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Tackling current production of HAp and HAp-driven biomaterials

The present work explores the latest trends in Hydroxyapatite (HAp)-based biomaterials. HAp has a well-established clinical track record, and recent efforts have focused on enhancing its properties by combining it with other materials, growth factors, and cells. 3D printing has emerged as a promising technology for these biomaterials, offering the potential for complex structures and on-demand production. However, manufacturing reproducible materials and ensuring scalability remains challenging. The limitations associated with HAp-based materials are addressed as well as the importance of continuous production for achieving uniform properties and enabling research-clinical-industry transition.

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Tackling current production of HAp and HAp-driven biomaterials

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Hydroxyapatite (HAp)-based biomaterials are well-established for biomedical applications due to their extensive research work and clinical track record. Recent efforts have been focusing on the development of enhanced HAp systems, through combination with other materials, growth factors, and cells. However, manufacturing reproducible materials and process scalability are still major challenges. 3D printing emerged in the last decade as a technology that allows obtaining complex structures, using HAp as a core material or by incorporating it in other organic or inorganic matrices to obtain high resolution and on-demand production. While this approach has potential, there are limitations associated with the HAp characteristics (such as particle size distribution, size, crystallinity and morphology) used during printing that need to be overcome. In this context, manufacturing high volumes of HAp with uniform properties can be achieved using continuous production, which allows for the development of highly tailored materials that can be used for 3D-printing. This review discusses the latest trends in HAp production-derived performance materials. Moreover, it fills the gap in current papers by exploring the steps required for research–clinical–industry transitions.

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1. Introduction

Calcium phosphates (CaPs) are well-known bioceramics with impressive osteoconductive and osteoinductive characteristics.¹ Their bioactivity is totally dependent on the CaPs mineral phase (e.g., hydroxyapatite (HAp) and tricalcium phosphate (TCP)), making them a versatile alternative for several applications since it is possible to tune different properties such as ion release, stability, solubility, and mechanical properties.² Synthetic HAP, in particular nano-HAP, has been widely studied in the field of biomaterials, especially for bone regeneration and replacement purposes due to its wide similarities with natural bone (e.g., 1.67 calcium to phosphorus molar ratio, thermodynamic stability at pH 4.3 and above, crystallinity and size).³

Bone implants and fillers composed of HAp show proper stabilization and fixation to the adjacent tissues, with the formation of a new, biologically active, carbonate apatite layer.⁴

However, HAP can present downsides, such as its low biodegradation rate,^{5,6} and its brittle nature (unsuitable for use in weight-supporting locations).^{7,8} Conventional approaches to improve the overall properties of HAP-based biomaterials consist in its conjugation with different atomic or molecular species. In this context, ion-doping has gained popularity as a chemical approach to modify the structure of HAP. This is because the crystal structure of HAP is amenable to isomorphic substitution of cations and anions with ease, leading to a wide range of potential applications in the field of biomedicine.⁹ Additionally, the combination with bioactive factors (such as BMP-2, VEGF) can also enhance HAP osteoinductivity, improve stability by promoting bone growth and integration, and enable controlled release over time for the optimization of the therapeutic effect.^{10–12}

In the development of HAP and HAP-hybrid biomaterials, wet synthesis methods are commonly employed as they offer a degree of control over the reaction parameters. However, these systems have some limitations associated with the lack of reproducibility.^{13–15} Stirred tank reactor systems are often characterized by heterogeneous compositional distribution in the reaction medium, resulting from the lack of control of parameters such as temperature and concentration and an overall non-efficient micromixing, which affects the steadiness

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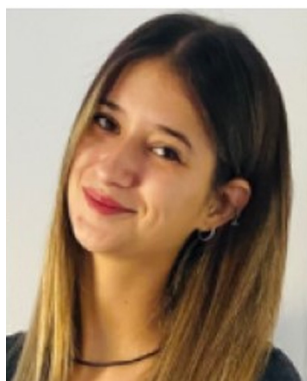
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of the final product.^{16,17} Consequently, a primary focus in current research endeavors is the establishment of new methodologies that allow reproducibility and scalability.¹⁸ In this context, continuous production offers advantages in terms of efficiency, quality control, scalability, cost-effectiveness, and the ability to tailor product characteristics. These benefits make continuous production an attractive option for industry and researchers involved in CaP synthesis and its diverse applications.^{19,20}

In the pharmaceutical industry, reactors working in continuous mode are considered cutting-edge technology for end-to-end production in a single, uninterrupted process line.²¹ Continuous processes in microreactors have also been successfully applied to obtain HAp and HAp-based nanoparticle materials with controlled properties.^{17,19} More recently, meso-oscillatory flow devices have been reported. In this system, the fluid is oscillated through a series of compartments or



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channels evenly separated by baffles. Oscillation of the fluid coupled to the baffles induces periodic changes in flow direction and intensity, creating a unique flow pattern. This oscillatory flow pattern promotes efficient mixing and enhances transport phenomena (including mass and heat transfer) by continually ensuring that reactants come into contact. Mixing intensity is governed by the oscillation amplitude and frequency.²¹

Although this approach using continuous flow reactors is gaining a lot of relevance, there are still only a few studies available to demonstrate its full potential.^{17,19,21} Such a degree of control over nano-HAp production will contribute to achieving higher quality in subsequent processing steps into a bulk material or 3D construct.

In this context, three-dimensional (3D) printing has become a promising tool in tissue engineering as it allows for the creation of highly precise and customizable scaffolds to support the growth of cells into functional tissue. The processing of HAp using 3D printing allows for the creation of highly porous and intricate structures with controlled microarchitecture.²² The ability to precisely control the composition and microstructure of HAp-composite materials opens new avenues for optimizing their biological and mechanical properties for specific medical applications.^{22,23} The shift towards the implementation of large-scale 3D printing technology has been a

gradual process. Nevertheless, a satisfactory solution has not yet been achieved to effectively integrate existing approaches for producing uniform and scalable HAp particles into 3D matrices. Hence, continuous productions can play a crucial preceding step in the development of customized 3D HAp constructs.

In this review, the latest developments in HAp-based biomaterials are comprehensively discussed by addressing the critical requirements and manufacturing processes involved in the light of 3D printing technology. Additionally, the significant challenges and prospective outlook of this rapidly advancing field are discussed in detail.

2. Opportunities for advanced HAp production

HAp is produced on a great scale per year to be used directly in patients or to fabricate medical devices. The batch production of this inorganic material is quite industrialized and reliable.²⁰

The final properties of HAp depend on several factors such as the initial concentration of reagents used, the Ca/P molar ratio, the synthesis temperature and pH, and mixing efficiency.^{11,16} In standard precipitation systems, a Ca/P of 1.67 (pure stoichiometric



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Table 1 Common aspect ratio ranges and physicochemical alterations for HAp synthesized through precipitation

Aspect ratio range	Morphological structure	Surface area	Porosity	Mechanical properties	Cell interaction	Reported applications	Ref.
Low	Nearly spherical	Moderate	Limited	Minimal impact	Interaction with limited area	Drug delivery, scaffold fabrication	30, 31
Medium	Elongated, rod-like	Increased	Variable	Can enhance anisotropy	Enhanced attachment	Bone regeneration, drug delivery	26, 27
High	Needle-like	Increased	Increased	Potential for increased strength	Enhanced alignment	Scaffold reinforcement, bone regeneration	29

mineral HAp), physiological temperature of 37 °C, pH between 4 and 12, and low concentrations of calcium and phosphate precursor solutions promote the precipitation of the less soluble CaP (HAp) and minimize the precipitation of other calcium phosphate phases.^{24,25} The resulting HAPs are typically nano-sized agglomerated particles, with low crystallinity and a medium to high aspect ratio range, which refers to the ratio of its length to its width or diameter.^{26–29} Spherical HAPs are also reported in the literature, which have more uniform mechanical properties, lower surface areas, and subsequently lower dissolution rates^{30,31} (Table 1).

The incorporation of other materials can lead to changes in these properties, which can require further optimization of the operating parameters.²⁴ To improve the batch-to-batch variability and fine-tuning of HAp crystallinity and nanoparticle size, continuous production can be an effective strategy for an easy scale-up.^{20,32}

In terms of operation mode, continuous processes offer advantages such as the efficient use of reagents, enhanced control over operating parameters, and consistent product quality. These processes also present lower costs and footprint, while being more energy-efficient due to reduced empty-fill and heat-up cycles. Moreover, an improved yield and capacity of continuous production allow higher productivity to be achieved.^{33,34} Continuous HAp precipitation methods offer the advantage of being easier to monitor and automate. However, a key challenge in implementing such methods is to achieve a perfectly mixed flow to reduce non-uniformities in the reaction medium and ensure particles suspension by minimizing concentration and temperature gradients. The attainment of a perfectly mixed flow is fundamental to the manufacturing of HAp particles with uniform attributes (*i.e.*, morphology, particle size distribution, among others).^{17,35}

2.1. Continuous stirred tank reactors (CSTRs)

Plug-flow reactors (PFRs) are widely used in continuous processes and are characterized by a tubular geometry with openings on each end for reactants and products to flow through. In these systems, the contents flow like plugs, from inlet to outlet, moving along at the same speed with minimized/reduced backmixing (Fig. 1(A)).³⁶ The lack of axial fluid interaction is a major drawback in precipitation systems where mixing intensity plays a crucial role in the creation of a homogeneous supersaturated solution and particle suspension. Furthermore, clogging issues may occur due to the deposition and sedimentation of solids on the reactor walls,

especially for long-run and high solid density experiments.¹¹ Thus, these reactors are more adequate for applications of gas or liquid phase systems.³⁶

In continuous stirred tank reactors (CSTRs), although the reactants are also continually introduced into the reactor, and the products are concurrently collected after the reaction, the reactor contents are continuously stirred.^{37–39} In similarity to mixed suspension mixed product removal (MSMPR) systems, CSTRs are equipped with stirring impellers and turbines, along with an integrated temperature control system (Fig. 1(B)).⁴⁰

CSTRs offer a stable supersaturation due to their steady-state operation.³⁷ These reactors often have a high working volume and can be connected in series to conduct multiple-step processes CSTRs in cascade.^{38,39} In recent years, double CSTRs set-ups to precipitate HAp have been implemented.⁴¹ However, the deposition of reaction products, which occurs after some time of operation, is also frequent, leading to the recurrent need to stop the process and wash the reactor.⁴¹ In addition, CSTRs are characterized by slow heat and mass transfer, *i.e.*, heterogeneous distributions of parameters such as temperature and concentration, leading to the heterogeneous distribution of supersaturation in the reaction medium. Despite achieving macromixing through stirring in CSTR and MSMPR systems, micromixing at the molecular level is not effectively controlled. While this issue is more pronounced in batch systems, it still exists in continuous setups, thereby negatively impacting the reproducibility of the process and the quality of the resulting product.⁴²

The aforementioned issues are further exacerbated during scale up, described as “how to design a pilot or industrial reactor able to replicate through a standard methodology the results obtained in the laboratory”.⁴³ CSTR mixing is intrinsically dependent on scale and laboratory scale experiments cannot reliably predict the behavior of large-scale plants, where stagnant zones and excessive shear rates at the impeller tip can limit the effectiveness of the reactor.⁴³ Due to these limitations, several reactors have been developed to better control the reaction conditions, thereby optimizing the final product's characteristics.⁴⁴

2.2. Microreactors

Microreactors are scaled-down systems (dimensions in the sub-millimeter range) that have been used to produce HAp, overcoming some of the reported drawbacks of the CSTR systems. Transfer processes, as well as residence time, occur faster in these reactors, owing to their high surface-area-to-volume ratio,



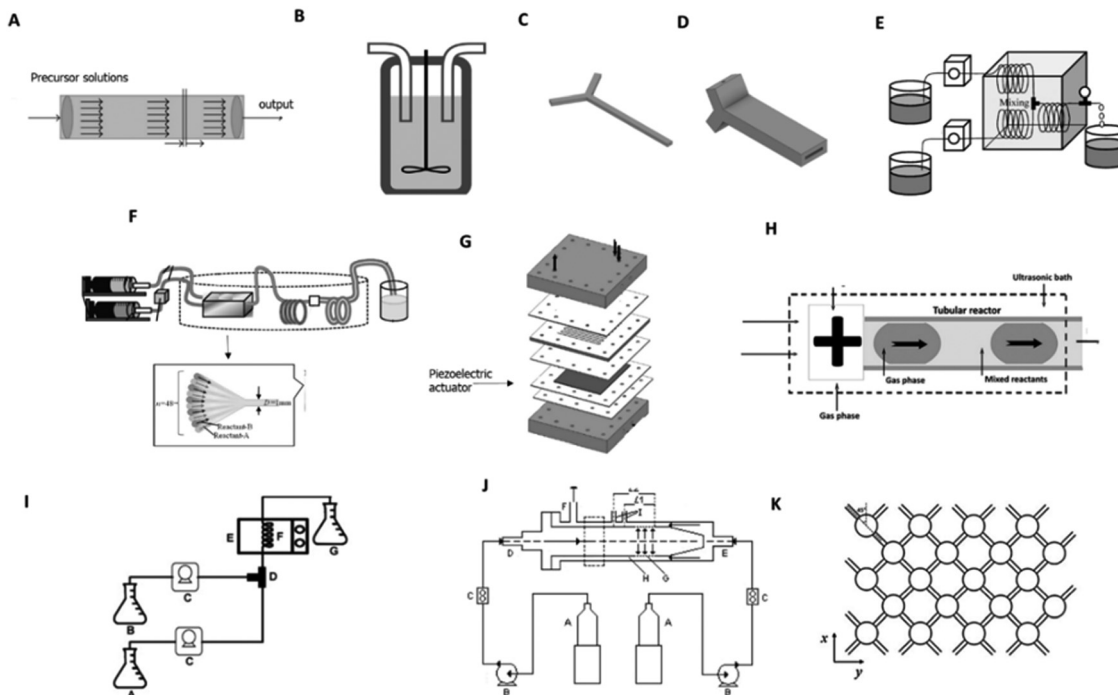


Fig. 1 Reactors applied for the continuous production of HAp: (A) plug-flow reactor; (B) conventional CSTRs;^{38,39} (C) tubular Y-shaped reactor;⁴⁷ (D) scaled tubular Y-shaped reactor;⁴⁸ (E) tubular T-connection reactor;⁴⁹ (F) microreactor with 48 mixing channels;⁵⁰ (G) tubular reactor with ultrasonic system;²⁸ (H) segmented gas-liquid flow tubular reactor;²⁸ (I) CSTRs system with microwave aiding apparatus;^{53,54} (J) MTMCR⁵⁶ and (K) NETmix reactor.^{58,59}

making these reactors interesting tools for high-throughput processing.⁴⁵ The reduced energy and reagent requirements make these devices ideal for research and development activities. Moreover, process optimization and scalability are also considerably easier, either by increasing the running time or by running reactions in parallel (a process known as scale-out).^{13,46} For these reasons, several microreactor configurations have been developed and applied in HAp-manufacturing procedures.

2.2.1. Tubular reactors. Various types of tubular reactors have been documented, with the prevalent configuration featuring two input channels and one reaction channel that come together at a T-junction or Y-shaped junction. At this juncture, the reactants collide before progressing through the reaction channel as the chemical reaction takes place.^{47–49}

Y-shaped reactors using 3D printing technology have been recently reported to obtain structures with precise geometry and channel dimensions. In the reported works a fused deposition modeling (FDM) technique was implemented using acrylonitrile butadiene styrene (ABS) polymer, which is stable under the conditions of the HAp precipitation and provides the necessary durability for the pressure used in the reaction system.^{47,48}

Latocha *et al.*⁴⁷ used two Y-shaped continuous reactors of different lengths, operating at atmospheric pressure and room temperature to produce nano-HAp and spherical, lecithin-modified nano-HAp (Fig. 1(C)). Although the methodology used was successful, the small dimensions of the reactor resulted in

a low production rate. In the work of Wojasiński *et al.*,⁴⁸ this problem was minimized by scaling up and numbering up this reactor (Fig. 1(D)).

Some alternative setups have been proposed to guarantee the mixing of the precursor solutions. Fujii and co-authors proposed a continuous flow tube reactor in which two plunger pumps are connected to a T-shaped mixing unit and a polytetrafluoroethylene (PTFE) reaction tube. In this system, the mixing of the precursor solutions is achieved before and after precipitation by using a coiled structure (Fig. 1(E)).⁴⁹ This approach is considered a passive mixing mechanism since it focuses on changing the reactor geometry to improve its mixture. Nano-HAPs were also effectively obtained by adding a microreactor unit with 48 channels before a coiled structure, each with a width of 1 mm to enable uniform mixing of suspensions on the micrometer order (Fig. 1(F)).⁵⁰

A different approach followed by Castro *et al.*,²⁸ to assure appropriate mixing and minimize HAp aggregation, was the implementation of an integrated piezoelectric element in a microreactor to allow direct transmission of ultrasound (Fig. 1(G)). The reported reactor led to a significant reduction in particle aggregation.²⁸ Lorenzo *et al.*,⁵¹ achieved a similar effect by immersing a tubular reactor in a bath with a sonicator.

Segmented flow reactors are also an alternative for HAp production and can narrow residence time distribution (Fig. 1(H)). In these reactors, reactants are segregated by immiscible fluids to form bubbles (gas in liquid) or droplets (dispersed liquid in a carrier liquid).⁵² Microwave heating has also been used for



continuous HAp production, as an optional energy source (the application of high frequency electric fields induces heating through the dipolar polarization of polarized molecules resulting from chemical bonds) (Fig. 1(I)).^{53,54}

Although the continuous processes implemented have been successful in the synthesis of HAp, the need for special types of equipment and complex set-ups can restrict its use to a laboratory scale.²¹

2.2.2. Industrial scale microreactors. Scaling up of microreactors is the focus of many researchers that aim to increase the throughput while preserving the reaction physics and channel flow hydrodynamics to obtain particles with more uniform characteristics at higher quantities, more suitable for industrial processes. One of the most important issues when stepping out of laboratory-scale is how to ensure flow equidistribution with minimal pressure loss in each microchannel.⁵⁵

Yang *et al.*,⁵⁶ adopted a microporous tube-in-tube microchannel reactor (MTMCR), which consists of two coaxial tubes, *i.e.*, the outer tube and the inner tube, forming an annular microchannel. The liquid in the inner tube through the annular micropore belt is dispersed into several small streams, followed by the high-speed impinging with the laminar flow in the chamber between inner and outer tubes. The structure of the annular microchannel greatly enlarges the throughput of MTMCR, compared to other microreactors (Fig. 1(J)).

The NETmix Reactor is another example of modified microreactors that is currently used in FLUIDINOVA, a company that focuses on the production of HAp-driven materials for medical, oral, and food applications.⁵⁷ This device consists of a network of several mixing chambers interconnected by channels oriented at a 45° angle from the main flow direction. Reactants are injected at the inlet channels. Mixing occurs in these chambers when the two streams coming from each channel meet (Fig. 1(K)).^{58,59}

Nevertheless, the reported reactors are associated with some disadvantages. Microreactors are usually characterized by geometries with a low Reynolds number (Re), which is the ratio of inertial forces to viscous forces. When Re is large, inertial forces dominate over the viscous forces, the fluid is flowing faster and the flow is turbulent. On the other hand, when Re is low, the viscous forces are dominant, and disturbances in the flow are damped out by viscosity. In this case, fluid particles are kept in line generating a laminar flow. In microreactor channels, laminar flow is dominant and is characterized by a parabolic velocity profile, leading to residence time distributions. The parabolic velocity profile and obstruction, by wall attachment of particles and by particle aggregation, limits the use of these reactors.¹⁶ This hinders the handling of solid particles and the overall transportation efficiency of the process.

2.3. Oscillatory flow reactors (OFRs)

2.3.1. Conventional OFRs. Oscillatory flow reactors (OFRs) have been proposed in the literature as an alternative to stirred tank and tubular reactors due to their superior mixing and efficient heat and mass transfer capabilities, particularly in

liquid–liquid systems or for homogeneously suspending solid particles.^{19,32,34}

OFRs are operated under oscillatory flow mixing (OFM), which occurs when the reactor is comprised of uniformly spaced orifice deflectors called baffles, and a liquid or a multi-phase mixture is oscillated in the axial direction by means of diaphragms, bellows or pistons. The generated flow accelerates and decelerates according to a sinusoidal velocity–time function. The interaction of the flow with the baffles generates vortex rings, which are formed downstream of the baffles. When the flow decelerates these vortices move into the centre of the reactor, where a well-mixed and uniform region is generated.^{60,61}

The intensity of mixing can be varied by tuning the oscillatory conditions (amplitude (x_0) and frequency (f) of oscillation) and several different mixing regimes have been identified, ranging from laminar to fully turbulent flow. The ability to generate radial mixing gives a unique form of control with respect to the intensity of mixing, axial dispersion and transfer processes.^{62,63}

Mixing is influenced by a combination of different geometric parameters and operational parameters (baffle area, baffle spacing, baffle thickness, oscillation frequency, and amplitude). The major advantage over tubular reactors, where mixing is dependent on the applied flow rate, is that in OFR mixing is essentially controlled by the oscillation frequency and amplitude. Hence, since mixing is independent of flow rate, it implies that OFRs can be used in processes with longer kinetics without the need to use larger tubes. In the case of tubular reactors, to increase the residence time, the flow rate must be decreased, which compromises mixing, or a tube length increase is required.⁶⁴

2.3.2. OFR with smooth periodic constrictions (SPCs). Despite their advantages, conventional OFRs present limitations associated with the existence of dead zones or stagnate regions near the angle between the baffles and the reactor. To overcome this problem Reis *et al.*,⁶⁵ proposed an OFR with smooth periodic constrictions (SPCs) suitable for biotechnological processes at the mesoscale, reducing dead volume areas and shear stress that can be detrimental to bacterial and other cell cultures.⁶⁵

Regarding HAp synthesis, OFRs with SPCs allows process intensification with nano-HAp particles being obtained 4 times faster than in stirred tank batch reactors, and with no intermediate CaP phases being formed.¹⁹ In the work of Castro *et al.*,³² HAp with controlled properties was obtained in the scaled-up OFR operating in continuous mode, demonstrating the feasibility of this technology and its innovative potential. The particle obtained also promoted cell proliferation of osteoblastic-like (Saos-2) cells³² (Fig. 2(A)).

However, secondary nucleation, agglomeration, clogging, and solid deposition can occur.^{34,61,65} To overcome the previously stated issues, and prevent solids deposition, a new design (modular oscillatory flow plate reactor) (US20190111402A1) was developed and the common circular cross-section was replaced with a rectangular one which is less prone to particle



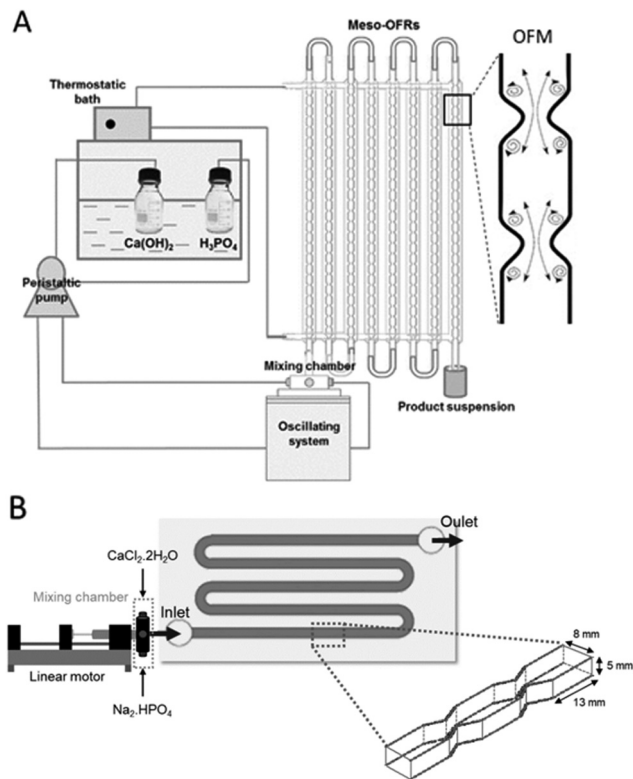


Fig. 2 (A) Scaled-up meso-OFRs and oscillatory flow mixing inside a reactor and (B) MOFPR used for continuous manufacturing of HAP particles.^{14,17}

accumulation close to the constrictions compared with the circular cross-section of conventional oscillatory flow reactors with sharp baffles.⁶⁶ According to Cruz and the research team,⁶⁴ contrary to the circular cross-section, which has an undulating base where the particles can be trapped while flowing along the tube, the rectangular cross-section has a flat base where the particles can flow in a less constricted path.

2.3.3. Modular oscillatory flow plate reactor (MOFPR). The optimized reactor, that fulfils the gaps of the described OFRs is the modular oscillator flow plate reactor (MOFPR).⁶⁶ Its 2D-SPCs and rectangular cross-section with optimized geometry and dimensions enable reduced shear stress and improved solid suspension and transportation.⁶⁶ The plate reactor can be assembled and disassembled easily for cleaning, which is bound to be necessary at some point in reactors where precipitation is carried out. Furthermore, the MOFPR technology is suitable for multiphase systems applications such as HAP manufacturing.^{34,67–69}

This system has been used to produce CaP particles with tailored properties for tissue engineering applications (in terms of morphology, particle size distribution and crystallinity) simply by changing the oscillation parameters (f , x_0). Moreover, other parameters such as the initial reagents concentration, initial Ca/P molar ratio, and temperature can also be easily modified in the applied process, leading to the synthesis of specific phases of CaPs in high amounts.¹⁴ The MOFPR has also been applied in the development of hybrid HAP-particles

such as HAP-sericin and HAP-sericin-cerium materials, that improve cell viability of human dermal fibroblasts.⁷⁰ Hence, MOFPRs showed promising results for the continuous production of HAP and HAP composite powders with high potential for both bone and skin-tissue engineering (Fig. 2(B)).^{70,71}

This technology offers the potential to customize the production of HAP and HAP-hybrid systems, ensuring a reproducible system that serves both academic and industrial purposes. This includes synthesizing significant quantities of uniform particles essential for characterization techniques and *in vitro* viability assessments in academia, as well as ensuring batch consistency for industrial applications.

3. Advanced HAP and HAP-hybrid systems

While HAP has been widely used in tissue engineering, certain limitations, such as low mechanical strength, poor resorption properties, processing challenges, and limited bioactivity⁴ have driven researchers to explore innovative approaches, such as ion doping and incorporation of bioactive factors. Ion doping aims to enhance HAP's properties by introducing various ions into its crystalline structure. This approach, driven by the ion exchangeability of HAP, brings forth new characteristics such as osteogenic activity and conferring antimicrobial properties and magnetic properties.^{72–74} On the other hand, a combination of biological cues introduces the addition of growth factors and cells into the HAP framework. This strategy aims to create synergistic effects that enhance HAP's potential in tissue engineering such as promoting bone tissue formation and angiogenesis.¹⁰

3.1. Ion-doping

3.1.1. Enhanced osteogenic and antimicrobial activity HAP. The literature describes two HAP crystal forms: a monoclinic form in the $P2_1/b$ space group and a hexagonal form in the $P6_3/m$ space group.⁴ The hexagonal phase is more common due to the susceptibility of the monoclinic form to destabilization by impurities. The hexagonal structure has two Ca sites: Ca1 (aligned with OH^- channels) and Ca2 (arranged in a staggered triangular pattern). These sites are within a phosphate assembly with parallel OH^- ion-filled channels along the crystallographic c -axis.⁷⁵ The introduction of foreign ions into the crystal lattice of HAP during its synthesis involved the following steps: (1) preparation of precursor solutions containing the ions that will be incorporated into the HAP lattice.^{76,77} These precursor solutions can include calcium and phosphate sources along with the doping ions. During the precipitation process, ions from the precursor solutions are incorporated into the growing HAP crystal lattice. This can occur through two main mechanisms: (2) ion exchange or ion substitution in which OH^- , Ca^{2+} , and PO_4^{3-} can be replaced by different ions (Fig. 3). (3) As the precipitation reaction progresses, the introduced ions become integrated into the crystal lattice of HAP. (4) The doping ions occupy lattice sites that would have been



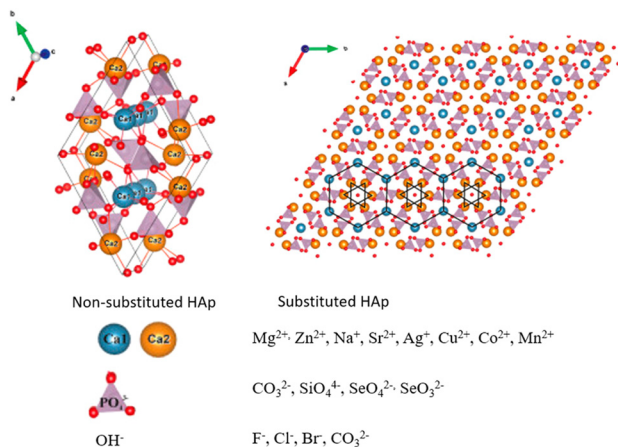


Fig. 3 Ion substitution (Ca^{2+} , PO_4^{3-} and OH^-) in the HAP lattice. Adapted from ref. 75 and 78.

occupied by the host ions if undoped HAP were forming. (5) The crystal growth of doped HAP continues as more ions are added to the lattice.⁷⁸ After substitution, HAP retains its hexagonal structure.^{75–78}

These substitutions result in unique characteristics of the final molecule and are possible due to the high ion exchangeability of HAP.⁷⁹ For the bulk material, these differences can be perceived in the macro crystal morphology, mechanical properties and even biological behavior (Table 2).⁴

When considering doped HAP for biomedical purposes, two major applications can be encountered: one relating to bone healing, in the way that many substitutions enhance HAP's osteogenic activity, promoting cell proliferation and differentiation into an osteoblastic phenotype;^{72,73} and other topics relate to substitutions that give rise to antimicrobial HAP.^{91,92} Many times, the ions chosen for calcium substitution led to both these characteristics (enhancement of biological response and antimicrobial properties).^{72,93,94}

The ions that exhibit enhanced osteogenic activity are widely acknowledged in the literature, among which strontium (Sr^{2+}) and zinc (Zn^{2+}) are the most frequently mentioned, and often combined.^{93–95} Strontium alone is said to improve the mechanical properties of HAP scaffolds, and its release into cell medium can activate transduction pathways that lead to osteogenesis.^{96,97} Jiang *et al.*,⁹⁶ doped HAP with Sr^{2+} ions in different percentages, from 2.5 to 20 wt%, which promoted

osteogenic activity *in vitro* and led to both bone and vasculature regeneration *in vivo* and, in both scenarios, the best results were achieved with 10% Sr^{2+} .⁹⁶

Zn^{2+} is the best example of a multipurpose ion since it is reported as an inducer of matrix mineralization towards new bone formation while presenting antimicrobial and anti-inflammatory properties.^{4,91} Zn^{2+} presents osteogenic properties associated with its participation in many regulatory pathways, being a co-factor of many enzymes, and inhibiting osteoclastic activity, hence, bone resorption.⁹¹

Other ions that are applied in HAP substitutions that lead to enhanced bone regeneration are silicon,⁹⁸ magnesium,⁴ lithium⁹⁹ and selenium¹⁰⁰ (Si^{2+} , Mg^{2+} , Li^+ and Se^{2-} , respectively).

Leu Alexa *et al.*,⁴ doped HAP with Mg^{2+} and Zn^{2+} , separately, and developed 3D-printed scaffolds with each of the doped powders, dispersed in a polymeric matrix. They found that Mg^{2+} induced a better behavior at 5%, while Zn^{2+} did at 0.5%. Ultimately, both improved cell viability and osteogenic behavior when compared to the control (with non-doped HAP). Nonetheless, the scaffold containing Mg^{2+} -doped HAP exceeded the effects of the one containing Zn^{2+} .^{4,94}

Concerning antimicrobial activity, ions such as fluoride,⁷² silver,^{92,101,102} zinc,^{91,103} magnesium and copper (F^- , Ag^+ , Zn^{2+} , Mg^{2+} , Cu^{2+} , respectively) were revealed to be a valid alternative when combined with HAP.

Although Zn^{2+} has presented osteogenic abilities, the local delivery of this ion can also act as a substitute for conventional antibiotics (it can destabilize cell membranes and deactivate enzymes⁹¹), which is always an advantage since bacterial resistance is a rising worldwide issue. The drawback of Zn^{2+} , as with other antimicrobial ions, is its cellular toxicity. Consequently, the integration of Zn^{2+} ions into HAP molecules represents a valuable strategy, as it enables controlled delivery and promotes appropriate bioavailability, thereby mitigating any potential toxic effects.¹⁰³

Despite the numerous advantages of single ion substitutions, recent advancements have shifted towards the implementation of co-substitutions, where two ions are incorporated into the crystalline structure of HAP.^{4,93,95,98,104,105} Another approach involves the fusion of two single substituted HAP into a single three-dimensional material. Silver is widely used as an antibacterial/antifungal substance and is present in many

Table 2 Changes that occurred with ion doping in HAP-precipitation systems

Properties	Changes that occurred with ion doping	Ions	Ref.
Mechanical strength	Ion substitution can increase compressive strength, fracture toughness, and hardness.	Sr^{2+} , Fe^{3+}	80–82
Biocompatibility	Dopants can enhance cell attachment and proliferation.	Sr^{2+} , Zn^{2+} , Li^+ , Se^{4+}	83–85
Crystallinity and crystal structure	Substituting ions with different sizes and charges can affect crystallinity and lattice stability.	Zn^{2+} , Ba^{2+} , F^- , Fe^{3+}	77,82,86
Surface properties	Ion doping can alter surface charge and wettability, affecting interactions with biological tissues and fluids.	Sr^{2+} , Zn^{2+}	87,88
Optical properties	Certain dopants can introduce coloration or luminescence, which is potentially useful in imaging and diagnostics.	Fe^{2+} , Fe^{3+} , Gd^{3+} , Eu^{3+} , Co^{2+}	89
Thermal stability	Dopants can influence the material's response to temperature changes, affecting its thermal stability.	Al^{3+}	90



commercially available products, such as wound dressings. Their antibacterial effect is related to its electrostatic interaction with cell membranes resulting in its rupture.⁹² Ag⁺ is also highly referred to in the literature of doped HAp.^{92,101,102} Sinulingga *et al.*,¹⁰⁶ made two variations of 2.5% Ag⁺-doped HAp, one with Zn²⁺ (2.5%) and another with Mg²⁺ (2.5%),⁹² in order to try to mitigate the cytotoxic effect found with previously reported 5% Ag⁺-doped HAp.¹⁰⁶ The presence of Zn²⁺ and magnesium enhanced the antimicrobial activity (when compared to the sample with 2.5% Ag⁺), since both 2.5 Ag⁺/2.5 Mg²⁺ and 2.5 Ag⁺/2.5 Zn²⁺ killed 99 ± 1% *E. coli* (similarly to the 5% Ag⁺HAp), while presenting lower cytotoxicity towards the tested pre-osteoblastic cells.

3.1.2. Magnetic HAp. The number of routes for producing magnetic nanoparticles in the biomedical field has increased steadily over the years due to their paramagnetic and superparamagnetic behavior. In the first case particles are magnetized by an external magnetic field, aligning parallel to it in a permanent manner, while in the second, particles behave similarly to in the paramagnetic case but their magnetization ceases alongside the magnetic field.¹⁰⁷

Magnetic micro and nanosystems have garnered significant attention from the scientific community due to their potential applications in various fields, including bioimaging, hyperthermia treatments, tissue regeneration, and controlled drug release.

The integration of magnetic nanoparticles into drug delivery systems has been shown to offer several advantages over conventional delivery methods. One key advantage is the ability to precisely guide the particles to the target site through the application of external magnetic fields. This ensures that the carried drug is distributed specifically to the intended region, reducing the likelihood of side effects associated with the drug, particularly in cases where the drug is toxic (such as chemotherapy drugs). In addition to targeted delivery, the release pattern of the drug is another important factor to consider. The sudden release of a high dose of the drug can lead to toxic effects, which can be mitigated by controlling the rate and pattern of drug release. Magnetic drug delivery systems have the potential to overcome this issue, as external magnetic fields can be used to modulate the release of the drug from the magnetic nanoparticles.⁷⁴ This enables the release pattern to be fine-tuned, ensuring that the drug is delivered at the optimal rate to minimize toxicity and maximize therapeutic efficacy.

Iron (Fe) doped HAp has been widely studied as a less toxic substitute for magnetite SPIONS which are, to this day, the most used magnetic nanoparticles for biomedical applications.^{107–110} The cationic substitution of calcium with iron (Fe²⁺ or Fe³⁺) induces deformations in the HAp lattice, and alters the magnetic properties of the material, from diamagnetic (pure HAp), to superparamagnetic (iron doped HAp).⁷⁹ Also, Fe-HAp can have different grain sizes, enhanced thermal stability and mechanical properties and higher bioactivity.

3.2. Combination of biological cues

3.2.1. Bioactive factors. Regarding the development of advanced HAp-based systems, the incorporation of bioactive

factors is a commonly employed approach. Due to the short half-life and rapid diffusion of growth factors by body fluids, combining them with CaP can lead to the creation of controlled release systems. This is particularly important for medium to large-sized bone defects that require long-term active bone healing processes. Of the various members of the bone morphogenetic proteins (BMP) family, BMP-2 has been shown to play a crucial role in promoting osteogenesis and chondrogenesis in several clinical trials. Currently, this protein is applied in non-union bone fractures, spinal fusions, and periodontal bone regeneration.¹⁰

The simplest approaches to produce these biomaterials include the deposition of BMP-2 onto CaP-based scaffolds (Fig. 4(A)). Under physiological conditions of temperature (37 °C) and pH (6–8), the solubility of CaPs and, consequently, their degradation *in vivo* is higher for β-TCP than for HAp.^{11,12}

According to the literature, the degradation rate of HAp can range from several months to years^{11,116} depending on factors such as the size of the implant and the presence of any impurities or dopants. On the other hand, the degradation rate of β-TCP can range from several weeks to several months, with the material gradually dissolving and being replaced by new bone tissue.^{117,118} As a result of the greater degradation rate, β-TCP is usually selected to develop 3D scaffolds that can effectively control BMP-2 release.^{12,112,113,119,120} Moreover, the incorporation of BMP-2 leads to the formation of new bone with denser connective tissue and well-supplied with cellular components (Fig. 4(A1)),¹¹² which can be associated with the expression of key genes in osteoblast differentiation (Smad1, RUNX2 and OCN) (Fig. 4(A2)).¹¹³

In Xu's research work,¹¹¹ it was proven that a HAp-scaffold coated with polymeric materials can be established to overcome the issues associated with HAp low degradation. Since growth factors are adsorbed at the surface of the CaPs, its loading is limited. Coating the HAp structure with other materials can increase their incorporation as a result of increased surface area (more binding sites) and tuneable affinity (targeted binding enables a higher degree of growth factor attachment).^{121,122} In this study, HAp porous scaffolds were coated with poly(L-lactide) (PLLA) fibers and supplemented with BMP-2 into the dorsal muscles of canines. The formation of bone was confirmed after a period of four weeks (Fig. 4(B3)). Similar results in a calvarial defect *in vivo* model were achieved by coating with ε-polycaprolactone (PCL),¹²³ or *in vitro* by modifying a HAp scaffold with collagen and BMP-2 conjugation.¹²⁴ The scaffold microstructure and resulting degradation rate can be manipulated by using biphasic calcium phosphate scaffolds, such as HAP/TCP systems.^{125,126}

The use of polymeric matrixes in combination or incorporated with CaP particles is another route proposed to combine CaP with growth factors (Fig. 4(B)).¹²⁷ In the work of Li *et al.*,¹¹⁴ a PCL scaffold with nano-HAp was coated with a polydopamine layer to facilitate covalent conjugation with BMP-2. The resulting material was used as a bone filler in rabbit calvaria defects *in vivo*, leading to substantial bone growth after 6 weeks and complete bone formation after 15 weeks (Fig. 4(B1)). In recent



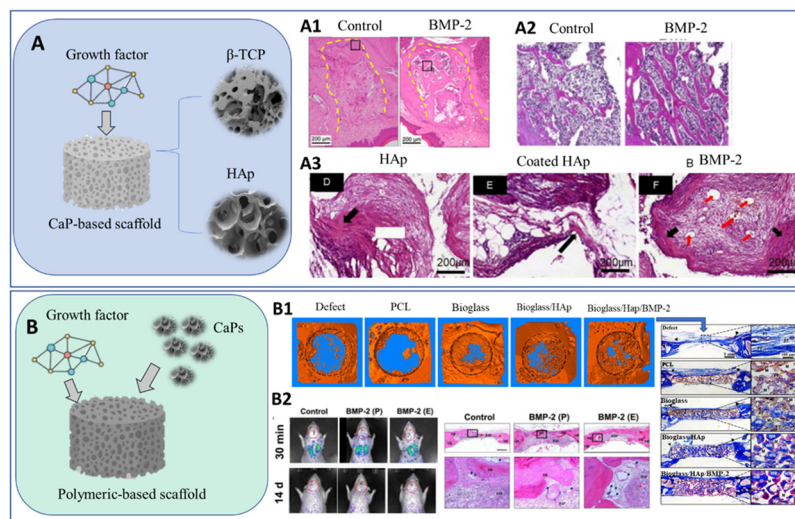


Fig. 4 Schematic representation of conventional CaP systems with growth factors using (A) β -TCP and HAp as structural scaffolds^{12,111} and (B) applying other polymeric-based scaffolds loaded with CaP particles. Histology results after implantation: (A1) bone formation in tooth extraction sockets, 3 weeks after implantation;¹¹² (A2) bone formation 12 weeks after implantation in a femur bone defect;¹¹⁵ (A3) bone and vessel formation (red arrows), 12 weeks after implanting in a subcutaneous model;¹¹¹ (B1) micro-CT and histology 6 weeks after implantation (blue and red represent immature bone matrix and maturing bone);¹¹⁴ (B2) migration of BMSCs after injection of the biomaterial through rat tail veins and bone formation after 12 weeks of implantation (BMP-2(P): scaffold with physically adsorbed BMP-2 and BMP-2 (E): scaffold with BMP-2 encapsulated into silk fibroin microspheres).¹¹⁵

studies, natural polymers such as silk fibroin have been explored for the same purpose, exhibiting sequential and sustained release of BMP-2 (Fig. 4(B2)).¹¹⁵

Although BMP-2/HAp composites have shown promising results in various *in vivo* studies, the intricate healing processes involved in bone tissue engineering may necessitate the inclusion of additional growth factors. Therefore, dual-releasing systems with melatonin (which modulates bone formation and resorption as it promotes the differentiation of osteoblast precursors toward osteoblasts),¹²⁸ SDF-1 (Stromal cell-derived factor-1) (which triggers the migration of stem cells and progenitor cells to injury sites during the acute phase of bone repair),¹¹⁵ nerve growth factor (NGF) (regulating the functional attributes of central and peripheral neurons, such as development, differentiation, growth, and regeneration, which can effectively enhance osseointegration around implants),¹²⁹ TGF- β 1 (Transforming growth factor beta) (causes induction of localized chondrogenesis when injected subperiosteally)¹³⁰ and platelet-derived growth factors (PDGF) (which increases the distance of cell infiltration into scaffolds)¹³¹ have been reported to recreate natural biological cues.

The majority of published research on the incorporation of growth factors in bone tissue engineering has centered on BMP-2/vascular endothelial growth factor (VEGF) combinations. VEGF is known to play a role in the regulation of vascular development and angiogenesis, while also indirectly stimulating the differentiation of mesenchymal stem cells (MSCs) into the osteogenic lineage by inducing BMP-2 expression. Thus, the combination of BMP-2 and VEGF has been reported to obtain ceramic biomaterials with both osteogenic and angiogenic properties.^{132,133} In fact, *in vitro* tests showed that dual systems have higher cell attachment and proliferation than HAp-based

scaffolds with single growth factors;¹³⁴ while when implanted *in vivo*, it can benefit bone mineralization and expression of important bone matrix proteins.¹³⁵ Critical size defects in New Zealand white rabbits were also successfully treated using a DNA-loaded nano-CaP material with BMP and VEGF.¹³⁶

3.2.2. Cell incorporation. The combination of HAp with MSCs can also be used as an approach to provide optimal performance in bone-tissue engineering, due to their ability to differentiate into osteoblasts or bone-forming cells. These cells can be isolated from different sources, such as human adipose-derived stem cells (ADSCs), which have the particularity of being largely available compared with other sources such as bone marrow and umbilical cord.¹³⁷ This incorporation is conventionally achieved by seeding the cells on top of the CaP scaffolds,^{138,139} or by mixing particles with the cells prior to *in vitro* implantation or injection.^{140,141}

Some studies report that the addition of these cells does not result in significant benefits in the enhancement of new bone formation *in vivo* models.^{138,142} In the work of Supphaprasitt *et al.*,¹³⁸ although a PCL-biphasic-CaP scaffold seeded with ADSCs enhanced *in vitro* cell proliferation and differentiation, when implanted *in vivo* the results were similar to a cell-free scaffold.

However, according to recent developments in this field, CaP-based biomaterials in combination with cells can result in an enhanced clinical response. Brennan and co-authors¹³⁹ developed HAp and TCP scaffolds and showed that their bone formation capacity was highly increased with the addition of MSCs. The differences in the effects of the combinatory approach (cells-CaPs) can result in differences in microporosity and specific surface area, which ultimately lead to optimal cell invasion, interaction, and ingrowth.^{139,143}



In addition to tailoring the CaP scaffold, the number of incorporated cells can affect the biological outcome. According to Mankani *et al.*,¹⁴⁰ the extent of bone formation can be improved by increasing the number of cells incorporated in the HAp/ β -TCP system, and a minimal MSCs dose is required for optimal bone formation (between 0.3 and 1.0 million cells per 40 mg transplant).¹⁴⁰ Another approach includes the encapsulation of cells in degradable hydrogel microbeads in combination with CaP material.¹⁴¹

The combination of growth factors and cell incorporation can be used to develop next-generation therapeutic tools. In Overman's work,¹⁴⁴ a short pre-treatment with BMP-2 had a strong effect on the osteogenic differentiation of ADSCs seeded in a HAp/TCP (60/40) scaffold. Promoted osteodifferentiation was also reported by Zhao,¹⁴⁵ after the addition of BMP-2 in a HAp/TCP encapsulated with MSCs. In applications that require strong mechanical properties, such as spinal fusion defects in rabbits, BMP-2/cell systems can also enhance biomechanical stiffness, as a result of more rapid bone formation and consolidation.¹⁴⁶

3.3. Conventional methodologies

Ion doping and growth factor incorporation in the context of a wet chemical precipitation process can involve different methodologies. In what concerns ion-doping, sequential precipitation, substitution during precipitation, ionic exchange, solid-state mixing, and impregnation are reported techniques in which the HAp particles are first formed and then a subsequent step is implemented to introduce the doping ions.^{147,148} However, the controlled introduction of foreign ions into a material during precipitation, also known as co-nucleation doping, is the most common approach. In this process, a solution containing precursor components is mixed with another solution containing reactants under specific conditions. The precursor ions are then incorporated into the growing crystal lattice of the precipitate as the reaction proceeds. As the reaction progresses, new solid particles form and precipitate out of the solution.¹⁴⁹

Stirred tank batch reactors and microreactors are used to carry out doped HAp synthesis.^{150,151} However, doping ions can affect the nucleation and growth kinetics or crystallization dynamics, leading to alterations in the final morphology of HAp particles.^{152,153} Hence, on one hand, the size and shape of the particles may be tuned by adjusting the concentration of the doping metal ions; on the other hand, it sets up an obstacle toward the production of HAp particles with the same morphology and size with different doping ratios.¹⁵²

Bioactive factors can also be added to HAp through various techniques to functionalize the biomaterial for specific biological applications. In the adsorption method, growth factors can be adsorbed onto the surface of HAp particles by immersing the HAp particles in a solution containing the growth factor. The growth factor molecules adhere to the HAp surface due to electrostatic interactions and other binding forces.¹⁵⁴ Growth factors with charged regions can also be electrostatically attracted to HAp particles, leading to their adherence on

the particle surfaces.¹⁵⁵ In addition, the inclusion of growth factors into coatings applied onto the surface of HAp particles is also possible using techniques like layer-by-layer deposition or spray coating.^{156,157} Moreover, recent studies have encapsulated growth factors and cells in carriers, such as nanoparticles, microspheres, or hydrogels to protect the growth factors from degradation and provide controlled release when placed in contact with HAp or tissues.¹⁵⁸ In these processes, the reactor is used to produce HAp, which is then submitted to a post-processing step to introduce the growth factor.

Incorporation of growth factors during synthesis can also be achieved by incorporating them directly into the solution during the HAp precipitation process. According to the literature, the addition of growth factors can result in alterations in precipitation kinetics and the final particle's properties.^{158,159}

4. HAp going 3D

Under physiological conditions, the tissues of the human body exhibit a complex 3D network structure. The development of alternative strategies that can accurately replicate the 3D anatomy and functions of tissues and organs remains a major challenge for bioengineers. The advancement of HAp-based scaffolds for clinical use necessitates a shift from traditional two-dimensional approaches towards more sophisticated 3D approaches. Additive manufacturing (AM) technologies provided hope for overcoming this critical challenge by allowing tailoring of the 3D scaffolds/implants in terms of: (1) shape; (2) architecture; (3) porosity; (4) pore interconnectivity; (5) material spatial distribution; (6) cell's spatial distribution; and, most importantly, (7) reduced fabrication times and reproducibility.¹⁶⁰ AM applied to tissue engineering and regenerative medicine presents a well-established workflow with 4 essential steps:

- Generation of the 3D model.
- 3D model processing in AM software.
- Produce the 3D scaffold by AM.
- 3D scaffold implantation (tissue replacement).

A vast number of materials have been processed by AM to produce HAp-based 3D scaffolds. HAp has been combined with several materials, both synthetic-based and natural-based materials, to develop suitable inks.²² Inks require specific attributes that make them good contenders to be applied in AM strategies. These characteristics include suitable rheological properties such as viscosity, shear-thinning behavior, surface tension, *etc.*¹⁶¹ However, it is important to consider the specific requirements and limitations of each AM technology when developing a HAp-based ink. Various AM technologies have been employed to fabricate such scaffolds, each with its own distinct requirements for the ink and a unique set of advantages and disadvantages. Nowadays, several 3D printed products incorporating HAp have been translated into the market. OsteoFab[®] is a 3D printed HAp-based implant that was developed by the medical technology division of the company 3D Systems. These implants are designed to assist



in bone regeneration and integration using a combination of polymer and HAp. OsteoFab[®] implants are created using additive manufacturing techniques, specifically 3D printing, which allows the production of patient-specific implants with complex geometries. The incorporation of HAp into the implant composition aims to enhance bone integration and promote long-term stability. MyBone[®], the innovative product developed by Cerium company, exhibits exceptional quality and remarkable efficacy in bone regeneration. This commercially available product has successfully promoted bone formation, highlighting its potential as a promising solution for bone tissue engineering applications.¹⁶²

4.1. Additive manufacturing (AM)

The AM methods that can be applied for the fabrication of tissue engineered HAp-based 3D scaffolds can be divided into four main groups: inkjet-based, laser-based, extrusion-based and indirect-based.¹⁶³ The rheological properties of the ink have a significant impact on the success of the AM process, particularly with regards to the viscosity increase that is commonly imposed by the inclusion of HAp (more interactions within the ink, leading to increased viscosity and a thicker consistency). These challenges manifest in various forms, depending on the specific AM method employed (Table 3).

Table 3 Additive manufacturing techniques

Technique	Process	Description	Advantages	Disadvantages	Ref.
Inkjet-based techniques	Thermal	Inkjet thermal 3D printing, also referred to as “Drop-on-Demand (DOD) 3D printing, uses an inkjet printhead to deposit droplets of a build material (often a photopolymer or a wax-like material) onto a build platform”. The droplets are expelled onto specific areas of the previous layer and are typically thermally cured or solidified to form a solid structure.	Speed and high resolution; multi-material printing; smooth surface finish.	Material limitations; post-processing required; size limitations.	164
	Piezoelectric	Inkjet piezoelectric 3D printing relies on piezoelectric crystals within the printhead. When an electrical voltage is applied to these crystals, they deform, causing a pressure wave that ejects a small droplet of material through a nozzle. The precise control over these piezoelectric actuators enables the deposition of materials in a highly controlled and precise manner.	Speed and high resolution; multi-material printing.	Material limitations; post-processing required; cost.	164
Laser-based techniques	Stereolithography (SLA)	SLA is one of the earliest and most widely used 3D printing technologies, known for its ability to produce high-resolution and highly detailed objects. It operates based on a process called photopolymerization using a UV laser or other light sources to selectively solidify a liquid photopolymer resin layer by layer.	Speed and high resolution.	Material limitations; post-processing required; cost.	165
	Selective laser sintering (SLS)	SLS is a powerful and versatile additive manufacturing technique known for its ability to produce strong, functional, and complex 3D-printed objects from a variety of materials. SLS employs a high-power laser to fuse powdered materials layer by layer.	Material versatility; geometry complexity; multi-part/object printing.	Surface finish; post processing required; material cost.	166
Extrusion-based techniques	Fused deposition modeling (FDM)	FDM is one of the most popular and accessible 3D printing techniques, known for its simplicity, versatility, and affordability. FDM works by melting and extruding a thermoplastic filament through a heated nozzle, which deposits the material layer by layer to build a three-dimensional object.	Affordability; material versatility; easy to use.	Surface finish; low resolution.	167
	Robotcasting	Robotcasting is an innovative 3D printing technique that combines aspects of both 3D printing and robotics to create large-scale structures and objects. This technique is based on a combination of robotics and traditional 3D printing principles. It involves a robotic arm with an attached nozzle that extrudes a material using a mechanical or electromagnetic actuator in layers to build structures.	Speed, material versatility, design freedom.	Material limitations; complexity.	167
Indirect-based techniques	Indirect 3D printing	Indirect 3D printing refers to methods where a temporary or sacrificial mold is 3D printed and then used as a template for creating the final object through various manufacturing processes. These techniques are used for applications where traditional 3D printing may not be the most suitable or cost-effective choice.	Material versatility; smooth surface finish; cost-effectiveness.	Low design freedom; time-consuming; material waste.	166



Inkjet-based printing allows controlled manufacturing using a droplet-based approach. The presence of HAP in the ink formulation (high viscosity) can be a huge barrier and this method is mostly avoided since it is only compatible with low viscosity inks (<10 mPa s).^{164,168} However, a considerable quantity of publications on the use of inkjet printing in the manufacture of ceramics, including HAP, have shown promising results.¹⁶⁸

The laser-based method uses as a major component a laser beam that focuses on the ink to selectively sinter it or concentrates its energy to deposit the ink in the substrate (receiver surface).¹⁶⁵ In the first case, since the beam is focused directly in the ink, it is incompatible to use cell-laden HAP bioinks due to the high energy density employed. In contrast, when the beam is used to deposit a thick layer of ink on the receiver surface, the direct printing of cell-laden bioinks is enabled without significantly affecting its viability.¹⁶⁹

The extrusion-based methods have been the most used and consensual among the scientific community as the most promising methods to fabricate HAP-based 3D scaffolds. These methods consist in dispensing the ink through a nozzle coupled to an extruder that is steered through a mechanical or electromagnetic actuator. The dispensing mechanism can use a robotic arm (robocasting) that moves in different coordinates (X ; Y ; Z) to deposit the ink. Robocasting allows the ink extrusion at low (*e.g.*, 15 °C) and high temperatures (*e.g.*, 90 °C) enabling the printing of a wide range of biomaterials, either cell-free or cell-laden.¹⁶⁷ In addition, robocasting allows the deposition of more than one ink within the 3D scaffold. Another extrusion-based method, referred to as fused deposition modeling (FDM), is limited in its application to synthetic-based HAP inks as it requires the material to be printed at high temperatures (140 – 250 °C) in a melted state.¹⁷⁰ Consequently, it is not suitable for printing bioinks containing living cells.

Finally, the indirect-based method consists in using the above-mentioned methods to produce personalized molds in which through mold-cast HAP-based materials can be treated to produce 3D scaffolds.¹⁶⁶ The advantage of this method relies on the non-dependence of viscosity to produce the scaffold. However, mold-cast processes that usually involve methods like freeze-drying or critical point drying, require a subsequent post-printing process that uses strong chemicals to remove or dissolve the mold. This additional step does not allow the use of cell-laden HAP-based bioinks, limiting the use of mold-cast processes in tissue engineering applications. Furthermore, despite being a fast-setting method, it only allows the fabrication of simplistic 3D scaffolds losing the full potential of the AM methods.

4.1.1. HAP incorporation in 3D printed structures. The incorporation of HAP into 3D-printed scaffolds can enhance their biocompatibility and osteoconductivity, allowing for improved bone regeneration and tissue repair. However, the high viscosity imposed by HAP can present challenges during the 3D printing process, particularly in extrusion-based methods such as fused deposition modeling. To overcome these challenges, various strategies have been employed, including the

blend with both synthetic and natural-based materials and the utilization of HAP granulate.

The use of synthetic biomaterials has emerged as a promising strategy for incorporating HAP into 3D printed scaffolds for medical applications.¹⁷¹ In the context of HAP-based 3D printed scaffolds, synthetic biomaterials can be utilized to enhance the biocompatibility and osteoconductivity of the scaffold, as well as to address the processing challenges posed by the high viscosity of HAP.^{23,172} Furthermore, the composition and properties of synthetic biomaterials can be precisely controlled, allowing for fine-tuning of the mechanical and biological properties of the scaffold.

The incorporation of HAP into 3D printed scaffolds can also be achieved using natural biomaterials. These materials are commonly derived from vegetal or animal sources and are attractive for their biocompatibility and biodegradability.¹⁷³ When combined with HAP, natural biomaterials can provide the necessary mechanical properties for tissue regeneration and repair, while also promoting cell adhesion and proliferation. Additionally, the use of natural biomaterials can mitigate the challenges posed by the high viscosity of HAP, providing an alternative to the use of synthetic biomaterials while providing biological cues and a better biological environment for tissue engineering applications. However, it is important to note that the composition and properties of natural biomaterials can vary greatly, depending on the source and processing methods used.

The HAP production method can also influence the quality and suitability of HAP materials for 3D printing scaffolds. The most common methods used to produce HAP are: (1) sol-gel synthesis; (2) precipitation; and (3) mechanical milling. Each method has its unique characteristics and suitability for scaffold fabrication. Sol-gel synthesis transforms precursor solutions or gels into solid HAP through chemical reactions, yielding HAP powder known for its high purity and homogeneity.⁶³ However, this method is highly complex, time-consuming and the chemicals and equipment required for sol-gel synthesis can be expensive. Mechanical milling methods produce HAP powders with fine particle sizes and controlled properties.^{171,174} This approach stands out because of its simplicity and cost-effectiveness. It offers good reproducibility, displaying high yields with easy management of reaction conditions. Moreover, it can be easily integrated into large-scale industrial processes.^{37,175}

Despite being a method that allows control of the final particle size by tuning the process parameters, this can be a challenge and there is a high risk of particle agglomeration. Thus, continuous precipitation systems with improved liquid-solid transportation and enhanced mass-heat transfer^{65,176} have been increasingly applied in both research and pharmaceutical companies.^{20,177}

4.1.2. HAP combined with synthetic-based polymers. Polylactic acid (PLA) and HAP is a common combination used to produce 3D printed scaffolds for tissue engineering applications. PLA is a biodegradable and biocompatible polymer that can be processed into 3D printing filaments and used to



fabricate scaffolds with controlled pore size, porosity, and mechanical properties.¹⁷⁸ The combination of PLA and HAp in 3D printed scaffolds offers a number of advantages, including improved mechanical strength, biocompatibility, and osteoconductivity. Russias *et al.*,¹⁷⁹ developed PLA-based constructs reinforced with HAp powder using an extrusion-based 3D printing approach. The 3D structures were successfully printed using ink composed of 70% HAp. The scaffolds presented suitable porosity (~55%) and acceptable mechanical performance with values of elastic modulus ranging from 84 to 150 MPa. The mechanical properties are still far from the minimal requirements for bone applications and in this study no biological tests were performed. The incorporation of natural-based biomaterials in combination with PLA and HAp to produce 3D printed structures has shown promising biological outcomes in terms of cell attachment and proliferation. Additionally, 3D-printed PLA/HAp scaffolds that combine the benefits of antimicrobial and bone regeneration properties have been developed and tested *in vivo* (Fig. 5(A)).¹⁸⁰ The results showed that the 3D-printed scaffold effectively inhibited bacterial growth and promoted bone regeneration in infected bone defect models. Additionally, the scaffold provided a suitable environment for bone cell proliferation and differentiation, leading to improved bone regeneration.

Poly(L-lactic) acid (PLLA) has also been used to develop printed HAp structures. Shuai and co-authors blended PGA into a HAp/PLLA scaffold by laser 3D printing to accelerate degradation, owing to the degradation of high hydrophilic PGA and the subsequent accelerated hydrolysis of PLLA chains. The developed 3D construct exhibited good osteoblast adhesion, spreading and proliferation of MG-63 human osteoblast-like cells. *In vivo*, when used in bone defects in New Zealand white rabbits, the scaffolds resulted in bone defect repair and blood vessel tissue formation after 8 weeks of implantation.¹⁸¹ In another study, the authors blended PLLA with Hap-graphene oxide (GO), in which the oxygen-containing groups in GO served as anchor sites for the chelation of Ca^{2+} and then Ca^{2+} absorbed HPO_4^{2-} *via* electrovalent bonding to form homogeneously dispersed HAp nanorods.¹⁸² The resulting scaffolds exhibited good bioactivity by inducing the formation of an apatite layer and increased viability of MG-63 after 5 days of cell culture. Moreover, the incorporation of HAp-GO improved the compressive properties of the PLLA scaffold.

Another common synthetic polymer used in combination with HAp is polycaprolactone (PCL), a biodegradable and biocompatible polymer that has been widely used in the fabrication of 3D printed scaffolds due to its good mechanical properties and ability to degrade into non-toxic by-products.¹⁸³ Therefore, PCL has been combined with HAp to develop reinforced scaffolds employing different AM techniques. Recently, Petretta *et al.*,¹⁸⁴ developed and characterized 3D-printed magnetic scaffolds using a composite material made of PCL and HAp. The results showed that the addition of HAp to PCL improved the scaffold's mechanical properties, making it more suitable for use as a bone scaffold. The scaffolds exhibited good biodegradability and provided a supportive environment

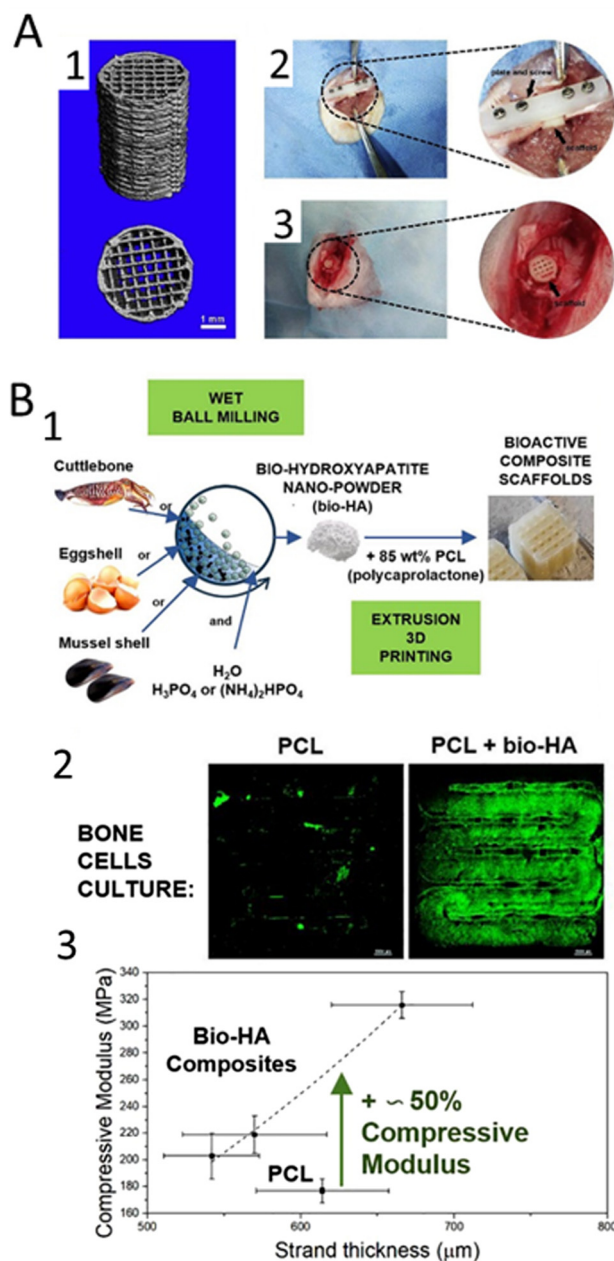


Fig. 5 Synthetic-based biomaterials incorporated with HAp used for 3D printing of scaffolds. (A) PLA/HAp 3D printed scaffold;¹⁸⁰ (A1) micro-CT reconstructed images of 3D-printed PLA/HAp scaffolds; (A2) femoral shaft critical segmental bone defect and (A3) femoral condyle local bone defect implanted with PLA/HAp scaffolds. (B) 3D-printed magnetic scaffolds using a composite material made of PCL and HAp.¹⁸⁵ (B1) Confocal images of the cells seeded on the PCL and PCL/HAp 3D printed scaffolds; (B2) compressive modulus vs. strand thickness curve of the PCL and PCL/HAp 3D printed scaffolds.

for the growth and proliferation of osteoblastic cells. Additionally, the addition of magnetic particles improved the MRI visibility of the scaffold, making it more suitable for tracking it *in vivo*. In another interesting work, the authors used HAp produced from biogenic sources¹⁸⁵ (Fig. 5(B1)). Again, the addition of HAp improved the biological and mechanical properties of the



scaffolds, making them more suitable for use as a bone scaffold (Fig. 5(B2)).

Another synthetic polymer that can be combined with HAP for the production of 3D printed scaffolds is poly(methyl methacrylate) (PMMA). PMMA is a synthetic polymer that is commonly used in a variety of medical and dental applications due to its transparency, toughness, and biocompatibility products.¹⁸⁶ PMMA-based polymers represent a group of resins commonly applied in clinics and are used for reconstructive surgery applications such as *in situ* formed bone cement or pre-surgically bone implant in the craniofacial area.^{187,188} When used in 3D printing, PMMA can be processed into a fine powder or liquid and then fused into solid object layer-by-layer products. However, PMMA is not biodegradable, meaning that it does not degrade over time and may require removal once the tissue engineering process is complete. A conventional strategy to improve the overall biological activity and bone ingrowth properties, that allow the stabilization of the structure *in vivo* is the incorporation of HAP, encouraging better interactions with living tissues. A recent study showed that the incorporation of PCL surface-modified HAP into PMMA increased cell proliferation and mineralization as well as enhanced bone ingrowth properties of PMMA, as demonstrated by increased bone density and bone-to-implant contact¹⁸⁷ (Fig. 6). This study revealed that the modification of HAP particles with PCL is a promising alternative for bone regeneration. This combination has also been applied in 3D printing approaches and showed promising results.¹⁸⁸

4.1.3. HAP combined with natural polymers. In contrast to synthetic polymers, natural-polymers are largely represented by hydrogel-based materials that are primarily processed through extrusion-based AM techniques. This results from the fact that

they are typically either thermosetting or elastomeric in nature, therefore, not suitable for thermal processing. When dissolved and combined with HAP, they present a fluidic nature, being typically printed in a liquid-viscous state. After printing, a post-processing step is often required to enhance the long-term stability of the printed structures under physiological conditions. These procedures are highly dependent on the printed material and significant efforts have been made to identify the most suitable cross-linking strategy for each candidate material.

Furthermore, strong and stable interfacial bonding is essential to ensure that the composite material retains its mechanical integrity, biocompatibility, and functionality and is critical for the development of bioinks.¹⁹⁰ This interfacial bonding can be achieved by chemical bonding, physical interactions, cross-linking agents, or surface modifications, where the choice of the natural polymer, the method of processing, and the specific application will influence the type and strength of the interfacial bonding.^{191,192}

The growing interest in printing HAP-reinforced natural-based biomaterials stems from the potential to simultaneously incorporate cells and/or bioactive molecules within the hydrogel matrix, in addition to HAP reinforcement, to promote cell proliferation/differentiation or elicit a desired *in vivo* response.¹⁹³

Alginate is a natural polymer derived from brown seaweed and is commonly used in medical applications due to its biocompatibility and biodegradability. It has excellent gelation properties and can be easily processed into a gel-like substance, making it a suitable material for 3D printing.¹⁹⁴ The combination of alginate and HAP in 3D printed scaffolds offers several advantages, including improved biocompatibility, biodegradability, and osteoconductivity.¹⁹⁵

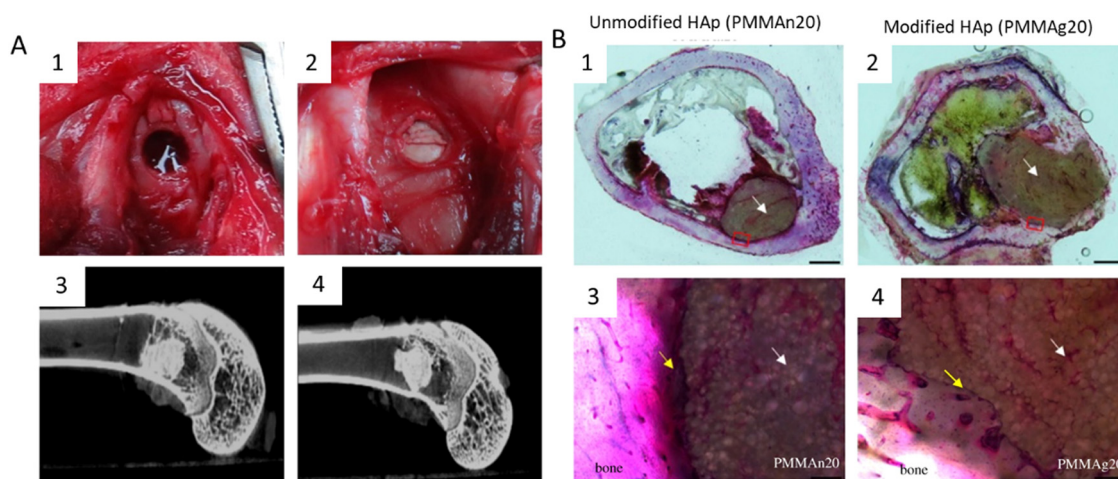


Fig. 6 HAP was modified with PCL and further incorporated into PMMA to improve the functionality of PMMA cement in bone regeneration.¹⁸⁹ (A) PMMA/HAP composites implanted into bone tissue. (A1) Appearance of the femoral condyle bone defect; (A2) PMMA/HAP composite implanted in the bone defect; (A3) micro CT image of the defect site after three months of being implanted with unmodified HAP; (A4) micro CT image of the defect site after three months of being implanted with modified HAP. (B) Histological images of bone tissue with the PMMA/HAP composite implanted for three months. (B1) and (B2) Histological images of bone after 3 months of PMMA incorporation with unmodified HAP and PCL modified HAP, respectively (scale bar 1 mm); (B3) and (B4) magnified images of the red squares on both composites, showing that the unmodified HAP composite was separated from the host bone by a layer of fibrous tissue (yellow arrow); whereas, HAP modified with PCL had a tight connection to the surrounding bone tissue, which revealed better bone ingrowth (scale bar 20 μ m).



Alginate has also been used as a bioink. Improved osteogenesis of stem cells encapsulated in HAP-reinforced alginate hydrogels has been proved due to alteration in DNA methylation with the consequent modification of gene expression of an osteogenic phenotype.¹⁹⁶ For instance, Wang *et al.*,¹⁹⁷ conducted a study that showed that human adipose-derived stem cells (hASCs) exhibited elevated expression of genes related to osteogenesis (OCN and RUNX2) in nano-HAP-loaded alginate hydrogels when compared to plain alginate hydrogels.¹⁹⁷

Collagen is a natural protein that is also used with HAP to produce 3D printed scaffolds. This protein is widely found in the body, including in bones, skin, and connective tissues. It is biocompatible and has a similar composition to the extracellular matrix (ECM) of tissues, making it an attractive material for tissue engineering.¹⁹⁸ The main challenge in the fabrication of collagen/HAP scaffolds is the high viscosity of the composite. This can partially be addressed by using chemical solvents, but the use of solvents also makes it challenging to incorporate bioactive molecules.¹⁹⁸ As a result, the most widely pursued approach for producing collagen/HAP structures is indirect-based AM methods.¹⁹⁹ Lin *et al.*,²⁰⁰ explored a novel approach to fabricate 3D constructs through the utilization of robocasting. The authors used a mixture of HAP and collagen to mimic the natural composition of bones and fabricate the scaffolds using a filament-free printing technique at a temperature of 4 °C. The resulting structures have a diameter range of 300–900 μm and are cross-linked using a genipin solution for stability. The results showed that the scaffolds produced using this technique show potential for use in bone regeneration applications due to their biomimetic composition and controlled structure.

Gelatin is a protein derived from collagen that has a gelling capacity and is often used in 3D printing. It is biocompatible and has been used in various biomedical applications, including as a scaffold material for 3D printing. This protein is water-soluble and presents several advantages mostly relying in the exposure of cell-ligand motifs (*i.e.*, RGD) that endorse cell adhesion to gelatin and target sequences for metalloproteinases (mitochondrial processing peptidase) that leads to *in vivo* enzymatic degradation and ECM remodeling.²⁰¹

In the context of extrusion-based printing strategies, the control of temperature is generally required. Literature has described the printing of gelatin/HAP hydrogels by using a printing cartridge that is kept at a temperature greater than the sol-gel transition temperature and a printing plate with a controlled temperature that is lower than the sol-gel transition temperature. This is due to the fact that the thermoresponsive properties of gelatin hydrogels have been shown to not be affected by the addition of HAP.²⁰² Additionally, the addition of HAP to the ink has been demonstrated to enhance the rheological response as well as improving the structural properties of the printed scaffolds. The addition of HAP derivatives to the gelatin scaffold, as observed in other soft biocomposites such as alginate, has been shown to promote cell osteogenesis through increased expression of osteogenic genes and inorganic matrix deposition.²⁰³ In the present study, gelatin

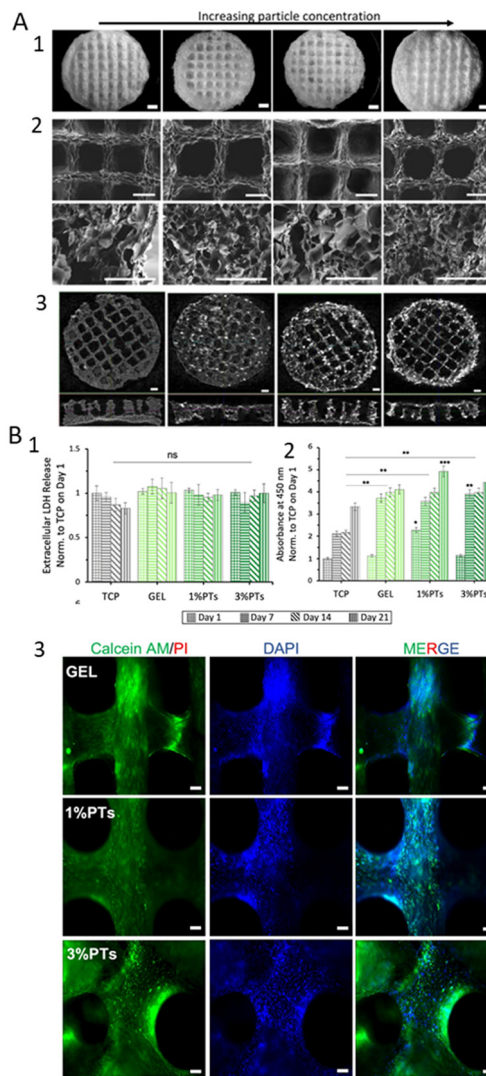


Fig. 7 Gelatin/dbPTs 3D printed scaffolds.²⁰³ (A) Morphological analysis of the Gelatin/dbPTs 3D printed scaffolds. (A1) Light microscopy images of the scaffolds after freeze-drying (scale bars: 1 mm); (A2) SEM images of the scaffolds showing the surface morphology of the scaffolds on the top view and pore structure in the cross-sectional area (scale bars: 500 μm); (A3) micro CT cross-section image of the scaffolds (scale bar: 1 mm). (B) *In vitro* cytocompatibility assessment of the Gelatin/dbPTs 3D printed scaffolds. (B1) Quantification of the extracellular lactate dehydrogenase release assay (LDH) during 21 days of incubation; (B2) Cell viability assay of MC3T3-E1 cells on the Gelatin/dbPTs scaffolds ($n = 6$), statistically significant differences were determined between groups depending on each time point. All data are represented as mean \pm SD. $**p < 0.01$, $***p < 0.001$ indicate a statistically significant difference among means; (B3) fluorescence microscopy images of MC3T3 cells on the scaffolds cultured for 14 days. Calcein AM (green), PI (red), DAPI (blue) staining represent live, dead cells and cell nuclei respectively (scale bars: 100 μm).

was combined with decellularized bone particles (dbPTs, ~100 μm diameter) to produce 3D printed scaffolds. These scaffolds were further subjected to freeze-drying, and the obtained results revealed a decrease in pore size with an increase in HAP concentration (Fig. 7(A)). Additionally, the incorporation of dbPTs was revealed to slightly enhance the



scaffolds biological properties in terms of LDH levels release and cell viability (Fig. 7(B)).

5. Conclusions and future perspectives

HAp is generated on a large scale annually to produce medical devices, and is widely used in bone and dental implants, as drug delivery systems, and tissue engineering. The industrialized batch production of this inorganic material is well-established, but there are still several drawbacks in conventional implemented methodologies to produce HAp and HAp composite particles, as well as complex 3D HAp-based structures.

Despite wet chemical precipitation being a method that allows the control of the final particle size by tuning the process parameters, batch operations can be a challenge and there is a high risk of particle agglomeration and wide distribution of properties. Thus, continuous precipitation systems with improved liquid–solid transportation and enhanced mass and heat transfer^{65,176} have been increasingly applied by both research and pharmaceutical companies.^{20,177}

As demonstrated in this review, new methodologies are being developed to improve reproducibility and scalability, such as continuous processes in microreactors and meso-oscillatory flow devices. These processes allow for efficient utilization of reagents, precise control over operating parameters, and consistent product quality with a smaller facility footprint, while being more energy-efficient due to reduced empty-fill and heat-up cycles.

In addition to guaranteeing the production of high-quality HAp, researchers are also focusing on enhancing the biological response of the resulting materials. Ion-doping, the combination with bioactive factors and the development of composite systems can enhance HAp's biological activity. As discussed in the present review these strategies can address some of the limitations of HAp in tissue engineering applications, by improving its mechanical properties, stability, cell response and antimicrobial activity. Several ions have been shown to enhance osteogenic activity, such as strontium and zinc. On the other hand, ions such as fluoride, silver, and copper have been used for their antimicrobial properties. From a bone tissue engineering viewpoint, incorporating bioactive factors in HAp is a common approach. Growth factors like BMP-2 play a crucial role in promoting osteogenesis and chondrogenesis. The simplest method involves the adsorption of BMP-2 onto HAp scaffolds or particles. Another approach is combining HAp with growth factors using polymeric matrices. Dual-releasing systems with multiple growth factors, such as BMP-2/VEGF combinations, have shown promising results in promoting bone formation and angiogenesis. Cell incorporation, particularly mesenchymal stem cells (MSCs), in HAp-based scaffolds can enhance bone formation. Combining growth factors and cell incorporation can lead to improved bone tissue engineering outcomes leading to next generation biotherapies.

In the context of ion-doping and incorporation of bioactive factor, MOFPR emerges as an optimal system for the continuous production of doped HAp particles, particularly within the scope of biomaterial synthesis for tissue engineering purposes. The micromixing efficiency, diminished shear stress, and improved suspension of solids of this system ensure the synthesis of homogeneous particles and can enhance the uniform integration of dopant ions throughout the HAp lattice.^{14,34} Moreover, this reactor offers the possibility of controlling the final physicochemical properties (such as crystallinity, mean particle size, particle size distribution and morphology) using the oscillation frequency and amplitude.^{14,70} In addition, the doping ions may form impurity phases during the growth of HAp nanocrystals.¹⁵² According to published data, the use of meso-OFRR for HAp precipitation can reduce the reaction time by four times while avoiding the formation of intermediate CaPs phases.^{17,19}

Thus, changing the oscillatory flow conditions in oscillatory flow-based reactors can open new avenues in the development of tailored HAp-growth factor particles by allowing better control over nucleation and crystal growth rates.²⁰⁴ Given the necessity of balancing optimization with the preservation of HAp structure, the thorough control provided by this type of reactor can be particularly advantageous. This technology can play a role in bridging the gap between academic exploration and large-scale industrial production, presenting an adaptable foundation for the consistent fabrication of composite HAp particles on a large scale.

The produced HAp and HAp composites can be further used to develop complex 3D structures. AM-based technology, particularly 3D printing and bioprinting are currently the most cutting-edge approaches. These processes have shown high potential in tissue engineering by allowing for customized patient-specific structures and grafts, with successful outcomes in regenerating hard tissues using tissue-specific printed constructs.

The choice of HAp production method should rely on various considerations, including the intended application, desired particle characteristics, and compatibility with additive manufacturing technologies. In these technologies, methods that generate HAp powders with fine particles, high purity, and suitable rheological properties are preferred. Firstly, HAp particle size and size distribution are essential for even distribution during 3D printing processes like stereolithography (SLA) and selective laser sintering (SLS).^{205,206} Inconsistent particle sizes or aggregates produced by certain methods can lead to printing difficulties and result in uneven scaffold structures. Secondly, the rheological properties of HAp suspensions or pastes, such as viscosity and shear-thinning behavior, influence extrusion-based methods like fused deposition modeling and robocasting.²⁰⁷ These properties are sensitive to the production method used, necessitating tailored HAp formulations for specific 3D printing techniques. In this context, continuous synthesis approaches, in particular oscillatory flow technology, can reduce the particle size distribution and particle aggregation degree, improving process reproducibility.^{17,28}



Moreover, the control over the operation parameters (such as frequency (f) and amplitude (x_0) oscillation, residence time (τ)), allows optimization of the final physicochemical and rheological properties. Thus, this methodology is particularly well-suited for producing HAP powders for 3D printing.^{32,208,209} Furthermore, binder materials, employed in methods like inkjet printing, interact with HAP and affect scaffold mechanical strength and porosity.²¹⁰

Regarding the effect of HAP concentration in printed biomaterials, higher HAP concentration often enhances the mechanical strength and structural integrity of printed constructs, making the printed structures more robust and capable of mimicking natural tissues.^{211,212} However, high concentrations typically increase the viscosity of the HAP suspension²⁰⁷ and can also negatively impact the biological activity due to changes in pH, osmolarity, and ion concentrations.²¹² While HAP is often considered biocompatible, high calcium concentrations might trigger unwanted immune responses or affect interactions between cells and the substrate.¹⁷⁵ Moreover, high concentrations can lead to nozzle clogging or other printing issues, affecting print resolution and accuracy.²¹³

In summary, it all comes down to obtaining HAP particles with specific properties (*i.e.* applying a synthesis methodology and a reactor that allow for greater particle control, reproducibility, and scale-up) so that these can be used in 3D printing, and according to the application combined with other materials, and optimized to match the desired properties. However, there is still a long way to go, since HAP particles and composites utilized in 3D printing applications are acquired through commercial sources rather than in-house production. Current works fail to address these issues and optimize HAP particles in the development of new biomaterials.

Appropriate optimization of inks/bioink design and development is crucial for native-like tissue regeneration, including implant integration, remodeling, vascularization, and maturation. Understanding biomaterials, cells, printing methods, and the *in vivo* environment is necessary. Printing more complex architectures with multiple materials and cell types for *in vivo* cellular diversity and functionality is also an opportunity to better meet the native tissue but it brings new technical barriers that will take time to overcome.

Finally and due to the increasing demand for HAP in biomedical applications, it is crucial to explore in the future the natural and renewable resources for this bioceramic. Thus, extracting HAP from a natural source could alleviate the problem of producing synthetic HAP. Animal bones and scales, for instances, can be a good source for producing HAP, ensuring sustainability, allowing for nutrient recovery of waste materials transforming them into added value materials.

Studying and comprehending the properties of HAP, such as phase purity and thermal stability, as well as enhancing extraction processes, are crucial steps in guaranteeing the ultimate quality of the obtained HAP. These measures are critical for the successful introduction of bio-HAP to the market as a dependable bioceramic suitable for medical applications.

Author contributions

A. V: conceptualization, writing – original draft; S. M.: writing – original draft; J. B. C: conceptualization, writing – original draft, review & editing; F. C: writing – review & editing, supervision, validation; F. R: writing – review & editing, supervision, validation; A. L. O: conceptualization, supervision, writing – review & editing, validation.

Conflicts of interest

There are no conflicts to declare.

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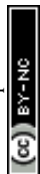
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