


Cite this: *Nanoscale Adv.*, 2020, 2, 3244

A high-performance and flexible thermoelectric generator based on the solution-processed composites of reduced graphene oxide nanosheets and bismuth telluride nanoplates†

Defang Ding,[‡]^a Fengming Sun,[‡]^a Fan Xia ^{*a} and Zhiyong Tang ^{bc}

The fabrication of a flexible thermoelectric generator (TEG) with both high power output and good flexibility has drawn considerable attention. Solution-processed inorganic nanocrystals have good processability in interface to retain excellent electrical properties of nanocrystals and can be processed into thin films on a flexible substrate by an easy scale-up printing or coating method. However, a high-performance TEG device based on inorganic solution-processed materials also poses challenges when it comes to flexibility of the whole device. Herein, flexible planar TEG devices are fabricated by printing an ink mixture comprising solution-processed bismuth telluride (Bi_2Te_3) nanoplates with reduced-graphene oxide (rGO) nanosheets onto flexible polyimide substrates. The interface treatment by hot ethylenediamine and the appropriate amount of rGO contribute to the high electrical properties of the material. Also, when rGO nanosheets of 1% mass ratio are added, the optimum power output of the corresponding rGO/ Bi_2Te_3 TEG device with six elements reaches $\sim 1.72 \mu\text{W}$ at a temperature difference of 20 K. Moreover, owing to the contribution from flexible rGO nanosheets, the suitable thickness of each element, and the artful connection of elements with a soft copper wire in the devices, the 1% rGO/ Bi_2Te_3 TEG device was found to be robust, and its electrical resistance merely changes by 2% after bending 1000 cycles on 5 mm in bending. These inorganic-based TEGs with both high performance and good flexibility will promote the development of new generation energy devices in the field of flexible electronics.

Received 12th February 2020
Accepted 8th June 2020

DOI: 10.1039/d0na00118j

rsc.li/nanoscale-advances

Introduction

Along with the rapid development of flexible electronics, particular attention has been paid to the flexible thin-film-based thermoelectric generator (TEG) devices because of their intriguing self-powering merit with no need of additional power managing components.^{1–4} An emerging method has been developed based on the solution-processed nanostructures,^{5–8}

which are easily processed into thin films on the flexible substrates by low-temperature, low-cost, convenient and efficient means such as coating or printing.^{9–12} Moreover, bulk binary chalcogenides such as bismuth telluride (Bi_2Te_3), antimony telluride (Sb_2Te_3), and their alloys, which are the preferred choice for most flexible thermoelectric applications due to their excellent thermoelectric materials near room temperature,^{13–16} are usually difficult to be integrated into flexible devices because of the poor processability.^{17–19} In this case, the coating or printing method based on solution-processed Bi_2Te_3 or Sb_2Te_3 nanocrystals should also be adopted. Also, it has been widely reported that these solution-processed films could retain the excellent electrical properties in high-performance electronic and optoelectronic devices on flexible substrates.^{20–23} Solution-processed inorganic nanocrystals have good processability in the interface to retain the excellent electrical properties of the nanocrystals. In our previous study about the interface engineering in solution-processed nanocrystal thermoelectric thin films, we also demonstrated that the hot ethylenediamine can effectively strip the insulating organic ligands from the surface of the nanocrystals that hinder the charge transfer.²⁴ Moreover, it is a facile and mild wet-chemical

^aEngineering Research Center of Nano-Geomaterials of Ministry of Education, Faculty of Materials Science and Chemistry, China University of Geosciences (CUG), 388 Lumo Road, Wuhan 430074, P. R. China. E-mail: xiafan@cug.edu.cn; xiafan@hust.edu.cn

^bCAS Key Laboratory of Nanosystem and Hierarchical Fabrication, CAS Center for Excellence in Nanoscience, National Center for Nanoscience and Technology, No. 11, Beiyitiao, Zhongguancun, Beijing, 100190, P. R. China

^cUniversity of Chinese Academy of Sciences, 19A Yuquan Rd, Shijingshan District, Beijing, 100049 P. R. China

† Electronic supplementary information (ESI) available: Experimental detail, some supporting characterizations about the rGO and devices, two tables about Hall testing results and state-of-art excellent examples of printed planar TEG devices, and a video that displays the real time measurement of open circuit voltage ($\Delta T = 20$ K) of the thermoelectric generator device made from pure Bi_2Te_3 nanoplates. See DOI: 10.1039/d0na00118j

‡ These authors contributed equally.

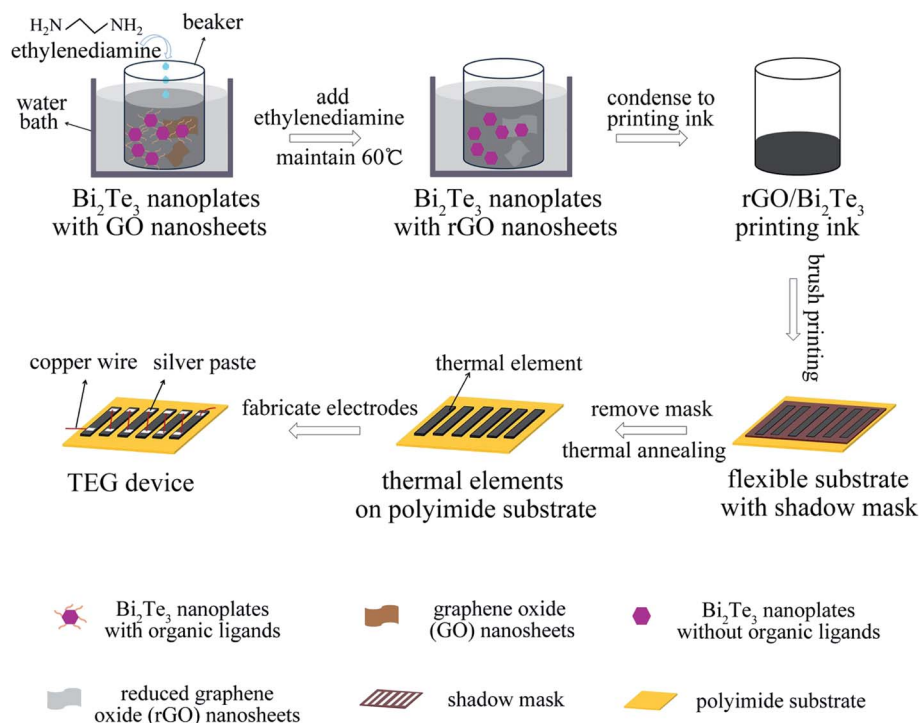


method. Therefore, coating or printing based on Bi_2Te_3 or Sb_2Te_3 solution-processed nanocrystals is a compatible method to prepare flexible thin-film-based TEG devices.

The power output of TEG devices depends on multiple parameters including the Seebeck coefficient (S) of materials, electrical conductivity (σ) of materials, thermal conductivity (κ) of materials, number of thermal elements (m), the temperature difference across the device (ΔT), and internal resistance of the device (R_i). In order to achieve a considerable power output, the thermoelectric properties of the materials including S , σ and κ must be mutually coordinated since high figure-of-merit ($ZT = S^2\sigma T/\kappa$,^{25–27} T is the absolute temperature) of material is crucial for power output.^{1–4} In addition, R_i of TEG devices must be low. However, low R_i and good flexibility are usually difficult to be simultaneously achieved in one flexible TEG device. First, for example, organic semiconductors are suggested to be used in flexible thin-film-based TEG devices due to their intrinsic flexibility,²⁸ but they encounter several limitations, *e.g.* rather low electrical conductivity and high contact resistance with the metal electrodes, leading to high R_i .²⁹ On the contrary, the inorganic thermoelectric materials are able to resolve those problems and ultimately higher power output is expected, but the inorganic-based materials are usually rigid.^{29,30} Second, the rational design on the geometry of the thermal element is a prerequisite for decreasing R_i of the TEG device. For example, thermal elements should be characteristic of short length and large cross-section area for low R_i , but it is difficult to maintain

ΔT when the length of the elements is extremely short. A large cross-section area demands high thickness and width of the element. However, increasing the width of the element would decrease the number of elements per unit area, giving rise to reduced power output. Also, extremely increasing the thickness of the element would increase the cracking risk during bending, leading to poor flexibility. Thereby, there must be a trade-off between the geometry of thermal elements and power output.³¹ In addition to thermal elements, the flexibility of a TEG device also is subjected to its connection of thermal elements, particularly the materials of connection, which are actually crucial for the whole device.^{32,33} Even if the thermal elements possess good flexibility, the whole TEG device with rigid connection materials could not tolerate cycle bending. However, not enough attention is paid to this issue. Altogether, in order to acquire flexible thin-film based TEG devices with high performance, a systematic design principle needs to be established, which includes a judicious selection of active materials, precise structural control on the thermal elements, appropriate design of connection manner, and convenient and efficient preparation of thin films.

Herein, we demonstrate the facile fabrication of the flexible planar TEG devices containing six elements by printing the solution-processed two-dimensional (2D) Bi_2Te_3 nanoplates with reduced-graphene oxide (rGO) nanosheets in different mass ratio onto the polyimide substrates with shadow masks (Scheme 1). The hot ethylenediamine was employed to remove



Scheme 1 Schematic of the fabrication of flexible six-element planar TEG devices by printing. Bi_2Te_3 nanoplates with GO nanosheets were treated by hot ethylenediamine to remove the organic ligands of Bi_2Te_3 , and the GO transformed to rGO with the hot ethylenediamine. Subsequently, rGO/ Bi_2Te_3 was slowly condensed to inks for printing on polyimide substrates with the shadow mask. After removing the shadow mask and thermal annealing process, the copper wires connected to the thermal elements by silver paste were used as electrodes in the six-element devices.



the organic ligands of Bi_2Te_3 nanoplates,^{24,34} where the GOs transformed into rGOs due to the reduction from hot ethylenediamine.³⁴ Also, the thermal elements are connected by the soft copper wires with silver paste in these six-element devices. Impressively, the 1% rGO/ Bi_2Te_3 thin film achieves the ZT of 0.4 at 400 K. The optimum power output (P_{max}) of the 1% rGO/ Bi_2Te_3 TEG devices, with six elements of the size of $10 \text{ mm} \times 3 \text{ mm} \times 40 \text{ }\mu\text{m}$, reaches $\sim 1.72 \text{ }\mu\text{W}$ at $\Delta T = 20 \text{ K}$ and $\sim 15.22 \text{ }\mu\text{W}$ at $\Delta T = 60 \text{ K}$. More importantly, this 1% rGO/ Bi_2Te_3 TEG device is robust upon repeated flexing to bending radius of down to 5 mm, highlighting the application potential of the solution-processed nanomaterials in flexible electronics.

Experimental

Materials

Sodium tellurite (NaTeO_3 , >99.99%), bismuth nitrate pentahydrate ($\text{Bi}(\text{NO}_3)_3 \cdot 5\text{H}_2\text{O}$, >90%), sodium hydroxide (NaOH , >90%), poly(vinyl pyrrolidone) (PVP, $\text{MW} \approx 40\,000$) and ethylene glycol (EG) were all purchased from Alfa Aesar. Ethylenediamine ($\text{C}_2\text{H}_8\text{N}_2$) (99%) was obtained from Acros. Graphite powder (325 mesh), potassium permanganate (KMnO_4 , >90%), concentrated sulfuric acid (H_2SO_4 , 98%), hydrogen peroxide (H_2O_2 , 30%), hydrochloric acid (HCl , 37%), anhydrous acetone and isopropanol were supplied by Sinopharm. All the chemicals were used as received without further purification.

Synthesis of bismuth telluride (Bi_2Te_3) nanoplates

Bi_2Te_3 nanoplates were synthesized according to the literature.¹⁹ Typically, 0.194 g $\text{Bi}(\text{NO}_3)_3 \cdot 5\text{H}_2\text{O}$ (0.4 mmol), 0.133 g NaTeO_3 (0.6 mmol), 0.32 g NaOH (8 mmol) and 0.4446 g PVP (4 mmol) were dissolved in 20 mL ethylene glycol in a 50 mL three-neck flask equipped with a thermocouple and reflux condenser. The mixture was stirred and heated to $60 \text{ }^\circ\text{C}$ on a heating mantle until the solution became transparent. Then the temperature was raised to $190 \text{ }^\circ\text{C}$ in 30 min, and the solution immediately turned black. After holding the temperature for 3 h, the heating mantle was removed and the mixture was allowed to cool down to room temperature naturally. The mixture was separated into two 50 mL centrifuge tubes, and 20 mL isopropanol and 10 mL acetone were added into both of the tubes, and then centrifuged them at 10 000 rpm for 8 min. The supernatant was discarded, while the solid was dispersed in fresh 40 mL isopropanol and then centrifuged them at 8000 rpm for 5 min. The washing steps were repeated with isopropanol for two more times. The final products were collected and dispersed in isopropanol for further use.

Synthesis of the composites of graphene oxide (GO) nanosheets and Bi_2Te_3 nanoplates

GO was synthesized by the Hummers' method.³⁵ The as-synthesized GO was freeze-dried for further use. A certain mass ratio of GO powder was ultrasonically dispersed in ethylene glycol for 1 h. And then, the solution of GO in ethylene glycol was used as a precursor being added into the above synthesis process of Bi_2Te_3 nanoplates. For example, 1% mass ratio of

dried GO ($\sim 2.5 \text{ mg}$), 0.194 g $\text{Bi}(\text{NO}_3)_3 \cdot 5\text{H}_2\text{O}$ (0.4 mmol), 0.133 g NaTeO_3 (0.6 mmol), 0.32 g NaOH (8 mmol) and 0.4446 g PVP (4 mmol) were dissolved in 20 mL ethylene glycol, and the following synthesis steps were similar to the above process.

Organic surfactant removing

Before printing or coating on the substrate, PVP polymer on Bi_2Te_3 nanoplate surfaces (or in GO/ Bi_2Te_3 composites), which would hinder the charge carrier transport, was stripped. Typically, $\sim 0.4 \text{ mmol}$ nanoplates were dispersed in 15 mL solution of ethylenediamine in isopropanol (1 M), and then the mixture was stirred at $60 \text{ }^\circ\text{C}$ for 12 h. After that, the products were centrifuged at 8000 rpm for 5 min. The supernatant was discarded, while the solid was dispersed in fresh 40 mL isopropanol, the washing steps were repeated with isopropanol for two more times. The final products were dispersed in isopropanol. Here it needs to be pointed out that GO turned to reduced-GO (rGO) during the treatment with hot ethylenediamine, as shown in Raman spectroscopy results (Fig. S3[†]). Therefore, the composites were mixtures of rGO nanosheets and Bi_2Te_3 nanoplates.

Fabrication of flexible six-element planar TEG device and testing

To print the solution on the polyimide substrate to obtain flexible planar TEG devices, the solution of Bi_2Te_3 nanoplates or rGO/ Bi_2Te_3 composites after the PVP removal was slowly condensed to ink in a vacuum ($\sim 4 \text{ mmol}$ nanoplates in 1 mL isopropanol). Subsequently, the ink was treated in an ultrasonic bath to keep the high dispersion of nanoplates. Then, the TE inks were printed on polyimide substrates with a shadow mask to form six $10 \text{ mm} \times 3 \text{ mm} \times 40 \text{ }\mu\text{m}$ thermal elements using write printing, followed by annealing at $380 \text{ }^\circ\text{C}$ for 1 h to form solid thick films. Each thermal element was connected in series with a copper wire ($50 \text{ }\mu\text{m}$ in diameter) using conductive silver paste, as shown in Scheme S1.[†] In order to reduce the contact resistance, the TEG devices with silver paste were cured at $60 \text{ }^\circ\text{C}$ for 4 h in a vacuum oven.

Two thermocouples of $50 \text{ }\mu\text{m}$ in diameter were pasted on the hot and cold sides of the device to measure the temperature difference. The temperature difference across the TEG devices was created by heating the hot-side while maintaining the cold-side temperature, held by a temperature controller.

Results and discussion

Morphology and structure characterization of the rGO/ Bi_2Te_3 composite films

The morphologies of the rGO/ Bi_2Te_3 composite films, *i.e.* the thermal element, from the six-element planar TEG devices are discerned by SEM (Fig. 1). From the surface SEM images, it is clear that with the increasing percentage of rGO from 0 to 2%, the rGO nanosheets cover the Bi_2Te_3 nanoplates like "gossamer", and the rGO is uniformly distributed in the Bi_2Te_3 nanoplate films. The thickness of all the films is controlled by about $40 \text{ }\mu\text{m}$. In theory, the thicker the films, the larger the



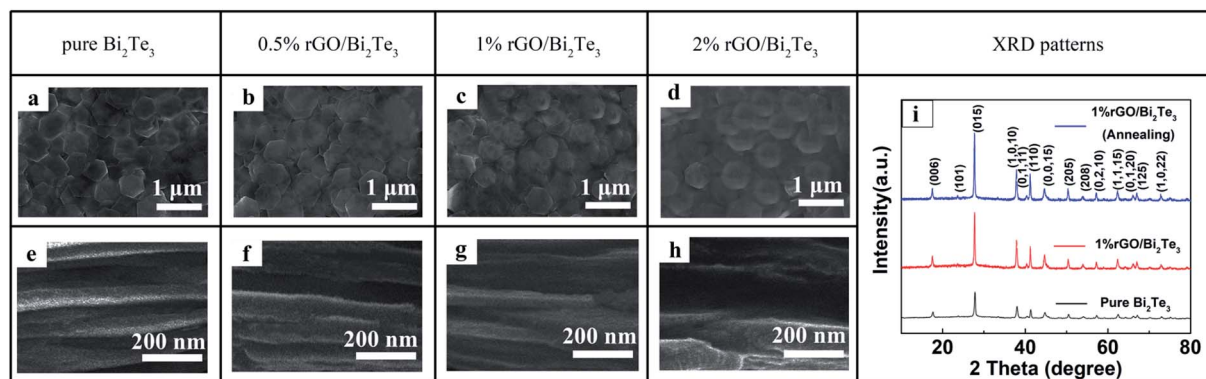


Fig. 1 Surface and cross-sectional SEM images of Bi_2Te_3 nanoplate films with different mass ratio of rGO nanosheets: (a) and (e) pure Bi_2Te_3 , (b) and (f) 0.5% $\text{rGO}/\text{Bi}_2\text{Te}_3$, (c) and (g) 1% $\text{rGO}/\text{Bi}_2\text{Te}_3$, and (d) and (h) 2% $\text{rGO}/\text{Bi}_2\text{Te}_3$; (i) XRD patterns of the pure Bi_2Te_3 and 1% $\text{rGO}/\text{Bi}_2\text{Te}_3$ composites films before and after annealing.

cross-section area of the films, which can efficiently decrease the internal resistance of the device. However, the thickness of the films should coordinate with the flexibility of the device, after all, excessively thick films crack easily during bending. Hence, we controlled the films (elements) to a moderate thickness, $\sim 40 \mu\text{m}$. The XRD patterns of the pure Bi_2Te_3 and 1% $\text{rGO}/\text{Bi}_2\text{Te}_3$ composites films before and after annealing are shown in Fig. 1i. The three XRD plots show similar patterns. The peaks can be indexed to PDF no. 08-0021 of Bi_2Te_3 . Moreover, there is no characteristic peak of rGO in the pattern of 1% $\text{rGO}/\text{Bi}_2\text{Te}_3$ film. The rGO's peaks should be hindered by the strong peaks of

Bi_2Te_3 due to the low weight percentage of rGO in the composite films. There is no obvious difference between 1% $\text{rGO}/\text{Bi}_2\text{Te}_3$ composites films before and after annealing.

Thermoelectric properties of the $\text{rGO}/\text{Bi}_2\text{Te}_3$ composite films

Fig. 2 summarizes the temperature-dependent thermoelectric properties of the films made from Bi_2Te_3 nanoplates or $\text{rGO}/\text{Bi}_2\text{Te}_3$ composites. It is seen that all the films possess negative S (Fig. 2a), manifesting that films are all n-type in nature. With the increasing percentage of rGO, the absolute value of S of the composite films is reduced, suggesting that the introduction of rGO is harmful to S of n-type Bi_2Te_3 nanoplate films. Since pure rGO is known to be p-type,^{36,37} and its positive S was experimentally confirmed by us, as shown in Fig. S5a.† Therefore, the addition of rGO would counteract the whole S of the composite films due to the opposite sign of S .³⁸ Fig. 2b displays that σ of $\text{rGO}/\text{Bi}_2\text{Te}_3$ composite films are significantly enhanced with the addition of rGO. The electron mobility (μ) is largely improved from $59.08 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ to $138.79 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ by adding rGO, whereas the change in the electron concentration (n) is small (Table 1). So, the improved μ should be responsible for the enhanced σ according to the approximate relationship $\sigma = en\mu$,³⁹ where e is the charge per carrier. rGO in the composite likely provides extra transfer channels for electrons,^{40,41} which results in the increased μ . Meanwhile, n changes little with the addition of rGO, which means that the n plays little role in the S . The $S^2\sigma$ of all the films are then calculated (Fig. 2c). Notably, the

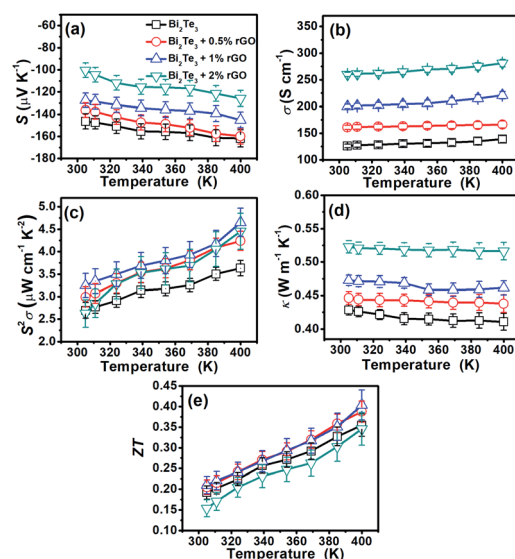


Fig. 2 Temperature dependent (a) Seebeck coefficient (S), (b) electrical conductivity (σ), (c) power factor ($S^2\sigma$), (d) thermal conductivity (κ) and (e) figure-of-merit (ZT) of Bi_2Te_3 nanoplate films with different mass ratio of rGO nanosheets. The legends in (a) are applied to all the figures. Error bars are estimated from repeatability of the experimental results: for each material-making condition, three samples are prepared and measured. With the increasing percentage of rGO, the S of the $\text{rGO}/\text{Bi}_2\text{Te}_3$ composite films is reduced but the σ and κ are increased. Finally, the 1% $\text{rGO}/\text{Bi}_2\text{Te}_3$ composite films exhibit the best ZT value calculated by the three factors ($ZT = S^2\sigma T/\kappa$).

Table 1 Carrier concentration (n), Hall mobility (μ_H), and electrical conductivity (σ) of as-prepared $\text{rGO}/\text{Bi}_2\text{Te}_3$ composite films and rGO film at 300 K

Samples	n (10^{19} cm^{-3})	μ_H ($\text{cm}^2 \text{ V}^{-1} \text{ s}^{-1}$)	σ (S cm^{-1})
Bi_2Te_3	1.33	59.08	125.7
0.5% $\text{rGO}/\text{Bi}_2\text{Te}_3$	1.26	79.79	160.8
1% $\text{rGO}/\text{Bi}_2\text{Te}_3$	1.21	103.27	199.9
2% $\text{rGO}/\text{Bi}_2\text{Te}_3$	1.17	138.79	259.8
rGO film	3.59	48.25	277.3



1% rGO/Bi₂Te₃ composite films exhibit the best performance of 3.2 $\mu\text{W cm}^{-2}$ at 300 K and 4.6 $\mu\text{W cm}^{-2}$ at 400 K due to the balance between S and σ . The thermal conductivity (κ) of the rGO/Bi₂Te₃ composites is increased with the increase in the percentage of rGO (Fig. 2d). We also tested the κ of pure rGO films (Fig. S5d†). It is about 14.3 $\text{W m}^{-1} \text{K}^{-1}$, which is much higher than that of the Bi₂Te₃ nanoplate film ($\sim 0.42 \text{ W m}^{-1} \text{K}^{-1}$). The rGO nanosheets will facilitate the heat conduction,⁴² thus increasing the κ of the rGO/Bi₂Te₃ composites. After the integration of S , σ and κ into the ZT value, the 1% rGO/Bi₂Te₃ composite films exhibit the best ZT of 0.2 at 300 K and 0.4 at 400 K as shown in Fig. 2e.

A transport mechanism in the rGO/Bi₂Te₃ composite films is proposed as follows. There are tremendous interfaces between rGO and Bi₂Te₃. Moreover, the energy barrier at the interfaces can filter the low-energy carriers and only high-energy carriers can pass through the barrier,⁴³ which increases the whole S of samples. The energy filtering effect on the S also reduces the harmful effect from the p-type rGO, which, thus, maintains the S of rGO/Bi₂Te₃ film to relatively high n-type values. Simultaneously, the interface of Bi₂Te₃ nanoplates and the interface between rGO and Bi₂Te₃ can efficiently scatter phonons with relative wavelength.⁴⁴ Therefore, we obtain low κ values for all the samples. Owing to the large number of interfaces between rGO, the κ of rGO/Bi₂Te₃ films are still at a relatively low value though the high κ of the added rGO.

The power output conclusion of all rGO/Bi₂Te₃ TEG devices

The thermoelectric output performances of the rGO/Bi₂Te₃ TEG devices are characterized. Fig. 3 summarizes the power output and operating voltage of TEG devices as a function of electrical current, which were tested by varying the external load resistance (R_e). Note that P_{max} is acquired when the external load

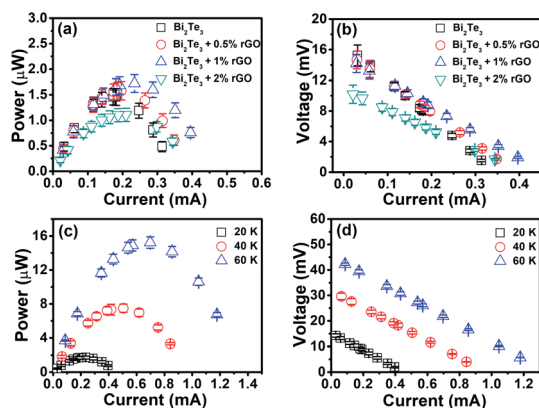


Fig. 3 Output characterization: (a) and (b) power output and operating voltage of flexible TEG devices made from Bi₂Te₃ nanoplates with different mass ratio of rGO nanosheets at temperature difference of 20 K; (c) and (d) power output and operating voltage of 1% rGO/Bi₂Te₃ composite TEG devices at temperature difference of 20 K, 40 K and 60 K. Error bars are estimated from repeatability of the experimental results: for each condition, three samples are prepared and measured. The 1% rGO/Bi₂Te₃ TEG devices possess the highest P_{max} . And larger ΔT brings higher power output and voltage.

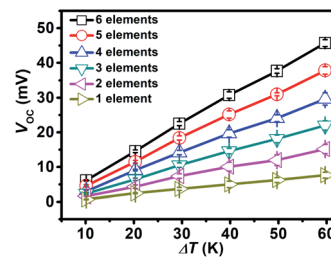


Fig. 4 Open circuit voltage (ΔV_{oc}) against temperature difference (ΔT) responsible for number of thermal elements of flexible TEG devices made from pure Bi₂Te₃ nanoplates. Error bars are estimated from repeatability of the experimental results: for each material-making condition, three samples are prepared and measured.

resistance matches the internal resistance of the TEG devices. As indicated in Fig. 3a, with the increasing percentage of rGO from 0 to 0.5%, 1% and 2%, P_{max} at $\Delta T = 20$ K is 1.45 μW , 1.58 μW , 1.72 μW and 1.09 μW , respectively. Fig. 3b presents the corresponding operating voltage plots of four TEG devices. They are 7.98 mV, 7.95 mV, 7.32 mV and 5.85 mV, respectively ($\Delta T = 20$ K) at their P_{max} . Remarkably, the 1% rGO/Bi₂Te₃ TEG devices possess the highest P_{max} .

The P_{max} of the TEG devices is defined by the equation $P_{\text{max}} = V_{\text{oc}}^2 / (4R_i)$ when $R_i = R_e$,⁹ where V_{oc} ($V_{\text{oc}} = mS\Delta T$) is the open circuit voltage of the TEG devices without the load and R_i is the internal resistance of the TEG devices. We also examined the V_{oc} of TEG devices made from pure Bi₂Te₃ nanoplates as shown from the data in Fig. 4. V_{oc} is approximately linear with ΔT . A video about the real-time measurements of V_{oc} of the TEG is

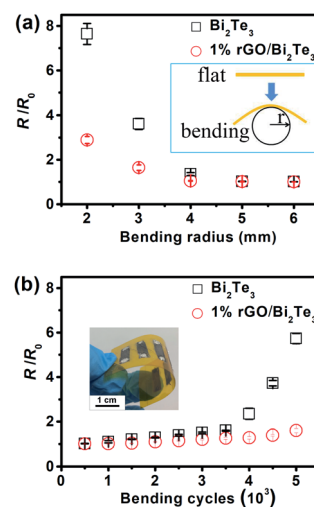


Fig. 5 Flexibility testing: the change rate of the electrical resistance of pure Bi₂Te₃ nanoplates and 1% rGO/Bi₂Te₃ composite-based TEG devices as a function of different bending radius (a) and bending cycles (b) on 5 mm in bending radius. R_0 and R are resistances of devices before and after bending. Error bars are estimated from repeatability of the experimental results: for each condition, three samples are prepared and measured. The R of 1% rGO/Bi₂Te₃ device increases by 2% after 1000 bending cycles with 5 mm of bending radius, while that of Bi₂Te₃ device increases about 10%. It reveals that the addition of rGO improved the flexibility of 1% rGO/Bi₂Te₃ device.



supplied in Video S1.† As displayed in the video, V_{oc} of a TEG device is over 14 mV at $\Delta T = 20$ K. Simultaneously, there is no significant change in the V_{oc} in 24 hours as shown in Fig. S6,† which means a good stability for the TEG device.

To further explore the influence of ΔT on the power output and operating voltage, the TEG devices made from 1% rGO/Bi₂Te₃ nanoplate inks were investigated at ΔT of 20 K, 40 K and 60 K, as shown in Fig. 3c, with ΔT increasing, P_{max} values are 1.72 μ W, 7.48 μ W and 15.22 μ W, respectively, while the corresponding operating voltages at P_{max} are 7.32 mV, 15.29 mV and 21.81 mV, respectively (Fig. 3d). Both results reveal that larger ΔT brings higher power output and voltage.

The flexibilities of rGO/Bi₂Te₃ TEG devices

Finally, the flexibilities of the pure Bi₂Te₃ and 1% rGO/Bi₂Te₃ TEG devices were inspected with a bending test. Fig. 5 presents the change rate of room-temperature electrical resistance (R/R_0) with different bending radii and bending cycles, R_0 and R are the resistances of devices before and after bending. The R of 1% rGO/Bi₂Te₃ device is almost unchanged when the bending radius was decreased from 6 mm to 4 mm as shown in Fig. 5a, and starts to significantly increase after the bending radius goes down to 2 mm. In contrast, the R of pure Bi₂Te₃ device starts to rapidly increase when the bending radius decreased from 4 mm to 2 mm. At the same time, as shown in Fig. 5b, the R of 1% rGO/Bi₂Te₃ device increases by 2% after 1000 bending cycles with

5 mm of bending radius, but the R of Bi₂Te₃ device increases by almost 10% by such a bending operation. Both results reveal that the flexibility of the 1% rGO/Bi₂Te₃ device is improved with the introduction of rGO. As mentioned above, the rGO nano-sheets covered the Bi₂Te₃ nanoplates like “gossamer” as shown in Fig. 1. The gossamer-like rGO may favor releasing the strain and improve the flexibility of the composite film.⁴⁵

Putting the comparison aside, the changes of R for both pure Bi₂Te₃ and 1% rGO/Bi₂Te₃ device are within 10% after 1000 bending cycles with 5 mm of bending radius. This result shows the good electrical stability of the two devices. We believe the reasons are like this: firstly, the suitable thickness of thermal elements (about 40 μ m) prevents the thermal elements from cracking during bending. Secondly, the soft copper wires rather than some deposited metal layers are served as the connection material and these copper wires, which connect thermal elements end to end, have excellent resistance to bend. All these structural designs of the devices contribute to the stable electrical properties during bending.

Compared with other related literature shown in Table 2, our rGO/Bi₂Te₃ TEG device with just six planar thermal elements possesses good power output, which should be further increased with an increasing number of thermal elements. Moreover, our TEG device shows excellent flexibility, thanks to the good flexibility of both rGO/Bi₂Te₃ elements and the soft connection of thermal elements.

Table 2 State-of-art excellent examples of printed planar TEG devices

Materials	$S^2\sigma$, μ W cm ⁻¹ K ⁻²	Element number	P_{max} at various ΔT	V_{oc} at various ΔT	Thickness of elements	Bending radius	Bending times/increase percentage of resistance
This work	3.2–4.6	6	20 K: 1.72 μ W 60 K: 15.22 μ W	20 K: 14 mV 60 K: 45 mV	40 μ m	5 mm	1000 cycles/2%
Bi ₂ Te _{2.8} Se _{0.2} film ¹¹	4.68	5	20 K: 0.4 μ W 60 K: 6.1 μ W	20 K: 12 mV 60 K: 41 mV	10 μ m	7 mm	150 cycles/1.4%
Bi ₂ Te ₃ film ⁹	2–3.4	62	20 K: 25 μ W	20 K: 210 mV	120 μ m	—	—
Carbon nanotube ¹	15	3 p–n couples	27.5 K: 2.51 μ W	27.5 K: 11 mV	150 μ m	—	—
WS ₂ (n-type)	WS ₂ : 0.05	3 p–n couples	60 K: 2.58 nW	60 K: 13 mV	100 nm	—	—
NbSe ₂ (p-type) ⁴	NbSe ₂ : 0.26						
RGO/Bi ₂ Te ₃ (n-type) and SWCNTs/Sb ₂ Te ₃ (p-type) ⁴³	RGO/Bi ₂ Te ₃ : 1.08 SWCNTs/Sb ₂ Te ₃ : 0.5	10 couples	70 K: 23.6 μ W	70 K: 135 mV	25–30 μ m	1.8 mm	800 cycles/25%
PEDOT-Tos/TTF-TCNQ ⁴⁶	3.24	54	10 K: 0.128 μ W	—	3–30 μ m	—	—
PEDOT:PSS ⁴⁷	2.84	16	10 K: 10.59 nW	10 K: 12.75 mV	1.1 mm	—	—
CNT/P3HT ⁴⁸	3.25	41	10 K: 32.7 nW	10 K: 41.8 mV	1.1–1.5 mm	—	—
Bi ₂ Te ₃ /CF ⁴⁹	2.5–3.75	12 pairs	15 K: 0.2 μ W	15 K: 2 mV	30 μ m	4 mm	100 cycles/6%
Bi ₂ Te ₃ (n-type) and Bi _{0.5} Sb _{1.5} Te ₃ (p-type) NWs ⁵⁰	1.8	5 pairs	32.5 K: 127 nW	32.5 K: 45 mV	0.62–0.89 μ m	11 mm	50 cycles/2.5%
SWCNT/PEDOT:PSS coated Te nanorod ⁵¹	1.04	6	44 K: 53.6 nW	44 K: 5.6 mV	10 μ m	—	—
Bi ₂ Se _{0.3} Te _{2.7} (n-type) and Bi _{0.5} Sb _{1.5} Te ₃ (p-type)/paper composite ⁵²	Bi ₂ Se _{0.3} Te _{2.7} : 0.44 Bi _{0.5} Sb _{1.5} Te ₃ : 0.11	10 p–n couples	35 K: 10 nW	35 K: 8 mV	200 μ m	20 mm	1000 cycles/<5%
BiSbTe films with 8 wt% Te ⁵³	3	4	80 K: 54 μ W	80 K: 64 mV	27 μ m	10 mm	1000 cycles/3%
Bi ₂ Te ₃ -based inorganic paints ⁵⁴	—	5 couples	50 K: 60.7 μ W	50 K: 79.4 mV	50 μ m	—	—
Bi ₂ Te ₃ -based inks ⁵⁵	—	—	39 K: 1.62 mW	39 K: 27 mV	1.5–2.0 mm	—	—



Conclusions

In summary, high-performance flexible TEG devices are successfully constructed by a systematic strategy including the selection of rGO/Bi₂Te₃ composites as active materials, design of thickness of thermal elements, artful connection manner, and the adoption of a simple printing method. The P_{\max} of the TEG devices on flexible polyimide substrates made from 1% rGO/Bi₂Te₃ composites is considerably enhanced to 1.72 μW at $\Delta T = 20$ K, compared with 1.45 μW at $\Delta T = 20$ K for pure Bi₂Te₃ nanoplate based TEG devices. Furthermore, P_{\max} of the 1% rGO/Bi₂Te₃ composite based TEG devices is up to 15.22 μW at $\Delta T = 60$ K. Significantly, the electrical resistance of the 1% rGO/Bi₂Te₃ composite based TEG devices merely changes by 2% after bending 1000 cycles on 5 mm in bending radius, which discloses good flexibility. The thermal elements are all n-type, next efforts will be made on solution-processed p-type materials with good ZT value, and a TEG device with several n-p thermal couples would be printed for higher power output. These inorganic-based TEGs with both high performance and good flexibility will largely promote the application of functional inorganic materials into flexible electronics through a mild wet-chemical method combined with compositional and structural design.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

The authors acknowledge financial support from by the National Basic Research Program of China (973 Program, 2015CB932600), the National Key R&D Program of China (2017YFA0208000), the National Natural Science Foundation of China (21525523, 21802130, 21874121, 21722507 and 21574048), the National Key Basic Research Program of China (2014CB931801 and 2016YFA0200700, Z. Y. T.), National Natural Science Foundation of China (21475029 and 91427302, Z. Y. T.), Frontier Science Key Project of the Chinese Academy of Sciences (QYZDJ-SSW-SLH038, Z. Y. T.), Instrument Developing Project of the Chinese Academy of Sciences (YZ201311, Z. Y. T.), CAS-CSIRO Cooperative Research Program (GJHZ1503, Z. Y. T.), "Strategic Priority Research Program" of Chinese Academy of Sciences (XDA09040100, Z. Y. T.) and K. C. Wong Education Foundation, Fundamental Research Funds for the Central Universities, China University of Geosciences (Wuhan) (No. CUG170669).

References

- 1 W. B. Zhou, Q. X. Fan, Q. Zhang, L. Cai, K. W. Li, X. G. Gu, F. Yang, N. Zhang, Y. C. Wang, H. P. Liu, W. Y. Zhou and S. S. Xie, *Nat. Commun.*, 2017, **8**, 1.
- 2 Z. Li, C. Xiao, H. Zhu and Y. Xie, *J. Am. Chem. Soc.*, 2016, **138**, 14810.

- 3 C. H. Zhang, Z. P. Peng, Z. Li, L. G. Yu, K. A. Khor and Q. H. Xiong, *Nano Energy*, 2015, **15**, 688.
- 4 J. Y. Oh, J. H. Lee, S. W. Han, S. S. Chae, E. J. Bae, Y. H. Kang, W. J. Choi, S. Y. Cho, J. O. Lee, H. K. Baik and T. Il Lee, *Energy Environ. Sci.*, 2016, **9**, 1696.
- 5 S. K. Li, T. J. Fan, X. R. Liu, F. S. Liu, H. Meng, Y. D. Liu and F. Pan, *ACS Appl. Mater. Interfaces*, 2017, **9**, 3677.
- 6 C. Xiao, K. Li, J. J. Zhang, W. Tong, Y. W. Liu, Z. Li, P. C. Huang, B. C. Pan, H. B. Su and Y. Xie, *Mater. Horiz.*, 2014, **1**, 81.
- 7 B. A. Xu, T. L. Feng, M. T. Agne, L. Zhou, X. L. Ruan, G. J. Snyder and Y. Wu, *Angew. Chem., Int. Ed.*, 2017, **56**, 3546.
- 8 C. H. Zhang, H. Ng, Z. Li, K. A. Khor and Q. H. Xiong, *ACS Appl. Mater. Interfaces*, 2017, **9**, 12501.
- 9 D. Madan, Z. Q. Wang, A. Chen, R. C. Juang, J. Keist, P. K. Wright and J. W. Evans, *ACS Appl. Mater. Interfaces*, 2012, **4**, 6117.
- 10 J. Hirotsu, S. Kishimoto and Y. Ohno, *Nanoscale Adv.*, 2019, **1**, 636.
- 11 T. Varghese, C. Hollar, J. Richardson, N. Kempf, C. Han, P. Gamarrachchi, D. Estrada, R. J. Mehta and Y. L. Zhang, *Sci. Rep.*, 2016, **6**, 1.
- 12 S. R. Tulsani, A. K. Rath and D. J. Late, *Nanoscale Adv.*, 2019, **1**, 1387.
- 13 C. B. Vining, *Nat. Mater.*, 2009, **8**, 83.
- 14 G. J. Snyder and E. S. Toberer, *Nat. Mater.*, 2008, **7**, 105.
- 15 S. S. Zhou, L. Gan, D. L. Wang, H. Q. Li and T. Y. Zhai, *Nano Res.*, 2018, **11**, 2909.
- 16 S. Shin, R. Kumar, J. W. Roh, D. S. Ko, H. S. Kim, S. I. Kim, L. Yin, S. M. Schlossberg, S. Cui, J. M. You, S. Kwon, J. L. Zheng, J. Wang and R. K. Chen, *Sci. Rep.*, 2017, **7**, 7317.
- 17 C. L. Ou, A. L. Sangle, A. Datta, Q. S. Jing, T. Busolo, T. Chalklen, V. Narayan and S. Kar-Narayan, *ACS Appl. Mater. Interfaces*, 2018, **10**, 19580.
- 18 F. Suarez, D. P. Parekh, C. Ladd, D. Vashaee, M. D. Dickey and M. C. Ozturk, *Appl. Energy*, 2017, **202**, 736.
- 19 J. Gao, C. Y. Liu, L. Miao, X. Y. Wang, Y. Peng and Y. Chen, *RSC Adv.*, 2016, **6**, 31580.
- 20 A. C. Arias, J. D. MacKenzie, I. McCulloch, J. Rivnay and A. Salleo, *Chem. Rev.*, 2010, **110**, 3.
- 21 Y. G. Sun and J. A. Rogers, *Adv. Mater.*, 2007, **19**, 1897.
- 22 Z. Y. Lin, Y. Chen, A. D. Yin, Q. Y. He, X. Q. Huang, Y. X. Xu, Y. A. Liu, X. Zhong, Y. Huang and X. F. Duan, *Nano Lett.*, 2014, **14**, 6547.
- 23 S. R. Thomas, P. Pattanasattayavong and T. D. Anthopoulos, *Chem. Soc. Rev.*, 2013, **42**, 6910.
- 24 D. F. Ding, D. W. Wang, M. Zhao, J. W. Lv, H. Jiang, C. G. Lu and Z. Y. Tang, *Adv. Mater.*, 2017, **29**, 1.
- 25 C. Han, Z. Li, G. Q. Lu and S. X. Dou, *Nano Energy*, 2015, **15**, 193.
- 26 C. Han, Q. Sun, Z. X. Cheng, J. L. Wang, Z. Li, G. Q. Lu and S. X. Dou, *J. Am. Chem. Soc.*, 2014, **136**, 17626.
- 27 C. Han, Q. Sun, Z. Li and S. X. Dou, *Adv. Energy Mater.*, 2016, **6**, 1600498.
- 28 C. L. Wan, X. K. Gu, F. Dang, T. Itoh, Y. F. Wang, H. Sasaki, M. Kondo, K. Koga, K. Yabuki, G. J. Snyder, R. G. Yang and K. Koumoto, *Nat. Mater.*, 2015, **14**, 622.



- 29 S. J. Kim, J. H. We and B. J. Cho, *Energy Environ. Sci.*, 2014, **7**, 1959.
- 30 C. Han, Z. Li and S. X. Dou, *Chin. Sci. Bull.*, 2014, **59**, 2073.
- 31 W. Glatz, S. Muntwyler and C. Hierold, *Sens. Actuators, A*, 2006, **132**, 337.
- 32 S. Hong, Y. Gu, J. K. Seo, J. Wang, P. Liu, Y. S. Meng, S. Xu and R. K. Chen, *Sci. Adv.*, 2019, **5**, eaaw0536.
- 33 S. Hong, S. Shin and R. Chen, *Adv. Funct. Mater.*, 2020, **30**, 1909788.
- 34 C. S. S. Sandeep, J. M. Azpiroz, W. H. Evers, S. C. Boehme, I. Moreels, S. Kinge, L. D. A. Siebbeles, I. Infante and A. J. Houtepen, *ACS Nano*, 2014, **8**, 11499.
- 35 Y. X. Xu, H. Bai, G. W. Lu, C. Li and G. Q. Shi, *J. Am. Chem. Soc.*, 2008, **130**, 5856.
- 36 X. M. Li, J. Yin, J. X. Zhou, Q. Wang and W. L. Guo, *Appl. Phys. Lett.*, 2012, **100**, 183108.
- 37 Y. Zhao, G. S. Tang, Z. Z. Yu and J. S. Qi, *Carbon*, 2012, **50**, 3064.
- 38 H. R. Yang, J. H. Bahk, T. Day, A. M. S. Mohammed, G. J. Snyder, A. Shakouri and Y. Wu, *Nano Lett.*, 2015, **15**, 1349.
- 39 H. Ju and J. Kim, *Dalton Trans.*, 2015, **44**, 11755.
- 40 R. Pasricha, S. Gupta and A. K. Srivastava, *Small*, 2009, **5**, 2253.
- 41 H. Ju, M. Kim and J. Kim, *J. Mater. Sci.: Mater. Electron.*, 2016, **27**, 3427.
- 42 G. Anoop, J. R. Rani, J. Lim, M. S. Jang, D. W. Suh, S. Kang, S. C. Jun and J. S. Yoo, *Sci. Rep.*, 2016, **6**, 1.
- 43 B. Wu, Y. Guo, C. Y. Hou, Q. H. Zhang, Y. G. Li and H. Z. Wang, *Adv. Funct. Mater.*, 2019, **29**, 1900304.
- 44 S. Shin, J. W. Roh, H. S. Kim and R. K. Chen, *J. Appl. Phys.*, 2018, **123**, 205106.
- 45 D. Chen, L. H. Tang and J. H. Li, *Chem. Soc. Rev.*, 2010, **39**, 3157.
- 46 O. Bubnova, Z. U. Khan, A. Malti, S. Braun, M. Fahlman, M. Berggren and X. Crispin, *Nat. Mater.*, 2011, **10**, 429.
- 47 E. J. Bae, Y. H. Kang, K. S. Jang and S. Y. Cho, *Sci. Rep.*, 2016, **6**, 1.
- 48 C. T. Hong, Y. H. Kang, J. Ryu, S. Y. Cho and K. S. Jang, *J. Mater. Chem. A*, 2015, **3**, 21428.
- 49 Q. Jin, W. B. Shi, Y. Zhao, J. X. Qiao, J. H. Qiu, C. Sun, H. Lei, K. P. Tai and X. Jiang, *ACS Appl. Mater. Interfaces*, 2018, **10**, 1743.
- 50 B. L. Chen, M. Kruse, B. Xu, R. Tutika, W. Zheng, M. D. Bartlett, Y. Wu and J. C. Claussen, *Nanoscale*, 2019, **11**, 5222.
- 51 Q. F. Meng, K. F. Cai, Y. Du and L. D. Chen, *J. Alloys Compd.*, 2018, **778**, 163.
- 52 X. Zhao, W. J. Han, C. S. Zhao, S. Wang, F. G. Kong, X. X. Ji, Z. Y. Li and X. A. Shen, *ACS Appl. Mater. Interfaces*, 2019, **11**, 10301.
- 53 T. Varghese, C. C. Dun, N. Kempf, M. Saeidi-Javash, C. Karthik, J. Richardson, C. Hollar, D. Estrada and Y. L. Zhang, *Adv. Funct. Mater.*, 2019, **30**, 1905796.
- 54 S. H. Park, S. Jo, B. Kwon, F. Kim, H. W. Ban, J. E. Lee, D. H. Gu, S. H. Lee, Y. Hwang, J. S. Kim, D. B. Hyun, S. Lee, K. J. Choi, W. Jo and J. S. Son, *Nat. Commun.*, 2016, **7**, 13403.
- 55 F. Kim, B. Kwon, Y. Eom, J. E. Lee, S. Park, S. Jo, S. H. Park, B. S. Kim, H. J. Lm, M. H. Lee, T. S. Min, K. T. Kim, H. G. Chae, W. P. King and J. S. Son, *Nat. Energy*, 2018, **3**, 301.

