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Pyrene-based metallocycles and metallocages: more than fluorophores

Xu-Qing Wang,^a Qing-Hui Ling,^a Wei Wang^{*a} and Lin Xu^{id *ab}

Attributed to their synthetic accessibility, unique structural and intriguing photophysical features, pyrene moieties have recently been introduced into supramolecular coordination complexes (SCCs) such as metallocycles and metallocages as either functional groups or key panels. The resultant pyrene-based SCCs displayed not only attractive emission properties but also impressive binding abilities towards diverse guests, endowing them with wide applications in sensing, host–guest chemistry, light harvesting etc. In this review, a comprehensive summary of pyrene-based metallocycles and metallocages is presented, and the emphasis will be on the intriguing photophysical features and unique host–guest chemistry that the pyrene moieties bring to the SCCs.

Introduction

During the past few decades, on the basis of the relatively rigid yet dynamic metal–ligand coordination bonds, coordination-driven self-assembly has evolved as a general and powerful strategy towards the construction of diverse delicate and elegant two-dimensional (2D) and three-dimensional (3D) supramolecular coordination complexes (SCCs) such as metallocycles and metallocages.¹ Taking advantage of their well-defined geometry and well-tuned cavities as well as well-arranged functional groups

within their skeletons, the resultant SCCs have shown wide applications in diverse fields such as host–guest chemistry, catalysis, drug delivery, functional materials, and so on.² Recently, increasing interest in the construction of fluorescent SCCs has been witnessed since they can serve as privileged platforms for sensing study, the mimicking of natural photosynthetic processes, the fabrication of photoelectric devices, etc.³ In general, the fluorophores can be introduced as either functional groups that are attached to the key building blocks or key panels that are incorporated into the main skeletons of the targeted fluorescent SCCs.

As one of the most common and useful fluorophores, pyrene is an excellent candidate for the construction of novel fluorescent SCCs.⁴ First of all, the synthetic accessibility of pyrene derivatives with diverse functional groups and well-defined structural features lays the foundation for the construction of

^a Shanghai Key Laboratory of Green Chemistry and Chemical Processes, School of Chemistry and Molecular Engineering, East China Normal University, 3663 North Zhongshan Road, Shanghai 200062, People's Republic of China. E-mail: wwang@chem.ecnu.edu.cn, lxu@chem.ecnu.edu.cn

^b State Key Laboratory of Fine Chemicals, Dalian University of Technology, Dalian 116024, P. R. China



Xu-Qing Wang

Xu-Qing Wang received her BS degree in chemistry from Northwest A & F University in 2014. She obtained her PhD degree from East China Normal University in 2019 under the supervision of Prof. Hai-Bo Yang. Then she joined Professor Yang's group as a Postdoctoral Fellow in June 2019. Her current research interest is focused on well-controlled self-assembly of supramolecular coordination architectures.



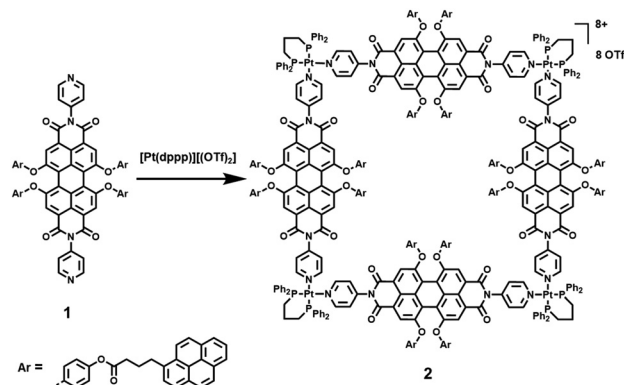
Qing-Hui Ling

Qing-Hui Ling received her BS degree in applied chemistry from Anhui University of Science and Technology in 2019. She is now pursuing her MS degree in Prof. Hai-Bo Yang's group at East China Normal University. Her research interests focus on the design and synthesis of functional fluorescent metallocages.

diverse SCCs with pyrene moieties either as functional groups or building blocks; secondly, compared with other fluorophores, pyrene derivatives usually displayed intriguing photophysical properties, such as high fluorescence quantum yield, relatively long excited-state lifetime, unique excimer emission *etc.* Notably, the fluorescence bands for the monomer and excimer of pyrene derivatives reveal exceptional distinction and, moreover, their excitation spectra are very sensitive toward the changes of microenvironments.⁵ Taking advantage of all these attractive photophysical features, pyrene derivatives can serve as versatile fluorophores for the incorporation of expected photophysical functions into the resultant SCCs. Last but not least, the relatively large conjugated π -electron-rich surface of pyrene derivatives makes them excellent candidates as panels for the construction of SCCs with well-defined and confined pockets for the investigations of host-guest chemistry. During the past few years, a growing number of reports on the construction of pyrene-based SCCs has been witnessed, and thus it is time to summarize these developments and achievements. Herein, divided by the different types of SCC cores, pyrene-based 2D metallocycles and 3D metallocages will be discussed, with an emphasis on the intriguing photophysical features and unique host-guest chemistry that the pyrene moieties bring to the SCCs.

Pyrene-based 2D metallocycles

Aiming at mimicking natural light harvesting systems with circular dye arrays, Würthner and coworkers⁶ reported the construction of a well-defined molecular square **2**, in which sixteen pyrene and four perylene bisimide chromophores are arranged in a circular way (Scheme 1). In the resultant metallocycle, the outer pyrene moieties served as antenna dyes, from which highly efficient and fast photo-induced energy transfer as well as very fast and efficient electron transfer to the inner perylene bisimide chromophores was demonstrated. Such



Scheme 1 The construction of molecular square **2** with sixteen pyrene and four perylene bisimide chromophores arranged in a circular way.

photophysical processes were clearly confirmed by the combination of steady-state and time-resolved emission as well as femtosecond transient absorption studies. More impressively, compared with the free ligand **1**, the metallocycle **2** revealed substantially faster and more efficient electron-transfer processes, making it a novel monodisperse nanoaggregate with well-defined multiple-chromophore arrangements.

On the basis of the coordination-driven self-assembly strategy, by employing various pyrene-modified ligands, Yang and coworkers⁷ demonstrated a simple and highly efficient approach for the construction of multipyrene hexagonal metallocycles with well-defined shape and size in 2013. As shown in Scheme 2, a mixture of pyrene-functionalized 120° Pt(II) acceptor **3** with a 120° or 180° dipyriddy donor resulted in the formation of corresponding multipyrene hexagons **4** or **5** with three or six pyrene moieties, respectively. In addition, the combination of pyrene-functionalized 120° dipyriddy acceptor **6** and a 120° or 180° Pt(II) acceptor could also give rise to similar tris- and hexakis(pyrene) hexagonal metallocycles **7** and **8**, respectively. Interestingly, preliminary spectroscopic study of



Wei Wang

interlocked macromolecules and topological chirality.

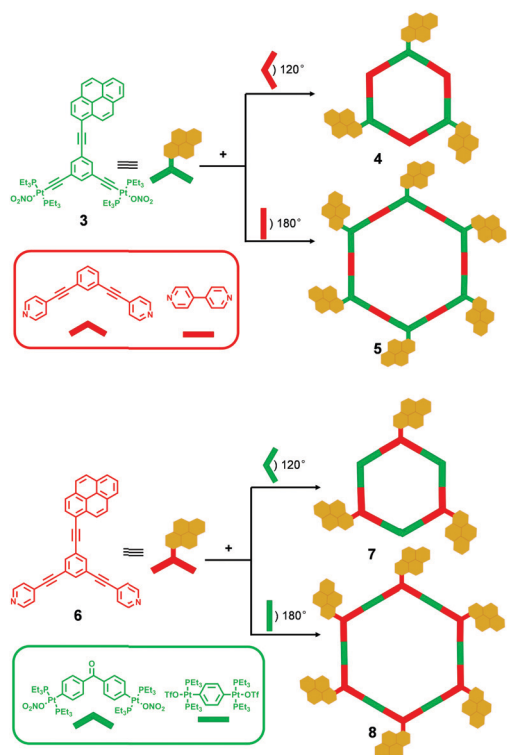
Wei Wang obtained his BSc degree from Sichuan University in 2011 and PhD degree from the East China Normal University in 2016 under the supervision of Prof. Hai-Bo Yang. He then joined Prof. A. Dieter Schlüter's group at ETH Zürich to pursue postdoctoral study (2016–2019). After that, he has become a research professor (Zijiang Young Scholar) at East China Normal University. His research interest includes



Lin Xu

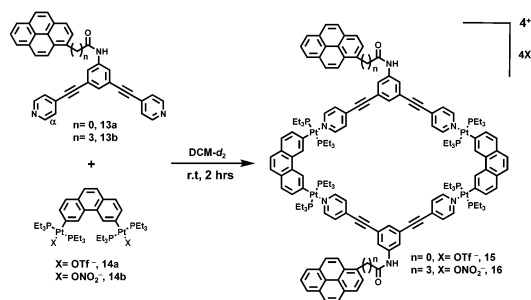
University of Cambridge with Prof. Jonathan Nitschke during 2015–2017. His research interests focus mainly on supramolecular fluorescent chemistry.

Lin Xu received his PhD degree in Chemistry from the East China University of Science and Technology (ECUST, Shanghai) in 2012 under the supervision of Professor Xuhong Qian. Subsequently, he joined the East China Normal University (ECNU, Shanghai) as an assistant professor and was promoted to an associate professor and a full professor in 2014 and 2018, respectively. He carried out two-years postdoctoral research at the



Scheme 2 Coordination-driven self-assembly of multipyrene hexagonal metallocycles **4**, **5**, **7**, and **8**.

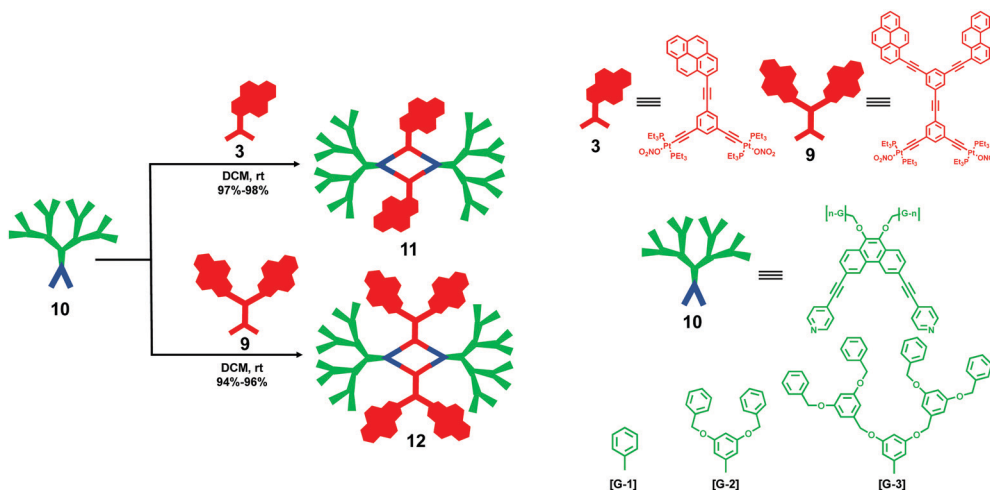
these resultant hexagonal metallocycles revealed dramatically different emission behaviours. For instance, hexagons **7** and **8** displayed a longer fluorescence emission band at λ_{max} 550 nm in CH_2Cl_2 , which corresponded to the excimer emission of the pyrene chromophore. However, hexagons **4** and **5** could not form excimers at the same conditions. The results indicated that the combination mode of building blocks played an important role in the excimer formation of the resulting self-assemblies, which might be reasonably explained by the varied



Scheme 4 Synthesis of pyrene-functionalized molecular rhomboids **15** and **16**.

charge densities of the resultant hexagons originating from the different combinations. Furthermore, by employing the same directional metal–ligand bonding strategy, the same group⁸ reported a series of pyrene-functionalized metallodendrimers **11** and **12** derived from pyrene-containing 120° Pt(II) acceptors (**3** and **9**) and 60° dendritic donors **10** (Scheme 3). Notably, compared to the pyrene-containing precursors, improved fluorescence quantum yields of the corresponding pyrene-modified dendrimers were observed, which might be due to the fact that the steric hindrance of dendrons inhibited the aggregation of pyrenes.

Similarly, Stang *et al.*⁹ prepared two rhomboids **15** and **16** by mixing pyrene-functionalized 120° donor **13** and 60° Pt(II) acceptor **14** (Scheme 4) via [2+2] self-assembly. Due to the existence of pyrene groups, both rhomboids were emissive and exhibited excimer emission bands, which derived from intermolecular interactions. Additionally, it was found that both the length of the spacers between the metallocycles and pyrene units and the solvent polarity had a great influence on the emission properties. For instance, as revealed by the emission spectra, rhomboids **15** and **16** displayed very different emission bands in sole CH_2Cl_2 solution. Meanwhile, the increase of the fraction of CH_3OH in CH_2Cl_2 could further remarkably modulate the emission behaviours of both rhomboids.



Scheme 3 Coordination-driven self-assembly of multipyrene hexagonal metallocycles **11** and **12**.

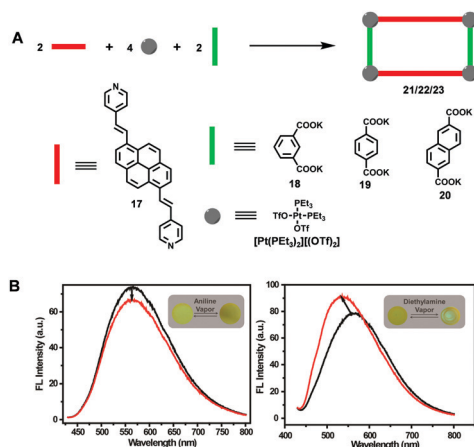


Fig. 1 The construction of pyrene-based metallocycles **21–23** via coordination-driven self-assembly (A) and the sensing performance of **22**-based fluorescent films towards aniline (left) and diethylamine (right) (B). Adapted with permission from ref. 10. (Copyright 2019 American Chemical Society.)

As shown in Fig. 1, the three-component coordination-driven self-assembly of pyrene-cored dipyriddy donor **17**, a 90° di-Pt(II) acceptor $[\text{Pt}(\text{PEt}_3)_2][(\text{OTf})_2]$ and dicarboxylates **18–20** resulted in the formation of three pyrene-based metallocycles **21–23**, respectively, which exhibited enhanced solubility in common organic solvents compared with the pyrene-cored ligand **17**. The resultant three metallocycles displayed similar emission properties in the solution state, but different emission wavelengths in the aggregated states. Interestingly, it was found that the solution emission of these metallocycles could be well-tuned through the modulation of the temperature and solvent, which is expectable since both factors could remarkably influence the aggregation states of the metallocycles mainly formed through the intermolecular π - π interaction between the pyrenyl moieties. Notably, taking advantage of the excellent photochemical stability, a homemade aromatic amine sensing platform with ligand **17**-based or metallocycle **22**-based fluorescent films as the core structure was built. More importantly, regarding the response speed and recovery time, the metallocycle **22**-based one revealed superior sensing performance (Fig. 1B), indicating that the formation of the metallocycles contributed a remarkable enhancement effect in the film state.¹⁰

In 2019, Stang *et al.*¹¹ developed a new strategy for the construction of a cross-linked supramolecular polymer based on platinum(II) metallocycles. As shown in Fig. 2, self-assembled [2+2] metallocycle **26** was prepared by mixing pyrene-functionalized donor **24** with 180° di-Pt(II) acceptor **25**, which contained two pyrene moieties at the periphery. Attributed to the existence of C-H... π interactions, host-guest interactions between the metallocycle core and the pyrene periphery were observed, as indicated by solid-state single-crystal structure and solution-state NMR study, respectively. Moreover, based on such an interesting driving force, a cross-linked supramolecular polymer was spontaneously formed in

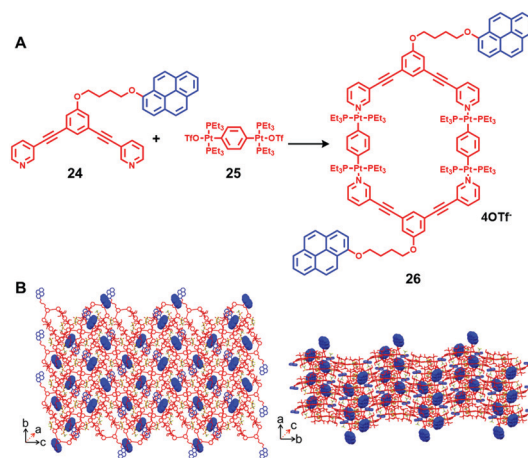
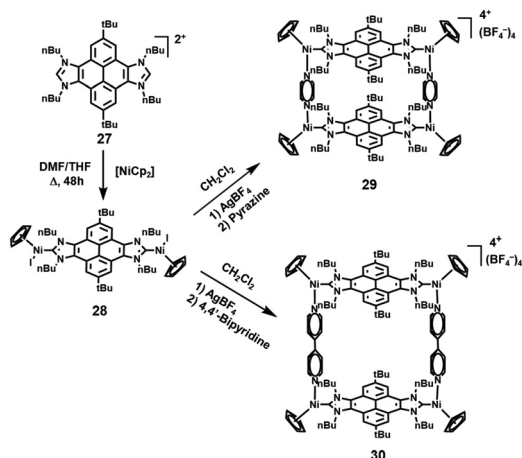


Fig. 2 The construction of metallocycles **26** bearing two pyrene moieties via coordination-driven self-assembly (A) and the single-crystal structure of the cross-linked supramolecular polymer of **26** (B). Adapted with permission from ref. 11. (Copyright 2019 American Chemical Society.)

both the solid and solution state. Similar to other platinum metallocycle-based supramolecular polymers,¹² the resultant cross-linked supramolecular polymer displayed interesting thermo-responsiveness and ion responsiveness. For instance, reversible thermo-induced phase transitions of the gel-like viscous liquids prepared from a highly concentrated solution of **26** were observed. In addition, attributed to the breaking and reforming of the Pt-N bonds in metallocycle **26**, a reversible transition between the gel-like supramolecular polymer and turbid solution was successfully achieved through the addition and removal of bromide anions.

Based on their continuous interest in poly-N-heterocyclic carbenes (NHCs),¹³ in 2017, Peris and coworkers¹⁴ demonstrated the preparation of a novel dimetallic complex **28** by mixing pyrene-linked di-NHC ligand **27** with nickelocene. Then, based on a metal-directed self-assembly strategy, in the presence of AgBF_4 , the reaction of **28** with pyrazine or 4,4'-bipyridine resulted in the successful synthesis of nickel-cornered metallo-rectangles **29** and **30**, respectively (Scheme 5). The resultant molecular rectangles were then tested to serve as receptors for the recognition of polycyclic aromatic hydrocarbons (PAHs). Due to the varied length of one of their sides, rectangle **29** could host one guest molecule within its cavity, whereas rectangle **30** was able to encapsulate up to two guest molecules, and the binding affinities strongly depended on the dimensional matching between the metallocycle hosts and guests. Notably, both rectangles could be used as efficient PAH scavengers in heptane though the formation of insoluble host-guest complexes.

Starting from the same pyrene-linked di-NHC ligand, the same group¹⁵ further prepared a palladium-cornered metallo-square **31** in 2019 *via* the metal-directed self-assembly strategy. Due to the existence of the flat polyaromatic surface of the four pyrene moieties, the resultant 3D cubic box with a large internal cavity could encapsulate both C_{60} and C_{70} fullerenes. The molecular structures of **31**, $\text{C}_{60}@\mathbf{31}$ and $\text{C}_{70}@\mathbf{31}$ were determined by single-crystal X-ray diffraction (Fig. 3). More



Scheme 5 The synthesis of nickel-cornered metallo-rectangles **29** and **30**.

importantly, as revealed by these solid-state structures, the resulting metallo-square was size-flexible and adaptable, which was able to adapt its size to the size of guest molecules. Furthermore, taking advantage of the excellent spin-converting properties of fullerenes, Peris *et al.* employed both inclusion complexes $C_{60}@31$ and $C_{70}@31$ as photochemically stable photosensitizers to efficiently generate singlet oxygen, which could oxidize a series of cyclic and acyclic alkenes at room temperature.¹⁶ In addition, considering that the distance between the palladium atoms of metallosquare **31** is about 13 Å, which is large enough to host three polyaromatic guests, metallo-square **31** was further employed for the encapsulation of three heteroguests including two electron-rich PAH molecules such as pyrene, triphenylene or coronene and one electron-poor *N,N'*-dimethyl-naphthalene-tetracarboxy diimide (NTCDI) molecule, giving rise to the formation of interesting quintuple D–A–D–A–D stacks (Fig. 4).¹⁷

Recently, as an extension of their previous work, Peris *et al.*¹⁸ demonstrated the synthesis of a novel tetragold metallo-rectangle **34** by the reaction of the pyrene-bis-imidazolydene-gold(i)

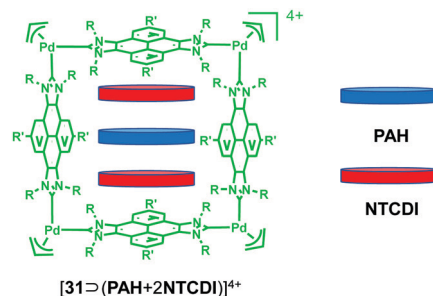


Fig. 4 The formation of the quintuple D–A–D–A–D stacks based on palladium-cornered metallosquare **31**.

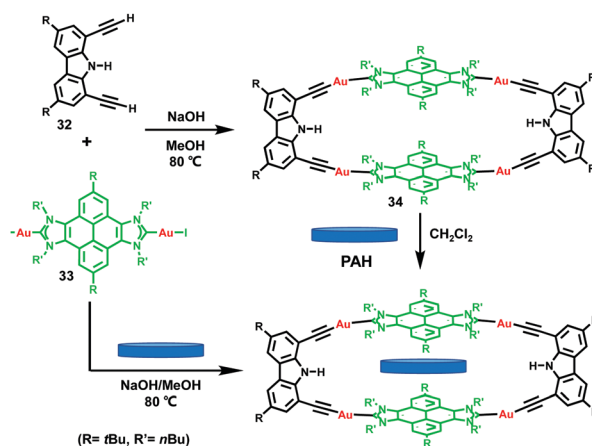


Fig. 5 The synthesis of the tetragold metallo-rectangle **34** and its complexes with diverse PAHs.

complex **33** and di-*tert*-butyl-diethynyl-carbazole **32** in the presence of NaOH as a base. Due to the existence of two cofacial pyrene panels separated by almost exactly 7 Å within its structure, the resultant tetragold metallo-rectangle was able to encapsulate a series of PAHs, such as anthracene, pyrene, triphenylene, perylene, coronene and corannulene (Fig. 5). It was found that for the planar PAH guests, the binding affinities increased with the number of electrons, which derived from dimensional matching. Meanwhile, for a bowl-shaped guest such as corannulene, a mutual induced-fit arrangement in the inclusion complex with a remarkable compression of the guest molecule and the concomitant expansion of the metallo-host was revealed, which was accompanied by a reduced binding affinity.

Pyrene-based 3D metallocages

Similar to the cases of 2D metallocycles as described above,¹⁹ pyrene could also be introduced into 3D metallocages as functional groups that endowed novel photophysical properties such as attractive emission properties as described below to the resultant SCCs. For instance, Johnson and coworkers²⁰ demonstrated the preparation of a novel class of polymer network, *i.e.* star PolyMOCs, in which metal-organic cages (MOCs) served as junctions that were cross-linked by polymer chains.

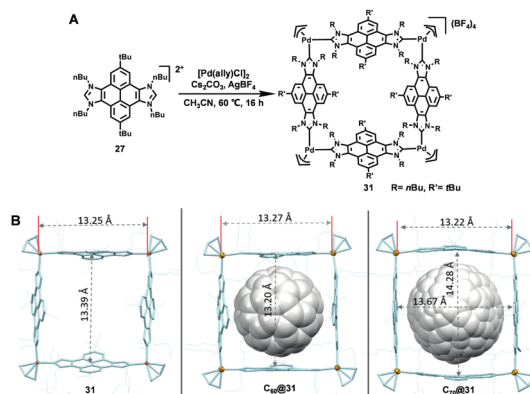


Fig. 3 The preparation of palladium-cornered metallosquare **31** (A) and the X-ray diffraction structures of **31** (left), $C_{60}@31$ (middle), and $C_{70}@31$ (right) (B). Adapted with permission from ref. 15. (Copyright 2019 Wiley-VCH.)

Starting from bispyridine-functionalized tetra-arm star polymers **35**, small bispyridine ligands **36**, and Pd^{2+} ions, a three-component assembly approach led to the modular synthesis of star polyMOC gels with Fujita's famous $\text{M}_{12}\text{L}_{24}$ MOCs as junctions (Fig. 6A). More importantly, by varying the amounts of the small bispyridine ligands and Pd^{2+} ions while keeping the concentration of polymer constant, a series of polyMOC gels with well-tailored structures, mechanical properties and dynamics was successfully constructed. In order to further install functionality into the resultant polyMOC gels, the pyrene-modified bispyridine ligand **37** was introduced as the fourth component to take part in the self-assembly process, which gave rise to novel fluorescent metallogels. As expected, upon the increase of the amount of the incorporated pyrene-modified bispyridine ligand, the fluorescence intensity of the corresponding metallogels is qualitatively proportional (Fig. 6B).

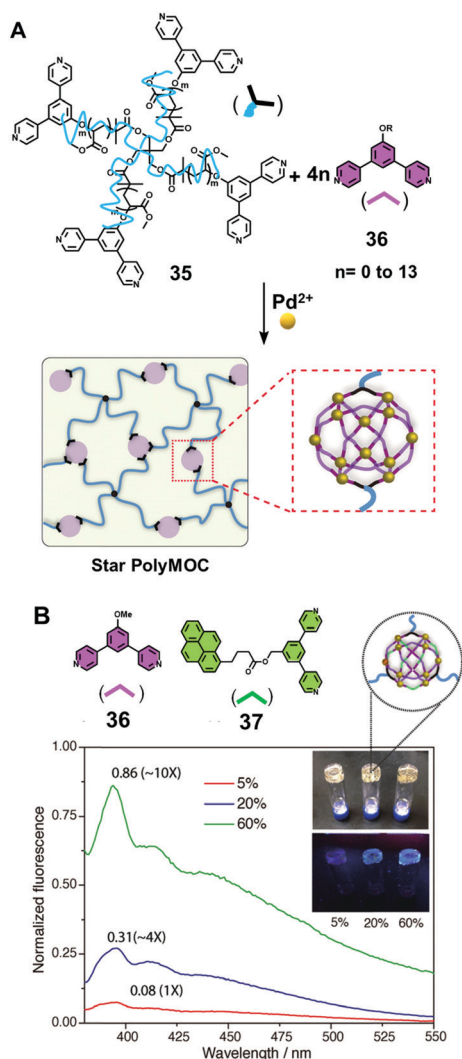


Fig. 6 The preparation of the star PolyMOCs via a three-component assembly approach (A) and the incorporation of pyrene moieties as functional groups (B). Adapted with permission from ref. 20. (Copyright 2017 Wiley-VCH.)

By using bis(aminophenyl)BODIPY and pyrene-appended formylpyridine as key subcomponents, a novel fluorescent metallo-supramolecular capsule **38** was successfully prepared by Nitschke and coworkers²¹ through the subcomponent self-assembly approach (Fig. 7A). In the resultant metallogel, the BODIPY moieties arranged along the edges and the pyrene moieties located at the vertices of the cage. Upon the binding of anionic guests such as acetate anions, a significant enhancement in the fluorescence of BODIPY was observed, while the pyrene fluorescence remained unchanged, which indicated that the binding of the anions was within the cage cavity. More interestingly, white light emission was observed when mixing metallogel **38** and perylene in a 1:3 ratio. Upon interacting with perylene, the emission intensity of the BODIPY moieties in the red region was remarkably enhanced, which played a key role in the generation of white emission (Fig. 7B). A control experiment, in which the mixture of the corresponding subcomponents as well as perylene only resulted in fluorescence in the blue and green regions rather than white emission, confirmed the importance of the cage architecture and its guest binding behaviour for white light emission.

Aiming at the design of novel fluorescent receptors for explosive detection, Mukherjee and coworkers²² demonstrated the construction of the first example of a platinum(II) tetragonal prism **41** through the two-component self-assembly of a new star-shaped pyrene-based planar Pt_4 acceptor **39** and an amide-based “clip” donor **40** (Fig. 8). Attributed to the existence of highly conjugated tetraethynylpyrene moieties, the resultant self-assembled prism displayed an excellent fluorescence quenching response even at parts per billion (ppb) concentrations of TNT.

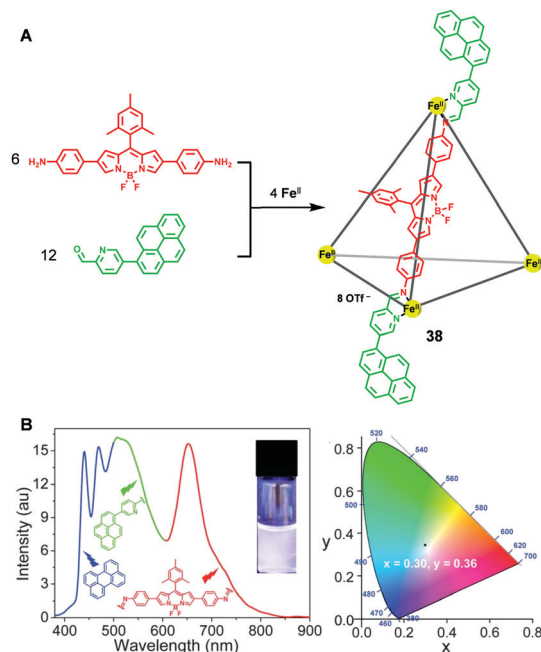


Fig. 7 The construction of the tetrahedral metallogel **38** via the subcomponent self-assembly approach (A) and the white light emission of the complex of the resultant metallogel and perylene (B).

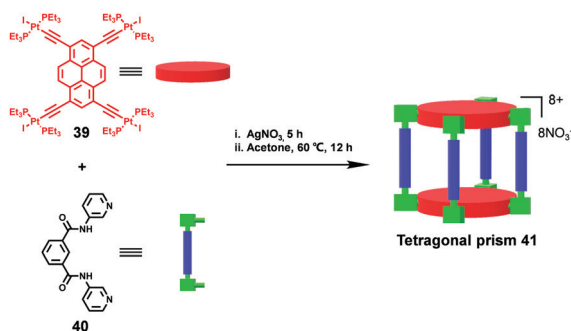


Fig. 8 The construction of the tetragonal prism **41** via coordination-driven self-assembly for nitroaromatic sensing.

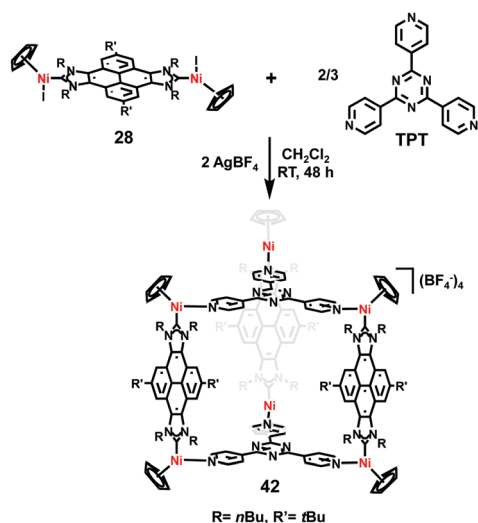
More importantly, compared with the planar tetratopic acceptor **39**, superior selectivity of prism **41** towards the nitro-aromatic explosives (NAEs) was observed, thus highlighting the important role of the π -electron-rich confined nanopocket for NAE encapsulation and detection. It is worth noting that, by employing a C–H-activation-directed assembly approach, Jin and coworkers²³ demonstrated the successful synthesis of a cuboid-shaped metallosupramolecular structure, in which two pyrene moieties also served as the floor and roof.

Through the self-assembly of aforementioned bis-nickel-pyrene-di-imidazolylidene complex **28** with 1,3,5-tripyrindyl-triazine (TPT) in the presence of AgBF_4 , Peris *et al.*²⁴ demonstrated the construction of a nickel-conjoined molecular prism **42** in 2018, whose molecular structure was determined by X-ray diffraction (Scheme 6). Similar to previously described metallosquare **31**, the resultant metallocage was also size-flexible and adaptable, which could also encapsulate both C_{60} and C_{70} fullerenes. Moreover, as evaluated by the binding constants, the metallocage preferred complexing with C_{70} over C_{60} , which was further confirmed by competitive experiments.

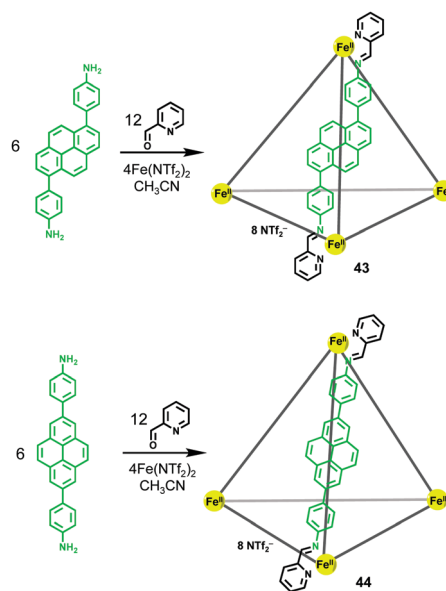
Considering the privileged role of pyrenes as planar and electron-rich panels for guest binding, two novel pyrene-edged

metallo cages **43** and **44** based on 1,6-pyrene and 2,7-pyrene scaffolds, respectively, were prepared by Nitschke and coworkers²⁵ though the subcomponent self-assembly approach (Scheme 7). As revealed by the ^1H NMR spectra, both metallocages consisted of three diastereomers, *i.e.* homochiral T ($\Delta\Delta\Delta\Delta/\Lambda\Lambda\Lambda\Lambda$), heterochiral C_3 ($\Delta\Delta\Delta\Lambda/\Lambda\Lambda\Lambda\Delta$) and achiral S_4 ($\Delta\Delta\Lambda\Lambda$). The ratios of these diastereomers for both metallocages were similar, both of which were close to the expected statistical distribution. Guest binding tests indicated that these two isomeric pyrene-edged metallocages displayed very different host–guest properties. For metallocage **43** based on a 1,6-pyrene scaffold, attributed to its well-enclosed cavity, it could serve as an excellent host for diverse guests ranging from large hydrophobic guests including fullerenes and PAHs to large, structurally complex natural products such as steroids. Interestingly, along with the fullerene binding processes, altered diastereomer distributions of **43** were observed, leading to the amplification of the diastereomers that can maximize the binding affinity. Meanwhile, in the case of metallocage **44** based on a 2,7-pyrene scaffold, due to the existence of an open cavity, it was unable to bind neutral guests. In a following paper,²⁶ the same group further examined the important role of the cavity enclosure for strong guest binding. In this study, besides pyrene, other PAHs such as triphenylene, naphthalene, anthracene and chrysene were introduced as the edges, leading to a series of aromatic-paneled metallocages. By evaluating the binding affinities of these resultant metallocages towards diverse guests, metallocages with large, offset aromatic panels such as chrysene that could maximize the cavity enclosure were proven to be the most effective hosts.

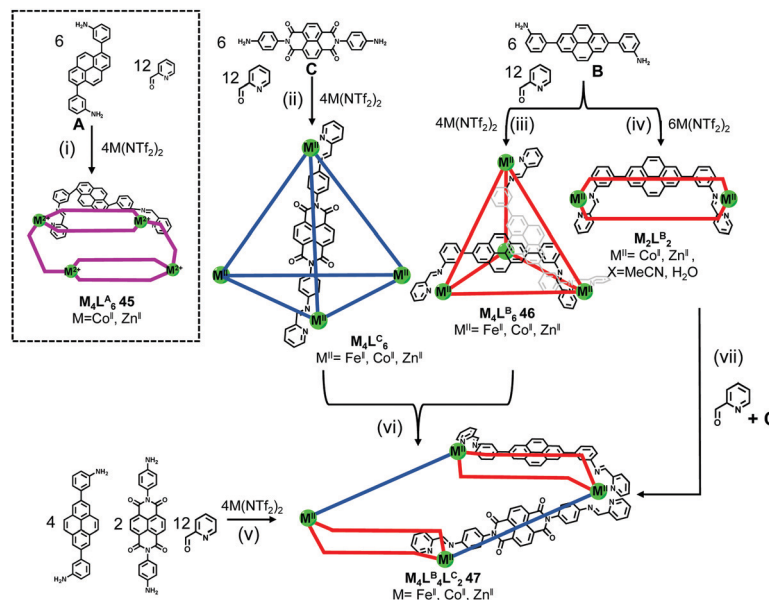
By using two isomeric bis(3-aminophenyl)pyrene derivatives **A** and **B**, rather than the isomeric bis(4-aminophenyl)pyrene derivatives described above, as key subcomponents, two unique



Scheme 6 The synthesis of nickel-cornered metallocage **42** for the encapsulation of fullerenes.



Scheme 7 The synthesis of pyrene-edged metallocages **43** and **44** via subcomponent self-assembly.



Scheme 8 The preparation of pyrene-based homeleptic and heteroleptic metallocages via subcomponent self-assembly.

metallocages, *i.e.* a C_2 -symmetric $M_4^{\text{II}}L_6$ assembly **45** with meridionally coordinated metal centers, and a C_3 -symmetric self-included $M_4^{\text{II}}L_6$ assembly **46** with facially coordinated metal centers ($M^{\text{II}} = \text{Fe}^{\text{II}}, \text{Co}^{\text{II}}, \text{and Zn}^{\text{II}}$), were constructed by Nitschke and coworkers²⁷ through the subcomponent self-assembly approach.

As revealed by the solid-state structures, face-to-face π -stacking interactions between the pyrene moieties were observed in both structures, which helped to stabilize the skeletons. Interestingly, the self-assembly of these two diamines, 2-formylpyridine (12 equiv.) and $\text{Zn}(\text{NTf}_2)_2$ in a 3:3:12:4 ratio resulted in the formation of a clean mixture of the homoleptic assemblies **45**-Zn and **46**-Zn, indicating a narcissistic self-sorting process that might be due to the geometrical differences as well as aforementioned favorable π -stacking within each individual metallocage structure. Furthermore, considering the favourable donor-acceptor stacking interactions between the pyrene units and naphthalenediimide (NDI) units, NDI-based diamine **C** was additionally chosen as a key subcomponent for the construction of heteroleptic metallocages. As expected, through either the direct subcomponent self-assembly approach (path v) or supramolecular transformation approach (path vi) as shown in Scheme 8, heteroleptic assemblies **47** were successfully prepared, from whose solid-state structure pyrene-pyrene-NDI stacks as the most favorable arrangements were identified.

Conclusion and outlook

During the past decades, more and more attention has been paid to the construction of pyrene-based SCCs including 2D metallocycles and 3D metallocages. Starting from pyrene-functionalized building blocks or pyrene derivatives themselves as key building blocks, diverse pyrene-based SCCs have been successfully constructed. For pyrene-based 2D metallocycles,

more attention was paid to the investigation of their photophysical properties originating from the pyrene moieties. Meanwhile, in the case of pyrene-based 3D metallocages, due to the existence of well-defined cavities with pyrene panels, research interest mainly focusing on the host-guest chemistry of such novel metallo-hosts was witnessed. Attributed to the unique structural and intriguing photophysical features of the incorporated pyrene units, the resultant SCCs displayed not only attractive emission properties but also impressive binding abilities towards diverse guests, endowing them with wide applications in sensing, host-guest chemistry, light harvesting, and so on. Although great achievements have been made in this area, the research on this topic is still in its infancy, and more in-depth investigations are in great demand. To name a few, firstly, aiming at further biological applications, the design and synthesis of water-soluble and biocompatible pyrene-based SCCs will be of great importance. Secondly, considering the wide applications of pyrene-based materials for organic electronics, how to construct and employ novel pyrene-based SCCs for the fabrication of photoelectric devices will be worth being further explored. Thirdly, regarding pyrene-based metallo-hosts, the expansion of the scope of guests and the improvement of the selectivity are still needed, which will be essential for practical applications such as separation, sensing, controllable capture/release *etc.* Considering the great power of coordination-driven self-assembly in the facile and efficient construction of novel SCCs with increasing complexity, a bright future of pyrene-based metallocycles and metallocages will be anticipated.

Conflicts of interest

The authors declare no competing interest.

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