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Soft-template-assisted synthesis: a promising approach for the fabrication of transition metal oxides

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The past few decades have witnessed transition metal oxides (TMOs) as promising candidates for a plethora of applications in numerous fields. The exceptional properties retained by these materials have rendered them of paramount emphasis as functional materials. Thus, the controlled and scalable synthesis of transition metal oxides with desired properties has received enormous attention. Out of different top-down and bottom-up approaches, template-assisted synthesis predominates as an adept approach for the facile synthesis of transition metal oxides, owing to its phenomenal ability for morphological and physicochemical tuning. This review presents a comprehensive examination of the recent advances in the soft-template-assisted synthesis of TMOs, focusing on the morphological and physicochemical tuning aided by different soft-templates. The promising applications of TMOs are explained in detail, emphasizing those with excellent performances.

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1. Introduction

The advancement of nanotechnology has captivated the perception of researchers for the formulation of cutting-edge materials for varied applications.^{1–4} One of the preminent classes of such materials is transition metal oxides (TMOs), which have been deemed as functional materials over the

years.^{5–9} The partially filled d-shells of the positive metallic ions in TMOs render them phenomenal properties such as high dielectric constants,^{10,11} competent electrical characteristics,^{12,13} reactive electronic transitions¹⁴ and broader band gaps.^{15,16} These diverse features together with reduced cost, environmental compatibility and easy availability¹⁷ have driven TMOs as a proficient material for a broad spectrum of applications including energy conversion and storage,^{17–22} catalysis,^{23,24} sensing,^{25,26} and bio-based applications.^{27,28} The regulation of the structure, morphology, stoichiometry, size and crystallinity

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of TMOs greatly control their properties and applications.^{29–31} The intricate surface chemistry and the occupancy of lattice vacancies also contribute to their properties by disrupting their electronic arrangement.³² Hence, continuous efforts have been employed for the controlled and scalable synthesis of TMOs with desired properties.^{33–36}

The synthesis of TMOs follows one of two approaches, namely, the top-down approach, or the bottom-up approach. The top-down approach focuses on the curtailing of the magnitude and dimension of the component from the bulk, while the bottom-up approach emphasizes the fabrication of nanomaterials from smaller atoms or molecules.³⁷ Of the various synthesis pathways, the template-assisted method is an advancing bottom-up approach for the preparation of highly crystalline mesoporous TMOs.^{38–41} Template-assisted synthesis methods can be broadly classified as hard-template and soft-template assisted methods,⁴² and colloidal templates are an emerging theme in this synthesis strategy.⁴³ While hard-templates are chiefly inorganic-based,⁴⁴ with silica being the most important among them,⁴⁵ soft-templates are primarily organic-based, like surfactants, block polymers, or flexible organic molecules.^{44,46} In comparison with the hard-templates, soft-templates are favored by virtue of their low cost and easy synthesis without the need for tedious processes for the removal of the template.⁴⁶

This review is intended to give a sketch of the recent progress in the state-of-the-art research that converges on the soft-template-assisted synthesis strategies for TMOs. Though earlier reviews have presented exquisite summaries of various synthesis methods of TMOs^{37,47–49} and various template-assisted synthesis approaches,^{42,50,51} this review explicitly focuses on the

synthesis of TMOs using soft-templates and their diverse applications. The review opens with a brief description of the general synthesis methods of TMOs, including top-down and bottom-up approaches, and advances with the meticulous examination of template-assisted synthesis methods. Of the various template-assisted synthesis schemes, the soft-template assisted approach is focused on the preparation of TMOs and different soft-templates have been discussed comprehensively by taking into account the distinct property tuning achieved using this method. The promising applications of TMOs are explained in detail, highlighting those with superior performance. An outline of the prevailing research status and development is given, along with the future perspectives and challenges in this upcoming area.

2. General synthesis strategies of TMOs

The fabrication of TMOs with superior properties for peculiar applications is often a tedious effort. The surface instability generated by the strong surface polarization due to the absence of solid interlayer interactions endures as a hurdle for the synthesis of TMOs with specific morphologies.⁵² However, these obstacles can be overcome by undertaking pertinent synthesis methods for the preparation of TMOs with notable lattice relaxations and pronounced structural reconstruction.^{53,54} For such an effective synthesis, comprehensive insight into the interactions between the precursor materials is required, along with their absolute distribution.⁵⁵ The methods for TMO synthesis can be generally sorted into two broad classes: the top-down approach and the bottom-up approach. The schematic



Fig. 1 Schematic representations of the top-down and bottom-up approaches with typical examples: (Bottom left) schematic representation of the exfoliation of hydrous chlorides as an example of the top-down approach. Adapted with permission from ref. 58. Copyright 2016 Springer Nature publishing. (Top right) Schematics for the molecular assembly of ultrathin 2D metal oxide nanosheets as an example of the bottom-up approach. Reproduced with permission from ref. 33. Copyright 2016 Springer Nature publishing.



representations of these methods along with typical examples are provided in Fig. 1. Some of the synthesis methods even follow a combined route of these two approaches, mainly for the synthesis of hybrid materials.^{56,57}

The top-down approach mainly implicates reducing the size and dimensions of bulk materials in order to design the targeted products. It makes use of the process of lithography, including physical and chemical etching along with other processes like mechanical milling and exfoliation.^{58,59} Being an accessible and efficient route without the demand for complicated methods and equipment, this technique can be employed for comparably large-scale production. Furthermore, the materials synthesized *via* this approach usually retain high crystallinity, which is acquired from the host material.⁶⁰ Notwithstanding these advantages, the top-down approach also has its own drawbacks, including the feeble quality distribution of the attained product, broad size distribution and varied sizes and shapes of the products, formation of internal stress, structural defects and deterioration, and most importantly, the requirement of appropriate host materials.⁶⁰

Nature's ability to exploit chemical or physical forces for the creation of self-assembled structures has inspired researchers to mimic this strategy for the preparation of a wide range of materials, giving rise to a new concept of the bottom-up approach. This approach adopts chemical or physical forces driving at the nanoscale to compile basic units into larger structures for multifunctional nanostructure systems and devices.⁶¹ The bottom-up approach affords the utmost limits of miniaturization by offering tunable, cost-effective, and readily controlled processes for the high yield of uniform products. However, the need for complex synthesis steps has limited its application to simple nanostructured materials.⁶⁰ The advantages and disadvantages of the top-down and bottom-up approaches are highlighted in Table 1.

The bottom-up approach is further sub-categorized as vapor-based, solution-based and template-based methods.⁶² When compared to the vapor deposition and solution-based methods,

the template-assisted synthesis is an advancing technique in the bottom-up approach for the controlled synthesis of nanostructures. It is a cost-effective technique capable of regulating the morphology and structure of the nanomaterials for customized applications.⁶³ The synthesis of highly ordered mesoporous nanostructures is the major highlight of this method.⁶⁴ It makes use of a template, which acts as a scaffold for guiding the growth of nanostructures for obtaining different geometries and morphologies.⁶⁵ Since the principal focus of this review is the soft-template-assisted synthesis of TMOs, the template-based synthetic approaches are discussed in detail in the forthcoming sections.

3. Template-assisted synthesis of TMOs

The major factor of a template-assisted synthesis is the "template", which can be any entity with nanostructured characteristics, and whose size, morphology and charge dissemination greatly influence the structure guiding properties.⁶⁶ The initial step in the template-assisted synthesis is the preparation of the template, followed by the fabrication of the desired material using the template and the elimination of the template, if necessary. The fabrication of the desired material may include physical methods such as surface coating or chemical methods like addition, elimination, substitution, or isomerization reactions. Once the reaction is completed, the template may be eliminated by physical methods like dissolution or chemical methods like calcination.⁴² The major benefit of the template-assisted method is its effective control over the structure, dimension and morphology of the end product. Different TMOs with varied morphologies including 1D, 2D and 3D structures have been synthesized using this technique.⁶⁷⁻⁷⁰ Based on the type of template present, the template-assisted synthesis can be further classified as a hard-template method, soft-template method, or colloidal-template method. A schematic representation of the fabrication of materials using these templates is provided in Fig. 2.

Table 1 Advantages and disadvantages of the top-down and bottom-up approaches for the synthesis of TMOs

Top-down approach	Bottom-up approach
Advantages	
Employed for the large-scale production of nano/microscale structures	Controlled tuning of the morphology, pore size, crystallinity and phase
Synthesized materials retain high crystallinity	Control of the reaction parameters
Non-requirement of chemical purification	Cost-effective technique
Disadvantages	
Structural defects and deterioration	Complex synthesis steps
Formation of internal stress	Difficult for large-scale production
Feeble morphology distribution of the products	Requires chemical purification

3.1 Hard-template method

The hard-template method, also termed as nanocasting, adopts a solid mold with a well-defined design, whose surfaces or pores are subsequently filled by the precursor molecules to form the final product. Once the reaction is completed, the mold is removed without affecting the properties of the product.^{51,71} There are different types of hard-templates like macroscopic structures and *in situ*-formed templates, which can be used for the synthesis processes. Macroscopic structures may consist of fibers, powders, or films. Typical examples are anodized aluminum oxide (AAO),^{72,73} biological materials such as organic macromolecules, tissues and biomolecules,⁷⁴ polymeric microspheres⁷⁵ and silica.⁷⁶ *In situ* hard templates refer to the templates formed by the *in situ* physical or chemical conversion of a compound in the precursors. Common examples include ice crystals, salt, or carbon.^{77,78} As the structural and morphological features of the acquired product immensely rely on the





Fig. 2 Schematic representation of the synthesis of materials using different types of templates.

quality and attributes of the template and their interaction with the precursors, a judicious selection of the template is crucial for the fabrication of desired products.⁷⁹ Though hard-template-assisted synthesis is a prominent method for the synthesis of crystalline oxides, certain features act as a challenge for its effective utilization. These may include their comparably less controllable pore size, lower yield resulting from the irresistible nucleation occurring outside the pore and their tedious processes with high cost.^{80,81}

3.2 Soft-template method

The soft-template method commonly makes use of flexible nanostructures as soft-templates, which are composed by intermolecular interactions. The soft-templates are comprised of soft matter, including surfactants, flexible organic molecules and block copolymers.⁸² The interactions of these templates with the precursors normally occurs by weak non-covalent bonds, like electrostatic or van der Waals interactions and hydrogen bonding.⁸³ Compared with the hard-template methods, soft-template-assisted synthesis imparts immense control on the geometry, morphology and size of the product by virtue of the controlled sol-gel transition of the precursors to their oxides.⁸⁴ Since soft-template-assisted synthesis is the core theme of this review, the detailed description of this method

along with its importance in the synthesis of TMOs will be presented in the forthcoming discussions.

3.3 Colloidal-template method

The enhancement of the thermal stability of the polymer-based template and the restriction of the crystal growth during heat treatment are the key challenges faced in the soft-template method, which has led to the introduction of another concept of the colloidal-template. In a colloidal-template, the stability of the template is enhanced by the incorporation of inorganic components into polymer templates for delivering increased thermal stability;⁸⁵ hence, it is an inorganic nanoparticle core bound with flexible polymer tails. Here, the polymer part acts as a soft-template for organizing the inorganic particles to form nanostructures *via* self-assembly, whereas the inorganic particles act as hard-template for the enhancement of stability and nanoconfinement for the production of desired materials;⁸⁶ thus, the colloidal-templates offer the synergetic features of both soft and hard templates. With the aid of colloidal-templates, ordered nanostructures with remarkable crystallinity and structural integrity with controlled pores have been formed, which can withstand higher calcination temperatures.⁸⁷ Moreover, highly crystalline mesoporous materials can be prepared without using thermal treatment by retaining their mesostructure.⁸⁸ This method has been employed for the



synthesis of a large number of TMOs nanoparticles and TMOs/noble metal hybrids for diverse applications.^{89–92}

4. Soft-template-assisted synthesis of TMOs

In the soft-template assisted synthesis of nanostructures, the inorganic precursor species are deposited on either the surface or the interior of the soft-template by techniques like precipitation, electrochemical methods, or similar routes for composing particles with well-defined sizes and shapes.⁹³ The soft-template is then removed by low-temperature chemical treatment or pyrolysis for obtaining the desired product.⁹⁴

4.1 The general synthetic strategy of the soft-template method

The addition of the precursor molecule into the soft-template can bring about considerable modifications in the self-assembly process leading to micelle/liquid crystal formation for templating the sol-gel process.⁹⁵ This regulation of the sol-gel conversion of the precursors during self-assembly is considered to be the decisive step in soft-template methods. The general steps included in the formation of nanostructures from the soft-template are discussed below and a schematic representation is given in Fig. 3.

4.1.1 Cooperative assembly. The initial step in the soft-template assisted synthesis is the assembly of precursors and templates, which occur concurrently in many of the reactions. This step is termed “cooperative assembly”, where small groups of precursors and soft matter held together by the above-mentioned weak intermolecular interactions are formed in the solution, which then integrates into bulkier structures.⁹⁴ Eventually, this leads to the development of a liquid-crystalline phase enclosed by the precursors, which precipitates out of the solution. By judiciously governing this cooperative assembly step, the desired product can be obtained by following the processes given below.

4.1.2 Liquid-crystal templating. This method resembles that of hard-template synthesis in that a liquid-crystalline phase is formed, cast around by the inorganic material. This process requires specific parameters such as a high concentration of surfactants, which induces higher viscosity in the solution, thereby limiting its application.⁹⁴ Nevertheless, it exists as an appropriate method for the preparation of certain products with precursors like chloroplatinic acid, which are compatible with these viscous solutions.⁹⁶

4.1.3 Evaporation-induced self-assembly (EISA). EISA is one of the most effective methods, especially for the synthesis of metal oxides. It involves three major steps:⁹⁷

(1) Preparation of a dilute solution (sol) consisting of the inorganic/organic precursors, soft-template and a volatile solvent with suitable stoichiometry.

(2) Dispersion of this solution over a vast area to attain the critical micelle concentration by evaporation of the solvent and the self-assembly of the precursors into the micelles owing to their feeble condensation, resulting in a state called a tunable steady state.

(3) Development of the final structure by further inorganic condensation of the precursor around the liquid crystalline phase.

The successful fabrication of the desired final structure is completed by further processing techniques for the removal of the soft-template without affecting the properties of the product. Soft-templates such as surfactants can be eliminated by several methods such as low-temperature chemical treatment, combustion or depolymerization *via* thermal treatment or calcination.⁴² Care should be taken in these steps to retain the properties of the product such as porosity and structure and appropriate methods should be adopted.

4.2 Synthesis of TMOs with a surfactant as the soft-template

Surfactants are amphiphilic molecules consisting of both hydrophobic and hydrophilic counterparts, which lower the surface and interfacial tension. Based on the charge possessed



Fig. 3 Schematic representation of the general steps included in the formation of nanostructures using the soft-template: cooperative assembly, liquid crystal (LC) templating and evaporation-induced self-assembly (EISA).



Table 2 Relationship between the packing parameters and micelle shapes and packing shapes

g value	The extent of the curvature of the interface between micelle and solution	Micelle shape	Packing shape
$g \leq 1/3$	High	Spherical micelle	Cone
$1/3 < g \leq 1/2$	Medium	Elliptical micelle	Truncated cone
$1/2 < g \leq 1$	Low	Disc-like micelle	Truncated cone
$g = 1$	Very low	Bilayers	Cylindrical

towards the negatively charged oxygen of the intermediate product resulted in the growth of petals in the desired direction for the formation of WO_3 microflowers. Li *et al.* came out with three interesting morphologies for nanostructured VO_2 for application in lithium-ion batteries. The three diverse morphologies, nanobelts, nanoflowers and nanoflakes, were obtained by making slight modifications in the CTAB surfactant used, along with the precursors and reaction time.¹⁰³ The effect of CTAB on the morphology of ZnO was also studied recently by Amrithavalli *et al.*, where different-shaped ZnO nanoparticles were synthesized by controlling the precursor: surfactant ratio.¹⁰⁴ The formation of ZnO nanoparticles was achieved by the interaction of the surfactant with the zincate oligomers formed in the course of the reaction and a change in morphology was observed from edge-rounded triangular to rough-edged quadrilateral with the increase in the precursor ratio. Thus, the introduction of the surfactant plays a major role in the morphology of the final products. It is also known that the concentration of the surfactant and the reaction conditions also regulates the shape of the micelles, which in turn acts as the predominant factor for the structure guiding properties of the surfactant.

Another important surfactant used for the fabrication of TMOs is PVP, which is categorized as a non-ionic surfactant. Wang *et al.* fabricated multi-shelled hollow spheres of Co_3O_4 as anode materials in lithium-ion batteries by PVP-assisted solvothermal method.¹⁰⁵ The optimal control of the concentration of PVP in the presence of cobalt glycolate resulted in the development of multilamellar PVP micelles with varying shell numbers, which caused the formation of multi-shelled hollow spheres on further annealing. Cao *et al.* utilized PVP for the synthesis of highly-ordered VO_2 hollow spheres.¹⁰⁶ PVP, in the initial stages, might serve as a shape controller or a ligand for the coordination of the precursor, but later forms spherical micelles for the adsorption of the precursor molecule. Under the control of the added PVP surfactant, the nanorods formed from the interconnected nanoparticles circle around to form the hollow microspheres. Cao *et al.* used the structure-directing properties of PVP for the preparation of polycrystalline belt-like ZnO nanowires.¹⁰⁷ Here, PVP serves as both a shape controller, by preventing the rapid growth of ZnO nanocrystallites into fractal morphologies, and as a controller of the ordered self-assembly of ZnO particles for the formation of ZnO belts. Triton-X and polyvinyl alcohol (PVA) are other important neutral surfactants employed for the synthesis of various TMOs. Dubal *et al.* synthesized MnO_2 thin-films by the surfactant-assisted electrodeposition method with the aid of Triton-X.¹⁰⁸

The presence of the surfactant resulted in the formation of uniform thin-films of higher surface area with increased numbers of pores. The surfactant also promoted the retainment of the amorphous character of the film with superhydrophilic nature. These features have resulted in the strengthening of the core by preventing capacity loss, leading to greater capacitance. Dubal *et al.* also extended their work to the synthesis of CuO thin-films grown on stainless steel substrates.¹⁰⁹ The surfactant-assisted chemical bath deposition method was employed for the synthesis, where Triton-X was used as the surfactant and the physicochemical properties of the obtained CuO were compared with the CuO synthesized using another surfactant, PVA. Though the product formed *via* both the surfactants retained the crystallinity, there was much variation in the surface properties and activity. PVA, with the help of its -OH group, was adsorbed on the smaller grown particles, leading to the formation of flower-like nanostructures, whereas, in the case of Triton-X, the nanoscale segregation of the organic-rich area on the growing CuO particles led to the formation of 3D curved flower-like nanostructures. Thus, the surface area and porosity of Triton-X aided CuO is greater than that of the PVA-aided CuO, which resulted in the enhanced electrochemical activity of Triton-X-synthesized CuO.

Another category of anionic surfactant used for the fabrication of TMOs is sodium dodecyl sulfate (SDS). Several TMOs with distinct properties have been synthesized using SDS as the soft-template. For example, Zhang *et al.* synthesized CuO nanoparticles, which were attached to SDS-functionalized macroporous carbon.¹¹⁰ The functionalization of macroporous carbon using the self-assembly of SDS resulted in the proper dissolution of macroporous carbon and the uniform and firm incorporation of CuO nanoparticles onto its surface by taking advantage of the electrostatic interactions between negatively charged SDS and Cu^{2+} ions. The surfactant-assisted synthesis of oxides of copper had also been studied by Lou *et al.*, where Cu_2O crystals were obtained by the surfactant-assisted electrodeposition method.¹¹¹ Here, SDS acted as both a mediator for inhibiting the aggregation of the products and a shape regulator for the fabrication of polyhedral crystals. Apart from these, there are a number of surfactants such as polyethylene glycol (PEG), tetraoctylammonium bromide (TOAB) and tetrabutylammonium hydroxide (TBAOH), which are used for the synthesis of TMOs. Akram *et al.* came up with a low-temperature surfactant-assisted synthesis of sub-10 nm TiO_2 nanocrystals with excellent photocatalytic activity.¹¹² Here, polyethylene glycol (PEG) was used as the surfactant to restrict the agglomeration and facilitate the development of





Fig. 5 Typical examples of TMOs with diverse morphologies synthesized with the aid of different surfactants. SEM images of (a) Cu_2O hollow spheres. Reproduced from ref. 159 with permission from The Royal Society of Chemistry. (b) $\text{VO}_2(\text{B})$ nanobelts. Reprinted from Li N., Huang W., Shi Q., Zhang Y. and Song L., A CTAB-assisted hydrothermal synthesis of $\text{VO}_2(\text{B})$ nanostructures for lithium-ion battery application, *Ceram. Int.*, **39**, 6199–6206. Copyright (2013), with permission from Elsevier. (c) TiO_2 nanosheets. Reproduced from ref. 212 with permission from The Royal Society of Chemistry. (d) Urchin-like MnO_2 . Reprinted with permission from ref. 183. Copyright (2007) American Chemical Society. (e) WO_3 microflowers. Reprinted from Yayapao O., Thongtem T., Phuruangrat A. and Thongtem S., CTAB-Assisted Hydrothermal Synthesis of Tungsten Oxide Microflowers, *J. Alloys Compd.*, **509**, 2294–2299. Copyright (2010), with permission from Elsevier. (f) Multigonal star-shaped ZnO . Reprinted from Hu H., Deng C. and Huang X., Hydrothermal Growth of Center-Hollow Multigonal Star-Shaped ZnO Architectures Assembled by Hexagonal Conic Nanotubes, *Mater. Chem. Phys.*, **121**, 364–369. Copyright (2010), with permission from Elsevier. TEM images of (g) Cu_2O hollow spheres. Reproduced from ref. 159 with permission from The Royal Society of Chemistry. (h) CuO nanowire bundle. Reprinted from Li Y., Yang X. Y., Rooke J., Tendeloo G. Van and Su B. L., Ultralong $\text{Cu}(\text{OH})_2$ and CuO Nanowire Bundles: PEG200-Directed Crystal Growth for Enhanced Photocatalytic Performance, *J. Colloid Interface Sci.*, **348**, 303–312. Copyright (2010), with permission from Elsevier. (i) TiO_2 nanosheet. Reproduced from ref. 212 with permission from The Royal Society of Chemistry. (j) MnO_2 nanosheet. Reproduced with permission under a Creative Commons CC-BY License from Liu Z., Xu K., Sun H. and Yin S., One-Step Synthesis of Single-Layer MnO_2 Nanosheets with Multi-Role Sodium Dodecyl Sulfate for High-Performance Pseudocapacitors, *Small*, **2015**, **11**, 2182–2191. Copyright 2015 John Wiley and Sons. (k) ZnO hollow star. Reprinted from Hu H., Deng C. and Huang X., Hydrothermal Growth of Center-Hollow Multigonal Star-Shaped ZnO Architectures Assembled by Hexagonal Conic Nanotubes, *Mater. Chem. Phys.*, **121**, 364–369. Copyright (2010), with permission from Elsevier. (l) Cu_2O nanoflowers. Reprinted from Cao S., Chen H., Han T., Zhao C. and Peng L., Rose-like Cu_2O Nanoflowers via Hydrothermal Synthesis and Their Gas Sensing Properties, *Mater. Lett.*, **180**, 135–139. Copyright (2016), with permission from Elsevier.





Fig. 6 Diverse morphologies of the same TMOs synthesized using different surfactants. (a) V_2O_5 . Reprinted from Hu B., Cheng H., Huang C., Aslam M. K., Liu L., Xu C., Chen P., Yu D. and Chen C., The Controlled Study of Surfactants on the Morphologies of Three-Dimensional Turbine-like V_2O_5 for the Application of High-Performance Lithium-Ion Storage, *Solid State Ionics*, **342**, 115059, Copyright (2019), with permission from Elsevier. (b) NiO. Reprinted from Rajendran V. and Anandan K., Different Ionic Surfactants Assisted Solvothermal Synthesis of Zero-, Three and One-Dimensional Nickel Oxide Nanostructures and Their Optical Properties, 203–208, *Mater. Sci. Semicond. Process.*, **38**, Copyright (2015), with permission from Elsevier. (c) Cr_2O_3 . Reprinted from Anandan K. and Rajendran V., Sheet, Spherical and Plate-like Chromium Sesquioxide (Cr_2O_3) Nanostructures Synthesized via Ionic Surfactants-Assisted Facile Precipitation Method, *Mater. Lett.*, **146**, 99–102, Copyright (2015), with permission from Elsevier. (d) NiO. "Reprinted with permission from ref. 197. Copyright (2010) American Chemical Society."

chelating agent.¹²⁴ The decomposition of the soft-template and the intermediate resulted in the formation of the highly ordered crystalline ZnO with desired pores for the sensing application of ethanol. The peculiar morphologies of TMOs synthesized via various block copolymer-assisted methods are demonstrated in Fig. 8.

4.4 Synthesis of TMOs with biological soft-templates

The use of biological substances as soft-templates is an incredible concept and various research efforts have been conducted in this area to utilize the readily available and inexpensive biological materials for the fabrication of TMOs with





Fig. 7 Schematic diagram of the general synthesis strategy of the block copolymer-assisted synthesis of TMOs.

controlled properties. Different categories of biological materials, including biomolecules, unicellular organisms, and complex tissues can be used as soft-templates (Fig. 9).⁴² The naturally existing unique structure of this biological template outweighs the synthetically made templates, and facilitates exceptional chemistry on the template, promoting the formation of TMOs with controlled morphology, composition and crystal structure.⁵⁰ Biomolecules and their naturally existing clusters like peptides, proteins and DNA can serve as soft-templates for the synthesis of TMOs. Peptides or protein template synthesis, owing to its simplicity, better biocompatibility and high control over the crystallinity have been widely employed for the fabrication of TMOs. Kim *et al.* came out with a rational synthesis technique for the synthesis of WO_3 nanofibres functionalized with nanoparticles by using a polar protein nanocage as the template.¹²⁵ The protein nanocage consisted of a hollow apoferritin for enclosing the nanoparticles whose bulk agglomeration was hindered by the protein template. The further decomposition of the protein shell resulted in the formation of WO_3 microstructures with desired pores for the detection of trace amounts of biomarkers in the exhaled breath. Kashyap *et al.* visualized the nucleation process of iron oxide nanoparticles facilitated by an acidic bacterial recombinant protein, Mms6.¹²⁶ The protein self-assembly and the micellar iron-binding inhibited the nucleation outside the micelles and the aggregation during the nucleation, thus providing an easy pathway for further nucleation by lowering the energy barrier. Han *et al.* employed bovine-serum albumin (BSA) as the protein-template for the fabrication of 2D MnO_2 nanoflakes for the colorimetric detection of glucose.¹²⁷ Since MnO_2 nanoflakes exhibit dual enzyme activity, a one-pot

strategy for the detection of glucose without the aid of enzymes was also introduced.

DNA molecules have also been distinguished for their ability to act as templates for nanostructure fabrication. As a reference, Hassanien *et al.* employed DNA strands as the templates for the self-assembly of conductive Cu_2O nanowires without the requirement of complex techniques.¹²⁸ Here, Cu_2O nanostructures were assembled on λ -DNA by undergoing a chemical reduction in the presence of Benedict's reagent and ascorbic acid. The transformation of Cu_2O from nanoparticles to nanowires was studied by means of atomic force microscopy, and it was observed that the Cu_2O nanoparticles attached initially to λ -DNA formed a beads-on-a-string morphology, which gradually transformed into smooth nanowires with time. Ede *et al.* developed a new approach for the evolution of different-shaped β - MnO_2 on a DNA scaffold for catalytic and supercapacitor applications.¹²⁹ Here, two diverse morphologies, wire-like and flake-like, were obtained by adjusting the DNA:precursor molar ratio along with the reaction conditions in the aqueous phase synthesis. The reaction advanced *via* the adsorption of the Mn^{2+} ions on the negatively charged phosphate group of DNA to form the Mn^{2+} -DNA complex, which later got converted into MnO_2 in the presence of NaOH and atmospheric oxygen. The lower concentration of DNA facilitated the MnO_2 particles growing into helical DNA structures, resulting in wire-like MnO_2 , whereas a higher concentration of DNA facilitated the aggregation of MnO_2 particles into flake-like structures.

Other important categories of biological templates include bacterial, viral, and fungal templating. These are certainly promising techniques for the large-scale formation of different micro/nanostructures. One of the typical bacteria used as bio-templates is *Bacillus subtilis*, which is a Gram-positive species.





Fig. 8 Typical examples of TMOs with diverse morphologies synthesized with the aid of different block copolymers. SEM images of (a) Cu_2O mesoporous spheres. Reproduced from ref. 252 with permission from The Royal Society of Chemistry. (b) MnO_2 nanorods. Reprinted from Nayak P. K. and Munichandraiah N., Rapid Sonochemical Synthesis of Mesoporous MnO_2 for Supercapacitor Applications, *Mater. Sci. Eng., B*, **177**, 849–854, Copyright (2012), with permission from Elsevier. (c) TiO_2 nanosheets. Reprinted from Sheng L., Liao T., Kou L. and Sun Z., Single-Crystalline Ultrathin 2D TiO_2 Nanosheets: A Bridge towards Superior Photovoltaic Devices, *Mater. Today Energy*, **3**, 32–39, Copyright (2017), with permission from Elsevier. (d) WO_3 nanosheet. Reprinted from Wang Z., Wang D. and Sun J., Controlled Synthesis of Defect-Rich Ultrathin Two-Dimensional WO_3 Nanosheets for NO_2 Gas Detection, *Sens. Actuators, B*, **245**, 828–834, Copyright (2017), with permission from Elsevier. (e) Spherical ZnO nanorods. Reproduced from Amornpitoksuk P., Suwanboon S., Sangkanu S., Sukhoom A. and Muensit N., Morphology, Photocatalytic and Antibacterial Activities of Radial Spherical ZnO Nanorods Controlled with a Diblock Copolymer, *Superlattices Microstruct.*, **51**, 103–113, Copyright (2012), with permission from Elsevier. (f) Fe_3O_4 nanoroses. Reprinted with permission from ref. 262. Copyright (2010) American Chemical Society. TEM images of (g) Cu_2O mesoporous spheres. Reproduced from ref. 252 with permission from The Royal Society of Chemistry. (h) MnO_2 nanorods. Reprinted from Nayak P. K. and Munichandraiah N., Rapid Sonochemical Synthesis of Mesoporous MnO_2 for Supercapacitor Applications, *Mater. Sci. Eng., B*, **177**, 849–854, Copyright (2012), with permission from Elsevier. (i) Co_3O_4 nanosheet. Reprinted from Yang J., Gao M., Lei J., Jin X., Yu L. and Ren F., Surfactant-Assisted Synthesis of Ultrathin Two-Dimensional Co_3O_4 Nanosheets for Applications in Lithium-Ion Batteries and Ultraviolet Photodetector, *J. Solid State Chem.*, **274**, 124–133, Copyright (2019), with permission from Elsevier. (j) WO_3 nanosheet. Reprinted from Wang Z., Wang D. and Sun J., Controlled Synthesis of Defect-Rich Ultrathin Two-Dimensional WO_3 Nanosheets for NO_2 Gas Detection, *Sens. Actuators, B*, **245**, 828–834, Copyright (2017), with permission from Elsevier. (k) Spherical ZnO nanorods. Reproduced from Amornpitoksuk P., Suwanboon S., Sangkanu S., Sukhoom A. and Muensit N., Morphology, Photocatalytic and Antibacterial Activities of Radial Spherical ZnO Nanorods Controlled with a Diblock Copolymer, *Superlattices Microstruct.*, **51**, 103–113, Copyright (2012), with permission from Elsevier. (l) Urchin-like MnO_2 hollow nanosphere. Reproduced from ref. 268 with permission from the Centre National de la Recherche Scientifique (CNRS) and The Royal Society of Chemistry.





Fig. 9 Schematic representation of the biological template-assisted synthesis of TMOs.

Shim *et al.* used this bacteria as a template for the fabrication of rod-type porous Co_3O_4 with enhanced surface area and electrochemical activity.¹³⁰ The electrostatic synergy between the functional surfaces of the bacteria and the cobalt ions was used for the uniform crystallization of Co_3O_4 on the surface of the bacterial template. They further applied this strategy using the same bacteria for the synthesis of rod- and hollow-shaped manganese oxide.¹³¹ Their group also reported 3D flower-like porous Co_3O_4 using another Gram-positive bacteria, *Micrococcus lylae*.¹³² Chen *et al.* employed bacterial cellulose as a template for the polyol synthesis of ZnO nanoparticles with enhanced mechanical properties and photocatalytic activity.¹³³ The template served as both a matrix for the quantitative inclusion of Zn^{2+} ions and as a nanoreactor for the formation of ZnO nanoparticles. Zhou *et al.* introduced the facile synthesis of hollow TiO_2 with lactobacillus bacteria as a template.¹³⁴ The use of two different species of lactobacillus resulted in the generation of diverse hollow-sphere and tube-like TiO_2 . The use of a virus as a soft-template for the synthesis of TMOs has also been a widely explored research area and various studies have been conducted in this field in the past decade. One of the most important viruses among them is the tobacco mosaic virus (TMV), which is a plant virus possessing a unique cylindrical shape with a high aspect ratio. Chu *et al.* adopted the TMV virus for the fabrication of hierarchical Ni/NiO electrodes with exceptional charge capacity.¹³⁵ The synthesis was facilitated by the thermal oxidation of electroless Ni-coated TMV, followed by a simple annealing technique. The role of TMV in the synthesis of ZnO nanostructures was studied by Atanasova *et al.*, where self-assembled TMV was used for the controlled synthesis of uniform ZnO nanowires as field-effect transistors.¹³⁶ TMV functioned both as a template for structure direction and as a facilitator for the

electron injection, resulting in the tuning of the properties of the semiconducting layer of the transistor. Another significant group of viruses used as bio-templates is M13, with excellent genetic tenability and replicability. Han *et al.* used genetically engineered M13 phages for the synthesis of MnO_2 nanowires for electrochemical sensing applications.¹³⁷ The nucleation and successive growth of MnO_2 crystals into nanowires was regulated by the surface charge of M13 mutants. The negatively charged M13 phages ended in the uniform distribution of MnO_2 crystals on the surface of the phage resulting in the formation of nanowires, whereas the weakly negative and positively charged phages ended in the irregular agglomeration of MnO_2 . Virus-templated synthesis was also used for the synthesis of mixed TMOs. Oh *et al.* came out with the M13-assisted synthesis of mixed TMOs, cobalt manganese oxide nanowires for lithium–oxygen batteries.¹³⁸ With the aid of the M13 template, three diverse compositions of cobalt manganese oxide were prepared with diverse morphologies. The cobalt dominated systems exhibited a particle shape growth and the manganese dominated systems exhibited a planar shape growth. Though less reported, fungus-templated synthesis has also been employed for the preparation of TMOs. As a reference, He *et al.* synthesized hollow Cu_2O microspheres by using yeast as a bio-template.¹³⁹ The reaction proceeded *via* the formation of yeast/ Cu^{2+} core–shell spheres by the coulombic interactions between negatively charged functional groups of the template and positively charged ions in the precursor. The yeast/ Cu^{2+} core–shell spheres subsequently formed Cu_2O spheres on the addition of OH^- and glucose solution. Some of the explicit examples for the attainment of specific morphology with the aid of biological templates are presented in Fig. 10.





Fig. 11 Schematic diagram of the liquid/liquid interface-assisted synthesis of 0D, 1D, 2D or 3D TMOs.

4.5 Synthesis of TMOs with interfaces as the soft-template

Interface-assisted synthesis is an emerging versatile approach for the controlled synthesis of nanostructures with tuned morphologies. Interfaces adopted in chemical reactions include air/liquid, air/solid, liquid/liquid and liquid/solid, of which the liquid/liquid interface-assisted synthesis predominates in the fabrication of TMOs.¹⁴⁰ The interface is acclaimed to afford a unique non-equilibrium environment with enhanced surface energy for the assembly of precursors to initiate the nucleation, leading to further growth of the intermediates to form the final structure with desired morphology (Fig. 11).¹⁴⁰ The simple synthesis procedure without the requirement of the removal of the template along with the retainment of high crystallinity has driven the liquid/liquid interface-assisted synthesis to be a major technique for the fabrication of TMOs.

Though there are fewer reports on the interface-assisted synthesis of TMOs, the last few decades have observed advancements in this area of synthesis. Wang *et al.* made use of the water/air interface for the fabrication of nanometer-thick large-area and single-crystalline ZnO nanosheets.¹⁴¹ The growth of nanosheets was facilitated by the epitaxial growth directed by the surfactant monolayer at the water/air interface. Among several interfaces, the liquid/liquid interface has been widely exploited for the synthesis of TMOs. The diverse morphologies of TMOs synthesized using the liquid/liquid interface as the soft-template are given in Fig. 12. Cheng *et al.* employed a benzene/water interface-aided solvothermal reaction for the synthesis of durian-like α -Fe₂O₃ hollow spheres for the removal of chromium in water.¹⁴² The presence of the liquid/liquid interface played a significant role along with PVP in the development of hierarchical hollow spheres consisting of single-crystalline polyhedral particles. The formation of





Fig. 12 Typical examples of TMOs with diverse morphologies developed using interface-assisted synthesis. SEM images of (a) MnO nanoparticles. Reproduced from ref. 143 with permission from The Royal Society of Chemistry. (b) MnO microbars. Reproduced from ref. 143 with permission from The Royal Society of Chemistry. (c) Urchin-like MnO₂. "Reprinted with permission from ref. 144. Copyright (2014) American Chemical Society." (d) CuO nanospheres. Reprinted from Zhu J. and Qian X., From 2-D CuO Nanosheets to 3-D Hollow Nanospheres: Interface-Assisted Synthesis, Surface Photovoltage Properties and Photocatalytic Activity, *J. Solid State Chem.*, **183**, 1632–1639, Copyright (2010), with permission from Elsevier. (e) Cauliflower-like Fe₂O₃. Reprinted from Cheng X. L., Jiang J. Sen, Jin C. Y., Lin C. C., Zeng Y. and Zhang Q. H., Cauliflower-like α -Fe₂O₃ Microstructures: Toluene-Water Interface-Assisted Synthesis, Characterization, and Applications in Wastewater Treatment and Visible-Light Photocatalysis, *Chem. Eng. J.*, **236**, 139–148, Copyright (2014), with permission from Elsevier. (f) Durian-like Fe₂O₃. Reproduced from ref. 142 with permission from The Royal Society of Chemistry. TEM images of (g) urchin-like MnO₂. "Reprinted with permission from ref. 144. Copyright (2014) American Chemical Society." (h) Cauliflower-like Fe₂O₃. Reprinted from Cheng X. L., Jiang J. Sen, Jin C. Y., Lin C. C., Zeng Y. and Zhang Q. H., Cauliflower-like α -Fe₂O₃ Microstructures: Toluene-Water Interface-Assisted Synthesis, Characterization, and Applications in Wastewater Treatment and Visible-Light Photocatalysis, *Chem. Eng. J.*, **236**, 139–148, Copyright (2014), with permission from Elsevier. (i) Durian-like Fe₂O₃. Reproduced from ref. 142 with permission from The Royal Society of Chemistry.

microrods in the absence of the benzene/water interface affirmed the requirement of the interface for the α -FeOOH to α -Fe₂O₃ phase transformation. Chen *et al.* proposed the tuning of the morphology of MnO by altering the solvent ratio of the *N,N*-dimethylformamide (DMF)/water bisolvent interface.¹⁴³ The alteration of the solvent ratio resulted in the formation of four different morphologies, including microrods, 2D flakes, microbars and irregular microparticles, out of which 2D flakes exhibited greater electrochemical activity. Liu *et al.* introduced a facile soft-interface method consisting of a dichloromethane/

water interface for the synthesis of hierarchical mesoporous MnO₂.¹⁴⁴ The as-synthesized MnO₂ exhibited an urchin-like hollow submicrosphere formed by the self-assembly of 1D nanorods at the bisolvent interface. Zhu *et al.* proposed the synthesis of hierarchical CuO hollow nanostructures with enhanced photocatalytic activity.¹⁴⁵ Here, the *n*-octanol/water interface along with the bubble template were used for the assembly of nanosheets into hollow nanostructures, which were formed *via* aggregation followed by the growth mechanism through the bubble template. The use of the liquid/liquid



Table 3 0D, 1D, 2D and 3D TMOs synthesized using different soft-templates

Soft-template category	TMOs	Soft-template used	Dimension	Morphology	Reference	
Surfactant	Co ₃ O ₄	Oleic acid	0D	Nanoparticles	147	
		CTAB		Nanoparticles	148	
		CTAB, PEG	1D	Nanowires	149	
		CTAB		Nanorods	149	
		CTAB	3D	Flower-like microcrystal	150	
	Cr ₂ O ₃	PEG	0D	Nanosphericals	151	
		SDS	2D	Nanosheet		
		CTAB		Nanoplates		
	CuO	CTAB	0D	Nanoparticle	152	
		CTAB/SDS		Spherical nanoparticles	153	
		TOAB	1D	Nanoribbon	113	
		PEG		Nanowire bundle	154	
		PEG		Belt-like	155	
		CTAB		Bamboo leaf-like		
		Cu(DS) ₂	2D	Leaf-like	156	
		Triton-X		Yarn balls	157	
		PVA		Cabbage slash		
		SDBS	3D	Butterfly-like	158	
	Cu ₂ O	CTAB	0D	Hollow sphere	159	
		CTAB	1D	Nanobelt	160	
		CTAB		Nanowhiskers	161	
		SDS		Nanowire	162	
		CTAB	3D	Rose-like nanoflower	163	
		CTAB		Nanoflower	164	
	Fe ₂ O ₃ /Fe ₃ O ₄	CTAB/PVP/SDBS	0D	Nanoparticle	165	
		SDBS		Nanoparticle	166	
		CTAB	1D	Nanowire	167	
		TEAOH		Nanorod	168	
		SDS	2D	Nanolamellar	169	
		TETA	3D	Nanoflower	170	
		MnO ₂ /Mn ₃ O ₄ /Mn ₂ O ₃	CTAB/PVP	0D	Nanocrystals	115
			CTAB		Nano hollow sphere	171
	CTAB			Nanoparticle	172	
	PEG		1D	Nanorod	173	
	SDS			Nanorod	174	
	PVP			Nanofiber	175	
	SDBS			Nanowire	176	
	CTAB			Nanorod	177	
	SDS		2D	Nanosheet	178	
	PVP		3D	Birnessite	179	
	Oleylamine			3D sphere	180	
	SDS			Colloidal sphere	181	
CTAB			Microflower	182		
SDS			Sea-urchin	183		
MoO ₂ /MoO ₃	PEG		0D	Sub-micron sphere	184	
	PEG		Hollow microsphere	185		
	CTAB/chitosan	1D	Nanorod	186		
	SDS		Nanoneedle	187		
	SDS		Nanorod	188		
	Cetyl alcohol		Nanobelt	189		
	Gemini surfactant	2D	Lamellar	190		
	CTAB		Microbelt	186		
	CTAB	3D	Urchin	191		
	PVP		Flake-flower	192		
	CTAB		Flower	193		
	CTAB		Nest-like	194		
NiO	PEG		Porous sponge	195		
	CTAB	0D	Spherical particles	196		
	CTAB/Triton-X		Ball-like	197		
	PEG		Microsphere	198		
	PEG		Nanoparticle	199		
	PEG	1D	Rod-like	196		
	CTAB		Pine-cone	200		



Table 3 (Contd.)

Soft-template category	TMOs	Soft-template used	Dimension	Morphology	Reference
	TiO ₂	SDS		Nanotubule	201
		SDS	2D	Nanoflake	197
		P123		Nanosheet	202
		SDS	3D	Hexagonal	196
		SDS		Microflower	203
		PVP		Nanoflower	204
		PEG	0D	Nanocrystal	112
		CTAB/PEG		Particles	205
		CTAB		Nanoparticles	206
		PEG/PVP		Nanocrystal	
		CTAB	1D	Nanofiber	207
		CTAB		Nanorod	208 and 209
		PEG		Nanowire	210
		TBAOH	2D	Nanosheet	114
		DDA		Nanosheet	211
	CTAB		Nanosheet	212	
	CTAB	3D	Dendrites	213	
	CTAB		Nanoflower	214	
	PEG		3D mesoporous	215	
	VO ₂ /V ₂ O ₅	CTAB	0D	Nanoparticle	216
		CTAB	1D	Nanobelt	103
		CTAB		Nanorod	216
		PVP		Nanowire	217
		Brij 30		Nanoribbon	218
		Triton-X		Nanowire	
		PVP		Nanorod	219
		PEG		Nanobelt	220
LAHC		2D	Nanosheet	221	
CTAB			Nanosheet	222	
Cetyl alcohol			Nanoflake	189	
PVP		3D	Hollow microsphere	106 and 223	
PVP/CTAB/SDS/SDBS		Turbine-like structure	224		
PVP		Hydrogel	217		
PVP		Flower-like	225		
SDS		3D network	226		
WO ₃	CTAB	1D	Nanorod	227 and 228	
	HTAB/PEG	2D	Thin film	229	
	PEG		Nanosheet	227	
	Triton-X		Disc-shaped	230	
	CTAB		Nanoplates	231	
	CTAB	3D	Microflower	102	
	CTAB		Urchin-like	232	
ZnO	CTAB	0D	Nanoparticles	104	
	PVP	1D	Nanobelt	107	
	Sodium oleate/hydrazine hydrate		Nanorod	233	
	CTAB		Nanotube	234	
	CTAB		Pencil-like nanorod	235	
	PEG		Nanowire	236	
	PEG		Micro/nanorod	237	
	SDS/PEG		Nanorod	238	
	SDS		Needle-like	239	
	SDS	2D	Nanosheet	240	
	CTAB		Nanoflake	241	
	Oleic	3D	Flower-like	242	
	SDS/PEG		Flower-like microsphere	240	
	Triton-X		Flower-like	243	
	PEG/PVP		Multigonal star-shape	244	
	CTAB		Flower-like	245	
	SDS		Flower-like	246	
Block copolymer	Co ₃ O ₄	P123	0D	Nanocrystals	247
		PS-PMA-PEG		Hollow nanosphere	248
		P123	1D	Channel structure	249
		P123	2D	Nanosheet	250 and 251



Table 3 (Contd.)

Soft-template category	TMOs	Soft-template used	Dimension	Morphology	Reference
	CuO/Cu ₂ O	P123	0D	Mesoporous spheres	252
		PS- <i>b</i> -PAA- <i>b</i> -PEG		Hollow nanosphere	253
	Fe ₂ O ₃ /Fe ₃ O ₄	P123	1D	Nanowire bundle	254
		P123		Nanowire	255
		PS- <i>b</i> -PEO	0D	Nanodot	256
		F127		Spherical nanoparticles	165
		PEO-PPO-PEO		Mesoporous spheres	257
		F127		Hollow microsphere	258
		Cellulose- <i>g</i> -(PAA- <i>b</i> -PS)	1D	Nanorod	259
		PIB-PEO	2D	Film	260
	MnO ₂ /Mn ₃ O ₄ /Mn ₂ O ₃	P123	3D	Rhombohedral	261
		P123		Nanorose	262
		P123	0D	Particles	263
		Pebax2533		Nanoparticles	264
		P123	1D	Nanorod	263
		P123		Nanowire	265
		PMPEGMA- <i>b</i> -PBA	2D	Film	266
		P123	3D	Spherical particle	267
		F127		Urchin-like	268
		F127		Flower-like	268
MoO ₂ /MoO ₃	PS-PMAPTAC-PEO	0D	Hollow nanosphere	269	
	PS- <i>b</i> -PVP- <i>b</i> -PEO	1D	Nanotube	270	
	KLE	2D	Film	121	
NiO	P123	0D	Nanoparticles	271	
	PS- <i>b</i> -P2VP	2D	Film	272	
TiO ₂	PS- <i>b</i> -P4VP	0D	Nanoparticles	273	
	PDMAEMA- <i>b</i> -PFOMA		Nanoparticles	274	
	F127		Microspheres	275	
	PS(46)-PMMA(21)	1D	Nanorod	276	
	PS- <i>b</i> -P4VP		Needle-like	277	
	P123		Hollow fiber	278	
	PS(80)-PMMA(80)	2D	Nanowall	276	
	P123		Nanosheet	279, 280 and 281	
WO ₃	P123/F127		Thin film	282	
	PS-PMAPTAC-PEO	0D	Hollow nanosphere	269	
	PS- <i>b</i> -PAA		Nanosphere	283	
	PS- <i>b</i> -P4VP	1D	Nanowire	284	
	P123	2D	Nanosheet	285 and 286	
	F127	3D	3D nanowall	287	
	PtBA ₄₅ PS ₉₂ PEO ₁₁₇		Peanut-shaped	288	
ZnO	PS- <i>b</i> -PAA	0D	Nanoparticles	289	
	PS- <i>b</i> -P2VP		Nanoparticles	290	
	P123	1D	Nanotube	291	
	PEO- <i>b</i> -PPO		Biprismatic shape	292	
	PEO- <i>b</i> -PPO		Nanorod	293	
	PS- <i>b</i> -P4VP	2D	Quasi hexagonally packed nanorings	294	
	F127	3D	Flower-like	295	
Biological templates	Co ₃ O ₄	Bacteria	1D	Hollow rod	130
		Bacteria	3D	Flower-like microsphere	132
Cu ₂ O	Yeast	0D	Hollow microsphere	139	
	DNA	1D	Nanowire	128	
	Bacterial cellulose	3D	Microtube lattices	296	
Fe ₂ O ₃ /Fe ₃ O ₄	Protein	0D	Nanoparticles	126	
MnO ₂ /Mn ₃ O ₄ /Mn ₂ O ₃	Bacteria	1D	Rods	131	
	DNA		Wire-like	129	
	Virus		Nanowire	137	
	DNA	2D	Flake-like	129	
	Protein		Nanoflake	127	
	l-Cysteine	0D	Spherical hedgehog	297	
NiO	Virus	1D	Nanorod	135	
	Fungi		Microtube	298	
TiO ₂	Alginate	0D	Nanoparticles	299	
	Nanocrystalline cellulose		Nanoparticles	300	



Table 3 (Contd.)

Soft-template category	TMOs	Soft-template used	Dimension	Morphology	Reference
Interface	V ₂ O ₅ WO ₃ ZnO	Protein fibrils	1D	Nanowire	301
		Yeast	0D	Nanoparticles	302
		Protein	1D	Nanofiber	125
		Bacterial cellulose	0D	Nanoparticles	133
		Virus	1D	Nanowire	136
		Rice	2D	Flake	303
		<i>n</i> -Octanol/water	3D	Hollow nanosphere	145
Interface	CuO Fe ₂ O ₃ MnO/MnO ₂ ZnO	Benzene/water	3D	Durian-like microsphere	142
		Toluene/water		Cauliflower-like	304
		DMF/water	0D	Nanoparticle	143
				Microbar	143
				Flake	143
		CH ₂ Cl ₂ /water	3D	Urchin-like	144
		Water/air	2D	Nanosheet	141

interface has also been extended to the formation of mixed TMOs. Our group has developed a bi-solvent interface-assisted synthesis mechanism for the preparation of one of the important mixed TMOs, nickel cobaltite.¹⁴⁶ Here, the *n*-butanol/water interface was employed for the synthesis of 1 nm thick flake-like nickel cobaltite with a single unit cell. The effect of the variation of the *n*-butanol/water solvent ratio on the morphology of the mixed TMOs was also studied in detail, where it was proved that

the 1 : 1 solvent ratio predominates in the morphology and physicochemical properties. Though the interface-assisted method for the fabrication of TMOs is less investigated, it is a promising technique for the future synthesis of TMOs.

A consolidated list of TMOs synthesized in the last decade using different soft-templates along with their morphology tuning is presented in Table 3.



Fig. 13 (a) Schematic representation of the degradation of MB dye using α -MoO₃ nanostructures. Photocatalytic degradation of MB dye solution over synthesized α -MoO₃ nanostructures. (b) Curves of C/C_0 versus time interval. (e) Degradation efficiency versus time. FESEM images of α -MoO₃ (c) nanoplates and (d) nanoneedles. Reprinted from Rathnasamy R. and Alagan V., A Facile Synthesis and Characterization of α -MoO₃ Nanoneedles and Nanoplates for Visible-Light Photocatalytic Application, *Phys. E*, 102, 146–152, Copyright (2018), with permission from Elsevier.



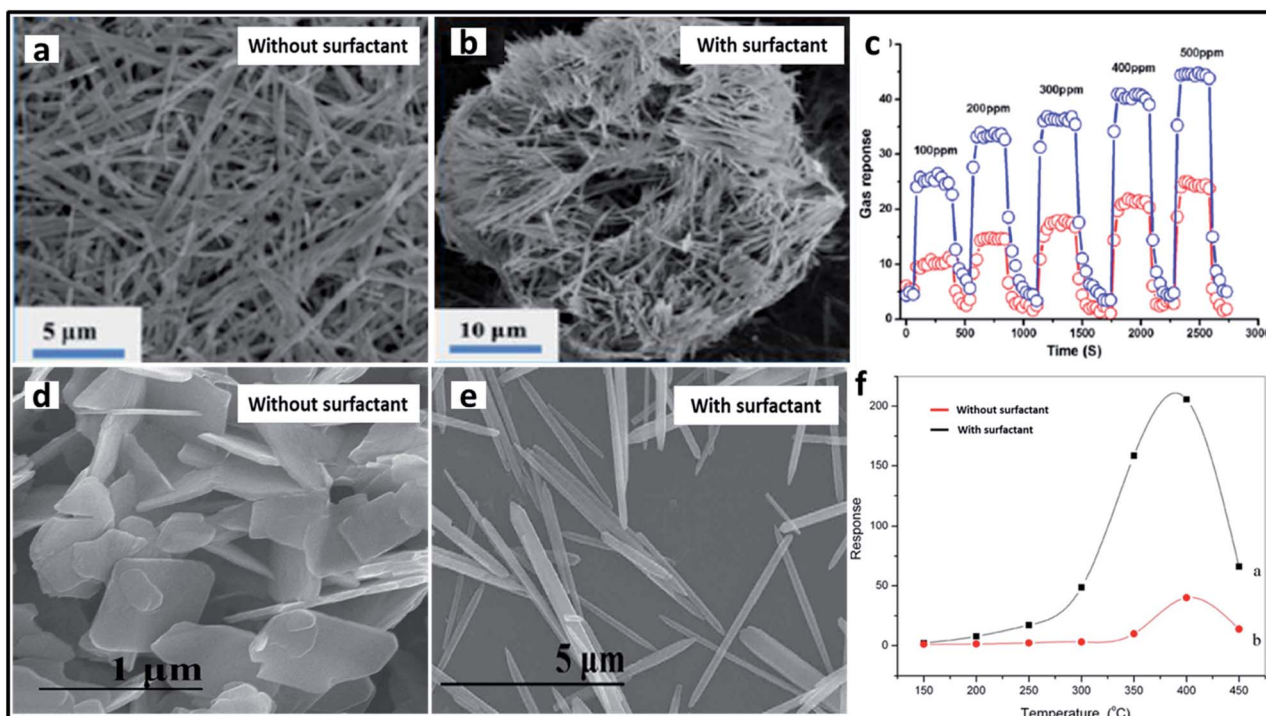


Fig. 14 SEM images of (a) monodispersed MoO_3 nanobelts synthesized without the use of surfactants; (b) nest-like MoO_3 synthesized using CTAB, (c) ethanol concentration vs. response property at 340°C . Blue lines represent nest-like MoO_3 and red lines represent monodispersed MoO_3 nanobelts. Reprinted from Li T., Zeng W., Zhang Y. and Hussain S., Nanobelt-Assembled Nest-like MoO_3 Hierarchical Structure: Hydrothermal Synthesis and Gas-Sensing Properties, *Mater. Lett.*, **160**, 476–479, Copyright (2015), with permission from Elsevier. SEM images of (d) ZnO nanosheets synthesized via the precipitation method and (e) ZnO nanorods synthesized using CTAB. (f) The effects of respective ZnO nanostructures on the response of 40 ppm NO_2 . Reprinted from Bai S., Liu X., Li D., Chen S., Luo R. and Chen A., Synthesis of ZnO Nanorods and Its Application in NO_2 Sensors, *Sens. Actuators, B*, **153**, 110–116, Copyright (2011), with permission from Elsevier.

5. Applications of TMOs synthesized via soft-templates

Over the last few decades, TMO materials have been broadly explored as noteworthy materials for a wide range of applications. The unique characteristics possessed by TMOs such as exquisite electrochemical activity, easy availability, wider band gaps and environmental compatibility render them the ability to serve as a promising candidate for a large number of applications. The soft-template-assisted method aids the synthesis of TMOs with desired properties such as controlled pore size, crystallinity and phase.⁴⁵ TMOs designed using soft-templates maintain well-defined crystal shapes, thereby enhancing their activity towards various applications to a greater extent.¹⁵⁴ The mesoporous structure synthesized by the use of soft-templates induces higher surface area and structure stability, which are the major requirements for various applications such as catalysis and electrochemical applications.³⁰⁵ However, controlling the shape and structure of the synthesized TMOs is rather important for tuning the exposed facets for facilitating different applications and soft-template-assisted synthesis has proved to be an important synthesis strategy for the development of shape and structure controlled TMOs.²¹² Some of the far-reaching applications of TMOs synthesized via soft-template are discussed briefly below:

5.1 Catalysis

One of the most researched applications that focus on the intrinsic properties of TMOs is catalysis, including photocatalysis and electrocatalysis. Various TMOs with diverse morphologies have been employed as catalysts, among which TiO_2 is the most abundant TMO to be used as a catalyst. TiO_2 materials with different morphologies synthesized by using different surfactants have been used for the photocatalytic degradation of many dyes. CTAB-assisted synthesis has resulted in the formation of diverse morphologies such as nanosheets, nanoflowers and nanorods, which have been used for the photocatalytic degradation of azo dyes,²¹² rhodamine²¹⁴ and for the photodecomposition of wastewater,²⁰⁸ respectively. The photocatalytic degradation of azo dye has also been studied by Patil *et al.* using a block copolymer F127-assisted synthesis of TiO_2 films.³⁰⁵ Zhang *et al.* synthesized TiO_2 nanowires using the surfactant PEG for the photocatalytic degradation of a hazardous herbicide, atrazine which is released into the soil.²¹⁰ CuO is another highly studied material for catalytic applications. Rao *et al.* studied the effects of two different surfactants, CTAB and SDS on the morphology of CuO nanoparticles and its ability in the photocatalytic degradation of methylene blue.³⁰⁶ CuO nanowire bundles were prepared by Li *et al.* using PEG for the photodegradation of rhodamine.¹⁵⁴ Certain other surfactant-assisted synthesized TMOs used for catalytic applications include Mn_3O_4 nanorods¹⁷³ and $\alpha\text{-MoO}_3$



nanoneedles¹⁸⁷ for the catalytic degradation of methylene blue. A schematic representation of the degradation of methylene blue dye using α - MoO_3 and the influence of the surfactant in the catalytic activity is provided in Fig. 13. Bai *et al.* studied the effect of the surfactants CTAB and PEG on the synthesis of Co_3O_4 for the catalytic combustion of toluene.¹⁴⁹ Apart from the surfactant, the interface has also been used as a soft-template for the synthesis of TMOs for catalytic applications. Cheng *et al.* synthesized flower-like Fe_2O_3 using a toluene–water interface for the effective photocatalytic degradation of rhodamine B.³⁰⁴

5.2 Sensors

TMOs have been used as effective sensors for the trace detection of large numbers of molecules. One of the important sensing applications of TMOs is comprised of the gas and humidity sensing. A plethora of research has been conducted in this field for the development of novel materials with the lowest detection limits. The major contribution in the field of gas sensing comes from molybdenum oxides. Various morphologies of molybdenum oxides have been developed using different soft-templates for the gas sensing application. As a reference, Xia *et al.* synthesized nanorod-assembled MoO_3 sponges with the aid of the surfactant PEG with enhanced gas sensing properties toward ethanol.¹⁹⁵ The gas sensing ability towards ethanol has also been studied using other molybdenum oxides. MoO_2 sub-

micron spheres synthesized using PEG,¹⁸⁴ flower-like MoO_3 synthesized using PVP,¹⁹² and nest-like MoO_3 synthesized using CTAB¹⁹⁴ were used as effective gas sensors with a high response towards ethanol (Fig. 14a–c). Some other surfactant-assisted synthesized TMOs gas sensors towards ethanol sensing include flower-like NiO synthesized using PVP,²⁰⁴ urchin-like WO_3 synthesized using CTAB,²³² and ZnO nanorods synthesized using SDS/PEG.²³⁸ Another important analyte in the gas sensing application is nitrogen oxide (NO_x). Cao *et al.* prepared 3D rose-like Cu_2O gas sensors using CTAB for the effective detection of NO_2 .¹⁶³ MoO_3 nanorods were fabricated by Bai *et al.* using SDS for the selective detection of NO_2 .¹⁸⁸ The sensing activity towards NO_x was also studied by Mehta *et al.* using WO_3 nanoplates synthesized using CTAB.²³¹ 1D ZnO nanorods were also employed for the detection of NO_2 , where ZnO nanorods synthesized using CTAB surfactant showed an enhanced response as compared to the ZnO nanosheets synthesized *via* the precipitation method (Fig. 14d–f).²³⁵ Nanocrystalline CuO was used as an efficient material for the sensing of ammonia gas. As an example, Bedi *et al.* studied the impact of the addition of CTAB on the preparation of thick CuO films for ammonia sensing.¹⁵² Their group also extended their studies to the use of other surfactant materials for the synthesis. In one study, they compared the role of SDS with that of CTAB in the effective sensing of ammonia and concluded that CuO prepared using CTAB surfactant showed a greater kinetic response towards ammonia.¹⁵³ TMOs have also been used for various



Fig. 15 (a) Schematic representation of a lithium metal battery using V_2O_5 nanomaterials and room temperature ionic liquid electrolyte. SEM images of diverse morphologies of V_2O_5 materials: (b) nanoribbons (c) nanowires (d) microflakes and their corresponding (e) specific capacities; (f) cycle life of V_2O_5 nanoribbon. "Adapted with permission from ref. 218. Copyright (2008) American Chemical Society."



types of humidity sensing. Gong *et al.* fabricated TiO₂ nanosheets with increased surface area and surface defects with the aid of P123 for the electrochemical detection of humidity.²⁸⁰ Parthibavarman *et al.* compared the sensing activity of WO₃ with different morphologies towards humidity, and concluded that WO₃ nanosheets prepared using PEG are preferred to the nanorods prepared using CTAB.²²⁷ The use of TMOs as sensors in biomedical applications has also gained major attention. TMOs have been used for the trace detection of various biomolecules, pathogens, and toxic compounds inside the human body. Among these, the detection of glucose has been widely studied in the past decade. Tanaka *et al.* reported mesoporous iron oxide synthesized using a block copolymer for both the colorimetric and electrochemical detection of glucose with a good detection limit.¹²² TMO nanostructures fabricated with the aid of biological templates were also used for the detection of glucose. For example, NiO prepared using L-cysteine²⁹⁷ and MnO₂ nanowires synthesized using M13 phages¹³⁷ were used for the successful selective and reproducible electrochemical detection of glucose. Studies were also conducted on the detection of toxic materials and pathogens in the human body. Zhu *et al.* came out with an effective chemiresistive sensor based on mesoporous WO₃ for the detection of a foodborne pathogen *Listeria monocytogenes*.¹²³ Ding *et al.* developed an amperometric sensor based on porous Mn₂O₃

nanofibres for the trace detection of one of the toxic compounds, hydrazine.¹⁷⁵ Kim *et al.* introduced an excellent sensor based on WO₃ nanofibres for the highly selective and ultra-trace detection of 10 different biomarker gases in the exhaled breath, which is a promising step in fabricating practical exhaled breath monitoring sensors.¹²⁵

5.3 Lithium ion batteries (LIB)

In light of the expeditious growth of the world economy, energy production and storage has become a critical issue. LIB is one of the leading candidates for energy storage, and developing electrodes with greater efficiency is a challenge. Of the numerous materials designed as electrodes for LIB, TMOs have been considered as competent materials owing to their excellent electrochemical performance. Vanadium oxides have gained greater attention in this area owing to their increased theoretical capacity, superior structural flexibility and reduced cost. Nanorod and flower-like VO₂ were developed by two different groups using the surfactant PVP as the cathode in LIB.^{219,225} Diverse nanostructures of V₂O₅ were also used as electrode materials for LIB. Cao *et al.* synthesized hollow V₂O₅ microspheres using PVP,²²³ and Chou *et al.* synthesized V₂O₅ with divergent morphologies using various types of surfactants, of which nanoribbons synthesized using Brij 30 served as an effective cathode material for LIB (Fig. 15).²¹⁸ Dai *et al.*



Fig. 16 Cyclic voltammograms of CuO: (a) Cu-bare, (b) Cu-TRX, and (c) Cu-PVA samples at different scan rates. The insets show their corresponding morphologies. (d) Plots of specific capacitance versus potential scan rates for corresponding CuO samples. Adapted from ref. 157 with permission from The Royal Society of Chemistry.





Fig. 17 (a) Schematic diagram of a dye-sensitized solar cell with a bifunctional photoanode consisting of a highly ordered mesoporous thin film and a scattering layer. (b) The thickness of P123 and F127-templated mesoporous TiO₂ thin-films as a function of the number of coatings; the insets present the low and high magnification cross-sectional images of a 30-layer P123-templated mesoporous thin film. (c) Photoconversion efficiency and the efficiency enhancement of solar cells with N-doped mesoporous TiO₂ thin film photoanodes. (d) Incident photon-to-current quantum conversion efficiency (IPCE) spectra of the solar cells with photoanodes consisting of P123-templated mesoporous TiO₂ thin films and Solaronix TiO₂ nanoparticle thin film. Reproduced from ref. 282 with permission from The Royal Society of Chemistry.

composed a 3D H₃V₃O₈ hydrogel with carbon nanotubes (CNT) to form H₃V₃O₈/CNT film, which served as an excellent electrode material.²¹⁷ Other examples of surfactant-assisted synthesized TMOs that serve as the anode material for LIB include Co₃O₄ hollow nanospheres,²⁴⁸ 3D flower-like Mn₃O₄,¹⁸² and mesoporous TiO₂.²⁰⁶

5.4 Supercapacitors

Supercapacitors are another significant solution for the energy storage challenges faced by the present generation. Owing to their increased capacitance, TMOs have been considered as efficient materials for supercapacitors. Various reports are available on the fabrication of TMOs using soft-templates for supercapacitor applications. Different oxides of manganese synthesized with the aid of surfactants were reported with diverse morphologies for competent supercapacitor applications.^{171,172,176–178,180} NiO is also an important electrode material for supercapacitor applications. Zhang *et al.* synthesized 2D NiO nanosheets with the aid of P123 to exhibit higher capacitance.²⁰²

Ci *et al.*²⁰³ and Meher *et al.*²⁰⁰ studied the effects of specific morphologies on the super capacitance behaviour of NiO. Porous Co₃O₄ synthesized with the aid of both block copolymers²⁴⁷ and bacteria templates¹³² were also reported for higher supercapacitance with excellent cycling stability. Gund *et al.* examined the process-structure-property relationship of two different morphologies of CuO synthesized using different surfactants.¹⁵⁷ As shown in Fig. 16, the CuO synthesized using Triton-X showed an increased specific capacitance when compared to the other samples. The effects of morphology on the capacitance activity was also stressed by Brezesinski *et al.*¹²¹ and Panigrahi *et al.*²²⁶ in their studies on mesoporous MoO₃ and 3D network of V₂O₅, respectively.

5.5 Solar cells

The depletion of fossil fuels has resulted in extended research in the area of renewable energy sources, and solar energy affords an effective solution to deal with the energy crisis. The fabrication of high-efficiency solar cells has been a challenge for



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