



Cite this: *Chem. Commun.*, 2018, 54, 4786

Received 28th February 2018,  
Accepted 16th April 2018

DOI: 10.1039/c8cc01661e

rsc.li/chemcomm

# Long-wavelength TCF-based fluorescence probes for the detection and intracellular imaging of biological thiols†

Adam C. Sedgwick,<sup>a</sup> Jordan E. Gardiner,<sup>a</sup> Gyoungmi Kim,<sup>b</sup> Maksims Yevglevskis,<sup>c</sup> Matthew D. Lloyd,<sup>c</sup> A. Toby A. Jenkins,<sup>a</sup> Steven D. Bull,<sup>a</sup> Juyoung Yoon<sup>b</sup> and Tony D. James<sup>a</sup>

Two 'turn on' TCF-based fluorescence probes were developed for the detection of biological thiols (TCF-GSH and TCFCI-GSH). TCF-GSH was shown to have a high sensitivity towards glutathione (GSH) with a 0.28  $\mu\text{M}$  limit of detection. Unfortunately, at higher GSH concentrations the fluorescence intensity of TCF-GSH decreased and toxicity was observed for TCF-GSH in live cells. However, TCFCI-GSH was shown to be able to detect GSH at biologically relevant concentrations with a 0.45  $\mu\text{M}$  limit of detection. No toxicity was found for TCFCI-GSH and a clear 'turn on' with good photostability was observed for the exogenous addition of GSH, Cys and HCys. Furthermore, TCFCI-GSH was used to evaluate the effects of drug treatment on the levels of GSH in live cells.

Glutathione (GSH), cysteine (Cys) and homocysteine (HCys) play a vital role in maintaining the biological redox homeostasis.<sup>1,2</sup> GSH is a natural tripeptide ( $\gamma$ -L-glutamyl-L-cysteinyl-glycine), which exists in the thiol reduced form (GSH) and disulphide-oxidised (GSSG) form.<sup>2</sup> GSH is the predominant form, which exists in millimolar concentrations in most cells where it functions as an antioxidant.<sup>3</sup> Elevated levels of GSH are common in the presence of oxidative stress and the susceptibility of a cell towards reactive oxygen or nitrogen species (ROS/RNS) largely depends on the concentration of intracellular GSH.<sup>4–7</sup> Therefore, the change in the level of GSH concentration has been associated with a number of diseases such as AIDS, liver damage, cancer and neurodegenerative disease (Alzheimer's disease).<sup>6,7</sup> Interestingly, it was reported that at early stages of cell proliferation (S, G<sub>2</sub> and M phases), GSH was found to localise at the nucleus. This was believed to prevent apoptosis and provide a reduced environment for transcription factors to bind to DNA.<sup>8</sup>

With our research, we are interested in the development of reaction based fluorescent probes for the detection of biologically relevant species to be used as powerful tools for the understanding of diseases.<sup>9–13</sup> Currently, a number of fluorescent probes exist for the detection of biological thiols.<sup>14–20</sup> However, long excitation/emission wavelength fluorescent probes are highly desirable as they allow deeper tissue penetration, minimal background autofluorescence from proteins and photodamage to the biological samples. Therefore, in this work we looked to develop TCF-based systems for the long wavelength detection of GSH.<sup>11</sup>

TCF-based fluorophores have an internal charge transfer (ICT) donor- $\pi$ -acceptor (D- $\pi$ -A) structure with long emission wavelengths (see ESI† – Scheme S1). As a result, TCF fluorophores have been used in many applications such as non-linear optic chromophores and fluorescent probes.<sup>21–25</sup> Hilderbrand *et al.* previously developed a 'turn on' sulfonamide based TCF fluorescent probe for the detection of biological thiols.<sup>26</sup> However, a PEG unit was required to provide aqueous solubility and cell permeability. The probe was successfully shown to detect biological thiols in 3T3 cells. We believed the synthesis of the analogous sulfonate ester would overcome the need for a PEG unit and provide a much simpler synthesis. The TCF fluorophore unit was synthesised as previously reported using the reaction of 3-hydroxy-3-methyl-2-butanone, malonitrile and NaOEt in EtOH. With the TCF unit in hand, the (D- $\pi$ -A) systems TCF-OH and TCFCI-OH were isolated in high yield using microwave reaction conditions.<sup>27</sup> The TCF phenols were then reacted with 2,4-dinitrobenzenesulfonylchloride to afford the desired fluorescent probes TCF-GSH and TCFCI-GSH in satisfactory yields (55% and 64%) (Fig. 1).

On the addition of GSH, both probes TCF-GSH and TCFCI-GSH change colour from yellow to purple (see ESI† – Fig. S1 and S2). We evaluated the fluorescence behaviour of TCF-GSH, in pH 8.0 buffer solution (20% v/v DMSO) (see ESI†–Fig. S3 and S4). Interestingly, 20% v/v DMSO was required for the reaction between the probe and the chosen biological thiol to take place. We then evaluated TCF-GSH for the detection of GSH, given that it is the most predominant biological thiol in cells.

<sup>a</sup> Department of Chemistry, University of Bath, Bath, BA2 7AY, UK.  
E-mail: t.d.james@bath.ac.uk, s.d.bull@bath.ac.uk

<sup>b</sup> Department of Chemistry and Nano Science, Ewha Womans University, Seoul 120-750, Korea. E-mail: jyoony@ewha.ac.kr

<sup>c</sup> Drug & Target Development, Department of Pharmacy & Pharmacology, University of Bath, Claverton Down, Bath, BA2 7AY, UK

† Electronic supplementary information (ESI) available: All data supporting this study. See DOI: 10.1039/c8cc01661e



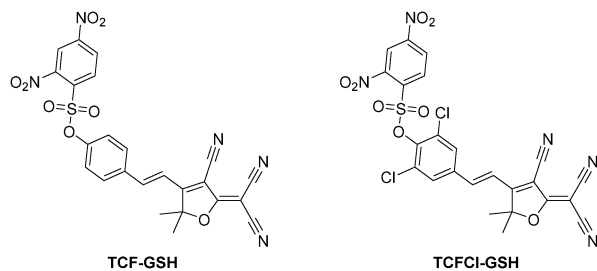


Fig. 1 TCF-based fluorescent probes for the detection of biological thiols (**TCF-GSH** and **TCFCl-GSH**).

Remarkably, **TCF-GSH** was very sensitive towards GSH producing a full 'turn on' fluorescence response in the presence of 25  $\mu\text{M}$  GSH. Unfortunately, at concentrations  $> 50 \mu\text{M}$  the fluorescence intensity of **TCF-GSH** began to drop dramatically. This is due to attack of the TCF fluorophores by nucleophiles (Fig. 2)<sup>28</sup> (see ESI† – Fig. S5–S11).

We then evaluated the selectivity of **TCF-GSH** towards other biologically relevant thiols and amino acids (see ESI† – Fig. S12 and S13). As predicted, **TCF-GSH** reacted with the other sulphhydryl (R-SH) compounds, Cys and HCys with Cys producing the largest fluorescent response. However, the overall concentrations of both Cys and HCys are low in comparison to GSH in cells.<sup>29,30</sup> **TCF-GSH** demonstrated an excellent selectivity for GSH against other amino acids. This excellent selectivity permitted the evaluation of **TCF-GSH** for the detection of exogenous and endogenously generated thiols in live cells. Sadly, despite **TCF-GSH** being sensitive towards GSH, we only observed a clear 'off-on' response for the exogenous addition of Cys in HeLa cells. Furthermore, **TCF-GSH** was shown to have toxicity in cell viability experiments (see ESI† – Fig. S16 and S17). **TCF-GSH** is therefore unsuitable for cell imaging experiments for the detection of biothiols. Interestingly, cellular imaging experiments did not require any additional additives to compensate for the 20% v/v DMSO required in the *in vitro* experiments.

We therefore turned our attention towards the fluorescence properties of **TCFCl-GSH**. In order to produce a fluorescence

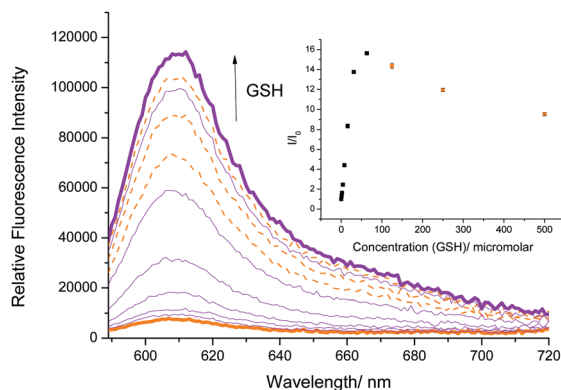


Fig. 2 Fluorescence spectra of **TCF-GSH** (5  $\mu\text{M}$ ) with addition of GSH (0–500  $\mu\text{M}$ ) and 15 min wait between additions in PBS buffer solution, 20% v/v DMSO, pH 8.00 at 25 °C.  $\lambda_{\text{ex}} = 560 \pm 15 \text{ nm}$ . Orange dashed lines indicate fluorescence decrease at high GSH concentrations.



Fig. 3 Fluorescence spectra of **TCFCl-GSH** (5  $\mu\text{M}$ ) with addition of GSH (0–750  $\mu\text{M}$ ) and 15 min wait between additions in PBS buffer solution, 20% v/v DMSO, pH 8.00 at 25 °C.  $\lambda_{\text{ex}} = 560 \pm 15 \text{ nm}$ .

response, **TCFCl-GSH** also required pH 8.0 buffer solution (20% v/v DMSO). However, **TCFCl-GSH** was shown to be less sensitive towards the biological thiols and no decrease in fluorescence intensity was observed at higher concentrations (Fig. 3).

We then evaluated the selectivity of **TCFCl-GSH** towards other biologically relevant thiols and amino acids (see ESI† – Fig. S14 and S15). As for **TCF-GSH**, **TCFCl-GSH** reacted with the R-SH containing amino acids Cys and HCys. While excellent selectivity for GSH was observed against other amino acids. This permitted the evaluation of **TCFCl-GSH** for the detection of exogenous and endogenous thiols in live cells. As shown in

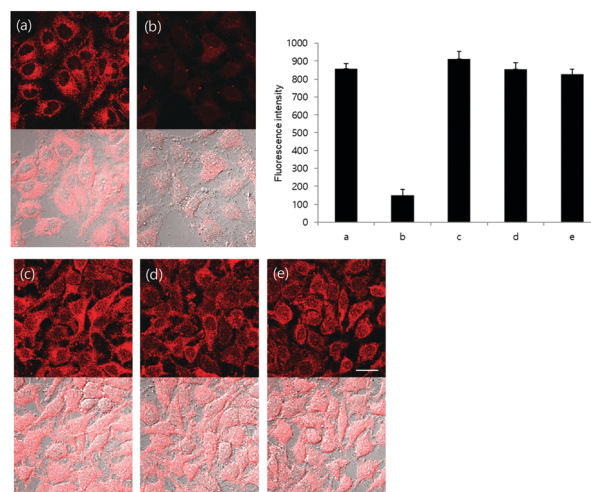


Fig. 4 Fluorescence imaging in live cells. HeLa cells were preincubated with 0.2 mM NMM for 20 min and washed with Dulbecco's phosphate-buffered saline (DPBS) and incubated with 200  $\mu\text{M}$  cysteine, homocysteine and GSH-MEE for 20 min. After washing with DPBS, cells were stained with 20  $\mu\text{M}$  **TCFCl-GSH** for 20 min and fluorescence images acquired by confocal microscopy. (a) Only **TCFCl-GSH**, (b) NMM + **TCFCl-GSH**, (c) NMM + cysteine + **TCFCl-GSH**, (d) NMM + homocysteine + **TCFCl-GSH** and (e) NMM + GSH-MEE + **TCFCl-GSH**. Top: Fluorescence image (ex. 559 nm/em. 575–675 nm), bottom: merged image with DIC. Scale bar: 20  $\mu\text{m}$ . Quantitative data of fluorescence intensity was calculated by FV10-ASW 4.0 software and measured per one cell. Results are expressed as mean  $\pm$  standard deviation of three independent experiments.



Fig. 4 (see ESI† – Fig. S18 and S19 for TCF-GSH), TCFCl-GSH displayed an already strong fluorescence response in live cells Fig. 4(a). This observation was due to the presence of endogenous thiols reacting with TCFCl-GSH. However, pre-treatment of HeLa cells with the thiol reactive *N*-methylmaleimide (NMM) led to the reduction of endogenous thiols and therefore low fluorescence intensity was observed when TCFCl-GSH was added Fig. 4(b). The addition of 200  $\mu\text{M}$  of an exogenous thiol (Cys, HCys, or GSH-Methyl ester) led to a clear change in fluorescence intensity demonstrating the ability of TCFCl-GSH to detect thiols in cells.

We then evaluated the ability of TCFCl-GSH to detect changes in the concentration levels of endogenous thiols through the addition of drugs and reactive oxygen species (ROS) such as  $\text{H}_2\text{O}_2$ . It is well known that GSH protects against drug induced toxicity and acts as a ROS scavenger. Therefore in Fig. 5 and 6

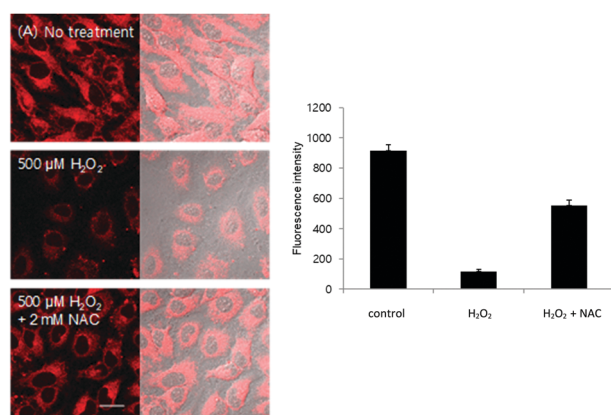


Fig. 5 Intracellular fluorescence change caused by drug treatment. (A) HeLa cells were incubated with 500  $\mu\text{M}$   $\text{H}_2\text{O}_2$  with or without 2 mM NAC for 6 h and stained with 20  $\mu\text{M}$  TCFCl-GSH for 20 min. Quantitative data of fluorescence intensity was calculated by FV10-ASW 4.0 software and measured per one cell. Results are expressed as mean  $\pm$  standard deviation of three independent experiments. Ex. 559 nm/em. 575–675 nm. Scale bar: 20  $\mu\text{m}$ .

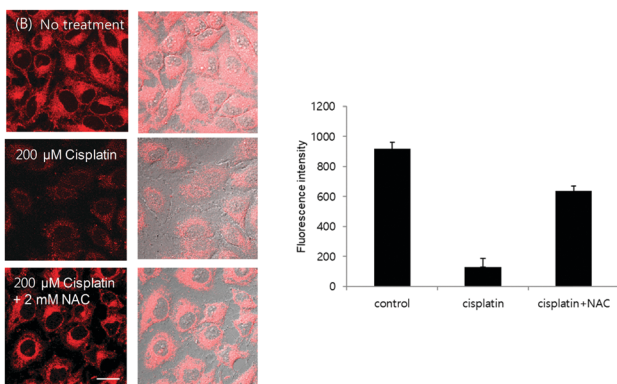


Fig. 6 Intracellular fluorescence change caused by drug treatment. (B) HeLa cells were incubated with 200  $\mu\text{M}$  cisplatin with or without 2 mM NAC for 6 h and stained with 20  $\mu\text{M}$  TCFCl-GSH for 20 min. Quantitative data of fluorescence intensity was calculated by FV10-ASW 4.0 software and measured per one cell. Results are expressed as mean  $\pm$  standard deviation of three independent experiments. Ex. 559 nm/em. 575–675 nm. Scale bar: 20  $\mu\text{m}$ .

(see ESI† – Fig. S20 and S21 for TCF-GSH), the addition of  $\text{H}_2\text{O}_2$  (500  $\mu\text{M}$ ) or Cisplatin (200  $\mu\text{M}$ ) resulted in the depletion of the endogenous thiols and consequently reduced fluorescence was observed when TCFCl-GSH was added. Subsequently, the addition of the GSH producing drug *N*-acetylcysteine<sup>31</sup> recovered the GSH levels resulting in a large increase in fluorescence intensity.

In summary, two ‘turn on’ TCF-based fluorescent probes have been developed for the detection of biological thiols (TCF-GSH and TCFCl-GSH). TCF-GSH was shown to have a high sensitivity towards glutathione (GSH). Unfortunately, at higher GSH concentrations the fluorescence intensity of TCF-GSH decreased and toxicity was observed in live cells making it unsuitable for cellular imaging. However, TCFCl-GSH was shown to be able to detect GSH at biological relevant concentrations. Also, no toxicity was observed for TCFCl-GSH and a clear ‘turn on’ response was observed upon the exogenous addition of GSH, Cys and HCys. Furthermore, TCFCl-GSH was able to evaluate the effects of drug treatment and the addition of ROS ( $\text{H}_2\text{O}_2$ ) on live cells, both of which resulted in a depletion of cellular GSH levels and a reduced fluorescence intensity. Subsequent, addition of NAC increased the GSH levels and enhanced the observed fluorescence intensity.

We would like to thank the EPSRC, the University of Bath and Prostate Cancer UK (PG14-009) for funding. ACS and JEG thank the EPSRC for studentships. TDJ wishes to thank the Royal Society for a Wolfson Research Merit Award. NMR characterisation facilities were provided through the Chemical Characterisation and Analysis Facility (CCAF) at the University of Bath ([www.bath.ac.uk/ccaf](http://www.bath.ac.uk/ccaf)). The EPSRC UK National Mass Spectrometry Facility at Swansea University is thanked for analyses. JY thanks the support from the National Research Foundation of Korea (NRF), which was funded by the Korea government (MSIP) (No. 2012RIA3A2048814). ACS, MY, MDL and TDJ are members of the Cancer Research@Bath (CR@B) network.

## Conflicts of interest

No conflicts of interest.

## Notes and references

- 1 L. B. Poole, *Free Radical Biol. Med.*, 2015, **80**, 148–157.
- 2 G. Y. Wu, Y. Z. Fang, S. Yang, J. R. Lupton and N. D. Turner, *J. Nutr.*, 2004, **134**, 489–492.
- 3 S. C. Lu, *Mol. Aspects Med.*, 2009, **30**, 42–59.
- 4 J. S. Armstrong, K. K. Steinauer, B. Hornung, J. M. Irish, P. Lecane, G. W. Birrell, D. M. Peehl and S. J. Knox, *Cell Death Differ.*, 2002, **9**, 252–263.
- 5 M. Valko, D. Leibfritz, J. Moncol, M. T. D. Cronin, M. Mazur and J. Telsler, *Int. J. Biochem. Cell Biol.*, 2007, **39**, 44–84.
- 6 C. B. Pocerich and D. A. Butterfield, *Biochim. Biophys. Acta*, 2012, **1822**, 625–630.
- 7 N. Ballatori, S. M. Krance, S. Notenboom, S. J. Shi, K. Tieu and C. L. Hammond, *Biol. Chem.*, 2009, **390**, 191–214.
- 8 J. Markovic, C. Borrás, A. Ortega, J. Sastre, J. Vina and F. V. Pallardo, *J. Biol. Chem.*, 2007, **282**, 20416–20424.
- 9 A. C. Sedgwick, R. S. L. Chapman, J. E. Gardiner, L. R. Peacock, G. Kim, J. Yoon, S. D. Bull and T. D. James, *Chem. Commun.*, 2017, **53**, 10441–10443.
- 10 A. C. Sedgwick, X. L. Sun, G. Kim, J. Yoon, S. D. Bull and T. D. James, *Chem. Commun.*, 2016, **52**, 12350–12352.



- 11 A. C. Sedgwick, H. H. Han, J. E. Gardiner, S. D. Bull, X. P. He and T. D. James, *Chem. Commun.*, 2017, **53**, 12822–12825.
- 12 A. C. Sedgwick, H.-H. Han, J. E. Gardiner, S. D. Bull, X.-P. He and T. D. James, *Chem. Sci.*, 2018, **9**, 3672–3676.
- 13 A. C. Sedgwick, A. Hayden, B. Hill, S. D. Bull, R. B. P. Elmes and T. D. James, *Front. Chem. Sci. Eng.*, 2018, DOI: 10.1007/s11705-017-1697-0.
- 14 D. Wu, A. C. Sedgwick, T. Gunnlaugsson, E. U. Akkaya, J. Yoon and T. D. James, *Chem. Soc. Rev.*, 2017, **46**, 7105–7123.
- 15 S. Q. Wang, S. L. Shen, Y. R. Zhang, X. Dai and B. X. Zhao, *Chin. J. Org. Chem.*, 2014, **34**, 1717–1729.
- 16 L. L. Yin, Z. Z. Chen, L. L. Tong, K. H. Xu and B. Tang, *Chin. J. Anal. Chem.*, 2009, **37**, 1073–1081.
- 17 Y. M. Yang, Q. Zhao, W. Feng and F. Y. Li, *Chem. Rev.*, 2013, **113**, 192–270.
- 18 Y. Qi, Y. Huang, B. W. Li, F. Zeng and S. Z. Wu, *Anal. Chem.*, 2018, **90**, 1014–1020.
- 19 J. Li, Y. Kwon, K. S. Chung, C. S. Lim, D. Lee, Y. K. Yue, J. Yoon, G. Kim, S. J. Nam, Y. W. Chung, H. M. Kim, C. X. Yin and J. H. Ryu, *Theranostics*, 2018, **8**, 1411–1420.
- 20 L. Chen, J.-S. Park, D. Wu, C.-H. Kim and J. Yoon, *Sens. Actuators, B*, 2018, **262**, 306–312.
- 21 Y. J. Wang, Y. Shi, Z. Y. Wang, Z. F. Zhu, X. Y. Zhao, H. Nie, J. Qian, A. J. Qin, J. Z. Sun and B. Z. Tang, *Chem. – Eur. J.*, 2016, **22**, 9784–9791.
- 22 W. Shu, L. G. Yan, Z. K. Wang, J. Liu, S. Zhang, C. Y. Liu and B. C. Zhu, *Sens. Actuators, B*, 2015, **221**, 1130–1136.
- 23 Y. H. Yang, J. L. Liu, H. Y. Xiao, Z. Zhen and S. H. Bo, *Dyes Pigm.*, 2017, **139**, 239–246.
- 24 T. Yu, G. X. Yin, P. Yin, Y. Zeng, H. T. Li, Y. Y. Zhang and S. Z. Yao, *RSC Adv.*, 2017, **7**, 24822–24827.
- 25 B. C. Zhu, H. Kan, J. K. Liu, H. G. Liu, Q. Wei and B. Du, *Biosens. Bioelectron.*, 2014, **52**, 298–303.
- 26 J. Bouffard, Y. Kim, T. M. Swager, R. Weissleder and S. A. Hilderbrand, *Org. Lett.*, 2008, **10**, 37–40.
- 27 M. Ipuay, C. Billon, G. Micouin, J. Samarut, C. Andraud and Y. Bretonniere, *Org. Biomol. Chem.*, 2014, **12**, 3641–3648.
- 28 W. L. Wu, Z. Y. Wang, X. Dai, J. Y. Miao and B. X. Zhao, *Sci. Rep.*, 2016, **6**, 25315.
- 29 R. Banerjee, *J. Biol. Chem.*, 2012, **287**, 4397–4402.
- 30 P. Ganguly and S. F. Alam, *J. Nutr.*, 2015, **14**, 6.
- 31 S. Y. Sun, *Cancer Biol. Ther.*, 2010, **9**, 109–110.

