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## Synthesis of 2-substituted quinazolines *via* iridium catalyst†

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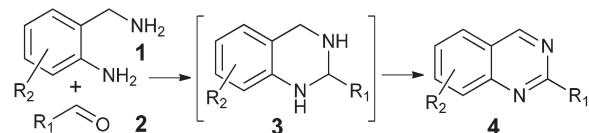
An iridium-catalyzed hydrogen transfer reaction was successfully applied in the synthesis of 2-substituted quinazolines in moderate yields starting from aldehydes or alcohols with 2-aminobenzylamines.

Quinazolines occur frequently in natural products and synthetic pharmaceuticals which exhibit important biological properties,<sup>1</sup> such as antidiabetic, antibacterial, anticonvulsant and anticancer activities. For example, prazosin was an effective medicine as  $\alpha$ -adrenergic blockers for the treatment of high blood pressure, panic disorder and anxiety,<sup>2</sup> and lapatinib was used to treat solid tumor and breast cancer.<sup>3</sup>

Syntheses of substituted quinazolines have been widely explored,<sup>4</sup> and many efficient methods have been developed recently. As shown in Scheme 1, one of the synthetic methods to quinazolines utilizes condensations between aldehydes **2** and 2-aminobenzylamines **1** followed by oxidation of the aminal intermediate **3**. However, stoichiometric or large excess amounts of toxic oxidants were required for this oxidation; *e.g.*, DDQ, *p*-chloranil,<sup>4c</sup> NaClO<sup>4k</sup> and MnO<sub>2</sub><sup>4l</sup> were used. In continuation of our work in the application of hydrogen transfer catalysis in the syntheses of quinazolinones,<sup>5</sup> we were interested to test if a hydrogen transfer catalyst<sup>6</sup> will catalyze the oxidation of aminal **3** to 2-substituted quinazoline **4** in one-pot as shown in Scheme 1.

Firstly, 2-aminobenzylamine **1a** with benzaldehyde **2a** was selected as the model substrate to test the one-pot reaction and the results are summarized in Table 1. We discovered that without a hydrogen acceptor, only 10% product **4a** was formed using  $[\text{Cp}^*\text{IrCl}_2]_2$  (2.5 mol%) as the catalyst ( $\text{Cp}^*$  = pentamethylcyclopentadienyl, entry 1). The major byproduct isolated was the *N*-benzylation product **5**<sup>7</sup> as shown in Scheme 2.

This byproduct formation could have originated from hydrogen transfer<sup>8</sup> to the imine intermediate **6**. Compound **5** could not be



Scheme 1 One-pot synthesis of quinazolines.

further transformed to the product quinazoline **4a** under hydrogen transfer catalysis, which accounted for the low yield of **4a** in this reaction. To improve the yields of **4a**, we decided to add a hydrogen acceptor to the reaction mixture. To our delight, the

Table 1 Optimization of conditions for the synthesis of quinazoline **4a** between **1a** and **2a**<sup>a</sup>

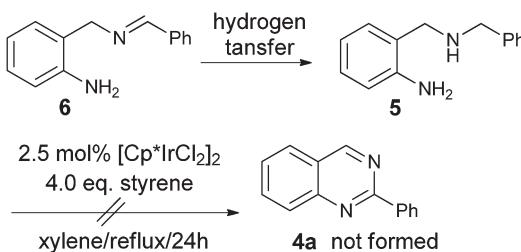
Entry	Catalyst	Additive	Acceptor	Solvent	Yield <sup>b</sup>
1	$[\text{Cp}^*\text{IrCl}_2]_2$	No	No	xylene	10%
2	$[\text{Cp}^*\text{IrCl}_2]_2$	No	styrene	xylene	66% <sup>c</sup>
3	$[\text{Cp}^*\text{IrCl}_2]_2$	No	<i>E</i> -crotonitrile	xylene	50% <sup>c</sup>
4	$[\text{Cp}^*\text{IrCl}_2]_2$	AcOH 0.2 eq.	styrene	xylene	43%
5	$[\text{Cp}^*\text{IrCl}_2]_2$	KOH 0.2 eq.	styrene	xylene	54%
6	$[\text{Cp}^*\text{IrCl}_2]_2$	<i>t</i> -BuONa 0.2 eq.	styrene	xylene	60%
7	$[\text{Cp}^*\text{IrCl}_2]_2$	$\text{K}_2\text{CO}_3$ 0.2 eq.	styrene	xylene	46%
8	$[\text{Cp}^*\text{IrCl}_2]_2$	No	styrene	toluene	35%
9	$[\text{Cp}^*\text{IrCl}_2]_2$	No	styrene	DMF	50%
10	$[\text{Cp}^*\text{IrCl}_2]_2$	No	styrene	xylene	57%
11	$\text{RuCl}_2(\text{PPh}_3)_3$	KOH 0.2 eq.	styrene	xylene	26%
12	$[\text{Ru}(p\text{-cymene})\text{Cl}_2]_2^d$	KOH 0.2 eq.	styrene	xylene	52%

<sup>a</sup> Conditions: **1a** (0.5 mmol), **2a** (0.5 mmol), catalyst (2.5 mol%), styrene (4.0 eq.) in refluxing temperature of the solvent listed (1 mL) under  $\text{N}_2$ , 24 h. <sup>b</sup> <sup>1</sup>H-NMR yield. <sup>c</sup> Isolated yield, 12% of byproduct **5** was also isolated in entry 2. <sup>d</sup> 2.5 mol% dpdp was added.

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† Electronic supplementary information (ESI) available: Experimental procedures and compound characterization data. See DOI: [10.1039/c2ra22278g](https://doi.org/10.1039/c2ra22278g)



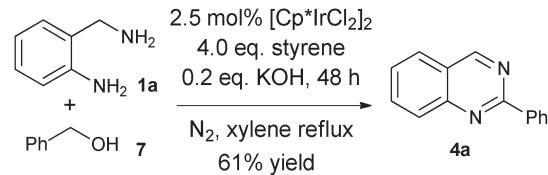
**Scheme 2** Possible pathway to **5** from hydrogenation of imine **6** and reaction of **5** under hydrogen transfer conditions.

yields of **4a** were improved to 66% with addition of styrene (entry 2) and 50% with *E*-crotonitrile (entry 3). Further optimizations of the reaction by using acid or base additives were also tried (entries 4 to 7), but the best yield of 60% obtained by addition of NaOtBu (entry 6) was inferior to the results of 66% without such additives in entry 2. The effects of solvents (entries 8 and 9) and catalysts (entries 10 to 12) were also examined briefly with no increase of the yield of **4a**. After examining the reaction profiles, we decided to select the conditions of entry 2 (2.5 mol%  $[\text{Cp}^*\text{IrCl}_2]_2$  in refluxing xylene with addition of 4.0 eq. styrene) for our investigations of the substrate scope of the reaction.

**Table 2** One-pot synthesis of quinazolines via Ir-catalyzed hydrogen transfers<sup>a</sup>

Entry	$R_1$	$R_2$	Yield <sup>b</sup>
1	H	$C_6H_5$	<b>4a</b> 66%
2	H	3-Cl- $C_6H_4$	<b>4b</b> 54%
3	H	3-Br- $C_6H_4$	<b>4c</b> 48%
4	H	3-NO <sub>2</sub> - $C_6H_4$	<b>4d</b> 58%
5	H	3-Me- $C_6H_4$	<b>4e</b> 54%
6	H	3-OMe- $C_6H_4$	<b>4f</b> 51%
7	H	4-F- $C_6H_4$	<b>4g</b> 51%
8	H	4-Br- $C_6H_4$	<b>4h</b> 55%
9	H	4-NO <sub>2</sub> - $C_6H_4$	<b>4i</b> 57%
10	H	4-Me- $C_6H_4$	<b>4j</b> 50%
11	H	Furyl	<b>4k</b> 55%
12	H	Benzyl	<b>4l</b> 49%
13	H	<i>n</i> -Pentanyl	<b>4m</b> 57%
14	F	$C_6H_5$	<b>4n</b> 56%
15	F	4-Br- $C_6H_4$	<b>4o</b> 60%
16	F	4-Me- $C_6H_4$	<b>4p</b> 62%
17	F	<i>n</i> -Pentanyl	<b>4q</b> 65%

<sup>a</sup> Conditions: Entries 1–13: **1a** (1.0 mmol), **2** (1.0 mmol), catalyst (2.5 mol%), styrene (4.0 eq.) in refluxing xylene (2 mL) under  $N_2$ , 24 h. Entries 14–17: **1b** (1.0 mmol), **2** (1.0 mmol), catalyst (2.5 mol%), styrene (4.0 eq.) in refluxing xylene (2 mL) under  $N_2$ , 24 h. <sup>b</sup> Isolated yield.

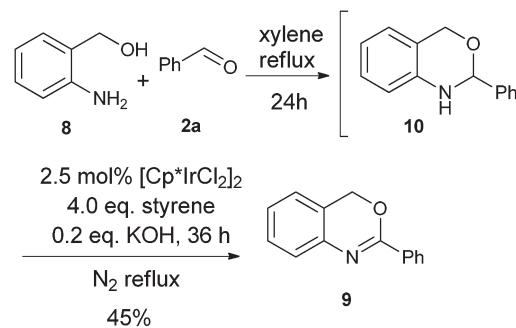


**Scheme 3** One-pot synthesis of 2-phenylquinazoline starting with benzyl alcohol.

Subsequently, a variety of substituted quinazolines were synthesized using our optimized conditions. As shown in Table 2, both aliphatic and aromatic aldehydes reacted with 2-aminobenzylamines to give the corresponding quinazolines **4** in moderate yields. Reactions between **1a** and aromatic aldehydes with either electron-withdrawing or electron-donating groups (entries 2 to 10) showed that the yields were not affected significantly in the range of 48% to 58%. Furthermore, the reactions also performed well when 2-furyl aldehyde (55% yield, entry 11), 2-phenylacetaldehyde (49% yield, entry 12) and hexanal (57% yield, entry 13) were involved. Investigations of 2-(amino-methyl)-3-fluoroaniline **1b** with several aldehydes again gave substituted quinazolines **4n** to **4q** in moderate yields (56% to 65%, entries 14 to 17).

It was our next interest to test the employment of benzyl alcohol **7** instead of benzaldehyde **2a** in the synthesis of quinazoline **4a**. The above described conditions using benzaldehyde did not give a satisfactory yield of **4a** (only 10%) when benzylalcohol **7** was used. Some optimizations (see supporting information, ESI†) identified that the addition of base additives, such as KOH (0.2 eq.) was necessary to increase the yield of **4a** to 61% (Scheme 3).

When 2-aminobenzyl alcohol **8** was used, the condensation with benzaldehydes **2a** gave 2-phenyl-4*H*-benzo[d][1,3]oxazine **9** in 45% yield as shown in Scheme 4.<sup>9</sup> The optimized conditions also involved the use of KOH (2 eq.) to give a better yield (see supporting information, ESI†).



**Scheme 4** One-pot synthesis of 2-phenyl-4*H*-benzo[d][1,3] oxazine between **8** and **2a**.

## Conclusion

We have demonstrated a one-pot synthesis of 2-substituted quinazolines between 2-aminobenzylamines **1** and aldehydes **2** via iridium-catalyzed hydrogen transfers using styrene as a hydrogen acceptor. The use of benzyl alcohol **7** instead of benzaldehyde also successfully gave a quinazoline product in moderate yield. Further extension for the synthesis of 4*H*-3,1-benzoxazine was also demonstrated by the example using 2-aminobenzyl alcohol **8**.

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- Compound **5** was formed in 5% under these conditions; intermediates of **3** and **6** were also detectable in LC-MS.
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- The assay yield of intermediate **10** is 62%, the rest of compound **8** decomposed under the reaction conditions, which accounted for the overall lower yield of compound **9**.