

REVIEW

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Recent advances in graphite carbon nitride-based nanocomposites: structure, antibacterial properties and synergies

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Bacterial infections and transmission threaten human health and well-being. Graphite carbon nitride ($g\text{-C}_3\text{N}_4$), a promising photocatalytic antibacterial nanomaterial, has attracted increasing attention to combat bacterial transmission, due to the outstanding stability, high efficiency and environmental sustainability of this material. However, the antibacterial efficiency of $g\text{-C}_3\text{N}_4$ is affected by several factors, including its specific surface area, rapid electron/hole recombination processes and optical absorption properties. To improve the efficiency of the antibacterial properties of $g\text{-C}_3\text{N}_4$ and extend its range of applications, various nanocomposites have been prepared and evaluated. In this review, the advances in amplifying the photocatalytic antibacterial efficiency of $g\text{-C}_3\text{N}_4$ -based nanocomposites is discussed, including different topologies, noble metal decoration, non-noble metal doping and heterojunction construction. The enhancement mechanisms and synergistic effects in $g\text{-C}_3\text{N}_4$ -based nanocomposites are highlighted. The remaining challenges and future perspectives of antibacterial $g\text{-C}_3\text{N}_4$ -based nanocomposites are also discussed.

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1. Introduction

Bacterial infections and their transmission pose a considerable threat to human health, usually leading to delayed wound healing and chronic intestinal diseases.^{1,2} Furthermore, pathogenic bacteria frequently contaminate water supplies and the soil, resulting in the death of animals and plants through environmental contamination.³ To address these problems, antibiotics are widely used to kill bacteria, but over-utilization of antibiotics can bring about resistance and secondary contamination.⁴⁻⁶ Therefore, the development of new strategies

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to inactivate bacteria without using antibiotics is urgently required as is the inactivation of drug-resistant bacteria. In recent years, semiconductor photocatalysis has attracted significant interest for applications in pollutant degradation⁷ and antimicrobial applications.⁸ Under sunlight irradiation, semiconductor photocatalysts react with water and oxygen to form reactive oxygen species (ROS), such as hydrogen peroxide (H_2O_2), hydroxyl radicals ($\cdot\text{OH}$) and superoxide ($\text{O}_2^{\cdot-}$), which are able to inactivate bacteria by oxidizing the phospholipid membrane, proteins and nucleic acids.^{9–13} Common photocatalysts include metallic oxides, sulfides, nitrides, and phosphides, and graphene and its derivatives,^{14,15} which show bactericidal activity against both Gram-positive and Gram-negative bacteria, as demonstrated in many studies. However, relatively narrow spectral absorption ranges result in low efficiencies.¹⁶ Together with other problems such as facile aggregation, potential toxicity, and low biocompatibility, practical application of these common photocatalysts is limited. As an emerging non-metallic photocatalyst, graphitic carbon nitride ($\text{g-C}_3\text{N}_4$) is easy to prepare and has an appropriate band structure and good biocompatibility,¹⁷ thus showing considerable potential as antibacterial materials.

Similar to the layered structure of graphene, $\text{g-C}_3\text{N}_4$ is a polymeric layered material which consists of carbon and nitrogen with some hydrogen (impurity). The conduction band (CB) and valence band (VB) position of $\text{g-C}_3\text{N}_4$ are ~ 1.13 and 1.57 eV, respectively. The appropriate band gap is about 2.70 eV, *i.e.* it is a medium band gap semiconductor, readily obtained from the pyrolysis of melamine, dicyandiamide or urea. Benefiting from the connection of tri-*s*-triazine units through tertiary amines (Fig. 1), $\text{g-C}_3\text{N}_4$ has a two-dimensional flake structure. The adjacent $\text{g-C}_3\text{N}_4$ flakes interact weakly with each other *via* van der Waals forces, displaying a layer gap of about 0.33 nm.^{18–22} Thus, when such forces between the bulk $\text{g-C}_3\text{N}_4$ layers are broken, paper-like $\text{g-C}_3\text{N}_4$ nanosheets are obtained.^{23–27} From a molecular prospective, C_3N_4 is considered to have two main molecular structures. One comprises a triazine with a tertiary nitrogen atom in the center of a planar triangle (Fig. 1a) that connects three separate triazine rings in an infinitely repeating pattern.²⁸ Another is a tri-*s*-triazine structurally

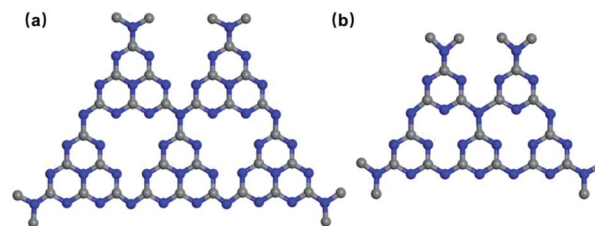


Fig. 1 (a) Tri-*s*-triazine and (b) *s*-triazine structure of $\text{g-C}_3\text{N}_4$.

connected to form $\text{g-C}_3\text{N}_4$ (Fig. 1b). This more stable tri-*s*-triazine structure is commonly used as the structural unit of $\text{g-C}_3\text{N}_4$.²⁹ These excellent structural features and properties make C_3N_4 a promising photocatalytic material. Since the first study of the photocatalytic activity of $\text{g-C}_3\text{N}_4$ by Wang *et al.* in 2009, it has become a prevalent photocatalytic material.³⁰ Due to the merits of the non-metallic $\text{g-C}_3\text{N}_4$ material, such as a wide visible light absorption range, excellent chemical stability and low toxicity, it has been widely studied to tackle environmental and energy related problems.^{31–35} Specifically, $\text{g-C}_3\text{N}_4$ has been used as a catalyst for photocatalytic water reduction and oxidation, contaminant degradation and carbon dioxide reduction.^{36–41} For photocatalytic degradation, the photo-produced electrons (e^-) and holes (h^+) can accelerate reduction and oxidation degradation reactions.^{42–47} In addition, the e^- and h^+ can react with surrounding H_2O and O_2 to generate ROS such as $\cdot\text{O}^{\cdot-}$ and $\cdot\text{OH}^{\cdot-}$. The generated ROS can further degrade pollutants, combining to achieve the removal of contaminants.^{48–51}

In this review, several tactics for enhancing the antimicrobial efficiency of $\text{g-C}_3\text{N}_4$ -based nanocomposites are discussed (Fig. 2), including the design of different $\text{g-C}_3\text{N}_4$ topologies,^{52–55} noble metal decoration,^{56–58} non-noble metal doping and heterojunction construction.^{59–61} These approaches have been shown to effectively boost the antibacterial activity of $\text{g-C}_3\text{N}_4$.^{62–67} The enhancement mechanisms and synergistic effects of $\text{g-C}_3\text{N}_4$ -based nanocomposites is highlighted. Additionally, photocatalytic mechanisms have been elucidated by analyzing the interactions between the nanomaterials and bacteria.



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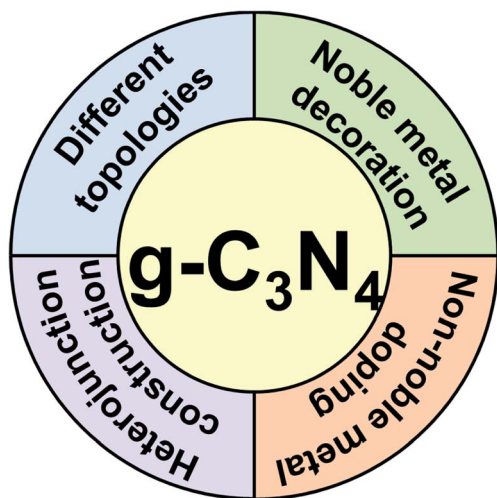


Fig. 2 Schematic illustration of antimicrobial enhancement of g-C₃N₄-nanomaterials.

Lastly, this review concludes by defining future prospects, opportunities and challenges in this exciting field.

2. g-C₃N₄-based materials and their photocatalytic antibacterial

Both the nitrogen and carbon atoms in g-C₃N₄ are sp² hybridized to form a conjugated structure with delocalized π -electrons, giving g-C₃N₄ the lowest bandgap compared to other phases of C₃N₄.⁶⁸ Compared to other conventional photocatalytic materials, g-C₃N₄ has a narrow band gap (2.7 eV), resulting in a wider spectral absorption range of up to 460 nm, and improving the photocatalytic window.⁶⁹ The photocatalytic reactions of g-C₃N₄ affect its antibacterial performance. When the energy of the visible light illuminating g-C₃N₄ is larger than the band gap energy of g-C₃N₄, e⁻ are promoted from the valence band (VB) to the conduction band (CB), producing active e⁻ and h⁺. However, the e⁻/h⁺ can recombine on the g-C₃N₄ surface. Alternatively, the e⁻/h⁺ pairs diffuse or are transported to the g-C₃N₄ interface by an electric field and undergo redox reactions with the surroundings. As shown in Fig. 3, under visible light illumination, g-C₃N₄ produces ROS that can destroy the bacterial membranes, causing cell membrane permeability, structural degradation and ultimately killing the bacterial.⁷⁰

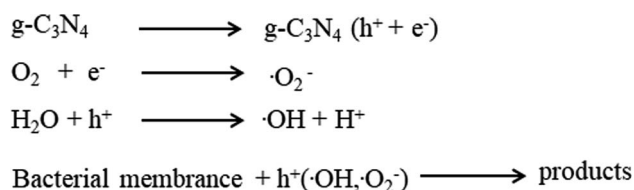


Fig. 3 Plausible reactions between bacterial and g-C₃N₄-based nanocomposite generated ROS.

2.1 Influence of topology on g-C₃N₄-based antibacterial photocatalysts

The antibacterial activity of g-C₃N₄ is influenced by its topology, in particular, the efficiency of bulk g-C₃N₄ is restricted by its small superficial area and rapid rate of recombination of photogenerated carriers.^{71–75} In contrast, mesoporous g-C₃N₄, g-C₃N₄ nanotubes and nanosheets absorb visible light more effectively and provide a larger contact area for reactants because of their larger specific surface areas and a larger number of active sites.^{76,77} In addition, these structures decrease the distance required for the transfer of the charge to the surface of the material, decreasing charge recombination.^{78,79} Moreover, due to the quantum size effects, g-C₃N₄ nanosheets are suited to charge transfer and separation processes.^{80,81} Therefore, g-C₃N₄ nanosheets and their composite materials show promising photocatalytic antibacterial properties. Li *et al.* developed a self-cleaning antibacterial membrane by simply filtering g-C₃N₄ nanosheets into polyacrylonitrile porous substrates (Fig. 4a), then forming a stable coating by cross-linking polyvinyl alcohol and glutaraldehyde.⁸² In contrast to membranes without g-C₃N₄ nanosheets, the membranes containing the g-C₃N₄ nanosheets (0.45 wt%) completely inactivated 1 × 10⁶ cfu mL⁻¹ *E. coli* under irradiation with visible light (Fig. 4b and c). The high superficial area of the g-C₃N₄ nanosheets in the membrane provides more active sites that produce ROS for sterilization. Meanwhile, the membranes with g-C₃N₄ nanosheets also showed good permeability to water and degraded dyes. Compared to g-C₃N₄ nanosheets, nanotubes have high aspect ratios, a topology that favors the migration of e⁻ along the axial direction and inhibits the lateral transfer of e⁻, thereby inhibiting the recombination of photogenerated carriers. Moreover, nanotubes usually have large specific surface areas, providing a higher density of active sites at their surface, which improves photocatalytic antibacterial performance.^{83,84} Xu *et al.* successfully synthesized microtubular nanoporous g-C₃N₄ with a layered structure and nitrogen defects using molecular self-assembly methods.⁸⁵ The hexagonal tubular structure promotes the multiple use of light, and provides a larger density active sites and a directional transfer channel for e⁻. Moreover, the nanoporosity of the material increases the specific surface area to provide rich charge transport paths. In addition, the nitrogen vacancies improve the light harvesting properties of the material ($\lambda > 450$ nm) and promote charge separation by trapping charge. Hence, microtubular nanoporous g-C₃N₄ completely inactivated 5 × 10⁶ cfu mL⁻¹ *E. coli* after 4 h of light illumination, compared to only 74% of *E. coli* sterilized by bulk g-C₃N₄. Hollow porous microspheres not only promote light penetration within the material and light absorption at the pore edges, but also provide sufficient contact area to accelerate interfacial charge transfer. In addition, the thinner pore wall structure reduces the distance (and time) required for charge transfer within the material, decreasing the recombination of photogenerated carriers.⁸⁶ Yang *et al.* successfully fabricated a self-cleaning, antimicrobial and antifouling membrane by integrating mesoporous g-C₃N₄ (MCN) into polyvinylidene fluoride



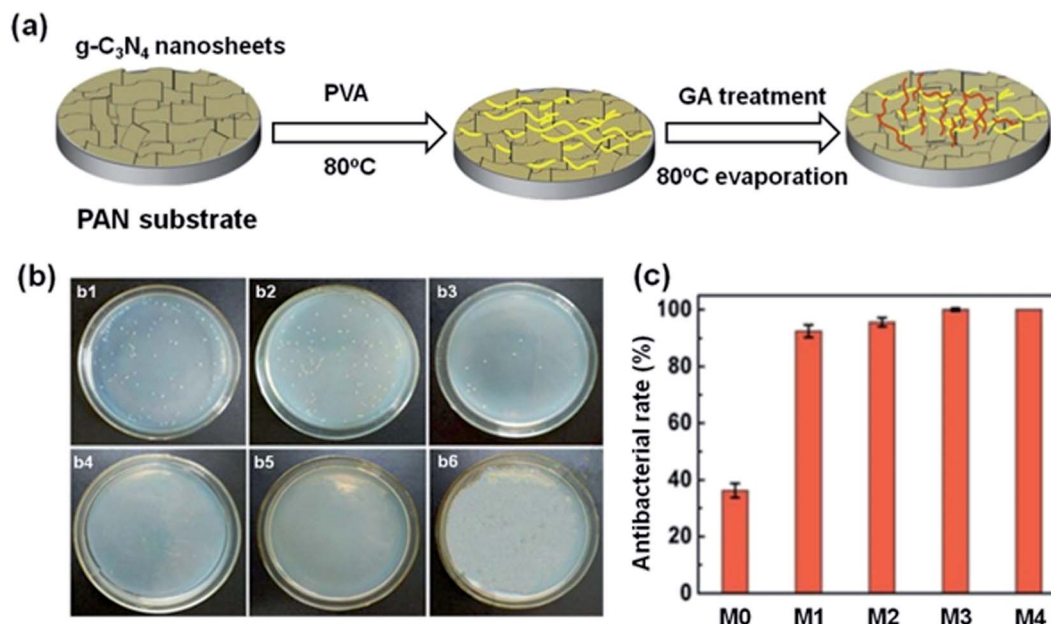


Fig. 4 (a) Preparation of g-C₃N₄ nanosheet-functionalized composite membranes. The concentration of g-C₃N₄ nanosheets in the membrane are 0, 0.15 wt%, 0.3 wt%, 0.45 wt%, 0.6 wt% in M0, M1, M2, M3, M4, respectively. (b) Antimicrobial activities against *E. coli* of (b1) control, (b2) M0, (b3) M1, (b4) M2, (b5) M3, (b6) M4. (c) Antibacterial rate of membranes M0, M1, M2, M3, M4. Reproduced from ref. 82 with permission from Royal Society of Chemistry, copyright 2017.

(PVDF).⁸⁷ The mesoporous structure promotes multiple reflections of incident light and enhances the capacity of the material for light capture, leading to an enhancement in the generation of h⁺ and ROS. The MCN-PVDF membrane showed a significant

reduction in the number of *E. coli* colonies under illumination with visible light over 4 hours, with approximately 3 log deactivation of *E. coli*.^{88,89} In contrast, an analogous experiment using a membrane-free material showed no significant decrease

Table 1 Antibacterial properties of g-C₃N₄ based materials with different topologies

Material	Preparation	Bacteria	Effect	Ref.
g-C ₃ N ₄ nanosheets	Acid etching and ultrasound	<i>E. coli</i>	~100%	82
Microtubular nanoporous g-C ₃ N ₄	Molecular self-assembly	<i>E. coli</i>	99.2%	85
g-C ₃ N ₄ nanosheets	Freezing and microwave-assisted	<i>E. coli</i>	100%	90
Porous g-C ₃ N ₄ nanosheets	Template-free	<i>E. coli</i>	100%	91
g-C ₃ N ₄ nanosheets	Bacterial etching	<i>E. coli</i>	3.65 log	92
g-C ₃ N ₄ nanosheets	Ultrasound	<i>E. coli</i> O157:H7	0.82 log	93
		<i>S. aureus</i>	0.85 log	
g-C ₃ N ₄ nanosheets	Chemical exfoliation	<i>E. coli</i> K-12	6.5 log	94
g-C ₃ N ₄ nanosheets	Ultrasound	<i>E. coli</i>	99%	95
bare g-C ₃ N ₄	Calcination	MS2	100%	96
Mesoporous g-C ₃ N ₄	Immersion-precipitation phase transformation	<i>E. coli</i>	3 log	87
Mesoporous g-C ₃ N ₄	Template method	<i>E. coli</i> K-12	99%	97
Mesoporous g-C ₃ N ₄	Thermal polymerization and selective dialysis approach	<i>E. coli</i>	99%	98
		<i>S. aureus</i>	90%	
Ag ₂ WO ₄ /mesoporous g-C ₃ N ₄	Polymerization	<i>E. coli</i>	100%	99
GO quantum dots/oxidized nanoporous g-C ₃ N ₄	Self-assembly	<i>E. coli</i>	99.6%	100
Nanomesh g-C ₃ N ₄	Template method	<i>E. coli</i> K-12	85%	101
CuInSe ₂ :Zn/g-C ₃ N ₄ /TiO ₂ nanowire	<i>In situ</i> growth	<i>S. aureus</i>	90%	102
Mesoporous g-C ₃ N ₄	Thermal, polycondensation	<i>E. coli</i>	Effective	103
Porous g-C ₃ N ₄	Calcination	<i>S. aureus</i>	99%	104
Mesoporous g-C ₃ N ₄	Calcination	<i>E. coli</i>	Effective	105
Nanostructured g-C ₃ N ₄	Calcination	<i>E. coli</i> O157:H	97.1%	106
		<i>S. aureus</i>	93.7%	
Mesoporous g-C ₃ N ₄	Hydrothermal	<i>E. coli</i>	Effective	107
g-C ₃ N ₄ powder	Calcination	<i>E. coli</i>	Effective	108
		<i>S. epidermidis</i>		



in *E. coli* communities over the same time period. Additionally, under visible light, MCN-PVDF degrades the antibiotic cefotaxime (CFX) with a degradation rate of over 97% over five cycles.

When g-C₃N₄ has a large specific surface area its photocatalytic performance is enhanced, e.g. in g-C₃N₄ nanosheets and nanotubes, which have a high density active sites. Furthermore, g-C₃N₄ nanocomposites inhibit e⁻/h⁺ pair recombination and have high charge transfer efficiency due to enhanced visible light absorption. Hence, discrepant topologies of g-C₃N₄ should have outstanding antibacterial performance and the discrepant

topologies of g-C₃N₄-based materials and their corresponding antibacterial properties are summarized in Table 1.

2.2 Noble metal decorated g-C₃N₄ nanocomposites

Another strategy used to increase the antibacterial performance of g-C₃N₄ involves modification with noble metal nanoparticles, including silver and gold nanoparticles.^{109,110} Surface Plasmon Resonance (SPR) of nanoparticles enhances antibacterial efficiency, by extending the spectral absorption range and promoting the formation of photogenerated carriers in g-C₃N₄.^{111–114} Additionally, noble metal nanoparticles act as electron traps, capturing free e⁻ and thus inhibiting the

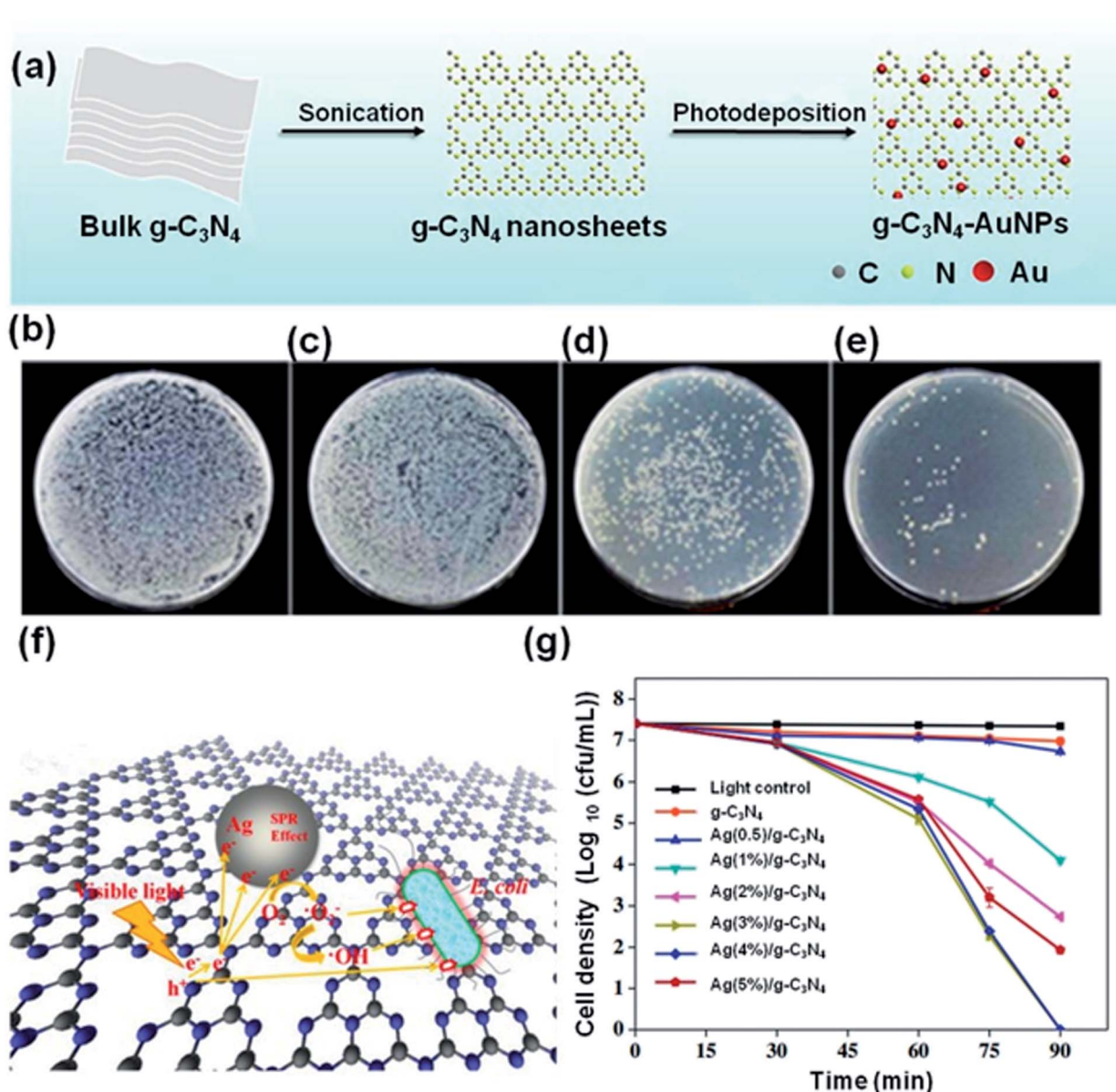


Fig. 5 (a) Preparation of g-C₃N₄-Au nanoparticle nanocomposites. (b) *E. coli* bacteria + control sample. (c) *E. coli* bacteria + g-C₃N₄-1.0% Au. (d) *E. coli* bacteria + g-C₃N₄-1.0% Au + 10 min irradiation. (e) *E. coli* bacteria + g-C₃N₄-1.0% Au + 20 min irradiation. Reproduced from ref. 123 with permission from American Chemical Society, copyright 2019. (f) The mechanism of *E. coli* inactivation in the presence of Ag/g-C₃N₄ under visible light. (g) Disinfection efficiencies of *E. coli* by the samples under visible light irradiation. Reproduced from ref. 119 with permission from Elsevier, copyright 2019.

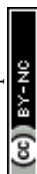
recombination of photogenerated carriers.^{115–118} Ma *et al.* developed Ag/g-C₃N₄ nanocomposites by combining thermal polymerization with photo-assisted reduction.¹¹⁹ A synergistic antibacterial efficiency was achieved with superior sterilization activity of the Ag/g-C₃N₄ nanocomposite compared to pure g-C₃N₄ (Fig. 5f). Ag(0.3 wt%)/g-C₃N₄ exhibited prominent antibacterial performance and suppressed *E. coli* replication (7.41 log) with only 1.25 h of visible light illumination. In contrast, pure g-C₃N₄ displayed very low inactivation, with only about (0.4 log) *E. coli* killed following 1.5 h of illumination by visible light (Fig. 5g). Notably, the loading of noble Ag nanoparticles on the g-C₃N₄ nanosheets significantly increases the visible light absorption region due to the SPR effect of the Ag nanoparticles and the charge transfer between the Ag and the g-C₃N₄ nanosheets.¹²⁰ Similarly, the strong and unique surface plasmon resonance (SPR) absorption of gold nanoparticles covers a wide range of spectra, including the visible and near-infrared light (NIR).^{121,122} As shown in Fig. 5a, Dai *et al.* utilized 5–10 nm sized Au nanoparticles to modify g-C₃N₄ via liquid-phase exfoliation of g-C₃N₄ combined with the photo-deposition of Au nanoparticles.¹²³ When a mixture of *E. coli* and the Au/g-C₃N₄ nanocomposite were irradiated at 670 nm the resulting ROS effectively kill the bacteria. The viability of the bacteria continually diminishes over the illumination period (Fig. 5b–e). The incorporation of Au nanoparticles into the g-C₃N₄ nanosheets strikingly improves photocatalytic ROS generation, due to the application of 670 nm light.¹²⁴ In general, noble metal/g-C₃N₄ nanocomposites significantly outperform unmodified g-C₃N₄ in antimicrobial experiments, and provide a viable photocatalytic disinfection method, see Table 2 for a summary.

2.3 Non-noble metal doped g-C₃N₄ nanocomposites

While g-C₃N₄ nanocomposites with noble metals have been shown to improve the photocatalytic antibacterial properties of the material, the high cost of noble metals prohibits widespread applications.^{125–129} Thus, g-C₃N₄ nanocomposites derived from inexpensive and abundant elements that are non-toxic would be advantageous.^{130–134} In such materials the g-C₃N₄ band gap may even be reduced to improve the separation efficiency of photo-generated carriers and the photoabsorption region may even be expanded to further improve the photocatalytic antibacterial performance.^{135–142} Surface engineering of carbon-based materials has been an effective tool for construction of materials with special functions.^{143–149} Advantageously, Lewis basic N-sites on the surface g-C₃N₄ allow strong interactions with Lewis acids, *i.e.* zinc ions, similar to that observed for other materials.^{150–157} For example, g-C₃N₄-Zn²⁺@graphene oxide (SCN-Zn²⁺@GO) were prepared using chemical vapor deposition (CVD).¹⁵⁸ The bidentate ligand, SCN, may coordinate to the Zn²⁺ ions to form cross-links with GO, and additionally changing the crystal structure of g-C₃N₄ and introducing defect sites (Fig. 6a). The resulting SCN-Zn²⁺@GO nanocomposite possessed excellent antibacterial activity. Irradiation at 808 nm (NIR) led to heating and irradiation at 660 nm resulted in the generation of ROS and the combination of photothermal and photodynamic processes effectively killed bacteria within a short time (almost quantitatively under the conditions employed). In Fig. 6b and d, the *E. coli* and *S. aureus* blank groups possess unbroken topologies, with glossy bacteria membranes and intact intra-cell structures. The membrane structures of both *E. coli* and *S. aureus* are ruptured under 808 nm and 660 nm light illumination. The intra-cell density decreases and part of cytoplasm overflows

Table 2 Antibacterial properties of noble metal decoration and non-noble metal doped g-C₃N₄ nanocomposites

Main component	Material	Preparation	Bacteria	Effect	Ref.
Noble metal decoration	Ag/g-C ₃ N ₄	Thermal polymerization and photo-assisted reduction	<i>E. coli</i>	7.41 log	119
	Ag/g-C ₃ N ₄	Single-pot, microemulsion	<i>E. coli</i>	9.95 log	166
	Ag/P/g-C ₃ N ₄	Pyrolysis and green reduction.	<i>E. coli</i>	7 log	167
	Ag/polydopamine/g-C ₃ N ₄	Ultrasound and freeze-drying	<i>E. coli</i>	Effective	168
	Ag/oxidized porous g-C ₃ N ₄	Photo-assisted, reduction	<i>S. aureus</i>	99%	169
	Ag/g-C ₃ N ₄	Biogenic, methodology	<i>E. coli</i>	Effective	174
			<i>S. aureus</i>		
			<i>P. aeruginosa</i>		
	Ag/g-C ₃ N ₄	Calcination	<i>E. coli</i>	Effective	175
			<i>S. aureus</i>		
Non-noble metal doping			<i>B. subtilis</i>		
			<i>P. aeruginosa</i>		
	Ag/g-C ₃ N ₄	Photo-deposition method	<i>E. coli</i>	Effective	177
			<i>P. aeruginosa</i>		
	Au/g-C ₃ N ₄	Liquid-phase, exfoliation and photodeposition	<i>E. coli</i>	Effective	123
	Au/g-C ₃ N ₄	Deposition-precipitation	<i>E. coli</i>	82%	170
			<i>S. aureus</i>	79%	
	Au/g-C ₃ N ₄	Calcination	<i>E. coli</i>	99%	176
	g-C ₃ N ₄ -Zn ²⁺ @graphene	Chemical vapor deposition and ultrasonic dispersion	<i>S. aureus</i>	99%	158
	CQDs/g-C ₃ N ₄	Impregnation	<i>S. aureus</i>	7 log	165
	Fullerene/g-C ₃ N ₄	Hydrothermal method	<i>E. coli</i> O157:H7	86%	171
	S-CQD/hollow tubular g-C ₃ N ₄	Self-assembly and ultrasound	<i>E. coli</i>	6.88 log	172
	Ag/P/Co/S/g-C ₃ N ₄	Calcination	<i>E. coli</i>	7 log	173



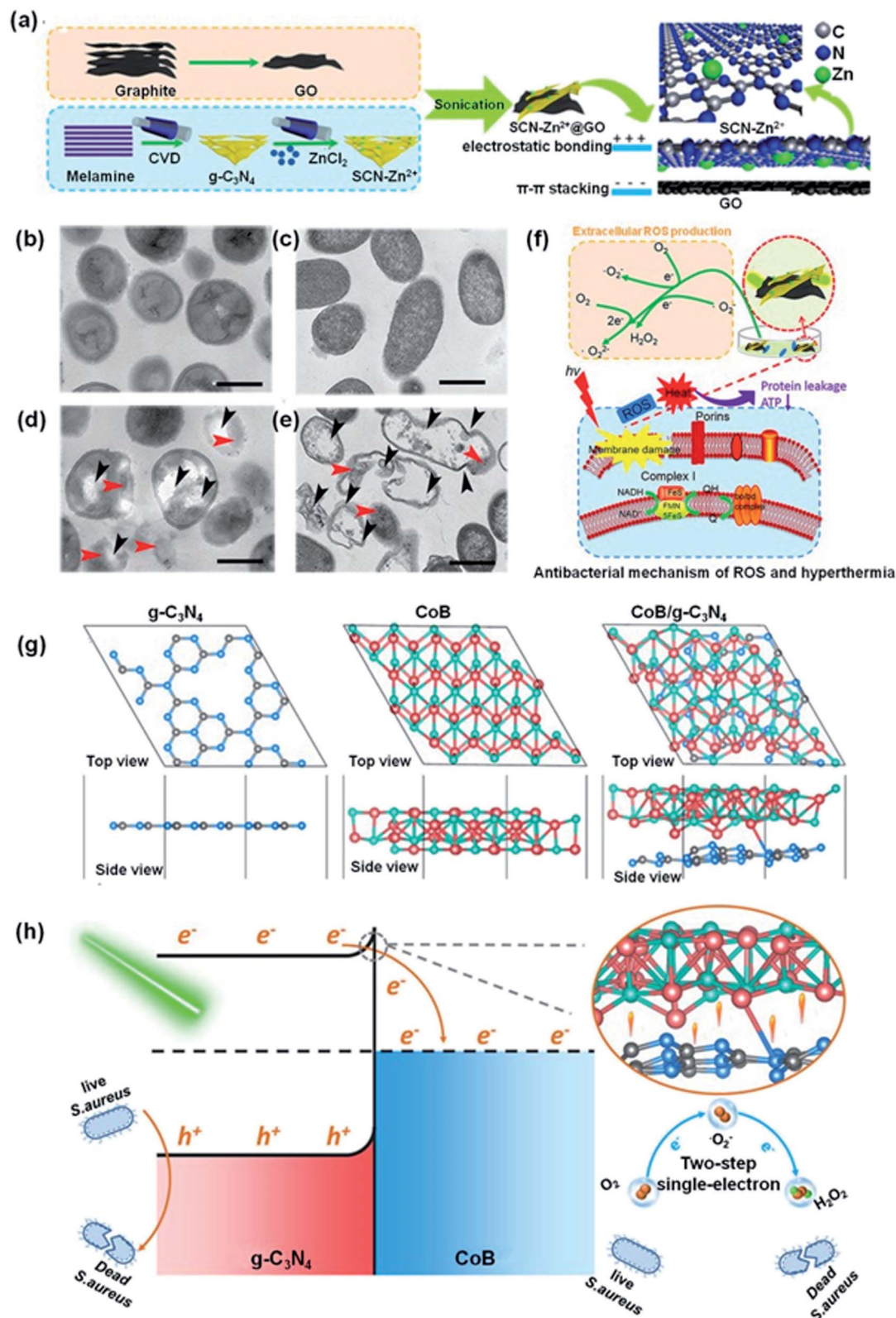


Fig. 6 (a) Preparation of SCN-Zn²⁺@GO. TEM topology of *S. aureus* (b) and *E. coli* (c) as control, (d) and (e) following treatment with SCN-Zn²⁺@GO 20% after 10 min irradiation with visible light (the red arrows indicate protein leakage and the dark arrows indicate rupture or ruffling of the bacterial membrane). (f) Antibacterial mechanism of SCN-Zn²⁺@GO 20% under 808 or 660 nm irradiation of ROS and hyperthermia. Reproduced from ref. 158 with permission from WILEY-VCH, copyright 2018. (g) Structural models of the g-C₃N₄ (001) surface, CoB-(010) surface and CoB/g-C₃N₄ after geometry optimization. (h) The mechanism of *S. aureus* bacteria inactivation in the presence of CoB/g-C₃N₄ under visible light. Reproduced from ref. 159 with permission from the American Chemical Society, copyright 2019.

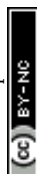


Table 3 Antibacterial properties of g-C₃N₄ heterojunction nanocomposites

Material	Preparation	Bacteria	Effect	Ref.
Bi ₂ S ₃ /g-C ₃ N ₄	Ultrasound	<i>E. coli</i> <i>S. aureus</i>	99.6% 99.2%	185
Red P/g-C ₃ N ₄	Sonochemical	<i>E. coli K-12</i>	7 log	186
Perylene diimide/oxygen-doped g-C ₃ N ₄	<i>In situ</i> electrostatic assembling	<i>S. aureus</i>	99.6%	196
Bi ₂ MoO ₆ /g-C ₃ N ₄	<i>In situ</i> solvothermal	<i>E. coli</i>	100%	199
MnO ₂ /g-C ₃ N ₄	Thermal vapor liquid-polymerization and redox	<i>E. coli</i> <i>S. aureus</i>	99.96% 99.26%	212
TiO ₂ /kaolinite/g-C ₃ N ₄	Sol-gel method	<i>S. aureus</i>	2.9 log	220
Ag/AgBr/g-C ₃ N ₄ /nitrogen-doped graphene aerogel	Hydrothermal and freeze-drying	<i>E. coli</i> <i>S. aureus</i>	~6 log ~1.2 log	252
Ag/AgBr/g-C ₃ N ₄	<i>In situ</i> deposition-precipitation	<i>E. coli</i>	7.9 log	253
Ag ₂ WO ₄ /g-C ₃ N ₄	Ultrasound	<i>E. coli</i>	90%	222
Ag ₂ PO ₄ /g-C ₃ N ₄	Co-precipitation and thermal pyrolysis	<i>E. coli</i> <i>S. aureus</i>	Effective	254
Ni ₂ P/g-C ₃ N ₄	<i>In situ</i> anchoring and hydrothermal	<i>E. coli K-12</i>	7 log	255
<i>m</i> -Bi ₂ O ₄ /g-C ₃ N	Hydrothermal	<i>E. coli K-12</i>	6 log	223
Vanadium modified g-C ₃ N ₄ /TiO ₂	Calcination and ultrasonic	<i>E. coli</i> <i>S. aureus</i>	Effective	256
BiVO ₄ /Ag/g-C ₃ N ₄	Photodeposition and hydrothermal	<i>E. coli</i>	6.5 log	225
Ag/ZnO/g-C ₃ N ₄	Thermal polymerization and solvothermal	<i>E. coli</i>	7.4 log	226
Ag/ZnO/g-C ₃ N ₄	Thermal polymerization and solvothermal	<i>E. coli</i>	6.19 log	257
Ag ₂ WO ₄ /Ag/g-C ₃ N ₄	Hydrothermal and situ reductive	<i>E. coli</i>	3.05 log	258
Bi ₂ MoO ₆ -Ag/g-C ₃ N ₄	Hydrothermal method	<i>E. coli</i> <i>S. aureus</i>	Effective	259
α-Fe ₂ O ₃ /CeO ₂ /g-C ₃ N ₄	Hydrothermal method	<i>E. coli</i> <i>S. aureus</i>	Effective	260
CuS/g-C ₃ N ₄	Electrostatic adsorption	<i>E. coli</i> <i>S. aureus</i>	99% 98%	250
Vanadate QDs/g-C ₃ N ₄	Sol-gel method	<i>Salmonella H9812</i>	96%	251
BiOI/g-C ₃ N ₄	<i>In situ</i> generation	<i>E. coli</i> <i>S. aureus</i>	96% 98%	261
Cu ₂ O/g-C ₃ N ₄	Chemical precipitation	<i>E. coli</i>	7 log	262
Cu ₂ O/g-C ₃ N ₄	Hydrothermal method	<i>E. coli</i> <i>S. aureus</i>	Effective	263
TiO ₂ nanofibers/Ag/g-C ₃ N ₄	Ultrasound	<i>E. coli</i> <i>S. aureus</i>	99% 83%	264
Ag/AgCl/g-C ₃ N ₄	<i>In situ</i> implanting and anchoring	Tetracycline-resistant bacteria	100%	265
γ-Fe ₂ O ₃ /Ag/AgCl/g-C ₃ N ₄	Solvothermal and photodeposition	<i>E. coli</i>	5.59 log	266
RGO/CA/g-C ₃ N ₄	Ultrasonification	<i>E. coli</i>	6.5 log	267
TiO ₂ /CuBA/g-C ₃ N ₄	Ultrasound	<i>E. coli</i> <i>S. aureus</i>	Effective	268
Ag ₂ ZrO ₃ /g-C ₃ N ₄	Co-precipitation	<i>E. coli</i> <i>B. subtilis</i>	97% 99%	189
AgBr/g-C ₃ N ₄	Adsorption deposition	<i>E. coli</i>	6.5 log	269
BiFeO ₃ /Cu ₂ O/g-C ₃ N ₄	Hydrothermally and ultrasonic	<i>E. coli</i> <i>S. aureus</i>	Effective	270
ZnO/g-C ₃ N ₄ /cellulose	Ultrasonic irradiation	<i>E. coli</i> <i>S. aureus</i>	Effective	271
CdS/g-C ₃ N ₄	Sonochemical	<i>E. coli</i> <i>S. aureus</i>	Effective	272
GO/g-C ₃ N ₄	Sonication	<i>E. coli</i>	97.9%	70
AgCl/CNTs/g-C ₃ N ₄	Deposition-precipitation	<i>E. coli</i>	Effective	273
MoO _{3-x} /g-C ₃ N ₄	Hydrothermal	<i>E. coli</i>	100%	274
AgO/g-C ₃ N ₄	Chemical oxidation	<i>S. aureus</i>	89%	275
Fe-2,5-thiophenedicarboxylic acid/g-C ₃ N ₄	Microwave-heating	<i>E. coli</i>	100%	276
Mg _{1.2} Ti _{1.8} O ₅ /g-C ₃ N ₄	Sol-gel method and calcination	<i>E. coli</i>	100%	277
CuWO ₄ /g-C ₃ N ₄	Sol-gel method	<i>E. coli</i> <i>S. aureus</i>	Effective	278
ZnBi ₂ O ₄ /g-C ₃ N ₄	Ultrasound-assisted chemical exfoliation	<i>E. coli</i> <i>S. aureus</i>	Effective	279
Cr-ZnO/g-C ₃ N ₄	Chemical coprecipitation method	<i>E. coli</i> <i>S. aureus</i>	Effective	280



Table 3 (Contd.)

Material	Preparation	Bacteria	Effect	Ref.
ZnBi ₂ O ₄ /g-C ₃ N ₄	Thermal polycondensation	<i>B. subtilis</i>		
TiO ₂ /Ag/g-C ₃ N ₄	Vacuum freeze-drying	<i>E. coli</i>	97%	281
AgBr/g-C ₃ N ₄	Calcination	<i>E. coli</i>	84%	282
TiO ₂ nanotubes/Ti/g-C ₃ N ₄ /SnO ₂	Dipping and calcination	<i>P. putida</i>	100%	283
BiOCl/g-C ₃ N ₄	Hydrothermal method	<i>E. coli</i>	Effective	284
NiFe ₂ O ₄ /g-C ₃ N ₄	Hydrothermal method	<i>E. coli</i>	96%	285
Perylene-3,4,9,10-tetracarboxylic diimide/g-C ₃ N ₄	<i>In situ</i>	<i>A. flavus</i>	90%	286
TiO ₂ /g-C ₃ N ₄	<i>In situ</i>	<i>E. coli</i>	Effective	287
ZnO/Mn/g-C ₃ N ₄	Chemical co-precipitation	<i>S. aureus</i>	65%	288
ZnTiO ₂ /S/g-C ₃ N ₄	<i>In situ</i>	<i>E. coli</i>	Effective	289
Ag-ZnO@g-C ₃ N ₄	Physical mixing method	<i>S. aureus</i>	Effective	290
		<i>E. coli</i>		
		<i>S. aureus</i>		
		<i>B. subtilis</i>		
Poly(vinyl alcohol)/g-C ₃ N ₄	Casting	<i>P. aeruginosa</i>	Effective	291
Polyaniline/g-C ₃ N ₄	<i>In situ</i> oxidative polymerization methodology	<i>E. coli</i>	Effective	292
PVA/Starch/Ag@TiO ₂ /g-C ₃ N ₄	Solution casting	<i>S. pneumoniae</i>		
		<i>E. coli</i>	Effective	293
		<i>S. aureus</i>		
Fe@ZnO/g-C ₃ N ₄	Chemical co-precipitation	<i>E. coli</i>	Effective	294
		<i>S. aureus</i>		
		<i>B. subtilis</i>		
		<i>S. salivarius</i>		
g-C ₃ N ₄ -based metal-free	Calcination	<i>E. coli</i>	Effective	295
		<i>B. subtilis</i>		
Ag ₂ S/g-C ₃ N ₄	Sonochemical	<i>E. coli</i>	Effective	296
		<i>S. aureus</i>		
		<i>B. subtilis</i>		
		<i>S. salivarius</i>		
TiO ₂ nanofibers/g-C ₃ N ₄	Electrospinning-calcination	<i>E. coli</i>	100%	297
		<i>S. aureus</i>		
Ag ₃ PO ₄ /g-C ₃ N ₄	Hydrothermal method	<i>E. coli</i>	Effective	298
Ag ₂ O/g-C ₃ N ₄	Chemical deposition method	<i>M. aeruginosa</i>	99%	299
TiO ₂ /g-C ₃ N ₄	Hydrothermal and calcination	<i>E. coli</i>	100%	300
Ag ₂ O/g-C ₃ N ₄	Physical mixing method	<i>M. aeruginosa</i>	100%	301
AgO/g-C ₃ N ₄	<i>In situ</i>	<i>E. coli</i>	Effective	302
Ag ₃ PO ₄ /g-C ₃ N ₄	Hydrothermal method	<i>Bacteriophage f2</i>	100%	303
AgBr/g-C ₃ N ₄	<i>In situ</i>	<i>E. coli</i>	100%	304
TiO ₂ /g-C ₃ N ₄	Hydrothermal	<i>E. coli</i>	100%	305
Porphyrin/g-C ₃ N ₄	<i>In situ</i>	<i>S. aureus</i>	63%	306
Ag/Ag/Ag/g-C ₃ N ₄ /BiVO ₄	<i>In situ</i>	<i>E. coli</i>	100%	307
ZnO-Cd/g-C ₃ N ₄	<i>In situ</i>	<i>E. coli</i>	Effective	308
		<i>S. aureus</i>		
RGO/S ₈ /g-C ₃ N ₄	<i>In situ</i>	<i>E. coli K-12</i>	100%	309

(Fig. 6c and e, red arrows indicate protein or intra-cell material leakage and the black arrows indicate bacterial membrane distortion). In comparison, when SCN-Zn²⁺@GO was exposed to either 808 nm or 660 nm illumination, the inactivation rate obtained was only 20–66%. Based on the above results, the antibacterial mechanism is proposed in Fig. 6f. The ROS pass through the cell membrane of the bacteria to oxidize intracellular proteins and interfere homeostasis, while hyperthermia weakens the activities of the proteins and reduces adenosine triphosphate synthesis, inactivating *E. coli* and *S. aureus* within 10 min. Similarly, novel CoB/g-C₃N₄ nanosheets were successfully prepared by an electrostatic self-assembly process coupled

with calcination.¹⁵⁹ The interfacial Co–N bond could act as an e[−] transport channel by accelerating the e[−] transfer from g-C₃N₄ to CoB, as supported by density functional theory (DFT) calculations and indirectly evidenced from antibacterial experiments (Fig. 6g). Consequently, the e[−] induced O₂ reduction process is promoted in CoB/g-C₃N₄, which boosts the generation of ROS (Fig. 6h). Notably, CoB/g-C₃N₄ exhibited superior disinfection efficacy of 100% against *S. aureus* with 125 min under visible light irradiation.

Quantum dots (QDs) are an important low-dimensional semiconductor materials. Because of their high reactivity and strong charge transfer abilities, QDs have been applied in



photocatalytic sterilization.¹⁶⁰ Carbon quantum dots (CQDs) were combined with g-C₃N₄ to enhance charge transfer and store e⁻.^{161–164} Tang *et al.* constructed a CQD/g-C₃N₄ photocatalyst by impregnation.¹⁶⁵ The addition of CQDs dramatically increased the disinfection performance, which was attributed to the increased ROS levels. The CQD/g-C₃N₄ nanocomposites exhibit a greatly enhanced bactericidal efficiency under illumination with visible light. In contrast, the CQDs alone showed no catalytic activity against *S. aureus* under comparable conditions. Hence, the interaction between the CQDs and g-C₃N₄ plays a significant role in increasing the bacterial inactivation efficiency.

2.4 g-C₃N₄ heterojunction nanocomposites

Heterojunctions could enable the directional migration of photoinduced charges, allowing the charge to be enriched in specific direction, a process that should reduce or even inhibit the recombination of photogenerated carriers.^{178,179} The antimicrobial properties of different g-C₃N₄ heterojunctions are discussed, including type I and type II heterojunctions, p–n heterojunctions, and Z-scheme heterojunctions.^{180–184} The photocatalytic antibacterial properties of g-C₃N₄ heterojunction nanocomposites are listed in Table 3.

2.4.1 Type I heterojunction nanocomposites. In general, type I heterojunctions are rarely considered as the optimal choice in photocatalysis because the photogenerated carriers can transfer to the interface with other semiconductors, reducing the redox capacity of the charge carriers. Nevertheless, under visible light irradiation, type I heterojunctions have the unique advantage, *i.e.* the e⁻ and h⁺ can be transferred from one semiconductor to another. If another semiconductor has a wide photoabsorption window, a broad-spectrum-response photocatalyst with minimal charge recombination can be obtained by creating a type I heterojunction (Fig. 7a). Li *et al.* developed zinc-doped g-C₃N₄ (g-C₃N₄-Zn) with Bi₂S₃ nanorod heterojunctions (g-C₃N₄-Zn/BiS), using ultrasonication.¹⁸⁵ In contrast to the precursors (g-C₃N₄-Zn and BiS), effective charge separation at the photocatalyst interface is achieved by adjusting the band gap, the density of the electronic distribution, and the oxygen adsorption capacity of the g-C₃N₄-Zn/BiS heterojunction. DFT calculations were employed to predict the stable crystal structure and the interface space between CN-Zn and BiS (Fig. 8a). The e⁻ and h⁺ were separated effectively by the energy band offset and the interface electric field, hence the g-C₃N₄-Zn/BiS heterojunction produces abundant ROS and shows excellent photocatalytic efficiency. Near-quantitative bactericidal efficiency towards *S. aureus* was achieved after only 10 min of NIR

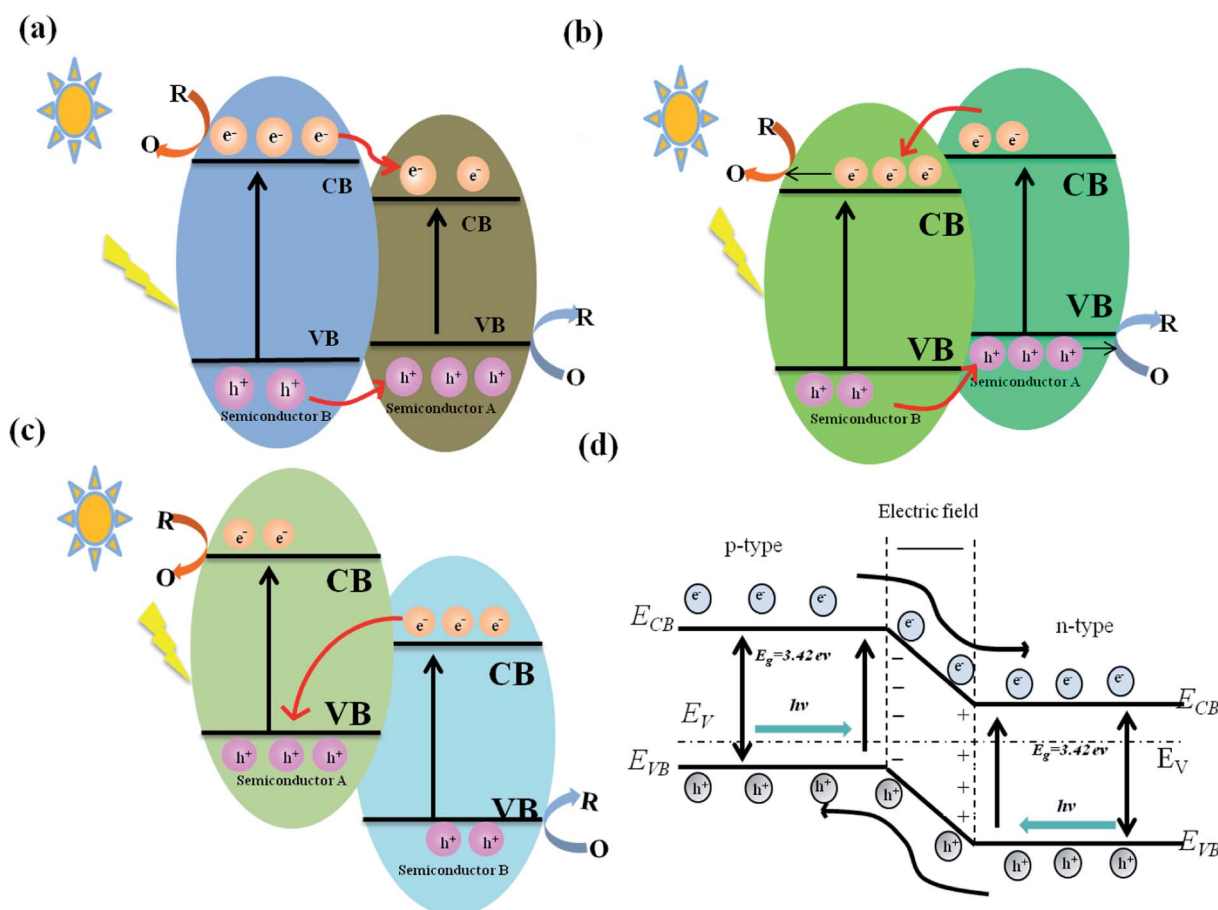


Fig. 7 Various types of heterojunctions. (a) Type I heterojunction model. (b) Type II heterojunction model. (c) Z-type heterojunction model. (d) p–n heterojunction model.



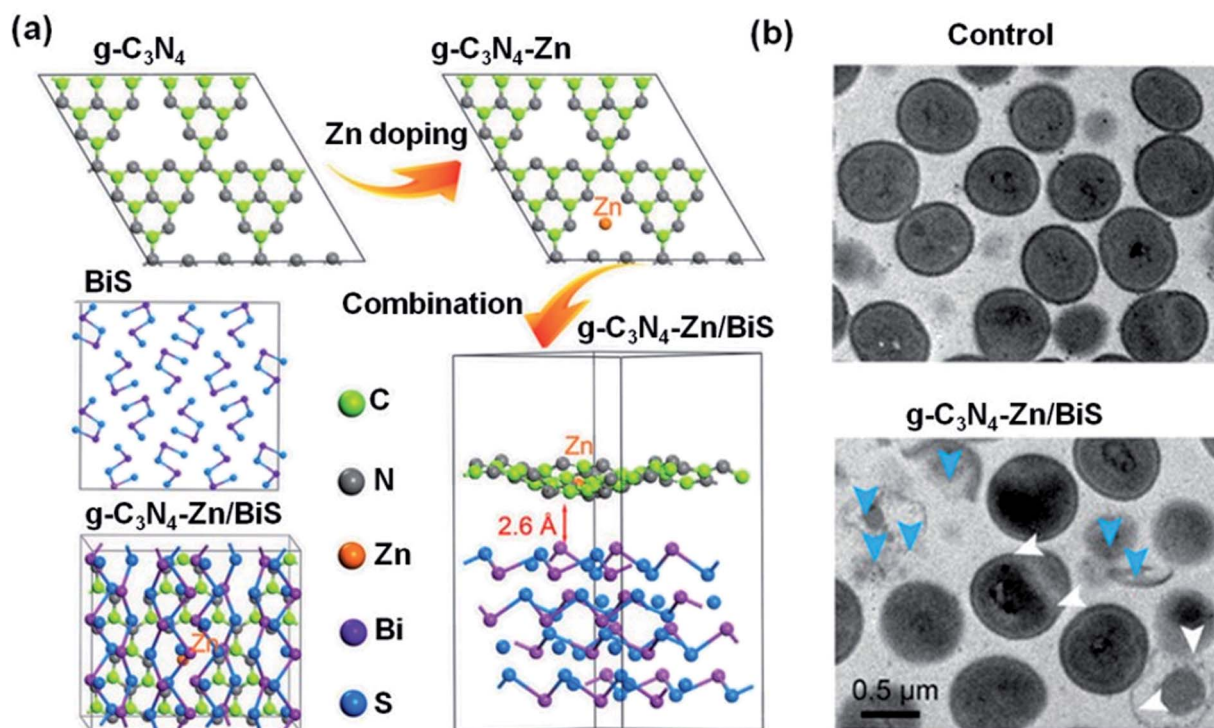


Fig. 8 (a) Structural models of g-C₃N₄, g-C₃N₄-Zn, BiS, and g-C₃N₄-Zn/BiS after geometry optimization. (b) TEM topology of *S. aureus* after treatment with control or g-C₃N₄-Zn/BiS after 10 min irradiation. The white arrows indicate twisted and broken cell membranes and the blue arrows point to intracellular matrix outflow. Reproduced from ref. 185 with permission from WILEY-VCH, copyright 2019.

irradiation (Fig. 8b). In addition, red P was a novel single-element photocatalyst, and its visible light response range is up to 700 nm. Efficient light harvesting is imperative for photocatalysts, and with this in mind Wang *et al.* developed a wide-spectral-response g-C₃N₄/red P photocatalyst using ultrasound.^{186–188} Ultrasonication was used to obtain nanosheets from bulk g-C₃N₄, and red P particles were anchored to the g-C₃N₄ nanosheets to construct close g-C₃N₄/red P heterojunctions. g-C₃N₄/red P may form a new wide-spectral-responsive photocatalytic system to fully utilize the solar energy. In addition, g-C₃N₄/red P was used as a dual activity center photocatalyst, exhibiting dramatically improved photocatalytic efficiency for sterilization under illumination with visible light. While the g-C₃N₄/red P nanocomposite showed 7 log cfu mL^{−1} bacterial inactivation after 1.3 h, the photocatalytic bacterial inactivation of pure g-C₃N₄ was limited, with < 1.5 log cfu mL^{−1} *E. coli* inactivation after 2 h of illumination.

2.4.2 Type II heterojunction nanocomposites. Type II heterojunctions g-C₃N₄-based nanocomposites have been widely reported as photocatalysts, *e.g.* Ag₂ZrO₃/g-C₃N₄,¹⁸⁹ Nb₂O₅/g-C₃N₄ and Bi₂MoO₆/g-C₃N₄.^{190,191} These materials have interlaced band gaps and appropriate VB and CB energies. Staggered heterojunctions are the most efficient type of heterojunctions due to highly efficient charge transfer,^{192–194} and therefore, type II heterojunctions are widely used.¹⁹⁵ Exposure of a type II heterojunction to visible light results in the transition of an e[−] from the VB to the CB, generating a corresponding h⁺ in the VB. When the CB of semiconductor A is higher in energy than the

CB of semiconductor B, the e[−] in the CB of semiconductor A is transferred to the CB of semiconductor B. Simultaneously, the h⁺ in the VB of semiconductor B is transferred to the VB of semiconductor A. Finally, they react with O₂ and H₂O in the surrounding media to produce ROS, leading to good antibacterial effects (Fig. 7b). Gao *et al.* prepared a perylene diimide (PDI)/oxygen-doped g-C₃N₄ nanosheet (PDI/O-g-C₃N₄) nanocomposites using an *in situ* electrostatic assembly method.¹⁹⁶ The PDI expanded the visible light range of the material, resulting in abundant photogenerated charge carriers and accumulation of ROS, boosting the oxidative capability. As a consequence, PDI/O-g-C₃N₄ demonstrated strong antibacterial oxidation activity under visible light with 96% of the *S. aureus* fully inactivated by PDI/O-g-C₃N₄ under 3 h of light irradiation, whereas only 62% of the *S. aureus* cells were inactivated by the control material. Coincidentally, Bi₂MoO₆ not only intersects the g-C₃N₄ band gap, but also has a very similar band gap energy (~2.7 eV). Consequently, Bi₂MoO₆ combines with g-C₃N₄ to afford neoteric and efficient nanocomposites.^{197,198} As shown in Fig. 9a, Li *et al.* developed Bi₂MoO₆/g-C₃N₄ heterojunctions using an *in situ* solvothermal approach.¹⁹⁹ The results showed that the photocatalyst completely inactivated 2.5 × 10⁷ cfu mL^{−1} *E. coli* after 3 h light irradiation (Fig. 9b).

2.4.3 Z-scheme heterojunction nanocomposites. Recently, Z-scheme heterojunctions have been widely studied as the structure accelerates the separation of photogenerated carriers. The e[−] in the VB of semiconductor A transfers to the CB of



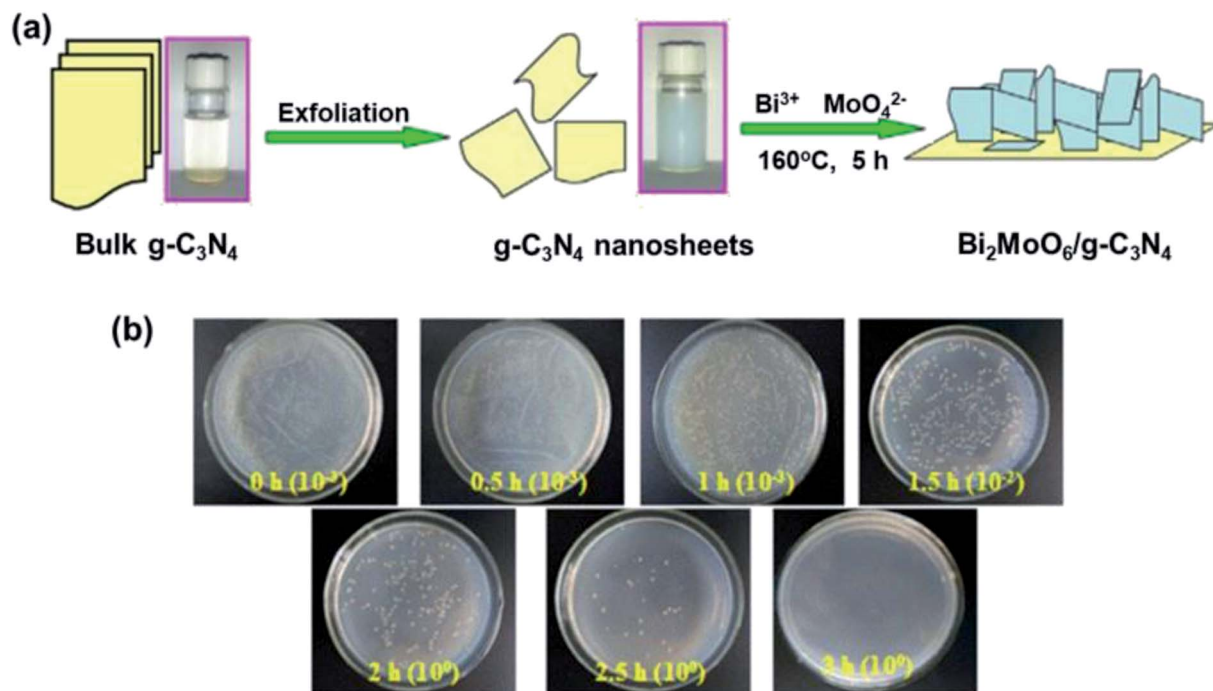


Fig. 9 (a) Preparation of the $\text{Bi}_2\text{MoO}_6/\text{g-C}_3\text{N}_4$ heterojunction. (b) *E. coli* re-cultured after treatment with 20% $\text{Bi}_2\text{MoO}_6/\text{g-C}_3\text{N}_4$ as a function of time. Reproduced from ref. 199 with permission from Elsevier, copyright 2016.

semiconductor B, and the remaining h^+ and e^- undergo redox reactions with the oxygen and water in the surroundings to generate ROS (Fig. 7c). As expected, Z-scheme heterojunctions exhibit excellent photocatalytic disinfection performance.^{200–204} MnO_2 is an inexpensive, abundant, biocompatible semiconductor that has a similar bandgap to $\text{g-C}_3\text{N}_4$.^{205–211} Wu *et al.* successfully constructed a $\text{MnO}_2/\text{g-C}_3\text{N}_4\text{-Ti}$ heterojunction using thermal vapor liquid-polymerization and redox methods²¹² (Fig. 10a). Contact between the $\text{g-C}_3\text{N}_4$ and MnO_2 formed a Z-scheme heterojunction. The $\text{MnO}_2/\text{g-C}_3\text{N}_4\text{-Ti}$ composite inactivates *S. aureus* and *E. coli* in near-quantitative yields (Fig. 10b and c). In addition, TiO_2 is an outstanding photocatalyst that binds with $\text{g-C}_3\text{N}_4$ for form a nanocomposite with high thermal stability.^{213–219} Li *et al.* constructed a $\text{g-C}_3\text{N}_4/\text{TiO}_2/\text{kaolinite}$ heterojunction using a sol-gel approach combined with self-assembly²²⁰ (Fig. 10d). Compared to bulk $\text{g-C}_3\text{N}_4$ or TiO_2 , the 3D structured $\text{g-C}_3\text{N}_4/\text{TiO}_2/\text{kaolinite}$ nanocomposite displayed increased adsorption-photocatalytic sterilization of *S. aureus* under light irradiation. The $\text{g-C}_3\text{N}_4/\text{TiO}_2/\text{kaolinite}$ composite inactivated 2.9 log *S. aureus* bacteria after 5 h of illumination, superior to $\text{g-C}_3\text{N}_4$ (1.6 log) and TiO_2 (0.8 log) alone (Fig. 10e). In addition, under visible light, the $\text{g-C}_3\text{N}_4/\text{TiO}_2/\text{kaolinite}$ nanocomposite exhibits heightened adsorption-photocatalytic degradation of ciprofloxacin, a broad-spectrum antibiotic. The antibacterial efficiency of the $\text{g-C}_3\text{N}_4/\text{TiO}_2/\text{kaolinite}$ composite may be attributed to both the improved light utilization and an increase in e^- transfer and separation efficiency (Fig. 10f). The visible light activated $\text{g-C}_3\text{N}_4/\text{TiO}_2/\text{kaolinite}$ composite is a useful material for pollutant decomposition and bacterial elimination.²²¹

2.4.4 Dual-path heterojunction nanocomposites. Generally, the charge migration paths observed in $\text{g-C}_3\text{N}_4$ heterojunctions are mostly type II and Z-scheme heterojunctions, which expedite the fast separation of photogenerated charges and intensify the antibacterial activity of semiconductor materials. Many type II and Z-scheme heterojunction nanocomposites have been shown to inactivate bacteria under irradiation with visible light, including $\text{m-Bi}_2\text{O}_3/\text{g-C}_3\text{N}_4$, $\text{AgWO}_4/\text{g-C}_3\text{N}_4$.^{222,223} Nevertheless, due to the relatively low redox potentials in type II and Z-scheme heterojunctions, these photocatalysts lack strong redox abilities.²²⁴ It is known that e^- accumulate in the CB of semiconductor A, which has a high reduction potential, and h^+ leave the VB of semiconductor B, which has a high oxidation potential. This not only effectively separates the e^-/h^+ pairs, but also produces the optimal redox properties. Therefore, the two models of ternary heterojunctions were also studied to further improve the antibacterial performance of photocatalytic heterojunctions. Zeng *et al.* constructed a ternary $\text{BiVO}_4/\text{Ag}/\text{g-C}_3\text{N}_4$ heterojunction using photo-deposition and hydrothermal methods.²²⁵ Based on heterojunction band gap energy level and surface chemistry, a dual Z-scheme photogenerated carrier transfer model was applied to $\text{BiVO}_4/\text{Ag}/\text{g-C}_3\text{N}_4$. Notably, the ternary $\text{BiVO}_4/\text{Ag}/\text{g-C}_3\text{N}_4$ heterojunction markedly strengthened the photocatalytic antibacterial capability, completely inactivating 6.5 log *E. coli* cells after 1 h of light illumination, whereas the binary $\text{BiVO}_4/\text{g-C}_3\text{N}_4$ heterojunction inactivated only 0.5 log *E. coli* under comparable conditions. The Ag and BiVO_4 nanoparticles on the $\text{g-C}_3\text{N}_4$ nanosheets inhibit recombination of the photogenerated carriers, thus promoting ROS generation. Ma *et al.* developed



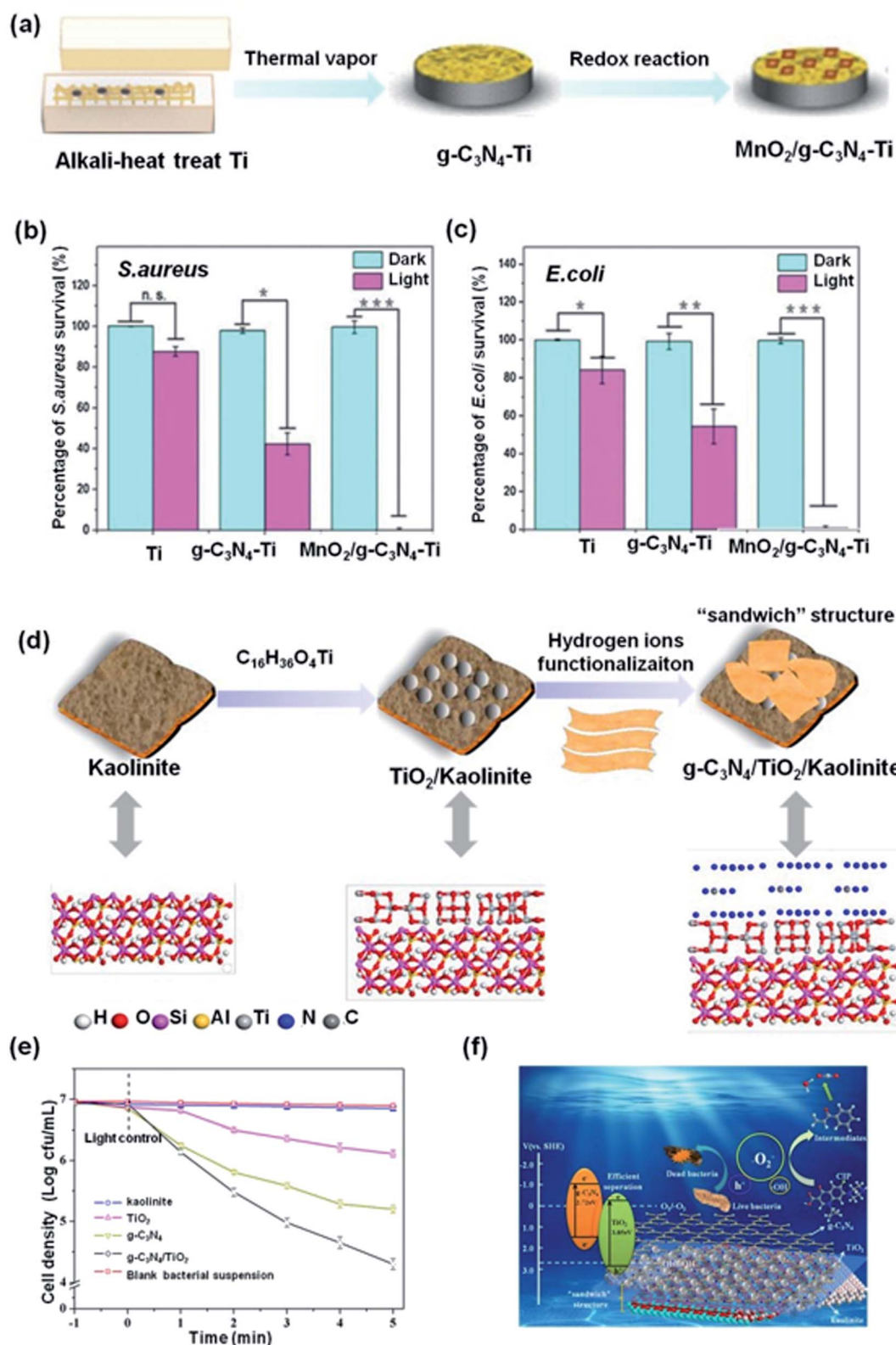
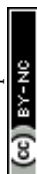


Fig. 10 (a) Schematic showing the preparation of $\text{MnO}_2/\text{g-C}_3\text{N}_4\text{-Ti}$. (b and c) The antibacterial effect of $\text{MnO}_2/\text{g-C}_3\text{N}_4\text{-Ti}$ irradiated for 20 minutes against *S. aureus* and *E. coli*, respectively. Reproduced from ref. 212 with permission from Elsevier, copyright 2019. (d) Schematic illustration of the preparation of $\text{g-C}_3\text{N}_4/\text{TiO}_2/\text{kaolinite}$. (e) Photocatalytic disinfection efficiency of *S. aureus* for different samples. (f) Schematic diagram of photocatalytic mechanism of the $\text{g-C}_3\text{N}_4/\text{TiO}_2/\text{kaolinite}$. Reproduced from ref. 220 with permission from Elsevier, copyright 2019.



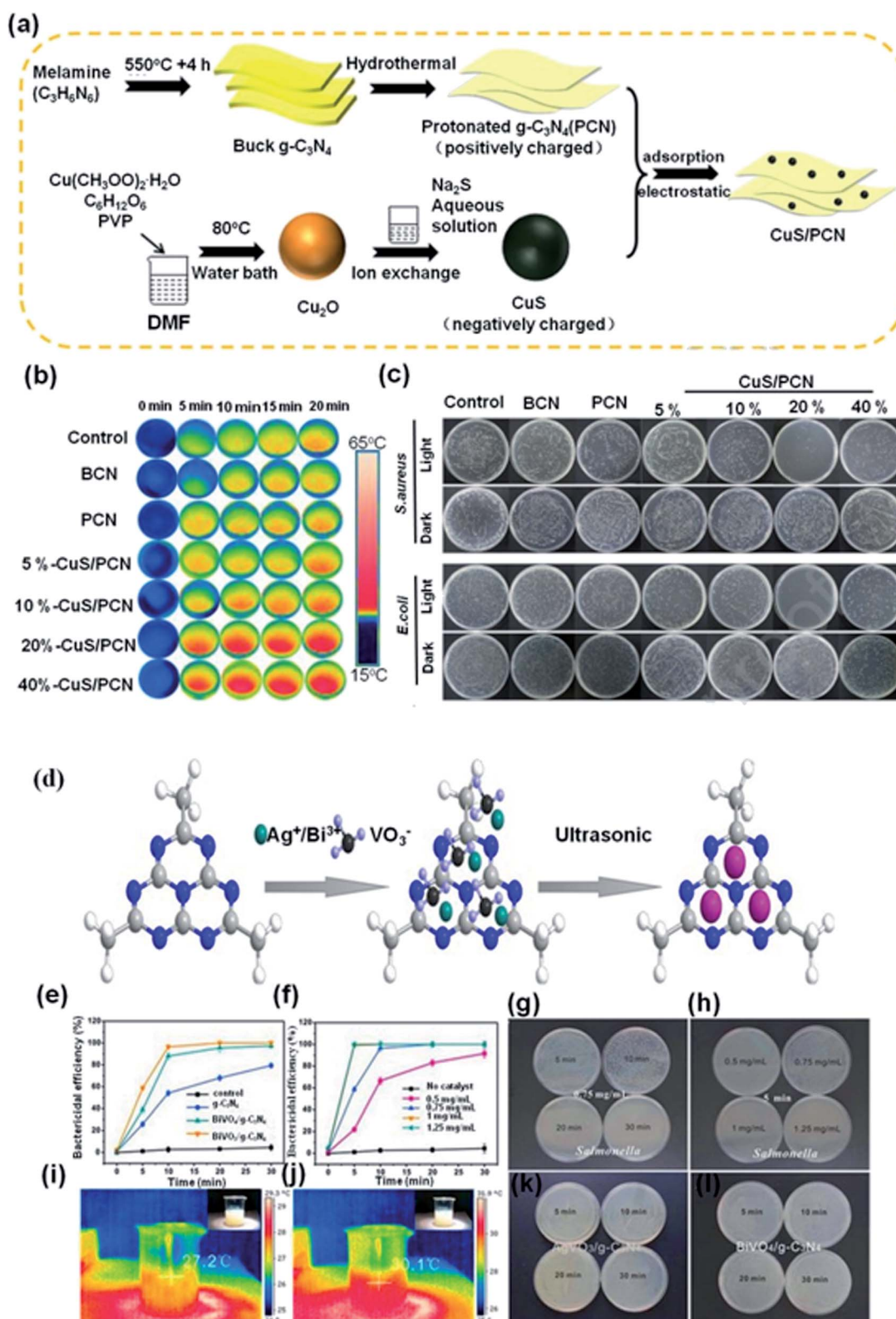


Fig. 11 (a) Preparation of CuS/g-C₃N₄. (b) Photothermal images following irradiation as a function of concentration and time. (c) *In vitro* antibacterial activity for *S. aureus* and *E. coli*. Reproduced from ref. 250 with permission from Elsevier, copyright 2020. (d) Formation mechanism of vanadate QDs/g-C₃N₄. (e) Photocatalytic disinfection efficiency of *Salmonella* with different samples. (f) Photocatalytic disinfection efficiency of *Salmonella* with AgVO₃/g-C₃N₄ at different concentrations. (g) Bacteria colony growth in the presence of AgVO₃/g-C₃N₄ with *Salmonella*. (h) Bacteria colony growth in the presence of AgVO₃/g-C₃N₄ with *Salmonella* at different concentrations. The corresponding thermal images of AgVO₃/g-C₃N₄ following irradiation for (i) 5 min and (j) 10 min. Bacteria colony growth with (k) AgVO₃/g-C₃N₄ and (l) BiVO₄/g-C₃N₄ in the dark. Reproduced from ref. 251 with permission from Elsevier, copyright 2017.

ZnO/Ag/g-C₃N₄ heterojunction using a solvothermal reaction.²²⁶ This composite was used to kill *E. coli* under illumination with visible light. The ZnO/Ag/g-C₃N₄ composite demonstrated significant visible light sterilization efficiency compared to g-C₃N₄, Ag/g-C₃N₄ and ZnO/g-C₃N₄ materials. Specifically, ZnO/Ag/g-C₃N₄ inactivated 7.4 log *E. coli* after 2 h light illumination. However, only 0.49 log and 2.61 log *E. coli* were inactivated by g-C₃N₄ and ZnO/g-C₃N₄. The interface of ZnO improves the sterilization performance by increasing the separation rate of charges because of the SPR effect of Ag and the similar band gap energies of ZnO to g-C₃N₄.^{227–231}

2.4.5 p–n heterojunction nanocomposites. The construction of p–n type heterojunctions can increase the spectral response range of photocatalytic semiconductors.^{232–234} The p–n type heterojunction needs to form at the interface of the space charge region and these heterojunctions form an internal potential that guides the e[−] and h⁺ in opposite directions.^{235–239} The e[−] transfers to the CB of the n-type semiconductor, whereas the h⁺ transfers to the VB of the p-type semiconductor (Fig. 7d). The separation effect of charges in p–n heterojunctions is higher than other heterojunctions leading to superior photocatalytic antibacterial activity.^{240–245} CuS is a p-type semiconductor and is the material of choice for photocatalysis due to its narrow band gap and excellent physicochemical stability.^{246–249} Ding *et al.* synthesized CuS/g-C₃N₄ heterojunction using a hydrothermal approach harnessing electrostatic adhesion.²⁵⁰ In the CuS/g-C₃N₄ heterojunction (Fig. 11a), the e[−] and h⁺ transfer in reverse directions between g-C₃N₄ and CuS, accelerating the separation of charges, thus producing high levels of ROS and increasing the photocatalytic antibacterial activity. In addition, the CuS/g-C₃N₄ heterojunction can transform visible light to heat (Fig. 11b). Hence, due to the synergistic influence of the ROS and thermal effects, the CuS/g-C₃N₄ composite inactivated *E. coli* and *S. aureus* bacteria near-quantitatively after 20 min of light irradiation. In contrast, g-C₃N₄ only inactivated 30% of the *E. coli* cells and 25% of the *S. aureus* cells (Fig. 11c).

Wang *et al.* fabricated vanadate (AgVO₃ and BiVO₄) QD/g-C₃N₄ nanocomposites using urea²⁵¹ (Fig. 11d). Due to the abundant production of ROS by the vanadate QDs and g-C₃N₄, the vanadate QDs/g-C₃N₄ composites exhibited high bactericidal efficiency, with 96% inactivation (AgVO₃ QDs/g-C₃N₄) and 87% inactivation (BiVO₄ QDs/g-C₃N₄) of *Salmonella* after only 10 min of light illumination (Fig. 11e and g). As shown in Fig. 11f and h, the photocatalytic disinfection efficiency increases with increasing photocatalyst concentration. Only 22% of *Salmonella* were killed with a AgVO₃/g-C₃N₄ composite concentration of 0.5 mg mL^{−1}. However, at the same period, the photocatalytic inactivation of *Salmonella* increased to 58% when the photocatalyst concentration reach 0.75 mg mL^{−1}. It is apparent from Fig. 11i and j that there is no significant change in temperature during the antibacterial tests. Furthermore, the bacteria grew well on the LB plate, meaning that photocatalyst does not kill the *Salmonella* (Fig. 11k and l). Considering the simplicity of the synthetic process, the chemical durability and the sterilization results, vanadate QDs/g-C₃N₄ are ideal photocatalysts for applications in environmental settings.

3. Conclusions and perspectives

Materials based on g-C₃N₄ are promising photocatalysts with excellent physico-chemical properties and have considerable promise in antibacterial applications. Nevertheless, the antibacterial applications of bulk g-C₃N₄ are limited by its narrow absorption of visible light and facile recombination of charges. Consequently, a variety of g-C₃N₄-based nanocomposites have been developed with high superficial areas, improved e[−]/h⁺ separation efficiencies and expanded visible light absorption ranges, combining to enhance their antibacterial activity. In this review, we highlighted the main strategies used to amplify the photocatalytic efficiency of g-C₃N₄-based nanocomposites and their antimicrobial properties, including different topologies, noble metal decoration, non-noble metal doping and heterojunction construction. The enhancement mechanisms and synergistic effects of g-C₃N₄-based nanocomposites was also discussed. Although g-C₃N₄ is an ideal photocatalyst for the construction of nanocomposites for antibacterial applications, there are still some issues to be solved and opportunities for further research:

(1) The antibacterial mechanism of g-C₃N₄-based nanocomposites include destroying cell membranes and cell walls, producing endotoxins, causing protein mutations, interfering with protein synthesis and oxidizing organics. However, the role of each process in the antibacterial activity has not yet been clearly defined, suggesting that future research on the antibacterial mechanisms of g-C₃N₄-based nanocomposites would be meaningful.

(2) Although g-C₃N₄-based nanocomposites have been extensively studied, the photocatalytic properties are not always predictable, and the performance between g-C₃N₄ and co-composites is often found to be additive and not synergistic. Therefore, molecular models that allow better composite design would be useful.

(3) When constructing g-C₃N₄ heterojunctions, a single heterojunction has many limitations in terms of light absorption and e[−] separation. Thus, the construction of dual heterojunctions, such as dual Z-type or combined Z-type and type II heterojunctions could enhance the photocatalytic effect of g-C₃N₄ nanocomposites and is a key topic for future research and development.

(4) The antibacterial efficiency of g-C₃N₄-based nanocomposite photocatalysts relies largely on ultraviolet and blue light. Extending the range to longer wavelengths would be advantageous.

(5) Most studies were carried at laboratory scales and synthetic strategies for large-scale production are challenging. The development of simple and large-scale green and sustainable synthetic methods that can be automated are required to facilitate commercial applications.

(6) ROS are also able to destroy viruses and therefore further research exploring the antiviral properties of g-C₃N₄ nanocomposites would be valuable.

Conflicts of interest

There are no conflicts to declare.

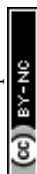


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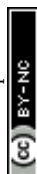
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