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Lanthanide doped luminescence nanothermometers in the biological windows: strategies and applications

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The development of lanthanide-doped non-contact luminescent nanothermometers with accuracy, efficiency and fast diagnostic tools attributed to their versatility, stability and narrow emission band profiles has spurred the replacement of conventional contact thermal probes. The application of lanthanide-doped materials as temperature nanosensors, excited by ultraviolet, visible or near infrared light, and the generation of emissions lying in the biological window regions, I-BW (650 nm-950 nm), II-BW (1000 nm-1350 nm), III-BW (1400 nm-2000 nm) and IV-BW (centered at 2200 nm), are notably growing due to the advantages they present, including reduced phototoxicity and photobleaching, better image contrast and deeper penetration depths into biological tissues. Here, the different mechanisms used in lanthanide ion-doped nanomaterials to sense temperature in these biological windows for biomedical and other applications are summarized, focusing on factors that affect their thermal sensitivity, and consequently their temperature resolution. Comparing the thermometric performance of these nanomaterials in each biological window, we identified the strategies that allow boosting of their sensing properties.

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Introduction

Several chemical, physical and biological processes are temperature-dependent. Thus, precise and accurate measurement of the temperature is of paramount importance in countless industrial and research applications. 1-7

Nowadays, temperature sensors account for 80% of the world-wide sensor market and expect to reach a value of \$8.8 billion by 2027, according to Grand View Research, Inc.8 However, most of these temperature sensors are based on contact thermometers, where the thermal reading is achieved by direct physical contact of an invasive probe material with the body in which the temperature should be determined. Contact thermometers require conductive heat transfer and need to reach a thermal equilibrium between the sensor and the object under study. This connection might disturb the measurement of the temperature at the object, leading to an inappropriate determination of the exact temperature in some cases. These measurements are especially affected when dealing with nanoscale dimensions, in which the size of the object under study is smaller than the sensor head of the

thermometer. 9-13 In addition, contact thermometers allow the performance of only surface measurements. Modern requirements for temperature readings in areas such as microelectronics, photonics, nanomedicine and diagnosis, and microfluidics, among others, have led to the replacement of contact thermometers and the development of non-contact nanoscale thermometers. 14-16

Non-contact nanoscale thermometers can provide feedback of the local temperature of a given system with micro and nanoscale spatial resolution.¹⁷ The precise determination of the temperature at the nanoscale has attracted great interest, especially in nanomedicine and diagnosis fields, as many functions of the human body, including cell division, gene expression or enzyme reactions, are temperature-dependent. 18 Diseases such as the cellular pathogenesis of cancer lead to heat generation.¹⁹ In addition, temperature changes can also be induced intentionally to kill locally infected cells.²⁰ Therefore, achieving precise temperature reading information is crucial to heal infected cells and simultaneously avoid the destruction of surrounding healthy tissues.

High-resolution temperature non-contact measurement techniques operating in the micro/nanoscale are divided into several types using distinct criteria based on their operating principle. These methods provide different advantages and drawbacks;21 hence a proper selection according to the required spatial, temporal and temperature resolutions is

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essential. Non-contact techniques for the determination of the temperature include infrared thermometry, thermoreflectance, Raman scattering, interferometry, and non-optical and luminescence nanothermometry. ¹⁷

The principle of infrared thermometry is to determine the temperature by the amount of the thermal radiation emitted by the target which is being monitored. 17 Knowing the amount of infrared energy emitted by the object and its emissivity, the target temperature can be determined. This noncontact method is a well-implemented commercial technique and grants a temperature image profile of the surface of the target, achieving spatial, temporal and temperature resolutions of \sim 10 μ m, \sim 10 μ s and \sim 0.1 K, 17 respectively. However, this technique requires an additional knowledge of the spatial resolution and the emissivity of the target at the micrometer scale.² These are serious drawbacks for the application of this technique at the nanoscale because, with the determination of the temperature on the surface and not inside the target, the precise discrimination of the temperature of a living cell, for instance, strongly impacts on the analysis of its pathology and physiology and, in turn, on the optimization of therapeutic processes (e.g. in hyperthermal tumor treatments and photodynamic therapy). 22,23

Thermoreflectance thermometry extracts temperature readings from the dependence of the refractive index of the material on the temperature. This technique offers high temperature ($\sim 0.01~\text{K}$) and temporal resolution ($\sim 0.1~\mu \text{s}$), and the combination of qualitative and quantitative measurements. Despite this, it requires the calibration of the refractive index of the material of analysis, and its spatial resolution is restricted by the light diffraction limit.

Raman scattering thermometry is based on the dependence of the position of the Raman modes of the material of interest on temperature. Raman thermometry can be applied to liquids and solids, including powders, achieving spatial, temporal and temperature resolutions of $\sim 1~\mu m$, $\sim 10^6~\mu s$ and $\sim 0.1~K$, respectively. Regardless of these good parameters, it is a time-consuming technique and the number of the materials to which this technique can be applied is very small because the materials need to have a large Raman efficiency if applied as successful temperature probes.

Optical interferometry provides dual information: local temperature and local deformation due to the thermal expansion of the material under analysis. Interferometers depict the differences in optical paths between light beams that bypass the test section and pass directly through it. Although this technique can be integrated in remote detection systems with spatial, temporal and temperature resolutions of $\sim\!1$ µm, $\sim\!0.001$ µs and $\sim\!10^{-5}$ K, respectively, drawbacks related to cross talk with other stimuli, such as strain/stress and bending, and low spatial resolution in the transverse direction, limit their application.

Non-optical thermometry methods include several techniques developed for nanoscale thermometric applications such as scanning thermal microscopy,²⁸ deposition of metallic temperature sensors by nanolithography,²⁹ carbon nanotube

thermometry,³⁰ and biomaterial thermometry.³¹ Scanning thermal microscopy uses a small thermocouple with a junction diameter of the order of 20 to 100 nm, formed at the probe tip of an atomic force microscope, which is scanned over the surface of interest.³² It can also consist of a combination of a thermocouple and a resistance temperature, 33 providing submicrometric spatial resolution.2 This technique is limited to solid samples and offers slow acquisition times, requiring at the same time fundamental knowledge of the tip-sample heat transfer mechanisms.2 Nanolithography can be applied to deposit a platinum strip to form a thermal sensor on the surface of the material of interest. However, ensuring the robustness of the sensor and its chemical consistency is an important challenge for this technique. 17 Carbon nanotubes as thermal probes were originally proposed using gallium as the thermometric liquid filling them, and with the support of a scanning electron microscope to observe their meniscus.³⁰ These nanotubes have been used also to provide good thermal contact between an electron microscope and the sample with nanoscale resolution,34 but they presents also some limitations, such as the thermal contact resistance formed between the tip and the surface, which can modify the probe response, high thermal conducting especially for samples.34 Biomaterials, such as living cells, sense temperature by using their components (proteins, nucleic acids and mRNAs).³¹ The change of temperature in these components is manifested either by modifying their conformational structure directly or undergoing complex reactions that can be exploited to extract the temperature at the nanoscale level. 35-37 However, the molecular complexity associated with this temperature sensing is not yet fully understood, and the thermal response occurs within the physiological range of temperatures. Furthermore, this technique often induces conformational changes that are subtle and reversible.31

Luminescence thermometry (also called thermographic phosphor nanothermometry¹) refers to the relationship between the temperature and the luminescence properties of the light emitted by the luminophore to achieve temperature sensing.10 The major boost for luminescence thermometry at the nanoscale is attributed to the high temperature sensitivity and the easy detection setups required for these signals. With the change of temperature, different luminescence characteristics such as the intensity of the emission, spectral position, decay and rise time, and band positions and widths, may all be modified, which gives rise to different luminescence thermometry classes.10 Luminescence thermometry provides high spatial resolution (<10 μm) in short acquisition times (<10 μs) and high thermal resolution (0.1 K).2 Additionally, luminescent thermometers can operate even in harsh conditions such as in biological fluids, strong electromagnetic fields, cryogenic temperatures and fast-moving objects, without restricting their resolutions. 10,13,38,39

1.1. Luminescence nanothermometry

1.1.1. Fundamental principles of luminescence nanother- mometry. Luminescence refers to the emission of light from

an excited electronic state of a given substance.⁴⁰ This substance (or phosphor) consists of a luminescence entity, molecule or activator ion, embedded in a host (or matrix) material, and sometimes accompanied by a second entity that favors the absorption of light and transfers this energy to the activator, called a sensitizer. Sometimes, this substance may consist of only an activator embedded in a host, without the presence of a sensitizer. The characteristic luminescence properties of a given substance can be obtained by doping the host with relatively small amounts of foreign ions (activator and sensitizer). An activator incorporated into a host lattice gives rise to an emitting center which can be excited to generate luminescence.

A sensitizer incorporated into a host lattice is capable of absorbing more efficiently the energy from the excitation source and transferring it in a very efficient way to a neighbor activator that at the end will generate luminescence. 41 When exposed to a temperature, the characteristics of the luminescence of the phosphor (intensity, spectral position, decay and/or rise time, the band position and width), may change. Exploring the way in which these characteristics change, with the ultimate goal of determining the temperature of the phosphor, gives rise to luminescent thermometry. Fig. 1 depicts a general schematic illustration of the basic mechanisms of luminescence thermometry. For the sake of simplicity, a single center emission phosphor is illustrated, displaying a change in the intensity of the emissions as the temperature increases. This figure illustrates the class of band-shape luminescence thermometry, probably the widest explored class. In the figure we show how the phosphor is excited with an energy source $(h\nu)$ from the ground state level (indicated as 0) to an excited state level (indicated as 3), from where, according to the principle of conservation of energy, it will decay back to an intermediate or to the ground state with the emission of light or through the release of energy via non-radiative processes. 40,41 The decay could occur directly from the excited state (3) back to the ground state (0) via a radiative relaxation process, generating an emission line (labelled as "Emission 1" with intensity I₁) or via a non-radiative relaxing process (shown with a golden wavy arrow) through a lower intermediate excited level (either 2 or 1). From here, the second emission line (labelled as

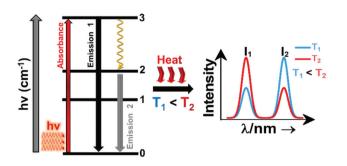


Fig. 1 Basic working principle of luminescence nanothermometry, illustrated *via* the change of the intensity of the emissions with the increase of temperature, as an example.

"Emission 2" with intensity I_2) can be generated when a second radiative relaxation process occurs to another lower level, or to the ground state. ⁴⁰ Please note that from the 2 and 1 intermediate excited states, additional emission lines can be generated, but for simplicity reasons, they are not illustrated in Fig. 1.

In order to be used in a luminescent nanothermometer, the generated emissions from the phosphor must fulfill several requirements. These requirements are related to the quality of the emission generated: they should exhibit high quantum vield and should be temperature-dependent. 10 The generated emissions of a phosphor are related to, among other different variables, the composition of the material, the purity level and the local temperature of the system. Thus, the principle of luminescence nanothermometry is to exploit the relationship between the properties of the luminescence emission process and the temperature to achieve temperature sensing by temporal or spectral analysis of the emission. 10 For example, in our simplified single center emission phosphor, upon increasing the temperature from T1 to T2 (presented in Fig. 1 by the red curved arrows labelled as "Heat"), this induces changes in the intensity of the emissions. Investigating the variation of the ratio of the intensities of the emissions $(I_1 \text{ and } I_2)$, the temperature can be determined before calibration of the luminescent thermometer.¹⁰ When the temperature changes are applied, the intensity of the emissions (and other luminescence properties as well) can vary due to the processes of electronic population or depopulation of the energy levels. 1,10

of luminescence 1.1.2. Classes nanothermometry. Temperature can affect the characteristics of the luminescence of the phosphor in different ways. Based on the particular parameter of luminescence whose temperature-dependence is analyzed, different classes of luminescence nanothermometry are developed. Fig. 2 illustrates six parameters that define the luminescence of a given phosphor: band-shape, intensity, bandwidth, spectral shift, lifetime and polarization, and their variation with an increase of temperature.10 Band-shape nanothermometry refers to the relative intensity between the different spectral lines of the luminescence spectrum (Fig. 2(a)). The changes induced by the temperature in the luminescence intensity are related to the thermal activation of processes of quenching and an increase in the probability of non-radiative mechanisms. This class of luminescence nanothermometry can be used in materials which possess radiative states with an energy gap (hereafter ΔE) of the order of 200 cm⁻¹ to 2000 cm⁻¹, *i.e*, the so-called "thermally coupled levels" (hereafter TCLs), and the extraction of thermal knowledge is achieved by the fluorescence intensity ratio (hereafter FIR), based totally on a Boltzmann type-distribution. 42 The main benefit of this class of luminescence nanothermometry is its independency of signal losses and the possible fluctuations in the excitation intensity. 10 TCLs with too small a ΔE (<200 cm⁻¹) will lead to a strong overlap of signals, while too large ΔE values (>2000 cm⁻¹) will result in the weak or null coupling of the electronic levels. 43 Furthermore, the performance of nanothermometers operating by these principles is

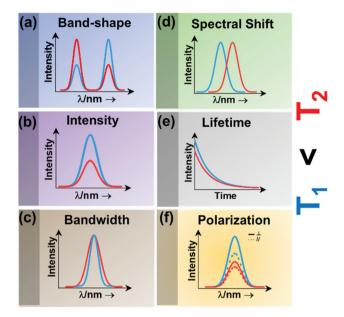


Fig. 2 Classes of luminescence nanothermometry: (a) band-shape, (b) intensity, (c) bandwidth, (d) spectral shift, (e) lifetime and (f) polarization.

restricted by the value of ΔE .⁴⁴ In addition, this class of luminescence nanothermometry includes single or dual emitting centers with a change in the intensity of at least two different emission bands in a material, even if they arise from non TCLs, for which different models for FIR have to be proposed, ²¹ as we will discuss in the following sections.

Intensity luminescence nanothermometry determines temperature through the analysis of the intensity of the luminescence generated by a single emission band (Fig. 2(b)). With the change of temperature, as for the case of the band-shape luminescence nanothermometry class, processes like quenching and the increase in probability of non-radiative mechanisms happening will display a decrease in intensity of the luminescence that can be correlated with the temperature. With the change of temperature, the intensity of the emission spectrum becomes less (or more) intense due to an overall change in the number of emitted photons. 42,46

Bandwidth luminescence nanothermometry extracts temperature information from the effect that the change of temperature has on the width of the emission lines of the luminescence spectrum (Fig. 2(c)).¹⁰ Generally, with an increase of the temperature, the emission lines become broader because of the thermal vibration of the luminescent center and its neighboring atoms/molecules.¹⁰ The degree of change of the emission line broadening is usually small, and as a result it can only be observed in systems showing inherent narrow emission lines and strongly temperature-dependent transitions.¹⁰

Spectral shift luminescence nanothermometry is based on the analysis of the spectral positions of the emission lines, which change due to their dependence on temperature. When the temperature increases, the energy value of the emitting level changes, manifested by a shift in the emission wavelength (Fig. 2(d)). This ΔE depends on a large variety of

temperature-dependent parameters of the emitting material, including the refractive index and the inter-atomic distances, among others. The shift in the spectral position is usually related to thermally induced strains in the environment of the thermal probes, occurring as a result of electron-phonon interaction.

Lifetime nanothermometry analyzes the temperature dependence of the luminescence lifetime (Fig. 2(e)). Luminescence lifetime studies the time interval in which the emitted intensity decays down to 1/e of its maximum value after a pulsed excitation. The experimental lifetime is related to the contribution of the radiative, non-radiative or multiphonon and quenching mechanisms. With the increase of temperature, the lifetime generally shortens due to the increase of the probability of quenching mechanisms happening. The lifetime is also related to the dimension of the material. The lifetime of nanocrystals is shorter when compared with that of macrometric particles. This difference is related to lattice distortion, and a bigger number of surface defects related to the higher surface-to-volume ratio in nanomaterials, which overall leads to an increase of the radiative rates and higher quenching rates.

Polarization luminescence nanothermometry is based on the influence of temperature in the polarization anisotropy parameter, which is the ratio between the luminescence intensities emitted at two orthogonal polarization states (Fig. 2(f)).¹⁰

In the following sections, we will analyze how these different luminescence nanothermometry classes have been used to determine temperature using lanthanide-doped luminescent nanothermometers operating in the biological windows, with special emphasis on the different strategies developed to improve their performance. For a detailed comprehensive understanding of the six luminescence thermometry classes and the determination of their thermal sensing capacity, the reader is requested to refer to the review of Jaque $et\ al.^{10}$ and Brites $et\ al.^{48}$

1.1.3. Materials used to develop luminescent nanothermometers. Different luminescent nanomaterials have been tested for providing a contactless temperature reading through luminescence nanothermometry (Fig. 3). In order to boost their potential nanothermometric applications in different fields, but specially in biomedicine, several factors, including strong luminescence, good photostability, no photobleaching effect, low or barely no toxicity, good dispersibility in biological media and outstanding optical properties that allow for deep tissue light penetration, ⁴⁹ should be taken into account.

Luminescent materials tested as potential nanothermometers include fluorescent proteins, ^{50–54} nanogels, ^{3,55–57} polymers, ^{58–61} organic–inorganic hybrids, ^{43,62–64} nanodiamonds, ^{65–67} metal organic frameworks (MOFs), ^{39,68–71} organic dyes, ^{72–76} semiconductor quantum-dots (QDs), ^{77–81} and lanthanide (Ln³⁺) doped nanomaterials. ^{10,21,48,82} Fig. 3 highlights the pros and cons of these materials as potential luminescent nanothermometers.

Fluorescent proteins, either in their simple form or genetically encoded, can be used to measure temperature at the intracellular level.^{50–54} These nanothermometers can be

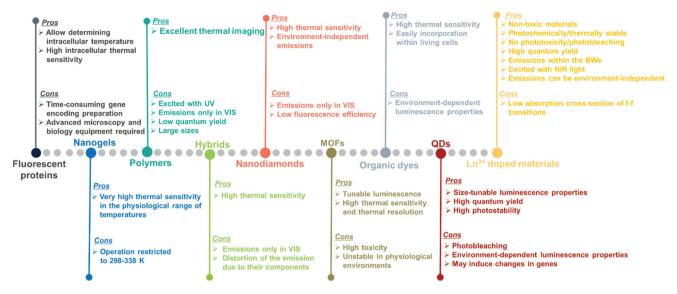


Fig. 3 Pros and cons of different types of material used to develop luminescent thermometers.

applied to obtain thermal images of heated single cells and to study thermogenesis, a fundamental process in cell biology. Regardless of the relatively high intracellular thermal sensitivity achieved with these nanothermometers, drawbacks, including time-consuming gene encoding, the need for specialized microscopy equipment and advanced molecular biology techniques for gene modification, ^{83,84} have prevented their future applications.

Nanogels portray nanoscale-sized three-dimensional hydrogel materials formed by crosslinked swellable polymer networks with a high capacity to hold water, without actually dissolving into the aqueous medium.85 Below a particular temperature, normally around 300 K, the structure of the nanogel swells by absorbing water into its interior, which causes a quenching of the fluorescence that is gradually recovered when the temperature increases above this temperature since the nanogel shrinks because of the release of water molecules.3,55-57 As luminescent nanothermometers, nanogel compounds present an important advantage: the most relevant changes in their fluorescence intensity occur within the physiological range of temperatures. A particular disadvantage is that the emission intensity of the nanogels becomes highly temperature sensitive only in a reduced temperature range (298-338 K).10

Polymers stands for large macromolecules composed of repeated structural units, called monomers. The presence of luminescent monomers within the polymers allows their use as luminescent thermometers. The luminescence of these compounds is usually found in the visible (VIS) range (500–600 nm) after being excited with ultraviolet (UV) light, 58–61 which can induce phototoxicity if applied to extract thermal information from biological tissues. Despite the good results demonstrated in thermal imaging, these thermometers usually exhibit drawbacks related to their low quantum yield, 7 relatively large sizes, and the dependence of the

luminescence not only on temperature, but also on the local concentration of emitting centers. 10

An organic–inorganic hybrid can be defined as a single particle that combines two different materials belonging to these categories. ⁸⁸ In the field of luminescence nanothermometry, these materials can be used by exploiting the complementary functionalities of their constituent materials to develop multifunctional platforms, such as temperature sensing and heat generation, for further applications. ^{62–64,89} Regardless of their good temperature-sensing properties, limitations are also encountered. These limitations are related to nanothermometers being operative only in the VIS region, which represents a low penetration depth in biological tissues and a distortion of the spectra recorded due to the non-negligible absorption of the different components encountered in them. ⁹⁰

The application of nanodiamonds as luminescent nanothermometers is attributed to the presence of particular local point defects (nitrogen vacancy centers) in their structure. These nanodiamonds are fluorescent and display a very sensitive temperature transition between two ground quantum spin states. The advantages of these nanothermometers rely on their high thermal sensitivity and on the fact that nitrogen vacancy centers are inner defects, hence, in theory, making these thermometers environment-independent. However, when applied to biological tissues, the light generated from these materials might be affected by the biological tissue components. Furthermore, their emissions are often located in the VIS, and the low fluorescence efficiency displayed is a limiting factor especially when acquiring cellular thermal images. Services

Metal organic frameworks (MOFs) have emerged as an important class of luminescent materials due to their multiple luminescent centers and tunable optical properties.^{39,68-71} The tunability of the optical properties of MOFs is related to the fact that the metal nodes, organic linkers, and guest molecules

within porous MOFs all can potentially generate luminescence. Furthermore, the inherent porosity and large surface areas of MOFs offer great opportunities for intermolecular interactions between the frameworks and guest molecules, influencing the coordination environment and the energy transfer processes that can occur among them, thus modulating their luminescence properties.⁷¹ However, it should be admitted that, to use these luminescent thermometers in biological environments, besides high sensitivity and spatial resolution, non-toxicity and stability in the physiological environment are required, which often lacking in MOF-based thermometers.71

Organic dyes are organic compounds with strong luminescence intensity, in general. 10 Despite their high temperaturesensing properties and the ability to be easily incorporated into living cells, the spectral properties of the luminescence band depend on many factors, including temperature, solvent used, concentration, and pH, among others. 10,92 Being specially dependent on the environment in which the organic dye is embedded makes it necessary to recalibrate the thermometers in each particular environment, and thus, the temperature-sensing ability will strongly depend on the medium where the compound is embedded.92

Quantum dots (QDs) are highly attractive nanomaterials for application as luminescent nanothermometers, due to their size-tunable spectroscopic properties,^{77–81} narrow emissions, high photostability and high quantum yields. 93,94 For years, concerns regarding their phototoxicity and cytotoxicity towards biological matter due to UV light excitation, 86,95 and emission wavelengths limited to the range shorter than 1000 nm, have resolved by the preparation of core@shell structures. 94,96,97 Despite this, limitations still remain concerning the variation of the luminescence parameters (intensity, lifetime and spectral shift) with the environment, including surfactants and ligands, which leads to a source of error in thermal readings, 98 and their genetically-induced changes. 95,97

Lanthanide (Ln³⁺) doped nanomaterials, due to their peculiar electronic configuration, exhibit stable and narrow emissions covering a wide range of the electromagnetic spectrum, upon proper selection of the dopants and the transparency of the host, with high emission quantum yields (>50% in the visible range). 99-102 In addition, these barely non-toxic materials¹⁰³ can generate their emission lines upon excitation via near infrared (NIR) light, which stands for a low-cost excitation source with no photobleaching, no autofluorescence and no phototoxicity upon biological matter. 21,48 Furthermore, the generation of NIR light and its use instead of VIS or UV light allows for deep-tissue penetration, 86 specifically when the wavelengths of the emissions fall within the so-called biological windows 104,105 (section 1.3). A drawback of these materials relates to the low absorption cross section of the 4f-4f electronic transitions because of their forbidden nature. 106 Different strategies have been implemented to increase the absorption properties of Ln3+-doped materials, including plasmonic enhancement, organic-dye sensitization, and coupling with semiconductors. 107,108 Thus, inspired by these properties

of Ln3+-doped luminescent nanomaterials, the rest of this review will be focused on the performance of these nanomaterials to extract information on the local temperature of a given system when the generated emissions fall in the biological window regions. Furthermore, the strategies explored to boost their ability to achieve better thermal readings will be analyzed.

1.2. Performance of a lanthanide-doped luminescent nanothermometer

Regardless of the luminescent nanothermometry class used to extract temperature information, the performance of a lanthanide-doped nanothermometer, as well as that of other thermometers based on other materials, can be evaluated and compared with other nanothermometers, based on their thermal sensitivity, temperature resolution, spatio-temporal resolution, repeatability and reproducibility. Detailed information about all these parameters is summarized elsewhere. 21,48 Thus, here we will emphasize, in general terms, how these parameters can be calculated.

The thermal sensitivity expresses the degree of change of the thermometric parameter (generally denoted by Δ) per degree of temperature. The thermal sensitivity can be expressed as the absolute thermal sensitivity and the relative thermal sensitivity. The absolute thermal sensitivity (S_{abs}) is defined as the rate of change of the thermometric parameter (Δ) with the temperature and is expressed as:¹⁰⁹

$$S_{\rm abs} = \frac{\partial \Delta}{\partial T} \tag{1}$$

 $S_{\rm abs}$ is expressed in units of K^{-1} . This parameter is strictly related to the experimental setup used and the sample characteristics; thus, its use is restricted to comparing only nanothermometers of the same nature and tested under the same conditions.

To compare the thermometric performance of different nanothermometers, independently of their nature or material employed, the relative thermal sensitivity (S_{rel}) can be used. S_{rel} expresses the maximum change in the thermometric parameter (Δ) for each temperature degree and it is defined as:^{38,42,110}

$$S_{\rm rel} = \frac{1}{\Delta} \left| \frac{\partial \Delta}{\partial T} \right| \times 100\%$$
 (2)

S_{rel} is expressed in units of % change per degree of temperature change (% K⁻¹).

The temperature resolution (also defined as temperature sensitivity), δT , represents the smallest temperature change that can be resolved in a given measurement.38

Depending on the experimental detection setup used, the acquisition conditions applied, and the signal-to-noise ratio in the experiment, δT might change.³⁸ δT is measured in kelvin (K) and it is expressed as:111

$$\delta T = \frac{1}{S_{\rm rel}} \frac{\delta \Delta}{\Delta} \tag{3}$$

where $\delta\Delta$ is the uncertainty in the determination of Δ that depends on the experimental detection setup used.

Another way of determining δT experimentally is by recording several consecutive emission spectra at a fixed temperature. By using the calibration curve of the nanothermometer, the range of variability of the calculated temperature is determined. Results display a gaussian distribution with a mean value and a standard deviation value, corresponding to the average temperature and uncertainty (δT) of the measurement. 21,112

The spatial resolution represents the minimum distance between points presenting a temperature difference higher than δT , when the temperature is measured at different spatial positions.²¹ The spatial resolution (δx) is expressed as:²¹

$$\delta x = \frac{\delta T}{|\vec{\nabla}T|_{\text{max}}} \tag{4}$$

where $|\vec{\nabla}T|_{\rm max}$ is the maximum temperature gradient in the system.

The temporal resolution of the measurement (δt) is defined as the minimum time interval between measurements presenting a temperature difference higher than δT and is expressed as:⁴⁸

$$t = \frac{\delta T}{\left|\frac{\mathrm{d}T}{\mathrm{d}t}\right|_{\mathrm{max}}} \tag{5}$$

where $\left|\frac{\mathrm{d}T}{\mathrm{d}t}\right|_{\mathrm{max}}$ is the maximum temperature change per unit of time.

The repeatability indicates the ability of a nanothermometer to provide the same temperature measurement under the same identical conditions, 113 and is defined as:

$$R = 1 - \frac{\max|\Delta_{c} - \Delta_{i}|}{\Delta_{c}} \tag{6}$$

where Δ_c is the mean thermometric parameter and Δ_i is the value of each measurement of the thermometric parameter.

Finally, the reproducibility refers to the variations of the same measurement in different modified circumstances, including different measurement methods, different equipment in use, different observers, or measurements made over a period of time within the true level of the already measured data. 114

From these performance parameters, the most reported ones are the thermal sensitivity and the temperature resolution, while the rest are scarcely used to determine the performance of a luminescent thermometer, despite their importance for the use of these thermometers in real applications.

1.3. Biological windows

A good use of luminescent thermometers in biological tissues requires knowledge about several parameters that influence the excitation of the nanothermometers by optical means, and also the detection of the emitted light by the materials. These parameters, including the emission characteristics, the optical path length through biological tissues and the volumetric energy distribution, are linked to the absorption and scattering

properties of the biological samples. 115,116 The selection of these parameters should meet the overall goal of achieving a high penetration depth in real biological samples, allowing deeper thermal readings, and favoring the development of an efficient and reliable luminescent material for early detection and diagnosis of diseases. Applying NIR light instead of UV or VIS as the excitation source results in a more efficient and deeper penetration in biological tissues due to the reduced scattering and absorption of light with longer wavelengths. 105 Spectral regions where both tissue absorption and scattering are minimized are known as biological windows (hereafter BWs). 105 The main absorbers of light in biological tissues include water, hemoglobin, melanin and lipids, and their size, composition and morphology are responsible for the scattering of light.117 Water absorbs in the NIR region at around 980 nm and it is nearly transparent in the visible region. 118 Hemoglobin (oxy-hemoglobin and deoxy-hemoglobin), from its side, is the component responsible for the absorption of visible light by blood. The highest absorbance of hemoglobin is in the visible region and it decreases above 600 nm. 119 The absorbance of melanin is inversely proportional to the increase of the wavelength from the visible to the NIR. 120 Regarding the absorbance of lipids in vivo, the data available in the literature are quite scarce; however as reported by Smith et al., the absorption contribution of fat components in biological tissues is less important than that of water and blood. 105 Scatterers of light in biological tissues include cell nuclei, mitochondria, cell membranes and whole cells.121 Light scattering decreases exponentially as the wavelength increases from the visible to the NIR. 122

Within these BWs, four distinctive wavelength regions have been established (Fig. 4(a)):

- First biological window (I-BW) (650-950 nm)
- Second biological window (II-BW) (1000-1350 nm)
- Third biological window (III-BW) or short-wavelength infrared (SWIR) window (1400–2000 nm)
 - Fourth biological window (IV-BW) (centered at 2200 nm)

The I-BW is limited at short wavelengths by the absorption of light by oxygenated blood and at long wavelengths by the absorption of water due to the second overtone of its stretching band, a combination of symmetric and antisymmetric stretching modes.¹²³ This biological window encounters limitations attributed to signal interference by biological tissue autofluorescence, which produces background noise, thus limiting the maximum tissue penetration to 1–2 cm.¹²⁰ This biological window is also known as the therapeutic window.¹²⁴

The II-BW provides an improved signal-to-noise ratio by a factor over 100-fold by effectively filtering out autoflorescence from biological tissues. The limits of the II-BW lie in the short wavelength by the second overtone of the stretching band of water and at longer wavelengths by the combination of the bending and stretching (symmetric and antisymmetric) modes of water also. 123

The III-BW or SWIR, despite the limitation from the absorption bands of water (stretching and bending modes), 123 offers improved image contrast and higher penetration depths

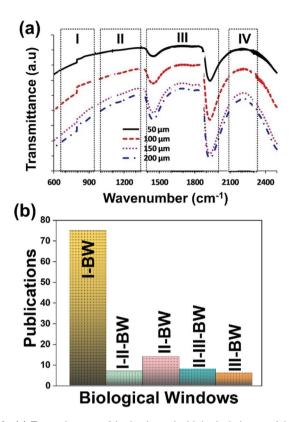


Fig. 4 (a) Transmittance of brain tissue in biological tissues. Adapted with permission from ref. 124, Copyright 2016, Wiley-VCH Verlag GmbH & Co. KGaA. (b) Number of publications of luminescence thermometry in every biological window, including also the thermometers whose performance is evaluated by emissions lying in two different BWs simultaneously.

compared with the other BWs due to the reduction of light scattering.

The IV-BW, centered at 2200 nm, is the least explored one due to the lack of sensitive detectors operating at these wavelengths. The advent of new photodetectors (InSb in single or array photodiodes) and excitation laser sources (supercontinuum lasers operating in the range from 400 nm to 2500 nm) now makes the IV-BW a viable spectral domain. The IV-BW shows similar values of light attenuation for skin and other soft tissues comparable with that of the II-BW, lower than the III-BW but better than the I-BW.

Fig. 4(a) shows the variation of transmittance of rat brain tissue as a function of the thickness of the biological sample, covering all four BWs.¹²⁴ The maximum transmittance is located in the III-BW, regardless of the thickness. The transmittances in the II-BW and the IV-BW are close to each other and are much higher than the transmittance in the I-BW, indicating a better potential application for deep imaging.¹²⁴ Fig. 4(b) presents the number of publications implementing lanthanide-doped materials as luminescent thermometers operating in the different biological window regions. Apart from the main four BWs, the regions in which the luminescent nanothermometer performance has been evaluated with emis-

sions lying partly in the I- and II-BWs (named as I-II-BW) and II- and III-BWs (named as II-III-BW) are also presented. The biological window with the highest number of publications is the I-BW, whereas the IV-BW represents, up to now, an unexplored region for luminescence thermometry. Recently, however, an increase has been observed in the number of publications regarding the II-BW and III-BW, mainly due to the benefits of better image contrast and higher penetration depths. The following sections have been structured according to the emission lines used for luminescent thermometry lying within any of these BWs, despite the fact that their excitation is provided in a spectral range lying in a different biological window.

Several reviews have been published about luminescence thermometry, and more specifically about the use of Ln3+doped materials devoted to this practice. Precisely, Brites et al.48 published in 2019 a review on these materials, focused on strategies to enhance the thermometric performance and on cutting-edge applications of Ln3+-doped materials mainly operating in the visible region. The present review unveils key guidelines to promote the application of Ln³⁺-doped materials only within the BW spectral regions. We stress the mechanisms behind the generation of the emission lines within the BWs, something not analyzed in detail in the review mentioned above, and emphasize the role of the different excitation sources. We point out the different fitting models applied to evaluate the thermometric performance of Ln³⁺doped luminescent thermometers. We highlight strategies towards the development of highly sensitive nanothermometers by stating the important role of the morphology of the host in which these Ln3+ ions are embedded, such as the shape of core@shell or multishell structures. We define the differences between the thermometric performances based on the combination of emissions arising from the same Ln³⁺ ion, different Ln3+ ions or Ln3+ ions combined with transition metal ions or quantum dots. Potential applications of these nanothermometers in the biological/biomedical field and possible applications outside of these fields are presented. Finally, we share our opinions on future challenges for luminescence nanothermometry in the BW spectral regions, and possible routes to overcome these.

2. Lanthanide-doped luminescent nanothermometers operating in the I-BW

The I-BW is the most explored biological window in relation to the performance of lanthanide-doped luminescent thermometers up to now. The most encountered lanthanide active ions are ${\rm Tm}^{3+}$ and ${\rm Nd}^{3+}$, mainly due to their emission bands in the 700–800 nm and 800–950 nm regions, respectively (Table 1). This section and the later ones will be arranged as follows: the lanthanide ion with the best thermometric performance (only the maximum value of $S_{\rm rel}$ and the associated

erature resolution (δT) is given at the same temperature. We indicate with an asterisk $S_{\rm rel}$ or δT values calculated by us using the parameters published in the corresponding references. The double line separation between rows stands for different types (single or dual emitting centers) of lanthanide-doped nanothermometers, as discussed in the corresponding subsections Table 1 Ln3+-doped luminescent thermometers operating in the I-BW. The table includes activators (A) and sensitizers (S). The excitation (A_{ex.}) and emission (A_{em.}) wavelengths are indicated in nanometers (nm), together with the corresponding electronic transition with which the emissions are associated. ΔT stands for the temperature range where the temperature reading was investi- $\Delta\lambda$ for spectral shift, and Δ for bandwidth The minimum temp-(S_{rel.}) is given at the temperature at which this value was obtained. gated. The thermometric parameter (Δ) indicates the luminescent nanothermometry class used in each case: FIR for band-shape, I for intensity ratio, The maximum relative thermal sensitivity ratio. VPR stands for valley-to-peak thermometry, respectively.

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A	S	Host	$\lambda_{ m exc}$ $({ m nm})$	$\lambda_{ m em} \left({ m nm} ight)$	Transitions	$\Delta T(\mathrm{K})$	۵	$S_{\rm rel}/T = (\% \ { m K}^{-1})/{ m K}$	$\delta T(K)$	Ref.
										ĺ
Nd^{3+}	Nd^{3+}	YAG	290	780-840, 870-920	1	77-850	$FIR_{780-840/870-920}$	1.02/200	0.49*	155
$^{+\varepsilon}$ PN	Nd^{3+}	${ m LiLuF}_4$	808	862, 866	\rightarrow $^{4}I_{9/2}$	77-575	I_{862}/I_{866}	0.62/77	9.0	162
Nd^{3+}	Nd^{3+}	SrF_2	573	800–950	$^{4}\mathrm{F}_{3/2} ightarrow ^{4}\mathrm{I}_{9/2}$	293-328	$FIR_{800-950(2)/(1)}$	0.61/293	2.1	163
Nd^{3+}	Nd^{3+}	${ m LiLuF_4}$ ${ m @LiLuF_4}$	793	883, 887	1	293-318	I_{887}/I_{883}	0.58/293	3.4	164
$^{+c}$ PN	$^{+e}$ PN	${ m LiLaP_4O_{12}}$	808	850-900	1	83-600	$\Delta \lambda_{850-900}$	0.47/323	1.05*	165
$^{+e}$ PN	Nd^{3+}	$RbLaP_4O_{12}$	808	850-900	$^4F_{3/2} \rightarrow ^4I_{9/2}$	83-600	$\Delta\lambda_{850-900}$	0.47/323	1.05*	165
$^{+c}$ PN	Nd^{3+}	$\text{KLaP}_4\text{O}_{12}$	808	850-900	1	83-600	$\Delta \lambda_{850-900}$	0.47/323	1.05*	165
$^{+e}$ PN	Nd^{3+}	$NaLaP_4O_{12}$	808	850-900	$^4F_{3/2} \rightarrow ^4I_{9/2}$	83-600	$\Delta\lambda_{850-900}$	0.42/323	1.2*	165
$^{+e}$ PN	Nd^{3+}	$LiLaP_4O_{12}$	808	850-900	1	83-600	$\Delta \nu_{850-900}$	0.32/323	1.5*	165
$^{+e}$ PN	Nd^{3+}	Y_2O_3	808	870,920	1	298-333	I_{870}/I_{820}	0.31/298	1	166
$^{3+}$	Nd^{3+}	$R\overline{b}L\overline{a}P_4O_{12}$	808	850-900	1	83-600	$\Delta \nu_{850-900}$	0.28/323	1.1^{*}	165
$^{3+}$	Nd^{3+}	YNbO_4	752	880,890	1	303-473	I_{880}/I_{890}	0.28/303	1.1	167
$^{+e}$ Nq $^{3+}$	Nd^{3+}	LaF	808	865, 885	_ ↑	283-333	I_{865}/I_{885}	0.26/283	1.95*	168
$^{+e}$ PN	Nd^{3+}	$NaLaP_4O_{12}$	808	850-900	$^4F_{3/2} \rightarrow ^4I_{9/2}$	83-600	$\Delta \nu_{850-900}$	0.24/323	2.1*	165
$^{+e}$ pN	Nd^{3+}	$\text{KLaP}_4\text{O}_{12}$	808	850-900	_ ↑	83-600	$\Delta \nu_{850-900}$	0.20/323	2.5*	165
$^{+e}$ PN	Nd^{3+}	YVO_4	808	879,887	1	298-333	I_{879}/I_{887}	0.19/298	2.6*	169
$^{+e}$ pN	Nd^{3+}	$\mathrm{Gd_{3}Sc_{2}Al_{3}O_{12}}$	908	936, 946	1	293-323	$FIR_{936/946}$	0.21/293	2.38*	170
Nd^{3+}	Nd^{3+}	YAG	808	938, 945	$^{4}\mathrm{F}_{3/2} o ^{4}\mathrm{I}_{9/2}$	283-343	$FIR_{938/945}$	0.15/283	0.3	171
Nd^{3+}	Nd^{3+}	BiVO_4	750	872, 902	1	310 - 523	$FIR_{872/902}$	0.14/310	3	172
$^{+e}$ PN	Nd^{3+}	$NaYF_4$	830	863, 870	1	273-423	$FIR_{863/870}$	0.12/273	0.1	173
$^{+\epsilon}$ PN	Nd^{3+}	$\mathrm{KGd}(\mathrm{WO}_4)_2$	808	883, 895	1	298-333	FIR _{883/895}	0.12/298	0.43	174
$^{+e}$ PN	$^{+e}$ PN	NaNdF ₄ @NaYF ₄ @NaYF ₄ :Nd	808	857, 863	1	293-333	FIR _{857/863}	0.11/293	4.5*	175
$^{+e}$ pN	$^{+e}$ PN	$RbLaP_4O_{12}$	808	850-900	1	83-600	$FIR_{880(2)/(1)}$	0.108/323	4.6*	165
$^{+e}$ PN	Nd^{3+}	$NaLaP_4O_{12}$	808	850-900	1	83-600	$FIR_{880(2)/(1)}$	0.104/323	4.8*	165
Nd ³⁺	Nd^{3+}	$\text{KLaP}_4\text{O}_{12}$	808	850-900	1	83-600	${ m FIR}_{880(2)/(1)}$	0.097/323	5.1*	165
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Na ³⁺	$y_{b^{3+}}$	Oxyfluoride glass	086	750, 863	$F_{7/2} \rightarrow 1_{9/2}$ $F_{3/2} \rightarrow 1_{9/2}$ $^{4}F_{-} \rightarrow ^{4}F_{-} \rightarrow ^{4}F_{-} \rightarrow ^{4}F_{-}$	303-623	F1R _{750/863}	3.27/303	0.15* $0.16*$	176
Nd ³⁺	$^{1}_{\text{Vh}^{3+}}$	SIWO4 CaWO.	980		$^{1}_{4}$ $^{1}_{7/2}$ $^{2}_{7}$ $^{2}_{19/2}$ $^{2}_{19/2}$ $^{4}_{19/2}$ $^{4}_{19/2}$ $^{4}_{19/2}$ $^{4}_{19/2}$	303-873	FIR-50/863	3/303	0.16*	178
Nd ³⁺	$^{rb}_{3+}$	Al, B,O,	977.7		$^{4}F_{r,0} \rightarrow ^{4}I_{c,0} (Nd^{3+}) \ ^{2}F_{r,0} \rightarrow ^{2}F_{r,0} (Vh^{3+})$	298-333	FTR 200/803	2.6/2.98	0.19*	179
Nd ³⁺	$^{1}_{ m V}$	$NaYF_A$	980		$^{4}F_{7/2} \rightarrow ^{4}I_{9/2}, ^{4}F_{3/2} \rightarrow ^{4}I_{9/2}$	293–333	$FIR_{750/863}$	2.4/293	0.20*	180
Nd^{3+}	Yb^{3+}	Oxyfluoride glass	086		$^{4}\mathrm{F}_{7/2} \rightarrow ^{4}\mathrm{I}_{9/2}, ^{4}\mathrm{F}_{5/2} \rightarrow ^{4}\mathrm{I}_{9/2}$		FIR _{750/863}	2.05/303	0.24*	176
Nd ³⁺	Yb^{3+}	Oxyfluoride glass	086		${}^{4}F_{5/2} \rightarrow {}^{4}I_{9/2}, {}^{4}F_{3/2} \rightarrow {}^{4}I_{9/2}$	303-623	FIR _{750/863}	1.95/303	0.25*	176
Nd ³⁺	$V_{b^{3+}}^{3+}$	La_2O_3	980	750, 803	$^{4}F_{7/2} \rightarrow ^{4}I_{9/2}, ^{4}F_{3/2} \rightarrow ^{4}I_{9/2}$		${ m FIR}_{750/803}$	1.6/293	0.1	181
NG	ID	$Al_4B_2O_9$	9///	864, 920	$F_{5/2} \rightarrow I_{9/2} (INC), F_{5/2} \rightarrow F_{7/2} (ID)$	299-333	F1K800/920	1.38/299	0.30	1/9
Nd^{3+} , Eu^{3+}	Nd^{3+} , Eu^{3+}	Ba ₂ LaF ₇	578.3	699 (Eu^{3+}) , 800 (Nd^{3+})	$^{5}_{\Sigma_{0}} \rightarrow ^{7}_{7-4} \left(Eu^{3+}_{2+}, ^{4}_{F_{3/2}} \rightarrow ^{4}_{19/2} \left(Nd^{3+}_{2+3+} \right) \right)$		I_{699}/I_{800}	2.2/290	0.22*	182
Nd., Eu	Nd., Eu.	YVO_4	290	696 (Eu~), 812 (Nd~)	$^{\circ}D_{0} \rightarrow ^{\circ}F_{4}\left(\mathrm{Eu^{\circ}}\right), ^{\circ}F_{3/2} \rightarrow ^{\circ}I_{9/2}\left(\mathrm{Nd^{\circ}}\right)$	299-466	I_{696}/I_{812}	0.79/299	1.4	183
Nd ³⁺ , Ti ⁴⁺		YAG	260	668 (Ti ⁴⁺), 870 (Nd ³⁺)	2 T ₂ \rightarrow O ²⁻ (Ti ⁴⁺), 4 F _{3/2} \rightarrow 4 I _{9/2} (Nd ³⁺)	123-773	I_{668}/I_{870}	3.70/473	0.13*	184
Nd^{3+} , Cr^{3+}	Nd ³		290	$690 (Cr^{3+}), 890 (Nd^{3+})$	$^{4}\text{T}_{1} \rightarrow ^{4}\text{A}_{2} \left(\text{Cr}^{3+} \right), ^{4}F_{3/2} \rightarrow ^{4}I_{9/2} \left(\text{Nd}^{3+} \right)$	77-850	I_{690}/I_{890}	3.49/200	0.14*	155
Nd°', Mn*		YAĞ	355	670 (Mn ⁻¹), 880 (Nd ²¹)	$^{2}E \rightarrow ^{4}A_{2} \; (Mn^{*+}), \; ^{4}F_{3/2} \rightarrow ^{4}I_{9/2} \; (Nd^{5/4})$	123-823	I_{670}/I_{880}	0.60/475	0.83*	185
Eu ³⁺ Fu ³⁺	Eu ³⁺ Fu ³⁺	Y_2O_3	590	690–715	${}^5\mathrm{D}_0 \to {}^7\mathrm{E}_4\left({}^7\mathrm{E}_1, {}^7\mathrm{E}_0\right)$ ${}^5\mathrm{D}_1 \to {}^7\mathrm{E}_1\left({}^7\mathrm{E}_1, {}^7\mathrm{E}_1\right)$	180-280	FIR ₆₉₀₋₇₁₄ /690-714	1.7/180	0.29*	186
3	3	200	1		20 4 (. 2; . 0)		****690-/14/690-/14		9	201
Er ³⁺ Fr ³⁺	Er ³⁺ Fr³+	SBN Naftf.@NaGdf.	532	800, 850 654, 806	$^{2}H_{11/2} \rightarrow ^{4}I_{13/2}, ^{4}S_{3/2} \rightarrow ^{4}I_{13/2}, ^{4}F_{6,6} \rightarrow ^{4}I_{1,6,6} \rightarrow ^{4}I$	300-700	FIR _{800/850}	1.39*/300	0.36*	161
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	S	Host	(nm)	$\lambda_{ m em} (m nm)$	Transitions	$\Delta T(\mathbf{K})$	4	$^{O_{\mathrm{rel}}/I}_{\mathrm{V}}$ (% K^{-1})/K δT (K)	$\delta T\left(\mathbf{K}\right)$	Ref.
± ± ±	Er^{3+} Er^{3+}	NaErF ₄ @Yb:NaYF ₄ @Nd:NaYbF ₄ Y_2O_3	800	652, 670 654, 660	$^{4}F_{9/2} \rightarrow ^{4}I_{15/2}$ $^{4}F_{9/2} \rightarrow ^{4}I_{15/2}$	305–425 200–1300	FIR _{652/670} FIR _{654/660}	0.22/305	2.2* 3.3*	188 189
3+, Mn ⁴⁺	${\rm Er}^{3+}$, ${\rm Mn}^{4+}$	YAP		$660 (Er^{3+}), 714 (Mn^{3+})$	$^4\mathrm{I}_{9/2} ightarrow ^4\mathrm{I}_{15/2} \left(\mathrm{Er}^{3+} \right), ^2\mathrm{E} ightarrow ^4\mathrm{A}_2 \left(\mathrm{Mn}^{4+} ight)$	300-550	I_{714}/I_{660}	1.95/530	0.25*	190
3+, Ho3+	Vb^{3+}	$NaLuF_4$		$817 (Er^{3+}), 887 (Ho^{3+})$	$^{4}I_{9/2} \rightarrow ^{4}I_{15/2}$ (Er ³⁺), $^{5}I_{6} \rightarrow ^{5}I_{8}$ (Ho ³⁺)	293-568	I_{887}/I_{817}	1.73/293	0.29*	191
3+	Yb^{3+}	YF_3		793,840	${}^{2}\mathrm{H}_{11/2} \rightarrow {}^{4}\mathrm{I}_{15/2}, {}^{4}\mathrm{S}_{3/2} \rightarrow {}^{4}\mathrm{I}_{15/2}$	293-473	FIR _{793/840}	0.98/293	0.51*	192
3+	Yb^{3+}	ZrO_2		660, 682	${}^4{ m F}_{9/2} o {}^4{ m I}_{15/2}$	293-573	$\mathrm{VPR}_{660/682}$	0.75/293	*99.0	193
3 ⁺ , Mn ⁴⁺	$4 \ln^{4+} \text{ Yb}^{3+}$	YAP	086	$660 (Ho^{3+}), 714 (Mn^{3+})$	$^{5}\text{F}_{5} \rightarrow ^{5}\text{I}_{8} \text{ (Ho}^{3+}), ^{2}\text{E} \rightarrow ^{4}\text{A}_{2} \text{ (Mn}^{4+})$	300-550	I_{714}/I_{660}	1.17/450	0.42*	190
)3+	Yb^{3+}	$\mathrm{KLu}(\mathrm{WO}_4)_2$		650, 660,	$^{5}\mathrm{F}_{5} ightarrow ^{5}\mathrm{I}_{8},$	297-673	$FIR_{650/660}$	0.38/297	1	194
)3+	Yb^{3+}	$\mathrm{Ba_2In_2O_5}$		653, 661,	$^{5}\mathrm{F}_{5} ightarrow ^{5}\mathrm{I}_{8}$	303-573	$FIR_{653/661}$	0.19*/303	2.5	195

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minimum value of δT will be given) will appear first. After this, single- or dual-doped materials and the excitation wavelengths associated with the luminescent thermometers will be presented, providing an overview of how these parameters influence the performance of that specific lanthanide active ion, in order to underline strategies for boosting thermometric performance.

2.1. Thulium-doped luminescent thermometers operating in the I-BW

Thulium (Tm³⁺) ion exhibits an electronic configuration 1s² 2s² 2p⁶ 3s² 3p⁶ 3d¹⁰ 4s² 4p⁶ 4d¹⁰ 5s² 5p⁶ 4f¹², and up to now, represents one of the most investigated active ions for luminescent nanothermometry in the I-BW due to, among others aspects, its NIR-to-NIR upconversion optical properties. Different Tm³⁺-doped luminescent nanothermometers have been developed including single Tm3+-doped materials, those using Tm³⁺ as an activator and Yb³⁺ as a sensitizer, and those that use Tm³⁺ together with another Ln³⁺ ion as activators in the presence or absence of Yb³⁺ as a sensitizer.

2.1.1. Single Tm³⁺-doped luminescent thermometers operating in the I-BW. Single Tm³⁺-doped luminescent nanothermometers operating in the I-BW without the presence of Yb³⁺ as a sensitizer are scarce. In fact, only two examples can be found in the literature, with orthorhombic yttrium orthoaluminate perovskite (YAlO₃, known as YAP)¹²⁷ and sodium niobate (NaNbO₃)¹²⁸ as hosts.

Tm³⁺:YAP nanothermometers (with a grain size of 35 nm) have been developed based on the temperature dependence of the emission lines located at 705 nm and 800 nm, arising from the ³F_{2,3} and ³H₄ TCLs decaying to the ³H₆ ground state, after excitation at 1210 nm. 127 These upconversion emissions can be generated either by sequential absorption of two excitation photons by a single Tm³⁺ ion through excited state absorption (ESA) or by the simultaneous absorption of two photons by two nearby Tm³⁺ ions via energy transfer upconversion (ETU) processes. 127,129 The process of upconversion refers to a nonlinear optical phenomenon known as anti-Stokes emission in which the sequential absorption of two or more low-energy photons leads to a high-energy luminescence emission. 129 From these two processes, the 3F2,3 excited state is populated. From this state, a radiative decay to the ground state will generate the emission line at 705 nm. A non-radiative relaxation from the ${}^{3}F_{2,3}$ level to the ${}^{3}H_{4}$ level, followed by a radiative decay to the ground state, generates the second emission line at 800 nm (Fig. 5(a)). 127,129

For the evaluation of the thermometric performance of the Tm3+:YAP nanothermometers, the FIR between the emission lines at 705 nm and 800 nm was used in the temperature range from 324 K-424 K. The FIR is defined using the emission intensities of the of $|2\rangle \rightarrow |0\rangle$ ($I_{02}\equiv I_2$) and $|1\rangle \rightarrow |0\rangle$ $(I_{01} \equiv I_1)$ transitions, where $|0\rangle$, $|1\rangle$, $|2\rangle$ denote the ground state and TCLs 1 and 2,42,196 respectively, as:

$$FIR = \frac{I_2}{I_1} = \frac{A_{02}h\nu_{02}N_2}{A_{01}h\nu_{01}N_1}$$
 (7)

Fable 1 (Contd.)

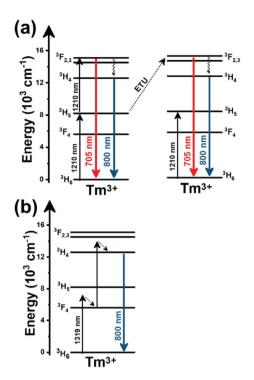


Fig. 5 Mechanisms of generation of the emission lines for single active ${\rm Tm}^{3+}$ -doped luminescent thermometers: (a) YAP after excitation at 1210 nm and (b) ${\rm NaNbO_3}$ after excitation at 1319 nm. Please note that despite the fact that other emission lines can be produced, for sake of simplicity, only the emission lines involved in the luminescence thermometer development are considered.

where N_2 and N_1 are the populations of the $|2\rangle$ and $|1\rangle$ levels, ν_{02} and ν_{01} are the frequencies of the $|2\rangle \rightarrow |0\rangle$ and $|1\rangle \rightarrow |0\rangle$ transitions, and A_{02} and A_{01} are the total spontaneous emission rates from levels $|2\rangle$ and $|1\rangle$ to level $|0\rangle$. Since the energy levels 1 and 2 are in thermal equilibrium, the electronic populations can be correlated by the following equation:

$$N_2 = \frac{g_2}{g_1} N_1 \exp\left(-\frac{\Delta E}{k_{\rm B}T}\right) \tag{8}$$

where g_2 and g_1 are the degeneracies of the two levels, and ΔE is the energy gap between the barycenters of the $|2\rangle \rightarrow |0\rangle$ and $|1\rangle \rightarrow |0\rangle$ emission bands. ^{21,42,196}

Thus, the expression for FIR can be rewritten as:

$$FIR = \frac{I_2}{I_1} = \frac{g_2 A_{02} h \nu_{02}}{g_1 A_{01} h \nu_{01}} \exp\left(-\frac{\Delta E}{k_{\rm B} T}\right) = B \exp\left(-\frac{\Delta E}{k_{\rm B} T}\right)$$
(9)

where $B = \frac{g_2 A_{02} h \nu_{02}}{g_1 A_{01} h \nu_{01}}$. By applying the expression for FIR to eqn.

(2) and (3), the corresponding expressions for $S_{\rm rel}$ and δT can be obtained as follows:^{21,42,196}

$$S_{\rm rel} = \left| \frac{\Delta E}{k_{\rm B} T^2} \right| \times 100\% \tag{10}$$

$$\delta T = \frac{k_{\rm B} T^2}{\Delta E} \frac{\delta {\rm FIR}}{{\rm FIR}} \tag{11}$$

Thus, considering the $|2\rangle \rightarrow |0\rangle$ as the electronic transition generating the emission line at 705 nm and the $|1\rangle \rightarrow |0\rangle$ as the electronic transition generating the emission line at 800 nm, the maximum $S_{\rm rel}$ of the Tm³⁺:YAP nanothermometer is 2.64% K⁻¹ at 324 K, whereas the δT was not reported by the authors. However, knowing the reported value of $\Delta E = 1926$ cm⁻¹, 127 and applying an experimental setup error, which is the most commonly used in the literature, 12 δT can be calculated according to eqn (11), with a minimum value of 0.20 K at 324 K that increases to 0.32 K at 424 K.

 ${\rm Tm^{3^+}:NaNbO_3}$ luminescent thermometer, from its side, was based on the change of the FIR of the emission peaks arising from different Stark sub-levels of the ${}^3{\rm H_4} \rightarrow {}^3{\rm H_6}$ transition with a broad emission band located at ${\sim}800$ nm, using in particular the emission lines located at 797 nm and 807 nm, after excitation at 1319 nm, in the temperature range 303–453 K. The generation of the 800 nm emission line is assigned to a multiphonon-assisted excitation from the ground state ${}^3{\rm H_6}$ to the ${}^3{\rm H_5}$ excited level, a rapid relaxation to the ${}^3{\rm F_4}$ level, followed by an excited-state absorption to the ${}^3{\rm H_4}$ level (Fig. 5(b)).

Instead of using the expression indicated above for FIR, to calculate the thermometric parameter in this case the authors propose to use the following equation:¹²⁸

$$R = A + B \times T \tag{12}$$

where R is the thermometric parameter used in this case (I_{707}/I_{807}), and A and B are parameters determined by the fitting of the experimental data. The authors preferred to use a linear function for this parameter, according to the trend followed by the experimental points obtained in the calibration of the thermometer, without taking into account that the different Stark sub-levels of a particular transition are, of course, under thermal equilibrium, and that the Boltzmann distribution proposed by the FIR model would be more convenient in this case. As a consequence, $S_{\rm rel}$, with a maximum value of 0.80% K^{-1} at 303 K, had to be calculated using again a different equation: $I_{\rm rel}$

$$S_{\text{rel}} = \frac{B}{A + B \times T} \times 100\% \tag{13}$$

This value is higher than the one that should be obtained when applying strictly the FIR model according to eqn (9) $(0.25\% \text{ K}^{-1})$, and according to that, the δT calculated according to eqn (11) (0.62 K) if the thermal sensitivity given by the authors is used, and 2 K if the thermal sensitivity according to the FIR model is used) is clearly overestimated.

Curiously, the two examples included here are based on upconversion mechanisms with excitation in the II-BW, despite the fact that the exploited emissions are in the I-BW, ensuring that both excitation and emission have a good penetration in the biological tissues. Upconversion nanomaterials offer the possibility of a relatively high thermal sensing in combination with high penetration depths and optical tunability. 130

In fact, Pereira et al. 128 could demonstrate that the penetration of light by exciting at 1319 nm was twice the one that could be obtained when exciting at 800 nm. Nevertheless, the excitation wavelength they used (1319 nm) induced a local heating in the environment where the Tm3+:NaNbO3 nanoparticles were used, of up to 8 K for a pump power of 350 mW, due to the fact that this wavelength is not resonant with any of the electronic levels of Tm³⁺. So part of the energy of the excitation source was lost in the form of heat, which the nanoparticles dispersed in the surrounding medium. 128

The only conclusion that can be extracted from these two examples is that to maximize S_{rel} , and consequently minimize the thermal resolution, the best strategy would be to maximize the separation between the emission lines used in the thermometer, but maintaining it below to 2000 cm⁻¹ to maintain them inside the defined range of TCLs.42

2.1.2. Tm³⁺,Yb³⁺-codoped luminescent thermometers operating in the I-BW. Lanthanide nanothermometers codoped with Tm³⁺ acting as an activator and Yb³⁺ acting as a sensitizer is, by far, the most explored model for luminescent thermometers operating in the I-BW. Probably this great interest is due to their NIR-to-NIR upconversion, more efficient than that obtained in single Tm3+-doped thermometers.148 This is a consequence of the bigger absorption cross-section of Yb3+ at the pumping wavelength of 980 nm, 196 the lower costs of NIR excitation sources emitting this wavelength, 197 and the generation of emission lines lying in the I-BW. However, it has to be mentioned here the strong heating effect that this pumping wavelength has in biological tissues, due to the efficient absorption of this wavelength by water. 122 This is why several authors considered that this approach should be avoided when performing luminescence thermometry for biomedical applications due to its non-discriminating effect with respect to the heating of biological tissues.

Another characteristic of the Tm³⁺,Yb³⁺-codoped luminescent thermometers operating in the I-BW is that most of them use the same emission lines to build the luminescent thermometer: the emission at around 700 nm, arising from the ${}^{3}F_{2,3} \rightarrow {}^{3}H_{6}$ transition, and the emission produced at around 800 nm, arising from the ${}^{3}\text{H}_{4} \rightarrow {}^{3}\text{H}_{6}$ transition, in both cases of Tm³⁺, as was shown in the first example of the previous subsection. However, in this case, due to the presence of Yb³⁺, the excitation and emission mechanism is different. Fig. 6(a) shows the excitation mechanism used to generate these emission lines after excitation at 980 nm. The 980 nm radiation is absorbed by Yb3+, promoting its electrons from the 2F7/2 ground state to the ²F_{5/2} excited state. From this excited state of Yb3+, via energy transfer (hereafter ET) processes, the excited states of Tm3+ are populated. Hence, three ET processes populate the ¹G₄, ²F_{2,3} and ³H₅ levels of Tm³⁺. From the ¹G₄ level, electrons relax non-radiatively to the ³F_{2,3} level, from which they fall back to the 3H6 ground state level, generating the 700 nm emission. From the ³F_{2,3} level, a non-radiative relaxation process to the ³H₄ level, followed by a radiative decay to the ground state, gives rise to the 800 nm emission. 131 It should be emphasized here that other mechanisms and emission lines might be involved in this process, but we are

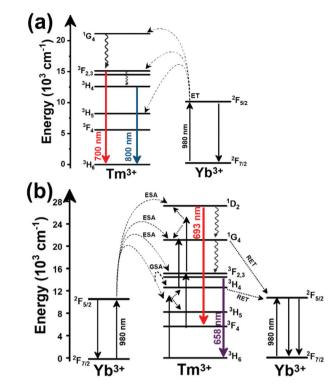


Fig. 6 Mechanisms of generation of the emission lines for: (a) Tm³⁺. Yb3+:GdVO4@SiO2 core@shell nanoparticles (as an illustrative example) after excitation at 980 nm, and (b) Tm³⁺,Yb³⁺-doped phosphate glass material after excitation at 980 nm.

presenting only the emission lines used to determine the thermometric performance.

In all these cases, since the ${}^{3}F_{2,3}$ and ${}^{3}H_{4}$ energy levels can be considered to be thermally coupled, the FIR model was used as the thermometric parameter. However, despite the big majority of the papers analyzed defining this FIR as the ratio between the intensity of the emission at 800 nm divided by the intensity of the emission at 700 nm, following the recommendation of Wade et al. in their seminal publication, 42 we encountered two recent publications (ref. 136 and 137) in which they defined it in the inverse way. Nevertheless, since eqn (10) does not take into account the way in which FIR has been defined, they could be compared with the rest of the luminescent thermometers analyzed in this section. If we analyze the maximum S_{rel} obtained for these kinds of luminescent thermometer, it is encountered in the range between 2–3% $\mbox{K}^{-1},$ as a result of having a ΔE between the Tm3+ 3F2,3 and the 3H4 energy levels of around 1785 cm⁻¹. This energy difference can change slightly from host to host, due to their different crystal fields, and this would be reflected in the different S_{rel} values reported by the authors and that are listed in Table 1. This would provide a δT of ~ 0.2 K, approaching the desired limit of 0.1 K for biomedical purposes.13 The temperatures at which these values are obtained are in all cases around room temperature, and all of them became worse as the temperature increased (Fig. 7).

In some cases the FIR model was modified, compared with the equation proposed by Wade et al., 42 to account for the sig-

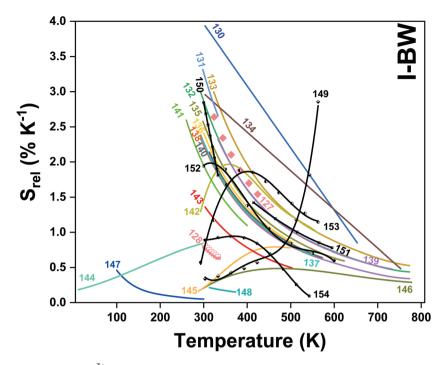


Fig. 7 Temperature dependence of S_{rel} of Tm^{3+} -doped luminescent thermometers, operating in the I-BW. In symbols only, in solid lines and in solid lines with symbols, single Tm^{3+} , $\text{Tm}^{3+}/\text{Yb}^{3+}$ and $\text{Tm}^{3+}/\text{Ln}^{3+}$ based thermometers, respectively. The numbers represent the corresponding references for each Tm^{3+} -based thermometer.

nificant overlapping between the two emission bands used in the luminescent thermometer, adding an additional summand to the equation as:

$$FIR = B \exp\left(-\frac{\Delta E}{k_{\rm B}T}\right) + D \tag{14}$$

where D is a fitting parameter. Then if S_{rel} is calculated according to eqn (10), the value is not correct. In fact, it is underestimated, since according to eqn (2), the right way to determine S_{rel} should be:

$$S_{\rm rel} = \frac{{\rm FIR} - D}{{\rm FIR}} \left| \frac{\Delta E}{k_{\rm B} T^2} \right| \times 100\% \tag{15}$$

It is worth mentioning here that in some cases the inner active cores have been coated with an inactive shell. This strategy has been used in the case of $GdVO_4@SiO_2$, 131 $NaYF_4@NaYF_4@SiO_2$, 135 $Bi_2SiO_5@SiO_2$, 141 and $NaYF_4@CaF_2$ nanoparticles, 148 using an inert layer constituted by amorphous silica (SiO_2), or cubic CaF_2 . The aim of coating the active core with this inert layer was to prevent the thermal degradation of the active core from potential oxidation, and at the same time prevent non-radiative relaxation processes of the active ions encountered on the surface of the nanoparticles

by their interactions with the surface-bound ligands in the case of colloidal dispersions. 135 Also, in the case of NaYF₄@NaYF₄@SiO₂, an inert NaYF₄ interlayer was used for further protection of the radiative properties of the active ions. 135 Thus, despite the fact that it seems evident that these core@shell structures help to increase the quantum yield of the luminescent nanoparticles, no conclusion can be extracted about their thermometric performance, since none of the papers published up to now for Tm³⁺,Yb³⁺-codoped systems operating in the I-BW reported a comparison between the same material coated and not coated with the inert layer. Also, by analyzing the data provided in Table 1 for the phosphates, it seems that when the phosphate material is based on a metal of the lanthanide series (Y or La), the S_{rel} achieved is higher than that obtained for a material based on a post-transition metal. In fact, S_{rel} values obtained for Tm^{3+} , Yb^{3+} : LaPO₄ (S_{rel} = 3% K^{-1}) and in Tm^{3+} , Yb^{3+} : YPO_4 ($S_{rel} = 2.33\%$ K^{-1}) nanoparticles132 are slightly higher than the ones obtained for $Tm^{3+}, Yb^{3+}:BiPO_4$ ($S_{rel} = 2.14\% K^{-1}$). However, other factors, such as the structure, can also affect the thermometric performance of these luminescent thermometers, since LaPO₄ crystallizes in the monoclinic system, YPO4 does so in the tetragonal system, and BiPO₄ is a hexagonal compound.

From another side, Xing *et al.*¹³³ studied the effect of using a single crystal or a polycrystal on the temperature-sensing performance of Tm^{3+} ,Yb³⁺:LiNbO₃ (labelled as S-LiNbO₃ and P-LiNbO₃, respectively, in Table 1). A higher $S_{\rm rel}$ was obtained for the single crystal (~3% K⁻¹ νs . ~2% K⁻¹ in the polycrystal). The authors assigned these differences in $S_{\rm rel}$ to the effect of

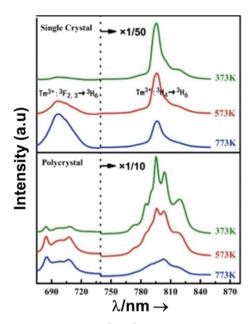


Fig. 8 Emission spectra of Tm³⁺/Yb³⁺-codoped LiNbO₃ polycrystal and single crystal at different temperatures under 980 nm excitation (1/10 and 1/50 represent the factor of division applied above the marked wavelength to visualize clear the different emission bands). Reprinted with permission from ref. 133. Copyright 2015, Elsevier.

the crystallinity of the material based on the degree of splitting

of the upconversion emission spectra. In fact, in Fig. 8 it can be seen that the spectrum recorded for the single crystal is narrower than the spectrum recorded for the polycrystalline material, and that the spectrum of the polycrystalline material has a richer peak structure. However, the intensity of the spectrum for the single crystal is much higher than that of the polycrystalline material (Fig. 8), which could generate also a saturation in the intensity of the spectrum that might affect also the thermometric performance of the luminescent thermometer. In fact, δT and, as a consequence, $S_{\rm rel}$ are greatly affected by the signal-to-noise $\left(\frac{dI}{I}\right)$ ratio of the emission spectra from which temperature is estimated.^{21,199} At lower intensity signals, the probability of having a high signal-tonoise ratio increases, hampering obtaining a good thermo-

A small number of examples applied the FIR method to the thermal evolution of the intensities of different emission lines arising from Stark sublevels of the ³H₄ \rightarrow ³H₆ transition of Tm³⁺, lying at around 800 nm, to develop thermometers based on Tm $^{3+}$ and Yb $^{3+}$ -codoped materials. 147,148 Since the $^{3}\text{H}_{4} \rightarrow$ ³H₆ transition generates the emission band with the maximum intensity in this region with Tm3+, it is logical to use it to develop luminescent thermometers since the optical sensitivity of the detection system is less strict, allowing the use of cheaper, and even portable detectors. However, as mentioned in the previous section, the fact that ΔE in this cases is small ($>300 \text{ cm}^{-1}$) makes the S_{rel} values obtained smaller (in the range 0.20-0.40% K⁻¹) than those obtained using other strategies, which gives δT values ~1 K. Hence, these luminescent thermometers are not competitive enough in front of the other strategies proposed.

A different example, based on the analysis of the intensity of two different emission lines, is that provided by an amorphous Yb3+ and Tm3+-codoped phosphate glass,130 and by Tm³⁺,Yb³⁺:Sr₂GdF₇ glass ceramics. 142 In both cases, the luminescent thermometer developed was based on the intensity ratio between two emissions arising from two different energy levels of Tm3+ (1G4 and 3F2,3) that are not thermally coupled, although the population of the ³F_{2,3} level is potentiated as the temperature increases by the non-radiative relaxations generated from the upper excited levels, including the 1G4 level (Fig. 6(b)). In these cases, the authors used two emission bands located at \sim 700 nm (generated by the $^{1}G_{4} \rightarrow {^{3}H_{5}}$ transition) and \sim 650 nm (generated by the $^3F_{2,3} \rightarrow ^3H_6$ transition) after excitation at 980 nm, although these two lines exhibited a very low intensity when compared with the 800 nm emission line used in the previous examples. The mechanism of generation of these emission lines is presented in Fig. 6(b). Under 980 nm excitation, Yb³⁺ is excited from the ²F_{7/2} ground state to the ²F_{5/2} excited state. The energy of this transition can be transferred to Tm3+, which excites its electron from the 3H6 ground state to the 3H5 excited state, which then relaxes rapidly and non-radiatively to the ³F₄ level. When the energy of a second photon of 980 nm is transferred from Yb3+ to the electrons that are in this excited state, they can promote to the ${}^{3}F_{2,3}$ level. From here, the radiative ${}^{3}F_{2,3} \rightarrow {}^{3}H_{6}$ transition can take place, generating the emission at 650 nm. The second emission is generated when part of the energy of the electrons lying in the ³F_{2,3} is dissipated non-radiatively, populating the lower ³H₄ energy level. From here, the transfer of the energy of a third photon of 980 nm from Yb3+ boosts these electrons to the ¹D₂ level that can relax radiatively to the ³F₄ level, generating the emission at 700 nm. Despite the fact that the two electronic levels from which these emissions arise are not thermally coupled, the authors fitted their intensity ratio to the FIR model, and from it they extracted a ΔE value (2493 cm⁻¹ and 2207 cm⁻¹ for the phosphate¹³⁰ and ceramic glass¹⁴² respectively) that allowed the calculation of S_{rel} according to eqn (13). Since the energy separation between the electronic levels involved in these emissions is of the order of 6000 cm⁻¹, higher than the upper limit defined for TCLs, the Boltzmann model cannot be used, and instead a phenomenological model fitting the experimental data to a conventional exponential curve should be taken into account. One might think that, according to that, the $S_{\rm rel}$ values they provided (3.94% K⁻¹ and 1.97% K^{-1} at 303 $K_1^{130,142}$ respectively) should not be considered, since they are based on a wrong model. But, surprisingly, by taking the FIR equation they provided as a phenomenological equation, of the form:

$$\Delta = \frac{I_2}{I_1} = C \exp\left(-\frac{D}{T}\right) \tag{16}$$

where C and D are fitting parameters, and calculating S_{rel} using eqn (2), the result is exactly the same, constituting the

metric performance of the material.

strategy that provides the highest S_{rel} among the other ones proposed for the Tm³⁺,Yb³⁺-codoped systems. However, the low intensity of the emissions used in these luminescent thermometers will hamper their practical use in real biomedical applications.

To end up this subsection, it is worth mentioning a totally different example, based on bandwidth thermometry instead of band-shape thermometry, as are all the luminescent thermometers reported up to now. Lu et al. synthesized rhombohedral Tm³⁺,Yb³⁺:YOF irregular nanoparticles and analyzed their temperature-sensing performance applying the bandwidth luminescence nanothermometry model to the 800 nm emission band of Tm3+, that as we said before is by far the band with the highest emission generated by Tm3+ in this region, after exciting the material at 980 nm in the temperature range from 12 to 300 K. 144 The temperature variation of the full width at half maximum (FWHMs) of the emission band was fitted to a simplified exponential equation:²⁰⁰

$$\Delta \nu = \Delta \nu_0 + A e^{R_0 T} \tag{17}$$

where $\Delta\nu_0$ represents the initial FWHM, and A and R_0 are empirical constants to be determined from the fitting model. The corresponding S_{rel} was deduced from the following equation:

$$S_{\rm rel} = \frac{1}{\Delta \nu} \left| \frac{\partial \Delta \nu}{\partial T} \right| \times 100\%$$
 (18)

The maximum S_{rel} obtained was 0.84% K^{-1} at the highest temperature analyzed. 144 Despite the fact that this example, strictly speaking, could not be used for biomedical applications since the temperature range analyzed was located at lower temperatures than those used in biological media, the fact that the maximum $S_{\rm rel}$ value was obtained at 300 K might suggest that it would be also useful in this temperature range. This is why we decided to include this example here.

Thus, the conclusion that can be extracted from the examples included in this subsection is that the most effective strategy to get a high S_{rel} is using the emissions arising from two non-thermally coupled electronic levels that are apart by a distance higher than the upper limit defining the thermally coupled range of energies (2000 cm⁻¹). Other strategies are not as effective as this one to increase $S_{\rm rel}$. However, the quantum yield of the materials and that of the emissions used has not been taken into account in any of the luminescent thermometers operating in the I-BW reported as lying in the category described here. In fact, the low intensity of the emissions considered at ~650 nm and ~700 nm, or the big difference of intensity existing between the two emissions located at ~700 nm and at ~800 nm, would imply using highly sensitive (and thus bulky and costly) detectors, or that recording one of the signals in the optimum conditions would affect the collection of the other one in terms of signal-to-noise ratio or saturation of the detector. Thus, other strategies that, a priori, have a lower $S_{\rm rel}$, like those involving different Stark sublevels of the

 ${}^{3}\text{H}_{4} \rightarrow {}^{3}\text{H}_{6}$ transition of Tm³⁺, lying at around 800 nm, might be more effective, due to the high intensity of this emission, which is the consequence of the ³H₄ level of Tm³⁺ being electronically fed by different radiative and non-radiative mechanisms. Nevertheless, this would lead to δT of around 1 K, one order of magnitude higher than that desired for biomedical applications.

2.1.3. Other Tm³⁺,Ln³⁺-codoped luminescent thermometers operating in the I-BW. Dual emission center luminescent nanothermometers, i.e. those luminescent thermometers in which there are two activators emitting at different wavelengths, 21 operating in the I-BW, have been based on the emissions arising from Tm3+ and another Ln3+ ion acting as activators, with or without Yb³⁺ acting as a sensitizer. Two groups of luminescent thermometers could be identified in this category: (i) those using Tm³⁺ and Ho³⁺ as activators, with the presence or absence of Yb³⁺ as a sensitizer; ¹⁴⁹⁻¹⁵¹ and (ii) those using Tm³⁺ and Er³⁺ as activators, using always Yb³⁺ as a sensitizer. 152-154

In the case of Ho³⁺, an important temperature-dependence of the emissions arising from the electronically linked energy levels of Ho³⁺ and Tm³⁺ has been reported, from which ET and back ET (hereafter BET) processes are possible. Despite the fact that the electronic levels of these different ions are not thermally coupled, a relationship between their intensities as the temperature changes can be established. Two different strategies have been proposed in this case. Savchuk et al. doped monoclinic potassium lutetium double tungstate KLu (WO₄)₂ nanoparticles with Ho³⁺ and Tm³⁺ and investigated their thermal performance in the physiological range of temperatures (303 K-333 K), after exciting the material at 808 nm. 150 In this case, Tm3+ acted both as a sensitizer, by absorbing the energy of the 808 nm laser source and transferring part of the energy to Ho³⁺, and as an activator, generating different emission lines, among which one located at ~700 nm attributed to the ${}^{3}F_{2,3} \rightarrow {}^{3}H_{6}$ transition. The laser radiation at 808 nm excites the Tm^{3+} electrons to the ${}^{3}\mathrm{H}_{4}$ level, from where they non-radiatively relax to the ³F₄ level. The absorption of a second photon at 808 nm promotes the electrons from the ³F₄ level to the ¹G₄ level, from which a non-radiative decay can take place to the ³F_{2,3} level, from which a radiative transition to the ground state can occur, generating the emission at 696 nm. The energy levels of Ho³⁺ can be populated through two different ET mechanisms: one from the Tm^{3+ 3}H₄ level to the Ho^{3+ 5}I₅ level, and a second one from the ³F₄ level of Tm³⁺ to the ⁵I₇ level of Ho³⁺, from which also a BET process to Tm³⁺ can take place. Besides the direct ET from Tm³⁺, the ⁵I₇ level of Ho³⁺ can also be populated by the non-radiative relaxation from its ${}^5\mathrm{I}_5$ energy level. The ET of another photon of 808 nm from Tm³⁺ promotes the electrons lying in the ⁵I₇ level of Ho³⁺ to the higher energy levels ³K₈ and ⁵F₃. From these levels, nonradiative decay processes populate the 5S2 and 5F4 levels of Ho³⁺ from which a radiative transition towards the ⁵I₇ level can take place, generating an emission at 755 nm. A non-radiative relaxation from the 5S2 and 5F4 levels to the 5F5 level, followed by a radiative transition to the ⁵I₈ ground state, generated an

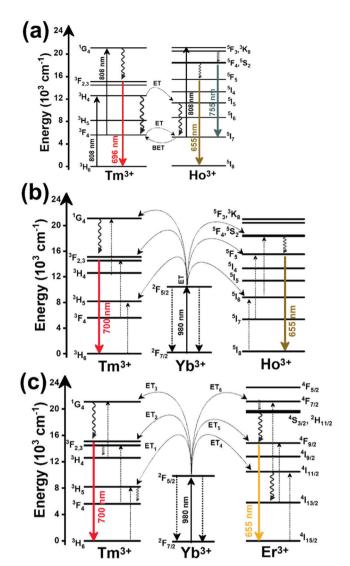


Fig. 9 Mechanisms of generation of the emission lines for: (a) Tm^{3+} , Ho^{3+} -doped $KLu(WO_4)_2$ nanocrystals after excitation at 808 nm, (b) Tm^{3+} , Ho^{3+} , Yb^{3+} -doped YPO_4 or $Gd_2(WO_4)_3$ nanoparticles after excitation at 980 nm, and (c) Tm^{3+} , Er^{3+} , Yb^{3+} : YF_3 nanoparticles embedded in a glass ceramic after excitation at 980 nm.

emission at 655 nm.¹⁵⁰ The detailed mechanism for this complex system is depicted in Fig. 9(a).

The temperature-sensing capacity was estimated by the integrated intensity ratio of the $^3F_{2,3} \rightarrow ^3H_6$ and the $^5S_2,^5F_4 \rightarrow ^5I_7$ transitions of Tm³+ and Ho³+, respectively, corresponding to the emission bands located at 696 nm and 755 nm, arising from non-thermally coupled electronic levels (NTCLs). Hence, the thermometric parameter (Δ) in this case was fitted to an empirical exponential growing equation of the type: 150

$$\Delta = \Delta_0 + B \exp(\alpha T) \tag{19}$$

where Δ_0 and B are constants to be determined from the fitting of the experimental data.

Thus, the corresponding S_{rel} was calculated by:

$$S_{\rm rel} = \frac{B\alpha \exp(\alpha T)}{\Delta_0 + B \exp(\alpha T)} \times 100 \tag{20}$$

The maximum $S_{\rm rel} = 2.84\%~{\rm K}^{-1}$ was achieved at 303 K. Savchuk *et al.*, also estimated the δT of this luminescent thermometer using the equation:¹⁵⁰

$$T = \frac{\delta \Delta}{B \exp(\alpha T)} \tag{21}$$

determining a value of 0.2 K at 303 K.

The second strategy developed in Tm³⁺,Ho³⁺-codoped systems in the presence of Yb3+ as a sensitizer is that demonstrated in Tm³⁺,Ho³⁺,Yb³⁺:YPO₄ microplates, ¹⁴⁹ and in Tm³⁺ Ho³⁺,Yb³⁺:Gd₂(WO₄)₃ nanocrystals. In these two cases, Yb³⁺ acted as a sensitizer to absorb the energy of the 980 nm excitation source and transferred it to Tm³⁺ and Ho³⁺, to generate an emission line at \sim 650-660 nm, corresponding to the Ho³⁺ $^5F_5 \rightarrow \, ^5I_8$ transition, and another one at ${\sim}700$ nm, corresponding to the $Tm^{3+} {}^3F_{2,3} \rightarrow {}^3H_6$ transition. Yb^{3+} absorbs the 980 nm excitation, exciting its electrons from the ²F_{7/2} ground state to the ${}^{2}F_{5/2}$ excited state (Fig. 9(b)). Subsequently, the energy of this sensitizer is transferred to the adjacent activators Ho3+ and Tm3+, and the sensitizer is relaxed back to the ground state. Ho³⁺ and Tm³⁺ activators promote their electrons to their excited states (Tm3+ to 3H5, 3F2,3 and 1G4 levels, and $\mathrm{Ho^{3+}}$ to ${}^{5}\mathrm{I}_{6}$, ${}^{5}\mathrm{F}_{5}$ and ${}^{5}\mathrm{S}_{2}$, ${}^{5}\mathrm{F}_{4}$ levels). In addition, the excited states involved in the generation of the red emissions (³F_{2,3} of Tm3+ and 5F5 of Ho3+) can be populated by non-radiative processes from the higher energy excited states of the corresponding ions (¹G₄ of Tm³⁺ and ⁵S₂, ⁵F₄ of Ho³⁺).^{149,151} From these states, decaying back radiatively to the corresponding ground states (${}^{3}F_{2,3} \rightarrow {}^{3}H_{6}$ of Tm $^{3+}$ and ${}^{5}F_{5} \rightarrow {}^{5}I_{8}$ of Ho $^{3+}$) generated the emissions used for temperature sensing.

Again, these emission bands do not arise from TCLs; hence the thermometric parameter could be deduced from the intensity ratio between these bands by fitting the experimental data to a linear equation, ¹⁴⁹ or to a polynomial function: ¹⁵¹

$$\Delta = X - S \times T \tag{22}$$

$$\Delta = C_1 + B_1 T + B_2 T^2 + B_3 T^3 \tag{23}$$

where X and S, and C_1 , B_1 , B_2 and B_3 were fitting parameters. From these equations the thermal sensitivities could be calculated, depending on the thermometric equation used, as:

$$S_{\text{rel}} = \left| \frac{S}{S - ST} \right| \times 100\% \tag{24}$$

$$S_{\text{rel}} = (B_1 + 2B_2T + 3B_3T^2) \times 100\% \tag{25}$$

The maximum $S_{\rm rel}=2.86\%~{\rm K}^{-1}$ at 563 K was obtained for Tm³⁺,Ho³⁺,Yb³⁺:YPO₄,¹⁴⁹ which allowed the calculation of a δT of 0.2 K. In the case of Tm³⁺,Ho³⁺,Yb³⁺:Gd₂(WO₄)₃, the authors reported a substantially lower $S_{\rm rel}$ of 1.43% K⁻¹ at 595 K, due to the different phenomenological model used to fit the experimental data. ¹⁴⁹

When the two emission centers consisted of the combination of Tm3+ and Er3+, in the presence of Yb3+ as a sensitizer, the same strategy was always used: the temperature reading was achieved by the intensity ratio between the 700 nm emission line of Tm³⁺ attributed to the ${}^{3}F_{2,3} \rightarrow {}^{3}H_{6}$ transition, and the 655-660 nm emission line of Er³⁺, arising from the ${}^4F_{9/2} \rightarrow {}^4I_{15/2}$ transition. ${}^{152-154}$

The mechanism of the generation of these emission lines is depicted in Fig. 9(c).

In short, Yb3+ acts as a sensitizer to absorb the energy of the 980 nm excitation source, and generates three ET processes towards Tm3+, populating its 3H5, 3F2,3, and 1G4 states after some non-radiative relaxation processes in some cases (Fig. 9(c)):

- $Yb^{3+}:^2F_{5/2} + Tm^{3+}:^3H_6 \rightarrow Yb^{3+}:^2F_{7/2} + Tm^{3+}:^3H_5$ (ET1)
- $Yb^{3+}:^{2}F_{5/2} + Tm^{3+}:^{3}H_{4} \rightarrow Yb^{3+}:^{2}F_{7/2} + Tm^{3+}:^{3}F_{2,3}$ (ET2)
- $Yb^{3+}:^{2}F_{5/2} + Tm^{3+}:^{3}H_{4} \rightarrow Yb^{3+}:^{2}F_{7/2} + Tm^{3+}:^{1}G_{4}$ (ET3)

On the other hand, a two-step ET processes from Yb3+ to Er3+ can also occur:

- $Yb^{3+}:^2F_{5/2} + Er^{3+}:^4I_{15/2} \rightarrow Yb^{3+}:^2F_{7/2} + Er^{3+}:^4I_{11/2}$ (ET4)
- Yb^{3+} : ${}^{2}F_{5/2} + Er^{3+}$: ${}^{4}I_{11/2} \rightarrow Yb^{3+}$: ${}^{2}F_{7/2} + Er^{3+}$: ${}^{4}F_{7/2}$ (ET5) followed by a non-radiative relaxation to populate the Er^{3+ 2}H_{11/2}, ⁴S_{3/2} and ⁴F_{9/2} electronic levels.

In addition, the population of the Er^{3+ 4}F_{9/2} state can be achieved through another ET process:

 $\bullet \ Yb^{3+}; ^2F_{5/2} + Er^{3+}; ^4I_{13/2} \to Yb^{3+}; ^2F_{7/2} + Er^{3+}; ^4F_{9/2} \left(ET6\right)$

Finally, a last ET process between Er³⁺ and Tm³⁺ can occur, populating the ³F_{2,3} level of Tm³⁺ (not presented in Fig. 9(c) for clarity reasons):

• $\mathrm{Er}^{3+}: {}^{4}\mathrm{F}_{9/2} + \mathrm{Tm}^{3+}: {}^{3}\mathrm{H}_{6} \to \mathrm{Er}^{3+}: {}^{4}\mathrm{I}_{15/2} + \mathrm{Tm}^{3+}: {}^{3}\mathrm{F}_{2,3}$ (ET7)

Once the ${}^3F_{2,3}$ level of Tm^{3+} and the ${}^4F_{9/2}$ level of Er^{3+} are populated, the emissions at 700 nm and at 666-660 nm are generated arising from the electronic transitions indicated

This strategy was used in Tm³⁺,Er³⁺,Yb³⁺:YF₃ nanoparticles embedded in a glass ceramic, 153 and in Tm3+,Er3+,Yb3+: LuF₃, ¹⁵⁴ and in oleic acid-capped Tm³⁺, Er³⁺, Yb³⁺: NaLuF₄ nanocrystals. 152 It is important to note here that due to the non-resonant energy matching between the ³F_{2,3} level of Tm³⁺ and the ${}^{4}F_{9/2}$ level of Er³⁺, the multiphonon-assisted ET rate (K_{ET}) can be described by the Mott-Seitz model:^{201,202}

$$K_{\rm ET} \propto \exp\left(-\frac{\Delta E}{k_{\rm B}T}\right)$$
 (26)

indicating the temperature-dependent behavior and the link between the two emissions involved in this thermometer. ΔE here stands for the energy gap between the excited and deexcited states.

In the case of Tm³⁺,Er³⁺,Yb³⁺:YF₃ nanoparticles embedded in a glass ceramic, 153 the one exhibiting the highest S_{rel} (1.89% K⁻¹ at 393 K) among these examples of luminescent thermometers, the authors fitted the experimental data to a FIR model of the form of eqn (14). However, as discussed previously, since the two energy levels from which the emissions considered are not thermally coupled since they do not belong to the same Ln3+, this model should not be used, and instead a phenomenological one should be implemented, exponential in that case. In the two other cases, phenomenological models were implemented, following polynomial equations in both cases, of second order in the case of oleic acid-capped Tm³⁺, Er³⁺,Yb³⁺:NaLuF₄ nanocrystals with the form: ¹⁵²

$$\Delta = C_1 - B_1 T + B_2 T^2 \tag{27}$$

and of third order in the case of Tm3+,Er3+,Yb3+:LuF3 mesocrystals with the form:154

$$\Delta = C_1 - B_1 T + B_2 T^2 - B_3 T^3 \tag{28}$$

The $S_{\rm rel}$ of these systems were obtained using eqn (2), with values of 0.76% K⁻¹ at 500 K, 152 and 0.95% K⁻¹ at 363 K, 154 respectively. The differences between Srel obtained can be attributed to the different phenomenological models implemented in each case.

In general, the strategies to develop luminescent nanothermometers analyzed in this section do not provide higher S_{rel} than those highlighted in the previous sections for Tm³⁺-doped systems, although in some cases they are pretty close. However, they use emission bands with a more similar intensity than those used previously in Tm3+,Yb3+codoped systems (Fig. 10), which would facilitate recording the two emission bands with enough guarantees with the available detection systems, facilitating their practical use in biomedical applications with a δT of ~ 0.2 K, closer to the ones demanded by the scientific community.

Table 1 summarizes all the information regarding Tm3+doped luminescent nanothermometers, and Fig. 7 depicts the evolution of the S_{rel} of every luminescent thermometer based on Tm3+ analyzed in this section operating in the I-BW in the range of temperatures studied. From this figure, it can be seen that in the majority of the cases analyzed, S_{rel} reaches its

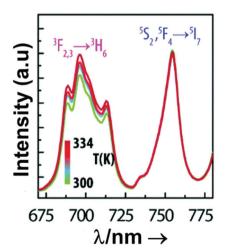


Fig. 10 Temperature dependence of the emission spectra of Ho,Tm: KLu(WO₄)₂ nanocrystals, demonstrating the similar intensity between the ${}^3F_{2,3} \rightarrow {}^3H_6$ of Tm³⁺ and ${}^5S_2, {}^5F_4 \rightarrow {}^5I_7$ of Ho³⁺. Reprinted with permission from ref. 150. Copyright 2018, The Royal Society of Chemistry.

maximum value at room temperature, and then it decreases as the temperature increases. However, there are three cases in which $S_{\rm rel}$ increases in the biological range of temperatures, decreasing later, and showing maxima at around 350–400 K. These luminescent thermometers, based on ${\rm Tm^{3+},Yb^{3+}}$: ${\rm Sr_2GdF_7},^{142}~{\rm Tm^{3+},Er^{3+},Yb^{3+}}$: ${\rm YF_3}$ nanoparticles embedded in a glass ceramics, ${\rm ^{153}}$ and ${\rm Tm^{3+},Er^{3+},Yb^{3+}}$: ${\rm LuF_3}$ mesocrystals, ${\rm ^{154}}$ would be the best options to perform luminescence thermometry for biological applications in the I-BW, particularly in the two first cases, since their maximum $S_{\rm rel}$ is around 2% K⁻¹, which might allow a δT of \sim 0.2 K, and the emission bands in which they are based have a similar intensity, simplifying the detection setup.

2.2. Neodymium-doped luminescent thermometers operating in the I-BW

Nd³⁺-doped luminescent thermometers have attracted significant attention due to their superior luminescent yields in the NIR spectral region. 158,173,203 Nd3+ ions can be excited with UV, VIS and NIR light to generate emissions in the I-BW, 204-206 with potential advantages for biomedical applications, including large penetration depths, minimal background interference, and little damage to the targeted samples, especially when excited in the NIR, enhancing their application as luminescent probes for various bioapplications as NIR-to-NIR luminescent nanoprobes. 207,208 Different Nd3+-doped luminescent thermometers operating in the I-BW have been designed based on three different pairs of TCLs that the electronic structure of this ion provides: ${}^{4}F_{7/2}/{}^{4}F_{3/2}$ (TCL 1), ${}^{4}F_{7/2}/{}^{4}F_{5/2}$ (TCL 2) and ${}^4F_{5/2}/{}^4F_{3/2}$ (TCL 3). In the field of luminescent nanothermometry, Nd3+-doped materials have been frequently used as sensitive thermometers in the physiological range of temperatures. An important number of papers have been also devoted to exploring luminescent thermometers based on the transitions from different Stark sublevels of the ⁴F_{3/2} excited state to the ⁴I_{9/2} ground state. Furthermore, other recent references explore the possibilities of developing dual emitting center luminescence thermometers combining Nd³⁺ with other Ln³⁺ ions, such as Eu³⁺, and transition metals like Ti⁴⁺. This section will focus first on the temperature-sensing properties of singledoped Nd³⁺ luminescent thermometers, then on dual emitting center Nd3+/Ln3+-codoped luminescent thermometers, and will conclude with dual emitting center Nd3+/transition metalcodoped luminescent thermometers, all of them operating in the I-BW.

2.2.1. Single Nd³⁺-doped luminescent thermometers operating in the I-BW. Contrary to single Tm³⁺-doped thermometers (section 2.1.1), single Nd³⁺-doped luminescent thermometers operating in the I-BW represent the widest explored class of luminescent thermometers operating in the I-BW due to the temperature dependence of the TCLs of Nd³⁺, lying in this spectral region. The thermometric performance of these thermometers, in general, is based on the FIR model, as can be seen in Table 1, because ΔE between these electronic levels lies in the range 200–2000 cm⁻¹. In single Nd³⁺-doped luminescent thermometers operating in this region, the TCL 1, TCL

2 and TCL 3, with theoretical ΔE 1910 cm⁻¹, 1020 cm⁻¹ and 960 cm⁻¹, 176 respectively, are applied to determine their thermometric performance.

The reported single Nd³⁺-doped luminescent thermometers operating in the I-BW have been excited with VIS and NIR light sources. In a typical example of exciting the Nd³⁺-doped particles with green light, the mechanism of the generation of the emission bands (Fig. 11(a)) is as follows: photons of the 580 nm excitation light source are absorbed by Nd³⁺, promoting its electrons from the 4I_{9/2} ground state to the 4G_{9/2} or 4G_{7/2} excited state, decaying then non-radiatively to the 4F5/2 and ⁴F_{7/2} states. From the ⁴F_{5/2} state, a radiative relaxation decaying back to the ground state generates an emission lying at 780-840 nm. The second emission band at 870-920 nm is generated after the non-radiative relaxation of the electrons from the ${}^4F_{5/2}$ state to the ${}^4F_{3/2}$ level, prior to a radiative relaxation to the ground state. 158 A third emission band at 730-770 nm is generated when electrons relax back to the ground state from the ⁴F_{7/2} state.

 ${
m Nd}^{3+}$ ions were embedded in hosts such as ${
m La_2O_2S},^{156}$ YAP, 157 Gd₂O₃, 158 LaPO₄, 159 fluorotellurite glass, 160 and strontium barium niobate (SBN) glass ceramic, 161 for this class of luminescent thermometers. In general, the performance of these luminescent thermometers was evaluated by analyzing the intensity ratio of the emissions corresponding to the $^4F_{5/2} \rightarrow ^4I_{9/2}$ and $^4F_{3/2} \rightarrow ^4I_{9/2}$ transitions (TCL 3). It should be

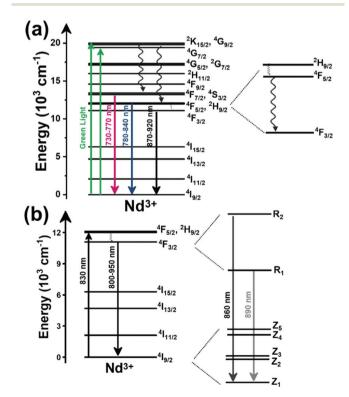


Fig. 11 Mechanisms of generation of the emission bands in single Nd^{3+} -doped nanoparticles after being excited with: (a) a green light source (the magnification detail depicts the thermally coupled ${}^4F_{5/2}$ and ${}^4F_{3/2}$ levels), and (b) a near infrared light source.

noted here that for La₂O₂S, the FIR model applied was that of eqn (14), in which a correction term accounting for the overlapping of the emissions was used although the emission bands were largely apart, whereas for the other cases, the classical FIR model, described by eqn (9), was applied. Nd³⁺-doped hexagonal La₂O₂S particles operated as a luminescent thermometer in the I-BW over the range of temperatures from 270 K to 600 K, exhibiting a maximum $S_{\rm rel}$ of 1.95% K⁻¹ at the lowest temperature analyzed, which is a little overestimated with respect to the other luminescent thermometers based on Nd³⁺ due to the fitting model used in its calibration. However, the big size of the particles obtained, and the temperature where this maximum value of S_{rel} was obtained, quite often might limit the potential applications of this thermometer.¹³ In fact, for this thermometer, by taking into consideration the value of the experimental ΔE reported (987 cm⁻¹), ¹⁵⁶ and the proper FIR equation (eqn (9)), the correct value of the S_{rel} that should be considered would be 1.15% K⁻¹ (at 293 K). In the case of the fluorotellurite glass, 160 and the SBN glass ceramic, 161 we calculated the values of the $S_{\rm rel}$ (and δT) by considering the ΔE provided by the authors and the temperature ranges analyzed. From all these results, we can conclude that the maximum $S_{\rm rel}$ of this class of luminescent thermometers would be in the range 1.15-1.83% K⁻¹, in all cases obtained at the lowest temperature analyzed. A higher S_{rel} was reported in Nd3+ embedded in YAG, when TCL 2 was taken into account. 155 If we compare the performance of the luminescent thermometers based on TCL 2 and TCL 3, it is clear that the best temperature-sensing properties were achieved for the former one, mainly due to the higher ΔE between the electronic levels. 155 Despite the higher $S_{\rm rel}$ (2.95% K^{-1}) this value was obtained at 200 K, whereas in the physiological range of temperatures, it drops down to 0.3% K⁻¹.155 The performance of the Nd3+-doped luminescent thermometers based on considering the Stark-sublevels of the $^4F_{3/2}$ \rightarrow $^4I_{9/2}$ transition resulted in a lower S_{rel} (0.11-0.62% K⁻¹) which implied a larger δT when determined using eqn (3) (>1 K), independently of the excitation source used. Nevertheless, in some cases in the bibliographic references δT was reported to be smaller, associated with better S_{rel} values reported that are impossible to get according to the E associated with the Stark sublevels involved in the emissions used to build the luminescent thermometer, of the order of 50-270 cm⁻¹ depending on the host in which the Nd3+ has been embedded. This energy difference, strictly speaking, does not allow for the application of the FIR model to those emission lines arising from energy levels apart by an energy difference below 200 cm⁻¹, so several reports fit the experimental data to an exponential function of the form of that shown in eqn (16), which might also explain why in some cases better S_{rel} values were reported.

Fig. 11(b) shows the mechanism of generation of the emission lines associated with the Stark-sublevels of the ${}^4F_{3/2} \rightarrow$ ⁴I_{9/2} transition after excitation in the NIR. In this way, the electrons in the Nd³⁺ ⁴I_{9/2} ground state are excited to the ⁴F_{5/2}, ²H_{9/2} states; then they relax via a non-radiative process populating the ⁴F_{3/2} metastable state. If instead, visible light in the green range is used to excite the sample, the electrons will be excited to the more energetic ${}^4G_{9/2}$ or ${}^4G_{7/2}$ excited states, as shown in Fig. 11(a), but from there, they relax non-radiatively to the ${}^4F_{5/2}$, ${}^2H_{9/2}$ state, from which the rest of the emission mechanism is the same. The ${}^4F_{3/2}$ metastable state may be radiatively depopulated, generating emission bands in the I-BW, assigned to the ${}^4F_{3/2} \rightarrow {}^4I_{9/2}$ transition. Since both levels (4F_{3/2} and 4I_{9/2}) are composed of several Stark sublevels $(R_1 \text{ and } R_2 \text{ for } {}^4F_{3/2}, \text{ and } Z_1-Z_5 \text{ for } {}^4I_{9/2}, \text{ as presented in }$ Fig. 11(b)), a plethora of emission bands lying in the 800-950 nm spectral range (Table 1) may be generated, which can be used to define different luminescent thermometers, depending on the host considered. Note that here we focused only on the generation of the emissions lines used to build this class of luminescent thermometers, without taking into account that Nd3+ can generate other emission lines in different spectral regions, as we will see in the following sections.

The case of Nd³⁺:LiLuF₄@LiLuF₄ should be mentioned here, where instead of the thermalization of the R2 and R1 levels, the contributions to peak shift and the width change of the intensity of the emissions were considered to build the luminescent thermometers, leading to higher sensitivity values (if only the thermalization is considered with an ΔE of 55 cm⁻¹, a maximum S_{rel} value of 0.09% K⁻¹ would have been achieved).164

Despite these limitations in this class of thermometer, Marciniak et al. investigated the effect of the Nd3+ concentrations and the alkali ion in the host on the thermometric performance of Nd³⁺-doped monoclinic tetraphosphates of the form ALaP₄O₁₂, where A = Li, K, Na and Rb. 165 The authors investigated the thermometric performance of these compounds, via three different classes of luminescence nanothermometry technique: band-shape, bandwidth and spectral position.

Regarding the performance of the luminescent thermometer via the band-shape thermometry technique, the experimental data were fitted to an exponential function of the form of that shown in eqn (16). The authors observed a tendency of the intensity ratio to increase as the temperature increased for all the investigated hosts. However, at higher temperatures, an inversion of this trend could be observed for the hosts containing Li and Na, due to the drastic lowering of the intensity of one of the emissions with respect to the other, due to the excited state absorption to higher energy levels (Fig. 12(a)). This effect reduces the usable temperature range for these luminescent thermometers. The highest S_{rel} was obtained for the LiLaP₄O₁₂ host while the lowest one was obtained for the KLaP4O12 host, while the other two, the RbLaP4O12 and NaLaP₄O₁₂ hosts, have similar values (Table 1), following a linear trend with the crystallographic b parameter of the materials, coinciding with the direction along which the (PO₄)³⁻ tetrahedral chains characteristic of these compounds run (Fig. 12(b)). 165 Here, the influence of the symmetric P-O-P vibration mode, that shifts towards lower energies as the mass of the alkali ion in the host increases due to the distance

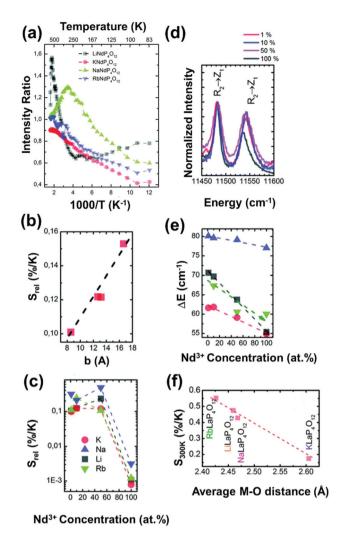


Fig. 12 (a) Effect of the temperature on the intensity ratio of Nd^{3+} -doped $ALaP_4O_{12}$ nanocrystals. (b) Variation of the relative sensitivity as a function of the cell parameter b of these compounds. (c) Change in $S_{\rm rel}$ as a function of the dopant concentration. (d) Comparison of the emission band positions for the $R_2 \to Z_1$ and $R_1 \to Z_1$ transitions for different concentrations of Nd^{3+} ions in $KLaP_4O_{12}$. (e) Effect of the dopant concentration on the energy separation between the R_1 and R_2 Stark sublevels for AP_4O_{12} nanocrystals. (f) Variation of $S_{\rm rel}$ with the M–O mean distance in Nd^{3+} -doped AP_4O_{12} nanocrystals. Reprinted with permission from ref. 165. Copyright 2016, The Royal Society of Chemistry.

between the alkali ion and the phosphorus ion, seems to play a major role. Also the concentration of Nd^{3+} in the hosts highly influences the S_{rel} value due to the dependence of the energy separation between the two Stark sublevels involved in the generation of the emissions used to build the thermometer (Fig. 12(c)). With the increase of the Nd^{3+} concentration, the position of R_2 and R_1 is shifted (Fig. 12(d)), using the KLaP₄O₁₂ host as an example. This tendency, with different rates, can be noticed for each of the other hosts. The decrease of the ΔE value was stronger for RbLaP₄O₁₂ (a decrease of 24 cm⁻¹), compared with LiLaP₄O₁₂ (15 cm⁻¹), NaLaP₄O₁₂ (6 cm⁻¹) and KLaP₄O₁₂ (3 cm⁻¹) (Fig. 12(e)). Since S_{rel} for these

nanothermometers is ruled by the Boltzmann distribution, it depends on ΔE , and it is clear that a decrease of this parameter will lead to a decrease in the thermal sensitivity.

Concerning spectral position thermometry, with the increase of the temperature a red-shift of the emission lines was observed for all the hosts analyzed, with $RbNdP_4O_{12}$ showing the strongest changes, while $KNdP_4O_{12}$ exhibited the smallest ones. This shift in the position of the emission lines is usually associated with the electron-phonon coupling effect that results from the introduction of random perturbations in the active-ion-surrounding environment by the host vibration modes at higher temperatures that can be thermally occupied, generating stronger electron–host interactions. This results in a decrease of the energy of the emitted phonons and thus a slight shift of the emission band towards higher wavelengths. The spectral change can be described by: 165,209

$$\Delta \lambda = \Delta \lambda^{S} + \Delta \lambda^{D} + \Delta \lambda^{M} + \Delta \lambda^{R} \tag{29}$$

where $\Delta \lambda^{S}$, $\Delta \lambda^{D}$, $\Delta \lambda^{M}$ and $\Delta \lambda^{R}$ represent the contributions to the spectral shifting of the crystal strain inhomogeneity, the single phonon processes, the multiphonon processes, and the electron-host interaction effect associated with Raman scattering, respectively. However, the three first parameters can be considered to be temperature independent, and thus only the electron-host interaction parameter should be considered as the one which controls the emission line shift as the temperature increases.

According to that, the change in the spectral position can be expressed as:^{209,210}

$$\delta\lambda = \delta\lambda_0 + \alpha \left(\frac{T}{\Theta_D}\right)^4 \int_0^{\Theta_D/T} \frac{x^3}{e^x - 1} dx$$

$$x = \frac{\hbar\Omega}{k_B T}$$
(30)

where $\delta\lambda_0$, α , Θ_D and $\hbar\Omega$ and represent the initial line position, the electron-host coupling parameter, the effective Debye temperature, and the phonon energy, respectively. The corresponding $S_{\rm rel}$, thus, can be calculated from the following equation:

$$S_{\text{rel}} = \frac{1}{\lambda_2} \frac{\lambda_2 - \lambda_1}{\Delta T} \times 100\% \tag{31}$$

where λ_2 and λ_1 represent the spectral shifts at temperatures T_2 and T_1 and ΔT is the result of $T_2 - T_1$.

 $S_{\rm rel}$ follows a linear inverse trend with the average distance between the metal and the oxygen host (Fig. 12(f)). After these results, the general conclusion that can be drawn is that the materials exhibiting the shortest metal–oxygen distance exhibited the highest thermometric performance. This is because the electron–host coupling parameter is proportional to the average metal–oxygen distance while the electron–phonon interaction strength increases with the covalence of the bond, which increases proportionally to the shortening of the chemical bond length. These results indicate that, to further enhance the thermometric performance of lumines-

cent thermometers based on Nd3+, host types with strong electron-host interaction parameters should be selected.

Using bandwidth luminescence nanothermometry, it was observed that the width of the emission band was broadened as the temperature increased, as a consequence of the electron-phonon interaction. To determine the thermometric performance, the bandwidth of the emission band at a particular temperature could be determined according to:²¹¹

$$\Delta \nu = \nu_0 \sqrt{\coth\left(\frac{\hbar \Omega}{2k_{\rm B}T}\right)} \tag{32}$$

where ν_0 stands for the full width at half-maximum at the initial temperature. The highest S_{rel} (determined using eqn (18)) was seen, yet again, for 1% Nd3+-doped LiLaP4O12 for a value of 0.32% K⁻¹. ¹⁶⁵ However, in that case, it was not possible to establish a correlation between the structural parameters of the host and $S_{\rm rel}$ for the case of the bandwidth luminescence nanothermometry.

As a final remark, among the three luminescence nanothermometry classes explored to study the thermometric performance of these compounds, the spectral position luminescence nanothermometry technique displayed the best temperaturesensing properties. 165 Despite this better performance, according to the authors the spectral position is much more difficult to accurately determine and requires high-resolution detection systems, which hampers the implementation of this technique to determine the temperature in real biomedical samples.¹⁶⁵ Thus, please note that Table 1 presents only the nanomaterial with the highest S_{rel} and the corresponding way of evaluating this performance. As can be observed from Table 1, the thermometric performance of this class of materials is still lower compared with other materials operating within this spectral region.

2.2.2. Dual Nd³⁺/Ln³⁺-codoped luminescent thermometers operating in the I-BW. Dual Nd3+/Ln3+ luminescent thermometers operating in the I-BW are mainly based on the incorporation of Yb3+ as a sensitizer due to its large absorption cross section at 980 nm, which would lead to brighter emissions. 196 This feature has led this element to be used together with Nd³⁺ and other Ln3+ ions in photon conversion processes involving ET mechanisms. For the case of Nd³⁺, the application of Yb³⁺ as a sensitizer allows the use of the three TCLs ($^4F_{7/2},\ ^4F_{5/2}$ and ⁴F_{3/2}) for temperature-sensing purposes (Table 1). However, as in the case of Tm³⁺/Yb³⁺-codoped luminescent thermometers, the overheating problem due to the strong water absorption band at around 980 nm makes nanothermometers containing Yb3+ and pumped at this wavelength non ideal for biomedical applications.²⁰⁶ The mechanism of the generation of the emissions in Nd3+/Yb3+-codoped thermometers is based on phononassisted energy transfer (hereafter PAET) processes. 179,181 In a typical process (Fig. 13(a)), under 980 nm excitation, electrons are excited from the Yb3+ $^2F_{7/2}$ ground state to the Yb3+ $^2F_{5/2}$ excited level. The transfer of energy from this state to the Nd3+ states is achieved through PAET processes. This allows the population of the ${}^4F_{3/2}$, ${}^4F_{5/2}$ and ${}^4F_{7/2}$ states of Nd³⁺ if the maximum phonon energy of the host is the suitable one for covering ΔE

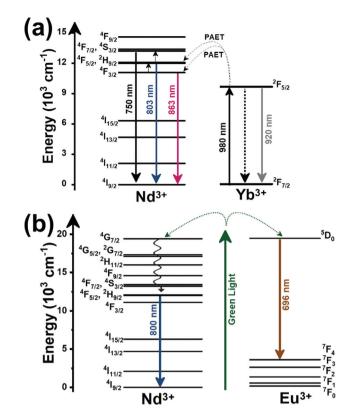
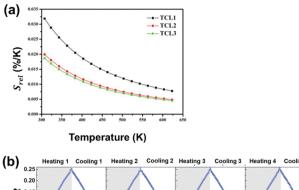
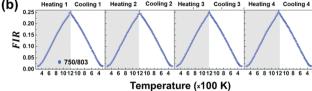


Fig. 13 Mechanisms of generation of the emission bands of nanoparticles codoped with Nd3+ and: (a) Yb3+ after 980 nm excitation, and (b) Eu3+ after excitation in the green range.

between these energy levels, of the order of 1750 cm⁻¹. Also, the PAET process becomes more effective as the temperature increases, 179,181 making these electronic levels thermally coupled. Hence, thermal population of the ${}^4F_{5/2}$ and ${}^4F_{7/2}$ levels can be achieved from the lower energy level. From these states, radiative decay to the ground state generates Nd3+ emissions in the I-BW window at 863 nm, 803 nm and 750 nm, respectively, as shown in Fig. 13(a). The performance of these Nd³⁺,Yb³⁺codoped nanoparticles was evaluated by the typical FIR model. The most used TCLs are the ${}^4F_{3/2}$ and the ${}^4F_{7/2}$ levels, which showed higher S_{rel} values than when the emission arising from the ⁴F_{5/2} level is used (Table 1). A full comparison between the temperature-sensing performance of the three TCLs in Nd3+ with emissions in the I-BW is provided by Nd3+,Yb3+-codoped oxyfluoride glass. 176 The results show that the thermometric performance follows the order: ${}^{4}F_{7/2}/{}^{4}F_{3/2} > {}^{4}F_{7/2}/{}^{4}F_{5/2} > {}^{4}F_{5/2}/{}^{4}F_{3/2}$ (Fig. 14(a)), which is in agreement with the experimental ΔE determined between these levels (2076 cm⁻¹, 1300 cm⁻¹ and 1216 cm $^{-1}$, respectively). The S_{rel} values obtained when using the ${}^4F_{3/2}$ and ${}^4F_{7/2}$ TCLs allows the calculation of a δT of ~0.15-0.2 K, approaching the desired values for biomedical applications,13 although this low limit is always attained at the minimum temperature analyzed, normally room temperature.

Another interesting example here is the luminescent thermometer constituted by Nd3+,Yb3+-codoped trigonal La2O3 microparticles, in which the thermometric performance was





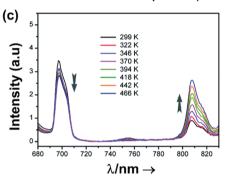


Fig. 14 (a) Evolution of S_{rel} as a function of the three TCLs in Nd³⁺, Yb^{3+} -codoped oxyfluoride glass: $TCL1 = {}^{4}F_{7/2}/{}^{4}F_{3/2}$, $TCL2 = {}^{4}F_{7/2}/{}^{4}F_{5/2}$ and TCL3 = ${}^{4}F_{5/2}/{}^{4}F_{3/2}$. Reprinted with permission from ref. 176. Copyright 2013, Elsevier. (b) Repeatability of the FIR 750 nm/803 nm over 4 heating and cooling cycles from 293 to 1233 K with a step of 20 K in Nd3+,Yb3+-codoped trigonal La2O3 microparticles. Adapted with permission from ref. 181. Copyright 2018, The Royal Society of Chemistry. (c) Emission spectra of Eu,Nd:YVO₄ nanoparticles under excitation at 590 nm in the temperature range of 299-466 K. Adapted with permission from ref. 183. Copyright 2020, The Royal Society of Chemistry.

explored in the widest temperature range among this class of thermometers, from room temperature to 1233 K. 181 Here, the repeatability of the measurements was tested via four heating and cooling cycles, revealing an excellent performance (Fig. 14(b)). Also, based on these heating and cooling cycles, the authors could calculate the temperature uncertainty as:

$$AT_{\min} = \frac{\sigma}{S_{\text{rel}}} \tag{33}$$

where σ represents the standard deviation of the intensity ratio $({}^{4}F_{7/2}/{}^{4}F_{3/2})$ in this case). The value obtained was 0.1 K in the temperature range below 400 K, whereas it increased for higher temperatures, reaching a maximum of ~3 K at 1230 K. 181

An interesting choice with these dual emitting center Nd³⁺, Ln³⁺-codoped luminescent thermometers operating in the I-BW is the choice of combining the emissions of Nd³⁺ with the emission of Yb3+, which besides acting as a sensitizer can also generate an emission line that can be used to measure the temperature. An example of this kind of thermometer is that

of Nd3+,Yb3+:Al4B2O9 orthorhombic nanoparticles excited at 977.7 nm. 179 The mechanism of the generation of the emissions is exactly the same as that described before in Fig. 13(a), including the emission generated by the ${}^2F_{5/2} \rightarrow {}^2F_{5/2}$ radiative relaxation in Yb $^{3+}$. The highest S_{rel} was obtained when calculating the 800 nm versus 920 nm intensity ratio of the emissions of Nd³⁺ and Yb³⁺, respectively (Table 1), as ΔE is larger when compared with the one between the 864 nm and 920 nm emissions. 179 The results obtained from this combination are comparable to that obtained for the TCL 3 when only the Nd³⁺ emissions are used, but are still lower than that of the TCL 1.

Besides Nd³⁺/Yb³⁺-codoped luminescent thermometers operating in the I-BW, another choice, rarely reported, is Nd³⁺/ Eu³⁺-doped materials. These luminescent thermometers are based on the emissions arising from non-thermally coupled levels (hereafter NTCLs) of Eu³⁺ at ~700 nm and Nd³⁺ at ~800 nm. 182,183 The reason why these Nd3+/Eu3+ co-doped thermometers were proposed was to overcome the drawback of the limitation in the maximum value of the S_{rel} that can be achieved operating with TCLs.212 Thus, the strategy of implementing emission lines generated from two different active centers was tested. The mechanism of the generation of these emission lines is depicted in Fig. 13(b). The excited energy states ⁵D₀ of Eu³⁺ and ⁴G_{7/2} of Nd³⁺ are very close in energy; therefore they can be simultaneously excited at 590 nm. From these excited states, the electrons of Eu3+ can fall back to the $^{7}F_{4}$ state, generating the emission located at ~700 nm. The electrons of Nd3+, from the 4G7/2 level, followed a series of nonradiative relaxations that populate the 4F5/2 level (among others). From this level, the radiative relaxation back to the ground state generates the emission at ~800 nm.182,183 It is worth noticing here again that we only focused on the mechanisms to generate the emissions involved in the luminescent thermometers, independently of the fact that other emission lines can be generated with these ions.

Two examples that apply this combination of Nd³⁺ and Eu³⁺ ions involve YVO4, 183 and Ba2LaF7, 182 as hosts. One of the interesting characteristics of these luminescent thermometers is that the emission that should be expected at around 750 nm for Nd³⁺, arising from the transition of the ⁴F_{7/2} level to the ground state, is very weak, compared with that observed in Nd³⁺/Yb³⁺ luminescent thermometers, so it cannot be used for luminescence thermometry in this case. Another interesting feature of this class of thermometers is that while the emission of Eu³⁺ changes slightly with temperature, the emission of Nd³⁺ is highly dependent on the temperature, and its intensity increases as the temperature increases (Fig. 14(c)). In the case of Ba₂LaF₇, the S_{rel} reported was calculated using eqn (1), corresponding to S_{abs} , so it cannot be compared with that of other luminescent thermometers reported in the literature. From the fitting function a ΔE of 1890.8 cm⁻¹ could be extracted, 182 a value that can be used to calculate S_{rel} using egn (10), obtaining a value of 2.2% K^{-1} and a δT of 0.22 K at the lowest temperature under investigation (290 K).

From Table 1 and Fig. 15(a) and (b), it can be noted that the purpose of these Nd³⁺/Eu³⁺-codoped luminescent thermoReview Nanoscale (b)_{3.5} (a) **Å** 184 2.5

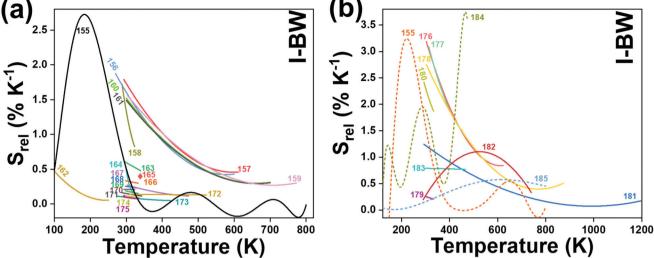


Fig. 15 Temperature dependence of S_{rel} for: (a) single doping and (b) codoping with other lanthanide ions (in solid lines) and transition metal ions (in dashed lines) for Nd3+ luminescent thermometers, operating in the I-BW. The numbers represent the corresponding references for each Nd3+based thermometer

meters to achieve better thermal sensing properties than that of thermometers based on TCLs is partially achieved. The performance is better when compared with that of the Nd3+/Yb3+ luminescent thermometers operating under TCL 3, similar to TCL 2, but still lower than that of TCL 1. However, an important aspect to consider here is that Nd³⁺/Eu³⁺-codoped luminescent thermometers are excited in the green region, whereas the ones based on Nd³⁺/Yb³⁺ are excited in the NIR. Thus, a limitation for this class of luminescent thermometers is the penetration depth that can be achieved with them in biological applications.⁸⁶

2.2.3. Dual Nd3+/transition metal-codoped luminescent thermometers operating in the I-BW. The combination of Ln³⁺ with transition metals in the same host lattice introduces new luminescence properties involving the emissions of both kinds of ion. 213 Mixed Ln3+/transition metal compounds with luminescence properties from both ions have emerged recently with especial interest for luminescence thermometry, attributable to the fact that the luminescence properties of the transition metals are drastically affected by local symmetry changes and the allowed character of the d-d transitions, which renders these ions as competitive thermal probes when compared with the f-f transitions of Ln³⁺. 155,184,185 Here, the intensity of the emissions arising from the Ln3+ change barely with temperature; thus, they can be considered as reference signals. 155,184,185 For the Nd³⁺-based luminescent thermometers operating in the I-BW, the transition metals involved for thermal sensing purposes include titanium, 184 chromium, 155 and manganese, 185 all embedded in yttrium aluminum garnet (YAG). To generate the emission lines, however, these materials have to be excited with UV (exciting Ti(IV) and Mn(IV)) and VIS (exciting Cr(III)) light (Table 1)), which will hamper their possible biomedical applications. As an illustrative example, Nd3+,Cr3+-co-doped YAG was excited at 590 nm and the multiple emissions generated are presented in Fig. 16(a).

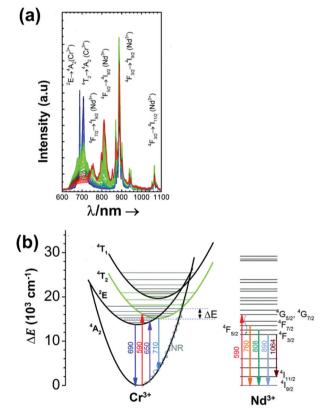


Fig. 16 (a) Emission lines and (b) mechanisms for their generation in Nd3+,Cr3+-codoped YAG, after being excited at 590 nm. Reprinted with permission from ref. 155. Copyright 2017, the Owner Societies.

Absorption of light at 590 nm allows for the excitation of electrons of Cr3+ from the 4A2 ground state to the 4T2 excited state, which was followed by nonradiative multiphonon relax-

ation processes leading to the population of the ²E state. Radiative relaxation from these two states led to narrow (690 nm) and broad emission bands (710 nm), assigned to the $^{2}\text{E} \rightarrow ^{4}\text{A}_{2}$ and $^{4}\text{T}_{2} \rightarrow ^{4}\text{A}_{2}$ transitions, respectively. ¹⁵⁵ In Fig. 16(b), the displacement of the position of the ⁴T₂ state with respect to that of the ⁴A₂ ground state is due to the distortion introduced by the crystal field, leading to the appearance of a crossing point between these parabolas. 155 Through this crossing point, electrons can be transferred to the ground state via nonradiative multiphonon relaxation processes, if the thermal energy provided to the system is sufficiently high. 155 Nd³⁺ emissions are generated after this ion absorbs the energy provided by the excitation source, promoting its electrons to the ${}^4G_{5/2}$, ${}^4G_{7/2}$ excited states, followed by a non-radiative relaxation to the ⁴F_{3/2} state, where either: (i) radiative relaxation to the ⁴I_{9/2} and ⁴I_{11/2} states, or (ii) thermalization of the higher energy levels ⁴F_{5/2} and ⁴F_{7/2}, can occur, giving rise to emission bands in the I-BW and II-BW. 155

The performance of this luminescent thermometer (and the Nd³⁺/Ti⁴⁺ and Nd³⁺/Mn⁴⁺ systems as well) was evaluated through the intensity ratio, defined as:¹⁵⁵

$$\Delta = \frac{I_{0 \text{ Cr}}}{1 + \exp\left(\frac{\Delta E_1}{k_{\text{B}}T}\right)} \frac{1}{I_{0 \text{ Nd}} - \left[1 + \exp\left(\frac{\hbar \omega}{k_{\text{B}}T}\right)\right]^p - \left(-\exp\left(\frac{\Delta E_2}{k_{\text{B}}T}\right)\right)}$$
(34)

The temperature dependence of this intensity ratio involves 3 mechanisms:

- (i) the luminescence thermal quenching with a cross-over point, where $I_{0 \text{ Cr}}$ and ΔE_{1} represent the initial intensity of the Cr^{3+} emission bands at low temperature, and the energy difference between the bottom of the excited state parabola and the energy of the crossing point of the excited and ground state parabolas (Fig. 16(b)).
- (ii) the temperature dependence of the intensity of the emission band arising from the ${}^4\mathrm{F}_{3/2} \to {}^4\mathrm{I}_{9/2}$ transition of Nd $^{3+}$, where I_0 Nd is the initial intensity of this transition, whose intensity decreased as the temperature increased due to multiphonon depopulation processes, and where $\hbar\omega$ and p represent the maximum phonon energy of the host material and the number of phonons involved in the process, respectively.
- (iii) the thermal depopulation of the ${}^4F_{3/2}$ state towards the ${}^4F_{5/2}$ state of Nd³⁺, lying $\Delta E_2 = 1000$ cm⁻¹ above the ${}^4F_{3/2}$ state.

 $S_{\rm rel}$ was calculated according to eqn (2). The temperature dependence of $S_{\rm rel}$ for each Nd³⁺/transition metal luminescent thermometer is presented in Fig. 15(b), whereas their maxima are listed in Table 1. A maximum $S_{\rm rel}$ of 3.49% K⁻¹ was achieved for this class of luminescent thermometers, being higher than the one that can be achieved in Nd³⁺/Yb³⁺ and single-doped Nd³⁺ luminescent thermometers. In fact, this value of sensitivity is approximately 3 times higher than the thermal sensitivity that can be obtained when using the intensity ratio of TCL 2 or TCL 3 in single-doped Nd³⁺ in YAG. The reason behind this improvement in the thermometric perform-

ance is the efficiency of the Cr³⁺ → Nd³⁺ ET process. ¹⁵⁵ This class of luminescent thermometers, constituted by the combination of the emissions arising from Nd3+ and a transition metal, definitely exhibit higher temperature-sensing performances, especially in the case of Nd³⁺/Ti⁴⁺ and Nd³⁺/Cr³⁺. Thus, their potential application as temperature sensors is attractive. Despite this, as can be observed in Fig. 15(b) and in Table 1, the maximum S_{rel} values are either achieved at lower (Nd³⁺/ Cr³⁺) or higher (Nd³⁺/Ti⁴⁺) temperatures than those similar to those that can be achieved with Nd3+/Yb3+ luminescent thermometers (for Nd^{3+}/Ti^{4+} this value is ~3.3% K^{-1} , ¹⁸⁴ for example). It should be emphasized as well that these systems are quite complex, and detailed studies to understand the effect of the factors that govern the temperature-sensing properties, such as the electron-phonon interaction, the crystal field strength and the physical processes responsible for temperature susceptibility, are highly needed. 184 Last, these materials are excited with UV and VIS light, at wavelengths that might induce autofluorescence, phototoxicity, and limited penetration depths when applied in biological tissues.86 Thus, novel strategies to excite these materials with NIR light sources are needed.

2.3. Europium-doped luminescent thermometers operating in the I-BW

The application of Eu³⁺-doped materials as luminescent thermometers operating in the I-BW is based on their emissions located in the deep red region, ~710 nm, assigned to the $^5D_0 \rightarrow {}^7F_4$ transition. These thermometers are rarely reported in the literature (Eu³⁺:Y₂O₃). However, the most significant aspect of the Eu3+:Y2O3 luminescent thermometer it that it offered for the first time the possibility of predicting the calibration curve of the thermometric parameter independently of the medium in which this thermometer operates, allowing the development of a luminescent primary thermometer based on Ln³⁺-doped materials. Primary thermometers are systems in which the temperature can be determined based on the knowledge of thermodynamic laws and quantities.214 In general, to extract the thermometric parameter, the calibration process requires an independent measurement of the temperature through a thermocouple or an IR camera. But when this thermometer operates in different media, a new calibration procedure is required. Hence, primary thermometers guarantee a single calibration procedure that is medium-free. The thermometric parameter, Δ , in Eu³⁺:Y₂O₃ was determined by the ratio between the emission intensity of the its ${}^5D_0 \rightarrow {}^7F_4$ transition, when the 5D0 state is excited through the 7F2 or the ⁷F₀ state, in the physiological range of temperature, or through the ⁷F₁ and ⁷F₀ states, for lower temperatures (down to 180 K). 186 Considering the physiological range of temperatures as a particular region of interest for biomedical applications, Δ is defined as:

$$\Delta = \frac{I_{23} \ (^{7}F_{2})}{I_{23} \ (^{7}F_{0})} = \frac{p \ (^{7}F_{2})}{p \ (^{7}F_{0})} \frac{W_{31} \ (^{7}F_{2})}{W_{31} \ (^{7}F_{0})}$$
(35)

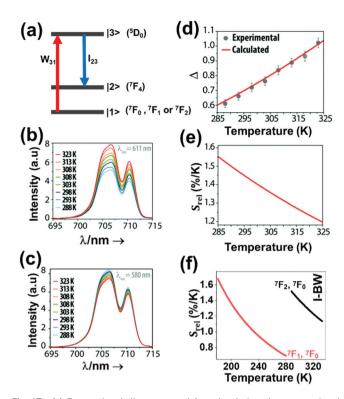


Fig. 17 (a) Energy level diagram used for stimulating the energy levels of Eu³⁺ involved in the luminescent thermometer operation. Deep red emission spectra of Eu³⁺:Y₂O₃ nanocrystals excited at: (b) 611 nm, and (c) 580 nm. (d) Thermometric parameter calculated in the physiological range of temperatures. Points are the experimental values of the Δ parameter obtained from the spectra in (b) and (c) after being corrected for the respective excitation intensity, whereas the line is the calculated curve obtained from eqn (37), corresponding to the primary thermometer. (e) Relative sensitivity in the physiological range of temperatures. (b)-(e) Reprinted with permission from ref. 186. Copyright 2016, The Royal Society of Chemistry. (f) The variation of S_{rel} of Eu^{3+} -doped nanothermometers operating in the I-BW. The red line is computed using eqn (35) and the black line using eqn (36).

where I_{23} is the intensity of the transition from level 3 (${}^{5}D_{0}$) to level 2 (${}^{7}F_{4}$) at the steady state regime (Fig. 17(a)), p stands for the Boltzmann population factor, and W_{31} is the absorption rate from level 1 (${}^{7}F_{0}$ or ${}^{7}F_{2}$) to level 3 (Fig. 17(a)).

 Δ is determined by measuring the integrated areas under the emission curves of the ${}^5D_0 \rightarrow {}^7F_4$ transition, excited resonantly through the 7F2 and the 7F0 levels, considering that the two different excitations towards the ⁵D₀ level are the same and are dielectric in nature, and that the emitting level (5D₀) and the 7F₀ ground state are nondegenerate:186

$$\Delta = \frac{S \left({}^{5}D_{0} \rightarrow {}^{7}F_{2}\right)}{S \left({}^{5}D_{0} \rightarrow {}^{7}F_{0}\right)} \exp\left(-\frac{\Delta E}{k_{\mathrm{B}}T}\right)$$
(36)

where, S ($^5D_0 \rightarrow {}^7F_2$) and S ($^5D_0 \rightarrow {}^7F_0$) represent the area under the red emission curves, excited resonantly through the $^{7}F_{2}$ and $^{7}F_{0}$ levels, respectively; and ΔE stands for the energy difference between the ⁷F₂ and the ⁷F₀ levels. To calculate the

thermometric parameter in the physiological range of temperatures, the theoretical expression of this parameter, predicted by considering the absorption rates in eqn (35), is expressed as:186

$$\Delta = 51 \exp\left(-\frac{875}{k_{\rm B}T}\right) \tag{37}$$

where $\Delta E = 875 \text{ cm}^{-1}$ and the pre-exponential factor 51 are determined from the energy difference between the barycenters when excited via the ⁷F₂ or ⁷F₀ level, and the ratio between the areas of the emissions rising from these levels, respectively. The experimental values of Δ were extracted from the integrated intensities of the ${}^5D_0 \rightarrow {}^7F_4$ transition. Two excitation wavelengths were used, 611 nm and 580 nm, to excite the 5D0 level resonantly from the 7F0 and the 7F2 levels, respectively. The red emission observed displayed a different behavior with the increase of the temperature, depending on the excitation wavelength used. When excited at 611 nm, the intensity of the red emission increased due to the increase of the thermal population of the Stark sublevels of the ⁷F₂ state (Fig. 17(b)), whereas when excited at 580 nm, the opposite tendency was detected, due to the thermal depopulation of the ${}^{7}F_{0}$ ground state (Fig. 17(c)).

The experimental values were in excellent agreement with those obtained from the equation defining the primary thermometer (eqn (37)) (Fig. 17(d)), displaying an error of 3%, 186 and demonstrating the successful development of a Ln³⁺-doped material as a primary luminescent thermometer. In terms of S_{rel} , it is in the order of 1.55% K⁻¹ at the physiological range of temperatures (Fig. 17(e)). Higher values of thermal sensitivity (~1.7% K⁻¹ at 180 K) were obtained when calculating the thermometric parameter from the red emission excited through the ⁷F₁ and ⁷F₀ levels. Also for this case, the principles of the primary thermometer could be applied, with a maximum error of 2%. 186 In that case, the thermometric parameter was defined as:

$$\Delta = \left[\frac{n^3 \left(\frac{n^2 + 2}{3} \right)}{n^5} \right] \frac{S \left({}^5 D_0 \to {}^7 F_1 \right)}{S \left({}^5 D_0 \to {}^7 F_0 \right)} \exp \left(-\frac{\Delta E}{k_{\rm B} T} \right)$$
(38)

where n stands for the refractive index of the host where the Eu³⁺ ions are embedded.

If we compare the performance of this single-doped Eu³⁺ luminescent nanothermometer with the dual emitting center Nd³⁺/Eu³⁺ ones, based as well in the red emission of Eu³⁺, the latter exhibit higher Srel in the physiological range of temperatures (Table 1).

2.4. Erbium-doped luminescent thermometers operating in the I-BW

The application of erbium (Er3+) doped materials as luminescent nanothermometers operating in the I-BW is mainly assigned to the presence of the red (~660 nm) and NIR (800 nm and 850 nm) emissions, either in single-doped materials, or in combination with other dopants, including other Ln3+ and transition metals.

2.4.1. Single Er^{3^+} -doped luminescent thermometers operating in the I-BW. Single Er^{3^+} -doped materials used as luminescent nanothermometers operating in the I-BW are based on the emissions located in the red and NIR. Upon excitation in the visible or NIR, these emission lines are generated, governed by the cross relaxation (CR) process (${}^4F_{7/2}$, ${}^4I_{11/2}$) \rightarrow (${}^4F_{9/2}$, ${}^4F_{9/2}$) (Fig. 18(a)). 187 In short, the electrons of Er^{3^+} , via absorption of two photons, are excited from the ground state to the ${}^4I_{11/2}$ level first, and further to the ${}^4F_{7/2}$ excited level at the end of the process. Also in short, the electrons of Er^{3^+} , via the absorption of two photons, are excited from the ground state to the ${}^4I_{9/2}$ level first, and further to the ${}^4F_{7/2}$ excited level at the end of the process. From this level, successive non-radiative decays lead to the population of the ${}^4I_{9/2}$ and ${}^4F_{9/2}$ levels. From these levels, when relaxing back to the ${}^4I_{15/2}$ ground

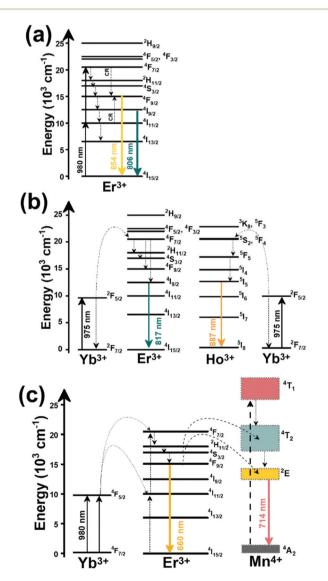


Fig. 18 Mechanisms of generation of the emission lines located in the I-BW of: (a) ${\rm Er}^{3+}$, (b) ${\rm Er}^{3+}/{\rm Ho}^{3+}/{\rm Yb}^{3+}$, and (c) ${\rm Er}^{3+}/{\rm Mn}^{4+}/{\rm Yb}^{3+}$ doped materials.

state, emissions lying in the red (654 nm) and in the NIR (806 nm) are generated (Fig. 18(a)). 187

The operation of these luminescent thermometers is based on the temperature dependence of the electronic population of the different Stark sublevels of the ${}^4F_{9/2}$ level, with emissions located in the red region, the intensity ratio of the emissions at ${\sim}800$ nm, or the intensity ratio between the red and NIR emissions (Table 1). Among these, the high $S_{\rm rel}$ extracted from the intensity ratio of the NIR emissions represents a promising strategy.

Er³⁺-doped strontium barium niobate (SBN) glass ceramic was investigated as a luminescent thermometer in a wide range of temperatures (300-700 K), by studying the intensity ratio in the NIR between the 800 nm and 850 nm emissions, attributed to the ${}^2H_{11/2} \rightarrow {}^4I_{13/2}$ and ${}^4S_{3/2} \rightarrow {}^4I_{13/2}$ electronic transitions, respectively. 161 The temperature dependence of this intensity ratio was fitted to the FIR equation (eqn (9)), achieving a maximum $S_{\rm rel}$ and a minimum δT of 1.39% K⁻¹ and 0.36 K, respectively, at the lowest temperature investigated. It should be noted here that the authors did not report these two parameters, but they provided an experimental value of ΔE (872.3 cm^{-1}) between the thermally coupled ${}^{2}\text{H}_{11/2}$ and ${}^{4}\text{S}_{3/2}$ levels, which allowed us to determine their performance. However, the excitation wavelength (532 nm) used to activate this luminescent thermometer, 161 located in the visible region, may hamper its biological/biomedical applications.

The $S_{\rm rel}$ obtained from the intensity ratio between the red and NIR emissions is higher when compared with that achieved using the emissions of the Stark sublevels of the ${}^4F_{9/2}$ level. An example of this class of luminescent thermometers is that of Er^{3+} incorporated within active core–inert shell NaErF₄@NaGdF₄ nanoparticles. ¹⁸⁷ The emissions of these nanocrystals, located at 654 nm and 806 nm, are generated from excitation either at 980 or 1530 nm. ¹⁸⁷ These two emissions arise from NTCLs, and their performance was extracted by fitting the intensity ratio with a second-order polynomial function, as presented in eqn (27).

When using emissions arising from a different Stark sublevel of the ⁴F_{9/2} level, red emissions (654 nm and 660 nm) were obtained after excitation at 800 nm. The S_{rel} of this class of luminescent nanothermometers was up to 9 times smaller than that obtained with the two other luminescent thermometer classes described in this subsection (Table 1). The reason for that is the low ΔE between the different Stark sublevels from which the emissions arose. 188,189 Moreover, the overlap of the two emission signals causes a low discriminability between the emission lines, and a large detection deviation. 189 Regardless of the poor performance of this class of luminescent thermometers (Fig. 19), it is worth mentioning here the idea behind heavily concentrated core@shell@shell Tm³⁺:NaErF₄@Yb³⁺: NaYF₄@Nd³⁺:NaYbF₄ nanostructures, in which Tm³⁺-mediated transient energy trapping coupled to Nd3+/Yb3+ cascade-sensitization is used to efficiently trigger a single-band red emission of Er³⁺, after careful optimization of the doping composition. ¹⁸⁸

2.4.2. Er³⁺/Yb³⁺-codoped luminescent thermometers operating in the I-BW. Materials codoped with Er³⁺ as an activator

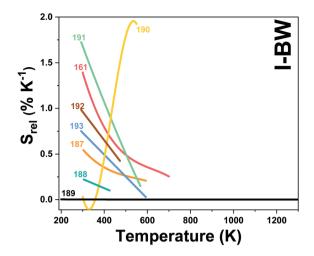


Fig. 19 Temperature dependence of S_{rel} of Er³⁺-doped luminescent thermometers, operating in the I-BW. The numbers represent the corresponding references for each Er³⁺-based thermometer.

and Yb3+ as a sensitizer are the most explored class of Er3+ luminescent thermometers operating in the I-BW. Yb³⁺ ions, due to the efficient absorption at 980 nm, allow the development of NIR-to-NIR based luminescent thermometers, besides improving the signal intensity of Er³⁺ emissions. 191-193 Er³⁺/ Yb3+-codoped materials can be applied as luminescent thermometers based on the emissions of Er3+, or in combination with other emissions arising from other Ln3+ or transition metals (Table 1). An effective strategy is the combination of emission arising from Er³⁺ and Ho³⁺ in the NIR region, in the presence of Yb3+ as a sensitizer, in hexagonal NaLuF4 microcrystals. 191 The emission bands used to determine the thermometric performance of these particles are located at 887 nm and 817 nm, assigned to the ${}^{5}I_{6} \rightarrow {}^{5}I_{8}$ transition of Ho^{3+} and the ${}^4\mathrm{S}_{3/2} \to {}^4\mathrm{I}_{13/2}$ transition of Er^{3+} , respectively. The mechanism of generation of these emissions is presented in Fig. 18(b). In short, Yb³⁺ absorbs the 975 nm excitation source and transfers the absorbed energy to the emitters Ho3+ and Er³⁺, exciting their excited states via ET mechanisms.¹⁹¹ From these excited states (${}^4F_{7/2}$ for Er^{3+} and 5S_2 , 5F_4 for Ho^{3+}), and via non-radiative decays, the ${}^4\mathrm{F}_{9/2}$ (Er $^{3+}$) and ${}^5\mathrm{I}_5$ (Ho $^{3+}$) electronic levels are populated, and from their radiative decays back to the ground states (${}^4I_{15/2}$ (Er $^{3+}$) and 5I_8 (Ho $^{3+}$)), the corresponding 817 nm and 887 nm emission lines are generated. Since these two emission lines arise from different emitting ions, the intensity ratio was fitted to a second-order polynomial function of the form of that in eqn (27). The maximum $S_{\rm rel}$ was ~1.73% K⁻¹ at room temperature. ¹⁹¹

When codoped Er³⁺,Yb³⁺ materials were investigated alone, without the addition of other ions, S_{rel} decreased around 2 times (Table 1). Two typical examples exploring this strategy are Er³⁺,Yb³⁺: ZrO₂, ¹⁹³ and Er³⁺,Yb³⁺:YF₃ nanoparticles. ¹⁹² In these cases, the temperature-sensing properties were explored by using the emissions located in the red (Er³⁺,Yb³⁺:ZrO₂) and NIR (Er³⁺,Yb³⁺:YF₃) spectral regions, by exciting the particles at 980 nm. Here, it is important to note that the luminescent

thermometer based on Er3+,Yb3+:ZrO2 nanoparticles worked under the so-called "valley-to-peak" model (VPR). 193 According to this model, and for emissions that originate from TCLs, if the emission lines are close enough, they will overlap, and as a consequence, a valley will be formed between them. 193,215 As the temperature increases, the linewidth of each emission peak broadens monotonically and the intensity valley formed overlaps also. 215 The VPR model is predicted to be a monotonic function of the temperature and is expressed as:193,215

$$VPR = a(T - T_0) + b \tag{39}$$

where a is the slope of the linear fitting, b is a constant, and T_0 and T represent the initial and final temperatures, respectively. $S_{\rm rel}$, described as the changing rate of VPR with temperature, is expressed as:193,215

$$S_{\rm rel} = \frac{1}{\rm VPR} \frac{\rm d\, VPR}{\rm d\, T} \tag{40}$$

The temperature evolution of this S_{rel} is presented in Fig. 19, together with that of the rest of the luminescent thermometers described in this subsection, showing a tendency to decrease as the temperature increases.

Er³⁺,Ho³⁺,Yb³⁺:NaLuF₄ luminescent thermometers exhibit a $S_{\rm rel}$ two times larger (Table 1), so that the combination of the emissions of Er3+ and those of other Ln3+, in the presence of Yb3+ as a sensitizer, is a promising strategy for luminescence thermometry.

An additional strategy towards better temperature-sensing properties is that of codoping materials with Er3+ and transition metals, such as Mn⁴⁺. Manganese ions generally display red to NIR luminescence assigned to the spin-forbidden $^2\mathrm{E} \rightarrow$ ⁴A₂ transition under excitation with UV or blue light, owing to their high effective positive charge and the influence of a strong local crystal-field. 216,217 Upon proper modification of the crystal-field environment, the spectral position of these ions can be tuned from 620 nm to 723 nm. 218,219 Hence, since the luminescence of transition metals is highly influence by the medium in which they are embedded, and often it drastically reduces, this makes them highly desirable in luminescence nanothermometry, in combination with Ln3+, in which the latter acts as a reference probe. 165,184,185,190

In comparison with the case of materials codoped with Nd3+ and transition metal materials, in which the excitation source was UV or VIS light (section 2.2.3), here, the incorporation of Yb3+ as a sensitizer has allowed for excitation in the NIR, overcoming the problems related to the phototoxicity and limited penetration depth exhibited by UV and VIS light.

Triply doped Er3+,Mn4+,Yb3+:YAP nanocrystals were investigated as luminescent temperature sensors operating in the I-BW over the temperature range 300-550 K. 190 They worked with the intensity ratio of the emissions arising from NTCLs located at 714 nm and 660 nm, corresponding to the ${}^{2}\text{E} \rightarrow {}^{4}\text{A}_{2}$ and ${}^{4}F_{9/2} \rightarrow {}^{4}I_{15/2}$ electronic transitions of Mn⁴⁺ and Er³⁺, respectively. The generation of these emission bands is presented in Fig. 18(c). Yb3+ absorbs the energy of the 980 nm

excitation source, and via two photon-assisted processes, the excited energy levels (4F_{7/2} and 4I_{11/2}) of Er³⁺ are populated. Through a non-radiative decay process, the ⁴F_{9/2} state is populated, which when relaxing back to the ⁴I_{15/2} ground state generates the red emission at 660 nm. Via two ET processes, from the ${}^4F_{7/2}$ and ${}^4F_{9/2}$ levels of Er^{3+} , the 4T_2 and 2E levels of Mn^{4+} are populated. The ²E energy level can relax back to the ⁴A₂ ground state, generating the emission centered at 714 nm. ¹⁹⁰

To calculate the dependence of the intensity ratio with the temperature and the thermometric performance the same model as in the case of materials codoped with Nd³⁺ and transition metals (eqn (34)) was applied. The maximum S_{rel} obtained was 1.95% K⁻¹ at 530 K.¹⁹⁰ Despite the relatively high thermal sensitivity reported, this value was obtained at a higher temperature than those used in the physiological range, for which S_{rel} reduces substantially. ¹⁹⁰ The overall temperature dependence of S_{rel} of this material is depicted in Fig. 19, together with those of all the other Er³⁺-doped luminescent thermometers operating in the I-BW.

To conclude with the section of Er³⁺-doped luminescent thermometers operating in the I-BW, it should be said that the most promising strategy identified for improving their thermometric performance involves dual emitting center nanoparticles, either with other Ln³⁺ or with transition metals. Concerning the use of red or NIR emissions, the better choice is using NIR emissions, not only because of the better performance, but also because deeper penetration depths can be achieved.

2.5. Holmium-doped luminescent thermometers operating in the I-BW

Luminescent temperature sensors based only on holmium (Ho³⁺) as an activator operating in the I-BW are very rare in the literature. The reason for that is because mainly Ho³⁺ is used together with Tm³⁺ for luminescence thermometry purposes exhibiting high relative thermal sensitivities in the I-BW (section 2.1.3) and in the III-BW (section 6.1).

The examples encountered involving Ho3+ as the main emitting ion incorporate Yb3+ as a sensitizer to enhance the red emission of Ho³⁺, lying at around 660 nm and attributed to the ${}^5F_5 \rightarrow {}^5I_8$ electronic transition of this ion. Monoclinic Ho³⁺,Yb³⁺:KLu(WO₄)₂ nanocrystals, ¹⁹⁴ and tetragonal Ba₂In₂O₅ particles, 195 have been investigated as luminescent thermometers operating in the I-BW. The red emissions, located at ~650 nm and ~660 nm, were generated after exciting these at 980 nm. By means of ground state absorption (GSA), Yb³⁺ absorbs the radiation from the excitation source, exciting its electrons from the ²F_{7/2} ground state to the ²F_{5/2} excited state. Via two ET processes, the electrons of Ho3+ are promoted to the 5I6 and 5S2, 5F4 energy levels. A nonradiative relaxation process from the ⁵S₂, ⁵F₄ energy levels populates the ⁵F₅ level, from which a radiative process generates the red emissions with two main peaks centered at ~650 nm and ~660 nm (Fig. 20(a)). A second path to explain the electronic population of the 5F5 energy level can also be postulated from a nonradiative process after the first ET transfer process from Yb³⁺,

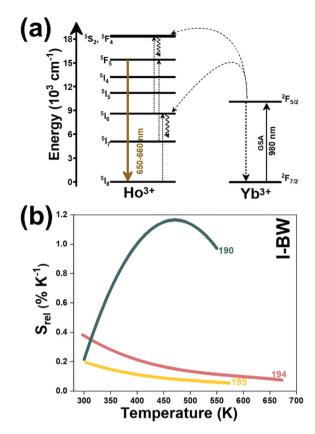


Fig. 20 (a) Mechanism of generation of the emission lines of $Ho^{3+}/$ Yb3+-doped luminescent thermometers. (b) Temperature dependence of S_{rel} of Ho³⁺-doped luminescent thermometers operating in the I-BW. The numbers represent the references for each Ho3+-based thermometer.

populating the ⁵I₇ energy level of the Ho³⁺ ion. The second ET from Yb3+ promotes these electrons directly to the 5F5 energy level of Ho³⁺.¹⁹⁴ These luminescent thermometers operate with emissions generated by TCLs. Nevertheless, for the case of Ho³⁺,Yb³⁺:KLu(WO₄)₂ nanocrystals, the authors take into consideration the overlapping factor for the emission peaks, such as the case of red emissions of Ho3+.194 Hence, in that case, eqn (14) was used to fit the experimental data of the thermometric parameter, whereas for the case of Ho³⁺,Yb³⁺: $Ba_2In_2O_5$ particles, the simple form of the FIR model (eqn (9)) was applied, which may lead to an underestimation of the performance of the luminescent thermometer. In fact, the S_{rel} of Ho³⁺,Yb³⁺:KLu(WO₄)₂ nanoparticles is approximately four times higher than that of Ho³⁺,Yb³⁺:Ba₂In₂O₅ particles (Table 1 and Fig. 20(b)). Nevertheless, the performance of Ho³⁺,Yb³⁺:KLu(WO₄)₂ nanoparticles is lower (~3 times) compared with that of Ho³⁺,Mn⁴⁺,Yb³⁺:YAP nanoparticles, ¹⁹⁰ (Fig. 20(b)). The red emission of Ho3+ was combined with the 714 nm emission arising from Mn⁴⁺, generated after NIR excitation ensured by the presence of Yb3+. However, in the physiological range of temperatures, the S_{rel} of this material does not differ much from that of Ho3+,Yb3+:KLu(WO4)2 nanoparticles.

3. Lanthanide-doped luminescent nanothermometers operating in the I-and II-BW simultaneously

The emissions of Ln³⁺ ions, upon proper excitation, could be located in a wide range of the electromagnetic spectrum. Besides lying in each of the four biological window regions, the emissions of Ln³⁺ ions used for thermometry can also be found in two different BWs. Hence, here, we focus our attention on the luminescent thermometers that use emissions which are placed in the I-BW and the II-BW. Although the number of publications that report luminescent thermometers working in this mixed region is reduced, the most representative examples are those of Nd³⁺-doped materials due to the emissions located at 850 nm and 1050 nm, covering simultaneously the I- and the II-BWs.

3.1. Nd^{3+} -doped luminescent thermometers operating in the I-BW and II-BW region simultaneously

Nd³⁺-doped materials operating in the I- and II-BWs simultaneously can be classified in single emitting Nd3+-doped luminescent thermometers, dual emitting center luminescent thermometers in combination with other Ln3+, and dual emitting center luminescent thermometers in combination with transition metals. Alkali earth fluorides such as CaF2, 220 and SrF₂, 163 have been studied as hosts for single emitting Nd³⁺ luminescent thermometers, due to their low phonon energies that reduce the probability of non-radiative relaxing processes happening, favoring in this way the luminescence efficiency.²²¹ They are also biocompatible, and nanoparticles of these materials with sizes below 10 nm can be produced, which makes them attractive for biomedical applications. 163,220 These water-dispersible nanocrystals were prepared by hydrothermal synthesis, 222 in which Gd3+ was introduced to enhance the emissions of Nd³⁺. 163,220 Upon excitation at 573 nm, these nanocrystals generate emissions at ~850 nm (I-BW) and $\sim\!\!1050$ nm (II-BW), attributed to the $^4F_{3/2}\to ^4I_{9/2}$ and $^4F_{3/2}\to ^4I_{11/2}$ transitions of Nd $^{3+}$, respectively. 163,220

Upon excitation at this wavelength, the ${}^4G_{5/2}$, ${}^4G_{7/2}$ energy levels of Nd $^{3+}$ are populated. Then, a non-radiative decay process takes place populating the ${}^4F_{3/2}$ state. From this energy level, a radiative decay to the ${}^4I_{11/2}$ level generates the emission located in the II-BW (Fig. 21(a)). Concerning the emission in the I-BW, from the lower lying Stark sublevels of the ${}^4F_{3/2}$ level (R₁ in Fig. 21(a)), a radiative decay to the lower lying Stark sublevel of the ${}^4I_{9/2}$ ground state (Z₁ in Fig. 21(a)) generates this emission

The thermometric parameter for these luminescent thermometers was a FIR between the 1050 nm and 850 nm emissions in the physiological range of temperatures, using eqn (9). The results, summarized in Table 2, show that the highest $S_{\rm rel}$ was obtained for Nd³⁺:SrF₂ nanoparticles with a value of 0.50% K⁻¹ and δT of 1.2 K at room temperature. ¹⁶³ The different thermometric performance of SrF₂ and CaF₂ can be due to the lower phonon energy of SrF₂ (366 cm⁻¹) com-

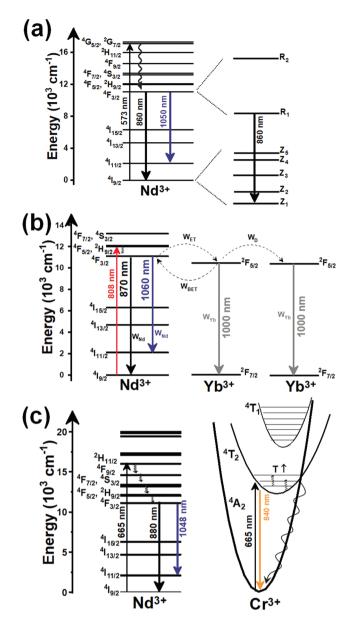


Fig. 21 Mechanisms of generation of the emission lines of: (a) single-doped Nd³⁺ luminescent thermometers excited at 573 nm, (b) dual emitting center Nd³⁺/Yb³⁺ luminescent thermometers excited at 808 nm, and (c) dual emitting center Nd³⁺/Cr³⁺ luminescent thermometers excited at 665 nm, operating simultaneously in the I- and II-BWs.

pared with CaF_2 (466 cm⁻¹).²³⁰ Thus, it seems that materials with low phonon energies maximize $S_{\rm rel}$. Nevertheless, it should be taken into consideration that the excitation wavelength used to activate these luminescent thermometers is not located in the BW spectral regions; instead it is located in the VIS, for which the absorption and scattering in the biological tissues is high, resulting in a limited penetration depth, not allowing for deep-tissue imaging.⁸⁶ Hence, this limitation should be surpassed by using other ions that are able to absorb excitation sources located within the BWs spectral regions.

cate with an asterisk $S_{
m cel}$ or δT values calculated by us using the parameters published in the corresponding references. The double line separation between rows stands for different types (single able 2 Ln³*-doped luminescent thermometers operating in the I and II-BWs. The table includes activators (A) and sensitizers (S). The excitation (J_{exc}) and emission (J_{exc}) and emission (J_{exc}) was investigated. The thermometric parameter (Δ) indicates the luminescent nanothermometry class used in each case: FIR for band-shape, / for intensity ratio, and τ for lifetime, respectively or dual emitting centers) of lanthanide-doped nanothermometers, as discussed in the corresponding subsections The maximum relative thermal sensitivity (S_{rei}) is given at the temperature at which this value was obtained.

	1	•								
A	S	Host	$\lambda_{ m exc} \left({ m nm} ight) = \lambda_{ m em} \left({ m nm} ight)$	$\lambda_{ m em} (m nm)$	Transitions	$\Delta T(K)$	Δ	S_{rel}/T (% K^{-1})/K δT (K) Ref.	$\delta T\left(\mathbf{K}\right)$	Ref.
Nd ³⁺	Nd ³⁺ Nd ³⁺	$\mathrm{Gd}^{3+}\mathrm{:SrF}_2$ $\mathrm{Gd}^{3+}\mathrm{:CaF}_2$	573 573	950, 1150 867, 1058	$ \begin{array}{l} {}^{4}F_{3/2} \rightarrow {}^{4}I_{9/2}, {}^{4}F_{3/2} \rightarrow {}^{4}I_{11/2} \\ {}^{4}F_{3/2} \rightarrow {}^{4}I_{9/2}, {}^{4}F_{3/2} \rightarrow {}^{4}I_{11/2} \end{array} $	293–338 294–338	${ m FIR}_{950/1150}$ ${ m FIR}_{1058/867}$	0.50/293 0.12/294	1.2	163 220
Nd ³⁺ , Yb ³⁺ Nd ³⁺ , Yb ³⁺ Nd ³⁺ , Yb ³⁺ Nd ³⁺ , Yb ³⁺	Nd^{3+} Nd^{3+} Nd^{3+}	Nd:LaF3@Yb:LaF3 Yb:LaF3@Nd:LaF3 LiLaP ₄ O ₁₂ Nd,Yb:LaF3	790 790 808 790	890 (Nd ³⁺), 1060 (Yb ³⁺) 890 (Nd ³⁺), 1060 (Yb ³⁺) 870 (Nd ³⁺), 1000 (Yb ³⁺) 890 (Nd ³⁺), 1060 (Yb ³⁺)	$ \begin{array}{l} {}^2F_{5/2} \rightarrow {}^2F_{7/2} \left(Yb^{3+}\right), \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \$	283–323 283–323 93–663 283–323	$I_{890}/I_{1060} \ I_{890}/I_{1060} \ I_{870}/I_{1000} \ I_{1300}/I_{1000}$	0.41/283 0.36/283 0.3/330 0.1/283	1.2 * 1.4 * 1.6 *	223 223 201 223
Nd^{3+} , Cr^{3+}	Nd^{3+}, Cr^{3+}	${\rm LiLaP_4O_{12}}$	999	810 (Cr^{3+}), 1048 (Nd^{3+})	$^{4}\mathrm{T}_{2} ightarrow ^{4}\mathrm{A}_{2} \left(\mathrm{Cr}^{3^{+}}\right), ^{4}\mathrm{F}_{3/2} ightarrow ^{4}\mathrm{I}_{11/2} \left(\mathrm{Nd}^{3^{+}}\right)$	293-323	I_{810}/I_{1048}	4.89/323	0.10*	224
${ m Er}^{3+}_{3}, { m Yb}^{3+}_{3+}_{{ m Yb}^{3+}_{3+}}_{{ m Yb}^{3+}_{3+}}_{{ m Yb}^{3+}_{3+}}$	${ m Yb}^{3+}_{3+}$ ${ m Nd}^{3+}_{3+}$ ${ m Yb}^{3+}_{3+}$	$NaYF_4@SiO_2$ $NaGdF_4$ $NaYF_4@Y@CaF_2$ $Bi_7F_{11}O_5$	975 808 800 808	810 ($\mathrm{Er^{3}}$), 1010 ($\mathrm{Yb^{3+}}$) 1012 980 1030	$\begin{array}{l} {}^{4}{}_{19/2} \rightarrow {}^{4}{}_{15/2} \left({\rm Er}^{3} \right), {}^{2}{}_{F_{5/2}} \rightarrow {}^{2}{}_{F_{7/2}} \left({\rm Yb}^{3+} \right) \\ {}^{2}{}_{F_{5/2}} \rightarrow {}^{2}{}_{F_{7/2}} \\ {}^{2}{}_{F_{5/2}} \rightarrow {}^{2}{}_{F_{7/2}} \\ {}^{2}{}_{F_{5/2}} \rightarrow {}^{2}{}_{F_{7/2}} \end{array}$	299–337 303–343 283–337 323–573	I_{1010}/I_{810} τ_{1012} τ_{980} τ_{1012}	1.64/337 1.59/343 1.4/283 0.24/323	0.76 0.31* 0.36* 2.1*	225 226 227 228
yb^{3+}, Cr^{3+} Yb^{3+}, Cr^{3+}	yb^{3+}, Cr^{3+} yb^{3+}, Cr^{3+}	$LiLaP_4O_{12}\\ LiLaP_4O_{12}$	665 + 920 665	820 ($\mathrm{Cr^{3+}}$), 975 ($\mathrm{Yb^{3+}}$) 820 ($\mathrm{Cr^{3+}}$), 975 ($\mathrm{Yb^{3+}}$)	$^{4}T_{2} \rightarrow ^{4}A_{2} \left(Cr^{3+} \right), ^{2}F_{5/2} \rightarrow ^{2}F_{7/2} \left(Yb^{3+} \right)$ $^{4}T_{2} \rightarrow ^{4}A_{2} \left(Cr^{3+} \right), ^{2}F_{5/2} \rightarrow ^{2}F_{7/2} \left(Yb^{3+} \right)$	100–475 100–475	$I_{820}/I_{975} \ I_{820}/I_{975}$	1.2/333 0.32/333	0.42* 1.6	229 229

An interesting alternative, was proposed by Marciniak et al. through the preparation of dual emitting center luminescent thermometer-based Nd3+,Yb3+:LiLaP4O12 nanoparticles, in which, under 808 nm NIR excitation, Nd3+ acts both as a sensitizer and an activator, while Yb3+ acts only as an activator.201 The reason for this selection is because the Nd³⁺ absorption cross section at 800 nm is higher than that of Yb3+ at 980 nm, and the water absorption at 800 nm is around 25 times lower than that at 980 nm, overcoming in this way also the overheating problems displayed by the 980 nm excitation.201 Upon 808 nm excitation, Nd3+ absorbs this energy and excites its electrons from the ⁴I_{9/2} ground state to the ⁴F_{5/2} excited state. Via a non-radiative process, the ${}^{4}F_{3/2}$ state is populated. From this state, two radiative processes towards the ⁴I_{9/2} and ⁴I_{11/2} levels generate the Nd3+ emissions at 870 nm and 1060 nm, respectively (Fig. 21(b)). From the 4F_{3/2} state of Nd3+, an ET process with $W_{\rm ET}$ probability can populate the ${}^2F_{5/2}$ state of Yb³⁺. A radiative decay from this state, back to the ²F_{7/2} ground state, allows for the generation of the 1000 nm emission band of Yb³⁺. ²⁰¹ The thermometric parameter of these nanocrystals was defined as the intensity ratio between the emissions at 870 nm (Nd³⁺) and 1000 nm (Yb³⁺) as function of the temperature over the range 93-663 K, and as a function of the concentration of Yb³⁺ ions (Fig. 21(a)). The thermometric performance of these nanocrystals is influenced by three processes: (i) with the increase of temperature, the electronic population of the Stark sublevels of the ²F_{5/2} manifold of Yb³⁺ increases gradually, leading to a BET process, with probability W_{BET} , towards the ⁴F_{3/2} state of Nd³⁺ (Fig. 21(b)); (ii) by increasing the concentration of Yb3+, the average distance between these ions decreases, and consequently, the energy diffusion between Nd^{3+} and Yb^{3+} ions, with probability W_D , and among Yb^{3+} ions (W^{Yb}) increases; and (iii) the energy difference between the ${}^4F_{3/2}$ state of Nd³⁺ and the ${}^2F_{5/2}$ level of Yb³⁺, that controls the ET processes, with probability $W_{\rm ET}$, is affected by the changes of temperature and the concentration of Yb3+.201 Hence, the intensity ratio between these two nonresonant energy matching ions, extracted from the phonon-assisted model of ET described by Miyakawa and Dexter,231 was correlated to these processes by the following equation:201

$$\Delta = \frac{W_{\text{BET}} - W_{\text{D}} - W^{\text{Yb}}}{W_{\text{ET}}} \tag{41}$$

From eqn (41) it can be deduced that as the doping concentration of Yb³⁺ increases, the effectiveness of the energy diffusion among the Yb³⁺ network gradually increases, while the impact of the BET process towards Nd³⁺ decreases, leading to changes in Δ . The highest $S_{\rm rel}$ was obtained for the highest concentration of Yb³⁺ (Fig. 22(a)), probably because the decrease in the average distance among the Nd³⁺-Yb³⁺ ions facilitated energy diffusion.²⁰¹ The $S_{\rm rel}$ of these luminescence thermometers is, however, very low, with a maximum of 0.3% K⁻¹ in the physiological range of temperatures (the full variation of the $S_{\rm rel}$ in the interval of temperatures analyzed is presented in Fig. 22(b)).

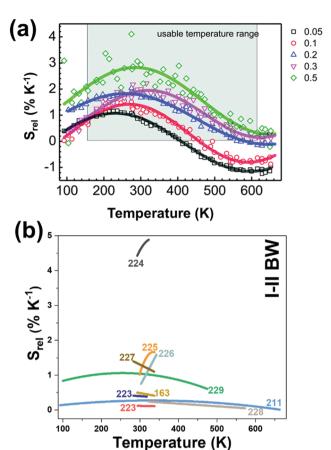


Fig. 22 (a) $S_{\rm rel}$ of Nd³⁺- and Yb³⁺-doped LiLaP₄O₁₂ as a function of Yb³⁺ concentration. Reprinted with permission from ref. 201. Copyright 2015, the Owner Societies. (b) Temperature dependence of $S_{\rm rel}$ of lanthanide-doped luminescent thermometers operating in the I-BW and II-BWs, simultaneously. The numbers indicate the corresponding references for each thermometer.

The same strategy was followed for the case of two different active core@active shell nanostructures, one having the core doped with Nd3+ and the shell with Yb3+ (Nd3+:LaF3@Yb3+: LaF₃), and the other constituted by a Yb³⁺-doped core and a Nd³⁺-doped shell (Yb³⁺:LaF₃@Nd³⁺:LaF₃).²²³ The performance of these two different core@shell structures was also compared with the corresponding nanoparticles containing the two ions in the same layer, revealing a four-fold enhancement of S_{rel} . In addition, the core@shell nanocrystals displayed an enhanced emission intensity, attributed to the reduction of intensity when the doping ions are located in the same layer due to selfquenching effects originating from non-radiative mechanisms such as cross-relaxation, energy migration, and energy-trap processes occurring between the emitting ions and the OHradicals present at the surface of the nanoparticles.²²³ These types of material, however, are more interesting for the emissions generated by Nd³⁺ in the II-BW (1060 nm and 1350 nm); hence they will be mentioned again in section 4.2.

When combining the emissions arising from Nd³⁺ and those of a transition metal, such as Cr³⁺, a substantial increase

on $S_{\rm rel}$ is observed (Table 2). In that case, the emission of Nd³⁺ located at ~1050 nm in the II-BW was combined with the emission at 810 nm of ${\rm Cr}^{3+}$ in the I-BW, both ions being embedded in LiLaP₄O₁₂.²²⁴ These emission bands were generated after exciting the nanoparticles at 655 nm, in the lower limit of the I-BW that allows the excitation of ${\rm Cr}^{3+}$ and ${\rm Nd}^{3+}$ at the same time. To describe the generation of the emission band of ${\rm Cr}^{3+}$, a low crystal field assumption for the low-symmetry host used was applied by the authors. Electrons of ${\rm Cr}^{3+}$ are excited after illumination with the 655 nm light from the ${}^4{\rm A}_2$ fundamental state to the ${}^4{\rm T}_2$ excited state (Fig. 21(c)).

When relaxing back to the 4A_2 state, the broad emission line centered at \sim 810 nm is generated. When the temperature increases, the electronic population of the higher vibrational levels of the 4T_2 state also increases, and the ${\rm Cr}^{3^+}$ emission can be quenched via non-radiative processes as soon as the thermally excited electrons reach the crossing point between the 4T_2 and 4A_1 energy level parabolas (Fig. 21(c)). 232 At the same time, the electrons of ${\rm Nd}^{3^+}$ are pumped from the $^4I_{9/2}$ ground state to the $^2H_{11/2}$ excited state. Then, a non-radiative relaxation to the $^4F_{3/2}$ state happens, from which the 880 nm ($^4F_{3/2}$ \rightarrow $^4F_{9/2}$) and the \sim 1050 nm ($^4F_{3/2}$ \rightarrow $^4I_{11/2}$) emissions are produced. 224 The intensity of the emission of ${\rm Nd}^{3^+}$ is barely influenced by temperature, so it can be used as a reference probe. A maximum $S_{\rm rel}$ of 4.89% K $^{-1}$ at 323 K (Fig. 22(b) was obtained in this case.

In Nd³⁺,Cr³⁺-doped LiLaP₄O₁₂ nanoparticles, as the size of the nanocrystals decreases, the intensity of the emissions decreases (\sim 0.81% nm⁻¹, Fig. 23(a)), the emission band position blue-shifts (\sim 0.066 nm grain size⁻¹), the temperature-sensing region narrows (from 300 to 600 K for particles with a size of 240 nm to 300–420 K for particles with a size of 20 nm), $S_{\rm rel}$ increases from 1% K⁻¹ to 5% K⁻¹ (Fig. 23(b)), and δT drops down to 0.03 K (Fig. 23(c)). ²³² The reason for that is that with

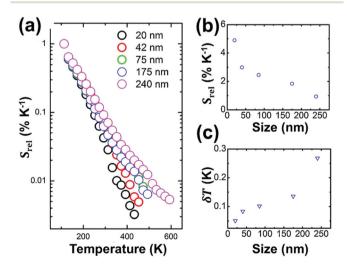


Fig. 23 The effect of the size of Nd^{3+} , Cr^{3+} : LiLaP₄O₁₂ nanoparticles on: (a) the intensity ratio between the emissions lines of Cr^{3+} and Nd^{3+} at 810 nm and 1048 nm, respectively, (b) S_{rel} , and (c) δT . Reprinted with permission from ref. 232. Copyright 2016, Elsevier.

the decrease of the size of the nanoparticles, non-radiative depopulation processes become more important, being the main processes responsible for the decreasing of the intensity of the emissions. Furthermore, this leads to a faster decrease in the intensity ratio used in the luminescent thermometer, which allows a higher $S_{\rm rel}$ and a lower δT to be obtained. ²³²

3.2. Yb³⁺-doped luminescent thermometers operating in the I- and II-BWs simultaneously

Another example of luminescent thermometers operating simultaneously in the I- and II-BWs is that of Yb^{3+} -doped materials, based on the ~1000 nm emission of this ion, combined with other emissions arising from other Ln^{3+} or transition metal ions.

Only a few publications explored the case of dual emitting center luminescent thermometers consisting of Yb³⁺ with another Ln³⁺ (such as Er³⁺) or a transition metal ion (such as Cr³⁺). For the Yb³⁺/Er³⁺-coemitting material, the host was an active core of hexagonal NaYF₄ nanorods coated with an amorphous silica shell, synthesized by a hydrothermal method.²²⁵ Under excitation at 975 nm, Yb³⁺ acts as a sensitizer, besides the role of activator, absorbing the energy of the excitation source and transferring this energy to Er³⁺ (Fig. 24(a)). By these ET processes, the $^4F_{7/2}$ state of Er³⁺ can be populated, from which a non-radiative decay can take place that populates

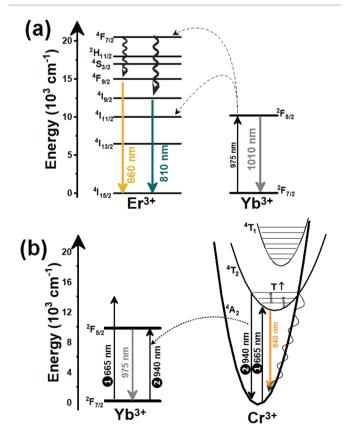


Fig. 24 Mechanisms of generation of the emission lines of dual-doped (a) ${\rm Er}^{3+}/{\rm Yb}^{3+}$ and (b) ${\rm Cr}^{3+}/{\rm Yb}^{3+}$ materials operating simultaneously in the I- and II-BWs.

the $^4F_{9/2}$ and $^4I_{9/2}$ energy levels of Er³⁺. From there, by relaxing back to the $^4I_{15/2}$ ground state, the red emission at 660 nm and the NIR emission located at 810 nm are generated, respectively. The 1010 nm emission of Yb³⁺ is generated by simply relaxing back the electrons to the $^2F_{7/2}$ ground state from the $^2F_{5/2}$ excited energy level. The thermometric performance of this material was extracted by studying the temperature dependence of the 1010 nm/810 nm and 1010 nm/660 nm intensity ratios, both modelled according to a second-order polynomial fitting equation, as expected for a luminescent thermometer based on NTCLs (eqn (27)). The results reveal that the 1010 nm/810 nm intensity ratio allows a $S_{\rm rel}$ two times higher than the 1010 nm/660 nm intensity ratio to be obtained. The maximum $S_{\rm rel}$ was 1.64% K⁻¹ and the minimum δT was 0.76 K, both obtained at 337 K.

In the case of Yb3+,Cr3+:LiLaP4O12,229 the intensity ratio between the emissions of Cr3+ located at 840 nm and Yb3+ located at 975 nm did not allow the obtaining of results as promising as those obtained for the combination of Nd³⁺/Cr³⁺. To generate these emission bands, a first alternative is excitation at 665 nm that allows the excitation of electrons of Cr³⁺ from the ⁴A₂ ground state to the ⁴T₂ excited state. A radiative relaxation from this level led to a broad emission band, centered at \sim 840 nm, assigned to the $^4T_2 \rightarrow ^4A_2$ transition, as explained before. Also, after thermalization of the higher energy levels of the 4T2 excited state, an emission at 940 nm can take place, when relaxing back radiatively to the ground state. For the excitation of Yb3+, the authors considered an ET process from Cr³⁺ to Yb³⁺ or a reabsorption process of the 940 nm emission band generated by Cr3+, that allowed excitation of the Yb³⁺ electrons to the ²F_{5/2} level (Fig. 24(b)), from where the radiative relaxation to the ground state generated the emission line at 975 nm used in the luminescent thermometer.

Here, the maximum S_{rel} obtained in the physiological range of temperatures was 0.32% K⁻¹, with a δT of 1.6 K.²²⁹ The authors tried to correlate the differences between the sensitivity of Yb³⁺/Cr³⁺ and Nd³⁺/Cr³⁺ luminescent thermometers to the population mechanism of the excited states in these ions. For the Nd³⁺/Cr³⁺ combination, the 655 nm excitation wavelength was chosen to excite simultaneously both ions.²²⁴ For the Yb³⁺/Cr³⁺ combination, the simple energy diagram of Yb³⁺ and the lack of high energy levels hampers the simultaneous excitation of both ions, which concurrently reduces the sensitivity of such luminescent thermometers. 229 A possible solution for the dually doped Yb³⁺/Cr³⁺ materials would be the use of two excitation beams simultaneously, one for Cr3+ (at 650 nm) and another one for Yb³⁺ (~940 nm). In terms of thermometric performance, this last approach achieved a three times higher S_{rel} (Fig. 25), with a value of 1.2% K^{-1} . ²²⁹ However, this approach is technically rather complex, and exhibits strong sensitivity fluctuations above 320 K (Fig. 25). 229

Recently, lifetime nanothermometry^{226–228} has been applied for materials operating within the I and II BWs simultaneously. Tan *et al.* studied the temperature dependence of the 980 nm emission of Yb³⁺ in a NaYF₄@NaYF₄:Nd³⁺,

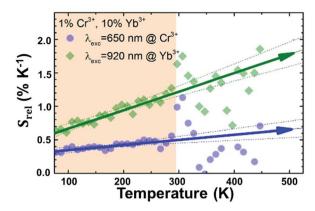


Fig. 25 S_{rel} of codoped Cr³⁺,Yb³⁺:LiLaP₄O₁₂ nanoparticles operating simultaneously in the I- and II-BWs, as a function of the excitation wavelength. Reprinted with permission from ref. 229. Copyright 2016, Elsevier.

Yb3+@CaF2 core-shell nanostructure with a size of 13.5 nm.227 This nanostructure does not only minimize the probability of deactivation processes happening, produced by crystal defects and surface quenching centers, but also maximizes the thermal sensitivity of the luminescent nanothermometer. The emission band of Yb3+ is generated after exciting the nanoparticles at 800 nm. Nd3+ ions absorb this energy to promote their electrons from the 4I9/2 ground state to the 4F5/2 excited state. From here, a non-radiative relaxation leads to the population of the ⁴F_{3/2} level. From this level, which is resonant in energy with the 2F_{5/2} level of Yb3+, ET and BET processes occur. ET processes populate the ${}^2F_{5/2}$ level of Yb^{3+} , from which, upon relaxation back to the ²F_{7/2} ground state, the 980 nm emission band is generated (as has been described in Fig. 21(b)).

The lifetime of this emission was recorded from 283 K to 337 K, after optimizing the concentration of the Ln³⁺ ions in these structures. This temperature dependence was governed by the temperature dependence of the ET and BET rates, the multiphonon process probabilities, and the energy migration that can take place among Yb3+ ions. The authors modelled the lifetime according to a polynomial function (as that presented in eqn (23)) and S_{rel} was expressed as follows:²²⁷

$$S_{\rm rel} = \left| \frac{1}{\tau} \frac{\mathrm{d}\tau}{\mathrm{d}T} \right| \times 100\% \tag{42}$$

The maximum $S_{\rm rel}$ was obtained at the lowest temperature under investigation with a value of 1.4% K⁻¹.

Similarly, Wang et al. 228 and Ji et al. 226 studied the temperature dependence of the lifetime of this emission embedded in Bi₇F₁₁O₅ and Nd³⁺:NaGdF₄ (see Table 2), respectively. Here the chosen fitting model was an exponential equation (as that presented in eqn (19)). Among these materials, Yb3+,Nd3+:NaGdF4 displayed the highest $S_{\rm rel}$ with a value of 1.59% K^{-1} at 343 K. The variation of S_{rel} with temperature is presented in Fig. 22(b). Lifetime nanothermometry, although not too much used within the BW regions, portrays a potential and very

powerful route towards biomedical applications of Ln³⁺-doped luminescent nanothermometers, as it is independent of the optical properties of the medium in which the nanoparticles are embedded.²²⁷ However, this technique requires the use of complex and expensive acquisition systems based on either fast detectors or time-gated detection procedures, which might hamper their implementation at a practical level.

Lanthanide-doped luminescent nanothermometers operating in the II-BW

Lanthanide-doped luminescent thermometers operating in the II-BW are based on Yb³⁺ and Nd³⁺-doped materials. The explanation for this relies on the ability of these two Ln³⁺ ions to absorb the energy of an excitation source located in the I-BW, and generate emissions in the II-BW, 48,90,233 providing deeper penetration depths into biological tissues, while maintaining a high spatial resolution. 86,90,233 The luminescent thermometers based on Yb3+ use the emission at ~1000 nm, either in singledoped materials by studying the Stark sublevels of the electronic transition that generate this emission, or combined with other emissions arising from other Ln3+, including Tm3+ and Nd³⁺. Luminescent thermometers based on Nd³⁺-doped materials used the emissions arising from the Stark sublevels of the 1050 nm and 1330 nm emission bands in single-doped materials, or the combination with emissions from other Ln³⁺ (Ho³⁺) in codoped materials, or from semiconductor quantum dots in dual emitting materials.

4.1. Yb³⁺-doped luminescent nanothermometers operating in the II-BW

4.1.1. Single Yb³⁺-doped luminescent thermometers operating in the II-BW. The number of reports based on single Yb³⁺-doped materials operating as luminescent thermometers in this spectral region is scarce, and their thermometric performance is relatively poor. For instance, a hybrid compound composed of mesostructured dipyridyl-pyridazine functionalized either with ethenylene bridged mesoporous organosilica (dppz-ePMO) or vinyl silica (vSilica) were both complexed with ytterbium(III) on 2-thenoyltrifluoroacetonate (Yb(tta)₃).²³⁴ The different peaks in the manifold corresponding to electronic transitions between the different Stark sublevels of the ${}^2F_{5/2} \rightarrow$ ²F_{7/2} transition of Yb³⁺ in dppz-ePMO@Yb(tta)₃ and dppzvSilica@ Yb(tta)3 particles, located at 990 nm and 1025 nm or 986 nm and 1015 nm, respectively, were used to determine the temperature through the FIR method. These emission lines have been generated in these materials via excitation with ultraviolet light. In this way, the ⁴f₁₃ electronic configuration of Yb³⁺ can easily gain one electron to reach the more stabilized ⁴f₁₄ configuration of the full shell, corresponding to Yb²⁺. This tendency of reduction to Yb2+ enables the Yb3+ ion in some hosts, such as is the case with these organic complexes, to receive an electron from the host's anion under high-energy

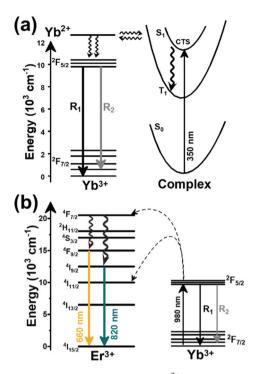


Fig. 26 Mechanisms of generation of Yb^{3+} emissions lines in: (a) organic complexes *via* charge transfer state (CTS) and (b) ZrO_2 codoped with Er^{3+} .

external excitation, such as UV light, forming a charge transfer state (CTS) (Fig. 26(a)). The CTS of Yb $^{3+}$ can transfer the excitation energy to the $^2F_{5/2}$ -emitting state via a non-radiative relaxation process, generating the emission bands located at around 980–990 nm (labelled as R_1) and 1015–1025 nm (labelled as R_2).

Nevertheless, for these hybrids, besides the low $S_{\rm rel}$ they exhibit (0.13–0.17% K⁻¹) with maximum values at cryogenic temperatures (110 K)²³⁴ (Table 3), they are excited with UV light, hampering significantly their implementation in possible biomedical applications.

Another choice to excite single Yb3+-doped luminescent thermometers operating in the II-BW is using NIR light at 980 nm. This approach has been investigated in ZrO2 nanocrystals codoped with Er³⁺, in which the authors investigated the performance of the thermometer in the I-, II- and III-BWs, using the different emissions of Er3+ and Yb3+.193 For the luminescent thermometer operating in the II-BW the linear model of VPR (eqn (39)) was applied between the different peaks in the manifold corresponding to the ${}^2F_{5/2} \rightarrow {}^2F_{7/2}$ electronic transition of Yb3+, with peaks located at 1023 nm (label R1 of Fig. 26(b)) and 1036 nm (label R2 of Fig. 26(b)). 193 The mechanism for the generation of these emission lines is similar to that reported in Fig. 24(a), in which Yb3+ not only acts as a sensitizer, but also as an activator. These nanocrystals exhibited a better S_{rel} with a value of 1.1% K⁻¹ at room temperature, making this approach a better strategy towards the development of single Yb³⁺-doped luminescent thermometers operating in the II-BW. Nevertheless, as pointed out before, the strong water absorption band at around 980 nm makes Yb³⁺-based luminescent nanothermometers pumped at this wavelength not ideal for biomedical or biological applications, as it might generate strong heating in the biological tissues when excessive pumping power densities are applied, which might damage biological tissues.²⁰⁶

4.1.2. Dual-doped Yb3+/Ln3+ luminescent thermometers operating in the II-BW. Dual-doped Yb3+/Ln3+ luminescent thermometers operating in II-BW are often based on the combination of the 1000 nm emission of Yb3+ with either the emission of Tm3+ located at 1230 nm or emissions of Nd3+ located at 1050 nm and 1300 nm. Here, generally, we underline the differences among incorporating a simple core, a core@shell, or a multishell structure in the nanoparticles used for luminescence nanothermometry (Table 3). In Table 3, as well, it can be observed that Yb³⁺ ions are not performing any more the role of sensitizers, as often reported in the previous sections, but instead they are acting only as activators. This approach uses excitation wavelengths ranging from 690-808 nm, matching the strong absorptions of Tm³⁺ and Nd³⁺ ions, ^{204,235} and may lead to a promising strategy to ensure biomedical applications, avoiding the heating problems generated by the 980 nm laser, as pointed out before, 206 since the absorption of water molecules, especially at around 800 nm, is quite weak.

If we fix our attention on the structure of the nanoparticles, besides their colloidal stability and target abilities in biomedical applications, they often exhibit limitations related to the drastic reduction of the luminescence efficiency due to the increase of non-radiative processes caused by the interactions with other molecules at their surfaces. 236 It is also difficult in a simple particle to implement multifunctionality. 237 To overcome these limitations, core@shell or multishell nanoparticles have emerged as a very promising strategy. The core@shell or multishell structure allows for flexible designs, incorporating efficient multifunctionality, and the facile incorporation of dopants with the desired spatial distribution, allowing the manipulation of ET processes among different ions located in different layers. 238,239 Hence, here we will highlight the advantages offered by core@shell nanocrystalline structures when compared with bare nanocrystals in luminescence thermometry.

First, we focus on luminescent nanothermometers based on the Yb³⁺ and Tm³⁺ emissions located in the II-BW, which display higher relative thermal sensitivities than Yb³⁺/Nd³⁺ thermometers (Table 3). A typical example that explores the differences between the core@shell and the simple particles is the case of Er³⁺,Yb³⁺:LaF₃@Tm³⁺,Yb³⁺:LaF₃ nanocrystals with sizes ~32 nm, synthesized *via* wet chemistry methodologies. ^{223,240} These active core@active shell nanocrystalline structures were compared, in terms of the intensity of the emissions generated after excitation at either 690 nm or 808 nm and in terms of the thermometric performance, with their corresponding simple cores. Upon excitation at 690 nm, the energy is absorbed by Tm³⁺, promoting its electrons from the ground state to the ³F_{2,3} excited state. From this state, a

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Table 3 Ln³⁺-doped luminescent thermometers operating in the II-BW. The table includes activators (A) and sensitizers (S). The excitation (λ_{exc}) and emission (λ_{em}) wavelengths are indicated in nanometers (nm), together with the corresponding electronic transition with which the emissions are associated. ΔT stands for the temperature range where the temperature reading was investination with which the emissions are associated.

Yb ³⁺ Yb ³⁺ Yb ³⁺ Yb ³⁺ Yb ³⁺ Tm ³⁺ Tm ³⁺ Tm ³⁺ Tm ³⁺ Yb ³⁺ Nd ³⁺ Nd ³⁺	:a).3 b:LaF.3 vd:NaYF.4 ?_2@, E::S:F2_@Nd:S:F2_2	1) \(\triangle \text{cm} \left(\text{IIII} \right) \) \(\triangle \text{cm} \right) \) \(\triangle \text{cm} \right) \)	Iransiuons	A TT (TZ)	(0/ 1/-1/17	J-G (A) L3
Yb ³⁺ Yb ³⁺ Yb ³⁺ Yb ³⁺ Yb ³⁺ Tm ³⁺ Tm ³⁺ Nd ³⁺ Nd ³⁺ Nd ³⁺ Nd ³⁺ Nd ³⁺ Nd ³⁺	$ ext{F}_3$ 4aYF $_4$. Er:Sr $ ext{F}_2$ @Nd:Sr $ ext{F}_2$		c			
Yb ³⁺ Tm ³⁺ Tm ³⁺ Tm ³⁺ Nd ³⁺ Nd ³⁺ Nd ³⁺ Nd ³⁺ Nd ³⁺ Nd ³⁺	$ ext{F}_3$ $ ext{ARYF}_4$ $ ext{Er:SrF}_2 ext{@Nd:SrF}_2$		$^{2}F_{5/2} \rightarrow ^{2}F_{7/2}$ $^{2}F_{5/2} \rightarrow ^{2}F_{7/2}$		9	*
Tm ³⁺ Tm ³⁺ Tm ³⁺ Nd ³⁺		990, 1025	$^2\mathrm{F}_{5/2} ightarrow ^2\mathrm{F}_{7/2}$	110-310 FIR _{990/1025}	0.13/110	3.8* 234
Tm., Nd., Nd., Nd., Nd., Nd., Nd., Nd., Nd			$^{2}F_{5/2} \rightarrow ^{2}F_{7/2} (Yb^{3+}), ^{3}H_{5} \rightarrow ^{3}H_{6} (Tm^{3+})$			
NG 3+ NG 3+ NG 3+ NG 3+ NG 3+ NG 3+ NG 3+		1000 (Yb ²⁺), 1230 (Tm ²⁺)	${}^{2}F_{5/2} \rightarrow {}^{2}F_{7/2} \text{ (Yb}^{2+}), {}^{3}H_{5} \rightarrow {}^{3}H_{6} \text{ (Tm}^{2+})$	$293-323 I_{1000}/I_{1230}$	30 1.3/293	0.38* 240
NG3+		$980 (Yb^{3+}), 1060 (Nd^{3+})$	$^{2}F_{5/2} \rightarrow ^{2}F_{7/2} (Yb^{3+}), ^{4}F_{3/2} \rightarrow ^{4}I_{9/2} (Nd^{3+})$	$103-443$ I_{1060}/I_{980}		0.24* 241
NG3+ NG3+ NG3+ NG3+ NG3+		$980 (Yb^{3+}), 1060 (Nd^{3+})$	$^{2}F_{5/2} \rightarrow ^{2}F_{7/2} (Yb^{3+}), ^{4}F_{3/2} \rightarrow ^{4}I_{11/2} (Nd^{3+})$	293–323		
Nd ³⁺ Nd ³⁺ Nd ³⁺		$1005 (Yb^{3+}), 1065 (Nd^{3+})$ $1005 (Yb^{3+}), 1052 (Nd^{3+})$	$ \begin{array}{c} -F_{5/2} \to {}^{-}F_{7/2} \left({\rm YB}^{-} \right), \ \ F_{3/2} \to {}^{-}I_{11/2} \left({\rm Nd}^{-} \right) \\ -F_{5/2} \to {}^{-}F_{7/2} \left({\rm YB}^{3+} \right), \ \ F_{2/2} \to {}^{4}I_{14/2} \left({\rm Nd}^{3+} \right) \end{array} $	$298-368 I_{1065}/I_{1005}$ $298-368 I_{1052}/I_{1005}$	0.53/368	1.09 243 0.08 243
Nd ³⁺ Nd ³⁺ Nd ³⁺	${ m Yb:LaF_3}$ 790	1000 (Yb ³⁺), 1300 (Nd ³⁺)	${}^{2}F_{5/2} \rightarrow {}^{2}F_{7/2} (Yb^{3+}), {}^{4}F_{3/2} \rightarrow {}^{4}I_{13/2} (Nd^{3+})$	283-323		
Nd3+ Nd3+ Nd3+		$1000 ({\rm Yb}^{3+}), 1060 ({\rm Nd}^{3+})$	$(2F_{5/2} \rightarrow (2F_{7/2} (Yb^{3+}), (4F_{3/2} \rightarrow (1_{11/2} (Nd^{3+}))))$	283-323		
Nd ³⁺		$1000 \text{ (Yb}^{3+}), 890 \text{ (Nd}^{3+})$	${}^2F_{5/2} ightarrow {}^2F_{7/2} \left({ m Yb}^{3+} ight), {}^4F_{3/2} ightarrow {}^4I_{9/2} \left({ m Nd}^{3+} ight)$	283-323		
דאת	Nd:LaF ₃ 790	$1000 \text{ (Yb}^{3+}), 1300 \text{ (Nd}^{3+})$	${}^{2}F_{5/2} \rightarrow {}^{2}F_{7/2} ({ m YB}^{2}), {}^{2}F_{3/2} \rightarrow {}^{3}I_{13/2} ({ m Nd}^{2})$	$283-323 I_{1300}/I_{1000}$	0.36/283	1.61 223
Nd3+		$1000 \text{ (Yb}^{3+}), 1000 \text{ (Nd}^{3+})$	$^{2}F_{\Xi',2} \rightarrow ^{2}F_{\Xi',2} (Yb^{3+}), ^{2}F_{\Xi',2} \rightarrow ^{4}I_{\alpha',2} (Nd^{3+}).$		0	
Nd3+	0	1000 (Yb ³⁺), 890 (Nd ³⁺)	$^{2}F_{5/2} \rightarrow ^{2}F_{7/2} (Yb^{3+}), ^{4}F_{3/2} \rightarrow ^{4}I_{9/2} (Nd^{3+})$			
Nd ³⁺			, ⁴ F _{3/2} →			
Nd ³⁺ , Yb ³⁺ Nd ³⁺ Yb,Nd:LaF ₃	3.3 790		${}^{2}F_{5/2} \rightarrow {}^{2}F_{7/2} \left(Yb^{3+} \right), {}^{4}F_{3/2} \rightarrow {}^{4}I_{13/2} \left(Nd^{3+} \right)$	$283-323 I_{1300}/I_{1000}$	0.1/283	1.61 223
	532	1348	$^4\mathrm{F}_{3/2} ightarrow ^4\mathrm{I}_{13/2}$	293-370 A ₁₃₄₈	3.3/293	0.37 157
		1316,	$^4\mathrm{F}_{3/2} ightarrow ^4\mathrm{I}_{13/2}$			
Nd ³⁺		1045,	$^4\mathrm{F}_{3/2} o ^4\mathrm{I}_{11/2}$			
Nd^{3+}		1064,	$^4\mathrm{F}_{3/2} ightharpoonup ^4\mathrm{I}_{11/2}$			
rcbN 2	808	1054,	$^{4}\mathrm{F}_{3/2} ightarrow ^{4}\mathrm{I}_{11/2}$			
Nd ³⁺	532	1053,	$^{*}F_{3/2} \rightarrow ^{*}I_{11/2}$			
Nd ³⁺	808	1064,	${}^{+}F_{3/2} \rightarrow {}^{+}I_{11/2}$			
Nd.			${}^{+}\mathrm{F}_{3/2} ightarrow {}^{+}\mathrm{I}_{11/2}$			2.7* 245
Nd^{3} Nd^{3} $KGd(WO_4)_2$	12 808	1067, 1075	$^{*}\mathrm{F}_{3/2} \rightarrow ^{*}\mathrm{I}_{11/2}$	298-333 FIR _{1067/1075}	0.16/298	* *

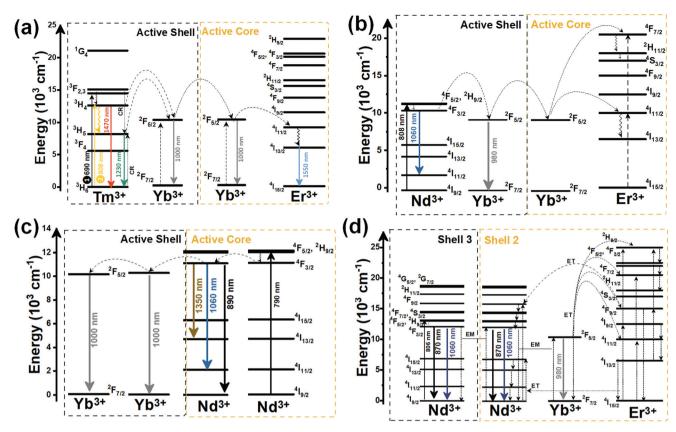


Fig. 27 (a) Mechanisms of generation of the emission lines of core@shell nanostructures:(a) Er^{3+} , Yb^{3+} : LaF_3 @ Tm^{3+} , Yb^{3+} : LaF_3 nanocrystals excited either at 690 nm or 808 nm, (b) Er^{3+} , Yb^{3+} : $NaYF_4$ @ Nd^{3+} , Yb^{3+} : $NaYF_4$ nanoparticles excited at 808 nm, (c) Nd^{3+} : LaF_3 @ Yb^{3+} : LaF_3 nanoparticles excited at 790 nm, and (d) Tm^{3+} , Yb^{3+} : SrF_2 @ Y^{3+} : SrF_2 @ Er^{3+} , Nd^{3+} , Yb^{3+} : SrF_2 @ R^{3+} , R^{3+} 0 nanocrystals excited at 806 nm.

cross relaxation process ${}^{3}F_{2,3}$; ${}^{3}H_{6} \rightarrow {}^{3}F_{4}$; ${}^{3}H_{5}$ (Fig. 27(a)) takes place leading to the population of the ³H₅ energy level, from which a radiative decay to the ground state generates the 1230 nm emission of Tm³⁺. From the ³H₅ energy level and from the ³H₄ energy level (populated by a non-radiative decay from the ³F_{2,3} state), ET processes to the energy-resonant ²F_{5/2} state of Yb3+ takes place, followed by a radiative relaxation to the ground state, which leads to the 1000 nm emission of Yb³⁺. Further ET processes between the active shell and the active core, as presented in Fig. 27(a), lead to the population of the ²F_{5/2} and ⁴I_{11/2} energy levels of Yb³⁺ and Er³⁺, respectively, present in the active core. The electrons in the 4I11/2 state of Er3+ can non-radiatively decay to the 4I13/2 state, prior to relaxing back radiatively to the ground state, leading to the generation of the 1550 nm emission. The intensity of the 1000 nm, 1230 nm and 1550 nm emissions of Yb3+, Tm3+ and Er3+ ions in this core-shell structure were compared with the ones obtained in the Tm3+, Er3+, Yb3+ nanocrystals in which the three ions are located in the same layer.

Under excitation at 690 nm, brighter emissions are obtained in the core@shell structures (Fig. 28(a)). This big difference between the intensity of the emissions is a direct effect of the core@shell architecture, according to the authors, leading to a spatial separation between Er³⁺ and Tm³⁺ ions in

such a way that the $Er^{3+} \rightarrow Tm^{3+}$ ET process, involving the ⁴I_{13/2} and ³F₄ energy levels of Er³⁺ and Tm³⁺, respectively, is avoided or reduced, as confirmed also from the fluorescence lifetime measurements performed.²⁴⁰ As can be observed also in Fig. 28(a), a significant effect on the signal of the emissions (and in the thermometric performance as well, as we will analyze in section 5) was produced by excitation at 808 nm in the core@shell structures, although in that case the emission of Tm³⁺ is almost non detected. A possible explanation for this might be that upon 808 nm excitation, the mechanism of generation of the Tm³⁺ emission is mainly governed by the deexcitation of the 3H4 state and not through the cross relaxation process. Hence, as presented in Fig. 27(a), upon 808 nm excitation, the electrons of Tm3+ will be excited to the 3H4 state, from where through a direct radiative decay to the ground state, the 1470 nm emission will be generated.

Since the 3H_5 energy level is not populated any more via cross relaxation processes, but through a simple non-radiative decay from the 3H_4 energy level (Fig. 27(a)), the emission at 1230 nm is almost inexistent. Here, the thermometric parameter was defined by the three intensity ratios that can be calculated combining the emissions of Yb $^{3+}$, Tm $^{3+}$ and Er $^{3+}$: (i) the 1000 nm emission of Yb $^{3+}$ combined with the 1230 nm emission of Tm $^{3+}$, both located in the II-BW; (ii) the 1000 nm

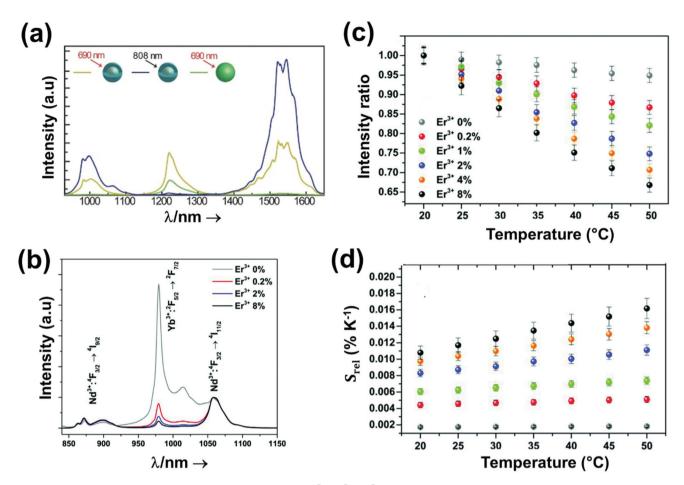


Fig. 28 (a) Room temperature emission spectra of simple core Er^{3+} , Tm^{3+} , Yb^{3+} :LaF₃ under 690 nm excitation (green) and of active core@active shell Er^{3+} , Yb^{3+} :LaF₃@Yb³⁺, Tm^{3+} :LaF₃ (concentrations of lanthanide ions Tm^{3+} , Yb^{3+} and Tm^{3+} are 10 mol%, 10 mol% and 2 mol%, respectively) under 690 (yellow) and 808 nm (blue) laser excitation. The green sphere and the grey core covered with a blue sphere represent the simple core and the active core@active shell, respectively. Reprinted with permission from ref. 240. Copyright 2017, Wiley-VCH Verlag GmbH & Co. KGaA. Effect of the Tm^{3+} concentration on: (b) the intensity of the emissions, (c) the intensity ratio, and (d) Tm^{3+} , Tm^{3+} , Tm^{3+} : Tm^{3+} , Tm^{3+} ; Tm^{3+}

emission of Yb³⁺ combined with the 1550 nm emission of Er^{3+} , located within the II- and III-BWs, that will be described in section 5; and (iii) the 1230 nm emission of Tm^{3+} combined with the 1550 nm emission of Er^{3+} , located within the II- and III-BWs, that will be also described in section 5. For the luminescent thermometer operating exclusively in the II-BW, a maximum S_{rel} of 3.9% K^{-1} was obtained at room temperature, Er^{240} three times higher than that corresponding to the nanoparticles in which the three Er^{3+} were located in the same layer (Table 3). This improvement was assigned to the efficient Er^{3+} and quenching rates in the core@shell structures.

Returning to the case of Yb³⁺ and Nd³⁺ emissions located in the II-BW, a hexagonal Er³⁺,Yb³⁺:NaYF₄@Nd³⁺,Yb³⁺:NaYF₄ core@shell structure was analyzed as a luminescent temperature sensor. Here, to determine the temperature, the authors used the emissions generated by Yb³⁺ and Nd³⁺ located in the outer shell (the emissions generated by the ions in the inner core were used for sensing in the visible range, which is out of the scope of this review).²⁴¹ By exciting the nanocrystals at

808 nm, the electrons are excited from the Nd3+ ground state to the ⁴F_{5/2} excited state, followed by a fast non-radiative decay to populate the ⁴F_{3/2} metastable state. At this stage, different radiative and non-radiative processes can take place (Fig. 27(b)), starting from the generation of the Nd³⁺ emission at 1060 nm, corresponding to the ${}^{4}F_{3/2} \rightarrow {}^{4}I_{11/2}$ electronic transition, and the ET between the Nd3+ and Yb3+ ions present in the shell, ruled by the Miyakawa-Dexter model.²³¹ From here, the emission of Yb3+ located at 980 nm, assigned to the 2F5/2 \rightarrow $^2F_{7/2}$ transition, is generated. Also, the energy migration from the Yb³⁺ ions in the shell to the Yb³⁺ ions in the core, followed by an ET between Yb3+ and Er3+, can take place, leading finally to the green emission of Er3+ in the visible range.241 The mechanisms for the generation of the green emissions of Er³⁺ are out of the scope of this review; thus, they are not discussed here. The intensity ratio, fitted to the model described by eqn (42) between the 1060 nm emission of Nd³⁺ and the 980 nm emission of Yb3+, was analyzed as a function of temperature, in the 103-443 K range, and the concentration of Nd³⁺

in the outer shell, tuned from 1 to 15 at% with respect to the concentration of $Yb^{3+}.^{241}$ The results show that a maximum $S_{\rm rel}$ of 2.1% K^{-1} at 370 K could be obtained.²⁴¹ In addition, this $S_{\rm rel}$ is proportional to the content of Nd^{3+} in the outer shell. This effect is related to the shortening of the average Nd^{3+} – Yb^{3+} distance, hence facilitating the Nd^{3+} to Yb^{3+} ET.²⁴¹

In simple Nd3+,Yb3+-doped LaF3 nanoparticles and two different active core@active shell nanostructures, one having the core doped with Nd3+ ions and the shell with Yb3+ ions (Nd3+:LaF3@Yb3+:LaF3) and the other constituted by a Yb3+doped core and a Nd3+-doped shell (Yb3+:LaF3@Nd3+:LaF3),223 Nd³⁺ displayed two additional emission bands at 890 nm (located in the I-BW) and 1350 nm (located in the II-BW), corresponding to the ${}^4F_{3/2} \rightarrow {}^4I_{9/2}$ and ${}^4F_{3/2} \rightarrow {}^4I_{13/2}$ transitions, respectively, when excited at 790 nm. The energy of this excitation source is absorbed by Nd3+ ions, promoting their electrons to the ⁴F_{5/2} excited state. Then, a rapid phonon-assisted relaxation to the metastable ${}^4F_{3/2}$ state takes place, from which three radiative decays to the ${}^4I_{9/2}$, ${}^4I_{11/2}$ and ${}^4I_{13/2}$ levels produce the emissions located at 890 nm, 1060 nm and 1350 nm, respectively (Fig. 27(c)). A further ET process to Yb³⁺ can lead to the excitation of the electrons of these ions from their ground state up to the 2F_{5/2} state, from which the 1000 nm emission is produced.

Here, all the possible intensity ratios considering the emissions of Nd^{3+} *versus* the emission of Yb^{3+} , which obey a linear function when plotted against temperature, were explored for temperature sensing by luminescence means. The same S_{rel} was found for all the three intensity ratios, either in the simple nanoparticles or in the core@shell structures (Table 3). Comparing the performance of the simple nanoparticles with that of the core@shell structures, a four-fold improvement in their S_{rel} was observed. Nevertheless, the S_{rel} values are relatively low, with a maximum of $\sim 0.41\%$ K $^{-1}$ in the physiological range of temperatures. 223

The thermal sensitivity of these Nd3+,Yb3+-doped LaF3 nanoparticles is comparable to that of Nd3+ and Yb3+codoped lanthanide complex with pyridine-3,5-dicarboxylate (hereafter PDC).²⁴³ Here, Nd³⁺ emission peaks at 1054 nm and 1065 nm are generated from the transitions between the different Stark sublevels in the ${}^4F_{3/2} \rightarrow {}^4I_{11/2}$ electronic transition, whereas the Yb3+ emission at 1005 nm corresponds to the ${}^{2}F_{5/2} \rightarrow {}^{2}F_{7/2}$ transition, after excitation at 808 nm. The intensity ratios among the three combinations (the combination of Nd3+ emissions is presented in section 4.2.1) of these emissions were analyzed for temperature-sensing purposes. The temperature dependence of the 1065 nm/1005 nm intensity ratio was fitted to eqn (14), while the 1054 nm/ 1005 nm intensity ratio was fitted to a linear equation. The highest $S_{\rm rel}$ was obtained for the 1054 nm/1005 nm ratio with a value of 0.48% K^{-1} and a δT of 0.08 $\text{K}^{.243}$ It should be underlined here that the uncertainty in the determination of the intensity ratio here was taken as 0.033%, 243 which of course, will imply a lower δT .

Cortelletti et al. reported the thermometric performance of a complex system in the form of a core@multishell nano-

structure composed of Tm³⁺,Yb³⁺:SrF₂@Y³⁺:SrF₂@Er³⁺,Nd³⁺, Yb3+:SrF2@Nd3+:SrF2,242 based on the intensity ratio between the emissions of Yb3+ at 980 nm and of Nd3+ located at 1060 nm (another emission of Nd³⁺ is generated at 870 nm, but the authors used only the one located at 1060 nm for temperature-sensing purposes due to its higher emission intensity). The performance of this luminescence nanothermometer was closely related to the amount of Er3+ ions in shell 2. In this structure, the energy of the 806 nm excitation source is absorbed by the Nd3+ ions present in the most external shell. This energy allows the promotion of the Nd³⁺ electrons to the 4F_{5/2} excited state, from which a non-radiative decay will populate the lower 4F_{3/2} excited state. From here, radiative decays to ${}^{4}I_{9/2}$ and ${}^{4}I_{11/2}$ states generate the emissions located at 870 nm and 1060 nm, respectively. From the ${}^4F_{5/2}$ excited state of Nd³⁺, an ET process can lead to the population of the Nd3+ ions located in shell 2 (Fig. 27(d)). Then, an ET process to Yb³⁺ can lead to the population of its excited state prior to a radiative decay to its ground state, generating the 980 nm emission. 242 Hence, between Nd3+ ions and Yb3+ ions, ET and BET processes may exist. However, this is true only when no Er3+ ions in shell 2 are found. With the addition of Er³⁺ ions, several ET processes between Yb³⁺ and Er³⁺ ions may occur, influencing in this way not only the intensity of the emissions generated, but also the thermometric performance of the system. With the increase of Er3+ concentration, the emission intensity of Yb3+ clearly decreases (Fig. 28(b)), due to the increase of the probability of the $Yb^{3+} \rightarrow Er^{3+}$ ET process. This ET process quenches the emission of Yb³⁺, which works in favor of the luminescence nanothermometry means. When considering the evolution of the intensity ratio between the emissions of Yb3+ and Nd3+ with temperature, its slope increases as the concentration of Er3+ increases (Fig. 28(c)), and so does S_{rel} (Fig. 28(d)), generating better luminescent thermometers. From these figures it can be concluded that the $S_{\rm rel}$ of the luminescent thermometer with the highest ${\rm Er}^{3+}$ concentration is approximately 5 times better compared with the material without this ion. Thus, efficient $Yb^{3+} \rightarrow Er^{3+}$ ET processes are beneficial for thermometric performance, although it is difficult to predict the effect of these processes in complex systems. Nevertheless, the maximum S_{rel} achieved (1.62% K⁻¹ in the physiological range of temperatures) is still lower when compared with the one achieved in Er3+,Yb3+: $LaF_3@Tm^{3+},Yb^{3+}:LaF_3,^{240}$ and $Er^{3+},Yb^{3+}:NaYF_4@Nd^{3+},Yb^{3+}:$ NaYF₄ nanostructures. 241

To conclude this subsection, it is clear that core@shell and core@multishell nanoarchitectures offer a boost in the brightness of the emissions generated and in addition, a significant enhancement of their temperature-sensing performance. However, it should be admitted that these structures constitute very complex systems either from the synthetic or the spectroscopic point of view. Additionally, most of them are hydrophobic because they are coated with organic surfactants; hence, for biomedical applications further modifications of their surfaces are needed to make them water dispersible.

4.2. Nd³⁺-doped luminescent nanothermometers operating in the II-BW

4.2.1. Single Nd3+-doped luminescent thermometers operating in the II-BW. Luminescent thermometers based on single Nd³⁺-doped nanoparticles operating in the II-BW are mainly based on the emissions generated by the Stark sublevels of the ${}^{4}\mathrm{F}_{3/2} \rightarrow {}^{4}\mathrm{I}_{11/2}$ electronic transition located at ~1064 nm. Hence, to determine the performance of these thermometers, the fitting model based on FIR was applied, using eqn (9). Another option for temperature sensing with this ion relies on the use of the emission at the limit of the II-BW spectral region located at ~1350 nm. When the 1064 nm emission band is used, $S_{\rm rel}$ is quite low, ranging from 0.16% K⁻¹ to 0.45% K⁻¹ (Table 3). Despite this, from these 1064 nm emission-based luminescent thermometers, some useful strategies can be extracted. For instance, the incorporation of an amorphous silica coating can tailor the performance of the luminescent nanothermometers. as explored in Nd3+:YVO4@SiO2 nanoparticles.244 The luminescent thermometric performance of this material was tested as a function of the thickness of the silica layer and compared with

that of the uncoated material. The results show that there exists a proportional relationship between S_{rel} and the thickness of the silica shell (Fig. 29(a)).

In Nd³⁺:YAP nanoparticles, the 1350 nm emission band was used for temperature-sensing purposes by analyzing the full width at half maximum $(\Delta \nu)$ of the emission located at 1348 nm as a function of the temperature, from room temperature to 370 K. 157 Upon green light excitation, Nd 3+ promotes its electrons from the ${}^4I_{9/2}$ ground state to the ${}^4G_{9/2}$ or ${}^4G_{7/2}$ excited states, followed by a non-radiative decay process to the $^4\mathrm{F}_{5/2}$ and $^4\mathrm{F}_{7/2}$ states. The $^4\mathrm{F}_{3/2}$ level can also be populated by a non-radiative decay from the ${}^4F_{5/2}$ level. From this state, a radiative decay to the 4I13/2 level generates the emission located at 1348 nm (Fig. 30(a)). As the temperature increased, $\Delta \nu$ became broader and its dependence on temperature could be fitted to a second-order polynomial function of the form: 157

$$\Delta v = a + bT + cT^2 \tag{43}$$

where a, b and c are constants to be determined from the fitting of the experimental data. The maximum S_{rel} for this

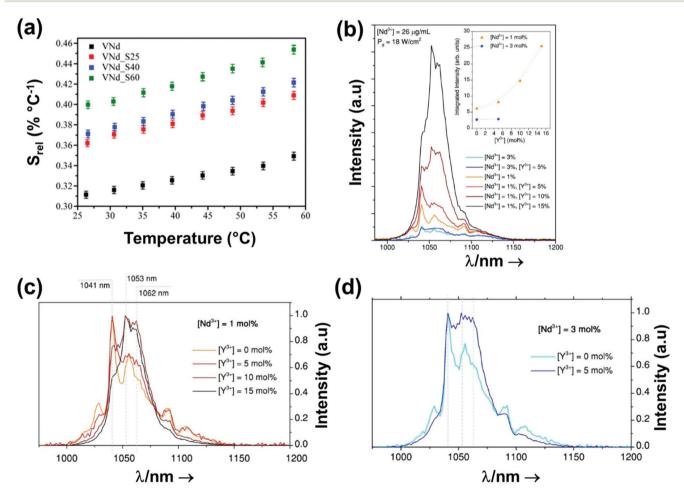


Fig. 29 (a) Effect of the thickness of the silica shell on the thermal performance of Nd3+-doped YVO4@SiO2, where S25, S40, S60 stand for the thickness of the silica in nm. Reprinted with permission from ref. 244, Copyright 2016, Elsevier. Effect of Y^{3+} impurities in Nd^{3+} -doped CaF_2 on: (b) the intensity, and (c) and (d) the spectral line shape of the ${}^4F_{3/2} \rightarrow {}^4I_{11/2}$ emission band. Reprinted with permission from ref. 245. Copyright 2018, American Chemical Society.

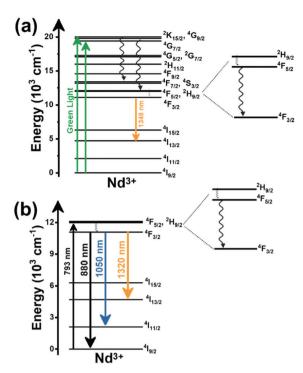


Fig. 30 Mechanisms of generation of the: (a) 1348 nm emission band in ${\rm Nd}^{3+}$:YAP nanoparticles under green light excitation, and (b) 1050 nm and 1320 nm emission bands in ${\rm Nd}^{3+}$:LiLuF₄@LiLuF₄ nanoparticles under 793 nm excitation.

luminescence thermometer is 3.3% K⁻¹ and the minimum $\delta T = 0.37$ K at room temperature. ¹⁵⁷

An additional strategy towards tuning the performance of these luminescent thermometers is the introduction of impurities in the host, such as the incorporation of Y3+ ions in Nd³⁺:CaF₂ cubic nanoparticles.²⁴⁵ The presence of Y³⁺ avoids the formation of clusters of Nd³⁺ ions in these nanoparticles, and thus it prevents the quenching of their emissions, leading to a brighter luminescence upon using the proper doping ratio (Fig. 29(b)).245 A brighter emission lead to a more accurate and reliable performance of the luminescent thermometer. 13 Nevertheless, the S_{rel} of these nanocrystals, determined as a function of the amount of Y³⁺ on the structure, is relatively low, since the emissions used in this luminescent thermometers arise from TCLs with a low ΔE between them, in the range of 49–98 cm⁻¹, depending on the Y³⁺ content.²⁴⁵ Besides the generation of a brighter emission, the introduction of Y³⁺ generated changes in the shape of the emission band (Fig. 29(c) and (d)), due to the modification of the environment surrounding the Nd3+ ions and to a different distribution of luminescent centers in the nanoparticles.

Skripka *et al.* demonstrated that upon intermixing and discriminating different factors such as the absolute intensity, the spectral shift and the line broadening of the emissions generated by different Stark sublevels, contributing to the shape of the Nd³⁺ emissions, a better thermometric performance can be obtained that is based only on the thermalization of the Stark sublevels.¹⁶⁴ Hence, upon excitation at 793 nm,

the Nd³⁺ ions in Nd³⁺:LiLuF₄@LiLuF₄ core-shell nanoparticles can absorb the energy and excite its electrons to the ⁴F_{5/2} state, from which the ⁴F_{3/2} state can be populated due to non-radiative decays (Fig. 30(b)). From here, successive radiative decays can lead to the generation of emissions lines located at 880 nm (explained in section 2.2.1), 1050 nm and 1320 nm. The corresponding FIRs between different emission peaks coming from different Stark sublevels of each of these emissions were analyzed to develop luminescent thermometers with these nanomaterials. Hence, the FIR of these emission bands provided $S_{\rm rel}$ values ~0.49% K⁻¹ in the physiological range of temperatures. The authors strongly emphasized that if only the thermalization of the Stark sublevels would have been taken into account, $S_{\rm rel}$ would have been $\sim 0.09\%~{\rm K}^{-1}$, since the ΔE between the different Stark sublevels was around 55 cm⁻¹.¹⁶⁴ Thus, by resolving the fine Stark structure of the Nd³⁺ spectra with the adequate detector systems (InGaAs NIR sensors in this case) and the proper choice and design of the host material, an improvement in the thermal sensitivity based on single Nd³⁺ nanothermometers can be achieved.

Overall, concerning the thermometric performance of single Nd^{3^+} -doped luminescent thermometers, the use of the 1350 nm emission band results in a higher S_{rel} . However, if advanced detector systems and proper host materials are chosen, the 1050 nm emission band can stand as a promising alternative. Nevertheless, the luminescent thermometers based on the 1350 nm emission band rely on excitation in the visible range, which will, of course, limit the biomedical applications of these temperature sensors. ^{13,86,90} On the other hand, normally the 1050 nm emission is obtained after excitation in the I-BW, which despide their low S_{rel} , facilitates their use in biological applications.

4.2.2. Dual emitting doped Nd^{3+} luminescent thermometers operating in the II-BW. Although there are not too many examples of dual emitting luminescent thermometers combining Nd^{3+} with other materials or Ln^{3+} ions apart from Yb^{3+} , the ones reported exhibit a high S_{rel} .

For example, Cerón et al.63 developed a high-resolution temperature sensor consisting of Nd3+:NaGdF4 dielectric nanoparticles and semiconductor PbS/CdS/ZnS quantum dots (QDs) combined in a poly(lactic-co-glycolic acid) (PLGA) hybrid nanostructure, excited within the I-BW (808 nm) and emitting in the II-BW. This luminescent thermometer worked on the basis of the different temperature responses of its components. While the intensity of the emission of Nd³⁺, located at 1060 nm, remained unchanged in the physiological range of temperatures, the intensity of the 1300 nm emission band of the QDs, assigned to their first exciton, decreased linearly due to photon-assisted processes (Fig. 31(a)). Hence, they developed a highly sensitive thermometer based on the change in the emission intensity of the quantum dots and using the emission of the Ln3+ ion as a reference probe. The maximum value of $S_{\rm rel}$ and the minimum value of δT , obtained at 303 K, were 2.5% K⁻¹ and 0.2 K, respectively.⁶³

A water-soluble multishell complex nanostructure based on hexagonal $NaGdF_4$ has been also proposed as a temperature

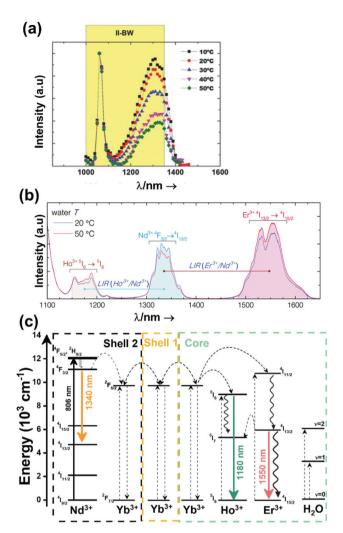


Fig. 31 (a) Evolution of the intensity of the emissions of Nd:NaGdF4 and PbS/CdS/ZnS nanoparticles combined in PGLA hybrid nanostructures. Reprinted with permission from ref. 63. Copyright 2015, Wiley-VCH Verlag GmbH & Co. KGaA. (b) Emissions of Er3+,Ho3+, Yb3+@Yb3+@Nd3+,Yb3+@NaGdF4 multishell nanostructures (Reprinted with permission from ref. 246. Copyright 2017, The Royal Society of Chemistry), and (c) mechanisms of generation of these emission lines.

probe in the II-BW (and in the II- and III-BW regions, as will be discussed in section 5). This multishell nanostructure, formed by an active core, a Yb3+ doped inner shell, a second active shell containing Nd3+, and a final inert shell, Er³⁺,Ho³⁺,Yb³⁺:NaGdF₄@Yb³⁺:NaGdF₄@Nd³⁺,Yb³⁺: NaGdF₄@NaGdF₄, was synthesized via a thermal decomposition process, leading to oleate-capped nanoparticles.²⁴⁶ To make these nanostructures water dispersible, they were encapsulated in polyethylene glycol-grafted phospholipid (PEG-DOPE) micelles. After excitation at 806 nm emissions located in the II-BW and III-BW (Fig. 31(b)) were generated. The energy of the excitation source is absorbed by the Nd³⁺ ions located in shell 2 that subsequently generate the emission at 1340 nm, after a non-radiative decay process to the $^4F_{3/2}$ level. Part of this energy is also transferred, via several ET

processes through Yb3+ ions in shell 1, to the core, leading to the population of the ⁵I₆ and ⁴I_{11/2} levels of Ho³⁺ and Er³⁺, respectively (Fig. 31(c)). The 1180 nm emission of Ho³⁺ is generated after a radiative decay to the ground state ⁵I₈. From the excited ⁴I_{11/2} level of Er³⁺, a non-radiative decay process leads to the population of the lower excited level ⁴I_{13/2}, prior to relaxing to the ground state, where the 1550 nm emission located in the III-BW is produced. Therefore, the role of the inert shell 1 is to suppress the interionic quenching of the Er³⁺ and Ho³⁺ emissions by Nd³⁺ ions, while the presence of Yb³⁺ in the core, shell 1 and shell 2 facilitates the energy migration across these multiple layers. 246 Hence, from this nanostructure, three emissions are generated, two of them located in the II-BW (1180 nm of Ho³⁺ and 1340 nm of Nd³⁺) and one in the III-BW (1550 nm of Er³⁺). In this section, we will review only the luminescence thermometric possibilities of the emissions located in the II-BW. The temperature determination from this luminescent thermometer was done by studying the temperature dependence of the intensity ratio between the emission at 1180 nm of Ho³⁺ (located in the core) and the 1340 nm emission of Nd³⁺ (located in shell 2) in the physiological range of temperatures. With the increase of the temperature, the intensity of the emission of Ho3+ increased, due to the phononassisted Yb³⁺ (${}^{2}F_{5/2}$) \rightarrow Ho³⁺ (${}^{5}I_{6}$) process, 247 whereas the emission of Nd3+ decreased, due to the phonon-assisted ET Nd3+ ${}^{(4}F_{3/2}) \rightarrow Yb^{3+} ({}^{2}F_{5/2})$ process^{201,246} (Fig. 31(b) and (c)). Since these multishell nanocrystals were dispersible in water, additional de-excitation channels associated with the presence of water molecules on the surfaces of these nanocrystals were expected. In order to prevent these de-excitation channels, the third passivating undoped shell was introduced. The maximum S_{rel} obtained for the 1180 nm/1340 nm intensity ratio was $1.17\% \text{ K}^{-1}$.

Despite its relatively good temperature-sensing performance, the core@shell and multishell materials presented in this subsection are based on fluoride-doped hosts synthesized (mostly) through the thermal decomposition of trifluoroacetate precursors to control the size and the morphology of the nanoparticles, which is accompanied by highly toxic by-products of the reactions.²⁴⁸ Nevertheless, in terms of the thermometric performance, clearly complex nanoarchitectures, such as core@shell and multishell materials, exhibit high relative thermal sensing compared with the other forms of materials presented up to now (Fig. 32(a) and (b)) and fulfill the conditions to be used in biomedical applications. They can be excited in the BWs and generated highly temperature-dependent emissions, also located within the BWs spectral regions.

Lanthanide-doped luminescent nanothermometers operating in the II and III-BWs simultaneously

Based on the emission at 1550 nm, Er3+ ions represent one of the most explored lanthanide luminescent ions for thermome-

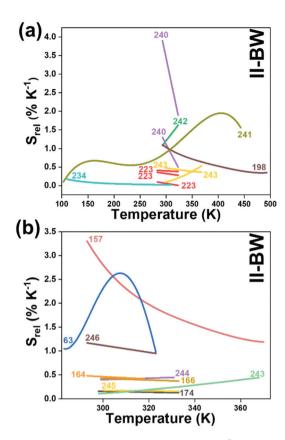


Fig. 32 Temperature dependence of $S_{\rm rel}$ of: (a) Yb³⁺- and (b) Nd³⁺- doped luminescent thermometers operating in the II-BW region. The numbers indicate corresponding references for each thermometer.

try operating in the II- and III-BWs simultaneously, by combining them with other emissions located in the II-BW arising either from other ${\rm Ln^{3^+}}$ ions or transition metals. Concerning the combination of ${\rm Er^{3^+}}$ with ${\rm Ln^{3^+}}$ ions, typical examples involve ${\rm Yb^{3^+}}$, ${\rm Nd^{3^+}}$ and ${\rm Ho^{3^+}}$ emissions. For the combination with transition metals, only one example involves the application of the ${\rm Er^{3^+}}$ emission with the emission of nickel(II), located in the II-BW. A few cases involve also single center emitting ${\rm Ln^{3^+}}$ ions (${\rm Nd^{3^+}}$ and ${\rm Tm^{3^+}}$) as potential thermometers operating in this region.

5.1. Er^{3+} -doped luminescent nanothermometers operating in the II- and III-BWs simultaneously

Several luminescent nanothermometers have been reported using the 1550 nm emission of $\mathrm{Er^{3+}}$ together with the emission of other $\mathrm{Ln^{3+}}$ ions, including $\mathrm{Yb^{3+}}$, $\mathrm{Ho^{3+}}$ and $\mathrm{Nd^{3+}}$, operating in this way in the II- and III-BWs simultaneously. Among these combinations, pairing it with the 1000 nm emission of $\mathrm{Yb^{3+}}$ stands out as one of the most promising strategies to develop highly sensitive luminescent thermometers, especially when sensitized via $\mathrm{Tm^{3+}}$ (Table 4). For example, $\mathrm{Er^{3+}}$, $\mathrm{Yb^{3+}}$: $\mathrm{LaF_{3}}$ ambs $\mathrm{LaF_{3}}$ ambs $\mathrm{LaF_{3}}$ ambs $\mathrm{LaF_{3}}$ core–shell nanocrystals (also explored in the II-BW) display an outstanding S_{rel} with a value of 5% $\mathrm{K^{-1}}$ at room temperature. This high S_{rel} was extracted from the

intensity ratio between the 1000 nm emission of Yb3+ and the 1550 nm emission of Er3+, generated after the core@shell nanocrystals were excited at 690 nm (the mechanism of generation of the emission lines is presented in Fig. 26(a)). Compared with the performance of the Er³⁺,Tm³⁺,Yb³⁺:LaF₃ nanoparticles, the core@shell nanocrystals have a 7-fold higher S_{rel} (Table 4).²⁴⁰ Furthermore, the authors investigated the effect of the excitation laser wavelength applied (690 nm and 808 nm) on the temperature-sensing properties of these nanocrystals. The results suggest that the high S_{rel} obtained when exciting the nanoparticles at 690 nm is due to the contribution of the ³H₅ level of Tm³⁺ ions to the overall emission spectrum generated by the nanoparticles and/or when the ³H₅ (Tm^{3+}) , ${}^2F_{5/2}$ $(Yb^{3+}) \rightarrow {}^3H_5$ (Tm^{3+}) , ${}^2F_{5/2}$ (Yb^{3+}) ET process is involved, implying that excitation at 690 nm, contrary to the excitation at 808 nm, highly favors this ET process. 240 For the case of excitation at 808 nm, the maximum S_{rel} was only $\sim 0.80\%$ K⁻¹. Hence, an important conclusion that can be drawn from this reference is that engineering core@shell nanostructures is not always a key feature for producing highly sensitive luminescent nanothermometers, but one has to properly select the most suitable excitation wavelength.

Er3+,Yb3+:Na2K(Lu3Si6O18) nanoparticles, with a triclinic crystalline structure, and synthesized by an autoclave-assisted hydrothermal process, were also tested as luminescent nanothermometers over a wide range of temperatures (12-450 K). 249 The emission lines, located at 930-1125 nm corresponding to electronic transitions of Yb3+ and 1425-1640 nm corresponding to electronic transitions of Er³⁺, were generated after excitation at 903 nm. Yb3+ acted here also as a sensitizer, by absorbing the energy of the excitation source and promoting its electrons to the ²F_{5/2} level. From this level, radiatively decaying to the ground state can generate the emission of these ions. In addition, from the excited level of Yb³⁺, an ET process can populate the ⁴I_{11/2} level of Er³⁺. From this level, a non-radiative decay can populate the ⁴I_{13/2} level, prior to a radiative decay to the ground state, generating the emission of Er3+. A typical mechanism for the generation of these emission lines is similar to the ones presented in Fig. 24(a) or Fig. 26(b).

To extract the intensity ratio and eventually $S_{\rm rel}$, the authors correlated the total transition probabilities to the inverse of the lifetime using the classic Mott–Seitz model. ^{250,251} Besides this, this lifetime, considered as the sum of the radiative (assumed to be temperature independent) and non-radiative lifetime (expressing an Arrhenius type of temperature dependence), is proportional to the intensity of the emissions, as follows: ^{250,251}

$$\frac{I(T)}{I_0} = \frac{\tau(T)}{\tau_0} \tag{44}$$

where I(T) and I_0 stand for the intensity at temperature T and 0 K, respectively, whereas $\tau(T)$ and τ_0 stand for the lifetime at T and 0 K (the radiative lifetime), respectively. Taking into account two deactivation channels arising from two different

Table 4 Ln ³⁺ -doped luminescent thermometers operating simultaneously in the II and III-BWs regions. In the table are shown the activators (A) and sensitizers (S). The excitation (λ_{exc}) and emis-
sion (λ_{em}) wavelengths are shown in nanometers (nm), together with the corresponding transitions of the emissions. ΔT stands for the temperature range where the temperature reading was
investigated. The thermometric parameter (4) indicates the luminescent nanothermometry class used: I for intensity ratio thermometry. The maximum relative thermal sensitivity (S _{rel.}) and the
minimum temperature resolution (3T) are presented at the temperature where these values were obtained. We indicate with an asterisk the values of Srei or 3T calculated by us using the para-
meters published in the corresponding references. The double line separation between rows stands for different types of lanthanide-doped nanothermometers, as discussed in the corresponding
subsections

Ą	ω.	Host	$_{\rm nm)}^{\lambda_{\rm exc}}$	$\lambda_{ m exc}$ (nm) $\lambda_{ m em}$ (nm)	Transitions	$\Delta T (K)$	◁	$S_{\rm rel}/T \qquad \delta T \\ (\%~{\rm K}^{-1})/{\rm K} ({\rm K})$	δT (K)	Ref.
Er ³⁺ , Yb ³⁺ Er ³⁺ , Yb ³⁺ Er ³⁺ , Yb ³⁺ Er ³⁺ , Yb ³⁺	Tm ³⁺ Yb ³⁺ Tm ³⁺ Tm ³⁺	Er, Yb:LaF3 @Tm, Yb:LaF3 Na5 K[Lu3Sl ₆ O ₁₈] Er, Yb:LaF3 @Tm, Yb:LaF3 Er, Tm, Yb:LaF3	690 808 690	1000 (Yb ³⁺), 1550 (Er ³⁺) 1125 (Yb ³⁺), 1640 (Er ³⁺) 1000 (Yb ³⁺), 1550 (Er ³⁺) 1000 (Yb ³⁺), 1550 (Er ³⁺)	$\begin{array}{l} {}^{2}F_{5/2} \rightarrow {}^{2}F_{7/2}\left(Yb^{3^{*}}\right), {}^{4}I_{13/2} \rightarrow {}^{4}I_{15/2}\left(Er^{3^{*}}\right) \\ {}^{2}F_{5/2} \rightarrow {}^{2}F_{7/2}\left(Yb^{3^{*}}\right), {}^{4}I_{3/2} \rightarrow {}^{4}I_{15/2}\left(Er^{3^{*}}\right) \\ {}^{2}F_{5/2} \rightarrow {}^{2}F_{7/2}\left(Yb^{3^{*}}\right), {}^{4}I_{3/2} \rightarrow {}^{4}I_{15/2}\left(Er^{3^{*}}\right) \\ {}^{2}F_{5/2} \rightarrow {}^{2}F_{7/2}\left(Yb^{3^{*}}\right), {}^{4}I_{3/2} \rightarrow {}^{4}I_{15/2}\left(Er^{3^{*}}\right) \end{array}$	293–323 12–450 293–323 293–323	$\begin{array}{lll} 293-323 & I_{1000}/I_{1550} \\ 12-450 & I_{1125}/I_{1640} \\ 293-323 & I_{1000}/I_{1550} \\ 293-323 & I_{1000}/I_{1550} \end{array}$	5/293 2.6/26.8 0.80/293 0.7/293	0.3 0.08 0.62* 0.71*	240 249 240 240
Er3+ Ho3+	Yb 3+ Yb 3+ Yb 3+ Yb 3+ Yb 3+ Yb 3+ Yb 3+ Yb 3+ Yb 3+ Yb 3+	NaYF ₄ NaYF ₄ Gd ₂ O ₃ Y ₂ O ₃ YAG BaTiO ₃ NaLuF ₄	980 980 980 980 980 975	1150 (Ho ³⁺), 1550 (Er ³⁺) 1150 (Ho ³⁺), 1550 (Er ³⁺) 1149 (Ho ³⁺), 1545 (Er ³⁺) 1149 (Ho ³⁺), 1545 (Er ³⁺) 1149 (Ho ³⁺), 1545 (Er ³⁺) 1177 (Ho ³⁺), 1545 (Er ³⁺) 1177 (Ho ³⁺), 1545 (Er ³⁺)	$\begin{array}{c} 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{152} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{152} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{152} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{152} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{152} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{152} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{152} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{1522} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{1522} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{1522} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{1522} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{1522} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{1522} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{1522} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{1522} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{1522} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{1522} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{1522} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{1522} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{1522} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{1522} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{1522} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{1522} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{1522} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{1522} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{1522} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{1522} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{1522} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_{1522} \left(\text{Er}^3 \right), \\ 5_{16} \rightarrow 5_{18} \left(\text{Ho}^3 \right), \ 4_{132} \rightarrow 4_$	299-319 298-319 313-573 313-573 313-573 313-573 298-568	299–319 I ₁₁₅₀ /I ₁₅₅₀ 313–573 I ₁₅₅₀ /I ₁₁₄₀ 313–573 I ₁₅₅₀ /I ₁₁₄₀ 313–573 I ₁₅₅₀ /I ₁₁₄₀ 313–573 I ₁₅₅₀ /I ₁₁₄₀ 313–573 I ₁₅₅₀ /I ₁₁₄₀ 228–568 I ₁₁₇₇ /I ₁₅₄₅ 313–573 I ₁₅₅₀ /I ₁₁₄₀	2.17/299 1.87/298 0.82/498 0.80/523 0.48/573 0.21/298	0.23* 0.2 0.61* 0.62* 1.05* 1.25* 2.38*	252 253 254 254 254 254 191 254
Er^{3+} , Nd^3	Nd^{3+}	$Er, Ho, Yb: NaGdF_4 @Yb: NaGdF_4 @Nd, Yb: NaGdF_4 @NaGdF_4 \\$	908	$1340({\rm Nd}^{3+}),1550\;({\rm Er}^{3+})$	$1340(Nd^{3+}),\ 1550\ (Er^{3+}) {}^4F_{3/2} \rightarrow {}^4I_{13/2}\ (Nd^{3+}),\ {}^4I_{13/2} \rightarrow {}^4I_{15/2}\ (Er^{3+}) 293-323 I_{1550}/I_{1340} 1.1/293$	293-323	I_{1550}/I_{1340}	1.1/293	8.0	246
$\mathrm{Er}^{3+},\mathrm{Ni}^{2+}$	$\mathrm{Er}^{3+},\mathrm{Ni}^{2+}$ SrTiO ₃	SrTiO ₃	375	$1245~(\mathrm{Ni}^{2+}),1540~(\mathrm{Er}^{3+})$	$1245\; (Ni^{2+}), 1540\; (Er^{3+}) ^3T_{2g} \rightarrow ^3A_{2g}\; (Ni^{2+}), ^4I_{13/2} \rightarrow ^4I_{15/2}\; (Er^{3+})$	123-483	$123-483 I_{1540}/I_{1245}$	5.8/483	0.08*	255
Tm ³⁺ Nd ³⁺	Tm^{3+} Nd^{3+}	LaF_3 Gd_2O_3	808	1230, 1470 1315, 1350	$^{3}\mathrm{H}_{5} \rightarrow ^{3}\mathrm{H}_{6} ^{3}\mathrm{H}_{4} \rightarrow ^{3}\mathrm{F}_{4}$ $^{4}\mathrm{F}_{3/2} \rightarrow ^{4}\mathrm{I}_{13/2}$	297-361 303-393	$\begin{array}{lll} 297 - 361 & I_{1470}/I_{1230} \\ 303 - 393 & I_{1315}/I_{1350} \end{array}$	1.90/297 0.23/303	0.26*	256 257

emissions, from which one dominates over the other, such as in the case of this thermometer, the thermometric parameter, related to the ratios between two intensities, is expressed as:

$$\Delta = \frac{I_1}{I_2} = \frac{\Delta_0}{1 + \alpha_1 \exp\left(-\frac{\Delta E_1}{k_B T}\right) + \alpha_2 \exp\left(-\frac{\Delta E_2}{k_B T}\right)} \tag{45}$$

where Δ_0 is the thermometric parameter at 0 K, α is the ratio between the radiative and non-radiative rates, and ΔE_1 and ΔE_2 are the activation energies of the two deactivation channels. For the case of $\mathrm{Er^{3+},Yb^{3+}:Na_2K(Lu_3Si_6O_{18})}$ nanoparticles these two channels, responsible for thermal sensing due to $\mathrm{Er^{3+}-to-Yb^{3+}}$ ET associated with ΔE_1 and ΔE_2 , are, respectively, the $\mathrm{Er^{3+}-to-Yb^{3+}}$ ET through the resonant excited levels, and the energy migration between two distinct neighboring $\mathrm{Yb^{3+}}$ sites. 249 S_{rel} calculated over the temperature range from 12 K to 450 K reached a maximum of 2.6% K $^{-1}$ at 26.8 K. 249 Despite this high value, if we observe the temperature dependence of S_{rel} in the physiological range of temperatures, the value is close to zero (Fig. 33).

A second combination of emissions for luminescent thermometers operating simultaneously in the II- and III-BWs is that formed by the emissions of $\rm Er^{3+}$ and $\rm Ho^{3+}$, located at $\sim\!1550$ nm and $\sim\!1150$ nm, respectively. These emission lines are generated in the presence of $\rm Yb^{3+}$ as a sensitizer; hence special attention should be paid to the overheating problem when pumping at 980 nm. 206

For Er³⁺,Ho³⁺,Yb³⁺:NaYF₄ luminescent nanothermometers operating within these spectral regions, we would like to stress that by using Yb³⁺ as a sensitizer, which has to be excited at 980 nm, and Er³⁺ as an emitter at 1550 nm, ^{253,254} an overlap with the water absorption bands is produced, generating heat and hampering their potential biomedical applications. ^{258,259} For temperature sensing, the intensity ratio between the 1150 nm emission of Ho³⁺ and the 1550 nm emission of Er³⁺

Fig. 33 Temperature dependence of $S_{\rm rel}$ of all nanothermometers operating simultaneously in the II- and III-BW regions. Numbers represent the corresponding literature references for each thermometer.

was used. These emissions lines are generated after the absorbance of the energy of the 980 nm excitation source by Yb³⁺, which promotes its electrons to the 2F_{5/2} excited state. From here, the ${}^4I_{11/2}$ level of Er^{3+} and the 5I_6 level of Ho^{3+} can be populated by ET or photon-assisted ET processes (Fig. 34), respectively. From the ${}^4\mathrm{I}_{11/2}$ level, Er^{3+} can be relaxed nonradiatively to the 4I13/2 level, prior to decaying radiatively to the ⁴I_{15/2} ground state, generating the emission at 1550 nm. ^{253,254} The 1150 nm emission of Ho³⁺ can be generated by the direct relaxation from the ⁵I₆ level to the ⁵I₈ ground state (Fig. 34). In terms of S_{rel} , Er^{3+} , Ho^{3+} , Yb^{3+} : NaYF_4 nanocrystals 253,254 show a 9-fold higher sensitivity than Er³⁺,Ho³⁺,Yb³⁺:NaLuF₄ nanocrystals, 191 both being based on the intensity ratio between the 1150 nm emission of Ho³⁺ and 1550 nm of Er³⁺ (Fig. 33 for a comparison of the evolution of the S_{rel} of these two materials and Table 4 for the maximum S_{rel} values). We observed that the thermometric parameter of the Er³⁺,Ho³⁺,Yb³⁺:NaYF₄ nanocrystals was fitted to a linear function (as in eqn (22)), 253,254 whereas for NaLuF4, nanocrystals it was fitted to a second-order polynomial function (as in eqn (23)). 191 This would explain the differences observed in S_{rel} .

Similar S_{rel} values for Er^{3+} , Ho^{3+} , Yb^{3+} : $NaYF_4$ nanocrystals were also achieved by Wortmann et al.252 The authors evaluated the performance of these nanocrystals by tuning their sizes, dispersing agent (distilled water versus an apolar organic solvent such as cyclohexane) and concentration of the emitting ions. Three trends were observed: (i) S_{rel} increased as the size of the nanocrystals decreased, attributed to the increase of the number of emitting ions in the surface of the nanoparticles due to the increase of the surface-to-volume ratio as the size decreased; (ii) Srel was higher when the nanocrystals were dispersed in water than in cyclohexane, attributed to the O-H vibrational stretching mode of water molecules which leads to quenching of the luminescence; 260,261 and (iii) S_{rel} depended on the concentration of Ho³⁺ ions, due to the influence of the amount of this ion on the ET and CR rates that might take place between the sensitizer and the activators. 252 These three

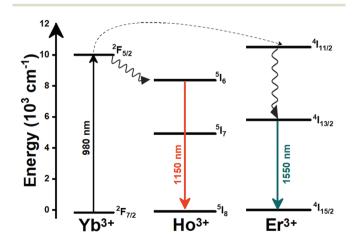


Fig. 34 Mechanism of generation of the emissions lines located in the II- and III-BWs of $\rm Er^{3+}, Ho^{3+}, Yb^{3+}: NaYF_4$ nanoparticles under 980 nm excitation.

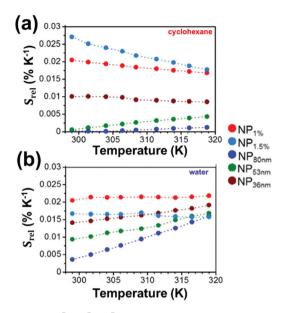


Fig. 35 S_{rel} of Er^{3+} , Ho^{3+} , Yb^{3+} : $NaYF_4$ nanocrystals as a function of dispersing agents: (a) cyclohexane and (b) distilled water. Also the effect of the Ho³⁺ concentration (1.5 mol% and 3 mol%) and the size of the nanocrystals (36 nm, 53 nm and 83 nm) can be observed. Adapted with permission from ref. 252. Copyright 2016, Elsevier.

tendencies can be seen in Fig. 35. The maximum $S_{\rm rel}$ (2.17% K⁻¹) was obtained for nanoparticles with a size ∼48 nm, dispersed in cyclohexane and with a Ho3+ concentration of 1.5 mol%. 252

Continuing the combination between the emissions of Er³⁺ and Ho3+, Jia et al.254 reasoned that phonons can play an essential role in allocation of harvested energy via two types of phonon-assisted processes: multi-phonon relaxation (MPR) between the $^4I_{11/2}$ and $^4I_{13/2}$ states of $\mathrm{Er}^{3+}\text{,}$ and PAET from the ²F_{5/2} state of Yb³⁺ to the ⁵I₆ state of Ho³⁺, that will govern the electronic population of the energy states of Er³⁺ and Ho³⁺ (Fig. 34). The temperature dependence of the probability rates of these processes happening can be described as: 254,262

$$WPR_{MPR} = WPR_{MPR}(0) \left[\frac{exp\left(\frac{h\nu}{k_{B}T}\right)}{exp\left(\frac{h\nu}{k_{B}T}\right) - 1} \right]^{\frac{\Delta E_{1}}{h\nu}}$$
(46)

$$WPR_{PAT} = WPR_{PAT}(0) \left[\frac{exp\left(\frac{h\nu}{k_{B}T}\right)}{exp\left(\frac{h\nu}{k_{B}T}\right) - 1} \right]^{\frac{a\nu_{2}}{h\nu}}$$
(47)

where WPR_{MPR}(0), WPR_{PAT}(0), hv, ΔE_1 and ΔE_2 represent the MPR rate at 0 K, the PAT rate at 0 K, the phonon energy of the host, the energy gap between the ${}^4I_{11/2}$ and ${}^4I_{13/2}$ states of Er³⁺ and the energy gap between the ${}^2F_{5/2}$ electronic level of Yb³⁺ and the 5I6 electronic state of Ho3+, respectively. The thermometric parameter, defined as the intensity ratio between the emissions of Er^{3+} and Ho^{3+} , and the corresponding S_{rel} can be calculated according to: 254,262

$$\Delta = \frac{I_{\text{Er}}}{I_{\text{Yb}}} = \frac{N_{\text{Yb}} W_{\text{R}} W_{\text{MPR}} A_1}{N_{\text{Yb}} W_{\text{PAT}} A_2} = a \left[\frac{\exp\left(\frac{h v}{k_{\text{B}} T}\right)}{\exp\left(\frac{h v}{k_{\text{D}} T}\right) - 1} \right]^b$$
(48)

$$S_{\text{rel}} = \left| \frac{1}{\Delta} \frac{d\Delta}{dT} \right| = \frac{\Delta E_1 - \Delta E_2}{k_B T^2} \frac{1}{\exp\left(\frac{h\nu}{k_B T}\right) - 1}$$
(49)

where $a=rac{W_{
m R}W_{
m MPR}(0)A_1}{W_{
m PAT}(0)A_2}$, $b=rac{\Delta E_1-\Delta E_2}{h
u}$, $N_{
m Yb}$ represents the electronic population of the ${}^2F_{5/2}$ state of Yb³⁺, W_R is the probability rate of the ET process, and A_1 and A_2 represent the spontaneous emission rates of the corresponding emission transitions.

For temperature-sensing purposes, the effect of these two mechanisms was tested in different hosts, including BaTiO₃, Gd₂O₃, Y₂O₃, YAG and YVO₄.²⁵⁴ Theoretically, the different phonon modes of the host would contribute differently to the MPR and PAT processes; however one of them should play a major role.²⁶³ In this study, the authors investigated the effect of the dominant and cutoff phonons on the thermometric performance of these luminescent thermometers, and determined that the dominant phonons have a major contribution to the MPR and PAT processes, and as a consequence, also to their temperature-sensing performance. Hence, the term $h\nu$ in eqn (48) and (49) was assigned to the dominant phonons presented in each of the tested hosts.²⁵⁴ S_{rel} was inversely proportional to the dominant phonon energy (Fig. 36), assuming identical $\Delta E_1 - \Delta E_2$ in the different hosts. The highest S_{rel} was obtained for Er3+,Ho3+,Yb3+:Gd2O3 nanoparticles, which had the lowest dominant phonon energy. It should be noticed that the maximum $S_{\rm rel}$ of these materials was usually obtained at

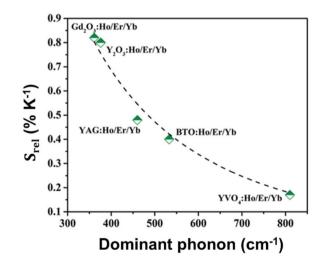


Fig. 36 S_{rel} of Er^{3+} , Ho^{3+} , Yb^{3+} -co-doped nanoparticles as a function of the dominant phonon energy. Reprinted with permission from ref. 254. Copyright 2020, Wiley-VCH Verlag GmbH & Co. KGaA.

the maximum temperature investigated (Table 4), contrary to the general trend observed in the other luminescence thermometers considered. In Table 4, however, it can be also observed that these phonon-assisted luminescent thermometers do not offer an improvement of thermal sensitivity, when compared with the other thermometers (for example Er³⁺,Ho³⁺,Yb³⁺: NaYF₄ nanoparticles^{252,253}).

For the combination of the emissions arising from Er³⁺ and Nd³⁺ ions, located at 1550 nm and 1340 nm, respectively, we water-soluble Er³⁺,Ho³⁺,Yb³⁺@Yb³⁺@Nd³⁺, Yb³⁺@NaGdF₄ core@shell nanoparticles in PEG-DOPE micelles commented on previously in section 4.2.2, as the only example that could be found in the literature. 246 The mechanism of the generation of these emission lines is presented in Fig. 31(c). S_{rel} , calculated from the intensity ratio of the emissions at 1550 nm and 1340 nm, is 1.1% K⁻¹ at room temperature. 246 This value is relatively low compared with other types of core@shell nanoparticles, such as is the case for Er³⁺,Yb³⁺: LaF₃@Tm³⁺,Yb³⁺:LaF₃ nanocrystals, but still higher than that of most of the luminescent nanothermometers based on the combination of emissions arising from Er³⁺ and Ho³⁺ (Table 4). Despite this, the temperature-sensing properties of this multishell structure are good for biomedical applications since they are water dispersible and are excited within the BW spectral regions.

Combining the emission of Er³⁺ with a transition metal, Ni^{2+} in this case, resulted in a S_{rel} slightly higher than the one obtained with Er3+,Yb3+:LaF3@Tm3+,Yb3+:LaF3 nanocrystals. Er3+,Ni2+:SrTiO3 nanocrystals were excited at 375 nm, and two emission bands were generated at 1540 nm, corresponding to Er3+, and 1240 nm, corresponding to Ni2+.255 Upon excitation at 375 nm, Er3+ ions are excited from the ground state to the ⁴G_{11/2} state. From here, successive non-radiative decay processes can populate the 4S_{3/2} and 4I_{13/2} states, and a radiative relaxation from the latter to the ground state results in the generation of the 1540 nm emission band. Similarly, the 1245 nm emission of Ni²⁺ is generated by the absorption of the energy of the 375 nm excitation source, promoting its electrons to the ³T_{1g} (P) excited state. From here, non-radiative decay processes can populate the ${}^{3}T_{1g}$ (F) and ${}^{3}T_{2g}$ (F) metastable states, prior to relaxing radiatively to the ground state, which results in the generation of the corresponding emission band of this transition metal ion. In addition, from the $^{3}\mathrm{T}_{1\mathrm{g}}$ (F) of Ni²⁺ an ET process to the $^{4}\mathrm{S}_{3/2}$ state of Er³⁺ can take place (Fig. 37), which might contribute to increasing the population of the 4I13/2 state, eventually contributing to the generation of the emission band at 1540 nm. The emission located at 1245 nm, as for all the emissions of Ln3+ ions in luminescent thermometers combined with transition metal ions, is temperature insensitive, serving as a reference, whereas the emission at 1240 nm drastically changes as the temperature increases.255

The maximum S_{rel} for this luminescent thermometer was 5.8% K⁻¹, obtained at the highest temperature under investigation (483 K). This value is comparable to that reported for Er3+,Yb3+:LaF3@Tm3+,Yb3+:LaF3 nanocrystals, despite the sig-

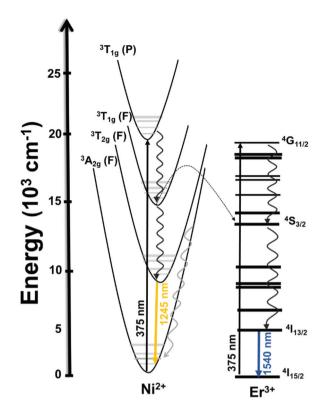


Fig. 37 Mechanism of generation of emission lines in Ni²⁺,Er³⁺:SrTiO₃ nanoparticles under 375 nm excitation.

nificant differences in terms of excitation wavelengths. However, to have a proper comparison with other luminescent thermometers as potential temperature sensors for biomedical applications, S_{rel} should be compared in the physiological range of temperatures. By doing so, the S_{rel} for this luminescent thermometer is reduced to $\sim 0.80\%$ K⁻¹ (Fig. 33).²⁵⁵ In addition, another significant limitation for the luminescent thermometers based on the combination of Er3+ and Ni2+ is the need for excitation in the UV at 375 nm, which reduces significantly the penetration depth that can be achieved in biological tissues, and more important, this light induces phototoxicity in these kinds of sample. 13,86,90

5.2. Other Ln³⁺-doped luminescent nanothermometers operating in the II- and III-BWs simultaneously

Single-doped lanthanide materials based on the emission of Tm³⁺ and Nd³⁺ have been also reported as thermometers operating simultaneously in the II- and III-BWs.

For the Tm³⁺-doped luminescent thermometers, their performance is based on the intensity ratio between the emissions located at 1230 nm and 1470 nm, generated after excitation either at 690 nm or 790 nm, as reported in Tm³⁺:LaF₃ nanoparticles, arising from the 3H4 and 3H5 levels of Tm3+, respectively.²⁵⁶ The mechanism of generation of these emission bands is based on cross-relaxation (CR) and multiphonon decay processes (Fig. 38(a)). The population of the ${}^{3}H_{5}$ level is favored by a CR process, whereas the population of the ³H₄

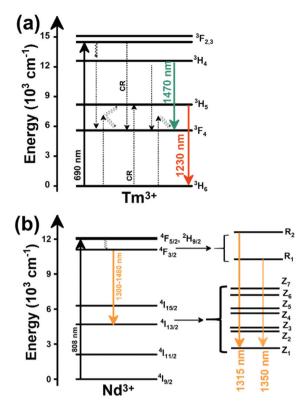


Fig. 38 Mechanism of the generation of the emission lines on: (a) Tm³⁺-doped LaF₃ nanocrystals, and (b) Nd³⁺:Gd₂O₃ nanospheres, operating in the II-BW and III-BW regions simultaneously.

level is due to multiphonon decay processes after electrons are excited to the 3F2,3 levels by absorption of the energy of the 690 nm excitation source. From the ³H₅ level, a radiative decay to the 3H6 ground state generates the emission located at 1230 nm, whereas through the ${}^{3}\text{H}_{4} \rightarrow {}^{3}\text{F}_{4}$ transition, the emission band located at 1470 nm is generated.²⁵⁶

Concerning the temperature-sensing properties of this Tm³⁺ luminescent thermometer, the thermometric parameter was calculated from a linear fitting of the intensity ratio between the 1230 nm and the 1470 nm emission bands of Tm3+ with temperature. However, several parameters were analysed before determining the optimal S_{rel} . First, two different excitation wavelengths (690 nm and 790 nm) were tested with the goal of determining which one generated the brightest emissions. Upon 690 nm excitation, both emissions were brighter than when 790 nm excitation was used (Fig. 39(a)), demonstrating a more efficient excitation path. In fact, when excited at 790 nm, the 1230 nm emission band shows a considerably lower intensity compared with the other emission band, mainly due to the inefficient population of the ³H₅ level due to the small branching ratio of the ${}^{3}H_{4} \rightarrow {}^{3}H_{5}$ transition and the low phonon energy of the LaF3 host (<400 cm⁻¹), ²⁶⁴ which hampers multiphonon decay processes in this large ΔE . When excited at 690 nm, this level is populated through a CR process (${}^{3}F_{3}$; ${}^{3}H_{6} \rightarrow {}^{3}F_{4}$; ${}^{3}H_{5}$) that could be probed by the increase of the intensity of the 1230 nm emission band when the concentration of Tm³⁺ increased. Second, by tuning the concentration of Tm³⁺ from 1 mol% to 5 mol%, the Tm³⁺ concentration that displayed the highest change in the intensity of the emissions as the temperature increased was selected. It can be observed in Fig. 39(b) that the intensity of the 1230 nm

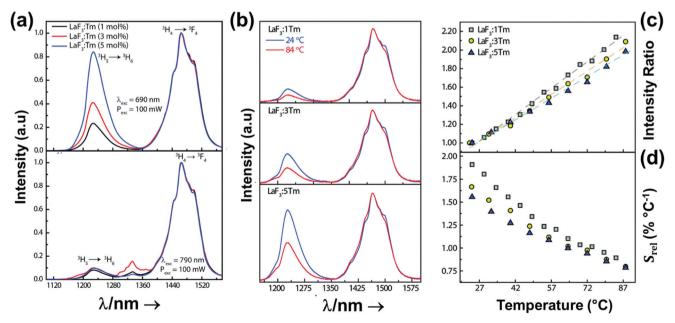


Fig. 39 (a) Emission spectra in Tm³⁺-doped LaF₃ nanocrystals as a function of the concentration of the Tm³⁺ ions (1, 3, 5 mol%) and the excitation source (690 and 790 nm). (b) Variation of the intensity of the emissions in Tm³⁺-doped LaF₃ nanocrystals with the change of the temperature. Variation of the: (c) intensity ratio and (d) S_{rel} of Tm³⁺-doped LaF₃ nanocrystals as a function of the concentration of Tm³⁺ ions and temperature. Reprinted with permission from ref. 256. Copyright 2019, The Royal Society of Chemistry.

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emission band changed substantially as a function of the concentration of Tm³⁺, whereas the band at 1470 nm was almost insensitive, upon increasing the temperature from 297 K to

361 K. The intensity ratio decreased slightly as the Tm³⁺ concentration increased (Fig. 39(c)), and the maximum $S_{\rm rel}$, 0.90% K⁻¹, was obtained for 1 mol% Tm³⁺ (Fig. 39(d)).²⁵⁶ This value is quite high and comparable to that of dual emitting center luminescent thermometers operating in these spectral regions (Table 4). These results, combined with a δT of ~ 0.2 K, clearly underline the potential of this luminescent thermometer for biomedical applications.

For the Nd³⁺ luminescent thermometers operating in these spectral regions, the only example that could be found in the literature is that of Nd3+:Gd2O3 nanoparticles, with the shape of nanospheres with a mean diameter of 108 nm ± 21 nm, synthesized via a precipitation method.²⁵⁷ The performance of this material as a temperature sensor is based on the FIR from the emission lines of the Stark sublevels of the ${}^4F_{3/2} \rightarrow {}^4I_{13/2}$ transition, located in the 1300 nm-1480 nm region, while excited at 808 nm. This excitation source promotes Nd3+ electrons from the ground state to the 4F_{5/2} excited state. From there, the ⁴F_{3/2} level, and its different Stark sublevels, are populated through non-radiative decay processes (Fig. 38(b)). Radiative decays from these Stark sublevels to the lower Stark sublevel of the 4I_{13/2} state generate the two emissions located at 1315 nm and 1350 nm used for temperature sensing in this particular case, with a $S_{\rm rel}$ of 0.23% K⁻¹. 257 According to the authors, this value was independent of the concentration of Nd^{3+} and was exclusively governed by the δT between the Stark sublevels $(150 \pm 20 \text{ cm}^{-1})$, ²⁵⁷ which is relatively low, providing this low value of S_{rel} .

Lanthanide-doped luminescent nanothermometers operating in the III-BW

The third biological window (III-BW), or the short wavelength infrared (SWIR) region as it was originally named, 126 started gaining the attention of researchers when Naczynski et al. reported that longer wavelengths than those lying in the I- and II-BWs transmitted up to three times more efficiently through oxygenated blood and melanin-containing tumors, and as a consequence for luminescent thermometry this would allow for deeper thermal readings in biological samples due to the reduction of tissue scattering and absorption, compared with the other BWs. 126 Moreover, up to now, the luminescent thermometers reported operating in the III-BW are all excited at wavelengths lying in the I-BW (Table 5), providing an important advantage for their possible application in biomedicine. Typical Ln3+ ions applied for temperature sensing in the III-BW are mainly based on the emissions arising from Tm³⁺ located at ~1450 nm and ~1800 nm, the emission of Er3+ located at ~1550 nm, and the emission of Ho3+ located at ~1960 nm. For the case of Tm³⁺-doped luminescent thermometers operating in this spectral region, their performance is determined by the temperature dependence of the intensity ratio of their two emissions, or as more often encountered, combined with the emission of Ho3+, using or not using Yb3+ a as sensitizer, and for the case of Er3+-doped thermometers, the intensity ratio of their two emissions, or as more often encountered, combined with the emission of Ho3+, using or not using Yb3+ as a sensitizer. For the case of Er3+-doped

Table 5 Ln3+-doped luminescent thermometers operating in the III-BW. In the table are shown the activators (A) and sensitizers (S). The excitation (λ_{exc}) and emission (λ_{em}) wavelengths are shown in nanometers (nm), together with the corresponding electronic transition producing these emissions. ΔT stands for the temperature range where the temperature reading was investigated. The thermometric parameter (Δ) indicates the luminescent nanothermometry class used in each case: FIR for band-shape and I for intensity ratio thermometry, respectively. The maximum relative thermal sensitivity (S_{rel}) and the minimum temperature resolution (δT) are presented at the temperature where these values were obtained. Marked by an asterisk are the values of S_{rel} or δT calculated by us, using the parameters published in the corresponding references. The double line separation between rows stands for different types of lanthanide-doped nanothermometers, as discussed in the corresponding subsections

A	S	Host	$\frac{\lambda_{\mathrm{exc}}}{(\mathrm{nm})}$	$\lambda_{\mathrm{em}} \left(nm \right)$	Transitions	$\Delta T (K)$	Δ	$S_{\rm rel}/T$ (% K ⁻¹)/K	δ <i>T</i> (K)	Ref.
Tm ³⁺ Tm ³⁺ Tm ³⁺ Tm ³⁺	Tm ³⁺ Yb ³⁺ Yb ³⁺ Yb ³⁺	KLu(WO ₄) ₂ NaYF ₄ KLu(WO ₄) ₂ KLu(WO ₄) ₂	808 980 980 808	1470, 1740 1470, 1740 1470, 1740 1470, 1740	$\begin{array}{c} {}^{3}H_{4}^{3}F_{4},{}^{3}F_{4}^{3}H_{6} \\ {}^{3}H_{4}^{3}F_{4},{}^{3}F_{4}^{3}H_{6} \\ {}^{3}H_{4}^{3}F_{4},{}^{3}F_{4}^{3}H_{6} \\ {}^{3}H_{4}^{3}F_{4},{}^{3}F_{4}^{3}H_{6} \end{array}$	298-333 298-333 298-333 298-333	$I_{1470}/I_{1740} \\ I_{1470}/I_{1740} \\ I_{1470}/I_{1960} \\ I_{1470}/I_{1740}$	0.08/298 0.6/298 0.22/298 0.06/298	6.2* 0.8* 2.2* 8.3*	265 265 265 265
Tm ³⁺ ,Ho ³⁺ Tm ³⁺ ,Ho ³⁺	Tm^{3+} Yb^{3+}	KLu(WO ₄) ₂ KLu(WO ₄) ₂	808 980	1800 (Tm ³⁺), 1960 (Ho ³⁺) 1480 (Tm ³⁺), 1960 (Ho ³⁺)	$^{3}H_{4} \rightarrow ^{3}F_{4} (Tm^{3+}),$ $^{5}I_{7} \rightarrow ^{5}I_{8} (Ho^{3+})$ $^{3}H_{4} \rightarrow ^{3}F_{4} (Tm^{3+}),$ $^{5}I_{7} \rightarrow ^{5}I_{8} (Ho^{3+})$	293-333 298-333	I_{1800}/I_{1960} I_{1480}/I_{1960}	0.90/293 0.52/298	0.7 0.9*	266 265
Er^{3+} Er^{3+} Er^{3+} Er^{3+} Er^{3+} Er^{3+} Er^{3+} Er^{3+} Er^{3+}	Yb^{3+} Yb^{3+} Yb^{3+} Yb^{3+} Yb^{3+} Yb^{3+} Yb^{3+} Yb^{3+} Yb^{3+}	$\begin{array}{c} LuVO_4\\ LuVO_4 @SiO_2\\ NaY_2F_5O\\ BaMoO_4\\ BaMoO_4\\ KLu(WO_4)_2\\ Lu_2O_3\\ NaYF_4 \end{array}$	980 915 980 980 980 980 980 980	1637, 1660 1469, 1527 1535, 1554 1521, 1531 1504, 1531 1535, 1554 1535, 1554 1535, 1554	$ \begin{array}{c} ^4I_{13/2} \rightarrow \ ^4I_{15/2} \\ ^4I_{13/2} \rightarrow \ ^4I_{15/2} \end{array} $	298-523 298-523 298-333 293-553 293-553 298-333 298-333	FIR _{1637/1660} FIR _{1469/1527} FIR _{1535/1554} FIR _{1521/1531} FIR _{1504/1531} FIR _{1535/1554} FIR _{1535/1554} FIR _{1535/1554}	0.54/523 0.18/298 0.15/298 0.13/293 0.095/293 0.095/298 0.09/298 0.06/298	0.92* 2.7* 3.3* 3.8* 5.2* 5.2* 5.5* 8.3*	267 268 265 269 269 265 265 265

thermometers, the performance is entirely based on the temperature dependence of emissions arising from TCLs constituted by the Stark sublevels of the ${}^4I_{13/2} \rightarrow {}^4I_{15/2}$ transition, using Yb³⁺ as a sensitizer.

6.1. Tm³⁺-doped luminescent nanothermometers operating in the III-BW

Tm³⁺-doped luminescent thermometers operating in the III-BW were either excited at 980 nm or 808 nm, depending on whether Yb3+ has been used as sensitizer or not. When Yb3+ was used as a sensitizer, Tm³⁺-doped luminescent thermometers were excited at 980 nm (Table 5). These thermometers can be formed by nanoparticles doped with Tm³⁺ or co-doped with Ho³⁺, using Yb³⁺ as the ion that absorbed the energy of the excitation source. Upon 980 nm excitation, the luminescence spectrum of the Tm3+-doped nanoparticles in the III-BW was composed of two emissions at ~1450 nm and ~1800 nm, whereas when Ho³⁺ was also present, a third emission band at ~1960 nm was observed. Yb3+ ions acting as a sensitizer absorb the 980 nm excitation light, promoting their electrons from the ${}^{2}F_{7/2}$ ground stated to the ${}^{2}F_{5/2}$ excited state. Due to the energy resonance between this level of Yb³⁺ and the ³H₅ level of Tm³⁺, an ET process can take place, populating this level of Tm³⁺. From the ³H₅ level, electrons can relax nonradiatively to the ${}^{3}F_{4}$ level.

From this state, the energy of a second excited electron of Yb3+ is transferred to Tm3+ while its electrons are in the ³F₄ excited state, promoting them to the ³F_{2,3} state. From this state, a second non-radiative process can take place towards the ³H₄ energy level, from which the emission band located at 1450 nm is generated through the ${}^{3}\text{H}_{4} \rightarrow {}^{3}\text{F}_{4}$ electronic transition (Fig. 40(a)). Finally, a radiative relaxation of the electrons from the ³F₄ level to the ground state gives rise to the second emission of Tm³⁺ located at 1800 nm.²⁶⁵

With the addition of Ho3+, an additional ET process can take place from Yb3+ to the 5I6 level of Ho3+ that is resonant in energy with the ²F_{5/2} excited state of Yb³⁺. From it, the ⁵I₇ level of Ho³⁺ can be populated through a non-radiative relaxation process, prior to a radiative transition to the ground state (⁵I₈), generating the emission at 1960 nm. Another mechanism for the generation of the Ho3+ emission band is a direct ET process from the ${}^{3}H_{5}$ level of Tm^{3+} to the resonant ${}^{5}I_{6}$ level of Ho³⁺ (Fig. 40(a)). This mechanism, however, has a lower probability of happening due to the high concentration of Yb3+ in the nanoparticles, up to ten times higher than that of Tm³⁺ which makes more feasible the ET between Yb3+ and Ho3+.265

The addition of Ho3+ provided positive effects not only in the intensity of the luminescence of the whole spectrum, which was improved, but also in the temperature-sensing properties.

Hence, the presence of Ho³⁺ plays an important role in transferring additional energy towards Tm3+ ions through the ⁵I₆ and ³H₅ levels of Ho³⁺ and Tm³⁺, respectively (Fig. 40(a)), apart from that already transferred by Yb3+. The intensity ratio between the two emissions of Tm3+ ions was considered for luminescence thermometry purposes in Ho³⁺,Tm³⁺,Yb³⁺:KLu

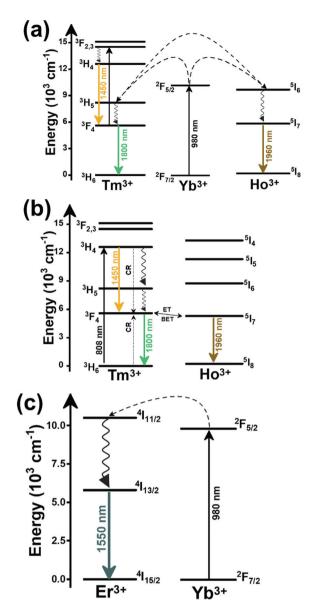


Fig. 40 Mechanism of the generation of the emission lines in: (a) Ho³⁺, $Tm^{3+}, Yb^{3+}: KLu(WO_4)_2$, (b) $Ho^{3+}, Tm^{3+}: KLu(WO_4)_2$, and (c) Er^{3+}, Yb^{3+} doped materials operating as luminescent thermometers in the III-BW.

 $(WO_4)_2$ nanoparticles, obtaining a two-fold higher S_{rel} in the nanoparticles containing Ho3+ when compared with those containing only Tm³⁺ (Table 5).

Besides the benefits demonstrated by this kind of luminescent thermometer, one thing that should be considered is the fact that they were excited at 980 nm, which would hamper their potential biomedical applications due to the absorption of this radiation by water and the subsequent heat generation.206 To surpass this drawback, the possibility of exciting these materials at 808 nm through the direct absorption of energy by Tm³⁺ was considered as a solution.

Hence, in the absence of Yb³⁺ ions, an illustrative example is Ho3+,Tm3+:KLu(WO4)2 nanocrystals. Upon excitation at 808 nm, Tm³⁺ promotes its electrons from the ³H₆ ground

state to the 3H4 excited state. The electrons decay radiatively to the ${}^{3}F_{4}$ manifold, generating the emission line at 1.45 μ m. From the ³F₄ level, a second radiative relaxation to the ³H₆ ground state generates the emission line at 1.8 µm. Tm³⁺ might undergo a CR process ${}^{3}H_{4}$, ${}^{3}H_{6} \rightarrow {}^{3}F_{4}$, ${}^{3}F_{4}$, which contributes to the higher intensity observed for the emission band at 1.8 µm. Also, due to the energy resonance between the ³F₄ level of Tm³⁺ and the ⁵I₇ level of Ho³⁺, ET and BET processes might take place between these electronic levels, promoting the electrons of Ho3+ to this excited state from the ground state. Then, the electrons of Ho³⁺ relax radiatively to the 5I8 ground state, giving rise to the emission band at 1.96 µm (Fig. 40(b)). 270 The temperature-sensing properties of Tm³⁺:KLu(WO₄)₂, Tm³⁺,Yb³⁺:KLu(WO₄)₂ and Ho³⁺,Tm³⁺:KLu (WO₄)₂ nanocrystals were compared upon excitation at 808 nm in the physiological range of temperatures. It was observed that when the temperature increased, the intensity of the 1450 nm band remained unchanged, while the intensity of the 1800 nm band decreased for Tm3+ and Tm3+,Yb3+ particles. Instead, for Tm3+,Ho3+ nanoparticles, the intensity of both the 1450 nm and 1960 nm bands decreased as the temperature increased, while the intensity of the 1800 nm band increased (Fig. 41(a)), indicating that the BET mechanism (Fig. 40(b)) between these two ions is promoted as the temperature increased.²⁶⁵

As a consequence, the $S_{\rm rel}$ of ${\rm Ho}^{3+}$, ${\rm Tm}^{3+}$: ${\rm KLu}({\rm WO}_4)_2$ nanoparticles is approximately 5 times higher than that of Tm³⁺: KLu(WO₄)₂ and Tm³⁺,Yb³⁺:KLu(WO₄)₂ nanoparticles, by considering either the 1450 nm/1800 nm or the 1800 nm/1960 nm intensity ratio, which provided the same S_{rel} since the electronic populations of the two emitting levels of Tm³⁺ from which the emissions at 1450 and 1800 nm are generated are linked through the CR process ${}^{3}H_{4}$, ${}^{3}H_{6} \rightarrow {}^{3}F_{4}$, ${}^{3}F_{4}$ (Fig. 41(b)). 266 Then, the effects of other parameters like the excitation power, the different lanthanide doping concentrations, and the choice between the 1450 nm/1800 nm, 1450 nm/1960 nm or 1800 nm/1960 nm intensity ratios were analyzed. The 1800 nm/1960 nm intensity ratio changed the most as the excitation power increased (Fig. 41(b)). The linear variation of these intensity ratios with the excitation power represents a very important point when envisaging the application of these luminescent nanothermometers in biomedical fields. During biomedical applications, there is no real control of the excitation power reaching the nanoparticles within the biological sample; hence this might lead to inaccuracy that can be reduced with this linear variation, while always maintaining it at a value low enough to avoid damage to the biological tissues. The optimal doping concentration to maximize the luminescence intensity was found at 3 at% of Ho³⁺ and 5 at% of Tm³⁺. 266 The doping concentration that allowed obtaining

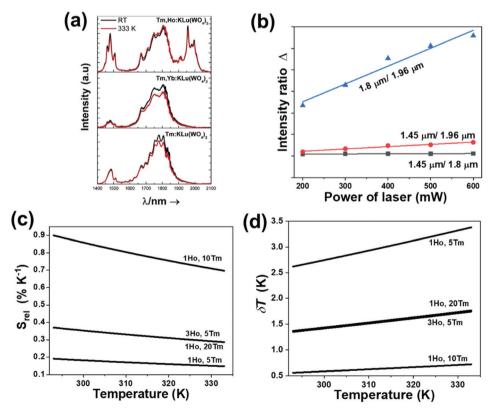


Fig. 41 (a) Effect of temperature in the III-BW emissions for the $KLu(WO_4)_2$ nanoparticles doped with Tm^{3+} ; Tm^{3+} , Yb^{3+} and Ho^{3+} , Tm^{3+} ions after pumping at 808 nm. Reprinted with permission from ref. 265. Copyright 2016, Elsevier. (b) Evolution of the intensity ratios in Ho^{3+} , Tm^{3+} : $KLu(WO_4)_2$ nanoparticles as a function of the excitation power. Effect of different lanthanide doping concentrations in Ho^{3+} , Tm^{3+} : $KLu(WO_4)_2$ nanoparticles on: (c) S_{rel} , and (d) temperature resolution. (b)–(d) Adapted with permission from ref. 266. Copyright 2020, The Royal Society of Chemistry.

the highest S_{rel} (0.90% K⁻¹), and subsequently the minimum δT (0.55 K at 293 K), was found at 1 at% of Ho³⁺ and 10 at% of Tm3+ (Fig. 41(c) and (d) and Table 5). Thus, different doping concentrations of Tm³⁺ and Ho³⁺ imply different efficiencies for the electronic population processes, especially those concerning the ET and BET processes between the ³F₄ level of Tm³⁺ and the ⁵I₇ level of Ho³⁺, and consequently different thermometric performances.²⁶⁶ We underline as an important conclusion from this section the combination of Tm³⁺ and Ho³⁺ emissions to generate highly sensitive thermometers.

6.2. Er³⁺-doped luminescent nanothermometers operating in the III-BW

Er3+ doped luminescent nanothermometers operating in the III-BW exhibit, in general, a very low $S_{\rm rel}$ (Table 5). Their performance, based entirely on the emissions generated from the different Stark sublevels of the ${}^4I_{13/2} \rightarrow {}^4I_{15/2}$ transition, has been evaluated through the FIR model (eqn (9)), being strictly related to the ΔE between the Stark sublevels involved in the generation of the emissions in which these luminescent thermometers are based (50-130 cm⁻¹), which in general are

Er³⁺-doped luminescent thermometers operating in the III-BW used Yb³⁺ as a sensitizer, which implies excitation at 980 nm. The mechanism of generation of this emission band is depicted in Fig. 40(c). After the absorption of a photon at 980 nm by Yb^{3+} , its electrons are promoted to the ${}^{2}F_{5/2}$ level. Then, an ET to Er³⁺ occurs, populating its ⁴I_{11/2} level. After a multiphonon relaxation to the 4I13/2 energy level followed by a radiative decay to the ground state through the ${}^4I_{13/2} \rightarrow {}^4I_{15/2}$ transition, the emission centered at 1550 nm is generated. 265,267,268

The intensity of the emission generated and the thermometric performance of different Er³⁺-doped luminescent thermometers operating the III-BW was analyzed by Savchuk et al. as a function of different hosts, including oxyfluorides (NaY₂F₅O), fluorides (NaYF₄), simple oxides (Lu₂O₃) and complex oxides (KLu(WO₄)₂).²⁶⁵ In terms of intensity, the fluoride host exhibited the highest one, mainly attributed to its low phonon energy (Fig. 42(a)). Comparing their temperaturesensing properties in the physiological range of temperatures, calculated from eqn (9), the oxyfluoride host (NaY2F5O) exhibited the highest S_{rel} , with a value of 0.15% K⁻¹ at room temperature, almost 3 times higher than that obtained for the other hosts,265

In another study, in which BaMoO4 was studied as a host for Er³⁺, the authors demonstrated that the performance of these Er3+-doped luminescent thermometers is influenced by

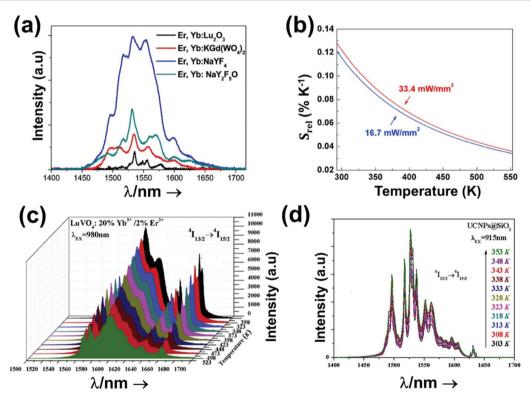


Fig. 42 (a) Intensity of the Er³⁺ 1550 nm emission in different hosts after excitation at 980 nm. Reprinted with permission from ref. 265. Copyright Elsevier. (b) Effect of the excitation power density at 980 nm in S_{rel} of Er³⁺, Yb³⁺:BaMoO₄ nanoparticles. Reprinted with permission from ref. 269. Copyright 2018, Elsevier. (c) Temperature dependence of the intensity of the Er3+ 1550 nm emission in Er3+, Yb3+: LuVO4 nanoparticles, excited at 980 nm. Reprinted with permission from ref. 267. Copyright 2018, Elsevier. (d) Temperature dependence of the intensity of the Er³⁺ 1550 nm emission in LuVO4@SiO2 core-shell nanoparticles, excited at 915 nm. Reprinted with permission from ref. 268. Copyright 2019, American Chemical Society.

the excitation power density.²⁶⁹ The authors tested two different intensity ratios from the Stark sublevels of the 4I_{13/2} \rightarrow $^{4}I_{15/2}$ transition (1504 nm/1531 nm and 1521 nm/1531 nm) under two different excitation power densities (16.7 mW mm⁻² and 33.4 mW mm⁻²). The 1504 nm/1531 nm intensity ratio was almost insensitive to the variation of the excitation power density, whereas the 1521 nm/1531 nm intensity ratio allowed a slightly higher S_{rel} to be obtained at higher excitation power densities (Fig. 42(b)).

All the thermometers mentioned above calculated their performance based on the Boltzmann model; however, in some cases another phenomenological fitting model was applied. This is the case for Er^{3+} , Yb^{3+} : $LuVO_4^{267}$ and Er^{3+} , Yb^{3+} : LuVO₄@SiO₂ nanoparticles, ²⁶⁸ where the thermometric parameter was fitted to the phenomenological equation:

$$\Delta = B \exp(a + b + cT^2) \tag{50}$$

where B, a, b and c are all constants to be determined by the fitting.

According to that, S_{rel} could be calculated as:

$$S_{\text{rel}} = |b + cT| \times 100\% \tag{51}$$

The maximum S_{rel} (0.54% K⁻¹) was obtained for the uncoated Er3+,Yb3+:LuVO4 nanoparticles, at the highest temperature under study (523 K). 267 For the silica-coated nanoparticles, a lower $S_{\rm rel}$ (~0.18% K⁻¹) was obtained at room temperature, although they were excited at a different wavelength.268

An additional difference between these two materials was observed in the evolution of the intensity of the emission as the temperature increased. For the uncoated nanoparticles, excited at 980 nm, the intensity of the emission decreased as the temperature increased (Fig. 42(c)), while for the SiO₂coated nanoparticles a general increase of the intensity of the emission band as the temperature increased was observed (Fig. 42(d)), assigned, according to the authors, to the improvement of the multiphonon relaxation process from the 4I_{11/2} level to the ⁴I_{13/2} level as the temperature increased.

A final remark for this subsection should be devoted to the fact that the excitation wavelength used for these luminescent thermometers has the drawback of causing overheating, hampering their use in biomedical applications. 206 Even though in Er3+,Yb3+:LuVO4@SiO2 core-shell nanoparticles excitation at 915 nm was applied to overcome this problem, the absorption cross section of Yb3+ at this wavelength is only half of that at 980 nm, which generates weaker emissions and might make their practical use as luminescent thermometers difficult.²⁷¹

In summary, the variation of S_{rel} with the increase of temperature for the luminescent thermometers operating in the III-BW is presented in Fig. 43(a) and (b). In general, for these thermometers a decrease of $S_{\rm rel}$ is observed as the temperature increases, with the exception of the thermometer based on Er³⁺,Yb³⁺:LuVO₄ nanoparticles, due to the different model to which the experimental data were fitted. In addition, the S_{rel} of the Er³⁺-doped luminescent thermometers operating within

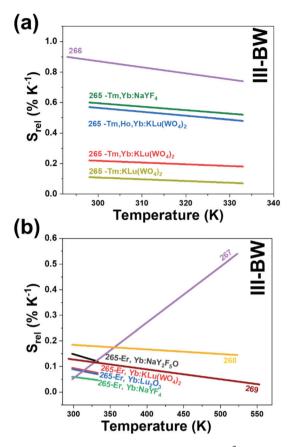


Fig. 43 Temperature dependence of S_{rel} of: (a) Tm^{3+} - and (b) Er^{3+} doped thermometers operating in the III-BW. The numbers indicate the corresponding references for each thermometer.

the III-BW is relatively low compared with the Tm3+-doped ones. When compared with the luminescent thermometers operating in the other BWs, the relative thermal sensitivities of the thermometers operating in the III-BW are lower, implying that additional strategies to boost their performances need to be developed.

Applications of lanthanide-doped luminescent nanothermometers operating within the BWs

The development of highly sensitive lanthanide-doped luminescent nanothermometers by means of tunable synthetic strategies and the combination of different emission bands has accelerated their application as potential tools for thermal sensing in ex vivo, in vitro and in vivo trials, including 2D subcutaneous dynamic thermal imaging as well as controlled photothermal therapy experiments. This section covers most of the examples published up to now devoted to the applications of these Ln3+-doped luminescent nanothermometers operating within the different BW spectral ranges. It concludes with some other applications, distinct from those in the bio-

medical/biological fields in which these luminescent thermometers have also been used.

7.1. Biological/biomedical applications of lanthanide-doped luminescent nanothermometers operating within the BWs

7.1.1. Ex vivo thermometry using lanthanide-doped luminescent nanothermometers operating in the BWs. Lanthanidedoped luminescent nanothermometers, upon suitable excitation and analysis of their emissions, can sense the temperature within biological tissues. The temperature-dependence properties of the luminescence of the Ln³⁺-doped materials allows their applicability as nanothermometers, despite the fact that a recalibration is required when they are embedded in a new medium, 13,47,214,259 which in most cases is neglected. Here, we cover the examples published in the literature of Ln³⁺-doped luminescence nanothermometers acting as thermal sensing agents in ex vivo experiments. Ex vivo refers to experimentation or measurements done in or on tissue from a biological organism in an external environment with minimal alteration of its natural conditions.272 The luminescent thermometers most used in biological/biomedical applications are those based on Nd3+. This ion can be excited with costeffective NIR laser diodes with emissions in the I-BW (790 nm and 808 nm). Among these, 808 nm has been demonstrated to be a risk-free wavelength for biological applications. 199,273 Additionally, all the characteristic emission bands of Nd³⁺, located at ~890 nm, ~1060 nm and ~1350 nm, lie within the BWs, respectively in the I-, II- and III-BWs. Hence, excitation and emissions within the BWs will allow higher penetration depths and deeper temperature sensing to be achieved.

Benayas et al. 171 reported the ability of Nd3+:YAG luminescent nanothermometers for ex vivo temperature monitoring, based on their emissions located in the I-BW (Table 1). The temperature-sensing properties of these nanothermometers were based on the emission lines from different Stark sublevels of the ${}^4F_{3/2} \rightarrow {}^4I_{11/2}$ transition, located at 938 nm and 945 nm, after pumping at 800 nm with a power of 100 mW. 100 μL of a 0.1 wt% water dispersion of Nd³⁺:YAG nanoparticles were injected into chicken breast tissue at a depth of 5 mm. The biological tissue was externally heated using a hot air flow (343 K) for 90 s, prior to letting it to cool down to room temperature naturally, as illustrated in Fig. 44(a). The luminescence and the intensity ratio of the particles during the heating/cooling cycles was monitored for a period of more than 8 min. The time evolution of the intensity ratio between the two emission lines and the corresponding subtissue temperature determined from it revealed that during the heating process a linear increase of the subtissue temperature with time was produced, reaching a maximum temperature change of about 328 K (Fig. 44(b)).¹⁷¹

When the hot air flow was turned off, an exponential decrease of the temperature due to thermal diffusion was observed (Fig. 44(b)).

Based on heating-cooling cycle dynamics, Skripka et al. used water-dispersible Nd3+:LiLuF4@LiLuF4 nanocrystals as ex vivo luminescent temperature sensors. 164 These nanocrys-

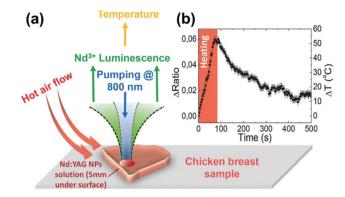


Fig. 44 Ex vivo temperature sensing in the I-BW with Nd3+:YAG nanoparticles. (a) Scheme of the experimental procedure used to measure the time evolution of the subtissue temperature of a chicken breast sample in which an aqueous dispersion of the luminescent nanothermometers was injected, externally heated by a flow of hot air. (b) Time evolution of the subtissue temperature determined from the ratio between the intensities of the 938 nm and 945 nm emission bands of Nd3+. Reprinted with permission from ref. 171. Copyright 2015, Wiley-VCH Verlag GmbH & Co. KGaA.

tals, in principle, can work as ex vivo thermal sensors in the three BWs (they generate emissions located at 880 nm, 1050 nm and 1320 nm). However, the authors selected the emission band located in the II-BW (${}^{4}F_{3/2} \rightarrow {}^{4}I_{11/2}$ transition, 1050 nm) due to its higher intensity. The heating-cooling cycles generated in a pork fat tissue sample when a 980 nm laser (power density ~37 W cm⁻²) was turned on and off were measured with an aqueous dispersion of these Nd³⁺: LiLuF₄@LiLuF₄ nanocrystals with a concentration of 5 mg mL^{-1} as a function of the tissue thickness (Fig. 45(a)). To record the emission generated by the nanocrystals they were continuously excited at 793 nm with a power density ~49 W cm⁻² for a period of 12 min. The recorded photoluminescence and the calculated intensity ratio based on the different emission lines arising from the different Stark sublevels that generate the 1050 nm emission of Nd3+ were used to determine the temperature. A temperature increase was induced in the pork fat tissue sample by turning ON the 980 nm laser after 2 min from the beginning of the experiment (2 min after the 793 nm laser was turned ON to excite the luminescent nanoparticles) and heating the sample for 5 min, after which it was switched OFF (Fig. 45(a)). After switching OFF the 980 nm laser, the cooling profile was also monitored through the emission generated by the nanoparticles over the next 5 min with the 793 nm laser still ON. In addition, a thermocouple was used to verify the temperature measurements. Based on these heatingcooling cycles, the heat dissipation time constant of the nanoparticles in the surrounding medium, δT , and the temporal uncertainty of the luminescent nanothermometers were determined at different biological tissue thicknesses. The results revealed that as the biological tissue becomes thicker, the values for the heat dissipation time constant determined from the spectra collected with the nanoparticles began to deviate from those measured with the thermocouple. 164 Also, a signifi-

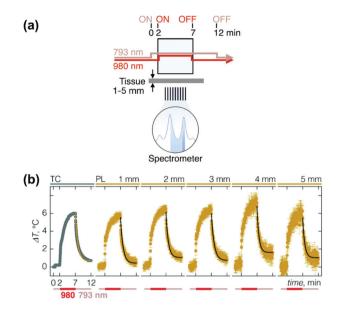


Fig. 45 (a) Scheme of the heating-cooling experiment measurements in pork fat tissue. (b) Heating-cooling cycles measured with an aqueous dispersion of Nd:LiLuF4@LiLuF4 nanocrystals measured with a thermocouple (TC) and via the intensity ratio among the emissions generated by the different Stark sublevels of the 1050 nm emission of Nd³⁺, as a function of the pork fat tissue thickness. Reprinted with permission from ref. 164. Copyright 2019, The Royal Society of Chemistry.

cant error was observed in the determination of the temperature difference induced solely by the 980 nm light, and calculated between the OFF point of the 980 nm laser and the end of the cycle, attributed to the photoluminescence extinction of the biological tissue. With the increase of the thickness of the biological tissue from 1 to 5 mm, the δT of the luminescent thermometer changed from 0.19 to 1.23 K, while the temporal uncertainty changed from 0.52 ± 0.37 to 4.26 ± 6.60 s. ¹⁶⁴ As a conclusion, and as can be observed in Fig. 45(b), the reliability of these nanothermometers is highly compromised for a tissue thickness above 2 mm.

Another example of a luminescent nanothermometer applied in an ex vivo experiment is that of water-dispersible hybrid Nd3+:NaGdF4 nanoparticles and the semiconductor PbS/CdS/ZnS QDs in PGLA, operating in the II-BW using the intensity ratio between the 1060 nm emission of Nd3+ and the 1220 nm emission of QDs (Table 3).⁶³ The ability of these particles to sense the temperature in biological tissues was tested by injecting subcutaneously 100 μL of a 1 mg mL⁻¹ dispersion of these particles in distilled water into a chicken breast sample at a depth of ~2 mm. The biological tissue was externally heated using a hot air current at 323 K for 20 s, prior to letting it to cool down to room temperature naturally. The emissions of the Nd3+:NaGdF4 nanoparticles and the QDs, generated within the tissue after excitation at 808 nm with a 1 W cm⁻² power density, were collected by an optical fiber and spectrally analyzed through a high-resolution spectrometer (Fig. 46(a)). The temperature profile of these emissions displayed a different behavior during heating/cooling cycles

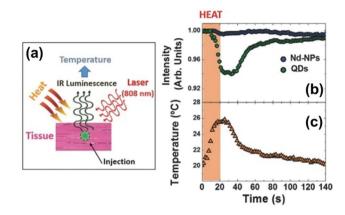


Fig. 46 Ex vivo temperature sensing in the II-BW with hybrid Nd³⁺: NaGdF₄ and QDs in PGLA particles. (a) Scheme of the experimental procedure. (b) Evolution of the intensity of the 1060 nm (Nd3+) and 1220 nm (QDs) emissions, normalized to their room temperature values, of the hybrid particles injected into the chicken breast tissue during a heating/cooling cycle. (c) Sub-tissue temperature evolution during the heating/cooling cycle determined from the intensity ratio of the emissions of the hybrid Nd3+:NaGdF4 and QDs in PGLA particles. Reprinted with permission from ref. 63. Copyright 2015, Wiley-VCH Verlag GmbH & Co. KGaA.

(Fig. 46(b)). During the heating cycle (shadowed area), the emission of Nd3+ remained unchanged, while the emission of the QDs drastically decreased due to thermal quenching effects.⁶³ During the cooling process, the emission of the QDs recovered its initial value. The variation of temperature of the tissue during the heating/cooling cycles was determined by calculating the intensity ratio between these two emissions (Fig. 46(c)). The results showed a temperature increase of approximately 6 K at the end of the heating pulse. 63 During the cooling cycle, a pseudo-exponential decreasing profile was observed, monitored for two minutes until the initial room temperature was recovered. From the magnitude of the random fluctuations in the temperature profile (Fig. 46(c)), δT was estimated to be 0.2 K (Table 3).63

Ex vivo lanthanide nanothermometers operating in the III-BW have been also reported. Savchuk et al. investigated the intensity ratio between the 1470 nm and 1711 nm emissions of Tm³⁺ in Ho,Tm:KLuW nanoparticles to sense the temperature inside a chicken breast sample at a depth of 2 mm, after being excited at 808 nm with a power of 200 mW.265 For the experiment, the nanoparticles were deposited on a microscope glass slide, on top of which the slice of chicken breast was placed. A heating gun, with a power of 1 W and fixed to a horizontal moving stage to control its movement, was used to induce the heating of the chicken breast sample. The heating gun was moved away from its initial position, close to the chicken breast sample, to generate a decrease of temperature in the biological tissue. The maximum temperature reached in the chicken breast sample was maintained below 318 K to avoid exceeding the denaturalization temperature of biological tissues. The excitation beam was focused on the nanoparticles by using a 40× microscope objective with a numerical aperture

(hereafter N.A.) of 0.6. The photoluminescence generated by the Ho,Tm:KLuW nanoparticles was collected with a Yokogawa AQ6375 optical spectrum analyzer and the temperature was determined through the intensity ratio indicated above. In addition, close to the nanoparticles, and inside the chicken breast sample, a small Pt-100 thermocouple was placed to verify the changes of temperature (the schematic representation of the experimental setup is in Fig. 47(a)). The temperature profile resulting from moving the heating gun away from the chicken breast sample is shown in Fig. 47(b), showing a temperature drop of 1.5 K cm⁻¹. The temperature determined by the nanoparticles displayed a difference of 0.8 K with respect to the one determined by the Pt-100 thermocouple.²⁶⁵ This difference was assigned to the different thermal conductivity of the two materials (nanoparticles and thermocouple) or also to the different location of the thermal probes within the chicken breast sample.265

7.1.2. Ex vivo photothermal experiments combined with luminescence thermometry using lanthanide-doped luminescent nanothermometers operating in the BWs. Lanthanidedoped luminescent nanothermometers, upon laser irradiation, not only can sense temperature, but they can also generate heat under special conditions. These two processes are generated due to the peculiar electronic configuration of the Ln³⁺ ions, which gives rise to radiative and non-radiative processes, upon the proper selection of the excitation source. These two

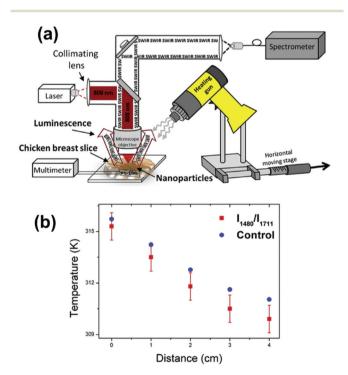


Fig. 47 Ex vivo temperature sensing in the III-BW with Ho, Tm:KLu $(WO_4)_2$ nanoparticles. (a) Schematic representation of the experimental setup. (b) Temperature profile determined from the intensity ratio of the 1480 nm and 1711 nm emission bands of Tm³⁺ in Ho,Tm:KLu(WO₄)₂ particles (red squares) and by a Pt-100 thermocouple (blue spheres). Reprinted with permission from ref. 265. Copyright 2018, Elsevier.

processes are the main ones responsible for the production of luminescence and heat in Ln3+-doped materials, respectively. Thus, Ln3+-doped materials have the potential of being implemented as photothermal agents within biological/biomedical media.

Photothermal therapy, which employs light-absorbing agents to convert photoenergy into heat to achieve local hyperthermia, is regarded as a minimally invasive and highly efficient method for targeted cancer treatment. 4,49,274 Typical conditions that have to be fulfilled by a material to act as a potential photothermal agent include high light-to-heat conversion efficiency, nanometric sizes, excitation and emission within the BW spectral ranges, and real-time temperature feedback.²⁷⁵ Here, we cover the examples published in the literature which involve the application of Ln³⁺-doped luminescence as nanothermometers combined with materials that can generate heat, and the so-called self-assessed Ln³⁺-doped nanothermometers, which can simultaneously sense the temperature but also generate heat within the medium where they are embedded.

Nd³⁺,Y³⁺:CaF₂ nanoparticles, combined with Au nanorods, have been used in ex vivo photothermal experiments.245 The nanoparticles that could sense the temperature $(Nd^{3+}, Y^{3+};$ CaF₂) were combined with nanoparticles that could generate heat (Au nanorods). It must be emphasized that the Nd3+doped particles alone would have also the ability to generate heat upon NIR excitation due to their ladder-like electronic configuration. 15,199,204 In fact, the heat generated by the Nd 3+doped nanoparticles is a consequence of both the increase of the absorbed pump power when irradiated at 808 nm (since the absorption coefficient of these nanoparticles is proportional to the Nd³⁺ concentration), and the decrease of the fluorescence quantum yield due to Nd3+-Nd3+ interactions (such as cross relaxations and energy migrations when the distance among these ions is short enough). Therefore, it would be possible to tailor the balance between light and heat generation through an appropriate choice of the Nd3+ concentration inside the nanoparticles. Nevertheless, the authors chose Au nanorods as photothermal agents because the use of Nd³⁺-doped nanoparticles would require higher illumination doses to reach the targeted temperature. 245,276 Au nanorods, upon excitation on their longitudinal plasmon mode, can efficiently generate heat.4,277 Hence, nanorods with a local surface plasmon resonance maximum at ~810 nm were used. Concerning the Nd³⁺,Y³⁺:CaF₂ luminescence temperature sensors, they operated in the II-BW using different emission lines of the band located at 1050 nm through the FIR model (Table 3). For the ex vivo experiment, 1 mol% Nd³⁺ and 10 mol% Y³⁺ co-doped CaF₂ nanoparticles were chosen due to their S_{rel} and good emission intensity. A solution of these nanoparticles with a concentration of 1.6 mg mL⁻¹ was combined with a solution of the Au nanorods with a concentration of 1.3 \times 10⁻³ mM.²⁴⁵ A TEM image of the combined nanoparticle dispersion is in Fig. 48(a). Chicken breast samples with different thicknesses were placed on the top of the cuvette containing the nanoparticle suspension. On the

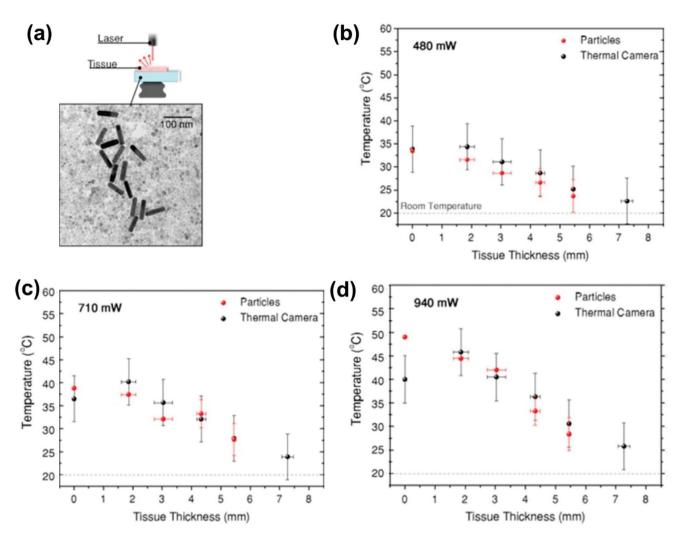


Fig. 48 (a) Ex vivo photothermal experiment in chicken breast using a colloidal solution containing gold nanorods as photothermal agents and Nd³⁺,Y³⁺:CaF₂ nanoparticles as luminescent thermometers (a TEM image of the mixed colloid is presented as well). Temperature determined by using the intensity ratio of the Stark sublevels of the 1050 nm emission of Nd3+ (in red) in the Nd3+, V3+: CaF2 nanoparticles, and by a thermal camera (in black) as a function of the power of the 808 nm laser applied: (b) 480 mW, (c) 710 mW and (d) 940 mW. Adapted with permission from ref. 245. Copyright 2018, American Chemical Society.

bottom of the cuvette, a thermal camera was used to monitor also the generation of heat (Fig. 48(a)). The luminescence generated by the Nd3+,Y3+:CaF2 nanoparticles, after 808 nm excitation, was recorded by an InGaAs spectrometer, and then the temperature was determined by calculating the corresponding intensity ratio. For the experiment, the power of the excitation laser was maintained below 0.94 W to avoid any overheating in the biological tissue. Also, a low concentration of Nd3+ in the nanoparticles was used to avoid the generation of heat by them; thus, the generated heat could be only attributed to the Au nanorods.²⁴⁵ Under these conditions, the temperature read by the thermal camera and the temperature determined by the intensity ratio between the emission lines coming from different Stark sublevels of the emitting state in the 1050 nm emission (${}^{4}F_{3/2} \rightarrow {}^{4}I_{11/2}$ transition) of the Nd ${}^{3+}$,Y ${}^{3+}$:CaF₂ nanoparticles were compared as a function of the power of the laser applied (Fig. 48(b)-(d)), showing a good agreement for thick-

nesses of the chicken breast below 5.5 mm. 245 For a thickness of 2 mm, a δT of 0.2 K was obtained, fulfilling the requirements for a nanothermometer to be operative in biomedical applications. This δT was determined using eqn (3) after recalibrating the luminescent nanothermometer in the new aqueous dispersion medium of the mixture of the Nd3+,Y3+: CaF_2 and Au nanoparticles.²⁴⁵ Above this thickness, δT increased, reaching a value of 3.5 K for a thickness of 5.5 mm, clearly jeopardizing the reliability of this luminescent thermometer, and indicating that a new recalibration of the luminescence thermometer is required, probably inside the chicken breast.

Another reported ex vivo photothermal experiment is the one in which so-called self-assessed photothermal agents were used.²⁶⁶ These self-assessed photothermal agents are materials that can simultaneously generate heat and sense the temperature. Here, these self-assessed photothermal agents were com-

posed of only Ln3+-doped nanoparticles, which were the responsible for sensing the temperature and generating heat at the same time. As an example, we can mention Ho³⁺,Tm³⁺:KLu (WO₄)₂ nanoparticles, ²⁶⁶ operating as luminescent nanothermometers in the III-BW, based on the intensity ratio between the 1800 nm emission of Tm³⁺ and the 1960 nm emission of Ho³⁺, that generated heat simultaneously due to the existence of non-radiative processes within the relaxations between the different electronic levels and non-resonant ET processes between Ln3+ ions (Fig. 49(a)). In the ex vivo experiment, the nanoparticles were placed in between two pieces of chicken breast, ensuring a complete medium homogeneity, with a maximum thickness of 2 mm. Close to the nanoparticles, a thermocouple was also located to verify the temperature determined. Upon excitation at 808 nm, a fast increase of temperature during the first 100 s was observed by monitoring the temperature with a thermocouple, prior to reaching a plateau, produced only by the Ho³⁺,Tm³⁺:KLu(WO₄)₂ (Fig. 49(b)). At this point, the temperature was determined through the intensity ratio of the emissions generated also by the luminescent nanoparticles (inset of Fig. 49(b)) and compared with the

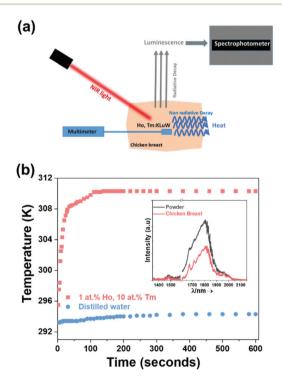


Fig. 49 (a) Schematic representation of the experimental procedure for ex vivo temperature determination in chicken breast using self-assessed 1 at% Ho, 10 at% Tm:KLuW photothermal agents that allow the generation of heat and the measurement of temperature at the same time, with the luminescent thermometer operating in the III-BW. (b) Timedependent temperature profiles of distilled water and aqueous dispersion of 1 at% Ho, 10 at% Tm:KLuW nanocrystals when illuminated with the 808 nm laser. The inset stands for the emissions in the III-BW of 1 at% Ho, 10 at% Tm:KLuW nanocrystals as powder in air and covered by a 2 mm thick chicken breast piece of meat. Adapted with permission from ref. 266. Copyright 2020, The Royal Society of Chemistry.

temperature measured with the thermocouple. The results showed a difference of up to 0.8 K between these two temperatures, probably due to the different thermal conductivities of the nanoparticles (a dielectric material) and the thermocouple (a metal), and the different medium in which the nanoparticles were embedded during the calibration procedure (air) and during the experiment (chicken breast). Hence, this ex vivo experiment demonstrated that Ho³⁺,Tm³⁺:KLuW nanoparticles could be used as self-assessed photothermal conversion agents.

7.1.3. In vitro thermometry using lanthanide-doped luminescent nanothermometers operating in the BWs. Lanthanidedoped luminescent nanothermometers have also been used in in vitro media. In vitro stands for investigation performed with microorganisms, cells, or biological molecules outside their normal biological context.²⁷⁸ In vitro nanothermometry (also known as cellular nanothermometry^{3,16,55,279}) has been explored principally in the visible spectral region because of the sufficient penetration depth achieved with VIS light in in vitro cell culture to analyze the whole thickness of the sample, and also because this light can be easily visualized using conventional optical microscopes, which produces an easier location of the nanoparticles and recording of their luminescence spectra within the cell culture. Typical examples of lanthanide-doped in vitro luminescent nanothermometers operating within the VIS involve Er3+,Yb3+-co-doped nanoparticles particles pumped in the NIR region, ^{22,280} and Eu³⁺ complexes with organic molecules excited with UV light. 62,281,282 Clearly, these examples are out of the scope of this review. For a more comprehensive understanding of in vitro luminescent nanothermometry operating in the VIS region, the reader is referred to other references. 279

By working with luminescent thermometers operating within the BW spectral regions, higher thermal sensitivities can be achieved when compared with those operating in the VIS, and thermal maps with higher spatial resolution can be recorded, due to the lower scattering at these wavelengths.²⁷⁹ Nevertheless, the number of Ln³⁺-doped nanoparticles operating within the BWs used as in vitro luminescent nanothermometers is very limited. In fact, we could identify only one in vitro luminescent nanothermometer based on a hybrid nanostructure that combines plasmonic Au nanoparticles as an efficient optical heater with Nd3+,Y3+:CaF2 nanoparticles acting as luminescent thermometers.276 These materials can both be excited with the same wavelength, 808 nm, located in the I-BW, while the emissions of the lanthanide nanoparticles used to determine the temperature are located within the II-BW. Hence, both functionalities (heating and temperature sensing) can be triggered by using a single excitation source. The luminescent thermometer constituted by the Nd^{3+}, Y^{3+} : CaF₂ nanoparticles was based on the FIR of different emission lines generated by different Stark sublevels of the ${}^4F_{3/2} \rightarrow {}^4I_{11/2}$ transition of Nd3+, located at ~1050 nm. Y3+ ions were intentionally added to break the energy migration paths between Nd³⁺ ions, which would otherwise quench the emission intensity. Concerning the nanoheater, the authors chose Au nano-

stars, which exhibited a higher light-to-heat conversion capacity when compared with the Au nanorods used previously in ex vivo thermometry (section 7.1.1).²⁷⁷ To properly prepare the combination of these nanoparticles for in vitro luminescence nanothermometry, the Nd3+,Y3+:CaF2 nanoparticles and Au nanostars were co-assembled into hybrid structures by using polysterene beads with diameters ~500 nm as the colloidal support that were finally coated with a silica shell (~7 nm thick) to prevent their dissociation when used in biological environments.²⁷⁶ TEM images of the hybrid structures can be seen in Fig. 50(a) and (b). Finally, these hybrids were internalized into a 3D spheroid tumor cell model prepared using a U-87MG human glioblastoma cell line. The hybrids, dispersed in water with a concentration of 100 μg mL⁻¹, were added to the cell culture medium and incubated for 48 hours at room temperature to allow internalization. The location of the hybrid particles inside the 3D tumor cells could be monitored through the observation of the Nd³⁺ luminescence, using a modified Lightsheet microscope to properly record the emission band at 1050 nm. The emission of Nd³⁺ (presented in a red color) was observed surrounding the DAPI (4,6-diamidino-2-phenylindole, used to label cell nuclei) emission, indicating that the hybrid beads were located in the cell cytoplasm (Fig. 50(c)).

The ability of these hybrids to emit light and generate heat was monitored in the cell medium, applying the experimental procedure schematically shown in Fig. 50(d) with an aqueous dispersion of 0.02 wt% of these hybrids. The hybrid particles were illuminated at 808 nm with a fiber-coupled laser diode using different powers and the beam was collimated to a 3 mm diameter spot. The emissions were collected with a lens, filtered with a longpass filter cut-off at 850 nm to remove the laser signal, and finally recorded by an InGaAs spectrometer. The generated heat was monitored also by a thermal camera. In the in vitro experiment, the local temperature inside the 3D spheroids tumor cells was investigated as a function of the different concentrations of the hybrids and different excitation power densities, during a period of time of 10 min to maintain the spheroids alive.²⁷⁶ The recorded temperature data, after transforming each of the intensity ratios of the emissions of the Nd³⁺,Y³⁺:CaF₂ nanoparticles into temperatures, are shown in Fig. 50(e). The temperature increased fast initially until thermal equilibrium was reached at around 100 s. Then, it remained stable for all the samples, with the only exception of the hybrid nanoparticles excited at 31 W cm⁻² and loaded with a concentration of 70 µg mL⁻¹, for which, according to the authors, the boiling point of water was likely reached prior to cooling due to the expansion and death of the 3D spheroids. δT was found to be between 3 K and 4 K, although it could reach up to 9 K for the sample with the lowest concentration of the hybrids, excited at the lowest power density.²⁷⁶

The average temperature at equilibrium increased proportionally to the increase in the concentration of the hybrids and the laser power density (Fig. 50(f)). It was also observed that the temperature measured by the hybrid particles was higher than that recorded by the thermal camera, in agreement with

the fact that the particles measure the temperature in the local proximity of the Au nanostars, while the thermal camera monitors the temperature at the surface of the cell medium. A final cell viability test (Fig. 50(g)) revealed that when the local temperature surpassed 328 K, the cells were almost completely dead, with a viability below 1% for a treatment reaching 334 K, and 0.3% above this temperature.²⁷⁶

It is important to note here that this luminescent thermometer was based on the FIR model. According to this model, and eqn (9) that is used to fit the experimental data, ΔE , *i.e.* the energy gap between the electronic energy levels from which the emissions are generated (in this case the different Stark sublevels of the $^4F_{3/2}$ and $^4I_{11/2}$ levels of Nd $^{3+}$), is independent of the external environment due to shielding of the 4f orbitals by the more external 5s orbitals of the Ln $^{3+}$ ions.

Nevertheless, the pre-exponential factor B can be understood as a correction parameter related to detected differences in the intensity of the emissions when embedded in different media. According to that, the intensity of the emission of the Nd³⁺,Y³⁺:CaF₂ nanoparticles, combined in the hybrid structures, can change when located in air or immersed in the tumor cell culture. To correct the value of the B parameter in eqn (9), fortunately a full re-calibration was not needed. Instead it was sufficient to record several spectra at different excitation powers inside the cells. 214 By doing so, a linear fit to the data allowed a calculation of the intensity ratio at zero power, which can be assumed to be the value of FIR at room temperature. 214,276 By introducing this correction, the results obtained were in agreement with the temperature measured by the thermal camera (differences below 1.5 K).²⁷⁶ Despite the positive results obtained in this in vitro experiment, there are, however, some aspects to be improved. One would be the size of the hybrid particles used, with a diameter of 830 \pm 30 nm, clearly above the desired dimensions of nanometers in the biomedical field.¹³ Additionally, the number of the lanthanidedoped temperature sensors per Au nanoparticle should be increased, to favor a higher intensity of the emission of Nd³⁺ in order to improve the signal-to-noise ratio and to reduce the time required to record the spectra. Also the distance between the Au nanoparticles and the Ln³⁺ ions should be larger than 5 nm, in order to avoid quenching of the Nd3+ emissions. Another issue to take into consideration is the low S_{rel} of these luminescent thermometers, 0.18% K⁻¹ at room temperature (Table 3).

7.1.4. *In vivo* thermometry using lanthanide-doped luminescent nanothermometers operating in the BWs. Lanthanide-doped luminescent nanothermometers can be used also in *in vivo* media. *In vivo* stands for investigations performed on whole living organisms, ²⁸³ basically mice in this review.

Nd³⁺:LaF₃ nanoparticles have been used as a luminescent nanothermometer for *in vivo* applications operating in the I-BW. In addition, these nanoparticles can release heat under excitation at 808 nm at a power density of 4 W cm⁻² when the doping level inside the nanoparticles is high enough, such as 5.6 at%. As described in section 7.1.1, it is possible to tailor the balance between light and heat generation through an

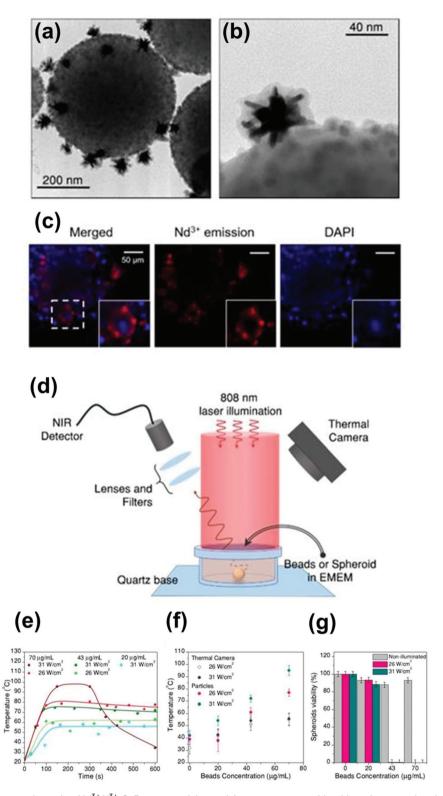


Fig. 50 In vitro temperature sensing using Nd³⁺,Y³⁺:CaF₂ nanoparticles and Au nanostars, combined in polysterene beads, and coated with a silica shell, forming a hybrid structure. (a) and (b) TEM images of the hybrid nanostructures. (c) Images of treated 3D spheroid tumor cells, showing the Nd³⁺ emission apparently located in the cytoplasm. (d) Schematic representation of the experimental procedure used to determine the temperature inside the 3D spheroid tumor cells. (e) Temperature measurements inside the spheroids during photothermal treatment as a function of time, using different excitation powers and different concentrations of the hybrid beads. (f) Average temperature at thermal equilibrium compared with the temperature determined by the thermal camera. (g) Viability of the 3D spheroid tumor cells 24 h after photothermal treatment. Adapted with permission from ref. 276. Copyright 2019, lvySpring International Publisher.

appropriate choice of the Nd³⁺ content inside the nanoparticles. Carrasco et al. used these Nd³⁺:LaF₃ nanoparticles for in vivo photothermal treatment with continuous temperature monitoring in a mouse model. 168 The authors chose the intensity ratio between the emission lines at 865 and 885 nm, assigned to two Stark sublevels of the ${}^4F_{3/2} \rightarrow {}^4I_{9/2}$ transition of Nd³⁺ (Table 1). Fig. 51(a) shows a schematic representation of the experimental procedure used in this investigation. Tumorbearing mice, with one tumor per flank, were injected with a dispersion of Nd3+:LaF3 nanoparticles in phosphate-buffered saline (PBS) at a concentration of 10% in mass into one of the tumors, while the other one was used as a control (Fig. 51(b)). The injected volume was equal to $\frac{1}{2}$ of the estimated tumor volume, from which it was estimated that around 7×10^{13} nanoparticles were injected. Intratumor injection was chosen to avoid the preferential take up of the nanoparticles by the liver and spleen observed in intravenous injection. Only the

tumor containing the nanoparticles exhibited luminescence within the I-BW (Fig. 51(c)), and heating capability (Fig. 51(d)) recorded with a thermographic camera. During the photothermal therapy experiment, the intratumoral temperature was monitored (Fig. 51(e)), together with the temperature determined at the surface of the skin of the mouse (being obtained by infrared thermographic imaging) in the treated tumor and the control tumor (injected only with the equivalent amount of pure PBS and illuminated also with the 808 nm laser). The photothermal treatment lasted for 4 min. Fig. 51(e) shows the discrepancy between the intratumoral temperature, determined with the luminescent nanothermometers internalized in the tumor, and the surface temperature, determined by thermographic imaging, indicating the importance of an accurate temperature control at the injection site to minimize collateral damage due to overheating. The photothermal therapy treatment had a successful result, with a decrease of the size

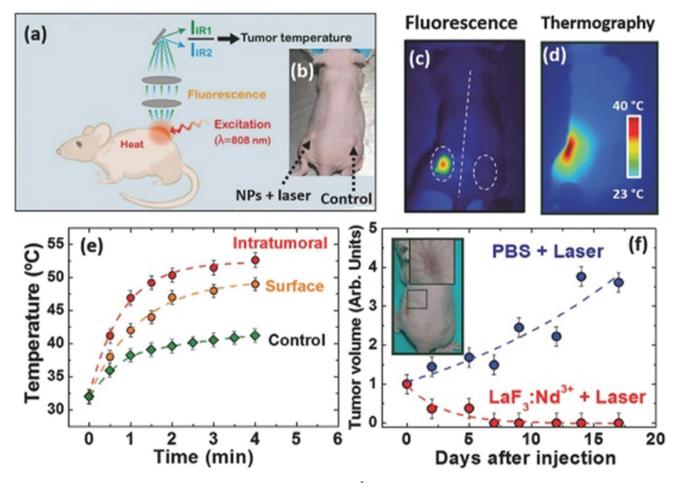


Fig. 51 Temperature-controlled photothermal therapy in living mice with Nd³⁺:LaF_τ nanoparticles operating in the I-BW. (a) Schematic representation of the experimental procedure for the dynamic controlled photothermal treatment of tumors in mice by using $Nd^{3+}:LaF_3$ nanoparticles. (b) Optical image of a tumor-bearing mouse. The left tumor was treated with Nd3+:LaF3 nanoparticles and laser, while the one on the right was used as a control. (c) Luminescent and (d) thermal images collected after a 4 min long irradiation treatment. (e) Time evolution of the intratumoral temperature determined from the analysis of the luminescence of the Nd3+:LaF3 nanoparticles, and tumor surface temperature (both treated and control tumor) as measured by infrared thermal imaging. (f) Post-treatment size evolution of both treated and control tumors. The inset shows the optical image of the surface scar left at the tumor site 15 days after the therapy. Reprinted with permission from ref. 168. Copyright 2015, Wiley-VCH Verlag GmbH & Co. KGaA.

of the tumor after treatment to finally leave only a surface scar behind (Fig. 51(f)). Meanwhile, the growth of the control tumor did not stop.

Real-time subcutaneous thermal sensing has been proposed as a powerful tool for the study of the thermal dynamics of biological tissues during in vivo experiments. Such studies would, in turn, provide access to the basic properties of biological tissues from which possible alterations related to incipient diseases could be detected.²⁸⁴ Ximendes et al. used Nd3+:LaF3@Yb3+:LaF3 core@shell nanoparticles to measure subcutaneous thermal transients as a potential theranostics tool. 223 The core@shell nanoparticles were excited at 790 nm, and by using the intensity ratio between the emissions of Nd3+ at 1.3 μ m ($^4F_{3/2} \rightarrow {}^4I_{13/2}$) and Yb $^{3+}$ at around 1 μ m ($^2F_{5/2} \rightarrow$ $^{2}F_{7/2}$), the temperature could be determined (Table 2). The authors used these luminescent nanothermometers to measure subcutaneous thermal transients, since the cooling dynamics strongly depend on the biological tissue properties. By measuring these cooling relaxation profiles it was possible to determine the characteristic thermal relaxation time that is unequivocally related to the fundamental properties of a particular biological tissue, and that additionally can provide information about the tissue status. The detection of small variations in the subcutaneous tissue relaxation times would allow for the identification of possible alterations of its thermal diffusivity, specific heat, thermal conductivity and density associated with diseases and, hence, could be used to detect anomalies caused by incipient diseases, such as dehydration, inflammation, ischemia or, even, tumor growth. To do that, the authors injected subcutaneously 200 μL of Nd³⁺: LaF₃@Yb³⁺:LaF₃ core@shell nanoparticles dispersed in PBS with a concentration of 1% in mass over the right flank of a CD1 mouse at a depth of 2 mm approximately. The mouse was anesthetized by isoflurane inhalation, and placed in a small animal imaging chamber equipped with a body temperature controller so that the mouse's temperature was maintained at 307 K \pm 1 K. A moderate temperature increment was induced at the injection site by illuminating the mouse with a 808 nm laser beam with a power density of 0.7 W cm⁻² for 4 min that activated the release of heat by the nanoparticles. After that, a low-power 790 nm laser diode at a power of 30 mW was also focused on the nanoparticles through an infrared long working distance microscope objective to generate the luminescence of the nanoparticles from which temperature could be determined. After 4 min of irradiation, the 808 nm laser was turned off and the mouse tissues gradually recovered their initial temperature. Fig. 52(a) shows a schematic representation of the experimental procedure used for this experiment. The luminescence of the nanoparticles was used to monitor the dynamics of the subcutaneous thermal relaxation. Just after the end of the heating cycle, the subcutaneous temperature was determined to be 312 K \pm 1 K. The authors were able to reveal clear differences between the thermal relaxation curves corresponding to the subcutaneous and skin temperatures (Fig. 52(b)). According to the authors, those differences arise mainly due to the different physical mechanisms respon-

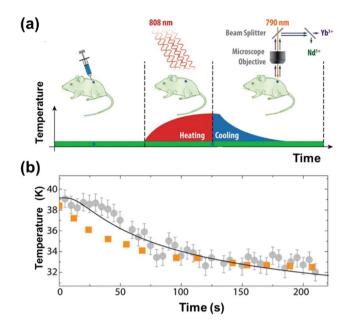


Fig. 52 Measurement of in vivo subcutaneous thermal transients in biological tissues using luminescence thermometry within the BWs. (a) Schematic representation of the experimental procedure used to measure subcutaneous thermal relaxation. (b) Time evolution of the temperature measured by the subcutaneous luminescent thermometers (grey and orange) and the IR camera (blue). Dots correspond to experimental subcutaneous (circles) and skin (squares) temperatures. Reprinted with permission from ref. 223. Copyright 2016, American Chemical Society.

sible for heat dissipation: whereas for the skin the dissipation may be essentially convective, for the subcutaneous tissues it could be assumed to be mainly conductive, due to the lack of any physical contact with air. Another reason for these different thermal relaxation curves is the minimum blood flow in the hypodermis, several orders of magnitude lower than in internal organs.

Through the analysis of the subcutaneous thermal relaxation dynamics, which allowed access to the characteristic thermal relaxation time of a given subcutaneous tissue, it was possible to compute the thermal diffusivity of the mouse tissue, resulting in a value of $0.13 \pm 0.04 \text{ mm}^2 \text{ s}^{-1}$, which was similar to that reported for chicken breast tissue determined by frequency-domain photon migration.²⁸⁵ Additionally, the subcutaneous thermal monitoring allowed the quantification of the local thermal dose administrated in a hyperthermia treatment which in that case was 54 \pm 8 J. This parameter is considered nowadays to be one of the most important factors which influences the efficiency of hyperthermia treatments. Since the total energy given by the 808 nm laser in the tissue was 560 \pm 63 J, the biological tissue absorbed only 10% of this energy. Thus, the absorption coefficient of the biological tissue was 0.23 ± 0.05 cm⁻¹, also in agreement with previous reported data in the literature. 285

Despite this big advance, accurate thermal-based diagnosis would require, at least, the dynamic acquisition of 2D sub-

cutaneous thermal images. To do that it is necessary to acquire low-noise luminescence thermal images with commercially available shortwave infrared cameras, and therefore to reconstruct accurate bidimensional thermal images. For this purpose, Ximendes et al. designed and synthesized Er³⁺,Yb³⁺: LaF₃@Yb³⁺,Tm³⁺:LaF₃ core@shell nanoparticles capable of producing 2D subcutaneous dynamic thermal imaging when excited at 690 nm (in the I-BW), emitting in the II- and III-BWs.²⁴⁰ In these Ln³⁺-doped nanoparticles, Tm³⁺ and Er³⁺ ions were spatially separated, with Tm3+ acting as the sensitizer, while Er3+ and Yb3+ were used as activators. The emissions used for the luminescent thermometer in this case were those of Er³⁺ at 1550 nm (III-BW), and Yb³⁺ at 1000 nm (II-BW) (Table 4). The authors used these core@shell nanoparticles for the acquisition of 2D thermal videos unveiling heat diffusion at the subcutaneous level. These videos allowed the acquisition of 2D patterns of tissue thermal properties that, in turn, could be used to identify and localize damaged (or non-healthy) biological tissues in vivo. Moreover, this analysis made possible the quantification of the thermal dose administered in the whole area of injection during a hyperthermia treatment. For that, the authors subcutaneously injected 200 µL of a dispersion of the Er³⁺,Yb³⁺:LaF₃@Yb³⁺,Tm³⁺:LaF₃ core@shell nanoparticles in PBS in a CD1 mouse model. Then, the temperature at the injection site was increased by illuminating at 808 nm with an excitation density of 0.7 W cm⁻², and the temperature was measured through the luminescence of the nanoparticles excited at 690 nm with a power density of 0.1 W cm $^{-2}$, low enough to avoid additional heating. 240 A schematic representation of the experimental procedure used can be seen in Fig. 53(a). From the videos recorded, the authors extracted thermal images at different time delays during the heating and cooling cycles (Fig. 53(b)–(c)), providing detailed 2D images of the subcutaneous spatial distribution of temperature with a time resolution better than 1 s. From these data the authors computed the time evolution of the subcutaneous temperature during both heating and cooling cycles (Fig. 53(d)). The thermal diffusivity of the tissue was determined to be 0.15 \pm 0.02 mm 2 s $^{-1}$ for the heating process, and 0.12 \pm 0.02 mm 2 s $^{-1}$ for the cooling one.

7.2. Other applications of lanthanide-doped luminescent nanothermometers operating within the BWs

Lanthanide-doped luminescent materials can be used as nanothermometers also outside the biological/biomedical fields. An example involves the application of lanthanide-doped luminescent materials as thermal probes on the surface of a copper microwire to visualize the Joule induced heating effect. 171 Nd $^{3+}$:YAG luminescent nanothermometers, operating in the I-BW based on the thermally coupled Stark sublevels of the $^4\mathrm{F}_{3/2} \rightarrow ^4\mathrm{I}_{9/2}$ electronic transition of Nd $^{3+}$, located at $\sim 940-950$ nm (Table 1), have been used for this purpose. A drop of an aqueous dispersion of these nanoparticles (1 wt%) was deposited onto a 100 $\mu \mathrm{m}$ thick copper wire fixed on a glass

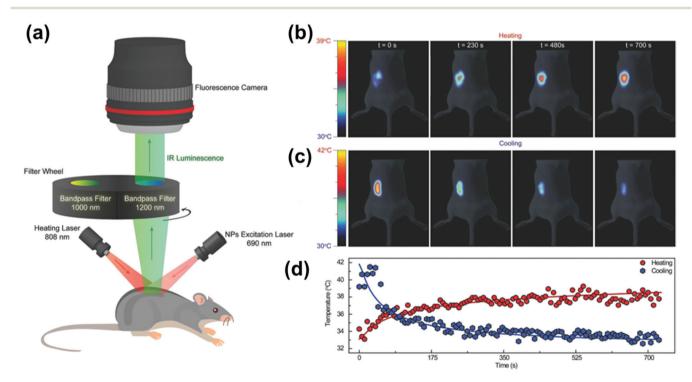


Fig. 53 In vivo acquisition of 2D subcutaneous thermal images. (a) Schematic representation of the experimental procedure used in the *in vivo* experiment. (b) Thermal images obtained during the heating and relaxation (cooling) cycles. (c) Time evolution of the average temperature in the injection area as measured by the subcutaneous lanthanide nanothermometers during heating and thermal relaxation cycles. Reprinted with permission from ref. 240. Copyright 2017, Wiley-VCH Verlag GmbH & Co. KGaA. (d) Time evolution of the average temperature of the injection area as measured by subcutaneous luminescence nanothermometry during heating and thermal relaxation processes.

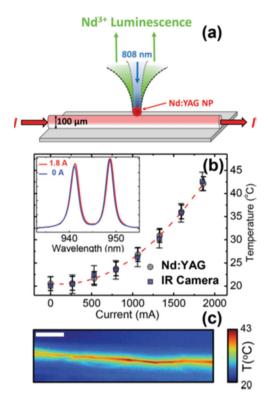


Fig. 54 Temperature sensing in metallic microwires heated by the Joule effect using lanthanide-doped luminescent nanothermometers operating in the I-BW. (a) Scheme of the experimental procedure used. (b) Temperature of the microwire as a function of the applied current, obtained from the intensity ratio of the emissions generated by the nanoparticles and from an infrared thermographic camera. The emission spectra generated by the nanoparticles placed on the wire obtained at the two extreme values of the electrical current (0 and 1.8 A) can be seen in the inset. (c) Thermal image of the microwire when conducting the maximum electrical current applied (1.8 A), as obtained with the thermographic camera (scale bar is 2 cm long). Adapted with permission from ref. 171. Copyright 2015, Wiley-VCH Verlag GmbH & Co. KGaA.

plate. After the evaporation of water, an 808 nm laser diode was focused onto the wire using a 10× magnification microscope objective with a N.A. = 0.55, which produced a laser spot of approximately 1.8 µm in diameter on the wire. Using the same microscope objective, the luminescence arising from the Nd³⁺:YAG nanoparticles was collected (Fig. 54(a) for the schematic representation of the experimental procedure used). The increase of the temperature caused by the ohmic heating of the microwire was monitored by an infrared thermographic camera and by the emissions of the luminescent particles. The temperature calculated from the intensity ratio of the emission generated by the nanoparticles presented a parabolic dependence on the different electrical currents applied (Fig. 54(b)), as expected since the Joule heating effect is directly proportional to the square of the intensity of the electrical current applied.171 These results were validated with those recorded with the infrared thermographic camera (Fig. 54(b) and (c)).

Nd³⁺:YAG nanoparticles with the same emissions as in the previous example have also been used as thermal probes in

microfluidics. 171 Taking advantage of the water dispersibility of these Nd³⁺:YAG nanoparticles, thermal imaging of a simple optofluidic device consisting of a 100 µm thick microchannel was recorded. In this microchannel, a single-mode fibercoupled 1480 nm diode laser with a maximum power of 200 mW was used as a heating source. This laser was focused on the channel containing the dispersion of the nanoparticles in water through a $10 \times$ microscope objective with a N.A. = 0.25. To extract the temperature distribution pattern generated by the 1480 nm laser irradiation, strongly absorbed by water, the microfluidic channel was filled with a 0.1 wt% aqueous dispersion of the nanoparticles, while being monitored with a thermal camera (Fig. 55(a) for the schematic representation of the experimental procedure used). Nd³⁺:YAG nanoparticles were excited with a 808 nm laser, which was focused onto the optofluidic device using a 10× microscope objective with NA = 0.55, that produced a laser spot of approximately 1.8 µm in diameter on the channel.

The luminescence generated by the nanoparticles was collected using the same microscope objective. Fig. 55(b) shows the temperature increment at the focus of the 1480 nm heating laser spot as a function of its power. The measurements of temperature were performed by overlapping both the focused heating and probe (808 nm) laser beams. It can be seen that for laser powers below 60 mW, the "on-focus" temperature increased linearly, whereas above this value the relationship did not obey a linear function any more. 171 The authors considered that at high laser powers additional

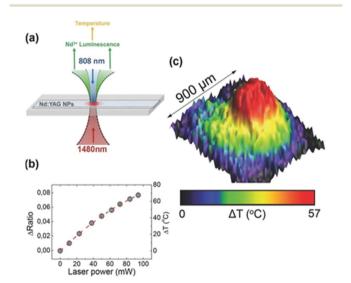


Fig. 55 Temperature sensing in optofluidics using luminescent nanothermometers operating in the I-BW. (a) Scheme of the experimental procedure used to record the temperature distribution within an optical trap created by a 1480 nm focalized laser beam on a microchannel filled with an aqueous dispersion of Nd3+:YAG nanoparticles. (b) Temperature increment as a function of the power of the 1480 nm laser. (c) All-optical thermal imaging (2D temperature distribution map) of the area where the optical trap was created under pumping with the 1480 nm laser. Adapted with permission from ref. 171. Copyright 2015, Wiley-VCH Verlag GmbH & Co. KGaA.

heating was produced, which could lead to the creation of relevant convection currents and the formation of bubbles. By applying a power of 60 mW, a 2D thermal image of the microchannel in the surroundings of the 1480 nm laser spot was recorded by scanning the 808 nm laser along a $900 \times 900 \ \mu m^2$ area around the heating laser spot. It was observed that the laser-induced heating caused an increase of the temperature in a large area surrounding the laser spot (Fig. 54(c)), far from the microchannel itself, due to thermal diffusion. 171

Luminescence nanothermometers operating in the I-BW can be also used as a tool for unveiling the thermal properties of the nanoparticles themselves, such as the determination of their thermal resistance. Thermal resistance is a key parameter to model the heat transfer at the nanoscale. In order to achieve this, Savchuk *et al.* applied Ho³⁺, Tm³⁺: KLu(WO₄)₂ nanoparticles, excited with an 808 nm laser, as upconversion luminescent nanothermometers based on the 696 nm emission of Tm³⁺ and the 755 nm emission of Ho³⁺, both located in the I-BW (Table 1).

The 808 nm laser was focused on the surface of the nanoparticles, generating, simultaneously, upconversion luminescent emissions and heat, due to radiative and non-radiative processes, respectively. A beam splitter was used to redirect a

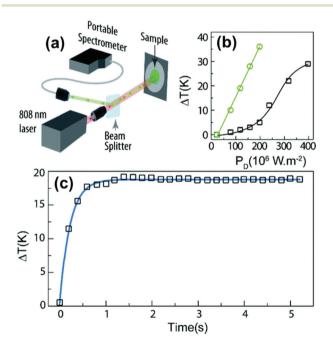


Fig. 56 Determination of the thermal resistance of nanoparticles by lanthanide-doped luminescent nanothermometers operating in the I-BW. (a) Scheme of the experimental procedure used. (b) Temperature increase induced as a function of the different laser powers applied, and as a function of the concentration of the two emitting ions in the nanoparticles: 1 at% $\rm Ho^{3+}$, 5 at% $\rm Tm^{3+}$ (black squares) and 1 at% $\rm Ho^{3+}$, 15 at% $\rm Tm^{3+}$ (green circles) in $\rm KLu(WO_4)_2$. (c) Heating curve obtained for the 1 at% $\rm Ho^{3+}$, 5 at% $\rm Tm^{3+}$ -doped $\rm KLu(WO_4)_2$ nanoparticles excited with the 808 nm laser operating with a power density of 318 \times 10 6 W m $^{-2}$. The blue line corresponds to the best fit of experimental data following the Fourier law. Reprinted with permission from ref. 150. Copyright 2018, The Royal Society of Chemistry.

part of the emitted signal to a portable spectrophotometer that recorded the upconversion emission generated by the nanoparticles (Fig. 56(a) for the schematic representation of the experimental procedure used). The intensities of these emissions were recorded as a function of time and were converted into temperature by calculating the intensity ratio between the emissions of Ho3+ and Tm3+, following eqn (19). The temperature increment $(\Delta T(t))$ generated in the nanoparticles by the excitation laser was measured as a function of the concentration of the emitting ions within the host (1 at% Ho³⁺, 5 at% Tm3+ (black squares) and 1 at% Ho3+, 15 at% Tm3+ (green circles) in Fig. 56(b)). The results showed that $\Delta T(t)$ generated was smaller for the nanoparticles with a smaller amount of Tm³⁺ ions, suggesting a positive correlation between the heat generated due to non-radiative Tm3+-to-Ho3+ ET and Ho3+-to-Tm³⁺ BET (Fig. 9(a)) and the Tm³⁺ concentration.

In addition, the temperature plateau observed at high power densities above $300 \times 10^6~W~m^{-2}$ and not observed in the sample with a higher Tm^{3+} concentration seemed to be related to a saturation effect on the Tm^{3+} absorption at 808 nm.

The heat dissipation could be modelled following the classical Fourier law, resulting in a temperature increase given by: 286,287

$$\begin{split} \Delta T(t) &= \Delta T_{\rm m} \left[1 - \exp \left(-\frac{hA}{mc} t \right) \right] = \\ \Delta T_{\rm m} \left[1 - \exp \left(-\frac{t}{RC} \right) \right] &= \Delta T_{\rm m} \left[1 - \exp \left(-\frac{t}{\tau} \right) \right] \end{split} \tag{52}$$

where $\Delta T_{\rm m}$ is the temperature increase in the steady-state regime (Fig. 56(c)), h is the convective heat transfer coefficient, A is the thermal contact area, R=1/hA is the convective thermal resistance and $\tau=RC$ stands for the system's thermal time constant.

By fitting this equation to the heating curve of the 1 at% ${\rm Ho^{3+}}$, 5 at% ${\rm Tm^{3+}}$:KLu(WO₄)₂ particles in contact with air, presented in Fig. 56(c), a temperature increase $\Delta T_{\rm m}=18.3\pm0.2$ K and a thermal time constant $\tau=0.223\pm0.004$ s were obtained, corresponding to a thermal resistance of $R=(9.50\pm0.17)\times10^7$ K W⁻¹. ¹⁵⁰ In addition, the extraction of τ allowed for the determination of the temporal resolution of the luminescent nanothermometers, $\delta T=0.0033\pm0.0007$ s, calculated using eqn (5), (2)–(4), orders of magnitude faster than the thermometers based on scanning thermal microscopy (0.1 s, ²⁸⁸ and 1.5 s), ^{289,290} and Raman spectroscopy (90 s). ²⁹¹

8. Concluding remarks

Lanthanide-doped luminescent materials have triggered the development of highly sensitive luminescent thermometers, particularly if the operating regime of their emissions is located within the BWs. In this review, several strategies towards obtaining better temperature-sensing performances within the BWs have been analyzed.

Hence, for the choice of the excitation wavelength, the selection of a NIR source is highly recommended, allowing for

deeper penetration depths and better temperature readings, implying no phototoxicity or photobleaching effects, enhancing significantly the interest in these luminescent thermometers in biological/biomedical applications. Related to the choice of hosts for the Ln³⁺ ions, a strategy that results in highly sensitive thermometers is the use of active core@active shell nanostructures. Another choice is the combination of the emissions of Ln³⁺ ions with those arising from transition metal ions or quantum dots, acting as a reference and as highly thermally quenched probes, respectively. Also, coating Ln³⁺-doped nanothermometers with an inert shell (including silica) can result in higher intensities and better temperaturesensing performances due to the reduced thermal quenching resulting from the influence of surface defects and ligands. Related to the choice of the Ln³⁺ ions, a proper selection of the concentration of the ion to be used and the operating biological window in which one wants to work is required. The intentional introduction of impurities inside the hosts (such as in the case of Y³⁺ in Nd³⁺:CaF₂, ^{245,276} or Gd³⁺ in Nd³⁺:SrF₂)¹⁶³ might lead to reduced thermal quenching effects, and upon a proper selection of the concentration of these impurities, to a better thermometric performance. In terms of Ln³⁺ ions included as emitters, Nd³⁺ and Tm³⁺ are the ones that attracted most of the attention (Table 1 for example). Nevertheless, a better choice for efficient luminescent nanothermometers is that constituted by dual emitting centers, since they provide better thermal sensitivity and allow the incorporation of multifunctionalities in these nanoparticles.

Despite the big number of works devoted to the development of luminescent thermometers operating in the BW spectral ranges, and the demonstration of amazing applications for them, there are still challenges that remain unsolved. From the synthetic point of view, a vast amount of research is devoted to the synthesis of fluoride-doped hosts, either in the form of bare, core@shell or multishell nanostructures, through the thermal decomposition of trifluoroacetate precursors to achieve control over their sizes and shapes. This synthetic approach suffers from the toxicity of the by-products and final products generated in the reactions, 103 and an additional surface functionalization of the obtained nanoparticles is required for their potential use in biomedical applications. 103 Therefore, new synthetic routes towards biocompatible nanomaterials are highly desirable. Furthermore, new choices for hosts for the Ln³⁺ ions should also be proposed.

Concerning the thermometric performance of the Ln³⁺doped luminescent nanothermometers, clearly some discrepancies in the evaluation of their performances (calculation of the thermometric parameter or S_{rel}) appear. In this way, it would be recommended that all authors follow the same procedures in order to be able to compare properly the performance of the different luminescent thermometers proposed. Additional strategies to boost the thermal sensitivity of the Ln³⁺-doped nanothermometers, such as the use of induced structural stress by lattice self-adaptation in heterojunctions producing a lattice distortion at the interface between two materials, 292 dopant-induced local site symmetry distortions, 293 and the effect of anisotropic interfacial strain in mismatched core-shell nanoparticles, 294 which are sensitive to temperature changes, are really interesting. Up to now, these strategies have only been applied for luminescence thermometry in the visible range, and thus attempts to implement these strategies within the biological window spectral regions should be considered.

In addition, besides S_{rel} and δT , a very low number of publications reported about their resolution (either spatial or temporal), reproducibility and repeatability. These parameters are also important to facilitate the implementation of the Ln³⁺doped luminescent thermometers in real biological/biomedical applications outside of laboratories. Another very important parameter that should be taken into account when developing a novel luminescent thermometer is the quantum yield, both absolute and relative. A good quantum yield would allow for an easier detection scheme for the emissions generated by the luminescent thermometers by using cheaper detectors and lower laser excitation powers, simplifying also the protocols of use of these luminescent nanoparticles, and facilitating their use by non-experts in these subjects. As can be seen in this review, the absolute quantum yield has not been yet reported for any Ln3+-doped luminescent thermometer. Thus, an additional effort is required to take this parameter into account.

From Tables 1-5 presented in this review, it can be noted that an important number of the luminescent thermometers presented are based on using Yb3+ ion as a sensitizer. Lanthanide ions such as Er³⁺ or Tm³⁺ are typically encountered in materials co-doped with Yb³⁺ in order to enhance the intensity of their photoluminescence since Yb3+ presents a strong absorption band at around 980 nm, with an absorption cross section higher than that of these ions. However, the use of this excitation source is not suitable for biological/biomedical applications due to the self-heating effects generated by the absorption of the energy of the excitation source by water present in the biological tissues through the absorption band of this molecule at the same wavelength. 148 Different strategies have been developed to substitute this excitation source in order to make lanthanide-doped nanomaterials more biological tissue-friendly. It is worth mentioning the replacement of Yb3+ by Nd3+ or Tm3+ ions acting as sensitizers. The absorption cross section of Nd³⁺ at 800 nm ($\approx 10^{-19}$ cm²) is one order of magnitude higher than that of Yb3+ at 980 nm $(\approx 10^{-20} \text{ cm}^2)$. Furthermore, at this wavelength the absorption coefficient of water is around 20 times smaller than that at 980 nm. 206 In this way, the overheating effect in biological tissues can be highly reduced. Concerning the choice of Tm³⁺ as a sensitizer, a successful strategy has been successfully developed by co-doping these systems with Ho3+.150,266 Tm3+ has a strong absorption band at 808 nm that can be directly pumped by diode lasers, and the energy transfer processes between Tm3+ and Ho3+ allow use of the emissions of these two ions for luminescence thermometry purposes.²⁹⁵ Hence, these combinations of Ln3+ ions are highly attractive for biomedical applications.

From the biological/biomedical applications point of view, the studies published up to now are mostly focused on the II-BW spectral region because of the deeper penetration depths that can be achieved. However, according to Naczynski *et al.*¹²⁶ and other recent publications, this parameter should be even better at longer wavelengths,²⁹⁶ in the III- and IV-BW spectral ranges. Therefore, these spectral regions should be further explored in the future.

In this context, luminescence thermometers fully operative within the BWs, especially those excited within the II, III and IV BWs, and generating emissions in the same spectral regions, would be very important to advance towards their real application in the biomedical field, because they will allow for a deeper penetration depth in biological tissues 105,123,124 that would allow for a full diagnosis and treatment of diseases. Nevertheless, the research work done in this area is very scarce, and the number of lanthanide ions that can be used for this purpose is limited. In fact, the Ln³⁺ ions which can fulfill these conditions are mainly Tm³⁺ and Er³⁺ due to their ability to absorb energy within these BW regions (Tm³⁺ at 1210 nm, ²⁹⁷ 1319 nm, ¹²⁸ and 1640 nm, ²⁹⁷ and Er³⁺ at 1500–1550 nm). ²⁹⁸ This strategy has been successfully applied in in vivo biosensing applications by combining Er³⁺ with other lanthanide ions, such as Ho3+ and Nd3+, in NaErF4@NaYF4:Ho@NaYF4 or NaErF4@NaYF4:Nd@NaYF4 core-shell nanoarchitectures, and

taking advantage of their emissions located in the II-BW (1180 nm of Ho³⁺, 1060 nm of Nd³⁺, and 980 nm of Er³⁺) after the energy of the 1530 nm excitation source was absorbed by Er³⁺.²⁹⁸ Furthermore, it has been confirmed that the penetration depth in biological tissues (in chicken pectoral muscle in that particular case) is two times higher when using an excitation source at 1319 nm than when using a 800 nm excitation source in Tm³⁺-doped nanoparticles.¹²⁸

In addition, for the biological/biomedical applications, the biological tissues in which these luminescent thermohave been applied are different (chicken breast, 63,166,171,174,245,266,299 fat, 164,300 pork phantom tissue, 157,163,301 or mice 126,223,302), leading to different and non-comparable penetration depths (Fig. 57) due to the different responses to the interaction with the light of their biological components. Thus, extensive research should be developed to understand the similarities and differences in the interaction of light with the different biological tissues to extract reliable conclusions in this area. Nevertheless, and regardless of the nature of the biological tissue, when operating at longer wavelengths, in general, the penetration depth increased, considering only those examples in which chicken breast was used as a subject of study, as can be seen in Fig. 57. Also from this figure, it can be seen that the majority of the luminescent nanothermometers developed up to now operat-

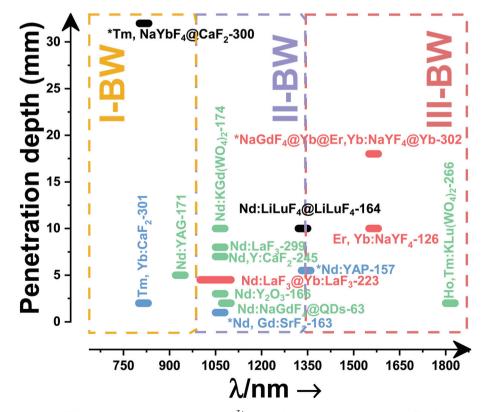


Fig. 57 Penetration depth in the BWs spectral regions reported for Ln^{3+} -doped luminescent nanomaterials. The intervals represent the operating wavelength regimes for the corresponding nanothermometer, whereas the asterisk indicates only one reported wavelength. The black, blue, red and green colors stand for the different types of biological medium: black = pork tissue, blue = phantom tissue, red = mice and green = chicken tissue. The numbers are the corresponding references.

ing within the BW spectral ranges, including also these materials used only to determine the penetration depth in biological tissues, were based on Nd3+-doped materials, when other choices are also possible, like the combination of Ho³⁺ and Tm³⁺ ions, for instance, to develop multifunctional nanoparticles such as self-assessed photothermal agents.

Nevertheless, there are some current challenges for luminescent nanothermometers when used inside biological tissues that should be addressed for future biomedical applications, such as the distortion of the shape of the spectra recorded for these luminescent nanothermometers produced when they are recorded through the biological tissues.303 Possible solutions for this drawback could be the calibration of the luminescent nanothermometers when already embedded in the biological medium, as proposed recently by Shen et al. 303 or the use of lifetime luminescence nanothermometry, which is not affected by changes in the optical properties of the surrounding medium, including biological tissues, by the concentration of the luminescent nanothermometers or by the excitation power density used.²²⁷

Furthermore, in in vitro applications, concepts related to the temperature flocculation at the nanoscale, the disagreement between the calculated and the experimental temperature, and the understanding of the thermal conductivity in a cell,304 should be addressed. In fact, the discrepancies in in vitro applications could also affect the in vivo applications.304

Finally, another research direction that should be further taken into account would be exploring the application of these luminescent thermometers outside the biological/biomedical fields, since their reported applicability in this direction is

As a final note, we would like to highlight, according to all the data analyzed in this review, which strategies would provide a better performance for the luminescent nanothermometers operating in the BWs. It is clear that to maximize the relative thermal sensitivity of the luminescent thermometer, and consequently minimize the temperature resolution, the most effective strategy consists of maximizing the separation between the emission lines used in the thermometer when the band-shape luminescent thermometry method is used. If this is done by using emissions arising from two non-thermally coupled electronic levels that are apart a distance higher than the upper limit defining the thermally coupled range of energies, the performance of the thermometer would be even better. In fact, by combining two different emission centers, and especially if emissions arising from transition metals are involved, the highest thermal sensitivities reported up to now are achieved, although these systems are quite complex, and we still have a low understanding of the factors that govern the temperature-sensing properties. Nevertheless, there are other aspects that have to be taken into account to develop efficient and practical luminescent nanothermometers, such as the use of emission lines with similar intensities in the thermometer, that would facilitate their recording and analysis with enough guarantees. Another important aspect to consider, especially

when developing these luminescent thermometers for biological/biomedical applications, is that the maximum of their relative thermal sensitivity should be in the range of 350-400 K, and not at room temperature, as happens with the majority of the luminescent thermometers developed up to now. All these aspects are important to analyze when considering developing luminescent thermometers operating in the III- and IV-BW, where the S_{rel} achieved up to now is still low, although the number of publications devoted to developing luminescence thermometers in this spectral range is still very low, or null in the case of the IV-BW.

After this analysis, we believe that working with Nd³⁺ and Ho³⁺,Tm³⁺-doped and codoped materials is a promising approach to not only promote the thermometric performance of the luminescent nanothermometers, but also to develop multifunctional platforms that can combine thermal sensing and photothermal therapy, for instance. Furthermore, using lifetime thermometry in these systems would allow current challenges on the biomedical application of these materials to be addressed, such as the spectral distortion. Finally, novel strategies towards the challenge of increasing the low brightness of Ln³⁺-doped materials should also be addressed. Up to now core-shell and core-multishell nanoarchitectures offered a boost in the brightness of the emissions generated by lanthanide ions and a significant enhancement of their temperaturesensing performance. However, other strategies that can improve this brightness, such as pairing them with plasmonic structures, use of organic-dye sensitization, and/or coupling them with semiconductors, 107,108 might be explored.

Conflicts of interest

There are no conflicts to declare.

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