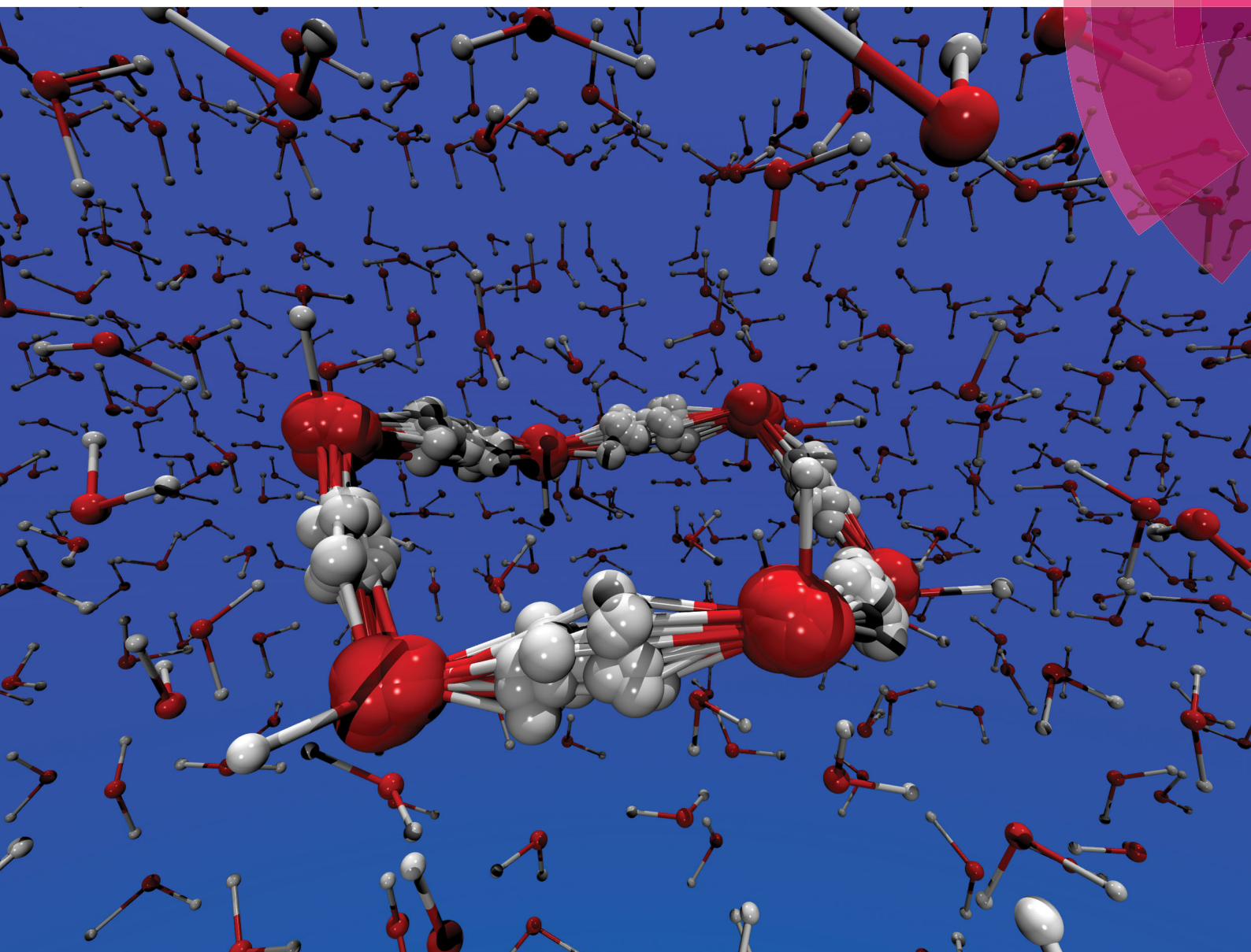


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PERSPECTIVE

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Collective proton transfer in ordinary ice: local environments, temperature dependence and deuteration effects

Christof Drechsel-Grau* and Dominik Marx*

The transfer of multiple protons in hydrogen-bonded networks usually occurs one proton at a time. At sufficiently high temperatures, each proton transfers via thermally activated hopping along its hydrogen bond, thereby moving a charge defect through the network. At low temperatures, quantum-mechanical tunnelling might set in instead, thus avoiding hopping over the energy barriers. In the case of several transferring protons, independent thermal hopping or quantum tunnelling of the individual protons becomes less favourable because of a significant creation of charge defects. In individual molecules or hydrogen-bonded molecular complexes, for instance, double proton transfer is often found to be concerted. Multiple proton transfer that avoids charge defects can occur in cyclic topologies built from several hydrogen bonds that allow for directional chains of proton transfer. This requires perfect proton order within these rings, which imposes handedness and thus chirality, and changes parity upon transfer of all protons. Ordinary ice, which is hexagonal ice I_h , is the most stable form of crystalline ice obtained upon freezing liquid water at ambient pressure and consists of interconnected six rings of oxygen atoms that host six protons each. These hexagonal rings remain proton disordered even down to low temperatures, as heralded by the residual entropy of ice I_h . However, owing to combinatorics, a certain number of these six rings is proton ordered in macroscopic crystals. These chiral hexameric rings might support coherent tunnelling of the hosted protons. Indeed, there is some evidence in the recent literature, both experimental and simulational, that correlated tunnelling of all six protons might be possible in proton-ordered six rings in ice I_h if temperatures are low enough. In this Perspective, the key ideas and previous findings will be reviewed in the light of relevant experiments with a focus on available *ab initio* path integral simulation work supplemented with additional data provided herein.

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1 Background: nuclear quantum effects and multiple proton transfer

Proton tunnelling belongs to the hallmarks of quantum mechanics. It represents the classically forbidden motion of a proton through the potential energy barrier that separates the final from the initial state – instead of the classically allowed thermally activated process of hopping over the top of that barrier. Although this intriguing phenomenon has attracted attention since the beginning of quantum mechanics, it is the exception rather than the rule in standard condensed phase chemical reactions at not too low temperatures. Much more common quantum effects in chemical kinetics are quantum delocalisation and zero-point motion. Proton tunnelling gets strongly suppressed upon increasing the height and/or the width of potential energy barriers and typically requires

unusually favourable circumstances (such as short hydrogen bonds and stiff molecular skeletons) to contribute appreciably to chemical dynamics.

Liquid water and aqueous solutions first come to mind when looking for quantum effects in hydrogen-bonded networks, and in particular for proton tunnelling.¹ Despite these expectations, molecular simulations that explicitly include nuclear quantum effects in terms of path integrals disclosed that both proton and hydroxide transfers in water under ambient conditions are mainly driven by specific hydrogen-bond fluctuations around these charge defects and, therefore, are essentially thermally activated processes.^{1–5} These fluctuations dramatically lower the energy barrier, which is unacceptably high at the typical hydrogen-bond distances in such systems of ~ 2.8 Å but is greatly reduced if the donor–acceptor distance for proton transfer is decreased, for instance as a result of thermal motion or extreme compression.¹ Therefore, the reduction of the intermolecular O...O distance in conjunction with proton transfer is often observed for aqueous systems and goes hand in hand

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The strength of hydrogen bonds is a key feature influencing the possibility of proton tunnelling and depends sensitively on the potential energy surface.¹⁰ Given the complexity of hydrogen bonding interactions already at the level of an individual water dimer,^{11,12} it is unsurprising that the wetting behaviour of various surfaces¹³ and a variety of nanostructures on metals¹⁴ indicate a vast richness of hydrogen-bond strengths. Proton transfer to and from water molecules on a metal oxide surface, for example, has been identified as the pathway for hydrogen migration.¹⁵ Although double proton transfer is a coherent tunnelling process in the formic-acid dimer at low temperatures, as evidenced by the measured tunnelling splitting in the gas phase,¹⁶ the transfer of the two protons *via* zero-point motion is

quantum mechanically uncorrelated at higher temperatures.¹⁷ In contrast, double proton transfer is stepwise in hydroporphyrins.¹⁸ In biologically relevant systems, hydrogen-bonded chains create pathways for proton transfer,¹⁹ and proton-ordered, hydrogen-bonded, cyclic topologies might contribute to glycine formation in the interstellar medium.²⁰ These examples illustrate the delicate balance of the strength of hydrogen bonds in a network between the limits of a sufficient coupling for strong donor-acceptor interactions and sufficient flexibility for donor-acceptor fluctuations.

Hexagonal ice, dubbed ice I_h , provides a rich playground for the study of nuclear quantum effects owing to its peculiar arrangement governed by the ice rules,^{21,22} giving rise to its residual entropy.^{23,24} Nuclear quantum effects affect the molecular dipole-moment distribution in hexagonal ice²⁵ and explain the anomalous lattice expansion upon $H \rightarrow D$ substitution.²⁶ Ice I_h is the most stable pure phase of water at low temperature and ambient pressure.²² Cubic ice I_c , denoted as I_c , is its metastable variant but never transforms to ice I_h upon cooling.²² A fully proton-ordered form of ice I_h , called ice XI, only exists in the presence of dopants such as KOH and is thus not a pure phase of solid H_2O .²² Hexagonal ice is, therefore, the most natural phase of solid water to study nuclear quantum effects at low temperature and ambient pressure. Nuclear quantum effects also underlie the low-temperature dynamics of non-crystalline water near the glass transition temperature.²⁷



Christof Drechsel-Grau is a chemist interested in the thermodynamics, kinetics, and mechanisms of chemical reactions. He has been particularly fascinated by the mechanisms of charge-transfer processes in condensed-phase environments. A particular focus of his work has been the influence of aqueous environments on electron and proton transfer via thermal and quantum fluctuations to elucidate specific aspects of the quantum-classical boundary.



Dominik Marx studied chemistry and physics at Universität Mainz and the University of California at Irvine. He received his Diplom (MSc) in Chemistry (1990) at MPI für Chemie and his PhD (1992) at Institut für Physik (Universität Mainz). Thereafter, he worked as a Postdoctoral Fellow at IBM Zurich Research Laboratory (Rüschlikon), as a staff scientist at MPI für Festkörperforschung, and obtained the Habilitation in Theoretical Physics at Universität zu Köln. He received a full Professorship at Ruhr-Universität Bochum. Several Chairs in Germany and abroad and he is currently the Distinguished Visiting Professor of Chemistry and the Head of the Center for Molecular Simulation at Ruhr-University Bochum. Professor Marx has been inspired by the multifaceted problems of the interplay of physics and chemistry of complex molecular systems. He tackled using the utmost "realistic" theoretical and computational techniques. Leitmotifs of the Marx Group are the development of novel quantum and computational techniques for molecular many-body problems, their application in terms of High-Pressure Chemistry and High-Pressure Physics.

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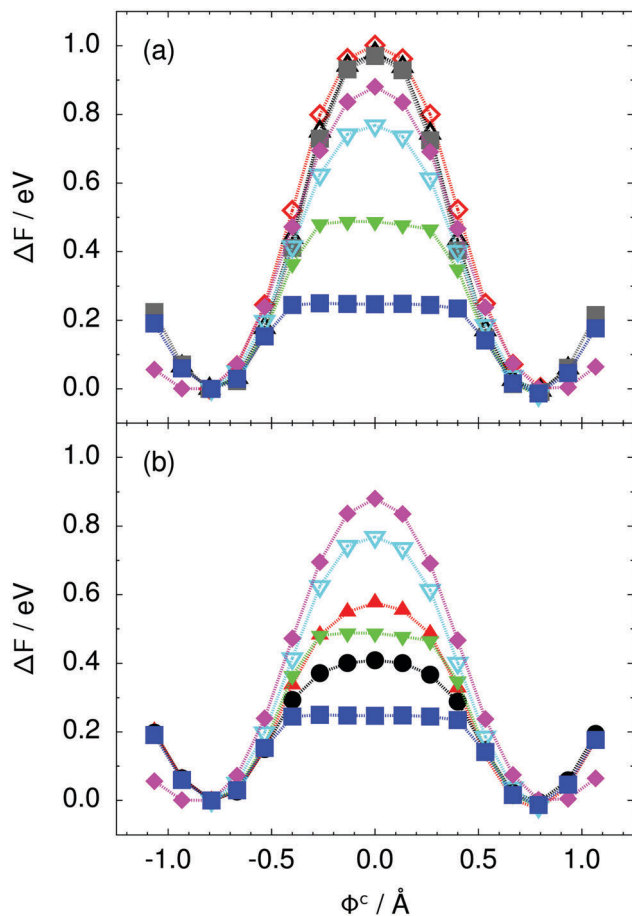


Fig. 3 Free energy profile along the constrained collective proton transfer coordinate Φ^c in proton-ordered six rings (see Fig. 1) for (a) the pure H_2O systems and for (b) all quantum systems. Simulations that treat all nuclei classically are shown in (a) for 50 K (filled grey squares), 100 K (empty black up triangles), and 300 K (empty red diamonds). Simulations that treat all or most nuclei quantum mechanically are shown in (b) for non-deuterated (filled blue squares: 50 K; filled green down triangles: 100 K; empty cyan down triangles: 200 K; filled pink diamonds: 320 K) and partially deuterated (filled black circles: 50 K and one deuterium; filled red up triangles: 50 K and one deuterium per six ring) systems. See Table 1 for symbol definitions.

Key analysis underlying the previous conclusion⁴³ that all six protons tunnel in a perfectly concerted manner in the sense of a delocalised quasi-particle is provided by the data in Fig. 2c; the interested reader is referred to additional so-called world-line analysis of individual tunnelling paths in Fig. 3 of ref. 43 and to the representative videos present in its ESI. The many-body concerted nature of the tunnelling process at low temperature is disclosed by comparing in a statistical sense the imaginary time evolution of the individual six local proton transfer coordinates ϕ_j (as depicted in Fig. 2d for ϕ_1 in terms of its correlation function) to the one of the collective proton transfer coordinate Φ . This is established by computing the average difference

$$\Delta(\gamma, \tau) = \left(\sum_{j=1}^6 |\Phi(\gamma, \tau) - \phi_j(\gamma, \tau)|^2 / 6 \right)^{1/2} \quad \text{between the collective and all six local proton transfer coordinates, i.e. deviations between } \Phi(\tau) \text{ and all } \phi_j(\tau) \text{ at each sampled configuration } \gamma \text{ and}$$

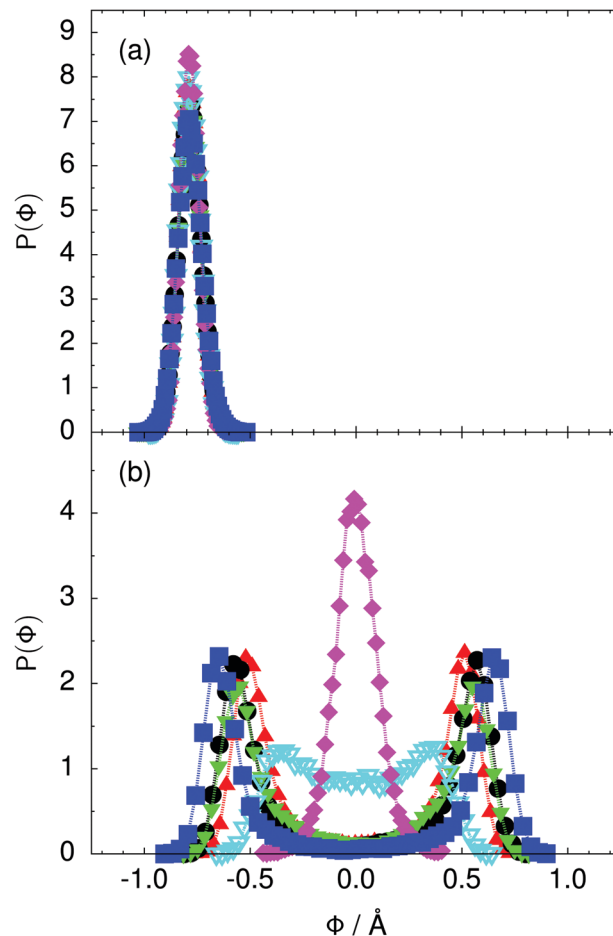


Fig. 4 Distribution of the collective proton transfer coordinate (see the text), $\Phi = \sum_{j=1}^6 \phi_j / 6$, (a) in the initial state and (b) in the transition state for collective proton transfer in a proton-ordered six ring (see Table 1 for symbol definitions).

all imaginary times $\tau \in [0, \hbar\beta]$ as represented by the discretised Trotter beads s given in the caption of Fig. 7 (see Section 6 for background). Thus, $\langle \Delta \rangle \equiv 0$ Å if all local proton transfer coordinates $\phi_j(\tau)$ always exactly follow the global $\Phi(\tau)$ for all imaginary times τ . For the initial state depicted in panel (a) of Fig. 7, the distribution $P(\Delta)$ at 50 K is sharply peaked at a small value Δ_{max} that quantifies ordinary quantum broadening at low temperatures due to zero-point motion. Importantly, its shape in the transition state at 50 K is statistically identical to that in the initial state, see panel (b), thereby unveiling that all six protons must move collectively as a quasi-particle at 50 K.

The concerted proton motion at 50 K also leads to a negligible number of formal charge defects in the transition state, as evidenced by the bimodal distribution of the number of transferred protons peaked at 0 and 6 in Fig. 6, and to a negligible probability of observing $\Phi = 0$ (Fig. 2b). All protons are statistically identical within the sampling error, as seen from Fig. 8 and 10, which shows the distribution of and the correlation between representative local proton transfer coordinates.



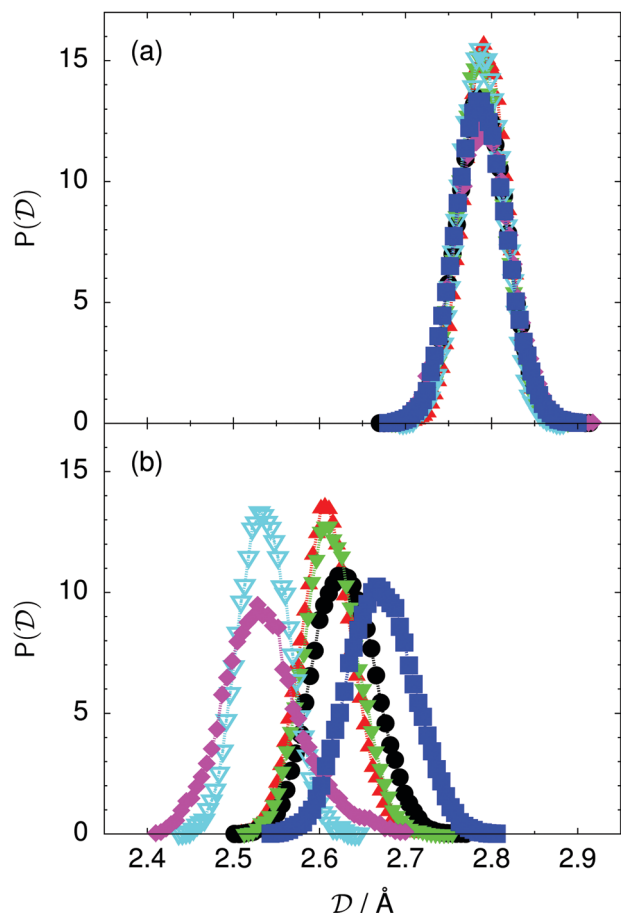


Fig. 5 Distribution of the average donor–acceptor distance within the chair-like proton-ordered six ring (a) in the initial state and (b) in the transition state for collective proton transfer (see Table 1 for symbol definitions).

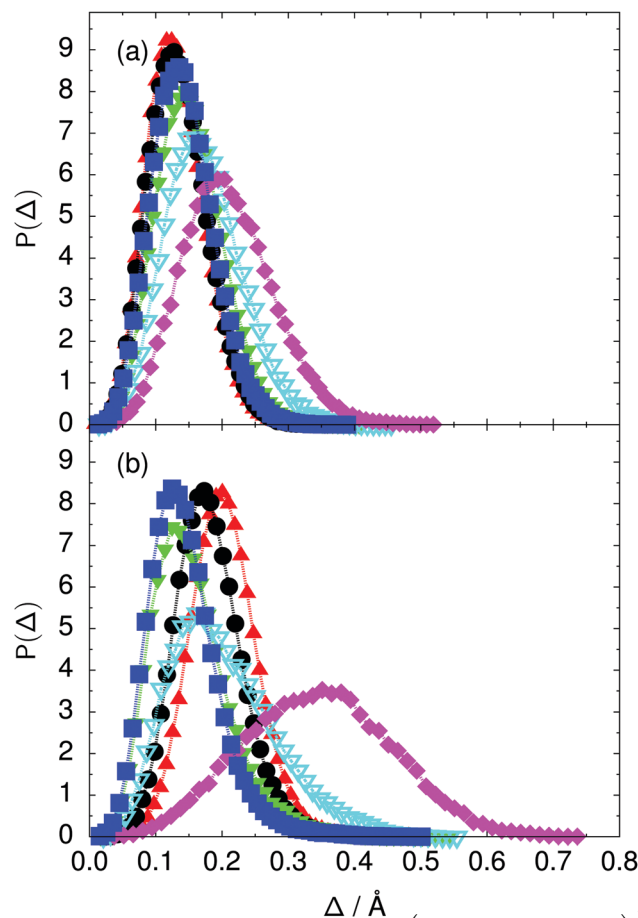


Fig. 7 Distribution of the deviation $\Delta \equiv \Delta(\gamma, s) = \left(\sum_{j=1}^6 |\Phi(\gamma, s) - \phi_j(\gamma, s)|^2 / 6 \right)^{1/2}$ of the local proton transfer coordinates ϕ_j from the collective proton transfer coordinate Φ for each computed configuration γ and at all imaginary times $\tau \in [0, \hbar\beta]$ as represented by the discrete Trotter bead index $s = 1, \dots, P$ (a) in the initial state and (b) in the transition state for collective proton transfer (see Table 1 for symbol definitions). Note that $\Delta = 0$ indicates perfect correlation among the global coordinate Φ and all local proton transfer coordinates ϕ_j at that particular configuration and Trotter slice.

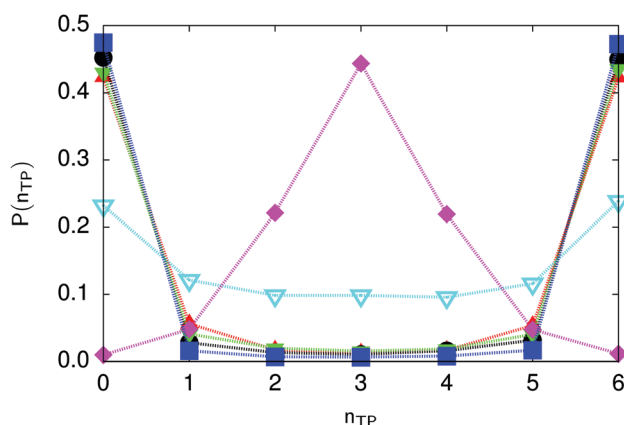


Fig. 6 Distribution of the number of transferred protons in the transition state for collective proton transfer (see Table 1 for symbol definitions). Data points at 50 K overlap on the scale of this figure for $P(n_{\text{TP}}) \sim 0$.

Hydrogen bond 4 is located opposite of hydrogen bond 1 in the six ring, and hydrogen bonds 2 and 3 are symmetrically equivalent to hydrogen bonds 6 and 5, as illustrated in Fig. 1. The strong correlation among protons in the transition state is

also essentially identical for collective proton transfer in chair-like and boat-like six rings (see Fig. 2c). The rather small contraction of the six-ring skeleton for the boat-like conformation resembles that for the chair-like conformation, as seen from the distribution of the mean donor–acceptor distance in Fig. 5, and results in a similarly low free energy barrier, displayed in Fig. 2a.

The strong similarity between the full quantum simulation results for collective proton transfer in chair-like and boat-like six rings at 50 K shows that the conformation of proton-ordered six rings within hexagonal ice, and thus the local environment, has no effect on the computed observables. What matters most for collective proton transfer at sufficiently low temperatures in non-deuterated hexagonal ice is the cyclic arrangement of the six water molecules within the crystal and the local proton order within the six ring.

Having disclosed the negligible influence of the local six-ring conformation on collective proton transfer above, we focus



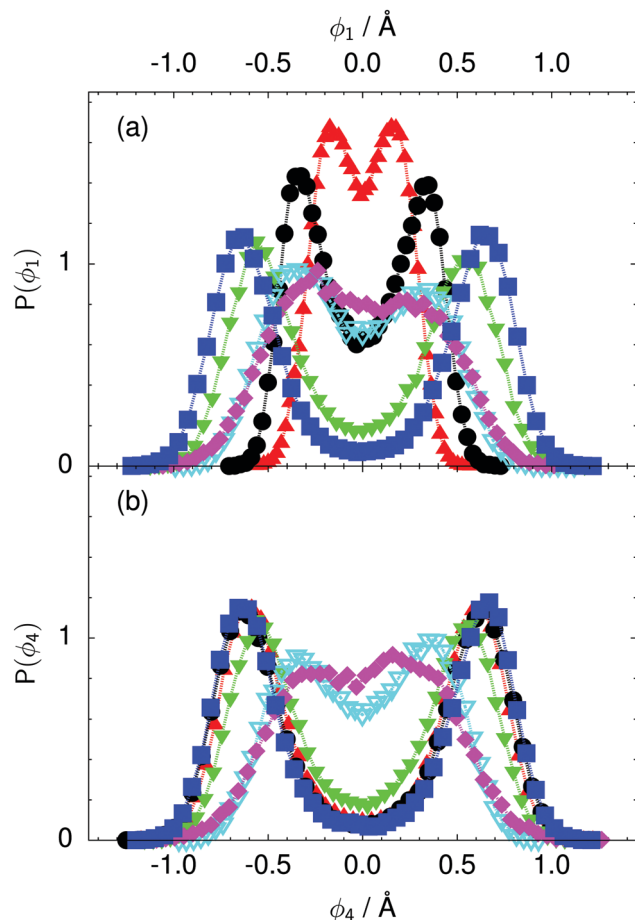


Fig. 8 Distribution of the local proton transfer coordinate (a) ϕ_1 and (b) ϕ_4 in the transition state for collective proton transfer (see Fig. 1 for the numbering of the hydrogen bonds and Table 1 for symbol definitions). Black circles, red up triangles, and blue squares in (b) overlap on the scale of this figure.

on chair-like proton-ordered six rings throughout the remainder of this Perspective.

2.2 Influence of temperature

Prior work on hexagonal ice^{43,44} has only established two limiting cases, namely concerted quantum tunnelling at 50 K (filled squares in all figures) and sequential classical proton transfer at 320 K (diamonds in all figures). Importantly, the high-temperature quantum simulations were found to share all qualitative features of classical simulations at 300 K. The shapes of the free energy profiles as well as the resulting barrier heights are seen to be essentially identical at 300, 100, and 50 K provided that the protons are treated as classical particles (see Fig. 3). Thus, the free energy barrier for quantum simulations at 320 K approaches that of these classical simulations, which demonstrates that thermal fluctuations do not influence the system if nuclear motion is treated classically in this temperature window. This implies that the proton transfer barrier is very high at the scale of the thermal energy; note that $1 k_B T \approx 0.026$ eV or ≈ 0.6 kcal mol⁻¹ at 300 K. We thus consider herein

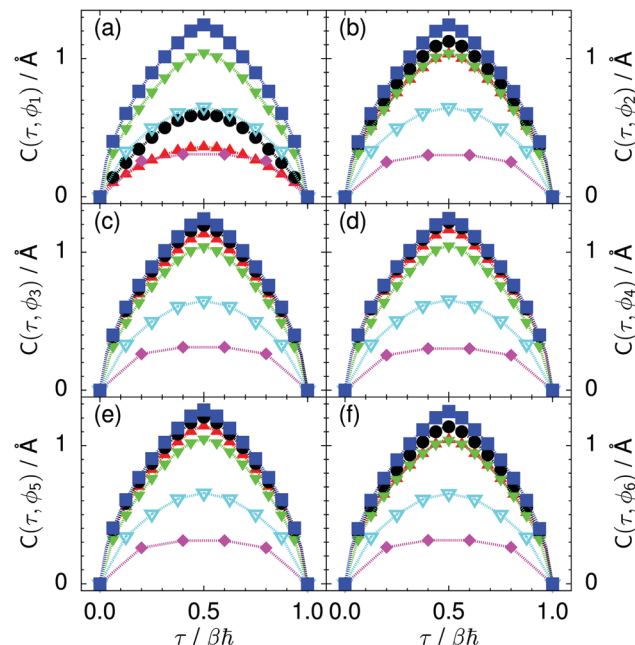


Fig. 9 Imaginary time correlation function of all local proton transfer coordinates $\phi_1, \phi_2, \dots, \phi_6$, defined as $C(\tau, \phi_j) = \langle |\phi_j(\tau) - \phi_j(0)|^2 \rangle^{1/2}$, in (a) to (f) in the transition state for collective proton transfer (see Fig. 1 for the numbering of the hydrogen bonds and Table 1 for symbol definitions). Filled red up and green down triangles overlap in (b) and (f); filled black circles and blue squares overlap in (c)–(e).

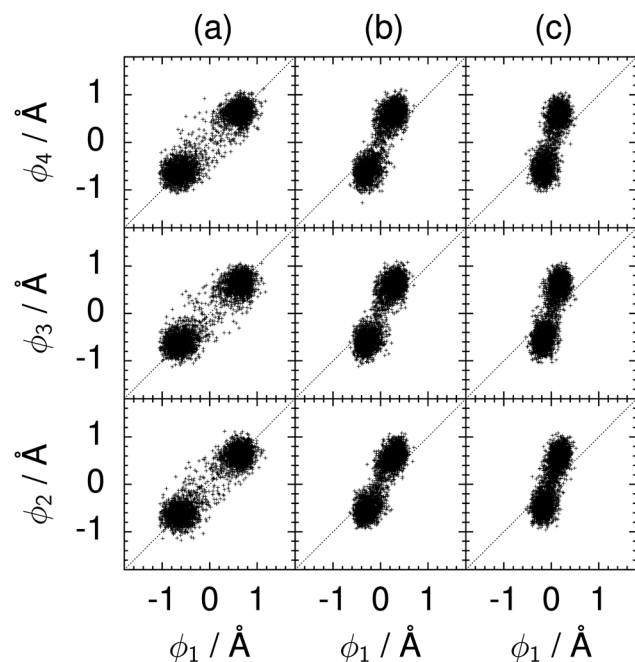


Fig. 10 Correlation between the local proton transfer coordinates ϕ_1 and ϕ_j with $j = 2, 3, 4$ in the transition state for collective proton transfer in a chair-like proton-ordered six ring at 50 K (see Fig. 1 for the numbering of the hydrogen bonds). Panel (a) represents pure H₂O. Panels (b) and (c) represent systems with one deuteron and with one deuteron per six ring, respectively. For clarity, only every hundredth data point is displayed.

non-classical proton jumps.⁴² The present work thus provides specific characteristics of the transition temperature region and its relation to the limiting high- and low-temperature scenarios.

2.3 Influence of the deuteration fraction

Replacing, on average, one proton per six ring by a deuteron (corresponding to a deuteration fraction of about 20%) has been shown to lead to the breakdown of collective proton tunnelling at low temperature both in the QENS measurements⁴² and in the *ab initio* path integral simulations.⁴⁴ This deuteration represents a significant perturbation because it not only replaces about one fifth of all protons in the system but also destroys on average the symmetry in the proton-ordered six rings. At variance with experiment, computer experiments allow us to replace protons by deuterons in a controlled manner. Here, we study the extreme case of replacing only one proton by a deuteron in the proton-ordered six ring to investigate the effect of this much milder perturbation of the entire system.

We found that even a single deuteron in the entire system, located in the proton-ordered six ring (filled black circles), strongly perturbs the collective proton tunnelling process. The free energy barrier almost doubles compared to the fully protonated system (filled blue squares), as seen from Fig. 3. One deuteron in the six ring also leads to significant deviations from the all-proton system in the distributions $P(\Delta)$, $P(\phi_1)$, and $C(\tau, \phi_1)$, and in the correlation between local proton transfer coordinates in the transition state, see Fig. 7–10, respectively. Although deuteration hardly affects the distributions of other collective variables, these deviations suggest that the replacement of a single proton by a deuteron in the entire system, if taking place in the proton-ordered six ring, induces localisation around this impurity very similar to the case of one deuteron per six ring.⁴⁴

Having discussed the effects of temperature and deuteration separately, we compare them in the following section.

The free energy profiles in Fig. 3b show that the activation free energy of the fully protonated system at 100 K is higher than

According to the above criteria extracted from our simulation data, yet being of a purely phenomenological nature, multiple proton transfer in hexagonal ice occurs for fully protonated systems at temperatures up to about 100 K as a concerted many-body tunnelling process that involves all protons within proton-ordered hydrogen-bonded cyclic hexamers in the spirit of a quasi-particle. In contrast, temperatures above 100 K or one deuteron impurity in locally proton-ordered six rings at 50 K already induce a clear breakdown of this peculiar collective proton tunnelling process. This assessment is consistent with our earlier finding from Section 2.2 of a broad crossover regime from quantum to classical behaviour somewhere between 100 and 200 K.

The simulations presented herein predict that the mechanism of the transfer of six protons in a locally proton-ordered six ring in hexagonal ice depends on temperature and isotopic composition as discussed in the previous section. This greatly extends the previous findings that, at 50 K, all six protons in proton-ordered six rings tunnel collectively in a perfectly concerted manner akin to a quasi-particle, which is destroyed by either partial deuteration or heating the system up to room temperature.^{43,44} These conclusions hinge on a set of approximations in order to make the underlying large-scale quantum simulations computationally practical. First of all, the simulations assume proton transfer along hydrogen bonds exclusively within the proton-ordered six rings by explicitly preventing proton transfer of the ordered six ring with its environment *via* hydrogen bonds (see Section 6 for details). They also induce the collective proton transfer within the proton-ordered hexamer by means of a constraint, as explained in Section 6 (which, however, does not impose the collective nature of the transfer but allows for defect formation as indeed produced in the high-temperature limit). Another approximation is the use of the classical limit of nuclei to assess the effect of (partial) deuteration by representing their paths by points in space. This approach more clearly brings out the consequences of deuteration given limited statistical sampling

Recent experiments probed collective proton motion in locally ordered water rings: a study using incoherent quasi-elastic neutron scattering (QENS) investigated bulk hexagonal ice⁴² and a scanning tunnelling microscopy (STM) study manipulated a hydrogen-bonded water tetramer adsorbed on a sodium chloride film supported on gold.⁴⁰ The QENS study discovered proton motion in proton-disordered hexagonal ice down to 5 K. The quasi-elastic signal disappeared both in fully proton-ordered ice VIII and in partially deuterated hexagonal ice, suggesting that the signal arises only in macroscopically proton-disordered phases of ice and involves the correlated quantum motion of more than one proton. The QENS study suggested a picture of six collectively tunnelling protons in locally proton-ordered water hexamers as they occur naturally in bulk hexagonal ice at thermal equilibrium. Open questions include the high rate constant of the collective proton motion, the transition temperature to the classical limit, the reason for the complete vanishing of the signal upon partial deuteration, and the role of the model and its implied temperature dependence in interpreting the raw data.

The STM study investigated four cyclically arranged water molecules with proton-ordered hydrogen bonds.⁴⁰ The proton-ordered water tetramer exists in two chiral arrangements: clockwise and anti-clockwise hydrogen bonds within the ring by close analogy with the proton-ordered cyclic hexamers in ice.⁴¹ The tunnelling current changed between two values for a given distance of the STM tip and the water tetramer, suggesting the collective proton transfer of all four protons in the four-ring to go from one chiral state of the tetramer to the other. The switching rates between the chiral states are virtually independent of temperature, tunnelling current, and bias potential, but they depend both on the distance of the STM tip from the water tetramer and on the position of the tip relative to the centre of the water tetramer.

Despite a wealth of stimulating insights obtained so far from simulations and theory, refined and extended studies are required that probe more directly the consequences of the different scenarios in terms of experimental observables. Of prime interest would be the full dynamical (quantum) structure factor, thus including both the coherent and incoherent contributions, as a function of both temperature and deuteration fraction as well as the associated timescales of all underlying processes.

To date, the most direct evidence for collective proton tunnelling stems from a QENS study of hexagonal ice as a function of temperature compared to proton-ordered ice VIII and to partially deuterated samples.⁴² Further supporting evidence for the concept of collective proton tunnelling in cyclic hydrogen-bonded topologies comes from an STM study that attributed the chirality switching of a water tetramer on a substrate to collective tunnelling.⁴⁰ Indeed, path integral simulations disclosed the possibility that all protons in proton-ordered water six rings occurring in ordinary ice can tunnel in a perfectly concerted manner like a delocalised quasi-particle at low temperatures, whereas both increasing thermal fluctuations and deuteration fraction destroy this phenomenon and lead to the usual activated sequential hopping of the six individual protons instead.^{43,44} Recent lattice gauge theoretical investigations, on the other hand, propose a quantum proton liquid ground state with vanishing entropy for hexagonal ice, being a coherent superposition of an exponentially large number of proton configurations obeying the usual ice rules.⁴⁵ All these studies contribute to our understanding of the collective transfer of multiple protons in hydrogen-bonded systems in the low-temperature quantum regime. However, despite this growing yet partially contradicting evidence in favour of some sort of collective quantum proton transfer in condensed systems at low temperature, possibly *via* concerted quasi-particle tunnelling in localised cyclic structures or in the spirit of a coherent quantum liquid encompassing the entire system, no truly conclusive evidence prevails to date. This urgently clearly calls for intensified research in the future using experimental, theoretical and simulational approaches with the sole aim of either strengthening, modifying, or refuting this emerging concept.

Appendix: computational methods and technical details

Defect-free hexagonal ice I_h exhibits a regular oxygen sublattice, whereas the hydrogen sublattice is disordered but satisfies the ice rules.^{21–23} Our model of such ordinary ice (shown in Fig. 1) consisted of 48 water molecules placed in an orthorhombic simulation cell of hexagonal symmetry²² subject to periodic boundary conditions. The volume of the simulation cell was $13.54 \times 15.64 \times 7.37 \text{ \AA}^3$. A purpose-built software (used in ref. 43 but unpublished) produced overall proton-disordered crystal structures with zero dipole moment and minimal quadrupole moment subject to both the usual ice rules⁵⁶ and the presence of a central proton-ordered six-membered ring (see Fig. 1). To represent the proton disorder of hexagonal ice, we selected from a set of stochastically generated structures one among those with maximum randomness of the protons.⁵⁶ Based on this set-up, we studied collective proton transfer, *i.e.* the transition from the initial to the final state, as illustrated in Fig. 1a.

Proton transfer reactions require a quantum-mechanical description of the electronic structure that allows for the breaking and the formation of covalent chemical bonds. We used Kohn–Sham density functional theory with the BLYP exchange–correlation functional^{57,58} in conjunction with a plane wave basis set expansion and norm-conserving pseudopotentials.⁵⁹ The cutoff for the plane wave expansion of the valence Kohn–Sham orbitals at the Γ -point of the Brillouin zone was 70 Ry, see *e.g.* ref. 48 for general background on such plane wave pseudopotential electronic structure calculations. It is stressed that the same electronic structure methodology is used for what we call “classical” and “quantum” (*ab initio*) simulations in order to represent classical and quantum nuclei, respectively, as specified in more detail below. The accuracy of these plane wave pseudopotential BLYP density functional calculations has been most carefully assessed in comparison to all-electron and frozen-core MP2 and CCSD(T) reference calculations using Gaussian basis sets for representative proton transfer processes sampled in hexagonal ice, as shown in the ESI (see in particular Fig. S8 and S9) of ref. 43.

The present work has focussed on the temperature and mass dependence of collective proton transfer in hexagonal ice. We captured thermal fluctuations of the atoms through *ab initio* molecular dynamics simulations⁴⁸ according to the Car–Parrinello algorithm⁶⁰ with a timestep and a fictitious electron mass of 4 a.u. ≈ 0.1 fs and 400 a.u., respectively. In both classical and quantum simulations, the temperature was controlled by coupling each nuclear positional degree of freedom to a separate Nosé–Hoover-chain thermostat,⁶¹ thus implementing so-called “massive thermostating”, to ensure efficient sampling in particular for *ab initio* path integral simulations^{47,48,62} in order to capture nuclear quantum effects. In addition, the electronic degrees of freedom have been thermostatted using a separate Nosé–Hoover chain. For brevity, we call standard *ab initio* molecular dynamics using classical nuclei and *ab initio* path integral simulations using quantum nuclei “classical” and “quantum” simulations, respectively, throughout this work. Following a

large body of earlier work on nuclear quantum effects in molecular systems, we employed simulations using classical nuclei (either all of them or specific subsets to be defined below) to qualitatively assess the deuteration effects.

The path integral formalism of quantum statistical mechanics maps the many-body quantum system onto an equivalent extended classical system in which each quantum particle is represented by a classical cyclic polymer of P discretised Trotter (or imaginary-time) slices (or beads or replicas), s , where $s = P + 1 = 1$, in order to time-slice the full imaginary time interval $\tau \in [0, \hbar\beta]$ of these so-called Euclidean path integrals in terms of $s = 1, 2, \dots, P$ replicas of the fully interacting system.⁶³ The extended classical system corresponds exactly to the quantum system in the limit of $P \rightarrow \infty$. According to quantum statistical mechanics, the average of an operator B in the canonical ensemble is approximated as $\langle B \rangle = \text{Tr}(B e^{-\beta H}) / \text{Tr}(e^{-\beta H}) \approx \frac{1}{M} \sum_{k=1}^M \frac{1}{P} \sum_{s=1}^P B(s, t_k)$, which becomes exact as $P \rightarrow \infty$. Above, Tr is the trace of its argument, $\beta = 1/(k_B T)$ with T being the temperature and k_B Boltzmann’s constant, H denotes the Hamiltonian, and M denotes the number of sampled closed Feynman paths; as usual, the indistinguishability of identical particles and thus exchange are neglected in our path integral simulations. The “on the fly” combination of path integral sampling of nuclear quantum effects and concurrent electronic structure calculations of the electrons lead to the so-called *ab initio* path integral simulation technique,^{46–48,62,64,65} which is used here. Despite using molecular dynamics techniques to evaluate numerically the path integrals, time is fictitious, which implies that these simulations provide exclusively thermal averages of time-independent observables and, therefore, neither quantum dynamics nor physical timescales of processes. Such quantum simulations at 50, 100, 200, and 320 K used $P = 32, 16, 8$, and 5 imaginary time slices, respectively, to keep $\hbar\beta/P$ and, hence, the discretisation error along imaginary time roughly constant. In order to assess this level of Trotter discretisation, selected quantum simulations at 50 K using $P = 64$ Trotter slices were performed to investigate the convergence of observables with respect to P ; see ESI of ref. 43. The results for $P = 64$ and 32 beads were qualitatively identical and quantitatively very close so that 32 replicas were employed to keep the computational effort tractable. Classical reference simulations are obtained in the $P = 1$ limit and were performed at 50, 100, and 300 K to assess the influence of nuclear quantum effects.

The two partially deuterated systems at 50 K used one proton in the entire system, being located in the proton-ordered six ring, and one proton per six ring, respectively, with one imaginary time slice of these particles in order to mimic the effect of deuteration when compared to the fully protonated system. This approach closely follows earlier work on single-particle proton transfer along an intramolecular hydrogen bond.⁶⁶ Note that this leads formally to a hybrid QM/MM simulation at the level of the nuclei where the deuteration effect is approximated by invoking the classical limit of a subset of protons whereas all other nuclei are treated fully quantum mechanically throughout.



Proton transfer from or to a water molecule that is a member of the proton-ordered six ring to a hydrogen-bonded water molecule outside the ring would formally lead to the formation of an OH^- or H_3O^+ defect, respectively, within the six ring. Thus, the six ring would formally remain proton ordered, but it is expected that the creation of such a net charge defect within the ring would lead to strong structural distortions. It could be speculated that they are at least comparable to (but probably even stronger because more asymmetric than) the intrinsic charge defects that occur at high temperatures due to the proton hopping mechanism. Thus, it is expected that fluctuations in the hydrogen-bonded network that host the proton-ordered six ring leading to net (de)protonation of the six ring

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