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Negative thermal expansion in high pressure layered perovskite Ca_2GeO_4 [†]

Wei-Tin Chen,^{ib ab} Chris Ablitt,^{ib c} Nicholas C. Bristowe,^{ib d} Arash A. Mostofi,^{ib ce}
Takashi Saito,^{fg} Yuichi Shimakawa^{ib fg} and Mark S. Senn^{ib *h}

We report the high pressure synthesis of a layered perovskite Ca_2GeO_4 which is found to have the Ruddlesden–Popper structure with $I4_1/acd$ symmetry. Consonant with our recent predictions [Ablitt *et al.*, *npj Comput. Mater.*, 2017, 3, 44], the phase displays pronounced uniaxial negative thermal expansion over a large temperature range. Negative thermal expansion that persists over a large temperature range is very unusual in the perovskite structure, and its occurrence in this instance can be understood to arise due to both soft lattice vibrations associated with a phase competition and the unusually compliant nature of this structure, which effectively couples thermal expansion in the layer plane to lattice contractions perpendicular to the layering direction via a “corkscrew” mechanism.

Recently there has been renewed interest in layered perovskite Ruddlesden–Popper systems due to the improper ferroelectricity discovered in the $n = 2$ family member.^{1–5} In this structure, octahedral rotation and out-of-phase octahedral tilt modes couple to drive polar distortions leading to a ferroelectric ground state.⁶ A phase that competes thermodynamically with the improper ferroelectric phase is one with a single out-of-phase rotation of the BO_6 octahedra about the layering axis, c . While this phase is not ferroelectric it does exhibit pronounced uniaxial negative thermal expansion (NTE)⁷ that may be tuned systematically through chemical control.⁸ We have recently shown that this thermal expansion arises precisely due to this phase competition and the high compliance of the octahedrally rotated phase. Dynamic, low

frequency tilts of octahedra about in-plane axes provide the thermodynamic driving force for large in-plane thermal expansion and slight uniaxial contraction at low temperatures. However, thanks to a corkscrew mechanism strongly coupling ab plane, the frozen rotation angle, θ , and c^9 (Fig. 1 inset), these modes drive substantial NTE along c over a wide temperature range.¹⁰ By studying the nature of the compliance matrix in the Ruddlesden–Popper series $\text{Ca}_{n+1}\text{Ge}_n\text{O}_{3n+1}$ as a function of n we recently predicted that the magnitude of NTE should be maximised for $n = 1$ ¹¹ in which an analogous phase competition to that observed in the $n = 2$ structure occurs (see Fig. 1). With this in mind, we have synthesised Ca_2GeO_4 using high pressure techniques and show that it has the most pronounced NTE over the largest temperature range recorded to date in these systems.

Olivine Ca_2GeO_4 precursor was prepared with stoichiometric amounts of CaCO_3 (99.9965%, Alfa) and GeO_2 (99.999%, Alfa). The starting materials were mixed well and heated at 1200 °C for 12 hours, and after intermediate grinding, the mixture was

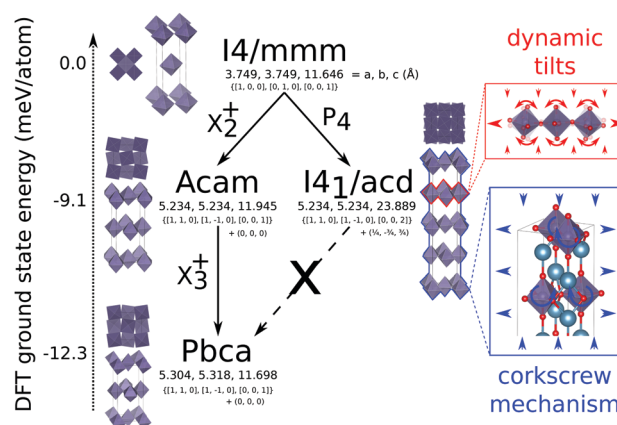


Fig. 1 The structure of Ca_2GeO_4 in aristotypal symmetry ($I4/mmm$), experimental ($I4_1/acd$) and thermodynamically competing and DFT ground state phase ($Pbca$), illustrating the different possible octahedral rotations and tilts. Lattice parameters and DFT ground state energies are also shown. Inset: The corkscrew mechanism illustrating how in the $I4_1/acd$ phase, the in-plane (ab) and layering (c) axes are strongly coupled such that dynamic tilts driving large in-plane PTE also drive NTE along c .

^a Center for Condensed Matter Sciences, National Taiwan University, Taipei 10617, Taiwan

^b Taiwan Consortium of Emergent Crystalline Materials, Ministry of Science and Technology, Taipei 10622, Taiwan

^c Department of Materials and the Thomas Young Centre, Imperial College London, London SW7 2AZ, UK

^d School of Physical Sciences, University of Kent, Canterbury CT2 7NH, UK

^e Department of Physics, Imperial College London, London SW7 2AZ, UK

^f Institute for Chemical Research, Kyoto University, Uji, Kyoto 611-0011, Japan

^g Integrated Research Consortium on Chemical Sciences, Uji, Kyoto 611-0011, Japan

^h Department of Chemistry, University of Warwick, Gibbet Hill, Coventry, CV4 7AL, UK. E-mail: m.senn@warwick.ac.uk

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then sintered at 1500 °C for a further 24 hours. The obtained precursor was sealed into a platinum capsule, and heated at 1000 °C at 15 GPa for 30 minutes in a Kawai type 2-stage high pressure apparatus. After the heating treatment the sample was quenched to room temperature over 5 minutes and the pressure slowly released. The sample was studied by temperature dependent synchrotron X-ray powder diffraction on beamline 09A of the Taiwan Photon Source, National Synchrotron Radiation Research Center (TPS, NSRRC) with a 15 keV incident beam and MYTHEN detector. The obtained Ca_2GeO_4 powder sample was sealed in a 0.3 mm diameter borosilicate capillary and 0.3 mm quartz capillary for high temperature experiment which was kept spinning during data collection for better averaging. A DynaFlow cryostat (designed and manufactured by ESRF) with dynamic cooling flow of exchange gas was utilised to provide low temperature diffraction data, for which the sample was cooled first to base temperature (10 K) and collect every 2 K on warming to 150 K. A Cryostream 800 Plus (Oxford Cryosystems) was utilised for data collection from 90 K to 480 K, at intervals of every 5 K to 150 K, every 10 K to 300 K, every 20 K to 480 K. A hot air gas blower (FMB Oxford) was utilised for data collection from 400 K to 800 K, at intervals of every 20 K to 600 K and every 50 K to 850 K. A broadening of diffraction peaks was observed around 800 K and the sample was fully decomposed to the olivine ambient pressure phase by 850 K. The collected patterns were analysed with the Rietveld method using TOPAS.¹² First-principles calculations were performed using plane-wave density functional theory (DFT) as implemented in CASTEP v7.0.3¹³ using the PBESol functional to describe exchange and correlation.¹⁴ A 1400 eV plane wave cutoff energy was employed for representing the wavefunctions in all calculations, and a $7 \times 7 \times 2$ grid of k -points were used for the $I4/mmm$ unit cell. This grid was scaled appropriately for larger unit cells to approximately maintain the density of grid points in k -space. Norm conserving pseudopotentials were generated using CASTEP v16.0 and the details may be found in the supplementary output files. In all geometry optimisation calculations, forces were relaxed to below $0.1 \text{ meV } \text{\AA}^{-1}$ and stresses to below 10 MPa.

The preliminary X-ray powder diffraction analysis revealed that the Ca_2GeO_4 precursor has the olivine type structure in which Ge are tetrahedrally coordinated, with space group $Pnma$. A temperature induced phase transition of olivine phase into glaserite type structure with transition temperature 1453 °C has previously been reported as the only other ambient pressure phase of this system.¹⁵ The desired meta-stable high pressure phase with K_2MgF_4 (Ruddlesden–Popper $n = 1$) type structure has previously been reported under the conditions 900–1000 °C at 10–12 GPa for 3–5 min,^{16,17} and crystallising in the space group $I4/mmm$. Our initial high pressure synthesis attempts using the olivine Ca_2GeO_4 as precursor with 1000 °C at 6 GPa for 30 min and 1200 °C at 9.5 GPa for 30 min conditions with DIA-type cubic anvil high pressure apparatus proved unsuccessful. In the present study we found that we had to use a much higher pressure of 15 GPa, afforded to us by the Kawai type 2-stage high pressure apparatus, to achieve the desired phase transformation to the K_2MgF_4 type structure.

The synchrotron PXRD data was refined in TOPAS using the Rietveld method. The main-phase corresponds to Ruddlesden–Popper $n = 1$ phase of Ca_2GeO_4 with aristotypical symmetry $I4/mmm$.



Fig. 2 Rietveld fit to 300 K high resolution X-ray powder diffraction data of Ca_2GeO_4 collected at $\lambda = 0.82661 \text{ \AA}$. Upper tick marks (cyan) indicate allowed reflections for the main $I4_1/acd$ phase and lower tick marks (yellow) indicate those associated with the olivine impurity phase. Inset: Superstructure peaks indexed with respect to the $I4_1/acd$ cell and impurity peaks from the olivine phase that significantly convolute this region.

Closer inspection of the room temperature diffraction pattern reveals very weak and broad superstructure peaks (Fig. 2) ($\left(hk\frac{l}{2}\right)$ with respect to the aristotype unit cell) indicating that a structure with an out-of-phase octahedral rotation about the c -axis is adopted ($I4_1/acd$, $a = 5.236339(5) \text{ \AA}$, $c = 23.81675(3) \text{ \AA}$ – see Fig. 1). These peaks are significantly broadened on account of the short coherence length of the octahedral rotations along the layering axis. The presence of an impurity with 10% phase fraction by weight corresponding to the ambient pressure phase of Ca_2GeO_4 ($Pnma$, $a = 11.404194 \text{ \AA}$, $b = 6.794365 \text{ \AA}$, $c = 5.242485 \text{ \AA}$), significantly convolutes the region of the diffraction pattern $20\text{--}23^\circ$ (see inset of Fig. 2), meaning that these peaks cannot be used to unambiguously assign the symmetry. However, a comparison of the fitting statistics for $I4/mmm$ ($R_{\text{wp}} = 2.07$, for two internal degrees of freedom) and $I4_1/acd$ ($R_{\text{wp}} = 2.00$, for three internal degrees of freedom), along with physically unrealistic atomic displacement parameters in the prior refinement, and DFT lattice parameters (see Table 1) and a ground state energy of $I4_1/acd$ which we find to be 9.1 meV per atom more stable than the $I4/mmm$ phase (see Fig. 1), all support this assignment.

$I4_1/acd$ (with a frozen rotation about c) is one of several known phases of A_2BO_4 Ruddlesden–Popper oxides¹⁸ that include the aristotype $I4/mmm$ (no rotations or tilts) and competing $Pbca$ phase (rotation about c and in-plane tilts) shown in Fig. 1. It is an interesting question why Ca_2GeO_4 crystallises as $I4_1/acd$ and not alternative phases, despite DFT showing $Pbca$ to be lower in energy. The answer likely relates to vibrational entropy favouring $I4_1/acd$ over the energetic ground state, however, this discussion is beyond the scope of the current study. It is noteworthy that there is no direct transition between $I4_1/acd$ and $Pbca$ without unravelling the anti-phase P_4 rotations to transform them to in-phase X_2^+ rotations. It is this so-called symmetry trapping that we have previously argued prevents the transition to the energetic ground state⁷ thus allowing octahedral tilts that have been shown to drive NTE^{8,10} to remain low in frequency without condensing.



Table 1 Structural measurements of $I4_1/acd$ Ca_2GeO_4 lattice parameters, unspecified Wyckoff coordinates in $\text{Ca}(0, 1/4, z)$; $\text{Ge}(0, 1/4, 3/8)$; $\text{O1}(3/4 + y, y, 1/8)$; $\text{O2}(0, 1/4, z)$ and frozen octahedral rotation angle, θ , compared to DFT simulations. For the full structures see supplementary cif files (ESI)

Method	T (K)	a (Å)	c (Å)	$\text{Ca}(z)$	$\text{O1}(y)$	$\text{O2}(z)$	θ (°)
X-ray (Rietveld)	24	5.217182(7)	23.85752(4)	0.050876(19)	0.0393(3)	−0.04491(7)	8.93(17)
X-ray (Rietveld)	300	5.236339(4)	23.81675(3)	0.051094(14)	0.0270(4)	−0.04454(5)	6.16(23)
DFT	0	5.234	23.89	0.0503	0.0466	−0.0446	10.6



Fig. 3 Lattice parameters of the $I4_1/acd$ phase of Ca_2GeO_4 as a function of temperature. The temperature devices used for different measurements are indicated in the legend. Inset: The single Einstein mode model plotted against the low temperature data to which it was fitted. Sum of the squared error (SSE) measures of fit shown for each lattice parameter.

Lattice parameters were extracted from the temperature dependent powder X-ray diffraction data by performing sequential Pawley refinement and are shown in Fig. 3. No discontinuities or anomalies in the lattice parameters were observed. Data collected on warming using the cryostat in the region 10–140 K shows a slight discontinuity with the data collected on cooling (despite normalisation using the volumes at 120 K) evidencing some hysteresis in the thermal expansion behaviour of the sample. This likely arises as a result of individual domains and grains expanding and contracting at vastly different rates along different directions, resulting in substantial macroscopic strain and hysteresis in the sample. Indeed, asymmetric peak broadening and shifts are observed at lower temperature. Fitting a Scherrer particle size Gaussian peak broadening model to the (hkl) $l = \text{odd}$ reflections gives an estimate of the domain size of 16.7 (1.4) nm.

Fitting the low temperature a and c lattice parameter data with a single Einstein mode model of the form

$$x(T) = x_0 + \frac{k_1^x}{\exp\left(\frac{k_2}{T}\right) - 1}, \quad (1)$$

we can obtain an estimate for the characteristic frequency of the vibrations that dominate contributions to the thermal expansion in this material. This model for a and c is plotted against the cryostat data to which it is fitted in the inset to Fig. 3.

We have previously shown that the character of the low energy vibrations giving rise to NTE along c are the same as those giving rise to PTE along a .¹⁰ Therefore a and c were fitted in tandem using least squares optimisation where the optimisation function was the sum of the errors normalised by lattice parameter for both a and c . This meant that a single characteristic frequency could be applied to both models. The characteristic phonon frequency, k_2 , is found to be 155 K (13.4 meV, 108 cm^{-1}). Fitting the same single Einstein mode model to a and c from our previous measurements of the isosymmetric system Ca_2MnO_4 ¹⁰ over an equivalent temperature range gives a k_2 of 261 K (22.5 meV, 182 cm^{-1}) which is substantially higher than in Ca_2GeO_4 (see Table 2 for full model parameters for both compounds). This is evident both in that the flattening of the NTE occurs at much lower temperatures in Ca_2GeO_4 , and that it is more pronounced compared to that observed in Ca_2MnO_4 . The average thermal expansion coefficient along the c direction, α_c , in Ca_2GeO_4 (100–400 K) is -6.3 ppm K^{-1} , which is substantially larger than for Ca_2MnO_4 measured over the same temperature range ($\alpha_c = -4.4 \text{ ppm K}^{-1}$).¹⁰

It is worth noting that in our previous computational studies of NTE in these systems,¹⁰ we used Ca_2GeO_4 as a substitute for Ca_2MnO_4 by applying a 4.3 GPa biaxial pressure to account for the in-plane exchange striction associated with Mn sites. The effect of this was to harden the soft phonon modes in our calculations with respect to the unstrained Ca_2GeO_4 phase. It is hence to be expected that in the experiment Ca_2GeO_4 shows more pronounced NTE than Ca_2MnO_4 , and the resulting trend is consistent with this picture. Ca_2GeO_4 and Ca_2MnO_4 ^{10,19} are not the only $n = 1$ family compounds exhibiting uniaxial NTE. Other examples found in the literature include Sr_2IrO_4 ^{20–22} and Sr_2RhO_4 ,^{23,24} both of which crystallise in the $I4_1/acd$ phase. However, in this instance the magnitudes of α_c are much smaller with an average of -2.4 ppm K^{-1} for Sr_2RhO_4 (calculated for 10–300 K from ref. 23) and -2.1 ppm K^{-1} in Sr_2IrO_4 (calculated for 13–295 K from ref. 20). It should be noted that Ca_2RuO_4 ²⁵ and related materials, which crystallise in an orthorhombic space group, exhibit NTE as a result of electronic instabilities and Jahn–Teller ordering, which can be very pronounced effects but which tend to be confined to a relatively narrow temperature range.

Table 2 Optimised parameters for single Einstein mode models for low temperature lattice parameters (see eqn (1)) fitted to experimental data for $I4_1/acd$ A_2BO_4 compounds up to 150 K

Compound	a_0 (Å)	c_0 (Å)	k_1^a (Å)	k_1^c (Å)	k_2 (K)
Ca_2GeO_4	5.2175	23.859	0.01119	−0.03101	154.9
Ca_2MnO_4 ¹⁰	5.1684	24.153	0.02907	−0.06214	261.2



Recently, it has been proposed that the quantity χ_{α_V} given by the product of the average volume thermal expansion coefficient α_V and the temperature range over which NTE is observed, can be used as a metric for classifying different regimes of volume NTE²⁶ and to help identify systems that exhibit remarkable NTE properties. Within this picture, Ca_2GeO_4 exhibits linear NTE with $\chi_\alpha = 0.55\%$ measured across the full 10–800 K window of stability of the $I4_1/acd$ phase. This compares to $\chi_{\alpha_V} = 2.74\%$ ($\chi_\alpha = 0.913\%$) in the prototypical NTE material ZrW_2O_8 ²⁷ (note the factor of 3 between linear α and volume α_V). It is, however, as yet unclear whether this metric may be used to meaningfully compare systems exhibiting volume (bulk) NTE and those exhibiting only uniaxial NTE. A recent report of pronounced uniaxial negative linear thermal expansion as large as $\alpha = -40 \text{ ppm K}^{-1}$ for the orthorhombic ($Cmc2_1$) phase in the (110)-cut perovskite LaTaO_4 represents possibly one of the largest linear CTEs in hard ceramic materials to date.²⁸ This phase, which we have recently shown to be very compliant with respect to coupling expansion along the PTE axis to contraction along NTE axes,¹⁰ has a $\chi_\alpha = 0.68\%$, which is larger yet comparable to Ca_2GeO_4 , except here NTE persists over a much narrower temperature range (<300 K). By tuning the chemistry, it should be possible to synthesise an A_2BO_4 layered perovskite with $I4_1/acd$ symmetry which is stable to higher temperatures thus extending the temperature range of NTE. Even down to cryogenic temperatures we find that Ca_2GeO_4 does not transform into the competing $Pbca$ phase. Therefore it should be possible to tune the tolerance factor to bring the system closer to this transition – thus increasing the thermodynamic driving force for NTE. If either (or both) of these enhancements could be achieved by chemical substitution, it may be possible to engineer a Ruddlesden–Popper possessing a significantly greater magnitude of NTE than LaTaO_4 by this χ_α metric.

In summary, we have successfully synthesised the layered perovskite Ca_2GeO_4 showing that it crystallises in the $I4_1/acd$ rather than $I4/mmm$ space group. In accordance with our predictions this structure shows pronounced uniaxial NTE over a large temperature range. This phenomenon can be understood to be driven by low energy lattice vibrations related to a thermodynamically competing $Pbca$ phase (shown in Fig. 1) in which these octahedral tilts condense, and the unusually compliant nature of the $I4_1/acd$ structure. Single Einstein mode fits of the a and c lattice parameters as a function of temperature provide an upper estimate for the phonon frequency of 155 K (108 cm^{-1}) for the characteristic modes driving NTE. The unusually wide temperature range over which these soft phonon modes persist may make them useful in controlling properties associated with thermal expansion, dielectric response and other phonon-mediated properties such as thermal transport, and the thermoelectric effect.

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Conflicts of interest

There are no conflicts to declare.

Notes and references

- 1 M. J. Pitcher, P. Mandal, M. S. Dyer, J. Alaria, P. Borisov, H. Niu, J. B. Claridge and M. J. Rosseinsky, *Science*, 2015, **347**, 420–424.
- 2 R. Zhang, M. S. Senn and M. A. Hayward, *Chem. Mater.*, 2016, **28**, 8399–8406.
- 3 S. Yoshida, H. Akamatsu, R. Tsuji, O. Hernandez, A. S. Gupta, A. S. Gibbs, K. Mibu, S. Murai, J. M. Rondinelli, V. Gopalan, K. Tanaka and K. Fujita, *J. Am. Chem. Soc.*, 2018, **140**, 15690–15700.
- 4 H. Zhang, S. Gao, Q. Zhang, J. Wu, J. Liang, C. Dong, L. Gu, S. Dong, J. Sun, F. Liao, J. Lin, R. Zou and G. Li, *Chem. Mater.*, 2017, **29**, 9840–9850.
- 5 T. Zhu, T. Cohen, A. S. Gibbs, W. Zhang, P. S. Halasyamani, M. A. Hayward and N. A. Benedek, *Chem. Mater.*, 2017, **29**, 9489–9497.
- 6 N. A. Benedek and C. J. Fennie, *Phys. Rev. Lett.*, 2011, **106**, 107204.
- 7 M. S. Senn, A. Bombardi, C. A. Murray, C. Vecchini, A. Scherillo, X. Luo and S. W. Cheong, *Phys. Rev. Lett.*, 2015, **114**, 035701.
- 8 M. S. Senn, C. A. Murray, X. Luo, L. Wang, F.-T. Huang, S.-W. Cheong, A. Bombardi, C. Ablitt, A. A. Mostofi and N. C. Bristowe, *J. Am. Chem. Soc.*, 2016, **138**, 5479–5482.
- 9 C. Ablitt, M. S. Senn, N. C. Bristowe and A. A. Mostofi, arXiv, 2018, arxiv:1810.02697.
- 10 C. Ablitt, S. Craddock, M. S. Senn, A. A. Mostofi and N. C. Bristowe, *npj Comput. Mater.*, 2017, **3**, 44.
- 11 C. Ablitt, A. A. Mostofi, N. C. Bristowe and M. S. Senn, *Front. Chem.*, 2018, **6**, 455.
- 12 A. A. Coelho, *Topas v4.2: General Profile and Structure Analysis Software for Powder*, 2009.
- 13 S. J. Clark, M. D. Segall, C. J. Pickard, P. J. Hasnip, M. J. Probert, K. Refson and M. Payne, *Z. Kristallogr. - Cryst. Mater.*, 2005, **220**, 567–570.
- 14 J. P. Perdew, A. Ruzsinszky, G. I. Csonka, O. A. Vydrov, G. E. Scuseria, L. A. Constantin, X. Zhou and K. Burke, *Phys. Rev. Lett.*, 2008, **100**, 136406.
- 15 W. Eysel and T. Hahn, *Z. Kristallogr.*, 1970, **131**, 322–341.
- 16 A. E. Ringwood and A. F. Reid, *Earth Planet. Sci. Lett.*, 1968, **5**, 67–70.
- 17 A. F. Reid and A. E. Ringwood, *J. Solid State Chem.*, 1970, **1**, 557–565.
- 18 P. V. Balachandran, D. Puggioni and J. M. Rondinelli, *Inorg. Chem.*, 2014, **53**, 336–348.
- 19 J. Takahashi and N. Kamegashira, *Mater. Res. Bull.*, 1993, **28**, 565–573.
- 20 Q. Huang, J. L. Soubeyroux, O. Chmaissem, I. Natali Sora, A. Santoro, R. J. Cava, J. J. Krajewski and W. F. Peck, *J. Solid State Chem.*, 1994, **112**, 335–361.
- 21 M. K. Crawford, M. A. Subramanian, R. L. Harlow, J. A. Fernandez-Baca, Z. R. Wang and D. C. Johnston, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1994, **49**, 9198.
- 22 B. Ranjbar and B. J. Kennedy, *J. Solid State Chem.*, 2015, **232**, 178–181.
- 23 T. Vogt and D. J. Buttrey, *J. Solid State Chem.*, 1996, **189**, 186–189.
- 24 B. Ranjbar and B. J. Kennedy, *Solid State Sci.*, 2015, **49**, 43–46.
- 25 M. Braden, G. Andre, S. Nakatsugi and Y. Maeno, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1998, **58**, 847–861.
- 26 C. S. Coates and A. L. Goodwin, *Mater. Horiz.*, 2018, DOI: 10.1039/C8MH01065J.
- 27 T. a. Mary, J. S. O. Evans, T. Vogt and A. W. Sleight, *Science*, 1996, **272**, 90–92.
- 28 K. J. Cordrey, M. Stanczyk, C. A. L. Dixon, K. S. Knight, J. Gardner, F. D. Morrison and P. Lightfoot, *Dalton Trans.*, 2015, **44**, 10673–10680.

