Faraday Discussions



# Multi-State Pair-Density Functional Theory

Journal:	Faraday Discussions		
Manuscript ID	FD-ART-03-2020-000037.R1		
Article Type:	Paper		
Date Submitted by the Author:	21-Apr-2020		
Complete List of Authors:	Bao, Jie; Department of Chemistry, University of Minnesota Zhou, Chen; Department of Chemistry, University of Minnesota Varga, Zoltán; University of Minessota, Chemistry Kanchanakungwankul, Siriluk; University of Minessota, Chemistry Gagliardi, Laura; University of Minnesota, Department of Chemistry and Supercomputing Institute Truhlar, Donald; Department of Chemistry, University of Minnesota		

SCHOLARONE<sup>™</sup> Manuscripts

#### **Faraday Discussions**

revised the Faraday Discussions volume on New Horizons in Density Functional Theory (2020)

## Multi-State Pair-Density Functional Theory

# Jie J. Bao,<sup>&</sup> Chen Zhou,<sup>&</sup> Zoltan Varga, Siriluk Kanchanakungwankul, Laura Gagliardi,\* and Donald G. Truhlar\*

Department of Chemistry, Chemical Theory Center, and Minnesota Supercomputing Institute, 207 Pleasant Street SE, University of Minnesota, Minneapolis, MN 55455-0431, USA

Abstract. Multiconfiguration pair-density functional theory (MC-PDFT) has previously been surfaces (PESs) in a region near a conical intersection. We have recently proposed state-<ii>interaction pair-density functional theory (SI-PDFT) to treat nearly degenerate states by creating</i> a set of intermediate states with state interaction; although this method is successful, it is inconvenient because two SCF calculations and two sets of orbitals are required and because it puts the ground state on an unequal footing with the excited states. Here we propose two new methods, called extended-multi-state-PDFT (XMS-PDFT) and variational-multi-state-PDFT ٰ(VMS), the properties of the <br/>The former uses the intermediate states proposed by Granovsky for extended multiconfiguration quasidegenerate perturbation theory (XMC-QDPT); the latter obtains the intermediate states by maximizing the sum of the MC-PDFT energies for the intermediate states. We also propose a Fourier series expansion to make the variational optimizations of the VMS-PDFT method convenient, and we implement this method (FMS-PDFT) both for conventional configurationinteraction solvers and for density-matrix-renormalization-group solvers. The new methods are ồỉ. <br/>for all of them except the mixed-valence case. Since both XMS-PDFT and VMS-PDFT are less expensive than XMS-CASPT2, they will allow well-correlated calculations on much larger systems for which perturbation theory is unaffordable.

<sup>†</sup>Electronic supplementary information (ESI) available: It contains geometries, HNCO potential curves, systems where XMS-PDFT fails, and absolute energies in hartrees. See DOI: 10.1039/[to be filled in by Faraday Discussions]

<sup>&</sup>These two authors contribute equally to this paper.

\*Email: gagliard@umn.edu, truhlar@umn.edu

# 1. Introduction

<b dkd<br/>
Although KS-DFT has lower accuracy for strongly correlated states than for weakly correlated ones, for large molecules it is much less expensive correlated ones, for large molecules it is much less expensive than wave function denotes for large molecules it is much less expensive than wave function denotes for large molecules it is much less expensive than wave functional theory (MC-PDFT) as a method that builds on a multiconfiguration on the formation on the more functional theory of correlated ones, for large molecules it is much less expensive than wave functional theory of compared with the molecules it is more formation and the molecules of the m

When states are nearly degenerate and have the same symmetry, they interact strongly with each other, and they should be treated and have the same symmetry, they interact strongly with each other, and they are nearly degenerate and have the same symmetry, they interact strongly with each other, and they same are nearly degenerate and have the same symmetry, they interact strongly with each other, and they should be treated and the nearly of the same strong the same strong the same state of the same state set of the same state of the same state of the same state of the same state of the same matrix (hence they interact through the off-diagonal elements of the same matrix). Similarly, complete active same strong of the same matrix (the same matrix) and state state of the same matrix.

energies for the intermediate states. We approximate the VMS-PDFT method by using a Fourier series expansion; using this method for VMS-PDFT is called the Fourier-multi-state-PDFT (FMS-PDFT). method.

A key aspect of all the above-mentioned MS methods is that they determine a model space spanned by the states to be treated as strongly interacting. Similar to XMC-QDPT or XMS-CASPT2, XMS-PDFT and VMS-PDFT build up a model space that spans the *N* lowest-energy states optimized in a state-averaged CASSCF (SA-CASSCF) calculation. (Generalizations to incomplete active spaces and smaller model spaces are straightforward but are not considered here.) The model space states are called the intermediate basis and are obtained by unitary transformation from the SA-CASSCF states.

Section 2 explains the two new methods and the Fourier-based approximation of VMS-PDFT. Section 3 specifies computational details for several test systems, including those that were previously studied by SI-PDFT. Section 4 presents applications of the new methods to these test systems and evaluates their performances. Section 5 has concluding remarks.

### 2. Theory

#### 2.1 MC-PDFT

The MC-PDFT method may be based on single-state CASSCF (SS-CASSCF) calculations on on SA-CASSCF calculations. In the present article we consider the latter type of calculation, in which case one starts with a reference wave function obtained by performing an SA-CASSCF

$$|\Psi_l\rangle = \sum_i c_i^l |\text{CSF}_i\rangle,\tag{1}$$

៤) where is a sequence of the properties of the pr

$$E_I^{\rm MC-PDFT} = T_{\rm e} + V_{\rm elec} + E_{\rm ot}(\rho_I, \Pi_I), \qquad (2)$$

$$\Pi_{I}(\mathbf{r}) = \int \Psi_{I}^{*}(\mathbf{r}_{1}, \mathbf{r}_{2}, ..., \mathbf{r}_{N_{e}}) \Psi_{I}(\mathbf{r}_{1}, \mathbf{r}_{2}, ..., \mathbf{r}_{N_{e}}) d\mathbf{r}_{3} ... d\mathbf{r}_{N_{e}}|_{\mathbf{r}_{1} = \mathbf{r}_{2} = \mathbf{r}}$$
(3)

#### 2.2 Multi-state MC-PDFT

To obtain the correct topography of PESs at conical intersections, we have proposed the SI-

In general, the intermediate states are obtained by a unitary transformation:

$$|\Phi_I\rangle = \sum_J U_{JI} |\Psi_J\rangle = \sum_{Ji} U_{JI} c_i^J |\text{CSF}_i\rangle, \tag{4}$$

where  $|\Phi_I\rangle$  is an intermediate state, and  $|\Psi_J\rangle$  is an SA-CASSCF state. The Hamiltonian is diagonal in the SA-CASSCF states but not in the intermediate basis.

We construct an effective Hamiltonian in the intermediate-state basis with diagonal elements defined as

$$H_{II}^{\text{eff}} = E_I^{\text{MC} - \text{PDFT}},\tag{5}$$

where  $E_I^{MC-PDFT}$  is the MC-PDFT energy for the intermediate state  $|\Phi_I\rangle$ . The off-diagonal elements of the effective Hamiltonian are defined as

$$H_{II}^{\text{eff}} = \langle \Phi_I \mid H \mid \Phi_I \rangle \tag{6}$$

with I, J = 1, 2, ..., N, where N is the number of states in the model space. (In the present work, the number of states in the model space is always the same as the number of states averaged in the SA-CASSCF calculation.) The effective Hamiltonian is then diagonalized to give the multi-state.

Following the above scheme, we next introduce two strategies (XMS-PDFT and VMS-PDFT) to generate the matrix U, yielding  $U^X$  and  $U^V$ , respectively.

# 2.3 XMS-PDFT

ॅ <list-item> <table-cell>

In MS-CASPT2, the unperturbed Hamiltonian is defined as

$$H_0 = PFP + QFQ, \tag{7}$$

where

$$P = \sum_{I} |\Psi_{I}\rangle \langle \Psi_{I}|$$

is the projection operator onto the SA-CASSCF state space and

$$Q = 1 - P$$

is the projection operator onto the complementary state space. In MS-CASPT2, the state Fock operator of states is defined as

$$F = \sum_{pq} f_{pq} E_{pq} = \sum_{pq} f_{pq} a_p^{\dagger} a_q,$$
(8)

where  $E_{pq} = a_p^{\dagger} a_q$  is a single-excitation operator,  $a_p^{\dagger}$  and  $a_q$  are creation and annihilation operators on molecular orbitals *p* and *q*, respectively, and  $f_{pq}$  is an element in the orbital Fock matrix

$$f_{pq} = h_{pq} + \sum_{rs} d_{rs} (J_{pq}^{rs} - \frac{1}{2}K_{pq}^{rs}),$$
(9)

where  $h_{pq}$  contains the electronic kinetic energy and electron-Coulomb interaction,  $d_{rs}$  is a state-averaged density matrix element, and  $J_{pq}^{rs}$  and  $K_{pq}^{rs}$  are two-electron integrals. The matrix elements of the state Fock matrix are defined as

$$F_{IJ} = \langle \Psi_I | F | \Psi_J \rangle = \sum_{pq} \sum_{ij} f_{pq} c_i^l c_j^j \langle \text{CSF}_i | E_{pq} | \text{CSF}_j \rangle.$$
(10)

$$(\mathbf{U}^{\mathrm{X}})^{\dagger} \mathbf{F} \mathbf{U}^{\mathrm{X}} = \tilde{\mathbf{F}}.$$
(11)

The **U**<sup>X</sup> matrix determined this way then yields the intermediate states defined by  $|\Phi_I\rangle = \sum_J U_{JI}^X |\Psi_J\rangle,$ (12)

where  $\Phi_I$  is an intermediate state in XMS-PDFT (and also in XMS-CASPT2). With the same transformation, we get a Hamiltonian matrix in the intermediate basis,

$$\mathbf{U}^{\mathrm{X}})^{\dagger}\mathbf{H}\mathbf{U}^{\mathrm{X}} = \tilde{\mathbf{H}},\tag{13}$$

where **H** is the Hamiltonian matrix in the basis of the SA-CASSCF reference states, and  $\tilde{\mathbf{H}}$  is in the basis of the intermediate states.

After the intermediate states are obtained, XMS-PDFT defines an effective Hamiltonian in the intermediate basis such that diagonal element  $H_{II}^{\text{eff}}$  is the MC-PDFT energy of intermediate state  $\Phi_I$ , and the off-diagonal element  $H_{IJ}^{\text{eff}}$  is  $\tilde{H}_{IJ}$ . The XMS-PDFT energies ( $E_I^{\text{XMC} - \text{PDFT}}$ ) and eigenvectors are obtained by diagonalizing the effective Hamiltonian matrix.

some mixed-valence<sup>19</sup> systems (see Fig. S3, where sections and figures with the prefix S are in Supporting Information) even when the states have the same symmetry, and we will see that XMS-PDFT does not always give good results for such systems. Next we present the VMS-PDFT method that does not have this problem (but it is more expensive).

#### 2.4 VMS-PDFT

The trace of the effective Hamiltonian defined above is given by

$$\operatorname{Tr}(\mathbf{H}^{\operatorname{eff}}) = \sum_{I} E_{I}^{\operatorname{MC} - \operatorname{PDFT}}$$
(14)

ॅ <list-item><list-item><text>

At present, we do not have an analytic procedure to find the transformation matrix  $\mathbf{U}^{V}$  that completely maximizes eqn (14). Instead, we propose here a numerical way to approximate the maximization in a practical and smooth way by fitting eqn (14) to a Fourier series. We call this implementation the FMS-PDFT method.

$$\mathbf{U}_{IJ}^{V}(\theta_{IJ}) = \begin{pmatrix} \cos\theta_{IJ} & \sin\theta_{IJ} \\ -\sin\theta_{IJ} & \cos\theta_{IJ} \end{pmatrix},\tag{15}$$

$$(\Psi_{I},\Psi_{J})\mathbf{U}_{IJ}^{\mathsf{V}}(\boldsymbol{\theta}_{IJ}) = (\Phi_{I},\Phi_{J}).$$
(16)

Now consider applying  $\mathbf{U}(\theta + \frac{\pi}{2})$  to the two states; this yields

$$(\Psi_I, \Psi_J) \mathbf{U}_{IJ}^{\mathsf{V}} \left( \theta_{IJ} + \frac{\pi}{2} \right) = (-\Phi_J, \Phi_I)$$
(17)

$$\Gamma r(\mathbf{H}^{\text{eff}}) = \frac{a_0}{2} + \sum_{n=1}^{\infty} [a_n \sin(4n\theta_{IJ}) + b_n \cos(4n\theta_{IJ})].$$
(18)

We keep only the terms with n = 1 in the sum on the right-hand side of eqn (18); then the

equation can be parameterized as

$$Tr(\mathbf{H}^{en}) = A + Bsin (4\theta_{IJ}) + Ccos (4\theta_{IJ}),$$
 (19)  
and the unknown parameters A, B, and C can be obtained by a three-point fitting. In this paper,  
three values (0°, 30°, and 60°) for  $\theta_{IJ}$  are applied to determine these three parameters for each  
single-point energy calculation. (We use these same three angles in all cases.) Then rotation  
angle is taken as the one that maximizes eqn (19).

For FMS-PDFT calculations with N states (where N is greater than 2), we write  $\mathbf{U}^{V}$  as a product of transformation matrices

$$\mathbf{U}^{\mathrm{V}} = \mathbf{U}_{12}\mathbf{U}_{23}\cdots\mathbf{U}_{I(I+1)}\cdots\mathbf{U}_{(N-1)N},\tag{20}$$

$$\boldsymbol{\Phi}^{(l)} = \boldsymbol{\Phi}^{(l-1)} \mathbf{U}_{l(l+1)}, \tag{21}$$

ॅ <list-item> <table-cell>

$$\tilde{D}_{pq}^{II} = \sum_{JK} U_{JI} U_{KI} D_{pq}^{JK} = \sum_{JK} U_{JI} U_{KI} \langle \Psi_J \mid E_{pq} \mid \Psi_K \rangle, \qquad (22)$$

$$\tilde{D}_{pqrs}^{II} = \sum_{JK} U_{JI} U_{KI} D_{pqrs}^{JK} = \sum_{JK} U_{JI} U_{KI} \langle \Psi_J | E_{pq} E_{rs} - \delta_{qr} E_{ps} | \Psi_K \rangle,$$
(23)

 where **D**<sup>JK</sup> and **d**<sup>JK</sup> are one-body and two-body transition density matrices between the reference states J and K, and **D**<sup>II</sup> and **d**<sup>II</sup> are one-body and two-body density matrices for the intermediate state I.

#### 3. Computational details

The calculations are performed in OpenMolcas v18.09, tag 548-g19e2926-dirty,<sup>29</sup> with codes

modified to perform XMS-PDFT and FMS-PDFT calculations. The DMRG calculations are performed with the QCMaquis software suite25<sup>,30,31,32</sup> in OpenMolcas v18.11, tag 17-g792ff65dirty, which is modified to perform FMS-PDFT/DMRG calculations.

Table 1 presents the wave function symmetry, basis set, number of averaged states, number of active electrons, and identities of active MOs for each system studied. The internal coordinates that are scanned for each system are shown in Table 2. The geometries are available in Section S1.

For the HNCO calculations with bond length r(NC) = 2.0–2.5 Å and torsion t(HNCO) = 150° (discussed in section 4.3), we carried out the NCO = 2.0–2.5 Å and torsion torsion = 150° (discussed in section and length of l

љ

System	Sym	<b>Basis set</b>	N <sub>states</sub>	n	active MOs
LiF	$C_1$	jun-cc-pVQZ <sup>34,35</sup>	2	8	2p <sub>z</sub> of F, 2s of Li
LiH	$C_{2v}$	aug-cc-pVQZ <sup>34</sup>	4	2	2s, 2p <sub>z</sub> , 3s, 3p <sub>z</sub> of Li, 1s of H
HNCO	$C_1$	cc-pVDZ <sup>34</sup>	2	16	Valence shell (2s and 2p of C, N, and O
					atoms and 1s of H atom)
CH <sub>3</sub> NH <sub>2</sub>	$C_1$	$6-31++G(d,p)^{36,37}$	2	6	$2 \sigma$ , $1 \sigma^*$ , $2p_z$ , $3s$ , and $3p_z$ of N
C <sub>6</sub> H <sub>5</sub> OH	$C_1$	jul-cc-pVDZ 34,35	2 or 3	12	$3 \pi, 3 \pi^*, \sigma_{OH}, \sigma_{OH}^*, \sigma_{CO}, \sigma_{CO}^*$ and $p_z$ of O
$O + O_2 (^{3}A')$	$C_{\rm s}$	cc-pVTZ <sup>34</sup>	6	12	9 2p orbitals
$O_3(^{3}A')$	$C_{\rm s}$	cc-pVTZ <sup>34</sup>	6	12	9 2p orbitals
Spiro	$C_{2v}$	$6-31G(d)^{38}$	2	11	See Ref. 17

Table 2.	Systems	studied	and the	e internal	coordinates	scanned	for r	otential	energy	curves
	-1								6.7.7	

System	Internal coordinates scanned
LiF	r(LiF) = [1.0 - 9.0]  Å
LiH	r(LiH) = [1.0 - 12.0]  Å
HNCO	$r(NC) = [1.25 - 3.00] \text{ Å and } \tau(HNCO) = [180 - 130]^{\circ}$
CH <sub>3</sub> NH <sub>2</sub>	$r(NH) = [0.8 - 3.6]$ Å and $\tau(H6-C4-N1-H3) = 0, 90, 95, or 100^{\circ}$
C <sub>6</sub> H <sub>5</sub> OH	$r(OH) = [0.5 - 3.0] \text{ Å and } \tau(C-C-O-H) = 1 \text{ or } 10^{\circ}$
$O + O_2 (^{3}A')$	r(O1O3) = [1.0 - 2.5]  Å
$O_3(^{3}A')$	$\alpha(O2O1O3) = [60 - 180]^{\circ}$

Spiro See Section 4.8

#### 4. Results and discussion

#### 4.1 Lithium fluoride (LiF)

Lithium fluoride has an avoided crossing of the ground state and first excited state that has been widely studied has an avoided crossing of the ground state and first excited state that an avoided crossing is ground state and first excited has an Avoided crossing is crossing is control of the ground state and first excited has an Avoided crossing is control of the ground state has an Avoided crossing is about 7.4 Å.<sup>39</sup> However, theoretical calculations usually underestimate the region by 1.0 Å, with an exception being the calculation in Ref. 41.

ॅ <list-item><list-item>

$R_{\text{Li-F}}(\text{\AA})$	Rotation angle (deg)	$\Delta E (eV)$
0.8	26.34	-0.0016
1.6	5.06	0.0078
2.4	15.72	0.0008
3.2	28.66	-0.0026
4.0	38.46	-0.0015
4.8	13.00	-0.0102
5.6	4.10	0.0033
6.4	1.37	0.0020
7.2	0.42	0.0007

Table 3. The rotation an	gles for various	Li-F bond length	s and the differe	ence of trace of the
effective Hamiltonian of	btained by fitting	g and by specific	calculation at th	at rotation angle.



Fig. 1. Comparison of XMS-PDFT, FMS-PDFT, MC-PDFT, and XMS-CASPT2 for the potential energy curves of the two electronic states of LiF. The area near the avoided crossing (indicated by a small box) for each curve is also shown magnified.

# 4.2 Lithium hydride (LiH)In ALL

ٰ
productude

The first (red) and second (blue) states calculated by XMS-PDFT and FMS-PDFT agree

ƙhiii} Wiji} Wijii} Wiji} Wijii} Wijii} Wijii Wiji Wi Wiji Wiji Wiji Wi Wiji Wiji Wiji Wi Wiji Wiji Wiji Wi Wij



Fig. 2. Comparison of XMS-PDFT, FMS-PDFT, MC-PDFT, and XMS-CASPT2 for the potential energy curves of LiH. The zoomed-in area near the avoided crossing is shown for the

ionic state and the highest covalent state calculated.

# 4.3 Isocyanic acid (HNCO)

ં



## 4.4 Methylamine (CH<sub>3</sub>NH<sub>2</sub>)

္ both exects exec

The calculated potential energy curves along the four paths are plotted in Fig. 4. The potential energy curves calculated by both new methods show correct topographies for avoided

Although MC-PDFT does not diagonalize an effective Hamiltonian matrix in the last step, Although MC-PDFT does not diagonalize an effective Hamiltonian matrix in the last step, we note that MC-PDFT does not diagonalize an effective Hamiltonian matrix in the last step, we note that MC-PDFT does not diagonalize an effective Hamiltonian matrix in the last step, we note that MC-PDFT does not diagonalize an effective Hamiltonian matrix in the last step, we note that MC-PDFT does not diagonalize an effective Hamiltonian matrix in the last step, we note that MC-PDFT does not diagonalize an effective Hamiltonian matrix in the last step, we note that MC-PDFT does not diagonalize an effective Hamiltonian matrix in the last step, we note that MC-PDFT does not diagonalize an effective Hamiltonian matrix in the last step, we note that MC-PDFT does not diagonalize an effective Hamiltonian matrix in the last step, we note that MC-PDFT does not diagonalize an effective Hamiltonian matrix in the last step, we note that MC-PDFT does not does not does not does not we not does not we not does not we not does not we not does not we not does not does



Fig. 4. potential energy curves for methylamine with the four dissociation paths of (a) eclipsed-

H3, (b) staggered, (c) 95° and (d) 100° conformations calculated by XMS-PDFT compared with
 HC-PDFT. potential energy curves for methylamine with the staggered conformation calculated by FMS-PDFT
 MC-PDFT and (f) XMS-CASPT2.

Method	$R_{ m N-H}( m \AA)$	Energy separations (eV)
	E	lipsed-H3
XMS-CASPT2	2.01	0.007
MC-PDFT	2.07	0.010
XMS-PDFT	2.07	0.010
	S	taggered
XMS-CASPT2	1.91	0.20
MC-PDFT	1.97	0.18
XMS-PDFT	1.97	0.23
FMS-PDFT	1.97	0.18
		95°
XMS-CASPT2	1.92	0.33
MC-PDFT	1.98	0.30
XMS-PDFT	1.98	0.28
		100°
XMS-CASPT2	1.95	0.73
MC-PDFT	2.00	0.68
XMS-PDFT	2.02	0.65

Table 4. The N-H bond length (Å) and the energy separations (eV) at the avoided crossing point for the four N-H fission paths.

# 4.5 Phenol (C<sub>6</sub>H<sub>5</sub>OH)

The O-H bond dissociation in phenol has been well studied in the past and it can be used as a model system for testing whether a method gives a proper description of potential energy curves for photodissociation. We tested MC-PDFT, XMS-PDFT, FMS-PDFT, and FMS-PDFT/DMRG for the O-H dissociation in phenol with the H-O-C-C dihedral angle being 1° (nearly planar) or 10°.

္In regions that we get we get

ॅ <list-item><text>





္<table-cell><text>

$R_{\text{O-H}}$ (Å)	$\Delta\theta$ (deg)	$\Delta E_1 (\mathrm{meV})$	$\Delta E_2 (\mathrm{meV})$
0.8	-0.0004	0.007	0.026
1.0	0.0005	-0.002	-0.012
1.2	0.0001	0.008	0.007
1.4	-0.0003	-0.040	-0.021
1.6	0.0001	0.016	-0.007
1.8	0.0004	0.008	-0.005
2.0	0.0000	0.017	-0.005
2.2	-0.0004	0.022	0.031

Table 5. Differences between rotation angles ( $\theta$ ), energies ( $E_1$  and  $E_2$ ) calculated by FMS-PDFT/DMRG and those by FMS-PDFT for each state

2.4	0.0000	-0.007	0.013
2.6	0.0007	-0.004	0.005
2.8	0.0000	0.004	-0.006
3.0	0.0001	-0.014	0.003
3.2	0.0000	-0.006	0.010

## 4.6 Oxygen atom plus oxygen molecule collision in triplet state (O + O<sub>2</sub>)

The first example considered corresponds to an O<sub>2</sub> + O collision with the atom, labeled O3, impinging on the O1 end of the corresponds to an O<sub>2</sub> + O collision with the atom, labeled O3, impinging on the O1 end of the corresponds to an O<sub>2</sub> + O collision with the atom, labeled O3, impinging on the O1 end of the corresponds to an O<sub>2</sub> + O collision with the atom, labeled O3, impinging on the O1 end of the corresponds to an O<sub>2</sub> + O collision with the atom, labeled O3, impinging on the dot end of the corresponds to an O<sub>2</sub> + O collision with the collision with the end of the the end of the constant of the corresponds to an O<sub>2</sub> + O collision with the end of the e



Fig. 7. Six potential energy curves of triplet O + O2 collisions calculated by XMS-CASPT2 and XMS-PDFT.

#### 4.7 Triplet ozone (O<sub>3</sub>)

亜mmmle Mannes Mann



#### 4.8 Spiro cation

ٰIn this section, we tested the tested teste

$$Q_{\gamma}(\xi) = \left(\frac{1}{2} - \xi\right) Q_{\gamma}^{A} + \left(\frac{1}{2} + \xi\right) Q_{\gamma}^{B}, \qquad \gamma = 1, 2, ..., 3N_{\text{atoms}}$$
(24)

For this very difficult test case, the XMS-PDFT curves resemble the MC-PDFT curves, and the XMS-CASPT2 difficult test case, the XMS-PDFT curves resemble the MS-CASPT2 estimates the test case, the XMS-CASPT2 curves and the test case the test case, the XMS-CASPT2 curves which is the test case the test case the test case test case test case test cases the test cases test cases the test cases the

great improvement of FMS-PDFT compared with MC-PDFT again shows the value of FMS-PDFT.

Section S3 in the Supporting Information shows some other mixed-valence cases where XMS-PDFT fails to give a correct topography of PESs.





# 5. Conclusions

A general scheme for multi-state MC-PDFT is proposed in this paper. In this scheme, the CASSCF reference states are rotated to a set of intermediate states via a unitary transformation, nd an effective Hamiltonian matrix in the intermediate-state basis is constructed using the PDFT method for the diagonal elements and wave function theory for the off-diagonal ones. Two practical methods, XMS-PDFT and VMS-PDFT, for the unitary transformations are noposed in this paper, and they are tested on eight systems exhibiting avoided crossings of two QDPT; VMS-PDFT chooses the transformation that maximizes the trace of the effective Hamiltonian. We implemented the VMS-PDFT method using a convenient Fourier series čexpansion, and the resulting method is called FMS-PDFT. Tests are performed on systems (mainly in the FMS-PDFT version), to state-specific MC-PDFT and the more expensive multistate method, XMS-CASPT2. We find that FMS-PDFT, like our earlier but less convenient SI-PDFT, gives reasonable potential energy curves for all test cases examined, and it shows great improvement over MC-PDFT. Similarly XMS-PDFT gives good results for all systems except the mixed-valence spiro cation. Since XMS-PDFT is less expensive than VMS-PDFT and since

# Acknowledgments

This work was supported by the National Science Foundation under grant CHE-1746186.

- 1 W. Kohn, A. D. Becke and R. G. Parr, Density functional theory of electronic structure, *J. Phys. Chem.*, 1996, **100**, 12974-12980.
- 2 H. S. Yu, S. L. Li and D. G. Truhlar, Perspective: Kohn-Sham density functional theory descending a staircase, *J. Chem. Phys.*, 2016, **145**, 130901.
- 3 W. Domcke, H. Köppel and L. S. Cederbaum, Spectroscopic effects of conical intersections of molecular potential energy surfaces, *Mol. Phys.*, 1981, **43**, 851-875.
- 4 A. W. Jasper, B. K. Kendrick, C. A. Mead and D. G. Truhlar, Non-Born-Oppenheimer Chemistry: Potential Surfaces, Couplings, and Dynamics, in *Modern Trends in Chemical Reaction Dynamics: Experiment and Theory* (Part 1), ed. X. Yang and K. Liu, World Scientific, Singapore, 2004, pp. 329-391.
- 5 R. Maurice, P. Verma, J. M. Zadrozny, S. Luo, J. Borycz, J. R. Long, D. G. Truhlar and L. Gagliardi, Single-ion magnetic anisotropy and isotropic magnetic couplings in the metal–organic framework Fe<sub>2</sub>(dobdc), *Inorg. Chem.*, 2013, **52**, 9379-9389.
- 6 M. Atanasov, J. M. Zadrozny, J. R. Long and F. Neese, A theoretical analysis of chemical bonding, vibronic coupling, and magnetic anisotropy in linear iron(II) complexes with single-molecule magnet behavior, *Chem. Sci.*, 2013, **4**, 139-156.
- 7 G. Li Manni, R. K. Carlson, S. Luo, D. Ma, J. Olsen, D. G. Truhlar and L. Gagliardi, Multiconfiguration pair-density functional theory, *J. Chem. Theory Comput.*, 2014, **10**, 3669-3680.
- 8 L. Gagliardi, D. G. Truhlar, G. Li Manni, R. K. Carlson, C. E. Hoyer and J. L. Bao, Multiconfiguration pair-density functional theory: a new way to treat strongly correlated systems, *Acc. Chem. Res.*, 2017, **50**, 66-73.
- 9 S. Ghosh, P. Verma, C. J. Cramer, L. Gagliardi, and D. G. Truhlar, Combining wave function methods with density functional theory for excited states, *Chem. Rev.*, 2018, **118**, 7249-7292.
- 10 K. Hirao, Multireference Møller-Plesset method, Chem. Phys. Lett., 1992, 190, 374-380.
- 11 H. Nakano, Quasidegenerate perturbation theory with multiconfigurational self-consistent-field reference functions, *J. Chem. Phys.*, 1993, **99**, 7983-7992.
- 12 A. A. Granovsky, Extended multi-configuration quasi-degenerate perturbation theory: the new approach to multi-state multi-reference perturbation theory, *J. Chem. Phys.*, 2011, **134**, 214113.
- 13 K. Andersson, P. A. Malmqvist, B. O. Roos, A. J. Sadlej and K. Wolinski, Second-order perturbation theory with a CASSCF reference function, *J. Phys. Chem.*, 1990, **94**, 5483-5488.
- 14 J. Finley, P.-Å. Malmqvist, B. O. Roos and L. Serrano-Andrés, The multi-state CASPT2 method, *Chem. Phys. Lett.*, 1998, **288**, 299-306.
- 16 A. M. Sand, C. E. Hoyer, D. G. Truhlar and L. Gagliardi, State-interaction pair-density functional theory, *J. Chem. Phys.*, 2018, **149**, 024106.

- 17 S. S. Dong, K. B. Huang, L. Gagliardi and D. G. Truhlar, State-interaction pair-density functional theory can accurately describe a spiro mixed valence compound, *J. Phys. Chem. A*, 2019, **123**, 2100-2106.
- 19 M. B. Robin and P. Day, Mixed Valence Chemistry: A Survey and Classification, in *Advances in Inorganic Chemistry and Radiochemistry*, eds. H. J. Emeléus and A. G. Sharpe, Academic Press 1968, vol. 10, pp. 247-422.
- 20 S. R. White, Density matrix formulation for quantum renormalization groups, *Phys. Rev. Lett.*, 1992, **69**, 2863-2866.
- 21 S. R. White, Density-matrix algorithms for quantum renormalization groups, *Phys. Rev. B*, 1993, **48**, 10345-10356.
- 23 Y. Kurashige and T. Yanai, Second-order perturbation theory with a density matrix renormalization group self-consistent field reference function: theory and application to the study of chromium dimer, *J. Chem. Phys.*, 2011, **135**, 094104.
- 24 R. Olivares-Amaya, W. Hu, N. Nakatani, S. Sharma, J. Yang and G. K.-L. Chan, The *ab-initio* density matrix renormalization group in practice, *J. Chem. Phys.*, 2015, **142**, 034102.
- 25 S. Knecht, E. D. Hedegård, S. Keller, A. Kovyrshin, Y. Ma, A. Muolo, C. J. Stein and M. Reiher, New approaches for *ab initio* calculations of molecules with strong electron correlation, *Chimia*, 2016, **70**, 244–251.
- 26 P. Sharma, V. Bernales, S. Knecht, D. G. Truhlar and L. Gagliardi, Density matrix renormalization group pair-density functional theory (DMRG-PDFT): singlet-triplet gaps in polyacenes and polyacetylenes, *Chem. Sci.*, 2019, **10**, 1716-1723.
- 27 C. Zhou, L. Gagliardi and D. G. Truhlar, Multiconfiguration pair-density functional theory for iron porphyrin with CAS, RAS, and DMRG active spaces, *J. Phys. Chem. A*, 2019, **123**, 3389-3394.
- 28 S. Wouters, W. Poelmans, P. W. Ayers and D. Van Neck, CheMPS2: A free open-source spin-adapted implementation of the density matrix renormalization group for ab initio quantum chemistry, *Comput. Phys. Commun.*, 2014, **185**, 1501-1514.
- 29 I. Fdez. Galván, M. Vacher, A. Alavi, C. Angeli, F. Aquilante, J. Autschbach, J. J. Bao, S. I. Bokarev, N. A. Bogdanov, R. K. Carlson, C. Angeli, F. Aquilante, J. Autschbach, J. J. Bokarev, N. A. Bogdanov, R. K. Carlson, L. Angeli, F. Aquilante, J. Autschbach, J. Bokarev, N. A. Bogdanov, R. K. Carlson, L. F. Chibotaru, J. Creutzberg, N. Dattani, M. Bokarev, N. A. Bogdanov, R. K. Carlson, L. F. Chibotaru, J. Creutzberg, N. Dattani, M. Bogdanov, R. K. Carlson, L. Bokarev, N. A. Bogdanov, R. K. Carlson, L. F. Chibotaru, J. Creutzberg, N. Dattani, M. Bogdanov, R. K. Carlson, J. C. Bogdanov, R. K. Bogdanov, R. K. S. S. Dong, A. Dong, A. Malmavis, P. Manani, M. Lundberg, Y. Ma, S. Maliado, P. Malhado, P. A. Malmavis, P. Manani, M. Lundberg, Y. Ma, S. Maliado, S. Malhado, P. Malhado, P. A. Malmavis, P. Malmavis, P. Manani, M. Lundberg, Y. Mason, S. Malhado, P. Malhado, P. Malhado, P. Manni, M. Lundberg, Y. Malson, S. Malhado, P. Malhado, P. Malhado, P. Manni, M. Lundberg, Y. Malavis, J. Malhado, P. Manni, M. Lundberg, N. Malhado, P. Malhado, P
- 30 S. Keller, M. Dolfi, M. Troyer and M. Reiher, An efficient matrix product operator representation of the quantum chemical Hamiltonian, *J. Chem. Phys.*, 2015, **143**, 244118.

- 31 S. Keller and M. Reiher, Spin-adapted matrix product states and operators, *J. Chem. Phys.*, 2016, **144**, 134101.
- 32 Y. Ma, S. Knecht, S. Keller and M. Reiher, Second-order self-consistent-field densitymatrix renormalization group, *J. Chem. Theory Comput.*, 2017, **13**, 2533-2549.
- 33 G. Ghigo, B. O. Roos and P.-Å. Malmqvist, A modified definition of the zeroth-order Hamiltonian in multiconfigurational perturbation theory (CASPT2), *Chem. Phys. Lett.*, 2004, **396**, 142-149.
- 34 T. H. D. Jr., Gaussian basis sets for use in correlated molecular calculations. I. The atoms boron through neon and hydrogen, *J. Chem. Phys.*, 1989, **90**, 1007-1023.
- 35 E. Papajak, J. Zheng, X. Xu, H. R. Leverentz and D. G. Truhlar, Perspectives on basis sets beautiful: seasonal plantings of diffuse basis functions, *J. Chem. Theory Comput.*, 2011, 7, 3027-3034.
- 36 D. Feller, The role of databases in support of computational chemistry calculations, *J. Comput. Chem.*, 1996, **17**, 1571-1586.
- 37 K. L. Schuchardt, B. T. Didier, T. Elsethagen, L. Sun, V. Gurumoorthi, J. Chase, J. Li and T. L. Windus, Basis set exchange: a community database for computational sciences, *J. Chem. Inf. Model.*, 2007, **47**, 1045-1052.
- 38 P. C. Hariharan and J. A. Pople, The influence of polarization functions on molecular orbital hydrogenation energies, *Theor. Chim. Acta*, 1973, **28**, 213-222.
- 39 L. R. Kahn, P. J. Hay and I. Shavitt, Theoretical study of curve crossing: *ab initio* calculations on the four lowest  ${}^{1}\Sigma^{+}$  states of LiF, *J. Chem. Phys.*, 1974, **61**, 3530-3546.
- 40 B. J. Botter, J. A. Kooter and J. J. C. Mulder, Ab-initio calculations of the covalent-ionic curve crossing in LiF, *Chem. Phys. Lett.*, 1975, **33**, 532-534.
- ] 2020]
- 42 J. P. Finley and H. A. Witek, Diagrammatic complete active space perturbation theory: calculations on benzene, N<sub>2</sub>, and LiF, *J. Chem. Phys.*, 2000, **112**, 3958-3963.
- 43 J. Meller, J.-P. Malrieu and J.-L. Heully, Size-consistent multireference configuration interaction method through the dressing of the norm of determinants, *Mol. Phys.*, 2003, **101**, 2029-2041.
- 44 Ö. Legeza, J. RÖDer and B. A. Hess, QC-DMRG study of the ionic-neutral curve crossing of LiF, *Mol. Phys.*, 2003, **101**, 2019-2028.
- 45 C. Angeli, R. Cimiraglia and J.-P. Malrieu, A simple approximate perturbation approach to quasi-degenerate systems, *Theor. Chem. Acc.*, 2006, **116**, 434-439.
- 46 M. Hanrath, Multi-reference coupled-cluster study of the ionic-neutral curve crossing LiF, *Mol. Phys.*, 2008, **106**, 1949-1957.
- 47 F. Sasaki and M. Yoshimine, Configuration-interaction study of atoms. II. electron affinities of B, C, N, O, and F, *Phys. Rev. A*, 1974, **9**, 26-34.
- 48 D. G. Truhlar and C. A. Mead, Relative likelihood of encountering conical intersections and avoided intersections on the potential energy surfaces of polyatomic molecules, *Phys. Rev. A*, 2003, **68**, 32501.

- 49 H. Nakamura and D. G. Truhlar, Extension of the fourfold way for calculation of global diabatic potential energy surfaces of complex, multiarrangement, non-Born-Oppenheimer systems: application to HNCO(S<sub>0</sub>,S<sub>1</sub>), *J. Chem. Phys.*, 2003, **118**, 6816-6829.
- 50 J. V. Michael and W. A. Noyes, The photochemistry of methylamine, *J. Am. Chem. Soc.*, 1963, **85**, 1228-1233.
- 51 E. Kassab, J. Gleghorn and E. Evleth, Theoretical aspects of the photochemistry of methanol, methylamine, and related materials, *J. Am. Chem. Soc.*, 1983, **105**, 1746-1753.
- 52 G. Waschewsky, D. Kitchen, P. Browning and L. Butler, Competing bond fission and molecular elimination channels in the photodissociation of CH<sub>3</sub>NH<sub>2</sub> at 222 nm, *J. Phys. Chem.*, 1995, **99**, 2635-2645.
- 53 C. L. Reed, M. Kono and M. N. R. Ashfold, Near-UV photolysis of methylamine studied by H-atom photofragment translational spectroscopy, *J. Chem. Soc., Faraday Trans.*, 1996, **92**, 4897-4904.
- 54 K. M. Dunn and K. Morokuma, Ab initio study of the photochemical dissociation of methylamine, *J. Phys. Chem.*, 1996, **100**, 123-129.
- 55 S. J. Baek, K.-W. Choi, Y. S. Choi and S. K. Kim, Spectroscopy and dynamics of methylamine. I. Rotational and vibrational structures of CH<sub>3</sub>NH<sub>2</sub> and CH<sub>3</sub>ND<sub>2</sub> in à states, *J. Chem. Phys.*, 2003, **118**, 11026-11039.
- 56 M. H. Park, K.-W. Choi, S. Choi, S. K. Kim and Y. S. Choi, Vibrational structures of methylamine isotopomers in the predissociative à states: CH<sub>3</sub>NHD, CD<sub>3</sub>NH<sub>2</sub>, CD<sub>3</sub>NHD, and CD<sub>3</sub>ND<sub>2</sub>, *J. Chem. Phys.*, 2006, **125**, 084311.
- 57 D.-S. Ahn, J. Lee, J.-M. Choi, K.-S. Lee, S. J. Baek, K. Lee, K.-K. Baeck and S. K. Kim, State-selective predissociation dynamics of methylamines: The vibronic and H/D effects on the conical intersection dynamics, *J. Chem. Phys.*, 2008, **128**, 224305.
- 58 C. Levi, R. Kosloff, Y. Zeiri and I. Bar, Time-dependent quantum wave-packet description of H and D atom tunneling in N–H and N–D photodissociation of methylamine and methylamine-d2, *J. Chem. Phys.*, 2009, **131**, 064302.
- 59 R. Marom, C. Levi, T. Weiss, S. Rosenwaks, Y. Zeiri, R. Kosloff and I. Bar, Quantum tunneling of hydrogen atom in dissociation of photoexcited methylamine, *J. Phys. Chem. A*, 2010, **114**, 9623-9627.
- 60 D.-S. Ahn, J. Lee, Y. C. Park, Y. S. Lee and S. K. Kim, Nuclear motion captured by the slow electron velocity imaging technique in the tunnelling predissociation of the S<sub>1</sub> methylamine, *J. Chem. Phys.*, 2012, **136**, 024306.
- 61 J. O. Thomas, K. E. Lower and C. Murray, Observation of NH  $X^3\Sigma^-$  as a primary product of methylamine photodissociation: Evidence of roaming-mediated intersystem crossing? *J. Phys. Chem. Lett.*, 2012, **3**, 1341-1345.
- 62 H. Xiao, S. Maeda and K. Morokuma, Theoretical study on the photodissociation of methylamine involving S<sub>1</sub>, T<sub>1</sub>, and S<sub>0</sub> states, *J. Phys. Chem. A*, 2013, **117**, 5757-5764.
- 63 M. Epshtein, Y. Yifrach, A. Portnov and I. Bar, Control of nonadiabatic passage through a conical intersection by a dynamic resonance, *J. Phys. Chem. Lett.*, 2016, **7**, 1717-1724.
- 64 T. A. Halgren and W. N. Lipscomb, The synchronous-transit method for determining reaction pathways and locating molecular transition states, *Chem. Phys. Lett.*, 1977, **49**, 225-232.