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Recent advances of zwitterionic-based topological polymers for biomedical applications

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Zwitterionic polymers, comprising hydrophilic anionic and cationic groups with the same total number of positive and negative charges on the same monomer residue, have received increasing attention due to their distinctive physico-chemical properties, including high hydration, good biocompatibility, good antifouling and good antibacterial properties, and thus show great potential as a novel type of biomedical material. Unique topological polymers with block, star, dendritic, comb, brush and cyclic structures not only meet the requirements of material structure accuracy but also provide a theoretical research template for the development of materials science. Topological structures can facilitate polymers with precise and adjustable structures, hydrodynamic size, lower viscosity, more active sites, less intermolecular entanglement, and increased colloidal stability, which can effectively achieve drug encapsulation, improve the drug-loading efficiency, achieve faster internalization, and reduce the *in vitro* cytotoxicity. Therefore, the synthesis of topological polymers with precise structures has become an inevitable requirement for the development of materials and their biomedical applications. The article mainly reviews the properties and application of zwitterionic polymers and their derivatives with different topological structures. In particular, the recent progress of these polymers in drug delivery, antitumor properties, biomedical diagnosis and antifouling coatings are described and introduced. Finally, the current problems in these applications are discussed, and the potential research prospects of zwitterionic-based topological polymers in current research are put forward.

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1. Introduction

In recent years, zwitterionic polymers have become a hot subject of research in the field of biomaterials because of their unique structure, easy functionalization and ultra-low contamination. Zwitterionic materials are charge-neutral ionic systems, which possess both anionic and cationic components with equal charge on the same monomer residue.^{1–4} Fig. 1 illustrates the most commonly used zwitterionic moieties, including quaternary ammonium,⁵ phosphate,^{6,7} carboxylate or sulfonate.^{8–10} Compared with non-ionic hydrophilic polymers (such as polyethylene glycol, polyvinyl alcohol and poly(2-methyl-2-oxazoline)), zwitterionic polymers can bind a large number of water molecules *via* strong electrostatic force-induced hydration, endowing the material surface with superhydrophilicity. As for zwitterionic species, charge neutrality and high surface hydration are generally considered to be the key to resisting non-specific protein adsorption, thus effectively preventing contamination from proteins.¹¹

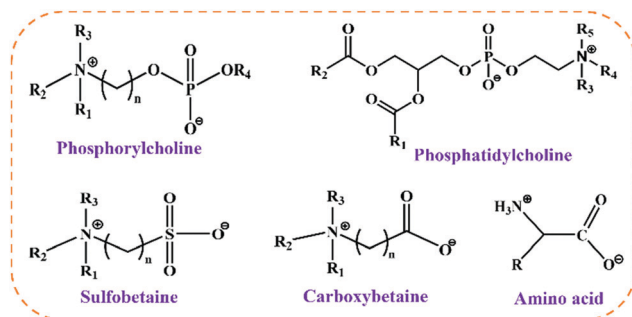


Fig. 1 Zwitterionic moieties and their chemical structures.

Polyethylene glycol (PEG) is a kind of non-toxic and biocompatible polymer. PEG can reduce protein immunogenicity and improve pharmacokinetics, becoming the gold standard of antifouling materials and being widely used in biomedical research.^{12–15} However, it is reported that PEG is easily oxidized and decomposed. Besides, it has poor long-term stability and difficult natural degradability. Unfortunately, these drawbacks greatly limit the practical application of PEG.^{16–18} Moreover, PEGylated drugs can induce the production of anti-PEG antibodies and lead to the rapid clearance of drugs *in vivo*.^{19,20} Several research groups have proved that anti-PEG antibodies

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are closely related to the accelerated blood clearance (ABC) of the second injection of PEG liposomes.^{21,22} In addition, several animal experiments have shown that PEG liposomes can activate the complement system of animals and may cause allergic reactions.^{23,24} Therefore, specific attention is given to the immunogenicity of PEG. In recent decades, the emerging zwitterionic polymers have been considered to be the most meaningful substitutes to PEG. Compared with the amphiphilic nature of PEG, zwitterionic polymers have the advantages of superhydrophilicity, good thermal stability and biocompatibility. At present, the most studied zwitterionic polymers including phosphorylcholine (PC), carboxybetaine (CB), sulfobetaine (SB), *etc.*,^{25–28} which can maintain their stability, prolong the circulation time *in vivo*, avoid immunogenicity and produce antibodies, demonstrate good application prospects in the fields of drug delivery, antitumor therapy and antifouling coating.^{29–31}

Traditional linear zwitterionic polymers can be prepared through the direct polymerization of zwitterionic monomers. Although its synthesis is simple and straightforward, the function of linear zwitterionic polymers is specific. They may self-assemble to form micelles, vesicles and nanoparticles with a large size, poor stability, and challenging controllability; thus, it is difficult to obtain more defined structures. With the development of polymerization technology, in order to meet the ever-increasing performance requirements of polymer materials, a variety of zwitterionic-based topological polymers, such as block-based, star-shaped, hyperbranched, dendritic, brush and cyclic polymers have sparked the attention of researchers'. This type of polymer, with the merits of controllable synthesis, good stability, small size, and convenient and efficient multi-functionalization, has been widely used in biomedicine. However, the existing zwitterionic polymer-based materials can no longer meet the complex and changeable microenvironment, which limits their applications in some specific fields. Thereby, how to further elegantly integrate zwitterionic materials with other stimuli-responsive materials to create novel materials with virtues from both moieties has become the focus of scientific research, and zwitterions with other functional monomer-based topological polymers could be a good choice. At present, although many reviews have summarized zwitterionic polymers for biomedical applications, to the best of our knowledge, there has been no systematic review of the research progress in zwitterionic-based topological polymers. In this review, we introduce the zwitterionic-based topological polymers with different topological structures, describe their preparation process, and clarify their biomedical applications. In addition, the problems existing in the current research are also discussed, and their prospects are considered.

2. Zwitterionic-based topological polymers

Traditional linear polymers do not have branch points and their molecular chains are easy to entangle, the viscosity of the system increases rapidly with the increase in relative molecular

weight, the stability of self-assembled micelles is poor, and they are easy to disassemble after intravenous injection because the concentration is below the critical micelle concentration (CMC), resulting in the early release of encapsulated drugs, which greatly limits their functional application. Topological polymers, including but not limited to linear block,^{32,33} star-shaped,³⁴ dendritic³⁵ and brush structures,³⁶ have more precise structures, a lower viscosity, and smaller size (such as dendrimers), for which it is easy to achieve cell internalization and they can be used as drug-delivery carriers. In addition, their abundant terminal functional groups increase the active sites, making it easy to achieve targeting. Moreover, topological polymers have an increased colloidal stability (such as cyclic brush polymers), which can achieve good antitumor effects. With the development of biomaterials science, the synthesis of zwitterionic polymers with different topological structures has attracted extensive attention, which has had an important influence on the polymer performance, regulated the topological structure in polymer synthetic chemistry, and further promoted the development of the biomedical field. In the following, we will focus on the characteristics and synthesis of each topological structure in the remainder of Section 2 and clarify their biomedical applications in Section 3.

2.1 Zwitterionic-based block polymers

As the demand for material properties increases, polymers are gradually moving towards the direction of refined structure synthesis, and the preparation of polymers with precise structures through controllable synthesis has become the development trend. In the past few decades, the self-assembly of zwitterionic block copolymers has attracted great attention in the scientific community. For example, Ye *et al.*³⁷ synthesized the controlled architecture of a dual-functional block polymer by sequentially grafting a quaternary ammonium salt polymer and a zwitterionic polymer *via* surface-initiated atom transfer radical polymerization (SI-ATRP). Quaternary ammonium salt polymers reduce the excessive propagation of bacteria, thus exerting antibacterial activity to prevent the formation of biofilms as well as the adhesion of biofouling, so that a comprehensive “defense” and “attack” against biofouling can be conducted. Fu *et al.*^{38–40} designed and constructed two polymer brushes with opposite hierarchical structures (SiPQAs-*b*-PSBMA and Si-PSBMA-*b*-PQAs) *via* the grafting of poly(sulfobetaine methacrylate) (PSBMA) and cationic bactericidal poly(quaternary ammonium salts) (PQAs). They found there were synergistic antibacterial effects between different alkyl chain lengths and different hierarchies. PSBMA also has good blood compatibility, especially anticoagulant properties, and thus plays a pivotal role as an anticoagulant material. Based on this, Zhao *et al.*⁴¹ prepared block and random zwitterionic polymers (P(SBMA-*b*-NaSS)) and (P(SBMA-*co*-NaSS)) for the first time by grafting PSBMA and sodium phenylethylsulfonate (NaSS) from a polysulfone (PSf) membrane through SI-ATRP (Fig. 2a). The results showed that, compared with PSf and PSBMA membranes, the modified zwitterion membranes had significantly enhanced anticoagulant properties and lower protein adsorption properties. This study

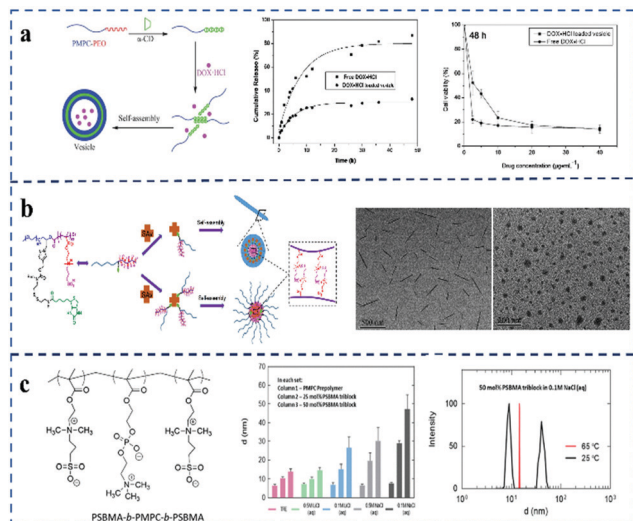


Fig. 2 Synthesis of zwitterionic-based block polymers. (a) Study on blood compatibility of the zwitterion-modified membrane.⁴¹ Reproduced from ref. 41 with permission from Elsevier. (b) Zwitterionic diblock bioconjugates.⁴⁹ Reproduced from ref. 49 with permission from American Chemical Society. (c) Average particle size of ZI-ABA in different solvents and 50 mol% PSBMA triblock copolymer in 0.1 M NaCl solution at different temperatures.⁵⁰ Reproduced from ref. 50 with permission from American Chemical Society.

has laid a foundation for the development of zwitterionic copolymer-modified membranes and has potential biomedical applications.

2-Methacryloyloxyethylphosphocholine (MPC) is currently one of the most widely used zwitterions, and the synthesis of biocompatible and non-toxic gene-delivery vectors is the key to the treatment of genetic diseases. For this purpose, Narain *et al.*⁴² copolymerized MPC with cations to synthesize di-block copolymers and random copolymers, which were then combined with DNA. Due to the different morphologies of DNA vectors formed *via* different polymerization methods, the gene-delivery efficiency was further affected. The results confirmed that the prepared block-based zwitterionic copolymer could deliver a gene vector more effectively in the presence or absence of serum *in vitro*. Therefore, polymer vesicles prepared from MPC-based materials have potential prospects as drug-delivery vehicles. Based on this, Ji *et al.*^{43,44} prepared a novel biocompatible polymer vesicle. Specifically, they first prepared PEO-PMPc diblock copolymers *via* ATRP and loaded doxorubicin (DOX) after reacting with α -cyclodextrin (α -CD). The *in vitro* results confirmed that the drug-loaded polymer vesicles can significantly reduce the cytotoxicity of the original drug and effectively release DOX at the tumor site.

Owing to the fact that phosphonic acid/ester can form a stable coordination with the metal/metal oxide surface, which provides a highly stable anchoring ability on various substrates and demonstrates excellent antifouling properties, phosphorus-group-based zwitterionic polymers are also of great significance in drug delivery, dental biomaterials and tissue engineering. Shen *et al.* developed a series of zwitterionic-phosphonate block polymers *via* controlled RAFT polymerization for antifouling

coatings on metal substrates.⁴⁵ The zwitterionic polymers not only have good synthesis controllability and use mild reaction conditions but can be stably coated on biomedical metals through PO-metal coordination bonds due to the existence of pDEMMP. They further systematically studied the relationship between the antifouling performance of the polymer-coated TC4 substrate and the chain lengths of pSBMA and pDEMMP in zwitterionic block polymers. The research results found that pDEMMP with only 5 chain lengths could form a stable zwitterionic polymer coating on the TC4 substrate, while pSBMA with a chain length of more than 40 could effectively inhibit protein adsorption, and cell and bacterial adhesion. Therefore, the biocompatible zwitterionic/phosphate block copolymer constructed by selecting the polymer with the optimal number of segments has excellent antifouling abilities and can be widely used in biomedical implants.

Zwitterionic poly(sulfobetaine) (PSB) is one of the most typical thermoresponsive polymer materials. The upper critical solution temperature (UCST) behavior of PSB is opposite to the lower critical solution temperature (LCST) behavior of many non-ionic polymers in aqueous solution. Moreover, the phase-transition temperature of PSB (CP_{PSB}) is very sensitive to the addition of salt to the polymer solution, while CP_{PNIPAM} is relatively insensitive to the addition of salt. Orthogonal dual thermoresponsive diblock copolymers (DBC) made from LCST-type PNIPAM or PNIPAM blocks and UCST-type PSB blocks have attracted attention due to their self-assembly behavior. In solution, they can form core-shell micelles, whose structure can be effectively reversed when the temperature is changed.^{46,47} Based on this dual stimulus-sensitive behavior, Kreuzer *et al.*⁴⁸ prepared a PSB-based DBC thin film (PSBP₂₄₅-*b*-PNIPAM₁₀₅) from saline solutions. Biological conjugates prepared by combining amphiphilic polymers and natural proteins have both temperature-dependent changes and enzymatic activity, and have been widely used in biological separation, drug delivery and enzyme biological engineering. However, the self-assembly behavior of bioconjugates has not been studied. Based on this, Zhao *et al.*⁴⁹ synthesized a zwitterionic block copolymer with biotin groups at the junction point through the combination of click chemistry, the thiol-disulfide exchange reaction and ATRP (Fig. 2b). The bioconjugates exhibit UCST-type thermosensitivity and can further self-assemble when induced by temperature. At temperatures lower than the UCST, the bioconjugates self-assembled in aqueous solution to form rod-like structures and spherical micelles, depending on the number of block copolymers grafted onto the protein molecules. When the conjugate contains 1.3 block copolymer chains, rod-like structures can be formed, while spherical micelles can be formed when the conjugate contains 2.9 copolymer chains. Therefore, this work provided a platform for the synthesis of conjugates with various topological structures. Panzer and co-workers synthesized a distinctive fully zwitterionic triblock copolymer (ZI-ABA) for the first time (Fig. 2c).⁵⁰ The fully ZI-ABA triblock polymers were synthesized through free radical polymerization with MPC as the middle block, and SBMA as the symmetrical end blocks. The ZI-ABA

copolymers exhibit self-assembly behavior, and thermal- and salt-responsive behavior in solution due to the strong aggregation of SBMA.

Poly(amino acid)s, with both cations and anions, belong to zwitterionic polymers, and also demonstrate wide biological applications. Wu *et al.* reported a series of dimethylaminopropyl aminated poly(α,β -L-aspartic acid) (DMPA-*g*-PASP) materials with anti-biofouling ability and good biocompatibility.⁵¹ Liu *et al.* developed a serine-based zwitterionic polymer (pSerMA), which was further grafted onto a gold surface *via* surface-initiated photopolymerization.⁵² The pSerMA-grafted surface can not only strongly inhibit the adsorption of proteins but also inhibit the adsorption from whole human serum and plasma. In addition, Gitsov and Schutz reported the “green” synthesis of unnatural poly(amino acid)s catalyzed by enzymes.⁵³ Specifically, they formed DL-tyrosine by polymerizing linear-dendritic laccase complexes with water. Studies have confirmed that the formed materials are completely water-soluble and exhibit typical polyzwitterionic pH-responsive behavior.

2.2 Star-shaped zwitterionic polymers

Star polymers are complex macromolecular polymers that have more than three linear connections radially from one branch point. Compared with linear analogues of the same molecular weight, star-shaped polymers possess unique hydrodynamic properties and encapsulation capabilities due to their three-dimensional spherical structure. In addition, star-shaped polymers exhibit a lower solution viscosity and smaller dynamic mechanical size in dilute solution due to less arm entanglement. Moreover, their internal and peripheral structures provide a convenient way to introduce different active functional groups, which has gradually become a research hotspot in the field of polymer science.⁵⁴ Up until now, star polymers have been widely used in the design of biomedical materials for drug delivery and imaging diagnosis.⁵⁵ For example, poly(amidoamine) (PAMAM) dendrimers have been widely studied in biomedical applications due to their good biocompatibility.⁵⁶ However, owing to the lack of any shielding effect on the positive charge of the dendrimer, the amino-terminated cationic PAMAM dendrimers exhibited concentration-dependent cytotoxicity. The cytotoxicity can be effectively reduced by modification with polymer, such as PEG, acryloxyethyl phosphorylcholine (APC) and carboxybetaine acrylamide (CBAA).^{57–62} Based on this, Li *et al.*⁶³ synthesized a novel, non-toxic, high-anticancer-activity star polymer anticancer drug-delivery vehicle (PAMAM-*g*-PDMAPS) by grafting PAMAM with poly[3-dimethyl(methacryloyloxyethyl)ammonium propanesulfonate] (PDMAPS).

High molecular weight polycations such as poly(2-dimethylamino)ethyl methacrylate have been used to construct gene-transfer vectors due to their strong DNA aggregation ability.⁶⁴ However, the cytotoxicity of the polycation limits its application. Compared with linear polycations, star-shaped polycations have a lower *in vitro* cytotoxicity but a higher transfection efficiency. For example, star-shaped polycations grafted with polyhedral oligomeric silsesquioxanes (POSS) have

good biocompatibility and a well-defined structure.^{65,66} Moreover, the hydrophobic part promotes gene aggregation, and improves drug loading and endocytosis. It has been reported that zwitterionic materials (such as PC, SB, CB, *etc.*) and nanoparticles based on these materials can stabilize gene vectors in complex-environment media (such as serum, blood and body fluids) to achieve ‘zero’ pollution.^{67,68} In addition, the superhydrophilic properties of zwitterions form a strong polar contrast with the hydrophobic core of POSS, and possess excellent stability, significantly enhancing the drug-loading efficiency. At the same time, nanoparticles coated with zwitterionic materials can effectively avoid rapid blood clearance during repeated injections by inhibiting the production of polymer-specific antibodies. Based on this, Li *et al.*⁶⁹ synthesized star-shaped zwitterionic cationic copolymers *via* the ATRP copolymerization of zwitterionic poly[*N*-(3-(methacryloylamino)propyl)-*N,N*-dimethyl-*N*-(3-sulfopropyl) ammonium hydroxide] (PMPDSAHA) and cationic poly[2-(dimethylamino)ethyl methacrylate] (PDMAEMA) (Fig. 3a). These star-shaped copolymers can self-assemble to form core-shell corona micelles in aqueous solution, and further encapsulate the tumor-suppressor gene p53 and DOX to evaluate their synergistic tumor suppression effect *in vivo*. The results showed that the POSS-based star-shaped polymer can form stable micelles in aqueous solution, even in the presence of serum. In addition, the DOX and p53-gene-loaded complex has excellent biocompatibility and can effectively deliver DOX and the p53 gene. *In vivo* results further confirmed that the complex not only has an enhanced antitumor effect but can also minimize the toxic side effects. Therefore, this work provides a potential clinical application for cancer treatment. Polymethacrylates are another cationic disinfectant that have

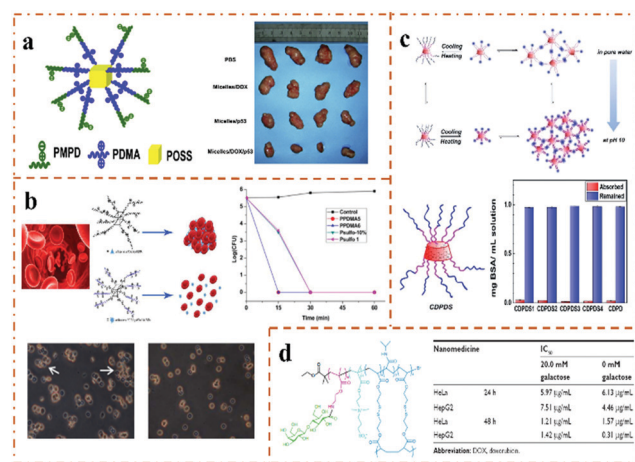


Fig. 3 Synthesis of zwitterionic star-shaped polymers. (a) Antitumor effect of POSS-based star-shaped polymer *in vivo*.⁶⁹ Reproduced from ref. 69 with permission from Elsevier. (b) Antibacterial study of star-shaped POSS-*g*-(PDMA-*r*-PSB) copolymers.⁷⁰ Reproduced from ref. 70 with permission from American Chemical Society. (c) Thermo-responsive and BSA adsorption behavior of CDPD and CDPDS star polymer.⁷⁴ Reproduced from ref. 74 with permission from The Royal Society of Chemistry. (d) *In vitro* cytotoxicity of dual cored hepatoma-specific star-shaped nanogel.⁷⁸ Reproduced from ref. 78 with permission from Dovepress.

been widely studied for their good antibacterial activity. However, they usually cause toxicity to serum, and various strategies to improve their non-contaminating properties or biocompatibility often result in reduced bactericidal activity. So far, the antibacterial work of PDMA has been reported, but star-shaped polymers are rarely used as antibacterial compounds. Based on the designability of the above three-dimensional (POSS) structure, Pu *et al.*⁷⁰ synthesized star polymers (POSS-*g*-PDMA) *via* ATRP with non-toxic POSS as the core and poly[2-(dimethylamino)ethyl methacrylate] (PDMA) as the arm (Fig. 3b). SB was further copolymerized with POSS-*g*-PDMA to obtain random (POSS-*g*-(PDMA-*r*-PSB)) and block (POSS-*g*-(PDMA-*b*-PSB)) arm structures. Furthermore, they compared the hemolytic and antibacterial activity of block and random zwitterionic cationic polymers. The results confirmed that both random and block zwitterionic polymers could effectively kill *Escherichia coli* within 30 min. More importantly, compared with random zwitterionic polymers, block zwitterionic polymers not only significantly enhanced the serum stability but also reduced the aggregation of red blood cells in the blood.

PSBs are the most popular UCST polymers in aqueous solution and have been widely applied in smart materials such as thermoresponsive gels, multilayer microspheres and nanoparticles.^{71–73} In addition, PSB is also used to construct antifouling surfaces under broad pH ranges. Based on the advantages of zwitterionic PSB, Zhang *et al.*⁷⁴ synthesized a star polymer (CDPDS) with polyzwitterionic blocks and hyperbranched polycationic (PDMAEMA) blocks (Fig. 3c). The star-shaped polymers exhibit adjustable UCST behavior and good biocompatibility. The research results showed that the PSBMA blocks have a good shielding effect on the cationic PDMAEMA blocks, so that the CDPDS star polymer only showed UCST behavior. In addition, the higher density of PDMAEMA significantly enhanced the solubility of the star polymers. Moreover, star polymers have enhanced thermal responsiveness, low toxicity and could effectively reduce the adsorption of BSA. Therefore, CDPDS star polymers can be well applied in the field of biomedicine.

In addition to the star-shaped zwitterionic polymers studied above, the recently reported star-shaped polymer nanogels, also known as the intelligent drug-delivery systems, have shown themselves to be potential tools for the efficient diagnosis and treatment of cancer or neurodegenerative disorders.⁷⁵ In fact, it has been reported that star-shaped polymer nanogels prepared *via* controlled/living free radical polymerization are in principle suitable for delivering multifunctional molecular cargoes, controlling cellular uptake, and triggering intracellular drug release. In addition, the star polymer nanogels also exhibit stable physicochemical properties, which is good for their prolonged systemic retention.⁷⁶ Moreover, the hydrophobic interaction between the drug and the core of the star polymer nanogel can achieve a high drug-carrying capacity, and the “arm” can be modified using chemically and biologically active ligands to achieve active and targeted delivery. Glycosylation is a promising method of active targeted drug delivery.⁷⁷ The combination of glycosylation targeting ability and the tumor microenvironment has potential prospects for targeted drug

delivery. Lou *et al.*⁷⁸ prepared a star-shaped thermoresponsive and reductase-cleavable polymer nanogel *via* an “arm-first” ATRP method for targeted liver cancer drug delivery (Fig. 3d). The thermo- and redox-sensitive core, composed of poly(*N*-isopropylacrylamide) (PNIPAM) cross-linked by disulfide bonds (–S–S–), has amphiphilic and anti-adsorption properties with PSBMA. The arm conjugate (PSBMA) formed by the ionic copolymer block is used to reduce the interaction of the polymer nanogel with plasma proteins and cells. Poly(2-lactose diaminoethyl methacrylamide) (PLAMA) is used as a targeted fragment for recognizing ASGP-Rs in hepatocytes. Compared with HeLa cells, star-shaped nanogels have lower IC₅₀ values in human hepatoma cells (HepG2).

2.3 Dendritic zwitterionic polymers

Dendrimers are polymers that have hyperbranched three-dimensional structures, are nano-sized and have high geometric symmetry.^{79–82} Owing to their internal cavities, surface functionalities, small size (<20 nm), and easy removability, they have been considered as ideal nanocarriers for drug or gene delivery. Therefore, massive studies have proposed the use of dendrimers as nanomaterial carriers. Due to steric hindrance, the PEG crown of PEGylated dendritic nanocarrier hinders its interaction with the cell membrane, resulting in a decrease in the endocytosis efficiency.⁸³ The zwitterionic PSB can not only resist protein absorption through electrostatic repulsion but can also avoid steric hindrance accumulation through the formed hydration shell. Therefore, PSB has been widely used in the preparation of dendritic macromolecules instead of PEG and its derivatives.⁸⁴ Based on the advantages of zwitterionic PSB in drug-delivery systems for cancer or inflammation treatment, Wu *et al.*⁸⁵ reported a novel dendrimer containing pH-degradable acetal bonds and dendritic polyacetal, where the end of the dendritic polyacetal is connected with an SB-type zwitterion. The self-assembly behavior, pH responsiveness and drug-release behavior of the modified zwitterionic dendritic polyacetal in water were studied, and it was confirmed that zwitterion-modified dendritic polyacetal has the function of charge reversal and an excellent drug-release performance. Furthermore, by combining the advantages of the polyacetal and the dendrimer, Wu *et al.*⁸⁶ designed and synthesized a dendritic zwitterionic polymer with CD as the host molecules and adamantane-terminated poly(sulfobetaine) (Ada-PSB) as the guest molecules (Fig. 4a). Then, the amphiphilic supramolecular LDBC were successfully synthesized *via* the host-guest recognition of β -CD and Ada. Micelles and vesicles were obtained by adjusting the hydrophilic/hydrophobic ratio of the LDBC. *In vitro* drug release and cytotoxicity results showed that drug-loaded dendritic nanoparticles could achieve over 90% drug release under intracellular acidic conditions, and their cytotoxicity is comparable to free DOX. Therefore, dendritic zwitterionic polymers could be used as potential anticancer drug-delivery carriers.

Dendritic nanocarriers with a definite structure are promising for tumor targeting and anticancer therapy due to their enhanced drug-carrying capacity and stability.⁸⁷ Zwitterionic materials, such as phosphorylcholine (PC), not only reduce

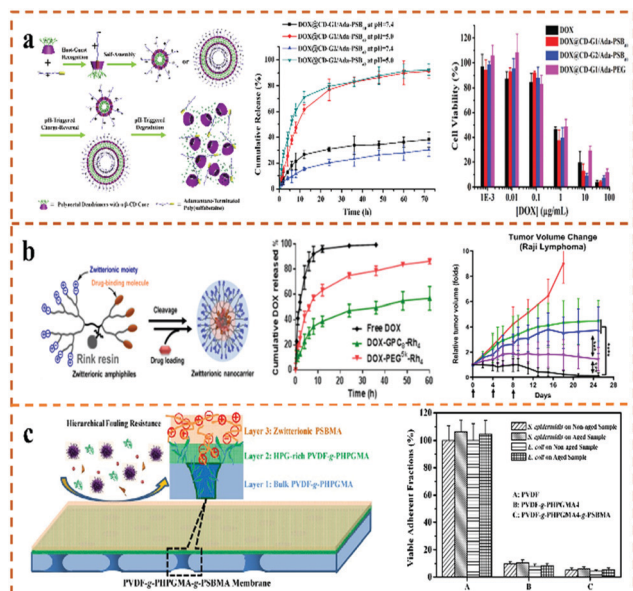


Fig. 4 Synthesis of dendritic zwitterionic polymers. (a) The pH-triggered charge reversal, degradation behavior and *in vitro* drug-release profile of dendritic supramolecular micelles and vesicles.⁸⁶ Reproduced from ref. 86 with permission from the Royal Society of Chemistry. (b) *In vivo* tumor treatment of amphiphilic dendritic zwitterionic polymer.⁹⁰ Reproduced from ref. 90 with permission from American Chemical Society. (c) Viable adherent fractions of dendritic zwitterionic (PVDF-*g*-PHPGMA-*g*-PSBMA) hierarchical antifouling membranes.⁹¹ Reproduced from ref. 91 with permission from American Chemical Society.

protein adsorption, but also show more effective antifouling properties by providing stealth properties to the nanoparticles.^{88,89} Luo *et al.*⁹⁰ developed a novel Janus dendrimer (JD) amphiphilic zwitterionic polymer based on their previously designed terminal dendrimer nanocarriers to enhance the delivery of anticancer drugs (Fig. 4b). In order to avoid the tedious synthesis and purification of liquid-phase dendrimers, they used solid-phase peptide chemistry (SPPC) to construct JD amphiphilic polymers with precise chemical structures. Janus dendrimers self-assemble to form zwitterionic JD nanoparticles, and then DOX is loaded through hydrophobic interactions. Compared with linear PEGylated nanoformulations, drug-loaded zwitterionic JD nanoparticles show higher stability, sustained drug release *in vitro*, prolonged blood circulation and better anti-tumor properties.

The practical applications of fluoropolymers are greatly restricted due to their inherent inertness. To develop environmentally friendly antifouling coatings that do not release pollutants, Li *et al.*⁹¹ used polyvinylidene fluoride (PVDF) as a model fluoropolymer to construct a layered antifouling surface polymer (Fig. 4c). Specifically, they synthesized the PVDF-*g*-PHPGMA copolymer of pentafluorophenyl methacrylate (PFMA) with the ozone-pretreated PVDF *via* RAFT polymerization. PVDF-*g*-PHPGMA membranes and zwitterionic SBMA were further functionalized *via* SI-ATRP to obtain the PVDF-*g*-PHPGMA-*g*-PSBMA membranes. The overall antibacterial adhesion and antifouling performance of the PVDF-*g*-PHPGMA-*g*-PSBMA membrane was obtained through the good synergistic

effect between the dendritic architecture of PHPGMA and the superhydrophilic PSBMA brush.

2.4 Brush zwitterionic polymers

Brush polymers are a kind of polymer with a nanometer-sized and hyperbranched structure, and have a simpler preparation procedure than dendritic polymers. Over the past few decades, polymer brushes have become lubricants with excellent stability, and antifouling and lubrication abilities due to their dense segment grafted structure and resistance to interpenetration under compression, which provide enhanced wear protection.⁹² With the increased demand for functional surface coatings, zwitterionic polymer brushes with excellent lubricity and low-pollution coatings have attracted considerable attention. Recent studies have reported that the low interpenetration of bottle brush (BB) polymers, zwitterionic brush (PZWs) and surface grafted polymer brush can provide good surface lubrication.^{93–95} Among them, the most lubricious brush formulations belong to PZWs. PZW brushes have been widely used on various surfaces due to their high lubricating properties and inherent inertness to biological contaminants.^{96–98} Yan *et al.*⁹⁹ prepared brush-like PZW brushes with different structures, such as linear PMPC brushes, graft copolymer brushes and bottlebrushes. Exposing the PZW molecular brush to the interface showed good lubricity and resistance to protein and cell contamination. PMPC is an attractive alternative to PE brushes.¹⁰⁰ Based on this, Faivre *et al.*¹⁰¹ designed lubricating and wear-protection fluids *via* the synergistic mixture of the natural linear polyelectrolyte hyaluronic acid (HA) and a water-soluble BB polymer (Fig. 5a). The results show that, compared with salt solution, the strong cohesion of the topological polymer BB and HA mixed solution can significantly improve the wear protection of polymer surfaces under wide range of shear conditions. PMPC zwitterionic brushes, whether grafted on the surface or physically adsorbed on the surface in the form of bottle-brush polymers, have excellent lubrication and wear-resistance properties. The zwitterionic bottlebrush polymer film shows unprecedented antifouling properties to charged proteins by adsorbing charged substances. Inspired by the excellent lubrication and wear-protection properties of zwitterionic bottlebrush polymers, Adibnia *et al.*¹⁰² reported that zwitterionic bottle-brush polymers can be physically adsorbed in the presence of multivalent cations, and further found that the zwitterionic bottle-brush polymer still had great lubricating and anti-wear properties under the concentration of multivalent ions owing to the unique hydration properties of the zwitterionic part (Fig. 5b). In addition, these polymers are generally biocompatible, making them potential candidates for clinical applications such as artificial joint interfaces. However, these brush polymers will cleave from the surface when they are immersed in aqueous or acidic and alkaline solutions for several days, which severely limits their applications. To improve the long-term stability of the brush-like zwitterionic polymer in water and salt solutions, Yu *et al.*¹⁰³ proposed a handy way for manufacturing robust and homogeneous PMPC brushes. Specifically, they coupled a 2-bromoisobutyl bromide

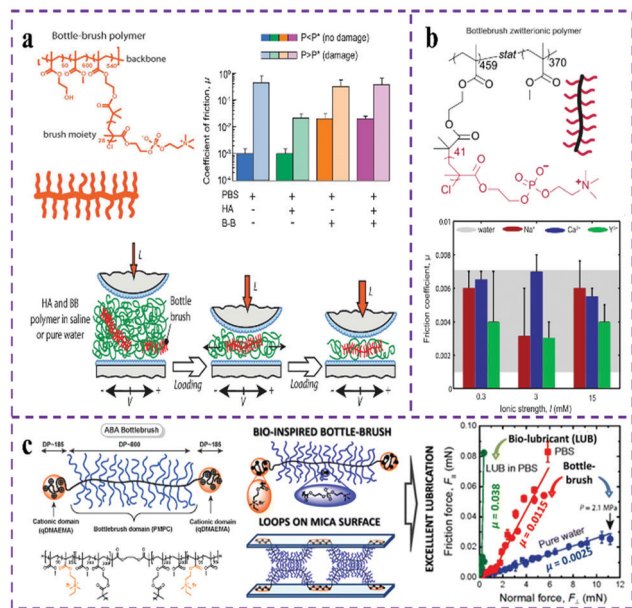


Fig. 5 Schematic representation of the zwitterionic bottlebrush (BB) polymer. (a) Friction coefficient measurement and wear protection mechanism of HA and BB polymer mixtures.¹⁰¹ Reproduced from ref. 101 with permission from American Chemical Society. (b) Super-lubricating behavior of BB polymer in the presence of multivalent ions.¹⁰² Reproduced from ref. 102 with permission from American Chemical Society. (c) Low friction behavior of bioinspired BB polymer PDMEMA-g-PMPC.¹⁰⁸ Reproduced from ref. 108 with permission from American Chemical Society.

(BIBB) initiator to a polyglycidyl methacrylate (PGMA)-modified substrate and used SI-ATRP to polymerize PMPC brushes. The prepared PGMA-PMPC brush shows excellent stability and lubricity in water and salt solutions, indicating their potential applications in low-fouling membrane coatings and biomedical engineering.

Manufacturing hierarchical polymer brushes with antifouling and bactericidal functions can effectively improve the antibacterial effect of biological materials, which is a promising way to resist infections associated with biological materials.^{104,105} Since the antifouling and bactericidal graft segments of the hierarchical polymer brush are constructed in different spatial layers, this not only greatly reduces the mutual interference of the same-level grafted segments, but also promotes the synergistic effects between the layered grafted segments. Although Fu *et al.*¹⁰⁶ have reported a dual-function antibacterial surface that prevents bacterial adhesion and kills bacteria, they have not studied in depth how the surface functional hierarchical structures affect the anti-adhesion and antibacterial properties of the surface. In order to construct a more effective antibacterial surface, this team successfully constructed three hierarchical polymer brushes (Si-PQAs-*b*-PSBMA, Si-PSBMA-*b*-PQAs and Si-PQAs-*r*-PSBMA) *via* the SI-ATRP polymerization of different alkyl chain lengths and zwitterionic antifouling PSBMA polymer blocks and cationic bactericidal poly(quaternary ammonium salt) (PQA) polymers. This work confirmed that the construction of hierarchical structure polymer brushes with different PQA alkyl chain lengths can effectively improve the antifouling and bactericidal properties.¹⁰⁷

Studies have shown that, due to the highly complex architecture of proteins, the natural glycoprotein lubricin (LUB) is key to the lubrication, antiadhesion and wear protection of articular cartilage surfaces. Inspired from the brush-like structure of LUB, Israelachvili *et al.*¹⁰⁸ prepared two bottle-brush polymer brushes (PDMEMA-g-PMPC and PMMA-g-PMPC) *via* the combination of ATRP and different post-modification techniques (Fig. 5c). The polymer block is composed of a hydrophobic (methyl methacrylate, MMA) monomer and a positively charged amine monomer (qDMAEMA). The two components are strongly adsorbed on the surface of mica through electrostatic and hydrophobic interactions. The polymer bottle-brush block consists of a flexible MMA main chain decorated with PMPC, and possesses the excellent biocompatibility and lubricity of the zwitterionic branch. Studies show that the ABA bottle-brush polymer has excellent lubrication properties.

2.5 Cyclic zwitterionic polymers

In recent decades, owing to the absence of polymer chain termini, cyclic polymers have been shown to have unique physical and chemical properties, such as a lower intrinsic viscosity, a smaller hydrodynamic volume and mean square radius of gyration, a higher glass transition temperature as well as a higher thermal stability compared with linear analogues; thus, they have attracted widespread attention from scientists. In addition, the unique steric hindrance structure of cyclic polymers can effectively resist entanglement between chains, such that the cyclic polymers show advantages in nanomedicine delivery that include but are not limited to enhanced colloidal stability and drug loading, slower degradation behavior, prolonged circulation time, lower *in vitro* cytotoxicity, and improved cell uptake efficiency and therapeutic effects.^{109–111} However, the current preparation methods for cyclic polymers are still very limited, and the synthesis and purification processes are very complicated. Therefore, it is necessary to develop a straightforward and efficient strategy for the preparation of cyclic polymers. Although there have been related reports on synthesizing cyclic polymers, these have been limited to the preparation of low-to-medium molecular weight (<10.0 kDa) cyclic polymers, and the cyclization of high molecular weight linear precursors usually produces topological by-products, such as knots or chains that arise from chain coupling. In order to alleviate the entropy challenge of coupling high molecular weight linear polymer end groups, innovative methods such as the cyclization of polymer carriers and zwitterionic ring-opening polymerization (ZROP) have emerged. N-Heterocyclic carbenes (NHC) are an effective nucleophilic reagent and a relatively poor leaving group,¹¹² and these two characteristics are vital for the preparation of high molecular weight polymers (>1200 kDa).¹¹³ Tezuka first proposed a new strategy for the synthesis of high molecular weight cyclic polymers *via* ZROP (Fig. 6a).¹¹⁴ Zhang *et al.*¹¹⁵ synthesized cyclic poly(*N*-substituted glycine) *via* NHC-mediated *N*-substituted *N*-carboxyanhydride ZROP (Fig. 6b). The excellent biocompatibility and enhanced proteolytic stability of these polymers could be used in various biotechnologies such as drug delivery and

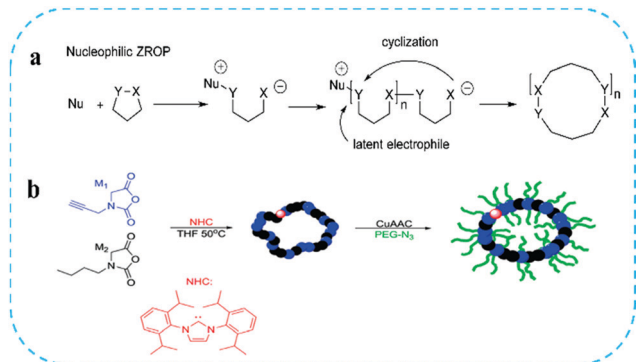


Fig. 6 (a) Zwitterionic ring-opening polymerization (ZROP) to synthesize cyclic polymers and (b) cyclic brush polymers.^{114,115} Reproduced from ref. 114 with permission from American Chemical Society.

bioactive coatings. At the same time, they synthesized a cyclic polymer backbone composed of poly(*N*-propargyl glycine) (*c*-PNPG) through NHC-mediated *N*-propargyl *N*-carboxyanhydride ZROP. However, up to now, scientists have only proposed ZROP, and the cyclic or cyclic brush polymer materials of zwitterionic have not been reported. Based on the superlubricity of zwitterionic polymers and the enhanced steric stability of the cyclic topology, facilitating zwitterionic polymers with cyclic topology can be a promising research direction. The higher dimensional stability of the cyclic brush zwitterionic polymer material endows its super-lubricating behavior, which can provide potential application advantages for antifouling materials and joint-lubrication treatment.

2.6 Other polymers

In addition to the topological zwitterionic polymers mentioned above, there are other types of polymer, such as spherical brushes, granular polymers, nano-microspheres, block-combs and so on.^{116–119} For example, Jiang and co-workers have done a lot of work on initiating the polymerization of zwitterions on a variety of planes and nanoparticle surfaces (Fig. 7a).¹²⁰ They modified the ATRP initiator with siloxane groups on silicon nanoparticles, and then added different concentrations of the carboxylic betaine monomer (CBAA) to initiate polymerization. Then, two kinds of silicon nanoparticle with different particle sizes were obtained, which could be stable in protein solution for more than 72 hours. Similarly, they prepared ultra-stable PCBMA-modified gold nanoparticles. Under mild conditions, Xu and co-workers introduced the PCB copolymer on the surface of silica nanoparticles for the first time and prepared a spherical mixed charged silica brush (MCB), which could regulate the adsorption and release of protein without inactivation.¹¹⁶ Because the zwitterionic CBMA component has adjustable adsorption for acidic and basic proteins, the spherical charged brushes have great potential for drug loading, controlled release and immobilized enzymes. Zhou *et al.*¹²¹ prepared temperature-sensitive spherical polymer microgels (SB-*g*-NBrMGs) by grafting PSPMK onto poly(*N*-isopropylacrylamide) (P-NIPAM) microgels (Fig. 7b). The microgel showed an ultra-low coefficient of friction and temperature-triggered

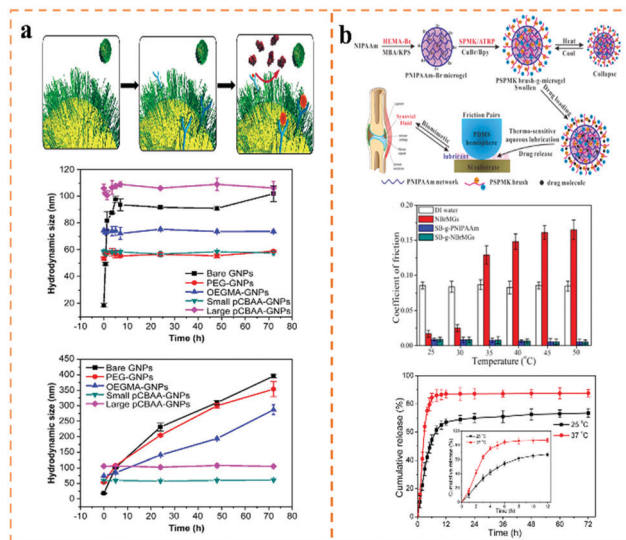


Fig. 7 (a) Non-specific protein adsorption of PCBAAs-GNPs.¹²⁰ Reproduced from ref. 120 with permission from Elsevier. (b) Temperature-sensitive hairy spherical microgel and release profiles of aspirin-SB-*g*-NBrMGs in PBS.¹²¹ Reproduced from ref. 121 with permission from American Chemical Society.

drug-release behavior. This microgel with both lubrication and controlled-drug release capabilities can be used as a joint synovial fluid for the treatment of arthritis.

In this section, we mainly introduced the various topological structures based on zwitterionic polymers, which can also generate materials with different morphologies, such as micelles, nanoparticles, films, and hydrogels. Among them, zwitterionic hydrogels are widely used in biomedicine, gel electrolytes, electronic sensing and other fields because of their high ionic group density, and strong hydrophilicity, affinity, and unique anti-polyelectrolyte effects. For example, Wei *et al.* reported two zwitterionic dextran hydrogels (CB-Dex and SB-Dex).¹²² Their research work confirmed that zwitterionic modified dextran can minimize bacterial adhesion, prevent wound infection, and accelerate wound healing. In addition, Lv and Wei *et al.*¹²³ recently developed an anti-swelling conductive hydrogel (DN-FT-HCl) with high toughness, high compressive modulus, and high ionic conductivity, and developed an underwater strain sensor based on this hydrogel. The hydrogel sensor can accurately and reliably identify underwater motion signals. Moreover, based on the potential biomedical advantages of hydrogels, Saha *et al.*¹²⁴ reviewed the synthesis and applications of zwitterionic microgels or nanogels developed in recent years. In the next section, we will focus on various biological applications of zwitterionic-based topological polymers.

3. Potential biomedical applications of zwitterionic-based topological polymers

Topological zwitterionic polymers have been widely used in the biomedical field due to their controllable structure, high

widely used by researchers. Based on the low disulfide ($-S-S-$) bond energy, Dong and Pang *et al.*¹³³ developed a reduction-responsive biodegradable PSBMA zwitterionic polymer nanogel with a long circulation time for the effective delivery of DOX at tumor sites. Based on the ultra-low contamination and functionalization properties of amino acid-derived zwitterionic polymers (PAAZ), the zwitterionic polymer nanogel can prolong the blood-circulation time *in vivo*. Compared with the non-degradable PSBMA nanogel, the zwitterionic nanogel showed faster DOX release under a reductive environment, indicating that the degradation of the nanogel greatly accelerates the drug release. In addition, Peng and co-workers first studied hypoxia-responsive degradable poly(phosphobetaine)-based zwitterionic nanogels.¹³⁴ The nanogels showed controlled DOX-release characteristics and a negligible accelerated blood clearance (ABC) phenomenon. Moreover, Lou *et al.*¹³⁵ synthesized thermo- and redox-sensitive star-shaped nanogels with disulfide cross-linked PNIPAM with zwitterion PSBMA as the core and PLAMA as the arm (Fig. 8b). *In vitro* release studies showed that the star-shaped drug-loaded nanogels had a significantly increased DOX release under high GSH and high temperature conditions, as well as increased cytotoxicity to human hepatoma cells (HepG2). Therefore, the dual-response star-shaped nanogels modified by lactose have a hepatoma-targeting ability and can be used as a specific anti-cancer drug-delivery carrier for cancer treatment. Topological zwitterionic polymers are an ideal drug-delivery material, which can effectively enhance the solubility of drugs, improve the encapsulation efficiency, prolong the drug circulation time *in vivo*, and reduce the toxic side effects. Furthermore, rapid release at the lesion site and ideal therapeutic effects are also pursued in practical applications.

3.2 Antitumor effects

A tumor is a medical problem that can seriously endanger human life. Tumors show great heterogeneity, genetic instability, and can have a complex pathogenesis. Tumor-treatment methods include surgery, chemotherapy, radiotherapy, targeted drug therapy, gene therapy, immunotherapy, and so on.¹³⁶ In addition, differentiation-inducing agents are widely used in the treatment of various precancerous lesions and can effectively prevent the occurrence of tumors. However, this differentiation-inducing therapy is only suitable for patients with early cancerous transformation, and the preparation of the differentiation-inducing agent is difficult; the stability and safety of this therapy therefore need to be confirmed further. Moreover, it is difficult to resist the complex and variable tumor environment using a single therapy, and the development of combined anti-tumor therapies is inevitable. At present, although tumor surgery has prolonged the lives of some patients to a certain extent and has improved the quality of life with a limited life span, surgery still has great limitations. Based on the shortcomings of tumor treatment mentioned above, the developed zwitterionic materials have broad application prospects in anti-tumor therapy. Based on this, Gu *et al.*¹³⁷ synthesized zwitterionic triblock multi-acid-sensitive antitumor drug-loaded “stealth” micelles [pCBMA-*b*-p(ADA-TMDP)], which consist of pCBMA as the hydrophilic shell and p(ADA-TMDP)

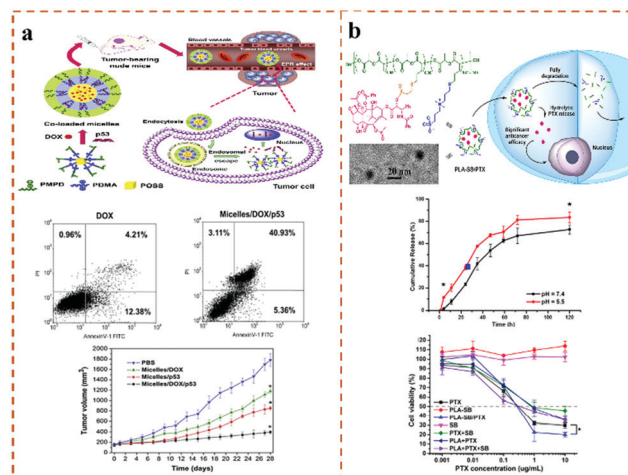


Fig. 9 (a) Study on drug delivery and tumor suppression of star-shaped amphiphilic copolymer micelles.⁶⁹ (b) The drug release and cytotoxicity of the biodegradable zwitterionic conjugate PLA-SB/PTX.¹⁴¹ Reproduced from ref. 69 and 141 with permission from Elsevier.

components as the hydrophobic core. The drug-loaded micelles could significantly accelerate the drug release (3.26-fold) under intracellular acidic conditions (pH 5.0) compared with the neutral environment (pH 7.4), which can effectively improve the anti-tumor effects of chemotherapy. Liu *et al.*⁶⁹ constructed star zwitterionic polymers for the co-delivery of DOX and p53 tumor-suppressor genes (Fig. 9a). The star-shaped gene carrier had a strong DNA binding ability and an increased drug-encapsulation ability. Zwitterionic PMPD can effectively improve the serum stability and biocompatibility of the star complex. In addition, the integrated delivery vector was effectively delivered to MCF-7 cells, which significantly increased tumor cell apoptosis and an improved anti-tumor efficacy *in vivo*. Based on the multiple anti-tumor effects of stimulus-responsive polymeric drug carriers, Wu and co-workers developed a pH/redox dual-responsive zwitterionic polymer.¹³⁸ The functional self-assembled spherical micelles can effectively encapsulate DOX, reduce the toxic and side effects, and enhance the antitumor efficacy *in vivo*. Moreover, Luo *et al.*¹³⁹ synthesized a four-arm star zwitterionic polymer (4sPCLDEAS) loaded with DOX, which can induce the apoptosis of tumor spheroids and effectively inhibit their growth, where the toxicity is even equivalent to that of free DOX. This work provided a reasonable design idea for the manufacture of star-shaped drug-loaded micelle systems with high anti-tumor activities. Moreover, they further developed a pH-sensitive four-arm star zwitterionic polymer (4s-PCL-PDEASB) with lower cytotoxicity compared with linear polymers.¹⁴⁰ The star copolymer loaded with curcumin can prolong the retention time of curcumin circulating in the blood, to effectively reduce the damage of curcumin to the liver and lungs, and showing excellent antitumor effects.

For tumor treatment, whether drug-loaded micelles can successfully enter tumor cells is the key. Tu *et al.*¹⁴¹ developed a hydrophilic star-shaped six-arm poly- ϵ -propiolactone-*b*-poly-(2-methacryloxyethyl phosphorylcholine) (6sPCL-*b*-PMPC) block polymer, loaded with the anticancer drug paclitaxel (PTX) after

self-assembly (Fig. 9b). Compared with ordinary PCL-*b*-PEG micelles, tumor cells can adsorb the star 6sPCL-*b*-PMPC micelles more effectively, and this carrier has a higher toxicity to human cervical cancer cells. Cheng *et al.*¹⁴² synthesized a biodegradable zwitterionic polymer with the corresponding paclitaxel (PTX) coupled with PLA-SB/PTX with biodegradable polylactide (PLA) as the main chain. This polymer drug conjugate can release PTX through biodegradation, thereby killing tumor cells and achieving the purpose of an anticancer drug.

The adsorption of serum proteins on the surface of nanoparticles will seriously affect the stability of the complex physiological and pathological microenvironment *in vivo* after intravenous injection. Therefore, drug-delivery systems require superhydrophilic surfaces to effectively reduce the formation of serum proteins, prolong long-term circulation *in vivo*, and improve the treatment of diseases *via* nano-formulations. Zwitterionic polymer materials have been proved to have excellent “stealth” functionality. For example, Luo and co-workers developed a Janus dendritic multifunctional zwitterionic polymer drug-delivery carrier through solid-phase peptide chemistry (SPPC).⁹⁰ The high hydrophilicity of zwitterionic GPC provides DOX nano-formulations with excellent stability both *in vitro* and *in vivo*, significantly increases tumor accumulation and improves the anticancer effects. Jiang and Wu *et al.*¹⁴³ prepared two water-soluble worm-like brush polymers with different structures using poly-3-phenylthiophene as the main chain and PEG and PCB as the side chains. The worm-like brush polymer not only effectively improved the water solubility of the conjugated polymer poly-3-phenylthiophene and prevent fluorescence aggregation-induced quenching in the aqueous phase, but also its luminescent properties endow the brush polymer with inherent biological functionality. In addition, the high-density of PEG and PCB hydrophilic side chains on the brush-shaped polymers can effectively prevent the non-specific adsorption of proteins, extend the blood circulation time, and improve the passive targeting of tumors. Aiming at the problem that it is difficult for drugs to penetrate in solid tumors, Mao *et al.*¹⁴⁴ proposed a zero-waste multifunctional cooperative active drug-delivery system, which is a nanomotor PSBMA-*l*-Cys/*a*-CHCA(PCA) that is driven by zwitterion H₂S (Fig. 10). Zwitterionic nanomaterials endow the nanomotor with a long cycle time through anti-protein adhesion properties. Over-expression of H₂S can cause tumor-cell acidosis. At the same time, *a*-CHCA can further aggravate acidosis. Therefore, the synergistic effect of H₂S and *a*-CHCA-loaded drugs induces multiple acidosis and apoptosis of tumor cells and can achieve the good inhibition of tumor growth by the zwitterionic H₂S-driven nanomotor. Early and correct diagnosis and the targeted treatment of malignant tumors are the key to successful tumor treatment. Zwitterionic polymers play a vital role in the early diagnosis and treatment of malignant tumors because of their unique structures and properties. Zwitterionic polymers are functionally bonded with targeting ligands to improve the specific recognition of tumor cells, enable the early diagnosis and treatment of tumors, inhibit tumor growth, control the rapid deterioration of diseases, and greatly reduce mortality.

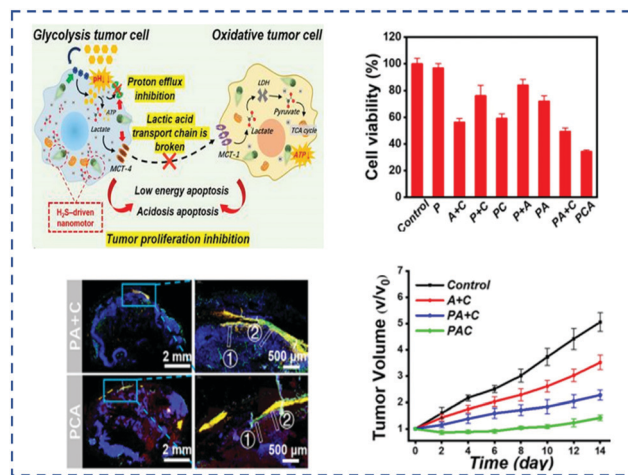


Fig. 10 Synthesis process of PCA nanomotor and the treatment strategy of tumor metabolism symbiosis.¹¹⁴ Reproduced from ref. 114 with permission from John Wiley and Sons.

Zwitterionic polymer materials have made great contributions to human health in the treatment of tumors.

3.3 Biomedical diagnostics

Biomedical diagnosis and treatment are the foundation of ensuring human health and improving people's quality of life. However, the existing biomedical diagnostic and therapeutic preparations have high costs, complex preparation processes, and lack any guarantee of biological safety, which greatly restrict the development of biomedical diagnostics and therapeutics. Zwitterionic polymer materials are low cost, require mild reaction conditions and have easy surface modification. In particular, topological polymers have stable chemical structures, which is expected to solve the current problems of unguaranteed biosafety and the single limitation of existing diagnosis and treatment. At the same time, they have huge scientific research and application value in the field of biomedical diagnosis and treatment. Zhang *et al.*¹⁴⁵ proved for the first time that pCBMA has both anti-biological pollution properties and easy modification based on carboxylic acid groups to fix other functional groups or biological ligands, so that pCBMA-based materials have good application prospects in medical diagnosis. Iwasaki *et al.*¹⁴⁶ prepared a copolymer of thiol-terminated methacrylic acid and MPC, then modified it on a gold-coated surface plasmon resonance (SPR) chip (Fig. 11a). Biotin was introduced into the carboxyl group of the copolymer, which not only provided a good biocompatible surface but also meant that antibiotic proteins could be detected at concentrations as low as 100 mg mL⁻¹ in plasma. Mitsubayashi *et al.*¹⁴⁷ copolymerized MPC and 2-ethylhexyl methacrylate (EHMA), and then immobilized glucose oxidase (GOD) on a porous polytetrafluoroethylene (PTFE) membrane to form an enzyme membrane. They further fixed the enzyme membrane on a hydrogen peroxide electrode to prepare a flexible biosensor to achieve the characteristics of the glucose sensor. This study showed that the glucose sensor has good

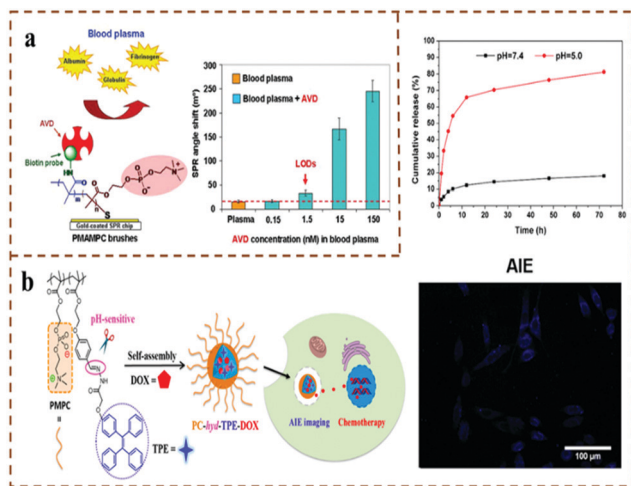


Fig. 11 (a) PMAMPC brush probe detects complex matrices (such as blood plasma and clinical samples).¹⁴⁶ Reproduced from ref. 146 with permission from American Chemical Society. (b) PMPC-hyd-TPE conjugate for AIE imaging and pH-responsive drug delivery.¹⁵¹ Reproduced from ref. 151 with permission from American Chemical Society.

glucose responsiveness, where the current output range (of 0.05–1.00 mmol L⁻¹) and the calibration range (0.14–0.23 mmol L⁻¹) are equivalent to the glucose level in normal tears. Therefore, this work is expected to produce a non-invasive contact lens biological detector for the continuous monitoring of glucose content in tear fluid.

Cancer is one of the diseases that endangers human life and health. Advanced cancer not only consumes a lot of financial resources, but also the adaptability of patients is poor. Therefore, the early diagnosis of cancer is extremely important.¹⁴⁸ Studies have shown that the level of the activated leukocyte adhesion molecule (ALCAM) in cancer patients is higher than the normal level of ALCAM in serum.¹⁴⁹ The movement of ALCAM onto a surface makes the material-specific detection of ALCAM possible. Jiang *et al.*¹⁵⁰ grafted PCBAA onto silica nanoparticles to prepare PCBAA-SiP *via* ATRP. The results showed that the PCBAA-modified nanoparticles could be stably stored in negative/positive protein solution for 72 hours, and the particle size was almost unchanged. More importantly, PCBAA-modified silica nanoparticles can easily be functionalized with ALCAM to diagnose cancer cells in complex media. The synergistic effect of AIE fluorescence imaging and drug delivery is of great significance for the diagnosis and treatment of cancer. Based on this, Jin *et al.*¹⁵¹ synthesized a new pH-sensitive zwitterionic polymer PMPC-hyd-TPE with the AIE-active molecule TPE (Fig. 11b). Due to the “stealth” performance of PC, zwitterionic self-assembled micelles showed excellent stability and low protein adsorption. *In vitro* results showed that TPE is cleaved under acidic conditions, which accelerated drug release *in vitro*, and the PMPC-hyd-TPE drug-loaded micelles were effectively internalized and had an AIE imaging ability in HepG2 cells; moreover, *in vitro* imaging confirmed that the PC-hyd-TPE-DOX micelles could effectively accumulate in the tumor site and specifically release DOX.

Therefore, the multifunctional micelles developed in this study with excellent antifouling, pH-sensitive drug delivery and *in vitro* AIE imaging properties have potential significance for the diagnosis and treatment of cancer in the future.

Zwitterionic polymer-modified magnetic nanoparticles have become potential materials for magnetic resonance imaging (MRI) diagnosis. Because zwitterionic pCBMA is rich in carboxyl groups, they can easily chemically immobilize antibodies and other biomolecules on the targeting ligand. Based on this, Zhang *et al.*^{152,153} developed the zwitterionic polymer pCBMA-coated magnetic nanoparticles pCBMA-MNPs. The pCBMA-MNPs can be stored stably in phosphate buffered saline (PBS) and undiluted serum for at least six months and can prolong the half-life of blood circulation in the body. Because MRI diagnosis can determine the location of the disease, timely and accurate diagnosis and treatment are possible. When the nanogel enters the reducing intracellular environment, Fe₃O₄ nanoparticles are released spontaneously, and the MRI properties are enhanced. These characteristics make zwitterionic polymer gels an ideal carrier for drug delivery and MRI diagnosis. Shen *et al.*¹⁵⁴ prepared a zwitterionic polyglycerol dendrimer (PGD), which was used as an extracellular dendritic MRI contrast agent (CA) for blood pool imaging. Xie *et al.*¹⁵⁵ modified the sensor of a dynamic blood glucose detector (CGMS) with a zwitterionic polymer, which effectively reduced the measurement noise and avoided frequent calibration. In addition, zwitterionic polymers can also be applied to sensors. Wiarachai *et al.*¹⁵⁶ designed the Au-PPgMA₆₅MPC₃₅-PNA zwitterionic polymer material, which can be used as a sensor for DNA detection and can detect the most DNA sequences. Takeuchi *et al.*¹⁵⁷ synthesized poly(2-methacryloxyethyl phosphorylcholine)-grafted gold nanoparticles (AuNPs). This nano-sensing system can effectively detect the concentration of CRP in human serum (as low as 1% w/w, with a detection limit of ~50 ng mL⁻¹). The high cost and low biological safety of existing biomedical diagnostic materials have greatly restricted the development of biomedical diagnostics. The novel zwitterionic biomedical diagnostic materials developed are not only low in price, have high biosafety and require a simple preparation process, but also have the function of magnetic imaging. Through molecular-targeted or antibody-targeted biological modification of zwitterions, the early localization, accuracy and integration of diagnosis and treatment are realized, which can promote the development of existing medical diagnosis and provide a greater guarantee for humans.

3.4 Antifouling

Biocontamination from proteins and bacteria has a great impact on biosensors, contact lenses, medical transplantation equipment and marine ships. Therefore, the inhibition of biofouling and the development of new functional antifouling materials are vital to real life. PEG and its derivatives are the 5th most widely used anti-biological pollution materials at present.¹⁵⁸ However, PEG materials are easily oxidized in the biological environment and have poor stability. Zwitterionic polymers have become an ideal substitute for PEG. The anti-biological

pollution mechanism of zwitterionic polymer materials has been widely studied and recognized. Zwitterionic polymers have large dipole moments and abundant hydrophilic groups (such as carboxyl, sulfonic acid, amine, *etc.*). A large number of water molecules can be gathered around these groups, that is, a tightly bonded hydration layer will be formed on the surface of the material, which can make the material superhydrophilic, and effectively prevent the adsorption of proteins and other biomolecules on the surface of the material.^{159,160} Moreover, zwitterionic polymers have been used to construct various biomedical materials, such as polyurethanes, poly(vinylidene fluoride), metal materials and so on, which endows the biological substrates with excellent antifouling properties. Therefore, zwitterionic polymers have become the representative of a new generation of antifouling materials. Wang *et al.*¹⁶¹ prepared a zwitterionic macromolecular cross-linked polyurethane antifouling coating with long-lasting tolerance and long-term antifouling performance. The hydrophilicity of the PCBMA-based macromolecular cross-linking agent coating can be maintained in flowing PBS for over two weeks, and the PU coating surface treated in PBS solution after one week could still effectively resist the adsorption of proteins and bacteria. Cai *et al.*¹⁶² developed a PCB-based hydrogel coating that can improve the blood compatibility of activated carbon (PAC). Furthermore, an antifouling ability is essential for the development of medical metal implants. Zwitterionic block copolymers can be anchored on different metal surfaces and impart good antifouling properties. This efficient antifouling strategy is used widely in biomedical metal implants and devices. Takai *et al.*¹⁶³ synthesized a series of poly(MPC-*b*-(MPTSSi-*r*-MPTMSi)) block copolymers, which were successfully applied to glass substrates as uniform polymer coatings. Zwitterionic block copolymers show excellent protein tolerance and a wide pH durability (pH = 2–9). Besides, the hydrophilic surface designed by the zwitterionic block copolymer coating does not require pretreatment, which indicates that it is suitable for medical and environmental purification equipment, such as biosensors, sanitary ware, and water purification. Based on the serious pollution problem of fluoropolymer membranes, Cai *et al.*¹⁶⁴ developed a PVDF-*g*-PHPGMA-*g*-PSBMA hierarchical antifouling and antibacterial membrane on a polyvinylidene fluoride (PVDF) membrane through the synergistic effect of the dendritic structure of PHPGMA branches and the “superhydrophilic” properties of PSBMA brushes. The PVDF-*g*-PHPGMA-*g*-PSBMA membranes have potential value in biomedical applications because of their excellent anti-protein and bacterial adhesion ability. This research is a universal and effective method for manufacturing antifouling fluoropolymer membranes in a controlled and green manner. Moreover, Fu and co-workers *et al.*¹⁰⁷ designed and constructed three different structures of hierarchical polymer brushes with antifouling and sterilization functions (Si-PQAs-*b*-PSBMA, Si-PSBMA-*b*-PQAs and Si-PQAs-*r*-PSBMA). The comprehensive antibacterial ability of biomaterials can be effectively improved by the polymer hierarchical brush. Kwon *et al.*¹⁶⁵ successfully grafted (sultaine-*co*-tyramine) (PSBTA) on the surface of polyurethane *via* a tyrosinase-catalyzed reaction (Fig. 12a). The PSBTA-modified coating surface can effectively prevent the adsorption

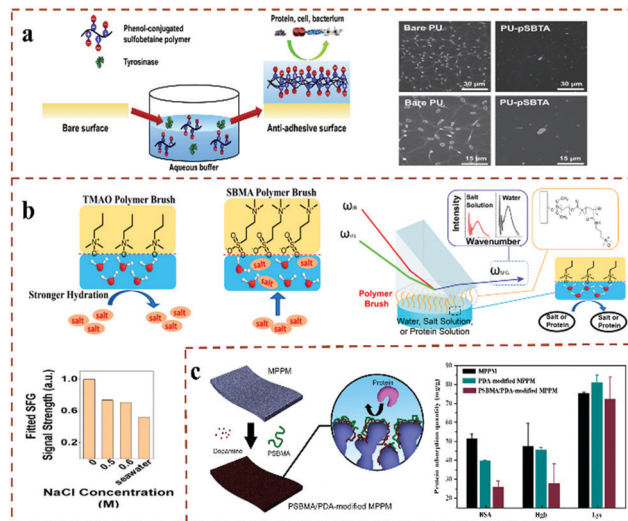


Fig. 12 Zwitterionic antifouling applications. (a) Fluorescence microscope image of *in vitro* platelet adhesion on bare and pSBTA-grafted PU substrate.¹⁶⁵ Reproduced from ref. 165 with permission from American Chemical Society. (b) Study on salt resistance of new type non-polluting zwitterionic polymer.¹⁶⁸ Reproduced from ref. 168 with permission from Elsevier. (c) PSBMA/PDA co-deposition preparation of anti-fouling film surface and protein adsorption quantity.¹⁶⁹ Reproduced from ref. 169 with permission from Elsevier.

of fibrinogen, fibroblasts and *Staphylococcus aureus*. The PSBTA polyurethane coating imparts good biological inertness to the material, which can greatly reduce the probability of implantation failure for artificial medical devices. Mi *et al.*¹⁶⁶ synthesized a thermosensitive multifunctional ABA triblock copolymer wound-dressing hydrogel. The middle block B contains the positively charged and hydrolyzable betaine ester as well as antibacterial drugs. The copolymer solution was applied to the wound and rapidly turned into gel at the body temperature. The release of antibacterial drugs can reduce the risk of wound infection, and the positive charge can promote cell adhesion during tissue regeneration.

With the increasing demand for biomedical antifouling, there have been many reports on the preparation of antifouling coatings based on zwitterions. For example, Zhou *et al.*¹⁶⁷ prepared an imidazole-based zwitterionic polymer antifouling coating *via* electrochemical surface atom transfer radical polymerization (e-SIATRP). Jiang's research group designed and synthesized various types of zwitterionic antifouling coating. For example, they developed a novel zwitterionic polymer brush oxide (TMAO) antifouling material using the advantages of zwitterionic polymer hydration antifouling (Fig. 12b).¹⁶⁸ Compared with other zwitterionic polymer antifouling materials, the pTMAO antifouling material can effectively improve the adverse effect of the addition of salt molecules on the hydration surface of the zwitterionic polymer. Even if exposed to high salt solutions, such as in seawater, the surface hydration is only slightly reduced. In addition, the pTMAO material can completely resist the adhesion of proteins in seawater. Therefore, the pTMAO material with excellent antifouling properties developed by this work can effectively resist the damage of salt and proteins on the

hydrated surface. In addition, Zhou *et al.*¹⁶⁹ prepared a zwitterionic polymer-modified polypropylene antifouling microporous membrane (MPPM) material *via* the one-step co-deposition of PSBMA and polydopamine (PDA) (Fig. 12c). Compared with the UV-induced surface-grafting method, the MPPM microporous membranes showed significantly enhanced antifouling and permeability. Based on the continuous development of zwitterionic polymer coatings, Lu and Yu *et al.* have also done a lot of research on this, confirming that zwitterionic coatings can be used as potential antifouling materials.¹⁷⁰

Zwitterionic polymers also play an important role in marine biological antifouling materials. For example, Zhang *et al.*¹⁷¹ successfully developed an environmentally friendly fouling-resistance marine coating based on PSBMA for the first time. Rajan *et al.*¹⁷² prepared the ABA triblock copolymer PSBMA-*b*-PDMS-*b*-PSBMA through the three-step polymerization process of ring-opening polymerization (ROP), Michael addition and ATRP, and then directly incorporated it into a polyurethane coating and evaluated the stain-release performance of the coating. Studies found that the modified polyurethane coating shows an excellent fouling-release effect on bacteria and diatoms, which further proved that the migration of the PSBMA segment on the surface has an important impact on the adhesion of marine fouling organisms. Therefore, in future work, it is necessary to further adjust the composition of PSBMA to obtain a coating with wide decontamination properties. Cheng *et al.*¹⁷³ proposed a scheme that integrates antifouling, antibacterial and biocompatibility properties, that is, they added the mild cationic antibacterial agent salicylate (SA) into the antifouling pCBMA hydrogel to prepare the poly(*N,N*-dimethyl-*N*-(ethylcarbonylmethyl)-*N*-[2-(methacryloyloxy)-ethyl] ammonium salicylate) (pCBMA-1 C2 SA) hydrogel. This hydrogel can achieve the long-term release of SA through ester hydrolysis, thereby inhibiting the growth of planktonic bacteria. The CBMA ester becomes an antifouling surface after complete hydrolysis, which can prevent protein adsorption or bacterial aggregation. In addition, the pCBMA-1 C2 SA hydrogel can inhibit 99% of the growth of *E. coli* and *Staphylococcus epidermidis*. Moreover, SA is a widely used anti-inflammatory drug, so the hydrogel developed in this research may also reduce skin wounds caused by inflammatory tissue and promote wound healing.

In recent years, some multifunctional zwitterionic polymer antifouling materials with stimuli responsiveness have become research hotspots. Zheng *et al.*¹⁷⁴ synthesized zwitterionic polymers with salt ion responsiveness and modified it on a silicon wafer substrate to obtain a surface with adjustable friction and antifouling properties. Zuilhof *et al.*^{175,176} not only developed a new kind of zwitterionic polymer crosslinked network (ZPN) coating with self-healing and antifouling properties, but also obtained a zwitterionic coating with anti-protein and dual self-healing properties through coating zwitterionic polymer colloidal particles on different surfaces for the first time. In addition, they further systematically reviewed the recent progress of self-healing antifouling materials.¹⁷⁷ In short, his research group has made outstanding contributions to the development of zwitterionic self-healing coatings.

As a new generation of antifouling materials, zwitterionic-based topological polymers are favored for their efficient antifouling properties and broad application prospects. Through the development of functional zwitterionic antifouling coatings, the aging time of a material can be delayed, and thus the durability of the material is effectively improved and the antifouling time is prolonged. In a word, these new materials have great potential to create benefits for human life and society.

3.5 Other polymers

In addition to the above scope of applications, zwitterionic polymers are also widely used in many research fields such as protein modification,^{178,179} gene carriers,^{180–182} anticoagulant materials,^{183,184} hemodialyzers,^{185,186} cardiovascular stents,¹⁸⁷ orthopedic materials,^{188,189} tissue engineering and so on.¹⁹⁰ For example, Gao *et al.*¹⁹¹ used a site-specific *in situ* growth (SIG) strategy to efficiently link the zwitterionic polymer PMPC to the protein interferon- α (IFN- α) to prepare a new type of site-specific and highly active conjugate (INF-PMPC). Compared with PEG-modified interferon-alpha (PEGASYS), INF-PMPC showed a prolonged circulation half-life and excellent therapeutic efficacy. In addition, PMPC conjugates can completely inhibit the growth of tumors. Therefore, this method of synthesizing PMPC protein conjugates has provided a general platform for the development of next-generation protein therapeutic agents. Lam *et al.*¹⁹² synthesized the 2-(dimethylaminoethyl methacrylate and 2-methacryloyloxyethyl phosphorylcholine) (DMAEMA-MPC) diblock copolymer in which MPC blocks were used as a biocompatible steric stabilizer to maintain the stability of the complex, thus further effectively compressing DNA *via* a cationic block. Kim *et al.*¹⁸⁷ successfully prepared a phospholipid/poly(L-lactic acid) mixture (PLLA/PMB30W) by mixing polylactic acid (PLLA) and a water-soluble amphiphilic phospholipid polymer (PMB30W). The results of the study showed that even if the (PLLA/PMB30W) blend was implanted into rat subcutaneous tissue for 6 months, it still had a high density of phosphocholine groups on the surface of the tissue without significant bio-absorption. In addition, the enzyme-linked immunosorbent assay of PLLA/PMB30W determined that this mixture can quantitatively reduce pro-inflammatory cytokines. Since the selected phosphorylcholine group could significantly reduce tissue response *in vivo* and *in vitro*, the PLLA/PMB30W blend becomes a promising material for temporary cardiovascular stent devices. Ishihara *et al.*¹⁸⁹ prepared PMPC-*g*-(Co-Cr-Mo) copolymers on the surface of a metallic cobalt-chromium-molybdenum alloy (Co-Cr-Mo) by introducing biocompatible MPC and silane through surface grafting and surface initiation, respectively. The biocompatibility, lubricity and graft stability of the surface before and after modification were studied. The results showed that the silanized surface was soaked in PBS for 12 weeks without hydrolysis, and the lubricity was similar to that of the original joint. Therefore, the metal modified using the above method is expected to become an artificial bone-replacement material.

4. Conclusion and perspective

In this review, we have introduced the design and synthesis of zwitterionic-based polymer materials with different topologies, as well as their biomedical applications in drug delivery, antitumor therapy, biomedical diagnosis and antifouling coatings. In terms of drug delivery, block and star zwitterionic polymer drug carriers have high drug loadings and encapsulation efficiencies, low clearance rates and long circulation times. In addition, the responsive drug release of the polymer can be realized through functional modification combined with other materials that have pH, GSH, salt and other stimulus responsiveness. For antitumor activity, the smaller size and higher stability of star and dendritic zwitterionic polymers enable the rapid internalization of cells, resulting in a higher cell uptake and better antitumor effects. For biomedical diagnosis, the introduction of imaging, carbon dots and magnetic materials into zwitterionic topological polymer structures can be used for the early diagnosis and treatment of diseases. In terms of antifouling coatings, the nanoscale size and high graft density show excellent stability and lubricity, promoting the development of antifouling coatings and wear-resistant materials. To summarize, zwitterionic-based topological polymer materials have played a non-negligible role in biomedical research.

Although considerable progress in topological zwitterionic materials for biomedical applications has been well demonstrated, substantial challenges remain to be addressed in the future. First, only a few of these strategies have achieved clinical success, indicating that there are still potential limitations, such as complex synthesis processes, unclear mechanisms of drug delivery, and so on. Second, as a drug-delivery system, unexpected disintegration of drug carriers may cause an explosive release of drugs. In order to solve this problem, it is a great concern to construct a controlled-release drug system with multiple binding forces. For tumor treatment, how to achieve more accurate targeted regulation without damaging normal tissues remains a big challenge. For biomedical diagnosis, improving the sensitivity and accuracy of imaging diagnosis is still being explored. Based on the advantages of molecular brushes, brush-shaped zwitterionic polymers have made great progress in anti-fouling applications. However, their anti-fouling times are limited, and their industrial development has not yet been implemented. Hence, it is urgent to develop a sustainable antifouling coatings in complex media. Last but not least, although zwitterionic-based topological polymer materials have been greatly developed, the synthesis of cyclic and cyclic brush zwitterionic polymers is still a blank. Based on the unique advantages of cyclic polymers and zwitterionic materials, we envisage combining the cyclic topology with zwitterions and designing stimulus-responsive cyclic brush zwitterionic polymer materials, which could promote the potential development of topological zwitterionic polymer materials in the field of biomedicine. In the future, the development of zwitterion functionalization is unstoppable. With the goal of improving the multifunctional application of zwitterionic materials, reducing the costs, simplifying the

processes, and promoting the clinical transformation of zwitterionic-based topological polymer systems are expected to reduce human suffering and create unlimited benefits for the economy and society.

Abbreviations

PEG	Polyethylene glycol
PC	Phosphorylcholine
CB	Carboxybetaine
SB	Sulfobetaine
PSBMA	Poly(sulfobetaine methacrylate)
MPC	2-Methacryloyloxyethylphosphocholine
CBAA	Carboxybetaine acrylamide
ATRP	Atom transfer radical polymerization
SI-ATRP	Surface-initiated atom transfer radical polymerization
RAFT	Reversible addition-fragmentation chain transfer
ROP	Ring-opening polymerization
UCST	Upper critical solution temperature
LCST	Lower critical solution temperature
POSS	Polyhedral oligomeric silsesquioxanes
PDMAEMA	Poly[2(dimethylamino)ethyl methacrylate]
PNIPAM	Poly(<i>N</i> -isopropylacrylamide)
CD	Cyclodextrin
Ada	Adamantane
DOX	Doxorubicin
PDA	Polydopamine
PAMAM	Poly(amidoamine)
HA	Hyaluronic acid
BB	Bottle brush
PU	Polyurethane
GSH	Glutathione
ROS	Reactive oxygen species
DTT	Dithiothreitol
Dex	Dextran
SA	Succinic acid

Author contributions

The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

Conflicts of interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Notes and references

- S. Chen, L. Li, C. Zhao and J. Zhen, *Polymer*, 2010, **51**, 5283–5293.
- A. B. Lowe and C. L. McCormick, *Chem. Rev.*, 2002, **102**, 4177–4190.
- C. Leng, S. Sun, K. Zhang, S. Jiang and Z. Chen, *Acta Biomater.*, 2016, **40**, 6–15.
- A. Laschewsky and A. Rosenhahn, *Langmuir*, 2019, **35**, 1056–1071.
- J. Meng, Z. Cao, L. Ni, Y. Zhang, X. Wang and E. Liu, *J. Membr. Sci.*, 2014, **461**, 123–129.
- S. Chen, J. Zheng, L. Li and S. Jiang, *J. Am. Chem. Soc.*, 2005, **127**, 14473–14478.
- V. Gaberc-Porekar and I. Zore, *Curr. Opin. Drug Discovery*, 2008, **11**, 242–250.
- Z. Cao, L. Zhang and S. Jiang, *Langmuir*, 2012, **28**, 11625–11632.
- W. Lee and Y. Chen, *J. Appl. Polym. Sci.*, 2003, **89**, 1884–1889.
- Y. Terayama, M. Kikuchi, M. Kobayashi and A. Takahara, *Macromolecules*, 2010, **44**, 104–111.
- X. Zhu, M. Fryd and B. Wayland, *Polym. Degrad. Stab.*, 2013, **98**, 1173–1181.
- E. Ostuni, R. G. Chapman, R. E. Holmlin, S. Takayama and G. M. Whitesides, *Langmuir*, 2001, **17**, 5605–5620.
- K. Avgoustakis, *Curr. Drug Delivery*, 2004, **1**, 321–333.
- J. Cheng, B. A. Tepy, I. Sherifi, J. Sung, G. Luther, F. Gu, E. LevyNissenbaum, A. F. Radovic-Moreno, R. Langer and O. C. Farokhzad, *Biomaterials*, 2007, **28**, 869–876.
- Z. Cao and S. Jiang, *Nano Today*, 2012, **7**, 404–413.
- S. N. S. Alconcel, A. S. Baas and H. D. Maynard, *Polym. Chem.*, 2011, **2**, 1442–1448.
- A. J. Keefe and S. Jiang, *Nature Chem.*, 2012, **4**, 60–64.
- Y. Li, M. Gerth and H. Zuilhof, *Langmuir*, 2012, **28**, 12509–12517.
- P. R. Garay, R. E. Gewely and J. Armstrong, *Expert. Opin. Drug. Delivery*, 2012, **9**, 1319–1323.
- T. Shida and H. Kiwada, *Biol. Pharm. Bull.*, 2013, **36**, 889–891.
- E. T. M. Dams, P. Laverman, W. J. G. Oyen, G. Storm, G. L. Scherphof, J. W. M. Van der Meer, F. H. M. Corstens and O. C. Boerman, *J. Pharmacol. Exp. Ther.*, 2001, **292**, 1071–1079.
- T. Shida, S. Kashima and H. Kiwada, *J. Controlled Release*, 2008, **126**, 162–165.
- I. Hamad, A. C. Hunter and J. Szebeni, *Mol. Immunol.*, 2008, **46**, 225–232.
- M. Steiner, A. Attarbaschi and U. Kastner, *Pediatr. Blood. Cancer*, 2007, **49**, 640–642.
- L. Zheng, H. S. Sundaram, Z. Wei, C. Li and Z. Yuan, *React. Funct. Polym.*, 2017, **118**, 5161.
- T. Hasegawa, Y. Iwasaki and K. Ishihara, *J. Biomed. Mater. Res.*, 2002, **63**, 333–341.
- L. R. Carr, H. Xue and S. Jiang, *Biomaterials*, 2011, **32**, 961–968.
- L. R. Carr, G. Cheng, H. Xue and S. Jiang, *Langmuir*, 2010, **26**, 14793–14798.
- R. S. Kane, P. Deschatelets and G. M. Whitesides, *Langmuir*, 2003, **19**, 2388–2391.
- J. Ladd, Z. Zhang, S. Chen, J. C. Hower and S. Jiang, *Biomacromolecules*, 2008, **9**, 1357–1361.
- Y. Wang, Y. Luo and O. Zhao, *ACS Appl. Mater. Interfaces*, 2016, **8**, 19899.
- K. H. Jung, H. J. Kim, M. H. Kim and J. C. Lee, *J. Membr. Sci.*, 2021, **618**, 118677.
- A. Noguchi, T. Masuda, C. Chen, S. Yoshizawa, N. Isu and M. Takai, *Mater. Adv.*, 2020, **1**, 2737–2744.
- C. Cao and X. Luo, *Biomaterials*, 2014, **35**, 4517–4524.
- Y. Han, Y. Qian, X. Zhou, H. Hu, X. Liu, Z. Zhou, J. Tang and Y. Shen, *Polym. Chem.*, 2016, **7**, 6354–6362.
- L. Liu, D. Zhang, B. P. Ren, X. Gong, L. Xu, Z. Feng, Y. Chang, Y. He and J. Zheng, *J. Mater. Chem. B*, 2020, **8**, 3814–3828.
- G. Ye, J. Lee, F. Perreault and M. Elimelech, *ACS Appl. Mater. Interfaces*, 2015, **7**, 23069–23079.
- Y. He, X. Wan, W. Lin, J. Li, Z. Li, F. Luo, J. Li, H. Tan and Q. Fu, *Biomater. Sci.*, 2020, **8**, 6890.
- Y. He, X. Wan, K. Xiao, W. Lin, J. Li, Z. Li, F. Luo, H. Tan, J. Li and Q. Fu, *Biomater. Sci.*, 2019, **7**, 5369–5382.
- X. Wan, Y. Zhang, Y. Deng, Q. Zhang and Q. Fu, *Soft Matter*, 2015, **11**, 4197–4207.
- T. Xiang, L. Zhang, R. Wang, Y. Xia, B. Su and C. Zhao, *J. Colloid Interface Sci.*, 2014, **432**, 47–56.
- R. Narain, M. Ahmed, N. Bhuchar and K. Ishihara, *Bioconjugate Chem.*, 2011, **22**, 1228–1238.
- G. Liu, Q. Jin, X. Liu, L. Lv, C. Chen and J. Ji, *Soft Matter*, 2011, **7**, 662–669.
- G. Liu, C. Chen and J. Ji, *Soft Matter*, 2012, **8**, 8811–8821.
- Y. Liu, W. Peng, X. Zhang, J. Peng, P. Liu and J. Shen, *J. Mater. Sci. Technol.*, 2021, **62**, 96–106.
- V. Vasantha, S. Jana, A. Parthiban and J. G. Vancso, *Chem. Commun.*, 2014, **50**, 46–48.
- V. Hildebrand, A. Laschewsky, M. Päch, P. Müller-Buschbaum and C. M. Papadakis, *Polym. Chem.*, 2017, **8**, 310–322.
- L. P. Kreuzer, T. Widmann, C. Geiger, P. Wang, A. Vagias, J. E. Heger, M. Haese, V. Hildebrand, A. Laschewsky and C. M. Papadakis, *Langmuir*, 2021, **37**, 9179–9191.
- J. Wang, L. Wang, X. Ji, L. Liu and H. Zhao, *Macromolecules*, 2017, **50**, 2284–2295.
- M. E. Taylor, S. J. Louder, A. Asatekin and M. J. Panzer, *ACS Mater. Lett.*, 2020, **2**, 261–265.

- 51 X. Wang, H. Gu and G. Wu, *J. Dispers. Sci. Technol.*, 2020, **18**, 1805331.
- 52 Q. Liu, S. Anuradha and L. Liu, *Biomacromolecules*, 2013, **14**, 226–231.
- 53 I. Gitsov, L. Wang, N. Vladimirov, A. Simonyan, D. J. Kiemle and A. Schutz, *Biomacromolecules*, 2014, **15**, 4082–4095.
- 54 X. Pang, L. Zhao, M. Akinc, J. Kim and Z. Lin, *Macromolecules*, 2011, **44**, 3746–3752.
- 55 W. Lin, G. Ma, F. Ji, J. Zhang, L. Wang, H. Sun and S. Chen, *J. Mater. Chem. B*, 2015, **3**, 440–448.
- 56 Y. Li, H. Guo, J. Zheng, J. Gan, Y. Zhang, X. Guan, K. Wu and M. Lu, *RSC Adv.*, 2014, **4**, 54268–54281.
- 57 W. Cao and L. Zhu, *Macromolecules*, 2011, **44**, 1500–1512.
- 58 H. Hu, X. Fan and Z. Cao, *Polymer*, 2005, **46**, 9514–9522.
- 59 N. Malik, R. Wiwattanapatapee, R. Klopsch, K. Lorenz, H. Frey, J. Weener, E. Meijer, W. Paulus and R. Duncan, *J. Controlled Release*, 2000, **65**, 133–148.
- 60 C. Kojima, K. Kono, K. Maruyama and T. Takagishi, *Bioconjugate Chem.*, 2000, **11**, 910–917.
- 61 L. Jia, J. Xu, H. Wang and J. Ji, *Colloids Surf., B*, 2011, **84**, 49–54.
- 62 L. Wang, Z. Wang, G. Ma, W. Lin and S. Chen, *Langmuir*, 2013, **29**, 8914–8921.
- 63 L. Li, Y. Wang, F. Ji, Y. Wen, J. Li, B. Yang and F. Yao, *J. Biomater. Sci.: Polym.*, 2014, **25**, 1641–1657.
- 64 L. Qiu, R. Wang, C. Zheng, Y. Jin and L. Jin, *Nanomedicine*, 2010, **5**, 193–208.
- 65 D. Cordes, P. Lickiss and F. Rataboul, *Chem. Rev.*, 2010, **110**, 2081–2173.
- 66 Y. Yang, X. Wang, Y. Hu, H. Hu, D. Wu and F. Xu, *ACS Appl. Mater. Interfaces*, 2014, **6**, 1044–1052.
- 67 K. Knop, R. Hoogenboom, D. Fischer and U. S. Schubert, *Angew. Chem., Int. Ed.*, 2010, **49**, 6288–6308.
- 68 S. Jiang and Z. Cao, *Adv. Mater.*, 2010, **22**, 920–932.
- 69 Y. Li, B. Xu, T. Bai and W. Liu, *Biomaterials*, 2015, **55**, 12–23.
- 70 Y. Pu, Z. Hou, M. Khin, R. Zamudio-Vazquez, K. Poon, H. Duan and M. B. Chan-Park, *Biomacromolecules*, 2017, **18**, 44–55.
- 71 J. Seuring and S. Agarwal, *ACS Macro Lett.*, 2013, **2**, 597–600.
- 72 N. Morimoto, K. Muramatsu, T. Wazawa, Y. Inoue and M. Suzuki, *Macromol. Rapid Commun.*, 2014, **35**, 103–108.
- 73 S. Chen and S. Jiang, *Adv. Mater.*, 2008, **20**, 335–338.
- 74 M. Zhang, W. Shen, Q. Xiong, H. Wang, Z. Zhou, W. Chen and Q. Zhang, *RSC Adv.*, 2015, **5**, 28133–28140.
- 75 J. A. Syrett, D. M. Haddleton, M. R. Whittake, T. P. Davis and C. Boyer, *Chem. Commun.*, 2011, **47**, 1449–1451.
- 76 J. K. OhD, J. Siegwart and H. I. Lee, *J. Am. Chem. Soc.*, 2007, **129**, 5939–5945.
- 77 Y. Zhang, W. Poon, A. J. Tavares, I. D. McGilvray and W. Chan, *J. Controlled Release*, 2016, **240**, 332–348.
- 78 S. Lou, X. Zhang, M. Zhang, S. Ji, W. Wang, J. Zhang, C. Li and D. Kong, *Int. J. Nanomed.*, 2017, **12**, 3653–3664.
- 79 S. Mignani, S. El Kazzouli, M. M. Bousmina and J. P. Majoral, *Chem. Rev.*, 2014, **114**, 1327–1342.
- 80 J. Bugno, H. J. Hsu and S. Hong, *Biomater. Sci.*, 2015, **3**, 1025–1034.
- 81 S. Svenson, *Chem. Soc. Rev.*, 2015, **44**, 4131–4144.
- 82 Q. Sun, M. Radosz and Y. Shen, *J. Controlled Release*, 2012, **164**, 156–169.
- 83 C. Stoffelen and J. Huskens, *Nanoscale*, 2015, **7**, 7915–7919.
- 84 X. Pang, Y. Jiang, Q. Xiao, A. W. Leung, H. Hua and C. Xu, *J. Controlled Release*, 2016, **222**, 116–129.
- 85 D. Huang, F. Yang, X. Wang, H. Shen, Y. You and D. Wu, *Polym. Chem.*, 2016, **7**, 6154–6158.
- 86 D. Huang, Y. Wang, F. Yang, H. Shen, Z. Weng and D. Wu, *Polym. Chem.*, 2017, **8**, 6675–6687.
- 87 Y. Yuan, C. Mao, X. Du, J. Du, F. Wang and J. Wang, *Adv. Mater.*, 2012, **24**, 5476–5480.
- 88 Z. Cao and S. Jiang, *Nano Today*, 2012, **7**, 404–413.
- 89 L. Wang, C. Shi, X. Wang, D. Guo, T. M. Duncan and J. Luo, *Biomaterials*, 2019, **215**, 119233.
- 90 L. Wang, X. Ji, D. Guo, C. Shi and J. Luo, *Mol. Pharmaceutics*, 2021, **18**, 2349–2359.
- 91 X. Li, X. Hu and T. Cai, *Langmuir*, 2017, **33**, 4477–4489.
- 92 S. Lee and N. D. Spencer, *Science*, 2008, **319**, 575.
- 93 U. Raviv, S. Giasson, N. Kampf, J. F. Gohy, R. Jerome and J. Klein, *Nature*, 2003, **425**, 163–165.
- 94 J. Klein, E. Kumacheva, D. Mahalu, D. Perahia and L. J. Fetters, *Nature*, 1994, **370**, 634–636.
- 95 M. Chen, W. H. Briscoe, S. P. Armes and J. Klein, *Science*, 2009, **323**, 1698–1701.
- 96 M. Kobayashi, Y. Terayama, N. Hosaka, M. Kaido, A. Suzuki, N. Yamada, N. Torikai, K. Ishihara and A. Takahara, *Soft Matter*, 2007, **3**, 740–746.
- 97 W. Feng, J. L. Brash and S. Zhu, *Biomaterials*, 2006, **27**, 847–855.
- 98 M. Kobayashi, Y. Terayama, M. Kikuchi and A. Takahara, *Soft Matter*, 2013, **9**, 5138–5148.
- 99 W. Yan, S. N. Ramakrishna, N. D. Spence and E. M. Benetti, *Langmuir*, 2019, **35**, 11255–11264.
- 100 X. Banquy, J. Burdyska, D. W. Lee, K. Matyjaszewski and J. Israelachvili, *J. Am. Chem. Soc.*, 2014, **136**, 6199–6202.
- 101 J. Faivre, B. R. Shrestha, J. Burdyska, G. Xie, F. Moldovan, T. Delair, S. Benayoun, L. David, K. Matyjaszewski and X. Banquy, *ACS Nano*, 2017, **11**, 1762–1769.
- 102 V. Adibnia, M. Olszewski, G. D. Crescenzo, K. Matyjaszewski and X. Banquy, *J. Am. Chem. Soc.*, 2020, **142**, 14843–14847.
- 103 Y. Yu, G. J. Vancso and S. Beer, *Eur. Polym. J.*, 2017, **89**, 221–229.
- 104 X. Ding, S. Duan, X. Ding, R. Liu and F. Xu, *Adv. Funct. Mater.*, 2018, **28**, 1802140.
- 105 J. Zhang, W. Zhu, B. Xin, S. Lin, L. Jin and H. Wang, *Biomater. Sci.*, 2019, **7**, 3795–3800.
- 106 W. He, Y. Zhang, J. Li, Y. Gao, F. Luo, H. Tan, K. Wang and Q. Fu, *Sci. Rep.*, 2016, **6**, 32140.
- 107 Y. He, X. Wan, W. Lin, J. Li, Z. Li, F. Luo, J. Li, H. Tan and Q. Fu, *Biomater. Sci.*, 2020, **8**, 6890–6902.
- 108 X. Banquy, J. Burdyska, D. W. Lee, K. Matyjaszewski and J. Israelachvili, *J. Am. Chem. Soc.*, 2014, **136**, 6199–6202.

- 109 X. Qiu, F. Tanaka and F. M. Winnik, *Macromolecules*, 2007, **40**, 7069–7071.
- 110 M. Zhang, Y. Liu, J. Peng, Y. Liu, F. Liu, W. Ma, L. Ma, C. Yu and H. Wei, *Polym. Chem.*, 2020, **11**, 6139–6148.
- 111 H. Wei, L. R. Volpatti, D. L. Sellers, D. O. Maris, I. W. Andrews, A. S. Hemphill, L. W. Chan, D. S. H. Chu, P. J. Horner and S. H. Pun, *Angew. Chem., Int. Ed.*, 2013, **52**, 5377–5381.
- 112 T. Yamamoto and Y. Tezuka, *Polym. Chem.*, 2011, **2**, 1930–1941.
- 113 W. Jeong, E. J. Shin, D. A. Culkun, J. L. Hedrick and R. M. Waymouth, *J. Am. Chem. Soc.*, 2009, **131**, 4884–4891.
- 114 Y. Tezuka, *Acc. Chem. Res.*, 2017, **50**, 2661–2672.
- 115 S. H. Lahasky, W. K. Serem, L. Guo, J. C. Garno and D. Zhang, *Macromolecules*, 2011, **44**, 9063–9074.
- 116 K. Chen, F. Hu, H. Gu and H. Xu, *J. Mater. Chem. B*, 2017, **5**, 435–443.
- 117 P. Saha, R. Ganguly, X. Li, R. Das, N. K. Singha and A. Pich, *Macromol. Rapid Commun.*, 2021, **42**, 2100112.
- 118 M. Ajmal, S. Demirci, M. Siddiq, N. Aktas and N. Sahiner, *Colloids Surf.*, 2015, **486**, 29.
- 119 Z. Ding, W. He, J. Tao, W. Jiang, L. Li and T. Pan, *J. Polym. Sci. Pol. Chem.*, 2011, **49**, 2783–2789.
- 120 W. Yang, L. Zhang, S. Wang, A. D. White and S. Jiang, *Biomaterials*, 2009, **30**, 5617–5621.
- 121 G. Liu, Z. Liu, N. Li, X. Wang, F. Zhou and W. Liu, *ACS Appl. Mater. Interfaces*, 2014, **6**, 20452–20463.
- 122 X. Qiu, J. Zhang, L. Cao, Q. Jiao, J. Zhou, L. Yang, H. Zhang and Y. Wei, *ACS Appl. Mater. Interfaces*, 2021, **13**, 7060–7069.
- 123 J. Ren, Y. Liu, Z. Wang, S. Chen, Y. Ma, H. Wei and S. Lv, *Adv. Funct. Mater.*, 2021, 2107404.
- 124 P. Saha, R. Ganguly, X. Li, R. Das, N. K. Singha and A. Pich, *Macromol. Rapid Commun.*, 2021, **42**, 2100112.
- 125 J. Fang, H. Nakamura and H. Maeda, *Adv. Drug Delivery Rev.*, 2011, **63**, 136–151.
- 126 G. Liu, P. Lv, C. Chen, X. Hu and J. Ji, *Macromol. Chem. Phys.*, 2011, **212**, 643–651.
- 127 L. Wu, L. Zhang, G. Shi and C. Ni, *Mater. Sci. Eng., C*, 2016, **61**, 278–285.
- 128 W. Lin, G. Ma, F. Ji, J. Zhang, L. Wang, H. Sun and S. Chen, *J. Mater. Chem. B*, 2015, **3**, 440–448.
- 129 X. Chen, S. McRae, S. Parelkar and T. Emrick, *Bioconjugate Chem.*, 2009, **20**, 2331–2341.
- 130 M. Cai, M. Leng, A. Lu, L. He, X. Xie, L. Huang, Y. Ma, Y. Chen and X. Luo, *Colloids Surf., B*, 2015, **126**, 1–9.
- 131 Y. Wen, Z. Zhang and J. Li, *Adv. Funct. Mater.*, 2014, **24**, 3874–3884.
- 132 L. Jia, J. Xu, H. Wang and J. Ji, *Colloids Surf., B*, 2011, **84**, 49–54.
- 133 Y. Men, S. Peng, P. Yang, Q. Jiang, Y. Zhang, B. Shen, P. Dong and Z. Pang, *ACS Appl. Mater. Interfaces*, 2018, **10**, 23509–23521.
- 134 S. Peng, Y. Men, R. Xie, Y. Tian and W. Yang, *J. Colloid Interface Sci.*, 2019, **539**, 19–29.
- 135 S. Lou, X. Zhang, W. Wang, J. Zhang, C. Li and D. Kong, *Int. J. Nanomed.*, 2017, **12**, 3653–3664.
- 136 J. Chen, Z. Jiao, L. Lin, Z. Guo, C. Xu, Y. Li, H. Tian and X. Chen, *J. Polym. Sci.*, 2015, **33**, 830–837.
- 137 J. Ma, K. Kang, Q. Yi, Z. Zhang and Z. Gu, *RSC Adv.*, 2016, **6**, 64778–64790.
- 138 Z. Wu, Z. Gan, B. Chen, F. Chen, J. Cao and X. Luo, *Biomater. Sci.*, 2019, **7**, 3190–3203.
- 139 J. Cao, X. Xie, A. Lu, B. He, Y. Chen, Z. Gu and X. Luo, *Biomaterials*, 2014, **35**, 4517–4524.
- 140 Z. Wu, M. Cai, X. Xie, L. He, L. Huang, Y. Chen and X. Luo, *RSC Adv.*, 2015, **5**, 106989–107000.
- 141 S. Tu, Y. Chen, Y. Qiu, K. Zhu and X. Luo, *Macromol. Biosci.*, 2011, **11**, 1416–1425.
- 142 H. Sun, M. Y. Z. Chang, W. Cheng, Q. Wang, A. Commisso, M. Capeling and C. Cheng, *Acta Biomater.*, 2017, **64**, 290–300.
- 143 Y. Zhang, Z. Zhang, C. Liu, W. Chen, C. Li, W. Wu and X. Jiang, *Polym Chem.*, 2017, **8**, 1672–1679.
- 144 M. Wan, Z. Liu, T. Li, H. Chen, Q. Wang, T. Chen, Y. Tao and C. Mao, *Angew. Chem., Int. Ed.*, 2021, **60**, 16139–16148.
- 145 Z. Zhang, S. Chen and S. Jiang, *Biomacromolecules*, 2006, **7**, 3311–3315.
- 146 P. Akkhat, S. Kiatkamjornwong, S. I. Yusa, V. P. Hoven and Y. Iwasaki, *Langmuir*, 2012, **28**, 5872–5881.
- 147 M. Chu, H. Kudo, T. Shirai, K. Miyajima, H. Saito, N. Morimoto, K. Yano, Y. Iwasaki, K. Akiyoshi and K. Mitsubayashi, *Biomed. Microdevices*, 2009, **11**, 837–842.
- 148 J. L. Jameson and D. L. Longo, *Obstet. Gynecol. Surv.*, 2015, **70**, 612–614.
- 149 H. Vaisocherova, W. Yang, Z. Zhang, Z. Cao, G. Cheng, M. Piliarik, J. Homola and S. Jiang, *Anal. Chem.*, 2008, **80**, 7894–7901.
- 150 G. Jia, Z. Cao, H. Xue, Y. Xu and S. Jiang, *Langmuir*, 2009, **25**, 3196–3199.
- 151 Y. Chen, H. Han, H. Tong, T. Chen, H. Wang, J. Ji and Q. Jin, *ACS Appl. Mater. Interfaces*, 2016, **8**, 21185–21192.
- 152 L. Zhang, H. Xue, C. Gao, L. Carr, J. Wang, B. Chu and S. Jiang, *Biomaterials*, 2010, **31**, 6582–6588.
- 153 L. Zhang, H. Xue, C. Gao, A. Keefe, J. Wang and S. Jiang, *Biomaterials*, 2011, **32**, 4604–4608.
- 154 Y. Han, Y. Qian, X. Zhou, H. Hu, X. Liu, Z. Zhou, J. Tang and Y. Shen, *Polym. Chem.*, 2016, **7**, 6354–6362.
- 155 X. Xie, J. C. Doloff, V. Yesilyurt, A. Sadraei, J. J. McGarrigle, M. Commis, O. Veiseh, S. Farah, D. Isa and S. Ghanis, *Nat. Biomed. Eng.*, 2018, **2**, 894–906.
- 156 O. Wiarachai, T. Vilaivan, Y. Iwasaki and V. P. Hoven, *Langmuir*, 2016, **32**, 184–1194.
- 157 Y. Kitayama and T. Takeuchi, *Anal. Chem.*, 2014, **86**, 5587–5594.
- 158 Y. Xie, M. Liu and J. Zhou, *Appl. Surf. Sci.*, 2012, **258**, 153–8159.
- 159 S. Chen, J. Zheng, L. Li and S. Jiang, *J. Am. Chem. Soc.*, 2005, **127**, 14473–14478.
- 160 Y. Xie, M. Liu and J. Zhou, *Appl. Surf. Sci.*, 2012, **258**, 8153–8159.
- 161 W. Wang, Y. Lu, J. Xie, H. Zhu and Z. Cao, *Chem. Commun.*, 2016, **52**, 4671–4674.
- 162 N. Cai, Q. Li, J. Zhang, T. Xu, W. Zhao, J. Yang and L. Zhang, *J. Colloid Interface Sci.*, 2017, **503**, 168–177.

- 163 A. Noguchi, T. Masuda, C. Chen, S. Yoshizawa, N. Isu and M. Takai, *Mater. Adv.*, 2020, **1**, 2737–2744.
- 164 X. Li, X. Hu and T. Cai, *Langmuir*, 2017, **33**, 4477–4489.
- 165 H. J. Kwon, Y. Lee, L. T. Phuong, G. M. Seon, E. Kim, J. K. Park, H. Yoon and K. D. Park, *Acta Biomater.*, 2017, **61**, 169–179.
- 166 L. Mi, H. Xue, Y. Li and S. Jiang, *Adv. Funct. Mater.*, 2011, **21**, 4028–4034.
- 167 W. Zhao, Q. Ye, H. Hu, X. Wang and F. Zhou, *J. Mater. Chem. B*, 2014, **2**, 5352–5357.
- 168 H. Huang, C. Zhang, R. Crisci, T. Lu, H. C. Hung, M. S. J. Sajib, P. Sarker, J. Ma, T. Wei, S. Jiang and Z. Chen, *J. Am. Chem. Soc.*, 2021, **143**, 16786–16795.
- 169 R. Zhou, P. Ren, H. Yang and Z. Xu, *J. Membr. Sci.*, 2014, **466**, 18–25.
- 170 C. Zhang, J. Lu, Y. Hou, W. Xiong, K. Sheng and H. Lu, *ACS Appl. Mater. Interfaces*, 2018, **10**, 17463–17470.
- 171 Z. Zhang, J. A. Finlay, L. Wang, Y. Gao, J. A. Callow, M. E. Callow and S. Jiang, *Langmuir*, 2009, **25**, 13516–13521.
- 172 R. B. Bodkhe, S. J. Stafslie, J. Daniels, N. Cilz, A. J. Muelhberg, S. E. M. Thompson, M. E. Callow, J. A. Callow and D. C. Rajan, *Prog. Org. Coat.*, 2015, **78**, 369–380.
- 173 G. Cheng, H. Xue, G. Li and S. Jiang, *Langmuir*, 2010, **26**, 10425–10428.
- 174 J. Yang, H. Chen, S. Xiao, M. Shen, F. Chen, P. Zhong and J. Zheng, *Langmuir*, 2015, **31**, 9125–9133.
- 175 Z. Wang, A. E. Van, S. P. Pujari, H. Feng, J. A. Dijkstra, M. M. J. Smulders and H. Zuilhof, *J. Mater. Chem. B*, 2017, **5**, 6728–6733.
- 176 Z. Wang, G. Fei, H. Xia and H. Zuilhof, *J. Mater. Chem. B*, 2018, **6**, 6930–6935.
- 177 Z. Wang, L. Scheres, H. Xia and H. Zuilhof, *Adv. Funct. Mater.*, 2020, **30**, 1908098.
- 178 S. Liang, Y. Liu, X. Jin, G. Liu, J. Wen, L. Zhang, J. Li, X. Yuan, S. Irvin, Y. Chen and W. Chen, *Nano Res.*, 2016, **9**, 1022–1031.
- 179 L. Zhang, Y. Liu, G. Liu, D. Xu, S. Liang, X. Zhu, Y. Lu and H. Wang, *Nano Res.*, 2016, **9**, 2424–2432.
- 180 M. Ukawa, H. Akita, T. Masuda, Y. Hayashi, T. Konno, K. Ishihara and H. Harashima, *Biomaterials*, 2010, **31**, 6355–6362.
- 181 F. Dai and W. Liu, *Biomaterials*, 2011, **32**, 628–638.
- 182 F. Dai, P. Wang, Y. Wang, L. Tang, J. Yang, W. Liu, H. Li and G. Wang, *Polymer*, 2008, **49**, 5322–5328.
- 183 W. Feng, S. Zhu, K. Ishihara and J. L. Brash, *Biointerphases*, 2006, **1**(1), 50–60.
- 184 Y. J. Shih and Y. Chang, *Langmuir*, 2010, **26**, 17286–17294.
- 185 T. Hasegawa, Y. Iwasaki and K. Ishihara, *J. Biomed. Mater. Res.*, 2002, **63**, 333–341.
- 186 Y. Iwasaki, N. Nakabayash and K. Ishihara, *J. Artif. Organs*, 2003, **6**, 260–266.
- 187 H. I. Kim, K. Ishihara, S. Lee, J. H. Seo, H. Y. Kim, D. Suh, M. U. Kim, T. Konno, M. Takai and J. S. Seo, *Biomaterials*, 2011, **32**, 2241–2247.
- 188 M. Kyomoto, T. Moro, K. Saiga, M. Hashimoto, H. Ito, H. Kawaguchi, Y. Takatori and K. Ishihara, *Biomaterials*, 2012, **33**, 4451–4459.
- 189 M. Kyomoto, T. Moro, K. Saiga, F. Miyaji, H. Kawaguchi, Y. Takatori, K. Nakamura and K. Ishihara, *Biomaterials*, 2010, **31**, 658–668.
- 190 Y. Hong, S. H. Ye, A. Nieponice, L. Soletti, D. A. Vorp and W. R. Wagner, *Biomaterials*, 2009, **30**, 2457–2467.
- 191 J. Hu, G. Wang, W. Zhao and W. Gao, *J. Controlled Release*, 2016, **237**, 71–77.
- 192 J. K. W. Lam, Y. M. S. P. Armes, A. L. Lewis, T. Baldwin and S. Stolnik, *J. Controlled Release*, 2004, **100**, 293–312.
- 193 Z. Xiong, C. S. Alves, J. Wang, A. Li, J. Liu, M. Shen, J. Rodrigues, H. Tomas and X. Shi, *Acta Biomater.*, 2019, **99**, 320–329.