in situ</i> CO surrogate, bench stable, <i>ex situ</i> solid, and easy to handle (2) First report for carbonylative Sonogashira coupling by using CO-surrogate under Ni catalyst	Shaifali et al. <sup>120</sup>
R + R'\ CO R"	$O \bigvee_{R'}^{R''} R'$	Pd(OAc) <sub>2</sub>	EtOAc	[NBu <sub>4</sub> ]I, 80 ° C, O <sub>2</sub> in N <sub>2</sub>	6 h	52- 92%	<ol> <li>The method avoids the use of dangerous 1,4-Dioxane</li> <li>Used O<sub>2</sub> as the terminal oxidant</li> <li>Does not require the use of ligands</li> <li>Develop synthesis of 2-ynamides through oxidative carbonylation</li> <li>A variety of alkynes and amines can be used</li> </ol>	Hughes et al. <sup>121</sup>
H I CO	R' O	Pd(II) APTES@K10	DME	NEt <sub>3</sub> 80 °C	7 h	60- 92%	(1) Method gave easy and simple access toward dibenzoylmethane (2) The electron withdrawing aryl halides can be efficiently used as an electrophile (3) Recovery of catalyst by simple filtration	Chavan et al. <sup>122</sup>
H Ar—I   +   CO	Ar O	Pd/C	Toluene	Et <sub>3</sub> N, 130 °C	4 h	63- 97%	<ul><li>(1) Mild conditions</li><li>(2) No need to add toxic phosphine ligands</li><li>(3) Easy catalyst recycling</li></ul>	Liu et al. <sup>123</sup>
H Ar—I   +   CO   R	Ar O	P(DVB-IL)-Pd	${ m H_2O}$	Et₃N, 130 °C	6 h	27- 95%	<ol> <li>The catalyst is easily separated from the products.</li> <li>Reuse of catalyst without activity loss</li> <li>Environmental synthesis of α, β-alkynyl ketones <i>via</i> the heterogeneous carbonylation reaction</li> <li>The method avoids the use of toxic phosphine ligands</li> </ol>	Wang et al. <sup>124</sup>

Table 6 (Contd.)

Reactants	Product	Catalyst	Solvent	Method employed	Reaction time	n Yield Limitations or progress	Ref.
I R + CO	R'————————————————————————————————————	Pd/Fe <sub>3</sub> O <sub>4</sub>	Tolueno	e Et₃N, 130 °C	4 h	<ol> <li>Highly efficient catalyst for the carbonylative Sonogashira coupling reaction (2) Catalyst can be reused with sustained selectivity and activity</li> <li>(3) Recoverability of catalyst</li> <li>(4) Catalyst could be reused seven times without remarkable loss in activity and selectivity</li> </ol>	Liu et al. <sup>125</sup>

Sonogashira coupling product 295 in 8% yield after 12 h and did not produce the desired 289 (Scheme 120C). Also, 1-iodo-2naphthyl acetate 296 under the standard reaction conditions generated just Sonogashira coupling product 297 in 98% after 4 h and did not produce the desired 289 or other products (Scheme 120D). A summary of the recent reaction of alkynes with CO and also limitations and progress of the reactions have been shown in Table 6.

# Conclusion

The need to improve the reaction conditions in the synthesis of heterocyclic compounds is an important and constant issue, and thus researchers are devoting their efforts to finding optimal and better methods and trying to design methods that solve the related problems. Among them, alkynes, which are a very important and widely use groups in chemistry, are used in important multi-component reactions given that they possess two active sites to carry out the reaction. Due to this characteristic of alkynes, it is challenging for researchers to design and activate a specific position. Meanwhile, researchers use ligands, catalysts, solvents, and even different methods such as microwave heating, which is a priority to perform the desired reactions using environmentally friendly conditions because both the reaction time is short and a low temperature is required, and also non-toxic materials are used, which help to improve the reaction conditions. We present our vision for improving the C≡C and C-H activation of alkynes in the various reactions that lead to a variety of important structures.

The amount of catalyst used in the investigated reactions showed that the amount of catalyst was very effective in the reactions and the alkyne used acted selectively. Furthermore, the presence of electron-withdrawing groups in alkynes (most of the reactions) reduces the efficiency of the reactions, and also aromatic alkynes participate in reactions better than aliphatic alkynes and have higher efficiency, although it should be noted that in some reactions aliphatic alkynes did not react at all. Also, the type of solvent is decisive in the reactions and plays a key role, especially in the reaction of alkynes with CO<sub>2</sub> or CO. In addition, the used molar ratios of alkynes are effective in the synthesis of compounds and different products are created by using different molar ratios.

This article highlighted some applications of the C≡C and C-H positions of alkynes in reactions with other important compounds in multicomponent reactions for the synthesis of important compounds and heterocycles and focused on recent advances in the use of alkynes. Also, the articles highlighted in this review are impressive and extensive and provide valuable insight into the activation of C≡C and C-H alkynes.

We hope that this review will encourage researchers to do more research in this field and cover the gaps in previous research, as well as offer insight into the design of new reactions and new chemical transformations of alkynes to prepare important structures.

#### 4.1 Future remarks

In the selectivity of alkynes, the use of a suitable catalyst is very important, and thus we suggest the use of efficient catalysts in this field, such as nanocatalysts, photocatalysts as well as enzymes. Also, in this case, the used ligands are very effective, and the use of a suitable ligand and catalyst combination helps to optimize the reactions, the selectivity of alkynes and also helps to carry out the reaction of all types of alkynes. Therefore, we suggest using or synthesizing ligands and catalysts that can activate the C≡C or C-H position with the highest selectivity.

## Conflicts of interest

There are no conflicts to declare.

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