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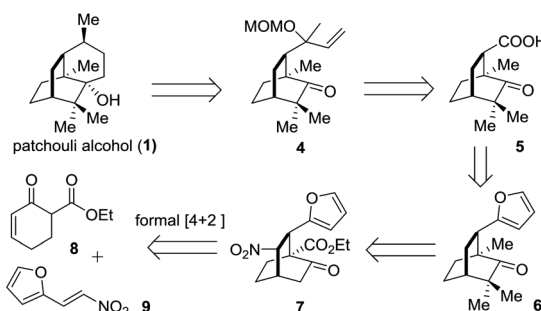
Concise asymmetric total synthesis of (–)-patchouli alcohol†

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The asymmetric total synthesis of (–)-patchouli alcohol was accomplished in a concise manner. Key reactions include a highly diastereo- and enantioselective formal organocatalytic [4 + 2] cycloaddition reaction, a radical denitration reaction, and an oxidative carboxylation reaction. The formal synthesis of norpatchoulenol was achieved as well.

The sesquiterpene patchouli alcohol (**1**, Fig. 1), the major component of patchouli oil, has been known for one and a half centuries since Gal and co-workers reported its isolation from *Pogostemon cablin* (Blanco) Benth in 1869.¹ The molecular structure of patchouli alcohol (**1**), however, had been unknown until its establishment in 1963 through single crystal X-ray diffraction analysis.² It possesses a compact tricyclo[5.3.1.0^{3,8}]undecane carbon skeleton decorated with three contiguous quaternary centers, closely akin to norpatchoulenol³ (**2**) and seychellene⁴ (**3**) (Fig. 1). Biologically, patchouli alcohol exhibits various activities including calcium ion antagonism,⁵ antibacterial,⁶ anti-inflammatory,⁷ and anti-tumor⁸ activities. The combination of the synthetic challenge and the biological interest makes patchouli alcohol a highly attractive target for synthetic chemists in the past three decades.⁹ In this paper, we report a concise asymmetric total synthesis of patchouli alcohol based on a new synthetic strategy.

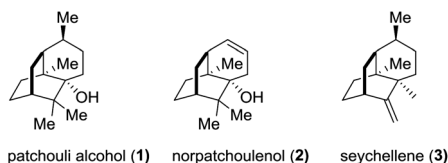
The retrosynthetic analysis of **1** is depicted in Scheme 1. In the light of Bertrand's procedure,^{9g} an intramolecular reductive radical coupling of **4** was envisioned to accomplish the target molecule. The MOM-protected carbinol in **4** could be traced back to a carboxylic acid group and further to a furan

Scheme 1 Retrosynthetic analysis of patchouli alcohol (**1**).

ring, thereby defining **5**^{9j} and **6** as successive precursors. By incorporating a nitro group into the framework, the key intermediate **7** was envisaged and a denitration reaction could convert **7** to **6**. Based on the formal organocatalytic [4 + 2] cycloaddition reaction developed by us and later successfully applied in natural product synthesis,¹⁰ compound **7** would be assembled from the reaction of **8** and **9**.

The synthetic journey commenced with the execution of the critical [4 + 2] reaction of **8** and **9** (Scheme 2). The one-pot reaction of **8** and **9** was first catalyzed by 5 mol% **10** before being treated with 0.3 equivalents of DBU at room temperature, providing **7** in 61% yield and 94% ee with a diastereomeric ratio of 20/1 (Scheme 2).

Mechanistically, this [4 + 2] reaction proceeded in two steps (Scheme 3). In the first step, **7** may form a complex with **10** via hydrogen bonding interactions, thereby discriminating the two faces of the reactive enol. The nitroolefin approaches the enol selectively from the less hindered α -face with the substituents of the two bond-forming carbons positioned in a staggered arrangement (**11**), delivering **12**. In the second step, when DBU is used as the base, **13** might be operative to minimize the steric repulsion between the nitro and the phenyl groups, leading to the formation of **7** as the major product. Remarkably, this reaction could be readily scaled up to a ten-gram scale.

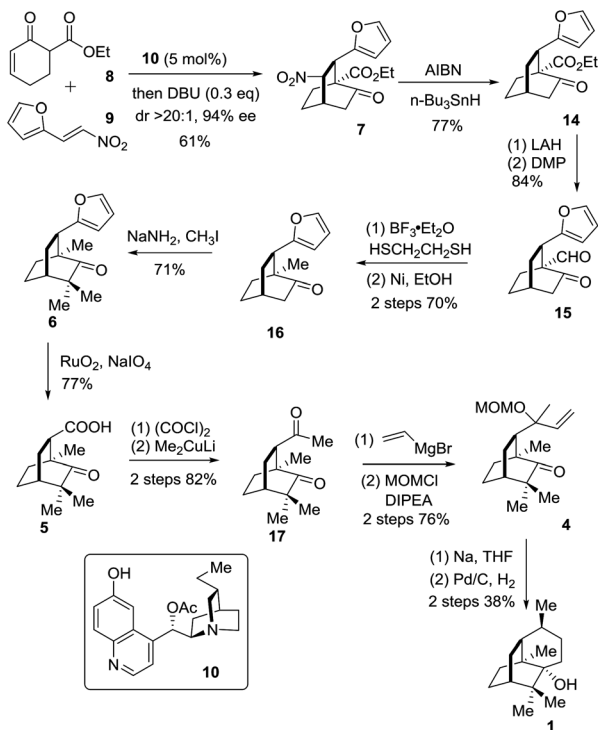
Fig. 1 Patchouli alcohol (**1**), norpatchoulenol (**2**), and seychellene (**3**).

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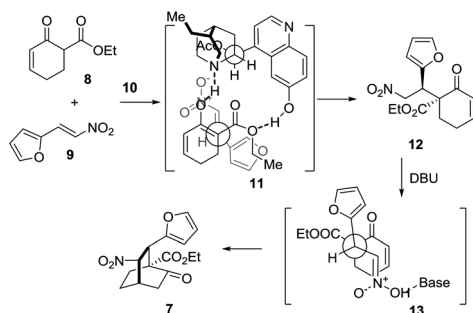
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† Electronic supplementary information (ESI) available: Experimental procedures, spectroscopic data, and copies of ¹H, ¹³C and 2D NMR spectra. See DOI: 10.1039/c7qo00459a





Scheme 2 Total synthesis of patchouli alcohol (1).



Scheme 3 Proposed stereocontrol model for the formal [4 + 2] cycloaddition reaction of 8 and 9 catalyzed by 10.

The strategic incorporation of the nitro group into the skeleton enabled us to assemble the bicyclic framework efficiently in a highly enantioselective manner. The key denitration was next investigated. As listed in Table 1, the conversion of 5 to 8 was examined with various solvents and temperatures. The optimal set of conditions involved AIBN/*n*-Bu₃SnH in mesitylene at 150 °C that provided 14 in 77% yield after a reaction time of less than half an hour (entry 4).

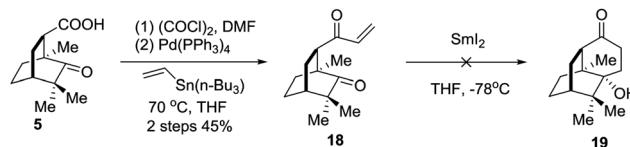
Table 1 Denitration of 7 to 14

| Entry | Conditions | Eq. (<i>n</i> -Bu ₃ SnH) | Yield (%) |
|-------|----------------------------|--------------------------------------|-------------|
| 1 | Toluene, 110 °C, 10 h | 2.5 | No reaction |
| 2 | Xylenes, 120 °C, 10 h | 2.5 | No reaction |
| 3 | Mesitylene, 150 °C, 10 h | 2.5 | 50 |
| 4 | Mesitylene, 150 °C, 25 min | 5 | 77 |

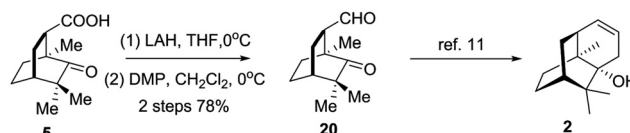
With the chiral [2.2.2] bicyclic skeleton 14 in hand, we proceeded to the next synthetic stage (Scheme 2). Reduction of the keto ester 8 with LAH followed by oxidation with DMP afforded the keto aldehyde 15 in 84% overall yield. Compound 15 was deoxygenated to give 16 in an overall yield of 70% via a two-step sequence involving thioacetalization with ethane-dithiol and the succeeding desulfurization with RANEY® nickel. The geminal dimethyl groups were next installed by treating 16 with NaNH₂ and CH₃I to give 6 in 71% yield. The oxidation of 6 with NaIO₄ in the presence of a catalytic amount of RuO₂·H₂O (1 mol%) in CCl₄/MeCN/H₂O underwent smoothly to afford the carboxylic acid 5 in a yield of 77%. Compound 5 was converted into 17 in 82% overall yield via the addition of lithium dimethylcuprate to the preformed acid chloride.^{9j} The Grignard addition of vinylmagnesium bromide occurred on the less hindered carbonyl group in 17 providing the tertiary allyl alcohol in a chemoselective manner. Exposure of the resultant tertiary allyl alcohol to methoxymethyl chloride (MOMCl) and diisopropylethylamine (DIPEA) in CH₂Cl₂ at 0 °C for 24 hours gave 4 in 76% overall yield.¹¹ The eventual single electron mediated 6-endo-trig cyclization reaction using Bertrand's conditions successfully delivered the annulated product which upon a catalytic hydrogenation reaction on Pd/C engendered 1 in 38% overall yield. The specific optical rotation of the synthetic sample ($[\alpha]_D^{24} -95.0$ (*c* = 0.4, CHCl₃)) matched that of the natural compound ($[\alpha]_D^{20} -94.6$ (*c* = 1.21, CHCl₃)).^{9e}

An alternative strategy involving SmI₂ mediated radical cyclization was attempted (Scheme 4). Thus, compound 5 was transformed into the vinyl ketone 18 through a Stille coupling reaction. To our disappointment, treatment of 18 with SmI₂ did not generate the anticipated intramolecular reductive cyclization product 19 (Scheme 4). Instead, the dimerization product was identified by LRMS and HRMS techniques. Considering that the lower carbonyl group was trapped by two neighboring quaternary carbon centers, the grave steric hindrance might account for the deviation of the cyclization reaction.

Eventually, compound 5 was employed to synthesize 20 via a redox procedure in a yield of 78% (Scheme 5). Compound 20 was a known synthetic precursor to norpatchoulenol (2).¹²



Scheme 4 Synthesis of 18.



Scheme 5 Formal synthesis of (-)-norpatchoulenol.



Conclusions

In summary, we have realized the asymmetric total synthesis of (–)-patchouli alcohol with a new strategy. Key features include a highly enantioselective and diastereoselective organocatalytic [4 + 2] approach to the [2.2.2] bicyclic core, an efficient *n*-Bu₃SnH mediated reductive denitration and an oxidative carbonylation reaction. The total synthesis spans 14 steps in a good overall yield from simple starting materials. The newly developed synthetic route may readily lend itself to the total synthesis of relevant natural products. Endeavours along this line are underway and will be reported in due course.

Acknowledgements

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- Compound **4** is a 1 : 1 diastereomeric mixture isomeric at the allylic carbon center. In the ESI⁺ the ¹H and ¹³C NMR data were read from the spectra of the mixture.
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