

## CORRECTION

View Article Online  
View Journal



Cite this: DOI: 10.1039/d5ee90077h

## Correction: Carbon footprint of oil produced through enhanced oil recovery using carbon dioxide directly captured from air

Antonio Gasós,<sup>a</sup> Ronny Pini,<sup>b</sup> Viola Becattini<sup>a</sup> and Marco Mazzotti<sup>\*a</sup>

DOI: 10.1039/d5ee90077h

rsc.li/ees

Correction for 'Carbon footprint of oil produced through enhanced oil recovery using carbon dioxide directly captured from air' by Antonio Gasós *et al.*, *Energy Environ. Sci.*, 2025, <https://doi.org/10.1039/d5ee01752a>.

In section 2.3 of the manuscript, immediately following eqn (5), the text contained an error in the following paragraph.

'Here,  $\rho_j$  and  $\rho_{\text{CO}_2}$  are the densities of phase  $j$  and of  $\text{CO}_2$  at relevant temperature and pressure levels, respectively, while  $M_j$  and  $M_{\text{CO}_2}$  are their molar masses, in mass per mole of carbon. We use  $M_o = 14 \text{ g mol}^{-1}$  (for  $\text{CH}_2$ , the building block of oil),  $M_g = 16 \text{ g mol}^{-1}$  (methane), and  $M_w = 0 \text{ g mol}^{-1}$  (water, being carbon-free).'

This should instead read as follows.

'Here,  $\rho_j$  and  $\rho_{\text{CO}_2}$  are the densities of phase  $j$  and of  $\text{CO}_2$  at relevant temperature and pressure levels, respectively, while  $M_j$  are the molar masses, in mass of  $j$  per mole of carbon contained in  $j$ . We use  $M_o = 14 \text{ g mol}^{-1}$  (for  $\text{CH}_2$ , the building block of oil),  $M_g = 16 \text{ g mol}^{-1}$  (methane), and  $M_w = \infty \text{ g mol}^{-1}$  (water, being carbon-free).'

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

<sup>a</sup> Institute for Energy and Process Engineering, ETH Zurich, 8092 Zürich, Switzerland. E-mail: marcom@ethz.ch

<sup>b</sup> Department of Chemical Engineering, Imperial College London, SW7 2AZ London, UK

