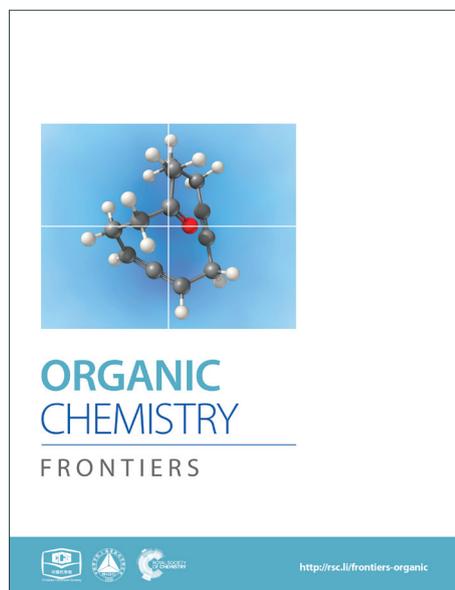
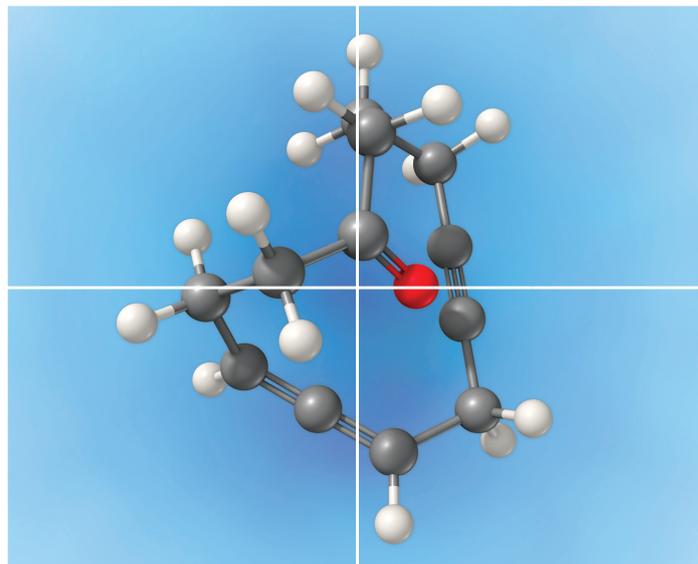


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Cleavage of the C-C Triple Bond of the Ketoalkynes: Synthesis of 4(3*H*)-Quinazolinones

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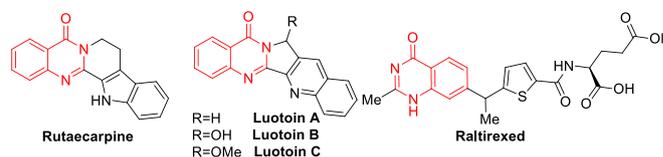
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A novel protocol for the synthesis of 4(3*H*)-quinazolinones via selectively cleavage of the triple bond of ketoalkynes under oxidant-, metal-, and ligand-free conditions has been developed. Various 4(3*H*)-quinazolinones were obtained through fragmentation of C-C triple bond and formation of two C-N bonds.

Quinazolinones widely present in natural products, such as Rutaecarpine from a Chinese herbal drug named Wu-Chu-Yu was used to treat headache and cholera;¹ Luotonin A from a Chinese plant (*Peganum nigellastrum*) showed cytotoxicity to the murine leukemia P388 cell line;² and Raltitrexed (brand name: Tomudex) have played an effective role in clinical therapy.³ They also exhibit wide range of bioactivities, such as anticonvulsant,⁴ antiviral,⁵ anti-inflammatory,⁶ antifungal,⁷ antimicrobial,⁸ and antimalarial.⁹ As a result, tremendous synthetic efforts have been made for building such a core. The condensation of aldehydes with *o*-aminobenzamides followed by oxidation has been used in the industries.¹⁰ Recently, the transition metal-catalyzed synthesis of quinazolines has made great progress. Zhou and Fang built such a structure via Iridium-catalyzed hydrogen transfers.¹¹ Yokoyama and Hidemasa described a Pd-catalyzed benzylic C-H amidation with benzyl alcohols.¹² Zhu and other groups developed Pd-catalyzed CO insertion to construct quinazolin-4(3*H*)-ones.¹³ Very recently, copper or iron-catalyzed C-N coupling of 2-halobenzoic acids or 2-halobenzamides with amidines to form quinazolines has been developed by Shafir, Buchwald and other groups.¹⁴ These procedures rely on metal catalysts and excessive oxidant or base.

Suitable ligands or microwaves were also necessary in some cases. The reaction of β -diketones with *o*-aminobenzamide to produce 4(3*H*)-quinazolinones has been mentioned in literatures.¹⁵ However, the generality of the β -diketones was limited and low efficiency were observed when 1,3-diarylpropane-1,3-diones were used as the starting materials. We reasoned that 1,3-diarylpropane-1,3-dione perhaps hardly generate enaminone intermediate. Recently, we easily synthesized a series of heterocycles, such as pyrroles, chromones and triazoles,¹⁶ from the ketoalkynes through an enaminone intermediate, which could be generated *in situ* from the reaction of ketoalkyne with *o*-aminobenzamide in the presence of suitable additives. Herein, we report a new strategy for the synthesis of 2-arylquinazolin-4(3*H*)-one from ketoalkynes *via* C-C triple bond cleavage.

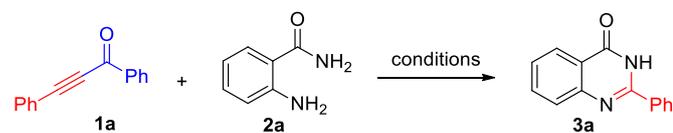


Scheme 1. Examples illustrating the importance of quinazolinones

In our initial studies, *o*-aminobenzamide **1a** and 1,3-diphenylprop-2-yn-1-one **2a** were chosen as model substrates to

optimize the reaction parameters. The desired product 2-phenylquinazolin-4(3*H*)-one **3a** were obtained by using TFA as an additive in THF, as well acetophenone as a byproduct. No product was detected in the absence of TFA. Different solvents were tested

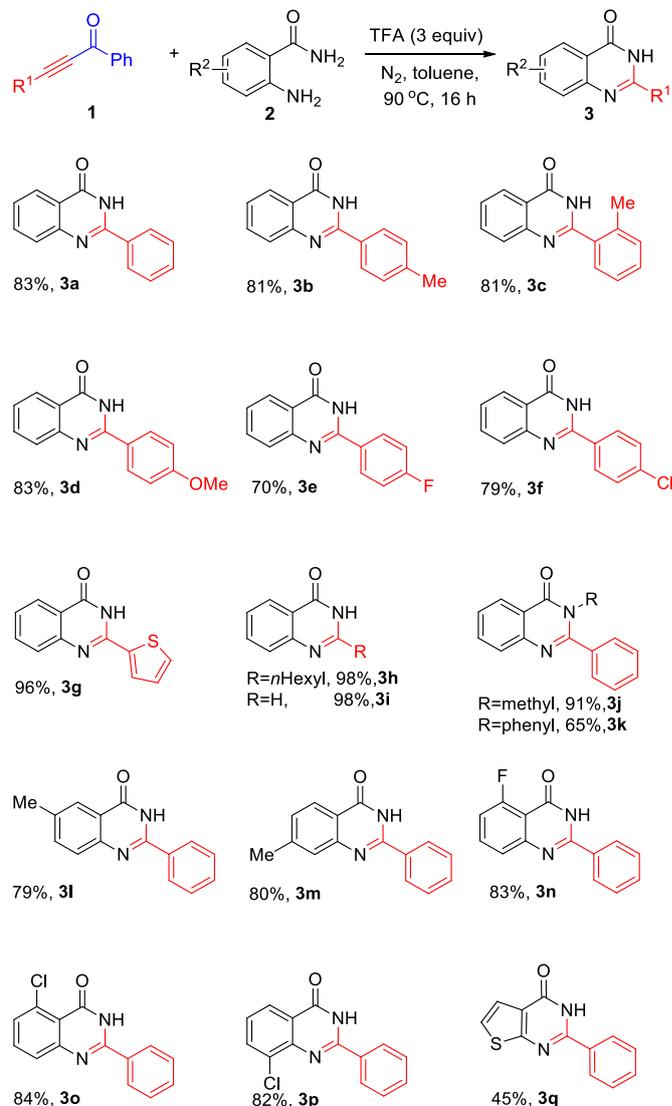
Table 1. Optimizing Reaction Parameters for the Condensation of *o*-Aminobenzamides **1a** with 1,3-Diphenylprop-2-yn-1-one **2a**^a



Entry	Additive (equiv)	Solvent	Yield ^b (%)
1	TFA (5)	THF	78
2	TFA (5)	DMSO	79
3	TFA (5)	ethanol	79
4	TFA (5)	DCM	71
5	TFA (5)	DMF	40
6	TFA (5)	H ₂ O	38
7	TFA (5)	toluene	83
8	TFA (5)	NMP	55
9 ^c	TFA (5)	DCM	67
10	TFA (4)	toluene	82
11	TFA (3)	toluene	83
12	TFA (2)	toluene	78
13 ^e	TFA (3)	toluene	62
14 ^d	TFA (5)	toluene	30
15	E ₃ N (5)	toluene	0
16	LiOtBu (5)	toluene	0

^a Reaction conditions: **1a** (0.65 mmol), **2a** (0.5 mmol), dry solvent (2 mL), 90 °C, overnight, N₂. ^b Isolated yields based on **2a**. ^c 60 °C, 16 h. ^d RT ^e **1a** : **2a** = 1:1.

Scheme 2. Substrate scope.^a



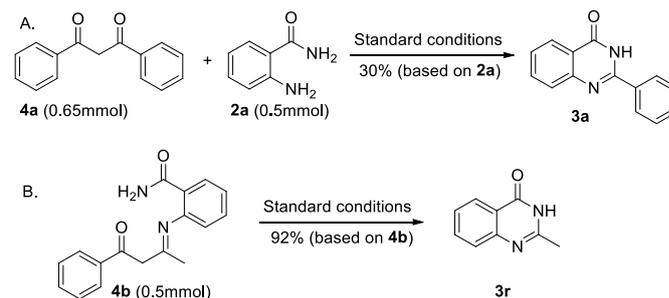
^a Reaction conditions: **1a** (0.65 mmol), **2a** (0.5 mmol), TFA (1.5 mmol), dry toluene (2 mL), 90 °C, 16 h. Isolated yields based on **2a**.

with 5 equiv of TFA at 90 °C (Table 1, entries 1–9). The highest yield was obtained in toluene (83%). Ethanol and DMSO led to **3a** in 79% yield (Table 1 entries 2, 3). The yield decreased significantly when NMP, DMF, and H₂O were used as the reaction media. Then the additive loading was investigated. This transformation was not affected when the acid additive was decreased to 3 equiv (Table 1, entry 11). However, further decreasing loading of TFA resulted in

relatively lower yield of **3a** (Table 1, entry 12). 1 Equiv of **1a** made the yield of **3a** decrease to 62% (entry 13). When the reaction temperature was decreased to room temperature, only 30% yield of 2-aryl-4(3*H*)-quinazolinone was obtained (Table 1, entry 14). The base, such as E_3N and $LiOtBu$, did not favour this transformation (Table 1, entries 15-16). Therefore, the optimal reaction conditions were identified as follows: 3 equiv of TFA at 90 °C in toluene, overnight.

Having the optimal reaction conditions in hand, the scope and generality of the substrates were investigated. As shown in Scheme 2 (**3a–o**), good to excellent yields were provided for the most substrates examined. 1,3-Diphenylprop-2-yn-1-one and diarylprop-2-yn-1-one bearing electron-donating groups in the 3-aryl ring, such as methyl and methoxyl, proceeded well, affording 2-arylquinazolin-4(3*H*)-ones **3a–d** in good yields (Scheme 1). Meanwhile, *o*-methyl-substituted substrates conducted smoothly and resulted in the desired 4(3*H*)-quinazolinones in 81% yield (Scheme 2, **3b vs 3c**), implying that influence of steric hindrance of ketoalkynes was not significant. In addition, ketoalkynes with electron-withdrawing groups (fluoro- and chloro-) also reacted well and furnished the desired products **3e**, **3f** in 70% and 79% yield, respectively. Heteroaryl- and aliphatic ketoalkyne were also applicable under the standard reaction conditions. For instance, 1-phenyl-3-(thiophen-2-yl)prop-2-yn-1-one, 1-phenylnon-2-yn-1-one and 1-phenylprop-2-yn-1-one gave the corresponding quinazolin-4(3*H*)-ones **3g**, **3h**, **3i** in 96%, 98%, and 98% yield, respectively. We next examined reactions of *N*-substituted 2-aminobenzamides, such as 2-amino-*N*-methylbenzamide and 2-amino-*N*-phenylbenzamide. The corresponding products **3j**, **3k** could be afforded in 91% and 65% yields, respectively. Next, the scope of the coupling reaction of substituted *o*-aminobenzamides **2l–p** was further examined. We were pleased to find that *o*-aminobenzamide with 3-chloro, 4-methyl, 5-

methyl, 6-chloro, and 6-fluoro groups also worked well and provided quinazolin-4(3*H*)-one derivatives **3l–p** in 79%–84% yields. To our delight, 2-aminothiophene-3-carboxamide could be a suitable substrate and resulted in the desired product in 45% yield (Scheme 2, **3o**).



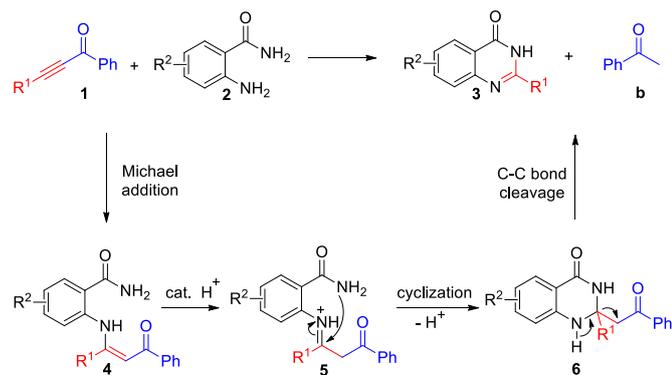
Scheme 3. Control Experiment

To clarify the reaction mechanism, 1.5 equiv of 1,3-diphenylpropane-1,3-dione (Scheme 3, **4a**) was treated with *o*-aminobenzamide under the standard reaction conditions. 4(3*H*)-Quinazolinone **3a** was isolated only in 30% yield. These results indicated that β -diketones as the possible intermediates of this metal-free C–C triple bond cleavage reaction was excluded. Meanwhile, a relatively accessible key intermediate (E)-2-(4-oxo-4-phenylbutan-2-ylideneamino) benzamide was successfully synthesized and resulted in the desired 4(3*H*)-quinazolinones in 92% yield under the standard reaction conditions. Based on the results obtained above and the literature¹⁷, a plausible reaction mechanism was proposed and shown in Scheme 4. With the help of TFA, the condensation reaction of **1** with **2** would generate the enaminone intermediate **4**. Then, intramolecular nucleophilic addition of **4** produced the intermediate **6**. Then the C–C bond cleavage reaction occurred to generate the product **3**.

In summary, we have demonstrated that the novel metal-free C–C triple bond cleavage reaction proceeded efficiently with TFA, affording 2-aryl(alkyl)-quinazolin-4(3*H*)-ones in moderate to excellent yields. In addition, the scope of this reaction was

successfully expanded to heteroaryl ketoalkyne and 2-aminothiophene-3-carboxamide. Currently, we are exploring the application of the proposed cleavage of the C–C triple bond with TFA to the synthesis of the pharmaceutical molecules.

Scheme 4. Plausible Reaction Mechanism



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Notes and references

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- Chen, A. L.; Chen, K. K. *J. Am. Pharm. Assoc.* **1933**, 22, 716.
- Ma, Z. -Z.; Hano, Y.; Nomura, T.; Chen, Y.-J. *Heterocycles*. **1997**, 46, 541.
- Nozoe, Y.; Ogata, Y.; Araki, Y.; Sasatomi, T.; Fukumori, H.; Shirouzu, K. *Anticancer res.* **2002**, 23, 4663.
- Aly, M. M.; Mohamed Y. A.; El-Bayouki, K. A.; Basyouni, W. M.; Abbas, S. Y. *Eur. J. Med. Chem.* **2010**, 45, 3365.
- Wang, Z. W.; Wang, M. X.; Yao, X.; Li, Y.; Tan, J.; Wang, L. Z.; Qiao, W. T.; Geng, Y. Q.; Liu, Y. X.; Wang, Q. M. *Eur. J. Med. Chem.* **2012**, 53, 275.
- (a) Lowe III, J. A.; Archer, R. L.; Chapin, D. S.; Cheng, J. B.; Helweg, D.; Johnson, J. L.; Koe, B. K.; Lebel, L. A.; Moore, P. F.

- Nielsen, J. A.; Russo, L. L.; Shirley, J. T. *J. Med. Chem.* **1991**, 34, 624.
- (b) de Laszlo, S. E.; Quagliato, C. S.; Greenlee, W. J.; Patchett, A. A.; Chang, R. S. L.; Lotti, V. J.; Chen, T.-B.; Scheck, S. A.; Faust, K. A.; Kivlighn, S. S.; Schorn, T. S.; Zingaro, G. J.; Siegl, P. K. S. *J. Med. Chem.* **1993**, 36, 3207.
- Liverton, N. J.; Armstrong, D. J.; Claremon, D. A.; Remy, D. C.; Baldwin, J. J.; Lynch, R. J.; Zhang, G. X.; Gould, R. J. *Bioorg. Med. Chem. Lett.* **1998**, 8, 483.
- (a) Habib, O. M.; Moawad, E. B.; Girges, M. M.; El-Shafei, A. M. *Boll. Chim. Farm.* **1995**, 134, 503. (b) Ibrahim, S. S.; Abdel-Halim, A. M.; Gabr, Y.; El-Edfawy, S.; Abdel-Rahman, R. M. *J. Chem. Res. Synop.* **1997**, 5, 154.
- (a) Jang, C. S.; Fu, F. Y.; Wang, C. Y.; Huang, K. C.; Lu, G.; Thou, T. C. *Science*. **1946**, 103, 59. (b) Chou, T.-Q.; Fu, F. Y.; Kao, Y. S. *J. Am. Chem. Soc.* **1948**, 70, 1765. (c) Murata, K.; Takano, F.; Fushiya, S.; Oshima, Y. *J. Nat. Prod.* **1998**, 61, 729.
- (a) Zhan, D.; Li, T. B.; Zhang, X. P.; Dai, C.; Wei, H. D.; Zhang, Y. Y.; Zeng, Q. L. *Synthetic Commun.* **2013**, 43, 2493. (b) Davoodnia, A.; Allameh, S.; Fakhari, A. R.; Tavakoli-Hoseini, N. *Chin. Chem. Lett.* **2010**, 21, 550. (c) Naleway, J. J.; Fox, C. M.; Robinhold, D.; Terpetschnig, E.; Olson, N. A.; Haugland, R. P. *Tetrahedron Lett.* **1994**, 35, 8569. (d) Bakavoli, M.; Sabzevari, O.; Rahimizadeh, M. *Chin. Chem. Lett.* **2007**, 18, 1466.
- Zhou, J.; Fang, J. *J. Org. Chem.* **2011**, 76, 7730.
- Hikawa, H.; Ino, Y.; Suzuki, H.; Yokoyama, Y. *J. Org. Chem.* **2012**, 77, 7046.
- (a) Ma, B.; Wang, Y.; Peng, J. L.; Zhu, Q. *J. Org. Chem.* **2011**, 76, 6362. (b) Wu, X. F.; He, L.; Neumann, H.; Beller, M. *Chem. Eur. J.* **2013**, 19, 12635. (c) Li, H. Q.; He, L.; Neumann, H.; Beller, M.; Wu, X.-F.; *Green Chem.* **2014**, 16, 1336.
- (a) Liu, X. W.; Fu, H.; Jiang, Y. Y.; Zhao, Y. F. *Angew. Chem. Int. Ed.* **2009**, 48, 348. (b) Zhang, X. D.; Ye, D. J.; Sun, H. F.; Guo, D. L.; Wang, J.; Huang, H.; Zhang, X.; Jiang, H. L.; Liu, H. *Green Chem.* **2009**, 11, 1881. (c) Yang, D. S.; Fu, H.; Hu, L. M.; Jiang, Y. Y.; Zhao, Y. F. *J. Comb. Chem.* **2009**, 11, 653. (d) Xu, W.; Jin, Y. B.; Liu, H. X.; Jiang, Y. Y.; Fu, H. *Org. Lett.* **2011**, 13, 1274. (e) Xu, L. T.; Jiang, Y. W.; Ma, D. W. *Org. Lett.* **2012**, 14, 1150.
- (a) Lu, L.; Zhang, M. M.; Jiang, H.; Wang, X. S. *Tetrahedron Lett.* **2013**, 54, 757. (b) Lessel, J. *Arch. Pharm.* **1994**, 327, 571.
- (a) Shen, J. H.; Cheng, G. L.; Cui, X. L. *Chem. Commun.* **2013**, 49, 10641. (b) Wang, X. S.; Cheng, G. L.; Cui, X. L. *Chem. Commun.* **2014**, 50, 652. (c) Cheng, G. L.; Zeng, X. B.; Shen, J. H.; Wang, X. S.; Cui, X. L. *Angew. Chem. Int. Ed.* **2013**, 52, 13265.
- Roy, S.; Davydova, M. P.; Pal, R.; Gilmore, K.; Tolstikov, G. A.; Vasilevsky, S. F.; Alabugin, I. V. *J. Org. Chem.* **2011**, 76, 7482.