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Recent progress in monolithic two-terminal perovskite-based triple-junction solar cells

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Multi-junction solar cells are the most famous approach to overcome the power conversion efficiency (PCE) limit of single-junction solar cells. Metal halide perovskite absorber materials offer low-cost fabrication and tunable bandgap that make them suitable candidates for multi-junction applications. Perovskite-based dual-junction solar cells have already shown impressive PCE improvement in recent years. As a next step, triple-junction structures can allow for further PCE increase; however, research on triple-junction solar cells with perovskite materials is still at an early stage. In this review, we present the status of three monolithic perovskite-based triple-junction technologies available: perovskite/perovskite/silicon, perovskite/perovskite/perovskite, and perovskite/perovskite/organic solar cells. First, an overview of the potential of these solar cell types based on simulation is given. Further, a summary of scientific and experimental challenges in the field as well as strategies to achieve suitable absorber materials and contact layers that can be implemented in these structures are presented. Finally, global cell characterization of triple-junction solar cells together with a guideline on correct measurement of this type of solar cells are discussed.

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Broader context

Tackling climate change requires transformation of global energy from conventional fossil fuels toward cleaner sources. This is achievable through electrification of energy demand as well as increasing the share of renewable sources for electricity production. Among all the renewable energy sources, photovoltaics (PV) plays a crucial role. Even though the electricity share of the PV is only 4.5%, its world installed capacity has reached 1 TWdc and is further growing. Entering the terawatt scale raises the need for reducing the cost of PV technology. Therefore, increasing the efficiency of the PV module is more important than ever. To do so, multi-junction solar cells have proven to be the most attractive option, yet development of efficient and low-cost subcells is necessary. In this regard, metal halide perovskite is one of the most promising candidates with perovskite/silicon holding the dual-junction record efficiency. Improving this efficiency further is possible by increasing the number of junctions to three as it has been demonstrated by III-V semiconductor materials. Realizing the triple-junction solar cell with perovskite material comes with several challenges and requirements which we have outlined in this review.

1. Introduction

The most famous approach to overcome the efficiency limit of single-junction solar cells and achieve efficient usage of the solar spectrum is the use of multi-junction solar cells. In this approach, solar cells with different bandgaps are stacked on top of each other in a bandgap descending order with the highest

bandgap solar cell facing the sun. This way, the top solar cell absorbs the high energy photons up to its bandgap and the lower energy photons are transmitted to reach the underlying subcells with lower bandgaps. This reduces the thermalization losses and leads to better harvesting of the solar spectrum. Fig. 1a shows this principle for the case of triple-junction solar cells.

According to simulation by Philipps *et al.*, the highest theoretical PCE that can be achieved in a detailed balance limit (Auger recombination is not taken into account) by a dual-junction solar cell is 45.9% under Air Mass 1.5 global (AM 1.5g) spectrum.¹ This PCE limit can be exceeded by increasing the number of junctions in a solar cell. However, as shown in Fig. 1b the incremental increase becomes lower towards more junctions. Furthermore, the additional junctions introduce new

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Fig. 1 (a) Use of the solar spectrum by a triple-junction solar cell consisting of a silicon bottom cell, a 1.50 eV middle cell (e.g. GaAsP), and a 2.00 eV top cell (e.g. AlGaInP). The transmission and thermalisation losses are reduced compared to single-junction and dual-junction solar cells. (b) Theoretical efficiency of multi-junction solar cells in the radiative limit as a function of the number of junctions. The optimum bandgap combination is given for the maximum PCE under the AM 1.5g spectrum. The gain in efficiency becomes lower by increasing the number of junctions. Data taken from ref. 1.

challenges in terms of processing, choice of materials and interaction between the layers. Thus, triple-junction solar cells are a reasonable approach to achieve a good balance between efficiency potential, processing, and device complexity.

So far, highly efficient triple-junction solar cells are made of III–V semiconductor materials and are commercially available for concentrator photovoltaic and space application. Triple-junction solar cells using III–V materials reached 39.5% efficiency for a gallium indium phosphide (GaInP) top cell, gallium indium arsenide (GaInAs)/gallium arsenide phosphide (GaAsP) quantum well middle cell, and GaInAs bottom cell under the AM 1.5g spectrum.² This exceeded the previous record of a sextuple-junction solar cell and made the triple-junction solar cell the most efficient non-concentrated terrestrial PV technology to date. Cells made from III–V compound semiconductor absorbers are highly efficient thanks to the low defect density that can be achieved in the single-crystalline materials and due to the direct

nature of their bandgap. However, they require expensive deposition techniques such as metalorganic or hydride vapor phase epitaxy (MOVPE, HVPE) or molecular beam epitaxy (MBE) to achieve the required high crystal quality. The high production costs prevent these technologies from entering the terrestrial photovoltaic markets and consequently limit them to concentrator photovoltaic and space application where the high efficiency is the decisive factor. Several works have been done to couple III–V materials on a silicon bottom cell to reduce the cost and to benefit from the fact that the photovoltaic (PV) market is dominated by crystalline silicon. For the first time in 2013 Derendorf *et al.* reported successful integration of GaInP/GaAs/silicon in a triple-junction structure using wafer bonding.³ A record efficiency of 35.9% was achieved in 2021 for this type of solar cell.⁴ Technologically, wafer bonding is not favorable as it is an expensive serial process and requires an extra chemical-mechanical polishing (CMP) step.⁵ However, direct growth of III–V materials



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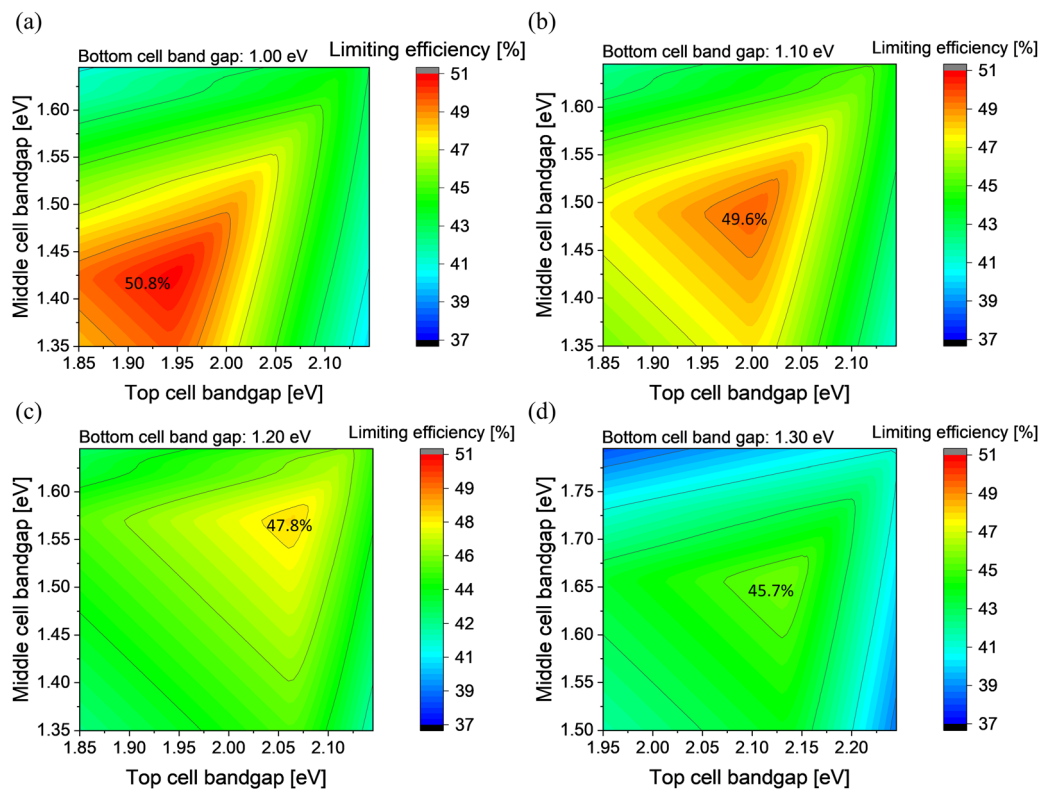


Fig. 3 Theoretical efficiencies in the radiative limit for triple-junction solar cells for a bottom cell absorber with a bandgap of (a) 1.00 eV, (b) 1.10 eV, (c) 1.20 eV, and (d) 1.30 eV. The optimum top and middle cell bandgaps in each case are (a) 1.94 eV and 1.42 eV, (b) 2.00 eV and 1.49 eV, (c) 2.06 eV and 1.57 eV, and (d) 2.13 eV and 1.66 eV.

Table 1 The optimum bandgap combinations to achieve highest theoretical PCE in a triple-junction solar cell with respect to the choice of the bottom cell

Bottom cell (eV)	Middle cell (eV)	Top cell (eV)
1.00	1.42	1.94
1.10	1.49	2.00
1.20	1.57	2.06
1.30	1.66	2.13

be current matched in the field as the spectra and temperature variations affect the bandgaps of the subcells in different ways.^{30–32} Furthermore, in a two-terminal configuration, the voltage of the final device is the sum of the voltages of all subcells. This means that to ensure highest voltage in the multi-junction solar cell, first the voltage deficit (*i.e.*, $V_{\text{deficit}} = E_g/q - V_{\text{OC}}$, where E_g is the bandgap, q is the electron elementary charge and V_{OC} the open-circuit voltage) of the individual subcells should be minimized and second lossless interconnection layers between the subcells are required.

So far, this introduction covered the potential of triple-junction solar cells based on simulation. Realization of such photovoltaic devices with maximum possible output in real lab conditions demands consideration of practical factors and limitations. Beyond optimization of individual layers, the processing sequence and layer interactions of the whole multilayer stack strongly influence the final device performance. Thus, a thorough analysis and comparison of the approaches taken for realizing the proof of

concepts of these new technologies as well as follow-up studies are needed to allow fast technological development. This paper discusses the progress in engineering of monolithic perovskite-based triple-junction solar cells comprehensively. First, device architectures and interlayers used in current reported perovskite-based triple-junction solar cells including their photovoltaic parameters are summarized. Then, a review of perovskite absorbers suitable for middle cell and top cell together with details on bandgap tuning of perovskites, different deposition techniques, challenges associated with each type of the perovskite absorber, as well as possible strategies to tackle them are presented. In addition, we discuss the available materials and methods that are used for the recombination layers between perovskite subcells. Lastly, precise characterization of triple-junction solar cells based on the publications from current state-of-the-art III-V triple-junction technologies is addressed. Special emphasis is placed on potential uncertainties arising from measurements, along with suggestions on how to avoid them in research laboratories.

2. Perovskite-based triple-junction solar cells: categories and status

A monolithically integrated multi-junction solar cell consists of many layers processed sequentially on top of each other with different deposition methods. Process compatibility of each layer with its underlying layer is therefore of great importance.





Fig. 4 Simulated external quantum efficiency (EQE) and current density–voltage (jV) curves for maximum practical PCE (a) 36.6% for an all-perovskite triple-junction solar cell with perovskite's bandgaps of 1.22 eV, 1.58 eV, and 2.04 eV for the subcells and (b) 38.8% for a perovskite/perovskite/silicon triple-junction solar cell with bandgaps of 1.44 eV and 1.95 eV for perovskite subcells. Reproduced with permission from ref. 27, Copyright © 2017, American Chemical Society.

For example, solvent orthogonality, possible processing damage as well as thermal budget of the bottom layers should be considered when processing subsequent layers. Moreover, addition of each subcell increases the number of layers, which consequently introduces new challenges.

Multi-junction solar cells can be divided into superstrate or substrate configuration (Fig. 5) depending on the type of bottom solar cell technology. In case of thin film bottom solar cells, such as in all-perovskite and perovskite/perovskite/organic solar cells, the superstrate configuration is used; solar cells are deposited on a glass (or foil) which is flipped and facing the sun in the final device. In case of wafer-based bottom solar cells, such as in perovskite/perovskite/silicon, the substrate configuration is used; the bottom cell wafer itself acts as the substrate and subsequent layers are processed on top. Consequently, the processing order of the absorbers is different. In the superstrate configuration, perovskites are processed in bandgap descending order and highest restrictions concerning process compatibility apply for the low bandgap perovskite absorber. In the substrate configuration, perovskite middle and top cells are processed successively on top of the bottom solar cell wafer. Highest

restrictions concerning compatibility, such as annealing temperature and possible solution damage to the underlying layers apply for the processing of high-bandgap perovskite top cell. One example is high-temperature annealed nickel oxide (NiO_x), that cannot be implemented as charge transport layer on top of another perovskite subcell. In addition, the first processed subcell (high bandgap in the superstrate and low bandgap in substrate configuration) must sustain several harsh conditions involved in the deposition of the top layers such as annealing steps, sputtering, *etc.* More discussion on fabrication of respective perovskite layers is presented in Section 3 and 4.

Moreover, perovskite solar cells can be classified by their order of deposition of the electron (n) and hole (p) transport and the perovskite absorber (i) layers into n–i–p (electron transport layer (ETL) deposited first) and p–i–n (hole transport layer (HTL) deposited first) device architectures. The work on perovskite/silicon tandem solar cells started on n–i–p architecture.³³ However, the research focus changed toward the p–i–n structure due to an optically more favorable front contact, ease of fabrication and upscaling.

Similarly, the very first report on all-perovskite triple-junction solar cells in 2019 by McMeekin *et al.* had an n–i–p





Fig. 5 Schematic illustration of monofacial perovskite-based triple-junction solar cells with (top) superstrate configuration and (bottom) substrate configuration. Perovskite subcells are in p–i–n architecture. Some of the common materials for each layer together with their deposition techniques are shown. The processing of substrate configuration starts with bottom cell (lowest bandgap subcell) while for superstrate configuration the top cell (the highest bandgap subcell) is processed first. The main limitations of the processes are highlighted.

configuration implementing $\text{FA}_{0.83}\text{Cs}_{0.17}\text{Pb}(\text{Br}_{0.70}\text{I}_{0.30})_3$ (1.94 eV), MAPbI_3 (1.55 eV) and $\text{MAPb}_{0.75}\text{Sn}_{0.25}\text{I}_3$ (1.34 eV) perovskite absorbers.²³ This work remains the only perovskite-based triple-junction solar cell with n–i–p architecture. Another exceptional approach regarding solar cell fabrication was that all the layers except for the silver electrode were solution-processed; even the indium tin oxide (ITO) recombination layer which is usually deposited *via* sputtering was replaced by spin coated ITO nanoparticles. Their strategy to overcome solvent compatibility issue was to change the conventional mixture of dimethylformamide/dimethyl

sulfoxide (DMF/DMSO) as perovskite solvent to a new acetonitrile/methylamine (ACN/MA) solvent system for both low bandgap and middle bandgap perovskites. However, the change in solvent system yielded a reduced PCE. The triple-junction solar cell reached a PCE of 6.7% with an open-circuit voltage (V_{OC}) of 2.70 V. The limited PCE mainly arose from low short-circuit current density (j_{SC}) of 8.3 mA cm^{-2} limited by the low bandgap bottom cell, as well as a low fill factor (FF) of 43.0%. Other possible factors limiting the performance of this structure could be high parasitic absorption, especially in the 60 nm thick



phenyl-C₆₁-butyric acid methylester (PCMB) as ETL layer, and a high V_{OC} deficit of the high bandgap perovskite top cell.

A year after this initial proof of concept, Wang *et al.* demonstrated an all-perovskite triple-junction solar cell in a p-i-n configuration based on Cs_{0.10}(FA_{0.66}MA_{0.34})_{0.90}PbI₂Br (1.73 eV), FA_{0.66}MA_{0.34}PbI_{2.85}Br_{0.15} (1.57 eV), and FA_{0.66}MA_{0.34}Pb_{0.50}Sn_{0.50}I₃ (1.23 eV) perovskite absorbers.²⁵ They employed a two-step deposition method for processing of all three perovskite layers. Another noticeable improvement in their work came from introducing an optimized interconnection layer stack of tin oxide (SnO_x)/gold (Au)/poly-(3,4-ethylenedioxythiophen)-poly-(styrolsulfonat) (PEDOT:PSS) between the subcells which showed good solvent barrier function. In addition, they improved the ETL and replaced the 80 nm thick solution processed PCBM with a thinner layer of C₆₀ (20 nm) deposited *via* thermal evaporation, which reduced parasitic absorption and charge accumulation in the interconnecting layers due to higher mobility of C₆₀ compared to PCBM. As a result, the champion cell exhibited 16.8% PCE with a j_{SC} of 7.3 mA cm⁻², a V_{OC} of 2.79 V, and a FF of 82.0%. The cell showed good stability during 300 s measurement at maximum power point. The high FF was the main factor that contributed to the PCE improvement compared to the previous work. The j_{SC} of their device was limited by non-optimized bandgaps of the subcells as well as high parasitic absorption in ITO substrate and PEDOT:PSS layer.

Also in 2020, Xiao *et al.* reported a 20.1% efficient all-perovskite triple-junction solar cell by employing perovskites with Cs_{0.2}FA_{0.8}PbI_{0.9}Br_{2.1} (1.99 eV), Cs_{0.05}FA_{0.95}PbI_{2.55}Br_{0.45} (1.60 eV), and FA_{0.7}MA_{0.3}Pb_{0.5}Sn_{0.5}I₃ (1.22 eV).²⁴ The bandgap of the top cell was higher compared to the two previous publications which is more suitable for all-perovskite triple-junction solar cells. The champion device showed 20.1% PCE with a j_{SC} of 8.8 mA cm⁻², a V_{OC} of 2.80 V, and a FF of 81.0%. The sample showed acceptable stability during 180 s of measurement at maximum power point in N₂ environment. However, a longer measurement for 4000 s led to degradation of performance. A strategy developed in their work was the introduction of a double hole transport layer of nickel oxide (NiO_x)/poly-[bis-(4-phenyl)-(2,4,6-trimethylphenyl)-amin] (PTAA) for the middle cell. Perovskite films typically feature a rough surface^{34,35} and it is challenging to form thin and conformal solution processed charge transport layers on top of them. Uncovered areas and pinholes can lead to shunting paths. Use of NiO_x alone and in combination with organic HTL is found to be beneficial to fully cover the rough perovskite surface. In their triple-junction solar cell, 80 nm NiO_x was deposited by spin coating of a pre-synthesized NiO_x nanocrystal dispersion followed by 5 minutes annealing at 100 °C. The NiO_x layer smoothed the surface and thus enabled improved spin coating of the thin PTAA layer on top. This strategy had been previously employed in a perovskite/CIGS tandem solar cell³⁶ where Jost *et al.* deposited a 10 nm conformal NiO_x layer by atomic layer deposition (ALD) without further annealing steps prior to deposition of PTAA, to avoid shunting in the perovskite cell processed on the rough CIGS surface.³⁶ It is worth mentioning that some other perovskite-based triple-junction solar cells, from all three types, have also incorporated

either NiO_x alone or double hole transport layers with NiO_x in their structure^{14,16,17} (Fig. 6).

Recently, Wang *et al.* published the first certified perovskite-based triple-junction solar cell.¹⁴ In this work, an all-perovskite triple-junction solar cell was made using Rb_{0.15}Cs_{0.85}PbI_{1.75}Br_{1.25} (2.00 eV), Cs_{0.05}FA_{0.90}MA_{0.05}Pb(I_{0.90}Br_{0.10})₃ (1.60 eV), and Cs_{0.05}FA_{0.70}MA_{0.25}Pb_{0.50}Sn_{0.50}I₃ with 5% SnF₂ (1.22 eV) perovskite absorbers. This cell showed a certified PCE of 23.3%, a j_{SC} of 9.6 mA cm⁻², a V_{OC} of 3.20 V and a FF of 76.2%. The encapsulated sample maintained 80% of its initial efficiency after 450 h of measurement in an ambient atmosphere. The main factor leading to this 23.3% efficient triple-junction cell was development of a rubidium (Rb), cesium (Cs) all-inorganic high bandgap perovskite absorber with improved stability. Details on this high bandgap perovskite are summarized in Section 4.3. Here, also a double HTL layer of NiO_x/Me-4PACz was employed for both top and middle subcells. The high V_{OC} of 3.2 V partially originated from the high-quality of the perovskite/charge transport layer interface. Me-4PACz has been shown to have a lossless interface with perovskite.³⁷ In addition, in this work the perovskite/PCBM interface was passivated with phenethylammonium iodide and ethane-1,2-diammonium iodide (PEAI-EDAI₂) to reduce the recombination loss at this interface. Moreover, to better protect the underlying layers, polyethyleneimine ethoxylated (PEIE) was spin coated on PCBM layer, prior to deposition of SnO_x similar to the approach that has been previously reported by Palmstrom *et al.*³⁸ As recombination layer between the top and middle cell, 20 nm of sputtered ITO was used whereas 1 nm evaporated Au served as the recombination layer between the middle and bottom cells. This work is the only perovskite-based triple-junction solar cell with certified efficiency.

The proof of concept for a perovskite/perovskite/silicon triple-junction solar cell was reported 1 year earlier than the all-perovskite triple-junction in 2018 by Werner *et al.* using a both-side textured silicon heterojunction (SHJ) bottom cell.²² In this work, perovskite solar cells had a p-i-n configuration. For deposition of perovskite layers on textured surfaces, a hybrid two-step deposition method was used, which the same group reported earlier for a both-side textured perovskite/silicon tandem solar cell.³⁹ This hybrid method is a combination of co-evaporation of a scaffold of the inorganic precursors lead iodide (PbI₂) and cesium bromide (CsBr) and subsequent spin coating of the organic precursors formamidinium iodide (FAI) and formamidinium bromide (FABr) from solution. Adaptation of the FAI/FABr ratio in the second step allows for control of the bandgap. The final perovskites had the general formula of Cs_xFA_{1-x}Pb(I_yBr_{1-y})₃ with non-specified values for *x* and *y* but bandgaps of 1.53 eV and 1.80 eV for middle and top cell absorbers, respectively. In this work, 150 nm indium zinc oxide (IZO) was used as a recombination layer between the two perovskite subcells. The champion cell exhibited 14.0% PCE with a FF of 68.0%, a V_{OC} of 2.69 V and a j_{SC} of 7.7 mA cm⁻². The current of the final device was limited by the j_{SC} of the middle cell. In order to estimate the ideal bandgap and thickness ranges for their material stack configuration, they performed optical simulations which showed that when the





Fig. 6 Structure evolution and cross sectional SEM images of perovskite-based triple-junction solar cells. Reproduced with permission from ref. 14–17, 22–26. Copyright © 2018, American Chemical Society. Copyright © 2019 Published by Elsevier Inc. Copyright © 2020, Wang *et al.*, Published by Springer Nature, Copyright © 2020, American Chemical Society, Copyright © 2022, American Chemical Society, Copyright © 2022, American Chemical Society, Copyright © 2023, Wang *et al.* under exclusive licence to Springer Nature Limited, Copyright © 2023, American Chemical Society, Copyright © 2023 Heydarian *et al.*, Published by American Chemical Society.

bandgap of the top cell is kept at 1.80 eV, the middle cell bandgap should be lowered to ~ 1.40 eV to achieve current matching at a j_{SC} of ~ 12.2 mA cm $^{-2}$ for a flat silicon bottom cell and a $j_{SC} > 13.0$ mA cm $^{-2}$ for a both-side textured bottom cell.

In 2022, Zheng *et al.* achieved 20.1% PCE for a perovskite/perovskite/silicon triple-junction solar cell.¹⁷ In contrast to the previous work, they employed solution-based spin coated perovskite absorbers on a SHJ bottom cell with flat front side and textured rear surface. Their champion device was based on a FA_{0.9}CS_{0.1}PbI₃ (1.55 eV) middle cell absorber and FA_{0.80}CS_{0.20}Pb(I_{0.45}Br_{0.55})₃ (1.90 eV) top cell absorber and reached 20.1% PCE, 86.0% FF and 8.5 mA cm $^{-2}$ j_{SC} with a V_{OC} of 2.74 V. The cell showed acceptable stability measured for more than 40 s at maximum power point. In this work, 1 nm of Au was used as recombination layer between the perovskite subcells. The top perovskite absorber layer had a higher bandgap compared to the previous publication. This bandgap combinations for perovskite middle and top cells are closer to the optimum values suggested by simulations. As an antireflection layer, textured polydimethylsiloxane (PDMS) was used at the front side of the device. Overall, current matching was still not achieved in this work and the perovskite middle cell limited the j_{SC} of the device.

Very recently, Choi *et al.* reported on a 22.2% perovskite/perovskite/silicon triple-junction solar cell¹⁵ on a flat front side

and textured rear side SHJ. Perovskite middle and top cells in this work were Cs_{0.10}FA_{0.85}MA_{0.05}PbI₃ (1.56 eV) and MAPb(I_{0.50}Br_{0.35}Cl_{0.15})₃ (1.96 eV), respectively. To avoid the solvent damage while processing the perovskite top cell, they followed a similar approach as the first all-perovskite triple-junction solar cell by McMeekin *et al.*²³ They employed ACN/MA solvent system instead of DMF/DMSO for the top perovskite which was removed immediately after spin coating and prevented solvent penetration to the underlying layers. By taking advantage of this optimization, the perovskite top cell was fabricated without the ALD SnO_x protection layer. Similar to the record all-perovskite triple-junction work,¹⁴ here PEIE was spin coated on top of C₆₀. In addition, they employed a high-quality high bandgap perovskite achieved *via* additive engineering by adding urea into their perovskite precursor solution. With these two optimizations, the best performing cell showed 22.2% PCE, 78.6% FF and 10.2 mA cm $^{-2}$ j_{SC} with a V_{OC} of 2.78 V. The cell showed satisfactory stability for 600 s measured at maximum power point in ambient atmosphere at room temperature. The j_{SC} of the final device was limited by the middle cell.

Our team recently fabricated perovskite/perovskite/silicon triple-junction solar cell with 20.0% PCE. A SHJ cell with flat front and textured rear side was employed as the bottom cell and perovskite middle and top cells were in p–i–n configuration





Fig. 7 (a) ABX₃ structure of perovskite material where A is a monovalent organic or inorganic cation such as methylammonium (MA), formamidinium (FA), and cesium (Cs). B is typically divalent lead (Pb) or tin (Sn) and X is a halide, mostly iodide (I) and bromide (Br) or a combination of them. (b) Perovskite bandgap tunability in the range from 1.2 eV to 3.1 eV. The bandgap range suitable for top cell, middle cell and bottom cell are highlighted by blue, green and red background accordingly. Adapted with permission from ref. 40 and 47 Copyright © 2020 Elsevier Inc and ref. 47 available from: <https://doi.org/10.1021/nl5048779>, Copyright © 2019 American Chemical Society.

(MA), formamidinium (FA), and cesium (Cs). B is typically divalent lead (Pb) or tin (Sn) and X is a halide, mostly iodide (I) and bromide (Br) or a combination of them.⁷ One outstanding property of metal halide perovskite is their bandgap tunability, which makes them ideal to be used in multi-junction solar cells. The perovskite's bandgap can be tuned from 1.2 eV to 3.1 eV⁴⁰ by compositional engineering *via* cation or halide substitution (Fig. 7).^{34,40–43}

Primarily, the perovskite's bandgap is determined by the B- and X-site ions forming a [BX₆] octahedral framework. The valence band maximum (VBM) consists of B-s and X-p orbital coupling, while CBM consists mainly of B-p orbital contribution.⁴⁴ The A-site cation has ionic character and its electronic states do not contribute to the VBM or CBM.⁴⁵ Nonetheless, the A cation can influence the [BX₆] framework sterically or electrostatically. In this way, expansion or tilting of the lattice structure can lead to changes in the bandgap.⁴⁶ Depending on the B-site cation used (Pb or Sn), smaller cations on the A-site (Cs < MA < FA) can either increase or reduce the bandgap.⁴⁶ Most commonly, A- and X-sites are tuned at the same time to adjust the bandgap and ensure phase stability of the final composition.

The first reported perovskite solar cell was MAPbI₃ absorber with a bandgap of 1.55 eV,⁴⁸ however it is not regarded as promising middle cell absorber due to poor moisture and thermal stability above 85 °C.⁴⁹ Improved performance as well as chemical and thermal stability have been achieved for multi cation (Cs/MA/FA) multi halide (Br/I) perovskite compositions that are now widely used in the community.⁵⁰

3.1 Middle perovskite subcell in current triple-junction solar cells

The record PCE of p–i–n perovskite single-junction solar cells is 24.7% using a triple cation double halide absorber with a Cs_{0.05}(FA_{0.95}MA_{0.05})_{0.95}Pb(I_{0.95}Br_{0.05})₃ composition and a 1.55 eV bandgap.⁵¹ Currently the perovskite middle cells in reported triple-junction solar cells have bandgaps in the range of 1.55 eV to 1.62 eV. Table 3 summarizes compositions and bandgaps of

the middle perovskite absorbers implemented in current triple-junction solar cells, as well as device structures and *jV* performances of respective single-junction solar cells.

As mentioned before, so far, the reported perovskite/perovskite/silicon triple-junction solar cells suffer from current mismatch in a way that the middle cell is limiting the overall current of the device and highest current is being generated in the silicon bottom cell. For improved current matching, reducing the bandgap of the middle cell perovskite absorber is required. In the following sections, we discuss the two most promising perovskite candidates with bandgap lower than 1.50 eV.

3.2 Strategies for stable FAPbI₃ perovskite with bandgap of 1.47 eV

Compared to MAPbI₃, implementing FA on the A-site allows to form a nearly cubic structure, a reduced bandgap (1.47 eV), and improved thermal stability.^{52,53} Moreover, its bandgap is ideal to be used as middle cell absorber in a triple-junction structure with a 1.12 eV bottom cell. The work on FAPbI₃ solar cell has mostly been in the n–i–p architecture which holds the record single-junction perovskite solar cell with a PCE of 25.8%.⁵⁴ Despite the relatively fewer works on the FAPbI₃ based p–i–n solar cells, recently an impressive PCE of 24.1%⁵⁵ has been reported, which is promising for multi-junction solar cell's application.

The major challenge of FAPbI₃ perovskite lies in its limited structural stability. Due to disordered FA-I ion interaction, the asymmetrical FA cation takes an off-centered position, which results in the formation of a trigonal instead of a cubic structure. Experimentally, a 1D yellow non-perovskite polymorph (yellow δ-phase) is formed for usual low-temperature annealing around ~100 °C. Annealing at 160 °C allows to form the desired cubic phase (black α-phase),⁵⁶ which could harm temperature-sensitive underlying layers. Moreover, exposure to ambient environment, particularly high humidity, can trigger the phase transition from black α- to yellow δ-phase.⁵⁶

Fig. 8 shows the PCE evolution of FAPbI₃ perovskites for n–i–p and p–i–n structures.



Table 3 j_V performance, perovskite composition, bandgap, absorber layer thickness, and device structure of perovskite single-junction solar cells with middle perovskite absorber used in perovskite-based triple-junction solar cells in literature. Single-junction solar cell values for the first perovskite/perovskite/silicon solar cell are not available (due to the fully textured design and respective optimization of processes on textured silicon bottom solar cells)²²

Year	2019	2020	2020	2022	2022	2023	2023	2023
Composition	MAPbI ₃	F _{A0.66} MA _{0.34} [†] PbI _{2.85} Br _{0.15}	C _{S0.05} FA _{0.95} [†] PbI _{2.55} Br _{0.45}	C _{S0.1} FA _{0.9} PbI ₃	C _{S0.15} MA _{0.15} FA _{0.70} [†] PbI _{2.85} Br _{0.15} [‡]	C _{S0.05} FA _{0.9} MA _{0.05} [†] PbI _{2.95} Br _{0.1} [‡]	C _{S0.1} FA _{0.85} MA _{0.05} [†] PbI ₃	C _{S0.05} (FA _{0.9} MA _{0.1}) _{0.95} [†] PbI _{2.95} Br _{0.05} [‡]
Bandgap (eV)	1.57	1.57	1.60	1.55	1.62	1.60	1.56	1.56
Structure	Glass/FTO/SnO ₂ /perovskite/PTAA/PCBM/perovskite/C60/BCP/Al	Glass/ITO/PTAA/perovskite/C60/BCP/Al	Glass/ITO/NiO _x /PTAA/perovskite/C60/BCP/Cu	Glass/ITO/HTL/perovskite/LiF/C60/BCP/Cu	Glass/ITO/NiO _x /Me-4PACz/perovskite/PEAI-EDAI ₂ /PCBM/PEIE/SnO ₂ /Ag	Glass/ITO/PTAA/PFN-perovskite/C60/PTAA/PFN-perovskite/C60/PTAA/PFN-perovskite/C60/SnO _x /Ag	Glass/ITO/PTAA/PFN-perovskite/C60/PTAA/PFN-perovskite/C60/SnO _x /Ag	Glass/ITO/PTAA/PFN-perovskite/C60/PTAA/PFN-perovskite/C60/SnO _x /Ag
Molarity (M)	1.03 : 1 MAI:PbI ₂	NA	1.40	1.50	1.25	1.50	1.20	1.20
Thickness (nm)	530	450	280	NA	NA	780	450	450
PCE (%)	15.3	16.5	19.4	18.1	21.0	21.1	20.1	15.6
FF (%)	70.0	76.0	80.2	78.0	79.4	77.3	80.3	77.2
J_{SC} (mA cm ⁻²)	20.8	21.0	22.4	23.3	22.4	22.5	24.5	19.8
V_{OC} (V)	1.06	1.03	1.08	1.01	1.18	1.20	1.04	1.02
Top cell in	Pero/pero/pero	Pero/pero/pero	Pero/pero/pero	Pero/pero/silicon	Pero/pero/organic	Pero/pero/pero	Pero/pero/silicon	Pero/pero/silicon
Ref.	23	25	24	17	16	14	15	26

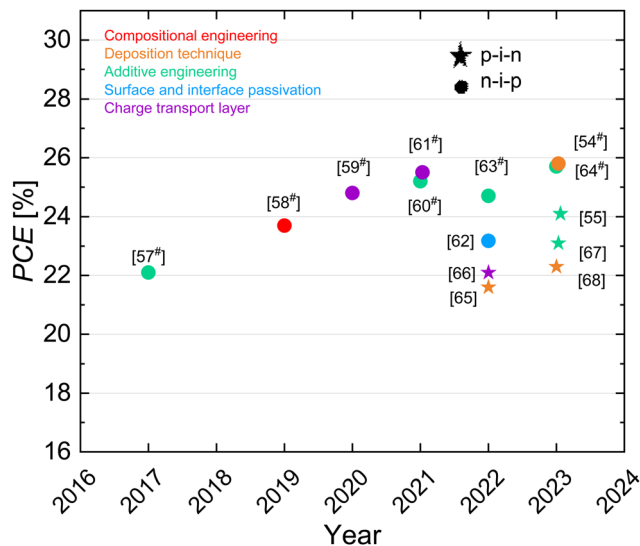


Fig. 8 PCE evolution of FAPbI₃ solar cells in p-i-n and n-i-p configuration discussed in this review for different strategies that resulted in efficient solar cells and improved stability (red: compositional engineering, orange: deposition technique, green: additive engineering, blue: surface and interface passivation and purple: charge transport layer). Data for n-i-p configuration are taken from ref. 54 and 57–64 and for p-i-n configuration from ref. 55 and 65–68. Certified values are marked with #.

In order to improve the stability of FAPbI₃ and achieve the black perovskite phase at low annealing temperature several strategies have been reported in the literature, which we briefly summarize below:

(1) Compositional engineering: the most famous approach to stabilize the black α -phase is alloying a small concentration of cations and anions with smaller ionic radius such as MA, Cs and Br to the composition. For example, Lee *et al.* showed that by adding a small amount of CsI to FAPbI₃, a black perovskite phase can be formed even prior to the annealing step.⁶⁹ This approach however has the drawback of increasing the bandgap of perovskite. Min *et al.* reported stable α -FAPbI₃ by incorporation of small amount of methylenediammonium dichloride (MDACL₂) into perovskite lattice. The film annealed for 10 minutes at 150 °C, showed almost identical bandgap compared to the control film, and the corresponding solar cell delivered 23.7% certified efficiency. This perovskite solar cell retained 90% of its PCE for 20 h in air under 150 °C with no encapsulation.⁵⁸ Moreover, non-stoichiometric modified precursors have shown several advantages in film quality of different perovskite compositions. For FAPbI₃ perovskite, excess FAI resulted in formation of the black film at low annealing temperature of 60 °C⁷⁰ or even without annealing at room temperature.⁷¹ Zhang *et al.* systematically studied the effect of excess PbI₂ in FAPbI₃ precursor solution and found that it reduced charge trap densities and prolonged charge carrier lifetimes.⁶⁶ Previous studies have confirmed the impact of excess PbI₂ in passivating the grain boundaries of perovskite.⁷²

(2) Deposition technique: the fabrication method employed to deposit the perovskite layer is known to influence film crystallization. Instead of common one-step spin coating

deposition, a two-step sequential deposition is found to form stable α -phase FAPbI₃ at low annealing temperature of 80 °C.⁷³ In this technique, PbI₂ is first deposited and then FAI solution is processed on top. It is crucial to obtain a highly porous thin film of PbI₂ in first place to form stable phase-pure α -FAPbI₃.⁷⁴ Xu *et al.* showed that by using DMF/N-methyl-2-pyrrolidone (NMP) solvent mixture for PbI₂, a porous morphology can be achieved. In addition, they introduced a dynamic spin coating method in which a pure α -phase FAPbI₃ film formed even prior to the annealing step.⁷⁴ Huang *et al.* used a mixture of isopropanol/hexafluorobenzene solvent for the organic salt of the second step and achieved a very high efficiency of 25.8% which retained 94% of its efficiency for more than 1000 h of MPP tracking.⁵⁴ Vacuum deposition is an alternative method to conventional solution processing. Borchert *et al.* showed that by co-evaporation of FAI and PbI₂, homogeneous and pinhole-free FAPbI₃ film can be formed. They demonstrated that in contrast to the standard solution processing which requires long annealing steps, here only a very short annealing time of 1 minute at 170 °C was needed to convert δ -phase of FAPbI₃ to the desired α -phase.⁷⁵

(3) Additive engineering is reported to enhance phase stability of FAPbI₃ by improving perovskite crystallization and defect passivation. For example, adding methylammonium chloride (MACl) into the perovskite precursor solution has been reported in both inverted⁶⁷ and regular^{54,63,76} architectures. Park *et al.* reported a 25.7% PCE by adding propylammonium chloride (PACl) as a secondary additive to the FAPbI₃ with MACl additive.⁶⁴ Jeong *et al.* added 2% formamidinium formate (FAHCOO) which is a pseudo-halide additive into the precursor solution and reported a certified PCE of 25.2% for a FAPbI₃ perovskite solar cell with improved operational stability over 450 h with no encapsulation.⁶⁰ Jiang *et al.* incorporated methylamine formate (MAFa) ionic liquid into FAPbI₃ perovskite which resulted in crystallinity and morphology improvement and increase of charge carrier lifetime. As a result, record PCE of 24.1% for inverted architecture was achieved. In addition, this strategy led to prolong stability against moisture with only 10% loss of PCE after storing the sample for 1000 h in ambient without encapsulation.⁵⁵ Furthermore, adding CsPbBr₃⁶³ and MAPbBr₃⁵⁷ into FAPbI₃ solution are reported to reduce the deep level defects concentration and increase the performance.

(4) Interfacial and surface passivation: interfacial treatment by formation of thin two-dimensional perovskite capping layers, between the 3D perovskite and the subsequent charge transport layer can protect FAPbI₃ from moisture penetration and stabilize its α -phase. Wang *et al.* demonstrated that post treatment of the perovskite surface with cyclopropylcarbamide hydrochloride (CPAH) resulted in formation of a hydrophobic 2D layer of (CPA)₂PbI₂Cl₂ on top of the perovskite surface which significantly improved the stability of the solar cells against humidity. CPAH treated samples retained 74% of the initial PCE after 150 h storage in ambient environment.⁷⁷ In a similar approach Kareem *et al.* treated FAPbI₃ surface with a 2-(4-fluorophenyl) ethyl ammonium iodide (FPEAI). The PCE of the not-passivated device maintained only 46% of its initial

PCE, while the device with FPEAI showed 80% of its efficiency during more than 1000 h measurement due to the formation of 2D/3D heterostructure.⁶²

(5) Charge transport layer: the charge transport layer plays an important role on phase stability of FAPbI₃ films. Roß *et al.* studied the effect of the HTL on phase stability of co-evaporated FAPbI₃ in p-i-n configuration and demonstrated that free phosphonic acids groups of the self-assembling molecule MeO-2PACz significantly improved the stability of the black FAPbI₃ perovskite phase.⁷¹ Zhang *et al.* also compared the effect of different commonly used HTL materials in p-i-n configuration: PEDOT:PSS, PTAA, NiO_x, and MeO-2PACz in FAPbI₃ solar cells. Similarly, they reported that devices with MeO-2PACz showed higher PCE and much longer stability as 92% of the PCE was retained after 800 h storage at room temperature and high humidity with no encapsulation.⁶⁶ Min *et al.* reported a 25.5% FAPbI₃ solar cell which was achieved by chemical passivation of perovskite/ETL interface in an n-i-p configuration. By coating a Cl-containing perovskite on a Cl-bonded SnO_x, a FASnCl_x interlayer was formed between the perovskite and the underlying ETL. The successful passivation not only resulted in improved charge extraction and reduced non-radiative recombination but also led to better long term stability (90% of the initial PCE was retained after 500 h of MPP tracking) of the device with no encapsulation.⁶¹ A study by Jeong *et al.* compared Spiro-OMeTAD, Spiro-mF and Spiro-oF as HTLs and showed that the fluorination of Spiro-OMeTAD helps to achieve high-efficient FAPbI₃ cells with improved stability (Fig. 9).⁵⁹

More detailed reviews on FAPbI₃-based perovskites as well as strategies to overcome the challenges associated with them are presented in recent review articles.^{53,56,78,79}

3.3 Strategies for stable Pb–Sn perovskite with bandgaps below 1.47 eV

Further lowering the bandgap of perovskite is possible by complete or partial substitution of Pb with Sn in B-site cation, which directly alters the conduction band.^{44,80,81} Sn-containing perovskites with mixture of halides in the composition (Br/I) cover a broad bandgap range from 1.2 eV to 2.0 eV.⁸² The reduction of bandgap however does not follow a linear trend when substituting Pb with Sn,^{80,81} studies have shown that in mixed Sn–Pb perovskites the bandgap continuously reduces by increasing Sn content and reaches its minimum at approximately 80% Sn content and then widens afterwards. Thus, MAPb_{0.20}Sn_{0.80}I₃ shows a lower bandgap of 1.19 eV compared to MASnI₃ with 1.28 eV bandgap.⁸⁰ Sn-based perovskite solar cells were first explored in 2012⁸³ and significant progress has been made ever since; in 2023 the PCE of Sn-containing perovskite increased to 23.7% using a Cs_{0.2}FA_{0.8}Pb_{0.5}Sn_{0.5}I₃ (1.29 eV) absorber.⁸⁴ Fig. 10 shows the PCE evolution of Pb–Sn perovskites along with their corresponding compositions and bandgaps.

Sn-containing perovskite absorbers have been widely used as low bandgap bottom cell in all-perovskites dual-junction and recently triple-junction devices. The community therefore has focused on perovskites with low bandgap ~1.20 eV; less research is available on Sn-based perovskites with bandgaps





Fig. 9 General approaches reported in literature to improve performance and stability of FAPbI₃ perovskite solar cells. (a) FAPbI₃ and FA_{0.9}Cs_{0.1}PbI₃ films with different annealing temperatures, (b) Top view SEM images of FAPbI₃ and FA_{0.9}Cs_{0.1}PbI₃ and (c) normalized PCE of their respective solar cells without encapsulation measured over time in ambient. Cs incorporation resulted in formation of black perovskite even at room temperature. FA_{0.9}Cs_{0.1}PbI₃ has improved film quality and stability compared to pure FAPbI₃. Reproduced with permission from ref. 69. Copyright © 2015 WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim. (d) Schematic illustration of perovskite processing of FAPbI₃ solar cells processed with 2-step deposition method using dynamic



spin coating for the organic salt and statistical photovoltaic parameters (j_{sc} , PCE, FF and V_{oc}) of FAPbI₃ solar cells comparing static and dynamic spin coating for organic salt deposition. Reproduced with permission from ref. 74. Copyright © 2022, Rights managed by AIP Publishing. (e) Schematic illustration of evaporation method for processing of FAPbI₃ and respective FAPbI₃ film processed on large area with this method. Black FAPbI₃ is formed after 1 minute annealing at 170 °C. Reproduced with permission from ref. 75 Copyright © 2017, American Chemical Society. (f) XRD measurement of fresh and aged FAPbI₃ films with and without MAFA additive and (g) operational and thermal stability of FAPbI₃ solar cells with and without MAFA additive. Adding MAFA into perovskite precursor leads to excellent phase stability for 500 h and the corresponding solar cell exhibited improved operational stability. Reproduced with permission from ref. 55. Copyright © 2023 Wiley-VCH GmbH. (h) Top view SEM images of FAPbI₃ films with and without passivation with CPAH, (i) XRD measurement of fresh and aged FAPbI₃ films with and without CPAH treatment and (j) operational and thermal stability of FAPbI₃ solar cells with and without passivation. Passivating the perovskite surface with CPAH resulted in phase stability for 48 h and the corresponding solar cell exhibited improved operational stability. Reproduced with permission from ref. 77. Copyright © 2021 Elsevier Ltd. All rights reserved. (k) PCE of FAPbI₃ solar cell in p-i-n configuration using different HTLs and varied annealing temperature along with the trap density calculated for perovskite films on different HTLs and (l) perovskite films with different HTLs kept under 70% RH and 120 °C condition. Solar cell with MeO-2PACz as HTL indicated highest PCE and lowest trap density. The FAPbI₃ perovskite deposited on MeO-2PACz was more stable compared to other HTLs. Reproduced with permission from ref. 66. Copyright © 2022 Wiley-VCH GmbH.



Fig. 10 Composition, bandgap and PCE evolution of Sn-containing perovskites discussed in this review for different strategies that resulted in efficient solar cells and improved stability (red: compositional engineering, orange: deposition technique, green: additive engineering, blue: surface and interface passivation and purple: charge transport layer). Data are taken from ref. 84–92. Certified values are marked with #.

above 1.35 eV suitable for the middle cell in triple-junction devices. Among them, pure Sn perovskite with FASnI₃ composition could be an attractive option. First, its bandgap of ~1.40 eV is suitable for the middle cell of triple-junction solar cell especially on top of silicon bottom cell, and secondly, employing such Pb-free perovskite in multi-junction solar cell relaxes the concern regarding the toxicity of Pb. However, the maximum certified PCE reported so far for FASnI₃ solar cell is 14.1%⁹² which lags behind the performance of pure Pb and Pb–Sn perovskites.

In general, there are two major challenges associated with Sn-containing perovskite. First, Sn²⁺ can easily be oxidized to Sn⁴⁺ in the presence of small amounts of oxygen leading to defect formation that reduces carrier lifetime.⁹³ Second, due to non-uniform nucleation and fast crystallization of Sn-containing perovskite, fabrication of high-quality, homogeneous, compact, and pinhole-free film is more challenging compared to pure Pb perovskite.^{91,94} Such non-ideal perovskite films with defects and large number of

grain boundaries in turn accelerate the oxidation of Sn²⁺. Klug *et al.* reported that the photoluminescence intensity of perovskite film decreased by orders of magnitude when replacing only 1% of the Pb with Sn in the composition, which proves the challenges on the way of bringing Sn-based perovskite to the level of pure Pb perovskite.⁹⁵

Some strategies reported in literature to overcome these challenges are:

(1) Compositional engineering: perovskite composition and solution preparation have a crucial impact on the quality of the perovskite film in general and consequently the device performance. This is particularly critical in the case of Sn-based perovskites since they are prone to defects as discussed before. Usually, the perovskite solution is prepared by dissolving the precursor powders in an appropriate solvent. Liao *et al.* introduced a new precursor system by preparing FASnI₃ and MAPbI₃ precursor solutions separately and mixed them subsequently at different molar ratios. The fabricated (FASnI₃)_{0.6}(MAPbI₃)_{0.4} perovskite films showed dense grains with no visible pinholes.⁹⁶ Li *et al.* demonstrated that incorporation of Br in mixed Pb–Sn perovskite improves perovskite film quality and effectively passivates the grain boundaries. Optimizing Br concentration to 6% led to 19.0% PCE.⁸⁹ In another approach incorporation of Cs into FA_{0.8}MA_{0.2}Sn_{0.5}Pb_{0.5}I₃ perovskite improved the operational stability and for the optimum Cs content of 2.5%, PCE of 18.2% was reported.⁹⁷

(2) Deposition technique: as discussed in the previous section, fabrication method can have a huge impact on morphology of the formed perovskite film. Abdollahi Nejang *et al.* introduced processing of Sn-containing perovskite films *via* the vacuum-assisted growth control (VAGC). In this method, instead of using an antisolvent, a vacuum chamber is used to remove the solvents from the film. As a result of process optimization, homogeneous pinhole-free perovskite layers with large grains were achieved.⁹⁴ Werner *et al.* employed the gas quenching method, in which a flow of nitrogen is used to remove the solvent from the film. Sn-containing perovskite films processed with N₂ quenching also showed uniform, pinhole-free morphology.⁹¹ The improvement in morphology was shown to reduce defects at the grain boundaries and is a crucial factor to improve the performance and stability of Sn-containing perovskites.⁹⁸

(3) Additive engineering: introducing inorganic Sn halide additives SnX₂ (X = F, Cl, Br, I) into the precursor solution has



widely been reported to limit the oxidation by reducing the formation of Sn vacancies. For example, employing SnF₂ additive resulted in better crystallization and formation of pinhole-free perovskite layer.⁹⁹ Incorporation of a small amount of metallic Sn powder not only helps to lower the oxidation of Sn²⁺ to Sn⁴⁺ but also is a step further and can convert Sn⁴⁺ back to Sn²⁺.⁸⁷ Organic additives such as Guanidinium thiocyanate (GuaSCN)¹⁰⁰ and methylammonium thiocyanate (MASCN)¹⁰¹ have also been reported to increase the perovskite grain size and reduce the grain boundaries. In addition Sn²⁺ and SCN⁻ interaction has shown to inhibit the degradation of Pb-Sn perovskite precursor.¹⁰¹ Xiao *et al.* added zwitterionic antioxidant into the perovskite precursor solution.⁸⁶ The zwitterionic molecules hindered Sn²⁺ oxidation and passivated defects at the surface and grain boundaries of mixed Pb-Sn perovskite films, leading to certified PCE of 20.7%.⁸⁶ Furthermore, introducing hydrazine sulfate (HS) additive into Sn-containing perovskite has been found to slow down the crystallization. This in turn resulted in homogeneous distribution of elements and high quality perovskite film which led to 23.2% PCE.¹⁰² Very recently 23.7% PCE was achieved for Pb-Sn perovskite by improving the film quality through addition of octyl ammonium tetrafluoroborate (OABF₄) into the perovskite precursor.⁸⁴

(4) Interfacial and surface passivation: oxidation of Sn mostly occurs at the perovskite surface. Therefore, passivating the surface with different treatment methods such as forming 2D/3D structure can protect the perovskite surface and reduce the degradation rate by blocking the diffusion of oxygen into the grains. Kapil *et al.* studied the effect of surface treatment of Pb-Sn perovskite with ethylenediamine (EDA). They showed that the EDA treated perovskite films had reduced Sn⁴⁺ concentration on their surface compared to the non-treated films. The champion cell with optimum EDA concentration showed 21.7% PCE.¹⁰³ Hu *et al.* passivated top surface of Pb-Sn perovskite with ethylenediammonium iodide (EDAI₂) and achieved 23.6% PCE.⁸⁵ Passivated samples showed improved stability with 80% of the PCE retained after measuring for 200 h in inert atmosphere. In another study, passivating the surface and grain boundaries of Pb-Sn perovskite with phenethylamine acetate (PEAAc) effectively hindered the oxidation of Sn²⁺. In addition, this surface modification improved the band alignment of perovskite and ETL.¹⁰⁴

(5) Charge transport layer: the charge transport layer plays an important role for the stability of Sn-containing perovskite solar cells. PEDOT:PSS is the most widely used HTL in p-i-n configuration for Pb-Sn perovskite solar cells. However, stability of solar cells employing this HTL is a concern due to its hygroscopic and acidic nature. Ghimire *et al.* employed a PEDOT:PSS/PTAA bilayer in their Pb-Sn perovskite which resulted in improved morphology and larger grain size. Moreover, due to hydrophobic nature of PTAA, devices incorporating this bilayer as the HTL, exhibited significantly improved stability when compared to those using PEDOT:PSS.¹⁰⁵ Kapil *et al.* achieved 23.3% efficiency by employing a 2PACz/methyl phosphonic acid (MPA) bilayer as HTL in Pb-Sn perovskite. Perovskite films with 2PACz/MPA showed much slower oxidation rate.⁹⁰ Pitaro *et al.* compared

the formation of Pb-Sn perovskite on PEDOT:PSS, 2PACz and Br-2PACz. Their findings revealed that Br passivates the halogen vacancies at the perovskite/HTL interface. In addition, perovskite layers deposited on Br-2PACz exhibited better crystallinity. Consequently, solar cells utilizing Br-2PACz demonstrated superior performance and stability compared to PEDOT:PSS and 2PACz.¹⁰⁶ In a later study they improved the wetting of the Sn-based perovskite on SAM by deposition of a carbazole alkylammonium iodide derivative (4CzNH₃I) layer on top of Br-2PACz (Fig. 11).¹⁰⁷

Several reviews have summarized the progress of Pb-Sn perovskite solar cells^{82,108} with focus on bandgap tuning and their application in all-perovskite tandem solar cells,¹⁰⁹ stability issues and how to overcome them¹¹⁰ as well as their optoelectronic properties.¹¹¹

4. Suitable perovskite absorbers for top cell application

The bandgap required for the top cell in a triple-junction device is in the range of 1.85 eV to 2.15 eV depending on the choice of the other two subcells (see Section 1). So far, the focus of the research on high bandgap perovskite (HBG) in perovskite community has been mostly on the bandgap range suitable for dual-junction application (1.60–1.70 eV) and compositions with bandgaps greater than that have been explored relatively little.

4.1 Strategies for stable mixed cation mixed halide perovskite with bandgaps above 1.75 eV

In a mixed cation (mainly Cs/MA/FA) mixed halide (mainly Br/I) perovskite composition, the most common practice to increase the perovskites' bandgap is by increasing the amount of Br and/or Cs in the composition, which results in around 0.06 eV and 0.02 eV increase in bandgap by each 10% increase in Br and Cs content, respectively.^{41,43} However, the effective Br tuning introduces defect-assisted photo-induced phase segregation, which can be regarded as one of the key challenges associated with HBG perovskite. For the first time Hoke *et al.* reported that HBG perovskites with halide mixture of Br and I in the composition with Br/I > 20% segregate into Br-rich and I-rich regions upon continuous illumination¹¹² (Fig. 12a). This phenomenon is also attributed to ion migration in mixed halide perovskite, where the generated electric field from electrical biasing breaks the halide's bond and leads to phase segregation as a result of this ionic movement.¹¹³ This goes along with a deficit in open-circuit voltage.¹¹² In addition to phase segregation, non-radiative recombination losses and the non-ideal energetic band alignment between the perovskite and the charge transport layers limit the V_{OC} of the final device¹¹⁴ (Fig. 12b and c). Therefore, high V_{OC} deficit is reported for HBG perovskites (Fig. 12d).

In order to tackle the above-mentioned issues and improve the V_{OC} in respective compositions several strategies have been reported in literature, which we briefly summarize below.

(1) Compositional engineering: the composition of HBG perovskites is an important factor influencing device performance





Fig. 11 General approaches reported in literature to improve performance and stability of Pb-Sn perovskite solar cells. (a) AFM images of Pb-Sn perovskite and (b) statistical photovoltaic parameters (J_{sc} , V_{OC} , FF, and PCE) of Pb-Sn perovskite with different Br concentrations. Br incorporation resulted in larger grain size, and the highest PCE is achieved for solar cells with 6% Br concentration. Reproduced with permission from ref. 89 Copyright © 2018 WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim. (c) Schematic illustration of deposition of Pb-Sn perovskite with VAGC and (d) top view and



cross sectional SEM images of perovskite films processed with antisolvent and VAGC methods. Perovskite processed with VAGC had larger grains compared to the film processed with antisolvent. Reproduced from ref. 94 Copyright © 2019 Abdollahi Nejand *et al.* Published by WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim under the Creative Commons CC BY. (e) Aging Pb–Sn perovskite precursor solution with and without MASCN additive. (f) Top view and cross sectional SEM images of perovskite films with different concentrations of MASCN and (g) normalized PCE of the solar cells with and without MASCN additive as a function of precursor solution aging time. Pb–Sn precursor solution with additive showed no oxidation after 50 minutes exposure to air. Enlarged grain size is achieved for the films with additive in the precursor. Reproduced with permission from ref. 101 Copyright © 2018 WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim. (h) XPS measurement of untreated and EDA coated Pb–Sn perovskite films, (i) top view SEM images of perovskite surface and (j) jV characteristic of solar cells with no passivation and passivated with different concentration of EDA. Non-treated perovskite contains higher concentration of Sn⁴⁺ compared to the treated surface. Better film quality is achieved with EDA treatment. 0.1 mM was found to be optimum EDA concentration resulting in maximum PCE. Reproduced with permission from ref. 103. Copyright © 2021 Wiley-VCH GmbH. (k) Top view SEM images of Pb–Sn perovskite deposited on PEDOT:PSS and PEDOT:PSS/PTAA bilayer. Hydrophobic nature of PTAA compared to PEDOT:PSS resulted in larger grain size formation. Reproduced with permission from ref. 105. Copyright © 2022 Wiley-VCH GmbH. (l) Statistical PCE and stability of Pb–Sn perovskite with PEDOT:PSS, 2PACz and Br-2PACz as HTL. Devices with Br-2PACz exhibited higher PCE and stability. Reproduced from ref. 106. Copyright 2023 © Royal society of chemistry under creative Commons Attribution-NonCommercial 3.0 Unported Licence.

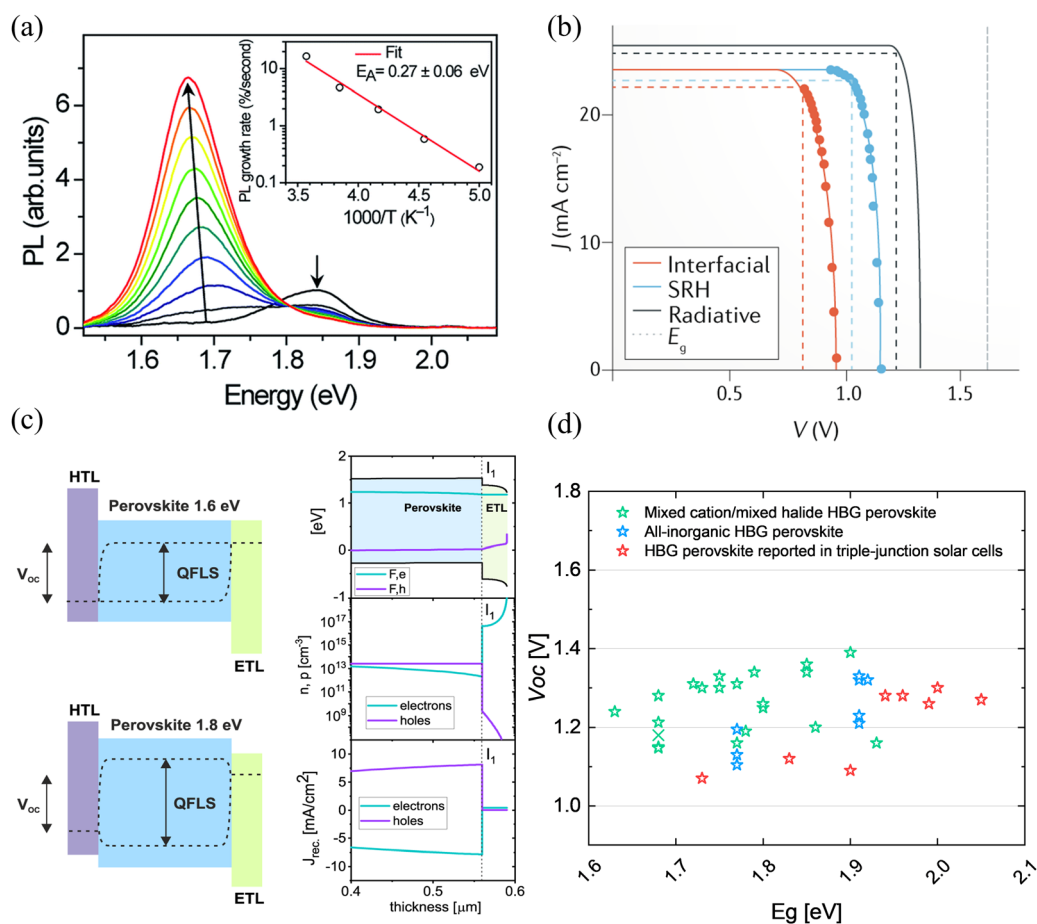


Fig. 12 Origin of V_{oc} limitation in HBG perovskite solar cells. (a) Phase segregation of mixed halide high bandgap perovskite. Reproduced from ref. 112 Copyright © 2015, The Royal Society of Chemistry used under a Creative Commons Attribution CC BY 3.0. (b) V_{oc} loss of a solar cell due to non-radiative recombination loss. Reproduced from ref. 115. Copyright © 2020, Springer Nature Limited. (c) Band diagram of perovskite and its charge transport layers showing a band misalignment for 1.8 eV perovskite due to the fact that the HTL and ETL were optimized for 1.6 eV perovskite. Reproduced from ref. 114 Copyright © 2023, Caprioglio *et al.* under a Creative Commons Attribution CC BY and (d) V_{oc} of some of the high bandgap perovskite solar cells discussed in this review as a function of the reported bandgaps. The discrepancy between V_{oc} and bandgap increases for high bandgap perovskite solar cells.

and stability as each composition may induce a different electronic and crystal structure. Therefore, perovskites with same bandgap and different compositions do not necessarily exhibit similar trap states in their films. Bush *et al.* studied a wide range of double

cation (FA/Cs)/double halide (Br/I) perovskite compositions and found that increasing the bandgap should not only be relied on Br content. Increasing the Cs content allows to reduce the Br content (to a certain extent) while maintaining similar bandgap and



improving photostability.⁴¹ Incorporation of other small cations such as rubidium (Rb)^{116,117} and potassium (K)¹¹⁸ into the A-site cation of the perovskite composition is another method practiced to slow down phase segregation. This is attributed to increased lattice distortion which introduce a barrier to ion migration.¹⁴

(2) Solution preparation and deposition technique: in addition, how the precursor solution is made can have impact on perovskite film quality and consequently its photostability and device efficiency. For example, the standard precursor materials used for preparation of a perovskite solution with double cation composition (e.g., Cs_xFA_{1-x}PbI_yBr_{1-y}) are typically FAI, CsI, PbBr₂ and PbI₂. Xiao *et al.* introduced a six precursor system of FAI, CsI, PbBr₂ and PbI₂, CsBr and FAb for preparation of HBG double cation perovskite (Cs_{0.2}FA_{0.8}PbI_{1.8}Br_{1.2}) and observed experimentally that perovskite solar cells prepared from the six precursor system showed improved photostability compared to the devices made from standard four precursor system.⁸⁶ The phase segregation phenomenon has also been attributed to defects present at grain boundaries, the bulk or surface of the perovskite that provide channels for ion migration. Therefore, improving the perovskite's bulk quality by increasing the crystallinity and the grain size could lead to enhanced photostability.¹¹⁹ This can be achieved through adaption of the processing method,¹²⁰ annealing time and annealing temperature.¹²¹ Jiang *et al.* employed a gentle gas quenching method instead of the conventional antisolvent method which resulted in better crystallization of their HBG perovskite and film morphology leading to more than 20.0% PCE for a 1.75 eV perovskite with a high V_{OC} of 1.33 V.¹²⁰

(3) Additive engineering: additive engineering is another strategy to improve the perovskite's quality as well as passivating the trap states at bulk and grain boundaries. Yu *et al.* added lead thiocyanate Pb(SCN)₂ to their HBG perovskite precursors and achieved much larger grain size compared to their reference perovskite.¹²² Kim *et al.* improved the morphology and crystallinity of the HBG perovskite by adding Pb(SCN)₂ and phenethylammonium iodide (PEAI) to the perovskite precursor.¹²³ Chloride-based additives such as MACl have also successfully been implemented in HBG perovskite which resulted in more homogeneous halide distribution and mitigation of phase segregation.¹²⁴⁻¹²⁶ Moreover, Thiesbrummel *et al.* added small amount of oleylamine into HBG perovskite which resulted in improvement of V_{OC}.¹²⁷ An *et al.* added phenylethylammonium acetate (PEAAc) additive to a variety of perovskites in the bandgap range of 1.72–1.92 eV. They showed that PEAAc resulted in more homogeneous halide distribution and hence reduced traps.¹²⁸ Urea as an additive has shown impressive improvement in film quality and performance of 1.96 eV perovskite¹⁵ which is discussed further in Section 4.3. It is worth mentioning that the amount of these additives must carefully be tuned with respect to the perovskite composition, bandgap and processing route.¹¹⁹

(4) Interfacial and surface passivation: employing surface and interfacial passivation with materials such as phenylethylammonium iodide (PEAI)¹²⁹ and butylammonium bromide (BABr),¹³⁰ which results in formation of a 2D layer on top of 3D perovskite, is an effective method to enable mitigation of halide segregation

through passivating defects and blocking ion migration.¹³¹ Dual passivation of grain boundaries and surface with 4-fluorophenylethylammonium iodide (FPEAI)¹³² or phenformin hydrochloride are also reported in literature.¹³¹ Surface modification of perovskite with electron-donating ligand trioctylphosphine oxide (TOPO) is also reported to reduce the rate of phase segregation.¹³³ Caprioglio *et al.* introduced a surface passivation between perovskite and PCBM using imidazolium bromide (ImBr) which resulted in growth of a low dimensional perovskite layer on top of perovskite and reduced the voltage loss of 1.80 eV perovskite.¹¹⁴ Similar to additive engineering, successful perovskite surface passivation requires thorough fine tuning of deposition parameters and controlled layer formation optimized for the respective device structure and perovskite composition.

(5) Underlying charge transport layer: proper choice of the charge transport layer can result in enhanced V_{OC}. Up to 2020, PTAA and NiO_x were the most widely used HTL of perovskite solar cells in p-i-n configuration, till a group of carbazole-based self-assembled monolayer (SAM) materials, namely 2PACz, MeO-2PACz and Me-4PACz, were introduced.^{134,135} HBG perovskite devices with SAM exhibited highly improved photostability compared to PTAA.³⁷ Ever since, most of the efficient HBG perovskite solar cells employed SAM in their structure.^{114,124,127,129,136} Recently, He *et al.* developed a new SAM (4-(7H-dibenzo[*c,g*]carbazol-7-yl)butyl)phosphonic acid (4PADCB) as HTL of 1.77 eV perovskite and demonstrated the lowest V_{OC} deficit for such HBG perovskite.¹³⁷ Currently the HTLs are not specifically optimized for perovskites with bandgap exceeding 1.80 eV, therefore there could be a mismatch between the HTL and perovskite energy levels which needs to be studied and addressed in future.

Fig. 13 summarizes the approaches that has been successfully employed to enhance the performance and stability of HBG perovskite solar cells.

4.2 Strategies for stable all-inorganic high bandgap perovskites with bandgaps above 1.77 eV

All-inorganic perovskite compositions with general formula of CsPbX₃ are promising to achieve bandgaps above 1.77 eV. They cover a bandgap range of 1.77 eV for CsPbI₃ to 2.30 eV for CsPbBr₃.⁴⁰ Although the performances of all-inorganic perovskites still lag behind the standard organic inorganic compositions, they feature better photo-¹⁴ as well as thermal stability.¹³⁸ Until now, the focus of research on all-inorganic perovskites has mostly been on the CsPbI₃ composition due to its suitable bandgap for application in dual-junction solar cells. Pure CsPbI₃ typically requires high annealing temperature of above 300 °C to form a black phase.¹³⁸ This becomes challenging in the case of triple-junction solar cells with substrate structure (see Fig. 5) as the high bandgap perovskite is the last solar cell in the deposition order. However, advancement in the field have reduced this high temperature annealing requirement which we explain in this section. In addition, similar to FAPbI₃ composition, CsPbI₃ experiences phase instability under ambient conditions (at room temperature in presence of moisture) which changes its α -phase to a yellow non-perovskite phase.



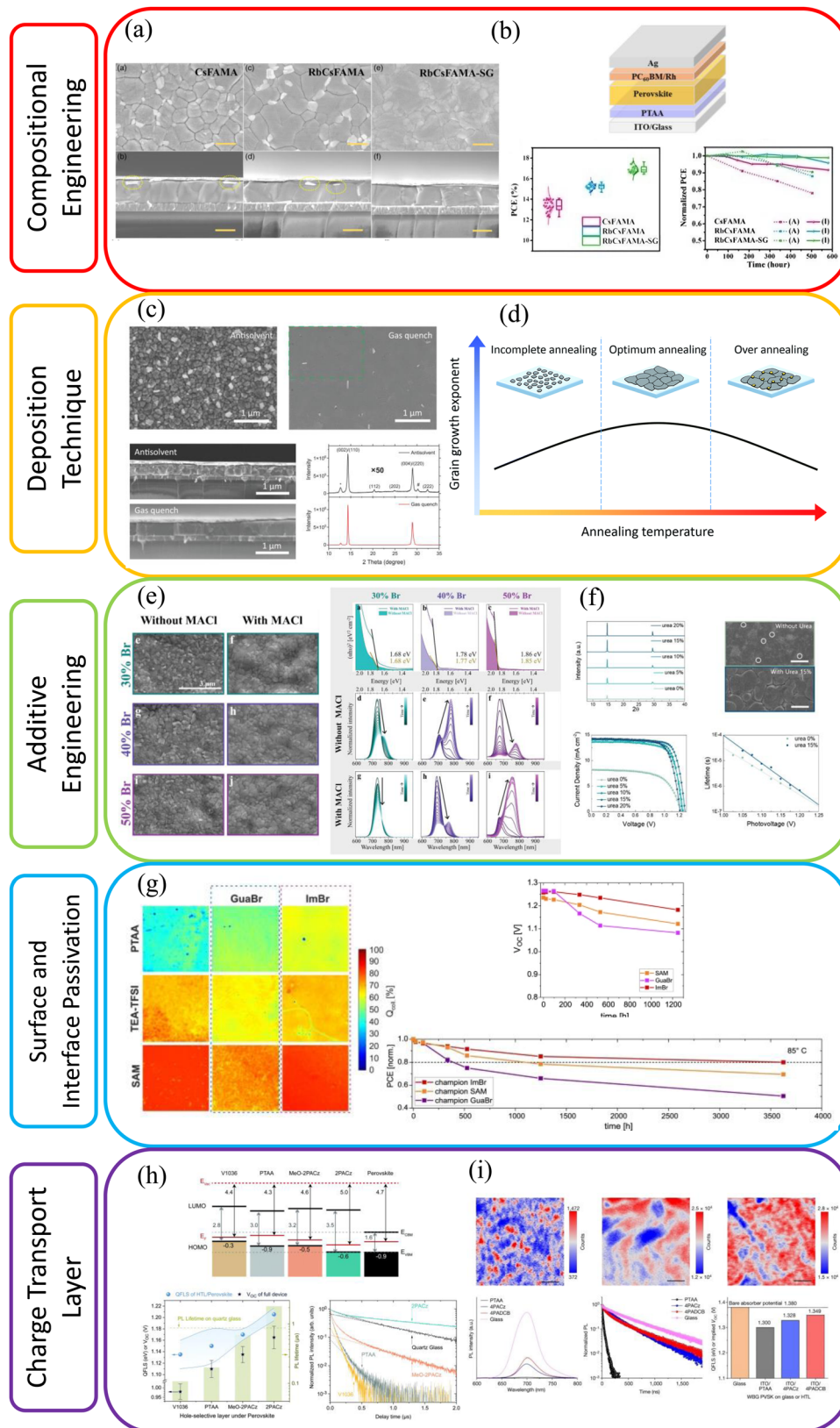


Fig. 13 Approaches reported in literature which successfully suppressed the light-induced phase segregation, non-radiative recombination and energetic band misalignment that improved the V_{OC} and consequently the performance of high bandgap perovskite solar cells. (a) Top view and cross-sectional SEM images of 1.78 eV perovskite films and (b) schematic structure of perovskite solar cell, statistical PCE as well as operational stability over time for perovskite solar cells with and without Rb in the composition. Reproduced with permission from ref. 116. Incorporation of Rb resulted in better film quality, increase in PCE and long term stability. Copyright © 2021 Wiley-VCH GmbH. (c) Top view and cross-sectional SEM images and XRD patterns



of 1.75 eV perovskite films deposited using antisolvent and gas quenching methods which shows smooth and dense perovskite film with gas quenching. Reproduced with permission from ref. 120. Copyright © 2022 Jiang *et al.* some rights reserved; exclusive licensee American Association for the Advancement of Science. (d) Schematic illustration of annealing temperature effect on grain growth of perovskite films. Reproduced with permission from ref. 121. Copyright © 2015, The Royal Society of Chemistry. (e) Top view SEM images and PL measurements over time for perovskite films with bandgap range of 1.68–1.86 eV with and without MACl additive. For HBG perovskite with up to 40% Br, MACl additive improved film quality and consequently suppressed light-induced phase segregation. Reproduced with permission from ref. 126. Copyright © 2023 Wiley-VCH GmbH. (f) XRD patterns, top view SEM images of perovskite films with 1.96 eV bandgap with and without urea additive together with *jV* curve and minority carrier lifetime of the corresponding solar cells. Better crystallinity and film quality for samples with urea additive. Highest PCE is achieved for 15% urea concentration. Reproduced with permission from ref. 15. Copyright © 2023, American Chemical Society. (g) PLQY images of 1.80 eV perovskite passivated with GuaBr and ImBr, V_{OC} over time along with the operational stability of their corresponding solar cells. GuaBr and ImBr passivations, increased the homogeneities and charge collection. Only ImBr passivation resulted in better stability. Reproduced from ref. 114 Copyright © 2023, Caprioglio *et al.* under a Creative Commons Attribution CC BY. (h) Energetic band alignment, QFLS and transient PL measurements of 1.68 eV perovskite on different HTLs. SAMs as HTL has better energetic alignment with tested perovskite. Higher QFLS and carrier lifetime is achieved for perovskites deposited on SAM compared to PTAA. Reproduced with permission from ref. 135. Copyright © 2019 The Royal Society of Chemistry. (i) PL mapping, PL and transient PL measurements and QFLS of 1.77 eV perovskite with different HTLs. HBG perovskite deposited on 4PADCB showed homogenous PL, longer carrier lifetime, higher PL intensity and consequently QFLS compared to PTAA and 4PACz. Reproduced with permission from ref. 137. Copyright © 2023, He *et al.*, under exclusive licence to Springer Nature Limited.

This is due to the small size of cesium ionic radii that results in non-ideal tolerance factor and structure distortion.¹³⁹ Recently, the performance of all-inorganic perovskite solar cells has been rapidly improved and a certified PCE of 20.1% was reported for CsPbI₃ with a V_{OC} of 1.18 V, FF of 83% and a j_{SC} of 11.5 mA cm⁻².¹⁴⁰ For CsPbBr₃ based solar cells the highest efficiency reported is 11.1%, with an impressive V_{OC} of 1.70 V, a FF of 83% and a j_{SC} of 7.9 mA cm⁻².¹⁴¹ However, the bandgaps of all-inorganic single halide perovskites (CsPbI₃ with 1.77 eV or CsPbBr₃ with 2.30 eV) do not lie within the optimum range of top cell in triple-junction solar cell. Hence, all-inorganic multi halide perovskites CsPbI_xBr_{3-x} are more suitable for this purpose.

In order to achieve suitable all-inorganic perovskite absorbers at low annealing temperature, and stabilize its phase in ambient condition, several approaches have been reported.

(1) Compositional engineering: incorporation of Br into CsPbI₃ lowers the required temperature for crystallization and therefore annealing temperature of CsPbI_xBr_{3-x} is lower than that of CsPbI₃.¹³⁸ For example, Beal *et al.* studied different all-inorganic compositions from pure CsPbBr₃ to pure CsPbI₃ and reported a structurally stable CsPbI₂Br with 1.90 eV bandgap achieved at 135 °C annealing temperature.¹⁴² Partial substitution of Cs with Rb can also improve photostability¹⁴ as well as moisture stability. For example Cs_{0.99}Rb_{0.01}PbI₂Br perovskite remained in black phase after 120 h exposure to ambient air.¹⁴³ Duan *et al.* introduced a new precursor system by mixing Cs formate (HCOOCs) with hydrogen lead trihalide (HPbI₃ and HPbBr₃) powders instead of standard CsI, PbI₂, and PbBr₂. With this optimized precursor system, they processed CsPbI₂Br without any antisolvent step and slowed down the crystallization process. Solar cells based on CsPbI₂Br with the new precursor system retained 92% of the original efficiency after more than 800 h storage in ambient air with no encapsulation.¹⁴⁴

(2) Solution preparation and deposition technique: Zhu *et al.* reported that, by aging the perovskite precursor solution up to 3 weeks a pure phase CsPbI₂Br film can be achieved at 100 °C annealing temperature.¹⁴⁵ Liu *et al.* processed CsPbI₂Br at room temperature by replacing the standard solvent of DMF/DMSO with NMP and combining it with vacuum-assisted extraction of the solvent instead of conventional antisolvent

method.¹⁴⁶ In addition, samples prepared at room temperature showed prolonged stability against humidity compared to samples prepared with high annealing temperature.¹⁴⁶ Mali *et al.* developed a hot-air assisted method for deposition of CsPbI₂Br, in which a hot air gun is used to remove the solvent during spin coating. Employing this method improved the uniformity and crystallinity of black perovskite.¹⁴⁷ Chen *et al.* fabricated CsPbBr₃ solar cells by evaporation of CsBr and PbBr₂. By optimizing the evaporation rate, high quality films with large grain size were achieved.¹⁴⁸ Evaporation of such inorganic high bandgap perovskite shows promising potential for its use as the top cell in a perovskite/perovskite/silicon triple-junction solar cell.

(3) Additive engineering: a stable black CsPbI₃ film at 100 °C annealing temperature was reported previously by Eperon *et al.* by adding small amount of hydroiodic acid (HI) into the perovskite precursor solution.¹⁴⁹ Currently HI is the most widely used additive for this type of solar cells.¹⁵⁰ Yu Han *et al.* studied the effect of adding calcium chloride (CaCl₂) into CsPbI₂Br perovskite and demonstrated that calcium can dope and passivate CsPbI₂Br resulting in higher stability in air. Their single-junction perovskite solar cell retained 90% of its initial PCE after more than 1000 h storage in air.¹⁵¹

(4) Interfacial and surface passivation: one of the most important methods to improve moisture-resistance of all-inorganic perovskite is passivation of surface and grain boundaries, which protects the perovskite against moisture penetration. Zhang *et al.* passivated CsPbI₂Br surface with guanidinium iodide (GuaI) and achieved 9.7% PCE.¹⁵² Wang *et al.* improved moisture-resistance of CsPbI₂Br by phenylethyl ammonium chloride (PEACl) treatment.¹⁵³ Similarly, post treatment of CsPbI₃ with phenyltrimethylammonium bromide (PTABr) resulted in surface passivation of perovskite and improved the stability against humidity. After exposing the perovskite films for 30 minutes to humidity of 80 ± 5% and temperature of 35 °C, the control films turned yellow while the treated samples remained in black phase.¹⁵⁴ Very recently Han *et al.* reported more than 19.0% efficient CsPbI₃ perovskite with sequential treatment of the perovskite's surface with ammonium benzenesulfonate (ABS) and phenylethylammonium iodide (PEAI) materials. Samples could retain around 70% of their PCE after 300 h exposure to ambient environment with no encapsulation.¹⁵⁵





Fig. 14 Approaches reported in literature which successfully improved performance and stability of all-inorganic perovskite. (a) PL measurements of $\text{CsPbI}_{1.75}\text{Br}_{1.25}$ and $\text{Rb}_{0.15}\text{Cs}_{0.85}\text{PbI}_{1.75}\text{Br}_{1.25}$ perovskites. Reproduced with permission from ref. 14. Copyright © 2023, Wang *et al.* under exclusive license to Springer Nature Limited. (b) Statistical PCE, stabilized j_{sc} as well as long term stability of the all-inorganic perovskite solar cells with and without Rb in the composition. Better performance, photostability and long-term stability is achieved by incorporation of Rb. Reproduced with permission from ref. 143. Copyright © 2020 WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim. (c) Schematic illustration of dynamic hot-air assisted deposition technique, optical and top view SEM images of CsPbI_2Br perovskite processed with and without hot air quenching indicating improved film quality for samples prepared by dynamic hot-air assisted method. Reproduced with permission from ref. 147. Copyright © 2020 Elsevier Inc. (d) Schematic illustration of evaporation deposition of CsPbBr_3 and top view SEM images of CsPbBr_3 with different evaporation rates. The best film with large grain size is achieved for 5 \AA s^{-1} evaporation rate. Reproduced with permission from ref. 148. Copyright © 2018 WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim. (e) Images of CsPbI_2Br films with different amount of CaCl_2 additive over time stored at humidity of 40%, together with stabilized j_{sc} and j/V curves of the corresponding solar cells. Samples with high CaCl_2 concentration are more stable under humidity. Highest PCE is achieved for 0.5% CaCl_2 amount. Reproduced with permission from ref. 151. Copyright © 2020 WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim. (f) Images of CsPbI_2Br films with different passivation over time stored at humidity range from 25% to more than 60% as well as long-term stability of their corresponding solar cells. Passivated films showed better stability compared to non-passivated films under humidity. Best stability is achieved for PEACl treatment. Reproduced with permission from ref. 153. Copyright © 2018 WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim.



Fig. 14 summarizes the approaches that have been successfully employed to enhance the stability and performance of all-inorganic HBG perovskite solar cells.

More details on strategies toward improving the PCE and stability of this type of solar cells have been well presented in recent review papers.^{139,150,156}

4.3 High bandgap perovskite top cells in current triple-junction solar cells

On a final note, most of the reported triple-junction solar cells implemented a mixed cation mixed halide composition as top cell absorber. Table 4 summarizes compositions and bandgaps of the top perovskite absorbers in current triple-junction solar cells, as well as device structures and *jV* performances of respective single-junction solar cells. This section is described in more detail compared to the middle cell absorber as greater focus has been dedicated to suitable high bandgap absorbers for top cell application in triple-junction devices and their development so far.

In the first reported all-perovskite triple-junction solar cell, a $\text{FA}_{0.83}\text{Cs}_{0.17}\text{Pb}(\text{Br}_{0.7}\text{I}_{0.3})_3$ perovskite with 1.94 eV bandgap was used. Steps taken to stabilize the high bandgap perovskite were incorporation of 2% potassium (K^+) in the precursor to help surpassing ion migration as well as utilizing hydroiodic and hydrobromic acids as additive, which enabled achieving apparent grains in micrometer range. In addition their solution was aged for 2 days prior to the processing.²³ As the top cell in the second all-perovskite triple-junction solar cell Wang *et al.* employed a two-step deposition method for fabrication of their $\text{Cs}_{0.1}(\text{FA}_{0.66}\text{MA}_{0.34})_{0.9}\text{PbI}_2\text{Br}$ perovskite (1.73 eV).²⁵ Xiao *et al.* employed $\text{Cs}_{0.2}\text{FA}_{0.8}\text{PbI}_{0.9}\text{Br}_{2.1}$ (1.99 eV) with high bromide content deposited *via* a one-step antisolvent method. To improve the perovskite quality and increase the grain size, they added 5 mol% $\text{Pb}(\text{SCN})_2$ to the perovskite precursor solution.²⁴ For fabrication of fully textured perovskite/perovskite/silicon solar cell, Werner *et al.* used a 1.80 eV double cation double halide perovskite composition achieved *via* sequential co-evaporation of PbI_2 (layer thickness of 180–200 nm) and CsBr continued by spin coating of FABr solution.²² In the perovskite/perovskite/silicon triple-junction solar cell by Zheng *et al.* a 1.90 eV perovskite with $\text{Cs}_{0.20}\text{FA}_{0.80}\text{Pb}(\text{I}_{0.45}\text{Br}_{0.55})_3$ composition was used. No strategy is reported for stabilizing the high bandgap perovskite in this work.¹⁷ The reported perovskite/perovskite/organic triple-junction had a $\text{Cs}_{0.15}\text{MA}_{0.15}\text{FA}_{0.70}\text{Pb}(\text{I}_{0.15}\text{Br}_{0.85})_3$ perovskite with a ~ 2.05 eV bandgap which is the highest bandgap employed in a perovskite-based multi-junction solar cell so far. The perovskite is spin coated from solution with antisolvent method.¹⁶ They passivated the top surface of their perovskite layer with phenformin hydrochloride acid, which was previously shown to suppress phase segregation.¹³¹ For the first time, Wang *et al.* used an all-inorganic high bandgap perovskite with $\text{Rb}_{0.15}\text{Cs}_{0.85}\text{PbI}_{1.75}\text{Br}_{1.25}$ composition that has a 2.0 eV bandgap in the record all-perovskite triple-junction solar cell. The stable all-inorganic perovskite was developed by partially replacing Cs with Rb. Incorporation of Rb resulted in increase in lattice distortion that reduced the ion migration and therefore

Table 4 *jV* performance, perovskite composition, bandgap, absorber layer thickness and device structure of high bandgap perovskite used in perovskite-based triple-junction solar cells in literature. Single-junction solar cell values for the first perovskite/perovskite/silicon solar cell are not available (due to fully textured design and respective optimization of processes on textured silicon bottom solar cells)²²

Year	2019	2020	2022	2022	2023	2023	2023	2023
Composition	$\text{FA}_{0.83}\text{Cs}_{0.17}\text{Pb}(\text{Br}_{0.7}\text{I}_{0.3})_3$	$\text{Cs}_{0.1}(\text{FA}_{0.66}\text{MA}_{0.34})_{0.9}\text{PbI}_2\text{Br}$	$\text{Cs}_{0.2}\text{FA}_{0.8}\text{Pb}(\text{I}_{0.45}\text{Br}_{0.55})_3$	$\text{Cs}_{0.15}\text{MA}_{0.15}\text{FA}_{0.70}\text{Pb}(\text{I}_{0.15}\text{Br}_{0.85})_3$	$\text{Rb}_{0.15}\text{Cs}_{0.85}\text{PbI}_{1.75}\text{Br}_{1.25}$	$\text{MAPb}(\text{I}_{0.3}\text{Br}_{0.35}\text{Cl}_{0.15})_3$	$\text{Cs}_{0.05}(\text{FA}_{0.55}\text{MA}_{0.45})_{0.95}\text{Pb}(\text{I}_{0.5}\text{Br}_{0.45})_3$	
Bandgap (eV)	1.94	1.73	1.90	2.05	2.00	1.96	1.83	
Structure	Glass/FTO/SnO _x /PCBM/perovskite/Spiro-OMeTAD/Ag	Glass/ITO/PTAA/perovskite/C60/BCP/Cu	Glass/ITO/HTL/perovskite/LiF/C60/BCP/Cu	Glass/ITO/2PACz/perovskite/PhenHCl/C60/BCP/Ag	Glass/ITO/NiO _x /Me-4PACz/perovskite/PEAI-EDAI ₂ /PCBM/PEIE/SnO _x /Ag	Glass/ITO/Me-4PACz/perovskite/C60/PEIE/Ag/MgF ₂	Glass/ITO/2PACz/perovskite/C60/SnO _x /Ag/MgF ₂	
Molarity (M)	0.75	$0.48(\text{CsI-PbI}_2) + 0.095(\text{FABr-MABr})$	1.00	1.10	0.80	0.75	1.20	
Thickness (nm)	350	100	180–240	100	170	NA	450	
PCE (%)	11.6	8.1	9.1	8.4	13.6	13.9	11.3	
FF (%)	76.0	76.0	71.0	70.4	84.7	76.6	74.6	
<i>J</i> _{sc} (mA cm ⁻²)	11.9	9.9	11.7	9.4	12.4	14.16	13.6	
<i>V</i> _{oc} (V)	1.28	1.07	1.09	1.27	1.30	1.28	1.12	
Top cell in Ref.	Pero/pero/pero	Pero/pero/pero	Pero/pero/silicon	Pero/pero/organic	Pero/pero/pero	Pero/pero/silicon	Pero/pero/silicon	
	23	24	17	16	14	15	26	



improved photostability under illumination.¹⁴ In the record perovskite/perovskite/silicon triple-junction solar cell, a triple halide perovskite composition of $\text{MAPb}(\text{I}_{0.50}\text{Br}_{0.35}\text{Cl}_{0.15})_3$ with 1.96 eV bandgap was employed. Urea was added to the perovskite solution to improve crystallinity and achieve a high-quality perovskite film with large grains without pinholes. Consequently, the performance of the high bandgap perovskite cell significantly improved from 6.4% to 13.9%. The main improvement came from increase in (j_{SC}) in the presence of the urea additive, which they attributed to the enhanced thin-film crystallinity and light absorption.¹⁵ We used a $\text{Cs}_{0.05}(\text{FA}_{0.55}\text{MA}_{0.45})_{0.95}\text{Pb}(\text{I}_{0.55}\text{Br}_{0.45})_3$ perovskite composition with 1.83 eV bandgap as top cell absorber of perovskite/perovskite/silicon triple-junction cell. A dynamic gas quenching method was used to prevent solvent damage to the already deposited layers. The perovskite film deposited by the adapted method showed homogeneous and compact morphology with improved crystallinity.²⁶

A comprehensive overview on high bandgap perovskite solar cells can be found in the recent review papers.^{157,158}

5. Recombination layer between the perovskite subcells

In monolithic multi-junction solar cells, the subcells are connected in series through recombination layers. The recombination layer has shown to be one of the key factors influencing the performance of tandem solar cells as its electro-optical properties directly affect all photovoltaic parameters (j_{SC} , V_{OC} and FF) of the device.^{159,160} In a perovskite solar cell, electrons are extracted through an ETL, and holes are extracted through an HTL. In electrically series-connected perovskite subcells without any additional layer, the ETL and HTL of the subcells get in contact. However, there is usually a difference between the work functions of the middle cell's ETL and top cell's HTL.¹⁶¹ Therefore, directly connecting these two layers from different subcells introduce a barrier at this interface^{160,161} and has shown to result in S-shape jV curves and reduced FF of the device.¹⁶² Hence, introducing an additional layer between the subcells with a suitable work function could reduce the mentioned gap and provide recombination sites for the collected charge

carriers at HTL and ETL of different subcells. There are several properties that a recombination layer between perovskite subcells must fulfill (Fig. 15) which can be categorized as follows:

1. Electrical requirements

- As the subcells are electrically connected in monolithic multi-junction solar cells, the recombination layer must have low contact resistance.

- It also requires low lateral conductivity to avoid current diffusion.

2. Optical requirements

- The parasitic absorption of this additional layer should be minimum to transmit the light to the underlying subcell. Given the bandgap combinations of the subcells in the dual-junction solar cell, the recombination layer needs to be transmitting especially in the infra-red (IR) range. This spectrum of interest is broader for the recombination layer between the top and middle cell in triple-junction solar cells and is not limited to IR region.

3. Processing requirements

- To avoid possible damages during the processing of the top layers, the recombination layer must protect the layers underneath.

- The processing of the recombination layer itself needs to be compatible in order not to damage the underlying layers.

Generally, there are two common approaches for the recombination layer between the subcells in perovskite-based multi-junction solar cells. Either employing a transparent conductive oxide (TCO) such as sputtered indium tin oxide (ITO) and indium zinc oxide (IZO) or ultrathin metal layer such as gold (Au) and silver (Ag).

A thick layer of sputtered ITO ($\sim 100\text{--}120\text{ nm}$) has been used as recombination layer between the perovskite subcells in early reports on all-perovskite dual-junction solar cells.^{163,164} In the first proof of concept perovskite/perovskite/silicon triple-junction cell, Werner *et al.* followed a similar approach and employed 150 nm IZO as the recombination layer between the perovskite middle cell and top cell.²² However, the thick TCO layers have several drawbacks such as low sheet resistance which has shown to result in shunting the solar cells.¹⁶⁵ Such thick layers are also optically poor due to high parasitic absorption and possible back reflection.



Fig. 15 Schematic of relevant electrical, optical and processing requirements for the monolithic connection of perovskite subcells.



Most of the recent, highly efficient all-perovskite dual-junction solar cells including the current certified record, employed 1 nm evaporated gold (Au) as recombination layer in their structure.^{10,120,166} The advantages of evaporated metal

layer are its damage-free deposition method compared to sputtering as well as its very high conductivity. However, this layer should be kept as thin as possible (~ 1 nm) to minimize parasitic absorption. Later, TCO layers with reduced thickness



Fig. 16 Replacing metal layer with thin TCO as recombination layer. (a) Schematic structure of perovskite/organic dual-junction solar cell with four types of recombination layers and their corresponding jV curves and (b) comparison of transmittance and reflectance of the tested recombination layers. Reproduced with permission from ref. 162. Copyright © 2022, Springer Nature. (c) Schematic structure of perovskite/organic dual-junction solar cell with different thicknesses of ALD deposited InO_x as recombination layer and their corresponding jV curves. (d) Diode characteristic of (SnO_x)/(InO_x)/(MoO_x). (e) Comparison of transmittance of ~ 1.5 nm of InO_x and 1 nm Ag, and EQE of the organic bottom cell with 1.5 nm of InO_x and 1 nm Ag as recombination layer. Reproduced with permission from ref. 129 Copyright © 2022, Springer Nature. (f) Schematic structure of perovskite/perovskite/silicon triple-junction solar cell with Au/PTAA and ITO/2PACz as interconnection layers and their corresponding jV curves and (g) comparison of transmittance, reflectance and absorbance of the 1 nm of Au with 15 nm of ITO. Reproduced with permission from ref. 26 Copyright © 2023 Heydarian et al., Published by American Chemical Society. Under a Creative Commons Attribution CC-BY-NC-ND 4.0. All examples indicate improved optical properties of the recombination layer upon replacing the metal layer with thin TCO. Improved electrical property is achieved by optimizing the thickness of the TCO layer.



were employed. Abdollahi Nejang *et al.* showed that 15 nm of ITO is still optically favorable compared to 1 nm Au, and their all-perovskite dual-junction solar cell exhibited 0.9 mA cm^{-2} increase in j_{SC} upon replacing Au with ITO as recombination layer.¹⁶⁵ We observed similar behavior in our perovskite/perovskite/silicon triple-junction solar cell where the j_{SC} of the device was severely limited by the middle cell. By replacing the Au with an ITO between the perovskite middle cell and top cell, more light was transmitted to middle cell which improved the j_{SC} of the final triple-junction solar cell (Fig. 16f and g).²⁶ Chen *et al.* developed a recombination layer based on a 4 nm IZO on perovskite/organic tandem solar cell (Fig. 16a and b).¹⁶² The perovskite/perovskite/organic triple-junction solar cell also employed only 2 nm IZO as the recombination layer between perovskite middle cell and top cell.¹⁶ Palmstrom *et al.* attempted to produce recombination layer-free all-perovskite dual-junction solar cells. However, based on their results at least 5–15 nm of IZO is needed to achieve high FF.³⁸ Brinkmann *et al.* demonstrated perovskite/organic dual-junction solar cell with only 1.5 nm indium oxide (InO_x) as recombination layer (Fig. 16c–e). Deposition by ALD instead of sputtering enabled formation of such thin and yet conformal InO_x layer.¹²⁹ Commonly, to avoid sputter damage, a buffer layer is deposited prior to sputtering of the TCO layers, which is usually a SnO_x layer deposited by ALD. This additional layer also helps to prevent the penetration of solvent during the solution-based processing of the subsequent layers.²⁶ Beyond this, Choi *et al.* fabricated perovskite/perovskite/silicon triple-junction solar cells without the SnO_x buffer layer.¹⁵ To enable fabrication of ALD free interconnection layer, Polyethylenimine ethoxylated (PEIE) in methanol was first spin coated on C_{60} similar to the approach outlined in ref. 14 and 38. Subsequently 20 nm ITO was sputtered on top of the PEIE coated stack. However, to remove the ALD buffer layer, firstly, development of a soft sputtering process is required. Secondly, in the absence of a solvent barrier layer, alternative strategies such as change of the perovskite solvent system^{15,23} or employing evaporation technique is required for processing of the perovskite layer on top.

Recently, a 23.7% efficient all-perovskite dual-junction solar cell with no TCO or metal based recombination layer was reported by directly contacting the SnO_x buffer layer and the HTL (PEDOT:PSS) of the low bandgap perovskite.¹⁶⁷ Even though this work highlights the potential of a simplified recombination layer, further study is needed to assess the possible formation of a Schottky barrier when the SnO_x and other HTL materials get in contact. Future work should focus on developing an ideal recombination layer between the subcells in multi-junction solar cells. In addition to the high parasitic absorption, taking the production cost into account, gold is the least favorable material to be used and material availability of both gold and indium are critical. Another motivation to avoid using a metal layer, especially in the substrate configuration (Fig. 5) where HBG perovskite is processed on top of the recombination layer, is that the common self-assembled monolayer (SAM) hole transport materials for HBG perovskites (see Section 4) cannot directly be deposited on metal as they require hydroxyl ($-\text{OH}$) groups to

bind to in a condensation reaction.¹⁶⁸ In that respect the surface of the recombination layer should provide sufficient $-\text{OH}$ groups to ensure homogeneous coverage of SAM.

Research on alternative indium-free recombination layers such as aluminum-doped zinc oxide (AZO) is ongoing. For example zinc oxide (ZnO)/AZO has been used as recombination layer in copper indium selenium (CuInSe_2)/perovskite dual-junction solar cell.¹⁶⁹ AZO has also been employed as front contact of SHJ solar cells¹⁷⁰ which opens up the window for replacing the common TCO recombination layer between silicon and perovskite subcells. In addition, according to Messmer *et al.* the final cost of the solar cell significantly drops by replacing ITO layer with AZO.¹⁷¹

Finally, it is important to mention that optimizing the recombination layer between two subcells in a triple-junction solar cell is complex as one cannot evaluate the quality of the recombination layer independent from the effect of the connection to the third subcell.

6. Characterization of triple-junction solar cells

Precise characterization of solar cells is highly important in research and development. Solar cell's performance data reported in literature are mainly from in-house measurements. While for single-junction solar cells the measurement approach and interpretation of data is more straightforward, a sophisticated procedure is needed for multi-junction solar cells, especially in two-terminal structures where no direct electrical access to the individual subcells is possible and additional effects such as luminescent coupling can occur for current mismatched subcells. This gets more challenging by adding to the number of junctions. Even though perovskite-based triple-junction solar cells are at the early stage of development, there is extensive research and well-established measurement standards for multi-junction solar cells based on III–V materials,^{172–177} from which the perovskite community can benefit.

6.1 External quantum efficiency (EQE) measurement

The general method for correct EQE measurement of monolithically series-connected solar cells has been first presented by Burdick and Glatfelter in 1986 for a-Si/Si dual-junction solar cells¹⁷⁸ and was extended for a $\text{GaInP}/\text{GaInAs}/\text{Ge}$ triple-junction solar cell by Meusel *et al.*¹⁷³ The method is used to characterize multi-junction devices independent of their material. In summary, the EQE of each subcell in a two-terminal triple-junction solar cell can be measured individually provided that the subcell of interest limits the overall current. To achieve this, the solar cell is illuminated with spectrally selective bias lights to saturate the other subcells so that they generate larger photocurrents than the subcell under test. The suitable selective bias light depends on the spectral responses of the subcells and can be provided by wavelength specific bias lights such as LEDs or the use of optical filters in combination with broadband light sources.¹⁷⁹ In addition, to determine the EQE correctly, the subcell under test needs to operate at its



short-circuit condition, therefore an external voltage equal to the sum of the voltages of the other two subcells under the bias light condition needs to be applied. Estimating the appropriate bias voltage value requires further effort as the exact values of the subcell voltages at this low irradiance are often not known and in this case the procedure described in International Electrotechnical Commission (IEC) should be followed.¹⁷⁹

One common challenge in EQE measurement of multi-junction solar cells are measurement artifacts.^{174,175} Subcell properties (e.g., low parallel resistance (R_p), early reverse breakdown and high radiative recombination) in combination with non-correct measurement settings (bias voltage, spectrally tuned bias light) can cause interaction between them during measurement which introduce undesired measurement artifacts such that the EQE of other subcells appears in the measured EQE-data of the subcell of interest. At the same time the absolute EQE of the measured subcell is lower than expected. This effect usually happens when the solar cells have low parallel resistance or low breakdown voltages leading to a change in operating voltage during measurement.¹⁷³ In addition, the luminescence coupling is another source of artifact in EQE measurement of multi-junction solar cells.^{176,180} However, EQE measurement of perovskite-based multi-junction solar cells with artifacts caused by luminescent coupling has not been reported yet. Therefore, in the following we will focus on artifacts caused by low parallel resistance or low breakdown voltage. This is one of the major challenges associated with EQE measurement of multi-junction solar cells that is widely discussed for III–V based multi-junction solar cells (Fig. 17a).¹⁷³ The measurement artifact can be corrected mathematically as described in ref. 174 and 175. However, it has been shown by Reichmuth *et al.* that the typical correction procedure based on the assumption of a uniform R_p might lead to deviations from the true EQE.¹⁸¹ These lateral effects may become more relevant for industrial-size solar cells.

For the case of perovskite/perovskite/silicon triple-junction solar cells, we have observed this effect in our lab when

measuring the EQE of a silicon bottom cell in which the artifact in the spectral response range of the top and middle cells (300–750 nm) is visible (Fig. 17b) and also the as-measured EQE curve of the middle cell shows an artifact in the spectral response range of the bottom cell (800–1200 nm) (Fig. 17b). The artifacts have been corrected with the method described in ref. 174. In summary, for the case of middle cell, a scaling factor (S) is determined to bring the EQE of the bottom cell to the level of the artifact at 800–1200 nm. Then the scaled curve is subtracted from the measured EQE to remove the artifact. Additionally, the measured EQE up to 800 nm is divided by $(1 - S)$. Similar procedure is applied to correct the EQE of the bottom cell.

Apart from the common challenges described above, several other factors and measurement conditions are shown to have an impact on the absolute EQE of perovskite solar cells due to their dynamic effects.¹⁸² Mundus *et al.* investigated the non-linearity of various single-junction perovskite solar cells and showed that the j_{sc} has non-linear behavior with respect to the irradiance leading to a dependency of the EQE on bias illumination intensity.^{183,184} Mercaldo *et al.* reported similar effect.¹⁸⁴ Other parameters such as chopper wheel frequency,^{184–188} preconditioning of the cell¹⁸⁷ and applied bias voltage are also reported to affect the absolute EQE.¹⁸⁷ Also, the common approach to measure the EQE with a monochromatic beam smaller than the active area of the solar cell can cause deviations in the signal compared to when illuminating the whole area including all metallization, the edges and spatially different absorber thickness, radiative properties *etc.*¹⁸¹ Therefore, calculating the generated current from measured EQE data obtained for individual subcells is not straightforward. Consequently, determining the current matching situation of the perovskite-based multi-junction solar cells from EQE measurement can result in misinterpretation.¹⁸⁹ An alternative approach based on spectrometric characterization for precise determination of current matching could in the future be used for perovskite-based triple-junction solar cells.¹⁷² This approach has been successfully applied on perovskite/silicon dual-junction solar cells.¹⁸⁹



Fig. 17 (a) Measurement artifact at spectral response (SR) range of top cell during EQE measurements of the germanium (Ge) bottom cell of a III–V/III–V/Ge triple-junction solar cell. With permission from ref. 173 Copyright © 2003 John Wiley & Sons, Ltd. (b) EQE curves of a perovskite/perovskite/silicon triple-junction solar cell measured at Fraunhofer ISE CaLab. Artifacts are visible for both middle cell and bottom cell EQEs. The EQE measurements with artifacts and the corrected curves are presented.



achieved for triple-junction solar cells is limited to 22.2%, 23.3%, and 19.4% on a silicon, perovskite, and organic bottom cell, respectively, well below their theoretical potential. We believe that the focus is now changing, and it is only a matter of time until the triple-junction solar cell development undergoes a fast efficiency increase similar to the trend of perovskite/silicon tandem solar cells or even faster as the learnings from single- and dual-junction can be transferred to triple-junction solar cells. In this review we summarized the recent advancements of perovskite-based triple-junction solar cells, their theoretical potential, and their challenges. Based on what has been discussed we identify the main losses and important topics for future work.

Regarding the V_{OC} , according to simulations,²⁷ a practical V_{OC} of 3.54 V for all-perovskite and 3.24 V for perovskite/perovskite/silicon triple-junction solar cells can be achieved, respectively. So far, the maximum V_{OC} values realized for these two types of solar cells are limited to 3.20 V for all-perovskite and 2.87 V for perovskite/perovskite/silicon triple-junction cells. Analyzing the V_{OC} values reported in the subcells of these two configurations, for middle bandgap single-junction cells, the V_{OC} deficit ranges from 0.40 V to 0.54 V, while for high bandgap cells, it is between 0.66 V and 0.81 V. Given the higher V_{OC} deficit in high bandgap perovskite subcells, there is a need for strategies to develop efficient and stable high bandgap perovskite in the optimum range of 1.85–2.15 eV with high voltage output. For this purpose, the high bandgap perovskite needs excellent material quality, which can be achieved by careful compositional engineering, additive engineering, processing control, surface treatment and interfacial passivation. In addition, so far even with the latest progress in the field, which resulted in high bandgap perovskite with satisfactory efficiency, the stability of such perovskite is not fully addressed. In terms of charge transport layer materials, the work function and band alignment of conventional materials with regard to adapted bandgap perovskites have not been fully studied and there has been little adjustment to the respective valence and conduction band levels. More work needs to be done on this topic, as it could bear a big leverage for V_{OC} improvement.

In addition, development of a lossless recombination layer between the middle perovskite and high bandgap perovskite with proper barrier function is a relatively new research field but critical for process compatibility and high voltage output.

In terms of j_{SC} , a practical maximum j_{SC} of 12.0 mA cm⁻² and 14.1 mA cm⁻² can be achieved for all-perovskite and perovskite/perovskite/silicon triple-junction solar cells, respectively.²⁷ However, the maximum j_{SC} reported for these two structures are still lower than expected with 9.6 mA cm⁻² for all-perovskite and 10.2 mA cm⁻² for perovskite/perovskite/silicon triple-junction solar cells. There are two main origins for this j_{SC} limitation. Firstly, triple-junction solar cells consist of many interlayers, which introduce parasitic absorption leading to lower j_{SC} of the final device. In order to maximize the j_{SC} , materials with high transparency need to be employed as electrodes, recombination and charge transport layers. In addition, the thickness of all layers except for the absorber layers should be optimized to

introduce minimum parasitic absorption. Secondly, current matching has not been achieved for most of the reported triple-junction solar cells. Even though proper evaluation of current matching condition requires further effort, it is clear that current perovskite/perovskite/silicon solar cells are using non-optimum bandgap combinations and have a current limitation caused by the middle cell. Up to now, bandgaps employed for the middle cell of this structure are in the range of 1.50–1.60 eV. By lowering the bandgap of the perovskite middle cell (employing FAPbI₃ or Sn-containing perovskite), a better current matching could be achieved in the final device.

In addition to what has been mentioned, depending on the technology, further development and adaptation is necessary.

- Perovskite/perovskite/silicon

Since the high bandgap perovskite cell is the last subcell processed in this structure, not only suitable bandgap with good stability is required, but also a deposition technique must be employed with no damage to the underlying layers. This adds to the challenges of high bandgap perovskite development. In this regard, fully evaporated high bandgap perovskite could be the next focus, as apart from the process compatibility, it has several other advantages *e.g.*, in terms of upscaling.

- Perovskite/perovskite/perovskite

Development of a stable low bandgap perovskite in this structure is of great importance. Moreover, since the low bandgap perovskite is processed in the last step, it requires a deposition technique that introduces no damage to the two-underlying perovskite subcells. Therefore, all-perovskite multi-junction solar cells could highly benefit from development of efficient and stable evaporated Sn-based perovskite.

- Perovskite/perovskite/organic

So far, efficient organic solar cells have mostly been realized with a bandgap of 1.30 eV. Employing the organic solar cell as the low bandgap subcell in triple-junction solar cells requires further reduction of its bandgap.

Finally, it is worth mentioning that precise characterization in parallel to the development of triple-junction solar cells is crucial to increase the reliability of the measurements and get better insight into the limitation of this new technology. For comparable jV measurements, it is especially important to correctly adjust the spectrum of the solar simulator in a way that all subcells generate the same current as they would do under the reference spectrum. Furthermore, along with PCE enhancement, stability and scalability of perovskite-based solar cells need to be addressed for future commercialization of this technology. With regards to the triple-junction solar cells, so far, the focus has been on improving the efficiency, while the stability of the reported solar cells is less studied.

Author contributions

M. Heydarian and M. Heydarian (conceptualization, data curation, formal analysis, visualization, writing – original draft, writing – review & editing), P. Schygulla (data curation, formal analysis, visualization; writing – review & editing), S. K. Reichmuth (data



curation, formal analysis, visualization, writing – review & editing), A. J. Bett (formal analysis, writing – review & editing), J. Hohl-Ebinger (project administration, supervision), F. Schindler (funding acquisition, project administration, supervision, writing – review & editing), M. Hermle (funding acquisition, project administration, supervision), M. C. Schubert (funding acquisition, project administration, supervision, writing – review & editing), P. S. C. Schulze (conceptualization, formal analysis, writing – review & editing, supervision), J. Borchert (conceptualization, funding acquisition, project administration, supervision, writing – review & editing), S. W. Glunz (conceptualization, funding acquisition, supervision, writing – review & editing).

Conflicts of interest

The authors declare no competing financial interest.

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