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# Liquid-crystalline circularly polarised TADF emitters for high-efficiency, solution-processable organic light-emitting diodes†

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Achieving a high emission efficiency and a large luminescence asymmetry factor ( $g_{\text{lum}}$ ) in a single molecule exhibiting circularly polarised thermally activated delayed fluorescence (CP-TADF) remains a formidable challenge. In this work, a proof-of-concept, liquid-crystalline CP-TADF molecule is proposed to realise high  $g_{\text{lum}}$  by taking advantage of the order inherent in liquid crystals. Employing a chiral dinaphthol-based CP-TADF molecule as the emissive unit, a pair of liquid-crystalline CP-TADF molecules (*R/S-4*) is synthesised *via* the introduction of six mesogenic moieties. The enantiomers show intense emission at about 520 nm which has clear TADF and liquid-crystalline characteristics. Both enantiomers display symmetrical electronic circular dichroism (CD) and circular polarisation luminescence (CPL) signals as thin films. Impressively, relatively large  $g_{\text{lum}}$  values of 0.11 are realised for the films. Solution-processed devices were fabricated using *R/S-4* as the dopants, with the TADF molecule CzAcSF as the sensitiser. The OLEDs so prepared show a very high maximum external quantum efficiency of 21.2%, revealing a novel strategy for realising large  $g_{\text{lum}}$  values in CP-TADF.

## New concepts

Circularly polarised emitters have attracted much attention in the field of organic semiconductors because they can emit circularly polarised luminescence directly. On account of the unit internal quantum efficiency and the fact that these materials do not require a heavy metal in order to access the triplet manifold, thermally activated delayed fluorescence (TADF) is a very attractive mechanism for CPL emission. Within this approach, two effective strategies can be identified, namely the use of intrinsic chirality and/or one of chiral perturbation. However, in order to realise large values of the luminescence asymmetry factor ( $g_{\text{lum}}$ ), the relatively large electric transition dipole moment and small magnetic transition dipole moment characteristic of organic molecules are not ideal. As such, achieving both high emission efficiencies and large values of  $g_{\text{lum}}$  in a single molecule is a formidable challenge. To address this issue, a proof-of-concept, liquid-crystalline CP-TADF molecule is proposed. Employing the chiral binaphthol-based CP-TADF molecule as the emissive unit, an enantiomeric pair of liquid-crystalline CP-TADF molecules (*R/S-4*) has been synthesised *via* functionalisation of the binaphthol core. Clear TADF and liquid-crystalline characteristics, as well as the rather large  $g_{\text{lum}}$  value of 0.11, are realised for the CP-TADF emitter. This research work for the first time obtains both high emission efficiency and large  $g_{\text{lum}}$  *via* integrating liquid crystal into the TADF emitter and considers the influence that liquid crystallinity may exert.

## Introduction

Circularly polarised luminescence (CPL) is a phenomenon leading to varying emission intensities of left- and right-circularly

polarised light, which can provide information about the excited states of chiral molecular systems.<sup>1</sup> Recently, circularly polarised light from molecular emitters has attracted much attention because of its applications in anti-glare displays, encrypted transmission, optical communication and three-dimensional (3D) displays.<sup>2–4</sup> Compared to traditional methods in which a polariser and waveplate are employed, luminescent circularly polarising materials can generate circularly polarised light directly and hence have higher efficiencies in devices. To date, many CPL materials, including small organic molecules and polymers, have been developed<sup>5,6</sup> and, depending on the emission mechanism, CPL materials are usually classified as emitting *via* fluorescence,<sup>7–11</sup> photoluminescence<sup>12–16</sup> or thermally activated delayed fluorescence (TADF).<sup>17–20</sup> Spin statistics demonstrate that fluorescence

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efficiency of the emitter. As such, achieving a balance between the emission efficiency and  $g_{\text{lum}}$  in CP-TADF materials remains a formidable challenge.

Recently, liquid crystals have aroused interest in the generation of circularly polarised light due to their excellent self-organisation properties<sup>34–36</sup> and some groups have reported a high  $g_{\text{lum}}$  value by doping a chiral emitter into an achiral liquid crystal, proving that the liquid crystal can amplify the  $g_{\text{lum}}$  value.<sup>35,37</sup> Very recently, Li *et al.* achieved a strong CPL signal of  $|g_{\text{lum}}| = 0.47$  using the orientational order parameter of achiral dyes in cholesteric liquid crystal media.<sup>38</sup> Our previous work also demonstrated that integrating liquid crystallinity and phosphorescence can lead to impressive values of  $g_{\text{lum}}$  of  $10^{-2}$  in an OLED device.<sup>39</sup> Encouraged by these results, a proof-of-concept, liquid-crystalline CP-TADF molecule was conceived in which the chiral TADF molecule (**R/S-1**) is employed as the emissive core due to its high emission efficiency and relatively simple preparation.<sup>19</sup> It is then functionalised with six mesogenic moieties which are integrated into the binaphthol unit as two 3,4,5-trisubstituted phenyl rings to give enantiomers **R/S-4** (Fig. 1). The pair of compounds shows symmetric CD and CPL signals as thin films, indicating their chirality in both the ground and excited states, and the  $g_{\text{lum}}$  values of the films prepared from **R/S-4** are  $10^{-1}$ . Employing the TADF material 10-(4-((4-(9H-carbazol-9-yl)phenyl)sulfonyl)-phenyl)-9,9-dimethyl-9,10-dihydroacridine (CzAcSF) as the sensitiser, the resulting solution-processed OLED showed an impressive device performance with the maximum external quantum efficiency (EQE<sub>max</sub>) of 21.2%.

## Results and discussion

### Synthesis and electrochemical properties

As depicted in Scheme S1 (ESI<sup>†</sup>), the key precursor – the di(bromophenyl) analogue of **1** – was prepared according to the previous publication.<sup>19</sup> A typical Suzuki coupling reaction between the triply functionalised mesogenic moiety and compound **1-Br<sub>2</sub>** yielded the chiral enantiomers **R/S-4** which were characterised by NMR spectroscopy, mass spectrometry and HPLC (Fig. S17–S23, ESI<sup>†</sup>). Both emitters possess good solubility in common organic solvents, such as toluene, dichloromethane and THF, and show good thermal stability, exhibiting a 5% weight loss at 385 °C (Fig. S1, ESI<sup>†</sup>). The redox behaviour of the materials was investigated by cyclic voltammetry (CV) in CH<sub>2</sub>Cl<sub>2</sub> (Fig. S2, ESI<sup>†</sup>) and an irreversible oxidation potential ( $E_{\text{ox}}$ ) of 0.84 V was found for **R-4** (as the enantiomers have identical physical properties, only **R-4** is discussed in the text). The highest occupied molecular orbital (HOMO) energy level can then be calculated as  $-5.19$  eV using the formula  $E_{\text{HOMO}} = -(E_{\text{ox}} + 4.8 - E_{1/2})$  in which ferrocene ( $E_{1/2} = 0.46$  V) is used, as in ref. 40. The lowest unoccupied molecular orbital (LUMO) energy level can be inferred from the HOMO energy and the optical bandgap ( $E_{\text{g}}^{\text{opt}}$ ) is calculated from the absorption onset. Based on the formula  $E_{\text{LUMO}} = E_{\text{HOMO}} + E_{\text{g}}^{\text{opt}}$ , the corresponding LUMO energy level is determined as  $-2.71$  eV.

### Theoretical calculations

Density functional theory (DFT) and time-dependent DFT (TD-DFT) calculations were carried out to optimise the ground and excited state geometries (to simplify the molecular structure, the alkoxy chains is omitted), using Gaussian 09 with the B3LYP functional and the def2-SVP basis set. The HOMO and LUMO density distributions, energy gaps, oscillator strengths, excited state energy and spin-orbit coupling matrix elements ( $\langle S/H_{\text{SOC}}/T \rangle$ ) so obtained are shown in Fig. 2. The enantiomers clearly exhibit a distorted molecular geometry, which is beneficial to the spatial separation between the FMO (frontier molecular orbitals). Thus, the HOMO is located predominantly on the carbazole units, while the LUMO is located mainly on the phthalonitrile. According to previous reports,<sup>41</sup> the 1,1'-binaphthyl moiety can utilise the chiral perturbation effect to realise circular polarisation, while the mesogenic unit has a negligible effect on the electronic distribution. To further understand the characters of the  $S_0 \rightarrow S_1$  and  $S_0 \rightarrow T_1$  transitions, natural transition orbital (NTO) analysis was performed for the excited state ( $S_n$  and  $T_n$ ) using TD-DFT (Fig. 2(b)). Substantial overlap occurs between holes and particles in the  $S_1$  and  $T_1$  states, indicating that the singlet and triplet states are mainly endowed with local excitation (LE) properties. The spatial separation of FMOs leads to the small value of  $\Delta E_{\text{ST}}$  required for TADF character, which is calculated to be 0.09 eV. A high spin-orbit coupling (SOC) matrix element of  $\langle S_1/H_{\text{SOC}}/T_1 \rangle$  was evaluated to be  $0.552 \text{ cm}^{-1}$ , implying different electronic structures of the  $S_1$  and  $T_1$  states and resulting in an effective reverse intersystem crossing (rISC) process between the  $T_1$  and  $S_1$  states (Table S1, ESI<sup>†</sup>). Notably, both  $\langle S_1/H_{\text{SOC}}/T_2 \rangle$  and  $\langle S_1/H_{\text{SOC}}/T_3 \rangle$  also have large SOC values, indicating that multiple rISC processes can occur in this molecule.<sup>42</sup>

### Liquid crystal behaviour

The liquid crystal properties of the emitters were investigated by polarising optical microscopy (POM) (Fig. 3 and Fig. S3, ESI<sup>†</sup>), differential scanning calorimetry (DSC) (Fig. S4, ESI<sup>†</sup>) and small-angle X-ray scattering (SAXS) (Fig. 3 and Fig. S3, ESI<sup>†</sup>). Importantly, the thermal behaviour of the two enantiomers is, as expected, essentially identical. Thus, on heating from room temperature, DSC shows two transitions at 164.5 °C ( $\Delta H \approx 7 \text{ kJ mol}^{-1}$ ) and at 184.8 °C ( $\Delta H \approx 11.5 \text{ kJ mol}^{-1}$ ), which equate to around  $1 \text{ kJ mol}^{-1}$  and  $2 \text{ kJ mol}^{-1}$  per mesogenic group, respectively (Table S2, ESI<sup>†</sup>). Despite the high molecular weight of the material ( $>3500 \text{ g mol}^{-1}$ ), the higher-temperature transition is perfectly reversible, while there is a small degree of supercooling in the lower-temperature transition. The absence of a large-enthalpy transition is consistent with the compound being in a mesophase at ambient temperature (*i.e.*, there is no readily apparent melting/crystallisation event).

The higher-temperature transition is identified as the clearing point, above which the material shows optical extinction under crossed polarisers, but slow cooling below this temperature allows development of a focal-conic fan texture punctuated with optically



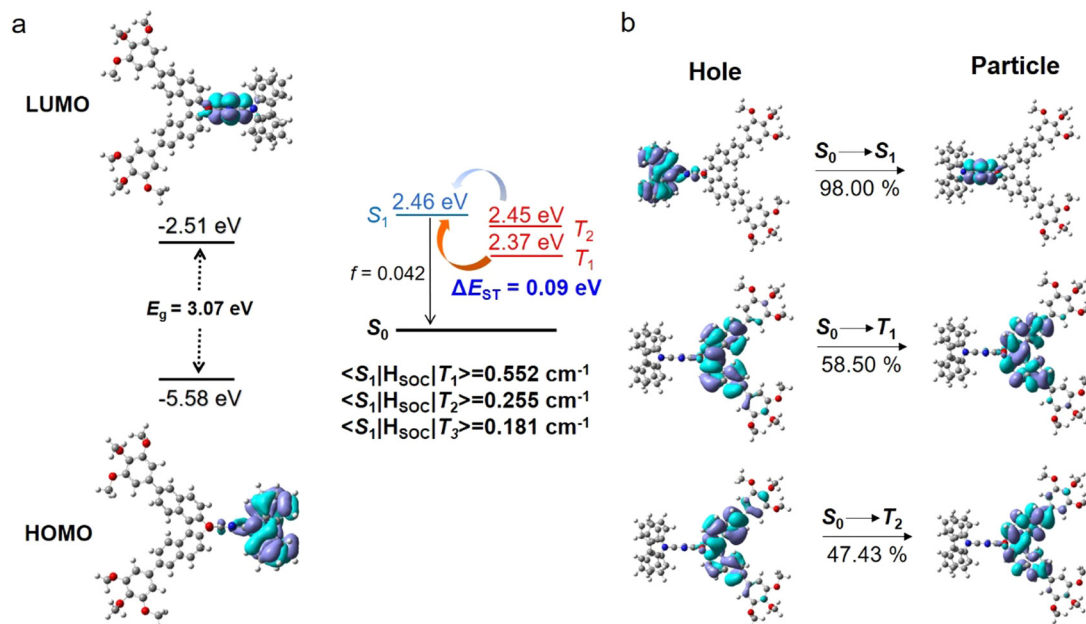


Fig. 2 Distributions of: (a) frontier molecular orbitals (FMOs) and (b) natural transition orbitals (NTOs, hole/particle wave functions), as well as energy levels, oscillator strength ( $f$ ), and spin-orbit coupling (SOC) constants for **R-4** (**S-4** is identical).

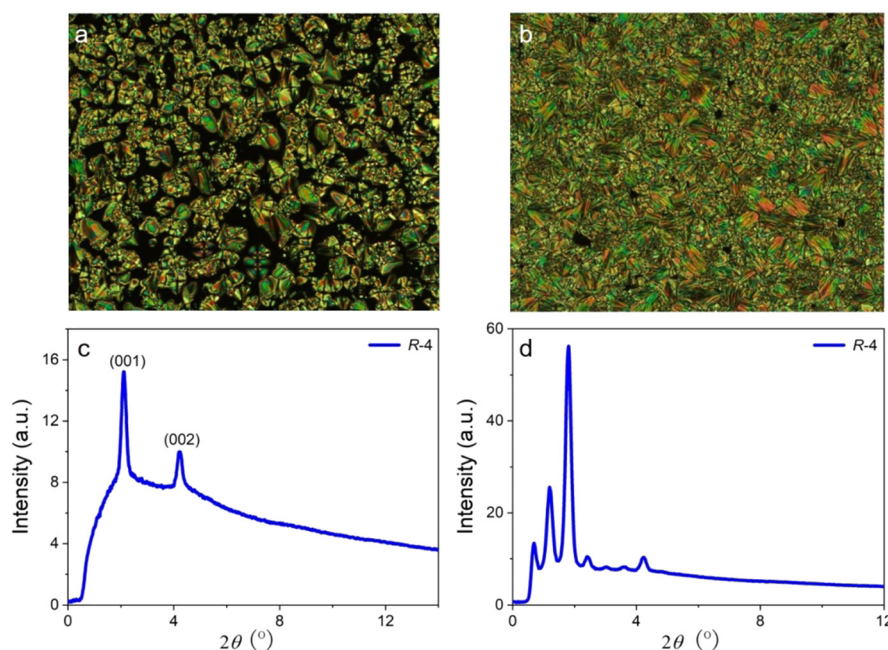


Fig. 3 POM image of **R-4** on cooling: (a) in the SmA\* phase at 183 °C and (b) in the more ordered mesophase at 155 °C. SAXS pattern of **R-4** on cooling: (c) in the SmA\* phase at 173 °C and (d) in the more ordered mesophase at 148.4 °C.

extinct domains of homeotropic orientation characteristic of a SmA\* phase (the phase is designated SmA\* rather than simply SmA on account of the chiral nature of the compound, although this has no impact on the observed texture) with clear elliptical and hyperbolic defects evident, although weak striations are observed across the back of the fans. Further cooling leads to a phase transition in which much stronger striations or deformations are observed across the back of the fan texture, indicative of a

mesophase with greater order. There is no change in the homeotropic domains indicating that molecular tilt has not been introduced, which may have explained the fragmented or striated focal conic texture at the phase transition. In addition, the lower temperature phase does not readily shear upon mechanical stress, again indicative of a phase with greater order.

SAXS data are consistent with the assignment of the upper phase as the SmA\* phase, and a primary reflection is observed





at  $2\theta = 2.12^\circ$ , corresponding to the (001) reflection and a spacing of 41.6 Å. In addition, there is a (002) reflection at  $2\theta = 4.21^\circ$  which corresponds to a spacing of 21.0 Å. The observation of a second-order reflection implies a greater degree of layer correlation than is commonly observed and is consistent with the weak striations in the optical texture. From published single crystal structures,<sup>43,44</sup> it is possible to estimate the fully extended length of the molecule to be around 50 Å, and so with a measured SmA\* spacing of *ca.* 42 Å, it is likely that some sort of bilayer organisation exists that may involve in the interdigitation of the two *ortho*-carbazole groups, which will be oriented perpendicular to the terephthalonitrile ring to which they are attached, perhaps also with some interdigitation and/or folding of the terminal pentyl chains.

For the lower-temperature phase, many more reflections are observed in the low-angle region in the range  $0.68^\circ \leq 2\theta \leq 4.23^\circ$ , corresponding to spacings of  $129.8 \text{ Å} \leq d \leq 20.9 \text{ Å}$ . However, their appearance is not accompanied by the development of reflections at mid or wider angles, which would imply three-dimensional order. These low-angle reflections indicate a much greater degree of order, although the fact that the widths of the reflections are very similar to those seen in the pattern for the SmA\* phase implies that this is a mesophase of sorts.

### Photophysical properties

As the two compounds are related as enantiomers, **R-4** is selected for detailed discussion. The UV-vis absorption and photoluminescent (PL) spectra of **R-4** were performed both in solution and doped in mCP films (Fig. 4); the related data are

summarised in Table 1. Two distinct absorption bands were detected in toluene solution, in which the very weak band at 425 nm is attributed to an intramolecular charge-transfer (ICT) transition, while the intense absorption band at 330 nm is ascribed as a  $\pi-\pi^*$  transition. The emission peak was detected at 525 nm for **R-4** in toluene solution and its broad, structureless profile indicates that it originates from a charge-transfer (CT) state. There is also evident solvatochromism and the 525 nm emission peak of **R-4** in toluene red shifts to 589 nm in a much more polar solvent DMF, consistent with the stabilisation of the charge-separated excited state by more polar solvents and related to the strong intramolecular charge transfer ability of this material (Fig. S5, ESI†).

The singlet ( $S_1$ ) and triplet ( $T_1$ ) energy values were obtained from low-temperature (77 K) fluorescence (Fl) and phosphorescence (Ph) measurements, respectively (Fig. S6, ESI†). According to the measured onset values, the  $S_1$  and  $T_1$  energy levels were evaluated to be 2.64 and 2.59 eV, respectively, so that the corresponding value of  $\Delta E_{ST}$  is calculated to be 0.05 eV in toluene. These values are consistent with the computational data reported above and the observation of TADF characteristics.

Compared with the PL in solution, the emission spectrum of **R-4** doped in a mCP film shows similar emission to the peak at 521 nm. According to the onset of both fluorescence and phosphorescence emission at low temperatures, the  $S_1$  and  $T_1$  energy levels of the doped film were evaluated to be 2.67 and 2.62 eV for **R-4**, respectively (Fig. 4(b)) in turn giving a value of 0.05 eV for  $\Delta E_{ST}$  (similarly results were obtained for **S-4**, shown

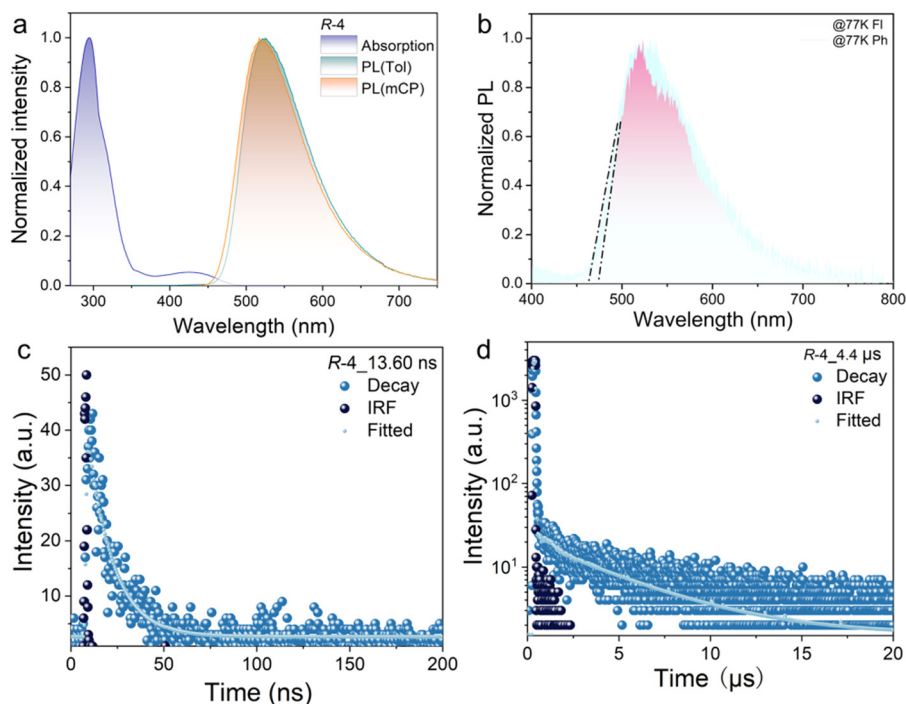


Fig. 4 (a) UV-vis spectra and PL spectra of **R-4** measured in toluene solution ( $10^{-5}$  M) and 5 wt% doped in mCP film ( $\lambda_{\text{ex}} = 420$  nm); (b) low-temperature fluorescence (Fl) and phosphorescence (Ph) of **R-4** doped in mCP film ( $\lambda_{\text{ex}} = 370$  nm, delayed 2 ms); and (c) and (d) transient PL decay curves ( $\lambda_{\text{ex}} = 420$  nm) of 5 wt% **R-4** doped in mCP film.



Table 1 Photophysical data for the emitters

	$\lambda_{\text{abs}}^a$ (nm)	$\lambda_{\text{em}}^a$ (nm)	$\lambda_{\text{Fl}}$ (nm)	$\lambda_{\text{Ph}}$ (nm)	$\Delta E_{\text{ST}}$ (eV)	$\tau_{\text{p}}^c$ (ns)	$\tau_{\text{d}}^c$ ( $\mu\text{s}$ )	PLQY <sup>c</sup> (%)	$g_{\text{PL}}^d$ ( $10^{-1}$ )	$E_{\text{g}}^{\text{opt}}$ (eV)
<b>R-4</b>	294, 431	524	525 <sup>b</sup> (523) <sup>c</sup>	533 <sup>b</sup> (518) <sup>c</sup>	0.05 <sup>b</sup> (0.05) <sup>c</sup>	13.60	4.4	51.3	0.8	2.48
<b>S-4</b>	292, 431	523	524 <sup>b</sup> (523) <sup>c</sup>	533 <sup>b</sup> (517) <sup>c</sup>	0.06 <sup>b</sup> (0.05) <sup>c</sup>	13.69	4.7	53.7	1.1	2.45

<sup>a</sup> Measured in toluene solution at room temperature. <sup>b</sup> Fluorescence and phosphorescence emission in toluene solution at 77 K and the calculated values for the onsets of fluorescence and phosphorescence. <sup>c</sup> In mCP films. <sup>d</sup> In film;  $E_{\text{g}}^{\text{opt}}$  was calculated via  $E_{\text{g}}^{\text{opt}} = 1240/\lambda$ , in which  $\lambda$  is collected from the band-edge absorption.

in Fig. S7, ESI<sup>†</sup>). This is good for rISC. To further confirm the TADF behaviour of the material, transient PL spectra of the doped mCP films were measured at room temperature. As shown in Fig. 4, **R-4** displays both a clearly prompt decay with an exciton lifetime ( $\tau_{\text{p}}$ ) of 13.6 ns and a delayed decay with an exciton lifetime ( $\tau_{\text{d}}$ ) of 4.4  $\mu\text{s}$ , indicating contributions from triplet excitons. Additionally, the temperature-dependent PL decay curves (recorded at 100, 200 and 300 K) showed an increase in the delayed component with increasing temperature, confirming the TADF mechanism (Fig. S8, ESI<sup>†</sup>). The photoluminescence quantum yield (PLQY) value of a film of 5 wt% **R-4** in mCP is 51.3%, similar to that found with **R/S-1**.<sup>19</sup> Based on the emission lifetime and PLQY, the photophysical constants of **R-4**, such as the rate constants of fluorescence ( $k_{\text{r}}$ ), nonradiative transition ( $k_{\text{nr}}$ ), intersystem crossing ( $k_{\text{ISC}}$ ), and

reverse intersystem crossing ( $k_{\text{rISC}}$ ), were calculated. As shown in Table S3 (ESI<sup>†</sup>), **R-4** shows a reasonable value of  $k_{\text{rISC}}$  of  $6.0 \times 10^4 \text{ s}^{-1}$  and  $k_{\text{d}}$  of  $2.27 \times 10^5 \text{ s}^{-1}$ , which demonstrates that there is a minor effect on the emission properties after the introduction of the mesogenic groups.<sup>19</sup>

In order to investigate the ground and excited state chirality of **R/S-4**, the electron circular dichroism (CD) and circular polarisation luminescence (CPL) spectra of **R/S-4** were evaluated in both solution and the as a film. Unfortunately, a very weak signal was obtained for the samples in toluene solution, whereas intense chiroptical signals were achieved for the films. As shown in Fig. S9 (ESI<sup>†</sup>), the symmetrical CD spectra indicate their enantiomeric relationship in the ground state and, in particular, the signal at ca 420 nm indicates successful chiral perturbation between the chiral unit and the TADF chromophore.<sup>19</sup> As shown in Fig. 5,

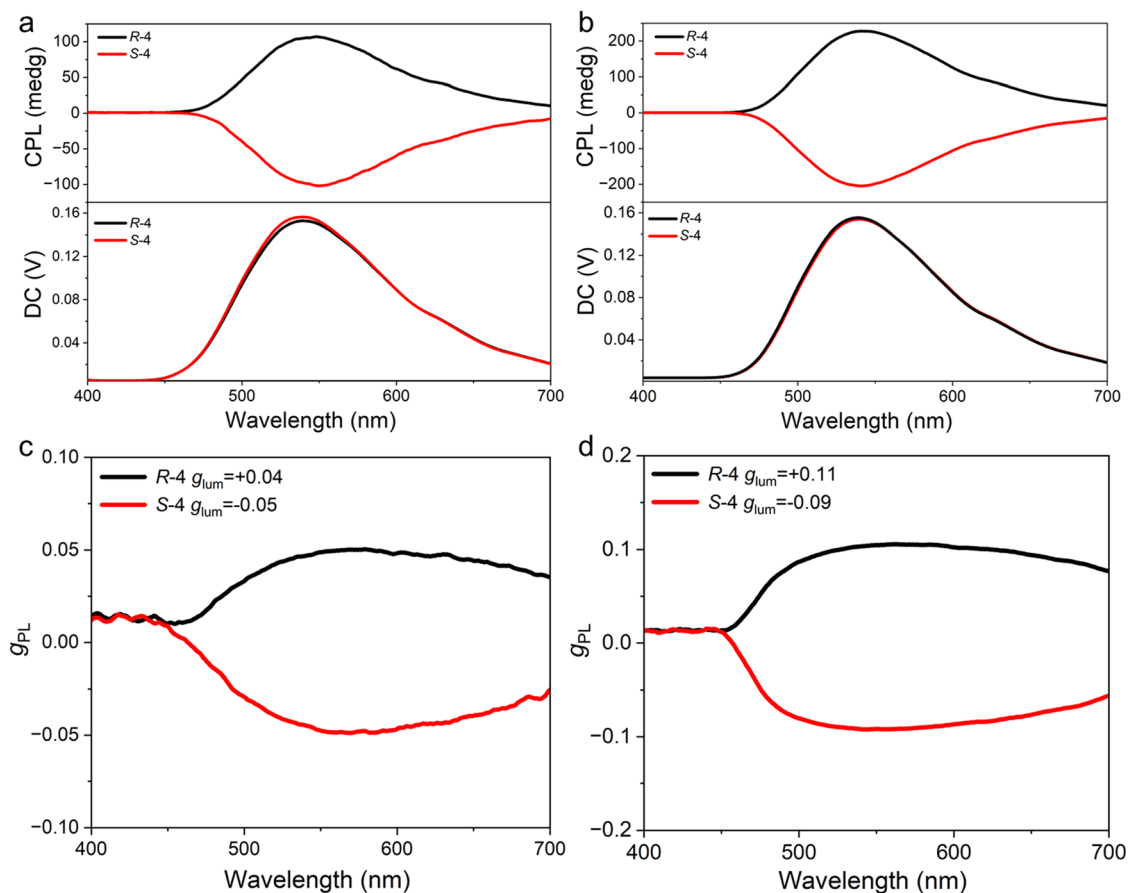


Fig. 5 (a) CPL spectra and (c)  $g_{\text{lum}}$  values of CPL versus wavelength of **R/S-4** in neat films; (b) CPL spectra and (d)  $g_{\text{lum}}$  values of CPL versus wavelength of **R/S-4** in doped mCP films ( $\lambda_{\text{ex}} = 340 \text{ nm}$ ).



the CPL spectra reflect the fact that this new TADF emitter can lead to chiral luminescence from the excited state. Impressive  $|g_{\text{lum}}|$  values as determined at the maximum emission peaks are about 0.04 and 0.05 for **R-4** and **S-4** in neat films, respectively, which are an order of magnitude greater than those of the parent core **R/S-1** ( $1.3 \times 10^{-3}$ ),<sup>19</sup> which could originate from the integrated liquid crystal concept. Interestingly, the mCP-doped films present clearly increased  $|g_{\text{lum}}|$  values of 0.11 and 0.09 for **R-4** and **S-4**, respectively (Fig. 5(c)). These increased values can be reasonably explained by suppressed intermolecular interaction. Thus, the strategy of introducing liquid crystallinity into the CP-TADF molecule is an effective method for amplification of the  $g_{\text{lum}}$  value.

### Device performance

In order to explore the electroluminescent (EL) properties of the materials, solution-processed devices were constructed with the following architecture: ITO/poly (3,4-ethylene-dioxythiophene):poly (styrene sulfonate) (PEDOT:PSS)/emitter layer (50 nm)/1,3,5-tris(3-pyridyl-3-phenyl)benzene (TmPyPB) (45 nm)/LiF (0.5 nm)/Al (120 nm) (Fig. S10a, ESI†). In this device structure, PEDOT:PSS and LiF are used as the hole-injection layer and the electron-injection layer, respectively, and TmPyPB is used as the electron-transport layer. Lithium fluoride (LiF) and aluminum (Al) were utilised as the electron-injection layer and cathode, respectively. The emission layer consists of the host mCP and the guest **R/S-4**, and the dopant concentrations were 5, 10 and 15 wt%. The relevant EL data are summarised in Table S4 (ESI†).

As shown in the inset to Fig. S9c (ESI†), all devices show their emission peak at 526 nm, and CIE coordinates are found consistently at (0.31, 0.56), with the EL spectra being invariant with dopant concentration. In addition, emission from the mCP host is absent, indicating complete energy transfer between host and dopant. The change in current density–voltage–luminance ( $J$ – $V$ – $L$ ) curves as a function of dopant concentration suggests that the carrier capture process is predominantly an emission mechanism in the devices and that increasing the dopant concentration apparently reduces the device performance significantly, likely due to concentration quenching. Thus, the device prepared with **R-4** exhibits its best performance at 5 wt% of the emitter with an EQE<sub>max</sub> of 10.2%, a maximum brightness ( $L_{\text{max}}$ ) of 444.9 cd m<sup>-2</sup> and a current efficiency (CE<sub>max</sub>) of 32.55 cd A<sup>-1</sup> (Table S4 and Fig. S10, ESI†).

### Performance of sensitised devices

To further improve the EL performance, the sensitiser concept was deployed. Previous work demonstrated that a TADF sensitiser can relieve exciton quenching and improve device performance *via* effective Förster energy transfer (FRET).<sup>45</sup> Therefore, the sensitisers and emitters should be chosen carefully for effective FRET, informed by the equation for the rate of Förster energy transfer ( $k_{\text{FET}}$ ):<sup>46</sup>

$$k_{\text{FET}} = \frac{1}{\tau_{\text{D}}} \times \frac{1}{r^6} \times \frac{9000(\ln 10)\kappa_{\text{p}}^2 Q_{\text{D}} I}{N_{\text{A}} 128\pi^5 n_{\text{D}}^4}$$

The magnitude of  $k_{\text{FET}}$  is, therefore, greatly influenced by  $\tau_{\text{D}}$ ,  $r$ ,  $Q_{\text{D}}$ , and  $I$ , where  $\tau_{\text{D}}$  is the excited-state radiative lifetime of

the sensitiser;  $r$  is the separation distance between sensitiser and emitter;  $\kappa_{\text{p}}$  is a parameter related to the relative orientation of the sensitiser and emitter dipoles;  $Q_{\text{D}}$  is the PLQY of the sensitiser;  $I$  is the spectral overlap integral between the emission of sensitisers and the absorption of the emitters;  $N_{\text{A}}$  is the Avogadro constant and  $n_{\text{D}}$  is the refractive index of the medium. Accordingly, decreasing  $\tau_{\text{D}}$  and  $r$ , and/or increasing  $Q_{\text{D}}$  and  $I$  are important in achieving effective energy transfer. The magnitude of  $r$  can be enhanced by increasing the doping concentration of the sensitisers or emitters, although most TADF materials show reduced PLQY at higher doping concentration due to strong concentration quenching. Therefore, to maintain the luminescence performance of the emitter, low doping concentrations of the emitter and high doping concentrations of the sensitiser are usually adopted to ensure effective energy transfer.<sup>47</sup>

Based on the above discussion, 10-(4-((4-(9H-carbazol-9-yl)phenyl)sulfonyl)phenyl)-9,9-dimethyl-9,10-dihydroacridine (CzAcSF),<sup>48</sup> a blue-emitting TADF luminophore with  $\Delta E_{\text{ST}} = 0.07$  eV was selected as the sensitiser. As depicted in Fig. S11 (ESI†), the PL spectrum of CzAcSF overlaps with the UV spectra of **R/S-4**, the prerequisite for FRET energy transfer. In order to reduce the high exciton density and related decay processes such as triplet–triplet annihilation (TTA) and triplet polaron annihilation (TPA), mCP is added to the emitter layer as the host to further increase the distance between the sensitiser and guest, and so suppress the Dexter energy transfer (DET).<sup>49</sup> Using **R-4** as the emitter, the emission band from CzAcSF gradually disappears with increasing dopant concentration (Fig. S12, ESI†), showing that there is efficient energy transfer between the TADF sensitiser and **R-4**. Furthermore, the transient PL decay curves of co-doped films show that the delayed component decreases with the increased dopant concentration, also implying an effective FRET process (Fig. S13, ESI†).<sup>49</sup> In addition, the doped film displays increased PLQY from 50.8% (mCP: 1.5 wt% **R-4**) to 55.2% (mCP: 30 wt% CzAcSF: 1.5 wt% **R-4**).

Based on the above observations, both sensitised and non-sensitised solution-processed devices were fabricated for comparison. The device structure of the non-sensitised device is ITO/PEDOT:PSS/mCP:  $x$  wt% **R-4** (50 nm)/TmPyPB (45 nm)/LiF (0.5 nm)/Al (120 nm) (where  $x = 0.5$  wt% (D1), 1 wt% (D2) and 1.5 wt% (D3)). The sensitised device structure is ITO/PEDOT:PSS/mCP: 30 wt% CzAcSF:  $x$  wt% **R-4** (50 nm)/TmPyPB (45 nm)/LiF (0.5 nm)/Al (120 nm) (where  $x = 0.5$  wt% (S1), 1 wt% (S2) and 1.5 wt% (S3)). Use of such a low dopant concentration effectively suppresses Dexter electron transfer in the device. The dopant concentration of sensitiser was varied from 10 wt% to 30 wt%, and the corresponding device performances are summarised in Table S5 (ESI†). The device achieved its best performance at 30 wt%.

As the concentration of **R-4** increases from 0.5 wt% to 1.5 wt%, the non-sensitised devices show values of EQE<sub>max</sub> between 3.6% and 3.2% (Table 2). In the sensitised device, the EL spectra exhibit maximum emission at 510 nm, similar to the non-sensitised device (Fig. 6(c) inset, Table 2), consistent with



Table 2 Device data for the compound

Device	$V_{\text{on}}$ (V)	$L_{\text{max}}$ ( $\text{cd m}^{-2}$ )	$CE_{\text{max}}$ ( $\text{cd A}^{-1}$ )	$\text{EQE}_{\text{max}}$	Peak (nm)	FWHM	CIE
D1	4.4	716	10.9	3.6	516	85.1	(0.26, 0.51)
D2	4.4	443.1	10.8	3.3	516	80.2	(0.26, 0.54)
D3	3.6	378	10.9	3.2	516	78.6	(0.26, 0.53)
S1	4.0	1894	45.4	16.7	504	97.1	(0.21, 0.39)
S2	4.4	2304	54.7	18.9	510	95.6	(0.24, 0.45)
S3	4.4	2570	62.9	21.2	510	95.6	(0.24, 0.45)

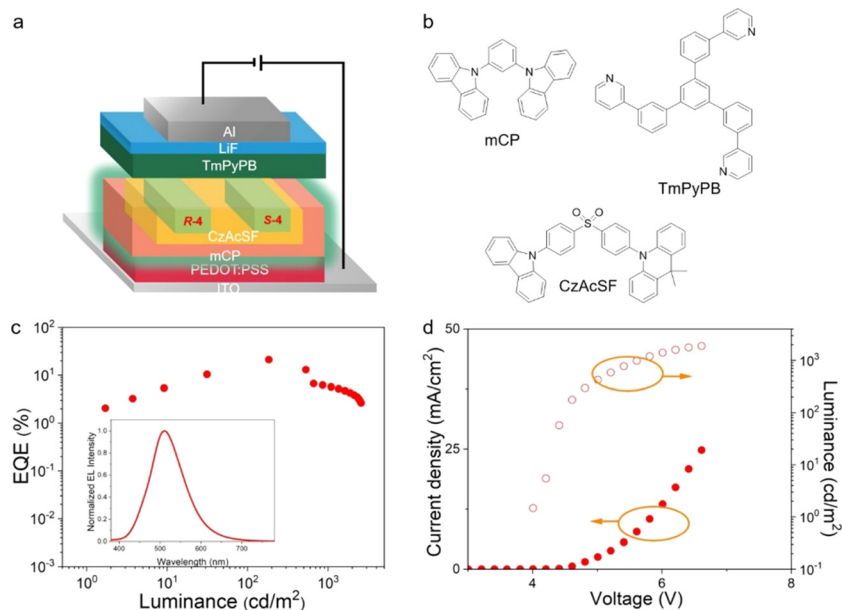


Fig. 6 (a) Device architecture and (b) materials used in the devices; (c) EQE–current density (inset: EL spectra of the devices) and (d) current density–voltage–luminance curves of the sensitised devices with 1.5 wt% dopant concentration.

the emission from the terminal emitter. This phenomenon also demonstrates that there is complete energy transfer between the host, sensitizer and dopant in the device. Relatively low turn-on voltages ( $V_{\text{on}}$ ) of  $4.2 \pm 0.2$  V were detected for the TADF-sensitized devices which, intriguingly, showed excellent performance with  $\text{EQE}_{\text{max}} = 21.1\%$ ,  $L_{\text{max}} = 2570 \text{ cd m}^{-2}$  and  $CE_{\text{max}} = 62.9 \text{ cd A}^{-1}$ . Data for the related *S-4* based devices are found in Fig. S14 and Table S6 (ESI<sup>†</sup>). The  $\text{EQE}_{\text{max}}$  of the sensitized device is, therefore, approximately six times higher than that of the non-sensitized device. The energy transfer and device emission processes are depicted graphically in Fig. S15 (ESI<sup>†</sup>). The electron–hole recombination on mCP produces singlet and triplet excitons. The singlet excitons in the host can transfer to both the sensitizer and terminal emitter by the FRET process, while the triplet exciton can transfer to the sensitizer *via* the DET process. Due to the rISC process, the triplet excitons of the sensitizer can be converted to singlet excitons, which can transfer to the terminal emitter by the FRET process. Eventually both singlet and triplet excitons can be achieved by the terminal emitter through the efficient FRET process and then achieve a high-performance sensitized fluorescent OLED. Unfortunately, the CPEL signal is very weak, which can be explained by the very low dopant concentration (Fig. S16, ESI<sup>†</sup>).

## Conclusions

In summary, an enantiomeric pair of liquid-crystalline CP-TADF molecules *R/S-4* was prepared using the chiral perturbation strategy. They exhibited intense green emission, similar to the CP-TADF emission core, showing that the introduction of the mesogenic moieties has a negligible effect on the emission properties. A delayed component of emission was clearly detected for the molecules in solution and as a neat film. *R/S-4* showed symmetrical CD and CPL signals as films and the  $g_{\text{lum}}$  values reached  $10^{-1}$  in the doped film – two orders of magnitude higher than that found for the parent chromophore. This phenomenon demonstrates that liquid-crystallinity has a positive effect on amplification of  $g_{\text{lum}}$ . Impressively, the solution-processed sensitized OLED using *R/S-4* as the dopants achieved a maximum EQE of 21.2%. As such, we report an effect strategy for amplifying the  $g_{\text{lum}}$  value of the CP-TADF molecule.

Through the modification of a known binaphthol emitter to confer liquid crystal properties, a material with an intrinsically very high luminescence asymmetry has been realised. Furthermore, its incorporation into a sensitized, solution-processed OLED leads to a high device external quantum efficiency, being approximately six times more efficient than that of the non-sensitized





device. Further work is required to understand the very positive effect of the liquid-crystalline modifications to *R/S-4*, but a hint may lie in the high value of the EQE compared to the device PLQY, which would suggest that, even at the rather low concentrations of *R/S-4* used.

## Author contributions

The project was conceived by YW, WZ and DWB. Experimental work was carried out by BH, QZ, QD, XY, SJC and WQ. Data were analysed by SJC, DWB, PD and YW. The manuscript was written by BH, QZ and DWB, and was edited by DWB, PD and YW.

## Conflicts of interest

There are no conflicts to declare.

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## Notes and references

- 1 F. S. Richardson and J. P. Riehl, *Chem. Rev.*, 2002, **77**, 773–792.
- 2 J. R. Brandt, F. Salerno and M. J. Fuchter, *Nat. Rev. Chem.*, 2017, **1**, 0045.
- 3 J. Kumar, T. Nakashima and T. Kawai, *J. Phys. Chem. Lett.*, 2015, **6**, 3445–3452.
- 4 R. Farshchi, M. Ramsteiner, J. Herfort, A. Tahraoui and H. T. Grahn, *Appl. Phys. Lett.*, 2011, **98**, 162508.
- 5 D. W. Zhang, M. Li and C. F. Chen, *Chem. Soc. Rev.*, 2020, **49**, 1331–1343.
- 6 J. Han, S. Guo, H. Lu, S. Liu, Q. Zhao and W. Huang, *Adv. Opt. Mater.*, 2018, **6**, 1800538.
- 7 J. Jimenez, L. Cerdan, F. Moreno, B. L. Maroto, I. Garcia-Moreno, J. L. Lunkley, G. Muller and S. de la Moya, *J. Phys. Chem. C*, 2017, **121**, 5287–5292.
- 8 E. M. Sanchez-Carnerero, F. Moreno, B. L. Maroto, A. R. Agarrabeitia, M. J. Ortiz, B. G. Vo, G. Muller and S. de la Moya, *J. Am. Chem. Soc.*, 2014, **136**, 3346–3349.
- 9 D. Q. He, H. Y. Lu, M. Li and C. F. Chen, *Chem. Commun.*, 2017, **53**, 6093–6096.
- 10 H. Sakai, T. Kubota, J. Yuasa, Y. Araki, T. Sakanoue, T. Takenobu, T. Wada, T. Kawai and T. Hasobe, *J. Phys. Chem. C*, 2016, **120**, 7860–7869.
- 11 M. Li, Y.-F. Wang, D.-W. Zhang, D. Zhang, Z.-Q. Hu, L. Duan and C.-F. Chen, *Sci. China Mater.*, 2020, **64**, 899–908.
- 12 T. Y. Li, Y. M. Jing, X. Liu, Y. Zhao, L. Shi, Z. Tang, Y. X. Zheng and J. L. Zuo, *Sci. Rep.*, 2015, **5**, 14912.
- 13 L. Norel, M. Rudolph, N. Vanthuyne, J. A. Williams, C. Lescop, C. Roussel, J. Autschbach, J. Crassous and R. Reau, *Angew. Chem., Int. Ed.*, 2010, **49**, 99–102.
- 14 Z. Jiang, J. Wang, T. Gao, J. Ma, Z. Liu and R. Chen, *ACS Appl. Mater. Interfaces*, 2020, **12**, 9520–9527.
- 15 G. Qian, X. Yang, X. Wang, J. D. Herod, D. W. Bruce, S. Wang, W. Zhu, P. Duan and Y. Wang, *Adv. Opt. Mater.*, 2020, **8**, 2000775.
- 16 J. Han, S. Guo, J. Wang, L. Wei, Y. Zhuang, S. Liu, Q. Zhao, X. Zhang and W. Huang, *Adv. Opt. Mater.*, 2017, **5**, 1700359.
- 17 Z. G. Wu, H. B. Han, Z. P. Yan, X. F. Luo, Y. Wang, Y. X. Zheng, J. L. Zuo and Y. Pan, *Adv. Mater.*, 2019, **31**, e1900524.
- 18 W.-L. Zhao, Y.-F. Wang, S.-P. Wan, H.-Y. Lu, M. Li and C.-F. Chen, *CCS Chem.*, 2022, **4**, 3540–3548.
- 19 S. Feuillastre, M. Pauton, L. Gao, A. Desmarchelier, A. J. Riives, D. Prim, D. Tondelier, B. Geffroy, G. Muller, G. Clavier and G. Pieters, *J. Am. Chem. Soc.*, 2016, **138**, 3990–3993.
- 20 T. Imagawa, S. Hirata, K. Totani, T. Watanabe and M. Vacha, *Chem. Commun.*, 2015, **51**, 13268–13271.
- 21 M. Pope, H. P. Kallmann and P. Magnante, *J. Chem. Phys.*, 1963, **38**, 2042–2043.
- 22 M. Baldo, D. O'Brien, M. E. Thompson and S. R. Forrest, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1999, **60**, 14422–14428.
- 23 K. S. Yook and J. Y. Lee, *Adv. Mater.*, 2012, **24**, 3169–3190.
- 24 M. Y. Wong and E. Zysman-Colman, *Adv. Mater.*, 2017, **29**, 1605444.
- 25 M. Li and C.-F. Chen, *Org. Chem. Front.*, 2022, **9**, 6441–6452.
- 26 S.-P. Wan, H.-Y. Lu, M. Li and C.-F. Chen, *J. Photochem. Photobiol., C*, 2022, **50**, 100500.
- 27 C. Liao, Y. Zhang, S. H. Ye and W. H. Zheng, *ACS Appl. Mater. Interfaces*, 2021, **13**, 25186–25192.
- 28 W. Yang, N. Li, J. Miao, L. Zhan, S. Gong, Z. Huang and C. Yang, *CCS Chem.*, 2022, **4**, 3463–3471.
- 29 Z.-P. Yan, T.-T. Liu, R. Wu, X. Liang, Z.-Q. Li, L. Zhou, Y.-X. Zheng and J.-L. Zuo, *Adv. Funct. Mater.*, 2021, **31**, 2103875.
- 30 F.-M. Xie, J.-X. Zhou, X.-Y. Zeng, Z.-D. An, Y.-Q. Li, D.-X. Han, P.-F. Duan, Z.-G. Wu, Y.-X. Zheng and J.-X. Tang, *Adv. Opt. Mater.*, 2021, **9**, 2100017.
- 31 L. Zhou, F. Ni, N. Li, K. Wang, G. Xie and C. Yang, *Angew. Chem., Int. Ed.*, 2022, **61**, e202203844.
- 32 E. M. Sanchez-Carnerero, A. R. Agarrabeitia, F. Moreno, B. L. Maroto, G. Muller, M. J. Ortiz and S. de la Moya, *Chem*, 2015, **21**, 13488–13500.
- 33 Z.-L. Gong, X. Zhu, Z. Zhou, S.-W. Zhang, D. Yang, B. Zhao, Y.-P. Zhang, J. Deng, Y. Cheng, Y.-X. Zheng, S.-Q. Zang, H. Kuang, P. Duan, M. Yuan, C.-F. Chen, Y. S. Zhao, Y.-W. Zhong, B. Z. Tang and M. Liu, *Sci. China: Chem.*, 2021, **64**, 2060–2104.
- 34 Y. Li, K. Liu, X. Li, Y. Quan and Y. Cheng, *Chem. Commun.*, 2020, **56**, 1117–1120.
- 35 K. Watanabe, I. Osaka, S. Yorozyua and K. Akagi, *Chem. Mater.*, 2012, **24**, 1011–1024.
- 36 B. A. San Jose, S. Matsushita and K. Akagi, *J. Am. Chem. Soc.*, 2012, **134**, 19795–19807.



- 37 D. Xu, X. Hua, C. Liu, J. Luo, W. H. Zheng and Y. Cheng, *ACS Appl. Mater. Interfaces*, 2023, **15**, 25783–25790.
- 38 Y. Li, Y. Chen, H. Li, C. Liu, L. Li, Y. Quan and Y. Cheng, *Angew. Chem., Int. Ed.*, 2023, e202312159.
- 39 P. Fan, Z. Fang, S. Wang, Q. Dong, C. Xiao, A. J. McEllin, D. W. Bruce, W. Zhu and Y. Wang, *Chin. Chem. Lett.*, 2023, **34**, 107934.
- 40 A. K. Gupta, W. Li, A. Ruseckas, C. Lian, C. L. Carpenter-Warren, D. B. Cordes, A. M. Z. Slawin, D. Jacquemin, I. D. W. Samuel and E. Zysman-Colman, *ACS Appl. Mater. Interfaces*, 2021, **13**, 15459–15474.
- 41 Z. Liu, Z. Jiang, C. He, Y. Chen and Z. Guo, *Dyes Pigm.*, 2020, **181**, 108593.
- 42 C. Lin, P. Han, S. Xiao, F. Qu, J. Yao, X. Qiao, D. Yang, Y. Dai, Q. Sun, D. Hu, A. Qin, Y. Ma, B. Z. Tang and D. Ma, *Adv. Funct. Mater.*, 2021, **31**, 2106912.
- 43 A. F. Suleymanova, M. Z. Shafikov, A. C. Whitwood, R. Czerwieńiec and D. W. Bruce, *J. Mater. Chem. C*, 2021, **9**, 6528–6535.
- 44 Y. Zhu, S. Zeng, B. Li, A. J. McEllin, J. Liao, Z. Fang, C. Xiao, D. W. Bruce, W. Zhu and Y. Wang, *ACS Appl. Mater. Interfaces*, 2022, **14**, 15437–15447.
- 45 D. Zhang, L. Duan, C. Li, Y. Li, H. Li, D. Zhang and Y. Qiu, *Adv. Mater.*, 2014, **26**, 5050–5055.
- 46 A. R. Clapp, I. L. Medintz and H. Mattoussi, *ChemPhysChem*, 2006, **7**, 47–57.
- 47 Y. Wang, J. H. Yun, L. Wang and J. Y. Lee, *Adv. Funct. Mater.*, 2021, **31**, 2008332.
- 48 I. H. Lee, W. Song, J. Y. Lee and S.-H. Hwang, *J. Mater. Chem. C*, 2015, **3**, 8834–8838.
- 49 X. Song, D. Zhang, Y. Lu, C. Yin and L. Duan, *Adv. Mater.*, 2019, **31**, e1901923.

