



Cite this: *Chem. Commun.*, 2019, 55, 2605

Received 3rd December 2018,
Accepted 6th February 2019

DOI: 10.1039/c8cc09612k

rsc.li/chemcomm

Ferri- and ferro-magnetism in CaMnMReO₆ double double perovskites of late transition metals M = Co and Ni†‡

Elena Solana-Madruga,^a Yu Sun,^a Ángel M. Arévalo-López^b and J. Paul Attfield^{ib}*^a

Two new double double perovskites of ideal compositions CaMnMReO₆ (M = Co, Ni) are reported. The M = Co material has refined composition CaMn_{0.7}Co_{1.3}ReO₆ and orders ferrimagnetically below T_C = 188 K with a relatively large saturated magnetisation of 4.5 μ_B. The M = Ni product, CaMn_{1.2}Ni_{0.8}ReO₆, is a remarkable example of a ferromagnetic oxide with four distinct spin sublattices all collinearly ordered below T_C = 152 K.

The versatility of the perovskite structure^{1–5} has led to the synthesis and study of many perovskite materials exhibiting a wide range of useful properties. 1 : 1 ordering of cations at the A and B site cations in the basic ABO₃ perovskite structure in layered, columnar or rock-salt arrangements⁶ leads to AA'B₂O₆^{7,8} or A₂BB'O₆⁹ double perovskites (DPv) but combinations of cation order in both sublattices are uncommon. These double double (or doubly ordered) perovskites usually combine layered and rock-salt motifs of A and B cations respectively.^{10–13} A new double double perovskite (DDPv) type was recently reported to combine a columnar arrangement of A-site cations with the rock-salt order of B and B' cations for RMnMnSbO₆ (R = rare earth).^{14,15} This structure crystallizes with the tetragonal P₄₂/n space group and has five independent cation sites according to the composition AA'_{0.5}A''_{0.5}BB'O₆; A (Wyckoff position 4e), A' (2a) and A'' (2b) sites with ten-fold, tetrahedral and square planar coordinations, respectively, and octahedral sites for B (4c) and B' (4d) cations. CaMnMReO₆ double double perovskites with the same structure were subsequently reported for M = Mn and Fe. These are found to have different charge distributions and magnetic properties, as CaMnFe³⁺Re⁵⁺O₆ is ferrimagnetic with Curie temperature T_C = 500 K, while

CaMnMn²⁺Re⁶⁺O₆ has a mix of ferro- and antiferro-magnetic spin sublattices ordered below 120 K.¹⁶ To discover the cation and charge distributions for late first row transition metal ions, which have not previously been incorporated into double double perovskites, we report here the syntheses, structure and properties of CaMnMReO₆ double double perovskites with M = Co and Ni, which are discovered to have ferromagnetic order over 3 or 4 spin sublattices.

Samples of ideal composition CaMnCoReO₆ and CaMnNiReO₆ have been treated under high pressure and high temperature conditions using a Walker-type multi anvil apparatus. A stoichiometric mixture of CaMnO₃ (obtained by heating a pellet of CaCO₃ and MnO in 1 : 1 ratio at 1473 K for 24 h under an O₂ flow with intermediate grinding and repelletising), ReO₂ and MO (M = Co, Ni) oxides, were packed in a Pt capsule and treated at 10 GPa and 1373 K (for M = Co) or 1473 K (M = Ni) during 20 minutes. The samples were quenched and slowly depressurized to ambient conditions. These synthesis temperatures were found to be optimal for production of the DDPv type phases, as syntheses at 1423–1473 K for M = Co lead to secondary Re phases, while products made at 1373–1423 K for M = Ni contained unreacted ReO₂ and NiO. However secondary phases were present for both materials and were not eliminated in repeated syntheses. Laboratory X-ray powder diffraction (XPD) patterns were collected using a D2 Bruker diffractometer. Magnetic susceptibilities (ZFC-FC under an applied magnetic field of 1000 Oe) and magnetization-field loops up to 7 T were measured using an MPMS SQUID magnetometer. Neutron powder diffraction (NPD) patterns collected at room temperature and 1.5 K on the WISH beamline at the ISIS facility were used to determine crystal and magnetic structures.

Fits to XPD (Fig. S1, ESI†) and NPD data showed that both CaMnMReO₆ (M = Co, Ni) samples adopt the P₄₂/n DDPv structure. XPD patterns reveal traces of secondary ReO₂ (1.7 and 3.0 weight% respectively for M = Co, Ni). The NPD patterns (Fig. 1 and ESI†) show the presence of a rocksalt type monoxide impurity (8.0 and 17.1% respectively for M = Co, Ni). In both cases the cation distribution in the latter refined to composition M_{0.67}Mn_{0.33}O,

^a Centre for Science at Extreme Conditions (CSEC) and School of Chemistry, The University of Edinburgh, EH9 3FD, UK. E-mail: j.p.attfield@ed.ac.uk

^b Univ. Lille, CNRS, Centrale Lille, ENSCL, Univ. Artois, UMR 8181 – UCCL – Unité de Catalyse et Chimie du Solide, F-59000 Lille, France

† Data that support the findings of this study have been deposited at <https://datashare.is.ed.ac.uk/handle/10283/838>.

‡ Electronic supplementary information (ESI) available: Supporting figures and tables. See DOI: 10.1039/c8cc09612k



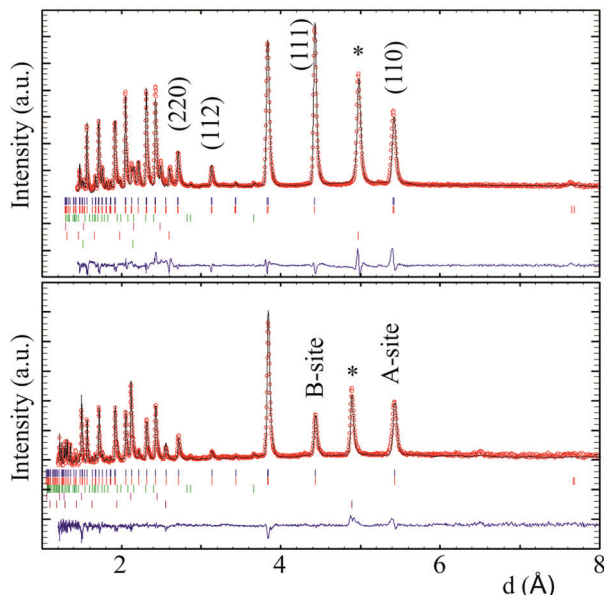


Fig. 1 Rietveld fits of NPD patterns collected at 1.5 K for CaMnCoReO₆ (top) and CaMnNiReO₆ (bottom). Bragg markers from top to bottom are the nuclear and magnetic structures of the DDPv phase, ReO₂ and the nuclear and magnetic structures of M_{0.67}Mn_{0.33}O. The main magnetic peaks are indexed on the top panel and the perovskite-superstructure peaks evidencing the A- and B-site cation orders are identified at the bottom. The main magnetic peak of M_{0.67}Mn_{0.33}O is marked *.

suggesting that high pressure stabilises a rocksalt-related M₂MnO₃ phase, but possible long range cation ordering was not detected in this study. Table 1 summarizes the refined DDPv structures of the CaMnMReO₆ (M = Co, Ni) compounds.

The varied neutron scattering lengths for the metals in these compounds (4.70, −3.73, 2.49, 10.3, 9.2 fm for Ca, Mn, Co,

Table 1 Sites showing Wyckoff labels, atomic coordinates, occupancies and BVS values resulting from the Rietveld fits of the room temperature NPD patterns of CaMnCoReO₆ (upper lines) and CaMnNiReO₆ (lower lines). The space group is *P4₂/n* and cell parameters are *a* = 7.666(1) and 7.6868(4) Å and *c* = 7.6899(4) and 7.693(1) Å respectively. *B*_{iso} values, constrained for cations and oxygen sites, are 0.2(1) and 1.0(1) Å² for M = Co and 0.8(1) and 1.7(1) Å² for M = Ni respectively. Residuals; *R*_p = 2.24%, *R*_{wp} = 2.08%, *R*_f = 5.63%, *R*_B = 4.69% and *R*_p = 2.59%, *R*_{wp} = 3.19%, *R*_f = 7.09%, *R*_B = 5.28% respectively

Site	<i>x</i>	<i>y</i>	<i>z</i>	Mn/M occ	BVS
A' (2a)	0.75	0.75	0.75	0.692/0.308(1)	1.9
				0.892/0.108(1)	1.9
A'' (2b)	0.25	0.25	0.75	0.612/0.388(1)	1.7
				0.788/0.212(1)	1.6
Ca (4e)	0.25	0.75	0.779(1)		2.0
			0.783(1)		
B (4c)	0	0.5	0.5	0.040/0.960(1)	2.0
				0.366/0.634(1)	1.7
Re (4d)	0	0	0.5		5.9
O1 (8g)	−0.055(2)	0.556(2)	0.239(1)		
	−0.049(1)	0.559(2)	0.24(1)		
O2 (8g)	−0.237(2)	−0.049(1)	0.571(1)		
	−0.240(3)	−0.049(1)	0.566(1)		
O3 (8g)	−0.263(1)	0.061(1)	−0.0298(5)		
	−0.265(2)	0.055(1)	−0.029(1)		

Ni and Re respectively)¹⁷ provide a high degree of contrast enabling cation mixing to be investigated across the five metal sites in the AA'_{0.5}A''_{0.5}BB'O₆ structure. The A = Ca (ten-fold) and B' = Re (octahedral) sites were found to be fully occupied in both materials within experimental error in the NPD refinements. CaMnCoReO₆ has 96% Co at the octahedral B site, but 30–40% of Co substitutes for Mn at the A' (tetrahedral) and A'' (square planar) sites. This leads to an overall Co-rich composition CaMn_{0.7}Co_{1.3}ReO₆ relative to the ideal formula. CaMnNiReO₆ has a mix of 37% Mn and 63% Ni at the B site but with 10–20% Ni substituting at the A' and A'' sites, leading to an overall Ni-poor composition CaMn_{1.2}Ni_{0.8}ReO₆. This is consistent with the higher proportion of observed M₂MnO₃ secondary phase than in the M = Co sample. Apparent lack of Ca in the products may be because this has formed amorphous phases, or crystalline products below limits of XPD detection. XPD data were fitted using the NPD structural models, and with M/Re disorder at the B-sites refined. Small antisite occupancies of 3.4(1) and 2.5(1)% for M = Co and Ni respectively show that the high charge contrast between (Mn/Co/Ni)²⁺ and Re⁶⁺ leads to a high degree of B-site ordering.

The cation oxidation states at each atomic position have been estimated from BVS calculations,¹⁸ as shown in Table 1, confirming that the formal charge distributions are Ca²⁺A'^{1.2+}A''²⁺_{0.5}B²⁺Re⁶⁺O₆ in both cases with A', A'', and B sites occupied by mixtures of Mn²⁺ and Co²⁺/Ni²⁺. This disorder reflects the similarity of ionic radii between the divalent transition metal cations. However, no substitutions at the Ca and Re sites were found in either material.

Magnetic susceptibility and field dependent magnetization measurements are depicted in Fig. 2. CaMnCoReO₆ and CaMnNiReO₆ show ferrimagnetic behaviour below *T*_C = 188 and 152 K respectively. A fit of the Curie–Weiss law to the high temperature inverse susceptibilities (shown in ESI†) results in Weiss constants *θ* = 188 and 151 K respectively, indicative of ferromagnetic exchange interactions for both compounds. The effective paramagnetic moments are 5.8 and 4.9 μ_B f.u.^{−1} for M = Co and Ni respectively. Predicted values are 6.9 and 7.2 μ_B f.u.^{−1} for the refined compositions CaMn_{0.7}Co_{1.3}ReO₆ and CaMn_{1.2}Ni_{0.8}ReO₆, assuming all cations are in high spin states. Both experimental values lie below the predicted values which most likely reflects the presence of impurity contributions to the susceptibilities. Low temperature *M*–*H* loops in Fig. 2 show that both materials have substantial saturated magnetic moments, of 4.5 and 1.8 μ_B f.u.^{−1} for M = Co and Ni respectively, indicating that they have ferri- or ferro-magnetic orders at low temperature. Their coercivities are small and the M = Co hysteresis loop shows a narrowing near *H* = 0 indicating that competing ferro- and antiferro-magnetic exchange interactions are present.

Magnetic diffraction peaks observed in the 1.5 K NPD patterns of CaMnCoReO₆ and CaMnNiReO₆ are indexed by propagation vector [0 0 0]. Rietveld fits (*R*_{mag} = 5.54% and 5.83% respectively) yield the magnetic structures shown in Fig. 3. Both CaMnCoReO₆ and CaMnNiReO₆ show a simultaneous order of A and B sublattices with spins along the *z* axis.



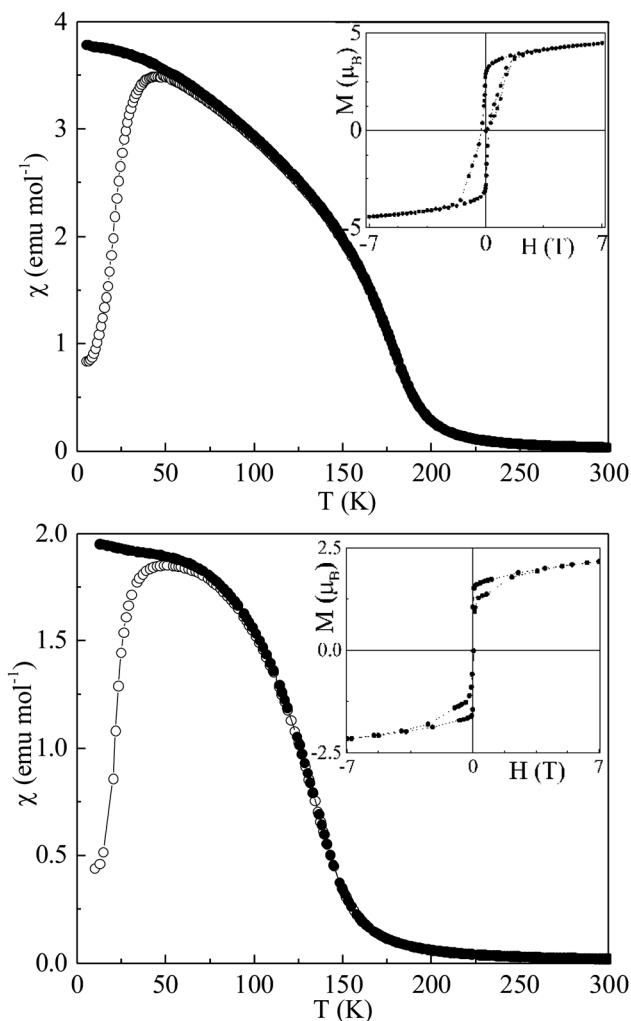


Fig. 2 Magnetic susceptibilities of CaMnCoReO₆ (top) and CaMnNiReO₆ (bottom) with insets showing magnetization-field loops at 7 and 2 K respectively.

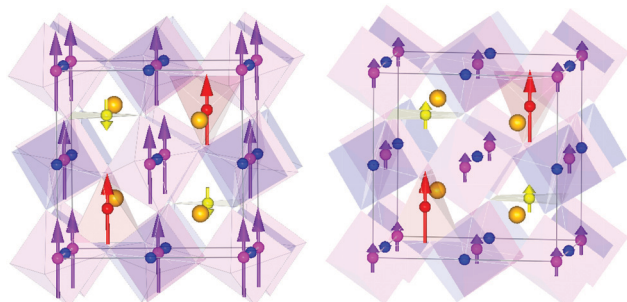


Fig. 3 Crystal and magnetic structures of CaMnCoReO₆ (left) and CaMnNiReO₆ (right) with the z-axis vertical. Orange, red, yellow, purple and blue spheres represent Ca, A', A'', B and Re sites respectively. Polyhedra show the tetrahedral, square planar and octahedral coordination of A', A'' and B sites, and arrows show the spin directions scaled to the refined magnetic moments.

All of the magnetic moments were refined independently, converging to a net ferrimagnetic arrangement of A and B sublattices

Table 2 Observed saturated magnetization at low temperatures M_{sat} and NPD refined magnetic moments for tetrahedral A', square planar A'', and octahedral B and B' sites, and the resulting net magnetic moment per formula unit μ , for CaMnMReO₆ with M = Co/Ni. All quantities are in μ_{B} units

M	M_{sat}	$\mu(A')$	$\mu(A'')$	$\mu(B)$	$\mu(B')$	μ
Co	4.5	2.9(1)	-1.3(1)	3.27(3)	0.3(1)	4.4
Ni	1.8	3.0(1)	1.2(2)	1.2(1)	0.2(1)	3.5

for CaMnCoReO₆ and their ferromagnetic alignment for CaMnNiReO₆, as summarised in Table 2. Other magnetic modes were not consistent with the data. See ESI† for further details of the magnetic symmetry analysis. An additional magnetic diffraction peak was assigned and fitted as the $(\frac{111}{222})$ peak of the M_{0.67}Mn_{0.33}O impurity phases. This peak is also observed in the 300 K pattern of CaMnNiReO₆. The high ordering temperature is consistent with the Néel temperatures of NiO ($T_{\text{N}} = 523$ K) and MnO ($T_{\text{N}} = 122$ K),¹⁹ so a transition near 390 K may be interpolated for Ni_{0.67}Mn_{0.33}O. An equivalent interpolation based on $T_{\text{N}} = 291$ K for CoO predicts spin ordering around 235 K for Co_{0.67}Mn_{0.33}O, consistent with the observed appearance of magnetic neutron diffraction peaks between 300 and 1.5 K.

It is notable that all four magnetic sublattices in these two double double perovskites order ferromagnetically. The Re moments are very small, consistent with those in other Re⁶⁺ oxides,¹⁶ but refine to be parallel to the other B sublattice spins whereas those in CaMnFe³⁺Re⁵⁺O₆ are antiparallel due to spin polarized conduction that leads to a high Curie temperature of 500 K. The T_{C} 's below 200 K and parallel alignment of B = Co, Ni and B' = Re spins in the CaMnMReO₆ (M = Co, Ni) materials indicate that they are insulating ferromagnets, although we have not yet obtained well-sintered ceramic pellets for direct conductivity measurements to confirm this. Ordered tetrahedral A' site moments are larger than those at square planar A'' sites in both materials, reflecting the greater degree of cation disorder at the latter sites. B site moments are relatively large (3.3 μ_{B}) for the M = Co material which has almost no cation disorder at this site, whereas substantial Mn/Ni mixing leads to a much smaller ordered moment of 1.2 μ_{B} for M = Ni. The reduction of the site moments compared to ideal 2S values reflects magnetic disorder resulting from a complex mix of antiferro- and ferro-magnetic exchange interactions between the different sites and cation dⁿ configurations. This may be sensitive to the precise compositions of the studied samples, both of which are off-stoichiometric as noted above.

CaMnCoReO₆ is a ferrimagnet overall as the ferromagnetic A'' sublattice spins are antiparallel to those of the A', B and B' sublattices. The net moment of 4.4 μ_{B} for M = Co predicted from the NPD results is in good agreement with the observed net magnetization from the hysteresis loop (4.5 μ_{B} at 7 K). The refined spin structure of CaMnNiReO₆ is remarkable as all four ferromagnetic sublattices are parallel so this material is a rare example of a ferromagnetic (and likely insulating) oxide. The observed magnetization is below the predicted value from the



refined NPD moments, most likely reflecting the presence of the antiferromagnetic $\text{Ni}_{0.67}\text{Mn}_{0.33}\text{O}$ impurity phase. Ferrimagnetism could also account for the smaller saturated moment, but attempts to refine such models against the neutron data always resulted in the ferromagnetic spin structure shown.

In conclusion, two new members of the double double perovskite family, of ideal compositions CaMnCoReO_6 and CaMnNiReO_6 , have been synthesized under high pressure and high temperature conditions. They retain the cation ordering pattern of CaMnFeReO_6 , but with more substantial Mn/M cation mixing across three of the five available cation sites leading to non-stoichiometric $\text{CaMn}_{0.7}\text{Co}_{1.3}\text{ReO}_6$ and $\text{CaMn}_{1.2}\text{Ni}_{0.8}\text{ReO}_6$ compositions. Ferromagnetic ordering within all of their spin sublattices occurs at a single magnetic transition, unlike previous studied $M = \text{Mn}$ and Fe analogues where two transitions were observed. CaMnCoReO_6 orders ferrimagnetically below $T_C = 188$ K with a relatively large saturated magnetization of $4.5 \mu_B$. CaMnNiReO_6 is a remarkable example of a ferromagnetic oxide with four distinct spin sublattices all collinearly ordered below $T_C = 152$ K.

We thank EPSRC for support, and STFC for provision of access to ISIS and Dr P. Manuel for assistance with data collection.

Conflicts of interest

There are no conflicts to declare.

Notes and references

- H. D. Megaw, *Proc. Phys. Soc.*, 1946, **58**, 326.
- C. J. Howard and H. T. Stokes, *Acta Crystallogr., Sect. B: Struct. Sci.*, 1998, **54**, 782–789.
- C. J. Howard, B. J. Kennedy and P. M. Woodward, *Acta Crystallogr., Sect. B: Struct. Sci.*, 2003, **59**, 463–471.
- M. Anderson, K. Greenwood, G. Taylor and K. Poeppelmeier, *Prog. Solid State Chem.*, 1993, **22**, 197–233.
- R. H. Mitchell, *Perovskites: Modern and ancient*, Almaz Press, Ontario, Canada, 2002.
- G. King and P. M. Woodward, *J. Mater. Chem.*, 2010, **20**, 5785–5797.
- K. Lienenweber and J. Parise, *J. Solid State Chem.*, 1995, **114**, 277–281.
- A. Aimi, D. Mori, K. Hiraki, T. Takahashi, Y. J. Shan, Y. Shirako, J. Zhou and Y. Inaguma, *Chem. Mater.*, 2014, **26**, 2601–2608.
- S. Vasala and M. Karppinen, *Prog. Solid State Chem.*, 2015, **43**, 1–36.
- M. C. Knapp and P. M. Woodward, *J. Solid State Chem.*, 2006, **179**, 1076–1085.
- G. King, S. Thimmaiah, A. Dwivedi and P. M. Woodward, *Chem. Mater.*, 2007, **19**, 6451–6458.
- P. Zuo, H. Klein, C. Darie and C. V. Colin, *J. Magn. Magn. Mater.*, 2018, **485**, 48–51.
- C. De and A. Sundaresan, *Phys. Rev. B*, 2018, **97**, 214418.
- E. Solana-Madruga, A. M. Arévalo-López, A. J. Dos santos-García, E. Urones-Garrote, D. Avila-Brandé, R. Sáez-Puche and J. P. Attfield, *Angew. Chem., Int. Ed.*, 2016, **55**, 9340–9344.
- E. Solana-Madruga, A. M. Arévalo-López, A. J. Dos santos-García, C. Ritter, C. Cascales, R. Sáez-Puche and J. P. Attfield, *Phys. Rev. B*, 2018, **97**, 134408.
- G. M. McNally, A. M. Arévalo-López, P. Kearins, F. Orlandi, P. Manuel and J. P. Attfield, *Chem. Mater.*, 2017, **29**, 8870–8874.
- V. F. Sears, *Neutron News*, 1992, **3**, 26–37.
- J. P. Attfield, *Solid State Sci.*, 2006, **8**, 861–867.
- W. L. Roth, *Phys. Rev.*, 1958, **110**, 1333–1341.

