

Cite this: *J. Mater. Chem. B*,  
2024, 12, 9566

## *In situ* biocatalytic ATP regulated, transient supramolecular polymerization†

Ananya Mishra, \*<sup>ab</sup> Angshuman Das<sup>a</sup> and Subi J. George \*<sup>a</sup>

Temporal control over self-assembly processes is a highly desirable attribute that is efficiently exhibited by biological systems, such as actin filaments. In nature, various proteins undergo enzymatically catalysed chemical reactions that kinetically govern their structural and functional properties. Consequently, any stimuli that can alter their reaction kinetics can lead to a change in their growth or decay profiles. This underscores the urgent need to investigate bioinspired, adaptable and controllable synthetic materials. Herein we intend to develop a general strategy for controlling the growth and decay of self-assembled systems *via* enzymatically coupled reactions. We achieve this by the coupling of enzymes phosphokinase/phosphatase with a bolaamphiphilic cationic chromophore (PDI) which selectively self-assembles with ATP and disassembles upon its enzymatic hydrolysis. The aggregation process is efficiently regulated by the controlled *in situ* generation of ATP, through enzymatic reactions. By carefully managing the ATP generating components, we realize precise control over the self-assembly process. Moreover, we also show self-assembled structures with programmed temporal decay profiles through coupled enzymatic reactions of ATP generation and hydrolysis, essentially rendering the process dissipative. This work introduces a novel strategy to generate a reaction-coupled one-dimensional nanostructure with controlled dimensions inspired by biological systems.

Received 16th July 2024,  
Accepted 17th August 2024

DOI: 10.1039/d4tb01558d

rsc.li/materials-b

## Introduction

Over the last few decades, research into the programmability and adaptivity of self-assembled polymers, with explicit structural and temporal control, is revolutionizing the field of supramolecular polymerization.<sup>1–3</sup> Traditionally, this field has relied on the passive self-assembly process,<sup>4–8</sup> but recent advances are steering it towards more active self-assembly techniques.<sup>9–11</sup> This shift enables the creation of highly programmable self-assembled structures, which hold the potential of constructing kinetically and structurally controlled supramolecular polymers with predetermined length and dispersity.<sup>12,13</sup> Additionally, these advanced techniques facilitate temporal programming resulting in supramolecular self-assembled structures achieved under non-equilibrium regimes.<sup>14–16</sup> The exceptional precision in the self-assembly process found in naturally occurring proteins has inspired supramolecular chemists to translate these properties to the synthetic realms. Among the various strategies

employed to achieve a temporal control over self-assembly, biocatalysis stands out as a particularly intriguing approach. Enzymatically controlled temporal self-assembly is ubiquitous in nature, directing essential processes such as cell motility, intracellular transport and maintaining the cell shape and structure.<sup>17–19</sup> Biocatalysis is known to trigger *in situ* temporal changes in the self-assembly process, a property that has been exploited to mimic the self-assembly of biomacromolecules such as the formation of collagen fibrils.<sup>20</sup> This has inspired researchers to investigate catalytic control over supramolecular polymerization which is crucial for combining structural and temporal control in synthetic systems.<sup>21</sup>

Recent approaches have leveraged biocatalysis to obtain temporal control over supramolecular polymerization, which typically follows a nucleation growth mechanism to induce molecular order.<sup>22</sup> In other examples, *in situ* hydrogel formation has also been explored through the integration of enzymatic networks with small molecules.<sup>23</sup> In most of these examples, carefully designed peptide chains have been utilized, which undergo enzymatically triggered covalent bond cleaving or formation, initiating the self-assembly process. However, the next challenge in this field is to realize a universal approach for the predictive design of a controlled supramolecular polymerization process. Recent attempts have managed to achieve control over the structure<sup>24–30</sup> and temporal<sup>31–34</sup> aspects of the self-assembly process but often in separate systems.

<sup>a</sup> Supramolecular Chemistry Laboratory, New Chemistry Unit and School of Advanced Materials (SAMat), Jawaharlal Nehru Centre for Advanced Scientific Research (JNCASR), Jakkur, Bangalore, 560064, India. E-mail: george@jncasr.ac.in

<sup>b</sup> Centre for Protolife Research, Centre for Organized Matter Chemistry, School of Chemistry, University of Bristol, Bristol BS81TS, UK.  
E-mail: ananya.mishra@bristol.ac.uk

† Electronic supplementary information (ESI) available. See DOI: <https://doi.org/10.1039/d4tb01558d>



Drawing inspiration from biological programming, designing reaction-driven assemblies provides a straightforward strategy to achieve desirable control over supramolecular polymerization.<sup>35</sup> With that idea, adenosine triphosphate (ATP) has recently been employed in multiple examples to provide a templating effect to the self-assembly process of supramolecular systems.<sup>36–39</sup> While ATP has most commonly been used for templated transient assembly formation,<sup>40–42</sup> there have been some reports towards the structural modulation of self-assembled systems with length dispersity control.<sup>43,44</sup> Our recent work on the nucleation–elongation followed by seeded self-assembly facilitated by the structural modification of the self-assembling unit exhibits the first example of ATP triggered living supramolecular polymerization.<sup>43</sup> However reaction regulated structural and temporal control similar to cytoskeletal self-assembly has so far not been achieved.

An interesting solution to this is to employ aggregating systems that can be controlled by biocatalysis, offering a more generalized strategy for constructing programmable supramolecular polymers. In a recent report from our group, we demonstrated the coupled nature of the enzymatic reaction with the growth of supramolecular polymers to realize structural control on the self-assembly.<sup>44</sup> However, the coupling of antagonistic enzymatic reactions for controlled and transient supramolecular polymerization akin to the tandem functioning of cofilin and profilin in the treadmilling process of actin assembly<sup>45</sup> by controlling ATP generation and decay rate remains unexplored. Herein we demonstrate a bioinspired strategy to achieve programmable supramolecular polymerization where an adenosine triphosphate (ATP) regulating, coupled enzymatic reaction network has been employed. We could successfully induce controlled supramolecular polymerization of a bolaamphiphilic dicationic molecule, perylene diimide functionalized with cationic trimethyl ammonium groups linked *via* ethylene spacers at the imide positions (**PDI**).<sup>46</sup> This molecule is shown to undergo a time-dependent self-assembly when ATP was generated enzymatically from the precursor adenosine diphosphate (ADP) followed by a disassembly on enzymatic hydrolysis of ATP by employing the phosphokinase/phosphatase cycle. The resultant coupling of both enzymatic reactions resulted in the formation of transient assemblies of **PDI**. Taken together, this work establishes a facile method to create adaptive and active supramolecular systems driven by ATP regulation (Scheme 1).

## Results and discussion

### ATP selective self-assembly of **PDI**

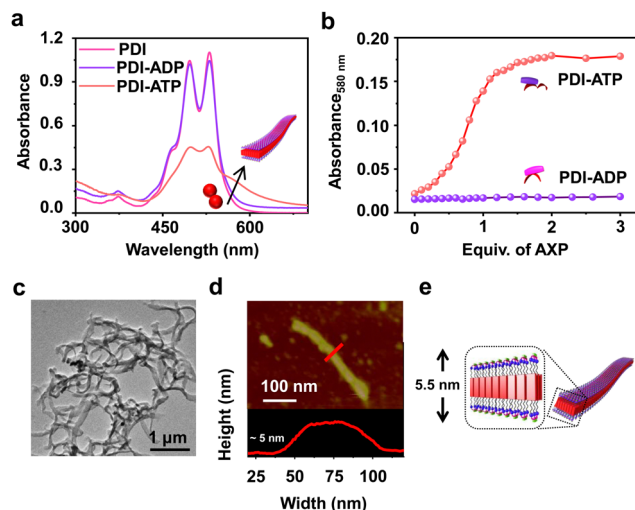
To realize bioinspired, enzymatically controlled supramolecular polymerization, we selected ATP as the enzyme-responsive template molecule capable of inducing integrative dynamic assembly. First, we investigated ATP selectivity in the supramolecular polymerization process of **PDI**. Due to its bolaamphiphilic nature, **PDI** ( $5 \times 10^{-5}$  M) remained in a monomeric state in 100% DMSO. However, as the percentage of HEPES



**Scheme 1** Biocatalytic ATP regulated temporal assembly of **PDI**. Molecular structure and schematic representation of (a) dicationic perylene diimide (**PDI**) and (b) enzymatic adenosine triphosphate (ATP) generation from adenosine diphosphate (ADP) in the presence of creatine phosphokinase (CPK) and phosphocreatine (PCr). This releases creatine (Cr) as a by-product. Enzymatic ATP hydrolysis to adenosine and  $\text{PO}_4^{3-}$  is triggered in the presence of calf intestinal alkaline phosphatase (CIAP). (c) Schematic representation of the overall biocatalytic controlled growth and seeding of ATP bound **PDI** through *in situ* generation of ATP followed by transient characteristics achieved through coupled enzymatic reactions and ATP regulation.

increased in the HEPES/DMSO mixture, **PDI** underwent aggregation, indicated by a reversal in the absorbance of vibronic bands ( $A_{492}/A_{530}$  from 0.64 to 1.48 for 100% DMSO and 100% HEPES, respectively) and the appearance of a new aggregate band at 580 nm (Fig. S1, ESI<sup>†</sup>). Therefore, we selected a 50/50 HEPES/DMSO (v/v) mixture for further experiments where **PDI** formed small nano-aggregates due to randomly arranged molecules that self-assemble through electrostatic interaction with ATP (Fig. 1 and Fig. S2, ESI<sup>†</sup>). As the equivalents of ATP were increased in the  $5 \times 10^{-5}$  M **PDI** solution (50% HEPES/DMSO mixture), a gradual increase in the aggregate band at  $\lambda_{580\text{nm}}$ ,





**Fig. 1** ATP selective self-assembly of **PDI**. (a) Comparative absorption spectra of **PDI** (pink), **PDI-ADP** (purple) and **PDI-ATP** (orange). (b) Comparative titration curves of **PDI** obtained from absorption ( $\lambda_{580\text{nm}}$ ) with increasing equiv. of ATP (red curve) and ADP (purple curve). (c) TEM image (stained with uranyl acetate, 1 wt% in water) of **PDI** with 2 equiv. of ATP. (d) AFM height image along with a height profile of 2 equiv. of ATP bound **PDI**. The red line corresponds to the region from where the height profile has been extracted. (e) Schematic representation of the cross-section of ATP bound **PDI** tapes. ( $[\text{PDI}] = 5 \times 10^{-5} \text{ M}$ ,  $\text{H}_2\text{O}$  or HEPES/DMSO, 50/50, (v/v)).

along with broadening of the absorption spectra, suggested ATP-templated self-assembly of **PDI**. Furthermore, the chiral induction to the assembly as evident in the CD spectra from the chiral ribose sugar and quenched emission corroborated this observation (Fig. S2–S5 and Table S1, ESI<sup>†</sup>). Detailed spectroscopic characterization of ATP-bound **PDI** indicated the formation of H-type aggregates resulting from  $\pi$ - $\pi$  stacking of the chromophores. The titration plot obtained by monitoring the absorbance at 580 nm followed a sigmoidal profile that saturated at 2.0 equivalents of ATP, suggesting 1:1 binding of **PDI** with ATP with an association constant ( $K_a$ ) of  $4.21 \times 10^5 \text{ M}^{-1}$  (Fig. 1b and Fig. S2 and S3, ESI<sup>†</sup>). In contrast, titrating **PDI** with increasing equivalents of ADP showed negligible changes in the absorption spectra (at 580 nm), indicating ATP-selective aggregation of **PDI** (Fig. 1a and b and Fig. S2d–f, ESI<sup>†</sup>). However, the slight quenching in the corresponding emission spectra suggests the weak binding of ADP with **PDI** (Fig. S2e and f, ESI<sup>†</sup>). The ATP-selective self-assembly of **PDI** was further evident from the extent of chiral induction in the **PDI-ATP/ADP** stack (Fig. S4, ESI<sup>†</sup>). The CD titration plot showed a CD silent state up to 0.8 equivalents of ATP, beyond which the CD signal increased gradually, saturating around 2.0 equivalents of ATP due to allosteric regulation. Conversely, no chiral induction in the CD signal was observed with increasing equivalents of ADP. From these spectroscopic experiments, we concluded that **PDI** exhibits ATP selectivity in self-assembly. For further experiments, we used 2.0 equivalents of ATP to maximize the extent of **PDI** self-assembly.

Furthermore, the ATP-selective self-assembly was confirmed by dynamic light scattering (DLS) experiments, which showed

an increase in size for ATP compared to ADP (Fig. S6a, ESI<sup>†</sup>). Transmission electron microscopy (TEM) images revealed the presence of nanoparticle-like assemblies ( $\sim 100 \pm 5 \text{ nm}$ ) of **PDI** in an  $\text{H}_2\text{O}/\text{DMSO}$  50/50 (v/v) mixture (Fig. S6b, ESI<sup>†</sup>). These assemblies transformed into 1D nano-tape-like structures with polydisperse lengths ( $> 5 \mu\text{m}$ ) and  $\sim 60 (\pm 10) \text{ nm}$  breadth upon interaction with 2.0 equivalents of ATP (Fig. 1c and d and Fig. S6d, ESI<sup>†</sup>). Atomic force microscopy (AFM) height analysis showed the height of the tape to be  $\sim 5 (\pm 0.3) \text{ nm}$ , obtained by tracing an individual tape across the breadth which matches the molecular dimensions of **PDI** bonded with ATP at both ends (Fig. 1e). This suggests that **PDI** molecules self-assemble through  $\pi$ - $\pi$  stacking along the tape's length, with hydrophobic interactions between the chromophores along the breadth where ATP is bound to the trimethylammonium cation exposed to the hydrophilic environment. Combined spectroscopic and morphological experiments indicate effective binding and self-assembly of **PDI** in the presence of trivalent ATP, resulting in elongated and entangled tape-like structures. In contrast, divalent ADP does not induce self-assembly of **PDI**, even though it binds to the molecule (Fig. S2 and S6c, ESI<sup>†</sup>). The uncontrolled elongation of the nanotape was attributed to the instantaneous self-assembly of **PDI** upon binding with 2 equivalents of ATP, as observed in time-dependent spectroscopic changes (Fig. S7, ESI<sup>†</sup>). We noticed an immediate increase in absorbance (at  $\lambda_{580} \text{ nm}$ ) and emission (at  $\lambda_{630} \text{ nm}$ ), indicating the spontaneous formation of ATP-bound **PDI** assemblies (Fig. 2a red plot and Fig. S7a, ESI<sup>†</sup>). To induce controlled self-assembly of **PDI**, we employed the biocatalytic *in situ* generation of ATP and further investigated the enzymatic modulation of temporal self-assembly of **PDI** (*vide infra*).

### Enzymatic *in situ* generation of ATP for controlled self-assembly of **PDI**

Having observed the uncontrolled growth of **PDI** with the direct addition of ATP, we next controlled the elongation process by *in situ* generation of ATP from ADP in the presence of phosphocreatine (PCr) as a substrate, using the creatine phosphokinase (CPK) enzyme (Scheme 1). The ATP selectivity facilitated the successful use of this enzymatic reaction for controlling supramolecular polymerization, where the CPK enzyme transfers the  $\text{PO}_4^{3-}$  group from PCr to ADP, generating ATP and releasing creatine (Cr) as a by-product. Thereby the ATP produced over time slowly induced the growth of **PDI-ATP** tapes. To a  $5 \times 10^{-5} \text{ M}$  solution of **PDI** in DMSO/HEPES 50/50 (v/v) containing  $15.0 \text{ U mL}^{-1}$  CPK and 10.0 equivalents of PCr, 2.0 equivalents of ADP were injected to initiate the biocatalytic reaction, generating ATP *in situ*. As a result of controlled ATP generation by CPK, kinetically controlled growth was observed, monitored at the 580 nm band in UV-vis spectroscopy, which saturated at 3000 s (blue plot, Fig. 2a), whereas emission showed quenching over time monitored at  $\lambda_{630\text{nm}}$  (Fig. S8, ESI<sup>†</sup>). This kinetic profile contrasts with the sudden jump observed for the direct addition of ATP to the **PDI** solution (Fig. S7a, ESI<sup>†</sup>), though the extent of aggregation for enzymatically produced 2.0 equivalents of ATP (from an equivalent





**Fig. 2** Biocatalytic generation of ATP driven temporally controlled self-assembly of **PDI**. (a) Comparative growth kinetics showing instantaneous self-assembly of **PDI** on addition of 2 equiv. of ATP from time dependent absorption changes ( $\lambda_{580\text{nm}}$ , red) and time dependent spectroscopic evolution of ATP-driven self-assembly of **PDI** as ADP enzymatically converts to ATP in the presence of CPK monitored by absorption ( $\lambda_{580\text{nm}}$ , blue) and scattering from DLS (green) and CD ( $\lambda_{500\text{nm}}$ , magenta). Schematic representation of ATP triggered and enzymatically generated ATP driven **PDI** self-assembly has also been shown. Time dependent absorption changes monitored at  $\lambda_{580\text{nm}}$  for enzymatically grown ATP bound **PDI** (b) for varying concentrations of ADP (CPK = 15 U mL<sup>-1</sup>, PCr = 10 equiv.), (c) for varying concentrations of PCr (ADP = 2 equiv., CPK = 15 U mL<sup>-1</sup>) and (d) for varying concentrations of CPK (ADP = 2 equiv., PCr = 10 equiv.). (e) TEM, (f) AFM and (g) FESEM images of enzymatically grown tapes of **PDI** (**[PDI]** =  $5 \times 10^{-5}$  M, ADP = 2 equiv., CPK = 15 U mL<sup>-1</sup>, PCr = 10 equiv., H<sub>2</sub>O or HEPES/DMSO, 50/50, (v/v), 25 °C).

amount of ADP) was less compared to the direct addition of 2.0 equivalents of ATP (Fig. S9, ESI†).

A similar kinetic profile was observed in DLS, showing an increase in scattering over time due to increased aggregation (green plot, Fig. 2a). However, the corresponding time-dependent change in CD signal followed sigmoidal kinetics with a 500 s lag phase (monitored at 500 nm, magenta plot, Fig. 2a), differing from the absorbance kinetics (blue plot, Fig. 2a). This suggests that supramolecular chirality induction sets in only after certain extent of growth of self-assembly. Interestingly, the enzymatically grown ATP-bound **PDI** assembly showed a narrower size distribution compared to **PDI** assemblies grown instantaneously, highlighting the importance of the controlled self-assembly process regulated by the enzymatic reaction (Fig. S8e, ESI†).

Further evidence of reaction-controlled growth was observed when the growth was influenced by various components

present in the enzymatic reaction, such as ADP, PCr, and CPK (Fig. 2b–d and Fig. S10–S12, ESI†). First, we varied the concentrations of ADP and monitored the changes spectroscopically. Since ADP enzymatically generates equivalent and quantitative amounts of ATP, increasing the concentration of ADP (0.5–2.5 equivalents) resulted in an increased self-assembly process. This was evidenced by a decrease in  $t_{50}$  values from 2700 ( $\pm 400$ ) s to 1436 ( $\pm 150$ ) s and an increase in  $\nu_{\text{max}}$  from 1.01 ( $\pm 0.5$ ) to 5.41 ( $\pm 0.2$ ) s<sup>-1</sup>, monitored through absorbance changes at 580 nm, due to faster production and higher concentration of ATP in solution (Fig. S11a, ESI† and Fig. 2b). Analysis through DLS showed a gradual increase in size from 290 ( $\pm 75$ ) nm to 2115 ( $\pm 94$ ) nm with increasing equivalents of ADP, with a linear increase characteristic of controlled supramolecular polymerization which saturated at 2 equiv. of ADP (Fig. S11, ESI†).

To further confirm the effects of enzymatically produced ATP on the self-assembly kinetics of **PDI**, we varied the concentrations of PCr and CPK and carefully monitored their effects on the resulting elongation kinetics of ATP-bound **PDI**. With increasing PCr concentrations from 5.0 to 10.0 equivalents at fixed ADP (2.0 equivalents) and CPK (15.0 U mL<sup>-1</sup>) in the **PDI** solution ( $5 \times 10^{-5}$  M), an increase in the growth rate ( $\nu_{\text{max}}$ ) was observed from 1.01 ( $\pm 0.3$ ) s<sup>-1</sup> to 5.12 ( $\pm 0.12$ ) s<sup>-1</sup> (Fig. 2c). Since PCr is the PO<sub>4</sub><sup>3-</sup>-generating substrate that enzymatically converts ADP to ATP, higher equivalents of PCr generate higher concentrations of PO<sub>4</sub><sup>3-</sup> quickly, thereby increasing the growth rate. These results are corroborated by the decrease in  $t_{50}$  from 3750 ( $\pm 250$ ) s to 1501 ( $\pm 150$ ) s (Fig. S12a, ESI†). Similarly, with increasing concentrations of CPK, a decrease in the total time, required for the completion of the growth process from, 14 000 s to 3000 s was observed, although the extent of growth remained similar (Fig. 2d). These observations were further reflected in the derived time parameter  $t_{50}$  values from 6667 ( $\pm 450$ ) s to 1359 ( $\pm 150$ ) s and rate of elongation  $\nu_{\text{max}}$  values from 0.97 ( $\pm 0.2$ ) to 5.01 ( $\pm 0.3$ ) s<sup>-1</sup> (Fig. S12b, ESI†). All these outcomes confirm that the rate of *in situ* ATP generation directly governs the kinetics of **PDI** elongation. The enzymatically grown ATP-bound **PDI** assemblies were further characterized in detail using various morphological techniques. TEM, FESEM, and AFM images of the enzymatically generated **PDI**-ATP assemblies showed the formation of 1-D nanotapes with uniform dimensions (length  $\sim 1 \pm 0.1$   $\mu\text{m}$ ) with a dispersity value of 0.02 obtained directly from DLS (Fig. 2e–g). Next, the enzymatic reaction-controlled generation of ATP was utilized to control the length of the **PDI**-ATP supramolecular polymer through a seeded approach, which is a characteristic feature of actin assembly (*vide infra*).

### Seeding *via* stepwise addition of ADP

Having demonstrated the enzymatic reaction-driven controlled formation of ATP-bound **PDI** nanotapes, we proceeded to control the degree of polymerization through a seeded assembly akin to actin polymerization (Fig. 3). In the previous section, we observed that the extent of tape growth could be controlled by the concentration of the substrate, ADP in the solution. Herein, we strategized the seeding process by stepwise addition





**Fig. 3** Seeded self-assembly of **PDI** by batch-wise addition of ADP. (a) Schematic representation of stepwise addition of ADP giving rise to enzymatically driven 1-D seeding on tapes where initially 0.5 equiv. of ADP is introduced to grow **PDI**-ATP followed by a second batch of 0.5, 1.0 or 1.5 equiv. of ADP addition. (b) Absorbance changes monitored at  $\lambda_{580\text{nm}}$  for multi-batch addition of 0.4 equiv. of ADP. (c) Absorbance changes monitored at  $\lambda_{580\text{nm}}$  for stepwise addition of ADP keeping the initial batch concentration of ADP at 0.5 equiv. and varying the second batch addition (numbers indicate the equiv. of ADP added in each batch). (d) Size changes obtained from DLS (pink) and dispersity from DLS for seed and seeded enzymatic generation of ATP leading to elongation of **PDI** by stepwise addition of ADP. Error bars are obtained from triplicate experiments. (e) TEM images obtained for two seeded growths of **PDI**-ATP through enzymatic generation of ATP (left: 0.5 + 0.5 equiv. of ADP, right: 0.5 + 1.5 equiv. of ADP) where ADP was introduced batchwise. Arrows indicate the addition of ADP into the solution ( $[\text{PDI}] = 5 \times 10^{-5} \text{ M}$ ,  $[\text{CPK}] = 15 \text{ U mL}^{-1}$ ,  $[\text{PCr}] = 10 \text{ equiv.}$ , HEPES/DMSO, 50/50 (v/v), 25 °C).

of ADP into the same solution, differing from the conventional seeding process where a fresh batch of monomer is added each time into the seed solution. As a result, the supramolecular polymer formed by the first batch of ADP addition acts as a seed for the second batch of ADP addition into the same **PDI** solution containing fixed CPK and PCr concentrations (Fig. 3a).

Since all binding sites of **PDI** ( $5 \times 10^{-5} \text{ M}$ ) were saturated with 2.0 equivalents of ADP, we added ADP in two sequential batches, keeping the total concentration of ADP fixed at 2.0 equivalents while maintaining a fixed  $15.0 \text{ U mL}^{-1}$  CPK and  $10.0$  equivalents of PCr (Fig. S13, ESI<sup>†</sup>). The seeded assembly was evident from the increased growth rate of **PDI** in the presence of the second batch of ADP (1.0 equivalent) compared to the first batch (1.0 equivalent), as monitored by absorbance at  $\lambda_{580\text{nm}}$  (Fig. S13c, ESI<sup>†</sup>). This was further supported by the increase in the extent of aggregation after each batch of ADP addition and the corresponding increase in size in the DLS. Although the nucleation step of **PDI**-ATP was not

evident from the absorption changes, as indicated by the absence of a lag phase in the initial growth of the seed, the seeded assembly from seed termini was evident from the increased growth rate for the second batch of ADP addition and the nonexistence of a bimodal distribution in DLS (Fig. S13b and c, ESI<sup>†</sup>).

We believe that the difference in the degree of polymerization during subsequent additions is a manifestation of *in situ* ATP generation being uniquely linked to seeded elongation characteristics. Since the subsequent additions of ADP are carried out in the presence of a preformed **PDI**-ATP seed, a faster elongation was observed due to the rapid generation of ATP in the solution. It has been previously reported by us that ATP-bound stacks tend to attract CPK, which can further facilitate positive feedback in the overall ATP generation and **PDI** stack seeded elongation process.<sup>44</sup> The efficiency of the seeded assembly was further demonstrated by multicycle seeding, performed by adding 0.4 equivalents of ADP five times (multiple batch additions), keeping the total concentration of ADP fixed at 2.0 equivalents where we noticed seeded behaviour when monitored by absorbance at  $\lambda_{580\text{nm}}$  (Fig. 3b).

A similar phenomenon was observed when the second batch of ADP addition was varied from 0.5 to 1.5 equivalents, keeping the initial batch of ADP constant at 0.5 equivalents (Fig. 3c). Interestingly, we noticed an increase in the growth rate along with an increase in the extent of aggregation due to the increased ADP equivalents in the second batch, indicating the seeded assembly of enzymatically controlled ATP-bound **PDI** (Fig. 3c). The increase in size from a seed to a seeded assembly was observed from scattering measurements for stepwise addition of ADP with varied equivalents (0.5, 1.0, and 1.5 equivalents) at the second step. This reiterates the seeded characteristic of **PDI**, which follows a linear fit with a good quality correlation curve (Fig. 3d, pink line and Fig. S14, ESI<sup>†</sup>). Moreover, the dispersity trends obtained directly from DLS indicate a controlled supramolecular polymerization process, as the values remain constant during the seeding experiments (Fig. 3d, green line). Although **PDI**-ATP structures are not spherical and the exact values obtained from DLS may not be accurate, they depict a qualitative trend in size increase during seeding experiments.

An increase in size was also observed when the seeded structures were visualized under an electron microscope (TEM), where the length of the **PDI**-ATP stack increased from  $250 (\pm 50) \text{ nm}$  to  $1100 (\pm 160) \text{ nm}$  with the increased ADP amount in subsequent batches with a polydispersity index close to 1.0 (Fig. 3e, S15 and S16, ESI<sup>†</sup>). This enzymatic reaction-driven, *in situ* generated ATP-triggered seeded supramolecular polymerization strategy provided controlled stack lengths, as evident from the low and consistent dispersity values, analogous to recently reported living supramolecular polymerization processes.<sup>43</sup>

### Enzymatic-reaction controlled transient assembly

In addition to the reaction-controlled seeded assembly, another crucial aspect of the natural self-assembly process of actin is the temporal regulation of the structure, which governs various



cellular functions. With an aim to mimic this, we formed a transient self-assembly of **PDI** using two antagonistic enzymes: CPK, which controls the growth by producing ATP in solution, and calf intestinal alkaline phosphatase (CIAP), which hydrolyses ATP to adenosine and phosphate, working in tandem (Fig. 4a). As shown previously, the self-assembly of **PDI** is selective to ATP. We envision that controlled enzymatic hydrolysis of ATP to phosphate will induce depolymerization of the tapes. First, we carried out control experiments where CIAP was introduced into a pre-grown ATP-bound **PDI** assembly, and we monitored the changes over time both spectroscopically and through scattering. We observed a decrease in the aggregation band over time, as indicated by the absorbance at 580 nm as well as the temporal changes in size obtained from DLS

(Fig. S17, ESI<sup>†</sup>). Interestingly, the supramolecular polymer disassembled at a faster rate with increasing CIAP concentration (Fig. S17a, ESI<sup>†</sup>). This indicates that enzymatic hydrolysis of ATP induces passive disassembly of **PDI** tapes.

To extend this passive enzyme response of the ATP-bound **PDI** assembly to active regimes, a careful standardization of the concentrations of both ATP-generating enzyme (CPK) and ATP-hydrolyzing enzyme (CIAP) is required to achieve significant growth of the assembly before the disassembly process dominates. This can be achieved by increasing the CPK concentration to 20.0 U mL<sup>-1</sup>, where a fast *in situ* generation of ATP would accelerate the growth process. Thus, the assembly reaches its maximum before disassembly begins due to enzymatic hydrolysis of ATP. By coupling 20.0 U mL<sup>-1</sup> of CPK with lower concentrations of CIAP (0.2–0.4 U mL<sup>-1</sup>), we could regulate the production and consumption of biofuel ATP *in situ*, rendering a transient self-assembly of **PDI**-ATP, as evidenced by the absorbance changes due to the differences in ATP generation and ATP hydrolysis rates. Interestingly, we noticed a decrease in the extent of growth, a shorter duration of the steady state, and faster disassembly when increasing the concentration of CIAP from 0.2 U mL<sup>-1</sup> to 0.4 U mL<sup>-1</sup> (Fig. 4b). The steady state could be due to the delay in disassembly of stable ATP-bound **PDI** assemblies due to the differential hydrolysis rate of free ATP in solution or the ATP bound to the **PDI**-ATP stack by CIAP enzyme. As a result, this prolongs the stability of **PDI**-ATP before initiating its disassembly (Fig. S18, ESI<sup>†</sup>).

We further analysed and quantified the assembly and disassembly rates ( $dA/dt$ ) and the lifetimes of assembly, steady state, and disassembly from the spectroscopic traces obtained from absorption changes monitored at  $\lambda_{580}$  nm. We noticed that the assembly rate remains constant ( $\sim 3.2 \times 10^{-4}$  min<sup>-1</sup>), but its lifetime decreases from 160 to 96 min with increasing concentrations of CIAP, whereas the disassembly rate increases from  $7.97 \times 10^{-5}$  to  $12.5 \times 10^{-5}$  min<sup>-1</sup> along with its lifetime (Table S2, ESI<sup>†</sup>). These observations result from faster hydrolysis kinetics with higher concentrations of CIAP. Higher units of CIAP also decrease the extent of growth of ATP-bound **PDI** during the assembly process. Moreover, introducing ADP again into the solution led to a second transient cycle (red trace, Fig. 4c).

These changes were further corroborated by monitoring scattering changes, which indicate the temporal increase in size (elongation), followed by a steady state and then a decrease (disassembly) when monitored through DLS (blue trace, Fig. 4c). These results confirm the enzymatic regulation of ATP-driven transient self-assembly of **PDI**. Time-dependent spectra taken for the transient assembly of ATP-bound **PDI**, as seen from the emergence of the aggregate band, whereas disassembly can be observed from the disappearance of the aggregate band over time (Fig. S19, ESI<sup>†</sup>). The initial spectra of assembly ( $t_{\text{initial}}$ , assembly) and the final spectra after the completion of disassembly ( $t_{\text{final}}$ , disassembly) could be overlaid, indicating a return to the same state. Furthermore, the transient



**Fig. 4** Enzyme-coupled transient supramolecular polymerisation of **PDI**-ATP. (a) Schematic representation of active self-assembly of ATP bound **PDI** where coupling of an ATP generating (CPK) and hydrolyzing (CIAP) enzyme regulates ATP in solution, rendering a transient self-assembly. (b) Transient assembly of **PDI** by coupling of CPK (constant) and CIAP (varying) as evident from the time-dependent absorbance changes monitored at  $\lambda_{580\text{nm}}$ . (c) Co-stacked absorbance ( $\lambda_{580\text{nm}}$ ) and scattering changes (CIAP = 0.4 U mL<sup>-1</sup>) showing multiple transient assembly cycles indicating growth followed by disassembly of ATP bound **PDI** on refuelling ADP into the solution. Arrows indicate the addition of ADP into the solution. TEM images showing the transient assembly of **PDI**-ATP at (d)  $t_{1\text{h}}$  (steady state) and (e)  $t_{4\text{h}}$  (after disassembly) (CIAP = 0.2 U mL<sup>-1</sup>) ([**PDI**] =  $5 \times 10^{-5}$  M, ADP = 2 equiv., CPK = 20 U mL<sup>-1</sup>, PCr = 10 equiv., HEPES/DMSO, 50/50, v/v, 25 °C).



self-assembly of the **PDI-ATP** assemblies was visualised at different time points by TEM, exhibiting the presence of elongated tapes at the end of 1 hour of the transient cycle, corresponding to the beginning of the steady state, and the absence of any elongated tapes at the end of 5 hours (Fig. 4d and e). Thus, we could achieve transient states of **PDI-ATP** through temporal control over the self-assembly followed by disassembly by coupling the enzymes CPK and CIAP.

## Conclusions

Herein, we present a bioinspired approach to achieve programmable supramolecular polymerization, mimicking reaction-controlled assembly, modulation of length through a seeded approach, and temporal control over the self-assembly process akin to biological systems. Drawing inspiration from the actin self-assembly process, we coupled ATP-generating and ATP-hydrolyzing enzymes, CPK and CIAP, respectively, within the same system. This coupling allowed for *in situ* ATP regulation, leading to the formation of transient assemblies. By adjusting the concentrations of molecular cues influencing the self-assembly process, we efficiently modulated the extent and kinetics of the assemblies. Additionally, we demonstrated the interdependency between the enzymatic reactions and the growth of the supramolecular polymer **PDI-ATP**. These enzymatic reaction-driven assemblies also exhibited the capability for seeded supramolecular polymerization, allowing for precise control over their length and dispersity. The enzymatic coupling of self-assembly and disassembly processes in **PDI-ATP** systems resulted in the formation of aggregates in out-of-equilibrium regimes. The development of these strategies holds the potential to significantly impact applications targeted by self-assembled structures, paving the way for the creation of next-generation materials with unprecedented control over their structural and functional properties.

## Materials and methods

### General

All chemicals were purchased from commercial sources and were used as such. ATP, ADP, PCr, CPK and CIAP were purchased from Sigma Aldrich. Spectroscopic grade solvents were used for all optical measurements.

### Spectroscopic measurements

Electronic absorption spectra were recorded on a Perkin Elmer Lambda 900 UV-vis-NIR Spectrometer and emission spectra were recorded on a Perkin Elmer Ls 55 Luminescence Spectrometer. Circular dichroism measurements were performed on a Jasco J-815 spectrometer. UV-vis, emission and CD spectra were recorded in a 5 mm path length quartz cuvette. Corresponding temperature dependent measurements were performed with a CDF-426S/15 Peltier-type temperature controller.

### Dynamic light scattering (DLS) experiments

The measurements were carried out using a NanoZS (Malvern UK) employing a 635 nm laser at a backscattering angle of 173°. The samples were measured in a 10 mm glass cuvette.

### Transmission electron microscopy (TEM)

TEM measurements were performed on a JEOL, JEM 3010 operated at 300 kV and a JOEL JEM2100 PLUS operated at 200 kV. Samples were prepared by placing a drop of the solution on carbon coated copper grids followed by drying at room temperature. The images were recorded at operating voltages of 300 kV and 200 kV depending on the instrument used. In order to get a better contrast, the sample was stained with uranyl acetate (1 wt% in water) before the measurements. For TEM, water was used instead of aq. HEPES solution to avoid masking of nanostructures due to HEPES deposition upon drying.

### Atomic force microscopy (AFM)

AFM was analysed using a Bruker Dimension Icon and the imaging was carried out in peak force tapping mode. Samples were prepared by spin coating a 2  $\mu$ L solution of the sample on silicon. Imaging was carried out under ambient conditions in tapping mode. The probe used for imaging was antimony doped silicon cantilever with a resonant frequency of 300 kHz and a spring constant of 40 N m<sup>-1</sup>. The software used for the analysis of the nanostructures was Nanoscope V7.30r1sr3. Water was used instead of aq. HEPES solution to avoid masking of nanostructures due to HEPES deposition upon drying.

### Field emission scanning electron microscopy (FESEM)

FESEM measurements were carried out on a ZEISS GeminiSEM 500. Samples were prepared by spin coating a 2  $\mu$ L solution of the sample on silicon. Water was used instead of aq. HEPES solution to avoid masking of nanostructures due to HEPES deposition upon drying.

**PDI** synthesis and characterization were carried out according to the literature procedure.<sup>47</sup>

### Sample preparation for self-assembly experiments of **PDI**

All samples for spectroscopic measurements were prepared by injecting the required volume of **PDI** (stock solution 10<sup>-3</sup> M in DMSO) into *N*-2-hydroxyethylpiperazine-*N'*-2-ethanesulphonic acid (HEPES) buffer solution (10 mM). The stock concentrations of ATP, ADP, and PCr were kept at 10<sup>-2</sup> M by dissolving the measured solid in appropriate amounts of HEPES. Aliquots of stock solution of CPK (50 U mL<sup>-1</sup>) were used for every measurement.

### Measurements for enzymatically triggered temporal self-assembly of **PDI**

For spectroscopic measurements of enzymatically triggered self-assembly of **PDI**, a 5  $\times$  10<sup>-5</sup> M solution of **PDI** in DMSO/HEPES 50/50 (v/v) is taken in a cuvette, to which the required components (PCr and CPK) are added, mixed by shaking and



the measurement is started within 30 seconds. ADP/ATP was introduced into the mixture in the end as the final component to trigger the temporal growth of PDI. All the measurements are carried out at 25 °C in cuvettes with a path length of 10 mm.

### Seeded self-assembly experiments of PDI

First, the seed solution was prepared by introducing  $x$  equiv. of ADP ( $x = 0.2, 0.5, 1.0$  or  $1.5$ ) into the reaction mixture in a cuvette containing the required amounts of PDI ( $5 \times 10^{-5}$  M), PCr (10 equiv.) and CPK ( $15 \text{ U mL}^{-1}$ ). The initial growth kinetics of PDI was monitored through absorption changes at  $\lambda_{580\text{nm}}$ . Subsequent batches of ADP ( $y = 0.2, 0.5, 1.0$  or  $1.5$  equiv.) were introduced into the preformed seed solution and the time dependent changes were monitored.

### Enzymatic ATP regulation leading to transient assembly of PDI

Both the enzymes, CPK ( $20 \text{ U mL}^{-1}$ ) and CIAP ( $0.2, 0.3$  or  $0.4 \text{ U mL}^{-1}$ ), are introduced into a solution containing the required amounts of PDI ( $5 \times 10^{-5}$  M). Measured volumes of the preformed mixture of PCr (10 equiv.) and ADP (2 equiv.) are introduced into the solution and the temporal measurements are initiated and monitored with absorbance at  $\lambda_{580\text{nm}}$  or through scattering in DLS. Refuelling was carried out by introducing a fresh feed of PCr (10 equiv.) and ADP (2 equiv.) into the solution and the changes were continued to be monitored.

## Author contributions

A. M. and S. J. G. conceived the project. A. M. designed and performed the experiments. A. M., A. D. and S. J. G. discussed the data and wrote the manuscript. All authors have given approval to the final version of the manuscript.

## Data availability

The authors confirm that the data supporting the findings of this study are included within the manuscript and its ESI.† Additionally, the raw data can be made available upon reasonable request.

## Conflicts of interest

There are no conflicts to declare.

## Acknowledgements

We thank JNCASR, Department of Science and Technology (DST), Government of India. The authors acknowledge SRISTI-BIRAC for funding provided through GYTI awards. A. M. thanks DST for the research fellowship. S. J. G. acknowledges the funding received from the Swarna Jayanti Fellowship award (DST/SJF/CSA01/2016-2017). A. D. thanks the CSIR for the fellowship. We thank SAMat research facilities and JNCASR for TEM experiments.

## Notes and references

- U. Alon, *An Introduction to Systems Biology: Design Principles of Biological Circuits*, Chapman and Hall Publishers, 2006.
- R. F. Service, *Science*, 2005, **309**, 95.
- G. M. Whitesides and B. Grzybowski, *Science*, 2002, **295**, 2418–2421.
- L. Brunsveld, B. J. Folmer, E. W. Meijer and R. P. Sijbesma, *Chem. Rev.*, 2001, **101**, 4071–4098.
- F. J. M. Hoeben, P. Jonkheijm, E. W. Meijer and A. P. H. J. Schenning, *Chem. Rev.*, 2005, **105**, 1491–1546.
- T. Aida, E. W. Meijer and S. I. Stupp, *Science*, 2012, **335**, 813–817.
- T. F. A. de Greef, M. M. J. Smulders, M. Wolffs, A. P. H. J. Schenning, R. P. Sijbesma and E. W. Meijer, *Chem. Rev.*, 2009, **109**, 5687–5754.
- E. E. Greciano, B. Matarranz and L. Sanchez, *Angew. Chem., Int. Ed.*, 2018, **57**, 4697–4701.
- J.-M. Lehn, *Proc. Natl. Acad. Sci. U. S. A.*, 2002, **99**, 4763–4768.
- J.-F. Lutz, J.-M. Lehn, E. W. Meijer and K. Matyjaszewski, *Nat. Rev. Mater.*, 2016, **1**, 16024–16038.
- D. van der Zwaag, T. F. A. de Greef and E. W. Meijer, *Angew. Chem., Int. Ed.*, 2015, **54**, 8334–8336.
- R. D. Mukhopadhyay and A. Ajayaghosh, *Science*, 2015, **349**, 241–242.
- X. Wang, G. Guerin, H. Wang, Y. Wang, I. Manners and M. A. Winnik, *Science*, 2007, **317**, 644–647.
- J. Boekhoven, A. M. Brizard, K. N. K. Kowligi, G. J. M. Koper, R. Eelkema and J. H. van Esch, *Angew. Chem., Int. Ed.*, 2010, **49**, 4825–4828.
- J. Boekhoven, W. E. Hendriksen, G. J. K. Koper, R. Eelkema and J. H. van Esch, *Science*, 2015, **349**, 1075–1079.
- C. G. Pappas, I. R. Sasselli and R. V. Uljijn, *Angew. Chem., Int. Ed.*, 2015, **54**, 8119–8123.
- H. Lodish, *Molecular Cell Biology*, 6th edn, 2007, W. H. Freeman, New York.
- M. Kasai, S. Asakura and F. Oosawa, *Biochim. Biophys. Acta*, 1962, **57**, 22–31.
- L. A. Lowery and D. V. Vactor, *Nat. Rev. Mol. Cell Biol.*, 2009, **10**, 332–343.
- E. Leikina, M. V. Merts, N. Kuznetsova and S. Leikin, *Proc. Natl. Acad. Sci. U. S. A.*, 2002, **99**, 1314.
- S. Dhiman and S. J. George, *Bull. Chem. Soc. Jpn.*, 2018, **91**, 687–699.
- C. G. Pappas, I. R. Sasselli and R. V. Uljijn, *Angew. Chem., Int. Ed.*, 2015, **54**, 8119–8123.
- N. Singh, B. Lainer, G. J. M. Formon, S. De Piccoli and T. M. Hermans, *J. Am. Chem. Soc.*, 2020, **142**, 4083–4087.
- S. Ogi, K. Sugiyasu, S. Manna, S. Samitsu and M. Takeuchi, *Nat. Chem.*, 2014, **6**, 188–195.
- T. Fukui, S. Kawai, S. Fujinuma, Y. Matsushita, T. Yasuda, T. Sakurai, S. Seki, M. Takeuchi and K. Sugiyasu, *Nat. Chem.*, 2017, **9**, 493–499.
- J. Kang, D. Miyajima, T. Mori, Y. Inoue, Y. Itoh and T. Aida, *Science*, 2015, **347**, 646–651.
- M. Wehner and F. Würthner, *Nat. Rev. Chem.*, 2020, **4**, 38–53.



- 28 S. Patra, S. Chandrabhas, S. Dhiman and S. J. George, *J. Am. Chem. Soc.*, 2024, **146**, 12577–12586.
- 29 S. Sarkar, A. Sarkar, A. Som, S. S. Agasti and S. J. George, *J. Am. Chem. Soc.*, 2021, **143**, 11777–11787.
- 30 R. Laishram, S. Sarkar, I. Seth, N. Khatun, V. K. Aswal, U. Maitra and S. J. George, *J. Am. Chem. Soc.*, 2022, **144**, 11306–11315.
- 31 S. Toledano, R. J. Williams, V. Jayawarna and R. V. Ulijn, *J. Am. Chem. Soc.*, 2006, **128**, 1070–1071.
- 32 R. J. Williams, A. M. Smith, R. Collins, N. Hodson, A. K. Das and R. V. Ulijn, *Nat. Nanotechnol.*, 2009, **4**, 19–24.
- 33 H. He, W. Tan, J. Guo, M. Yi, A. N. Shy and B. Xu, *Chem. Rev.*, 2020, **120**, 9994–10078.
- 34 K. Jalani, A. D. Das, R. Sasmal, S. S. Agasti and S. J. George, *Nat. Commun.*, 2020, **11**, 3967.
- 35 S. Dhiman, A. Sarkar and S. J. George, *RSC Adv.*, 2018, **8**, 18913–18925.
- 36 A. Mishra, S. Dhiman and S. J. George, *Angew. Chem., Int. Ed.*, 2021, **133**, 2772–2788.
- 37 J. Deng and A. Walther, *Adv. Mater.*, 2020, **32**, 2002629.
- 38 J. Deng and A. Walther, *Nat. Commun.*, 2020, **11**, 3658.
- 39 M. Kumar, P. Brocorens, C. Tonnelé, D. Beljonne, M. Surin and S. J. George, *Nat. Commun.*, 2014, **5**, 5793.
- 40 S. Dhiman, A. Jain, M. Kumar and S. J. George, *J. Am. Chem. Soc.*, 2017, **139**, 16568–16575.
- 41 S. Dhiman, A. Jain and S. J. George, *Angew. Chem., Int. Ed.*, 2017, **129**, 1349–1353.
- 42 S. Maiti, I. Fortunati, C. Ferrante, P. Scrimin and L. J. Prins, *Nat. Chem.*, 2016, **8**, 725–731.
- 43 A. Mishra, D. B. Korlepara, M. Kumar, A. Jain, N. Jonnalagadda, K. K. Bejagam, S. Balasubramanian and S. J. George, *Nat. Commun.*, 2018, **9**, 1295.
- 44 A. Das, S. Ghosh, A. Mishra, A. Som, V. B. Banakar, S. S. Agasti and S. J. George, *J. Am. Chem. Soc.*, 2024, **146**, 14844–14855.
- 45 D. Didry, M.-F. Carlier and D. Pantaloni, *J. Biol. Chem.*, 1998, **273**, 25602–25611.
- 46 T. Ma, C. Li and G. Shi, *Langmuir*, 2008, **24**, 43–48.
- 47 B. Russ, M. J. Robb, F. G. Brunetti, P. L. Miller, E. E. Perry, S. N. Patel, V. Ho, W. B. Chang, J. J. Urban, M. L. Chabinye, C. J. Hawker and R. A. Segelman, *Adv. Mater.*, 2014, **26**, 3473–3477.

