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A review on the cytotoxicity of graphene quantum dots: from experiment to simulation

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Graphene quantum dots (GQDs) generate intrinsic fluorescence and improve the aqueous stability of graphene oxide (GO) while maintaining wide chemical adaptability and high adsorption capacity. Despite GO's remarkable advantages in bio-imaging, bio-sensing, and other biomedical applications, many experiments and simulations have focused on the biosafety of GQDs. Here, we review the findings on the biosafety of GQDs from experiments; then, we review the results from simulated interactions with biological membranes, DNA molecules, and proteins; finally, we examine the intersection between experiments and simulations. The biosafety results from simulations are explained in detail. Based on the literature and our experiments, we also discuss the trends toward GQDs with better biosafety.

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1. Introduction

Graphene quantum dots (GQDs), a novel type of zero-dimensional luminescent nanomaterial, are small graphene fragments varying in size from 1 to 100 nm.^{1,2} Confined by the

quantum-size effect, GQDs exhibit extraordinary opto-electronic properties,³⁻⁶ because of which they have the potential to replace the well-known metal chalcogenide-based quantum dots. In recent years, GQD-based nanomaterials have become an important research topic and have attracted much attention.⁷⁻⁹ GQDs have achieved rapid growth as breakthrough tools for multiple purposes in various fields of science including photonics, composites, energy,^{10,11} and electronics.¹² In particular, GQD-based nanomaterials have shown much promise in biomedical fields, particularly for diagnostics,^{13,14} drug delivery,^{15,16} and bioimaging^{17,18} *in vitro* and *in vivo*. At the same time, questions about the biosafety of GQDs have also attracted much attention in recent years.¹⁹⁻²³ However, a paradox concerning the biosafety of GQDs has emerged from experiments.

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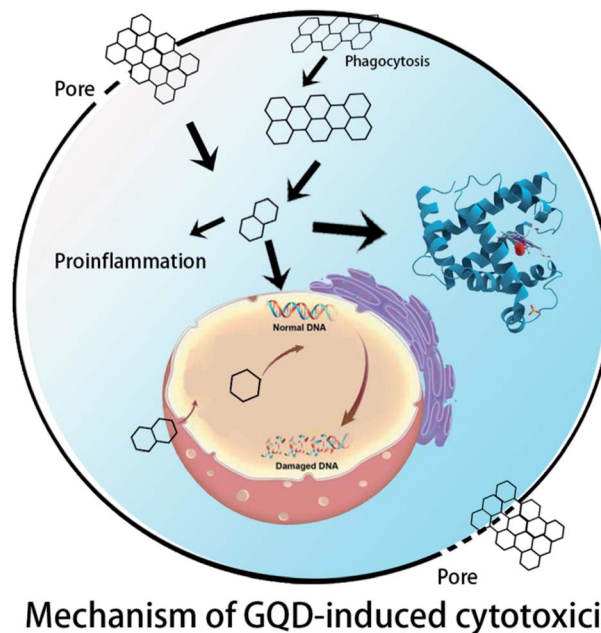
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Many studies have shown that GQDs have excellent biocompatibility and low cytotoxicity,^{24,25} but others have shown them to have high cytotoxicity.²⁶ In particular, the biological interaction mechanism of GQDs with biomolecules including DNA, proteins, and cell membranes is not well understood.

GQDs are unique and variable, and many properties including the degree of surface oxidation, surface charge density, doping status, or links with polymers can be dramatically different in different settings, yielding clearly different interaction behaviors with biomolecules. Therefore, experimental studies alone are not sufficient, and they have certain limitations for understanding the cytotoxicity of GQDs (especially at the atomic and molecular levels) which hinder the design and development of GQDs for biomedicine. As a complementary tool to experimentation, theoretical calculation covering the length scale from the atomic to the molecular level has been extensively used to uncover the cytotoxicity mechanism of carbon nanomaterials such as carbon nanotubes,^{27,28} graphene sheets,^{29–33} and fullerene.^{34,35} With the development of supercomputers, theoretical calculations have played a significant role in both providing atomistic pictures of



Mechanism of GQD-induced cytotoxicity

Fig. 1 Schematic diagram of the mechanism of GQD-induced cytotoxicity.

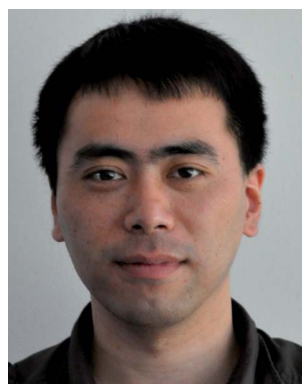


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In Section 2, experiments on the cytotoxicity of GQDs *in vitro* and *in vivo* are summarized. In Section 3, the simulation results of GQDs with different biomolecules including DNA, proteins, and lipid membranes are summarized. In Section 4, the intersections between experiment and theoretical calculation and the future and outlook of GQD biosafety are discussed (Fig. 1).

2. Experiments on the toxicity of GQDs

Nanomaterials can enter into cells and affect cell division, proliferation, apoptosis, and more. Klopfer *et al.* found that GQDs of less than 5 nm could enter into *E. coli* and *Bacillus subtilis* cells directly and produce toxic effects.³⁸ Chong *et al.* revealed that the cytotoxicity of GO is less than that of pristine graphene,³⁹ which agreed with Liao's result.⁴⁰ Akahvan *et al.* discovered that pristine graphene could greatly affect DNA even at very low concentrations.⁴¹ They also pointed out that the

genotoxicity of graphene in human stem cells depended on the size of graphene particles.⁴² This was also confirmed by Ros's experiment, where the small size of GQDs demonstrated good biocompatibility and ability to pass through the ultrafiltration barrier in an *in vitro* model.⁴³ In addition, due to the different oxidation extent and surface chemistry of GO sheets, GO shows low cytotoxicity in some experiments but high cytotoxicity in others.^{44,45} As shown in confocal laser scanning microscopy images (Fig. 2B) and consistent with the MTT assay results, cells incubated with a high concentration (200 $\mu\text{g ml}^{-1}$) of GO and rGO showed bright red fluorescence, indicating that severe cell death was induced; GO seemed to cause more cell death than rGO. In stark contrast, cells treated with GQDs, aminated GQDs (GQD-NH₂), carboxyl GQDs (GQD-COOH), and graphene oxide quantum dots (GOQDs) still showed high cell viability. GQD-COOH also showed low systemic toxicity for nanodrug-delivery system *in vivo* experiments.⁴⁶ From Wang's experiment, it was also confirmed that GQDs, nitro doped GQDs (GQD-N) and folic

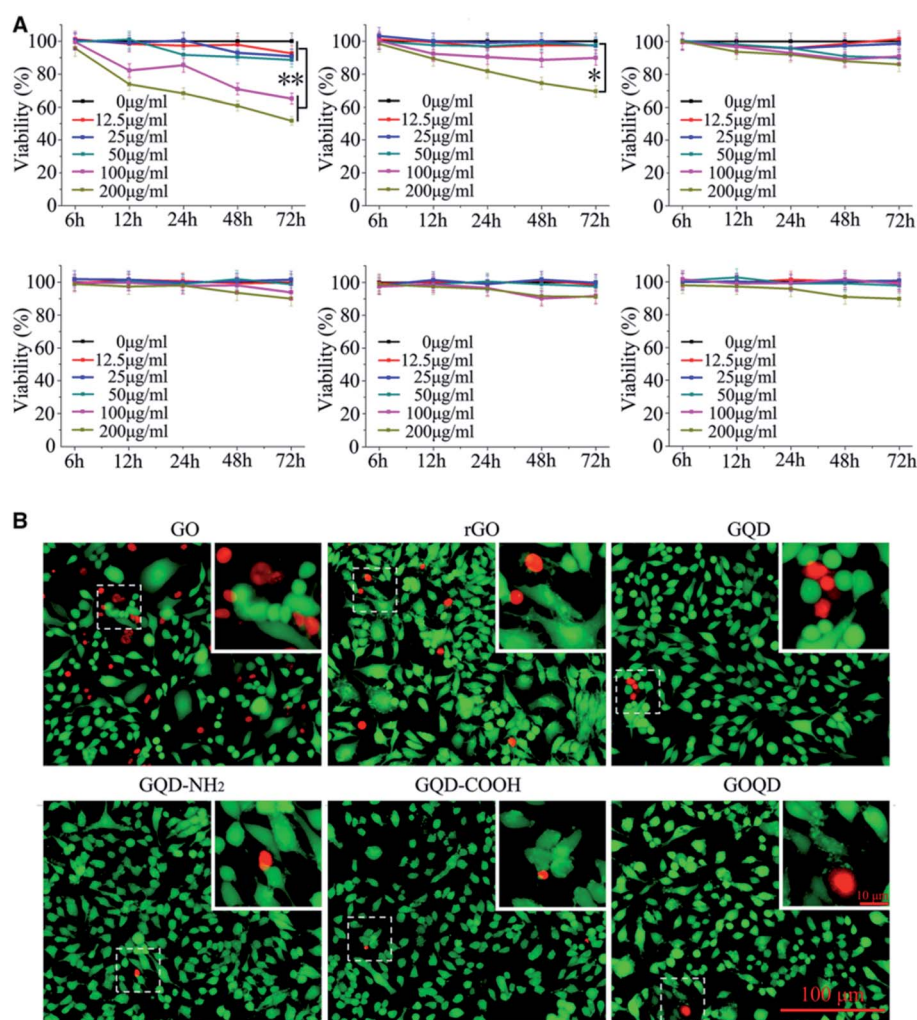


Fig. 2 Relative viability of tumor cells after incubation with GQDs. (A) HL-7702 cells incubated with various concentrations of GQDs for 6, 12, 24, 48, and 72 h. (B) Fluorescence images of HL-7702 cells treated with GQDs (200 $\mu\text{g ml}^{-1}$) for 24 h. Viable cells were stained with calcein AM, while dead cells were stained with PI. Inset: high magnification images of the area within the white dotted square. Reprinted with permission from ref. 48, Copyright© 2018, Oxford University Press.



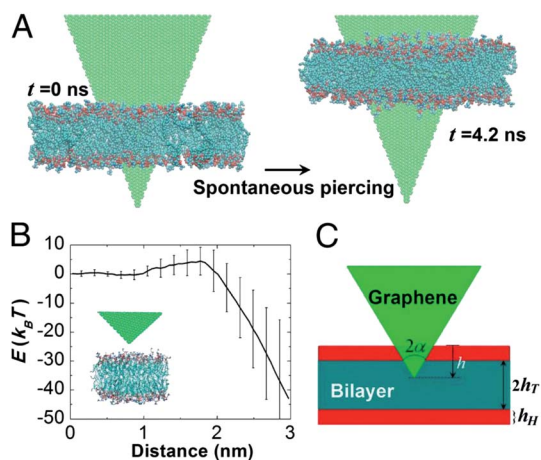


Fig. 3 All-atom molecular dynamics simulations of corner piercing of GQDs across a lipid bilayer. (A) Simulations directly showing that the corner piercing proceeds spontaneously. (B) GQD–bilayer interaction energy as a function of the penetration distance, showing the existence of an energy barrier of about 5 $k_B T$ associated with corner piercing. The mean value of interaction energy is obtained from 11 independent simulation runs and the error bars show the SD. The relatively large fluctuations of interaction energy at a large penetration distance are mainly due to random translational and rotational movements of GQDs relative to the bilayer membrane and random configurational changes of individual lipids adjacent to the GQD. (C) Analytical model of corner piercing. Reprinted with permission from ref. 62. Copyright (2013) National Academy of Sciences.

acid-modified GQDs (GQD-FA) are all safe by *in vitro* and *in vivo* tests.⁴⁷ There was no significant difference in cell death after treatment with different concentrations of GQD, GQD-NH₂, GQD-COOH, and GOQD. Taken together, these results indicate that the concentration and redox state are more important factors for cell viability.⁴⁸ It was also identified that GQD, GQD-NH₂, GQD-COOH, and GOQD could be safely used for biomedical applications as drug carriers.^{49–52}

Several experiments exhibited the interaction of GQDs with proteins. Deng *et al.* found that nitrogen-doped graphene quantum dots (N-GQDs) could perturb a redox-sensitive system *via* the selective inhibition of antioxidant enzyme activities in zebrafish.⁵³ However, Jiang *et al.* discovered that the toxicity of N-GQDs was much lower than that of GO by using surface-enhanced infrared absorption spectroscopy. They found that the adsorption of GO could destroy the integrity of a membrane by extracting the lipid bilayer, and N-GQDs could only disturb the membrane of red blood cells.⁵⁴ The experiments showed that GQDs exhibit very low cytotoxicity owing to their ultra-small size and high oxygen content *in vitro* and *in vivo*.⁵⁵ Wang *et al.* found that the permeability and transport mechanism of GQDs across the biological barrier depend on the GQD size.⁵⁶ Kim *et al.* pointed out that GQDs could inhibit fibrillization of α -synuclein and trigger their disaggregation.⁵⁷ This was confirmed with the results of Yang's work.⁵⁸ Besides membranes and proteins, GQDs also exhibit high affinity to

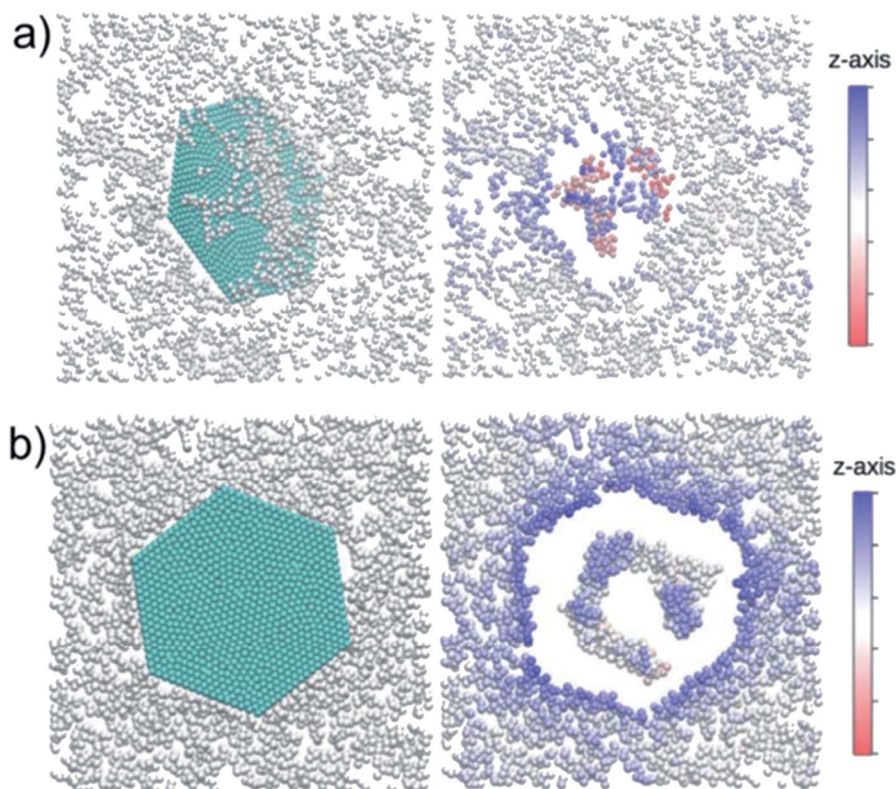


Fig. 4 Two views of the interaction of GQDs with the membrane are depicted corresponding to (a) piercing through and (b) adsorbing onto the membrane. The z-axis color code corresponds to the position of the phospholipid heads. The graphene flake locally affects the membrane structure. The empty spaces are occupied by the tails. Phospholipids are displaced with respect to the z-average position. Water molecules are not shown. Reprinted (adapted) with permission from ref. 63. Copyright (2015) American Chemical Society.



DNA molecules and a certain level of toxicity in DNA molecules in experiments.^{59,60} Moreover, although the viability of cells shows no significant decrease, the expression level of miR-21, miR-29a, Bax, Bcl2 and PTEN genes was affected greatly after treatment with large sized GQDs (50 nm).²³ In Yu's experiment, OH-GQDs, rGQDs, and NH₂-GQDs led to global DNA hypermethylation in zebrafish, and surface chemical groups of GQDs were critical for modulating DNA methylation.⁶¹

3. Computational studies

Computational studies on the interaction of GQDs and biomolecules including the lipid bilayer, proteins, and DNA molecules have been widely carried out. Li *et al.* found that the nearly orthogonal orientation of a sharp graphene corner with respect to the bilayer minimized the interactive free energy and is the thermodynamically preferred configuration even before penetration begins when using coarse-grained MD (CG-MD). They also found that pristine GQDs could enter into the lipid membrane spontaneously when they put the closed angle of GQDs in the membrane. To further explore the free energy barrier between GQDs and the membrane, the all-atomic molecular dynamics (AA-MD) simulation was used. As seen in Fig. 3B, the energy barrier of the closed angle of GQDs translocating through the membrane was approximately 5 k_BT based on their potential of mean force (PMF) calculation. After the insertion of the sharp corner of GQDs, the interaction between the hydrophobic tail of a lipid and hydrophobic GQD increased. Further studies revealed that the free energy barrier of translocation of GQDs through the membrane depends on four variables: h_H and h_T (thicknesses of the head and tail groups in the lipid monolayer, as shown in Fig. 3C) and γ_H and γ_T (interaction energy densities between one side surface of GQDs and head and tail groups of lipids relative to the concentration of GQDs in the solvent). The energy may be predicted using the following equation:

$$E = 2(1 - \gamma_H/\gamma_T)h_H^2 \times \gamma_H \times \tan \alpha \quad (1)$$

where 2α is an internal angle of GQDs. As predicted from this model, the energy increases with the increase of h when h is in the range from 0 to h_H . The energy decreases with the increase of h when h is larger than $h_H + 2h_T$. This model matches the data from the simulation well. It shows us that GQDs can insert themselves into a membrane by the sharp corner and can disturb the structure of the membrane.⁶²

Dallavalle *et al.* also found from CG-MD simulation that small GQDs (<5.2 nm) could enter into a membrane spontaneously. The large GQDs (>11.2 nm) mainly lay flat on the top of the bilayer, where they wreaked havoc on the membrane and created a patch of upturned phospholipids.⁶³ As seen in Fig. 4, the z coordination of phospholipid heads when the membrane is pierced by the GQD is more ordered than when the GQD is adsorbed onto the membrane. As shown in Table 1, the order parameter of the membrane is 0.77 with a 5.2 nm GQD piercing the membrane, and it is only 0.03 with a 5.2 nm GQD adsorbed onto the membrane. This large gap between order parameters

Table 1 Global versus local (dis-)order induced by GQDs piercing through or adhering to the membrane

Nanosheet size(nm)	GQD piercing through the membrane		GQD adhering to the membrane	
	S_{local}	S_{global}	S_{local}	S_{global}
0.9	0.72	0.69	—	—
2.7	0.72	0.69	—	—
5.2	0.77	0.68	0.03	0.66
8.1	0.34	0.65	-0.16	0.59
11.2	0.1	0.57	-0.16	0.52
13.3	—	—	-0.13	0.45

showed that GQDs adsorbed onto a lipid membrane can greatly disturb it. In addition, the order parameter of the membrane decreased with increasing GQD size. The parameter was 0.72 with the 0.9 nm GQD piercing the membrane, and it decreased to 0.1 with an 11.2 nm GQD. Thus, the larger GQD induced more local anti-alignment (S was negative for anti-alignment). This also indicated that a GQD adhering on the membrane could disrupt it much more than one piercing the membrane. A hydrophobic GQD adhered to the top of a membrane made the GQDs interact with the hydrophobic tail and the hydrophilic phospholipids of the layer directly under the capsized sheet. The anti-alignment was therefore truly related to the hydrophobic-hydrophobic interaction that allowed the sheet to adhere to the membrane. Importantly, the overturned phospholipids were able to impair cell function and disrupt the functioning of the membrane proteins. This may explain the cytotoxic activity of adhered GQDs, the so-called masking effect in experiments.⁶⁴⁻⁶⁷

The piercing behavior of small GQDs and the adhering behavior of large GQDs was also found in our research from AA-MD simulation.⁶⁸ By calculating the PMF of GQDs with different sizes including GQD7, GQD19, and GQD61, the translocation mechanism of GQDs through the lipid membrane was further interpreted. As seen in Fig. 5, the free energy of GQDs decreased as they moved from the water phase into the lipid membrane. This indicates that GQDs preferred to enter into the lipid membrane. The free energy differences between the lowest point in the membrane of GQDs to that of GQDs in the water phase of are $-56.3 \text{ kJ mol}^{-1}$, $-55.2 \text{ kJ mol}^{-1}$ and $-56.1 \text{ kJ mol}^{-1}$ for GQD7, GQD19 and GQD61. The positions corresponding to the lowest free energy in the z direction are -1.10 nm and 0.98 nm for GQD7 and GQD19. However, the position corresponding to lowest free energy in the z direction increased to 1.75 nm for GQD61. This implies that the location of the lowest free energy in the z direction moves farther from the middle of the lipid membrane as the GQD size increases from GQD19 to GQD61. The free energy barrier increased from 35.1 kJ mol^{-1} to 96.2 kJ mol^{-1} with the increase of GQD size from GQD19 to GQD61. The free energy barrier for GQDs passing through the lipid membrane was much higher with increased GQD size. The relatively small GQDs (GQD7 and GQD19) could pierce the membrane, and the large GQDs could stably adsorb on top of it.



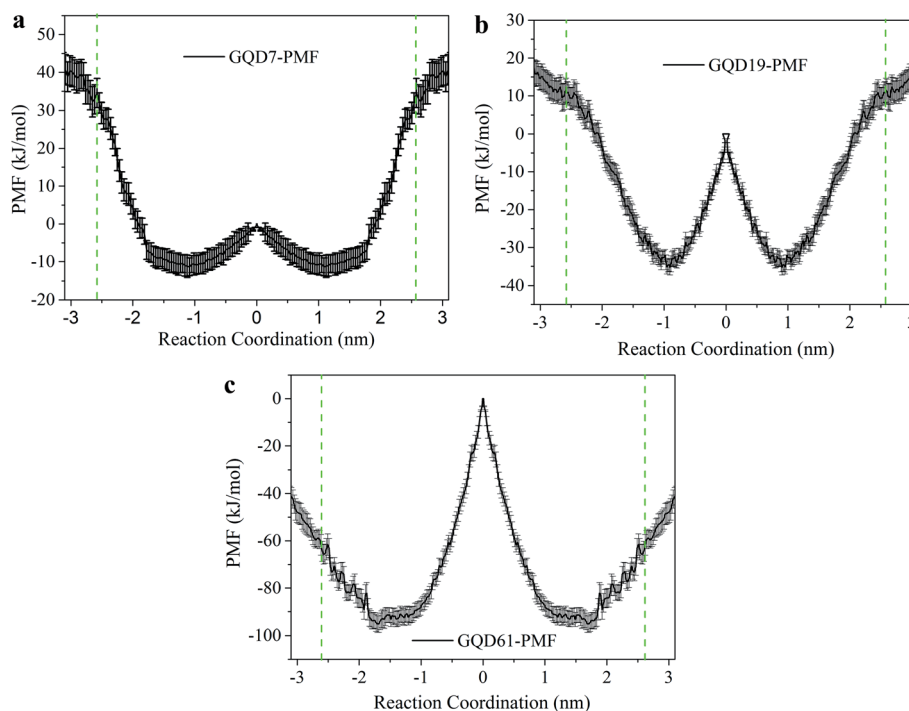


Fig. 5 Potential of mean force (PMF) of GQDs with different sizes translocating through the POPC membrane. (a) GQD7; (b) GQD19; (c) GQD61. Green dashed lines represent the location of two ends of the POPC membrane. Reprinted (adapted) with permission from ref. 68. Copyright (2016) American Chemical Society.

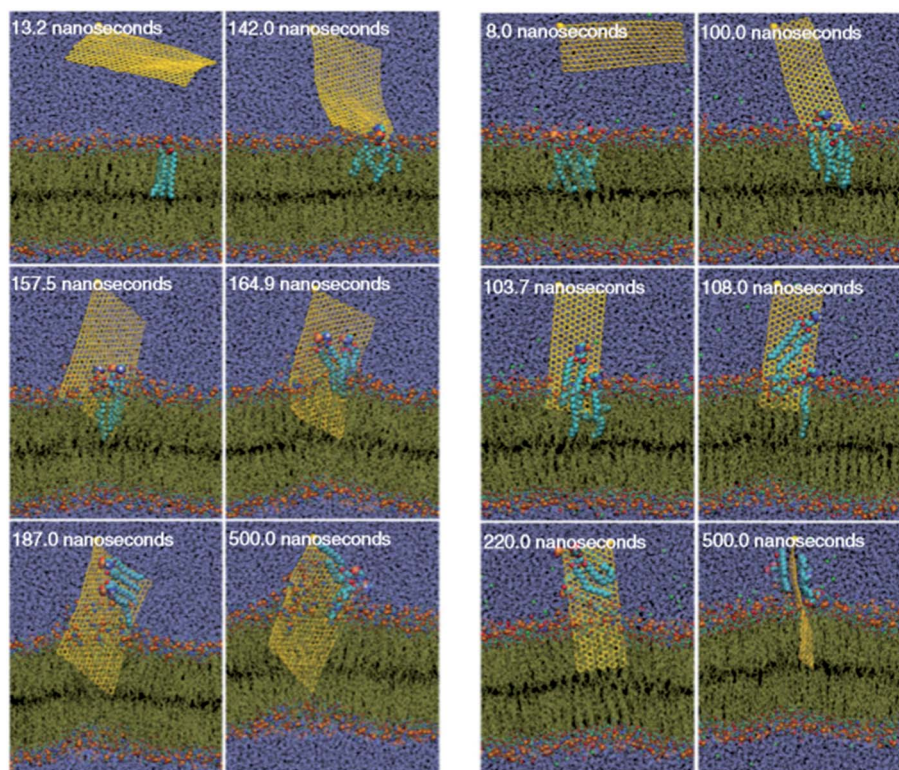


Fig. 6 Insertion of a GQD into the cell: (a) outer membrane and (b) inner membrane. Reprinted with permission from ref. 69, Copyright© 2013, Springer Nature.



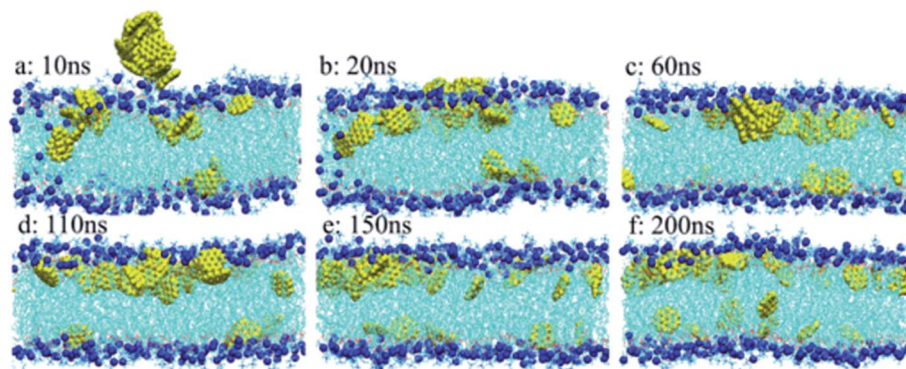


Fig. 7 Aggregated structures and distribution of GQD7-H3 during simulation: (a) 10 ns; (b) 20 ns; (c) 60 ns; (d) 110 ns; (e) 150 ns; and (f) 200 ns. GQDs are represented by the yellow vdW model, the POPC membrane is represented by the line model, and the P atoms in the membrane are represented by the blue vdW model. For clarity, water molecules are not shown. Reprinted (adapted) with permission from ref. 68. Copyright (2016) American Chemical Society.

Besides the piercing and adsorption models, Tu *et al.* further revealed that large GQDs could destroy the lipid bilayer by an extraction process.⁶⁹ As seen in Fig. 6, three distinguishable modes are observed during the process of the GQD spontaneously entering the outer and inner cell membranes. First, the swing mode, the initial stage in which the GQD underwent an unbiased orientation swing, swinging back and forth around the constrained atoms. This process lasted for a short period of time. Second, the insertion mode, in which the strong van der Waals force (vdW) between the GQD and lipid membrane caused the insertion of the GQD into the membrane. In the third mode (the extraction mode), the GQD started to extract phospholipid molecules on the cell membrane surface, destroying the cell membrane structure. This shows that GQDs can not only be inserted into a cell membrane by physical cutting, but can also destructively extract lipid molecules on the surface of the membrane through hydrophobic interaction. GQDs can interact strongly with the phospholipid molecules on the bacterial cell membrane, which causes a large number of phospholipid molecules to break away from the membrane and adsorb to the graphene surface.

Besides the effect of GQD size, the effect of GQD concentration on the membrane was also investigated. As seen in Fig. 7a, multiple small GQDs aggregated into clusters in the aqueous solution before translocation into the lipid membrane during the first 10 ns. This is similar to the permeation of fullerene into the POPC membrane.^{34,70} After this, the aggregates permeated into the POPC membrane at 20 ns, as shown in Fig. 7b. In Fig. 7c, most of the GQDs were still aggregated in one large cluster. However, as opposed to the aggregates at 10 ns, most GQDs in the aggregates tend to be parallel to the POPC lipid at 110 ns. The aggregate of GQD7 tends to dissociate and disperse to the position of energy minimum on both sides of the POPC membrane. The hydrophobic GQDs tend to adsorb on the tail of POPC lipids in the membrane, and GQD7 tends to be parallel to the main chain of POPC molecules. The simulation revealed that GQD7 tends to permeate into the membrane as aggregates but disperses in the membrane after permeation. This may cause disruption of the membrane (pore formation) in the piercing process of GQD7 under high concentration, but not

after permeation. It could also explain why small GQDs do not display any cytotoxicity at low concentrations but have high cytotoxicity at high concentrations.

In addition, simulation helps us to predict the average time of entry of nanoparticles in lipid membranes, using a combination of molecular dynamics simulations and statistical approaches.⁷¹ From the results of the theoretical model, researchers have found that different sizes and shapes of GQD molecules can affect the translocation time greatly, and this data was confirmed by experiments. Yang *et al.* found through all-atom molecular dynamics simulation (AAMD) that the dynamics of GQD translocation through a membrane could be modulated by the charge on the GQDs.⁷² Using coarse-grained MD (CGMD), as seen in Fig. 8, GQDs could be incorporated with the membrane in a sandwiched structure.⁷³ Fig. 8a illustrates how the micelle including GQDs and lipids merged. As the micelle slowly merges with the top lipid layer, some lipids become trapped below the attached micelle (Fig. 8b). At $t = 120$ ns, a neck-like protrusion formed on the bottom part of the bilayer (Fig. 8c). This initiated fusion of GQDs into the membrane at $t = 360$ ns, as shown in Fig. 8d. The process continued until the GQD was stabilized in the center of the bilayer membrane at $t = 516$ ns (Fig. 8f). The results were confirmed by both Brownian dynamics simulations and by experiments.⁷⁴

It is estimated that there are more than 1 million protein types in the human proteome. The interaction between materials and biomolecules (such as DNA and proteins) can interfere with biological functions and lead to cytotoxicity.⁷⁵ Graphene can break the structure of polypeptides, protein fragments, and proteins.^{76,77} Evidently GQDs can also interact with proteins and other biological macromolecules. Fang *et al.* revealed the interaction mechanism of GQDs and the ubiquitin protein by both experiments and MD simulations.⁷⁸ Our research group⁷⁹ used AAMD to reveal the effect of GQD size on the structure of the protein villin headpiece (HP35). With increased GQD size, more protein residues could adsorb on the GQDs and the adsorption capacity of protein on GQDs increased. Meanwhile, the damage to the secondary structure of proteins rose along with GQD size. Moreover, a remarkable capacity of GQDs in



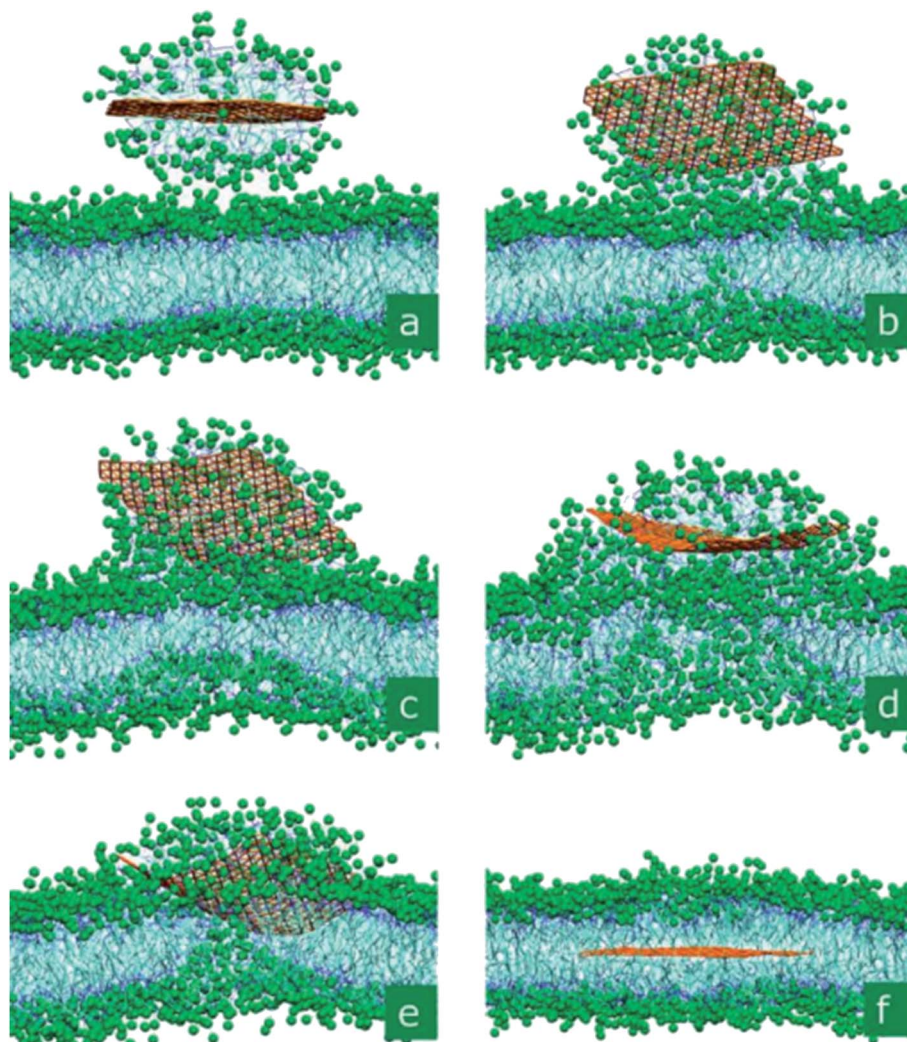


Fig. 8 Self-insertion of QGDs inside a phospholipid membrane. In the displayed process, a QGD micelle merges with a membrane and releases the monolayer, which penetrates into the membrane. The snapshots were taken at $t_{a-f} = 2.9, 52.4, 120.0, 299.2, 356.4,$ and 516.4 ns. Reprinted (adapted) with permission from ref. 73. Copyright (2010) American Chemical Society.

regulating aberrant protein expression was revealed through H-bonding and hydrophobic interactions⁸⁰ from discrete MD. QGDs were able to exhibit potential toxicity by disrupting the protein–protein interaction *via* the hydrophobic interaction.⁸¹ Fig. 9b–e illustrate the insertion process of the QGD into the protein–protein interaction from the trajectory defined as sim-1. Initially parallel and close to the dimer's interface, the QGD rotated and came into contact with one monomer after 2 ns (Fig. 9b). Because the graphene surface is hydrophobic, typically a protein molecule with hydrophobic residues on its surface can be adsorbed on the QGD. From 30 ns to 40 ns, the protein monomer began to rotate around its barrel axis, due to the preferred hydrophobic interaction between the QGD and non-polar residues at the dimer interface. After this, the QGD was fully inserted into the dimer and the dimer was separated. Other simulations confirmed that this insertion process was representative. In addition, the amyloid fibrils could be destroyed by penetration and extraction of the peptides by

QGDs due to the strong interaction between peptides and QGDs.^{82,83} In addition to lipid membranes and proteins, interaction with DNA molecules is also an essential topic for the evaluation of QGD toxicity. Pristine QGDs can interact with DNA molecules strongly.⁸⁴ As QGD oxidation increases, the interaction between DNA and QGDs decreases. For clarity, the simulations of QGD with biomolecules were summarized in Table 2.

4. Cross point between experiments and simulations

We have discussed the agreement among the experiments and theoretical work on the interaction of QGDs with biomolecules including lipids, proteins, and DNA molecules. In terms of the effect of QGD size on lipid membranes, the results from the simulations and the experiments correspond well for the lipid bilayer. Small QGDs can only penetrate the membrane, and simulation also gave us the atomic details of the penetration



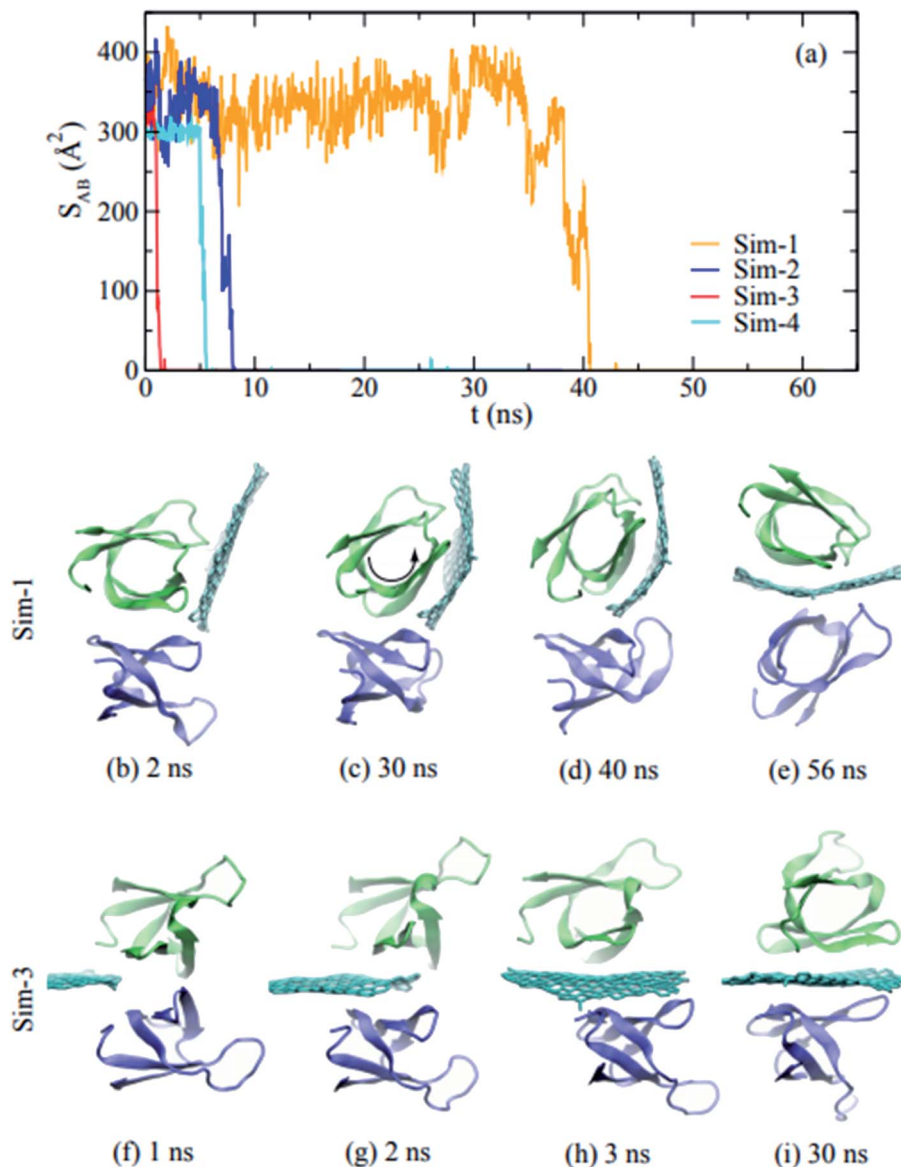


Fig. 9 (a) Time-dependent contact areas of the dimer during the insertion of a graphene sheet; (b)–(i) snapshots of the insertion process of a graphene sheet into the dimer. Reprinted (adapted) with permission from ref. 81. Copyright (2015) American Chemical Society.

process. Large GQDs were seen to wreak havoc on the lipid bilayer, in both simulations and experiments. Simulation allowed observation of the extraction process of lipids from the membrane by GQDs.

In terms of GQD concentration, the toxicity of GQDs increases along with concentration. With higher concentrations of GQDs, the GQDs aggregate into a larger cluster and can form pores in the lipid membrane. However, the time scale for the translocation of GQDs through the membrane increases exponentially with GQD size. Limited computational resources have prevented investigation of the translocation mechanism of large GQDs by AA-MD and CG-MD. However, a simplified model such as DPD could be used to describe the translocation mechanism of large GQDs.

Regarding interactions between GQDs and protein and DNA molecules, strong adsorption between GQDs and biomolecules

was observed in both experiments and simulations. Due to the strong hydrophobic interaction between GQDs and proteins, proteins can adsorb on GQDs strongly, and the conformation of proteins on the GQDs changes greatly. In particular, GQDs can insert themselves into the interior of proteins and disrupt protein–protein interactions. Due to the strong π – π interaction between GQDs and DNA molecules, GQDs can also change the structure of DNA molecules.

5. Summary of the intersection and outlook

In summary, the aim of this review was to bring together results from computer simulations and computational models relating to experiments on the cytotoxicity of GQDs. Simulations and experiments also assessed the interaction of lipids and



Table 2 Simulations of GQDs with biomolecules

Materials	Simulation Methods	Force field	Time scale	Results	References
GQDs	MD	GROMOS54A7	Hundreds of ns	GQDs tend to penetrate into the membrane perpendicularly	Vatanparast <i>et al.</i> ⁸⁵
GQDs	MD	CHARMM27	Hundreds of ns	Graphene quantum dots (GQDs) act as a potent inhibitor against the <i>in vivo</i> aggregation and toxicity of human islet amyloid polypeptide (IAPP)	Wang <i>et al.</i> ⁸⁶
GQDs	MD	AMBER99	Hundreds of ns	GQDs interact with human peroxidases and are degraded by them	Martín <i>et al.</i> ⁸⁷
GQDs	MD	CHARMM36	Hundreds of ns	The translocation of GQDs could be modulated by charge on the GQDs.	Tang <i>et al.</i> , ⁷²
GQD (sheet)	CGMD	—	Hundreds of ns	Graphene sheets could be hosted in the hydrophobic interior of biological membranes formed by amphiphilic phospholipid molecules	Titov <i>et al.</i> ⁷³
GQDs	MD	CHARMM27	Hundred of ns	The time of entry of nanoparticles in lipid membranes	Liu <i>et al.</i> ⁷¹
GQDs	MD	CHARMM27	Hundreds of ns	Small GQDs can translocate through membranes <i>via</i> drugs.	Xue <i>et al.</i> ⁸⁸
Graphene	Brownian MD	—	~hundreds of μ s	GQDs incorporated with membranes as a sandwiched graphene–cell membrane superstructure	Chen <i>et al.</i> ⁷⁴
GQDs GOQDs	MD	Universal force field	Dozens of ns	GQDs interact more strongly with DNA than GOQDs.	Wang <i>et al.</i> ⁸⁴
GQDs	MD	CHARMM27	Dozens of ns	The chirality of D-GQDs provides a stronger tendency to accumulate within the cellular membrane than that of L-GQDs.	Suzuki <i>et al.</i> ⁸⁹
GQDs	Discrete MD	CHARMM27	—	A remarkable capacity of GQDs in regulating aberrant protein expression through H-bonding and hydrophobic interactions	Faridi <i>et al.</i> ⁸⁰
GQDs	MD	CHARMM27	~Hundreds of ns	GQDs can destroy the lipid membrane by extracting the lipid molecules	Tu <i>et al.</i> ⁶⁹
GQDs	MD	CHARMM27	Hundreds of ns	GQDs can disrupt protein–protein interaction	Luan <i>et al.</i> ⁸¹
GQDs	MD	CHARMM27	1 μ s	Destruction of amyloid fibrils by graphene through penetration and extraction of peptides	Yang <i>et al.</i> ⁸²
GQDs	MD	CHARMM27	Hundreds of ns	Small GQDs cannot disturb membranes.	Liang <i>et al.</i> ⁶⁸
GQDs	MD	CHARMM27	Hundreds of ns	Small GQDs can penetrate membranes	Xue <i>et al.</i> ⁸⁸
GQDs	MD	CHARMM27	Hundreds of ns	Secondary structure of proteins can be destroyed by GQDs	Zhou <i>et al.</i> ⁷⁹
GQDs	MD	CHARMM27	Hundreds of ns	DNA fragment	Kong <i>et al.</i> ⁹⁰

functionalized GQDs. However, the results from these two approaches cannot be compared directly. The first reason is the relative diversity of functionalized GQDs in experiments compared to simulations. Also, the scale of GQDs in the experiments was larger than that in the simulations. Simulation with simplified functionalized GQDs could shed light on the effect of particular functional groups on the translocation of GQDs. To enhance the potential applications of GQDs in biomedicine, surface modification and functionalization are important methods to improve their properties. However, due to the diversity and complexity of surface modifications of GQDs, a systematic evaluation of the effect of surface modification of GQDs on their toxicity and function is still lacking both in experiments and in theoretical simulation. In

particular, the simulation of the biosafety effect of polymer modified GQD is still an interesting and challenging task.

Thus, the first task is to further elucidate the interaction of functionalized GQDs and biomolecules including lipids, proteins, and DNA molecules based on simulation and experiments. Several challenges also remain to be overcome in the research process itself. One is the scale gap between theoretical work and experiments. In almost all theoretical simulations, the size of GQDs is less than 10 nm, and larger GQDs (10–100 nm) should be investigated with simulation in order to allow comparison with experiments performed with larger GQDs. Another challenge is the time scale; the time scale of GQD cytotoxicity in experiments is based on hours or even days and years, but in simulation the time scale is measured in ns or μ s,



a tiny fraction in comparison. We recommend further development of accelerated MD simulation to resolve this mismatch. Finally, the force fields generally used in MD simulation are not sufficiently accurate. Most simulations cited in this review used the AMBER and CHARMM force fields to describe the potential energy of biomolecules. The performance of these two force fields is considered superior for describing biomolecules; however, we believe that more accurate force fields need to be developed in order to achieve a more comprehensive understanding of GQD cytotoxicity. Ideally, these force fields would precisely describe the cross-interaction between biomolecules and inorganic materials and consider the polarization effect.

In recent years, our group has attempted to answer open questions remaining in this field and provide a perspective on the future of GQD cytotoxicity research. With advances in nanotechnology and theoretical modeling, we anticipate that a nanoscale biosafety evaluation based on GQDs or other nanomaterials could be developed in the future. The studies we have summarized here represent the first steps toward this pressing goal.

Conflicts of interest

There are no conflicts of interest to declare.

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