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Lectin inspired polymers based on the dipeptide Ser-Asp for glycopeptide enrichment†

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Lectin inspired polymers were prepared through modification of silica microspheres with Ser-Asp (SD). This functional polymer showed distinct adsorption and retention towards different disaccharides and demonstrated high-efficiency enrichment of glycopeptides.

The covalent binding of glycans on proteins not only regulates the important biological processes but also reflects the state of cells. Typically, enzymes are equipped with certain glycans for recognition sites in gene transcription or protein translation.^{1,2} Aberrant protein glycosylation can change the physicochemical properties or structures of biomolecules, and then trigger diseases, *e.g.*, *N*-glycosylated tau protein can only be detected in biofluids in Alzheimer's disease.³ Therefore, it is important to identify protein glycosylation, which provides information about the protein sequence, glycosylation sites and glycan structures. However, it is difficult to characterize protein glycosylation with state-of-the-art mass spectrometry (MS) because of the low abundance of glycopeptides.⁴

In the last decade, several enrichment methods for glycopeptides have been developed, including lectin affinity chromatography,^{5–7} hydrophilic interaction chromatography,^{8,9} boronic acid affinity chromatography,^{10,11} *etc.* Among these, lectin affinity chromatography is widely used to enrich glycopeptides/glycoproteins.⁵ Lectin recognizes specific carbohydrate moieties through the reversible binding between the anomeric hydroxyl groups on carbohydrates and the hydrophilic groups from amino acid residues in lectin protein.^{12,13} Such interactions are interpreted as the synergistic effects of multiple forces, including hydrogen bonding, CH- π interactions, ion pairing, van der Waals forces, *etc.*^{14,15} Even

though the single peptide-carbohydrate interaction is weak, the multiple binding sites form a network and enhance these interactions. These peptide-carbohydrate interactions inspire us to design artificial lectins for glycopeptide enrichment.

In this paper, a lectin inspired polymer was prepared through modification of amino-silica microspheres with polymerized Ser-Asp (SD) (denoted as polySD-SiO₂ in Fig. 1a). The Ser-Asp was selected to mimic the amino acid residues in lectins and the polymer chain is used to amplify the stereo-



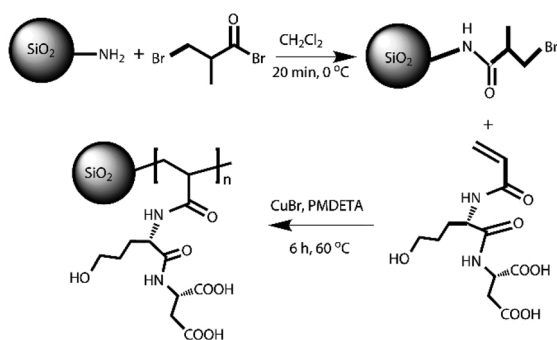
Fig. 1 The synthesized materials were characterized by X-ray photoelectron spectroscopy (a), scanning electron microscopy (silica before (c) and after modification (d)), thermogravimetry analysis (e) and zeta potential analysis (f).

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Scheme 1 The synthesis route of polySD-SiO₂.

selective recognition between Ser-Asp and glycans. The bio-inspired material polySD-SiO₂ was synthesized by surface initiated atom transfer radical polymerization (Scheme 1) because this method is robust and easy to prepare, and uses commercially available, inexpensive catalysts and initiators.^{16–19}

The synthesized materials were characterized by high resolution scanning electron microscopy (HRSEM), zeta potential analysis and thermogravimetry analysis (TGA), and used to enrich glycopeptides from different samples. HRSEM images reveal that polySD-SiO₂ materials preserve a similar morphology to that of silica microspheres. These data proved that polySD-SiO₂ materials are not degraded after coating with polymers on the silica surface. The nitrogen sorption isotherm of polySD-SiO₂ showed a typical type IV curve (Fig. S1†). The Barrett-Joyner-Halenda (BJH) pore size of polySD-SiO₂ was 26 nm compared with 33 nm of silica microspheres without modification (Fig. 1b), indicating a 3.5 nm polymer coated on the silica sphere. The elemental analysis and X-ray photoelectron spectroscopy data (Fig. 1a) imply the presence of carbon, nitrogen and oxygen on the polySD-SiO₂. TGA analysis was used to further measure the relative composition of polySD-SiO₂ (Fig. 1e). The sharp decline above 400 °C resulted from the weight loss of the polymer layer. The zeta-potentials of polySD-SiO₂ under different pH conditions were measured. As shown in Fig. 1f, the surface charge of polySD-SiO₂ is zero at solution pH 5.1, and negative at a pH higher than 5.1. The tunable surface charges are ascribed to the naturally smart zwitterions of Ser-Asp chemical composition. These results suggest that polySD-SiO₂ was successfully modified on the silica microspheres.

In order to exhibit the distinct discrimination of polySD-SiO₂ toward saccharides, the chromatographic separation of disaccharides and fructo-oligosaccharides was carried out. Because of the subtle differences in stereochemistry, chemoselective discrimination of disaccharides is very challenging. Five disaccharides (sucrose, maltose, leucrose, melibiose, and gentiobiose) with subtle differences in structure could be well separated with the column (Fig. 2a). The application of this column to oligosaccharides with different polymerization degrees was also investigated. The components of fructo-oligosaccharides were separated into 26 constituents with different

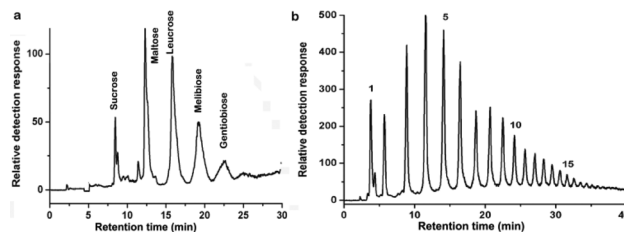


Fig. 2 Separation of disaccharides (a) and fructo-saccharides (b) with a polySD-SiO₂ packed column and the samples were detected with an evaporative light scattering detector (ELSD). The flow rate was 1.0 mL min^{−1} and the column temperature was 30 °C. The mobile phase A: 100% CH₃CN, the mobile phase B: 100% H₂O. Gradient for the separation of disaccharides: 0–12 min, 80%–65% A; gradient for the separation of fructo-oligosaccharides: 0–12 min, 80%–65% A; 12–27 min, 65%–55% A. ELSD conditions: Gas pressure 30 psi, tube temperature 70 °C, gain 10.

polymerization degrees, as labelled by numbers (Fig. 2b). The fructo-oligosaccharides with low to high polymerization degrees could be gradually eluted from the adsorbent using a solution with high to low acetonitrile content, which matches well with the typical characteristics of hydrophilic interaction chromatography. These data clearly demonstrate the powerful discrimination capacities of the polySD-SiO₂-based column toward different saccharides. This column effectively addresses the problem of saccharide separation and can be widely applied in carbohydrate chemistry.

The enrichment conditions were optimized. The loading solution was 80% CH₃CN/0.1% FA because most of the non-glycopeptides and all glycopeptides retained well under these conditions (Fig. S2†). The elution solution was 40% CH₃CN/H₂O/5 mM NH₄HCO₃ because of the weaker retention of glycopeptides at pH 8.0 and a lower CH₃CN content (Fig. S3†). Tryptic digests of bovine fetuin were chosen as the model sample, attributing to its definite glycosylation sites and glycan structures. In order to mimic the complexity of the bio-samples, the tryptic digests of bovine fetuin and bovine serum albumin were mixed at different molar ratios and treated with polySD-SiO₂. After enrichment with polySD-SiO₂, up to 43 glycopeptides were detected from the mixture of fetuin and BSA at a molar ratio of 1 : 100, in sharp contrast to no detectable glycopeptides in the mixture before enrichment (Fig. S4†). For comparison, only 7 glycopeptides were captured from the same sample by commercial zwitterionic hydrophilic interaction liquid chromatography (ZIC-HILIC) (Fig. 3c). Moreover, the selectivity of the material toward glycopeptides remained even in the case of fetuin and BSA at a molar ratio of 1 : 3000 (Fig. 3b). It should be noted that some glycopeptides enriched with polySD-SiO₂ originated from the impurities of BSA, as reported in the literature,⁹ indicating the ability of polySD-SiO₂ to enrich trace glycopeptides from a vast amount of interferents. These results demonstrate the high selectivity of polySD-SiO₂ for glycopeptides. The high selectivity of polySD-SiO₂ toward glycopeptides might be attributed to the reason that the monomer Ser-Asp binds glycans *via* hydrogen





Fig. 3 The performance of polySD-SiO₂ (a and b) and commercial ZIC-HILIC (c) toward glycopeptides using tryptic digests of bovine fetuin and bovine serum albumin (BSA) at molar ratios of 1:100 (a and c) and 1:3000 (b). Glycopeptides are labelled with their corresponding glycans or red asterisks.

bonds and the multiple sites of Ser-Asp in the polymer further enhance their interaction.

In addition to its high selectivity toward glycopeptides, polySD-SiO₂ also exhibited high recovery and adsorption capacity for glycopeptides, which are two important parameters to assess adsorbents. A stable-isotope dimethyl labelling method²⁰ was performed to determine the recovery of polySD-SiO₂ toward a standard glycopeptide (peptide sequence: KVANKT, N: glycosylation site). The recovery rate was $86.6 \pm 5.0\%$ ($n = 5$), contributing to the strong binding of glycopeptides on materials during sample loading and the thorough elution of targets from materials. The adsorption capability of polySD-SiO₂ toward the standard glycopeptide was 83 mg g^{-1} . These results demonstrated the high efficiency of polySD-SiO₂ materials for glycopeptide enrichment.



Fig. 4 Enriched glycopeptides from tryptic digests of bovine fetuin and bovine serum albumin (BSA) at a molar ratio of 1:100 treated with commercial ZIC-HILIC (a), at molar ratio 1:1000 (b) and 1:3000 (c) treated with polySD-SiO₂. Glycopeptides are labelled with glycans or red asterisks.

The established enrichment strategy was further applied to enrich glycopeptides from the HeLa S3 cell lysate. Three technical replicates were performed using 20 μg of total protein. With a false discovery rate of 1% at both the peptide and site levels, 175, 151, and 165 unique glycosylation sites were characterized after treatment with polySD-SiO₂, respectively (Fig. 4a). Furthermore, substantial overlap of the identified glycosylation sites could be found between two replicates (Fig S5 in the ESI†). The average selectivity of polySD-SiO₂ toward glycopeptides was 68% for three technique replicates. The detailed information of the identified glycosylation sites is shown in Table S1†. For comparison, *Sambucus nigra agglutinin* II (SNA), a lectin, was used to enrich glycopeptides from the same samples. The numbers of identified glycosylation sites were 23, 19 and 26, respectively. These numbers are substantially less than those obtained with our materials. This result could explain that functional monomer Ser-Asp in poly SD-SiO₂ could effectively bind various glycans in the glycopeptides, while the SNA is specific to only one subset of glycans. Gene ontology analysis indicated that 69.6% of the matched glycoproteins were annotated as binding proteins, and other glycoproteins were involved in the cell adhesion and receptor activity (Fig. 4b). These data exemplify the excellent potential of our material for use in glycoproteomic analysis.

Conclusions

In conclusion, lectin inspired materials showed high selectivity for glycopeptides, and could enrich low abundance glycopeptides from both the fetuin/BSA mixture and real biosamples. This material could also discriminate disaccharides with a



slight difference in their structures. These features well match the demand for fine discrimination of glycans and in-depth analysis of glycoproteomics. As a biomimetic and bioinspired approach employed in this study, lectin inspired materials pave the way for designing artificial saccharide receptors with high specificity.

Conflicts of interest

There are no conflicts to declare.

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Notes and references

- 1 M. Aebi, R. Bernasconi, S. Clerc and M. Molinari, *Trends Biochem. Sci.*, 2010, **35**, 74–82.
- 2 R. Kadirvelraj, J.-Y. Yang, J. H. Sanders, L. Liu, A. Ramiah, P. K. Prabhakar, G.-J. Boons, Z. A. Wood and K. W. Moremen, *Proc. Natl. Acad. Sci. U. S. A.*, 2018, **115**, 4637–4642.
- 3 S. A. Yuzwa, A. H. Cheung, M. Okon, L. P. McIntosh and D. J. Vocadlo, *J. Mol. Biol.*, 2014, **426**, 1736–1752.
- 4 D. F. Zielinska, F. Gnad, J. R. Wisniewski and M. Mann, *Cell*, 2010, **141**, 897–907.
- 5 M. Chen, X. Shi, R. M. Duke, C. I. Ruse, N. Dai, C. H. Taron and J. C. Samuelson, *Nat. Commun.*, 2017, **8**, 15487.
- 6 E. Ruiz-May, S. Hucko, K. J. Howe, S. Zhang, R. W. Sherwood, T. W. Thannhauser and J. K. C. Rose, *Mol. Cell. Proteomics*, 2014, **13**, 566–579.
- 7 E. Song, R. Zhu, Z. T. Hammond and Y. Mechref, *J. Proteome Res.*, 2014, **13**, 4808–4820.
- 8 X. Li, Y. Xiong, G. Qing, G. Jiang, X. Li, T. Sun and X. Liang, *ACS Appl. Mater. Interfaces*, 2016, **8**, 13294–13302.
- 9 G. Qing, X. Li, P. Xiong, C. Chen, M. Zhan, X. Liang and T. Sun, *ACS Appl. Mater. Interfaces*, 2016, **8**, 22084–22092.
- 10 Z. Bie, Y. Chen, H. Li, R. Wu and Z. Liu, *Anal. Chim. Acta*, 2014, **834**, 1–8.
- 11 X. Li, H. Liu, G. Qing, S. Wang and X. Liang, *J. Mater. Chem. B*, 2014, **2**, 2276–2281.
- 12 Y. Hang, X.-P. He, L. Yang and J. Hua, *Biosens. Bioelectron.*, 2015, **65**, 420–426.
- 13 J. M. Propster, F. Yang, S. Rabbani, B. Ernst, F. H. T. Allain and M. Schubert, *Proc. Natl. Acad. Sci. U. S. A.*, 2016, **113**, E4170–E4179.
- 14 W. I. Weis, K. Drickamer and W. A. Hendrickson, *Nature*, 1992, **360**, 127–134.
- 15 W. I. Weis, R. Kahn, R. Fourme, K. Drickamer and W. A. Hendrickson, *Science*, 1991, **254**, 1608–1615.
- 16 H. W. Ma, J. H. Hyun, P. Stiller and A. Chilkoti, *Adv. Mater.*, 2004, **16**, 338–341.
- 17 L. Hua, M. Y. Liu, L. C. Mao, Q. Huang, H. Y. Huang, G. J. Zeng, J. W. Tian, Y. Q. Wen, X. Y. Zhang and Y. Wei, *Appl. Surf. Sci.*, 2018, **434**, 204–221.
- 18 L. Hua, J. Wu, M. Y. Liu, L. C. Mao, H. Y. Huang, Q. Wang, Y. F. Dai, Y. Q. Wen, X. Y. Zhang and Y. Wei, *J. Colloid Interface Sci.*, 2017, **508**, 396–404.
- 19 L. C. Mao, X. H. Liu, M. Y. Liu, L. Hua, D. Z. Xu, R. M. Jiang, Q. Huang, Y. Q. Wen, X. Y. Zhang and Y. Wei, *Appl. Surf. Sci.*, 2017, **419**, 188–196.
- 20 F. Wang, R. Chen, J. Zhu, D. Sun, C. Song, Y. Wu, M. Ye, L. Wang and H. Zou, *Anal. Chem.*, 2010, **82**, 3007–3015.

