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Novel ethyl 2-hydrazineylidenethiazolidin-5-ylidene acetate clubbed with coumarinylthiazolyl pyrazole system as potential VEGFR-2 inhibitors and apoptosis inducer: synthesis, cytotoxic evaluation, cell cycle, autophagy, *in silico* ADMET and molecular docking studies†

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Novel derivatives of ethyl 3-substituted-2-(4-oxo-2-(2-((3-(2-oxo-2*H*-chromen-3-yl)-1-(4-phenylthiazol-2-yl)-1*H*-pyrazol-4-yl)methylene)hydrazineyl)thiazol-5(4*H*)-ylidene)acetate (**5a–h**) were synthesized and assessed for their cytotoxic potential against the liver cancer cell lines Huh-7 and HepG-2. Among these, compounds **5d** and **5g** demonstrated notable antiproliferative effects, which were benchmarked against the standard drug doxorubicin. To further understand the mechanisms behind their antiproliferative activity, compounds **5d** and **5g** were investigated for their impact on the cell cycle and their ability to induce apoptosis. They were found to induce significant cellular cycle arrest at the G1 phase. Besides, they potentially enhanced the cellular late apoptosis and reduced the cellular viability. In consent with the apoptosis results, compounds **5d** and **5g** displayed significant potential autophagic induction against the studied cancer cell lines. Further, both compounds **5d** and **5g** showed strong interactions with the VEGFR-2 receptor when they were studied using molecular docking. The ADMET prediction indicated that these bioactive compounds have the potential to serve as effective to fight liver cancer.

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1. Introduction

The thiazole ring is a five-membered heterocyclic structure that occurs naturally in various substances, such as vitamins, pigments, alkaloids, and components of both plant and animal cells.¹ The presence of the –S–C=N– moiety in the thiazole system has been demonstrated to enhance lipid solubility and hydrophilicity, which are critical factors enabling the structure to be efficiently metabolized through standard biochemical pathways.² Thiazole and its thiazolidinone derivatives, particularly those featuring a carbonyl group at the 2-, 4-, or 5-position, have been shown to possess significant pharmaceutical

potential.^{3,4} Notably, thiazolidin-4-one derivatives have displayed broad-spectrum anticancer activity against breast cancer (MCF-7) and lung cancer (A549) cell lines, as well as promising EGFR inhibitory effects^{5,6} (Fig. 1).

The thiazolidin-4-one containing ylidene acetate fragment at 5-position was synthesized by reaction of thioamides with dialkyl acetylene dicarboxylate (DAAD) under mild conditions.⁷ These compounds recorded pharmacological activities such as leukotriene B4 inhibitors,⁸ antidiabetic,⁹ antiparasitic,¹⁰ antibacterial¹¹ and anticancer.¹² On the other hand, numerous 2-hydrazinyl-thiazolidinone derivatives have been identified as potential chemotherapeutic agents, demonstrating a range of biological activities, including antimicrobial,^{13,14} antiviral,¹⁵ antiparasitic,¹⁶ and antitumoral effects.¹⁷ In particular, arylidenehydrazinethiazole pharmacophore structures have shown potent inhibitory activity against tyrosyl-DNA phosphodiesterase 1 (TDP1) in the nanomolar range. By inhibiting DNA repair mechanisms, these compounds serve as adjuvant therapies, enhancing the efficacy of chemotherapy drugs, especially in the treatment of drug-resistant tumors.^{18,19}

The coumarin ring as one of the most significant heterocyclic compounds exhibited numerous biological actions, such as

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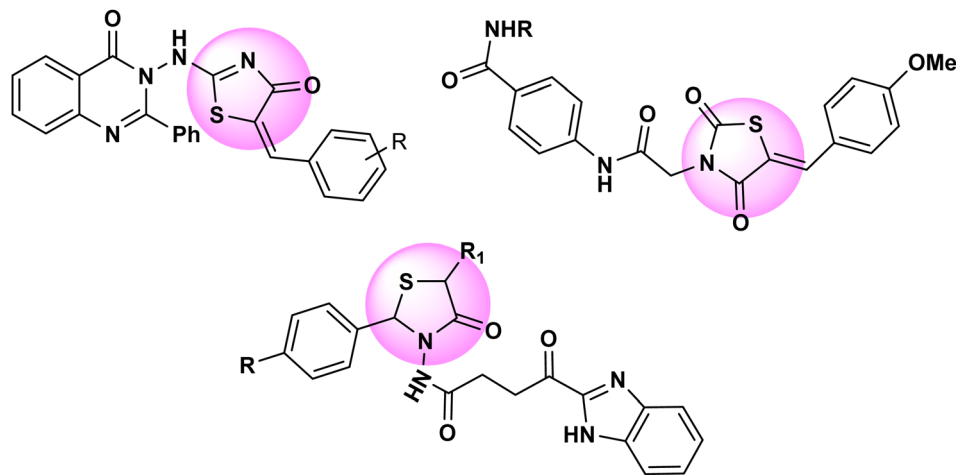


Fig. 1 Some thiazolidinone compounds exhibited anticancer effects against MCF-7 and A549 cell lines.

anticancer,²⁰ antituberculosis,²¹ anticholinesterase,²² antiviral,²³ anticoagulant,²⁴ antidepressant,²⁵ antioxidant²⁶ and antibacterial agent.²⁷ In the same manner, five-membered heterocyclic scaffolds containing pyrazole and thiazole, are the most important ones.^{28–30} Chemists have been interested in pyrazole and thiazole scaffolds because of their wide range of biological actions, including anti-inflammatory,³¹ antiviral,³² anticancer,³³ antibacterial³⁴ and anti-tuberculosis.³⁵ Several commercial drugs containing coumarin, pyrazole, or thiazole rings are illustrated in Fig. 2.

Cancer is one of the most serious diseases. After heart disease, cancer diseases are thought to be the second most common cause of mortality worldwide.^{36,37} One of the most important facets of cancer treatment remains chemotherapy. The most significant issue with the use of chemotherapy is its negative effects. Therefore, finding novel compounds as safe anticancer agents is a challenging goal that will help not only treat cancer but also overcome drug-resistant malignancy and prevent adverse medication reactions.

Depending on the previous and continuing our efforts to discover novel polyheterocyclic frames as a new antiproliferative

agents to inhibit of cancer cell growth with targeting inhibitory action,^{38–42} we have now designed a new series of ethyl 2-hydrazineylidenethiazolidin-5-ylidene acetate clubbed with coumarinylthiazolyl pyrazole system (Fig. 3). Using the SBR assay, the synthesized compounds were tested on a panel of liver cancer cell lines, namely Huh-7 and HepG2 and the results were compared with standard drug doxorubicin.

2. Results and discussion

A new series of coumarinyl-thiazolyl-pyrazole clubbed with hydrazineylidene-thiazolidin-5-ylidene acetate derivatives **5a–h** were synthesized by two-step reaction: first, a mixture of 3-acetylcoumarin (**1**), thiosemicarbazide and phenacyl bromide in DMF at 70 °C temperature was heated for 1 hour to give the corresponding coumarinyl-thiazolyl hydrazone **2**. Then, the latter hydrazone reacted with the dimethylformamide (DMF) in the presence of phosphoryl chloride under Vilsmeier–Haack reaction conditions to form 3-(2-oxo-2H-chromen-3-yl)-1-(4-phenyl-thiazol-2-yl)-1H-pyrazole-4-carboxaldehyde (**3**) (Scheme

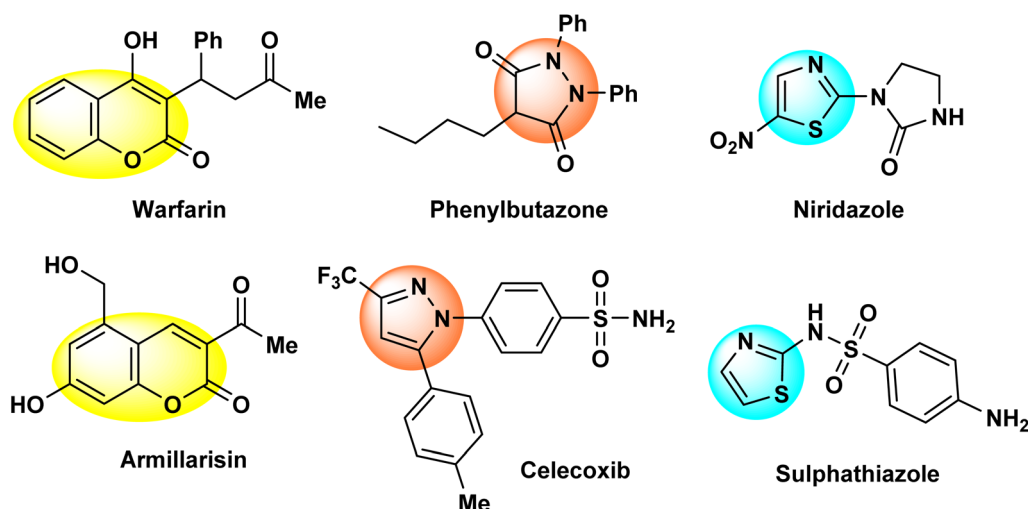


Fig. 2 Marketed drugs containing coumarin, pyrazole or thiazole scaffold.



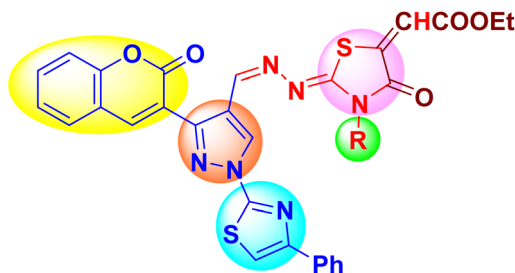


Fig. 3 The designed molecular frame for the title compounds.

1).⁴³ In the next step, one-pot three-component reaction of the aldehyde 3 with a series of 4-substituted thiosemicarbazide and diethyl acetylenedicarboxylate in glacial acetic acid afforded the corresponding ethyl 3-substituted-2-{4-oxo-2-[(3-(2-oxo-2H-chromen-3-yl)-1-(4-phenylthiazol-2-yl)-1H-pyrazol-4-yl)methylene]hydrazineyl}thiazol-5(4H)-ylidene}acetate (**5a–h**) (Scheme 2 and Fig. 4). The progress of the reaction was monitored at regular intervals using thin layer chromatography (TLC). The structures of the synthesized compounds (**5a–h**) were confirmed and characterized through various spectroscopic methods, including IR, MS and NMR spectroscopy, specifically ¹H-NMR and ¹³C-NMR spectra.

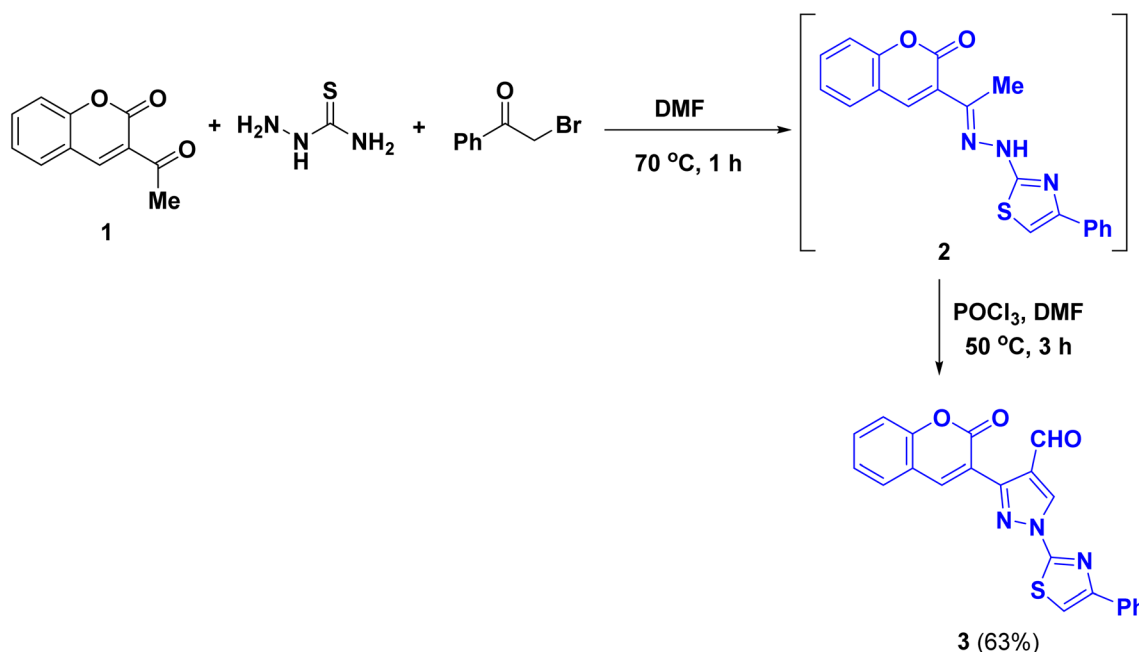
The IR spectroscopic analysis of target compounds **5a–h**, featured four absorption bands at 1706–1720, 1688–1704, 1716–1754 and 1626–1652 cm^{−1} corresponded to C=O_{coumarin}, C=O_{ester}, C=O_{thiazolidinone} and C=N stretching, respectively. Also, there is band for NH stretching band observed for the product **5a** at 3116 cm^{−1}. The ¹H-NMR spectra of target compounds **5a–h** present a similar pattern for the ylidene acetate unit. The most upfield signals, between δ 1.20–1.29 and 4.19–4.25 ppm,

corresponds to the ethoxy moiety (OCH₂CH₃). The singlets around δ 6.54–6.71 ppm were assigned to the olefinic protons (=CH). In addition, the other singlets in range δ 8.02–8.08, 8.45–8.64, 8.36–8.41 and 9.06–9.26 ppm due to the characteristic protons H-5_{thiazole}, H-4_{coumarin}, CH=N and H-5_{pyrazole}, respectively.⁴⁴ In the ¹³C-NMR spectra, the signals belonging to =CHCOOCH₂CH₃ moiety were observed at δ 13.6–15.1 (CH₃), 61.2–62.3 (OCH₂), 122.1–122.5 (=CH) and 160.2–161.2 (C=O) ppm.⁴⁴ Also, the carbon atoms of the CH=N, C=O_{thiazolidinone}, C=O_{coumarin} and C-2_{thiazolidinone} appeared in the regions δ 137.0–138.7, 159.0–159.8, 159.4–160.2 and 166.2–167.6 ppm, respectively. The expected molecular ions (M⁺) were found in the mass spectra of all target compounds **5a–h**.

3. Antiproliferative properties

3.1. Assessment of cytotoxicity effects

To assess the impact of the synthesized 3-substituted-2-{4-oxo-2-[(3-(2-oxo-2H-chromen-3-yl)-1-(4-phenylthiazol-2-yl)-1H-pyrazol-4-yl)methylene]hydrazineyl}thiazol-5(4H)-ylidene}acetate (**5a–h**) on the viability of cancer cells, the SRB assay⁴⁵ was conducted in two liver cancer cell lines such as Huh-7 and HepG2 and the results were compared with standard drug doxorubicin. Huh-7 and HepG2 are well-established human hepatoma cell lines employed extensively in hepatic research. They possess distinct characteristics that influence their utility in various scientific applications. HepG2 cancer cells originated from a hepatoblastoma, an embryonic liver tumor primarily affecting children, while Huh-7 cancer cells derived from a well-differentiated hepatocellular carcinoma (HCC), the most common form of adult liver cancer. The results were presented in Table 1 and provided valuable insights into the compounds' half-maximal inhibitory concentration (IC₅₀) values for each



Scheme 1 The synthetic pathway for the synthesis of the starting material 3.





mL⁻¹, respectively, whereas compounds **5b**, **5c**, **5e**, **5f** and **5h** showed weak activity. Both compounds **5d** and **5g** may serve as potential candidates for further optimization to treat hepatoma cancer.

The Structure–Activity Relationship (SAR) analysis of the prepared compounds against HepG2 and Huh-7 cancer cell lines revealed key trends. The unsubstituted compound **5a** (R = H) exhibited moderate activity against HepG2 but was highly potent against Huh-7, indicating greater sensitivity of Huh-7 to this compound. Introduction of a methyl group in **5b** resulted in complete loss of activity against both cell lines, suggesting that small alkyl groups like methyl hinder activity. Similarly, the

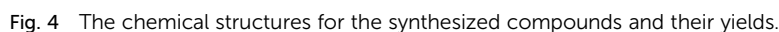


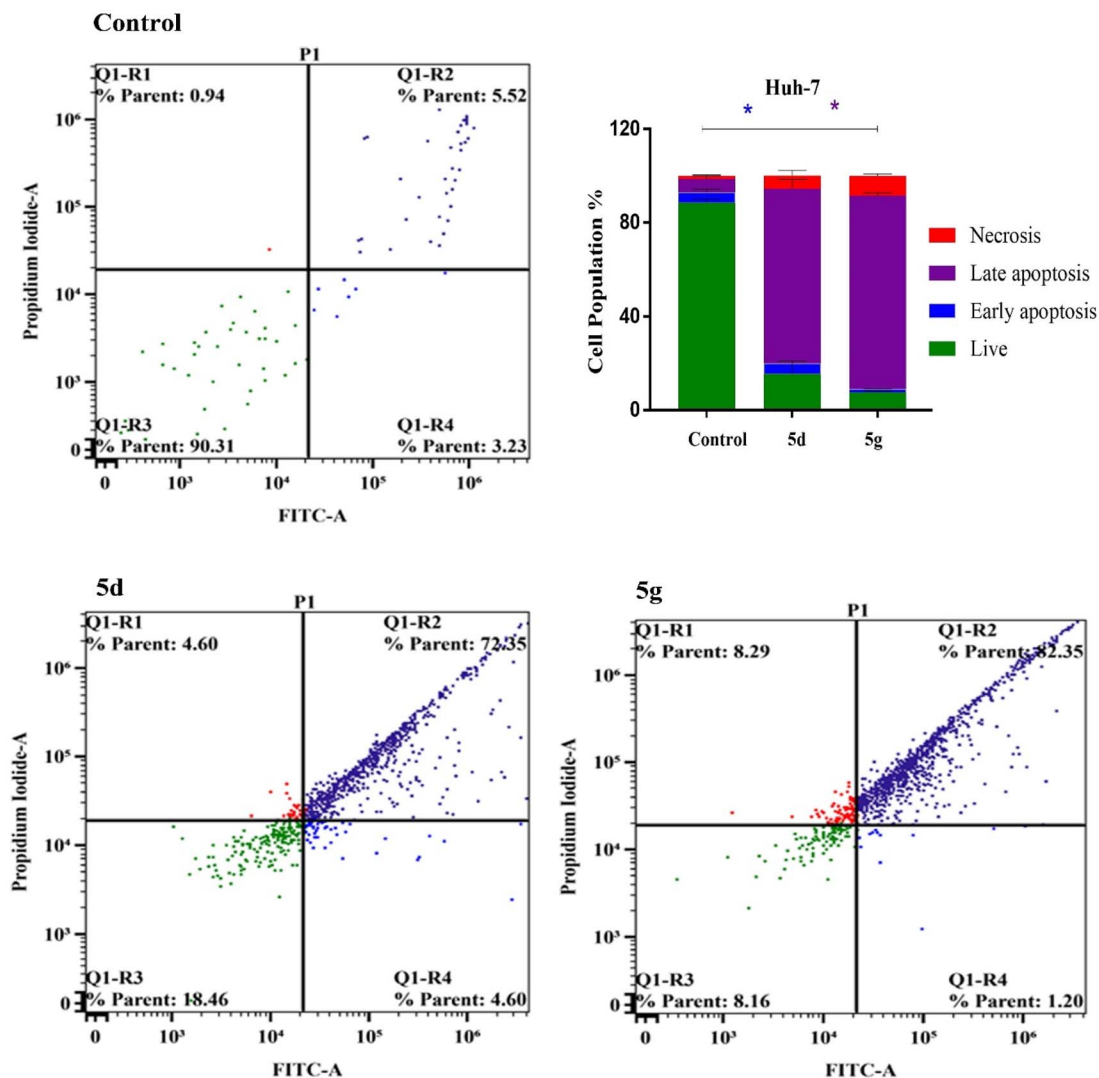
Table 1 *In vitro* anticancer activity of synthesized compounds 5a–h against Huh-7 and HepG2 cells

Compounds	IC ₅₀ ^a (μg mL ^{−1})	
	Huh-7	HepG2
5a	0.8 ± 0.1	28.9 ± 1.7
5b	≥100	≥100
5c	0.5 ± 0.8	≥100
5d	0.2 ± 0.01	4.2 ± 0.4
5e	≥100	≥100
5f	3.2 ± 0.8	≥100
5g	0.3 ± 0.03	2.3 ± 0.3
5h	≥100	≥100
Doxorubicin	1.5 ± 0.6	1.3 ± 0.4

^a IC₅₀ values are the mean ± SD of three separate experiments.

ethyl group in 5c led to inactivity against HepG2 but retained high potency against Huh-7, indicating that Huh-7 tolerates larger alkyl groups better than HepG2. The allyl group in 5d

significantly enhanced activity against both cell lines, particularly against Huh-7, suggesting that the presence of a double bond improves binding or cellular uptake. In contrast, the bulky adamantanyl group in 5e caused complete inactivity against both cell lines, highlighting the detrimental effect of steric hindrance. The phenyl group in 5f resulted in inactivity against HepG2 but retained moderate activity against Huh-7, indicating that aromatic groups are better tolerated by Huh-7. The benzyl group in 5g further enhanced activity against both cell lines, especially Huh-7, suggesting that a phenyl ring with a flexible methylene linker improves potency. However, the 4-chlorophenyl group in 5h led to complete inactivity, indicating that electron-withdrawing groups like chlorine negatively impact activity. In summary, the Huh-7 cell line is generally more sensitive to these compounds than HepG2. Small alkyl groups reduce activity, while unsaturated or aromatic groups with flexible linkers (*e.g.*, allyl, benzyl) enhance potency. Bulky or electron-withdrawing groups (*e.g.*, adamantanyl, 4-chlorophenyl) are detrimental, likely due to steric hindrance or

**Fig. 5** Apoptosis and necrosis evaluation of compounds 5d and 5g in Huh-7 cancer cells.

unfavorable electronic effects. These insights can guide further optimization, focusing on substituents that enhance activity against both cell lines while avoiding sterically hindered or electron-withdrawing groups.

3.3. Apoptosis analysis

To further characterize the apoptotic activity of the studied compounds **5d** and **5g**, we examined and quantified the percentage of apoptotic cells that was determined by Alexa Fluor-488/PI staining, and the stained cells were subsequently detected by flow cytometric analysis.⁴⁶ Our splitting data depicted that the quantification of annexin-V was remarkably increased in the late apoptosis quadrants of Huh-7 cells subjected to **5d** and **5g** (72.35% and 82.35%, respectively) compared to that of control (5.52%). Also, we observed no significant

changes for both **5d** and **5g** compounds in the early apoptosis. Furthermore, the cellular viability of Huh-7 cells was significantly decreased by treatments with compounds **5d** and **5g** (18.46% and 8.16%, respectively) compared to that of control (90.31%). Consistently, our results showed increased levels of necrotic cells in Huh-7 cells subjected to **5d** and **5g** (4.60% and 8.29%, respectively) compared with control (0.94%) (Fig. 5). Concurrently, our investigations of cellular late apoptosis within the HePG2 hepatic cells was remarkably increased using compounds **5d** and **5g** (39.38% and 54.14%, respectively) *versus* control (1.88%). Furthermore, compound **5d** induced a significant increase in the populations of necrotic cells (31.26%) compared to control (1.22%). Together with increased late apoptotic activity, the early apoptosis did not change for both **5d** and **5g** compounds. However, they successfully lower the

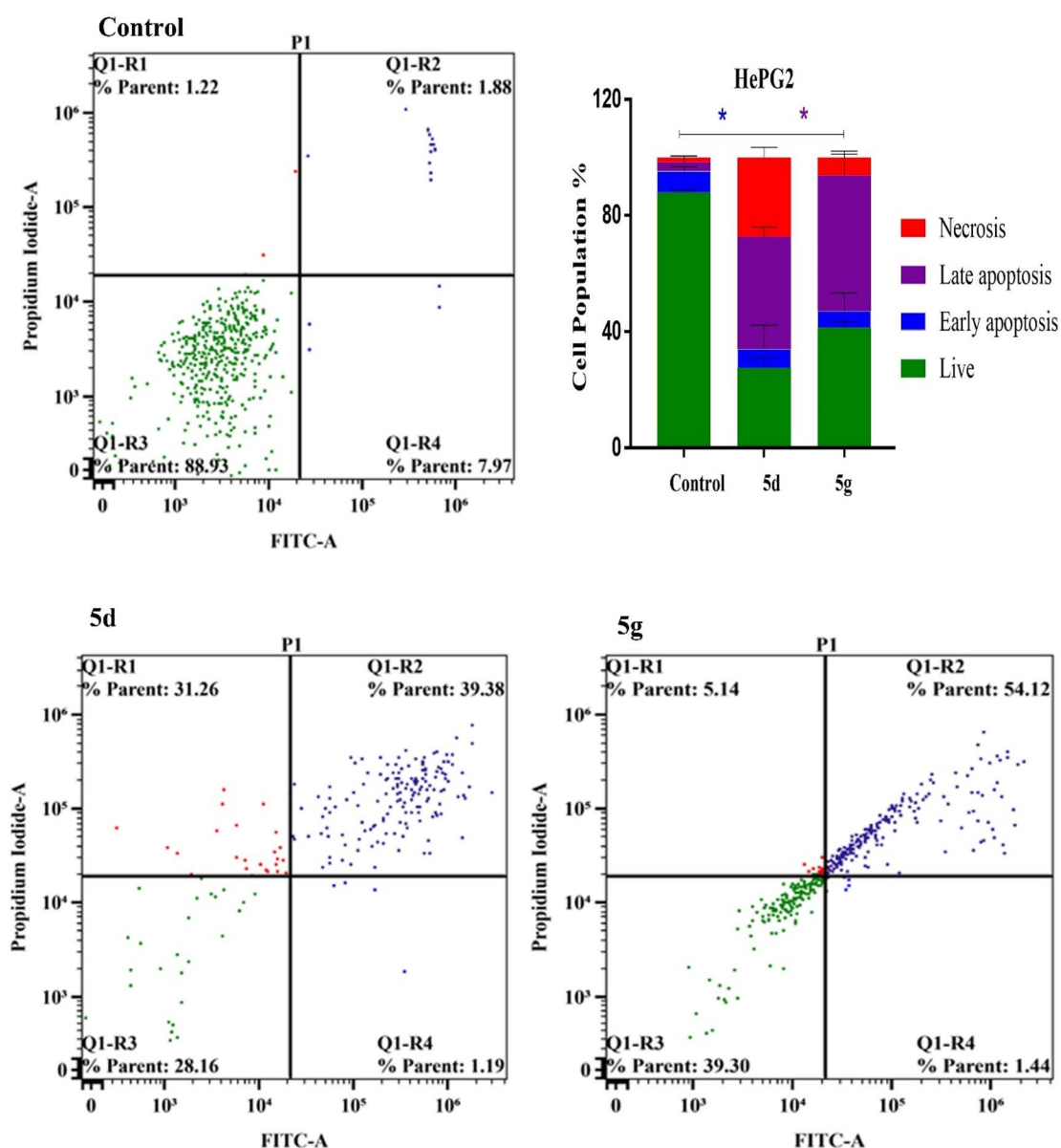


Fig. 6 Apoptosis and necrosis evaluation of compounds **5d** and **5g** in HepG2 cancer cells.



viability rates of HePG2 cells (28.16% and 39.30%, respectively) compared to control (88.93%) (Fig. 6). These results supported that these compounds exert their cytotoxic activities against both Huh-7 and HePG2 hepatic cancer cells and potentially enhance the cellular late apoptosis and reduce the cellular viability.

3.4. Cell-cycle distribution

To determine the potentials of activity of the bioactive compounds **5d** and **5g**, we analyzed their influence on the cell cycle of Huh-7 and HePG2 hepatic cancer cells using the standard flow cytometry technique.⁴⁷ In case of Huh-7 cancer cells, our records showed that both compounds **5d** and **5g** arrested the cell cycle at the G1 phase with average percentages of

(50.47% and 34.12%) with moderate significant changes compared to that of control (43.36%). Also, compound **5g** caused significant increase in the cellular distribution at the S and G2 phases (50.11% and 15.77%) compared to control (43.36% and 13.27%), respectively. However, compound **5d** did not cause significant changes compared to that of control at the S and G2 phases (Fig. 7). With respect to the HePG2 cells, our analyses showed that both compounds **5d** and **5g** significantly arrested the cell cycles at the G1 phase (55.13% and 57.13%, respectively) compared to that of control (46.43%). However, they displayed non-significant increase for cellular distribution at S and G2 phases in comparison with control (Fig. 8). Therefore, both compounds **5d** and **5g** showed promising high ability to arrest the cell cycle generally at G1 phase in Huh-7 and HePG2 hepatic cancer cells.

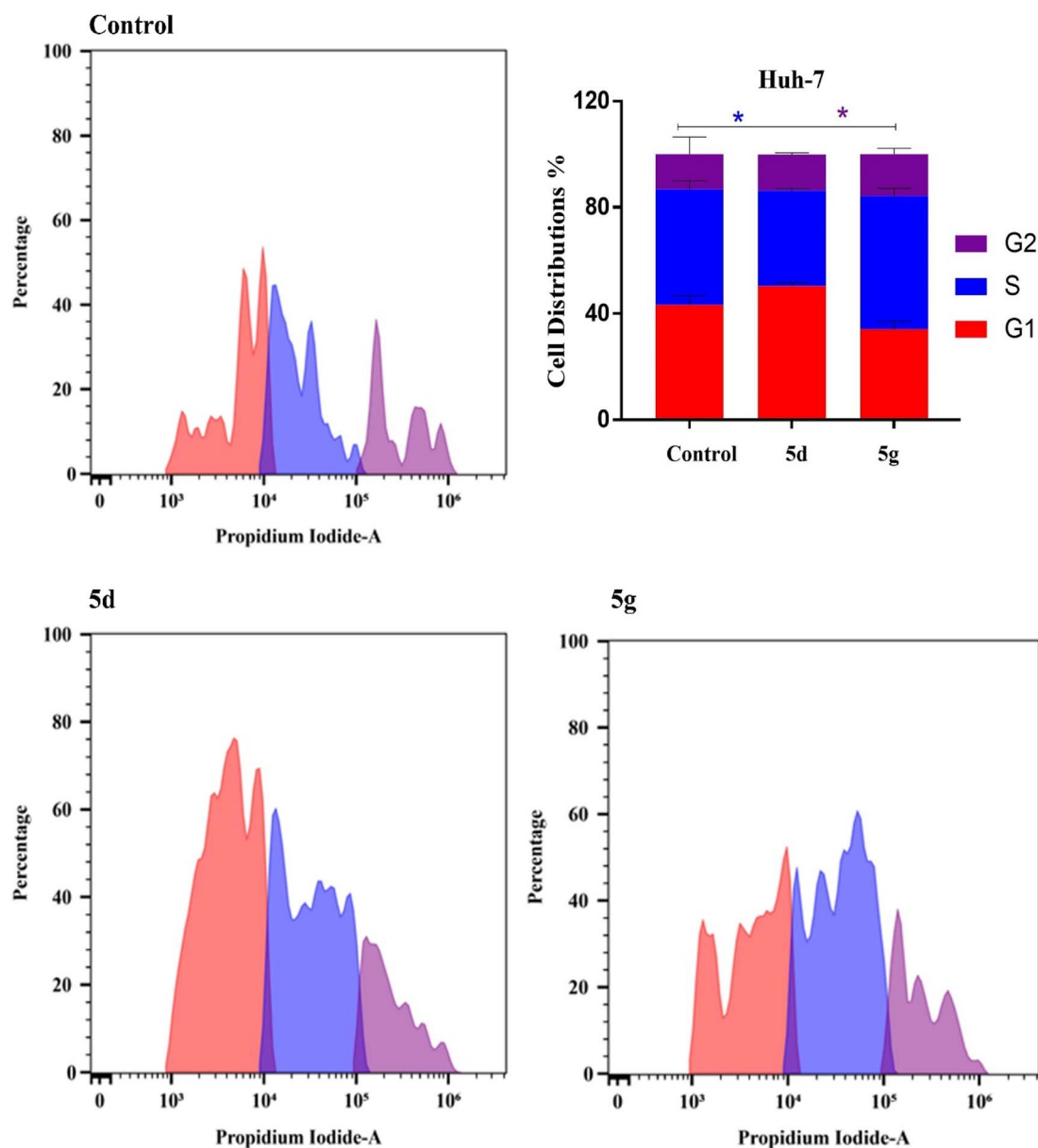


Fig. 7 Cell cycle distribution of compounds **5d** and **5g** was analyzed using DNA cytometry in Huh-7 cell lines.

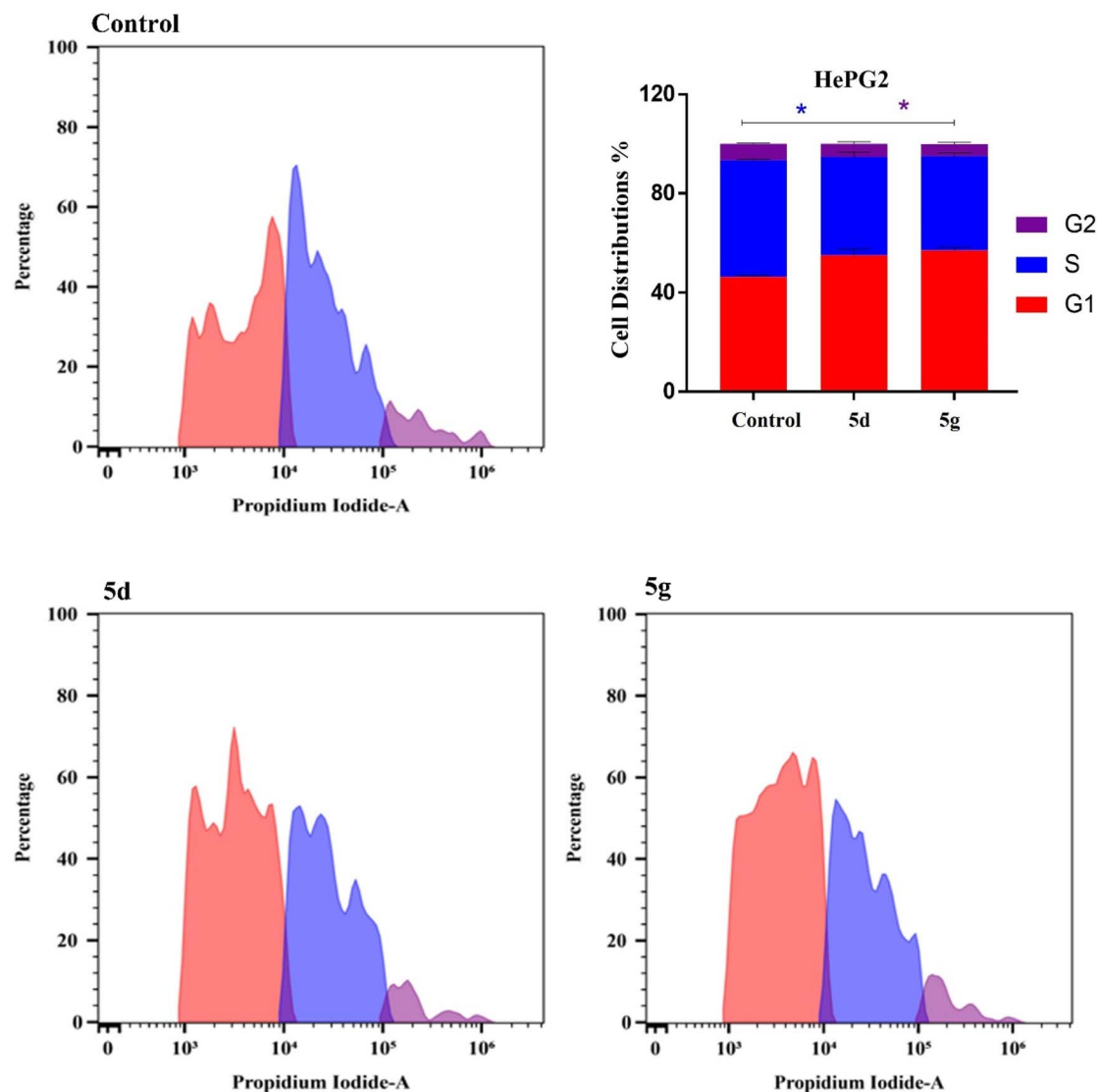


Fig. 8 Cell cycle distribution of compounds **5d** and **5g** was analyzed using DNA cytometry in HepG2 cell lines.

3.5. Autophagy assessment

We investigated the effect of bioactive compounds **5d** and **5g** on autophagy, a well-characterized cellular self-degradation pathway, in colon cancer cell lines. Acridine orange staining coupled with flow cytometry served as a quantitative measure for autophagic vacuole (AVO) formation, a hallmark of autophagy induction.⁴⁸ The net fluorescence intensity (NFI) percentage change in AVO population was determined following treatment with pre-established half-maximal inhibitory concentration (IC_{50}) doses of compounds **5d** and **5g**, alongside doxorubicin, a positive control for autophagy induction. Compounds **5d** and **5g** significantly elevated AVO formation in both Huh-7 and HepG2 liver cancer cell lines compared to the control group at the tested concentration. Notably, 42.08% and 45.46% increases in AVO formation were observed in Huh-7 cells treated with **5d** and **5g**, respectively, compared to the control (26.42%) (Fig. 9). Similarly, HepG2 cell lines displayed a substantial increase in autophagic signal upon

treatment with **5d** and **5g**, with NFI percentage changes of 36.53% and 18.42%, respectively, compared to the control (16.69%) (Fig. 10). These findings collectively suggest that compounds **5d** and **5g** have the potential to induce autophagic cell death in liver cancer cells.

3.6. Molecular docking

Vascular Endothelial Growth Factor Receptor (VEGFR) plays an important role in the process of angiogenesis. VEGFR-2 is critical in angiogenesis, a mechanism required for tumor growth and metastasis. Its overexpression has been linked to several malignancies, including breast, cervical, non-small cell lung, hepatocellular, and renal carcinoma. As a result, blocking VEGFR-2 has emerged as a possible anticancer therapy strategy.^{49–56} To identify achievable VEGFR-2 inhibitors, the bioactive compounds **5d** and **5g** were docked against VEGFR-2 (Juxtamembrane and Kinase Domains) (PDB ID:4ASE). These compounds were compared to tivozanib (AV-951), a recognized



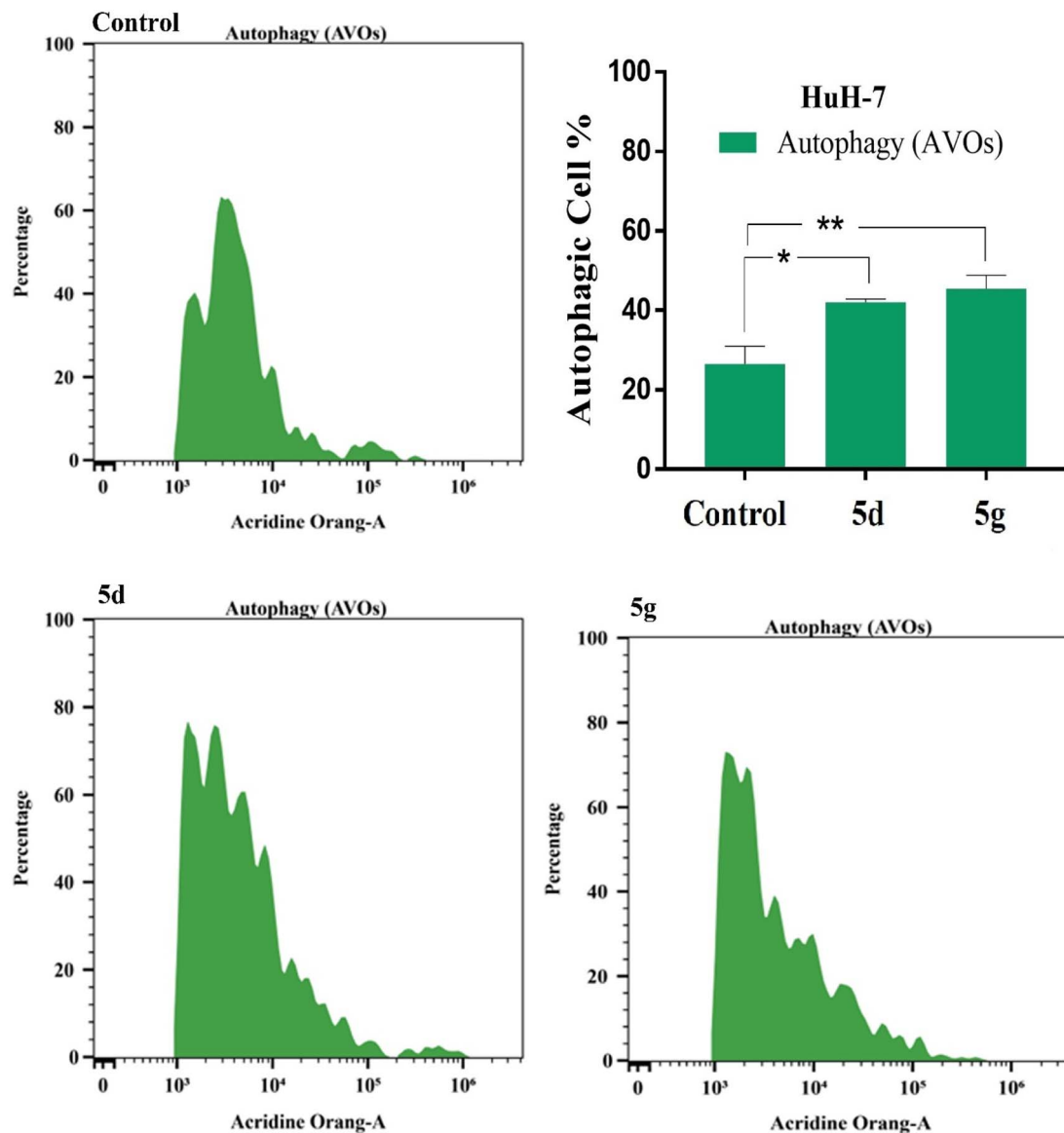


Fig. 9 Autophagic cell death evaluation of compounds 5d and 5g in Huh-7 cancer cells.

VEGFR-2 inhibitor. Table S1 (ESI[†]) displayed the binding affinity (kcal mol^{-1}) along with the specific interactions of each product. Tivozanib (AV-951), as VEGFR-2 inhibitor, was able to form strong interactions by forming three hydrogen bonds with CYS 919, GLU 885, and ASP 1046 with distances 1.96, 3.23 and 2.06 Å, respectively. Further, it formed a carbon H-bond interaction with CYS 1045 with distance 2.54 Å, with a binding energy of $-11.4 \text{ kcal mol}^{-1}$, as shown in Fig. 11. Pi interactions also observed with the key amino acids through planar aromatic rings, including π -alkyl (LEU 1035, ALA 866), π -sigma (LEU 840), and π - π T-shaped (PHE 1047). The products compounds 5d and 5g showed similar docking interactions, which pointed to tivozanib (AV-951) having a similar binding pocket.^{57–62} The product 5d demonstrated a binding mode with an affinity score of $-8.4 \text{ kcal mol}^{-1}$ against VEGFR-2 enzyme. It established three hydrogen bonds with ARG 1051 ($\text{C}=\text{O}_{\text{ester}}$), ASN 923 ($\text{C}=\text{O}_{\text{thiazolidinone}}$) and CYS 919 ($\text{C}=\text{O}_{\text{coumarin}}$), with distances measuring 2.76, 2.82 and 2.48 Å, respectively, as well as carbon hydrogen bond with the key amino acid residue GLY 841 through $\text{C}=\text{N}$ bond with distance measuring 2.37 Å (Fig. 12). In addition, it established several hydrophobic interactions, specifically π -cation (LYS 868), π -sulfur (CYS 1045), π -sigma (LEU 840), π - π stacked (PHE 1047) and π -alkyl (LEU 1035, ALA 866, VAL 848, 899, 916) interactions. In the same manner, product 5g had a binding energy of $-7.5 \text{ kcal mol}^{-1}$ towards VEGFR-2 enzyme. The compound had three hydrophobic π - π stacked (PHE 918) and π -alkyl (VAL 848, VAL 916, VAL 899, ALA 866 and CYS 1045) interactions. Additionally, it formed two hydrogen bonds and one carbon hydrogen bond with ARG 1051 (O_{ester}), CYS 919 ($\text{C}=\text{O}_{\text{coumarin}}$), and LEU 840 (pyrazole), with distances of 3.71, 2.50 and 4.94 Å, respectively. Moreover, it formed π -donor hydrogen bond with ASN 923 through benzyl

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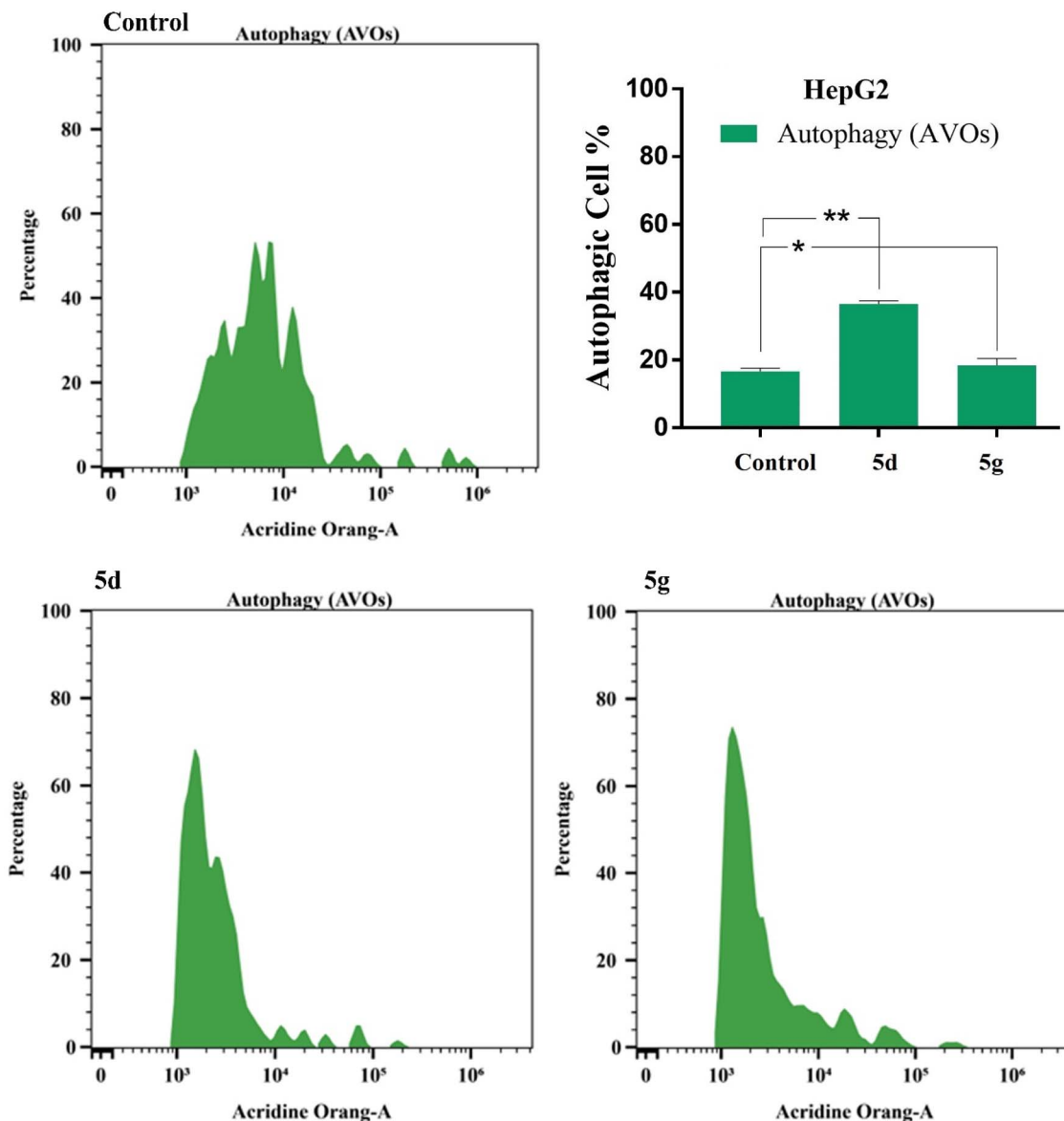


Fig. 10 Autophagic cell death evaluation of compounds 5d and 5g in HepG2 cancer cells.

ring with distance 3.08 Å (Fig. 13). The molecular docking study supported that both compounds **5d** and **5g** may be promising antiproliferative agents for further development as VEGFR-2 inhibitors. Their interactions with key amino acid residues of the receptors agreed with the cytotoxicity values, confirming their potential as anticancer agents (Fig. 14 and 15).

3.7. *In silico* ADMET prediction study

Analyzing the ADMET (Absorption, Distribution, Metabolism, Excretion, and Toxicity) properties of targeted drugs can offer valuable insights into the most appropriate therapeutic options. SwissADME, a freely accessible online tool, facilitated the efficient completion of this study.^{63–66} The selection of the most suitable drug for oral administration was guided by Lipinski's rule (molecular weight ≤ 500 , hydrogen bond acceptors ≤ 10 , hydrogen bond donors ≤ 5 , and $MlogP \leq 4.15$) and Veber's rule

(topological polar surface area $< 140 \text{ Å}^2$, rotatable bonds ≤ 10). Based on these criteria, the bioactive compounds **5d** and **5g** were found to comply with most rules, exhibiting only two violations (Table S2, ESI†). The synthesized compounds **5d** and **5g** were observed to fall within the optimal range (pink area) for six key parameters—lipophilicity, size, polarity, solubility, saturation, and flexibility as indicated by the bioavailability radar chart (Fig. 16). This suggested a favorable prediction for their oral bioavailability, with the exception of the saturation parameter.

The pharmacokinetic characteristics analysis for the synthesized compounds **5d** and **5g** are provided in Table S3 (ESI†) and Fig. 17. All products were away from both the white and yellow zones of the boiled-egg chart (Fig. 17). Therefore, these products can lead to a weak anticipated absorption in the gut and used to treat peripheral infections and are expected to





drug efflux transporter *P*-glycoprotein (*P*-gp), which is known to move drugs out of cells, may be a factor in drug tolerance. According to the SwissADME website's prognosis (red dots in

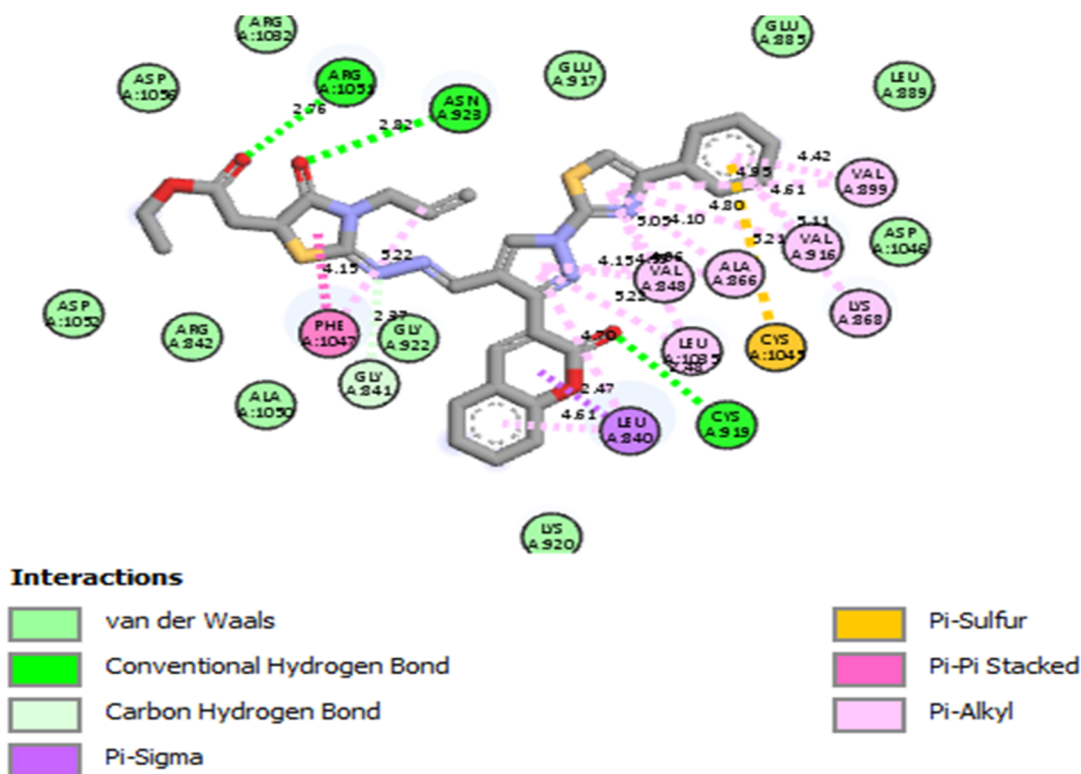


Fig. 12 The 2D interactions of compound 5d with VEGFR-2 receptor (PDB ID:4ASE).

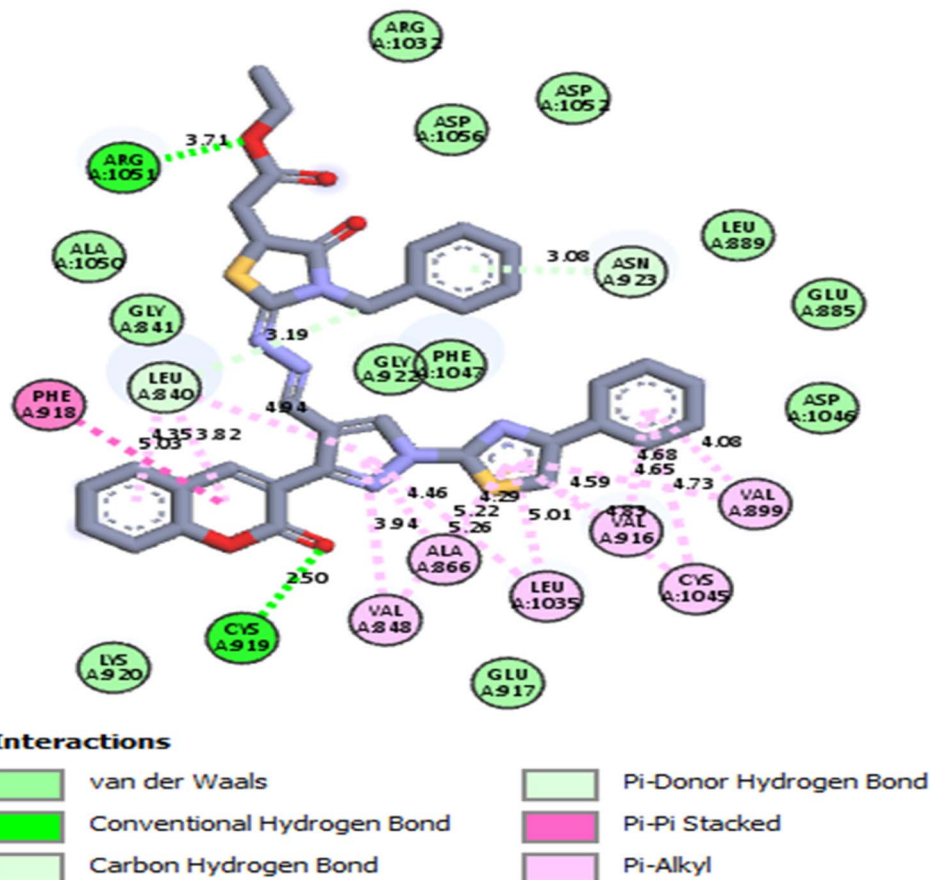


Fig. 13 The 2D interactions of compound 5g with VEGFR-2 receptor (PDB ID:4ASE).



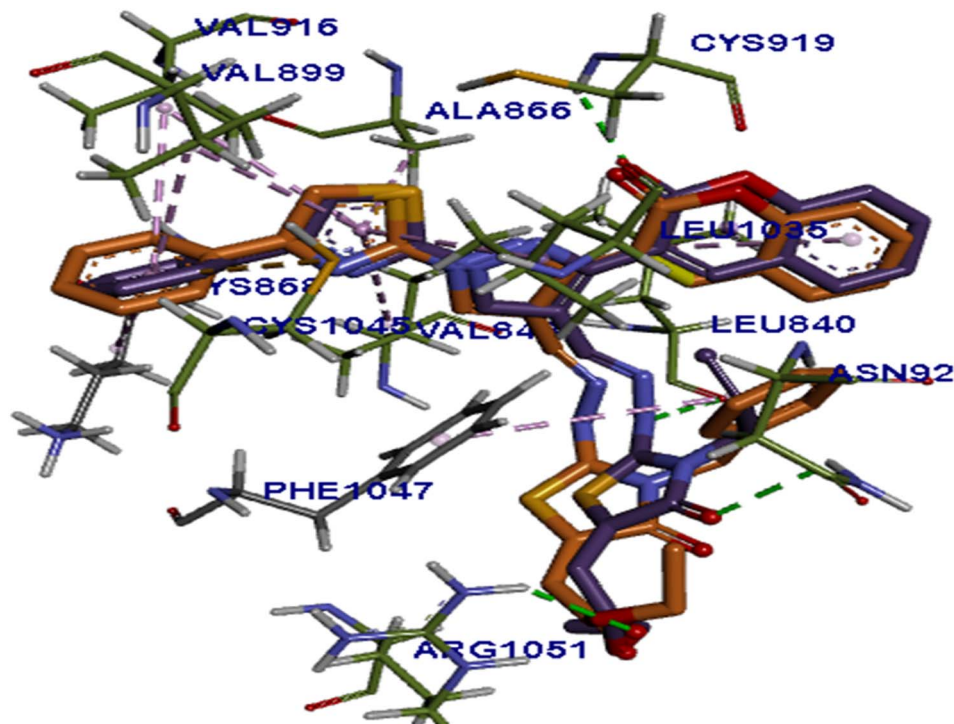


Fig. 14 Superimposition of compounds 5d and 5g inside the binding pocket of VEGFR-2 receptor (PDB ID:4ASE).

Fig. 17), these products under study are *P*-gp non-substrates, indicating that there is a small possibility of them effluxing out of the cell at maximal activity. The pharmacokinetic

properties of the synthesized compounds 5d and 5g are detailed in Table S3† and Fig. 17. All compounds were located outside the white and yellow zones of the boiled-egg chart (Fig. 17),

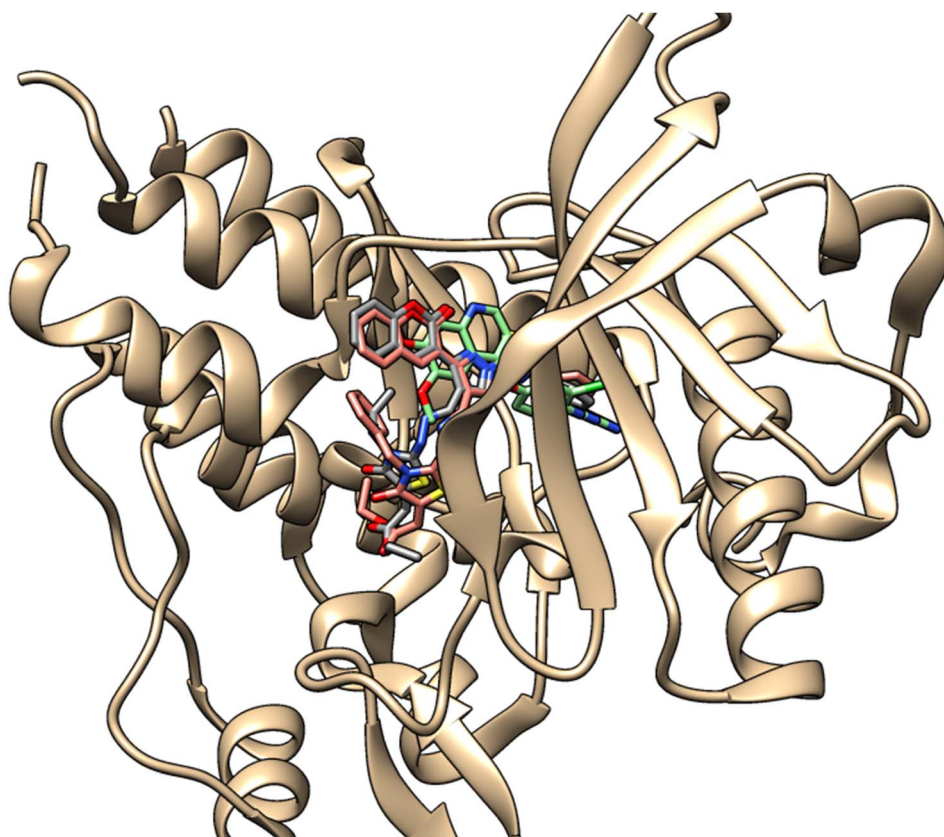


Fig. 15 Superimposition of compounds 5d and 5g with AV-951 inside the binding pocket of VEGFR-2 receptor (PDB ID:4ASE).





Fig. 16 Bioavailability radar charts for the synthesized compounds **5d**, **5g** and doxorubicin.

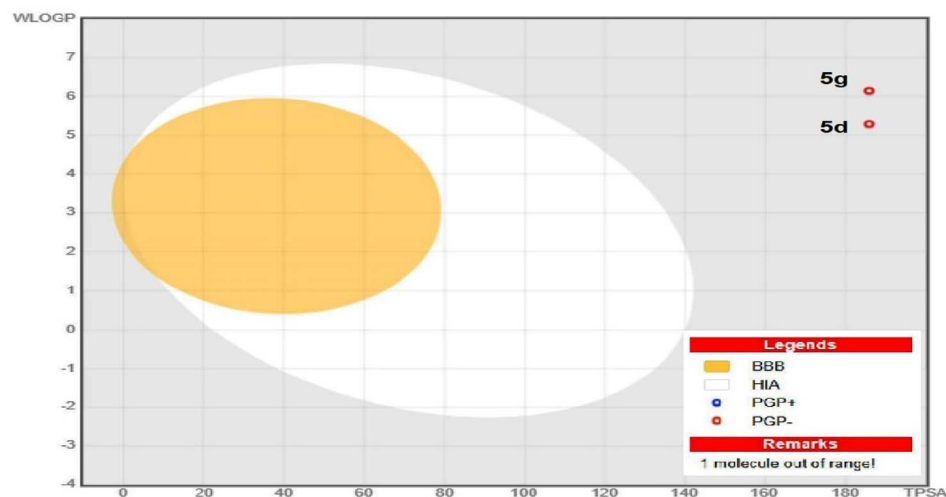


Fig. 17 A boiled-egg of the synthesized **5d**, **5g** and doxorubicin.

suggesting poor anticipated absorption in the gastrointestinal tract. As a result, these compounds are more suitable for treating peripheral infections and are likely to have a limited effect on the central nervous system. Additionally, all compounds exhibited a high bioavailability score of 0.17, which is unlikely to trigger a PAIN (pan-assay interference compounds) alert. The drug efflux transporter *P*-glycoprotein (*P*-gp), known for expelling drugs from cells, may contribute to drug resistance. However, according to the SwissADME prediction (represented by red dots in Fig. 17), the studied compounds are classified as *P*-gp non-substrates, indicating a low probability of being effluxed out of cells at peak activity levels.

The synthesized compounds **5d** and **5g** did not display toxicity characteristics associated with the inhibition of the human ether-a-go-go-related gene (hERG) (Table S4, ESI†), indicating no risk of cardiotoxicity or adverse cardiac effects. This is a critical factor in the clinical evaluation of drug candidates. Additionally, none of the tested compounds showed Ames toxicity, a significant finding in early drug discovery as it helps assess the potential genotoxicity of the target molecules. The acute oral toxicity calculations for all synthesized compounds yielded values ranging from 381 to 487 mg kg⁻¹,

placing them in the third toxicity category (500 mg kg⁻¹ < LD₅₀ ≤ 5000 mg kg⁻¹), which classifies them as relatively harmless. Furthermore, the carcinogenicity descriptor (CARC) values for these compounds ranged between 768 and 839 mg kg⁻¹ of body weight per day, suggesting they are likely non-carcinogenic and do not require further carcinogenicity testing.

4. Conclusion

In conclusions, a new set of ethyl 3-substituted-2-{4-oxo-2-((3-(2-oxo-2*H*-chromen-3-yl)-1-(4-phenylthiazol-2-yl)-1*H*-pyrazol-4-yl)methylene)hydrazineyl}thiazol-5(4*H*)-ylidene}acetate (**5a-h**) was synthesized and assessed for their *in vitro* antiproliferative activity against Huh-7 and HepG-2 liver carcinoma cell line. Most tested hybrids revealed significant antiproliferative activity. Among them, two products **5d** and **5g** presented potent antiproliferative activity against the examined cell lines. The cell cycle analysis showed evidence that **5d** and **5g** halted the Huh-7 and HepG2 at the G1 phase. Besides, they induced a significant increase in the populations of the cellular late apoptosis and necrotic cells with reducing the cellular viability. Additionally, compounds **5d** and **5g** demonstrated promising autophagic



induction potential compared control. Finally, both products demonstrated favorable binding affinities with the active sites of VEGFR-2 enzyme, as shown by molecular docking. ADMET profiling studies indicated that these bioactive compounds are likely to be safe and hold significant promise as anticancer agents. Based on the data, compounds **5d** and **5g** are promising inhibitor candidates for advancing innovative liver cancer treatments.

5. Experimental

5.1. The synthesis of 3-(2-oxo-2H-chromen-3-yl)-1-(4-phenylthiazol-2-yl)-1H-pyrazole-4-carboxaldehyde (3)

A mixture of 3-acetylcoumarin (**1**) (0.94 g, 5 mmol), thiosemicarbazide (0.46 g, 5 mmol) and phenacyl bromide (5 mmol) in absolute ethanol (20 ml) containing a few drops of glacial acetic acid, was heated under reflux for 4 h. The formed hydrazone **2** was filtered off and dried. A solution of DMF (15 ml) and POCl₃ (1.5 ml, 16 mmol) was allowed to stir at 0–5 °C for 15 minutes. After that the hydrazones **2** (4 mmol) was added at room temperature. The reaction mixture was warmed under stirring at 50 °C for 6 hours. The reaction mixture was poured into beakers containing crushed ice or ice-cold water and neutralized with sodium acetate. The formed solid was filtered off and washed several times with water. The crude product was dried and crystallized from acetic acid to afford the target aldehyde **3** as beige solid in 77% yield, mp 226–228 °C.⁴³ IR (KBr), ($\nu_{\max}$, cm⁻¹): 3093 (C–H_{arom}), 2987, 2899 (C–H_{aldehyde}), 1716 (C=O_{coumarin}), 1694 (C=O_{aldehyde}), 1605, 1538 (C=C), 1572 (C=N). ¹H-NMR (400 MHz, DMSO-d₆): δ 7.38–7.52 (m, 5H, Ph-H, H-6_{coumarin} and H-8_{coumarin}), 7.71 (t, 1H, J = 8.0 Hz, H-7_{coumarin}), 7.90 (d, 1H, J = 7.2 Hz, H-5_{coumarin}), 8.02 (d, 2H, J = 8.0 Hz, Ph-H), 8.10 (s, 1H, H-5_{thiazole}), 8.44 (s, 1H, H-4_{coumarin}), 9.37 (s, 1H, H-5_{pyrazole}), 9.98 (s, 1H, CHO). ¹³C-NMR (100 MHz, DMSO-d₆): δ 113.5 (C-5_{thiazole}), 119.2 (C-8_{coumarin}), 119.7 (C-3_{coumarin}), 120.8 (C-4_a_{coumarin}), 124.4 (C-4_{pyrazole}), 125.5 (C-6_{coumarin}), 126.5 (C-4_{phenyl}), 127.7 (C-2,6_{phenyl}), 128.6 (C-5_{coumarin}), 129.1 (C-1_{phenyl}), 129.5 (C-3,5_{phenyl}), 133.9 (C-7_{coumarin}), 146.9 (C-4_{coumarin}), 149.1 (C-5_{pyrazole}), 150.0 (C-4_{thiazole}), 152.3 (C-3_{pyrazole}), 154.2 (C-8_a_{coumarin}), 159.5 (C=O_{coumarin}), 166.9 (C-2_{thiazole}), 189.9 (C=O_{aldehyde}). MS (m/z , I %): 399 (M^+ , 17%). Anal. calcd for C₂₂H₁₃N₃O₃S (399.42): C, 66.16%; H, 3.28%; N, 10.52%; S, 8.03%. Found: C, 66.03%; H, 3.16%; N, 10.35%; S, 7.89%.

5.1.1. General procedure for the synthesis of products 5a–h. A mixture of 3-(2-oxo-2H-chromen-3-yl)-1-(4-phenylthiazol-2-yl)-1H-pyrazole-4-carboxaldehyde (**3**) (0.39 g, 1.0 mmol), thiosemicarbazide derivative **4a–h** (1 mmol) and diethyl acetylene dicarboxylate (1.1 mmol) in acetic acid (20 ml) was heated under reflux for 5 h. The reaction mixtures were cooled to room temperature. The formed solids were filtered off and crystallized from ethanol to afford the target compounds.

5.1.1.1. Ethyl 2-{4-oxo-2-(2-((3-(2-oxo-2H-chromen-3-yl)-1-(4-phenylthiazol-2-yl)-1H-pyrazol-4-yl)methylene)hydrazineyl)thiazol-5(4H)-ylidene}acetate (5a). Yellow solid in 67% yield, mp > 300 °C. IR (KBr), ($\nu_{\max}$, cm⁻¹): 3116 (NH), 3002 (C–H_{arom}), 2946, 2938 (C–H_{aliph}), 1726 (C=O_{thiazolidinone}), 1716 (C=O_{coumarin}), 1691

(C=O_{ester}), 1644 (CH=N_{exocyclic}), 1609, 1578, 1544 (C=N, C=C). ¹H-NMR (400 MHz, DMSO-d₆): δ 1.25 (t, 3H, J = 6.8 Hz, CH₃), 4.19 (q, 2H, J = 6.8 Hz, OCH₂), 6.54 (s, 1H, =CH_{exocyclic}), 7.40 (t, 2H, J = 6.8 Hz, Ph-H), 7.47–7.52 (m, 3H, Ph-H, H-6_{coumarin} and H-8_{coumarin}), 7.67 (t, 1H, J = 7.2 Hz, H-7_{coumarin}), 7.86 (d, 1H, J = 7.2 Hz, H-5_{coumarin}), 8.00–8.02 (m, 3H, Ph-H and H-5_{thiazole}), 8.36 (s, 1H, CH=N_{exocyclic}), 8.54 (s, 1H, H-4_{coumarin}), 9.09 (s, 1H, H-5_{pyrazole}), 12.69 (brs, 1H, NH). ¹³C-NMR (100 MHz, DMSO-d₆): δ 14.5 (CH₃), 61.6 (CH₂), 112.7 (C-5_{thiazole}), 117.2 (C-8_{coumarin}), 119.0 (C-3_{coumarin}), 119.8 (C-4_a_{coumarin}), 120.7 (C-4_{pyrazole}), 122.5 (=CH_{exocyclic}), 125.1 (C-4_{phenyl}), 126.4 (C-2,6_{phenyl}), 128.9 (C-6_{coumarin}), 129.3 (C-3,5_{phenyl}), 129.9 (C-5_{coumarin}), 130.4 (C-1_{phenyl}), 133.8 (C-7_{coumarin}), 137.4 (CH=N_{exocyclic}), 143.2 (C-5_{thiazolidinone}), 143.6 (C-4_{coumarin}), 148.6 (C-5_{pyrazole}), 151.3 (C-4_{thiazole}), 152.2 (C-3_{pyrazole}), 154.2 (C-8_a_{coumarin}), 159.2 (C=O_{thiazolidinone}), 159.6 (C=O_{coumarin}), 160.2 (C=O_{ester}), 165.5 (C-2_{thiazole}), 166.2 (C-2_{thiazolidinone}). MS (m/z , I %): 596 (M^+ , 25%). Anal. calcd for C₂₉H₂₀N₆O₅S₂ (596.64): C, 58.38%, H, 3.38%, N, 14.09%, S, 10.75%. Found: C, 58.30%, H, 3.24%, N, 14.01%, S, 10.66%.

Data availability

All data are available in the manuscript and ESI.†

Author contributions

Data curation: TEA and AAS. Formal analysis: AAS and AKA. Investigation: TEA, WAB, AKA, MYA and SEIE. Writing – original draft: TEA, MAA and WAB. Conceptualization: TEA, MAA and WAB. Supervision: TEA. Resources: TEA, WAB and SEIE. Software: TEA and SEIE. Methodology: WAB, AKA and SEIE. Writing – review & editing: TEA, MAA, WAB and SEIE. All authors read and approved the final version of the manuscript.

Conflicts of interest

Authors declare that they have no conflict of interest.

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