



Cite this: *Nanoscale Horiz.*, 2022, 7, 616

Received 7th October 2021,
Accepted 6th April 2022

DOI: 10.1039/d1nh00527h

rsc.li/nanoscale-horizons

Exploring seebeck-coefficient fluctuations in endohedral-fullerene, single-molecule junctions†

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For the purpose of creating single-molecule junctions, which can convert a temperature difference ΔT into a voltage ΔV via the Seebeck effect, it is of interest to screen molecules for their potential to deliver high values of the Seebeck coefficient $S = -\Delta V/\Delta T$. Here we demonstrate that insight into molecular-scale thermoelectricity can be obtained by examining the widths and extreme values of Seebeck histograms. Using a combination of experimental scanning-tunnelling-microscopy-based transport measurements and density-functional-theory-based transport calculations, we study the electrical conductance and Seebeck coefficient of three endohedral metallofullerenes (EMFs) $\text{Sc}_3\text{N}@C_{80}$, $\text{Sc}_3\text{C}_2@C_{80}$, and $\text{Er}_3\text{N}@C_{80}$, which based on their structures, are selected to exhibit different degrees of charge inhomogeneity and geometrical disorder within a junction. We demonstrate that standard deviations in the Seebeck coefficient σ_S of

New concepts

If high-thermoelectric-performance molecules could be identified, then they could be used as building blocks for organic thin-film materials able to convert waste heat into electricity. A key parameter governing thermoelectric performance is the Seebeck coefficient S , which determines the voltage generated when a temperature difference is applied to a single-molecule junction. In the literature, ensemble averaged values of S are typically reported. Here we demonstrate that the search for high-performance molecules can be accelerated by focussing attention on fluctuations in S . We study the electrical conductance and Seebeck coefficient of three endohedral metallofullerenes (EMFs) $\text{Sc}_3\text{N}@C_{80}$, $\text{Sc}_3\text{C}_2@C_{80}$, and $\text{Er}_3\text{N}@C_{80}$, which based on their structures, are selected to exhibit different degrees of charge inhomogeneity and geometrical disorder within a junction. These cause the molecules to exhibit a wide range of Seebeck coefficients, which means that if orientations corresponding to high values can be selected and controlled, then the molecules have the potential to exhibit high-performance thermoelectricity. Furthermore, the EMFs studied here exhibit distributions of Seebeck coefficients containing both positive and negative signs, which reveals that all these EMFs are bi-thermoelectric materials. The strategy of exploiting fluctuations is a new concept and will lead to new strategies for designing thermoelectric materials.

EMF-based junctions are correlated with the geometric standard deviation σ and the charge inhomogeneity σ_q . We benchmark these molecules against C_{60} and demonstrate that both σ_q , σ_S are the largest for $\text{Sc}_3\text{C}_2@C_{80}$, both are the smallest for C_{60} and for the other EMFs, they follow the order $\text{Sc}_3\text{C}_2@C_{80} > \text{Sc}_3\text{N}@C_{80} > \text{Er}_3\text{N}@C_{80} > C_{60}$. A large value of σ_S is a sign that a molecule can exhibit a wide range of Seebeck coefficients, which means that if orientations corresponding to high values can be selected and controlled, then the molecule has the potential to exhibit high-performance thermoelectricity. For the EMFs studied here, large values of σ_S are associated with distributions of Seebeck coefficients containing both positive and negative signs, which reveals that all these EMFs are bi-thermoelectric materials.

Introduction

During recent years, the search for non-toxic and easily processable thermoelectric materials has led several groups to explore

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† Electronic supplementary information (ESI) available: Synthesis, purification and MALDI-TOF data of the EMFs, HPLC graphs demonstrating the purification process for the $\text{Sc}_3\text{N}@C_{80}$; Tauc plots and UV-Visible spectra for $\text{Er}_3\text{N}@C_{80}$ and $\text{Sc}_3\text{N}@C_{80}$; EPR of the $\text{Sc}_3\text{C}_2@C_{80}$; details on the experimental methods, further examples of individual measurements and complementary analysis of the results; details on the theoretical methods, and complementary results from the theoretical calculations are available free of charge via the internet. See DOI: <https://doi.org/10.1039/d1nh00527h>

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the potential of molecular-scale devices for converting waste heat into electricity.^{1–41} This is achieved *via* the Seebeck effect, which converts a temperature difference ΔT into a voltage $\Delta V = -S\Delta T$, where the constant of proportionality is the Seebeck coefficient S . Measured values of S are currently too low to create an economically viable technology and therefore there is a need to develop strategies for rapidly assessing the ability of a given molecule to deliver high values of S . In the 1980s, studies of sample-to-sample variations in electrical conductance G of mesoscopic, phase-coherent solids led to remarkable discoveries such as universal conductance fluctuations (UCFs), in which the standard deviation σ_G in the conductance is $\sigma_G = aG_0$, where $G_0 = \frac{2e^2}{h} \approx 77 \mu\text{S}$ is the quantum of conductance.^{42,43} In this expression, a is a number of order unity, which is independent of the average value of G , and depends only on the presence of an applied magnetic field or spin-orbit scattering. Since that time, the field of single-molecule electronics has been established, in which the electrical conductor is a single molecule located between electrodes separated by a few nanometres. However, even though many experiments have confirmed that transport through single molecules is phase coherent,^{44–53} even at room temperature, information contained in fluctuations has been largely ignored. Here, our aim is to demonstrate that studies of fluctuations in single-molecule transport properties are of particular interest in the search for molecules with high thermoelectrical performance, because rare examples of junctions exhibiting extreme values of Seebeck coefficients are a proof of principle that such junctions can exhibit high thermopowers.

To demonstrate that fluctuations can be used to search for molecules with potential for high thermoelectric performance, we report a combined experimental and theoretical study of a family of endohedral metallofullerenes (EMFs), specifically chosen to exhibit large fluctuations. EMFs are chosen because they can form single-molecule junctions without the need for anchor groups, which in the literature, are often used to bind molecules to electrodes. Such anchor groups would restrict the number of binding configurations within a junction and reduce fluctuations, whereas in this study, we would like to explore the full phase space of molecular configurations within a junction.

EMFs represent a fascinating class of nanomaterials, whose optical,⁵⁴ electrochemical⁵⁵ or magnetic⁵⁶ properties are controlled by the type of metal atom(s) encapsulated by the fullerene cage. For example, erbium containing EMFs possess a characteristic 1520 nm emission associated with the erbium ion, which is of fundamental importance for telecommunication applications and the fabrication of erbium doped amplifiers.⁵⁷ Moreover, there is a broad spectrum of scandium-containing metallofullerenes, including the paramagnetic $\text{Sc}@\text{C}_{82}$ and $\text{Sc}_3\text{C}_2@\text{C}_{80}$, which can find applications as spin probes and quantum information processing components.⁵⁸ The latter exhibits a unique diamond shaped EPR, with respect to the three equivalent scandium atoms (see Fig. S4, ESI†). For the purpose of probing fluctuations in transport properties, these molecules are attractive, because they not only bind to electrodes with a range of different orientations,

but also by making different choices for the encapsulated moiety, they can possess highly inhomogeneous charge distributions (for example as in the case of $\text{Sc}_3\text{C}_2@\text{C}_{80}$) or possess a relatively homogeneous charge distribution (as for C_{60}). Therefore, a study of their transport properties is expected to reveal how charge distributions and geometry are correlated with fluctuations in their Seebeck coefficients and their ability to exhibit high values of S .

To investigate correlations between Seebeck coefficient, geometry and charge inhomogeneity, our aim is to compare transport properties of $\text{Sc}_3\text{N}@\text{C}_{80}$ and C_{60} (ref. 11), with those of two other trimetallic EMFs, namely, the paramagnetic $\text{Sc}_3\text{C}_2@\text{C}_{80}$ (see Fig. S4, ESI†) and a rare-earth-based EMF, $\text{Er}_3\text{N}@\text{C}_{80}$. Like $\text{Sc}_3\text{N}@\text{C}_{80}$, these EMFs are formed with the fullerene cage $I_h\text{-C}_{80}$, composed of 80 C atoms organised in an icosahedral structure (I_h symmetry also shared by C_{60}) (see Fig. 1a)^{57–61} and their cages possess sizeable charge inhomogeneity. Our main result is that the standard deviations of Seebeck coefficients of these four molecules are strongly correlated with standard deviations in the charge distributions on their fullerene cages, thereby establishing a structure–function relationship between Seebeck fluctuations and charge inhomogeneity. Furthermore, molecules with high charge inhomogeneity exhibit rare examples of high thermopower, which means that if these rare junction configurations could be isolated and controlled, then such molecules have the potential to deliver high-performance thermoelectricity.

Results and discussion

All molecules were deposited on gold (111) substrates by drop-casting from very dilute 1,2,4-trichlorobenzene (TCB) solutions of the corresponding EMF. The Au substrates were previously flame annealed to form atomically flat surfaces (see ESI† for more details on sample preparation). We perform our measurements in ambient conditions and at room temperature using a home-built scanning tunnelling microscope (STM) adapted to measure simultaneously the conductance G and Seebeck coefficient S of single-molecule junctions with our previously reported technique^{4,11} (see ESI† for more details).

To localize individual molecules on the Au surface, we proceed by recording STM images of successively smaller areas.^{4,11} Two typical STM images of individual scandium carbide and erbium nitride EMFs can be seen in Fig. 1b and additional images can be found in the ESI.† To form the molecular junction, the tip is then placed on top of the fullerene and is approached until a certain conductance threshold is reached. As the tip approaches, its motion is stopped every few picometers and a small voltage ramp (± 10 mV) is applied, recording I - V curves (measured current I vs. applied voltage V). When a temperature difference of $\Delta T = 34$ – 42 K is established between the tip and the substrate, I - V curves show a ΔT -dependent voltage offset at zero current, which is identified with the thermovoltage V_{th} of the fullerene junction and this one with the Seebeck coefficient $S = V_{\text{th}}/\Delta T$. Furthermore, from the slope





Fig. 1 Molecules studied and their STM characterization. (a) Schematic of the three endohedral metallofullerenes (EMFs) studied, namely, $\text{Sc}_3\text{N}@C_{80}$, $\text{Sc}_3\text{C}_2@C_{80}$, and $\text{Er}_3\text{N}@C_{80}$. (b) STM images of the new molecules investigated (top panel: $\text{Sc}_3\text{C}_2@C_{80}$; bottom panel: $\text{Er}_3\text{N}@C_{80}$) deposited on gold substrates. Image size: $8.8 \times 8.8 \text{ nm}^2$ (top panel); $8 \times 8 \text{ nm}^2$ (bottom panel). Bias voltage $V_{\text{bias}} = 0.8 \text{ V}$ and 0.5 V , respectively (tip grounded). (c and d) Two individual examples of simultaneous measurements of conductance G (in red) and Seebeck coefficient S (in blue) as a function of the tip displacement ΔZ for single-molecule $\text{Sc}_3\text{C}_2@C_{80}$ (c) and $\text{Er}_3\text{N}@C_{80}$ (d) junctions. Each example shows a complete approach-retraction cycle, as indicated by the arrows (approaches plotted in the top panels; retractions, in the bottom panels). Data points highlighted in yellow indicate the contact regime, that is, the tip displacement while the molecule is directly in contact with both electrodes. In both examples, $V_{\text{bias}} = 0.01 \text{ V}$. $G_0 = (2e^2)/h$ is the quantum of conductance and ΔT is the temperature difference applied.

of the I - V curves, we obtain simultaneously the electrical conductance, $G = I/V$ (see ESI† for a detailed description of the technique).

Two examples of simultaneous conductance G and Seebeck coefficient S approach-retraction cycles, as a function of the tip displacement ΔZ , are shown in Fig. 1c and d for single-molecule junctions of the new EMFs studied. As the tip approaches the molecule (top panels in Fig. 1c and d, ΔZ decreases from zero; G plotted in red; S , in blue), the logarithmic conductance of the junction increases linearly with the distance (in semilogarithmic scale) until a physical contact is made with the corresponding EMF and a sudden jump in the signal is measured. The tip moves towards the fullerene for a further $\sim 0.7 \text{ nm}$, to establish a better tip-molecule contact (the contact regime during the tip approach in Fig. 1c and d has been shaded yellow for clarity). Simultaneous thermovoltage measurements allow us to obtain the thermopower of each single-molecule junction as it is formed. Fig. 1c shows an example of a junction with an initially positive Seebeck coefficient, while the junction in Fig. 1d has an initially negative Seebeck coefficient. A jump in Seebeck coefficient is also typically observed when physical contact is established (see Fig. 1c). Additionally, in the curves shown in Fig. 1c and d, we observe a clear correlation between small jumps observed in the conductance and the simultaneous thermopower signals once the contact is formed, showing that both magnitudes are sensitive to small changes in the junction geometry, such as atomic rearrangements in the electrodes.^{62–64} During retraction, the curves remain almost the same, indicating that the junctions were unaltered (bottom panels in Fig. 1c and d). In particular, their conductance shows generally a change of the slope or a jump down, although some hysteresis is observed and the features of both G and S retraction traces are usually slightly shifted in ΔZ with respect to the approach curves,

possibly due to elastic deformations of the Au atoms in the electrodes.^{62–64} Further examples of individual junctions (see the ESI†) show both signs of the Seebeck coefficient in the case of the two EMFs analysed, demonstrating their bi-thermoelectric nature, with either positive or negative values of S depending on the single molecule studied. This behaviour is similar to the reported bi-thermoelectricity of $\text{Sc}_3\text{N}@C_{80}$ junctions and is in clear contrast to the consistently negative Seebeck coefficient values obtained for the C_{60} .¹¹

The bi-thermoelectric response of the scandium nitride, scandium carbide and erbium nitride are compared with the more conventional C_{60} behaviour in Fig. 2a and b, which shows 2D histograms of the Seebeck coefficient S as a function of the conductance G (the data of C_{60} and $\text{Sc}_3\text{N}@C_{80}$ junctions are those previously reported in ref. 11). While the histograms in Fig. 2a are formed with all the experimental data obtained from the complete approach traces (*i.e.* the whole tip movement from tunnelling until the selected conductance threshold), for the histograms in Fig. 2b, only G and S values at first-contact between the tip and molecule are collected, *i.e.* values measured within 0.1 nm after the contact in the $G(\Delta Z)$ curve, where pressure from the tip does not appreciably deform the fullerene-electrodes junction. These histograms, and especially those of first-contact values, show that $\text{Sc}_3\text{N}@C_{80}$ and $\text{Er}_3\text{N}@C_{80}$ present smaller G values than $\text{Sc}_3\text{C}_2@C_{80}$ and C_{60} systems. Furthermore, these two trinitrate molecules present both positive and negative Seebeck coefficients mostly centred at zero. In contrast, the Seebeck coefficient of $\text{Sc}_3\text{C}_2@C_{80}$ junctions seems to be preferentially displaced towards negative values, despite also showing both signs and a considerable dispersion. C_{60} junctions, as previously reported,^{4,11} present only negative values of S .

An additional striking feature of the histograms in Fig. 2b is the broad distribution of Seebeck coefficients obtained for





Fig. 2 Conductance G and Seebeck coefficient S histograms of the monomers of EMFs and C_{60} . (a) S vs. G 2D histograms built with all the data from the I – V curves measured during the complete approach of the tip, from the noise level until close to the metallic contact. The area shadowed in yellow corresponds approximately to the contact regime (highlighted also in yellow in Fig. 1c and d). (b) S vs. G 2D histograms built with first-contact values, *i.e.*, within 0.1 nm right after junction formation. In a and b, each column corresponds to one molecule, as indicated on top; from left to right: $Sc_3N@C_{80}$, $Sc_3C_2@C_{80}$, $Er_3N@C_{80}$ and the fullerene C_{60} . The colour scale accounts for the number of points in each histogram.

$Sc_3C_2@C_{80}$ and $Sc_3N@C_{80}$, compared with the narrower distributions obtained for $Er_3N@C_{80}$ and C_{60} . This can be more easily observed in the 1D histograms of G and S shown in Fig. 3a and b for all four molecules. Fig. 3b and 2b show that all the EMFs present a bi-thermoelectric behaviour, signalled by the presence of both positive and negative values of S , in clear contrast with the empty fullerene C_{60} that exhibits only negative Seebeck coefficients. The mean Seebeck coefficients \bar{S} of $Sc_3N@C_{80}$ and $Er_3N@C_{80}$ junctions are very small, while in the case of $Sc_3C_2@C_{80}$ (in green in Fig. 3b), two peaks appear in the 1D histogram, centred at $0 \mu V K^{-1}$ and $-20 \mu V K^{-1}$, which means that it is equally likely to connect a molecule with a small value of S close to zero or with a negative value similar to that of C_{60} . Finally, $Sc_3N@C_{80}$ and $Sc_3C_2@C_{80}$ have a larger dispersion of S values than $Er_3N@C_{80}$. The Seebeck coefficient measurements of the latter are spread only over a range of approximately $-20 \mu V K^{-1}$ to $+20 \mu V K^{-1}$, whereas the range of values of the EMFs with three scandium atoms is twice as wide. The standard deviation σ_S in the Seebeck coefficients obtained from Fig. 3b for all the compounds

Table 1 Columns 2 and 3 show standard deviations σ of the distance d , associated with rotations about axes θ , Φ . Standard deviations of charge (columns 4 and 5) and Seebeck coefficients (columns 6 and 7) of $Sc_3C_2@C_{80}$, $Sc_3N@C_{80}$, $Er_3N@C_{80}$ and C_{60} . The sixth column shows experimental values for σ_S from the Seebeck distributions in Fig. 3b. The right-most column shows theoretical values for σ_S from Table S6 of the ESI. Computation of a theoretical value for σ_S of $Er_3N@C_{80}$ was not possible, due to the presence of f -electrons

Molecule	σ_θ	σ_Φ	σ_q Hirshfeld	σ_q Voronoi	σ_S Exp. ($\mu V K^{-1}$)	σ_S Theor. ($\mu V K^{-1}$)
$Sc_3C_2@C_{80}$	1.47	1.09	113×10^{-4}	133×10^{-4}	19.2	36.0
$Sc_3N@C_{80}$	1.43	0.95	109×10^{-4}	119×10^{-4}	17.6	29.4
$Er_3N@C_{80}$	1.43	0.95	26×10^{-4}	27×10^{-4}	7.7	==
C_{60}	==	==	5.7×10^{-4}	8.5×10^{-4}	6.8	2.44

are shown in column 6 of Table 1, (Note that the error in the determination of the thermopower is less than $0.9 \mu V K^{-1}$).

To understand the origin of these differences in Seebeck fluctuations, we used density functional theory (DFT) to examine the binding energies of these junctions under various rotations. The basic theoretical principles of the method can be found in ref. 65 and 66. There are an infinite number of inequivalent orientations of the metallic moieties relative to their fullerene cage. Here we consider a selection of rotations about high-symmetry axes. Fig. 4 shows four axes of rotation θ , Φ , α and β considered below, while Fig. 5 shows how these axes pass through the metallic moieties. For each of the four axes of rotation, we investigate how the total energy varies with angle of rotation, both in presence and absence of the gold substrate. For each of these four axes, we consider one mode of rotation in the gas phase and three modes of rotation on a substrate.

In the gas phase, we consider rotation of the metallic moiety relative to a fixed fullerene cage. For this rotational mode, rotations about the β and Φ axes are equivalent, so in the gas phase, rotations about 3 distinct axes are investigated. In the presence of



Fig. 3 Experimental first-contact histograms of $\log(G/G_0)$ and S . (a) $\log(G/G_0)$ and (b) S 1D histograms for all the molecules, showing only the values right after junction formation (at first contact). The data shown in these 1D histograms is the same as in Fig. 2b.



Fig. 4 Illustration of the four rotation axes: θ and Φ are horizontal axes, α and β are vertical axes. This Figure shows how the axes pass through the I_h - C_{80} cage + metallic moiety.



Fig. 5 Illustration of how the four rotation axes θ , Φ , α and β , pass through the metallic moiety (in the absence of the cage).

the gold substrate, we consider three modes of rotation about each of the four axes: rotation of the bare metallic moiety (in the absence of the cage), rotation of the metallic moiety in the presence of a fixed cage, and rotation of both the metallic moiety and cage, such that their relative orientation is fixed. This means that in total, on a substrate, 12 distinct cases are considered (4 axes \times 3 modes of rotation). For simplicity of notation, in what follows, the same symbol (e.g. θ) is used to label both the rotation axis and the angle of rotation about the axis.

To quantify geometrical variations under rotation, it is convenient to define the distance d to be the smallest vertical distance between the top-most plane of the Au-substrate and the closest metal atom of the encapsulated moiety. Fig. S12 (ESI†) shows the variation of d with rotations about the above axes, while Table 1 (columns 2 and 3) shows the standard deviations σ_θ , σ_Φ in d under rotations about the axes θ , Φ . By symmetry $\sigma_\alpha = \sigma_\beta = 0$ and therefore these follow the order $\sigma_\theta > \sigma_\Phi > \sigma_\alpha, \sigma_\beta$. Furthermore, the values of both σ_θ and σ_Φ associated with $Sc_3C_2@C_{80}$ are higher than those associated with $Sc_3N@C_{80}$ and $Er_3N@C_{80}$ (for more detail see Table S2 in the ESI†).

To quantify the charge inhomogeneities associated with these molecules, we also computed the standard deviations

σ_q of the charge distributions on their cages, using two different definitions of charge (Hirshfeld and Voronoi).^{67,68} If the charge on atom i of the cage is $|e|n_i$, then the standard deviation σ_q in the charge is defined by:

$$\sigma_q^2 = \langle (n_i - \langle n_i \rangle)^2 \rangle$$

where angular brackets denote an average over all atoms on the cage. These are presented in Table 1 (columns 4 and 5) and reveal that $Sc_3C_2@C_{80}$ and $Sc_3N@C_{80}$ possess relatively high values of σ_q , followed by $Er_3N@C_{80}$. In contrast the charge distribution of C_{60} is relatively uniform. Comparison between the experimental values for σ_S in the sixth column and the theoretical values for σ_S in the right-most column (obtained as described below and copied from Table S6 of the ESI†) shows that standard deviations in the Seebeck coefficients are strongly correlated with the structural and charge standard deviations σ_θ , σ_Φ and σ_q . Table 1 is a key result of our study and reveals the origin of the differences in Seebeck fluctuations between the four studied molecules. Interestingly, the experimental and theoretical values of σ_S for C_{60} shown in Table 1 are comparable with those found for thiol-terminated oligophenylenes reported in ref. 69, which varied from 2.1 to 3.2 as the number of phenyl rings was increased from 1 to 3. They are higher than those found for thiol-terminated alkyl chains, which were found to be 0.3 ± 1 and almost independent of length.⁷⁰ Studies^{71–74} have also shown that increasing the applied temperature difference can lead to additional fluctuations in the thermovoltage.

As a first step in the calculations leading to Table 1, energy barriers to rotation about the above axes were computed, to obtain the preferred angles of rotation, which minimise the total energy. σ_S was then obtained by computing Seebeck coefficients using an ensemble of angles and a distribution of Fermi energies E_F , as described in sections 5.7 to 5.15 of the ESI†. Table 2 shows the computed gas-phase energy barriers $\Delta E(\theta)$, $\Delta E(\Phi)$, $\Delta E(\alpha)$ associated with rotations about θ , Φ , α of Sc_3C_2 , Sc_3N and Er_3N within the fixed I_h - C_{80} cage, along with the preferred orientation angles for the three rotation axes (see Section 5.5 in the ESI†). This reveals that for all three axes, $Sc_3C_2@C_{80}$ possesses the highest energy barriers to rotation, followed by $Sc_3N@C_{80}$, with $Er_3N@C_{80}$ possessing the lowest barriers. These correlate closely with their values of σ_q^2 and furthermore, the highest energy barrier associated with $Sc_3C_2@C_{80}$ correlates with the high value of the standard deviation in d .

Table 2 shows that for all EMFs, in the gas phase, the energy barriers to rotation follow the order $\Delta E(\theta) > \Delta E(\Phi) > \Delta E(\alpha)$. Fig. S21 (ESI†) shows that on a gold surface, the binding

Table 2 Gas phase energy barriers $\Delta E(\theta) > \Delta E(\Phi) > \Delta E(\alpha)$ associated with rotations about θ , Φ , α of Sc_3C_2 , Sc_3N and Er_3N within the fixed I_h - C_{80} cage, along with the preferred angles of rotation for the three rotation axes (according to Fig. S15, ESI)

EMF	$\Delta E(\theta)$ (meV)	Preferred orientations (θ)	$\Delta E(\Phi)$ (meV)	Preferred orientations (Φ)	$\Delta E(\alpha)$ (meV)	Preferred orientations (α)
$Sc_3C_2@C_{80}$	400	0°, 190° and 360°	350	0°, 110°, 180°, 210°, 270° and 360°	300	0°, 75°, 160°, 240° and 320°
$Sc_3N@C_{80}$	300	0°, 190° and 360°	250	10°, 95°, 200°, 280° and 340°	150	0°, 75°, 100°, 150°, 200°, 250°, 300° and 350°
$Er_3N@C_{80}$	130	0°, 190° and 360°	100	0°, 120°, 250° and 350°	60	0°, 80°, 150°, 225°, 320° and 360°



energies are also higher for the θ and Φ axes and follow the order $\Delta E(\theta) > \Delta E(\Phi) > \Delta E(\alpha) > \Delta E(\beta)$. Furthermore the barriers are higher for $\text{Sc}_3\text{C}_2@\text{C}_{80}$ and lower for $\text{Er}_3\text{N}@\text{C}_{80}$. On a gold substrate, the energy barriers significantly increase to approximately 0.6 eV for $\text{Sc}_3\text{C}_2@\text{C}_{80}$, which means that the configurational degrees of freedom of the endofullerene relative to the electrode exhibit a small number of preferred orientations corresponding to minima in the total energy of the endofullerene/electrode complex as shown in Fig. S21 (ESI†) (for more detail see section 5.6 of the ESI†). The energy barrier for $\text{Sc}_3\text{N}@\text{C}_{80}$ is 0.3 eV and the lowest barrier of 0.1 eV is obtained for $\text{Er}_3\text{N}@\text{C}_{80}$ (see Fig. S18, ESI†). (Note: Fig. S21, ESI† shows the four rotation axes, while Fig. S18, ESI† shows only θ).

To obtain the theoretical values of σ_S in Table 1 from the above distribution of orientations, we used a combination of the quantum transport code Gollum⁷⁵ and the density functional theory (DFT) code SIESTA,⁷⁶ to calculate the transmission coefficient $T(E)$ for electrons of energy E passing through the EMFs, when contacted to gold electrodes (a detailed description can be found in the ESI†). The energy dependence of transmissions curves for $\text{Sc}_3\text{C}_2@\text{C}_{80}$, $\text{Sc}_3\text{N}@\text{C}_{80}$ and C_{60} , and their Seebeck coefficients for all minimum-energy orientations, are shown in Fig. S23 to S44 (ESI†). Since the curves are rather smooth on the scale of $k_B T$ at room temperature, their electrical conductance G is approximated by $G \approx T(E_F)G_0$, where G_0 is the quantum of conductance, and $S(E_F) \approx -S_0 \left(\frac{d \ln T(E)}{dE} \right)_{E=E_F}$.⁶⁵ In the latter equation, if E is

measured in units of electron volts, then $S_0 = \alpha T$, which at $T = 300$ K, takes the value $S_0 = 7.3 \mu\text{V K}^{-1}$. Fig. S23 to S44 (ESI†) clearly show that rotations about the θ axis cause much more severe fluctuations in the transmission functions than rotations about the other axes. By sampling the transmission curves over a range of Fermi energies and preferred orientations (as described in the ESI†), we obtain the theoretical 2D histograms shown in Fig. S45, (ESI†) from which the theoretical values of σ_S in Table 1 were obtained. In general, depending on the value of E_F , the Seebeck coefficient can achieve high values when the distance d between the metallic moiety and the Au substrate is large (e.g., for $\theta = 90^\circ$, in Fig. S22, ESI†), whereas S tends to be lower when d is small (e.g., for $\theta = 0^\circ$, in Fig. S22, ESI†).

The high values of σ_S and the bi-thermoelectric response of these EMF junctions confirm our initial intuition that charge inhomogeneity leads to larger variability in the transport properties, compared with pristine C_{60} . This interesting behaviour is also observed in ref. 77, where $\text{Gd}@\text{C}_{82}$ and $\text{Ce}@\text{C}_{82}$ and the empty C_{82} were reported to present mainly negative Seebeck coefficients, with occasional positive values that were attributed possibly to *meta*-stable configurations of the junctions. The values reported for the Seebeck coefficient show an enhancement in the endohedral fullerenes with respect to the C_{82} . Compared to our investigated EMFs, the main difference with the systems studied in ref. 77 is the total number of metal atoms inside the fullerene cage, which in the case of ref. 77 is just one atom placed out of centre.

Of particular interest for quantifying the thermoelectric performance of molecular junctions is the power factor GS^2 .

Fig. 6 shows a comparison between the experimental and theoretical power factor distributions. These possess comparable shapes and, in both cases, reveal that rare values corresponding to particular configurations of the junctions can lead to very high-power factors.

Fig. 6 also shows that the high conductance and Seebeck coefficient values measured for the carbide EMF translate into quite high values of power factor GS^2 , with the highest power factor values of $\text{Sc}_3\text{C}_2@\text{C}_{80}$ significantly exceeding those of C_{60} , both in value and probability.

Lee *et al.*⁷⁷ noted that their EMF-based junctions possessed a high single molecule power factor of 16.2 fW K^{-2} for $\text{Gd}@\text{C}_{82}$, which corresponds to around $4 \times 10 \mu\text{W K}^{-2} \text{ m}^{-1}$ for a thin-film device consisting of a monolayer of $\text{Gd}@\text{C}_{82}$, which at the time of publication was the largest power factor obtained for a single molecule device. From the first-contact data in Fig. 3, we find even higher values up to 50 fW K^{-2} for $\text{Sc}_3\text{N}@\text{C}_{80}$ and $\text{Sc}_3\text{C}_2@\text{C}_{80}$ junctions, and even two examples of values up to $70\text{--}80 \text{ fW K}^{-2}$ (for these two EMFs). Statistically, we find larger values for the carbide compound ($\text{Sc}_3\text{C}_2@\text{C}_{80}$ junctions). Furthermore, considering all the measured conductance and Seebeck coefficient values (Fig. S9, ESI†), the power factor can be statistically increased when the junction becomes slightly compressed, because under compression, G increases and S becomes more negative.

Conclusions

Through a combination of experimental STM-based transport measurements and DFT-based transport calculations, we have studied the electrical conductance and Seebeck coefficient of three endohedral metallofullerenes (EMFs) $\text{Sc}_3\text{N}@\text{C}_{80}$, $\text{Sc}_3\text{C}_2@\text{C}_{80}$, and $\text{Er}_3\text{N}@\text{C}_{80}$, and found that all EMFs studied are bi-thermoelectric systems, exhibiting both positive and negative Seebeck coefficients, in contrast to the empty C_{60} . To understand the origin of these fluctuations in thermoelectric properties, DFT simulations of a selection of orientations were carried out, that allowed us to answer the following questions:

1) Can variations in the orientations of the fullerenes and their encapsulated moieties lead to fluctuations in Seebeck coefficients on the scale of those measured experimentally? DFT modelling shows that fluctuations on the scale of the measured values can indeed arise from variations in the orientations of the fullerenes and their encapsulated moieties and the degree of variation is characterised by the associated standard deviation σ , which is a purely geometric quantity. This is an important result, because it shows that such fluctuations are an intrinsic property of the junctions and not due to some unknown extrinsic effect.

2) Are there correlations between Seebeck fluctuations and the degree of charge inhomogeneity? To establish these correlations, in Fig. 4, 5 and Fig. S15, S21, Table S3, (ESI†) we explored a large number (360×3 axes in gas phase and 360×4 axes onto a gold surface) of different geometries and identified the most energetically favourable configurations, which are





Fig. 6 Power factor GS^2 analysis. a) Experimental 1D histograms of power factor at first contact, built with the data in Fig. 3. The inset zooms into the details of the main panel. b). Theoretical 1D histograms of power factor obtained from Fig. S46 of the ESI.†

relevant experimentally. These calculations demonstrate a clear correlation between Seebeck fluctuations and the degree of charge inhomogeneity, which can be characterised by a standard deviation σ_q .

We found that standard deviations in the Seebeck coefficients σ_S of EMF-based junctions are correlated with the geometric standard deviation σ and the charge inhomogeneity σ_q . We benchmarked these molecules against C_{60} and found that all of σ_q, σ_S are the largest for $Sc_3C_2@C_{80}$, all are the smallest for C_{60} and for the other EMFs, they follow the order $Sc_3C_2@C_{80} > Sc_3N@C_{80} > Er_3N@C_{80} > C_{60}$. This means that *external* measurements of fluctuations in the Seebeck coefficient provide insight into the *internal* structure and charge distribution of endohedral metallofullerenes.

This points the way to designs of molecular switches and bi-thermoelectric materials, because molecules with large values of σ_q possess large values of σ_S ; the former is a sign that a molecule can switch between orientation-dependent electrical conductances, whereas the latter indicates that a molecule can exhibit a wide range of Seebeck coefficients with both positive and negative signs. If the orientation and Fermi energy of such molecules can be controlled, then high thermoelectric performance is possible. On the other hand, if the aim is to minimise fluctuations, then our study suggests that molecules with low values of σ and σ_q should be selected.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

P. D. acknowledges support from the Hellenic Foundation for Research and Innovation (Grant number: 1468). A. K. I. acknowledges the Leverhulme Trust for Early Career Fellowship ECF-2020-638. A. K. I. and A. A. are also grateful for financial assistance from Tikrit and Anbar Universities (Iraq), and the Iraqi Ministry of Higher Education (SL-20). T. A. is grateful for financial assistance from Al-Jouf University (Saudi Arabia), and the Saudi Ministry of Education. A. K. I. and C. J. L. acknowledge financial support from the UK EPSRC, through Grant no. EP/M014452/1, EP/P027156/1 and EP/N03337X/1. This work was

additionally supported by the European Commission is provided by the FET Open project 767187–QuIET. N. A. and G. R.-B. acknowledge the Spanish MINECO (grants MAT2014-57915-R, MAT2017-88693-R, and MDM-2014-0377, and the María de Maeztu Programme for Units of Excellence in R&D MDM-2014-0377 and CEX2018-000805-M) and Comunidad de Madrid (grants NANOFrontMAG-CM, S2013/MIT-2850, and NANOMAGCOST-CM, P2018/NMT-4321). L. R.-G. acknowledges financial support from the Spanish MECD (Grant no. FPU14/03368). N. A. and L. R.-G. are also grateful for financial support from the Education and Research Council of the Comunidad de Madrid and the European Social Fund (Ref. PEJD-2019-POST/IND-16353).

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