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Abnormal anti-thermal quenching of Mn²⁺ and reverse thermal response of Mn²⁺/Mn⁴⁺ luminescence in garnet phosphor for wide-range temperature sensing†

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Herein, we explored the diverse temperature-dependent luminescence, thermal quenching, and decay kinetics of Mn²⁺/Mn⁴⁺ emission centers in a Y₃Al₂Ga₃O₁₂ garnet host. Emission color tuning was achieved from red to deep red region as a function of the Mn doping amount. The tuning of the Mn valence state as a function of doping concentration in Y3Al2Ga3O12 was investigated in detail, and the mechanism of self-reduction of Mn⁴⁺ has been discussed. Interestingly, the temperature-dependent photoluminescence study of $Y_xAl_yGa_xO_{12}$: $Mn^{2+/4+}$ revealed the abnormal anti-thermal quenching of the Mn²⁺ emission band with increasing temperature. In contrast, Mn⁴⁺ emission displayed thermal quenching with increase in temperature. The interplay of self-reduction and defects in the anti-thermal quenching of Mn²⁺ has been discussed. Benefiting from the reverse temperature-dependent luminescence behaviour of Mn^{2+/4+}, luminescence ratiometric and lifetime thermometry have demonstrated a promising wide temperature sensing range, from cryogenic conditions (10 K) to 490 K. A high relative sensitivity of $2.35\%~{\rm K}^{-1}$ (190 K) and $3.10\%~{\rm K}^{-1}$ (390 K) could be achieved for ratiometric and lifetime thermometry, respectively. The lifetime thermometer based on the lifetime of Mn²⁺ and Mn⁴⁺ ions offered good temperature resolution (<0.30 K). This work proposed the use of dual-emitting singledoped materials for highly sensitive wide sensing range thermometry and offers meaningful advancement in the design of orange-red emitting phosphors.

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1. Introduction

Phosphor materials have emerged as potential sources of solidstate lighting technology for indoor/outdoor illumination, displays, optoelectronics, and light-emitting diodes (LEDs) applications. 1-6 The transition metal manganese (Mn^{2+/4+} ions) is a well-known photoluminescence activator for various inorganic luminescent materials used in modern lighting sources owing to red emission with high luminous efficiency, low cost and non-rare earth-based red phosphors.⁷⁻¹¹ Mn²⁺-based phosphors have been widely investigated to achieve multimode fluorescence for the design of white LEDs, thermal sensing, and anti-counterfeiting applications. 12,13 Garnet phosphors (A₃B₂C₃O₁₂) are popular candidates for achieving the strong red to deep red emission of Mn^{2+/4+} ions as they can be stabilized in octahedral Al3+/Ga3+ sites.14,15 The high temperature solid-state synthesis of the garnet structure may result in the coexistence of Mn^{2+/4+} ions due to the defectinduced self-reduction of Mn⁴⁺ ions. The design of Mn^{2+/4+}activated materials offers the potential optimization of emission from red light (Mn²⁺) to deep red light (Mn⁴⁺) by controlling the composition and tuning the emission for indoor plant growth. Since the control of the Mn valence state is a great challenge for developing Mn-activated luminescent materials, few studies have reported dual-emitting Mn^{2+/4+} phosphors for suitable applications. 14-20 Therefore, understanding the photophysical properties, defect-induced self-reduction, charge transfer dynamics, decay kinetics and thermal behaviour is essential for designing Mn-based phosphors for targeted applications utilizing the multiple valence states of Mn. The mixed valence of Mn offers the development of multifunctional phosphors with emission tunability, which is promising for the development of phosphors for indoor plant lighting, anti-counterfeiting, optoelectronics and temperature sensing applications.

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Benefitting from the multifunctionality of phosphor materials, optical temperature sensors based on the thermal response of luminescence intensity ratio (LIR), lifetime, and bandwidth have attracted scientific attention due to their high accuracy, noncontact thermometry, and fast response. 21-26 Lifetime-based thermometers have gained popularity due to the low chances of erroneous temperature readout.²² Recently, dual-wavelength emission materials have emerged as perfect candidates for ratiometric or lifetime-based temperature sensing, which utilize temperature-dependent energy transfer between the dopant ions and charge transfer (CT) between the host matrix and dopant ions.²⁷⁻²⁹ Interestingly, the dual-wavelength emitting Mn^{2+/4+}activated phosphors can be studied for ratiometric and lifetime thermometry based on the diverse temperature-dependent photoluminescence (PL) properties and thermal quenching (TO) of $Mn^{2+/4+}$ ions.

Herein, we synthesized a series of Mn^{2+/4+}-activated Y₃Al₂-Ga₃O₁₂ (YAGG) phosphors using solid-state synthesis reported in our earlier work,3 which is detailed in the ESI.† We reported the stabilization of mixed valent states of Mn in the YAGG lattice and studied the photoluminescence properties of the dual-emitting Y₃Al₂Ga₃O₁₂:Mn^{2+/4+} phosphor at different excitation energies, doping amounts and temperature. The luminescence and application prospects of the dual-emitting Y₃Al₂Ga₃O₁₂:Mn^{2+/4+} phosphor have not been studied before to the best of our knowledge. The changes in the defects and oxidation state of Mn ions with composition were probed using positron annihilation lifetime spectroscopy (PALS) and electron paramagnetic resonance (EPR) spectroscopy. We investigated the temperature-dependent luminescence behaviour of Mn^{2+/4+} ions in a wide temperature range of 10 K to 490 K. The evolution of the Mn²⁺ emission band with temperature is discussed in detail with respect to temperature-dependent CT and defect-induced self-reduction of $Mn^{4+} \rightarrow Mn^{2+}$. The excitation wavelength-dependent decay kinetics was studied as a function of temperature to understand the depopulation mechanism and charge transfer dynamics in Mn^{2+/4+}-activated YAGG system. We demonstrate that the dual-wavelength emitting YAGG:Mn phosphor (620 nm/675 nm) can be utilized for ratiometric and lifetime thermometry based on the diverse thermal response of luminescence and decay of Mn^{2+/4+} ions. The characterization details are provided in the ESI† file (Section S1.2).

2. Results and discussion

Structural and morphological analysis

The XRD diffraction patterns of all the Mn-doped YAGG samples confirmed the phase purity and YAGG crystallized in a cubic system (Ia3d space group) (Fig. S1, ESI†). The garnet structure YAGG is composed of YO₈ dodecahedron, AlO₆/GaO₆ octahedron, and AlO₄/GaO₄ tetrahedron (Fig. S2, ESI†). Based on the ionic size, the Mn^{2+}/Mn^{4+} ions (0.67 Å/0.53 Å) preferably occupy the octahedral Al³⁺/Ga³⁺ (0.535 Å/0.62 Å) sites to form Mn²⁺/Mn⁴⁺ color centers.²⁷ The FTIR spectra show the

characteristic bands of Ga-O and Al-O stretching frequencies in the 580-850 cm⁻¹ region (Fig. S3, ESI†).³ The TEM image of YAGG:0.02Mn shows that the sample crystallized as particles with irregular morphology and size $\sim 1 \mu m$ (Fig. S4, ESI†). The EDS spectra confirmed the presence of Y, Al, Ga, Mn, and O elements in the YAGG:0.02Mn sample (Fig. S5, ESI†).

2.2 Photoluminescence study

The photoluminescence excitation (PLE) spectra show different profiles for 620 nm and 675 nm emissions (Fig. 1(a) and (b)). When excited at 280 nm, the emission spectra (Fig. 1(d)) show a broad red emission peaking at 620 nm along with weak sharp bands at 675 nm, which can be ascribed to the ${}^{4}T_{1} \rightarrow {}^{6}A_{1}$ transition of Mn^{2+} ions and the $^2E \rightarrow {}^4A_2$ transition of Mn^{4+} ions located in the AlO₆/GaO₆ octahedra, respectively. ^{21,23} Mn²⁺ emission depends on the crystal field strength and red emission of Mn²⁺ ions in the YAGG lattice is in accordance with the octahedral coordination of Mn²⁺ ions.^{6,30} Mn⁴⁺ ions with 3d³ electronic configuration occupy the octahedral sites due to the preferred octahedral coordination and size. 18,24 The absence of green emission of Mn²⁺ ions in the emission spectra negated the possible occupation of tetrahedral Al³⁺/Ga³⁺ sites. This suggested the presence of Mn^{2+/4+} ions in the YAGG:Mn samples, which was confirmed by EPR measurements. The presence of sextet EPR lines due to both Mn2+/4+ ions are marked in the EPR spectra (Fig. S6 and S7, ESI†). To further confirm the presence of Mn2+ and Mn4+ ions and investigate the relative amounts of $Mn^{2+/4+}$ ions in the samples with different Mn doping contents, X-ray photoelectron spectroscopy (XPS) measurements were performed on two samples, i.e., YAGG:0.02Mn and YAGG:0.06Mn. The XPS survey spectra of the two samples and Mn 2p_{3/2} spectra are presented in Fig. S8ac (ESI \dagger). The Mn 2p_{3/2} spectra for YAGG:0.02Mn and YAGG:0.06Mn can be deconvoluted into two peaks of Mn²⁺ at 640.9 eV and Mn⁴⁺ at 644.9 eV. 18,31 Though the peak positions remained similar, the relative peak areas of Mn2+ and Mn4+ show different order. In YAGG:0.02Mn, the area of the Mn²⁺ peak is much smaller than that of Mn4+, while in the YAGG:0.06Mn sample, the area of the Mn²⁺ peak is relatively higher than that of YAGG:0.02Mn. This indicated that the Mn²⁺ concentration increased compared to Mn⁴⁺ with the increase in the doping amounts of Mn ions in the YAGG matrix, which can be explained by the self-reduction of Mn⁴⁺ ions to Mn²⁺ at higher doping concentrations, as discussed later.

The color tunability from the red emission of Mn²⁺ in YAGG:0.002Mn to the deep red emission of Mn⁴⁺ ions under different excitation wavelengths is depicted by the emission photographs captured by a camera (see inset of Fig. 1(d) and (e)). The emission spectra (λ_{ex} = 320 nm) show the typical sharp band emission of Mn⁴⁺ ions with less intense Mn²⁺ emission (Fig. 1(e)). Interestingly, upon excitation at 280 and 320 nm, which can be ascribed to $O^{2-} \rightarrow Mn^{2+}$ and $O^{2-} \rightarrow Mn^{4+}$ CT transitions, the dual-band emission bands of both Mn²⁺ and Mn⁴⁺ were observed in the PL emission spectra of the samples at 620 nm ($^{4}T_{1} \rightarrow {^{6}A_{1}}$) and 675 nm ($^{2}E \rightarrow {^{4}A_{2}}$), respectively. Mn^{4+} emission dominates at λ_{ex} = 320 nm, and Mn^{2+} emission

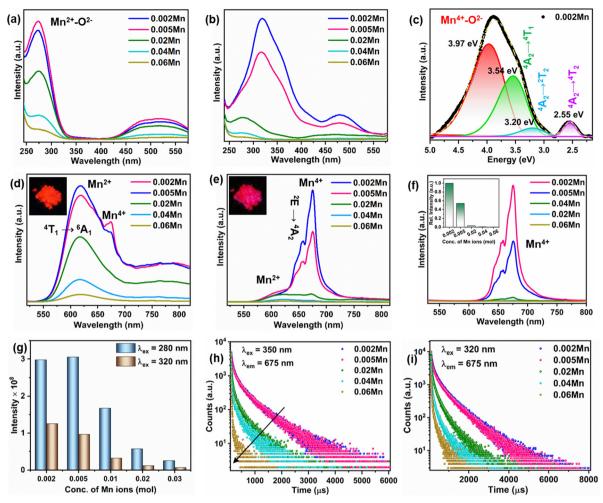


Fig. 1 PLE spectra of YAGG:xMn (x=0.002-0.06 mol) phosphors monitored at (a) $\lambda_{em}=620$ nm and (b) $\lambda_{em}=675$ nm. (c) Deconvolution of the PLE spectrum of YAGG:0.002Mn monitored at 675 nm. PL emission spectra for (d) $\lambda_{ex}=280$ nm and (e) $\lambda_{ex}=320$ nm (emission photograph of YAGG:0.002Mn in inset). (f) PL emission spectra for $\lambda_{ex}=350$ nm (inset showing the relative intensity with increasing Mn doping amounts). (g) Integral intensity as a function of Mn ions concentration. (h) Decay curves of Mn⁴⁺ ions at $\lambda_{ex}=350$ nm and $\lambda_{em}=675$ nm. (i) Decay curves of Mn ions at $\lambda_{ex}=320$ nm and $\lambda_{em}=675$ nm.

is more intense at $\lambda_{\rm ex}$ = 280 nm. The broad PLE spectrum ($\lambda_{\rm em}$ = 675 nm) of YAGG:0.002Mn is composed of four Gaussian bands centered at 312 nm (32051 cm⁻¹), 350 nm (28571 cm⁻¹), 388 nm (25839 cm⁻¹) and 487 nm (20533 cm⁻¹) due to the O²⁻ \rightarrow Mn⁴⁺ CT, $^4{\rm A}_2$ \rightarrow $^4{\rm T}_1$, $^4{\rm A}_2$ \rightarrow $^2{\rm T}_2$ and $^4{\rm A}_2$ \rightarrow $^4{\rm T}_2$ transitions of Mn⁴⁺ ions, respectively (Fig. 1(c)). The PL emission spectra of the samples under 350 nm excitation, *i.e.*, the $^4{\rm A}_2$ \rightarrow $^4{\rm T}_1$ transition of Mn⁴⁺ ions, showed single emission of Mn⁴⁺ ions at 675 nm only due to the $^2{\rm E}$ \rightarrow $^4{\rm A}_2$ transition (Fig. 1(f)). The presence of dual-band emission and single-band emission under 320 and 350 nm excitations can be observed in Fig. S9 (ESI†). We evaluated the crystal field strength ($D_{\rm q}$) and Racah parameter (B) of the Mn⁴⁺ sites in the YAGG:0.002Mn⁴⁺ sample using the following equations.

$$D_{\rm cl}/B = 15(x - 8)/x^2 - 10x \tag{1}$$

$$D_{q} = E(^{4}A_{2} - {}^{4}T_{2})/10$$
 (2)

$$x = E(^{4}A_{2} - {}^{4}T_{1}) - E(^{4}A_{2} - {}^{4}T_{2})/D_{q}$$
 (3)

The crystal field strength (D_q) , Racah parameters (B), and crystal field strength (D_q/B) for Mn^{4+} were calculated as 2053 cm⁻¹, 798 cm⁻¹, and 2.57, respectively.

The variation of emission intensity as a function of Mn concentration is displayed in Fig. 1(f) and (g) where the optimal concentration is 0.002 mol and 0.005 mol for 320/350 nm and 280 nm excitation, respectively. The tuning of Mn emission color with doping amounts can be visualized from the CIE diagram (Fig. S10 and S11, ESI†). The excitation wavelength-dependent color tunability of Mn emission is achieved due to the different absorption efficiency of Mn^{2+/4+} ions in the UV and near-UV regions (Fig. S12, ESI†). The concentration quenching resulted in the reduced emission intensity of both Mn^{2+/4+} ions at higher Mn doping amounts.³² The concentration quenching of both Mn^{2+/4+} ions is caused by electric dipole–dipole interactions as θ = 5.76 and 6.36 (close to 6) were obtained from the log(I/x) *versus* log(x) plot shown in Fig. S13a and b (discussion in ESI†), respectively.³²

With the increase in the Mn ions content, Mn⁴⁺ emission is quenched significantly and the broad emission of Mn²⁺ is

dominant in the YAGG samples with higher Mn content (Fig. S14 and S15, ESI†). This suggested the defect-induced self-reduction of $Mn^{4+} \rightarrow Mn^{2+}$ at higher Mn doping amounts.³³ Due to charge imbalance, the substitution of Al³⁺/ Ga³⁺ sites by Mn⁴⁺ ions will induce the formation of cation vacancies $(V_{Al/Ga})$. The nature of defects with changing doping amount of Mn ions was probed using PALS technique (Fig. S16, ESI†). Typical positron annihilation lifetime spectra are shown in Fig. S16a (ESI†). All the lifetime spectra could be fitted as the sum of three exponentials or three positron annihilation lifetimes with longest lifetime of 1.5 to 2 ns with intensity < 0.5%. The typical fitting of the lifetime spectra is given as inset of Fig. S16a (ESI \dagger). The long-lived component of <0.5% intensity is due to positronium formation on the surface of the powder particles. The other two positron lifetimes (τ_1 and τ_2), intensity of the second positron lifetime (I_2 and $I_1 \sim 100 - I_2$) and the intensity-weighted average positron lifetime (τ_{avg}) are given in Fig. S16b (ESI†).

The first positron lifetime (τ_1) is from the delocalised positron annihilation in the bulk of the sample while the second positron lifetime (τ_2) is from defects. The first positron lifetime in all the samples is in the range from 167 to 173 ps and is close to the lifetimes of 147-160 ps reported for the YAG samples in the literature.³⁴ With the doping of Mn, the first positron lifetime varied only marginally. A significant change is observed as a continuous reduction in the second positron lifetime, with the intensity of this component initially reducing and reaching a minimum at about 0.25-1% Mn doping and then increasing. Theoretical positron lifetimes for positron annihilations from various possible vacancies in YAG are reported.³⁵ As per the calculations, Y vacancies are expected to give positron lifetime of 341 ps, Al vacancy gives a lifetime of 275-279 ps, and oxygen vacancies about 238 ps. The second positron lifetime suggested that cation vacancies are formed initially at low Mn doping amount and finally lead to oxygen vacancies at higher Mn doping. Mn can be doped in variable oxidation states. It is expected that the doping of Mn⁴⁺ would lead to the formation of cation vacancies while Mn2+ would enhance the formation of oxygen vacancies for charge compensation. These results indicated that Mn might be doped as Mn⁴⁺ at lower concentrations while Mn²⁺ is favoured at higher concentrations. This result is in line with the XPS analysis; hence, it can be concluded that more Mn⁴⁺ ions get reduced to Mn²⁺ with the increase in the Mn doping amounts.

The presence of both Mn⁴⁺ and Mn²⁺ emission indicated abnormal self-reduction in the YAGG lattice. The charge compensation model has been used in earlier reports to explain the abnormal self-reduction from Mn⁴⁺ → Mn²⁺ ions. ¹⁵ The substitution of trivalent Al3+/Ga3+ sites by tetravalent Mn4+ ions in YAGG will induce the formation of positively charged $[Mn^{4+}]_{Al/Ga}^{\bullet}$ and negatively charged $V_{Al/Ga}^{\prime\prime\prime}$ defects. To neutralize the excess positive charge, four Al3+/Ga3+ ions will be replaced by three Mn^{4+} ions (eqn (1)). The generated $V_{Al/Ga}^{\prime\prime\prime}$ can act as electron donors, and the released electrons from the $V_{Al/Ga}^{\prime\prime\prime}$ defects can be captured by the electron acceptor $[Mn^{4+}]_{Al/Ga}^{\bullet}$ -defects. As a result, some of the Mn^{4+} ions are reduced to Mn²⁺. The above processes can be explained by the following equations.

$$3Mn^{4+} + 4Al^{3+}/Ga^{3+} \rightarrow [Mn^{4+}]^{\bullet}_{Al/Ga} + V'''_{Al/Ga}$$
 (4)

$$V_{Al/Ga}^{\prime\prime\prime} \rightarrow V_{Al/Ga}^{\times} + 3e^{-} \tag{5}$$

$$\left[Mn^{4+}\right]_{Al/Ga}^{\bullet} + 2e^{-} \rightarrow \left[Mn^{2+}\right]_{Al/Ga}^{'} \tag{6}$$

In addition, the YAGG:0.02Mn samples were synthesized under air atmosphere using MnCO3 and MnO2 as raw Mn source. The PLE and emission spectra (Fig. S17, ESI†) show similar spectral profiles, which confirm the self-reduction of Mn⁴⁺ to Mn²⁺ ions in the YAGG matrix.

The luminescence decay profiles for the Mn-doped YAGG phosphors for emission at 675 nm and 620 nm (λ_{ex} = 320/ 350 nm and 280 nm) are presented in Fig. 1(h), (i) and Fig. S18 (ESI†), respectively. The average lifetimes (τ_{avg}) of Mn ions were calculated using a bi-exponential decay function.

$$I(t) = I_0 + A_1 e^{-t/\tau_1} + A_1 e^{-t/\tau_2}$$
 (7)

where I_0 and I(t) represent the emission intensities at zero offset and at 't' time, τ_1 and τ_2 are the two lifetime components, while A_1 and A_2 are constants, respectively. The lifetime values for the Mn-doped YAGG phosphors are summarized in Tables S1 and S2 (ESI†). The average lifetimes of the 675 nm emission (λ_{ex} = 320 nm) peak decreased from 652 µs to 276 µs on higher Mn doping (x = 0.002 to 0.06), which is consistent with the PL intensity trend due to non-radiative energy migration between Mn⁴⁺ ions.³⁶ Similarly, the lifetime value of Mn⁴⁺ ions decreased from 500 µs to 200 µs on higher Mn doping at $\lambda_{\rm ex}$ = 350 nm. The lifetime values of 620 nm emission ($\lambda_{\rm ex}$ = 280 nm) firstly increased from 354 µs to 360 µs and then decreased to 160 µs at higher Mn doping. Time-resolved emission spectroscopy (TRES) was performed for 280 nm excitation (Fig. S19, ESI†). With an increase in the delay times, both Mn²⁺ and Mn4+ emissions can be observed while only Mn2+ emissions are dominant at lower delay times.

2.3 Temperature-dependent PL study

Since the excitations at 280 and 320 nm fall in the excitation range of both Mn4+ and Mn2+ ions, the temperature-dependent luminescence behaviour of the dual-emission bands of Mn^{2+/4+} ions can be investigated effectively using these two CT transitions. To understand the thermal behaviour of Mn^{2+/4+} luminescence, the temperature-dependent PLE and emission spectra of Mn^{2+/4+} under CT band excitation were recorded from 10 K to 490 K (Fig. 2(a)-(f)). The PLE spectra ($\lambda_{\rm em}$ = 620 nm) in Fig. 2(a) display an increase in the $O^{2-} \rightarrow Mn^{2+}$ CT band intensity initially up to 290 K, followed by decreased intensity at elevated temperatures (Fig. 3(a)). On the contrary, the PLE spectra (λ_{em} = 675 nm) shows a continuous decrease in the intensity of the $O^{2-} \rightarrow Mn^{4+}$ CT band with temperature (Fig. 3(a)). The temperature-dependent PL emission

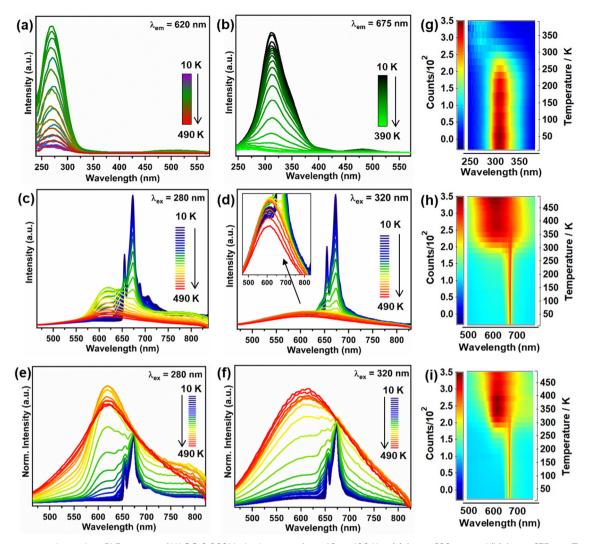


Fig. 2 Temperature-dependent PLE spectra of YAGG:0.002Mn in the range from 10 to 490 K at (a) λ_{em} = 620 nm and (b) λ_{em} = 675 nm. Temperaturedependent PL spectra at (c) λ_{ex} = 280 nm and (d) λ_{ex} = 320 nm (inset shows the zoomed-in emission spectra). Temperature-dependent normalized PL spectra at 675 nm under (e) λ_{ex} = 280 nm and (f) λ_{ex} = 320 nm. (g)-(i) Contour plots for the normalized intensity of the CT band (λ_{em} = 675 nm) and emission spectra (λ_{ex} = 320 and 280 nm), respectively.

spectra (λ_{ex} = 320 nm and 280 nm) demonstrated an abnormal trend in integral intensity change with temperature (Fig. 2(c) and (d)). The normalized emission spectra with respect to 675 nm emission shows that the spectral profile changes with temperature, where the Mn²⁺ emission becomes dominant at elevated temperatures (Fig. 2(e) and (f)). The broad emission at 620 nm (λ_{ex} = 280 nm) due to Mn²⁺ ions increased with temperature till 290 K and shows TQ above 310 K; however, Mn⁴⁺ emission at 675 nm undergoes continuous TQ (Fig. 2(g) and 3(a)). The above observations can be visualized from the contour plots (Fig. 2(g)-(i)). The TQ in the Mn⁴⁺ ions can be caused by the crossover of ${}^{2}E$ and ${}^{4}T_{2}$ parabolas (ΔE_{1}) or ${}^{4}T_{2}$ and 4 A₂ parabolas (ΔE_2), which results in non-radiative relaxation. ³⁷ The phonon-assisted crossover of 4T_1 and 6A_1 parabolas in the case of Mn²⁺ ions can result in non-radiative relaxation to the ground state (ΔE) .³⁸

The trend of total integral intensity under different excitation wavelengths is shown in Fig. S20 (ESI†). The trend of the $O^{2-} \rightarrow Mn^{2+}$ CT band is similar to the 620 nm peak intensity variation with temperature. This indicated that the release of electrons from the traps with increasing temperature may strength the $O^{2-} \rightarrow Mn^{2+}$ CT, which led to the anti-TQ of Mn²⁺ emission (<290 K). The temperature dependence of the Mn²⁺ ion showing both anti-TQ and normal TQ can be fitted by the Shibata et al. model discussed in the ESI† (Fig. S21).³⁹ The values of activation energies for the anti-TQ of Mn2+ emission below 290 K is 686 cm⁻¹, normal TQ of Mn²⁺ emission after 310 K is 1955 cm⁻¹, and energy barrier for TQ of Mn⁴⁺ ions is 774 cm⁻¹. This suggested that the anti-TQ of Mn²⁺ ions and TQ of Mn⁴⁺ ions may be controlled by the thermal population of the ${}^{4}\Gamma_{1}$ state of Mn²⁺ ions and the simultaneous depopulation of the ²E state of Mn⁴⁺ ions, respectively.

To get better insights into the depopulation and thermal quenching mechanisms, temperature-dependent decay curves were recorded at Mn^{2+/4+} emission wavelengths in the temperature range from 10 K to 490 K. The lifetime values of Mn²⁺

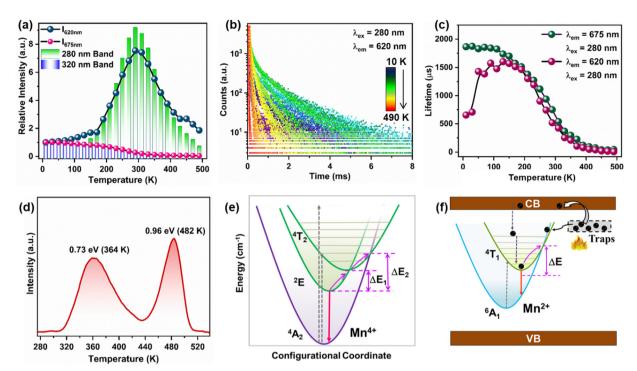


Fig. 3 (a) Variation in the peak intensity at 620 nm and 675 nm with temperature (λ_{ex} = 280 nm) and the CT band intensity for λ_{em} = 620 nm and 675 nm. (b) Luminescence decay profiles as a function of temperature at λ_{ex} = 280 nm and λ_{em} = 620 nm. (c) Temperature variation of lifetime for λ_{ex} = 280 nm. (d) TL spectra of YAGG:0.02Mn when irradiated with 280 nm light. (e) Schematic illustration of radiative transitions and possible thermal guenching mechanisms for Mn⁴⁺ emission in octahedral coordination. (f) Schematic illustration of electron release from the traps and enhanced Mn²⁺ emission with the increase in temperature

emission at 620 nm increases initially, which can be attributed to the thermal population of the 4T_1 state of Mn^{2+} ions, which modulates the PL intensity of Mn²⁺ emission (Fig. 3(b) and (c) and Fig. S22, S23, ESI†). 40 The lifetime increase indicated that with increasing temperature, the excited state population is increased by some thermally-activated processes. At higher temperature, the increased non-radiative pathways may result in the observed decrease in the lifetime values. In contrast, the lifetime values of 675 nm emission show a continuous decrease with temperature due to the increased non-radiative processes for the Mn⁴⁺ excited states (Fig. 3(c) and Fig. S22-S24, ESI†). The normal TQ of Mn⁴⁺ emission can be explained by the thermal cross-over of the excited states with the ground state of Mn^{4+} ions (Fig. 3(e)).

2.4 Mechanism of thermal quenching behaviour of Mn^{2+/4+} ions

To investigate the nature of defects, the thermoluminescence (TL) spectrum for the YAGG:0.002Mn^{2+/4+} was recorded sample upon irradiation with 280 nm light for 100 s. The TL spectra (Fig. 3(d)) shows two peaks at 364 K and 482 K, which can be attributed to the presence of V_O and $\left[Mn^{4+}\right]^{\bullet}_{Al/Ga}$ defects, respectively.³³ The trap depths of these defects were calculated as 0.73 eV and 0.96 eV, using the equation E = T/500 (eV). Fig. 3(f) shows the proposed mechanism for the anti-TQ of Mn²⁺ emission. The shallow traps can capture the electrons during UV excitation, which can be released to the conduction band (CB) on increasing the temperature. The electrons in the CB can populate the excited 4T_1 state of Mn²⁺. The increased thermal vibrations can also induce electron tunneling to the excited ⁴T₁ level of Mn²⁺ ions from the trap levels around the excited energy levels of Mn2+. The above thermally-induced population of Mn2+ excited levels resulted in the enhancement of Mn²⁺ emission with temperature increase to 290 K. However, thermal quenching dominates at higher temperatures due to the thermal crossover of the ${}^4\Gamma_1$ and 6A_1 states. The observed trend can also be explained by the thermallyinduced self-reduction of $Mn^{4+} \rightarrow Mn^{2+}$ ions as $V_{Ga/Al}$ can act as electron donors. 30,33 With the increase in temperature, the electrons are released from V_{Ga/Al} and transferred to the nearby Mn⁴⁺ ions. 18

Temperature sensing application 2.5

The diverse temperature-dependent behaviour of Mn^{2+/4+} emission upon 280 nm excitation can be utilized for LIR-based ratiometric and lifetime thermometry. The LIR (620 nm/ 675 nm) (Fig. 4(a)) from 10 K to 330 K can be fitted with the following exponential equation.41

$$LIR_{(620/675)} = A + B \exp(-\Delta E/kT)$$
(8)

Here, ΔE is the activation energy for thermal quenching, T is temperature (Kelvin), k is the Boltzmann constant, and A and B are constants. The absolute sensitivity (S_a) and relative sensitivity (S_r) values can be evaluated by the following

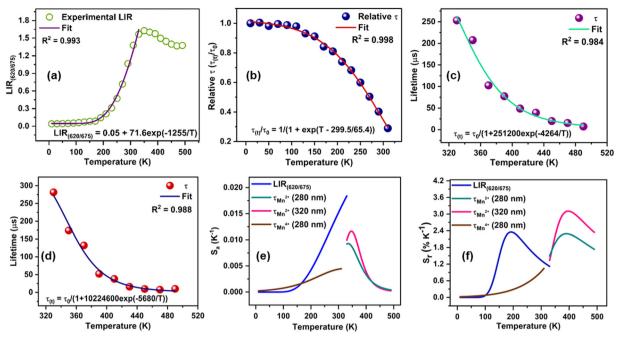


Fig. 4 (a) Temperature dependence of LIR_(620/675) and the fitting curve. Temperature dependence of relative τ values for (b) $\lambda_{\rm ex}$ = 280 nm and $\lambda_{\rm em}$ = 675 nm, (c) $\lambda_{\rm ex}$ = 280 nm and $\lambda_{\rm em}$ = 620 nm, (d) $\lambda_{\rm ex}$ = 320 nm and $\lambda_{\rm em}$ = 620 nm. (e) Corresponding $S_{\rm a}$ values as a function of temperature. (f) Corresponding $S_{\rm r}$ values as a function of temperature.

relations.

$$S_{a}\left(\mathbf{K}^{-1}\right) = \left|\frac{\mathbf{dLIR}}{\mathbf{d}T}\right| \tag{9}$$

$$S_{\rm r} \left(\% \text{ K}^{-1}\right) = \frac{1}{\text{LIR}} \left| \frac{\text{dLIR}}{\text{d}T} \right| \times 100$$
 (10)

The $S_{\rm a}$ and $S_{\rm r}$ values as a function of temperature are shown in Fig. 4(e) and (f), respectively. The maximum $S_{\rm a}$ value of 0.018 K⁻¹ (at 330 K) and high $S_{\rm r}$ of 2.35% K⁻¹ (190 K) could be achieved for the Mn^{2+/4+} LIR-based thermometer. An $S_{\rm r}$ value higher than 1.0% K⁻¹ was retained up to 330 K. The temperature uncertainty (δT) can be evaluated using the following expression.

$$\delta T (K) = \frac{1}{S_{\rm r}} \frac{\delta LIR}{LIR}$$
 (11)

where δ LIR/LIR can be evaluated as the uncertainty of the measured LIR values.⁴⁰ The δ *T* values in the range of 0.15–0.01 K were obtained for ratiometric thermometry.

The thermal dependence of the lifetime values of 675 nm emission under 280 nm excitation wavelength can be utilized for wide-range temperature sensing from 10 to 310 K. The sigmoidal temperature dependences of the relative lifetime $(\tau_{(t)}/\tau_0)$ of Mn⁴⁺ ions can be fitted using the Boltzmann function as follows.

$$\Delta_{\tau} = \frac{\tau_{(t)}}{\tau_0} = \frac{1}{1 + A \exp(T - T_0/dT)}$$
 (12)

where τ_0 and $\tau_{(t)}$ are the lifetime values at initial temperature T_0 and at 'T', respectively.

The temperature dependences of $\rm Mn^{2^+}$ lifetime ($\lambda_{\rm ex}$ = 280 nm/320 nm) can be fitted in the temperature range from 330 K to 490 K using an equation derived from the Mott–Seitz model.²⁷

$$\Delta_{\tau} = \tau_{(t)} = \frac{\tau_0}{1 + A \exp(-\Delta E/kT)}$$
 (13)

where k is Boltzmann constant, and ΔE is the activation energy for non-radiative processes.

The S_a and S_r values for lifetime thermometry can be evaluated by the following relations.

$$S_{\rm a}\left({\rm K}^{-1}\right) = \left|\frac{{\rm d}\varDelta_{\rm \tau}}{{\rm d}T}\right| \tag{14}$$

$$S_{\rm r} \left(\% \ \mathrm{K}^{-1} \right) = \frac{1}{\Delta_{\rm r}} \left| \frac{\mathrm{d} \Delta_{\rm r}}{\mathrm{d} T} \right| \times 100 \tag{15}$$

Lifetime thermometry based on the Boltzmann function using $\mathrm{Mn^{4+}}$ emission offers the maximum S_a value of 0.0045 K⁻¹ (at 310 K), and S_r of 1.05% K⁻¹ was attained at 310 K (Fig. 4(f)). Lifetime thermometry based on $\mathrm{Mn^{2+}}$ emission (λ_ex = 280 nm) displays the maximum S_a value of 0.009 K⁻¹ (at 330 K) and S_r of 2.29% K⁻¹ (390 K). Lifetime thermometry based on $\mathrm{Mn^{2+}}$ emission (λ_ex = 320 nm) displays the maximum S_a value of 0.012 K⁻¹ (at 350 K) and S_r of 3.10% K⁻¹ (390 K). The temperature uncertainty (δT) for lifetime thermometry can be

Table 1 Comparison of thermometric performance of our thermometer with similar reported thermometers

System	Туре	Temp. range	$S_{\rm r}~(\%~{ m K}^{-1})$	Ref.
LaZnGa ₁₁ O ₁₉ (LZG):Cr ³⁺ /Ni ²⁺	LIR	100-175 K, 200-475 K	2.4% K ⁻¹ at 475 K	42
Sr ₂ LuNbO ₆ :Mn ⁴⁺	LIR	12-523 K	0.55% K ⁻¹ at 136 K	43
	Lifetime	12-523 K	1.55% K ⁻¹ at 519 K	
SrGdLiTeO ₆ :Mn ⁴⁺ /Eu ³⁺	Lifetime	300-600 K	$0.229\%~{ m K}^{-1}~{ m at}~573~{ m K}$	44
$Ca_3Ga_2Ge_3O_{12}:Mn^{3+}/Mn^{4+}$	LIR	5-75 K	6% K ⁻¹ at 33 K	40
$ZnGa_{2-y}Al_yO_4:Mn^{2+}/Mn^{4+}$	LIR	100-475 K	4.345% K ⁻¹ at 350 K	17
$Lu_3Al_5O_{12}$: Ce^{3+}/Mn^{4+}	LIR	250-350 K	4.37% K ⁻¹ at 350 K	27
	Lifetime	250-350 K	$3.22\% \text{ K}^{-1} \text{ at } 350 \text{ K}$	
Mn ⁴⁺ -doped BaLaCa _{1-x} Mg _x SbO ₆	Lifetime	303-503 K	1.42% K ⁻¹ at 488 K	9
$Y_2O_3:Ho^{3+}/Mg_2TiO_4:Mn^{4+}$	LIR	RT-100 °C	4.6% °C ⁻¹ 0.9 at 473 K	45
	Lifetime	RT-100 °C		
$Gd_3Al_5O_{12}:Mn^{3+}/Mn^{4+}$	Lifetime	100-600 K	$2.08\% \text{ K}^{-1} \text{ at } 249 \text{ K}$	22
$Y_2O_3:Yb^{3+}/Er^{3+}$	Lifetime	298-338 K	$0.50\%~{\rm K}^{-1}~{\rm at}~310~{\rm K}$	46
$Y_3^{-}Al_2^{-}Ga_3O_{12};Mn^{2+}/Mn^{4+}$	LIR	10-330 K	2.35% K ⁻¹ at 190 K	This work
	Lifetime (Mn ⁴⁺)	10 to 310 K	1.05% K ⁻¹ at 310 K	
	Lifetime ₂₈₀ (Mn ²⁺)	330 to 490 K	2.29% K ⁻¹ at 390 K	
	Lifetime ₃₂₀ (Mn ²⁺)	330 to 490 K	$3.10\%~{\rm K}^{-1}~{\rm at}~390~{\rm K}$	

evaluated using the following expression

$$\delta T (K) = \frac{1}{S_{\rm r}} \frac{\delta \tau}{\tau}$$
 (16)

where $\delta \tau / \tau$ can be evaluated as uncertainty in lifetime measurements, which was determined as 0.4% for repeated decay measurements. The δT values within 0.12-0.30 K indicated high resolution for temperature sensing in the range from 330 K to 490 K for lifetime thermometry using Mn²⁺ emission (Fig. S25, ESI†). Temperature sensing with high precision can be achieved in this work based on LIR and the lifetime of Mn^{2+/4+} ions in the YAGG matrix.

The thermometric performance of our thermometer was compared with similar thermometers in Table 1. Hence, the lifetime thermometry displayed by the YAGG:Mn^{2+/4+} phosphor is relatively higher than the earlier reported lifetime thermometers, and S_r greater than 1.0% K⁻¹ can be achieved above 150 K. The reverse thermal response of Mn^{2+/4+} ions can be utilized for temperature sensing at low temperatures. The lifetime of the two valence states can be used in different temperature ranges for temperature detection with high precision. Hence, the mixed valence state YAGG:Mn^{2+/4+} phosphor is a potential candidate for a dual-mode temperature sensor. This work discussed the abnormal reverse thermal response of two ions, which can be utilized for targeted applications and provide useful insights into such mixed valent Mn^{2+/4+} orange-red emitting phosphors.

3. Conclusion

In conclusion, YAGG:Mn^{2+/4+} phosphors display different emission from red to deep red under different excitation wavelengths (280 and 320 nm). The luminescence tuning of orange-red emitting phosphors with doping concentration of Mn ions and excitation wavelength can be promising for designing orangered emitting materials. The Mn2+/Mn4+ ions display distinctive luminescence and decay time behaviour, which can be utilized for potential wide range optical thermometry with high temperature sensitivity. The ratiometric thermometer offers a high maximum $S_{\rm r}$ value of 2.35% K⁻¹ (190 K). Lifetime thermometry at two different excitation wavelengths, namely, 280 nm and 320 nm, presents temperature read-out with a high maximum S_r value of $3.10\% \text{ K}^{-1}$ (390 K) and $1.05\% \text{ K}^{-1}$ (310 K), respectively. This work reports a single metal (two oxidation states)-based dual emitting ratiometric and lifetime thermometer with wide sensing range. Additionally, this work demonstrated the self-reduction of Mn ions in Mn-doped garnet materials and abnormal temperaturedependent luminescence properties of Mn^{2+/4+} ions.

Data availability

The data supporting this article have been included as part of the ESI.†

Conflicts of interest

There are no conflicts of interests to declare.

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