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Triplet–triplet annihilation photon upconversion (TTA-UC) provides a milder alternative to traditional UVB/UVC photochemistry. However, suitable sensitizer-annihilator pairs are scarce, in particular in the high-energy UV regime. Herein, we present a benzene-based annihilator that, paired with a suitable sensitizer, generates upconverted emission approaching the UVC region for the first time.

Combining the energy of two photons into a single photon of higher energy offers significant advantages over traditional direct excitation. Using longer-wavelength incident light enhances penetration depth,^{1,2} which is relevant for biomedical applications, and enables the utilization of sub-bandgap photons in photovoltaics^{3,4} or photoelectrochemical cells⁵ thereby improving solar efficiency.⁶ Among the mechanisms for photon energy fusion,^{7,8} triplet–triplet annihilation upconversion (TTA-UC)^{9,10} stands out as the most promising, because it operates effectively under non-coherent and low-intensity excitation as that provided by commercial LEDs¹¹ or even sunlight.^{12,13} The first reported TTA-UC system employed anthracene as an annihilator, producing violet emission.¹⁴ Since then, the field has advanced considerably in converting lower-energy near-infrared (NIR)^{12,15–17} or red light into higher energy visible light. In contrast, progress in visible-to-UV TTA-UC^{10,13,18–20}—particularly below the UVA range (<315 nm)^{10,21–23}—has been much slower. This lag is notable because it does not align with the practical needs of light sources for photochemical reactions:^{24,25} modern visible/UVA LEDs (>370 nm)²⁶ are already highly efficient, and many photochemical reactions²⁷ can readily utilize sunlight, which contains a non-negligible proportion of UVA.²⁸ However, in the UVB (280–315 nm) and UVC (<280 nm) range, both are most inadequate. To this day, highly inefficient and hazardous UVB sources, such

as mercury²⁹ and excimer lamps,³⁰ remain the commonly used options, underscoring the urgent need for TTA-UC as an alternative for generating light in this spectral region. The slow progress in UVB TTA-UC can be attributed to several challenges. A key obstacle is the limited availability of UV annihilator structures. To date only benzene,¹¹ naphthalene,^{23,31} and biphenyl^{21,22} derivatives have been identified as being capable of approaching the UVB range while no structures are reported for the UVC range (Fig. 1). This absence is likely due to the inherently large singlet–triplet energy gap required for Vis-to-UVC TTA-UC and the limited understanding of how to precisely modulate these energy gaps in such molecules.³² The best results for UV annihilators have been achieved by modifying the annihilator scaffold with so-called TIPS-ethynyl ((triisopropylsilyl)ethynyl) groups that greatly reduced the triplet energy compared to the parent molecule.^{11,13,21,33} Recently, we reported a novel benzene-based annihilator substituted with two

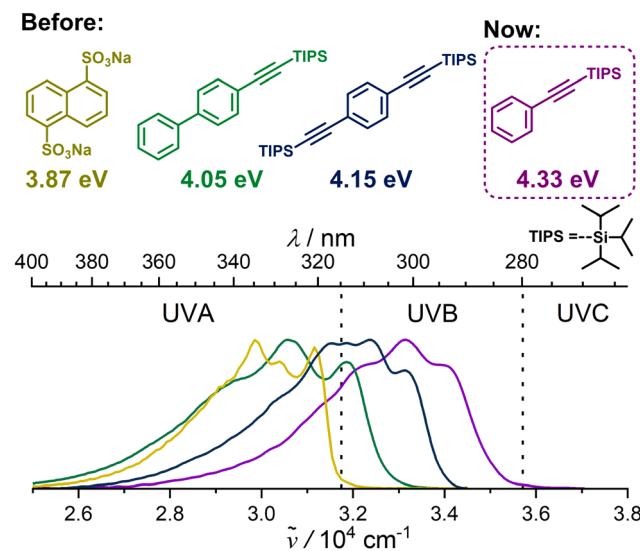


Fig. 1 Structures and emission spectra of high-energy UV annihilators in comparison to the novel UV annihilator **TIPS-Bz**.

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TIPS-ethynyl groups in *para*-position. That annihilator was capable of blue-to-UVB TTA-UC, achieving the highest UC emission energy recorded to date (Fig. 1).¹¹ In contrast, the monosubstituted benzene derivative (**TIPS-Bz**) could not be utilized in such a blue-light-driven UC system, as its triplet state energy (E_T) was estimated at 3.06 eV by DFT calculations and it is thus too high for sensitization. However, the high-energy emission of **TIPS-Bz** (Fig. 1) suggested its potential as a candidate for UVB/UVC photon upconversion when paired with an appropriate sensitizer. Yet, discovering suitable sensitizers for **TIPS-Bz** presents another significant challenge. This is because to enable sensitization of **TIPS-Bz** ($E_T > 3.0$ eV) by violet/UVA light (<400 nm, >3.1 eV) energy losses must be minimized. In general sensitizers employed for TTA-UC exhibit small triplet-singlet energy splitting, such as B,N-heteroarene based^{34,35} and cyanoarene^{11,18} based donor-acceptor chromophores or metal complexes capable of direct triplet-state excitation.³⁶ The triplet state energy of these sensitizers typically does not exceed ~2.8 eV,^{37,38} which is insufficient to effectively sensitize potential UVB/C annihilators. In contrast, the metal complex $[\text{Ir}(\text{CF}_3\text{-pmb})_3]$ (see Fig. 2A for its structure) demonstrates an exceptional triplet state energy of 3.18 eV.^{39,40} The high triplet state energy was exploited by one of our groups in a challenging De Mayo-type ring expansion reaction that is not feasible with commercial high triplet state Ir sensitizers.³⁹ We now report a novel sensitizer-annihilator pair based on **TIPS-Bz** as annihilator and the high triplet energy sensitizer $[\text{Ir}(\text{CF}_3\text{-pmb})_3]$ (**Ir**). A record-breaking upconverted UVB emission approaching the UVC range was achieved, with an unprecedented excited singlet state energy of 4.33 eV (Fig. 1). Mechanistic investigations are provided that highlight the advantages and challenges of this UC system. This study thus represents another breakthrough in the field of UV photon upconversion, advancing the practical limits of TTA-UC and paving the way for future developments.

The synthesis of the annihilator triisopropyl(ethynyl)-benzene and (**TIPS-Bz**) and *fac*- $[\text{Ir}(\text{CF}_3\text{-pmb})_3]$ (**Ir**) followed literature reports.^{11,39} The room-temperature absorption and emission spectra of **TIPS-Bz** and **Ir** are shown in Fig. 2B. **TIPS-Bz** has an excited singlet state energy of 4.33 eV, a low extinction coefficient of the long-wavelength absorption band with a maximum at 280 nm ($\varepsilon_{280} < 1000 \text{ M}^{-1} \text{ cm}^{-1}$), a fluorescence quantum yield of 0.19 ± 0.02 and a fluorescence lifetime of 19.9 ns, in line with the Strickler-Berg relationship.⁴¹ A direct comparison of **TIPS-Bz** with benzene and other benzene-derived annihilators is provided in Chapter S4 (ESI[†]). Generally, the substitution of benzene with TIPS-ethynyl increases the fluorescence quantum yield, reduces the excited-state singlet lifetime, and induces a red shift in both the absorption and emission spectra, a trend that has been observed for naphthalene as well.³³ We estimate a decreased triplet state energy of 3.06 eV from DFT calculations compared to the parent benzene molecule (3.9 eV).⁴² **Ir** readily absorbs UVA light and can be selectively excited with 355 nm laser pulses. The triplet-triplet absorption spectrum is presented in Fig. 2C, and time-resolved measurements indicate a triplet state lifetime of 4 μ s (Fig. 2D), which is sufficient to enable efficient bimolecular quenching. The high triplet state energy of 3.18 eV³⁹ makes

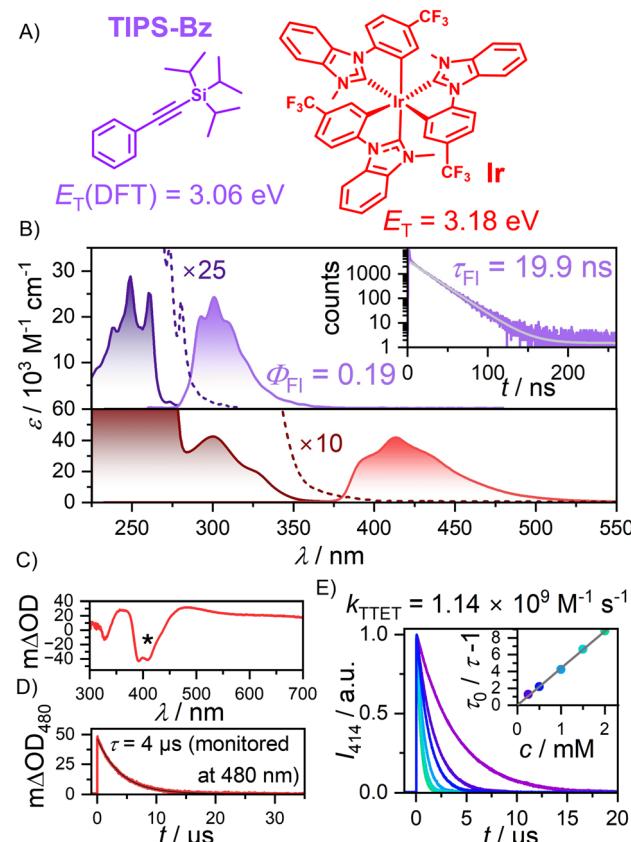


Fig. 2 (A) Molecular structures of **Ir** and **TIPS-Bz**. (B) Absorption and emission spectra of **TIPS-Bz** in cyclohexane (inset: fluorescence lifetime) and of **Ir** in benzene. (C) Transient absorption (TA) spectrum of **Ir** 100 ns after excitation with 355 nm laser pulses in benzene. The asterisk marks the emission of **Ir** (D) Kinetic TA trace of **Ir** in benzene. (E) Quenching studies of 25 μM **Ir** with different concentrations of **TIPS-Bz** in benzene. Inset: Stern–Volmer analysis.

triplet–triplet energy transfer (TTET) to **TIPS-Bz** (3.06 eV) thermodynamically feasible, qualifying **Ir** as a potential sensitizer for the upconversion system under study. Indeed, the phosphorescence of **Ir** in Ar-saturated benzene is quenched upon addition of increasing **TIPS-Bz** concentrations (Fig. 2E). A Stern–Volmer analysis yielded a quenching rate constant of $1.14 \times 10^9 \text{ M}^{-1} \text{ s}^{-1}$ (Fig. 2E, inset). To investigate the TTET process, a transient absorption (TA) spectrum of a solution containing **TIPS-Bz** and **Ir** was recorded after the initial deactivation of the sensitizer triplet. The relatively low sensitizer concentration minimizes filter effects at wavelengths $\lambda < 320$ nm. After quenching a TA band with a maximum at 304 nm is observed, which is attributed to the triplet-triplet absorption of ³**TIPS-Bz** (Fig. 3A). This assignment is supported by TD-DFT calculations, which predict a maximum oscillator strength at 291 nm. Time-resolved measurements show the simultaneous formation of ³**TIPS-Bz** as the luminescence of **Ir** is quenched (Fig. 3B). Under the selected conditions a triplet lifetime of 34 μ s is measured for **TIPS-Bz** (Fig. S1, ESI[†]). After complete deactivation of ³**TIPS-Bz** only baseline level is detected (Fig. S3, ESI[†]). This gives us unequivocal proof of triplet-triplet energy transfer between the sensitizer and



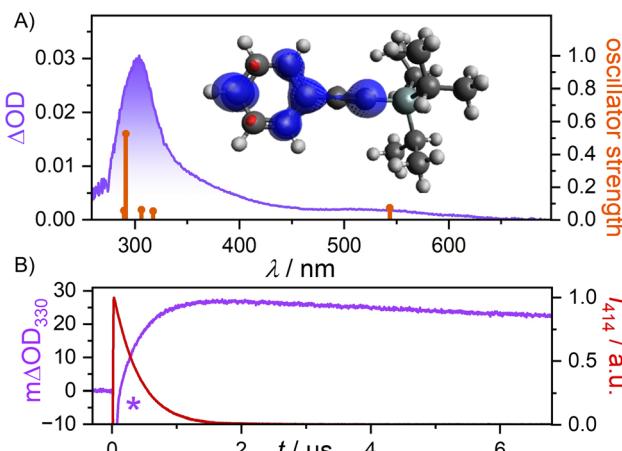


Fig. 3 TA measurements of **Ir** and **TIPS-Bz** in Ar-saturated benzene with 355 nm laser pulses (A) TA spectrum of **Ir** (6.25 μ M) and 1 mM **TIPS-Bz** recorded 4 μ s after excitation shown together with the oscillator strength calculated by TD-DFT (B3LYP/6-311+G(d,p) level of theory). (B) Kinetic measurements of **Ir** (25 μ M) emission and transient absorption of **TIPS-Bz** (2 mM).

TIPS-Bz. With the TTET step established, we investigated the triplet-triplet annihilation (TTA) process of **TIPS-Bz**. For this study, a solution containing 25 μ M **Ir** and 2 mM **TIPS-Bz** was selected, ensuring 95% deactivation of the excited sensitizer, a critical factor for efficient triplet-triplet annihilation.⁹ Reabsorption by **TIPS-Bz** does not play a significant factor since the lowest electronic transition (e.g. in the range of 280–320 nm) exhibits weak absorption presumably as it is symmetry forbidden.⁴³ This is advantageous for upconversion since reabsorption typically leads to a reduced upconversion quantum yield.¹⁸ Time-gated emission spectra were recorded throughout the complete deactivation of ³**TIPS-Bz** (integration window: 4–204 μ s after laser pulse excitation, Fig. 4A, blue spectrum). A signal in the UVB range is detected, which resembles the fluorescence spectrum of **TIPS-Bz** in diluted solution (Fig. 2B). Differences between these two spectra likely stems from inner filter effects of **Ir** owing to the large molar absorption coefficient ϵ of **Ir** in that range. Directly exciting the solution containing the UC system at 280 nm yields a practically identical emission spectrum in the UV (Fig. 4A, green spectrum). Further control experiments do not suggest the formation of excimers (Fig. S2, ESI[†]).³³ Additionally, we detect post-quenching emission of **Ir** (compare to direct excitation at 280 nm, red spectrum). We assume that this emission stems from back-TTET and singlet energy transfer from ¹**TIPS-Bz*** to **Ir**.¹¹ Time-resolved transient absorption (TA) measurements of ³**TIPS-Bz** and the emission of ¹**TIPS-Bz*** are shown in Fig. 4B. While strong artifacts are observed immediately after laser pulse excitation likely due to remaining stray light, the signals of both species concurrently reach a maximum, while the ¹**TIPS-Bz*** emission decays faster than ³**TIPS-Bz**.⁹ Time-gated emission spectra were recorded with increasing laser pulse excitation intensities (Fig. 4C). Integration over the UC emission spectrum revealed a nonlinear dependence on the laser power, consistent with the bimolecular nature of the TTA process. Altogether, these

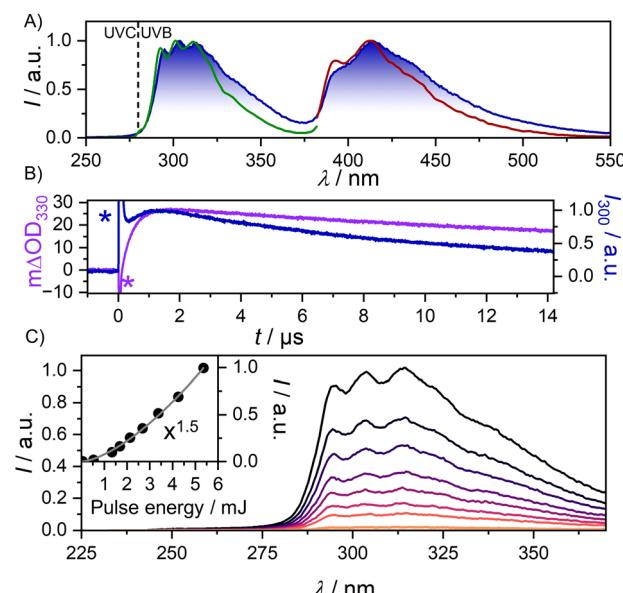


Fig. 4 Mechanistic investigations of 25 μ M **Ir** and 2 mM **TIPS-Bz** in Ar-saturated benzene. (A) Time-gated emission spectrum after 355 nm laser pulse excitation (blue, 4–204 μ s) and after 280 nm excitation (green and red, normalized to the maximum in the UV and visible region). (B) Kinetic traces of the ¹**TIPS-Bz*** emission and transient absorption of ³**TIPS-Bz**. (C) Excitation-intensity dependent delayed emission spectra. Inset: Emission maximum plotted against laser pulse intensity with a pulse duration of \sim 5 ns. See text for details.

findings provide clear evidence that the observed UVB emission originates from TTA-UC. Unfortunately, steady-state UC emission could not be detected with a 370 nm LED as light source. This absence is presumably due to the absorption characteristics of **Ir**, whose ϵ is two orders of magnitude lower at 370 nm than in the UVB region. Consequently, achieving sufficient concentrations of ³**TIPS-Bz** for TTA under mild 370 nm LED excitation necessitates a high concentration of **Ir**, which in turn enhances reabsorption losses. Ideally, the sensitizer should have an absorption minimum in the spectral region of UC emission.^{13,44} This highlights one of the major challenges of upconversion in the UVB/C range, where most photoactive compounds have large extinction coefficients ($\epsilon_{\text{UVB/C}} > 10^3 \text{ M}^{-1} \text{ cm}^{-1}$), which makes this a practically unavoidable loss channel in diffusion based UC in solution. Hence, under our UC conditions, considering the excitation and excited singlet state energy, the anti-Stokes shift (0.83 eV) remains below typical values of related systems.¹⁰ To address this challenge, **TIPS-Bz** was further modified by introducing a *tert*-butyl group at the *para*-position (**4tBu-TIPS-Bz**). This modification was intended to lower the triplet energy, potentially allowing sensitization by a broader range of sensitizers. However, the calculated triplet energy of 3.02 eV remains too high to provide significant benefits. While the UC system **4tBu-TIPS-Bz-Ir** remains functional under 355 nm laser pulse excitation, its energy transfer constant and photostability are notably lower than those of **TIPS-Bz**, making it a less favorable system overall (see Chapter S5 for details, ESI[†]).

This study demonstrates that a benzene annihilator functionalized with a TIPS-ethynyl group can be integrated into a TTA-UC



system, achieving emission near the UVC range. This was realized using the high-triplet-energy sensitizer *fac*-[Ir(CF₃-pmb)₃], which efficiently sensitizes the high-lying triplet state of **TIPS-Bz**. A thorough investigation of the TTA-UC mechanism provided unequivocal evidence of the process, pushing the feasibility of TTA-UC into higher-energy regions. However, these findings also highlight critical challenges for UVC upconversion. Future efforts towards Vis-to-UVC photon upconversion should prioritize increasing the singlet-triplet gap of the annihilator by appropriate functionalization, bringing it closer to the theoretical thermodynamic limit for TTA (*i.e.* $2 \times E_{T1} \sim E_{S1}$).^{32,45} This will effectively enable sensitization with sensitizers having lower triplet energies. Additionally, the development of an optimally matched sensitizer remains crucial—one that balances high absorption in the UVA/purple spectral range with minimal absorptivity in the UVB/C region to reduce losses.⁴⁶

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Data availability

All experimental data have been provided in the main text and the ESI.† The data sets shown in the main paper can be found under <https://doi.org/10.25358/openscience-12308>.

Conflicts of interest

The authors declare no conflict of interest.

Notes and references

- 1 M. R. Hamblin, *Dalton Trans.*, 2018, **47**, 8571–8580.
- 2 Q. Zhou, B. M. Wirtz, T. H. Schloemer, M. C. Burroughs, M. Hu, P. Narayanan, J. Lyu, A. O. Gallegos, C. Layton, D. J. Mai and D. N. Congreve, *Adv. Mater.*, 2023, **35**, 2301563.
- 3 Y. Shang, S. Hao, C. Yang and G. Chen, *Nanomaterials*, 2015, **5**, 1782–1809.
- 4 D. Beery, T. W. Schmidt and K. Hanson, *ACS Appl. Mater. Interfaces*, 2021, **13**, 32601–32605.
- 5 M. Barawi, F. Fresno, R. Pérez-Ruiz and V. A. De La Peña O'Shea, *ACS Appl. Energy Mater.*, 2019, **2**, 207–211.
- 6 T. Trupke, M. A. Green and P. Würfel, *J. Appl. Phys.*, 2002, **92**, 4117–4122.
- 7 M. Haase and H. Schäfer, *Angew. Chem., Int. Ed.*, 2011, **50**, 5808–5829.
- 8 M. Pawlicki, H. A. Collins, R. G. Denning and H. L. Anderson, *Angew. Chem., Int. Ed.*, 2009, **48**, 3244–3266.
- 9 Y. Murakami and K. Kamada, *Phys. Chem. Chem. Phys.*, 2021, **23**, 18268–18282.
- 10 M. Uji, T. J. B. Zähringer, C. Kerzig and N. Yanai, *Angew. Chem., Int. Ed.*, 2023, **62**, e202301506.
- 11 T. J. B. Zähringer, J. A. Moghtader, M. Bertrams, B. Roy, M. Uji, N. Yanai and C. Kerzig, *Angew. Chem., Int. Ed.*, 2023, **62**, e202215340.
- 12 W. Liang, C. Nie, J. Du, Y. Han, G. Zhao, F. Yang, G. Liang and K. Wu, *Nat. Photonics*, 2023, **17**, 346–353.
- 13 N. Harada, Y. Sasaki, M. Hosoyamada, N. Kimizuka and N. Yanai, *Angew. Chem., Int. Ed.*, 2021, **60**, 142–147.
- 14 C. A. Parker and C. G. Hatchard, *Proc. R. Soc. Lond. Ser. Math. Phys. Sci.*, 1962, **269**, 574–584.
- 15 Z. Huang, D. E. Simpson, M. Mahboub, X. Li and M. L. Tang, *Chem. Sci.*, 2016, **7**, 4101–4104.
- 16 T. N. Singh-Rachford, A. Nayak, M. L. Muro-Small, S. Goeb, M. J. Therien and F. N. Castellano, *J. Am. Chem. Soc.*, 2010, **132**, 14203–14211.
- 17 T. N. Singh-Rachford and F. N. Castellano, *Coord. Chem. Rev.*, 2010, **254**, 2560–2573.
- 18 A. Olesund, J. Johnsson, F. Edhborg, S. Ghasemi, K. Moth-Poulsen and B. Albinsson, *J. Am. Chem. Soc.*, 2022, **144**, 3706–3716.
- 19 H.-L. Lee, M.-S. Lee, H. Park, W.-S. Han and J.-H. Kim, *Korean J. Chem. Eng.*, 2019, **36**, 1791–1798.
- 20 Y. Murakami, A. Motooka, R. Enomoto, K. Niimi, A. Kaiho and N. Kiyoyanagi, *Phys. Chem. Chem. Phys.*, 2020, **22**, 27134–27143.
- 21 T. J. B. Zähringer, M.-S. Bertrams and C. Kerzig, *J. Mater. Chem. C*, 2022, **10**, 4568–4573.
- 22 L. Schmid, F. Glaser, R. Schaer and O. S. Wenger, *J. Am. Chem. Soc.*, 2022, **144**, 963–976.
- 23 X. Lin, Z. Chen, Y. Han, C. Nie, P. Xia, S. He, J. Li and K. Wu, *ACS Energy Lett.*, 2022, **7**, 914–919.
- 24 B. Lehmann, *Photochem. Photobiol.*, 2005, **81**, 1246–1251.
- 25 N. Zhang, G. Liu, H. Liu, Y. Wang, Z. He and G. Wang, *J. Hazard. Mater.*, 2011, **192**, 411–418.
- 26 H. Hirayama, S. Fujikawa and N. Kamata, *Electron. Commun. Jpn.*, 2015, **98**, 1–8.
- 27 T. J. B. Zähringer, M. Wienhold, R. Gilmour and C. Kerzig, *J. Am. Chem. Soc.*, 2023, **145**, 21576–21586.
- 28 Reference Air Mass 1.5 Spectra, <https://www.nrel.gov/grid/solar-resource/spectra.html>, (accessed March 24, 2025).
- 29 T. Chang, S. You, B. Yu and H. Kong, *J. Hazard. Mater.*, 2007, **141**, 784–792.
- 30 A. G. Griesbeck, N. Maptue, S. Bondock and M. Oelgemöller, *Photochem. Photobiol. Sci.*, 2003, **2**, 450–451.
- 31 B. Pfund, D. M. Steffen, M. R. Schreier, M.-S. Bertrams, C. Ye, K. Börjesson, O. S. Wenger and C. Kerzig, *J. Am. Chem. Soc.*, 2020, **142**, 10468–10476.
- 32 W. Zeng, C. Zhong, H. Bronstein and F. Plasser, *Angew. Chem., Int. Ed.*, 2025, e202502485.
- 33 A. Olesund, S. Ghasemi, K. Moth-Poulsen and B. Albinsson, *J. Am. Chem. Soc.*, 2023, **145**, 22168–22175.
- 34 J. Li, M. Zhang, L. Zeng, L. Huang and X. Wang, *Angew. Chem., Int. Ed.*, 2023, **62**, e202303093.
- 35 X. Cao, K. Pan, J. Miao, X. Lv, Z. Huang, F. Ni, X. Yin, Y. Wei and C. Yang, *J. Am. Chem. Soc.*, 2022, **144**, 22976–22984.
- 36 Y. Sasaki, S. Amemori, H. Kouno, N. Yanai and N. Kimizuka, *J. Mater. Chem. C*, 2017, **5**, 5063–5067.
- 37 F. Strieth-Kalthoff, M. J. James, M. Teders, L. Pitzer and F. Glorius, *Chem. Soc. Rev.*, 2018, **47**, 7190–7202.
- 38 S. Dutta, J. E. Erchinger, F. Strieth-Kalthoff, R. Kleinmans and F. Glorius, *Chem. Soc. Rev.*, 2024, **53**, 1068–1089.
- 39 T. O. Paulisch, L. A. Mai, F. Strieth-Kalthoff, M. J. James, C. Henkel, D. M. Guldi and F. Glorius, *Angew. Chem., Int. Ed.*, 2022, **61**, e202112695.
- 40 K. Tsuchiya, S. Yagai, A. Kitamura, T. Karatsu, K. Endo, J. Mizukami, S. Akiyama and M. Yabe, *Eur. J. Inorg. Chem.*, 2010, 926–933.
- 41 M.-S. Bertrams and C. Kerzig, *Chem. Commun.*, 2021, **57**, 6752–6755.
- 42 J. P. Doering, *J. Chem. Phys.*, 1969, **51**, 2866–2870.
- 43 J. N. Murrell and J. A. Pople, *Proc. Phys. Soc. Sect. A*, 1956, **69**, 245–252.
- 44 A. Monguzzi, J. Mezyk, F. Scotognella, R. Tubino and F. Meinardi, *Phys. Rev. B*, 2008, **78**, 195112.
- 45 L. Naimovičius and A. B. Pun, *Trends Chem.*, 2025, S2589597425000425.
- 46 Y. Zhou, F. N. Castellano, T. W. Schmidt and K. Hanson, *ACS Energy Lett.*, 2020, **5**, 2322–2326.

