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## A miniaturized push–pull-perfusion probe for few-second sampling of neurotransmitters in the mouse brain<sup>†</sup>

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**Table 1** Comparison of various techniques to measure neurotransmitter concentrations *in vivo*. Colored text indicates what in our opinion is an advantage (green) or drawback (red) of the probe technology

Technology	Temporal resolution	Sampling area	Lower concentration range	Range of molecules	Remarks
<b>Sampling probes</b>					
Dialysis probe <sup>11,20,21</sup>	Minutes	≥ 0.6 mm <sup>2</sup>	nM	Wide	
Push-pull perfusion probe <sup>5,6,15,22</sup>	Minutes	≥ 0.0003 mm <sup>2</sup>	nM	Wide	Difficult to use – risks of clogging
Micro push-pull perfusion probe	Seconds	0.004 mm <sup>2</sup>	nM	Wide	
<b>Sensing probes</b>					
Electrochemical probe <sup>23,24</sup>	Seconds	~0.003 mm <sup>2</sup>	μM	Limited	Reference electrode needed

Furthermore, a microfabricated needle with integrated droplet generator was used for the collection and delivery of nanoliter volumes of liquid with a particularly high temporal resolution (averaged 3.4 seconds).<sup>19</sup> To this end the fluorescent sample is pulled through 2.6 μm-wide ‘slits’ using negative outlet pressure, which produces a train of aqueous sample droplets that are captured in the oil phase that flows along the narrow channels.

Miniaturization is necessary to enable more localized measurements, which also should limit tissue damage inflicted during the measurement. As indicated in Table 1, push-pull types of probes can suffer from clogging,<sup>22</sup> a problem that can be solved by sampling through a porous aluminium oxide membrane located on top of the microfabricated probe needle.<sup>25</sup> Even more advanced, hybrid approaches that combine fluidic handling with electrical sensing for measurements of neuronal or ion activity, have been demonstrated in probes that combined EEG recordings for monitoring brain activity and drug dosing *via* microfluidic channels.<sup>26,27</sup> A hybrid probe combining direct sampling and droplet compartmentalization, and EEG readout has been published recently by the group of Renaud,<sup>14</sup> that allows tracking changes in local concentrations of trace metals, such as sodium, magnesium, potassium, calcium, copper and zinc, by laser-ablation – inductively coupled plasma – mass spectrometry.

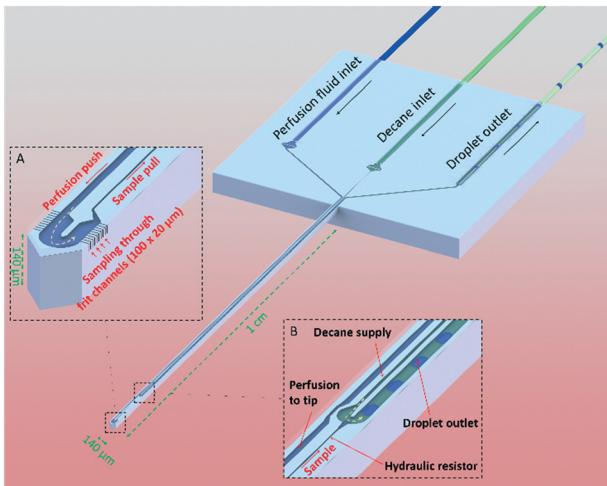
In contrast to the sampling techniques described earlier, electrochemical probes have the advantage of high temporal and spatial resolution. However, the concept of electrochemical probes relies on selectivity towards one specific type of molecule that is typically provided by including enzymatic reactions, which prove difficult to keep stable, and impossible to periodically calibrate, once placed in the brain.<sup>28</sup> Electrochemical probes have been used to measure a variety of ions and neurotransmitters such as glutamate,<sup>7,29</sup> dopamine,<sup>30</sup> and choline.<sup>10</sup> Direct electrochemical detection can be used for *e.g.* dopamine. However, a more robust sensing mechanism is based on an enzymatic conversion of the respective neurotransmitter to obtain sufficient selectivity, followed by amperometric detection. Alternatively, silicon-based probes can be equipped with different electrochemical

sensing areas that are functionalized using ionophores, achieving for example selectivity for K<sup>+</sup> on one sensing area, while another sensing area can be made pH-sensitive, thus enabling simultaneous measurements of K<sup>+</sup> and H<sup>+</sup> concentrations.<sup>31,32</sup> With such an ion-sensitive probe, *in vivo* K<sup>+</sup> variations were recorded in a mouse brain during induction of cortical spreading depolarization (CSD), a strong wave of neuronal and glial cell depolarization.<sup>23,33</sup>

Here we present a microfabricated push-pull perfusion sampling (μPPPS) probe capable of sampling with a spatial resolution of ~4000 μm<sup>2</sup>. In our probes, the probe tip (140 μm wide) is equipped with two rows of ten 3 μm-wide microfluidic sampling channels that act as a sort of ‘frit’ that allows molecules to be captured while larger extracellular matrix debris is kept outside. A novelty of our ‘frit’ design is that it allows for adjustment of the internal flow rate and pressure drop over these microfluidic sampling channels and tuning of mass transport between a more ‘diffusive’ *versus* a ‘convective’ mode. We designed two types of probes. In the first version (probe type 1), our microfabricated needles are equipped with a droplet generator inside the probe tip to capture dialysate in droplets immediately after it is sampled into the probe. This approach decouples sample collection from sample analysis, as the changes in concentration are stored in this ‘chemical memory’. In a second version of our probe (probe type 2) an external droplet generator is connected to a push-pull microprobe. Here, the dialysate and sample is combined with enzymes (glutamate oxidase and horseradish peroxidase) and a fluorophore precursor (Amplex Red) prior to droplet formation, in order to create a reaction mixture that produces a fluorescent signal (resorufin) that represents the amount of target neurotransmitters. A major benefit of our method compared to enzymatic electrochemical sensors, is that the enzymes and their reaction products are not in contact with brain tissue (thus preventing confounding effects of glutamate conversion at the site of measurement) while providing similar spatial and temporal resolution.

Sample analysis for both types of probes is done on-line, outside the sampling area, using a dedicated optical setup based on fluorescence detection. The optical filters and light





**Fig. 1** Schematic visualization of the  $\mu$ PPPS principle. Sample from tissue is collected through small sampling channels (see sampling tip inset, A). A push-pull configuration inside the needle ensures rapid transport of the sample to the external measurement equipment. A droplet generator is either integrated in the shaft of the probe (see droplet generator inset, B; probe type 1) or located externally (probe type 2), to prevent dispersion and to maintain a good temporal resolution.

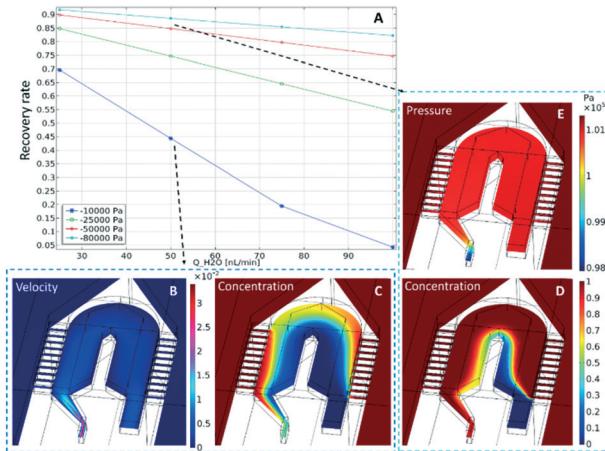
source are tuned for the detection of resorufin, a fluorophore frequently used in commercially available assay kits for *e.g.* glutamate or acetylcholine. The outlet fused silica capillary of the  $\mu$ PPPS probe is mounted in front of a LED source and photo multiplier tube, enabling the on-line detection of resorufin at concentrations  $<1$  nM.

Here we first describe the development and characterization of the two types of newly developed  $\mu$ PPPS probes<sup>34</sup> using fluorescein and glutamate, and then show the capability of our approach to measure – in an *in vivo* experiment in a mouse – glutamate levels in the brain in relation to electrically evoked neuronal network activity.

## Design

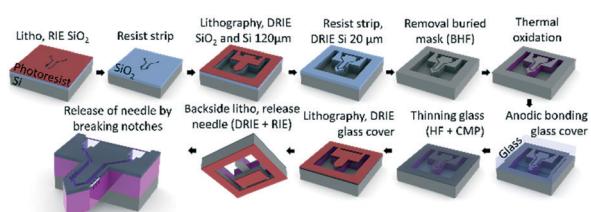
### Probe design

The operating principle of the  $\mu$ PPPS probe is illustrated in Fig. 1. The probe is operated using a dialysate inlet, a continuous phase inlet (in this case decane) and a droplet outlet. Dialysate flow is infused in the probe tip at a constant flow rate. A concentration gradient develops across the microfluidic sampling channels and molecules will diffuse inwards to be collected by the dialysate flow. This series of 10 sampling channels (3  $\mu$ m wide, 20  $\mu$ m height), spanning a length of 100  $\mu$ m on both sides of the probe tip, reduces the risk of probe clogging and provides a reasonable spatial resolution along the length of  $\sim$ 100  $\mu$ m. Note that the sampling channels are small open microfluidic channels providing transport of ions into the probe, which is different from porous materials such as membranes or hydrogels. The resulting diffusion coefficients of transported ions are therefore equally high as in bulk media. The cross-sectional area through

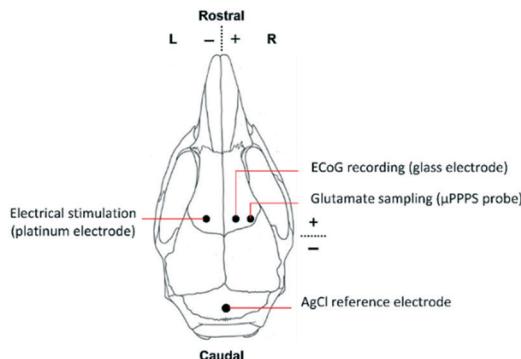


**Fig. 2** COMSOL simulations to determine (A) the recovery rate of the  $\mu$ PPPS probe tip as a function of flow rate (x-axis) and outlet pressure (legend). Velocity profile ( $\text{m s}^{-1}$ ) (B) and concentration profile (C) using 50  $\text{nL min}^{-1}$  and  $-10$  kPa outlet pressure. Pressure distribution (E) and concentration profile (D) for a flow rate of 50  $\text{nL min}^{-1}$  and  $-80$  kPa outlet pressure. Note that the pressure outside the probe is set to 1 bar (101.32 kPa).

which molecules enter the probe is 1200  $\mu\text{m}^2$ . Note however, that there are two rows of 10 sampling channels on each side of the tip, spatially separated by 140  $\mu\text{m}$ . Two types of probes were fabricated. Probe type 1 has a droplet generator in the needle of the probe (see Fig. 1, and 5A). Probe type 2 uses an external glass chip droplet generator (Fig. 5D-F) that was used to enable addition of an enzymatic assay kit to detect neurotransmitter (glutamate). A large hydraulic resistance in the form of a narrow channel (probe type 1: channel width 5  $\mu\text{m}$ , channel height 20  $\mu\text{m}$ , and 1 mm channel length; probe type 2: channel width 20  $\mu\text{m}$ , channel height 20  $\mu\text{m}$ , and 1 cm channel length) is present between the tip and the droplet generator to avoid the continuous phase (decane) reaching the tip in case of small imbalances in flow in practical applications. For probe type 1, following sample collection (from tissue) and the hydraulic resistor, the organic continuous phase comes in contact with the dialysate sample inside a curved T-junction to create a train of water droplets that preserves concentration fluctuations in time, providing a temporal resolution of just a few seconds, while enabling both on-line and off-line analysis without losing temporal information.



**Fig. 3** Schematic overview of the fabrication process of the  $\mu$ PPPS probe. Note that the probe and layer thicknesses are not drawn to scale. See text for a full description, and Table S6† for a more detailed process flow.



**Fig. 4** Schematic representation of a mouse skull indicating locations of the  $\mu$ PPPS probe in relation to the stimulation site, ECoG and reference electrode during *in vivo* experiments in an anesthetized mouse.

To promote efficient sample collection, a low negative pressure is applied to the droplet outlet. Ideally, the push-pull probe is operated in such a way that the push and pull flows are equal, which prevents withdrawing of liquid from or introducing liquid to the measurement volume (*i.e.* a sample cup, or mouse brain tissue). In practice, this is difficult to realize and the operating conditions are determined in such a way that a small inward flow is established in order to establish a stable sampling process.

An electrical equivalent circuit is used to determine the appropriate flow and pressure conditions to operate the probe and to predict the inward flow rate. Hydraulic resistances of the microfluidic channels and connected tubing, together with the flow and pressure sources are modelled in a circuit as shown in S1.† From this schematic, it can be calculated that an outlet pressure of  $-80$  kPa, a continuous phase flow of  $200$   $\text{nL min}^{-1}$  and perfusion flow of  $50$   $\text{nL min}^{-1}$  give rise to an inwards flow rate of  $126$   $\text{nL min}^{-1}$ .

## Numerical simulations

To predict the recovery rate of both types of our  $\mu$ PPPS probes, simulations were run on a 3D-model (see S10†) of the probe using COMSOL Multiphysics (version 5.3). Flow velocity ( $u$ ) and pressure ( $p$ ) variables were calculated using the Stokes flow approximation of the Navier-Stokes equation (creeping flow physics model) (1) and the continuity eqn (2), after which concentration ( $c$ ) distributions were calculated using the steady-state convection-diffusion equation (transport of diluted species physics model, with convection and without migration) (3):

$$\mu \nabla^2 u - \nabla p = 0 \quad (1)$$

$$\nabla \cdot u = 0 \quad (2)$$

$$u \cdot \nabla c - D \nabla^2 c = 0 \quad (3)$$

The viscosity is used from water at  $293$  K and the mass transport is studied of a substance with  $D = 0.9 \times 10^{-9}$   $\text{m}^2 \text{s}^{-1}$ ,

equal to the value for monosodium glutamate.<sup>35</sup> Boundary conditions are a perfusion inflow of  $25$  to  $100$   $\text{nL min}^{-1}$ , and an outlet pressure between  $-10$  kPa and  $-80$  kPa *vs.* atmospheric pressure (1 bar or  $101.32$  kPa). Note that this outlet pressure is set right after the tip of the needle using a laminar outflow boundary condition, including the small hydraulic resistance, but not including the droplet generator, or connecting tubing towards the detection area. Pressure around the probe tip was set to 1 bar (atmospheric pressure). All other boundaries carry a no-slip condition ( $u = 0$ ). An arbitrary concentration ( $c_0$ ) of  $1$  mM was defined at the boundaries around the probe tip and of  $0$  mM at the perfusion inlet. We have validated that any arbitrary concentration will lead to identical results for the recovery rate, as we do not take any adsorption effects into account. The concentration gradient normal to the outlet is set to zero and the flux normal to all other boundaries is zero.

The concentration distributions at the tip are shown in Fig. 2, for a perfusion flow rate ( $Q_{\text{H}_2\text{O}}$ ) of  $50$   $\text{nL min}^{-1}$  and an outlet pressure of either (C)  $-10$  kPa or (D)  $-80$  kPa. In the plot it can be seen that the sampling channels promote a significant uptake of compounds around the probe tip. To quantify this uptake, the recovery rate (the ratio between sampled concentration, and external concentration) is calculated at the channel outlet. The calculated values are plotted in Fig. 2A for varying perfusion flow rates and outlet pressures. As can be expected, the recovery rate increases if the negative outlet pressure is increased or the perfusion flow rate is decreased. To measure concentration changes at a reasonable temporal resolution, a certain inward flow is needed to provide a sufficient recovery rate. For example, the recovery rate under conditions similar to those discussed above (*i.e.* with a perfusion flow rate of  $50$   $\text{nL min}^{-1}$  and an outlet pressure of  $-80$  kPa) is  $88\%$ .

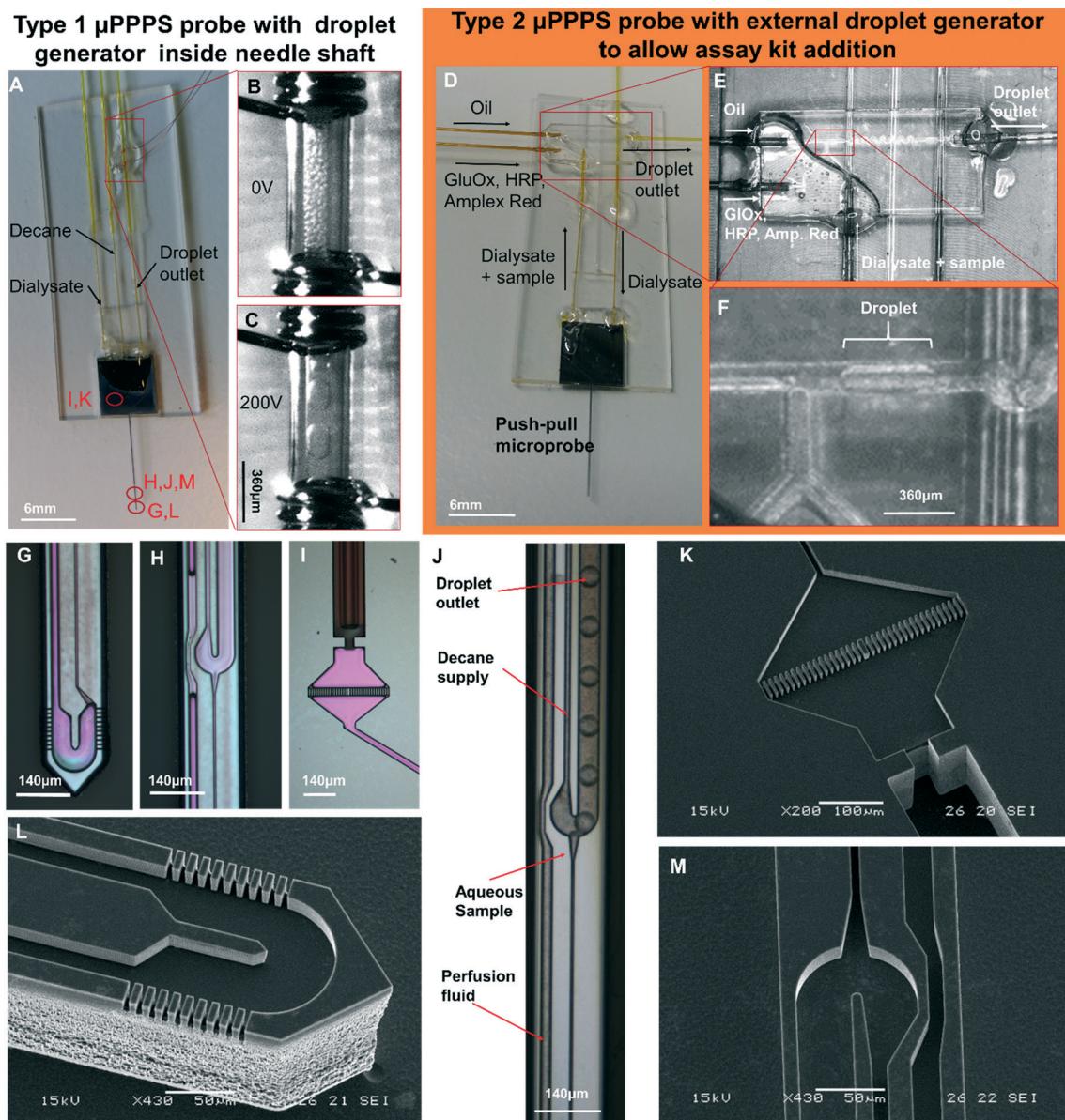
## Methods

### Probe fabrication

The  $\mu$ PPPS probes were fabricated in the Nanolab cleanroom of the MESA+ Institute of Nanotechnology, following the steps as illustrated in Fig. 3.

Silicon wafers are provided with a thermal  $\text{SiO}_2$  layer, which was used as a buried mask that is patterned with the fluidic channel structures. Following a second lithography step, fluidic inlets and outlets ( $140$   $\mu\text{m}$ ) and the channel structures ( $20$   $\mu\text{m}$ ) were etched in two consecutive deep reactive ion etching (DRIE) steps. The delicate sampling channel structures are protected by a thin thermally grown oxide layer to prevent damage during the last DRIE step to release the probe. The channels were covered with a  $500$   $\mu\text{m}$  thick borosilicate glass wafer by anodic bonding, etched with  $25\%$  hydrogen fluoride (HF) and subsequently polished back to  $\sim 20$   $\mu\text{m}$  thickness using chemical mechanical polishing. In the last two steps, the probes were released by etching slits through the glass and the silicon layers surrounding the structures. The probes were kept in place within the wafer





**Fig. 5** Push-pull microprobes. A: μPPPS probe type 1 with internal droplet generator located inside the shaft of the needle. B and C: Enlarged view of the electrode pair (spaced 1 mm) used for merging microdroplets (see Movies S9A and B†). Merging occurs as soon as the microdroplets are transferred from the 30  $\mu\text{m}$  inner diameter outlet capillary to the 150  $\mu\text{m}$  I.D. transport capillary. Droplets are visible inside the transport capillary before (B) and after (C) high-voltage is switched on (200 V<sub>pp</sub>, 10 Hz square wave). D: μPPPS probe type 2 with external glass chip droplet generator to allow the addition of an assay kit consisting of enzymes and a fluorescent indicator to detect neurotransmitter (glutamate). E: Photo of the external droplet generator used for probe type 2 (see Movie S8†). F: Close-up of the plug-shaped droplets forming inside the external droplet generator. G: μPPPS probe type 1 tip (sampling channels are identical for probe types 1 and 2). H: Droplet generator inside needle (probe type 1). I: Inlet of the probe containing a pillar array filter. A 130  $\mu\text{m}$  O.D. fused silica capillary is glued into the inlet channel. J: Droplet formation for probe type 1 (see Movie S7†). K: SEM photo of the inlet filter and capillary connecting channel. L: SEM photo of the tip of the μPPPS probe type 2, with the dialysate entry, return channel, and sampling channels visible. M: SEM photo of the droplet generator inside the needle (probe type 1). Microfluidic channel structures are all 20  $\mu\text{m}$  deep, the capillary connecting channel is 140  $\mu\text{m}$  deep.

with two small silicon notches that allow one to carefully detach a probe from the substrate using tweezers. A more detailed process flow is included in Table S6.†

Prior to their characterization, the probes were mounted on a micro-milled polystyrene substrate featuring recessed placeholders for the probe and capillaries to simplify alignment using cyanoacrylate glue (Pattex – super gel, Henkel,

Nieuwegein, Netherlands). Some 2.5 cm of 30  $\mu\text{m}$  inner diameter (ID)/130  $\mu\text{m}$  outer diameter (OD) fused silica capillary tubing was glued (type NOA83H, Norland Optical Adhesives, Cranbury (NJ), USA) in the probe inlets and outlet. The NOA83H glue is UV-curable, which was used to cure the glue before it enters the microfluidic channels by capillary forces. Subsequently, the cyanoacrylate was added on this joint to

add additional bonding strength. Curing of the cyanoacrylate was sped up using a spray can of CA accelerator (Zip Kicker, ZAP, Ontario (CA), USA). The other end of the 130  $\mu\text{m}$  OD tubing was subsequently glued by repeating the glue step with NOA83H and cyanoacrylate to 30 cm of 150  $\mu\text{m}$  ID/360  $\mu\text{m}$  OD-fused silica capillary tubing. Finally, the 360  $\mu\text{m}$  OD-fused silica capillary was fixed to the polystyrene substrate using epoxy (Loctite EA M-31CL, Henkel, Nieuwegein, Netherlands). See Fig. 5A for a photo of the assembled device.

A different probe type 2 was constructed for the glutamate level measurements, because an additional inlet was needed to supply the chemicals for an enzyme-based assay. This device is shown in Fig. 5D. In this case, a push-pull microprobe without internal droplet generator is connected to a dialysate inlet. The outlet of this probe is connected with a 2.5 cm long glass capillary of 30  $\mu\text{m}$  ID/130  $\mu\text{m}$  OD to an external borosilicate glass droplet generator. This droplet generator contains a Y-junction where the dialysate + sample flow is combined with the enzymes and fluorophore precursor. Droplets were generated from this mixture in decane at a T-junction, and the droplet contents were subsequently mixed in a meandering channel. In the first step, 100  $\mu\text{m}$  channels in this droplet generator were wet-etched (25% HF, etch rate  $\sim 0.9 \mu\text{m min}^{-1}$ ). In the second step, 380  $\mu\text{m}$  inlets were wet-etched for the glass capillaries to be glued in place.

## Chemicals

Sampling with the  $\mu$ PPPS probes was done by generating aqueous droplets in decane as the continuous phase. Decane has relatively low values for gas solubility compared to *e.g.* fluorinated alkanes, which prevents gas bubbles being formed inside the probe due to the negative outlet pressures.<sup>36,37</sup> *In vitro* characterization was done using fluorescein that was prepared in a 10 mM stock solution and titrated in de-ionized water to sample from varying concentrations. Water was purified using an Elga PureLab flex purification system. Water and decane were filtered using a 0.22  $\mu\text{m}$  pore size filter prior to use. Decane, resorufin sodium salt and fluorescein sodium salt were obtained from Sigma-Aldrich GmbH (Steinheim, Germany).

For *in vivo* recordings a sterile minimal perfusion fluid (MPF) solution, consisting of 140.3 mM NaCl, 3.0 mM KCl, 1.0 mM MgCl<sub>2</sub>·6H<sub>2</sub>O<sub>2</sub> and 1.2 mM CaCl<sub>2</sub>·2H<sub>2</sub>O<sub>2</sub> in MilliQ and filtered with a 0.2  $\mu\text{m}$  sterilizing filter, was used for the 'push' flow. Glutamate measurements were performed using an Amplex Red glutamic acid/glutamate oxidase assay kit (Thermo Fisher Scientific, Breda, The Netherlands). With this method, L-glutamic acid is oxidized by glutamate oxidase to produce  $\alpha$ -ketoglutarate, H<sub>2</sub>O<sub>2</sub> and NH<sub>3</sub>. Subsequently, H<sub>2</sub>O<sub>2</sub> reacts with Amplex Red in a 1:1 ratio catalysed by horseradish peroxidase to generate resorufin, a highly fluorescent product.

## Measurement setup

The microprobe was mounted in a custom-made holder and connected to a syringe pump (Nemesys, Cetoni GmbH,

Korbussen, Germany) to infuse dialysate and decane. A pressure pump (MFCS-EZ, Fluigent, Paris, France) was used to apply negative pressure for sampling. For the electrical merging of droplets, a waveform generator (Agilent 33250A, Santa Clara, USA) was connected to an amplifier with an amplification of 100 V/V (Trek model 2210, Trek Inc., Lockport (NY), USA). To monitor the applied signal, an Agilent DSO3062A oscilloscope (Santa Clara, USA) was installed. For droplet merging pictures, a stereomicroscope with a PTGREY grasshopper 3 camera (FLIR Integrated Imaging Solutions GmbH, Ludwigsburg, Germany) was used. The droplets recorded inside the probe needle were recorded with an inverted Olympus microscope (Hamburg, Germany) with the same grasshopper 3 camera.

The optical setup consists of a high-power LED (M530L3, Thorlabs, Munich, Germany) controlled by a LED driver (LEDD1B, Thorlabs). LED power was measured using a silicon photodiode (S121C, Thorlabs) connected to a power meter (PM100USB, Thorlabs). The LED was focused at a measurement tube, which is a 50  $\mu\text{m}$  ID/360  $\mu\text{m}$  OD glass capillary tube from which the polyimide coating is locally removed. Alignment and focus of the LED beam was verified using a CMOS camera (DCC1545M, Thorlabs). For fluorescence measurements, a right-angle mirror (DFM/M-P01, Thorlabs) was used to reflect the beam towards a photomultiplier tube (PMT) (H7422, Hamamatsu, Herrsching am Ammersee, Germany). The PMT signal was amplified with 1 V  $\mu\text{A}^{-1}$  using an amplifier unit (C7319, Hamamatsu) before being collected by a National Instruments myDAQ board (Woerden, Netherlands). See Fig. S2A† for more information on the measurement setup and a schematic overview of the optics.

## In vivo recordings

For the *in vivo* experiments in a mouse brain, first, the fluorescent signal intensity was calibrated *in vitro* using a 0  $\mu\text{M}$ , 5  $\mu\text{M}$  and 10  $\mu\text{M}$  glutamate solution, since the expected glutamate levels in the mouse cortex are between approximately 2 and 5  $\mu\text{M}$ .<sup>24</sup> Since auto-oxidation of Amplex Red and deterioration of glutamate oxidase are known problems with this assay kit, this calibration procedure was performed every day. Moreover, with these concentration steps the time delay between sampling and detection is measured as well, allowing one to link the timing of measured fluorescence signal to the cortical activity changes occurring in the brain.

Subsequently, the mouse was prepared for *in vivo* surgery for measuring cortical glutamate levels (Fig. S5B†). Animals used for the experiments were male C57BL/6J mice (Harlan, Venray, The Netherlands), kept under standard housing conditions with a 12 hour light/dark cycle with free access to water and food. All animal experiments were approved by local (LUMC Authority of Animal Welfare) and national (Central Authority for Scientific Procedures on Animals) ethical committees in accordance with recommendations of the European Communities Council Directive (2010/63/EU). All efforts



were made to minimize animal suffering. Mice were anesthetized with isoflurane (induction 4%; maintenance 1.5 to 2%; in pressurized air) and placed in a stereotaxic frame (David Kopf Instruments, Tujunga, CA, USA) on a homoeothermic heating pad (Stoelting, Wood Dale, IL, USA) with continuous monitoring and maintenance of the core temperature ( $37.0 \pm 0.5^\circ\text{C}$ ). A blood pressure transducer (AD Instruments, Inc., Colorado Springs, CO, USA) was connected to PE-10 tubing for a catheter (BD Intramedic, VWR, Radnor, PA, USA) placed in the right femoral artery for continuous blood pressure monitoring, and for regular blood sample collection for assessing systemic physiology based on plasma pH,  $p\text{CO}_2$  and  $p\text{O}_2$  values as assessed by blood-gas analysis, using a Rapidlab blood gas analyzer 248 (Siemens Healthcare Diagnostics, the Hague, Netherlands). After exposing the skull bone, cortical windows were made over the right cortex using a dental drill (Fine Science Tools Inc., Foster City, CA, USA), in the presence of cooled saline to prevent overheating. Three windows were drilled for placement of i) a concentric Pt/Ir bipolar stimulation electrode with 125  $\mu\text{m}$  tip diameter (FHC Inc., Bowdoin, ME, USA) for transcallosal electrical stimulation in the left somatosensory cortex (location in mm: 1.5 mm posterior/1.5 mm lateral from bregma and 0.3 mm ventral from dura), ii) a glass capillary electrode (FHC Inc., Bowdoin, ME, USA) filled with 150 mM NaCl for AC-potential recordings in the right somatosensory cortex (0.5 mm anterior/1.5 mm lateral from bregma and 0.3 mm ventral from dura), iii) the  $\mu\text{PPPS}$  probe in the right somatosensory cortex (0.5 mm anterior/2.5 mm lateral from bregma and 0.4 mm ventral from dura. See Fig. 4 for a schematic representation. AC-potential signals were used to record local field potentials from the cortex ("electrocorticogram", ECoG), that were measured with respect to an Ag/AgCl reference electrode placed subcutaneously at the neck and amplified 10 $\times$  (Molecular Devices, Sunnyvale, CA, USA). Data were recorded and analysed off-line using LabChart Pro (AD Instruments). An external output of the ECoG signal was also connected to the second channel of the National Instruments myDAQ board and recorded simultaneously with output of the PMT (glutamate-related fluorescence signal). The trigger output used for electrical stimulation was recorded simultaneously with signals from the glass recording electrode and the PMT (fluorescence signal), enabling synchronization of the electrical stimulation, evoked cortical activity changes and glutamate measurements. After filling the system with a fresh enzyme mixture and prior to probe insertion, the MPF flow was established at 100  $\text{nL min}^{-1}$  to form a droplet around the tip and avoid immediate clogging of the probe by the extracellular matrix. After probe insertion, to a depth of 0.4 mm, the decane flow (500  $\text{nL min}^{-1}$ ), enzyme mixture flow (200  $\text{nL min}^{-1}$ ) and outlet pressure (-50 kPa) were applied.

After setting up this experiment, a recovery and baseline recording period of 15 minutes was observed, or longer in case signals had not stabilized. Next, transcallosal electrical stimulation was performed using the bipolar microelectrode on the left hemisphere (contralateral to the ECoG recording

side) and a battery-driven isolated WPI stimulator (A395 isolated WPI stimulator) while measuring ECoG and glutamate signals in the right hemisphere as indicated above. Single-pulse stimulation (biphasic 1 ms pulses, 500  $\mu\text{s}$  negative – 500  $\mu\text{s}$  positive, *i.e.* 'cathodic first') was performed at 70% of the intensity yielding a maximum cortical response (which was 3.15 mA for the two mice) at 0.1 Hz, 6 stimuli. The block of 6 stimuli was repeated 3 times, at 10 minute interval. Current output intensity was monitored separately by a custom-made isolated current meter. The 3.15 mA stimulation intensity used for assessing the transcallosal cortical responses corresponds to a current density of  $\sim 0.26 \mu\text{A} \mu\text{m}^{-2}$ , given an exposed stimulation electrode surface area of approximately 12 200  $\mu\text{m}^2$ .

The raw glutamate signal obtained from the PMT was processed using Matlab (version R2017a, Mathworks Inc., Natick, MA, USA), first to remove all data points below the baseline background fluorescence, second, to apply a moving averaging filter of 500 points (equal to 0.5 seconds). The AC ECoG signal recorded on the second channel of the MyDAQ was obtained from the glass electrode using a 3rd order Butterworth band-pass filter with cut-offs at 0.5 and 200 Hz.

## Results and discussion

### Probe fabrication

Photos of mounted probes are shown in Fig. 5A and D, while SEM pictures of the probe without its glass cover are shown in Fig. 5K–M.

Electrodes connected to the transport capillary to fuse droplets to prevent dispersion of smaller droplets can be seen in Fig. 5A–C. Microscopy images of key features of the probe are shown in Fig. 5G–I. The purple coloring in the microscopy images arises from the silicon oxide passivation layer inside the channels. In Fig. 5G and L, the tip of the needle with its sampling channels is shown. Fig. 5H and J shows an inset of the middle part of the needle.

The delivery channel of perfusion fluid to the tip can be seen at the left, with the T-shaped droplet generator on the right. The narrow channel leading towards the droplet generator from the bottom side of the image is the hydraulic resistor that prevents the continuous phase (decane) from entering the tip if small imbalances in flow rate occur. At the dialysate and continuous phase inlet, small filters made of rectangular pillars with 5  $\mu\text{m}$  spacing are included to prevent clogging of the probe as shown in Fig. 5I and K.

### Droplet formation

First the possibility to generate droplets was tested for the two probe types (see Fig. 5J for type 1, and 5E and F for type 2). In Fig. 5J, droplet formation is visualized for a probe with the droplet generator embedded in the needle tip. In this experiment, the flow rate of oil was set to 200  $\text{nL min}^{-1}$ , perfusion with demineralized water was set to 50  $\text{nL min}^{-1}$ , and the outlet pressure was set to -50 kPa. From the high-speed video (400 frames per second (fps)), it was estimated that droplets were generated at a rate of 80 fps (see Movie S7†).



For the second probe type, a push-pull probe is connected to an external droplet generator that enables the mixing of a sample containing the neurotransmitter glutamate with glutamate oxidase (GLOx), horseradish peroxidase (HRP) and a fluorophore precursor (Amplex Red) to obtain a glutamate concentration-dependent fluorescent signal. This arrangement is shown in Fig. 5D-F. Here, the flow rate of decane was set to  $500 \text{ nL min}^{-1}$ , the flow rate of the GLOx/HRP/Amp. Red mixture was set to  $200 \text{ nL min}^{-1}$ , perfusion of DI water was set to  $100 \text{ nL min}^{-1}$  and the outlet pressure was set to  $-50 \text{ kPa}$ . Droplets are generated on average at a rate of 1 drop per second (see Movie S8†).

As soon as the microdroplets generated in probe type 1 are transferred from the outlet capillary to the much wider transport capillary, they form a dense packing at first (see Fig. 5B), but due to the parabolic flow velocity profile, droplets moving along the centerline will be faster than those closer to the wall again causing dispersion. Therefore, measurements will get distorted because the order of the droplets moving through the optical detector will be different from the order in which they were generated. As a result, the temporal resolution goes down to  $\sim 1$  minute, as shown in Fig. S3A.† To solve this, a pair of electrodes was installed right after the outlet capillary in order to electrically merge microdroplets into larger ones that fit the dimensions of the transfer capillary, thereby forming a stable segmented flow and preserving the temporal resolution of the concentration variations in the sample (see Fig. 5C).

Based on separate measurements of the generation rate and volume of the small and merged droplets, it is estimated that about 50 droplets are merged into one bigger droplet (see Movies S9A and B†). Although the segmented flow was restored by this merging step, theoretically this modification will reduce the temporal resolution only by the ratio of the original and merged droplet generation rates, which is estimated to be a factor of 50.

### Optical detection

The uPPPS probes are designed to sample neurotransmitter from the brain, which will be detected using fluorescence

spectroscopy. Our optical setup is designed for resorufin detection, which is a fluorescent dye used in a variety of commercially available assay kits (see S2† for more details).

To demonstrate the possibility to detect resorufin at a sufficiently low concentration, the optical setup is first characterized with respect to the limit of detection for resorufin sodium salt. Fig. 6 shows a linear response in the lower concentration range of resorufin from 2 to 20 nM. Analysis of the recorded signal from the photomultiplier tube (PMT) reveals that the limit of detection (LOD) of this system can be estimated to be about 2 nM, which corresponds to 3 times the standard deviation of the blank value (see Fig. S2B†). Note that this LOD is reported for resorufin to determine the lower limitations of the optical setup. Basal glutamate levels in the brain are typically 3 orders of magnitude larger fluctuating around a few  $\mu\text{M}$ .<sup>38</sup>

### In vitro sampling

Next, the sampling capability of probe type 1 was verified using fluorescein sodium salt. An aqueous stock solution of 10 mM fluorescein was prepared and a titration experiment was done in a 10 mL sample cup with a magnetic stirrer (500 rpm). Flow rates of  $200 \text{ nL min}^{-1}$  decane,  $50 \text{ nL min}^{-1}$  perfusion with demineralized water were used, and the outlet pressure was set to  $-80 \text{ kPa}$ . Stepwise addition of  $200 \mu\text{L}$  of stock solution with intervals of 100 seconds resulted in the fluorescence intensity plot of Fig. 7A. LED power was set to  $300 \mu\text{W}$ .

For glutamate measurements using the type 2 probe, a commercial assay kit based on glutamate oxidase and the fluorescent marker resorufin was used. The design of the uPPPS probes does not allow for the addition of the enzymes and amplex red to the droplets, and adding these to the dialysate flow before the sampling area is undesirable for obvious reasons. Therefore, the push-pull probe with external droplet generator was fabricated, which allows one to add the components of the assay kit to the sample flow just before the droplet-generating T-junction. Analogous to the fluorescein titration experiment, a 10 mL sample cup was used with 1× PBS and a magnetic stirrer (500 rpm). A stepwise addition of  $20 \mu\text{L}$  of 1 mM L-glutamate with intervals of 100 seconds resulted in the fluorescence intensity plot of Fig. 7C. Flow rates were set to  $500 \text{ nL min}^{-1}$  decane,  $200 \text{ nL min}^{-1}$  GLOx/HRP/Amp. Red mixture,  $100 \text{ nL min}^{-1}$  perfusion with sterile perfusion fluid and an outlet pressure of  $-50 \text{ kPa}$  was used. A typical titration curve for fluorescein is indicated in Fig. 7A, and is representative of comparable curves obtained from a total of  $n = 3$  experiments (see S3B†). From this curve it can be seen that a sudden increase of approximately  $180 \mu\text{M}$  in concentration can be measured within 3 seconds. This delay of 3 seconds can be explained in part by the time needed for the added fluorescein to mix in the sample cup, and in part by the Taylor dispersion occurring in the needle between sampling channels and droplet generator. The relation between fluorescein concentration and fluorescent readout is calculated by an algorithm that cuts off any data-point below

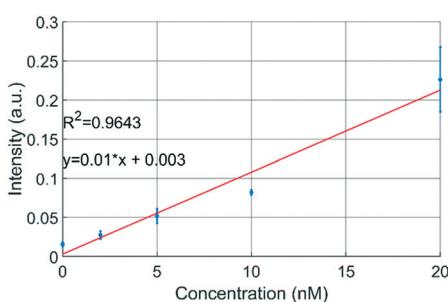
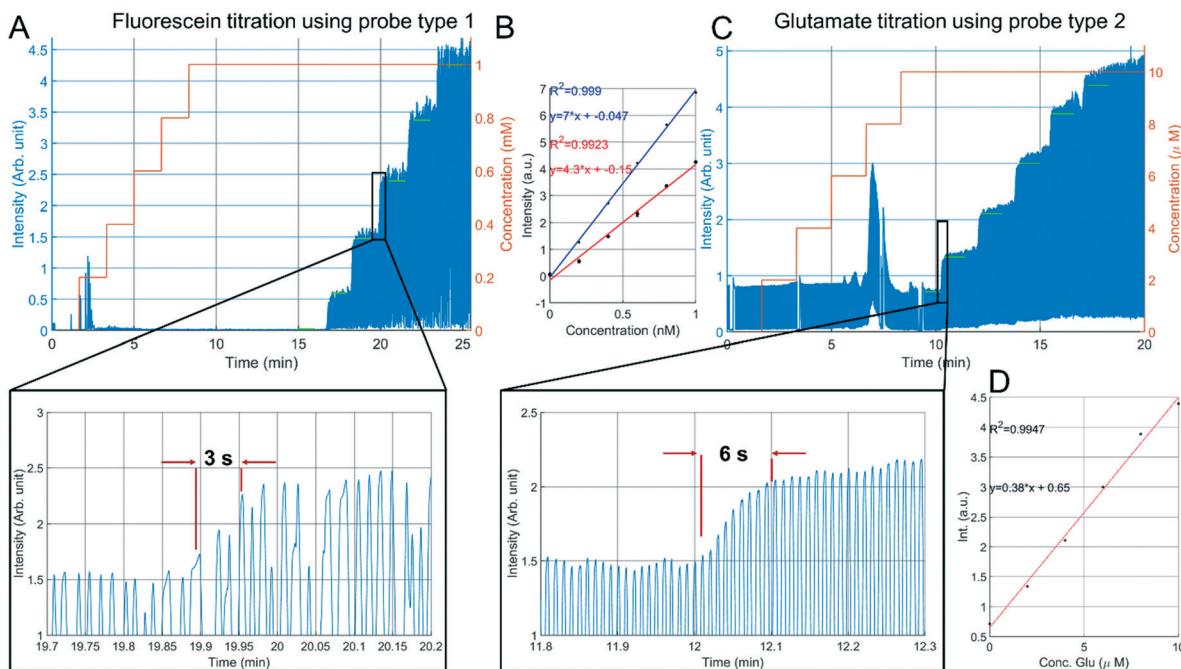


Fig. 6 Fluorescence measurements with resorufin sodium salt demonstrate that concentrations  $< 2 \text{ nM}$  can be measured using the developed optical detection setup (see text for details). Resorufin was injected directly into the optical detection capillary ( $n = 3$ ).





**Fig. 7** *In vitro* sampling from a 10 mL sample cup. A: Stepwise addition of fluorescein sampled with a probe type 1 having a droplet generator embedded in the needle. B: Relation between fluorescent readout and fluorescein concentration (red fitting curve) ( $n = 3$ ), extracted from the green plateaus from the titration curves of Fig. 7A and S3B.† The blue line indicates a calibration curve with fluorescein directly infused into the detection capillary. C: Stepwise addition of glutamate sampled with a probe type 2 having an external droplet generator in which glutamate oxidase, horseradish peroxidase and Amplex red are added. In the presence of glutamate, Amplex red is converted to highly fluorescent resorufin. The peaks at 2.5 min (Fig. 7A) and 7.5 min (Fig. 7B) are residual fluorophore resulting from earlier conducted staircase experiments. D: Relation between fluorescent readout and glutamate concentration, extracted from the green plateaus indicated in Fig. 7B.

a minimum threshold that is 80% of the maximum plateau value. Comparing the intensity values measured for the droplets ( $n = 3$ ) to a calibration curve recorded at the same LED power (see Fig. 7B), the recovery rate is estimated at 61%. However, as can be seen from the insets in Fig. 7, the fluorescent signal from the droplets never reach a plateau value. It is therefore not unlikely to assume that the real recovery rate might be even higher.

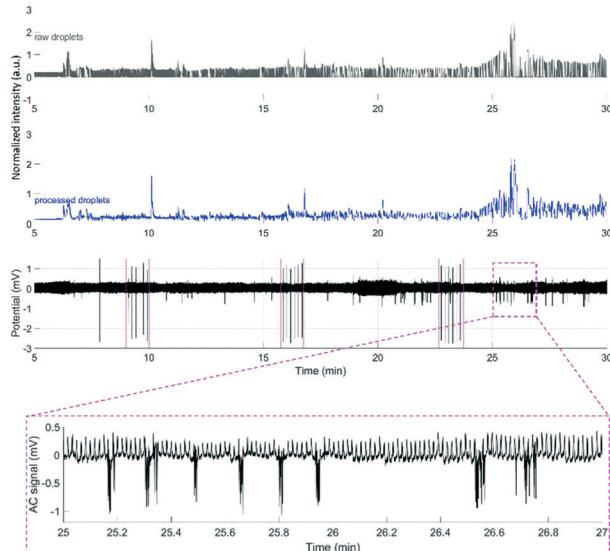
Glutamate titration using the probe type 2 (Fig. 7C) showed a slower response, roughly double the rise time observed with fluorescein. In addition to the time required for the glutamate to mix in the sample cup, the longer transfer distance between the tip of the probe and the external droplet generator causes some extra Taylor dispersion that adds to the total response time. Nonetheless, our results indicate that the compartmentalization in droplets are a significant improvement compared to the several minute time scale results shown using traditional dialysis probes combined with either flow injection analysis,<sup>20</sup> or fluorescent readout in continuous flow.<sup>21</sup>

The relation between glutamate concentration and fluorescent readout is calculated in an identical manner as for the fluorescein experiments. The algorithm cuts off any data-point below a minimum threshold that is 80% of the maximum plateau value. Next, the mean of each plateau (green lines in Fig. 7C) is calculated, as shown in Fig. 7D. The sensitivity is  $0.38 \text{ a.u. } \mu\text{M}^{-1}$ . Additional experiments ( $n = 2$  in S4†)

showed similar results, albeit with slightly different sensitivities of  $0.45$  and  $0.30 \text{ a.u. } \mu\text{M}^{-1}$ . Linearity was good with  $R^2$  values between  $0.9918$  and  $0.9969$ . A small offset is present at zero concentration, this could be due to known issues of self-formation of resorufin in the absence of glutamate with the Amplex Red assay kit.<sup>39</sup> Based on the calibration curves shown in Fig. 7D and S4† it is expected that our setup should be sensitive enough to detect both basal levels (few  $\mu\text{M}$ ) and bursts of glutamate ( $\sim 40 \mu\text{M}$ ) *in vivo*.<sup>24,38</sup> Compared to the fluorescein titrations, reproducibility was lower for experiments conducted with the glutamate assay kit, which was largely due to variable enzyme activities across (purchased) batches. The glutamate assay kit is sensitive to small changes in temperature during the droplet transfer, and to a lesser degree small variations in timing between the start of the reaction and the optical readout. Moreover, if resorufin is in direct contact with the walls of the channel adsorption can occur, leading to drift in the fluorescence intensity measurements.

### *In vivo* recordings

To demonstrate that the  $\mu$ PPPS probes are not only working in an *in vitro* environment, but are also functional in *in vivo* experiments, we used probe type 2 with external droplet generator to record changes in glutamate levels in the brain of an anesthetized mouse. As a proof of principle, such changes



**Fig. 8** Example recordings of glutamate during an *in vivo* experiment using the  $\mu$ PPPS probe with external droplet generator for measuring neurotransmitter glutamate changes during evoked cortical activity by electrical stimulation on the left sensorimotor cortex (periods including the stimulus pulses, visible on the ECoG trace, are indicated by magenta-colored dashed lines around 9, 17 and 23 minutes) in an anesthetized mouse. See Fig. 6 for a schematic of the experimental approach. Shown are the normalized raw optical droplet read-out (grey) from the right sensorimotor cortex, the processed droplet signal (blue), and alternating current (AC) (black) local field potential “electrocorticogram” (ECoG) recordings from a glass electrode inserted in superficial layers of the right sensorimotor cortex. For the first two stimulation sessions, the glutamate signal shows a transient increase directly following cortical stimulation, whereas in the third session the glutamate signal increased with a delay, in line with a pronounced burst of cortical activity that is evident on the AC ECoG recording (see additional inset showing details of the AC response at higher time resolution).

were recorded in the right sensorimotor cortex while cortical activity was evoked in the right sensorimotor cortex using transcallosal electrical stimulation in the left hemisphere<sup>40,41</sup> (see Fig. 8). It has been demonstrated by others that increased cortical network activity is paralleled by increased extracellular levels of glutamate *in vivo*.<sup>42,43</sup> Experiments were successfully performed in two mice, in which an electrical stimulation paradigm was repeated three times. Each stimulation paradigm consisted of a set of 6 biphasic 1 ms pulses at 70% of the maximum cortical response, delivered at 0.1 Hz. The resulting glutamate responses, as well as the recordings from a standard glass electrode are shown in Fig. 8. A simple staircase recording, similar to what is shown in Fig. 7 was conducted prior to insertion of the probe in the brain to check for proper functioning of the setup. From the delay observed between *in vitro* addition of glutamate and time of detection, the delay time was estimated to be approximately 4.25 minutes. This approximate delay time was used in *post hoc* analysis for shifting the droplet trace to the left (in case of the example in Fig. 8 by a shift of 4.25 minutes). Fig. 8 shows representative traces from one of the mouse experiments.

ments, with subsequent electrical stimulation of brain activity and related changes in fluorescent signal, as measured by the  $\mu$ PPPS probe, indicative of altered cortical glutamate levels. In the shown example, fluorescent signals from the sampled droplets increased following cortical electrical stimulation, either directly (at 10 and 17 min), or with a delay (25 min), in line with pronounced cortical burst activity (see inset). Observations are in line with expected rises in cortical glutamate levels during the evoked cortical activity.

## Conclusions and outlook

Here we have demonstrated the design and fabrication of a miniaturized push-pull perfusion sampling probe with a width of 140  $\mu$ m and a height of 140  $\mu$ m. Molecules (including neurotransmitters) enter the probe through 20 small sampling channels ( $20 \times 3 \mu\text{m}^2$ ) located at the end of the probe, effectively providing a sampling area of approximately 1200  $\mu\text{m}^2$ . A recirculating, U-shaped channel in the tip of the needle, operated in a push-pull-like fashion ensures fast transport of sample to the in-house developed external measurement setup. The detection method used for the *in vitro* and *in vivo* glutamate experiments in this study is based on fluorescence spectroscopy using a sensitive photomultiplier tube, and (in combination with  $\mu$ PPPS probes of type 2) the mixing of externally generated droplet samples with glutamate oxidase and fluorophore precursor Amplex Red. The detection limit for the fluorescent reaction product of the Amplex Red assay kit, resorufin, in our experiments was  $\sim 2 \text{ nM}$ . Immediately after sampling, the dialysate is encapsulated into water-in-decane droplets to prevent dispersion, and to enhance the temporal resolution. *In vitro* experiments using a solution containing fluorescein demonstrated a recovery rate of 61%, while step-wise changes in concentration could be measured within 3 seconds. As proof of principle, transient changes in brain glutamate levels were recorded *in vivo* in the cortex of anesthetized mice that were electrically stimulated on the contralateral cortical hemisphere. These experiments demonstrated a correlation between evoked cortical network activity and enhanced glutamate release, as indicated by rises in fluorescent signal.

Whereas our *in vivo* pilot data are encouraging, there is still a challenge with the  $\mu$ PPPS probe and fluorescent measurement approach. Small variations in droplet generation rate (in the order of a couple of seconds at most), which could occur due to technical changes in the fluidic system, will cause small fluctuations in the timing between brain activity changes and our detection of droplet signals, which can result in a larger cumulative variation due to the relatively long time between sampling and detection (4–5 minutes). This issue can be solved by detecting neurotransmitters immediately inside the body of the probe. This could be done optically (as we used in our study for detection outside of the probe), but, based on unpublished results from our team, it should also be possible to measure glutamate by electrochemical means inside droplets. In that scenario, inclusion



of a separate channel for calibration fluids would allow periodic recalibration of such electrochemical measurements. An additional benefit offered by switching to a different detection method is to prevent the issues observed with the glutamate assay kit obtaining reproducible activity, which make the glutamate assay kit not suitable for quantitative measurements.

In our current design we did not include additional functionality (*e.g.* measuring electrical activity) to the glass cover of the probe during fabrication. As we have shown in previous work,<sup>23</sup> it would be relatively straightforward to also include AC- and/or DC-EEG electrodes to create a hybrid probe that combines microfluidic sampling with brain activity recordings. Moreover, it would be feasible in the future to also include waveguides to enable optogenetic stimulation.

## Conflicts of interest

There are no conflicts to declare.

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## References

- 1 R. T. Kennedy, *Curr. Opin. Chem. Biol.*, 2013, **17**, 860–867.
- 2 T. I. F. H. Cremers, M. G. de Vries, K. D. Huinink, J. P. van Loon, M. v d Hart, B. Ebert, B. H. C. Westerink and E. C. M. De Lange, *J. Neurosci. Methods*, 2009, **178**, 249–254.
- 3 C. J. Watson, B. J. Venton and R. T. Kennedy, *Anal. Chem.*, 2006, **78**, 1391–1399.
- 4 N. A. Cellar, S. T. Burns, J. C. Meiners, H. Chen and R. T. Kennedy, *Anal. Chem.*, 2005, **77**, 7067–7073.
- 5 T. R. Slaney, J. Nie, N. D. Hershey, P. K. Thwar, J. Linderman, M. A. Burns and R. T. Kennedy, *Anal. Chem.*, 2011, **83**, 5207–5213.
- 6 S. Kottegoda, I. Shaik and S. A. Shippy, *J. Neurosci. Methods*, 2002, **121**, 93–101.
- 7 S. Qin, M. van der Zeyden, W. H. Oldenziel, T. I. F. H. Cremers and B. H. C. Westerink, *Sensors*, 2008, **8**, 6860–6884.
- 8 B. K. Day, F. Pomerleau, J. J. Burmeister, P. Huettl and G. A. Gerhardt, *J. Neurochem.*, 2006, **96**, 1626–1635.
- 9 J. J. Burmeister, K. Moxon and G. A. Gerhardt, *Anal. Chem.*, 2000, **72**, 187–192.
- 10 O. Frey, T. Holtzman, R. M. McNamara, D. E. H. Theobald, P. D. Van Der Wal, N. F. De Rooij, J. W. Dalley and M. Koudelka-hep, *Biosens. Bioelectron.*, 2010, **26**, 477–484.
- 11 Brainlink, <https://www.brainlink.nl/our-products/p/normal-cns-probes/1>, (accessed 23 November 2018).
- 12 R. T. Kennedy, J. E. Thompson and T. W. Vickroy, *J. Neurosci. Methods*, 2002, **114**, 39–49.
- 13 G. Taylor, *Proc. R. Soc. A*, 1953, **219**, 186–203.
- 14 G. Petit-Pierre, P. Colin, E. Laurer, J. Déglon, A. Bertsch, A. Thomas, B. L. Schneider and P. Renaud, *Nat. Commun.*, 2017, **8**, 1239.
- 15 T. Ngernsutivorakul, D. J. Steyer, A. C. Valenta and R. T. Kennedy, *Anal. Chem.*, 2018, **90**, 10943–10950.
- 16 M. Wang, T. Slaney, O. Mabrouk and R. T. Kennedy, *J. Neurosci. Methods*, 2010, **190**, 39–48.
- 17 M. Wang, N. D. Hershey, O. S. Mabrouk and R. T. Kennedy, *Anal. Bioanal. Chem.*, 2011, **400**, 2013–2023.
- 18 P. Song, N. D. Hershey, O. S. Mabrouk, T. R. Slaney and R. T. Kennedy, *Anal. Chem.*, 2012, **84**, 4659–4664.
- 19 S. Feng, G. Liu, L. Jiang, Y. Zhu, E. M. Goldys and D. W. Inglis, *Appl. Phys. Lett.*, 2017, **111**, 183702.
- 20 J. B. Gramsbergen, J. Skjøth-Rasmussen, C. Rasmussen and K. L. Lambertsen, *J. Neurosci. Methods*, 2004, **140**, 93–101.
- 21 A. Morales-Villagrán, K. Pardo-Peña, L. Medina-Ceja and S. López-Pérez, *J. Neurochem.*, 2016, **139**, 886–896.
- 22 W. H. Lee, T. R. Slaney, R. W. Hower and R. T. Kennedy, *Anal. Chem.*, 2013, **85**, 3828–3831.
- 23 M. Odijk, E. J. J. van der Wouden, W. Olthuis, M. D. D. Ferrari, E. A. A. Tolner, A. M. J. M. M. J. M. van den Maagdenberg and A. Van Den Berg, *Sens. Actuators, B*, 2015, **207**, 945–953.
- 24 K. N. Hascup, E. R. Hascup, F. Pomerleau, P. Huettl and G. A. Gerhardt, *J. Pharmacol. Exp. Ther.*, 2008, **324**, 725LP–731.
- 25 W. H. Lee, T. Ngernsutivorakul, O. S. Mabrouk, J. T. Wong, C. E. Dugan, S. S. Pappas, H. J. Yoon and R. T. Kennedy, *Anal. Chem.*, 2016, **88**, 1230–1237.
- 26 Y. Kim, H. J. Lee, D. Kim, Y. K. Kim, S. H. Lee and E. Yoon, in *Transducers conference*, 2013, pp. 876–879.
- 27 R. Rathnasingham, D. R. Kipke, S. C. Bledsoe and J. D. McLaren, *IEEE Trans. Biomed. Eng.*, 2004, **51**, 138–145.
- 28 M. Perry, Q. Li and R. T. Kennedy, *Anal. Chim. Acta*, 2009, **653**, 1–22.
- 29 E. R. Hascup, K. N. Hascup, M. Stephens, F. Pomerleau, P. Huettl, A. Gratton and G. A. Gerhardt, *J. Neurochem.*, 2010, **115**, 1608–1620.
- 30 M. D. Johnson, R. K. Franklin, M. D. Gibson, R. B. Brown and D. R. Kipke, *J. Neurosci. Methods*, 2008, **174**, 62–70.
- 31 J. Voipio, M. Pasternack and K. Macleod, in *Microelectrode techniques, The Plymouth workshop handbook*, ed. D. Ogden, The company of Biologists, Cambridge UK, 2nd edn, 1994, pp. 275–316.
- 32 E. Bakker, P. Bühlmann and E. Pretsch, *Chem. Rev.*, 1997, **97**, 3083–3132.



33 J. P. Seymour, F. Wu, K. D. Wise and E. Yoon, *Microsyst. Nanoeng.*, 2017, **3**, 16066.

34 F. van den Brink, A. Asthana, J. Bomer, E. Tolner, A. van den Maagdenberg and M. Odijk, *21st Int. Conf. Miniaturized Syst. Chem. Life Sci.*, 2017, pp. 215–216.

35 A. C. F. Ribeiro, M. M. Rodrigo, M. C. F. Barros, L. M. P. Veríssimo, C. Romero, A. J. M. Valente and M. A. Esteso, *J. Chem. Thermodyn.*, 2014, **74**, 133–137.

36 L. N. Mizerovsky and K. P. Smirnova, *Russ. Chem. Bull.*, 2010, **59**, 673–676.

37 A. M. A. Dias, R. P. Bonifácio, I. M. Marrucho, A. A. H. Pádua and M. F. Costa Gomes, *Phys. Chem. Chem. Phys.*, 2003, **5**, 543–549.

38 J. J. Burmeister and G. A. Gerhardt, *Anal. Chem.*, 2001, **73**, 1037–1042.

39 B. Zhao, F. A. Summers and R. P. Mason, *Free Radical Biol. Med.*, 2012, **53**, 1080–1087.

40 G. Tsenov and P. Mareš, *Physiol. Res.*, 2007, **56**, 485–491.

41 E. A. Tolner, D. W. Hochman, P. Hassinen, J. Otáhal, E. Gaily, M. M. Haglund, H. Kubová, S. Schuchmann, S. Vanhatalo and K. Kaila, *Epilepsia*, 2011, **52**, 104–114.

42 M. H. Millan, T. P. Obrenovitch, G. S. Sarna, S. Y. Lok, L. Symon and B. S. Meldrum, *Epilepsy Res.*, 1991, **9**, 86–91.

43 L. Medina-Ceja, K. Pardo-Peña, A. Morales-Villagrán, J. Ortega-Ibarra and S. López-Pérez, *BMC Neurosci.*, 2015, **16**(1), 11.

