

Cite this: *Chem. Sci.*, 2015, **6**, 4610

Oxalyl amide assisted palladium-catalyzed synthesis of pyrrolidones *via* carbonylation of γ -C(sp³)-H bonds of aliphatic amine substrates†

Chao Wang, Li Zhang, Changpeng Chen, Jian Han, Yingming Yao* and Yingsheng Zhao*

The first Pd-catalyzed regioselective γ -carbonylation of oxalyl amide protected aliphatic amines with carbon monoxide leading to synthesis of pyrrolidones has been developed. Both γ -methyl and cyclopropyl methylene C-H bonds are well activated to obtain the corresponding pyrrolidones in moderate to excellent yields. The role of 3-(trifluoromethyl)benzoic acid as an additive is critical as it helps in stabilizing the palladium intermediate formed during the catalytic cycle. The reaction scope is extended to benzylamine and allyl amine derivatives, thereby affording the corresponding products in good to excellent yields.

Received 10th February 2015
Accepted 16th May 2015

DOI: 10.1039/c5sc00519a
www.rsc.org/chemicalscience

Introduction

Recently, the development of practical methods for the construction of synthetically useful building blocks through direct C-H functionalization has emerged as an attractive and fruitful area in modern organic chemistry.¹ Among these methods, the transition-metal-catalyzed carbonylation of aryl C(sp²)-H bonds using CO as a carbon atom source to form an easily transformable carbonyl group has experienced landmark development.^{2–5} The first example was in the 1950s when Murahashi reported the cobalt-catalysed direct C-H functionalization/carbonylation of arenes with CO.⁶ In 1980, Fujiwara *et al.* also described a protocol involving Pd(OAc)₂-catalyzed direct carboxylation of some aromatic compounds with CO (15 atm) in an autoclave to give the corresponding carboxylic acids in poor to moderate yields.⁷ However, large excess of the arenes and lack of regioselectivity has hampered further application of this method in organic synthesis.

In the last decades, the challenge of selectivity in carbonylation of C(sp²)-H bonds has been overcome by the employment of directing groups,⁸ or where regioselectivity could also be achieved based on the substrates own intrinsic reactivity, *e.g.* amines,^{4d,8d,9} carboxylics,¹⁰ phenols,¹¹ and alcohols.¹² In spite of these great developments in the carbonylation of arenes, there is still limited success in the direct functionalization of

C(sp³)-H bonds with CO. In 2011, Chatani *et al.* described the Ru(0)-catalyzed regioselective carbonylation of unactivated C(sp³)-H bonds of aliphatic amides using bidentate directing groups (Scheme 1A). Similarly, Yu *et al.* reported the synthesis of succinimides *via* a Pd(II)-catalyzed β -carbonylation of aliphatic amide with CO (Scheme 1B).¹³ Recently, Gaunt *et al.* described a Pd-catalyzed carbonylation of secondary aliphatic amines at the β -position *via* a rarely reported four-membered palladacycle, leading to β -lactams in moderate to good yields (Scheme 1C).¹⁴ To the best of our knowledge, only these two examples of Pd-catalyzed carbonylation of C(sp³)-H bonds at specific positions have been reported. We speculated that during the catalytic cycle, Pd(II) might be easily reduced to Pd(0), thereby undergoing β -hydride elimination, and that there could be a competitive coordination with the palladium centre between CO and the specific C-H bond under CO atmosphere.¹⁵ For these reasons, the direct carbonylation of C(sp³)-H bonds remained a significant challenge.

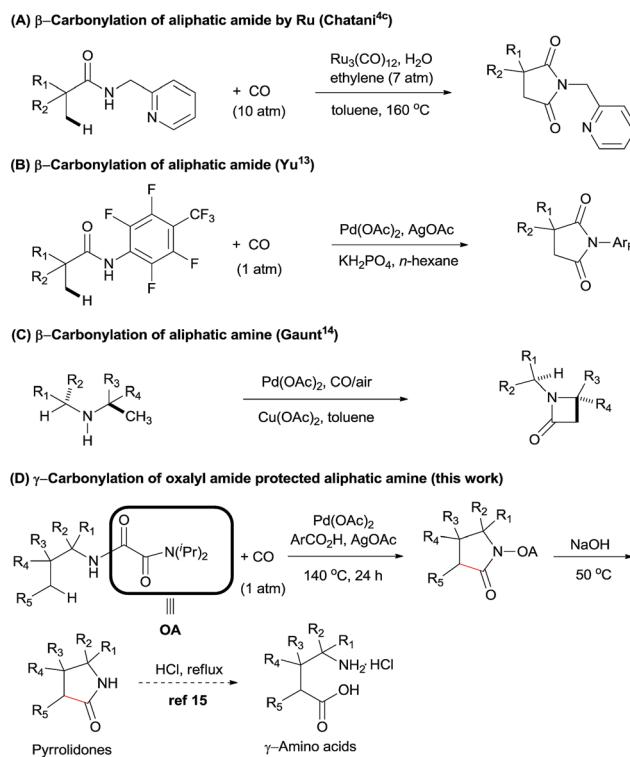
In 2005, Daugulis and co-workers reported the first use of picolinamide and aminoquinoline auxiliaries as bidentate directing groups in Pd-catalyzed regioselective functionalization of C(sp³)-H bonds.¹⁶ Later, several groups demonstrated that these groups have an immense ability for enabling a wide variety of C-H transformation.¹⁷ Inspired by these results, our group has developed another bidentate directing group, oxalyl amide, which could be easily synthesized from oxalyl chloride and diisopropylamine *via* an S_N type reaction. We found that it could be used as a powerful directing group in enabling a variety of C-H transformations for both unactivated C(sp³)-H and C(sp²)-H bonds.¹⁸

In order to expand the reaction scope of oxalyl amide assisted C-H functionalization and explore its potential application towards transformation not achievable by other auxiliary-

Key Laboratory of Organic Synthesis of Jiangsu Province College of Chemistry, Chemical Engineering and Materials Science Soochow University, Suzhou 215123, PR China. E-mail: yszhao@suda.edu.cn

† Electronic supplementary information (ESI) available: Experimental procedures, characterization data for all new compounds. CCDC 1048649. For ESI and crystallographic data in CIF or other electronic format see DOI: [10.1039/c5sc00519a](https://doi.org/10.1039/c5sc00519a)



Scheme 1 Carbonylation of $\text{C}(\text{sp}^3)\text{-H}$ Bonds.

assisted C-H functionalization, we therefore proceeded to address the unreported γ -carbonylation of aliphatic amine substrates in a bid to synthesize important structural motifs of

pyrrolidones.¹⁹ Herein, we report a novel protocol for the preparation of pyrrolidones by Pd(II)-catalyzed γ -carbonylation of oxaryl amide protected aliphatic amine substrates. Both γ -methyl and cyclopropyl methylene C-H bonds were well activated and gave the corresponding pyrrolidones in moderate to excellent yields. Furthermore, pyrrolidones could be transformed into synthetically useful γ -amino acids under acidic conditions,²⁰ which offers a potential route towards synthesizing unnatural amino acids (Scheme 1D). The substrate scope could also be further extended to oxaryl amide protected benzyl amine and allyl amine derivatives, affording the corresponding valuable synthons in good to excellent yields.

Results and discussion

We started our investigation with the reaction of oxaryl amide protected **1a** in the presence of $\text{Pd}(\text{OAc})_2$ catalyst and silver acetate as oxidant under atmospheric pressure of CO at 140°C in mesitylene for 24 h. Gratifyingly, the desired pyrrolidone **2a** was obtained in 22% yield. A variety of additives, such as AcOH, PivOH, $(^n\text{BuO})_2\text{PO}_2\text{H}$, Ac-Gly-OH, which are well known to improve the reactivity of a palladium center, were applied to this reaction,²¹ with all of them displaying different levels of positive effect on the product yield. Pivalic acid greatly aided the conversion of **1a**, affording **2a** in 57% yield, but benzoic acid had a better promoting effect. However, further screening revealed that 3-(trifluoromethyl)benzoic acid was the most effective additive, with an excellent conversion of **1a** giving 85% yield of **2a**. During optimization studies, reactions with various different additives proceeded cleanly and it was observed that

Table 1 Optimization of palladium-catalyzed carbonylation of $\gamma\text{-C}(\text{sp}^3)\text{-H}$ bonds

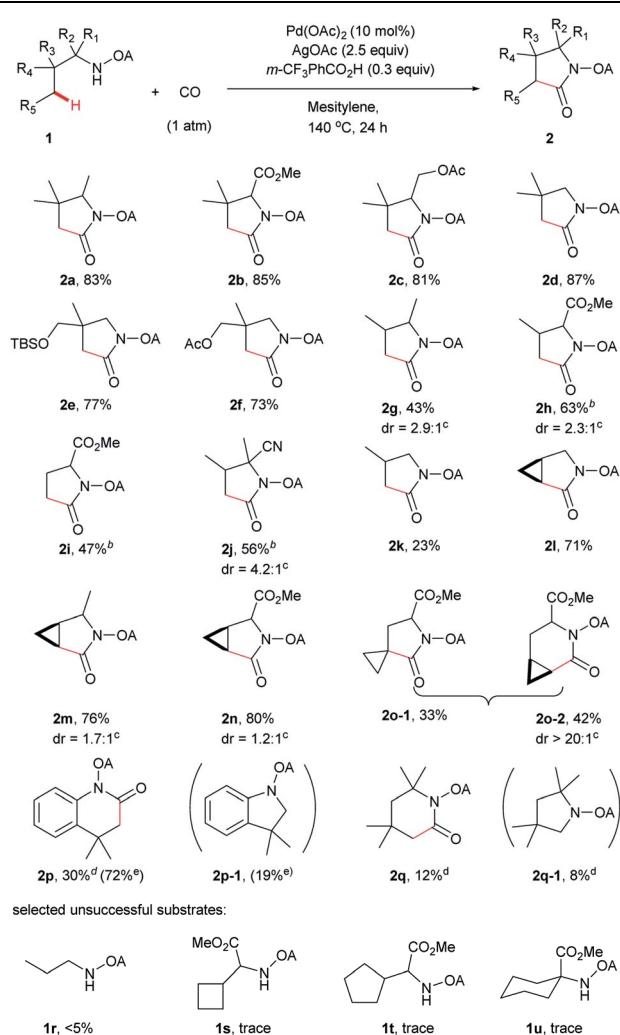
Entry	$\text{Pd}(\text{OAc})_2$ (mol%)	Oxidant (2.5 equiv.)	Additive (0.3 equiv.)	Conv ^a (%)	Yield ^{a,b} (%)
1	10	AgOAc		26	22
2	10	AgOAc	AcOH	21	19
3	10	AgOAc	PivOH	60	57
4	10	AgOAc	$(^n\text{BuO})_2\text{PO}_2\text{H}$	66	62
5	10	AgOAc	Ac-Gly-OH	12	8
6	10	AgOAc	PhCO_2H	76	75
7	10	AgOAc	<i>o</i> -PhPhCO ₂ H	27	18
8	10	AgOAc	<i>m</i> -CF _{3PhCO₂H}	87	85(83)
9	10	Ag_2CO_3	<i>m</i> -CF ₃ PhCO ₂ H	21	20
10	10	$\text{Cu}(\text{OAc})_2$	<i>m</i> -CF ₃ PhCO ₂ H	2	0
11	10	BQ	<i>m</i> -CF ₃ PhCO ₂ H	5	4
12	0	AgOAc	<i>m</i> -CF ₃ PhCO ₂ H	2	0

Reaction conditions: **1a** + CO (1 atm) $\xrightarrow[\text{Mesitylene, } 140^\circ\text{C, 24 h}]{\text{Pd}(\text{OAc})_2, \text{ oxidant, additive}}$ **2a**.

Yield data for **2a**:

- Entry 1: 22%
- Entry 2: 19%
- Entry 3: 57%
- Entry 4: 62%
- Entry 5: 8%
- Entry 6: 75%
- Entry 7: 18%
- Entry 8: 85(83)%
- Entry 9: 20%
- Entry 10: 0%
- Entry 11: 4%
- Entry 12: 0%

^a Reactions were carried out at 0.2 mmol scale under CO (1 atm), using mesitylene (0.3 mL) as the solvent; conversion and yield were based on GC using tridecane as the internal standard. ^b Isolated yield in parentheses. Ac = acetyl, Gly = glycine.

Table 2 Palladium-catalyzed carbonylation of γ -C(sp³)-H bonds^a

^a Yields based on isolated products on a 0.2 mmol scale under CO (1 atm), using mesitylene (0.3 mL) as the solvent. ^b 15 mol% Pd(OAc)₂ was used. ^c Determined by H NMR analysis. ^d At 100 °C. ^e 1 equiv. Pd(OAc)₂, without AgOAc.

Pd(II) was reduced to Pd(0) during the reaction thus rationalizing the low efficiency of some of these additives (Table 1, entries 2–8). Although the reason behind the promoting effect of 3-(trifluoromethyl)benzoic acid was not clear, one plausible explanation was that the palladium(0) stabilized by the electron-deficient arene during the catalytic cycle, could be further reoxidized to Pd(II). Meanwhile, it played the same role as pivalic acid during the C–H activation step.²² Other oxidants such as Cu(OAc)₂ or BQ were not effective in this transformation. Notably, alternative directing groups or protecting groups, such as Cbz, triflamide and picolinamide were also tested under optimized conditions. However, none could give the corresponding carbonylated products in more than 5% yield analyzed by LC–MS.

With the optimized conditions in hand, reactions involving a series of oxaryl amide protected propylamine derivatives were

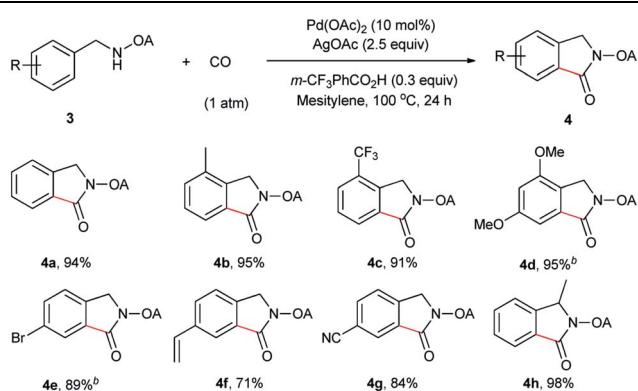
carried out. The carbonylation of substrates bearing two β -substituents proceeded smoothly to provide the corresponding pyrrolidines in good to excellent yields (Table 2, 2b–f). Ester derivatives were tolerated, giving the corresponding products in good yields (2b,c,f,h,i,n,o). Substrates 1g–k, bearing only one β methyl group, were less reactive and acceptable yields could only be obtained by slightly modifying the reaction conditions. Delightfully, substrate 1j, with an α -substituted cyano group, which is a highly valuable synthon in synthetic chemistry, also gave the desired carbonylated product 2j in 56% yield. The methylene C–H bonds in cyclopropane rings were all carbonylated giving good yields of products (2l–n). Interestingly, substrate 1o gave a mixture of five and six-membered carbonylated products in a total of 75% yield (2o), which could be easily separated by silica gel chromatography. Inspired by these result, substrate 1p was also tested, but a lower yield of product than expected was recovered, together with unreacted starting material. It was thought that the probable competitive carbonylation reaction happening at the aromatic ring was not detected by NMR and GC–MS.²³ However, when one equivalent of Pd(OAc)₂ was used, the δ -carbonylated product 2p was obtained in 72% yield, along with 19% yield of intramolecular aminated product 2p-1. Unfortunately, when the same reaction was attempted on substrate 1q, only 12% of the corresponding carbonylated product 2q could be isolated along with 8% of the product from intramolecular amination. Regrettably, the use of one equivalent of Pd(OAc)₂ as in the case of substrate 1p, did nothing to improve the yield of carbonylation and intramolecular amination products. However, it can be inferred from these results that an unstable alkyl seven-membered palladacycle²⁴ might have been formed during the catalytic cycle, which could open doors for the development of newer transformation. A series of other oxaryl amide protected amines (1r–u) were also tested, all failing to give acceptable yields.

Despite the great development in palladium-catalyzed C(sp²)-H activation/carbonylation,²⁵ there have been only few examples on the direct preparation of benzolactams from benzylamines substrates assisted by directing groups. Encouraged by the results obtained above, oxaryl amide protected benzylamines were also examined, and representative data were listed in Table 3. A wide variety of functional groups, such as CF₃, OMe, Br, vinyl, cyano, were compatible with this protocol, affording the corresponding products in good to excellent yields. Both electron-donating and electron-withdrawing substituents on the phenyl ring had little or no effect on the yield of products. Interestingly, substrates 3d and 3e needed slightly higher temperature for complete transformation.

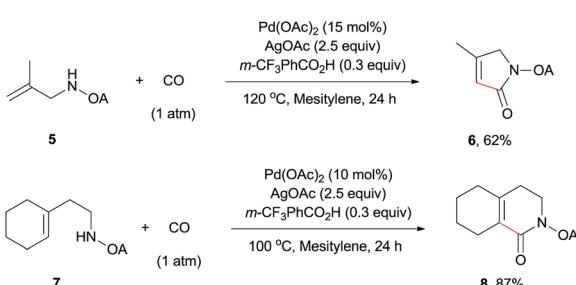
The scope of γ -carbonylation of oxaryl amide protected primary amine could be further extended to allylamine substrates. Substrates 5 and 7 smoothly reacted with CO to give the corresponding carbonylated products 6 and 8 in acceptable yield respectively, which could be used as synthons for a variety of further elaborations (Scheme 2).²⁶

To demonstrate the powerful use of this protocol, a gram scale reaction involving substrate 1a was carried out and the corresponding pyrrolidone was isolated in 72% yield (Scheme 3). The oxaryl amide directing group could be removed

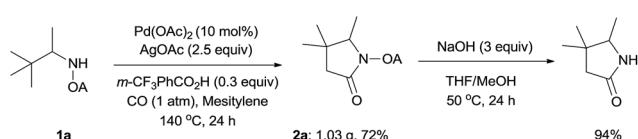


Table 3 Palladium-catalyzed carbonylation of γ -C(sp²)-H bonds^a

^a Yields based on isolated products on a 0.2 mmol scale under CO (1 atm), using mesitylene (0.3 mL) as the solvent. ^b At 120 °C.



Scheme 2 Carbonylation of allylamine reaction.



Scheme 3 Scaling up and auxiliary removal.

under basic conditions at 50 °C, thus underlining the potential utility of this newly developed synthetic method.

Conclusion

In conclusion, we have developed the first Pd-catalyzed γ -carbonylation of oxalyl amide protected aliphatic amine substrates by using oxalyl amide as a bidentate directing group with AgOAc, and ArCO₂H under carbon monoxide atmosphere. This method is a novel strategy for the preparation of pyrrolidones. The substrate scope can be extended to benzylamine and allyl amine derivatives, giving products in good to excellent yields. The use of 3-(trifluoromethyl)benzoic acid is important for this transformation, and a detailed mechanistic study is now being undertaken in our laboratory.

Acknowledgements

We gratefully acknowledge financial support from the Natural Science Foundation of Jiangsu Province of China (BK20130294), a start-up fund (Q410901212) from Soochow University and the Young National Natural Science Foundation of China (No. 2140213321402133) for support of this work. The PAPD and the Qing Lan Project are also gratefully acknowledged.

Notes and references

- For recent reviews: (a) K. Kakiuchi and S. Murai, *Acc. Chem. Res.*, 2002, **35**, 826; (b) K. Godula and D. Sames, *Science*, 2006, **312**, 67; (c) R. Giri, B.-F. Shi, K. M. Engle, N. Maugel and J.-Q. Yu, *Chem. Soc. Rev.*, 2009, **38**, 3242; (d) T. W. Lyons and M. S. Sanford, *Chem. Rev.*, 2010, **110**, 1147; (e) D. A. Colby, R. G. Bergman and J. A. Ellman, *Chem. Rev.*, 2010, **110**, 624; (f) L. McMurray, F. O'Hara and M. J. Gaunt, *Chem. Soc. Rev.*, 2011, **40**, 1885; (g) C. S. Yeung and V. M. Dong, *Chem. Rev.*, 2011, **111**, 1215; (h) B.-J. Li and Z.-J. Shi, *Chem. Soc. Rev.*, 2012, **41**, 5588; (i) J. J. Mousseau and A. B. Charette, *Acc. Chem. Res.*, 2012, **46**, 412; (j) K. M. Engle, T.-S. Mei, M. Wasa and J.-Q. Yu, *Acc. Chem. Res.*, 2012, **45**, 788; (k) P. B. Arockiam, C. Bruneau and P. H. Dixneuf, *Chem. Rev.*, 2012, **112**, 5879; (l) G. Rouquet and N. Chatani, *Angew. Chem., Int. Ed.*, 2013, **52**, 11726; (m) J. L. Roizen, M. E. Harvey and J. D. Bois, *Acc. Chem. Res.*, 2012, **45**, 911; (n) L. Ackermann, *Acc. Chem. Res.*, 2014, **47**, 281.
- (a) *Handbook of Organopalladium Chemistry for Organic Synthesis*, ed. E.-I. Negishi, Wiley-Interscience, New York, 2002; (b) R. Jazzaar, J. Hitce, J. Sofack-Kreutzer and O. Baudoin, *Chem.-Eur. J.*, 2010, **16**, 2654; (c) Y. Zhao, L. Jin and A. Lei, *J. Am. Chem. Soc.*, 2008, **130**, 9429; (d) R. H. Munday, J. R. Martinelli and S. L. Buchwald, *J. Am. Chem. Soc.*, 2008, **130**, 2754; (e) C. F. J. Barnard, *Organometallics*, 2008, **27**, 5402; (f) A. Brennfuhrer, H. Neumann and M. Beller, *Angew. Chem., Int. Ed.*, 2009, **48**, 4114.
- For select examples of Pd-catalyzed carbonylation: (a) J. Schranck, M. Burhardt, C. Bornschein, H. Neumann, T. Skrydstrup and M. Beller, *Chem.-Eur. J.*, 2014, **20**, 9534; (b) J. Chen, K. Natte, A. Spannenberg, H. Neumann, P. Langer, M. Beller and X.-F. Wu, *Angew. Chem., Int. Ed.*, 2014, **53**, 7579; (c) L. Wang, Y. Wang, C. Liu and A. Lei, *Angew. Chem., Int. Ed.*, 2014, **53**, 5657; (d) P. Xie, Y. Xie, B. Qian, H. Zhou, C. Xia and H. Huang, *J. Am. Chem. Soc.*, 2012, **134**, 9902; (e) H. Zhang, R. Shi, P. Gan, C. Liu, A. Ding, Q. Wang and A. Lei, *Angew. Chem., Int. Ed.*, 2012, **51**, 5204; (f) H. Li, G.-X. Cai and Z.-J. Shi, *Dalton Trans.*, 2010, 10442.
- (a) M. Tobisu, N. Chatani, T. Asaumi, K. Amako, Y. Ie, Y. Fukumoto and S. Murai, *J. Am. Chem. Soc.*, 2000, **122**, 12663; (b) S. Inoue, H. Shiota, Y. Fukumoto and N. Chatani, *J. Am. Chem. Soc.*, 2009, **131**, 6898; (c) N. Hasegawa, V. Charra, S. Inoue, Y. Fukumoto and





N. Chatani, *J. Am. Chem. Soc.*, 2011, **133**, 8070; (d) Y. Du, T. K. Hyster and T. Rovis, *Chem. Commun.*, 2011, **47**, 12074.

5 (a) Z.-H. Guan, Z.-H. Ren, S. M. Spinella, S. Yu, Y.-M. Liang and X. Zhang, *J. Am. Chem. Soc.*, 2009, **131**, 729; (b) I. S. noue, H. Shiota, Y. Fukumoto and N. Chatani, *J. Am. Chem. Soc.*, 2009, **131**, 6898.

6 S. Murahashi, *J. Am. Chem. Soc.*, 1955, **77**, 6403.

7 (a) Y. Fujiwara, T. Kawauchi and H. Taniguchi, *J. Chem. Soc., Chem. Commun.*, 1980, 220; (b) Y. Fujiwara, K. Takaki and Y. Taniguchi, *Synlett*, 1996, 591.

8 (a) W. Y. Yu, W. N. Sit, K. M. Lai, Z. Y. Zhou and A. S. C. Chan, *J. Am. Chem. Soc.*, 2008, **130**, 3304; (b) R. Giri, J. K. Lam and J.-Q. Yu, *J. Am. Chem. Soc.*, 2010, **132**, 686; (c) V. Rajeshkumar, T.-H. Lee and S.-C. Chuang, *Org. Lett.*, 2013, **15**, 1468; (d) L. Grigorjeva and O. Daugulis, *Org. Lett.*, 2014, **16**, 4688; (e) S. Li, G. Chen, C.-G. Feng, W. Gong and J.-Q. Yu, *J. Am. Chem. Soc.*, 2014, **136**, 5267.

9 (a) B. Haffemayer, M. Gulias and M. J. Gaunt, *Chem. Sci.*, 2011, **2**, 312; (b) B. Lopez, A. Rodriguez, D. Santos, J. Albert, X. Ariza, J. Garcia and J. Granell, *Chem. Commun.*, 2011, **47**, 1054.

10 (a) R. Giri and J.-Q. Yu, *J. Am. Chem. Soc.*, 2008, **130**, 14082; (b) B. R. Rosen, L. R. Simke, P. S. Thuy-Boun, D. D. Dixon, J.-Q. Yu and P. S. Baran, *Angew. Chem., Int. Ed.*, 2013, **52**, 7317.

11 S. Luo, F.-X. Luo, X.-S. Zhang and Z.-J. Shi, *Angew. Chem., Int. Ed.*, 2013, **52**, 10598.

12 Y. Lu, D. Leow, X. Wang, K. M. Engle and J.-Q. Yu, *Chem. Sci.*, 2011, **2**, 967.

13 E. J. Yoo, M. Wasa and J.-Q. Yu, *J. Am. Chem. Soc.*, 2010, **132**, 17378.

14 A. McNally, B. Haffemayer, B. L. Collins and M. J. Gaunt, *Nature*, 2014, **510**, 129.

15 (a) T. Nishiguchi, K. Nakata, K. Takaki and Y. Fujiwara, *Chem. Lett.*, 1992, **21**, 1141; (b) K. Nakata, Y. Yamaoka, T. Miyata, Y. Taniguchi, K. Takaki and Y. Fujiwara, *J. Organomet. Chem.*, 1994, **473**, 32.

16 V. G. Zaitsev, D. Shabashov and O. Daugulis, *J. Am. Chem. Soc.*, 2005, **127**, 13154.

17 For select references: (a) D. Shabashov and O. Daugulis, *J. Am. Chem. Soc.*, 2010, **132**, 3965; (b) Y. Feng, Y. Wang, B. Landgraf, S. Liu and G. Chen, *Org. Lett.*, 2010, **12**, 3414; (c) Y. Zhao and G. Chen, *Org. Lett.*, 2011, **13**, 4850; (d) G. He, C. Lu, Y. Zhao, W. A. Nack and G. Chen, *Org. Lett.*, 2012, **14**, 2944; (e) S.-Y. Zhang, Q. Li, G. He, W. A. Nack and G. Chen, *J. Am. Chem. Soc.*, 2013, **135**, 2124; (f) Y. Ano, M. Tobisu and N. Chatani, *Org. Lett.*, 2012, **14**, 354; (g) F.-R. Gou, X.-C. Wang, P.-F. Huo, H.-P. Bi, Z.-H. Guan and Y.-M. Liang, *Org. Lett.*, 2009, **11**, 5726; (h) B. V. S. Reddy, L. R. Reddy and E. J. Corey, *Org. Lett.*, 2006, **8**, 3391.

18 (a) C. Wang, C. Chen, J. Zhang, J. Han, Q. Wang, K. Guo, P. Liu, M. Guan, Y. Yao and Y. Zhao, *Angew. Chem., Int. Ed.*, 2014, **53**, 9884; (b) Q. Wang, J. Han, C. Wang, J. Zhang, Z. Huang, D. Shi and Y. Zhao, *Chem. Sci.*, 2014, **5**, 4962; (c) J. Han, P. Liu, C. Wang, Q. Wang, J. Zhang, Y. Zhao, D. Shi and Y. Zhao, *Org. Lett.*, 2014, **16**, 5682.

19 (a) G. He, S.-Y. Zhang, W. A. Nack, Q. Li and G. Chen, *Angew. Chem., Int. Ed.*, 2013, **52**, 11124; (b) C. V. Galliford and K. A. Scheidt, *Angew. Chem., Int. Ed.*, 2007, **46**, 8748; (c) D. E. A. Raup, B. Cardinal-David, D. Holte and K. A. Scheidt, *Nat. Chem.*, 2010, **2**, 766; (d) D. Q. Tan, K. S. Martin, J. C. Fettinger and J. Shaw, *Proc. Natl. Acad. Sci. U. S. A.*, 2011, **108**, 6781.

20 (a) S. Rousseaux, B. Liegault and K. Fagnou, *Chem. Sci.*, 2012, **3**, 244; (b) T. Saget and N. Cramer, *Angew. Chem., Int. Ed.*, 2012, **51**, 12842; (c) J.-X. Yan, H. Li, X.-W. Liu, J.-L. Shi, X. Wang and Z.-J. Shi, *Angew. Chem., Int. Ed.*, 2014, **53**, 4945.

21 (a) M. Lafrance and K. Fagnou, *J. Am. Chem. Soc.*, 2006, **128**, 16496; (b) D.-H. Wang, K. M. Engle, B.-F. Shi and J.-Q. Yu, *Science*, 2010, **327**, 315; (c) S.-Y. Zhang, G. He, W. A. Nack, Y. Zhao, Q. Li and G. Chen, *J. Am. Chem. Soc.*, 2013, **135**, 2124.

22 (a) M. Lafrance, S. I. Gorelsky and K. Fagnou, *J. Am. Chem. Soc.*, 2007, **129**, 14570; (b) S. Rousseaux, M. Davi, J. Sofack-Kreutzer, C. Pierre, C. E. Kefalidis, E. Clot, K. Fagnou and O. Baudoin, *J. Am. Chem. Soc.*, 2010, **132**, 10706.

23 (a) W. Li, Z. Duan, X. Zhang, H. Zhang, M. Wang, R. Jiang, H. Zeng, C. Liu and A. Lei, *Angew. Chem., Int. Ed.*, 2015, **54**, 1893; (b) W. Li, Z. Duan, R. Jiang and A. Lei, *Org. Lett.*, 2015, **17**, 1397.

24 (a) K. Orito, A. Horibata, T. Nakamura, H. Ushito, H. Nagasaki, M. Yuguchi, S. Yamashita and M. Tokuda, *J. Am. Chem. Soc.*, 2004, **126**, 14342; (b) K. Orito, M. Miyazawa, T. Nakamura, A. Horibata, H. Ushito, H. Nagasaki, M. Yuguchi, S. Yamashita, T. Yamazaki and M. Tokuda, *J. Org. Chem.*, 2006, **71**, 5951; (c) J. Albert, X. Ariza, T. Calvet, M. Font-Bardia, J. Garcia, J. Granell, A. Lamela, A. Lopez, M. Martinez, L. Ortega, A. Rodriguez and D. Santos, *Organometallics*, 2013, **32**, 649.

25 (a) M. Otsuka, T. Masuda, A. Haupt, M. Ohno, T. Shiraki, Y. Sugiura and K. Maed, *J. Am. Chem. Soc.*, 1990, **112**, 838; (b) J. S. Clark, P. B. Hodgson, M. D. Goldsmith and L. J. Street, *J. Chem. Soc., Perkin Trans. 1*, 2001, 3312.

26 Y. Hoshimoto, T. Ohata, Y. Sasaoka, M. Ohashi and S. Oogoshi, *J. Am. Chem. Soc.*, 2014, **136**, 15877.