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Hollow Co₃O₄ dodecahedrons with controlled crystal orientation and oxygen vacancies for the high performance oxygen evolution reaction†

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Rational design and preparation of advanced electrocatalysts for the oxygen evolution reaction (OER) are highly desirable, where various strategies have been applied to enhance the electrocatalytic activities, such as developing nanostructures, facet control, and creating oxygen vacancies. Herein, we have successfully integrated multiple strategies into one catalytic system for further enhancing the OER performance. A series of hollow Co₃O₄ dodecahedrons with controlled crystal orientation and oxygen vacancies were prepared by using ZIF-67 as a precursor and adjusting the atmosphere during calcination. Hollow Co₃O₄ dodecahedrons with both controlled exposure of the (111) facets and high content of oxygen vacancies showed an excellent OER performance with an overpotential of 307 mV at 10 mA cm $^{-2}$ and a Tafel slope as low as 55 mV dec $^{-1}$ and significantly superior to its counterpart with low content of oxygen vacancies or a broken Co₃O₄ dodecahedron being ground and losing the preferred facets of (111). The excellent OER performance should be attributed to the unique hollow structure and the effective control of the (111) facets and oxygen vacancies, which allows for more highly active sites and enhanced conductivity.

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1. Introduction

With the continuous excessive consumption of fossil energy, the related energy and environmental issues are increasingly prominent.¹⁻³ Electrochemical energy storage, such as electrical catalyzed water splitting,4 nitrogen fixation,5 and carbon dioxide reduction,6 are promising strategies to convert redundant electricity into chemical energy. 7-10 The oxygen evolution reaction (OER) is a half reaction of overall water splitting, which plays a critical role in various next-generation energy storage techniques including the use of fuel cells and metal-air batteries. 11-14 The OER is a four electron-proton coupled reaction but far from practical application due to the lack of proper catalysts to promote the sluggish kinetics. 15,16 Iridium and ruthenium oxides are proved to have excellent OER catalytic performance. However, the vulnerable stability and high cost prevent them from wide application in practical $production.^{17-21}\\$

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Cobalt-based compounds, such as oxides and hydroxides, are regarded as active non-precious metal-based OER catalysts with relatively low cost and easy availability. 22-26 Among them, Co₃O₄ has been frequently investigated due to its considerable OER activity and corrosion resistance in alkaline solution, ^{27–30} although its bulk material is less active for the OER because of the limited surface area and conductivity. Therefore, various strategies have been applied to enhance the OER activity of Co₃O₄, such as developing nanostructured materials for increasing surface area, facet control for generating more highly active sites, 31-33 creating oxygen vacancies for regulating electronic states and improving conductivity.30 For instance, Gupta et al.25 reported a flower-shaped Co₃O₄ nanostructure showing a low overpotential of 356 mV at 10 mA cm⁻² with a Tafel slope of 68 mV dec⁻¹. Liu et al.³⁴ synthesized four types of Co₃O₄ nanocrystals by predominantly exposing (111), (112), (110) and (001) facets, respectively, and illustrated the highest OER activity of the Co₃O₄ nanocrystals with exposed (111) facets due to the highest surface energy, largest dangling bond density and smallest free energy. Xu et al.35 presented a plasmaengraving strategy to create more oxygen vacancies on Co₃O₄ nanosheets and thus resulted in a higher atomic ratio of Co²⁺/ $Co^{3+}(1.2)$ than that (1.0) of those without plasma-engraving, which substantially improves the electronic conductivity and generates more active defects for the OER. Nevertheless, these strategies are still expected to be combined into one catalytic

[†] Electronic supplementary information (ESI) available: XRD patterns, nitrogen adsorption-desorption isotherms, EDS spectra, XPS spectra and electrochemical measurement results. See DOI: 10.1039/d0am00671h

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system for further enhancing the OER performance, which remains a great challenge.

Research Article

Hollow nanostructures possess some unique advantages for practical application in wide fields, ^{28,36–50} such as high specific surface area, abundant active sites, and multiple masstransferring channels. Among all the methods for fabricating hollow structured materials, the self-templating methods based on metal-organic frameworks (MOFs) are receiving more and more attention. 51-57 For example, Yao and coworkers 58 prepared a hollow Co₃O₄/C OER electrocatalyst through a simple two-step thermal cracking process by using the cobalt ion enriched MOF as a precursor, which exhibited a low overpotential of 349 mV at 10 mA cm⁻² and a low Tafel slope of 60 mV dec⁻¹, showing a better performance of hollow structures compared to its solid counterpart. Recently, Wang and coworkers⁵⁹ first reported a hollow multi-shelled Co₃O₄ dodecahedron templated from zeolitic imidazolate framework-67 (ZIF-67), which showed a dominant exposure of (111) facets and 3 times higher photocatalytic activity toward the reduction of CO₂ than that of its counterpart without facet control. Additionally, our previous work showed that such a hollow Co₃O₄ dodecahedron with dominant exposure of (111) facets could be also used as a promising electrocatalyst for glucose oxidation.²⁸ However, its direct use for the OER is rarely reported, which should deserve more attention.

Herein, we have prepared a series of hollow Co₃O₄ dodecahedrons with controlled crystal orientation and oxygen vacancies using ZIF-67 as a precursor and attempted to combine multiple strategies into one catalytic system for further enhancing the OER performance, as shown in Scheme 1. By finely adjusting the atmosphere (the volume ratio of argon to oxygen, V_{Argon}/V_{Oxygen}) during calcination, hollow Co₃O₄ dodecahedrons with controlled exposure of the (111) facets and high content of oxygen vacancies could be obtained, which substantively



Scheme 1 Schematic of the preparation of hollow Co₃O₄ dodecahedrons and oxygen content controlled dominant exposure of the (111) facet and oxygen vacancies for the high performance OER.

resulted in both an increased electrochemical active surface area (ECSA) and enhanced conductivity. As a result, a better OER performance with an overpotential of 307 mV at 10 mA cm⁻² and a Tafel slope as low as 55 mV dec⁻¹ were realized by using hollow Co₃O₄ dodecahedrons obtained at a relatively low oxygen partial pressure $(V_{Argon}/V_{Oxygen} = 9:1)$, which is significantly superior to those for its counterpart with low content of oxygen vacancies obtained at a relatively high oxygen partial pressure $(V_{Argon}/V_{Oxygen} = 7:2)$ or broken Co_3O_4 dodecahedrons being ground and losing the preferred facets of (111).

2. Experimental section

2.1. Materials

Cobalt(II) nitrate hexahydrate (Co(NO₃)₂·6H₂O₇ > 99.0%) and 2-dimethyl imidazole and potassium hydroxide (KOH, AR) were purchased from Aladdin Reagent. Methanol, anhydrous ethanol and other solvents were obtained from Sinopharm chemical reagent Co., Ltd (Shanghai, China). All reagents were used without further purification. High purity argon and oxygen were purchased from Yinchuan Jinfeng Ningfeng Oxygen Co., Ltd (Yinchuan, China). Ultrapure water with a resistivity of 18 M Ω was used in all experiments.

2.2. Synthesis of ZIF-67

In a typical procedure, 22.5 mmol of Co(NO₃)₂·6H₂O and 96.9 mmol of 2-dimethyl imidazole were dissolved in a mixture of 60 mL of methanol and 60 mL of ethanol, respectively. Then the solution of 2-dimethyl imidazole was poured into the solution of Co(NO₃)₂, and the mixed solution was aged for 24 hours at 40 °C. After centrifugation, the purple precipitate was obtained and dried under vacuum for 24 hours.

2.3. Synthesis of hollow Co₃O₄ dodecahedrons

The hollow Co₃O₄ dodecahedrons were obtained by the calcination of ZIF-67 in a tube furnace under different atmospheres. In a typical method, the tube furnace was purged by mixed gases (the V_{Argon}/V_{Oxygen} values were 7:2, 4:1 and 9:1, respectively) for 30 min before calcination. Then the temperature was raised to 425 °C with a heating rate of 0.5 °C and annealed for 1 hour.

Characterizations

The surface morphology of ZIF-67 was observed using a ZEISS EVO 18 scanning electron microscope (SEM) with an acceleration voltage of 15 kV. Transmission electron microscopy (TEM) images of hollow Co₃O₄ dodecahedrons were obtained using a Hitachi HT-7700 electron microscope with an operating voltage of 120 kV. High-resolution TEM images, selected area electron diffraction (SAED), high-angle annular dark field scanning TEM (HAADF-STEM) images and elemental mappings were recorded by an FEI Talos 200S electron microscope operated at 200 kV. The powder X-ray diffraction (XRD) patterns were recorded on an AXS D8 ADVANCE A25 diffractometer with an operating voltage of 40 kV and a scanning speed of 5° min⁻¹. The nitrogen adsorption-desorption isotherms under liquid

nitrogen (-196 °C) were measured on the Autosorb iQ adsorption analyzer. Thermo Scientific ESCALAB 250Xi X-ray photoelectron spectroscopy (XPS) was used to observe the chemical environment of the samples.

2.5. Electrochemical measurements

The electrochemical properties of the samples were determined by a PINE electrochemical workstation (Pine Research Instrumentation, USA) using a conventional three-electrode system. The rotation speed of the rotating-disk electrode (RDE) was set at 1600 rpm, and the disk electrode (RDE 5 mm) deposited with the samples was used as the working electrode, the Ag/AgCl electrode as the reference electrode and the Pt wire electrode as the counter electrode. Before testing, the disk electrode was polished by Al₂O₃ particles with different meshes, and the electrolyte (1 M KOH aqueous solution) was purged by argon for 30 min. The samples (3 mg) were dispersed in 1 mL of anhydrous ethanol under ultrasound treatment for 30 min. Then, 5 µL of the dispersions were dropped on the surface of the disk electrode and dried under ambient temperature conditions. This process was repeated 5 times, so 25 μL of the dispersions were deposited on the surface of the working electrode. Finally, 5 µL of 5 wt% Nafion solution was dropped as a binder on the surface of the electrodes.

The turnover frequency (TOF) calculation:²⁹ the values of TOF were calculated by using the following equation: TOF = $(j \times S)/(4 \times F \times n)$, where j (mA cm⁻²) is the measured current density at an overpotential of 300 mV, S is the working surface area of the glassy carbon electrode, the number 4 means a four-electron OER, F is Faraday's constant (96485.3 C mol⁻¹), and *n* is the moles of the coated Co atom on the electrode calculated from the loading weight and the molecular weight of the coated catalysts (Co₃O₄).

3. Results and discussion

Hollow Co₃O₄ dodecahedrons were synthesized using ZIF-67 as a precursor, which was prepared using a typical procedure.⁵⁹ As shown in Fig. 1a, the SEM image indicates the dodecahedron morphology of ZIF-67 with a uniform size of 2.5 μm. The XRD pattern in Fig. S1 (ESI†) further confirms the crystal structure of ZIF-67. After calcination in a mixture of argon and oxygen, the dodecahedron morphology of ZIF-67 was well-preserved to give samples of Co₃O₄-7:2, Co₃O₄-4:1 and Co₃O₄-9:1 based on the value of V_{Argon}/V_{Oxygen} . The TEM images in Fig. 1b-d demonstrate the hollow structure of Co₃O₄-7:2, Co₃O₄-4:1 and Co₃O₄-9:1 with a similar size of about 1.2 µm. Instead of the smooth surface of ZIF-67, the shell of the Co₃O₄ dodecahedrons was composed of small Co₃O₄ nanoparticles, resulting in a rough surface and high specific areas of about 36 m² g⁻¹, as illustrated in Fig. S2, (ESI†).

In order to investigate the detailed microstructure of the synthesized Co₃O₄ dodecahedrons, HR-TEM was performed to observe the crystal structure and chemical composition of Co₃O₄-9:1. As illustrated in Fig. 2a, the shell of the Co₃O₄-9:1

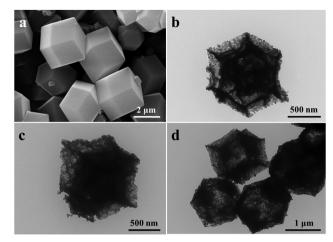


Fig. 1 (a) SEM image of ZIF-67 and (b-d) TEM images of Co₃O₄-7:2, Co₃O₄-4:1 and Co₃O₄-9:1, respectively.

dodecahedron was formed of small Co₃O₄ nanoparticles with various morphologies. The average diameter of the Co3O4 nanoparticles is measured to be 22.3 \pm 5.4 nm. The crystal facet of the spinel Co₃O₄ was determined by calculating the crystal plane spacing based on the HR-TEM image in Fig. 2b. The d-spacings of 0.462, 0.246 and 0.285 nm are ascribed to the (111), (311) and (220) planes, respectively. Besides, the SAED pattern in Fig. 2c also indicates the polycrystalline structure of the Co₃O₄ nanoparticles. The content of all elements of Co₃O₄-9:1 can be determined by EDS (Fig. S3, ESI†), and the atomic fraction of Co and O is 38.4% and 55.0%, in accordance with the stoichiometric ratio of Co₃O₄. Moreover, there is also residual carbon with an atomic fraction of 4.7%, which might increase the conductivity of the samples. The HAADF-STEM image and elemental mappings in Fig. 2d-f reveal the homogeneous distribution of Co and O elements.

It has been reported that the (111) plane dominant Co₃O₄ hollow multi-shell structures could be obtained by the calcination of ZIF-67 due to the topological arrangement of metal atoms in ZIF-67, which exhibited significantly enhanced catalytic activity

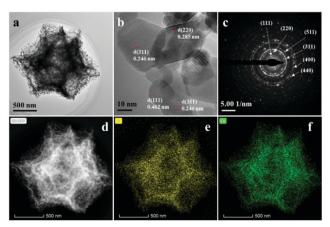


Fig. 2 (a) TEM, (b) HR-TEM image and (c) SAED of Co₃O₄-9:1; (d) HAADF-STEM image and (e and f) elemental mappings of Co₃O₄-9:1.

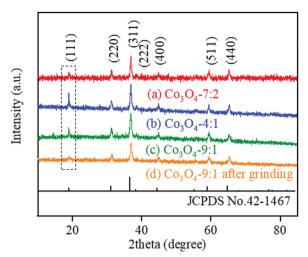


Fig. 3 XRD patterns of (a) $Co_3O_4-7:2$, (b) $Co_3O_4-4:1$, (c) $Co_3O_4-9:1$ and (d) Co₃O₄-9:1 after grinding

for CO₂ photoreduction.⁵⁹ Besides, the (111) plane also showed excellent catalytic performance toward the OER. 34,60 Therefore. we changed the calcination conditions to control the exposure of the (111) plane by changing the oxygen partial pressure during calcination. As shown in Fig. 3, XRD analysis was carried out to probe into the crystal structure of the Co₃O₄-7:2, Co₃O₄-4:1 and Co_3O_4 -9:1. The main diffraction peaks of (111), (220), (311), (222), (400), (511) and (440) planes are observed, which is in good accordance with the standard card and previous report.⁶¹ However, the intensity of the (111) plane of Co₃O₄-4:1 and Co_3O_4 -9:1 is abnormally strong $(I_{(111)}/I_{(311)} = 0.61, 0.42,$ respectively) compared to that of the (311) plane (curve (b and c) in Fig. 3). When the oxygen partial pressure increased to 7:2,

the selective exposure of the (111) facet disappeared, as shown in Fig. 3a. Besides, the orientation of the (111) facet also disappeared after grinding (Fig. 3d), demonstrating the importance of the hollow dodecahedron structure to maintain the selective exposure of the (111) facet. A previous study has also demonstrated that the dominant exposure of the (111) facet of hollow Co₃O₄ dodecahedrons was the result of the orientated growth and alignment of Co₃O₄ nanocrystals during the shell formation induced by the precursor ZIF-67.59 The calcination atmosphere might influence the oriented growth and arrangement of the Co₃O₄ nanocrystals.

To evaluate the difference of the surface components and the chemical environment of the elements of hollow Co₃O₄ dodecahedrons with different dominant exposure of the (111) plane, XPS was carried out. The survey XPS spectra in Fig. S4 (ESI†) confirmed the existence of Co, O and C in the samples, while the similar C 1s high resolution XPS spectra of Co₃O₄-7:2, Co₃O₄-4:1 and Co₃O₄-9:1 could be divided into three peaks, belonging to C=C/C-C (284.8 eV), C-O (286.3 eV) and C=O (288.8 eV), as shown in Fig. S5 (ESI†). Fig. 4a-c show the high resolution XPS spectra of Co 2p of Co₃O₄-7:2, Co₃O₄-4:1 and Co₃O₄-9:1, respectively, which present two prominent peaks at 780.2 and 795.1 eV, assigned to the Co 2p_{3/2} and Co 2p_{1/2} peaks, respectively. The Co 2p_{3/2} peak could be subdivided into two peaks centred at 779.6 and 780.4 eV, which were the characteristic peaks of Co3+ 2p3/2 and Co2+ 2p3/2, respectively. Similarly, the Co 2p_{1/2} peak could also be subdivided into two peaks that belonged to Co^{3+} $2p_{1/2}$ (794.7 eV) and Co^{2+} $2p_{1/2}$ (796.1 eV), respectively. 62-64 Calculating from the peak area of Co3+ and Co2+, the relative content of Co2+ of Co3O4-9:1 was as high as 75.9%, while the content of Co2+ of Co3O4-4:1 and Co₃O₄-7:2 was slightly reduced to 72.1% and 71.9%,

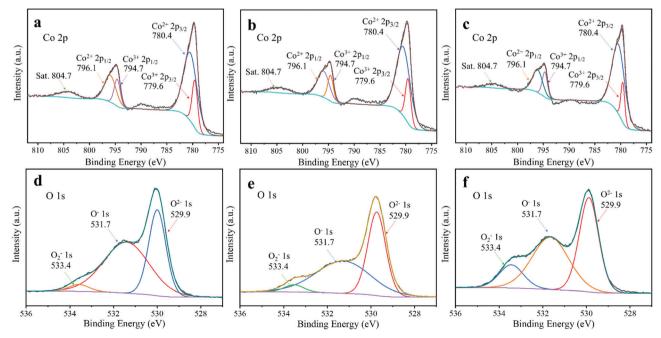


Fig. 4 High resolution (a-c) Co 2p and (d-f) O 1s XPS spectra of (a and d) $Co_3O_4-7:2$, (b and e) $Co_3O_4-4:1$ and (c and f) $Co_3O_4-9:1$.

respectively, which was consistent with the previous study that $\mathrm{Co^{2^+}}$ was the dominant species in the (111) plane. Besides, a high content of $\mathrm{Co^{2^+}}$ facilitates the formation of active sites for the OER.

The high resolution XPS spectra of O 1s of Co₃O₄-7:2, Co₃O₄-4:1 and Co₃O₄-9:1 were also obtained, which clearly displayed three oxygen contributions, as illustrated in Fig. 4d-f. The peak at around 529.9 eV is assigned to the metal oxygen bond, while the other two peaks centred at 531.7 and 533.4 eV belong to the chemosorbed oxygen species. 61 The chemosorbed species centred at 533.4 eV is ascribed to the high binding energy peak from surface oxygen defect species, implying the existence of oxygen vacancies on the surface of the samples.⁶⁶ Based on the peak areas of different oxygen species in Fig. 4f, we calculated that the relative content of oxygen vacancies of Co₃O₄-9:1 was 13%, which was much higher than that of Co₃O₄-4:1 (4.0%) and Co₃O₄-7:2 (4.2%). The high content oxygen vacancy of Co₃O₄-9:1 might be due to the high content of Co²⁺ resulting from the lack of oxygen during calcination.⁶⁰ Because the oxygen vacancy is the active site for the OER, 24,35 electrochemical measurements were conducted to evaluate the catalytic activities of Co₃O₄-7:2, Co₃O₄-4:1 and Co₃O₄-9:1.

The electrochemical measurements for the OER were carried out using a standard three-electrode system in oxygen-saturated 1.0 M KOH aqueous solution with the Ag/AgCl electrode as a reference electrode and the Pt wire electrode as a counter electrode. The sample deposited Pt rotating-disk electrode (RDE) with an area of 0.196 cm 2 was used as a working electrode. Linear sweep voltammetry (LSV) was performed at a scan speed of 0.1 mV s $^{-1}$ and a rotating speed of 1600 rpm to remove the generated oxygen. First of all, the optimal mass

loading was determined by depositing different amounts of Co_3O_4 -9:1 on the surface of the RDE, as shown in Fig. S6 (ESI†). The LSV curves and Tafel plots indicated the optimal mass loading of 0.38 mg cm $^{-2}$ (25 μL), showing a small overpotential of 307 mV and a Tafel slope of 55 mV dec $^{-1}$ (Fig. S6, ESI†). So the following test for Co_3O_4 -7:2, Co_3O_4 -4:1 and Co_3O_4 -9:1 was carried out at a mass loading of 0.38 mg cm $^{-2}$.

Fig. 5a shows the polarization curves of $\text{Co}_3\text{O}_4\text{-}7:2$, $\text{Co}_3\text{O}_4\text{-}4:1$ and $\text{Co}_3\text{O}_4\text{-}9:1$. The peaks at around 1.45 V νs . RHE can be ascribed to the oxidation of Co^{2^+} to Co^{3^+} or Co^{4^+} .⁶⁷ The $\text{Co}_3\text{O}_4\text{-}9:1$ shows a much lower onset potential (1.49 V νs . RHE) than that of $\text{Co}_3\text{O}_4\text{-}4:1$ (1.51 V νs . RHE) and $\text{Co}_3\text{O}_4\text{-}7:2$ (1.57 V νs . RHE). The overpotential of $\text{Co}_3\text{O}_4\text{-}9:1$ is 307 mV to reach a current density of 10 mA cm⁻², which is lower than that of $\text{Co}_3\text{O}_4\text{-}4:1$ (333 mV) and $\text{Co}_3\text{O}_4\text{-}9:1$ (371 mV), as well as most of the previous reports, as shown in Table S1 (ESI†). The OER kinetics of electrocatalysis was further analysed by Tafel plots. As shown in Fig. 5b, the Tafel slopes of $\text{Co}_3\text{O}_4\text{-}9:1$, $\text{Co}_3\text{O}_4\text{-}4:1$ and $\text{Co}_3\text{O}_4\text{-}7:2$ are 55, 55 and 60 mV dec⁻¹, demonstrating the fast OER catalytic reaction kinetics. The high content of oxygen vacancies of $\text{Co}_3\text{O}_4\text{-}9:1$ was believed to contribute to the enhanced catalytic activity.

The resistances of the samples were obtained by electrochemical impedance spectroscopic (EIS) measurement at a potential of 1.548 V νs . RHE, as shown in Fig. 5c. $R_{\rm ct}$ is the charge transfer resistance, which can be estimated from the diameter of the semicircles in the Nyquist plots. The $R_{\rm ct}$ value of ${\rm Co_3O_4}$ -9:1 is the smallest (271 Ω), much lower than that of ${\rm Co_3O_4}$ -4:1 (386 Ω) and ${\rm Co_3O_4}$ -7:2 (584 Ω), respectively, indicating that the OER is easier at the interface of the electrolyte/electrode of ${\rm Co_3O_4}$ -9:1. We believed that the best conductivity

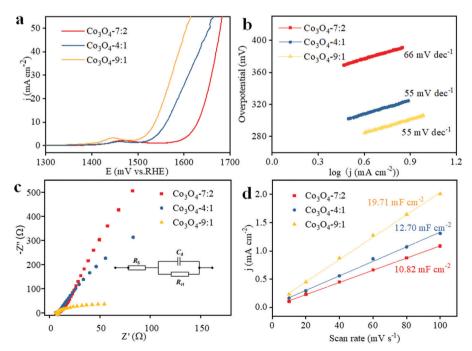


Fig. 5 Electrochemical performance toward the OER. (a) LSV curves, (b) Tafel plots, (c) EIS spectra and (d) capacitive currents as a function of the scan rate to give the double-layer capacitance (C_{dl}) of Co_3O_4 -9:1, Co_3O_4 -4:1 and Co_3O_4 -7:2.

of $\text{Co}_3\text{O}_4\text{-9:1}$ was contributed by the high content of oxygen vacancies, which supplies more charge carriers compared to the counterparts.

ECSA is a key parameter to determine the catalytic activity of an electrocatalyst, which can be roughly evaluated by the double-layer charging current $(C_{\rm dl})$. 68,69 As illustrated in Fig. 5d, the linear plots were obtained by the fitting of the current densities and scan rates, presenting a straight line. The current densities were measured at 1.3 V vs. RHE from the cyclic voltammetry (CV) curves at different scan rates $(10-100 \text{ mV s}^{-1})$ in the region of 1.148-1.348 V vs. RHE (no faradaic current is observed). The $C_{\rm dl}$ value of the samples could be estimated from the slope of the fitted lines, $^{15,70-72}$ which are calculated to be 19.71, 12.70, and 10.82 mF cm⁻² for Co_3O_4 -9:1, Co_3O_4 -4:1 and Co_3O_4 -7:2. According to the calculation equation (ECSA = $C_{\rm dl}/C_{\rm s}$, $C_{\rm s}$ is equal to 40 $\mu \rm F~cm^{-2}$ under alkaline conditions), 68,69 the ECSAs of Co_3O_4 -9:1, Co_3O_4 -4:1 and Co_3O_4 -7: 2 are calculated to be 129.7, 83.7 and 71.3 m² g⁻¹, respectively, considering that the mass loading of the samples is 0.38 mg cm^{-2} . Compared to Co_3O_4 -4:1 and Co_3O_4 -7:2, the ECSA of Co₃O₄-9:1 was significantly improved, indicating the more exposure of the catalytically active site contributed by oxygen vacancies, which was consistent with the results of XPS. The TOFs of Co₃O₄-9:1, Co₃O₄-4:1 and Co₃O₄-7:2 were calculated to be 4.17, 1.56 and 0.56 s⁻¹, respectively, further confirming the excellent OER performance of Co₃O₄-9:1.

In order to further confirm the contribution of the hollow structure and exposure of the (111) facet of hollow Co_3O_4 dodecahedrons to the OER catalytic activity, the electrochemical properties of Co_3O_4 -9:1 were tested after grinding, which destroyed the hollow structure and the dominant exposure of the (111) facet (Fig. 3d). ⁵⁹ As shown in Fig. S7 (ESI†), the overpotential of Co_3O_4 -9:1 increased to 344 mV at 10 mA cm⁻² after grinding. Besides, the Tafel slope also increased to 68 mV dec⁻¹. The EIS and double layer capacitance were also tested, showing an increased charge transfer resistance (851 Ω) and a decreased ECSA of 29.3 m² g⁻¹, much lower than that before grinding (129.7 m² g⁻¹). The results demonstrated that the hollow structure and exposure of the (111) facet of Co_3O_4 -9:1 supplied more active sites for the OER.

The ECSA-normalized LSV curves are presented in Fig. S8 (ESI†), showing similar results as presented in Fig. 5a. The potential of Co_3O_4 -9:1 is 1.544 V at 0.1 mA cm_{ECSA}^{-2} , which is 14 and 47 mV lower than that of Co_3O_4 -4:1 and Co_3O_4 -7:2, respectively. The results implied that the high content of oxygen vacancies might optimize the electron structure of Co₃O₄-9:1, which accelerated the charge transfer and the OER process.⁷³ Overall, the highest catalytic activity of Co₃O₄-9:1 to the OER was determined by several factors including the exposure of the (111) facet, the content of oxygen vacancies, charge transfer resistance and the ECSA. Moreover, Co₃O₄-9:1 also showed reasonable durability, as illustrated in Fig. S9 (ESI†). The overpotential increased 56 mV at a current density of 10 mA cm⁻² after 1000 cycles compared to the initial overpotential. However, the stability is still far from meeting the requirements of real-world applications. We speculate that the "soft" attached structure of the hollow Co₃O₄ dodecahedrons is not stable enough to guarantee the long term catalytic stability.⁵⁹

4. Conclusions

In summary, we have successfully prepared a series of hollow Co₃O₄ dodecahedrons with controlled exposure of the (111) facet and oxygen vacancies by using ZIF-67 as a precursor and adjusting the atmosphere during calcination, and realized the combination of multiple strategies into one catalytic system for further enhancing the OER performance. Hollow Co₃O₄ dodecahedrons with both controlled exposure of the (111) facets and high content of oxygen vacancies were obtained at a relatively low oxygen partial pressure, which showed an excellent OER performance with an overpotential of 307 mV at 10 mA cm⁻² and a Tafel slope as low as 55 mV dec⁻¹ which is significantly superior to its counterpart with low content of oxygen vacancies obtained at a relatively high oxygen partial pressure or a broken Co₃O₄ dodecahedron being grinded and losing the preferred facets of (111). The excellent OER performance should be attributed to the hollow structure and the effective control of the (111) facets and oxygen vacancies, which allow for more highly active sites and enhanced conductivity.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

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References

- 1 S. J. Davis, K. Caldeira and H. D. Matthews, Future CO₂ Emissions And Climate Change From Existing Energy Infrastructure, *Science*, 2010, 329, 1330–1333.
- 2 Z. Liu, D. Guan, W. Wei, S. J. Davis, P. Ciais, J. Bai, S. Peng, Q. Zhang, K. Hubacek, G. Marland, R. J. Andres, D. Crawford-Brown, J. Lin, H. Zhao, C. Hong, T. A. Boden, K. Feng, G. P. Peters, F. Xi, J. Liu, Y. Li, Y. Zhao, N. Zeng and K. He, Reduced Carbon Emission Estimates From Fossil

- Fuel Combustion And Cement Production In China, Nature, 2015, 524, 335-338.
- 3 S. Bilgen, Structure And Environmental Impact Of Global Energy Consumption, Renewable Sustainable Energy Rev., 2014, 38, 890-902.
- 4 Y. Wei, J. Wang, R. Yu, J. Wan and D. Wang, Constructing SrTiO₃-TiO₂ Heterogeneous Hollow Multi-Shelled Structures For Enhanced Solar Water Splitting, Angew. Chem., Int. Ed., 2019, 58, 1422-1426.
- 5 W.-L. Wang, J. K. Moore, A. C. Martiny and F. W. Primeau, Convergent Estimates Of Marine Nitrogen Fixation, Nature, 2019, 566, 205-211.
- 6 C. Dong, C. Lian, S. Hu, Z. Deng, J. Gong, M. Li, H. Liu, M. Xing and J. Zhang, Size-Dependent Activity And Selectivity Of Carbon Dioxide Photocatalytic Reduction Over Platinum Nanoparticles, Nat. Commun., 2018, 9, 1-11.
- 7 J. Y. Wang, Y. Cui and D. Wang, Design Of Hollow Nanostructures For Energy Storage, Conversion And Production, Adv. Mater., 2019, 31, 1801993.
- 8 E. H. M. Salhabi, J. L. Zhao, J. Y. Wang, M. Yang, B. Wang and D. Wang, Hollow Multi-Shelled Structural TiO2-x with Multiple Spatial Confinement For Long-Life Lithium-Sulfur Batteries, Angew. Chem., Int. Ed., 2019, 58, 9078-9082.
- 9 J. Y. Wang, J. W. Wan and D. Wang, Hollow Multishelled Structures For Promising Applications: Understanding The Structure-Performance Correlation, Acc. Chem. Res., 2019, **52**, 2169-2178.
- 10 D. Mao, J. W. Wan, J. Y. Wang and D. Wang, Sequential Templating Approach: A Groundbreaking Strategy To Create Hollow Multishelled Structures, Adv. Mater., 2019, 31, 1802874.
- 11 N. T. Suen, S. F. Hung, Q. Quan, N. Zhang, Y. J. Xu and H. M. Chen, Electrocatalysis For The Oxygen Evolution Reaction: Recent Development And Future Perspectives, Chem. Soc. Rev., 2017, 46, 337-365.
- 12 T. Rodenas, S. Beeg, I. Spanos, S. Neugebauer, F. Girgsdies, G. Algara-Siller, P. P. M. Schleker, P. Jakes, N. Pfaender, M. Willinger, M. Greiner, G. Prieto, R. Schloegl and S. Heumann, 2D Metal Organic Framework-Graphitic Carbon Nanocomposites As Precursors For High-Performance O₂-Evolution Electrocatalysts, Adv. Energy Mater., 2018, 8, 1802404.
- 13 C. Guan, A. Sumboja, H. Wu, W. Ren, X. Liu, H. Zhang, Z. Liu, C. Cheng, S. J. Pennycook and J. Wang, Hollow Co₃O₄ Nanosphere Embedded In Carbon Arrays For Stable And Flexible Solid-State Zinc-Air Batteries, Adv. Mater., 2017, 29, 1704117.
- 14 Z. Xue, Y. Li, Y. Zhang, W. Geng, B. Jia, J. Tang, S. Bao, H.-P. Wang, Y. Fan, Z.-W. Wei, Z. Zhang, Z. Ke, G. Li and C.-Y. Su, Modulating Electronic Structure Of Metal-Organic Framework For Efficient Electrocatalytic Oxygen Evolution, Adv. Energy Mater., 2018, 8, 1801564.
- 15 L. Liu, H. Wu, J. Li and L. She, Tuning Microstructures Of Iron-Nickel Alloy Catalysts For Efficient Oxygen Evolution Reaction, Chem. J. Chin. Univ., 2020, 41, 1083-1090.
- 16 H. Xu, J. Cao, C. Shan, B. Wang, P. Xi, W. Liu and Y. Tang, MOF-Derived Hollow CoS Decorated With CeOx Nanoparticles

- For Boosting Oxygen Evolution Reaction Electrocatalysis, Angew. Chem., Int. Ed., 2018, 57, 8654-8658.
- 17 E. Antolini, Iridium As Catalyst And Cocatalyst For Oxygen Evolution/Reduction In Acidic Polymer Electrolyte Membrane Electrolyzers and Fuel Cells, ACS Catal., 2014, 4, 1426-1440.
- 18 Y. Lee, J. Suntivich, K. J. May, E. E. Perry and Y. Shao-Horn, Synthesis And Activities Of Rutile IrO2 And RuO2 Nanoparticles For Oxygen Evolution In Acid And Alkaline Solutions, J. Phys. Chem. Lett., 2012, 3, 399-404.
- 19 J. Shan, C. Guo, Y. Zhu, S. Chen, L. Song, M. Jaroniec, Y. Zheng and S.-Z. Qiao, Charge-Redistribution-Enhanced Nanocrystalline Ru@IrOx Electrocatalysts For Oxygen Evolution In Acidic Media, Chemistry, 2019, 5, 445-459.
- 20 L. C. Seitz, C. F. Dickens, K. Nishio, Y. Hikita, J. Montoya, A. Doyle, C. Kirk, A. Vojvodic, H. Y. Hwang, J. K. Norskov and T. F. Jaramillo, A Highly Active And Stable IrOx/SrIrO3 Catalyst For The Oxygen Evolution Reaction, Science, 2016, 353, 1011-1014.
- 21 Y. Sun, L. Yuan, Z. Liu, Q. Wang, K. Huang and S. Feng, Optimization Of Oxygen Evolution Dynamics On RuO2 Via Controlling Of Spontaneous Dissociation Equilibrium, Mater. Chem. Front., 2019, 3, 1779-1785.
- 22 J. Mei, T. Liao, G. A. Ayoko, J. Bell and Z. Sun, Cobalt Oxide-Based Nanoarchitectures For Electrochemical Energy Applications, Prog. Mater. Sci., 2019, 103, 596-677.
- 23 L. Chen, X. Zuo, S. Yang, T. Cai and D. Ding, Rational Design And Synthesis Of Hollow Co₃O₄@Fe₂O₃ Core-Shell Nanostructure For The Catalytic Degradation Of Norfloxacin By Coupling With Peroxymonosulfate, Chem. Eng. J., 2019, 359, 373-384.
- 24 Z. Chen, B. Zhao, Y.-C. He, H.-R. Wen, X.-Z. Fu, R. Sun and C.-P. Wong, NiCo2O4 Nanoframes With A Nanosheet Surface As Efficient Electrocatalysts For The Oxygen Evolution Reaction, Mater. Chem. Front., 2018, 2, 1155-1164.
- 25 C. K. Ranaweera, C. Zhang, S. Bhoyate, P. K. Kahol, M. Ghimire, S. R. Mishra, F. Perez, B. K. Gupta and R. K. Gupta, Flower-Shaped Cobalt Oxide Nano-Structures As An Efficient, Flexible And Stable Electrocatalyst For The Oxygen Evolution Reaction, Mater. Chem. Front., 2017, 1, 1580-1584.
- 26 B. Y. Guan, Y. Lu, Y. Wang, M. Wu and X. W. Lou, Porous Iron-Cobalt Alloy/Nitrogen-Doped Carbon Cages Synthesized Via Pyrolysis Of Complex Metal-Organic Framework Hybrids For Oxygen Reduction, Adv. Funct. Mater., 2018, 28, 1706738.
- 27 J. H. Kim, S.-K. Park, Y. J. Oh and Y. C. Kang, Hierarchical Hollow Microspheres Grafted with Co Nanoparticle-Embedded Bamboo-Like N-Doped Carbon Nanotube Bundles As Ultrahigh Rate And Long-Life Cathodes For Rechargeable Lithium-Oxygen Batteries, Chem. Eng. J., 2018, 334, 2500-2510.
- 28 J. Yang, L. Wei, T. Zhao, T. Yang, J. Wang, W. Wu, X. Yang, Z. Li and M. Wu, Hollow Petal-Like Co₃O₄ Nanoflakes As Bifunctional Electrocatalysts through Template-Free Protocol And Structural Controlled Kinetics In Gas Evolution, Electrochim. Acta, 2019, 318, 949-956.

- 29 X. Wang, L. Yu, B. Y. Guan, S. Song and X. W. Lou, Metal-Organic Framework Hybrid-Assisted Formation of Co₃O₄/ Co-Fe Oxide Double-Shelled Nanoboxes For Enhanced Oxygen Evolution, Adv. Mater., 2018, 30, 1801211.
- 30 L. Li, T. Tian, J. Jiang and L. Ai, Hierarchically Porous Co₃O₄ Architectures with Honeycomb-Like Structures For Efficient Oxygen Generation from Electrochemical Water Splitting, J. Power Sources, 2015, 294, 103-111.
- 31 X. Zhou, Z. Liu, Y. Wang and Y. Ding, Facet Effect of Co₃O₄ Nanocrystals on Visible-Light Driven Water Oxidation, Appl. Catal., B, 2018, 237, 74-84.
- 32 Y. Xu, F. Zhang, T. Sheng, T. Ye, D. Yi, Y. Yang, S. Liu, X. Wang and J. Yao, Clarifying the Controversial Catalytic Active Sites of Co₃O₄ For the Oxygen Evolution Reaction, J. Mater. Chem. A, 2019, 7, 23191-23198.
- 33 Z. Chen, C. X. Kronawitter and B. E. Koel, Facet-Dependent Activity And Stability of Co₃O₄ Nanocrystals towards the Oxygen Evolution Reaction, Phys. Chem. Chem. Phys., 2015, 17, 29387-29393.
- 34 L. Liu, Z. Jiang, L. Fang, H. Xu, H. Zhang, X. Gu and Y. Wang, Probing the Crystal Plane Effect of Co₃O₄ For Enhanced Electrocatalytic Performance toward Efficient Overall Water Splitting, ACS Appl. Mater. Interfaces, 2017, 9, 27736-27744.
- 35 L. Xu, Q. Jiang, Z. Xiao, X. Li, J. Huo, S. Wang and L. Dai, Plasma-Engraved Co₃O₄ Nanosheets with Oxygen Vacancies And High Surface Area For the Oxygen Evolution Reaction, Angew. Chem., Int. Ed., 2016, 55, 5277-5281.
- 36 Z. Y. Wang, D. Y. Luan, F. Y. C. Boey and X. W. Lou, Fast Formation of SnO₂ Nanoboxes with Enhanced Lithium Storage Capability, J. Am. Chem. Soc., 2011, 133, 4738-4741.
- 37 X. Y. Lai, J. Li, B. A. Korgel, Z. H. Dong, Z. M. Li, F. B. Su, J. Du and D. Wang, General Synthesis And Gas-Sensing Properties of Multiple-Shell Metal Oxide Hollow Microspheres, Angew. Chem., Int. Ed., 2011, 50, 2738-2741.
- 38 Z. H. Dong, X. Y. Lai, J. E. Halpert, N. L. Yang, L. X. Yi, J. Zhai, D. Wang, Z. Y. Tang and L. Jiang, Accurate Control of Multishelled ZnO Hollow Microspheres For Dye-Sensitized Solar Cells with High Efficiency, Adv. Mater., 2012, 24, 1046-1049.
- 39 J. Y. Wang, N. L. Yang, H. J. Tang, Z. H. Dong, Q. Jin, M. Yang, D. Kisailus, H. J. Zhao, Z. Y. Tang and D. Wang, Accurate Control of Multishelled Co3O4 Hollow Microspheres As High-Performance Anode Materials In Lithium-Ion Batteries, Angew. Chem., Int. Ed., 2013, 52, 6417-6420.
- 40 J. Wang, H. Tang, L. Zhang, H. Ren, R. Yu, Q. Jin, J. Qi, D. Mao, M. Yang, Y. Wang, P. Liu, Y. Zhang, Y. Wen, L. Gu, G. Ma, Z. Su, Z. Tang, H. Zhao and D. Wang, Multi-Shelled Metal Oxides Prepared Via an Anion-Adsorption Mechanism For Lithium-Ion Batteries, Nat. Energy, 2016, 1, 16050.
- 41 M. Tian, L. Wang, J. Wan, X. Lai, X. Wang and D. Wang, Hollow Multi-Shelled Co₃O₄ Dodecahedra For Nonenzymatic Glucose Biosensor, Chin. Sci. Bull., 2019, 64, 3640-3646.
- 42 G. Lin, H. Wang, X. Lai, R. Yang, Y. Zou, J. Wan, D. Liu, H. Jiang and Y. Hu, Co₃O₄/N-Doped RGO Nanocomposites

- Derived from MOFs And Their Highly Enhanced Gas Sensing Performance, Sens. Actuators, B, 2020, 303, 127219.
- 43 J. Wang, J. Wan, N. Yang, Q. Li and D. Wang, Hollow Multishell Structures Exercise Temporal-Spatial Ordering And Dynamic Smart Behaviour, Nat. Rev. Chem., 2020, 4, 159-168.
- 44 H. Sun, Y. Zhu, B. Yang, Y. Wang, Y. Wu and J. Du, Template-Free Fabrication of Nitrogen-Doped Hollow Carbon Spheres For High-Performance Supercapacitors Based on a Scalable Homopolymer Vesicle, J. Mater. Chem. A, 2016, 4, 12088-12097.
- 45 J. Wang, H. Tang, H. Wang, R. Yu and D. Wang, Multi-Shelled Hollow Micro-/Nanostructures: Promising Platforms For Lithium-Ion Batteries, Mater. Chem. Front., 2017, 1,
- 46 B. Wang, L. Xu, G. Liu, P. Zhang, W. Zhu, J. Xia and H. Li, Biomass Willow Catkin-Derived Co₃O₄/N-Doped Hollow Hierarchical Porous Carbon Microtubes As an Effective Tri-Functional Electrocatalyst, J. Mater. Chem. A, 2017, 5, 20170-20179.
- 47 L. Zhang, T. Mi, M. A. Ziaee, L. Liang and R. Wang, Hollow POM@MOF Hybrid-Derived Porous Co₃O₄/CoMoO₄ Nanocages For Enhanced Electrocatalytic Water Oxidation, J. Mater. Chem. A, 2018, 6, 1639-1647.
- 48 L. Yu, X. Y. Yu and X. W. Lou, The Design And Synthesis of Hollow Micro-/Nanostructures: Present And Future Trends, Adv. Mater., 2018, 30, 1800939.
- 49 J. Tang, J. Yao, X. Zhang, L. Ma, T. Zhang, Z. Niu, X. Chen, L. Zhao, L. Jiang and Y. Sun, Multi-Shell Hollow FeP Microspheres As Efficient Electrocatalyst For Hydrogen Evolution at All pH Values, Chem. J. Chin. Univ., 2019, 40, 2340-2347.
- 50 H. Sun and X. Lai, Progress In Preparation And Gas-Sensing Application of Hollow Multi-Shell Structured Materials, Chem. J. Chin. Univ., 2020, 41, 855-871.
- 51 L. Oar-Arteta, T. Wezendonk, X. Sun, F. Kapteijn and J. Gascon, Metal Organic Frameworks As Precursors For the Manufacture of Advanced Catalytic Materials, Mater. Chem. Front., 2017, 1, 1709-1745.
- 52 S. Dang, Q.-L. Zhu and Q. Xu, Nanomaterials Derived from Metal-Organic Frameworks, Nat. Rev. Mater., 2018, 3, 1-14.
- 53 Y.-Z. Chen, R. Zhang, L. Jiao and H.-L. Jiang, Metal-Organic Framework-Derived Porous Materials For Catalysis, Coord. Chem. Rev., 2018, 362, 1-23.
- 54 H.-F. Wang, L. Chen, H. Pang, S. Kaskel and Q. Xu, MOF-Derived Electrocatalysts For Oxygen Reduction, Oxygen Evolution And Hydrogen Evolution Reactions, Chem. Soc. Rev., 2020, 1414-1448.
- 55 P. He, J. Zhou, A. Zhou, Y. Dou and J. Li, MOFs-Based Materials For Photocatalytic CO₂ Reduction, Chem. J. Chin. Univ., 2019, 40, 855-866.
- 56 J. Chang, G. Xu, L. Hui and Q. Fang, Quinone-Based Covalent Organic Frameworks For Efficient Oxygen Evolution Reaction, Chem. J. Chin. Univ., 2020, 41, 1609-1614.
- 57 X. Gao, H. Pan, C. Qiao, F. Chen, Y. Zhou and W. Yang, Construction of HRP Immobilized Enzyme Reactor Based on Hierarchically Porous Meta-Organic Framework And Its

- Dye Degradation Application, Chem. J. Chin. Univ., 2020, 41, 1591-1599.
- 58 Y. Wu, H. Chen, J. Wang, H. Liu, E. Lv, Z. Zhu, J. Gao and J. Yao, Metal-Organic Framework-Templated Hollow Co₃O₄/ C with Controllable Oxygen Vacancies For Efficient Oxygen Evolution Reaction, ChemNanoMat, 2020, 6, 107-112.
- 59 L. Wang, J. Wan, Y. Zhao, N. Yang and D. Wang, Hollow Multi-Shelled Structures of Co₃O₄ Dodecahedron with Unique Crystal Orientation For Enhanced Photocatalytic CO₂ Reduction, J. Am. Chem. Soc., 2019, 141, 2238-2241.
- 60 W. T. Koo, S. Yu, S. J. Choi, J. S. Jang, J. Y. Cheong and I. D. Kim, Nanoscale PdO Catalyst Functionalized Co₃O₄ Hollow Nanocages Using MOF Templates For Selective Detection of Acetone Molecules In Exhaled Breath, ACS Appl. Mater. Interfaces, 2017, 9, 8201-8210.
- 61 C. Zhang, F. Zheng, Z. Zhang, D. Xiang, C. Cheng, Z. Zhuang, P. Li, X. Li and W. Chen, Fabrication of Hollow Pompon-Like Co₃O₄ Nanostructures with Rich Defects And High-Index Facet Exposure For Enhanced Oxygen Evolution Catalysis, J. Mater. Chem. A, 2019, 7, 9059-9067.
- 62 C. Jiao, Z. Wang, X. Zhao, H. Wang, J. Wang, R. Yu and D. Wang, Triple-Shelled Manganese-Cobalt Oxide Hollow Dodecahedra with Highly Enhanced Performance For Rechargeable Alkaline Batteries, Angew. Chem., Int. Ed., 2019, 58, 996-1001.
- 63 C. Wang, J. Wang, W. Hu and D. Wang, Controllable Synthesis of Hollow Multishell Structured Co₃O₄ with Improved Rate Performance And Cyclic Stability For Supercapacitors, Chem. J. Chin. Univ., 2020, 36, 68-73.
- 64 Q. Liang, H. Jin, Z. Wang, Y. Xiong, S. Yuan, X. Zeng, D. He and S. Mu, Metal-Organic Frameworks Derived Reverse-Encapsulation Co-Nc@Mo₂C Complex For Efficient Overall Water Splitting, *Nano Energy*, 2019, **57**, 746–752.
- 65 Q. Liu, Z. Chen, Z. Yan, Y. Wang, E. Wang, S. Wang, S. Wang and G. Sun, Crystal-Plane-Dependent Activity of Spinel Co₃O₄ Towards Water Splitting And the Oxygen Reduction Reaction, ChemElectroChem, 2018, 5, 1080-1086.

- 66 W. Xu, W. Xie and Y. Wang, Co₃O_{4-x}-Carbon@Fe_{2-ν}Co_νO₃ Heterostructural Hollow Polyhedrons For the Oxygen Evolution Reaction, ACS Appl. Mater. Interfaces, 2017, 9, 28642-28649.
- 67 H. Liang, F. Meng, M. Cabán-Acevedo, L. Li, A. Forticaux, L. Xiu, Z. Wang and S. Jin, Hydrothermal Continuous Flow Synthesis And Exfoliation of NiCo Layered Double Hydroxide Nanosheets For Enhanced Oxygen Evolution Catalysis, Nano Lett., 2015, 15, 1421-1427.
- 68 P. Chen, K. Xu, Z. Fang, Y. Tong, J. Wu, X. Lu, X. Peng, H. Ding, C. Wu and Y. Xie, Metallic Co₄N Porous Nanowire Arrays Activated by Surface Oxidation As Electrocatalysts For the Oxygen Evolution Reaction, Angew. Chem., Int. Ed., 2015, 54. 14710-14714.
- C. G. Morales-Guio, L. Liardet and X. Hu, Oxidatively Electrodeposited Thin-Film Transition Metal (Oxy) Hydroxides As Oxygen Evolution Catalysts, J. Am. Chem. Soc., 2016, 138, 8946-8957.
- 70 H. Xu, Z.-X. Shi, Y.-X. Tong and G.-R. Li, Porous Microrod Arrays Constructed by Carbon-Confined NiCo@NiCoO2 Core@Shell Nanoparticles As Efficient Electrocatalysts For Oxygen Evolution, Adv. Mater., 2018, 30, 1705442.
- 71 M. Li, K. Duanmu, C. Wan, T. Cheng, L. Zhang, S. Dai, W. Chen, Z. Zhao, P. Li, H. Fei, Y. Zhu, R. Yu, J. Luo, K. Zang, Z. Lin, M. Ding, J. Huang, H. Sun, J. Guo, X. Pan, W. A. Goddard, P. Sautet, Y. Huang and X. Duan, Single-Atom Tailoring of Platinum Nanocatalysts For High-Performance Multifunctional Electrocatalysis, Nat. Catal., 2019, 2, 495-503.
- 72 J. Zhao, M. Yang, N. Yang, J. Wang and D. Wang, Hollow Micro-/Nanostructure Reviving Lithium-Sulfur Batteries, Chem. J. Chin. Univ., 2020, 36, 313-319.
- 73 H. Wang, J. Qi, N. Yang, W. Cui, J. Wang, Q. Li, Q. Zhang, X. Yu, L. Gu, J. Li, R. Yu, K. Huang, S. Song, S. Feng and D. Wang, Dual-Defects Adjusted Crystal-Field Splitting of $LaCo(1-x)Ni(x)O(3-\delta)Hollow$ Multishelled Structures For Efficient Oxygen Evolution, Angew. Chem., Int. Ed., 2020, **59**, 19691-19695.