


Cite this: *RSC Adv.*, 2020, 10, 37116

Received 16th May 2020
Accepted 18th September 2020

DOI: 10.1039/d0ra04388e

rsc.li/rsc-advances

Biomedical applications of dendritic fibrous nanosilica (DFNS): recent progress and challenges

Mina Shaban^{ab} and Mohammad Hasanzadeh *^a

Dendritic fibrous nanosilica (DFNS), with multi-component and hierarchically complex structures, has recently been receiving significant attention in various fields of nano-biomedicine. DFNS is an emerging class of mesoporous nanoparticles that has attracted great interest due to unique structures such as open three-dimensional dendritic superstructures with large pore channels and highly accessible internal surface areas. This overview aims to study the application of DFNS towards biomedical investigations. This review is divided into four main sections. Sections 1–3 are related to the synthesis and characterization of DFNS. The biomedical potential of DFNS, such as cell therapy, gene therapy, immune therapy, drug delivery, imaging, photothermal therapy, bioanalysis, biocatalysis, and tissue engineering, is discussed based on advantages and limitations. Finally, the perspectives and challenges in terms of controlled synthesis and potential nano-biomedical applications towards future studies are discussed.

1. Introduction

Scientists have been continuously trying to find new ways to serve the needs of medical science. Nanotechnology provides the ability to manipulate the structure of materials on the atomic and molecular scale to produce new materials with new features that can afford solutions to some medical challenges.¹ Recently, the development of structures of mesoporous nanomaterials has attracted attention in biomedical fields such as diagnostics, treatment, drug delivery and bioanalysis.^{2–4} Numerous nano-biomedical studies have been conducted on nanocomposites such as liposomes, polymers, nanoparticles, dendrimers, nano-micelles, nanogels, exosomes, magnetic nanoparticles (MNPs), mesoporous silica nanoparticles (MSNs), carbon nanotubes, and gold nanoparticles.^{5–7} Among these, a new generation of mesoporous nanomaterials, such as dendritic fibrous silica nanoparticles (DFNS), has received biomedical attention due to their special structural properties and intrinsic silica features. These features include easy synthesis, adjustable and uniform pore sizes, tunable particle sizes, unique large surface areas, great permeability, high-level pore volumes, low density, high thermal stability, great mechanical stability, highly selective functionalization of the pore surface or particle surface, low range of toxicity, and perfect biocompatibility.^{8,9} There are several types of mesoporous nanoparticles, including dendritic fibrous nanosilica (DFNS) particles, MCM-41 (Mobil

Composite Material), SBA-15 (Santa Barbara Amorphous), Stöber silica, mesosphere silica nanoparticles (MSNP), all of which have some advantages and disadvantages in nano-medicine application. Table 1 compares the physicochemical properties and cytotoxicities of different types of mesoporous nanoparticles.^{10,11} DFNS are fibrous spheres with higher temperature stability and smaller pore size, pore volume and size as compared to other mesoporous nanoparticles; cytotoxicity studies have shown that they have low toxicity.^{11–13} These properties allow them to be suitable in various nano-medical fields such as therapy, pharmacy, bioanalysis, imaging, biocatalysis and biotechnology. In recent years, several types of DFNS have been introduced for nano-biomedical applications due to their special structures. These include dendritic mesoporous silica nanoparticles (DMSNs),^{14–17} fibrous mesoporous silica microspheres (FMSMs),¹⁸ wrinkled silica nanoparticles (WSNs),^{19,20} wrinkled silica nanospheres (WSNs),²⁰ wrinkled mesoporous silica (WMS),^{21–23} wrinkled mesoporous silica nanoparticles (WMSNs),^{21,24} radial-like mesoporous silica (RMS),^{25,26} fibrous silicon dioxide spheres (FSSs),²⁷ fibrous silica nanoparticles or nanospheres (FSNs),^{28,29} dendrimer-like mesoporous silica nanoparticles with hierarchical pores (HPSNs),³⁰ hierarchical and radial mesoporous (HRM),^{31,32} hierarchically structured spherical mesoporous nanofibrous (HSMNFs),³³ wrinkle-structured periodic mesoporous organosilica (PMO),^{34,35} and mesostructured silica nanoparticles (MSNs).³⁶ The history of DFNS exploration indicates that novel DFNS or KCC-1 was introduced by Polshettiwar *et al.* at KAUST (King Abdullah University of Science and Technology) Catalysis Center. They named it KCC-1 according to the acronym of their institution.³⁷ Compared to other mesoporous silica nanocarriers,

^aPharmaceutical Analysis Research Center, Tabriz University of Medical Sciences, Tabriz, Iran. E-mail: hasanzadehm@tbzmed.ac.ir

^bFood and Drug Safety Research Center, Tabriz University of Medical Sciences, Tabriz, Iran



dendritic fibrous silica nanoparticles (DFNSs) with pore size ranging from 2 to 30 nm are perfect candidates for drug delivery and biomedical applications.³⁶ The fibrous surface of DFNS allows them to be easily loaded with several ranges of organic groups, organometallic complex ionic liquids, polymers, peptides, enzymes, DNA, genes, drugs, *etc.*¹⁰ They can also interact with metal nanoparticles, bimetallic nanoparticles, as well as with single atoms of metals, quantum dots, metal oxides and hydroxides. These unique properties of DFNS allow them to be useful in a variety of applications such as catalysis,^{38,39} biomedical fields,⁴⁰ analysis,⁴¹ industrial fields⁴² as adsorbents for water and air treatment,^{43,44} energy storage,¹⁰ *etc.* Therefore, in this review, we have probed the spherical shape of DFNS with tunable particle sizes ranging from 14 to 1100 nm, their fibrous structure, which provides better accessibility to the DFNS surface, and their open structure and accessibility, which also permit the high loading of active molecules that make them suitable for use in biomedical fields. The toxicity studies of DFNS were conducted in animals but clinical trials have not been done to date.⁴⁵ We also introduce DFNS as new nanocomposites with expanded biomedical applications. There are four main sections. Sections 1–3 are related to the synthesis and characterization of DFNS. In Section 4, the biomedical potential of DFNS, such as cell therapy, gene therapy, immune therapy, drug delivery, imaging, photothermal therapy, bioanalysis, biocatalysts, and tissue engineering, is discussed based on the advantages and limitations. Finally, perspectives and challenges in terms of the controlled synthesis and potential nano-biomedical applications towards future studies are discussed.

2.1. Hydrothermal synthesis

solutions at high vapor pressures, mostly used for the fabrication of silicone monomers on the surface of NPs. On the other hand, this method is used for growing crystals of many different materials such as quartz, malachite, *etc.* This technique has also been used to grow dislocation-free single crystal particles, and the grains formed in this process could have better crystallinity than those from other processes.⁴⁶ The process was based on minerals dissolving at high temperatures and pressures in water and subsequently crystallizing from dissolved materials.⁴¹ Polshettiwar *et al.* succeeded in synthesizing DFNS in their laboratory (KAUST Catalysis Center (KCC-1)), which were named KCC-1. Fig. 1 shows the stages of the synthesis of DFNS. KCC-1 was prepared using the microwave-assisted hydrothermal technique in a short time. Firstly, 0.012 mol solution of tetraethyl orthosilicate (TEOS) was dissolved in 30 ml solution of cyclohexane and 1.5 ml of pentanol. Then, another solution containing 0.0026 mol of cetyl pyridinium bromide (CPB) and 0.01 mol of urea in 30 ml of water was added to the first solution. This mixture was stirred for 30 min at 25 °C, and was placed in a teflon-sealed microwave (MW) reactor. The reaction mixture was exposed to MW irradiation (400 W maximum power) at 120 °C for 240 min. When the reaction was complete, the mixture was allowed to cool at room temperature and the silica formed was separated by centrifugation, then washed with distilled water and acetone, and air-dried for 24 h. The obtained material was calcined at 550 °C for 360 min in air.³⁷ Another synthesis protocol was introduced in 2019 by Polshettiwar *et al.*, who showed that the previous protocol had some limitations. According to them, increasing the size above 800 nm *via* the previous method is not possible, and some materials such as metal oxides and carbon 30 having DFNS morphology cannot be synthesized using this method so some experimental conditions such as co-surfactant and reaction conditions should be modified.⁴⁷

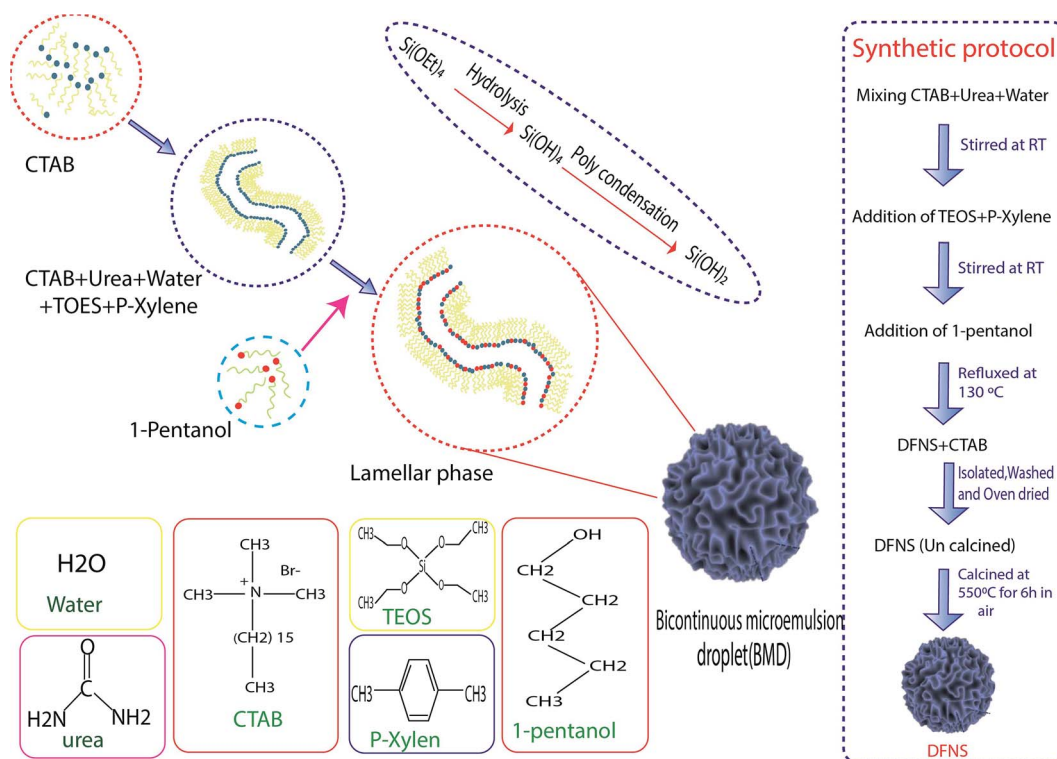


Fig. 1 Graphic illustration of DFNS synthesis.

2.2. Microemulsions

Microemulsions contain isotropic liquid mixtures of water, oil, surfactant, and co-surfactant, which were first reported by Hoar *et al.* In microemulsions, the surfactant molecules align themselves to form spherical aggregates in the continuous phase. This method, including water-in-oil (w/o) microemulsions, or reverse micelles, was used for the preparation of ultrafine silica NPs. The silica NPs were developed inside the water droplets of a w/o microencapsulation, catalyzed by the base-assisted hydrolysis of silicon

alkoxide. The major disadvantage reported in this approach is the high cost and difficulty in the removal of surfactants in the final products. Jin-Lee *et al.* showed that in the first stage, cetyl pyridinium bromide and urea were dissolved in water, then cyclohexane and iso-propanol were added to the solution. Next, TEOS was added dropwise to the mixed solution. After that, the obtained solution was heated to 70 °C for 16 h. In the final stage, the obtained NPs were washed with ethanol until the surfactants were removed from the surface of the NPs.³²

Table 2 A comparison of the different conditions used for the preparation of DFSN

Surfactant	Si-Precursor	Solvent	Cosurfactant	Base	Particle size (nm)	Reaction condition	Ref.
CTAB or CPB	TEOS ^e	Water : cyclohexane	1-Pentanol	Urea	90–120	120 °C, 4 h	37 and 122
CPB	TEOS	Water : cyclohexane	Isopropanol or <i>n</i> -butanol	Urea	50–500	70 °C, 16–20 h	32 and 49
CTAB ^a	TEOS	Water : toluene	<i>n</i> -Butanol	Urea	50–500	120 °C, 4 h	136
CTAT ^b	TEOS	Water	None	Organic amines	<200	80 °C, 2 h	48
CTAT	TEOS	Water	[BMIM] OTF ^g	Triethanolamine	40–100	80 °C, 3 h	50
CPB ^c	TEOS	Water : cyclohexane	Urea	Urea	200–500	120 °C, 20 h	52
CTAC ^d	TEOS	Water : 1-octadecene	Triethanolamine	Triethanolamine	180	60 °C, 12 h	137
CTAB	TEOS	Water : ethylether	Ethanol	Aqueous ammonia	100–320	20 °C, 4 h	138
CTAB	TEOS	Water : octane and styrene	Ethanol	Lysine, AIBA ^f	40–50	70 °C, 20 h	51
CPB	TEOS	Water/cyclohexane	1-Pentanol	Urea	14.78	120 °C, 6 h	53

^a Cetyltrimethyl ammonium bromide. ^b Cetyltrimethyl ammonium tosylate. ^c Cetylpyridinium bromide. ^d Cetyltrimethylammonium chloride.

^e Tetra ethyl ortho silicate. ^f 2,2'-Azobis(2-methylpropionamidine) dihydrochloride. ^g 1-Butyl-3-methylimidazolium trifluoro-methanesulfonate.



2.3. Sol-gel method

The sol-gel method is a wet-chemical technique that is widely explored for the synthesis of silica NPs with uniform size.^{40–42,47} In the coating process, it involves the hydrolysis and condensation of TEOS or inorganic salts such as sodium silicate (Na_2SiO_3) in the presence of a mineral acid or base as the catalyst, or under alkaline conditions in ethanol solution.⁴³ Zhang *et al.* illustrated the preparation DFNS by the sol-gel method. In the first stage, tetraethyl orthosilicate (TEOS) and cetyltrimethylammonium tosylate (CTA^+) were dissolved in water. Then, the reaction was carried out at 80 °C for 2 h, then the obtained NPs were filtered and dried.⁴⁸

2.4. Optimized conditions for the synthesis of DFNS

Several studies have been done for optimizing the synthesis protocol of DFNS. Table 2 compares the different conditions used for the preparation of DFNS with various sizes and textural properties by changing some reaction parameters. It has been found that precise control over the size, shape, and surface properties could be achieved by varying the reaction parameters, such as pH, temperature, and surfactant chemical nature, and controlling the condensation of the silica source. Table 2 also shows the type of surfactant and solvent used in the chemical reaction; the co-surfactant and reaction conditions are important parameters that can impact the size of the DFNS and inter-wrinkle distances by controlling them to process different sizes of DFNS. The particle size and inter-wrinkle distances are two main characterization parameters of DFNS for obtaining suitable types in biomedical studies. For example, Moon and Lee succeeded in synthesizing DFNS with smaller size (50–500 nm) by using isopropanol or n-butanol as the co-surfactant and increasing the time of reaction in the microemulsion model.^{32,49} Another parameter illustrated the use of organic amine (instead of urea) and cetyltrimethylammonium (CTA⁺) tosylate and bromide as templates to make DFNS with size <200 nm.⁴⁸ Following this study, small-sized DFNS (<50 nm) was also synthesized by using nonionic surfactants like 1-butyl-3-methylimidazolium trifluoromethane sulfonate ([BMIM] OTF), or pluronic F127, which can stop the growth size of silica particles.⁵⁰ Experimental data showed that using a cationic base such as lysine instead of urea and ethanol as co-surfactants made very small DFNS with size of 14 nm.⁵¹ On the other hand, for making large-sized DFNS, it is better to use cationic surfactants like CPB and non-ionic co-surfactants like urea.⁵² This study showed that increasing the temperature and reaction time could produce DFNS with small size of approximately 14 nm.⁵³

3. Characterization techniques

Different techniques have been the main methods for the characterization of DNFs. For instance, to verify the synthesis of silica NP-based mesoporous materials, we can use Fourier transform infrared spectroscopy (FTIR),⁵⁴ X-ray diffraction (XRD) analysis,⁵⁵ transmission electron microscopy (TEM),⁵⁶ scanning electron microscopy (SEM),³⁷ atomic force microscopy

(AFM), Brunauer–Emmett–Teller (BET) analysis,⁵⁷ dynamic light scattering (DLS),³⁷ thermogravimetric analysis (TGA),⁵⁰ X-ray photoelectron spectroscopy (XPS),⁵⁸ energy-dispersive X-ray fluorescence spectroscopy (ED-XRF),⁵³ vibrating sample magnetometer (VSM),^{59,60} solid-state ¹³C CPMAS,⁶¹ and ²⁹Si CPMAS NMR spectroscopy.^{53,62} The functional groups of nanocomposites can be identified by FTIR, and XRD (used for the determination of the crystalline structure and the phase purity of modified silica NPs). SEM, DLS, TEM, and AFM techniques are used to check the size, size distribution and morphology of the NPS. By utilizing more sophisticated microscopy techniques, surface images can be constructed *via* a physical probe that scans the specimen in order to view the chemical composition near the surface of the nanocomposites. TGA is used for measuring the thermal stability of the monomers and coating the surface of the nanomaterial, the moisture content and the possible decomposition of volatile materials in the nanostructured catalysts. XPS is a surface-sensitive quantitative spectroscopy technique that measures the elemental composition in the range of parts per thousand, and also helps to determine the oxidation state and chemical connectivity of the metal present in the nanomaterial. The metal present in the heterogeneous catalyst can be detected *via* ED-XRF, and the magnetic proration of NPs is assessed by VSM. Solid-state NMR spectroscopy is used to verify the covalent grafting of organic functionalities onto silica NP surfaces, and BET analysis is used to investigate the surface, pore size, and pore volume of DFNS.⁶³

4. Functionalization of DFNS for tumor targeting purposes

The surface of DFNS is a main factor that directly affects the interaction of the DFNS with tumor cells. DFNS has been applied in the biomedical and pharmaceutical fields in target therapy and drug delivery. The DFNS have special shapes that allow them to easily interact with damaged cells. For instance, the DFNS surfaces have large amounts of (Si-OH) groups that allow covalent conjugation with various types of functional groups, including carboxylates, amines, amines/phosphonates, polyethylene glycol, octadecyl, and carboxylate/octadecyl groups.⁶⁴ The solubility of the drug in biological environments is also an important factor in drug delivery. The shape of DFNS allows them to release highly water-soluble anticancer drugs in aqueous biological media. On the other hand, when the drugs have hydrophobic properties, the DFNS provide control over the drug release in aqueous surroundings, releasing the drug in contact with hydrophobic media. The functional groups on the surface of DFNS can be made to specifically target the DFNS with targeting ligands to mediate the internalization of DFNS into the cells. The greatest advantage of DFNS, other than being mesoporous nanoparticles, is the high loading capacity of the drug, which makes them suitable for tumor therapy. The blood circulation time and the ability of the DFNS to overcome biological barriers make them effective for targeting. Some properties such as size, shape, surface charge, and the composition of the DFNS will affect the penetration, tissue accumulation and

blood clearance (ABC), which may be activated after repeated administration of PEGylated particles, needs to be carefully controlled. The lipid coating modification of DFNS is another strategy for reducing the toxicity, which can also improve the pharmacokinetic (PK) properties of NPs attributable to the reduced quantity of non-specific protein bindings (plasma protein opsonization) and prolonged systemic circulation time.

The morphology of DFNS can also directly influence their biosafety. Studies have shown that the endocytosis of DFNS with spherical shape is relatively higher than the cylindrical MSNs with the same surface area. However, some other studies have shown that the level of endocytosis is highly dependent on the type of cells. The aspect ratio is another determinant for the biosafety of DFNS, in which the greater aspect ratio might lead to enhanced uptake and improved biological activity. The higher surface area is a suitable physical property for the encapsulation of cargo in MSNs, even though it may have adverse effects on the safety of DFNS through the increase in the generation of reactive oxygen species (ROS) as DFNS facilitates ROS generation *via* their exposed surface areas. Thus, the higher surface area may result in a higher toxicity of DFNS. In this context, spheres and short rod DFNS can cause greater mitochondrial damage, resulting in an intensified generation of intracellular ROS. Collectively, while various observations confirmed the pivotal role of the size, shape, surface morphologies, and structure on the biosafety of MSNs, more investigations need to be undertaken to address the precise effects of these factors on the biosafety of DFNS.

The safety of the nanoparticles has led to the need for the field of therapy to be characterized according to *in vivo* protocols based on the nanomaterials' absorption, distribution, metabolism, excretion, and toxicity properties; however, few studies have discussed their *in vivo* biodistribution and toxicity after oral administration. Yu *et al.* found that the appearance of DFNS, such as their porosity and surface characteristics, may affect the acute toxicity in immune-competent mice and found that there was *in vivo* toxicity after injection. Also, by increasing the dose of MSNs' (aspect ratio 1, 2, 8) at approximately 30–65 mg kg⁻¹, there was mechanical clogging of the vasculature followed by organ failure. These findings indicate that the MSNs' with larger hydrodynamic size have the maximum tolerated dose (MTD).⁶⁵ In another study done by Soleymani *et al.*, the cytotoxicity of Tb@KCC-1-NH₂-FA on colorectal cancer cells was evaluated by MTT assay, which confirmed their biocompatible nature.⁶⁶ This analysis showed that DFNS had no cytotoxic effects on colon cells. The next aspect to consider about DFNS is that their toxicity properties depend on the surface functionalization and amount injected into the body. To date, DFNS has not been evaluated in clinical trials; they have only been tested in different animal models.⁶⁷ All of the articles considered in this review confirmed the safety of DFNS in animal models.

The surface properties of DFNS are another factor that can directly affect their biosafety. Although the positive charge of the DFNS surface causes an enhanced uptake, it can definitely increase the cytotoxicity and biological damage as compared to anionic/neutral particles. Some modifications such as PEGylation of DFNS may prevent their aggregation and opsonization and hence, decrease their immune clearance and cytotoxicity while prolonging their systemic circulation period. The accelerated

Cell therapy involves the injection, grafting, or implanting of viable cells into a patient in order to achieve a therapeutic

The treatment of cancers using different types of DFNS armed with targeting moieties (*e.g.*, small molecules, peptides, proteins, antibodies, or Ap) seems to be one of the most promising cancer therapy modalities that might be practically

This journal is © The Royal Society of Chemistry 2020

led to DNA delivery, RNA delivery and some biological drugs such as IFN- α . In a new study, DFNS with an average particle size of 174 ± 17 NM was prepared, then the surface was functionalized with aldehyde groups. Fig. 2 shows the preparation of (DFNS-CHO). The release of bovine serum albumin (BSA) on the surface of (DFNS-CHO) was investigated at different pHs. Furthermore, the *in vitro* cytotoxicity and cell uptake of the DFNS-CHO nanoparticles were studied using RAW264.7 cells as the cellular system. The results indicated that the nanocarriers had a highly loaded capacity of approximately $136 \mu\text{g mg}^{-1}$. Moreover, in acidic conditions, BSA was released from the DFNS-CHO nanoparticles with no cytotoxicity. It was concluded that DFNS-CHO nanoparticles would be acceptable as novel nanocarriers for the pH-responsive delivery of protein drugs.⁹⁶

One of the main injected drugs for type 1 and type 2 diabetes is insulin; its use is sometimes quite painful for patients. Finding an oral type of insulin is important in pharmaceutical science. Kleitz *et al.* found a new oral drug delivery route for insulin *via* DFNS. In the first stage, DFNS with positive charges in various pore sizes have been designed and negatively charged insulin was loaded on the surface of DFNS. The surfaces of the nanoparticles were then modified with pH-sensitive protein (B-lactoglobulin). The data showed that with this method, insulin was release at pH 7.4.⁹⁷

6.4.3 Ayurvedic drug delivery using DFNS. The ayurvedic system of medicine is one of the oldest systems of traditional medicine. Many people prefer to use ayurvedic drugs instead of chemical drugs due to chemical drugs having many side effects. It is always time-consuming to find a new method for the delivery of ayurvedic drugs.⁹⁸ Nanoparticles such as DFNS can encapsulate enormous biomaterials and drugs in their structural pores and deliver them to the target. In this research

project, DFNS-based multifunctional nanoparticles were reported as a drug delivery system. Firstly DFNS was prepared by a previously reported method with slight modification;⁹⁹ the average size was 63 nm, and one of the optimal cancer-targeting ligands with a strong chemical attraction to folate receptors ($K_d = 10^{-10}$ M) is folic acid. In this project, folic acid (FA) was conjugated to the surface of DFNS for selectively targeting cancer cells, then biocompatible calcium hydroxide and curcumin (Cur) were immobilized on (DFNS-FA) (Fig. 3). Finally, the anti-cancer efficacy of obtaining the nanocomposite (Cur-Ca@DFNS-FA) was investigated by a battery of *in vitro* and *in vivo* experiments. The results show that the Cur-Ca@DFNS-FA was diffuse in aqueous solution. Cur was control-released by pH and efficiently targeted the MCF-7 cells of breast cancer. Cur-Ca@DFNS-FA effectively inhibited cell proliferation, enhanced intracellular ROS generation, reduced mitochondrial membrane potential, and increased cell cycle retardation at the G2/M phase, leading to greater apoptosis in MCF-7 as compared to free Cur. On the other hand, the western blotting analysis illustrated that Cur-Ca@DFNS-FA was more effective than free Cur through the suppression of PI3K/AKT/mTOR and Wnt/ β -catenin signaling, and the activation of the mitochondria-mediated apoptosis pathway. *In vivo* studies indicated that Ca@DFNS-FA exhibited good biocompatibility, the Cur concentration in blood serum and tumor tissues increased after 1 h when Cur was encapsulated in Ca@DFNS-FA and finally, all of the data showed that Cur-Ca@DFNS-FA was the best candidate against breast cancer.¹⁰⁰

In other research, the scientists focused on the anticancer effect of folic acid conjunction on both surfaces of mesopore nanoparticles, including KCC-1 and MCM. Fig. 4 shows the anticancer mechanism of DFNS in human liver carcinoma

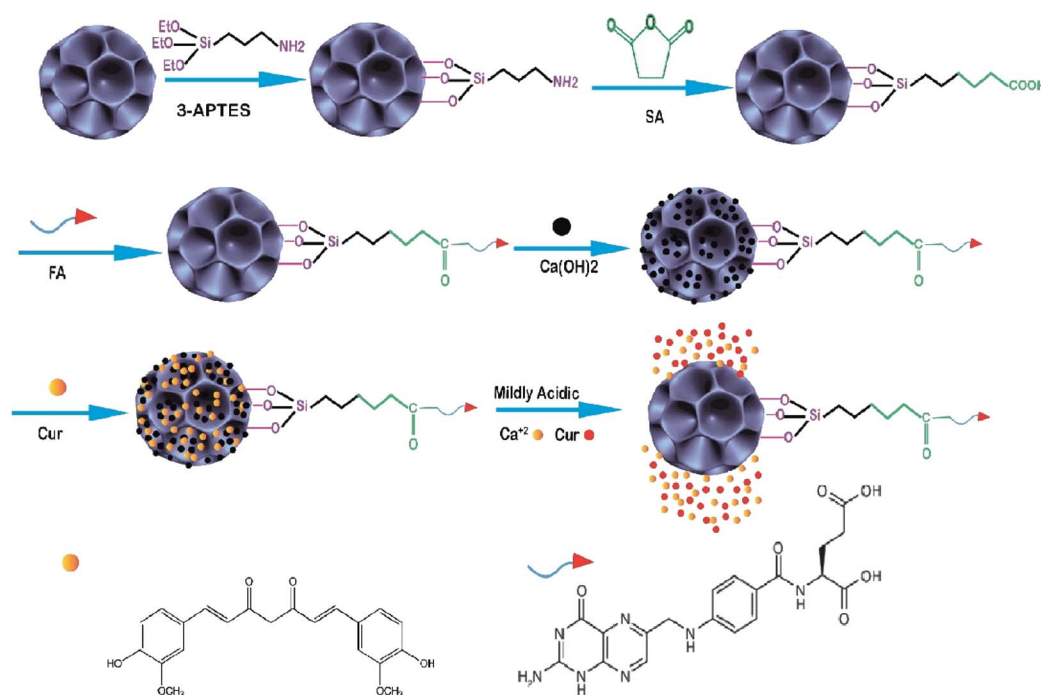


Fig. 3 Schematic illustration of the fabrication of Cur-Ca@DFNS-FA and the pH-stimuli-responsive release in tumors.¹⁰⁰

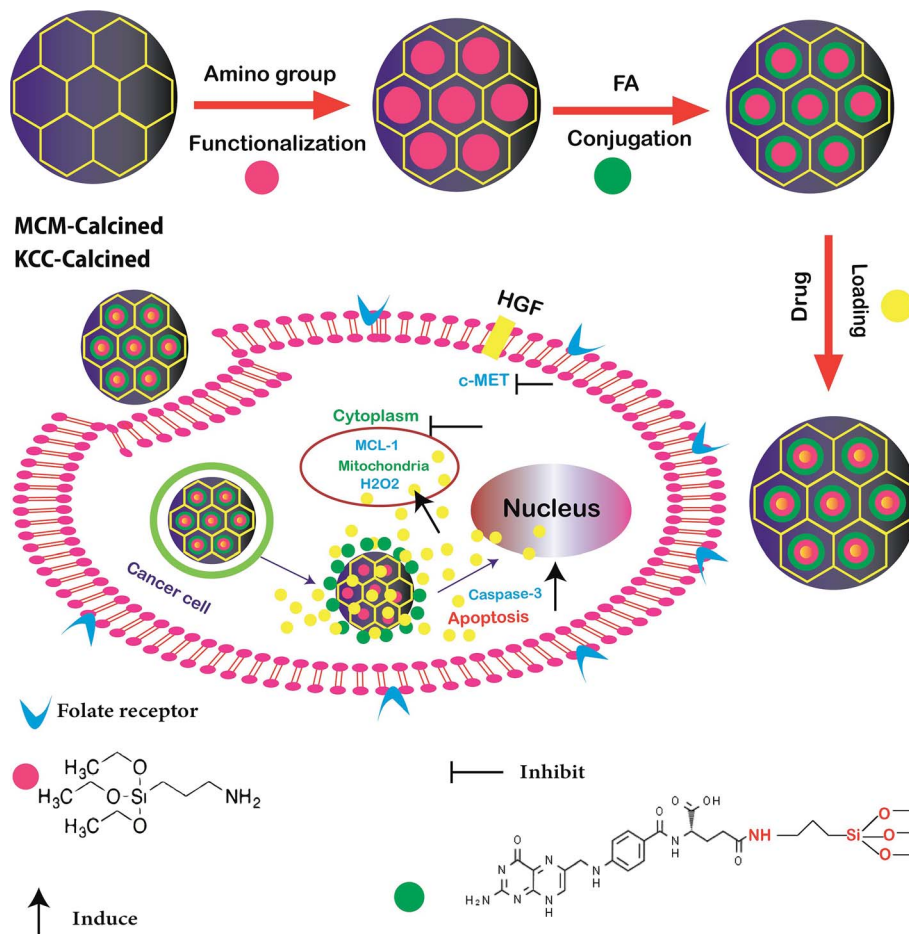


Fig. 4 Schematic presentation of the prepared nanosystem from preparation, internalization, and anticancer mechanism of action DFNS in human liver carcinoma (HepG2) cells.¹⁰¹

(HepG2) in this study; various factors, such as the surface of nanoparticles, timing, and size, which may influence the cellular uptake of NPs by cancer cells were investigated. The kind of surface modification (as-synthesized MSNs, amino-modified MSNs, and FA-conjugated MSNs), and incubation time (4, 12, 24, and 48 h) were investigated on two types of NPS including KCC-1 and MCM. All of the results show that the FA-conjugated KCC-1 has promise for controlling the release of curcumin, quercetin, and colchicine as compared to MCM.¹⁰¹

Following this work in 2016, Lojkowski synthesized a new type of mesoporous silica nanoparticles for the controlled release of curcumin (Cur) as an anticancer natural pro-drug in an *in vitro* experiment. Firstly, two types of nanocomposite including KCC-NH₂ and MCM-NH₂ were synthesized, then cur was loaded on the surface, and the release of Cur was investigated at several pH values (2.5, 5–7.5) under *in vitro* conditions. The results showed that the highest amount of Cur was released under acidic conditions (pH 2.5). This opens the way for their application in controlling the curcumin delivery in cancer due to the acidic tumor environment increasing its stability and leading to an increase in the Cur bioavailability. The KCC-NH₂ seems to be a more promising nanocarrier for the release of Cur as compared to the commonly used MCM-41 material.¹⁰²

6.4.4 Antimicrobial enzyme delivery using DFNS. The misuse of antibiotics has led to the appearance of antibiotic-resistant bacteria and dangerous threats to human health.¹⁰³ Lysozyme is an abundant enzyme in nature and has antimicrobial activity towards Gram-positive bacteria by separating the glycoside linkage in the bacterial cell wall. It is less effective towards Gram-negative bacteria and can be limited by its poor stability and weak binding affinity to the surface of bacteria; therefore, it requires a method that can increase its surface area. Yu *et al.* synthesized three types of DFNS with various sizes, including MSNs-FC2-R1, MSNs-FC2-R0.4, MSNs-FC8-R0.15. The results showed that MSNs-FC8-R0.15 had a smaller size and density as compared to other DFNP. The lysozyme-loaded MSNs-FC8-R0.15 carried on complete inhibition toward *E. coli* for five days. The bacterial killing activity was also confirmed by an agar plate test. The fibrous morphology of the nanocarriers improved the bacterial membrane adhesion.⁹⁹

6.4.5 Immobilization of *Candida rugosa* lipase by DFNS. By developing nanotechnology, nanocomposites such as nanoparticles, nanotubes, and nanofibrous silica particles were used as enzyme immobilization due to their special structures including their high surface area, high stability, and easy separation.^{104–106} Zhang *et al.* synthesized two types of



nanocomposite, namely, $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{hollow@fibrous SiO}_2$ (yolk shell-1), and $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{hollow@mesoporousSiO}_2$ (yolk shell-2), due to the immobilization of *Candida rugosa* lipase (CRL) through template-assisted selective etching. In the first stage, Fe_3O_4 nanoparticles were synthesized. Then, the SiO_2 shell was coated on the surface of Fe_3O_4 by a sol-gel reaction, and the prepared $\text{Fe}_3\text{O}_4@\text{SiO}_2$ composites were covered by the resorcinol formaldehyde (RF) shell supplied with mesoporous fibrous (KCC-1) in the yolk shell-1. Yolk shell-2 was prepared from yolk shell-1 but the structure of mesoporous silica was not fibrous and glutaraldehyde was used as a linker for the immobilization of *Candida rugosa* lipase. The results indicated that yolk shell-1, with large magnetization ($25\text{--}29.5 \text{ emu g}^{-1}$), highly accessible mesoporous channels ($20\text{--}50 \text{ nm}$), high surface area ($600\text{--}400 \text{ m}^2 \text{ g}^{-1}$), and good loading and enzyme activity as compared to yolk shell-2, was the best candidate for use due to the immobilization of *Candida rugosa*.¹⁰⁷

6.5. Imaging

6.5.1. Bioimaging by DFNS. Quantum Dots (QD) have several forms, one of which is cadmium telluride (CdTe), and they are used in various ways due to their unparalleled size-sensitive optoelectronic properties. However, their photostability and colloidal stability issues, as well as their toxicity, are somewhat limited. Zhang *et al.* used DFNS to stabilize CdTe and enhance its stability and cytotoxicity. 3-APTS was loaded on DFNS to manufacture $\text{mSiO}_2\text{--NH}_2$. In the next stage, CdTe was functionalized with $\text{mSiO}_2\text{--NH}_2$ by using the electrostatic attraction between the positively charged surface and the negatively charged CdTe. A modified DFNP was formed over it using Stöber's method to improve its stability. SEM analysis illustrated the dendritic fibrous nature of $\text{mSiO}_2@\text{CdTe}@\text{SiO}_2$,

and EDS elemental mapping showed an equal distribution of CdTe on the fibers of DFNS silica. These materials showed enhanced QD stability, low toxicity and strong fluorescence (used for bioimaging).¹⁰⁸

6.5.2. Real-time imaging using DFNS. Innovative mesoporous nanomaterials have provided new opportunities for cell imaging. They are interesting because of their sensitivity, modularity, capacity for many potentially varied ligands, and their potential for multifunctional imaging.¹⁰⁹ In a recent study, a novel DFNS was synthesized by a mild one-pot hydrothermal process. In addition to functionalization, the magnetism and fluorescence were amalgamated into a single mesoporous silica composite particle composed of 0.1 g of fluorescein dye and 0.05 g of Fe_3O_4 nanoparticles in DFNS. The obtained nanocomposite was able to load drugs and other large molecules. The synthesized nanocomposite was incubated with murine fibroblast L-929 cell lines to examine the cytotoxicity and ability for *in vitro* fluorescence labeling of living cells. These multifunctional nanocomposites with huge specific surface areas were accepted as ideal platforms due to targeted mass drug delivery and tracking based on their magnetic and luminescence properties.¹¹⁰ The cytotoxicity study illustrated that these materials can be safely used for live-cell imaging and MRI at concentrations less than 10 ppm. The fluorescence imaging showed that these NPs could be used for cell imaging, where green fluorescence was observed in the cytoplasm as well as inside the nuclei due to their internalization and equal distribution within the cells.

6.5.3. Photoacoustic imaging by DFNS. In recent years, phototheranostic imaging has received significant attention in modern nanomedicine dealing with cancer tumor imaging and therapy. Improving the signals from cancer cells is an important stage for diagnosis. Among the variety of nanocarriers systems,

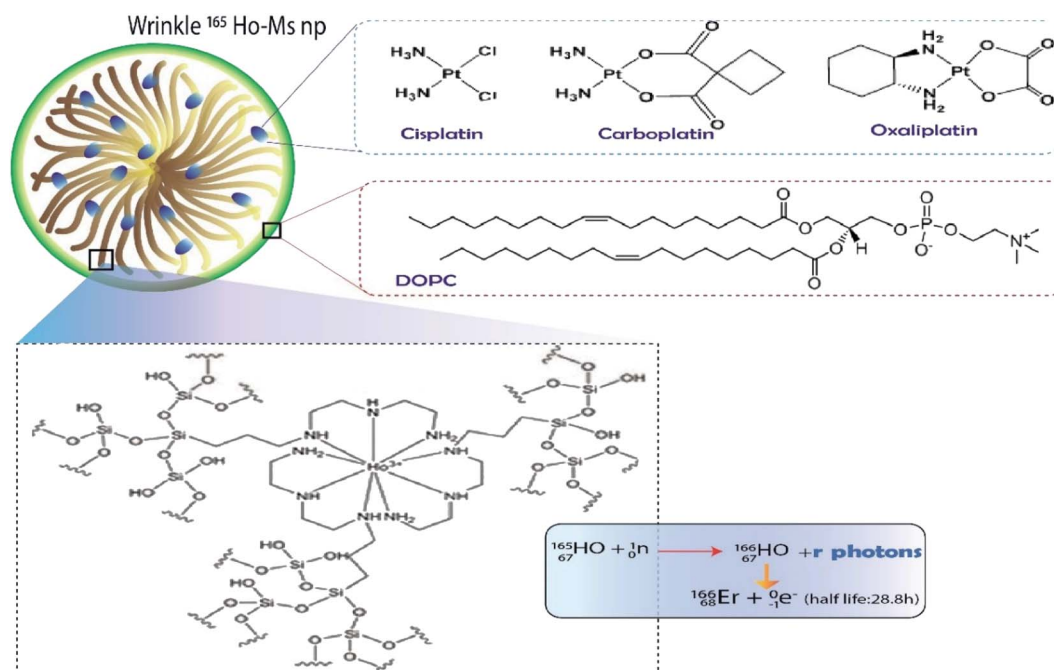


Fig. 5 DOPC lipid-coated platinum drug-loaded holmium-165 containing DFNS.²¹



toxic but after light activation, the photosensitizer has a toxic effect on the targeted tissue.¹¹⁷ Common organic photosensitizers (PSs) are still widely used in cancer treatment in PDT but they have some drawbacks including hydrophobicity, poor stability within the PDT environment and low cell/tissue specificity, which limit their clinical applications. In recent years, the combination of nanoparticles with the PDT method could reduce their side effects. One of the mesoporous nanoparticles introduced by Chengzhong Yu *et al.* for photodynamic therapy was the core-shell-structured dendritic mesoporous silica nanoparticles. In this study, C60 was dispersed in the silica core of DFNS due to photosensitizers and fluorescence agents for imaging. Then, C18 loading to DFNS was used to load a therapeutic mAb anti-pAkt. All of the modifications led to the generation of O₂ and the fluorescence light excitation in MCF-7 cancer cells. The results indicated the cancer cell inhibition by

reducing the levels of anti-apoptotic proteins required by using modified DFNS in combination with the PDT method.¹¹⁸

6.7.3 Mediated combination therapy with DFNS. Cancer therapy based on nanomaterials plays an important role in significantly increasing the therapeutic efficiency against cancer by utilizing a combination of chemotherapeutic agents with nanomaterials. Recent findings illustrated that the combination of two or more nanoparticle-mediated therapies can result in a synergistic therapeutic result to improve the current cancer treatments. Combination therapy has also been illustrated to increase tumor control and reduce undesired side effects of cancer drugs by improving the pharmacokinetics and targeted delivery of the drug payloads. The special structure of silica mesoporous nanoparticles allows them to be chosen as the best candidates in mediated combination therapy. Yan *et al.* succeeded at synthesizing hollow dendritic mesoporous silica nanoparticles capped with chitosan (GM-CS) then

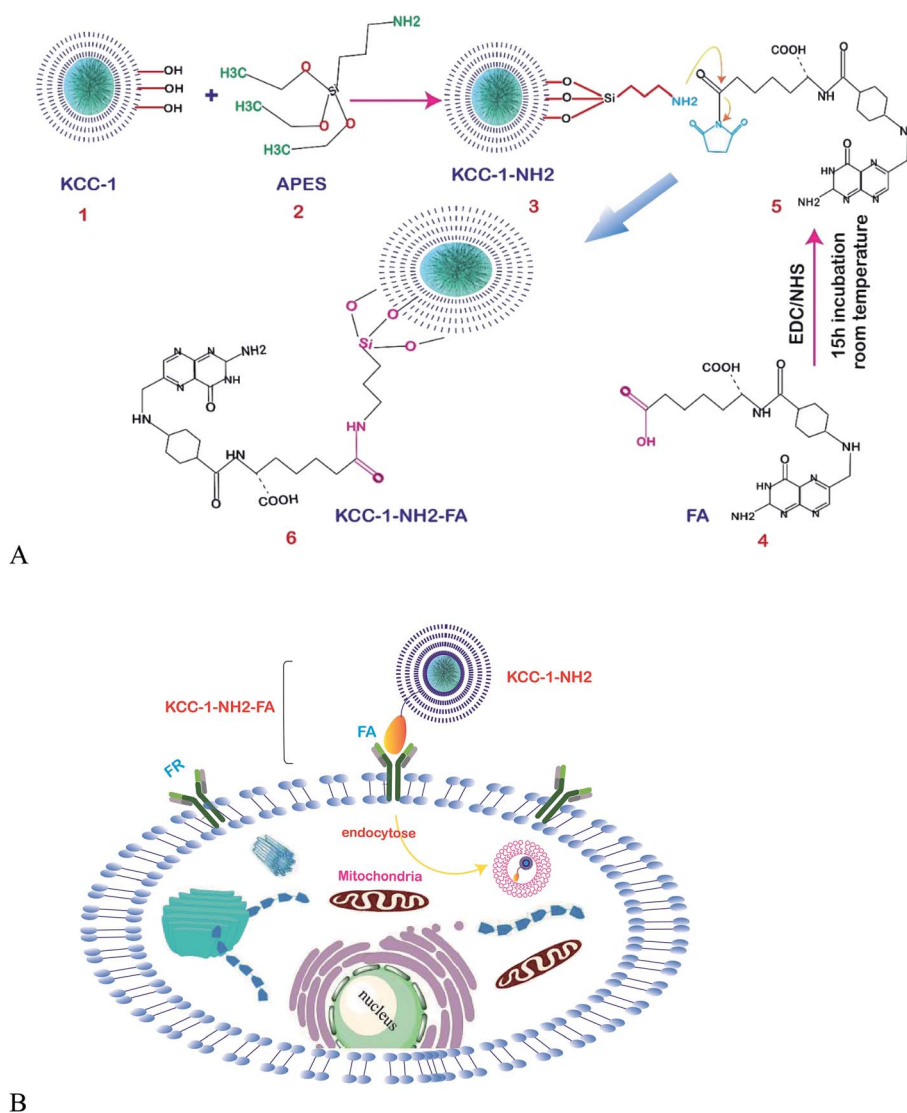
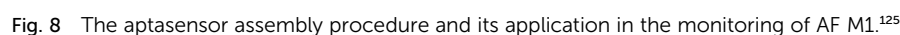


Fig. 7 (A) Schematic representation of the functionalization of KCC-1-NH₂. (B) Schematic illustration of the mechanism of the developed FR-positive cytosensor.¹²¹

One of the amino acids that were classified as a non-essential amino acid is L-proline (L-Pro); the biological and biotechnological applications are clear for any biomedical research program, and the introduction of sensitive and easy methods for its determination in biological samples is in demand. Mirzaie and colleagues succeeded at making an innovative electrochemical biosensor for the sensitive and specific detection of L-proline (L-Pro) in human plasma for the first time. KCC-1 was synthesized by the method by Bayal *et al.*,¹²² then functionalized with 3-APTES to prepare KCC-1-NH₂ as the enzyme encapsulating support. The fabrication of the surface of the electrode was done by encapsulating PRODH/POX on KCC-1-NH₂, and L-proline was finally determined by the ultra-sensitive electrochemistry methods of cyclic voltammetry and chronoamperometry in PBS at pH = 7. The results showed that the fabricated electrode can determine L-Pro in the range of 2.26 to 15 mM, with a LLOQ of 2.26 mM. Therefore, due to the proper function and high sensitivity of the prepared enzyme biosensor, this method can be used to encapsulate and increase the temperature stability of the enzyme as well as to detect L-Pro.¹²³

6.8.1. Biosensors. The highly sensitive sensing of cancer cells in a short time is very important in clinical studies. One of the main stages in biosensor analysis is the fabrication of the surface of the electrode. The unique physicochemical properties of these fibrous nanomaterials make them excellent candidates in electrochemical and optical biosensing. One of the biomarkers investigated in colon carcinoma is HT29.¹²⁰ In a new study, Soleymani *et al.* successfully determined HT29 in FR-positive center cells by folic acid functionalized with KCC-1-NH₂ nanoparticles. Fig. 7A shows the stages for functionalization and Fig. 7B shows the mechanism of the developed FR-positive cytosensor. Two sensitive electrochemistry techniques



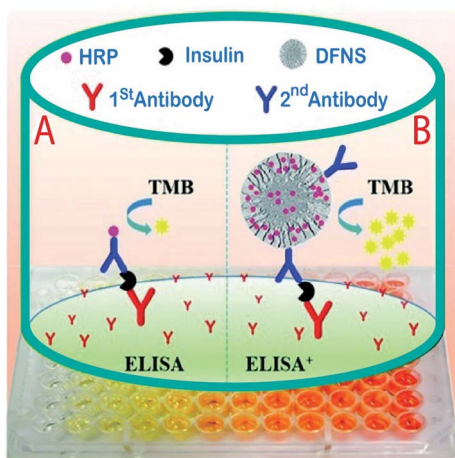


Fig. 9 A comparison of (A) traditional sandwich-like ELISA and (B) the ELISA+ using modified DFNS with high potential HRP loading and achievable dendritic channels to obtain ultrasensitive detection.¹²⁷

Aflatoxins are among the food pollutant items produced by fungi, the most toxic of which is aflatoxin M1 (AFM1). It is a human carcinogen that is found in milk products and may have potentially severe health impacts on milk consumers. In addition, the cancer risk has been reportedly caused by AFM1 in small amounts.¹²⁴ Aflatoxin M1 was determined by an aptamer-based biosensor whose surface was modified by dendritic fibrous nanosilica functionalized by amine groups. In this project, the research group attempted to design a novel, creative electrochemical biosensor for the rapid and sensitive quantification of AFM1 in milk samples. To fabricate this biosensor, GQD-CS and KCC-1-NH₂-Tb were synthesized, and then the nanocomposites were electrodeposited on the surface of GCE. Next, the special oligonucleotide sequence, as a probe aptamer

tag with toluidine blue as an indicator, was immobilized on the surface of GQDs-CS/KCC-1-NH₂-Tb (Fig. 8). Two ultra-sensitive techniques, namely, cyclic voltammetry (CV) and differential pulse voltammetry (DPV), were used for the determination of AFM1 in real samples. The results showed that the method is in the linear range of 0.1 μM to 1 fM with LLOQ of 10 fM.¹²⁵

6.8.2. Bio-analysis. Untreated water can cause many gastrointestinal illnesses. To have a safe, clean environment, the removal of the toxic organic compounds in water is vital. In recent years, many nanoadsorbents have been introduced to solve this problem. In a new study by Polshettiwar *et al.*, diamino-functionalized KCC-1 (DA-KCC-1) with micro-mesostructured fibers was chosen as a nanoadsorbent for the removal of Congo red from water samples. Essential parameters were investigated and optimized, such as the initial pH value, adsorbent dose, and initial dye concentration. The maximum uptake capacity was found to be $\sim 400 \text{ mg g}^{-1}$, according to the Langmuir model.¹²⁶

In another study for the determination of insulin in human serum samples, an ultrasensitive enzyme-linked immunosorbent assay (ELISA⁺) was fabricated by using dendritic fibrous silica nanoparticles (DFNS) with high horseradish peroxidase (HRP) loading and easily accessible pore channels for signal amplification. In this project, Yu *et al.* successfully developed three different pore diameters, 6.9 nm (DFNS-1), 14.9 nm (DFNS-2), and 34.2 nm (DFNS-3) by immobilizing the horseradish peroxidase (HRP) enzyme by treating it with aldehyde functionalized DFNS *via* Schiff base linkage (Fig. 9). The results showed DFNS as having a surface area of $484 \text{ m}^2 \text{ g}^{-1}$ and a pore volume of $1.39 \text{ cm}^3 \text{ g}^{-1}$, with an average pore size of 14.5 nm. There were huge loadings of the HRP enzyme, approximately 2000, and it was more sensitive than a commercial ELISA kit for insulin detection in serum. Also, the calibration parameter illustrated that insulin samples could be determined in low

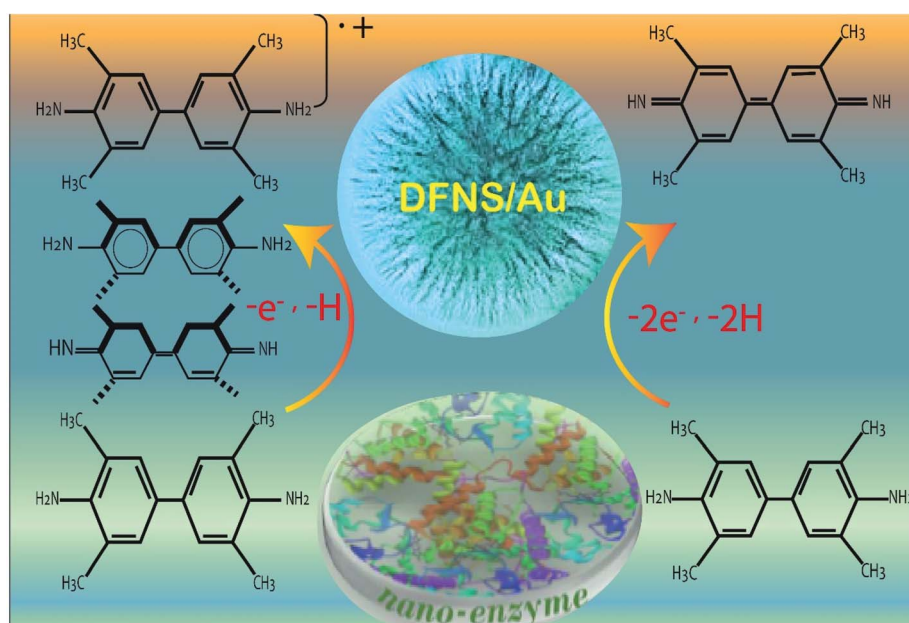


Fig. 10 Application of DFNS/Au as an artificial enzyme.¹³⁴



preparation of DFNS, and toxicity issues related to their medical use. We have compared the physicochemical properties of different mesoporous nanoparticles such as MCM-41, SBA-15, Stöber silica, DFNS, and MSN, and discussed the characterization methods used for investigating the structure of DFNS. The articles studied show that DFNS has unique structures with large numbers of active sites with extraordinarily high accessibility as compared to conventional mesoporous silica materials. Due to the radially oriented pores (channels), the size of DFNS can be manipulated. Biomaterials can quickly access the active sites on the channels of DFNS and increase their interactions with the functionalized organic substances on the surface of DFNS. The biomedical application of DFNS was investigated in drug delivery, therapy, bioanalysis, biocatalysis, bioimaging and biotechnologies. The obtained results showed that for real clinical applications, these nanoprobe must have excellent biocompatibility, good colloidal stability, long circulation time, and controllable size range.

8. Future perspectives

DFNS are an emerging class of mesoporous silica nanoparticles (MSNPs) and are used as biomaterial in various biomedical fields. Due to their unique structure, they can interact with a range of polymers, and macromolecules. They have great biocompatibility and cytocompatibility. Numerous biomedical applications of DFNS have been scientifically explored, such as cell therapy, gene therapy, immunotherapy, drug delivery, chemotherapy, photothermal therapy, imaging, bioanalysis, and biocatalysis. Future studies will focus on the development of these techniques toward clinical applications. Many of these studies are currently not sufficient and just involve the testing of lab animals. On the other hand, the interactions of DFNS with cells are not significant and several parameters such as pore size, channel size, pH and toxicity need to be investigated. We expect that omics-based approaches, including proteomics, transcriptomics and metabolomics will be investigated for the interactions of DFNS with cells. Many MSNPs are used to increase drug solubility but there are no similar studies on the application of DFNS in the solubility of drugs. Therefore, there is the opportunity for the investigation of drug solubility studies based on DFNS. Also, in chemotherapy, it has been illustrated that DFNS can reduce the amount of radiodrug used for therapy; *i.e.*, the side effects of radio drugs will be reduced by these methods. Computational modeling software can be used to estimate the interactions between DFNS with cells or biopolymers. By understanding these interactions, custom inks for 3D printing can be produced. Another exciting application of DFNS is in tissue engineering. It would be interesting to investigate the use DFNS for bone tissue, vascular tissue and wound healing. DFNS can interact with bone cells, skin cells and others, but it is not clear why and how DFNS have these properties. It would be valuable to investigate this phenomenon for further potential applications in tissue engineering.

Finally, novel DFNS has three-dimensional dendritic structures with large pore channels and highly accessible internal surface areas for various applications. We expect to increase the

The use of mesoporous silica nanoparticles in tissue engineering is a newly emerged field that has attracted much research interest¹³⁵ but studies were not accomplished for the investigation DFNS in tissue engineering until now.^{136–139}

This review has mainly focused on the biomedical applications of dendritic fibrous silica nanoparticles. We briefly discussed the synthesis methods and various conditions used for the

Review

cellular uptake and improve the interactions by changing the shape, size and charge of DFNS. All of the parameters facilitate the different biomedical applications of the nanoparticles in the near future.

Conflicts of interest

There is no conflict of interest.

Acknowledgements

This work was supported by Tabriz University of Medical Sciences, Tabriz -Iran.

References

- C. G. Yao and P. N. Martins, *Transplantation*, 2020, **104**, 682–693.
- D. W. Lee and B. R. Yoo, *J. Ind. Eng. Chem.*, 2016, **38**, 1–12.
- Y. Wang, Q. Zhao, N. Han, L. Bai, J. Li, J. Liu, E. Che, L. Hu, Q. Zhang and T. Jiang, *Nanomedicine*, 2015, **11**, 313–327.
- M. Hasanzadeh, N. Shadjou, M. de la Guardia, M. Eskandani and P. Sheikhzadeh, *TrAC, Trends Anal. Chem.*, 2012, **33**, 117–129.
- V. Agrahari, *Neural Regener. Res.*, 2017, **12**, 197.
- E. Solhi and M. Hasanzadeh, *TrAC, Trends Anal. Chem.*, 2019, **118**, 840–852.
- S. M. Vasudevan, N. Ashwanikumar and G. V. Kumar, *Biomater. Sci.*, 2019, **7**, 4017–4021.
- M. Hasanzadeh, N. Shadjou, M. Eskandani and M. de la Guardia, *TrAC, Trends Anal. Chem.*, 2012, **40**, 106–118.
- Y. Wang, X. Du, Z. Liu, S. Shi and H. Lv, *J. Mater. Chem. A*, 2019, **7**, 5111–5152.
- A. Maity and V. Polshettiwar, *ChemSusChem*, 2017, **10**, 3866–3913.
- M. Mahmudi, N. Shadjou and F. A. M. Hasanzadeh, *J. Electroanal. Chem.*, 2019, **848**, 113272.
- V. Tzankova, D. Aluani, Y. Yordanov, M. Valoti, M. Frosini, I. Spassova, D. Kovacheva and B. Tzankov, *Drug Chem. Toxicol.*, 2019, 1–12.
- M. A. Downing and P. K. Jain, in *Nanoparticles for Biomedical Applications*, Elsevier, 2020, pp. 267–281.
- M. Zahiri, M. Babaei, K. Abnous, S. M. Taghdisi, M. Ramezani and M. Alibolandi, *J. Cell. Physiol.*, 2019, **235**(2), 1036–1050.
- T. Li, S. Shi, S. Goel, X. Shen, X. Xie, Z. Chen, H. Zhang, S. Li, X. Qin and H. Yang, *Acta Biomater.*, 2019, **89**, 1–13.
- C. Xu, C. Lei and C. Yu, *Front. Chem.*, 2019, **7**, 290.
- T. L. Nguyen, Y. Choi and J. Kim, *Adv. Mater.*, 2019, **31**, 1803953.
- Y. Xie, J. Wang, M. Wang and X. Ge, *J. Hazard. Mater.*, 2015, **297**, 66–73.
- V. Califano, F. Sannino, A. Costantini, J. Avossa, S. Cimino and A. Aronne, *J. Phys. Chem. C*, 2018, **122**, 8373–8379.
- I. Munaweera, J. Hong, A. D'Souza and K. J. Balkus, *J. Porous Mater.*, 2015, **22**, 1–10.
- I. Munaweera, B. Koneru, Y. Shi, A. J. Di Pasqua, J. Balkus and J. Kenneth, *APL Mater.*, 2014, **2**, 113315.
- X. Huang and H. Townley, *Nanobiomedicine*, 2016, **3**, 4.
- Z. Wang and K. J. Balkus Jr, *Catal. Commun.*, 2017, **96**, 15–18.
- I. Munaweera, Y. Shi, B. Koneru, A. Patel, M. H. Dang, A. J. Di Pasqua and K. J. Balkus Jr, *J. Inorg. Biochem.*, 2015, **153**, 23–31.
- T. Qian, J. Li, X. Min, Y. Deng, W. Guan and L. Ning, *Energy*, 2016, **112**, 1074–1083.
- J. Gao, W. Kong, L. Zhou, Y. He, L. Ma, Y. Wang, L. Yin and Y. Jiang, *Chem. Eng. J.*, 2017, **309**, 70–79.
- D. Wan, C. Yan, Y. Liu, K. Zhu and Q. Zhang, *Microporous Mesoporous Mater.*, 2019, **280**, 46–56.
- F. Zhao, X. Wang, B. Ding, J. Lin, J. Hu, Y. Si, J. Yu and G. Sun, *RSC Adv.*, 2011, **1**, 1482–1488.
- M. Mogharabi-Manzari, M. H. Ghahremani, T. Sedaghat, F. Shayan and M. A. Faramarzi, *Eur. J. Org. Chem.*, 2019, **2019**, 1741–1747.
- Z. Liu, J. Ru, S. Sun, Z. Teng, H. Dong, P. Song, Y. Yang and H. Guo, *J. Mater. Chem. B*, 2019, **7**(21), 3446–3454.
- X. Du and J. He, *Langmuir*, 2011, **27**, 2972–2979.
- D.-S. Moon and J.-K. Lee, *Langmuir*, 2012, **28**, 12341–12347.
- X. Du, C. Zhao, M. Zhou, T. Ma, H. Huang, M. Jaroniec, X. Zhang and S. Z. Qiao, *Small*, 2017, **13**, 1602592.
- N. Mizoshita, T. Tani and S. Inagaki, *Chem. Soc. Rev.*, 2011, **40**, 789–800.
- S. S. Park, M. S. Moorthy and C.-S. Ha, *NPG Asia Mater.*, 2014, **6**, e96.
- Y. Wang, X. Du, Z. Liu and S. Shi, *J. Mater. Chem. A*, 2019, **7**(22), 5111–5152.
- V. Polshettiwar, D. Cha, X. Zhang and J. M. Basset, *Angew. Chem., Int. Ed.*, 2010, **49**, 9652–9656.
- S. Zhang, Y. Qian and W.-S. Ahn, *Chem. Sci.*, 2019, **11**, 5369–5403.
- A. Maity, R. Belgamwar and V. Polshettiwar, *Nat. Protoc.*, 2019, **14**, 2177–2204.
- X. Huang, Z. Tao, J. C. Praskavich Jr, A. Goswami, J. F. Al-Sharab, T. Minko, V. Polshettiwar and T. Asefa, *Langmuir*, 2014, **30**, 10886–10898.
- F. Shahangi, A. N. Chermahini and M. Saraji, *J. Energy Chem.*, 2018, **27**, 769–780.
- Z. Mohammadbagheri and A. N. Chermahini, *J. Ind. Eng. Chem.*, 2018, **62**, 401–408.
- R. Hasan, C. C. Chong and H. D. Setiabudi, *Bull. Chem. React. Eng. Catal.*, 2019, **14**, 196–204.
- R. Hasan, C. Chong, S. Bukhari, R. Jusoh and H. Setiabudi, *J. Ind. Eng. Chem.*, 2019, **75**, 262–270.
- A. Rahikkala, S. A. Pereira, P. Figueiredo, M. L. Passos, A. R. Araújo, M. L. M. Saraiva and H. A. Santos, *Adv. Biosyst.*, 2018, **2**, 1800020.
- T. Daou, G. Pourroy, S. Bégin-Colin, J.-M. Grenèche, C. Ulhaq-Bouillet, P. Legaré, P. Bernhardt, C. Leuvrey and G. Rogez, *Chem. Mater.*, 2006, **18**, 4399–4404.
- A. Maity, R. Belgamwar and V. Polshettiwar, *Nat. Protoc.*, 2019, **14**, 2177.



- 48 K. Zhang, L.-L. Xu, J.-G. Jiang, N. Calin, K.-F. Lam, S.-J. Zhang, H.-H. Wu, G.-D. Wu, B. I. Albela and L. Bonnevot, *J. Am. Chem. Soc.*, 2013, **135**, 2427–2430.
- 49 D.-S. Moon and J.-K. Lee, *Langmuir*, 2014, **30**, 15574–15580.
- 50 Y.-J. Yu, J.-L. Xing, J.-L. Pang, S.-H. Jiang, K.-F. Lam, T.-Q. Yang, Q.-S. Xue, K. Zhang and P. Wu, *ACS Appl. Mater. Interfaces*, 2014, **6**, 22655–22665.
- 51 H. Gustafsson, S. Isaksson, A. Altskär and K. Holmberg, *J. Colloid Interface Sci.*, 2016, **467**, 253–260.
- 52 H. Yang, S. Liao, C. Huang, L. Du, P. Chen, P. Huang, Z. Fu and Y. Li, *Appl. Surf. Sci.*, 2014, **314**, 7–14.
- 53 S. M. Sadeghzadeh, R. Zhiani, M. Khoobi and S. Emrani, *Microporous Mesoporous Mater.*, 2018, **257**, 147–153.
- 54 S. M. Sadeghzadeh, *Appl. Organomet. Chem.*, 2016, **30**, 835–842.
- 55 M. Y. Shahul Hamid, S. Triwahyono, A. A. Jalil, N. W. Che Jusoh, S. M. Izan and T. A. Tuan Abdullah, *Inorg. Chem.*, 2018, **57**, 5859–5869.
- 56 Z. S. Qureshi, P. B. Sarawade, M. Albert, V. D'Elia, M. N. Hedhili, K. Köhler and J. M. Basset, *ChemCatChem*, 2015, **7**, 635–642.
- 57 A. Aghakhani, E. Kazemi and M. Kazemzad, *J. Nanopart. Res.*, 2015, **17**, 386.
- 58 P. Gautam, M. Dhiman, V. Polshettiwar and B. M. Bhanage, *Green Chem.*, 2016, **18**, 5890–5899.
- 59 K. Yu, X. Zhang, H. Tong, X. Yan and S. Liu, *Mater. Lett.*, 2013, **106**, 151–154.
- 60 A. Hassankhani, S. M. Sadeghzadeh and R. Zhiani, *RSC Adv.*, 2018, **8**, 8761–8769.
- 61 A. Fihri, D. Cha, M. Bouhrara, N. Almana and V. Polshettiwar, *ChemSusChem*, 2012, **5**, 85–89.
- 62 A. S. Lilly Thankamony, C. Lion, F. Pourpoint, B. Singh, A. J. Perez Linde, D. Carnevale, G. Bodenhausen, H. Vezin, O. Lafon and V. Polshettiwar, *Angew. Chem., Int. Ed.*, 2015, **54**, 2190–2193.
- 63 R. Sharma, S. Sharma, S. Dutta, R. Zboril and M. B. Gawande, *Green Chem.*, 2015, **17**, 3207–3230.
- 64 S. A. P. Pereira, *Adv. Biosyst.*, 2018, **2**(7), 1800020.
- 65 T. Yu, K. Greish, L. D. McGill, A. Ray and H. Ghandehari, *ACS Nano*, 2012, **6**, 2289–2301.
- 66 J. Soleymani, M. Hasanzadeh, N. Shadjou, M. H. Somi and A. Jouyban, *J. Pharm. Biomed. Anal.*, 2020, **180**, 113077.
- 67 V. Mamaeva, C. Sahlgren and M. Lindén, *Adv. Drug Delivery Rev.*, 2013, **65**, 689–702.
- 68 K. Newick, S. O'Brien, E. Moon and S. M. Albelda, *Annu. Rev. Med.*, 2017, **68**, 139–152.
- 69 E. M. Andre, C. Passirani, B. Seijo, A. Sanchez and C. N. Montero-Menei, *Biomaterials*, 2016, **83**, 347–362.
- 70 P. L. Abbaraju, A. K. Meka, H. Song, Y. Yang, M. Jambhrunkar, J. Zhang, C. Xu, M. Yu and C. Yu, *J. Am. Chem. Soc.*, 2017, **139**, 6321–6328.
- 71 X. J. Loh, T.-C. Lee, Q. Dou and G. R. Deen, *Biomater. Sci.*, 2016, **4**, 70–86.
- 72 J. Liu, L. Song, S. Liu, Q. Jiang, Q. Liu, N. Li, Z.-G. Wang and B. Ding, *Nano Lett.*, 2018, **18**, 3328–3334.
- 73 H. Ye, C. Owh and X. J. Loh, *RSC Adv.*, 2015, **5**, 48720–48728.
- 74 X. J. Loh, Z.-X. Zhang, K. Y. Mya, Y.-I. Wu, C. B. He and J. Li, *J. Mater. Chem.*, 2010, **20**, 10634–10642.
- 75 X. J. Loh, S. J. Ong, Y. T. Tung and H. T. Choo, *Macromol. Biosci.*, 2013, **13**, 1092–1099.
- 76 X. Du, L. Xiong, S. Dai, F. Kleitz and S. Z. Qiao, *Adv. Funct. Mater.*, 2014, **24**, 7627–7637.
- 77 J. Li, H. Pei, B. Zhu, L. Liang, M. Wei, Y. He, N. Chen, D. Li, Q. Huang and C. Fan, *ACS Nano*, 2011, **5**, 8783–8789.
- 78 A. M. Krieg, *Annu. Rev. Immunol.*, 2002, **20**, 709–760.
- 79 Y. Xu, Y. Zhu, X. Li, H. Morita and N. Hanagata, *Mater. Express*, 2016, **6**, 116–126.
- 80 F. Benyettou, R. Rezgui, F. Ravoux, T. Jaber, K. Blumer, M. Jouiad, L. Motte, J.-C. Olsen, C. Platas-Iglesias and M. Magzoub, *J. Mater. Chem. B*, 2015, **3**, 7237–7245.
- 81 Y. Yang, Y. Lu, P. L. Abbaraju, J. Zhang, M. Zhang, G. Xiang and C. Yu, *Angew. Chem., Int. Ed.*, 2017, **56**, 8446–8450.
- 82 B. K. Nanjwade, A. B. Sarkar and T. Srichana, in *Characterization and Biology of Nanomaterials for Drug Delivery*, Elsevier, 2019, pp. 337–350.
- 83 A. K. Agrawal, F. Aqil, J. Jeyabalan, W. A. Spencer, J. Beck, B. W. Gachuki, S. S. Alhakeem, K. Oben, R. Munagala and S. Bondada, *Nanomedicine*, 2017, **13**, 1627–1636.
- 84 A. Altunbas, S. J. Lee, S. A. Rajasekaran, J. P. Schneider and D. J. Pochan, *Biomaterials*, 2011, **32**, 5906–5914.
- 85 F. D. Duman, M. Erkisa, R. Khodadust, F. Ari, E. Ulukaya and H. Y. Acar, *Nanomedicine*, 2017, **12**, 2319–2333.
- 86 F. A. Galaway and P. G. Stockley, *Mol. Pharmaceutics*, 2012, **10**, 59–68.
- 87 S. Hua and B. Vaughan, *Int. J. Nanomed.*, 2019, **14**, 2191.
- 88 B. R. Smith and E. Ghosn, US Patent App, 16/353743, 2019.
- 89 H. Thi, T. Thanh, D.-H. Nguyen Tran, L. G. Bach, H. Vu-Quang, D. C. Nguyen, K. D. Park and D. H. Nguyen, *Pharmaceutics*, 2019, **11**, 120.
- 90 S. P. Kuruvilla, G. Tiruchinapally, A. C. Crouch, M. E. ElSayed and J. M. Greve, *PLoS One*, 2017, **12**, e0181944.
- 91 A. Watermann and J. Brieger, *Nanomaterials*, 2017, **7**, 189.
- 92 E. V. Batrakova, M. Haney, N. Klyachko, Y. Zhao and A. Kabanov, *J. Extracell. Vesicles*, 2018, **7**, 266.
- 93 S. R. Prasad, T. S. Kumar and A. Jayakrishnan, *Nanotechnology*, 2017, **29**, 015101.
- 94 S. Gai, P. Yang, L. Wang, C. Li, M. Zhang and L. Jun, *Dalton Trans.*, 2012, **41**, 4511–4516.
- 95 X. Du, B. Shi, J. Liang, J. Bi, S. Dai and S. Z. Qiao, *Adv. Mater.*, 2013, **25**, 5981–5985.
- 96 Z. Tian, Y. Xu and Y. Zhu, *Mater. Sci. Eng. C*, 2017, **71**, 452–459.
- 97 E. Juère, R. Caillard, D. Marko, G. Del Favero and F. Kleitz, *Chem.-Eur. J.*, 2020, **26**(23), 5195–5199.
- 98 S. Farooq, Z. Mehmood, F. A. Qais, M. S. Khan and I. Ahmad, in *New Look to Phytomedicine*, Elsevier, 2019, pp. 581–596.
- 99 Y. Wang, Y. A. Nor, H. Song, Y. Yang, C. Xu, M. Yu and C. Yu, *J. Mater. Chem. B*, 2016, **4**, 2646–2653.
- 100 J. Wang, Y. Wang, Q. Liu, L. Yang, R. Zhu, C. Yu and S. Wang, *ACS Appl. Mater. Interfaces*, 2016, **8**, 26511–26523.
- 101 K. AbouAitah, A. Swiderska-Sroda, A. A. Farghali, J. Wojnarowicz, A. Stefanek, S. Gierlotka, A. Opalinska,



- A. K. Allayeh, T. Ciach and W. Lojkowski, *Oncotarget*, 2018, **9**, 26466.
- 102 K. E. AbouAitah, A. Farghali, A. Swiderska-Sroda, W. Lojkowski, A. Razin and M. Khedr, *J. Nanomed. Nanotechnol.*, 2016, **7**, 1–11.
- 103 Y. Wang, Y. Wang, X. Li, J. Li, L. Su, X. Zhang and X. Du, *ACS Sustainable Chem. Eng.*, 2018, **6**, 14071–14081.
- 104 W. Xie and J. Wang, *Energy Fuels*, 2014, **28**, 2624–2631.
- 105 T. Jamshaid, E. T. T. Neto, M. M. Eissa, N. Zine, M. H. Kunita, A. E. El-Salhi and A. Elaissari, *TrAC, Trends Anal. Chem.*, 2016, **79**, 344–362.
- 106 M. Shaban, M. Hasanzadeh and E. Solhi, *Anal. Methods*, 2019, **11**(44), 5661–5672.
- 107 Z. Ali, L. Tian, B. Zhang, N. Ali, M. Khan and Q. Zhang, *Enzyme Microb. Technol.*, 2017, **103**, 42–52.
- 108 S. Zhang, L. Wen, J. Yang, J. Zeng, Q. Sun, Z. Li, D. Zhao and S. Dou, *Part. Part. Syst. Charact.*, 2016, **33**, 261–270.
- 109 P. Wang, T. Kim, M. Harada, C. Contag, X. Huang and B. R. Smith, *Nanoscale Horiz.*, 2020, **5**(4), 628–653.
- 110 T. S. Atabaev, J. H. Lee, J. J. Lee, D.-W. Han, Y.-H. Hwang, H.-K. Kim and N. H. Hong, *Nanotechnology*, 2013, **24**, 345603.
- 111 C. Zhang, D. Li and X. Shi, in *Photonanotechnology for Therapeutics and Imaging*, Elsevier, 2020, pp. 23–43.
- 112 H. Dong, W. Dai, H. Ju, H. Lu, S. Wang, L. Xu, S.-F. Zhou, Y. Zhang and X. Zhang, *ACS Appl. Mater. Interfaces*, 2015, **7**, 11015–11023.
- 113 K. Zhang, X. Meng, Z. Yang, Y. Cao, Y. Cheng, D. Wang, H. Lu, Z. Shi, H. Dong and X. Zhang, *Adv. Mater.*, 2019, **31**, 1807888.
- 114 X.-D. Li, Z. Wang, X.-R. Wang, D. Shao, X. Zhang, L. Li, M.-F. Ge, Z.-M. Chang and W.-F. Dong, *Int. J. Nanomed.*, 2019, **14**, 3967.
- 115 S. Wang, Y. Yin, W. Song, Q. Zhang, Z. Yang, Z. Dong, Y. Xu, S. Cai, K. Wang and W. Yang, *J. Mater. Chem. B*, 2020, **12**(3), 125–129.
- 116 R. Chen, F. Yang, Y. Xue, X. Wei, L. Song and X. Liu, *RSC Adv.*, 2016, **6**, 38931–38942.
- 117 R. Falk-Mahapatra and S. O. Gollnick, *Photochemistry and Photobiology*, 2020.
- 118 P. L. Abbaraju, Y. Yang, M. Yu, J. Fu, C. Xu and C. Yu, *Chem.-Asian J.*, 2017, **12**, 1465–1469.
- 119 T. Yan, J. He, R. Liu, Z. Liu and J. Cheng, *Carbohydr. Polym.*, 2020, **231**, 115706.
- 120 H. Li, J. Su, J. Jiang, Y. Li, Z. Gan, Y. Ding, Y. Li, J. Liu, S. Wang and Y. Ke, *Int. J. Biol. Macromol.*, 2019, **131**, 886–895.
- 121 J. Soleymani, M. Hasanzadeh, M. H. Somi, N. Shadjou and A. Jouyban, *Biosens. Bioelectron.*, 2019, **132**, 122–131.
- 122 N. Bayal, B. Singh, R. Singh and V. Polshettiwar, *Sci. Rep.*, 2016, **6**, 24888.
- 123 A. Mirzaie, A. Saadati, S. Hassanpour, M. Hasanzadeh, M. Siahi-Shadbad and A. Jouyban, *Anal. Methods*, 2019, **11**(36), 4609–4619.
- 124 S. Ahlberg, D. Grace, G. Kiarie, Y. Kirino and J. Lindahl, *Toxins*, 2018, **10**, 348.
- 125 M.-H. Moosavy, M. Hasanzadeh, J. Soleymani and A. Mokhtarzadeh, *Anal. Methods*, 2019, **11**, 3910–3919.
- 126 R. Soltani, A. Marjani, M. R. S. Moguei, B. Rostami and S. Shirazian, *J. Mol. Liq.*, 2019, **294**, 111617.
- 127 C. Lei, C. Xu, A. Nouwens and C. Yu, *J. Mater. Chem. B*, 2016, **4**, 4975–4979.
- 128 J.-M. Liu and X.-P. Yan, *J. Anal. At. Spectrom.*, 2011, **26**, 1191–1197.
- 129 C. Hess, A. Thomas, M. Thevis, B. Stratmann, W. Quester, D. Tschoepe, B. Madea and F. Musshoff, *Anal. Bioanal. Chem.*, 2012, **404**, 1813–1822.
- 130 M. Zhao, C. Deng and X. Zhang, *Chem. Commun.*, 2014, **50**, 6228–6231.
- 131 Y. Yan, X. Zhang and C. Deng, *ACS Appl. Mater. Interfaces*, 2014, **6**, 5467–5471.
- 132 H. Zhong, X. Xiao, S. Zheng, W. Zhang, M. Ding, H. Jiang, L. Huang and J. Kang, *Nat. Commun.*, 2013, **4**, 1656.
- 133 Y. Hong, Y. Yao, H. Zhao, Q. Sheng, M. Ye, C. Yu and M. Lan, *Anal. Chem.*, 2018, **90**, 7617–7625.
- 134 R. Singh, R. Belgamwar, M. Dhiman and V. Polshettiwar, *J. Mater. Chem. B*, 2018, **6**, 1600–1604.
- 135 L. Chen, X. Zhou and C. He, *Wiley Interdiscip. Rev.: Nanomed. Nanobiotechnol.*, 2019, **11**, e1573.
- 136 E. Febriyanti, V. Suendo, R. Mukti, A. Prasetyo, A. Arifin, M. Akbar, S. Triwahyono and I. Marsih, *Langmuir*, 2016, **32**, 5802–5811.
- 137 D. Shen, J. Yang, X. Li, L. Zhou, R. Zhang, W. Li, L. Chen, R. Wang, F. Zhang and D. Zhao, *Nano Lett.*, 2014, **14**, 923–932.
- 138 X. Du, X. Li, H. Huang, J. He and X. Zhang, *Nanoscale*, 2015, **7**, 6173–6184.
- 139 A. Maity and V. Polshettiwar, *ACS Appl. Nano Mater.*, 2018, **1**, 3636–3643.

