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Design, *in silico* studies, and synthesis of new 1,8-naphthyridine-3-carboxylic acid analogues and evaluation of their H1R antagonism effects†

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New 1,8-naphthyridine-3-carboxylic acid derivatives were designed, synthesized and evaluated for their *in vivo* antihistaminic activity on guinea pig trachea by using chlorpheniramine as the standard drug. It was found that compound **5a1** displayed a promising bronchorelaxant effect in conscious guinea pigs using the *in vivo* model. A molecular docking study was performed to understand the molecular interaction and binding mode of the compounds in the active site of the H1 receptor. Furthermore, *in silico* computational studies were also performed to predict the binding modes and pharmacokinetic parameters of these derivatives. Prior to the start of experimental lab work, PASS software was used to predict the biological activities of these compounds. An *in silico* PASS, Swiss ADME assisted docking approach was found to be suitable to derive and synthesize effective antihistaminic agents for the present study.

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1 Introduction

Heterocyclic synthesis has become a powerful technique in organic synthesis for generating new molecules for drug discovery and development.^{1–8} Nitrogen atom-containing heterocyclic compounds provide highly functionalized scaffolds on which pharmacophoric features can be arranged to obtain effective and selective drugs.^{9–17} Histamine is one of the most important chemical mediators that influences immune regulation *via* an acute and chronic inflammatory response through 4 different types of G-protein coupled receptors, H1, H2, H3, and H4. It is involved in the pathophysiology of allergic disorders like urticaria, rhino conjunctivitis, and asthma.¹⁸ Although antihistaminic agents belong to several chemical classes, such as ethylene-diamines, aminoethyl ethers, propyl- and propenyl-arnines, phenothiazines, piperidines and piperazines, they show remarkable chemical similarities.¹⁹ Presently, antihistamines are widely prescribed for the management of various allergic symptoms, but these drugs show complex side effect profiles, which include sedation, light-headedness, motor incoordination, cardiovascular effects, diminished alertness, concentration difficulties, fatigue and a tendency to fall asleep.²⁰ A common structural feature for the classical first-generation antihistaminic drugs is an aryl group including phenyl, substituted phenyl and heteroaryl groups (2-pyridyl)

linked to a terminal amino group *via* a two or three carbon chain (chlorpheniramine maleate).²¹ In contrast, most of the non-sedative H1R inhibitors like azelastine, levocetirizine and fexofenadine²² have a minimal muscarinic effect and show a better BRA (benefit-to-risk ratio). These drugs cannot diffuse through the blood–brain barrier and thus have a weaker sedating effect and possess higher receptor selectivity and affinity,²³ and they are marked as ‘non-sedative’ H1R inhibitors. The ‘non-sedative’ second-generation drugs like terfenadine and levocetirizine also have many structural features belonging to old classic antihistamines. The condensed heterocyclic ring system of the second-generation H1-antihistamines (loratadine, fexofenadine, and azelastine) does not have the above-described pharmacophoric groups of H1-antihistamines. This concept and idea have enabled the discovery of many potent antihistaminic drugs like temelastine and mangostin. The currently used antihistaminic compounds are almost completely based on modifying the structure of some old H1 antagonists (Fig. 1).

The availability of the crystal structure of H(1)R, however, can enable new ways to study the binding of histamine and its antagonists and to find the important receptor–ligand interactions. It is also possible to compare the binding affinity of the synthesized analogues to predict their possible therapeutic activity.^{24,25}

Azelastine is a phthalazinone derivative and a H1R antagonist.²⁶ It is an antiallergic agent that inhibits the release of histamine and other mediators involved in the allergic response. It antagonizes histamine and leukotriene-induced bronchospasm in animal studies and reduces airway responsiveness to an inhaled antigen or distilled water and exercise challenges.²⁷ From a comprehensive literature search of 1,8-

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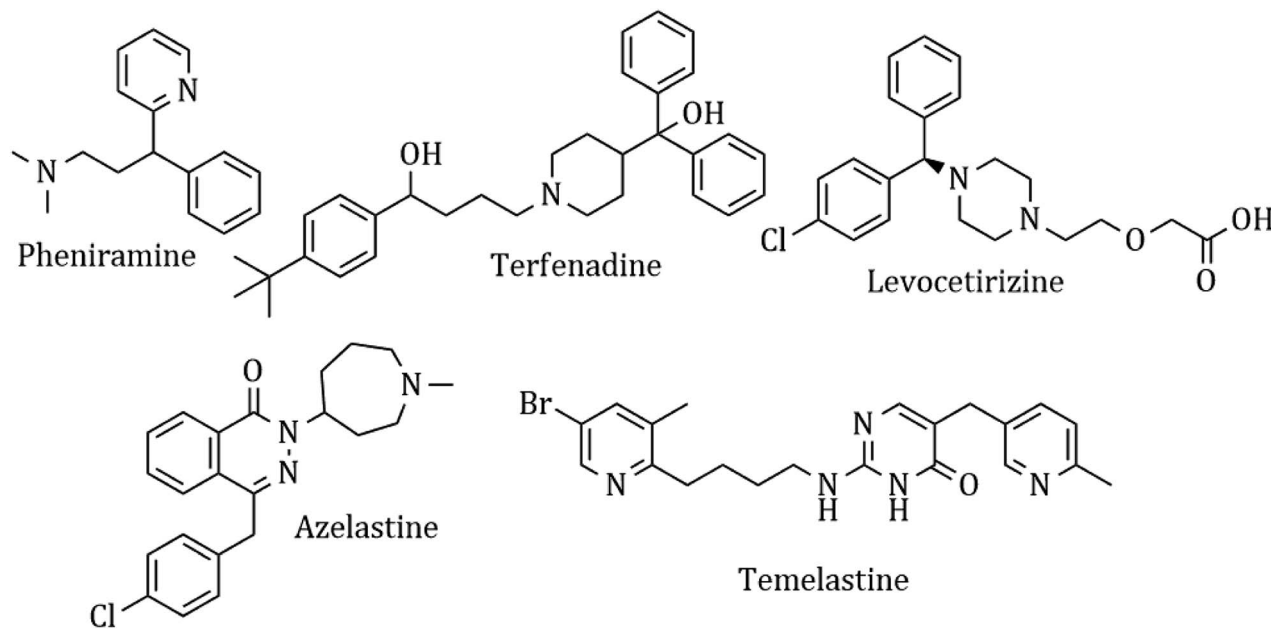


Fig. 1 Classical antihistamines.

naphthyridine, it has been found to be a potent antiallergic and antihistaminic agent.^{28,29} In addition, its derivatives are reported to have a broad spectrum of pharmacological activity. Notable amongst these are A1 and A2A adenosine receptor agonist,³⁰ antibacterial,³¹ analgesic and anti-inflammatory³² agents, cannabinoid receptor ligands³³ and phosphodiesterase (PDE 4) inhibitors.³⁴ A review of the literature also supports that azelastine as an asthma prophylactic has three main pharmacophoric groups: (i) phthalazinone nuclei, (ii) a *p*-chlorobenzyl moiety and (iii) a basic amino group, *N*-methyl azepine. These three parts are essential to be an effective H1R antagonistic compound.

In reference to our continuous interest in establishing potential non-sedating, nonclassical histaminergic (H1) blockers with low toxicity profiles, we have previously published a research paper where the synthesis of some substituted 1,8-naphthyridine derivatives with different cyclic amine substituents of 1,8-naphthyridine-3-carboxylic acid³⁵ has been described. Consequently, in this work, the key compound 1,8-naphthyridine-3-carboxylic acid is selected, which is susceptible to forming amide derivatives. In the present paper, we have reported the design, synthesis and *in vitro* antihistaminic activity of 1,8-naphthyridine-3-carboxylic acid derivatives that are modified at the terminal 3-carboxylic acid end (Fig. 2). In the

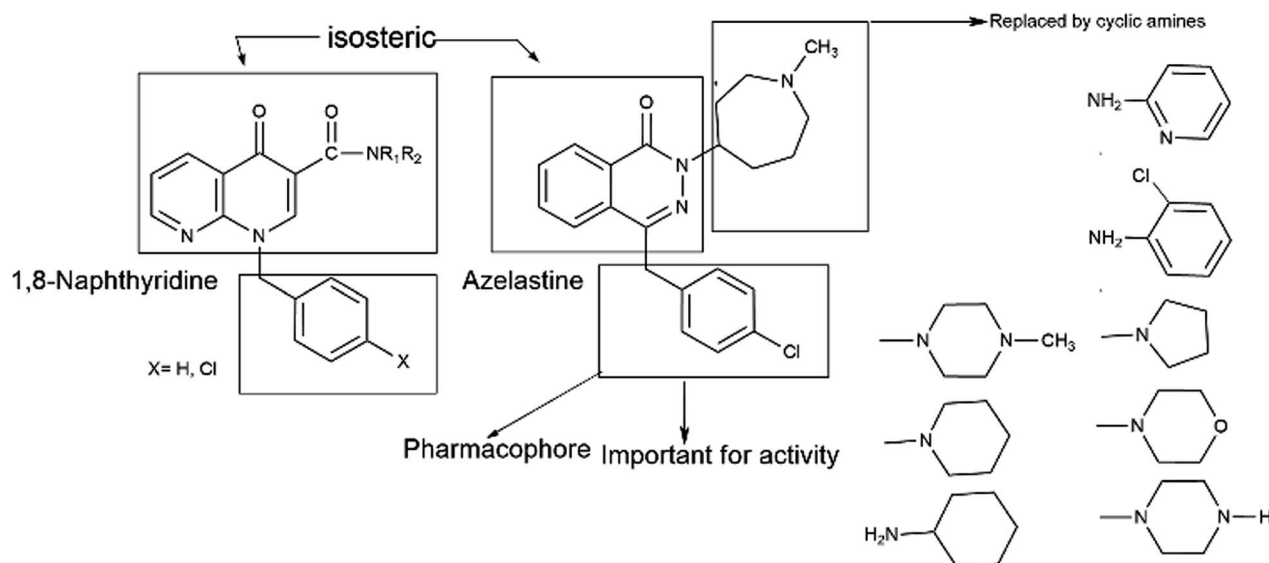
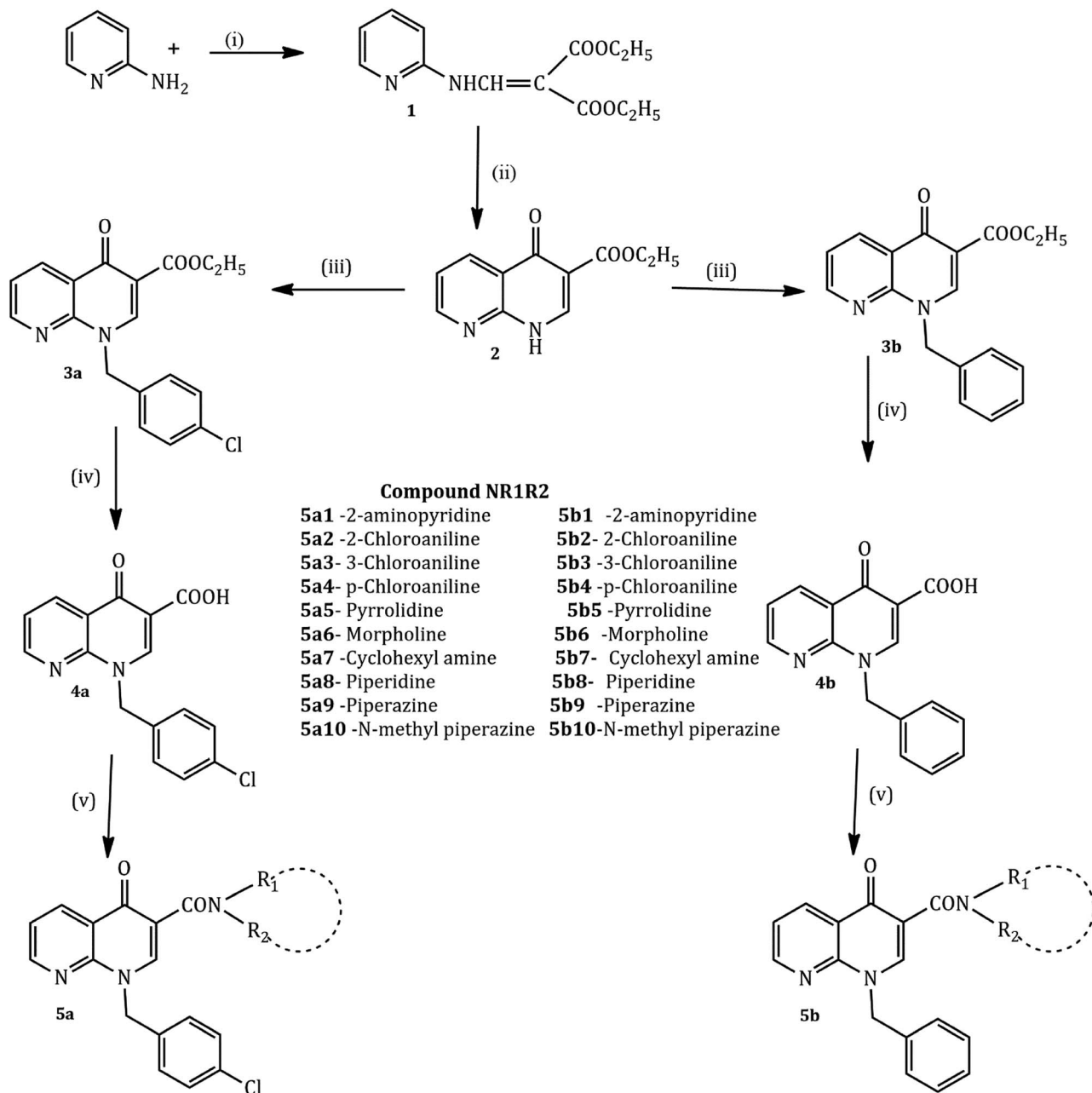


Fig. 2 Design of 1,8-naphthyridine derivatives.





Scheme 1 Preparation of 1,4-dihydro-4-oxo-1,8-naphthyridine-3-carboxylic acid derivatives. Reagents and conditions: (i) diethyl ethoxy methylene malonate, 120 °C, 1 h, (ii) diphenyl ether, 250 °C reflux, 4 h, (iii) *p*-chlorobenzylchloride/benzyl chloride, NaH, DMF, 90 °C reflux, 24 h, (iv) NaOH, EtOH, 100 °C reflux, 2 h and (v) R-NH R-NH₂, heat 25–100 °C in a sealed tube for 24 h.

current drug discovery process, the potential of a novel compound is frequently studied initially through virtual tools. The possibility of a compound to exhibit useful therapeutic activity (sometimes called 'drug-likeness') is predicted from its molecular structure.^{36,37} Prediction of bioavailability and bioavailability-related properties, such as solubility and lipophilicity, is significant before synthesis. This could be the best method to avoid depletion of chemicals, valuable time and above all possible environmental problems. Furthermore, we also performed *in silico* ADMET profile and PASS analyses of all compounds.

2 Materials and methods

2.1 General

All chemicals and reagents were purchased from commercial sources (Himedia, Sigma-Aldrich) having the best analytical-grade and used without further purification unless otherwise stated. Melting points were checked using open capillary methods on a Macro scientific apparatus and were uncorrected. All reactions were routinely checked by thin-layer chromatography (TLC) analysis on silica gel G plates eluted with a mixture of chloroform : methanol (4 : 1) as a solvent system. The IR



spectra were recorded using a Shimadzu FTIR spectrophotometer. ¹HNMR spectra were recorded using a Bruker (Avance III) 400 MHz system spectrometer with tetramethyl silane (TMS) as the internal standard and DMSO as the solvent. Chemical shifts were reported as parts per million (ppm). Signals were represented as follows: s: singlet, d: doublet, m: multiplet, and t: triplet. High-resolution mass spectra (HRMS) were obtained using the Electrospray Ionization (ESI) technique on a Bruker Fourier Transform Ion Cyclotron Resonance Mass Spectrometer.

The compounds were synthesized in a sequence of reactions from the starting material 2-aminopyridine (1). Diethyl ethoxy methylene malonate (2), *p*-chloro benzyl chloride/benzyl chloride (3) and 1,8-naphthyridine-3-carboxylic acid (4) were obtained using reported methods. The synthetic route to the title compound is illustrated in the following Scheme 1.^{35,38}

2-Aminopyridine (1 mmol) and diethyl ethoxy methylene malonate (1 mmol) were heated at 120–130 °C for 2 h producing crude malonate (1) that was purified by recrystallization from light petroleum ether. The crude ester (1, 0.017 mol) and diphenyl ether in excess were heated at 240–250 °C for one hour, cooled at room temperature and washed with petroleum ether. The resulting white powder of malonate (2) was recrystallized from dimethylformamide. Sodium hydride (NaH, 1.81 mmol, 50% in mineral oil) and a solution of malonate (2) (1.5 mmol) in 10 mL of dry DMF were stirred at room temperature. After one hour, *p*-chlorobenzyl chloride (for 3a) and chlorobenzyl chloride (for 3b) (1 mmol each) were added and the mixture was stirred for 24 h. The resultant solution was evaporated and the addition

of ethyl ether caused the precipitation of carboxylate 3a and 3b as pure solids. The 1,8-naphthyridine-3-carboxylic acid ethyl esters 3a and 3b (4.13 mmol) and a mixture of 10% sodium hydroxide (5 mL) and ethyl alcohol (5 mL) were refluxed for 2 h. After cooling, the solution was adjusted to pH 4 with aqueous 10% hydrochloric acid. The resulting precipitate of 4a and 4b was collected. A mixture of the 1,8-naphthyridine acids 4a and 4b (1 mmol) and appropriate amines (2-aminopyridine, 2-chloro aniline, cyclohexane amine, and *n*-phenyl piperazine) (10 mmol each) was heated in a sealed tube at 120 °C for 24 h. After cooling, the reaction mixtures were treated with ethyl ether to give the title compounds as a pure solid.

1-(4-Chlorobenzyl)-4-oxo-*N*-(pyridin-2-yl)-1,4-dihydro-1,8-naphthyridine-3-carboxamide (5a1). White crystalline powder (yield 76.6%); mp > 300 °C; (DMSO-*d*₆) δ ppm: 5.68 (s, 2H, CH₂-Ph), 8.2 (dd, 1H, aromatic, *J* = 4.4, Hz), 9.19 (d, 1H, aromatic), 7.36 (d, 4H, Ph-H, *J* = 8.24 Hz), 7.42 (d, 2H, aromatic, *J* = 8.4 Hz), 7.66 (d, 1H, aromatic, *J* = 8.8 Hz), 7.76 (d, 2H, aromatic), 8.51 (s, 1H, aromatic), IR (KBr, cm⁻¹): 3240 (NH), 3112.7, 3086.0 (C-H aromatic), 1686.4 (C=O keto), 1651.1 (C=O amide), 734 (C-Cl); mass (*m/z*): 390 (M⁺), chemical formula: C₂₁H₁₅ClN₄O₂, MW 390.82, anal calc.: C, 64.54; H, 3.87; N, 14.34; found: C, 64.44; H, 3.78; N, 14.30%.

1-(4-Chlorobenzyl)-*N*-(2-chlorophenyl)-4-oxo-1,4-dihydro-1,8-naphthyridine-3-carboxamide (5a2). White crystalline powder (yield 75%); mp 193–195 °C; (DMSO-*d*₆) δ, ppm: 8.25 (s, 1H, aromatic), 8.52 (dd, 1H, aromatic, *J* = 4.67 Hz), 7.57 (dd, 1H, aromatic, *J* = 7.77 Hz), 7.97 (d, 1H, aromatic, *J* = 7.77 Hz), 7.24 (m, 4H, aromatic), 5.68 (s, 2H, CH₂-Ph), 9.19 (d, 1H, NH), 7.24–

Table 1 Antihistaminic and sedative–hypnotic activity of the compounds (5a1–10 and 5b1–10)^a

Compd	Antihistaminic activity		Percent CNS depression		
	Time of onset of convulsion (s)	% Protection	1 h	2 h	3 h
5a1	371 ± 6.53*	61.45 ± 0.15*	13.02 ± 0.22**	13.60 ± 1.26**	9.81 ± 1.52**
5a2	352 ± 2.46*	59.38 ± 0.19*	9.12 ± 1.02**	9.15 ± 1.02**	6.00 ± 0.58**
5a3	352 ± 1.82*	59.37 ± 0.14*	10.14 ± 0.56**	11.16 ± 0.77**	8.03 ± 0.91**
5a4	341 ± 2.40*	58.06 ± 0.16*	7.11 ± 0.88**	8.16 ± 0.73**	5.01 ± 0.43**
5a5	311 ± 2.43*	54.01 ± 0.14*	8.16 ± 0.73**	8.19 ± 0.76**	6.00 ± 0.58**
5a6	328 ± 2.86*	56.40 ± 0.19*	7.11 ± 0.88**	7.17 ± 0.87**	5.01 ± 0.43**
5a7	305 ± 1.52*	53.11 ± 0.13*	7.11 ± 0.88**	9.81 ± 1.52**	6.00 ± 0.58**
5a8	340 ± 4.29*	57.94 ± 0.36*	6.00 ± 0.58**	7.17 ± 0.87**	5.18 ± 0.4**
5a9	316 ± 4.56*	54.74 ± 0.42*	8.18 ± 0.74**	10.13 ± 0.55**	6.07 ± 0.59**
5a10	311 ± 2.46*	54.01 ± 0.23*	9.61 ± 1.42**	10.11 ± 0.56**	7.11 ± 0.88**
5b1	354 ± 5.33*	59.60 ± 0.352*	8.58 ± 0.09**	11.17 ± 0.10**	5.29 ± 0.11**
5b2	353 ± 3.87*	59.49 ± 0.284*	9.53 ± 0.13**	11.2 ± 0.09**	5.16 ± 0.08**
5b3	366 ± 3.94*	60.92 ± 0.315*	6.43 ± 0.26**	8.10 ± 0.09**	4.67 ± 0.13**
5b4	354 ± 4.23*	59.60 ± 0.265*	8.28 ± 0.10**	9.42 ± 0.17**	5.09 ± 0.10**
5b5	355 ± 6.42*	59.71 ± 1.53*	8.23 ± 1.23**	12.1 ± 1.51**	6.1 ± 1.62**
5b6	304 ± 6.69*	52.96 ± 1.42*	9.23 ± 1.72**	13.1 ± 1.60**	9.1 ± 1.54**
5b7	305 ± 6.38*	53.11 ± 1.28*	11.10 ± 1.52**	14.1 ± 1.62**	7.2 ± 1.93**
5b8	359 ± 5.41*	60.16 ± 1.39*	12.9 ± 1.72**	15.1 ± 1.83**	9.1 ± 1.7**
5b9	301 ± 5.87*	52.49 ± 1.72*	6.21 ± 1.94**	12.1 ± 1.87**	6.1 ± 1.92**
5b10	361 ± 6.72*	60.38 ± 1.40*	8.29 ± 1.93**	10.1 ± 1.04**	8.1 ± 1.87**
Control	143 ± 3.29*	—	6.10 ± 0.49**	4.1 ± 0.59**	4 ± 0.91**
Chlorpheniramine	411 ± 4.43*	65.20 ± 0.33*	38.80 ± 0.32**	34.80 ± 0.72**	29.58 ± 0.72**

^a Each value represents the mean ± SEM (*n* = 6). Significance levels **p* < 0.001, ***p* > 0.05.



7.46 (m, 4H, aromatic), IR (KBr, cm^{-1}): 3112.7, 3086.0 (C–H aromatic), 1686.4 (C=O amide), 1651.1 (C=O ring), 797.70 (C–Cl); mass m/z 423 (M^+), chemical formula: $\text{C}_{22}\text{H}_{15}\text{Cl}_2\text{N}_3\text{O}_2$, MW 423.29, anal. calc.: C, 62.28; H, 3.56; N, 9.90; found: C, 62.21; H, 3.72; N, 9.81%.

1-(4-Chlorobenzyl)-N-(3-chlorophenyl)-4-oxo-1,4-dihydro-1,8-naphthyridine-3-carboxamide (5a3). Off white crystalline powder (yield 67%); mp > 300 °C; (DMSO- d_6) δ ppm: 9.19 (s, 1H, aromatic), 8.53 (dd, 1H, aromatic, $J = 4.67$ Hz), 7.57 (dd, 1H, aromatic, $J = 7.77$ Hz), 7.97 (d, 1H, aromatic, $J = 7.77$ Hz), 7.24 (m, 4H, aromatic), 5.68 (s, 2H, CH₂-Ph), 9.91 (d, 1H, NH), 7.24–7.46 (m, 4H, aromatic), IR (KBr, cm^{-1}): 3112.7, 3086.0 (C–H aromatic), 1686.4 (C=O keto), 1651.1 (C=O amide), 780 (C–Cl); mass (m/z): 423 (M^+), chemical formula: $\text{C}_{22}\text{H}_{15}\text{Cl}_2\text{N}_3\text{O}_2$, MW 424.28, anal. calc.: C, 62.28; H, 3.56; N, 9.90; found: C, 62.21; H, 3.72; N, 9.81%.

1-(4-Chlorobenzyl)-N-(4-chlorophenyl)-4-oxo-1,4-dihydro-1,8-naphthyridine-3-carboxamide (5a4). Off white crystalline powder (yield 66%); mp 193–195 °C; (DMSO- d_6) δ , ppm: 9.19 (d, 1H, NH), 9.29 (s, 1H, H₂), 8.52 (dd, 1H, aromatic, $J = 4.66$ Hz), 7.56 (d, 1H, aromatic, $J = 7.77$ Hz), 7.96 (d, 1H, aromatic, $J = 7.77$ Hz), 8.71 (s, 1H, aromatic), 7.24–7.46 (m, 4H, aromatic), 5.68 (s, 2H, CH₂-Ph), 7.32–7.47 (m, 4H, aromatic), IR (KBr, cm^{-1}): 3112.7, 3086.0 (C–H aromatic), 1686.4 (C=O keto), 1651.1 (C=O amide), 797.7 (C–Cl); mass m/z 423 (M^+), chemical formula: $\text{C}_{22}\text{H}_{15}\text{Cl}_2\text{N}_3\text{O}_2$, MW 424.28, anal. calc.: C, 62.28; H, 3.56; N, 9.90; found: C, 62.28; H, 16.71; N, 9.90%.

1-(4-Chlorobenzyl)-3-(pyrrolidine-1-carbonyl)-1,8-naphthyridin-4(1H)-one (5a5). Off white crystalline powder (yield 56%); mp 181–183 °C; (DMSO- d_6) δ , ppm: 9.19 (s, 1H, aromatic), 8.52 (dd, 1H, aromatic, $J = 4.7$ Hz), 7.56 (d, 1H, aromatic, $J = 7.8$ Hz), 7.97 (d, 1H, aromatic, $J = 7.8$ Hz), 7.24–7.48 (m, 4H, aromatic), 5.68 (s, 2H, CH₂-Ph), 8.71 (s, 1H, aromatic), 3.37 (t, 4H, pyrrolidine, $J = 13.2$ Hz), 1.67 (t, 4H, pyrrolidine, $J = 15.2$ Hz), IR (KBr, cm^{-1}): 3069.1, 2998.7 (C–H), 1714.0 (C=O keto), 1692.1 (C=O amide), 781 (C–Cl); mass (m/z): 367 (M^+), chemical formula: $\text{C}_{20}\text{H}_{18}\text{ClN}_3\text{O}_2$, MW 367.84, anal. calc.: C, 65.31; H, 4.93; N, 11.42; found: C, 65.22; H, 4.84; N, 11.41%.

1-(4-Chlorobenzyl)-3-(morpholine-4-carbonyl)-1,8-naphthyridin-4(1H)-one (5a6). Off white crystalline powder (yield 76.6%); mp 147–149 °C; (DMSO- d_6) δ ppm: 9.19 (s, 1H, aromatic), 8.52 (dd, 1H, aromatic, $J = 4.7$ Hz), 7.56 (d, 1H, aromatic, $J = 7.8$ Hz), 7.97 (d, 1H, aromatic, $J = 7.8$ Hz), 7.24–7.46 (m, 4H, aromatic), 5.68 (s, 2H, CH₂-Ph), 8.71 (s, 1H, aromatic), 1.91 (d, 1H, NH), 2.1–2.3 (m, 4H, morpholine, $J = 15.2$ Hz), 2.6–2.8 (m, 4H, morpholine, $J = 15.2$ Hz), IR (KBr, cm^{-1}): 3069.1, 2998.7 (C–H), 1714.0 (C=O keto), 1692.1 (C=O amide), 781.00 (C–Cl); mass (m/z): 383 (M^+); chemical formula: $\text{C}_{20}\text{H}_{18}\text{ClN}_3\text{O}_3$, MW 383.84, anal. calc.: C, 62.58; H, 4.73; N, 10.95; found: C, 62.51; H, 4.70; N, 10.86%.

1-(4-Chlorobenzyl)-N-cyclohexyl-4-oxo-1,4-dihydro-1,8-naphthyridine-3-carboxamide (5a7). Off white crystalline powder (yield 65.1%); mp 182–184 °C; (DMSO- d_6) δ , ppm: 9.29 (s, 1H, aromatic), 8.52 (dd, 1H, aromatic, $J = 4.7$ Hz), 7.56 (d, 1H, aromatic, $J = 7.8$ Hz), 7.97 (d, 1H, aromatic, $J = 7.8$ Hz), 7.24–7.46 (m, 4H, aromatic), 5.68 (s, 2H, CH₂-Ph), 8.71 (s, 1H, aromatic), 3.85 (m, 1H, CH), 2.17–1.59 (m, 10H, cyclohexyl, $J =$

13.2 Hz), IR (KBr, cm^{-1}): 3069.1, 2998.7 (C–H), 1714.0 (C=O keto), 1692.1 (C=O amide), 781.0 (C–Cl); mass (m/z): 395 (M^+), chemical formula: $\text{C}_{22}\text{H}_{22}\text{ClN}_3\text{O}_2$, MW 395.87, anal. calc.: C, 66.75; H, 5.60; N, 10.61; found: C, 66.62; H, 5.42; N, 10.52%.

1-(4-Chlorobenzyl)-3-(piperidine-1-carbonyl)-1,8-naphthyridin-4(1H)-one (5a8). Pale yellow crystalline powder (yield 76.7%); mp > 300 °C; (DMSO- d_6) δ , ppm: 9.29 (s, 1H, aromatic), 8.52 (dd, 1H, aromatic, $J = 4.7$ Hz), 7.56 (d, 1H, aromatic, $J = 7.8$ Hz), 7.97 (d, 1H, aromatic, $J = 7.8$ Hz), 7.24–7.46 (m, 4H, aromatic), 5.68 (s, 2H, CH₂-Ph), 8.71 (s, 1H, aromatic), 8.71 (s, 1H, aromatic), 1.5–1.8 (m, 6H, piperidine), 1.9–2.2 (t, 4H, piperidine), IR (KBr, cm^{-1}): 3112.7, 3086.0 (C–H aromatic), 1686.4 (C=O keto), 1651.1 (C=O), 780.10 (C–Cl); mass (m/z): 381 (M^+), chemical formula: $\text{C}_{21}\text{H}_{20}\text{ClN}_3\text{O}_2$, MW 381.87, anal. calc.: C, 66.05; H, 5.28; N, 11.00; found: C, 66.01; H, 5.22; N, 10.87%.

1-(4-Chlorobenzyl)-3-(piperazine-1-carbonyl)-1,8-naphthyridin-4(1H)-one (5a9). Pale yellow crystalline powder (yield 76.1%); mp 193–195 °C; (DMSO- d_6) δ , ppm: 9.29 (s, 1H, aromatic), 8.52 (dd, 1H, aromatic, $J = 4.7$ Hz), 7.56 (d, 1H, aromatic, $J = 7.8$ Hz), 7.97 (d, 1H, aromatic, $J = 7.8$ Hz), 7.24–7.46 (m, 4H, aromatic), 5.68 (s, 2H, CH₂-Ph), 8.71 (s, 1H, aromatic), 7.22–7.34 (m, 4H, aromatic), 1.5–1.7 (m, 4H, piperazine), 1.9–2.1 (m, 4H, piperazine), IR (KBr, cm^{-1}): 3112.7, 3086.0 (C–H aromatic), 1686.4 (C=O amide), 1651.1 (C=O ring), 797.8 (C–Cl); mass (m/z): 382 (M^+), chemical formula: $\text{C}_{20}\text{H}_{19}\text{ClN}_4\text{O}_2$, MW 382.85, anal. calc.: C, 62.74; H, 5.00; N, 14.63; found: C, 62.65; H, 5.01; N, 14.54%.

1-(4-Chlorobenzyl)-3-(4-methylpiperazine-1-carbonyl)-1,8-naphthyridin-4(1H)-one (5a10). Pale yellow crystalline powder (yield 55.10%); mp 181–183 °C; (DMSO- d_6) δ , ppm: 9.29 (s, 1H, aromatic), 8.52 (dd, 1H, aromatic, $J = 4.7$ Hz), 7.56 (d, 1H, aromatic, $J = 7.8$ Hz), 7.97 (d, 1H, aromatic, $J = 7.8$ Hz), 7.24–7.46 (m, 4H, aromatic), 5.68 (s, 2H, CH₂-Ph), 8.71 (s, 1H, aromatic), 1.5–1.7 (m, 4H, piperazine), 1.9–2.1 (m, 4H, piperazine), 2.88 (s, 3H, CH₃), IR (KBr, cm^{-1}): 3069.1, 2998.7 (C–H), 1714.0 (C=O keto), 1692.1 (C=O), 781.0 (C–Cl); mass (m/z): 396 (M^+), chemical formula: $\text{C}_{21}\text{H}_{21}\text{ClN}_4\text{O}_2$, MW 396.87, anal. calc.: C, 63.55; H, 5.33; N, 14.12; found: C, 63.48; H, 5.24; N, 14.09%.

1-Benzyl-4-oxo-N-(pyridin-2-yl)-1,4-dihydro-1,8-naphthyridine-3-carboxamide (5b1). Pale yellow crystalline powder (yield 76%); mp > 300 °C; (DMSO- d_6) δ , ppm: 8.51 (dd, 1H, aromatic, $J = 4.8$ Hz), 8.24 (dd, 1H, aromatic, $J = 4.7$ Hz), 5.68 (s, 2H, CH₂-Ph), 9.19 (dd, 1H, aromatic, $J = 7.8$ Hz), 7.36 (d, 4H, Ph-H, $J = 8.24$ Hz), 7.31–6.96 (m, 5H, aromatic), 7.40 (m, 2H, aromatic), 7.77 (dd, 1H aromatic, $J = 4.7$ Hz), 6.59–8.08 (m, 4H, pyridine), IR (KBr, cm^{-1}): 3112.7, 3086.0 (C–H aromatic), 1686.4 (C=O keto), 1651.1 (C=O ring), 780.10 (C–Cl); mass (m/z): 356 (M^+), chemical formula: $\text{C}_{21}\text{H}_{16}\text{N}_4\text{O}_2$, MW 356.38, anal. calc.: C, 70.77; H, 4.53; N, 15.72; found: C, 70.70; H, 4.46; N, 15.65%.

1-Benzyl-N-(2-chlorophenyl)-4-oxo-1,4-dihydro-1,8-naphthyridine-3-carboxamide (5b2). Pale yellow powder (yield 75%); mp 193–195 °C; (DMSO- d_6) δ , ppm: 8.25 (s, 1H, aromatic), 8.52 (dd, 1H, aromatic, $J = 4.67$ Hz), 7.57 (dd, 1H, aromatic, $J = 7.77$ Hz), 7.97 (d, 1H, aromatic, $J = 7.77$ Hz), 7.24–7.33 (m, 5H, aromatic), 5.68 (s, 2H, CH₂-Ph), 9.19 (d, 1H, NH), 7.14–8.06 (m, 5H, aromatic), IR (KBr, cm^{-1}): 3112.7, 3086.0 (C–H aromatic),



1686.4 (C=O amide), 1651.1 (C=O amide), 797.8 (C-Cl); mass (m/z): 389 (M^+), chemical formula: $C_{22}H_{16}ClN_3O_2$, MW 389.84, anal. calc.: C, 67.78; H, 4.14; N, 10.78; found: C, 67.71; H, 4.05; N, 10.70%.

1-Benzyl-*N*-(3-chlorophenyl)-4-oxo-1,4-dihydro-1,8-naphthyridine-3-carboxamide (5b3). Pale yellow crystalline powder (yield 76.2%); mp > 300 °C; (DMSO- d_6) δ , ppm: 8.25 (s, 1H, aromatic), 8.52 (dd, 1H, aromatic, $J = 4.67$ Hz), 7.57 (dd, 1H, aromatic, $J = 7.77$ Hz), 7.97 (d, 1H, aromatic, $J = 7.77$ Hz), 7.24–7.33 (m, 5H, aromatic), 5.68 (s, 2H, CH₂-Ph), 9.19 (d, 1H, NH), 7.14–7.99 (m, 4H, aromatic), IR (KBr cm^{-1}): 3112.7, 3086.0 (C–H aromatic), 1686.4 (C=O keto), 1651.1 (C=O amide), 780 (C–Cl); mass (m/z): 389 (M^+), chemical formula: $C_{22}H_{16}ClN_3O_2$, MW 389.83, anal. calc.: C, 67.78; H, 4.14; N, 10.78; found: C, 67.71; H, 4.05; N, 10.70%.

1-Benzyl-*N*-(4-chlorophenyl)-4-oxo-1,4-dihydro-1,8-naphthyridine-3-carboxamide (5b4). Pale yellow crystalline powder (yield 71%); mp 193–195 °C; (DMSO- d_6) δ , ppm: 8.25 (s, 1H, aromatic), 8.52 (dd, 1H, aromatic, $J = 4.67$ Hz), 7.57 (dd, 1H, aromatic, $J = 7.77$ Hz), 7.97 (d, 1H, aromatic, $J = 7.77$ Hz), 7.24–7.33 (m, 5H, aromatic), 5.68 (s, 2H, CH₂-Ph), 9.19 (d, 1H, NH), 7.14–7.99 (m, 5H, aromatic), IR (KBr, cm^{-1}): 3112.7, 3086.0 (C–H aromatic), 1686.4 (C=O keto), 1651.1 (C=O amide), 797.71 (C–Cl); mass (m/z): 389 (M^+), chemical formula: $C_{22}H_{16}ClN_3O_2$, MW 389.83, anal. calc.: C, 67.78; H, 4.14; N, 10.78; found: C, 67.71; H, 4.05; N, 10.70%.

1-Benzyl-3-(pyrrolidine-1-carbonyl)-1,8-naphthyridin-4(1*H*)-one (5b5). Pale yellow crystalline powder (yield 59%); mp 181–183 °C; (DMSO- d_6) δ , ppm: 9.19 (s, 1H, aromatic), 8.52 (dd, 1H, aromatic, $J = 4.7$ Hz), 7.56 (d, 1H, aromatic, $J = 7.8$ Hz), 7.97 (d, 1H, aromatic, $J = 7.8$ Hz), 7.24–7.48 (m, 5H, aromatic), 5.68 (s, 2H, CH₂-Ph), 8.71 (s, 1H, aromatic), 3.37 (t, 4H, pyrrolidine, $J = 13.2$ Hz), 1.67 (t, 4H, pyrrolidine, $J = 15.2$ Hz), IR (KBr, cm^{-1}): 3069.1, 2998.7 (C–H), 1714.0 (C=O keto), 1692.1 (C=O) cm^{-1} . Mass (m/z): 333 (M^+), chemical formula: $C_{20}H_{19}N_3O_2$, MW 333.39, anal. calc.: C, 72.05; H, 5.74; N, 12.60; found: C, 72.06; H, 5.75; N, 12.61%.

1-Benzyl-3-(morpholine-4-carbonyl)-1,8-naphthyridin-4(1*H*)-one (5b6). Yellow crystalline powder (yield 76.2%); mp 147–149 °C; (DMSO- d_6) δ , ppm: 9.19 (s, 1H, aromatic), 8.52 (dd, 1H, aromatic, $J = 4.7$ Hz), 7.56 (d, 1H, aromatic, $J = 7.8$ Hz), 7.97 (d, 1H, aromatic, $J = 7.8$ Hz), 7.24–7.48 (m, 5H, aromatic), 5.68 (s, 2H, CH₂-Ph), 8.71 (s, 1H, aromatic), 1.91 (d, 1H, NH), 2.1–2.3 (m, 5H, morpholine, $J = 15.2$ Hz), 2.6–2.8 (m, 4H, morpholine, $J = 15.2$ Hz), IR (KBr, cm^{-1}): 3069.1, 2998.7 (C–H), 1714.0 (C=O keto), 1692.1 (C=O); mass (m/z): 349 (M^+); chemical formula: $C_{20}H_{19}N_3O_3$, MW 349.37, anal. calc.: C, 68.75; H, 5.48; N, 12.03; found: C, 68.62; H, 5.41; N, 12.00%.

1-Benzyl-*N*-cyclohexyl-4-oxo-1,4-dihydro-1,8-naphthyridine-3-carboxamide (5b7). Yellow crystalline powder (yield 60%); mp 181–183 °C; (DMSO- d_6) δ , ppm: 9.19 (s, 1H, aromatic), 8.52 (dd, 1H, aromatic, $J = 4.7$ Hz), 7.56 (d, 1H, aromatic, $J = 7.8$ Hz), 7.97 (d, 1H, aromatic, $J = 7.8$ Hz), 7.24–7.46 (m, 5H, aromatic), 5.68 (s, 2H, CH₂-Ph), 8.71 (s, 1H, aromatic), 3.85 (m, 1H, CH), 2.17–1.59 (m, 10H, cyclohexyl, $J = 13.2$ Hz), IR (KBr, cm^{-1}): 3069.1, 2998.7 (C–H), 1714.0 (C=O keto), 1692.1 (C=O amide), 781 (C–Cl). Mass (m/z): 361 (M^+), chemical formula: $C_{22}H_{23}N_3O_2$, MW

361.44, anal. calc.: C, 73.11; H, 6.41; N, 11.63; found: C, 73.03; H, 6.32; N, 11.53%.

1-Benzyl-3-(piperidine-1-carbonyl)-1,8-naphthyridin-4(1*H*)-one (5b8). Yellow crystalline powder yield (%): 76.6; mp >300 °C; (DMSO- d_6) δ , ppm: 9.19 (s, 1H, aromatic), 8.52 (dd, 1H, aromatic, $J = 4.7$ Hz), 7.56 (d, 1H, aromatic, $J = 7.8$ Hz), 7.97 (d, 1H, aromatic, $J = 7.8$ Hz), 7.24–7.46 (m, 5H, aromatic), 5.68 (s, 2H, CH₂-Ph), 8.71 (s, 1H, aromatic), 8.71 (s, 1H, aromatic), 1.5–1.8 (m, 6H, piperidine), 1.9–2.2 (t, 4H, piperidine), IR (KBr, cm^{-1}): 3112.7, 3086.0 (C–H aromatic), 1686.4 (C=O keto), 1651.1 (C=O); mass (m/z): 347 (M^+), chemical formula: $C_{21}H_{21}N_3O_2$, MW 347.41, anal. calc.: C, 72.60; H, 6.09; N, 12.10; found: C, 72.54; H, 6.01; N, 12.04%.

1-Benzyl-3-(piperazine-1-carbonyl)-1,8-naphthyridin-4(1*H*)-one (5b9). Yellow crystalline powder yield: 75%; mp 193–195 °C; (DMSO- d_6) δ , ppm: 9.19 (s, 1H, aromatic), 8.52 (dd, 1H, aromatic, $J = 4.7$ Hz), 7.56 (d, 1H, aromatic, $J = 7.8$ Hz), 7.97 (d, 1H, aromatic, $J = 7.8$ Hz), 7.24–7.46 (m, 4H, aromatic), 5.68 (s, 2H, CH₂-Ph), 8.71 (s, 1H, aromatic), 7.22–7.34 (m, 5H, aromatic), 1.5–1.7 (m, 4H, piperazine), 1.9–2.1 (m, 4H, piperazine), IR (KBr, cm^{-1}): 3112.7, 3086.0 (C–H aromatic), 1686.4 (C=O keto), 1651.2 (C=O); mass (m/z): 348 (M^+), chemical formula: $C_{20}H_{20}N_4O_2$, MW 348.41, anal. calc.: C, 68.95; H, 5.79; N, 16.08; found: C, 68.86; H, 5.69; N, 16.02%.

1-Benzyl-3-(4-methylpiperazine-1-carbonyl)-1,8-naphthyridin-4(1*H*)-one (5b10). Off white crystalline powder (yield 69); mp 181–183 °C; (DMSO- d_6) δ , ppm: 9.19 (s, 1H, aromatic), 8.52 (dd, 1H, aromatic, $J = 4.7$ Hz), 7.56 (d, 1H, aromatic, $J = 7.8$ Hz), 7.97 (d, 1H, aromatic, $J = 7.8$ Hz), 7.24–7.46 (m, 5H, aromatic), 5.68 (s, 2H, CH₂-Ph), 8.71 (s, 1H, aromatic), 1.5–1.7 (m, 4H, piperazine), 1.9–2.1 (m, 4H, piperazine), 2.88 (s, 3H, CH₃), IR (KBr, cm^{-1}): 3069.1, 2998.7 (C–H), 1714.0 (C=O keto), 1692.1 (C=O); mass (m/z): 362 (M^+), chemical formula: $C_{21}H_{22}N_4O_2$, MW 362.43, anal. calc.: C, 69.59; H, 6.12; N, 15.46; found: C, 69.51; H, 6.03; N, 15.37%.

2.2 Antihistaminic activity study

The antihistaminic screening of the synthesized compounds was carried out through the process of protection from histamine-induced constriction on guinea pig trachea. The animals were maintained in a fully controlled animal facility at 23 ± 2 °C, a relative humidity (RH) of 45–55%, and under a 12 h light–dark cycle; and they had sufficient standard nutritional animal feed and purified water. All the animals were acclimatized to their environment for one week before carrying out the experiments. The Institutional Animal Ethics Committee approved the protocol adopted for the experimentation on animals. All animal procedures were performed in accordance with the Guidelines for Care and Use of Laboratory Animals of Pinnacle Biomedical Research Institute (PBRI), Bhopal, India and approved by the Animal Ethics Committee of the Institute (Approval reference number PBRI/IAEC/PN-17082). In all experimental procedures, efforts were made to minimize pain and suffering.

A modified Van Arman technique was adopted to determine the *in vivo* antihistaminic potential of the synthesized



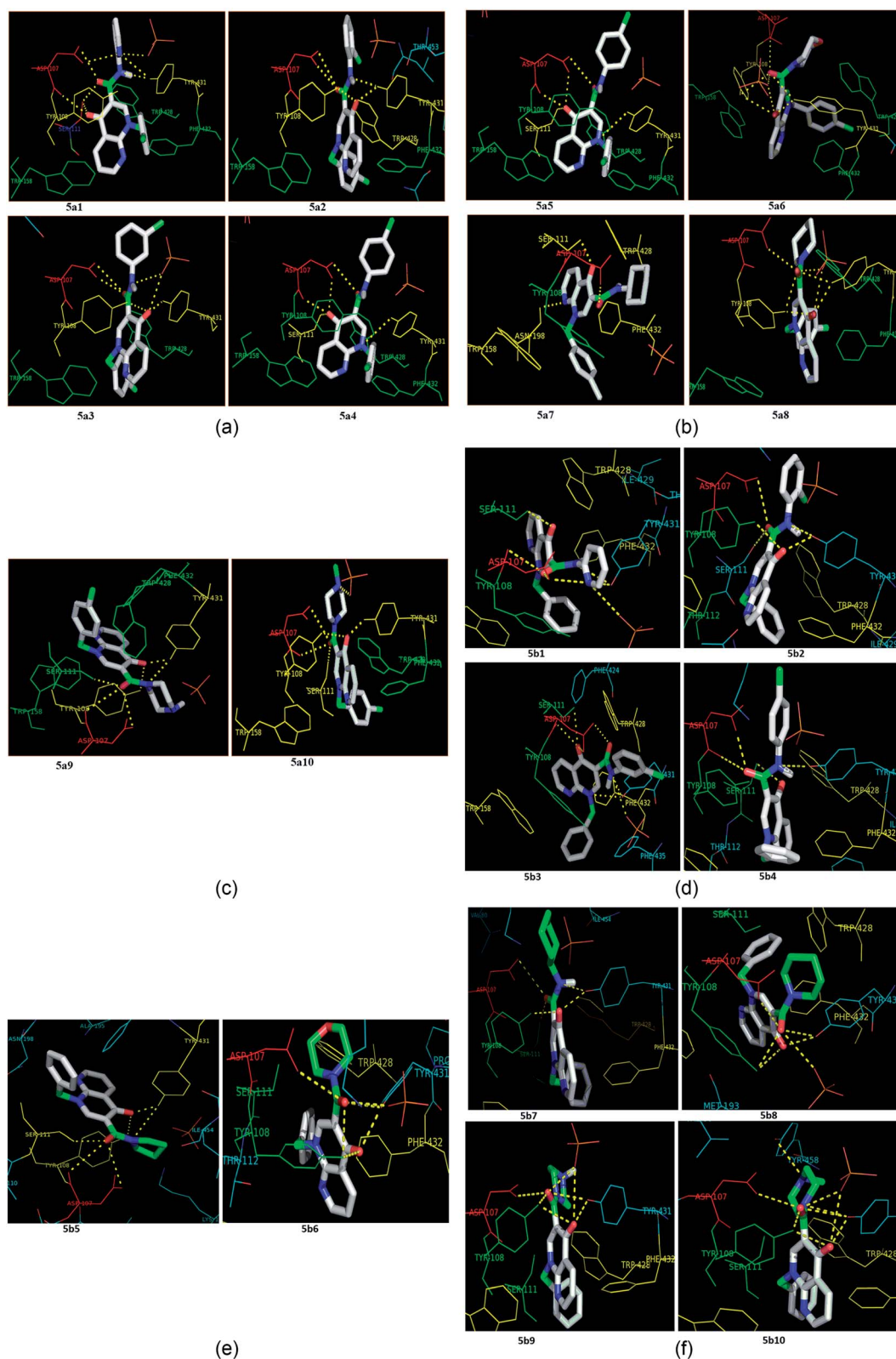


Fig. 4 (a) The docking poses of **5a1**–**5a4** in the binding site of the human histamine H1 receptor in the complex (PDB ID: 3RZE). (b) The docking poses of **5a5**–**5a8** in the binding site of the human histamine H1 receptor in the complex (PDB ID: 3RZE). (c) The docking poses of **5a9** and **5a10** in the binding site of the human histamine H1 receptor in the complex (PDB ID: 3RZE). (d) The docking poses of **5b1**–**5b4** in the binding site of the human histamine H1 receptor in the complex (PDB ID: 3RZE). (e) The docking poses of **5b5** and **5b6** in the binding site of the human histamine H1 receptor in the complex (PDB ID: 3RZE). (f) The docking poses of **5b7**–**5b10** in the binding site of the human histamine H1 receptor in the complex (PDB ID: 3RZE).



Table 2 Docking interaction of active compounds (5a1–a10 and 5b1–b10) with human histamine H1 receptor in complex

Comd	No. of H bonds	Docking score kcal mol ⁻¹	H-bond with amino acid
5a1	7	-7.9	A:Asp-107, Ser111, Tyr431
5a2	5	-9.5	A:Asp-107, Tyr431, Tyr108
5a3	5	-7.1	A:Asp-107, Tyr431, Tyr108
5a4	5	-7.5	A:Asp-107, Tyr431, Ser111
5a5	8	-8.2	A:Asp-107, Tyr431, Tyr108
5a6	7	-7.2	A:Asp-107, Tyr431, Tyr108
5a7	3	-7.4	A:Asp-107, Ser111, Asb198
5a8	7	-7.4	A:Asp-107, Tyr431, Tyr108
5a9	5	-7.6	A:Asp-107, Tyr431, Tyr108, Ser111
5a10	6	-7.1	A:Asp-107, Tyr431, Tyr108, Ser111
5b1	7	-8.0	A:Asp-107, Tyr431, Ser111
5b2	5	-9.8	A:Asp-107, Tyr431, Tyr108, Ser111
5b3	7	-7.4	A:Asp-107, Tyr431, Ser111
5b4	4	-7.7	A:Asp-107, Tyr431, Ser111
5b5	6	-8.4	A:Asp-107, Tyr431, Tyr108, Ser111
5b6	6	-7.2	A:Asp-107, Tyr431, Tyr108, Ser111
5b7	5	-8.2	A:Asp-107, Tyr431, Tyr108
5b8	6	-8.6	A:Asp-107, Tyr431, Tyr108
5b9	7	-7.4	A:Asp-107, Tyr431, Tyr108
5b10	8	-8.2	A:Asp-107, Tyr431, Tyr108, Tyr458
Azelastine	3	-10.6	A:Asp-107, Tyr431, Ser111

reference ligands were initially docked into the crystal structure of the enzymes.

2.6 ADME analysis

The success of a drug candidate is determined not only by its good potential but also by a satisfactory ADME profile. Because a wide variety of experimental media and high throughput *in vitro* ADME screens are available, it has the capacity to predict some important properties *in silico* and is valuable for analysis of the good qualities of the molecules. Nowadays, it is recommended that the employment of computational ADME, in combination with *in vivo* and *in vitro* predictions, should be done as early as possible in the drug discovery process so as to reduce the number of safety issues.⁴⁹ Lipinski's rule of five is helpful in describing molecular properties of a drug candidate, which is necessary to evaluate the important pharmacokinetic parameters such as ADME. The rule is helpful in drug design and development of a potential drug molecule.^{50,51} The ADME study was carried out using the SWISS ADME predictor for the present investigation. This is a free web tool to evaluate the ADMET properties, like aqueous solubility (log *S*), skin permeability (log *K_p*), synthetic accessibility score (SA), percentage absorption, pharmacokinetics, drug-likeness, and medicinal chemistry friendliness properties of small molecules.⁵² The criteria of molecular weight ≤500, ≤5 hydrogen bond donors (HBDs), 10 hydrogen bond acceptors (HBAs), and ≤10 rotatable bonds (RBs) were selected for the present study.⁵³

The search engine further gave a compiled result on lipophilicity and hydrophilicity of these molecules by integrating

Table 3 Physicochemical descriptors and ADME parameters^a

Compd	MW	R-bonds	H-A	H-D	TPSA	MR	<i>W</i> log <i>P</i> (lipophilicity)	ESOL log <i>S</i>	GI absorption	BBB permeant	log <i>K_p</i> (cm s ⁻¹)	Lipinski violations	PAINS alerts	Synthetic accessibility
5a1	390.8	5	4	1	76.88	108.77	3.55	-4.65	High	Yes	-6.28	0	0	2.67
5a2	424.3	5	3	1	63.99	115.99	4.81	-5.69	High	Yes	-5.52	0	0	2.59
5a3	424.3	5	3	1	63.99	115.99	4.81	-5.69	High	Yes	-5.52	0	0	2.58
5a4	424.3	5	3	1	63.99	115.99	4.81	-5.69	High	Yes	-5.52	0	0	2.56
5a5	367.8	4	3	0	55.2	106.07	2.95	-4.35	High	Yes	-6.24	0	0	2.46
5a6	383.8	4	4	0	64.43	107.16	2.19	-3.89	High	Yes	-6.95	0	0	2.58
5a7	395.9	5	3	1	63.99	111.68	4.16	-5.15	High	Yes	-5.61	0	0	2.69
5a8	381.9	4	3	0	55.2	110.88	3.34	-4.65	High	Yes	-6.07	0	0	2.57
5a9	382.8	4	4	1	67.23	112.79	1.38	-3.71	High	Yes	-7.14	0	0	2.61
5a10	396.9	4	4	0	58.44	117.69	1.72	-4.07	High	Yes	-6.9	0	0	2.72
5b1	356.4	5	4	1	76.88	103.76	2.9	-4.06	High	Yes	-6.51	0	0	2.65
5b2	389.8	5	3	1	63.99	110.98	4.16	-5.11	High	Yes	-5.75	0	0	2.56
5b3	389.8	5	3	1	63.99	110.98	4.16	-5.11	High	Yes	-5.75	0	0	2.55
5b4	389.8	5	3	1	63.99	110.98	4.16	-5.11	High	Yes	-5.75	0	0	2.53
5b5	333.4	4	3	0	55.2	101.06	2.3	-3.76	High	Yes	-6.48	0	0	2.41
5b6	349.4	4	4	0	64.43	102.15	1.54	-3.3	High	Yes	-7.19	0	0	2.54
5b7	361.4	5	3	1	63.99	106.67	3.51	-4.55	High	Yes	-5.84	0	0	2.65
5b8	347.4	4	3	0	55.2	105.87	2.69	-4.06	High	Yes	-6.31	0	0	2.52
5b9	348.4	4	4	1	67.23	107.78	0.73	-3.12	High	No	-7.38	0	0	2.57
5b10	362.4	4	4	0	58.44	112.68	1.07	-3.48	High	Yes	-7.13	0	0	2.68
CLPH	274.8	5	2	0	16.13	108.77	3.82	-3.82	High	Yes	-5.57	0	0	2.7
Azelastine	381.9	3	3	0	38.13	115.99	3.92	-5.20	High	Yes	-5.53	1	0	3.62

^a R bond = rotatable bond, H-A = hydrogen bond acceptor, H-D = hydrogen bond donor, TPSA = topological polar surface area, log *P* = lipophilicity, log *S* = water solubility, log *K_p* = permeability coefficient, PAINS = pan-assay interference structure, CLPH = chlorpheniramine, AZT = azelastine.



results obtained from various $\log P$ and S prediction programs called ILOGP, XLOGP3, WLOGP, ESOL, and SILICOS-IT. $\log P$, a measure of lipophilicity of a molecule, is the logarithm of the ratio of the concentration of a drug substance in two solvents in a unionized form. The lower the $\log P$ value, the stronger the lipophilicity of the chemical substance. The aqueous solubility of a compound significantly affects its absorption and distribution characteristics.⁵⁴ On the other hand, low water solubility often leads to bad absorption, and, therefore, the general aim is to avoid poorly soluble compounds. $\log S$ is a unit expressing solubility, and it is the 10-based logarithm of the solubility measured in mol L^{-1} . The distribution of $\log S$ in traded drugs reveals a value somewhere between -1 and -4 and will be optimized for better absorption and distribution of drugs in the body.

2.7 Prediction of activity spectra for substances (PASS)

We used PASS (Prediction of Activity Spectra for Substances, available at <http://www.pharmaexpert.ru/PASSonline/predict.php>) for computational screening of possible

biological effects⁵⁵ like antihistaminic and other related activities, *i.e.*, antiallergic, anti-asthmatic, histamine release inhibition, rhinitis treatment, IgE antagonism, bronchodilation, immunomodulation, IL antagonism, PDE inhibition, 5HT₃ antagonism, anti-inflammatory activities, *etc.* This software program designed a tool for evaluating the general biological potential of an organic drug-like candidate. PASS provides simultaneous predictions of a wide range of biological activity based on the structure of organic compounds. Thus, PASS can be used to estimate the biological activity profiles for virtual molecules, prior to their chemical synthesis and biological testing. It predicts the required pharmacological effect, as well as molecular mechanisms of action and generation of unnecessary side effects such as mutagenicity, teratogenicity, carcinogenicity, and embryotoxicity. This tool provided quantitative structure–activity relationships based on the decomposition of chemical structures using 2D and/or 3D descriptors, followed by the generation of models obtained from bioactive ligands.⁵⁶ The activity was estimated in terms of Pa (probable activity) and Pi

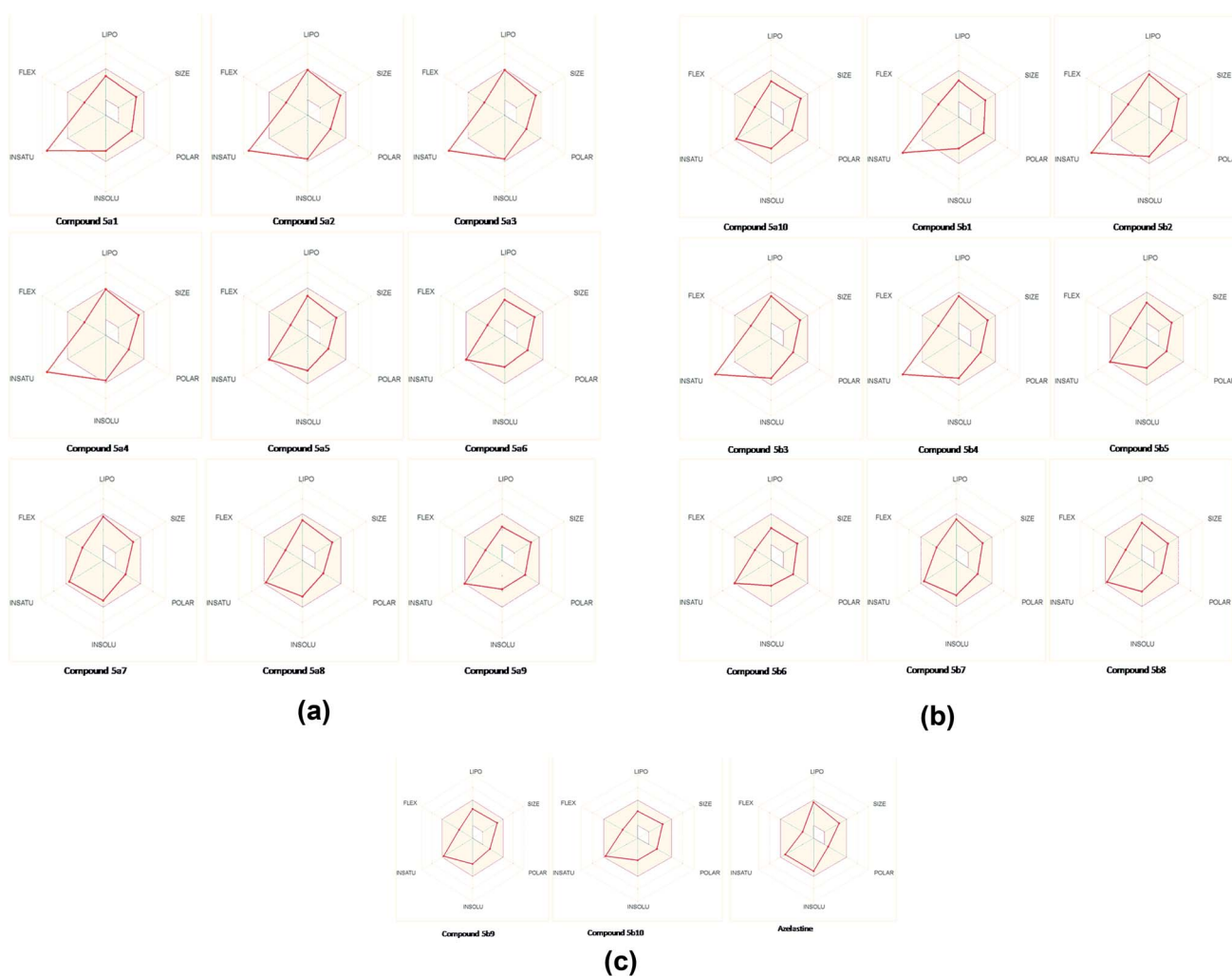


Fig. 5 (a) Bioavailability radar graph of 1,8-naphthyridine derivatives **5a1–a9** (pink area reflects the allowed values of drug likeness properties of the molecule). (b) Bioavailability radar graph of 1,8-naphthyridine derivatives **5a10–5b8** (pink area reflects the allowed values of drug-likeness properties of the molecule). (c) Bioavailability radar graph of 1,8-naphthyridine derivatives **5b9–5b10** (pink area reflects the allowed values of drug-likeness properties of the molecule).



(probable inactivity). Structures with Pa greater than Pi were the only compounds considered for a particular pharmacological activity.⁵⁷

3 Results and discussion

3.1 Synthetic chemistry

A series of cyclic amine substituted 1,8-naphthyridine-3-carboxylic acid analogues was efficiently synthesized based on the methods we had developed in our previous schemes.³⁵

The key compounds, 1-(4-chlorobenzyl)-4-oxo-1,4-dihydro-1,8-naphthyridine-3-carboxylic acid **4a** and 1-benzyl-4-oxo-1,4-dihydro-1,8-naphthyridine-3-carboxylic acid **4b**, were prepared as outlined *via* a four-step methodology in excellent yield.

The condensation reaction of 2-aminopyridine with ethoxy methylene malonate by Gould–Jacobs reaction yielded diethyl 2-((pyridine-2-ylamino)methylene) malonate **2** that was cyclized during refluxing with phenoxy ether to give ethyl 4-oxo-1,4-dihydro-1,8-naphthyridine-3-carboxylate **3**.⁵⁸ *N*-alkylation of **3** in anhydrous DMF with appropriate alkyl chloride in the presence of sodium hydride gave the 1-alkyl-4-oxo-1,4-dihydro-1,8-naphthyridine-3-carboxylic acid ethyl esters. Hydrolysis in 10% aqueous NaOH solution yielded the corresponding 1-alkyl-

4-oxo-1,4-dihydro-1,8-naphthyridine-3-carboxylic acids **4a** and **4b**. The coupling step was achieved in 24 h by heating the corresponding acids with the appropriate amines (2-aminopyridine, chloroaniline, morpholine, pyrrolidine, piperazine, *etc.*) in a dry DMF solution in a sealed tube. The target compounds **5a1–10** and **5b1–10** were purified by recrystallization from appropriate solvents.

The synthesized compounds were purified *via* column chromatography using a methanol : chloroform (10 : 40) mixture as the eluent. Characterization of the synthesized molecules was performed using FTIR spectroscopy, ¹H NMR spectroscopy, elemental analysis and mass spectrometry, which confirmed the formation of the structure. Further, the structures were established by mass spectral data in accordance with their molecular formula. Compound **5a**, for example, exhibited a molecular ion peak at *m/z* 390.2 (*M*⁺) in its mass spectrum. Its ¹H NMR spectrum included the characteristic signals at 8.02 (d, 1H, Ar-H), 9.11 (d, 1H, Ar-H), 7.66 (d, 1H, Ar-H), and 7.76 (d, 2H Ar-H), corresponding to the naphthyridine ring structure.

3.2 Antihistaminic activity study

A series of compounds containing the 1,8-naphthyridine-3-carboxylic acid scaffold was evaluated for their *in vivo*

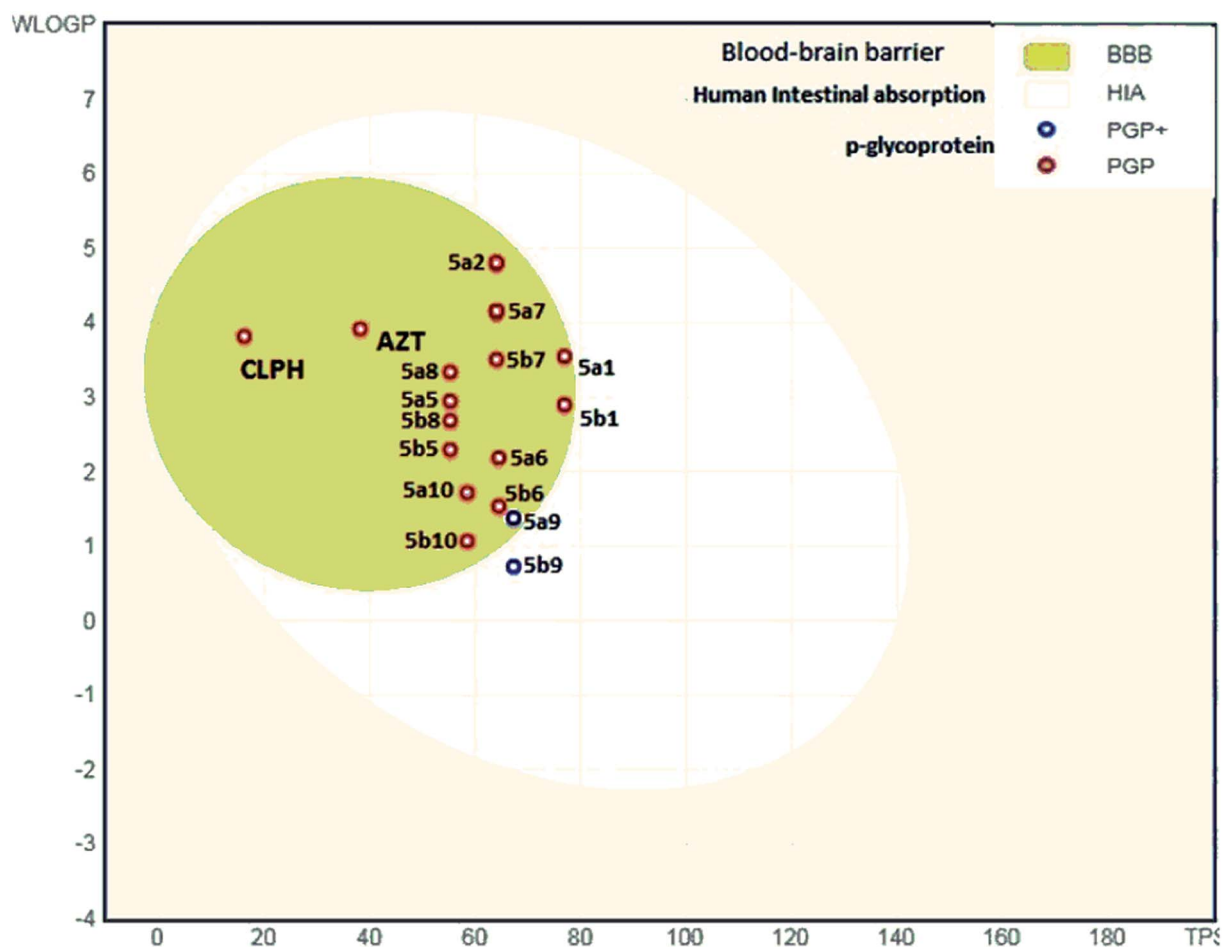


Fig. 6 ADME properties of compounds **5a1–a10** and **5b1–b10** and azelastine by graphical representation (boiled-egg) (predict gastrointestinal absorption and brain penetration of small molecules).



antihistaminic activity. All the test compounds were found to exhibit comparable antihistaminic activity (Table 1). The percentage protection result showed that all the test compounds of the series exhibited comparable protection in the range of 52–61%. The pharmacological studies of the test compounds indicated that different substituents present at the third position of 1,8-naphthyridine-3-carboxylic acid exert varied pharmacological activity. Replacement of the hydroxyl group of the acid with alicyclic amine (pyrrolidinyl and piperidinyl compounds **5a5**, **5a8**, **5b5**, and **5b8**, respectively) showed good antihistaminic activity. Replacement of the heterocyclic amine (morpholinyl, piperazinyl and *N*-methyl piperazinyl compounds **5a6**, **5a9**, **5a10**, **5b6**, **5b9** and **5a10**, respectively) showed comparable antihistaminic activity. Furthermore, replacement with aromatic amine (amino pyridinyl and cyclohexylamine compounds **5a1**, **5a7**, **5b1** and **5b7**, respectively) led to a further increase in activity. Amongst the series, compound **5a1** (pyridinyl derivative) was the most potent with the percentage protection of 61.45 that is comparable to that of the standard drug (65.20%).

3.3 Sedative-hypnotic exercise

Sedation is the most common side effect associated with more than 20% of antihistamines used by human beings. Hence, *in vivo* sedative-hypnotic activity was also evaluated for the test compounds. The percentage fall in photoactometer count was measured as the degree of sedation produced. The results of this study shown in Table 1 demonstrate that most of the tested compounds exhibited insignificant sedation (less than 10%). Chlorpheniramine maleate (first-generation H1R inhibitor) and Cetirizine (second-generation H1R inhibitor) were used as reference drugs for this study. Chlorpheniramine maleate showed approximately 30% sedation and cetirizine showed around 10–12% sedation. All the test compounds showed lower sedative potentials; while compared to the reference drug, the test compounds exhibited equipotent activity. Thus, this series can be developed as a clinically useful novel class of non-sedative antihistamines.

3.4 Molecular docking study

Molecular docking is a valuable technique in computational chemistry to deeply analyse ligand recognition and it has led to important breakthroughs in drug discovery and design in the field of medicinal chemistry.⁵⁹ Molecular docking methodology explores the binding mode and affinity of a small molecule within the binding site of the receptor target protein. The docked ligands were ranked according to their binding affinity in ligand–receptor complexes (Fig. 3 and 4a–f).

Molecular docking was performed on a set of test compounds, *viz.* **5a1–10** and **5b1–10**, against the H1 receptor in order to identify the critical ligand–protein interactions. The scores due to docking for the tested ligands determined that all the prepared admixtures possessed potential for interaction with one or more amino acids in the active site (binding pocket) of the receptor. Compound **5a1** was the most potent inhibitor of the receptor. In the docking, the binding of **5a1** was supported

Table 4 Predicted biological activities of compounds **5a1–a10**^a

Activity name	5a1		5a2		5a3		5a4		5a5		5a6		5a7		5a8		5a9		5a10		
	Pa	Pi	Pa	Pi	Pa	Pi	Pa	Pi	Pa	Pi	Pa	Pi	Pa	Pi	Pa	Pi	Pa	Pi	Pa	Pi	
Antiallergic	0.556	0.020	0.528	0.024	0.521	0.024	0.544	0.024	0.294	0.022	0.096	0.297	0.095	0.500	0.028	0.311	0.088	0.256	0.125	0.284	0.103
Antiasthmatic	0.490	0.029	0.479	0.030	0.450	0.035	0.468	0.035	0.193	0.032	0.180	0.204	0.167	0.447	0.036	0.209	0.162	0.192	0.180	NA	NA
Antihistaminic	NA	NA	NA	NA	0.237	0.210	NA	0.210	0.171	NA	0.053	0.140	0.079	NA	NA	0.185	0.045	0.145	0.073	0.188	0.043
AIFML	0.259	0.202	0.254	0.167	0.113	0.071	0.255	0.071	NA	0.164	NA	NA	NA	0.240	0.201	NA	NA	NA	NA	NA	NA
Histamine antagonist	NA	NA	NA	NA	NA	NA	NA	NA	0.112	0.071	0.071	NA	NA	NA	0.024	0.122	0.062	0.097	0.094	0.123	0.061
Bronchodilator	0.127	0.056	0.138	0.047	0.166	0.098	0.143	0.044	NA	0.044	NA	NA	NA	0.187	0.024	NA	NA	NA	NA	NA	NA
H1R agonist	0.161	0.104	0.166	0.098	0.359	0.115	0.165	0.165	0.179	0.099	0.083	0.078	0.070	0.238	0.040	0.171	0.092	0.208	0.057	0.165	0.100
H1R antagonist	NA	NA	NA	NA	NA	NA	NA	NA	0.100	0.045	NA	NA	NA	NA	NA	0.094	0.051	0.092	0.053	0.153	0.025
HRI	NA	NA	0.373	0.107	0.310	0.096	0.353	0.118	NA	0.118	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA
Hks	0.321	0.088	0.310	0.096	0.080	0.038	0.320	0.084	0.333	0.089	0.081	0.318	0.090	0.288	0.111	0.321	0.088	0.309	0.096	0.310	0.096
IgE antagonist	0.092	0.018	NA	NA	0.140	0.023	0.084	0.029	NA	0.029	NA	NA	NA	0.092	0.017	NA	NA	NA	NA	NA	NA
IMM	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	0.198	0.161	NA	NA	NA	NA	NA	NA
IL antagonist	0.141	0.139	0.158	0.127	0.152	0.124	0.152	0.135	NA	0.135	NA	0.225	0.044	NA	NA	NA	NA	NA	NA	NA	NA
PDE inhibitor	0.150	0.026	0.159	0.023	0.149	0.026	0.168	0.020	0.084	0.020	0.066	0.093	0.057	0.158	0.023	0.094	0.056	0.082	0.067	0.094	0.052
Rhinitis treatment	0.317	0.093	0.282	0.125	0.311	0.098	0.312	0.097	0.333	0.097	0.081	0.350	0.072	0.269	0.140	0.326	0.087	0.479	0.023	0.363	0.066

^a Hks – histidine kinase, PDE – phosphodiesterase, IgE – immunoglobulin E, H1R – histamine H1 receptor, HRI – histamine release inhibitor, IL – interleukin, Pa (probability “to be active”), Pi (probability “to be inactive”), NA – not available, AIFML – anti-inflammatory, IMM – immunomodulatory.



Table 5 Predicted biological activities of compounds 5b1–b10^a

Activity name	5b1		5b2		5b3		5b4		5b5		5b6		5b7		5b8		5b9		5b10	
	Pa	Pi	Pa	Pi	Pa	Pi	Pa	Pi	Pa	Pi	Pa	Pi	Pa	Pi	Pa	Pi	Pa	Pi	Pa	Pi
Antiallergic	0.575	0.018	0.536	0.023	0.521	0.024	0.534	0.023	0.280	0.106	0.283	0.104	0.505	0.027	0.296	0.095	0.240	0.141	0.269	0.114
Antiasthmatic	0.520	0.025	0.490	0.029	0.450	0.035	0.458	0.034	0.176	0.050	0.195	0.177	0.461	0.033	0.200	0.172	0.149	0.069	NA	NA
Antihistaminic	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	0.143	0.075	NA	NA	0.189	0.043	NA	NA	0.193	0.041
AIFML	0.245	0.218	NA	NA	0.237	0.210	0.255	0.165	NA	NA	NA	NA	0.257	0.161	NA	NA	NA	NA	NA	NA
Histamine antagonist	NA	NA	NA	NA	NA	NA	NA	NA	0.116	0.067	0.096	0.095	NA	NA	0.128	0.058	0.100	0.089	0.129	0.057
Bronchodilator	0.156	0.037	NA	0.041	0.113	0.071	0.140	0.046	NA	NA	NA	NA	0.219	0.017	NA	NA	NA	NA	NA	NA
H1R agonist	NA	NA	0.148	0.113	0.166	0.098	0.173	0.089	0.151	0.121	NA	NA	0.218	0.050	0.143	0.137	0.183	0.078	NA	NA
H1R antagonist	NA	NA	NA	NA	NA	NA	NA	NA	0.095	0.050	NA	NA	NA	NA	0.089	0.056	0.087	0.059	0.139	0.028
HRI	0.256	0.180	0.155	0.106	0.359	0.115	0.355	0.118	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA
Hks	0.261	0.138	0.375	0.107	0.310	0.096	0.321	0.088	0.271	0.126	0.260	0.139	0.232	0.175	0.261	0.138	0.250	0.151	0.250	0.151
IgE antagonist	0.105	0.008	0.294	0.057	0.080	0.038	0.085	0.028	NA	NA	NA	NA	0.107	0.008	NA	NA	NA	NA	NA	NA
IMM	0.222	0.132	NA	NA	NA	NA	0.196	0.164	NA	NA	0.194	0.167	0.236	0.115	NA	NA	NA	NA	NA	NA
IL antagonist	0.166	0.011	0.181	0.028	0.140	0.023	0.138	0.025	NA	NA	0.300	0.015	0.270	0.028	0.250	0.041	NA	NA	NA	NA
PDE inhibitor	0.194	0.015	0.127	0.020	0.149	0.026	0.069	0.034	0.109	0.046	0.118	0.040	0.120	0.039	0.068	0.037	0.106	0.048	0.113	0.043
Rhinitis treatment	0.294	0.113	0.057	0.133	0.311	0.098	0.309	0.099	NA	NA	0.241	0.176	0.300	0.107	0.474	0.025	0.336	0.080	0.336	0.080

^a Hks – histidine kinase, PDE – phosphodiesterase, IgE – immunoglobulin E, H1R – histamine H1 receptor Pa (probability “to be active”), Pi (probability “to be inactive”), NA – not available.

by three hydrophilic interactions between residues Asp107, Ser111, and Tyr431 and the ligand at bond distances of 2.84 and 3.08 and 3.18 Å. According to the dockings, compounds **5a2–10** and **5b1–10** shared a similar binding profile with **5a** and also showed two hydrophilic interactions with residue Asp107. The docked ligands were found to have a similar binding capacity to the co-crystallized ligands. Theoretically, the entire synthesized compounds showed moderate to good docking scores and binding energy to the selected protein target ranged from -7.1 to -10.6 kcal mol⁻¹, which is in good agreement with the observed antihistaminic activity. Most of the test ligands showed comparable docking scores in comparison to the reference drug azelastine, which was used as a reference for interpretations of biological study and analysis. Amongst these compounds, the complex of **5b2** (*o*-chloroaniline derivative) showed high binding affinities during the docking study in the active sites of the H1R receptor (PDB ID: 3RZE). The results are shown in Table 2.

3.5 ADME analysis

The results obtained from *in silico* studies clearly indicate that the compounds had drug-like candidate properties with no violation of any of the drug-likeness rules discussed above. It was interesting to note that the results of the SWISS ADME predictor values of log *P*, molar refractivity and the total polar surface area in these molecules were in excellent agreement with the most important rules of drug-likeness.

Though these compounds exhibited a good hydrophilic-lipophilic balance and the same predicted bioavailability, the halogen derivative with high lipophilicity was expected to show decent GI absorption. In addition, we calculated the total polar surface area (TPSA) since it is another key property that is related to drug bioavailability. Thus, passively absorbed molecules with TPSA >140 are thought to have low oral bioavailability. The results obtained from the Swiss ADME search engine are listed in Table 3.

The analysis indicates that the derivatives fell within the permissible range of standard drugs, as is evident from the boiled-egg diagram (Fig. 5a–c and 6).

It is also clear that compounds **5a1–5a10** and **5b1–5b10** cannot be affected by the P-glycoprotein of the CNS system. Compounds present in the yellow zone except **5a9** and **5b9** can permeate through the blood–brain barrier (BBB). All compounds present in the white area can be absorbed very easily by the gastrointestinal tract. In the present study, the synthesized ligand and its complexes were found to be in good agreement with the given criteria and can be said to possess good oral bioavailability.

3.6 Biological activity spectrum PASS analysis

The biological activity spectra of the synthesized compounds were determined by using an online version of PASS software. The results obtained (Tables 4 and 5) were interpreted and used in a flexible manner. The antihistaminic activities of the derivatives were only slightly changed by amidation at position 3. However, an additional chloro-substituent increased the activity remarkably, as was observed for compound **5a3**. Compound **5a3** showed the highest Pa for H1R inhibitory

activity (0.359), whereas compound **5b1** showed the highest Pa for antiallergic and antiasthmatic activity (0.575 and 0.520, respectively), compound **5a2** showed the highest Pa for histamine release inhibitor (0.373), compound **5a9** showed the highest Pa for rhinitis treatment (0.479), and compound **5b6** showed the highest Pa for IL antagonist activity (0.300).

4 Conclusions

The synthesis, characterization, *in silico* ADME and *in vivo* antihistaminic activity of a new series of 1,8-naphthyridine-3-carboxylic acid derivatives have been described in this manuscript. A combination of cyclic amines with a naphthyridine ring and possession of sufficient hydrogen donor-acceptor sites resulted in good antihistaminic activity.

These compounds showed significant H1R inhibitory (antihistaminic) activity through histamine-induced bronchoconstriction on conscious guinea pigs by the *in vivo* model. Out of the synthesized compounds, the pyridyl derivative **5a1** showed maximum antihistaminic activity with 61.45% protection. Further, the average sedative properties of the synthesised compounds were found to be very small (<16%) when compared to the standard drug chlorpheniramine maleate (38%). The molecules showed promising *in silico* results, as indicated by their significant scoring functions and high protein-ligand interaction energy, which simultaneously predicted the activity of the test compounds. In conjugation with the *in silico* results, the compounds showed very promising antihistaminic activities. The *in silico* ADME profiling, toxicity, drug-likeness, drug-scoring results, PASS analysis and *in vitro* antihistaminic activities suggested that the compounds are promising leads for the development of selective, safe and potent antihistamines. The noteworthy conclusion of the study is the development of 1,8-naphthyridine derivatives with pyrazolines, morpholine, piperidine and piperazine like cyclic amines as remarkable antihistaminic agents in general. Further study is in progress with these compounds and related heterocyclic scaffolds to get information about the chemical structure that efficiently modulates histamine release to discover new small molecule(s) endowed with noteworthy antihistaminic activities.

Conflicts of interest

The authors declare that there are no conflicts of interest in this manuscript.

Abbreviations

RPMC	Rat peritoneal mast cells
PDE 4	Phosphodiesterase 4
ADMET	Absorption distribution, metabolism, excretion, toxicity
PASS	Prediction of activity spectra for substances
TPSA	Total polar surface area

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