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## COMMUNICATION

## Amplifying vibrational circular dichroism by manipulation of the electronic manifold†

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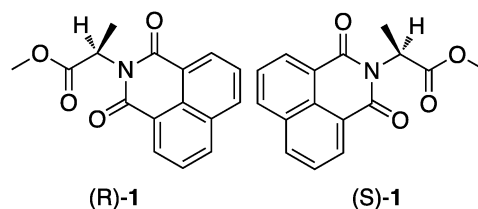
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Vibrational circular dichroism is a powerful technique to study the stereochemistry of chiral molecules, but often suffers from small signal intensities. Electrochemical modulation of the energies of the electronically excited state manifold is now demonstrated to lead to an order of magnitude enhancement of the differential absorption. Quantum-chemical calculations show that increased mixing between ground and excited states is at the origin of this amplification.

The development of tools to unambiguously assign the absolute configuration of naturally occurring and synthetic chiral molecules is of major importance for biochemical and pharmaceutical research. Vibrational circular dichroism (VCD), the infrared circular dichroism associated with vibrational transitions of chiral molecules, is a powerful spectroscopic method for this purpose since it makes it possible to determine the absolute configuration and conformation (or distribution of conformations) of a molecule without reference to any previous empirical rules. However, in many cases VCD measurements are difficult because of the small signal magnitudes (the differential absorption being typically  $10^{-5}$  of the absorption itself). Such small intensities can be measured routinely nowadays, but at the expense of long measuring times. Strategies that would enhance VCD signals and thereby improve the ability of accurately differentiating between optical isomers by means of their vibrational signatures could therefore be of substantial importance.

It has been reported previously that molecular systems with low-lying electronically excited states can exhibit strongly enhanced VCD signals.<sup>1–3</sup> The origin of this enhancement can be understood from the expression for the vibrational magnetic-dipole transition moment derived in the vibronic coupling theory for VCD.<sup>4,5</sup> Here, we present a novel approach to amplify VCD signals by modulation of the energies of the excited-state manifold in a controlled way using spectroelectrochemistry.<sup>6–8</sup>



**Fig. 1** Chemical structure of compounds (R)- and (S)-methyl 2-(1,3-dioxo-1H-benzo[de]isoquinolin-2(3H)-yl)propanoate, (R)-1 and (S)-1 respectively.

To investigate the effect of electrochemical reduction on the VCD response, we measured IR and VCD spectra of compounds (+)-1 and (–)-1 (Fig. 1) both in their neutral and radical anion forms (see ESI† for experimental details). Here, our main focus is on the comparison of the peak intensities in the VCD spectrum of the neutral and of the radical anion. A full conformational analysis of the IR and VCD spectra is given in the ESI.†

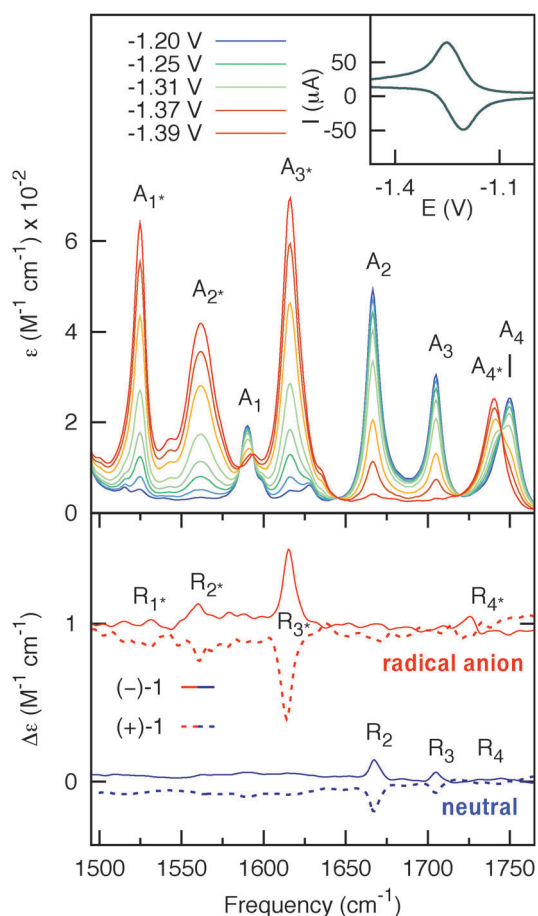
The FTIR spectrum of neutral (–)-1 is shown in Fig. 2 (blue curve). In the 1500–1780  $\text{cm}^{-1}$  frequency range four dominant vibrational modes ( $A_{1-4}$ ) can be identified and assigned by comparison with the literature<sup>9</sup> (*vide infra*). The FTIR spectrum of the radical anion (red curve in Fig. 2) was obtained upon complete reduction of a 35 mM solution of (–)-1 in  $\text{CD}_3\text{CN}$  ( $10^{-1}$  M TBAH) at 293 K. The conventional cyclic voltammogram for (–)-1 shows a completely reversible cathodic wave (see ESI†). The observed reduction potential ( $E_{1/2} = -1.3$  V vs.  $\text{Fc}/\text{Fc}^+$ ) is in good agreement with previously reported values for similar systems.<sup>9</sup> Full reversibility was also achieved with TL-CV on time scales of 2  $\text{mV s}^{-1}$  (inset of Fig. 2). Intermediate spectra were recorded in small potential steps across the cathodic wave, and IR spectral changes were monitored at each step.

Comparison of the IR spectra of neutral and anionic (–)-1 shows that the IR bands  $A_2$  and  $A_3$  of the neutral at 1704  $\text{cm}^{-1}$  and 1667  $\text{cm}^{-1}$ , assigned to the symmetric and asymmetric C=O stretching modes of the naphthalimide moiety, are red-shifted in the radical anion to 1615  $\text{cm}^{-1}$  ( $A_{2*}$ ) and 1562  $\text{cm}^{-1}$  ( $A_{3*}$ ), respectively. The naphthalimide aromatic C=C-stretching mode ( $A_1$ ) observed in neutral (S)-1 at 1589  $\text{cm}^{-1}$  is red-shifted to 1525  $\text{cm}^{-1}$  ( $A_{1*}$ ) in the radical anion. The methyl ester C=O-stretching mode of the amino-acid moiety ( $A_4$ )

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**Fig. 2** Upper panel: potential-dependent steady-state IR spectra of 35 mM (–)-1 (optical path = 200  $\mu\text{m}$ ). The blue and red curves represent the spectra of the initial (neutral) and final states (radical anion), respectively. Inset: thin-layer cyclic voltammogram of (–)-1 obtained using the IR OTTLE cell. Lower panel: VCD spectra of 7 mM (–)-1 (solid-line) and 7 mM (+)-1 (dashed-line) (optical path = 1.2 mm) for the neutral (blue) and radical anion (red) species. The spectra for the radical anion are vertically offset for clarity.

at 1750  $\text{cm}^{-1}$  is red-shifted to 1740  $\text{cm}^{-1}$  ( $A_{4*}$ ) in the radical anion. The VCD spectra of 7 mM solution of neutral and anionic (+)-1 and (–)-1 are displayed in the lower panel of Fig. 2. Spectra for both enantiomers have been obtained by averaging three consecutive sets of 20 min scans for the neutral, and three sets of 2 min scans from independent but consecutive CV cycles for the radical anion. For neutral (+)-1 and (–)-1, three features  $R_2$ ,  $R_3$  and  $R_4$  can be observed in the VCD spectra. These bands are red-shifted for the radical anion of (+)-1 and (–)-1 to the features  $R_{2*}$ ,  $R_{3*}$  and  $R_{4*}$ , respectively. Moreover, the spectra show another band  $R_{1*}$  that is not visible in the spectrum of the neutral. Importantly, we find that the VCD signals are enhanced up to one order of magnitude compared to the neutral species. As will be argued in the following, this enhancement is due to a strong vibronically induced mixing of low-lying electronic excited states with the ground state in the electrochemically generated radical anion species. The peak intensities in VCD spectra are proportional to the rotational strength, given by:<sup>10</sup>

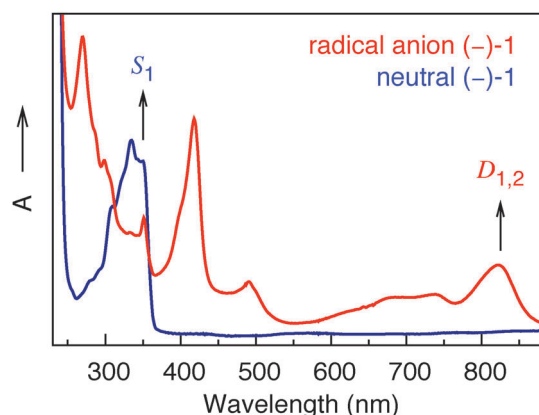
$$R = \text{Im}[\langle \Psi_i | \mu_{\text{el}} | \Psi_f \rangle \langle \Psi_f | \mu_{\text{mag}} | \Psi_i \rangle] \quad (1)$$

where  $\mu_{\text{el}}$  and  $\mu_{\text{mag}}$  are the electric and magnetic dipole moment operators, respectively, and  $|\Psi_i\rangle$  and  $|\Psi_f\rangle$  are the total wave functions for the initial and final states. It has been shown previously<sup>11</sup> that within the Born–Oppenheimer (BO) approximation, the vibrational magnetic transition dipole moment  $\mu_{\text{mag}}$  is identically zero, and that VCD can be described only if vibronically induced mixing of the BO wave functions of the electronically excited states and the ground state is taken into account.<sup>4,12</sup> In particular, it can be shown that when a molecule is in an electronic state  $|\psi_0\rangle$ , the magnetic transition dipole moment of a transition between the  $\nu = 0$  and  $\nu = 1$  levels of a vibrational mode is given to first order by:<sup>4,12</sup>

$$\langle \Psi_f | \mu_{\text{mag}} | \Psi_i \rangle = \langle \chi_{\nu=0} | \sum_{n \neq 0} \frac{\langle \psi_0 | \mu_{\text{mag}}^e | \psi_n \rangle}{E_n - E_0} (\langle \psi_n | T_{\text{nuc}} | \psi_0 \rangle - \langle \psi_0 | T_{\text{nuc}} | \psi_n \rangle) | \chi_{\nu=1} \rangle \quad (2)$$

where  $|\chi_{\nu=0}\rangle$  and  $|\chi_{\nu=1}\rangle$  are the nuclear wave functions of the  $\nu = 0$  and  $\nu = 1$  states in the electronic ground state  $|\psi_0\rangle$ , and  $T_{\text{nuc}}$  is the nuclear kinetic energy operator.  $\mu_{\text{mag}}^e$  is the electronic magnetic transition dipole moment,  $|\psi_0\rangle$  and  $|\psi_n\rangle$  are the BO electronic wave functions for the ground state and the  $n^{\text{th}}$  electronically excited state, with energies  $E_0$  and  $E_n$ , respectively.

For large energy gaps between ground and electronically excited states, the coefficients in the perturbation expansion,  $\langle \psi_0 | \mu_{\text{mag}}^e | \psi_n \rangle / (E_n - E_0)$  (see eqn (2)), are small due to large energy denominators. As a consequence, the VCD signals are very small. This is the case for nearly all closed-shell organic molecular systems. In open-shell species, on the other hand, the first electronically excited states are generally at much lower energies. Eqn (2) shows that this leads to a strong increase in the magnetic transition-dipole moments, and thus in the VCD amplitudes. Indeed, the UV-Vis absorption spectrum of the radical anion displayed in Fig. 3 shows absorption bands at much longer wavelengths than the corresponding spectrum of the neutral system. We have confirmed the presence of these low-energy states and identified their electronic characteristics using time-dependent density functional theory (TD-DFT) calculations on the most stable conformer of the radical anion (see ESI†). The low-energy nature of these excited states



**Fig. 3** UV-Vis spectra of (–)-1 in the neutral (blue) and radical anion (red) forms.

induces mixing of their wave functions into the ground state, increasing the total number of adiabatic electronic states contributing to the vibrational eigenfunctions, and leading to strongly enhanced VCD signals, as confirmed by our experimental results.

From the TD-DFT calculations we can assign two almost degenerate electronic states,  $D_1$  and  $D_2$ , contributing to the absorption band at 822 nm in Fig. 3. Interestingly, we find that the electronic magnetic transition-dipole moment  $\langle \psi_0 | \mu_{\text{mag}}^e | \psi_n \rangle$  between the ground state and  $D_2$  is one order of magnitude larger than for any other of the first ten electronically excited states. Eqn (2) thus tells us that  $D_2$  will be a dominant contributor to the vibrational magnetic transition dipole moment, not only because of the relatively small energy gap ( $E_2 - E_0$ ), but also because of the large value of the  $\langle \psi_0 | \mu_{\text{mag}}^e | \psi_n \rangle$  matrix element. It will therefore magnify the vibrational magnetic transition dipole moment, and thus the VCD intensity. The difference density plot between  $D_2$  and the ground state  $D_0$  shows that the electron density on the naphthalimide carbonyl groups is noticeably different in  $D_2$  and  $D_0$ , while the electron density on the carbonyl group near the methyl ester is the same in the two states. The dominant role of  $D_2$  in the expansion of the vibrational magnetic transition moment (eqn (2)) makes one therefore expect that the VCD of the naphthalimide C=O-stretching modes would be affected in particular. Our experimental results nicely confirm this expectation: we observe maximum VCD enhancement (one order of magnitude) for the symmetric naphthalimide C=O-stretching mode. It is satisfying to note that our observations are internally consistent in other respects, as well. From a purely energetic point of view,  $D_1$  might also be expected to make a significant contribution to the vibrational magnetic transition dipole moment. Difference density plots between  $D_1$  and  $D_0$  show, however, that the electron density on the naphthalimide carbonyl groups is hardly different in  $D_1$  and  $D_0$  (see ESI†). A dominant contribution of  $D_1$  in eqn (2) would thus be at odds with an enhancement of the naphthalimide C=O-stretching modes. Indeed, the influence

of  $D_1$  is greatly reduced in the end on account of the small electronic magnetic transition dipole moment from the ground state (see ESI†).

In conclusion, we have observed a strong enhancement of VCD signals upon the electrochemical reduction of a closed-shell organic compound in solution: the conversion from neutral molecule to radical anion increases the VCD signals by up to one order of magnitude. Such intensity-enhanced VCD arising from the presence of electrochemically “created” low-lying electronic states should be a general phenomenon in organic compounds. As such, electrochemical VCD can be a valuable tool for amplifying signals in molecules that have small VCD amplitudes, or that can be studied only at low concentrations.

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## Notes and references

- 1 C. Barnett, A. Drake, R. Kuroda, S. Mason and S. Savage, *Chem. Phys. Lett.*, 1980, **70**, 8–10.
- 2 Y. He, X. Cao, L. A. Nafie and T. B. Freedman, *J. Am. Chem. Soc.*, 2001, **123**, 11320–11321.
- 3 C. Johannessen and P. W. Thulstrup, *Dalton Trans.*, 2007, 1028–1033.
- 4 L. A. Nafie and T. B. Freedman, *J. Chem. Phys.*, 1983, **78**, 7108.
- 5 L. A. Nafie, *J. Phys. Chem. A*, 2004, **108**, 7222–7231.
- 6 W. E. Geiger, *Organometallics*, 2011, **30**, 28–31.
- 7 M. I. Bruce, N. N. Zaitseva, B. W. Skelton, A. H. White, M. A. Fox and P. J. Low, *Dalton Trans.*, 2010, 1222–1234.
- 8 D. R. Laws, D. Chong, K. Nash, A. L. Rheingold and W. E. Geiger, *J. Am. Chem. Soc.*, 2008, **130**, 9859–9870.
- 9 D. Jagesar, F. Hartl, W. Buma and A. Brouwer, *Chem.–Eur. J.*, 2008, **14**, 1935–1946.
- 10 D. J. Caldwell and H. Eyring, *The Theory of Optical Activity*, Wiley-Interscience, New York, 1971.
- 11 L. D. Landau and E. M. Lifschitz, *Quantum Mechanics, Non-relativistic Theory*, Butterworth-Heinemann, Oxford, 1999.
- 12 P. J. Stephens, *J. Phys. Chem.*, 1985, **89**, 748–752.