Chemical Science



EDGE ARTICLE

View Article Online
View Journal | View Issue



Cite this: Chem. Sci., 2018, 9, 6446

All publication charges for this article have been paid for by the Royal Society of Chemistry

Received 15th May 2018 Accepted 1st July 2018

DOI: 10.1039/c8sc02138d

rsc.li/chemical-science

N-Heterocyclic carbenes as chiral Brønsted base catalysts: a highly diastereo- and enantioselective 1,6-addition reaction†

Surojit Santra, Arka Porey, Barun Jana and Joyram Guin **

Highly diastereo- and enantioselective 1,6-addition of 1,3-ketoamides to p-quinone methides (p-QMs) using chiral NHCs as Brønsted base catalysts is developed. The reaction is based on the utilization of a 1,3-ketoamide having acidic N-H that forms a chiral ion-pair consisting of the enolate and the azolium ion. Different β -ketoamides and functionalized p-QMs are applicable to the reaction. Synthetic application of the method is demonstrated via the preparation of highly enantioenriched β and γ -lactam derivatives

Introduction

N-Heterocyclic carbenes (NHCs) are the most versatile organocatalysts. Consequently, asymmetric catalysis with chiral NHCs involves covalent interaction with the activated substrate, Lewis acid–base interaction with the reagent (Scheme 1a and 1b)¹ and hydrogen bonding interaction with the substrate using a proton shuttle (Scheme 1c).² Although asymmetric NHC catalysis *via*

a) covalent interaction

N Ar

Many enantioselective reactions

c) H-bonding interaction

Ar

N
Ar

Chiral NHC

enantioselective reactions

e) NHC-catalysis via ion-pair interaction (this study)

R

Chiral ion-pair

enantioselective reaction-unknown

e) NHC-catalysis via ion-pair interaction (this study)

Scheme 1 Different modes of substrate activation for asymmetric NHC-catalysis.

Department of Organic Chemistry, Indian Association for the Cultivation of Science, 2A & 2B Raja S. C. Mullick Road, Jadavpur, Kolkata-700032, India. E-mail: ocjg@iacs.res. in

† Electronic supplementary information (ESI) available. CCDC 1831345-1831347. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c8sc02138d

ion-pair interaction with substrates by utilizing the intrinsic Brønsted base characteristic of NHCs³ has drawn significant interest, it remains highly challenging (Scheme 1d).⁴ Herein, we present a catalytic and highly stereoselective 1,6-addition reaction of *p*-QMs using NHCs as Brønsted base catalysts (Scheme 1e).

p-QM scaffolds are found in many natural products⁵ and serve as key reactive intermediates in several chemical, medicinal and biological processes. 6 Consequently, efforts have been devoted to developing organocatalytic asymmetric nucleophilic addition to p-QM.7 However, attempts at developing enantioselective 1,6-addition of enolizable nucleophiles to p-QMs using chiral NHCs as Brønsted bases remains largely unsuccessful.8 To employ NHCs as Brønsted base catalysts for asymmetric 1,6addition of p-QMs, we anticipated that the substrate should have a lower pK_a value than the NHC. We thus envisioned that easily enolizable 1,3-ketoamides containing an acidic N-H group would be appropriate nucleophiles. It is expected that an in situ generated NHC having a p K_a value in the range of 17–19 ^{3a} may deprotonate the ketoamide (p $K_a \approx 10-12$), furnishing a chiral ion-pair comprising the enolate and the azolium ion. The in situ generated chiral enolate is expected to react with p-QM, thus providing enantioselectivity to the final product.

Results and discussion

Our studies were commenced using p-QM **1a** and the β -ketoamide **2a** as model substrates under various reaction conditions with a series of chiral NHCs (see the ESI†). Following initial optimization, a variety of amides possessing different aromatic amines were assayed using 20 mol% of **3g**, 16 mol% of LiHMDS, 20 mol% of HFIP and 4 Å MS in toluene (Table 1). The β -ketoamide derived from aniline afforded product **4a** with poor stereoselectivity (entry 1). Enantioselectivity of the reaction could not be improved using β -ketoamide **4b** or **4c** having

Edge Article Chemical Science

Table 1 Reaction development⁶

Entry	Amide	Temp. (°C)	h	dr	ee (%)	Conv. (%)
1	2a	0	12	64:36	44, 48	95
2	2b	0	12	73:27	27, ND	50
3	2c	0	12	65:35	30, 12	70
4	2d	0	12	80:20	68, 14	70
5	2e	0	12	70:30	70, 10	50
6	2f	0	12	67:33	66, 20	60
7	2g	0	12	83:17	70, 15	70
8	2h	0	12	95:5	73 ^b	50
9^c	2h	0	12	97:3	85 ^b	75
10^c	2h	-20	14	98:2	95^{b}	70
11 ^c	2h	-40	20	98:2	98^b	97^d
$12^{c,e}$	2h	-40	24	98:2	98^b	93^d

^a Reaction conditions: 1a (0.05 mmol), 2a-h (0.05 mmol), LiHMDS (16 mol%), HFIP (20 mol%), 4 Å MS (35 mg) in toluene (1.0 mL); dr determined by ¹H NMR and HPLC analysis; ee determined by HPLC analysis on a chiral stationary phase. ^b ee of the major diastereoisomer is given. c Reaction performed without using HFIP. Isolated yield. e Reaction performed on a 0.1 mmol scale using 15 mol% of 3g. ND = not determined.

a phenyl substituent at the para or meta position of the aniline ring (entries 2 and 3). However, introducing a phenyl group at the *ortho* position of the aniline moiety of β -ketoamide was found to be beneficial, delivering product 4d in 68% ee with the amide 2d (entry 4). Based on this result, different β-ketoamides 2e-h containing bulky substituents at the ortho position of the aniline unit were prepared and examined. No considerable improvement in the stereoselectivity of the products, 4e-g, was realized with the ketoamides 2e-g (entries 5-7). The sterically demanding ketoamide 2h, however, delivered product 4h with a slight improvement in stereoselectivity (entry 8). Interestingly, the reaction efficacy was improved further with the amide 2h, when the reaction was performed in the absence of HFIP (entry 9). This result may indicate that HFIP interferes in the formation of a tight ion-pair comprising the enolate and the azolium ion.11 By carrying out the reaction at reduced temperature, product 4h was obtained in excellent yield and stereoselectivity (entries 10 and 11). Importantly, the loading of precatalyst 3g could also be reduced to 15 mol% without diminishing the overall reaction efficiency (entry 12).

The substrate scope of the reaction was then evaluated varying both p-QMs and amides (Table 2). A series of p-QMs with different substituents at the aromatic ring and quinone moiety were first investigated with the amide 2h. Accordingly, the reaction furnished the desired products 5-7 in excellent yields and stereoselectivity with p-QMs having benzene, p-tolyl and pbiphenyl substituents. While a moderate yield of product 8 was obtained with o-tolyl substituted p-QM, the stereoselectivity remained excellent. p-QMs bearing F, Cl and Br substituents at any position of the aromatic ring were easily transformed into the corresponding products 9-11 in excellent yields and stereoselectivity. Substrates having electron rich substituents like OMe (12), NMe₂ (13), SMe (14), O-allyl (15) and OAc (16) or electron deficient functional groups such as CF3 (17), cyano (18), ester (19), keto (20) and nitro (21), irrespective of their position, were applicable to the reaction. Naphthyl substituted p-OM was converted to product 22. Heteroaromatic rings, for instance, pyridine (23), furan (24) and thiophene (25), were tolerated in this catalytic reaction. A slight decrease in ee values was realized by replacing the tert-butyl group of the quinone ring with methyl or isopropyl substituents (products 26 and 27). Interestingly, the catalytic asymmetric process could be extended to the challenging alkylated p-QM, providing the expected product 28 in good yield, albeit with moderate stereoselectivity. Substrate scope studies were then carried out using different amides with 1a. The amide 2i $(R^3 = Me \text{ and } Ar^1 =$ quinolin-8-yl) afforded product 29 in good yield and stereoselectivity. Likewise, different α-aroylacetamides 2j-m were found to be suitable for the reaction, furnishing the desired products 30-33 in good yields and enantioselectivity. Absolute configurations of 2S, 3R were determined for compound 11 using single-crystal X-ray analysis.12

To illustrate the synthetic utility of the reaction, a preparative scale experiment using 0.97 g of p-QM 1a and 0.68 g of amide 2i was performed, which resulted in 1.25 g of product 29 in good yield and stereoselectivity (dr = 98:2, ee = 97%, Scheme 2a). Furthermore, a variety of valuable enantioenriched compounds were prepared from product 29 through simple chemical transformations (Scheme 2b, see the ESI†). Accordingly, NaBH₄ reduction of ketone 29 afforded the alcohol 34a having three contiguous stereogenic centres in good yield and excellent ee along with a minor diastereoisomer 34b (see the ESI†). Compound 34b was separated by column chromatography. The major diastereoisomer 34a, whose structure was confirmed by single crystal X-ray analysis, was transformed into the spirocyclic γ -lactam 35 in one step *via* PIDA oxidation. The removal of the tert-butyl group from the alcohol 34a with AlCl₃ afforded the phenol 36 in good yield without diminishing its original stereoselectivity. The mesyl protection of the alcohol 34a provided compound 37 in good yield. The absolute configuration of compound 37 was unambiguously determined by singlecrystal X-ray analysis.12 Compound 37 was then converted to the enantioenriched β-lactam 38 in good yield with excellent stereocontrol. Furthermore, the synthesis of biologically relevant enantioenriched triarylmethane 39 was achieved with excellent yield and ee value upon the treatment of 29 with hydrazine hydrate.

^a Reaction conditions: 1a-y (0.1 mmol), amides (0.1 mmol), 3g (15 mol%), LiHMDS (12 mol%), 4 Å MS (70 mg) in toluene (2.0 mL); isolated yields. ^b Reaction performed at -78 °C. ^c Reaction performed with amide 2i. Diastereoisomeric ratio (dr) determined by ¹H NMR and HPLC analysis. Enantiomeric excess (ee) determined by HPLC analysis on a chiral stationary phase.

For a better understanding of the reaction mechanism, several other experiments were carried out (see the ESI†). It was observed that the reaction afforded the desired product with a similar reactivity and ee value using NaHMDS/KHMDS in lieu of LiHMDS as the base. These results indicate that the metal ion of the base may not have any considerable effect on the reaction outcome. Using a preformed NHC as the catalyst, the reaction afforded the desired product with an identical stereoselectivity albeit with a very low reactivity. The importance of the acidic N-H group of β-ketoamide was established by carrying out the reaction using N-methylated amide, which resulted in complete inhibition of the catalytic process (see the ESI†). Additionally, the in situ formation of the NHC under our reaction conditions was confirmed by performing the well-established oxidative annulation reaction between cinnamaldehyde and acetylacetone using precatalyst 3g (see the ESI†). This was further established by ¹H NMR spectroscopy studies. It was found that the ¹H NMR resonance at 10.76 ppm corresponding to the iminium C2-H of 3g disappeared when 3g was treated with LiHMDS. However, the ¹H NMR signal reappeared upon adding an equimolar amount of the β -ketoamide **2h** into the solution (see the ESI†). These observations may indicate that the initially

Reaction conditions: i) NaBH₄, toluene:methanol (3:1), 0 °C, 1 h; ii) PIDA, CF₃CH₂OH, 0 °C, 1 h; iii) AlCl $_3$, toluene, 60 °C, 1 h; iv) MsCl, Py, 0 °C, 2 h; v) NaH, DMF:DCM (1:4), 0 °C, 6 h; vi) N $_2$ H $_4$ +H $_2$ O, EtOH, 80 °C, 24 h.

Scheme 2 Synthetic applications.

Edge Article Chemical Science

Scheme 3 Proposed activation pathway

Fig. 1 Probable transition state of the reaction

formed NHC deprotonates the acidic N–H of **2h**, thus forming a chiral ion-pair involving the azolium ion and enolate.

For mechanistic consideration, two possible reaction pathways are proposed for the reaction (Scheme 3). The reaction may involve hydrogen bonding interaction between the NHC and the β -ketoamide (Scheme 3, **A**). Alternatively, it may proceed through a chiral ion-pair intermediate consisting of the enolate and the azolium ion (Scheme 3, **B**). By reconciling the reported pK_a values of the similar NHC and β -ketoamide with the results obtained in our preliminary mechanistic studies, we tend towards the ion-pair interaction between the NHC and the substrate. Finally, the activated nucleophile undergoes addition reaction with the p-QM to deliver the desired product with excellent stereocontrol. Further studies are surely required to establish the actual mode of NHC-catalysis for this reaction.

Based on the crystal structure analysis of product **11**, the observed high stereoselectivity of the addition product is explained through the proposed transition state (TS) in which the Re-face of p-QM **1h** approaches the chiral enolate due to hydrogen bonding activation and π - π interaction as shown in Fig. 1.

Conclusions

In conclusion, we have introduced a new mode of asymmetric NHC-catalysis through ion-pair interaction between a catalyst and a nucleophile. The novel catalytic method enables highly diastereo- and enantioselective 1,6-conjugate addition of 1,3-ketoamides to *p*-QMs using NHCs as Brønsted bases. The reaction furnishes the desired products with excellent stereo-selectivity (ee and dr) and yields. Synthetic application of the method is demonstrated by the preparation of several valuable materials with excellent stereoselectivity.

Conflicts of interest

There are no conflicts of interest to declare.

Acknowledgements

We gratefully acknowledge the generous financial support from the Science and Engineering Research Board, India (EMR/2016/ 006344) and CSIR for fellowship to SS and AP.

Notes and references

- 1 For some selected reviews on asymmetric NHC-catalysis with covalent interaction, see: (a) M. Zhao, Y.-T. Zhang, J. Chen and L. Zhou, Asian J. Org. Chem., 2018, 7, 54; (b) M. H. Wang and K. A. Scheidt, Angew. Chem., Int. Ed., 2016, 55, 14912; (c) S. R. Yetra, A. Patra and A. T. Biju, Synthesis, 2015, 47, 1357; (d) R. S. Menon, A. T. Biju and V. Nair, Chem. Soc. Rev., 2015, 44, 5040; (e) D. M. Flanigan, F. Romanov-Michailidis, N. A. White and T. Rovis, Chem. Rev., 2015, **115**, 9307; (f) J. Mahatthananchai and W. Bode, Acc. Chem. Res., 2014, 47, 696; M. N. Hopkinson, C. Richter, M. Schedler and F. Glorius, Nature, 2014, 510, 485; (h) S. J. Ryan, L. Candish and D. W. Lupton, Chem. Soc. Rev., 2013, 42, 4906; (i) S. De Sarkar, A. Biswas, R. C. Samanta and A. Studer, Chem.-Eur. J., 2013, 19, 4664; (j) N. Marion, S. Díez-González and S. P. Nolan, Angew. Chem., Int. Ed., 2007, 46, 2988; (k) D. Enders, O. Niemeier and A. Henseler, Chem. Rev., 2007, **107**, 5606.
- 2 (a) J. Chen and Y. Huang, Sci. China: Chem., 2016, 59, 251; (b)
 L. Wang, J. Chen and Y. Huang, Angew. Chem., Int. Ed., 2015, 54, 15414; (c) J. Chen, S. Meng, L. Wang, H. Tang and Y. Huang, Chem. Sci., 2015, 6, 4184; (d) J. Chen and Y. Huang, Nat. Commun., 2014, 5, 3437.
- 3 (a) R. S. Massey, C. J. Collett, A. G. Lindsay, A. D. Smith and A. C. O'Donoghue, J. Am. Chem. Soc., 2012, 134, 20421; (b)
 T. L. Amyes, S. T. Diver, J. P. Richard, F. M. Rivas and K. Toth, J. Am. Chem. Soc., 2004, 126, 4366; (c) Y.-J. Kim and A. Streitwieser, J. Am. Chem. Soc., 2002, 124, 5757.
- 4 (a) F. Perez, Y. Ren, T. Boddaert, J. Rodriguez and Y. Coquerel, J. Org. Chem., 2015, 80, 1092; (b) Q. Kang and Y. Zhang, Org. Biomol. Chem., 2011, 9, 6715; (c) T. Boddaert, Y. Coquerel and J. Rodriguez, Chem.-Eur. J., 2011, 17, 2266; (d) E. M. Phillips, M. Riedrich and K. A. Scheidt, J. Am. Chem. Soc., 2010, 132, 13179; (e) S. De Sarkar, S. Grimme and A. Studer, J. Am. Chem. Soc., 2010, 132, 1190; (f) M. Movassaghi and M. A. Schmidt, Org. Lett., 2005, 7, 2453; (g) J. A. Cowan, J. A. C. Clyburne, M. G. Davidson, R. L. W. Harris, J. A. K. Howard, P. Küpper, M. A. Leech and S. P. Richards, Angew. Chem., Int. Ed., 2002, 41, 1432.
- 5 (a) R. Jansen, K. Gerth, H. Steinmetz, S. Reinecke, W. Kessler,
 A. Kirschning and R. Müller, Chem.–Eur. J., 2011, 17, 7739; (b)
 H. J. Martin, T. Magauer and J. Mulzer, Angew. Chem., Int. Ed., 2010, 49, 5614.
- 6 (a) C. Sridar, J. D'Agostino and P. F. Hollenberg, *Drug Metab. Dispos.*, 2012, 40, 2280; (b) R. Dehn, Y. Katsuyama, A. Weber, K. Gerth, R. Jansen, H. Steinmetz, G. Höfle, R. Müller and A. Kirschning, *Angew. Chem., Int. Ed.*, 2011, 50, 3882.

Chemical Science Edge Article

- 7 For some recent examples on asymmetric organocatalytic 1,6-addition of p-QM, see: (a) W. Li, X. Xu, Y. Liu, H. Gao, Y. Cheng and P. Li, Org. Lett., 2018, 20, 1142; (b) T.-C. Kang, L.-P. Wu, Q.-W. Yu and X.-Y. Wu, Chem.-Eur. J., 2017, 23, 6509; (c) M. Chen and J. Sun, Angew. Chem., Int. Ed., 2017, 56, 4583; (d) K. Zhao, Y. Zhi, A. Wang and D. Enders, ACS Catal., 2016, 6, 657; (e) Y. F. Wong, Z. Wang and J. Sun, Org. Biomol. Chem., 2016, 14, 5751; (f) X. Li, X. Xu, W. Wei, A. Lin and H. Yao, Org. Lett., 2016, 18, 428; (g) L. Ge, X. Lu, C. Cheng, J. Chen, W. Cao, X. Wu and G. Zhao, J. Org. Chem., 2016, 81, 9315; (h) N. Dong, Z.-P. Zhang, X.-S. Xue, X. Li and J.-P. Cheng, Angew. Chem., Int. Ed., 2016, 55, 1460; (i) L. Caruana, F. Kniep, T. K. Johansen, P. H. Poulsen and K. A. Jørgensen, J. Am. Chem. Soc., 2014, 136, 15929; (i) W.-D. Chu, L.-F. Zhang, X. Bao, X.-H. Zhao, C. Zeng, J.-Y. Du, G.-B. Zhang, F.-X. Wang, X.-Y. Ma and C.-A. Fan, Angew. Chem., Int. Ed., 2013, 52, 9229.
- 8 (a) S. Santra, A. Porey and J. Guin, *Asian J. Org. Chem.*, 2018, 7, 477; (b) P. Arde and R. V. Anand, *RSC Adv.*, 2016, 6, 77111.
- 9 (a) M. d. M. Sanchez Duque, O. Baslé, N. Isambert, A. Gaudel-Siri, Y. Génisson, J.-C. Plaquevent, J. Rodriguez and T. Constantieux, Org. Lett., 2011, 13, 3296; (b) J. W. Bunting, J. P. Kanter, R. Nelander and Z. Wu, Can. J. Chem., 1995, 73, 1305.
- some selected reviews on asymmetric ion-pair 10 For organocatalysis, see: (a) D. Parmar, E. Sugiono, S. Raja and M. Rueping, Chem. Rev., 2017, 117, 10608; (b) X. Yang, T. Wu, R. J. Phipps and F. D. Toste, Chem. Rev., 2015, 115, 826; (c) S. Shirakawa, S. Liu, S. Kaneko, Y. Kumatabara, A. Fukuda, Y. Omagari and K. Maruoka, Angew. Chem., Int. Ed., 2015, 54, 15767; (d) T. James, M. van Gemmeren and B. List, Chem. Rev., 2015, 115, 9388; (e) M. Mahlau and B. List, Angew. Chem., Int. Ed., 2013, 52, 518; (f) K. Brak and E. N. Jacobsen, Angew. Chem., Int. Ed., 2013, 52, 534; (g) M. Terada, Synthesis, 2010, 1929; (h) D. Uraguchi, Y. Ueki and T. Ooi, Science, 2009, 326, 120; (i) C. Palomo, M. Oiarbide and R. Lopez, Chem. Soc. Rev., 2009, 38, 632; (j) J. Lacour and D. Moraleda, Chem. Commun., 2009, 7073; (k) T. Akiyama, Chem. Rev., 2007, 107, 5744.
- 11 The addition of a catalytic amount of a proton shuttle like HFIP was essential for the reported hydrogen bonding asymmetric NHC-catalysis (see ref. 2). In contrast, the addition of HFIP was found to be detrimental in this study. This may indicate that the reaction involves an alternative mode of asymmetric NHC-catalysis *via* ion-pair interaction.
- 12 CCDC 1831345, 1831346 and 1831347 contain the supplementary crystallographic data for 11, 34a and 37, respectively (see also the ESI†).