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Reinforcement of ultrahigh thermoresistant polybenzimidazole films by hard craters†

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Porous structures were formed in polybenzimidazole derivative films by a silica template method, followed by HF etching. As expected, the resulting ultrahigh thermoresistant films showed better elongation properties than did the non-porous films, with renewed strength and hardness. Microscopic analyses revealed that the pores were clearly formed, and craters with swollen edges and hardened surfaces were also observed.

Production of bioplastics from renewable biological resources is imperative for developing a sustainable society and achieving low-carbon technology.¹ One of the major research topics in this regard is the synthesis of highly heat-resistant bioplastics because carbon dioxide in the atmosphere can be immobilized in durable materials for a considerable amount of time. For instance, 3-amino-4-hydroxybenzoic acid produced by genetically modified *Escherichia coli* can be converted to polybenzimidazole and its copolymers, both of which have extremely high heat resistance. Despite having the highest thermal decomposition temperature on record, polybenzimidazole-co-polyamide (poly(DABA-co-ABA)) films are found to be brittle in nature due to their rigid chain structure. Numerous studies have been conducted to improve the flexibility of polybenzimidazole through molecular design,² inorganic composites,³ and porous structure formation,⁴ among others. Porous structure formation can influence the merits of lightweight, physical stability, and high resiliency, which has attracted the attention of researchers in the last few decades.⁵ Luo and coworkers prepared porous PBI films using a phase separation process induced by water vapour. Chaudhari and coworkers prepared these films using the immersion precipitation method.⁶ The

effects of the porous structure on the mechanical properties of the films are rarely discussed in those reports, but the results reveal that the formation of porous structures improves the elongation rate of the material while degrading their mechanical properties. On the other hand, the silica (SiO₂) hard-template method, which is primarily used for water-absorbable materials such as hydrogels does not alter the components or chemical structure of the polymer, thus being a promising strategy for the processing of materials.⁷ In our study, we discovered that high-performance polymer films are also water-absorbable plastics due to their polar structures.⁸ Therefore, porous structures were cast in poly(DABA-co-ABA) films, and the hard-template method using SiO₂ nanoparticles was adopted to improve their stretchability and hardness (Fig. 1).

To form pores in poly(DABA-co-ABA), SiO₂ having a particle size of 300 nm was added to a mixed solution of poly(DABA-co-ABA) in trifluoroacetic acid (TFA)/methanesulfonic acid (MSA) and stirred until dissolution. Then, a PBI-SiO₂ composite film was produced by casting the mixed solution onto a glass substrate. The obtained film was etched with hydrofluoric acid (HF) to obtain a poly(DABA-co-ABA) porous film. The Fourier transform infrared (FT-IR) spectrum of SiO₂ showed a broad absorption band at 1000–1180 cm⁻¹, which was assigned to the Si–O–Si groups (Fig. S2†). As for the pristine poly(DABA-co-ABA), the characteristic bands at 1636, 1553, and 1284 cm⁻¹ were assigned to the stretching vibration of the C=N, C=C, and C–N groups, respectively, and the broad absorption band at 3250–3450 cm⁻¹ corresponded to N–H groups. A medium-strong peak was observed at 1000–1180 cm⁻¹ in the spectrum

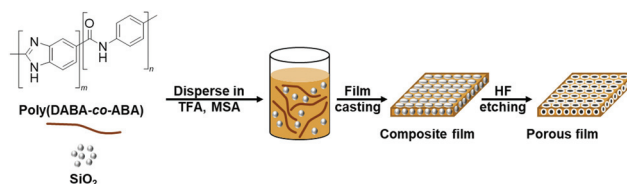


Fig. 1 Preparation of porous poly(DABA-co-ABA) films.

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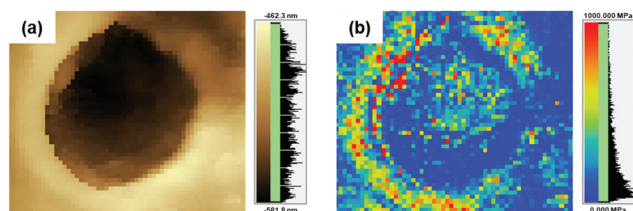


Fig. 3 SPM images of the (a) height and (b) elastic modulus of porous poly(DABA-co-ABA)-50% film.

area (Fig. 3a). Moreover, the elastic modulus of the white rim area (672 MPa), calculated from SPM force curve mode, is much higher than that of the area apart from the pore (72 MPa). With reference to the SPM images, it is evident that the elastic modulus of the white rim around the pore is higher than the rest of the area, thereby constituting an anomalistic behaviour.

The crystallinity and orientation of pristine and porous poly(DABA-co-ABA) were investigated by wide-angle X-ray diffraction (WAXD) to determine the causes of the improvements in mechanical properties (Fig. 4). The peak position (2θ) and d -spacing of pristine as well as porous poly(DABA-co-ABA) from WAXD were found to be 26.3 and 0.14 nm, respectively. However, the pristine poly(DABA-co-ABA) had a degree of orientation of 0.8 and a crystal size of 0.1 nm, while the porous poly(DABA-co-ABA) had a degree of orientation of 0 and a crystal size of 0.2 nm, implying that the formation of pores in poly(DABA-co-ABA), led to a decrease in its orientation and crystallinity, but an increase in crystal size. Furthermore, the degree of crystallinity for pristine and porous poly(DABA-co-ABA) were 7.8 and 4.9%, respectively. The decrease in the orientation and crystallinity possibly cause a reduction of modulus and strength, but the strength in elastic modulus of the rim areas brought a much greater effect on enhancing mechanical properties, as a result, the negative influence of decreased orientation and crystallinity was offset and a general increase in modulus and strength was manifested. This observation, along with the results of the SPM analysis demonstrates that pore formation improved the elongation ratio and mechanical properties of the entire poly(DABA-co-ABA) film.

The surface energy causes the poly(DABA-co-ABA) film covering the SiO_2 nanoparticles to form a crater-like shape in the holes formed by the SiO_2 nanoparticles removal. This formation may be attributed to the enhancement of intermolecular hydrogen bonds of poly(DABA-co-ABA) chains in the presence of HF

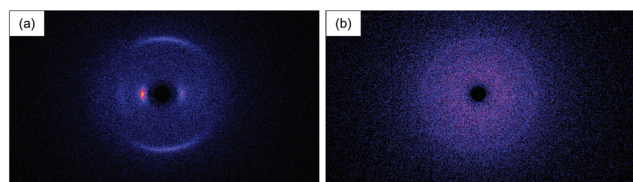


Fig. 4 WAXD images of the pores in (a) pristine poly(DABA-co-ABA) and (b) porous poly(DABA-co-ABA)-50% film.

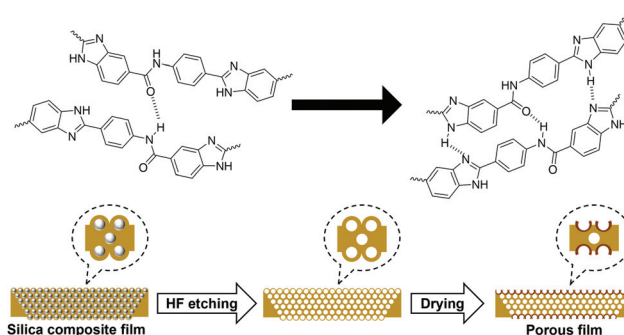


Fig. 5 The mechanism for the formation of hard craters in poly(DABA-co-ABA) film.

aqueous solution plasticizer, as well as the reduction of free energy due to the disappearance of SiO_2 nanoparticles (Fig. 5). In the case of using SiO_2 nanoparticles with a particle diameter of 20 nm, uniform pores were not observed due to aggregation of the SiO_2 nanoparticles (Fig. S5†). However, even in this case, where the particle size is not uniform, a white region around the pores was also observed as well, indicating the formation of “hard craters”. This result indicated that it is effective to improve the mechanical properties by SiO_2 addition and HF etching into the polymer.

To summarise, poly(DABA-co-ABA) films with a porous structure were prepared by means of a hard-template method using SiO_2 nanoparticles. The craters formed on the films measured diametrically from 200 to 300 nm and the porosity of the films was adjusted by changing the volume of added SiO_2 nanoparticles during the preparation stage. The porous poly(DABA-co-ABA) films thus prepared, showed 4.9 fold higher elongation at break than the pristine ones, attributed to their porous structure. Surprisingly, Young's modulus was also found enhanced in porous poly(DABA-co-ABA) films. The porous structure increased the toughness of the pristine poly(DABA-co-ABA) film by 11 fold as whole materials (including pore volume) and by 27 fold in matrix polymers (excluding pore volume). SEM analyses revealed a white edge circumscribing each of the pores and SPM analyses attested the hardness of these edges. WAXD analyses revealed an increased crystal size of poly(DABA-co-ABA) films due to the formation of pores. The findings indicate that the hard craters formed on the film surface helped to reinforce the mechanical properties of the film. The porous poly(DABA-co-ABA) films were extremely thermoresistant and qualified for applications such as catalyst support, gas storage, proton exchange, and so on. Moreover, the hard-template method using SiO_2 nanoparticles followed with etching by HF aqueous solution would be beneficial for similar experimentation with films composed of water-absorbable polar polymers such as polyimides, polyamides, polyurethanes, and polyesters.

Conflicts of interest

There are no conflicts to declare.



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