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Recent advances in electrochromic materials and devices for camouflage applications

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Electrochromic (EC) materials and devices, whose optical properties (color, reflectance, emittance, etc.) can be manipulated by switching to a low-voltage bias, show great promise in many applications such as smart windows, adaptive military camouflage, displays, and anti-glare rearview mirrors. In recent years, EC materials and devices for camouflage, including color camouflage and thermal camouflage, have attracted growing attention and research interest due to their many attractive properties such as rapid response, controllable optical modulation, low energy consumption (bistability), and flexibility. Despite the continuous development of related fields, the design and fabrication of EC devices that truly meet the requirements of camouflage applications still face tremendous challenges. Herein, in this review, we first briefly introduce the structures and main performance parameters of EC devices for camouflage. Then we comprehensively summarize the latest research progress of EC technology in the camouflage field from the aspects of color camouflage and thermal camouflage. Finally, the current challenges and future perspectives in the EC camouflage field are proposed and discussed. It is our hope that this review will accelerate the development of these promising EC materials and devices and provide guidance for developing mature camouflage devices.

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1. Introduction

With the rapid development of science and technology, camouflage technologies are developing toward lighter, more comfortable and more intelligent properties. In terms of adaptive chromic camouflage technology, there are three types of chromic technologies, including thermochromism,^{1–3}

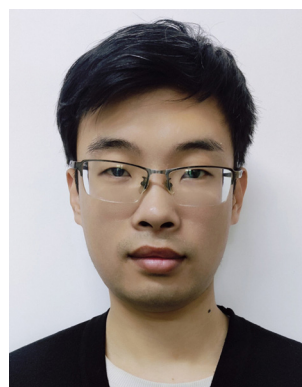
photochromism,^{4,5} and electrochromism,^{6–8} which control the optical properties of materials and devices in response to an external stimulus such as heat, light, or electricity. Among them, the application of thermochromic and photochromic materials and devices in adaptive camouflage is limited, because their optical properties strongly depend on the temperature and light intensity, and are often difficult to control and operate precisely.⁹ Moreover, the response speed and stability of photochromic materials and devices require further investigation and improvement. Among various camouflage technologies, electrochromic (EC) camouflage has received

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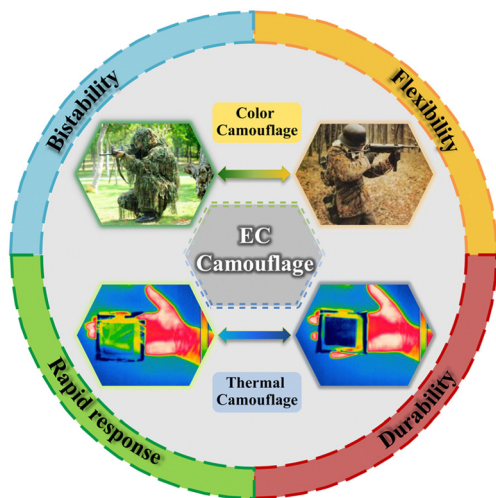


Fig. 1 An illustration of EC camouflage. Reproduced from ref. 13. Copyright 2018, American Chemical Society.

extensive attention and research interest due to its unique advantages and potential values. Some EC camouflage materials and devices with excellent properties have gradually emerged and been reported in detail.^{10–12} In recent years, EC camouflage technologies have made significant progress in both practical and theoretical terms, and are expected to be one of the next-generation camouflage technologies that people have long been awaiting.

The working principle of EC camouflage is based on the electrochemically driven redox process of EC materials to produce color/reflectivity/emissivity changes so that the objects exhibit adaptive changes to the surrounding environment and achieve perfect concealment from visual and infrared detection. Based on the differences in physical mechanisms, EC camouflage technology can be divided into two categories: color camouflage and thermal camouflage (Fig. 1). Compared with other camouflage materials and devices, EC materials and devices demonstrate many unique advantages in camouflage

applications. First, most existing EC materials show certain flexibility and processability while maintaining good compatibility with multiple substrates, including metal, plastic, and fibers, which meet wearable fabric requirements.^{14–16} Second, EC devices have good optical tunability, which can be tuned by matching the optical properties of each functional layer of the device to meet the needs of camouflage in different surroundings.^{17,18} Finally, the unique optical memory effect and fast response of EC devices are also highly favorable for camouflage applications.^{19–22} Therefore, EC materials and devices have great application potential in color camouflage and thermal camouflage. Compared to other EC applications (smart windows, EC displays, antiglare rearview mirrors, etc.), ideal EC camouflage devices have higher performance requirements, such as smaller chromatic aberration (ΔE_{ab}^*) with the surrounding environment to ensure concealment of camouflage, better flexibility to ensure comfort and practicality, and better stability and durability to support their life. Despite the remarkable progress that EC camouflage technology has achieved through the design of EC materials and devices, there are some critical issues that need to be addressed to boost their future research and applications, for instance, the design and development of EC camouflage devices which can independently control visible light (VIS) reflectivity and infrared (IR) emissivity, and the fabrication and research of high-performance wearable fabric EC camouflage devices. The current research and development process of EC camouflage devices is still in the early stage. And the practical application of relevant products still has a long way to go.

The study of EC camouflage materials and devices is of great significance to further expand the EC application fields. However, there is little systematic and overall design guidance. Thus, comprehensive summary and analysis of the recent advances in EC camouflage are necessary to provide directions for developing high-performance EC camouflage devices to push their real-world applications. This review will systematically summarize and discuss in detail the latest development



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application research in the field of organic optoelectronics and energy storage.

Prof. Cheng Zhang received his PhD degree in polymer materials at Zhejiang University in 1998. In the following year, he carried out research work at Tokyo Institute of Technology as a JSPS post-doctoral fellow. In 2002, he joined the College of Chemical Engineering, Zhejiang University of Technology as a professor. His research interests mainly focus on the design and synthesis of novel conjugated polymer materials, as well as their fundamental and



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of EC camouflage materials and devices to comprehensively understand their current research level and existing challenges for further development. First, we present the EC device structures and main performance parameters for camouflage applications. Then, the latest research progress of EC technology in color camouflage and thermal camouflage is summarized in detail. The advantages and limitations of each material and device or design strategy are analyzed and compared to guide the fabrication of high-performance EC camouflage devices. Finally, the challenges and perspectives for future research on EC camouflage materials and devices are proposed and discussed. It is our hope that this review will provide guidance for developing high-performance mature EC camouflage devices.

2. Electrochromism

Electrochromism refers to the reversible change of the optical properties (such as colors and transmittance) of materials under charge insertion/extraction or chemical reduction/oxidation induced by the application of an external voltage or current.^{23–25} The term “electrochromism” was first coined by Platt in 1961 when he discovered that amorphous WO_3 exhibited different colors under different external voltage conditions.²⁶ Since then, electrochromism has been attracting the interest of researchers.^{27–30} In 1984, the concept of energy-saving windows (smart windows) was proposed to make electrochromism known to people.^{31,32} During its development over 50 years, EC technology has been used in a wide range of applications, including smart windows,^{33–36} displays,^{37–39} and sunglasses,^{40–42} due to the tremendous efforts of outstanding pioneers.

2.1 Structure of EC devices for camouflage

For the application of EC technology, electrochromic devices (ECDs) as basic working modules need to be studied and manufactured. For different applications, ECDs can be divided into two categories: transmissive window-type devices (for smart windows and EC sunglasses, *etc.*) and reflective-type devices (for EC displays, adaptive camouflage, and EC antiglare mirrors, *etc.*).⁴³ It is worth noting that the EC camouflage devices discussed in this review are reflective-type devices. In the traditional design, ECDs are usually in a seven superimposed thin layers structure consisting of two substrates, two conducting electrodes, an EC layer, an electrolyte layer, and an ion storage layer, as shown in the schematic illustration (Fig. 2a). To meet future and sophisticated application needs, ECDs with a lateral structure in which the EC layer and the ion storage layer are arranged in a shoulder-by-shoulder manner have been proposed and manufactured (Fig. 2b).⁴⁴ In camouflage ECDs, the EC layer plays a core role; it undertakes the task of optical modulation or color change. In this regard, we will systematically discuss the EC layer in Section 3.1 and Section 4. The ion storage layer functions as a counter electrode layer, undergoing reversible electrochemical oxidation (reduction) to coordinate the reduction (oxidation) of the EC material to balance the charge shuttling from the EC layer through the

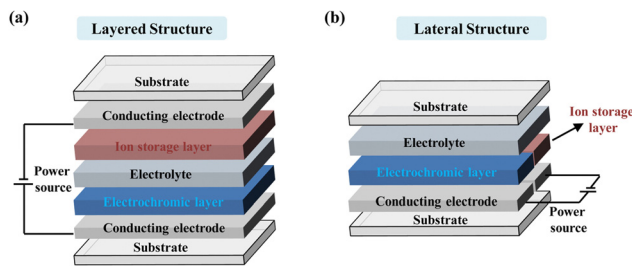


Fig. 2 Structures of the classic (a) layered and (b) lateral ECD.

electrolyte.^{45–47} The electrolyte layer sits in the center of the device, which can separate the EC layer and ion storage layer to prevent direct electrical contact and transport ions inside the device.⁴⁸ Substrates and conducting layers as conductive electrodes which determine the mechanical properties of the devices are commonly discussed together.⁴⁹

2.2 Performance parameters for camouflage

In order to promote the development of EC materials and devices, researchers have used a series of performance indexes to evaluate them, generally including optical and non-optical analyses. The optical analysis focuses on measuring changes in color and transmittance through techniques related to optical spectroscopy, including optical modulation, contrast ratio, optical memory, and chromatic aberration (ΔE_{ab}^*). Non-optical analyses include electrochemical and other conventional analyses for evaluating the electrochemical/kinetics properties of film and devices, including switching time, long-term stability and durability, and coloration efficiency (CE). Of course, for different applications, researchers are concerned with different performance parameters and requirements. For instance, EC materials and devices used for electronic displays must exhibit ultra-fast response rates (<40 ms); in contrast, when used for smart windows, the requirement for response time is not too high, but the requirements for optical modulation capabilities and memory effects are outstanding. As the field grows, it is essential to give precise definitions of individual performance metrics and to provide a standardized performance measurement for different EC applications. Here, we focus on the camouflage field. The camouflage performance of the EC materials and devices is usually evaluated by the main performance parameters of optical modulation, switching rates, cycling stability and weatherability, coloration efficiency, optical memory, and color difference. High-performance EC camouflage devices should exhibit excellent modulation capabilities, lasting cycling stability, and outstanding optical memory while maintaining a small ΔE_{ab}^* with the surrounding environment to ensure its camouflage.

3. EC materials and devices for color camouflage

Adaptive color camouflage requires the material to adjust its color to the color of the surroundings to blend with it and achieve perfect concealment of the target.^{50,51} In order to

survive and reproduce, many natural creatures, such as oceanic cephalopods and chameleons, are born with the ability to change their skin color in this way to blend in with the natural environment for camouflage. Currently, EC technologies have significant application potential for adaptive color camouflage due to their rich color and ease of control.

The critical point in EC color camouflage is to design and prepare high-performance EC materials and devices with camouflage color switching. Firstly, it is clear that the primary requirement for EC materials and devices for military color camouflage is the ability to switch between camouflage colors. Camouflage colors usually refer to yellow and green shades, such as sandy yellow and dark green. Secondly, to achieve an efficient camouflage effect, the excellent EC properties of the materials and devices are also required, such as fast switching time and robust stability. In this section, the efforts made by researchers to realize camouflage applications in terms of color modulation as well as performance optimization are summarized from both material and device design.

3.1 EC materials for color camouflage

To date, EC materials can be divided into inorganic and organic EC materials according to the difference in EC mechanisms. For inorganic materials, single color variation is their greatest limitation in applications,⁵² so is adaptive color camouflage. For example, WO_3 , the most widely used inorganic EC material, exhibits only a monochromatic color change from transparent to dark blue.⁵³ This problem can be effectively solved by the introduction of structural colors and the preparation of composite materials.^{54,55} The great advantage of organic EC

materials compared to inorganic materials is that the color can be tuned by means of molecular structural design.^{56–58} Of course, structural colors and color composites are also common means of color modulation in organic materials.^{59,60} Next, we discuss in detail the strategies that can be used to prepare EC materials for camouflage color conversions.

3.1.1 Introduced structural colors. Structural colors, well known from coloration in nature, refer to colors emerging from the interaction (refraction, diffuse reflection, or diffraction) of light with micro- and nano-structures with periodicities on the order of the wavelength,^{61–63} including Fano and Fabry–Perot (F–P) resonances,^{64,65} and localized surface plasmon resonances (LSPR)^{66–68} and so on. Different from dyes and pigments, structural colors require only very thin layers, or more precisely tiny volumes, to create brighter colors under sunlight that do not fade over time.⁶⁹ Combining electrically-driven chemical color changes with structural color changes (also known as physical color) is a viable strategy for realizing color modulation in EC materials and has received extensive attention.

In 2012, Geoffrey A. Ozin *et al.*⁷⁰ reported an electrochromic Bragg mirror (ECBM) based on multilayer photonic crystal structures of WO_3 and NiO with different refractive indices and achieved a reversible color change from dark green to yellow. The change in the refractive index of the film at different applied voltages is responsible for the color tuning of ECBM. To enhance the reflectance modulation of the ECBM, Xingyuan Liu and Ying Lv *et al.*⁷¹ fabricated electroresponsive WO_3 -based electrochromic distributed Bragg reflectors (ECDBRs) by alternating the deposition of WO_3 layers with various porosities (refractive indices) (Fig. 3a). By simply

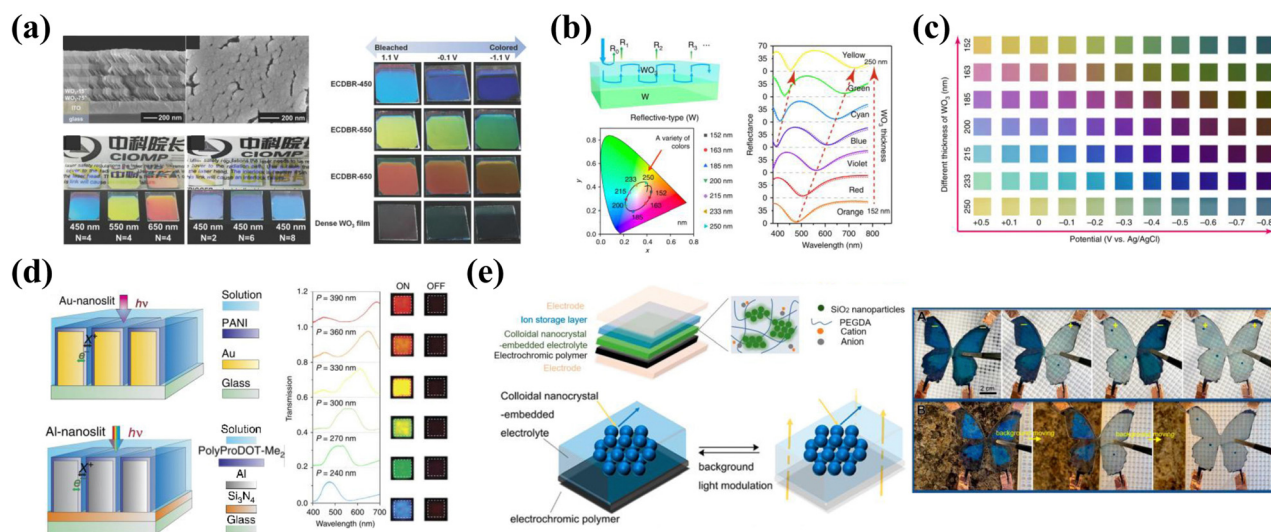


Fig. 3 (a) Left: cross-section, surface SEM images and photographs of the ECDBR. Right: EC properties of ECDBRs. Reproduced from ref. 71. Copyright 2017, Wiley-VCH GmbH. (b) Left: schematic and calculated color coordinates of an EC electrode using a metallic W layer as a current collector. Right: simulated (dashed lines) and measured (solid lines) reflection spectra with different thickness of WO_3 . (c) Color gallery obtained from an F–P nanocavity-type EC electrode at different applied potentials. Reproduced from ref. 65. Copyright 2020 (CC BY 4.0), Springer Nature. (d) Left: schematic of a plasmonic EC electrode with Au (top) or Al (bottom) nanoslit arrays. Right: optical transmission spectra of PolyProDOT- Me_2 -coated Al-nanoslit structures with different slit periods (P). Reproduced from ref. 73. Copyright 2016 (CC BY 4.0), Springer Nature. (e) Left: schematic of the photonic nanocrystal embedded EC. Right: the color appearance/disappearance of the butterfly device for both active and passive camouflage. Reproduced from ref. 75. Copyright 2021, American Chemical Society.

changing the deposition angle and thickness of each WO₃ layer, the Bragg wavelength of ECDBRs could be adjusted in the VIS range continuously. Among them, ECDBR-550 is capable of reversible switching between yellow-green and green at the applied voltage, as shown in Fig. 3a.

Color tuning in EC materials can also be achieved by constructing an F-P resonances structure.⁶⁴ Zhigang Zhao *et al.*⁶⁵ fabricated an ultra-compact asymmetric F-P nanocavity-type EC film by introducing metallic tungsten (W) between the WO₃ layers and Indium Tin Oxide (ITO) electrodes to induce strong interference resonance (Fig. 3b). By adjusting the thickness of the WO₃ layer, the F-P nanocavity-type EC electrode can display various structure colors. In addition, the EC electrode could also achieve multicolor displays by applying different voltages (Fig. 3c). In particular, when the thickness of WO₃ is 163 nm, the electrode could achieve reversible switching in yellow (−0.4 V) and green (−0.8 V) colors. This research result is an important development in the multicolor display of inorganic EC materials.

Furthermore, as the most frequent Plasmonic resonance structures, structural colors can also be introduced into organic EC materials for the purpose of color modulation.⁷² In 2016, A. Alec Talin *et al.*⁷³ fabricated a fast speed switching and multi-color EC electrode based on plasmonic nanoslit arrays (Fig. 3d). The required nanoslit arrays were obtained by focused ion beam (FIB) milling. Then, polyaniline (PANI) and poly(2,2-dimethyl-3,4 propylenedioxythiophene) (PolyProDOT-Me₂) (as EC material) were fabricated on nanoslit arrays by potentiodynamic cycling to a thickness of 15 nm. The peak spectral transmission of the PolyProDOT-Me₂ electrodes could be adjusted by altering the period of the array to achieve a full-color EC optical response, including yellow and green (Fig. 3d). And then, the ultra-thin film brought about by the introduction of structural color resulted in a fast response time (80 ms). A similar method of introducing structural color into PANI film was reported by Hui-Ying Qu and co-workers.⁷⁴ The authors have prepared a lamellar PANI/PS-*b*-P2VP thin film to enable the simulation of chromatophores and iridophores color changes using the combined EC effect of PANI and structural coloration of PS-*b*-P2VP.

In addition to constructing special structures in EC layers or electrodes, color tuning can also be achieved by introducing structural colors into the electrolyte. Jianguo Mei *et al.*⁷⁵ reported an electrolyte with a special structure by embedding silica nanoparticles into an ethylene glycol (EG)/poly(ethylene glycol) methacrylate (PEGMA) matrix. The voltage-controlled EC polymer enables the device to actively switch from a colored state to a high transmission state. On the other hand, the devices show background adaptive visual effects in a passive manner when the EC polymer is kept in a highly transmissive state. Finally, a butterfly ECD capable of both active and passive camouflage was prepared to demonstrate its application potential (Fig. 3e).

3.1.2 Modified molecular structure. Over the past few years, great achievements have been made in modulating the color of organic EC materials by modifying the molecular

structure,^{76–78} which is also considered to be one of the most important strategies to achieve color tuning of EC materials. Using this strategy, a lot of green and yellow EC materials for adaptive color camouflage have been designed and synthesized, such as small organic molecules,^{79–82} metal-organic complexes,^{83,84} and conjugated polymers.^{85–87}

Small organic molecules mainly include viologen and its derivatives,^{88,89} organic redox dyes,^{16,90} *etc.* Among various small molecule EC materials, the cathodic EC material 1,1'-disubstituted-4,4'-bipyridinium salts, commonly called viologens, is by far the most intensively investigated.^{91,92} The EC color properties of viologens could be readily tuned by altering the *N*-substituent group and counter anions.⁸⁸ C. Daniel Frisbie *et al.*⁷⁹ synthesized a viologen derivative (CPQ) with 4-cyanophenyl as the *N*-substituent group, which could reversibly switch from yellow to green (Fig. 4a). Later, Hong Chul Moon *et al.*⁸⁰ synthesized a viologen derivative (TFMFPhV(TFSI)₂) using 3-fluoro-4-(trifluoromethyl)phenyl as the *N*-substituent, which also exhibited a yellow-to-green color electrochromism. In comparison to CPQ, the TFMFPhV(TFSI)₂ exhibited an improved EC performance with better coloration/bleaching operational stability. And there are other strategies to achieve color modulation of viologen. Gang He's group proposed the extending *p*-conjugation method to stabilize the radical state and tune the EC properties of viologens.⁸¹ Based on this, a *p*-extended chalcogenoviologens (Se-PhV²⁺) with stable radical states and reversible color changes between yellow and dark green was successfully obtained (Fig. 4b).

Recently, Sean Xiao-An Zhang and Yu-Mo Zhang *et al.*⁸² reported a bromocresol green derivative that can also switch between yellow and green with different electrical stimulation (Fig. 4c). Unlike the EC mechanism of other small molecule materials, the EC mechanism of bromocresol green derivative is based on proton-coupled electron transfer, which is the reversible proton release/capture of molecules during the electrochemical oxidation/reduction process under voltage stimulation.⁹⁰

In addition to the above-mentioned relevant studies on small organic molecules, Olena V. Zenkina *et al.*⁸³ reported an impressive EC camouflage material based on metal-organic complexes. Due to metal-to-ligand charge transfer (MLCT), the EC properties of metal complexes were found to be greatly dependent on the electrochemical oxidation of the metal center, and its actual color is determined by the nature of the metal center and the structure of the ligand.⁹³ The authors synthesized a series of green-color EC materials by reacting a templating layer with pyridyl nitrogen atoms (Fig. 4d). The color of the EC foundation monolayer could be finely tuned by altering the *N*-quaternization ammonium group. The tests showed that these materials have different shades of green and become yellowish or clay under applied voltage. The accessible color gamut meets the requirements of military camouflage and can simulate various natural environments, including forests and sands (Fig. 4d).

Despite the late start of the research, electrochromic conjugated polymer (ECP) materials have also received remarkable

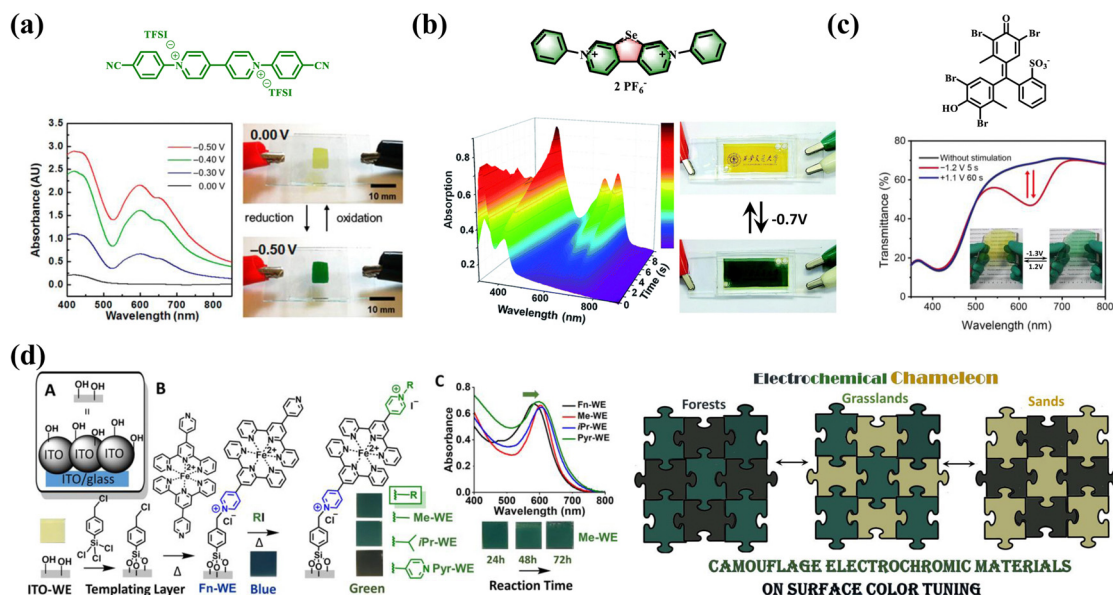


Fig. 4 (a) Molecular structure, spectra and photographs of viologen derivative CPQ in the bleached and colored states. Reproduced from ref. 79. Copyright 2016, American Chemical Society. (b) Molecular structure, UV-vis spectra, and photographs of p-extended chalcogenoviologens (Se-PhV²⁺). Reproduced from ref. 81. Copyright 2020, Royal Society of Chemistry. (c) Molecular structure, UV-vis spectra, and photographs of bromocresol green derivatives at -1.2 V and $+1.5$ V. Reproduced from ref. 82. Copyright 2021, Chinese Chemical Society. (d) Left: scheme demonstrating steps for the preparation of WEs. Right: schematic of adaptive military camouflage. Reproduced from ref. 83. Copyright 2021, American Chemical Society.

attention and shown attractive potential for adaptive military camouflage.¹⁰ Color tuning can be efficiently achieved by controlling the band gap of the ECP materials in neutral and doped states, which is the greatest advantage of ECP materials. Through the efforts of many related research and development projects, researchers have used structural design strategies to prepare many high-performance ECP materials that can be applied in military camouflage. Polyaniline (PANI), a classical ECP material, is a very desirable material for EC military camouflage due to the color switching ability between green and yellow.^{94,95} Xiaolong Weng *et al.*⁹⁶ prepared a coral-like PANI film with nanowire structures by the combination of galvanostatic and cyclic voltammetric electrodeposition techniques. The PANI film showed reversible color changes between brownish-yellow, green, and dark green, which is essential for adaptive camouflage in natural conditions (Fig. 5a). In addition, there are many factors that affect the EC properties of the PANI, such as acid dopants, thickness, and morphology of film, substrate, *etc.* Among them, acid dopants are a large and widely studied influence.⁹⁷ Commonly used acidic dopants are divided into inorganic acids such as HClO₄ and H₂SO₄ and organic acids such as dodecylbenzene sulfonate acid (DBSA) and camphorsulfonic acid (CSA).

Based on triphenylamine and carbazole structures similar to aniline, many ECP camouflage materials have also been synthesized and reported.^{85,86} As early as 2006, two ECPs, P1 and P2, were synthesized by Mario Leclerc *et al.*⁸⁵ by using the polymerization reaction between the 4-butyltriphenylamine unit and thiophene or 3,4-ethylenedioxythiophene (EDOT) (Fig. 5b). These two ECPs show good optical and EC properties for military camouflage with a color transition from pale yellow

to green. Subsequently, three new poly(2,7-carbazole) derivatives (P3–P5) (Fig. 5c) were also reported,⁸⁶ which reflect the green light in their neutral state and show the transmissive brown color after being oxidized. In addition, poly(2,7-carbazole) derivatives (P3–P5) show excellent solubility in common organic solvents such as chloroform, toluene, and tetrahydrofuran, which is important for their further commercial applications *via* solution processing.

Electrochemical polymerization is a well-established methodology to prepare polymer film directly on an electrode surface without unmanageable chemical polymerization reactions and complicated purifications.^{98,99} Cheng Zhang *et al.*⁸⁷ prepared a star-shaped thiophene derivative P6 consisting of one central core of phenyl and three arms of bithiophene by electrochemical polymerization (Fig. 5d). The P6 film can switch reversibly between orange-yellow, green and blue under applied voltage. Furthermore, the film demonstrated an outstanding EC switching speed, with the coloring and bleaching times of 1.5 s and 4.2 s between orange-yellow and green, respectively. These results show that P6 polymers have the potential to be used as a yellow-to-green EC material with rapid switching time for adaptive military camouflage. Similarly, Fatma Baycan Koyuncu *et al.*¹⁰⁰ synthesized an EC polymer based on the pyrrole–thiophene structure that can also switch reversibly between brown, green, and blue (Fig. 5e). Hong Meng, Wei Huang, and coauthors¹⁰ synthesized a donor–acceptor-type EDOT-based polymer that reversibly changes between green color and sand-tone colors by adjusting the alkyl chain on the side chain of the quinoline group (Fig. 5f).

In addition, metallic supramolecular polymers (MSPs) have been extensively studied as an emerging EC conductive polymer

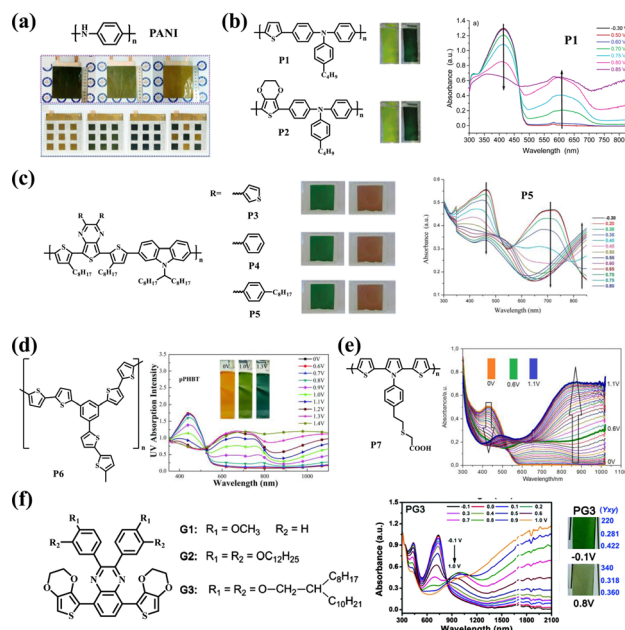


Fig. 5 (a) Molecular structure and photographs of PANI at different applied potentials. Reproduced from ref. 96. Copyright 2022, Elsevier Ltd. (b) Structure, photographs and spectra of polymers P1 and P2. Reproduced from ref. 85. Copyright 2006, American Chemical Society. (c) Structure and photographs and spectra of polymers P3–P5. Reproduced from ref. 86. Copyright 2009, American Chemical Society. (d) Structure, photographs and spectra of thiophene derivative P6. Reproduced from ref. 87. Copyright 2018, Royal Society of Chemistry. (e) Structure, photographs and spectra of polymers P7. Reproduced from ref. 100. Copyright 2018, Elsevier B.V. (f) Structure, photographs and spectra of EDOT-based polymer. Reproduced from ref. 10 (CC BY 3.0). Copyright 2016, Royal Society of Chemistry.

material.^{101–104} Metal-to-ligand charge transfer (MLCT) plays a key role in the EC behavior of MSPs.¹⁰⁵ Furthermore, the structure of the ligand also has significant influence on the EC dynamics and optical properties of MSPs.¹⁰⁶ Masayoshi Higuichi *et al.* have intensively investigated MSPs containing different central metal ions such as Fe(II), Ru(II), and Co(II), *etc.*^{107–109} They synthesized dual-redox metallo-supramolecular polymers with zigzag structures by using Fe(II) as the redox-active ion and 4,4-bis(2,2':6,2'-terpyridinyl)phenyl-triphenylamine as the redox-active ligand.¹¹⁰ Under applied voltage, the MSPs film shows a reversible color change from brown to light green, and exhibits high optical contrast and excellent cyclic stability. This kind of novel EC material has great application potential in color camouflage through the reasonable design of central metal ions and ligand structure.

Finally, the modifiability of organic EC materials allows for the design and synthesis of new materials with a variety of desired functions through the introduction of some functional groups.^{111,112} For instance, Chunyang Jia *et al.*¹¹³ reported a flexible and stretchable EC copolymer film based on triphenylamine and self-healing Diels–Alder groups with excellent self-healing properties. This EC flexible film with self-healing properties also has great potential for application in military camouflage.

3.1.3 Prepared composite materials. Considering the color switching abilities of different materials, the desired color switch can be achieved by mixing EC materials of complementary colors to form composites. According to the type of EC material, composites can be classified as inorganic/inorganic, inorganic/organic, and organic/organic types. Researchers have fabricated high-performance EC materials for adaptive military camouflage by manufacturing composite film. We will discuss each of these strategies systematically in this section to facilitate understanding.

The composite of two or more inorganic materials is an effective strategy for adjusting the color of the material.^{55,114} Shu-Hong Yu *et al.*¹¹⁵ reported multicolor inorganic EC composite film by coassembling $W_{18}O_{49}$ and V_2O_5 nanowires. By applying different voltages, the $W_{18}O_{49}/V_2O_5$ nanowires film shows dynamic colors of orange, green, and gray (Fig. 6a). Importantly, both the transmittance and color of the film could be easily controlled by manipulating the layers of coassembled nanowires and the ratios between the two nanowires. Subsequently, Rufan Zhang and coauthors improved the multicolor EC properties of vanadium oxide and the diversity of its camouflage environment by reconstructing a vanadium oxide containing various valence states of the V element.¹² The prepared composite film shows outstanding multicolor EC performance with five reversible color transformations (orange, yellow, green, gray, and blue) at voltages of less than 1.5 V (Fig. 6b). The $Au@V_xO_{2x+1}$ nanoflowers could be adapted as a functional coating material to camouflage warships in the ocean and tanks in different natural scenarios, which shows an excellent camouflage function (Fig. 6b).

In addition to color adjustment, the compound is also an effective strategy for improving EC performance. In 2018, Yucheng Wu *et al.*¹¹⁶ prepared WO_3 nanorod/ V_2O_5 dot hybrid arrays (Fig. 7a). Under the different applied voltages, the

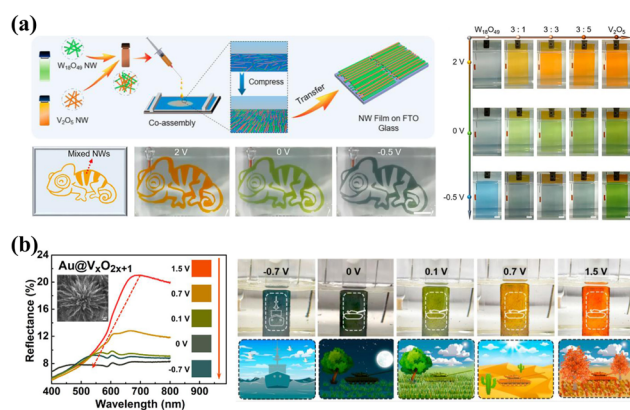


Fig. 6 (a) Left: schematic and photographs for coassembling $W_{18}O_{49}$ and V_2O_5 NWs. Right: optical images of pure $W_{18}O_{49}$ NWs, coassemblies, and pure V_2O_5 NWs under different applied voltages. Reproduced from ref. 115. Copyright 2021, American Chemical Society. (b) Left: optical reflection spectra and corresponding optical images of $Au@V_xO_{2x+1}$ NFs films at different applied voltages. Right: application scenarios of the $Au@V_xO_{2x+1}$ NFs in adaptive camouflage fields. Reproduced from ref. 12. Copyright 2022, American Chemical Society.

composite film could reversibly switch between black, green, and orange-yellow based on the color overlay (Fig. 7a). Meanwhile, this well-designed structure provides ion transport channels and significantly increases the active sites for electron and ion transfer. Thus, the nanorod arrays exhibit significantly enhanced EC properties with superior cyclic stability compared to the individual components. Compounding EC materials with transparent conducting nanomaterials provides a large electrochemically active surface area and an abundance of gaps, thus resulting in fast response times and excellent stability.¹¹⁵ Jiangping Tu and coauthors¹¹⁷ prepared a $\text{SnO}_2/\text{V}_2\text{O}_5$ core/shell composite film with a rough surface by combining a hydrothermal method and electrodeposition (Fig. 7b). By alternating the strength of the applied voltages, the composite film could reversibly transform between yellow, green, and blue, which allows the material to be used as an adaptive camouflage material in all conditions of sand, forest and sea for military applications. Meanwhile, the composite film demonstrated superior EC performance to the single V_2O_5 film in optical modulation and cyclic stability (Fig. 7b).

Another viable strategy is to combine inorganic with organic to create inorganic/organic hybrid electrodes to realize multi-color camouflage of composite films. Guofa Cai *et al.*¹¹⁸ fabricated a WO_3/PANI core/shell dual EC nanowire array film (Fig. 8a). Composite electrodes containing the cathodic coloration material WO_3 and the anodic coloration material PANI could lead to a dual-electrochromism effect by separating the electroactivity within the application potential window, enabling a reversible color change between green, yellow, and dark blue (Fig. 8b). Recently, to improve the color accessibility of V_2O_5 film, a high-performance V_2O_5 -methyl cellulose (VMC) composite film was fabricated by Quanyao Zhu, Jinyong Zhang, and coauthors (Fig. 8c).¹¹⁹ The as-prepared VMC composite film shows various colors (yellow, green, blue, chartreuse, emerald, medium, and dark green) at different applied voltages. Based on this, the as-prepared inorganic/organic hybrid

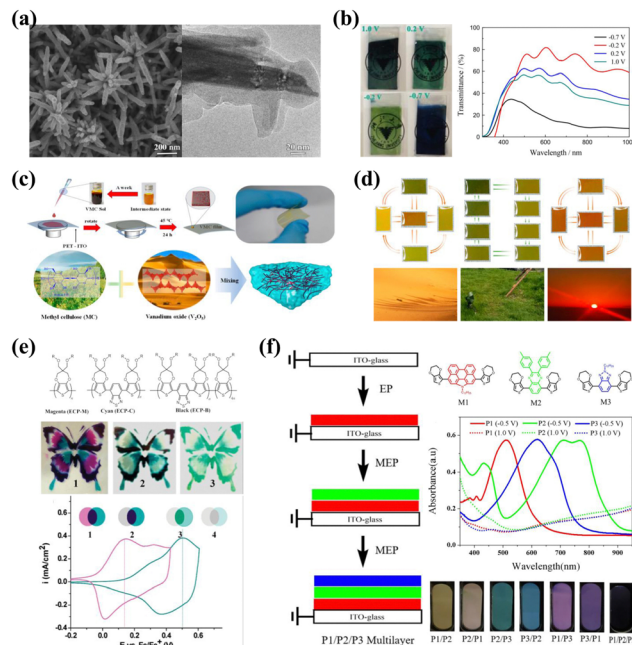


Fig. 8 (a) SEM and TEM images and (b) photographs and transmittance spectra of a WO_3/PANI core/shell nanowire array. Reproduced from ref. 118. Copyright 2014, Elsevier B.V. (c) Top: schematic of the spin-coating process. Bottom: morphology and structure of the VMC composite film. (d) Fully colored VMC RECDs that correspond to natural environments. Reproduced from ref. 119. Copyright 2022, American Chemical Society. (e) Repeat unit structures of the polymers, photographs of a multicolored "butterfly" ECD, and cyclic voltammograms of the polymers. Reproduced from ref. 123. Copyright 2011, American Chemical Society. (f) Left: schematic of the MEP for P1/P2/P3 Multilayers. Right: molecular structures of the monomers and absorbance spectra of polymers in the neutral state and oxidized state. Reproduced from ref. 126. Copyright 2021, Elsevier B.V.

film is able to simulate natural environments, including grassland and desert (Fig. 8d).

In addition, the preparation of composites using two or more organic materials by solution blending or stacking strategies is a promising and straightforward idea for color modulation.¹²⁰ The color mixing theory based on the subtraction model of cyan-magenta-yellow-black (CMY-K) and red-yellow-blue (RYB) has been widely used in the electrochromic field.¹²¹ It opens up endless possibilities for the use of organic EC materials, including military camouflage. John R. Reynolds's group has carried out many impressive studies on organic blend and stack materials.^{121,122} For example, they prepared "butterfly" EC composite films by stacking two ECP materials which display magenta and cyan, respectively.¹²³ The repeat unit structure of the two conjugated polymers is presented in Fig. 8e. Following "color-mixing theory" principles, the two polymer stacks appear in black. As the voltage increased, the butterfly device exhibited multicolor switching due to the different oxidation voltages of the polymer film (Fig. 8e). Similar approaches were also reported by Jianguo Mei and coauthors.¹²⁴ As reported in this paper, under ultraviolet light illumination, the EC polymer was cross-linked with the crosslinker to endow them with solvent-resistant properties,

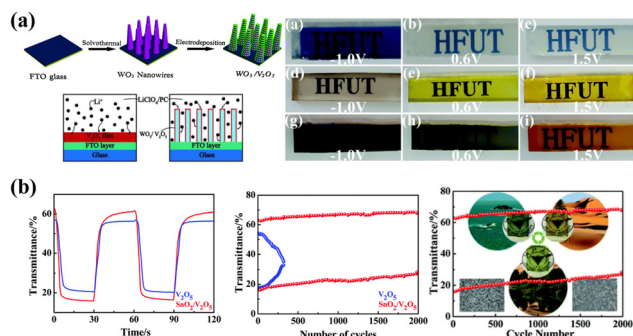


Fig. 7 (a) Top left: schematic of the formation process of $\text{WO}_3/\text{V}_2\text{O}_5$ hybrid films. Bottom left: a model of insertion/extraction of Li^+ ions in different structures. Right: digital photographs of films at different applied voltages. Reproduced from ref. 116. Copyright 2018, Royal Society of Chemistry. (b) Left: transmittance curves and cyclic stability of V_2O_5 and $\text{SnO}_2/\text{V}_2\text{O}_5$ films. Right: schematic of adaptive military camouflage. Reproduced from ref. 117. Copyright 2019, Royal Society of Chemistry.

resulting in the preparation of multilayers. Moreover, the authors constructed a full-color palette by varying the film thickness of each ECP layer. Finally, the goal of color tuning could also be achieved by using the solubility of the ECPs material for solution mixing.¹²⁵

In addition to the above-mentioned related studies, another classic method of tuning color is the preparation of multilayer composite films through multilayer electrochemical polymerization (MEP) technology, as reported by Cheng Zhang, Weijun Li, and coauthors (Fig. 8f).¹²⁶ As shown in that work, the authors chose three EDOT-based monomers (M1–M3), which exhibited red, green, and blue colors after electrochemical polymerization. The selection of the monomers should take into account the polymerization voltage window of the monomer and the EC voltage window of the polymer. Then, the multilayer films based on three individual polymers were successfully deposited onto ITO-glass by the MEP technology. By controlling the thickness of each polymer layer, multilayers of various colors have been successfully prepared, including pleasing black EC film (Fig. 8f).

3.2 EC devices for color camouflage

The design and manufacture of ECDs are key steps toward the application of materials. In the conventional design, ECD is usually a “sandwich” structure consisting of two substrates, two conductive electrodes, an EC layer, an electrolyte layer, and an ion storage layer, making up a total of seven layers. The complex device structure also provides a margin for color adjustment and performance improvement. In the following sections, we will focus on EC camouflage devices and discuss their design strategies and underlying issues.

3.2.1 EC camouflage devices based on camouflage material. Based on the existing yellow-to-green switching camouflage materials, scientists have designed and prepared many EC

camouflage devices and demonstrated their feasibility in camouflage applications. One of the most straightforward strategies for camouflage devices is to use a camouflage material as the EC layer, followed by a colorless electrolyte and a colorless transparent substrate. Based on this strategy, Mario Leclerc *et al.*⁸⁵ fabricated an EC camouflage cell by using conjugated polymer P1 as the EC layer and bare ITO as the counter electrode (Fig. 9a). The prepared device could achieve reversible change from yellow to green at applied voltages of 0.0 V and 2.4 V, which is the same color as that observed for thin film in solution. Finally, a hybrid plastic/textile EC cell was also prepared to demonstrate its application in wearable adaptive camouflage. However, the irreversibility of ITO redox leads to a large driving voltage and poor stability of the device.⁴⁶ This makes it unsuitable for industrial applications of EC camouflage devices.

To improve the EC performance of the camouflage device, an ion storage layer with excellent stability is required. Poly(3,4-ethylenedioxythiophene) (PEDOT) and poly(3,4-ethylenedioxythiophene)/poly(styrenesulfonate) (PEDOT: PSS) is commonly used as an ion storage layer due to its excellent electrochemical and chemical stability.^{127,128} Fatma Baycan Koyuncu *et al.*¹⁰⁰ constructed a complementary ECD by using camouflage material poly(SNSPA) (P7) as the EC layer and PEDOT: PSS as the counter layer. The prepared device has a fast switching time (about 0.5 s) and lasting cyclic stability (98% of the optical contrast is retained after 1000 steps). For camouflage devices, the use of colored ion storage layers not only reduces the contrast of the device, but it is also not easy for the device to achieve a specific camouflage color or hue. Using colorless ion storage materials can effectively solve this problem. Olena V. Zenkina *et al.*⁸³ assembled a solid-state EC camouflage device using TiO₂ anatase transparent paste as the counter electrode (Fig. 9b). The result demonstrated that the device retained the

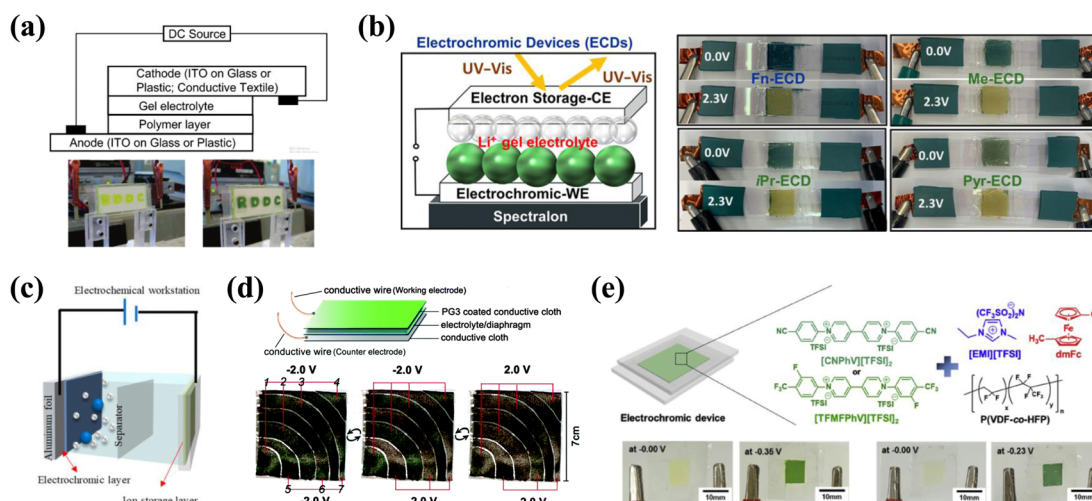


Fig. 9 (a) Schematic and photographs of sandwich-type EC cells. Reproduced from ref. 85. Copyright 2006, American Chemical Society. (b) Left: schematic of the solid-state ECDs. Right: photographs of the ECD at the colored and bleached states. Reproduced from ref. 83. Copyright 2021, American Chemical Society. (c) Schematic of VMC RECDs. Reproduced from ref. 119. Copyright 2021, American Chemical Society. (d) Schematic and photographs of fabric chameleonic ECD. Reproduced from ref. 10 (CC BY 3.0). Copyright 2016, Royal Society of Chemistry. (e) Top: ECD configuration and molecular structures of components in the EC gels. Bottom: photographs of the ECD. Reproduced from ref. 80. Copyright 2017, Elsevier B.V.

color switching of EC materials while also showing high long-term durability (95% retention of the contrast after 3300 steps), fast response time (less than 4 s) and excellent coloration efficiency. In addition, other colorless ionic storage layers, such as inorganic oxides,^{129,130} poly(2,2,6,6-tetramethylpiperidinyloxy-4-ylmethacrylate) (PTMA) derivatives,^{131–133} and conjugated polymers,^{43,46} could also be applied to camouflage devices. Notably, the matched electroactive voltage window of the ion storage material and the EC material is also a major influence on device performance.⁴⁵

In addition to the above-mentioned relevant strategies, Zhang Jinyong's group has reported a clever method for preparing reflective camouflage devices.¹¹⁹ In this work, a reflective camouflage device was fabricated with two pieces of VMC composite film, one of which served as the working electrode and the other was pre-built with Li^+ as the counter electrode. To isolate color interference with the counter electrodes and achieve a high optical reflectivity of the device, the authors placed a white diaphragm between the working electrode and the counter electrode (Fig. 9c). In addition, aligning the EC layer laterally with the ion storage layer was also a strategy for preparing camouflaged devices, as reported by Meng Hong *et al.* (Fig. 9d).¹⁰

Unlike other EC materials, viologen is usually mixed with electrolytes to prepare integrated gel devices due to its solubility differences in different states.⁷⁹ Hong Chul Moon *et al.*⁸⁰ prepared an EC ion gel based on viologen derivative CPQ and TFMFPPhV(TFSI)₂, and dimethyl ferrocene (dmFc) was used as a counter redox material (Fig. 9e). It has been demonstrated that the addition of Fc could decrease the operating voltage and improve the cycle stability of the EC device.¹³⁴ The prepared gel device achieves switching from yellow to green reversibly (Fig. 9e).

3.2.2 EC camouflage devices based on device design strategies. Preparation of devices for color camouflage by preparing EC materials for camouflage color switching is tedious. In contrast, the direct use of device structures to prepare high-performance devices for camouflage applications is a simple and effective method. The electrodes, electrolytes, and substrates could all be optimally designed to tune and enhance the color and performance of the device. In the following sections, we will discuss and summarize these device design strategies in detail.

3.2.2.1 EC electrode color complementary. The parallel connection of two or more devices using double-sided conductive electrodes is one of the most common means of achieving multicolor device preparation.^{135,136} John R. Reynolds *et al.*³⁸ reported a multicolor prototype EC device based on soluble conjugated polymer. In this work, a dual-active EC device based on subtractive color mixing utilizing the cyan-magenta-yellow (CMY) primary system was prepared (Fig. 10a). Essentially, the dual-active device was the integration of two independently controlled EC devices into a pseudo-three-electrode setup since the two counter electrodes share a substrate. By controlling the voltage applied across each film, there were various color states

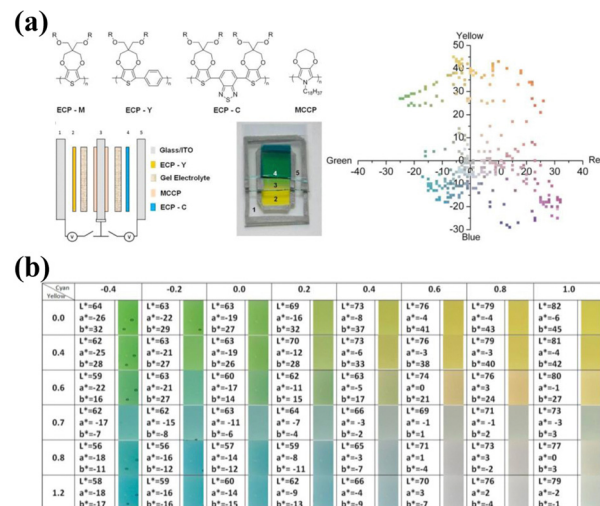


Fig. 10 (a) Left: repeat unit structures of ECPs, and exploded view and photographs of a dual-active ECD. Right: plot of a^*b^* values of all color points recorded with dual-active devices. (b) Photographs and $L^*a^*b^*$ coordinates dual-active (bottom) cyan + yellow ECDs. Reproduced from ref. 38. Copyright 2014, American Chemical Society.

accessible to the device. The dual-active structure device with ECP-cyan and ECP-yellow as working electrodes could be reversibly switched between a variety of yellows and greens (Fig. 10b). Similarly, there are other strategies to achieve this state, for instance, an employed three-stacked ECD reported by Jong S. Park, Sung Kyu Park and coauthors.¹³⁶ The multi-stacked ECDs were implemented by the combinatorial stacking of the coloration devices, exhibiting a wide range of color adjustments, including black.

Although the multilayer device structure allows access to a wide gamut of color states, its complex structure makes it not the desired route in many applications. A dual EC electrode device is a much simpler solution, particularly for color camouflage applications where a single, particular color state is desirable. Typically, the EC layer plays a central role in devices, undertaking the task of optical modulation or color change. The ion storage layer acts as a counter electrode layer that undergoes reversible electrochemical oxidation (reduction) to balance the charge shuttling from the EC layer through the electrolyte without color change.⁴⁵ However, in a dual EC electrode device, both electrodes are complementary to each other for the purpose of color adjustment.^{137,138} Based on the color stacking strategy, the device could achieve the desired multicolor switching capability through rational design. Additionally, the structure of the dual EC electrode device is conducive to enhancing the EC performance (optical contrast, stability, *etc.*) of the ECD through rational design.^{137,138}

A similar method has been widely used to prepare high-performance yellow-green switching camouflage devices. Yao Li *et al.*¹³⁹ prepared all-solid-state ECDs using WO_3 and V_2O_5 as complementary electrodes (Fig. 11a). According to the subtractive color mixture theory, the all-solid-state EC device appears yellow-green at 2.5 V due to the transparent WO_3 and yellow-green V_2O_5 in the bleached state. At -2.5 V, the all-solid-state

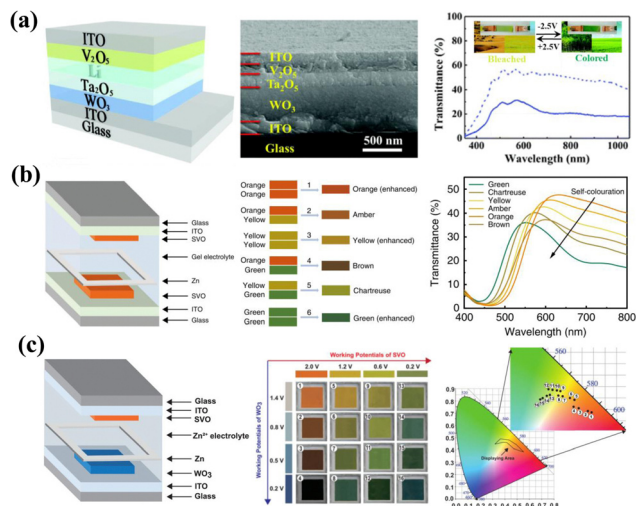


Fig. 11 (a) Structural schematic, cross-sectional SEM image, and transmittance curves of the ECD. Reproduced from ref. 139. Copyright 2020, Royal Society of Chemistry. (b) Schematic of the Zn-SVO ECD, and schematic and transmittance spectra of the color overlay effect. Reproduced from ref. 140 (CC BY 4.0). Copyright 2020, Springer Nature. (c) Schematic, digital images and CIE color coordinates of the SVO-Zn-WO₃ ECD. Reproduced from ref. 141. Copyright 2022, Wiley-VCH GmbH.

ECD appears emerald green due to the mixing of blue WO₃ and yellow V₂O₅ in the colored state. This well-designed ECD achieves a reversible color change between yellow-green and emerald green (Fig. 11a), as well as excellent cycle stability (> 5000 cycles). The two complementary electrodes could not be controlled independently in complementary ECD, which significantly limits its multicolor performance. A Zn-based ECD was reported to efficiently solve this problem by Abdulhakem Y. Elezzabi, Haizeng Li, and coauthors.¹⁴⁰ The two sodium vanadium oxide electrodes could be colored and bleached independently due to the introduction of the Zn sandwich

(Fig. 11b). Due to the color superposition of two electrodes, the prepared ECD achieved a reversible switch between six colors (including a variety of yellow and green colors) (Fig. 11b). To further broaden the color palettes of devices, WO₃ and sodium vanadium oxide films were selected as two asymmetric electrodes in the device.¹⁴¹ As shown in Fig. 11c, the Zn-anode-based EC devices of dual EC electrodes could switch between 16 colors based on the color overlay effect. In essence, the device could display all the color shades within the enclosed area, including a wide range of green and yellow hues (Fig. 11c).

In addition, it has been shown that dual polymer ECDs using two complementary ECP electrodes could enhance overall EC performance due to their electrochemical reversibility and compatibility with the polymer electrodes. Cheng Zhang and Weijun Li *et al.*¹¹ reported a dual conjugated polymer ECD. In this report, the authors synthesized two soluble EC materials that possess a yellow-to-transmissive and green-to-transmissive electrochromism, respectively (Fig. 12a). At the same time, the two ECPs prepared have approximate electrochromic potentials, which is an essential factor for the preparation of the desired dual conjugated polymer ECDs with good stability and the expected switch color change. Then, the dual conjugated polymer ECD with P1 and P2 as complementary electrodes was fabricated (Fig. 12a). This well-designed ECD presented yellow-to-green electrochromism, which showed excellent EC performances, including low driving voltage, fast switching speed, and lasting cyclic stability (Fig. 12b). Finally, the patterned and flexible dual-conjugated polymer devices demonstrated promising applications in color camouflage (Fig. 12c). This work is an important step towards enabling enhanced capabilities for EC color camouflage.

3.2.2.2 Electrolyte coloring engineering. Similarly, the color of the electrolyte layer can be introduced to broaden the palette of

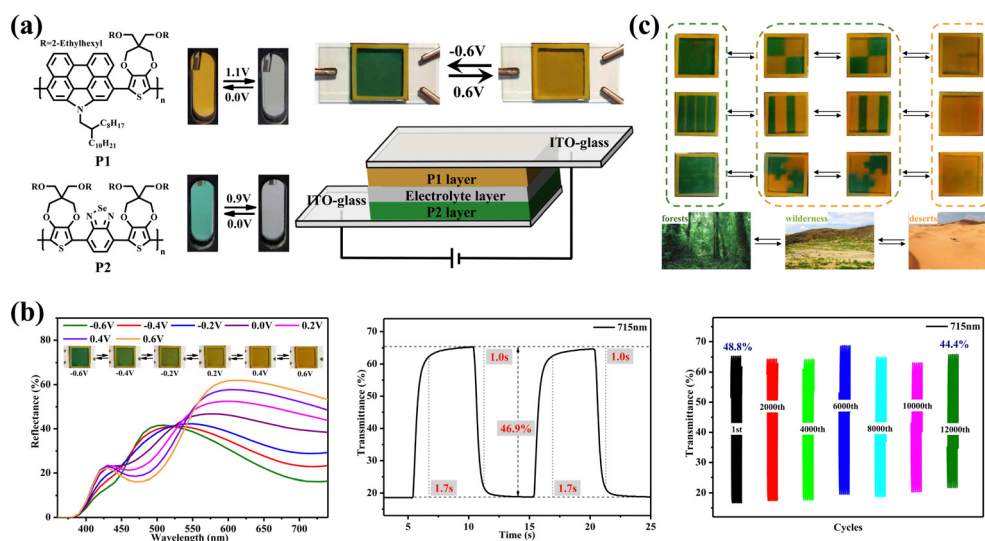


Fig. 12 (a) Left: structure of polymer and colors of the neutral and oxidized states of the film. Right: the structure and color of yellow-to-green ECDs. (b) EC properties of yellow-to-green ECDs. (c) Military camouflage ECDs. Reproduced from ref. 11. Copyright 2022, Elsevier B.V.

EC devices. Hong Meng, Xing Xing, and coauthors reported EC smart windows based on a temperature-stimulated phase hydrogel electrolyte.¹⁴² In this work, the electrolyte layer could achieve a highly scattered zero-transparent private state controlled by temperature. Thus, the EC devices could also switch between a nontransparent “white private state” (only scattered white light is transmitted) and an “absolute private state” (dark, zero-transmission state). It provides a new strategy for the design of EC device engineering. Furthermore, they proposed a strategy by introducing a thermochromic electrolyte so that the electrolyte layer and the EC layer share responsibility for the color change of the devices (Fig. 13a).¹⁴³ It not only provides a strategy for color tuning of ECDs but is also a viable idea for high-performance yellow-to-green switching camouflage.

In addition, the direct introduction of colored electrolytes without the function of color change to achieve color tuning in EC devices is a much simpler and more effective method, as reported by Aline Rougier and coauthors.¹⁴⁴ In this work, the authors tuned the color of the electrolyte by mixing four pigments into a gel polymer electrolyte, which resulted in four different colors of the electrolyte (white, yellow, red, and green) (Fig. 13b). In the white electrolyte device, the device could switch reversibly between blue and light blue, showing the color change of the PEDOT:PSS. Notably, the high opacity of

the white electrolyte isolates the two EC layers from color interference, and at the same time, produces a large reflection value, which is very favorable for reflective-type camouflage devices. Introducing a white diaphragm into the electrolyte could also achieve this goal.¹¹⁹ In other color electrolyte devices, depending on the electrolyte, different colors could be observed when the device is in the oxidized and reduced states. By changing the pigment color, a colored opacity electrolyte could be imagined and integrated into the camouflage device to achieve color tuning.

3.2.2.3 Complementary substrate colors. EC camouflage devices are reflective-type devices and do not require a transparent substrate layer like transmissive-type devices (e.g., smart windows). The color of the substrate layer of the device can also be cleverly used. The use of yellow gold/nylon 66 porous substrates in military camouflage devices was reported by Yao Li and coauthors.¹⁴⁵ In their work, a flexible bio-inspired chameleon skin EC device was assembled with WO_3 porous film (Fig. 14a). As we all know, WO_3 film can realize the reversible color switch between transparent and blue by applying voltage. According to the subtractive color mixture theory, the ECD they prepared could achieve a reversible transition from yellow to green. Meanwhile, under different applied voltages, various tones of green could be obtained according to the blue tones of WO_3 (Fig. 14b). They achieved the same result by using Prussian blue (PB) (transparent to blue) with a similar strategy.¹⁴⁶

3.2.3 Preparation of wearable fabric devices. Unlike other EC applications (e.g., smart windows, displays, etc.), the design and fabrication of high-performance large-area wearable fabric devices are critical for EC adaptive camouflage applications. Many practical and clever device structures and preparation techniques have been reported to achieve this goal. The commonly used strategies for preparing wearable fabric EC devices

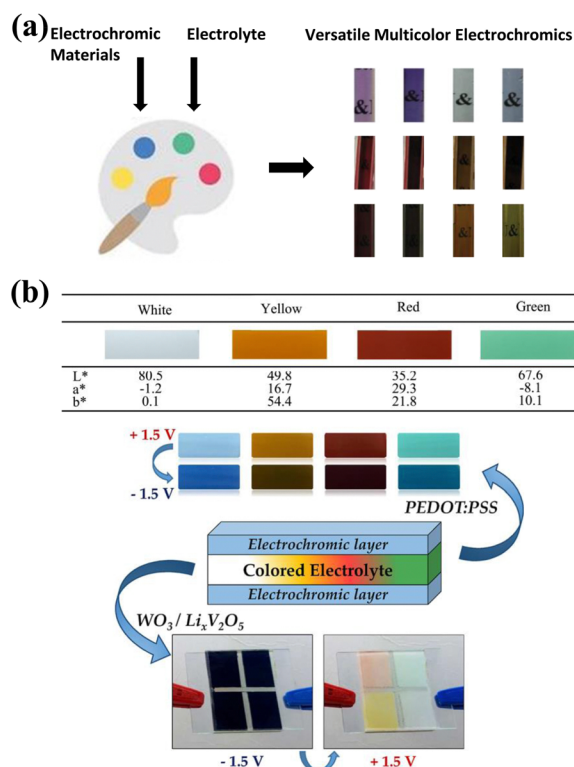


Fig. 13 (a) Digital images of the ECD with the thermochromic electrolyte. Reproduced from ref. 143. Copyright 2022, Wiley-VCH GmbH. (b) Top: photographs and $L^*a^*b^*$ coordinates of electrolytes. Bottom: digital images of the ECD with the colored electrolytes. Reproduced from ref. 144. Copyright 2022, Elsevier B.V.

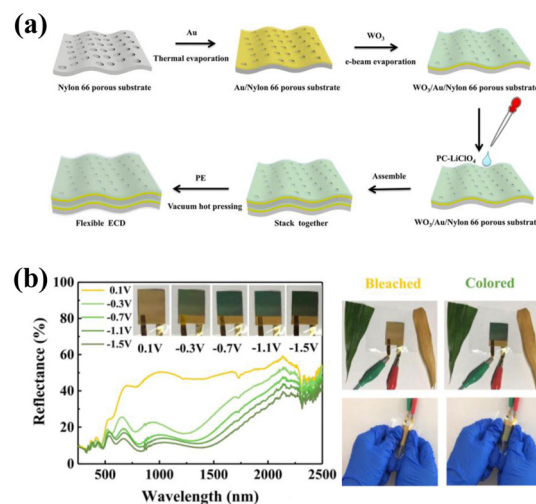


Fig. 14 (a) Schematic for the assembling of flexible ECD. (b) Reflectance spectra and digital photos of the flexible ECD at different voltages. Reproduced from ref. 145. Copyright 2021, Elsevier B.V.

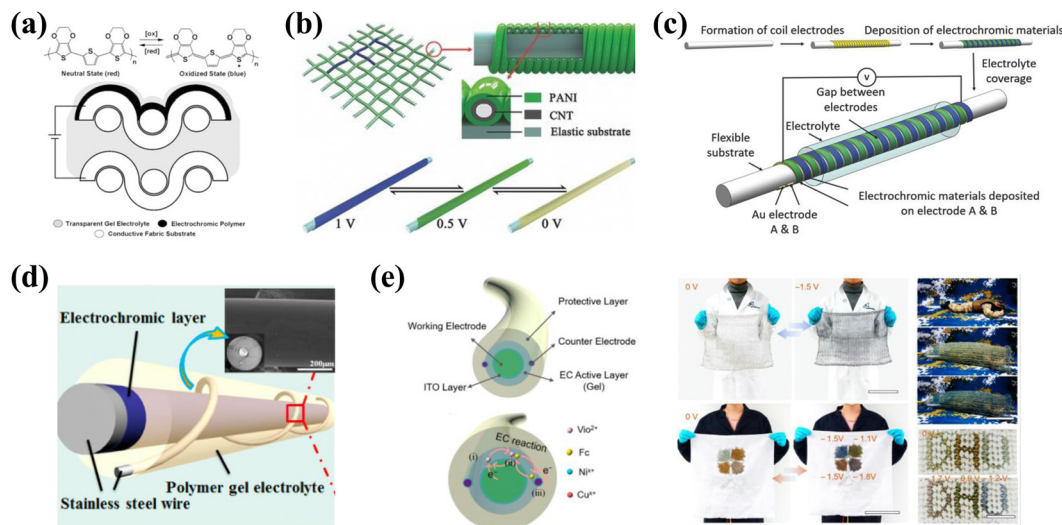


Fig. 15 (a) Top: structure of the EC polymer. Bottom: cross-section schematic of fabric ECD. Reproduced from ref. 152. Copyright 2010, American Chemical Society. (b) Schematic of the EC, wearable fiber-shaped supercapacitor. Reproduced from ref. 153. Copyright 2014, John Wiley and Sons. (c) Schematic of parallel coil electrodes. Reproduced from ref. 154. Copyright 2014, Wiley-VCH GmbH. (d) Structure diagram of the EC fibers. Reproduced from ref. 149. Copyright 2014, American Chemical Society. (e) Left: structural scheme and schematic representation of electrochemical reactions of the EC fibers. Right: multifunctional applications of EC fibers. Reproduced from ref. 150. Copyright 2014, American Chemical Society.

include (i) depositing EC film directly on fabric substrates^{147,148} and (ii) preparing fibrous EC devices.^{149,150}

Gregory A. Sotzing *et al.*¹⁵¹ reported a stretchable e-textile by simply soaking Spandex fabric in a PEDOT-PSS aqueous dispersion. The electrical conductivity of the fabric varies between 0.1 and 2.0 S cm⁻¹ due to the substrate; the lower conductivity also leads to the slow response (color change) of the EC fabric. To efficiently solve this problem and further improve the response of fabric devices, they developed a modified reflective-type fabric device architecture using steel counters as the two electrodes (Fig. 15a).¹⁵² In this device, the EC polymer switches between different colored states at a time of *ca.* 0.3 s due to the high conductivity (*ca.* 9800 S cm⁻¹).

Compared with conventional fabric-based structure devices, fiber-based EC devices offer many unique and promising advantages, such as a higher integration capability and the ability to be processed by conventional weaving, spinning, or knitting techniques. Huisheng Peng *et al.*¹⁵³ fabricated a fiber-shaped EC supercapacitor *via* electro-depositing polyaniline (PANI) onto sheets of aligned carbon nanotubes (CNT) as two electrodes (Fig. 15b). The as-prepared fiber-shaped EC supercapacitor could be further woven into fabrics to display designed patterns. Furthermore, fiber-based devices could make rapid and reversible transitions between blue, green, and light yellow under different applied voltages (Fig. 15b). Another approach worthy of attention is a design strategy based on lateral device structures, as reported by Tong Lin and coworkers.¹⁵⁴ The required dual helix electrodes were prepared by depositing tungsten oxide and poly(3-methylthiophene) onto two electrodes, respectively (Fig. 15c). The results showed that the lateral helix fiber devices could reversibly switch between dark red, green, and gold at the different applied potentials. Furthermore, the fiber devices demonstrated high flexibility

and fast response. This is expected to be useful for the development of smart apparel and adaptive military camouflage in the future.

In addition to the above-mentioned related studies, Hongzhi Wang's group performed many impressive studies on EC fiber-based devices.^{149,150} They prepared fiber devices using stainless steel wires (SSWs) as a substrate (Fig. 15d).¹⁴⁹ Three RGB EC materials were deposited on the surface of SSWs by electrochemical polymerization. The EC layer was then coated with a polymer gel electrolyte, and finally, another finer stainless steel wire was screwed on top as a counter electrode. These well-designed EC fibers could reversibly switch between sky blue and dark blue (PEDOT), blue and red (poly(3-methyl-thiophene) (P3MT)), and green and yellow (poly(2,5-dimethoxyaniline) (PDMA)), respectively. Furthermore, to make fiber devices more practical, they demonstrated the preparation of hundreds of meters of EC fibers based on a parallel dual-counter-electrode structure using custom-built equipment (Fig. 15e).¹⁵⁰ The use of a parallel dual-counter-electrode structure allows for a more even distribution of the electric field and faster electron transfer, and improved the time and uniformity of the color change. Finally, the preparation of large-area smart color-changing textiles demonstrates their potential application in the field of EC color camouflage (Fig. 15e).

4. EC materials and devices for thermal camouflage

With the joint efforts of many researchers, research on EC materials and devices is not limited to the visible light field. Recently, infrared (IR) EC materials and devices have been extensively studied due to their great potential for applications

in smart windows, thermal camouflage, *etc.*^{6–8,155} Unlike the color camouflage discussed earlier, thermal camouflage can achieve infrared concealment of a target by tuning the thermal radiation to be the same or similar to the background environment.¹⁵⁶ According to the Stefan–Boltzmann law, the regulation of thermal radiation can be achieved by tuning either the local temperature field or the local emissivity of the target. Among them, EC thermal camouflage is achieved by adjusting the emissivity. Due to atmospheric windows, most infrared detectors operate in the mid-infrared (MIR) range within two wavelengths ranges, 3–5 μm and 8–13 μm . Thus, the modulation range of an expected camouflage material should encompass these wavelengths and exhibit the greatest modulation capability. The currently reported IR emissivity modulated EC materials can be divided into inorganic EC materials such as tungsten oxide (WO_3),^{157–159} lithium titanate ($\text{Li}_4\text{Ti}_5\text{O}_{12}$),¹⁶⁰ and carbon materials,^{161,162} *etc.*, and organic EC materials such as polyaniline (PANI).^{163,164} In this section, we briefly describe recent advances in inorganic and organic EC materials for thermal camouflage EC materials and devices.

WO_3 is a promising candidate for thermal camouflage applications. In 1999, John A. Woollam *et al.*¹⁶⁵ studied the emissivity of semicrystalline WO_3 and NiO in their unintercalated and fully intercalated states within the range of 2–14 μm . The authors obtained a large modulation range and a high modulation of the IR emissivity ($\Delta\varepsilon = 0.538$) by adjusting the

thickness of the NiO and WO_3 film. A. Rougier *et al.*¹⁵⁸ studied the IR modulation properties of WO_3 film in detail. As reported in this paper, deposition conditions, including the duration of deposition, the substrate natures, and the chamber pressure, have an important influence on the IR properties of WO_3 films. Controlling the crystallinity and micro-structure are the two main ways to improve the emissivity modulation of WO_3 . Yao Li *et al.*¹⁵⁹ studied the effects of crystallinity on the IR emittance of WO_3 film by preparing amorphous WO_3 (a- WO_3) and crystalline WO_3 (c- WO_3). The results indicate that the c- WO_3 all-solid-state EC device showed a higher emittance regulation of 0.37 in the 8–14 μm range and 0.3 in the 2.5–25 μm range (Fig. 16a). J.P. Tu, G.F. Cai, and coauthors reported vertically aligned hierarchical WO_3 nano-architectures.¹⁶⁶ The nanostructured array film with thicknesses of about 1.1 μm show remarkable enhancement of the EC properties in the IR range.

In 1995, Tsutamu Ohzuku, Atsushi Ueda, and coauthors¹⁶⁷ reported the color change during the transition from the $\text{Li}_4\text{Ti}_5\text{O}_{12}$ phase (white) to the $\text{Li}_7\text{Ti}_5\text{O}_{12}$ phase (blue) upon lithium insertion. With further research, the light modulation properties of $\text{Li}_4\text{Ti}_5\text{O}_{12}$ extend from the VIS to the IR region, suggesting that $\text{Li}_4\text{Ti}_5\text{O}_{12}$ is also a promising EC material for thermal camouflage.¹⁶⁸ Yuan Yang *et al.*¹⁶⁰ reported the broadband EC properties of $\text{Li}_4\text{Ti}_5\text{O}_{12}$ and further investigated its potential in thermal camouflage applications. Upon Li^+ intercalation, the transition of LTO from the delithiated (DL) state

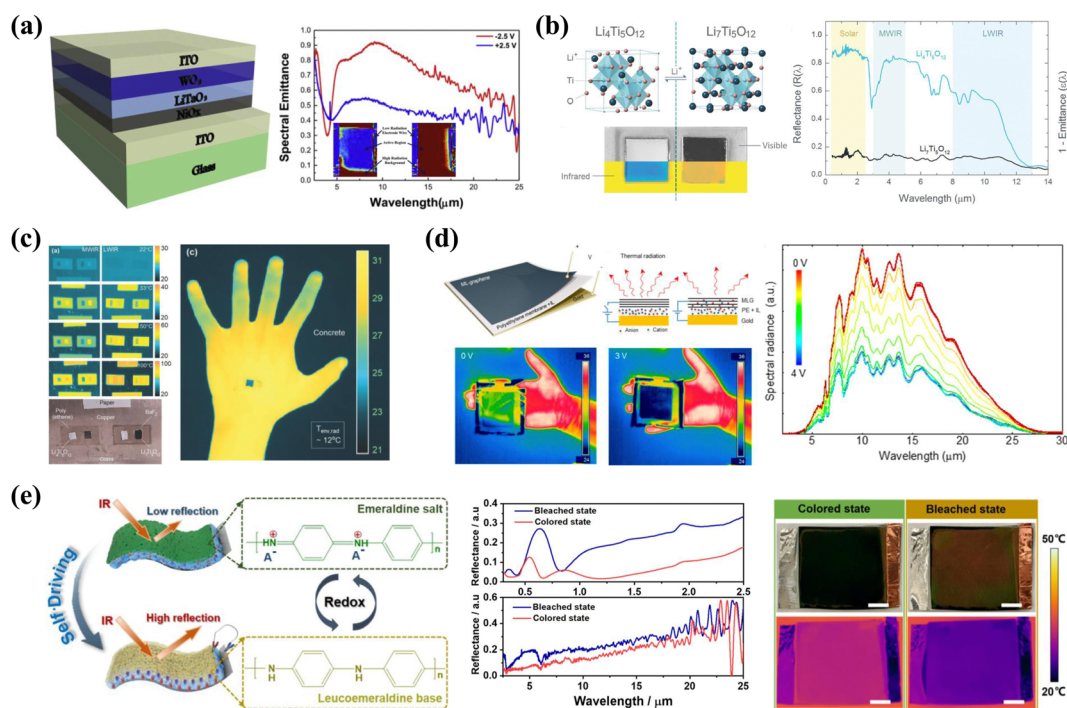


Fig. 16 (a) Left: schematic of ECD structure. Right: spectral emittance curves and IR optical photographs of ECDs. Reproduced from ref. 159. Copyright 2019, Elsevier B.V. (b) Crystal structures, photographs, and visible-to-infrared spectral reflectance of $\text{Li}_4\text{Ti}_5\text{O}_{12}$ and $\text{Li}_7\text{Ti}_5\text{O}_{12}$ nanoparticles. (c) Left: MWIR and LWIR thermographs of LTO. Right: LWIR thermograph of a hand, with PE-coated L and DL LTO, held against concrete. Reproduced from ref. 160. Copyright 2018, Wiley-VCH GmbH. (d) Left: schematic drawing and the working principle schematic of the active thermal surface. Right: spectra of the thermal radiation from the device at different bias voltages. Reproduced from ref. 13. Copyright 2018, American Chemical Society. (e) Self-driven behavior and IR reflectance management of the IR-ECD. Reproduced from ref. 164. Copyright 2021, American Chemical Society.

($\text{Li}_4\text{Ti}_5\text{O}_{12}$) to the lithiated (L) state ($\text{Li}_7\text{Ti}_5\text{O}_{12}$), resulting in a drastic change in the electromagnetic properties and thus in the optical properties between the two states in the VIS to IR range (Fig. 16b). The results demonstrated that the $\text{Li}_4\text{Ti}_5\text{O}_{12}$ nanoparticles electrodes showed large emissivity tunability of 0.74, 0.68, and 0.30 in the solar reflectance, mid-wave infrared emittance (3–5 μm), and long-wave infrared emittance (8–13 μm), respectively (Fig. 16b). Finally, the authors also tested the temperature regulation range of the LTO-based electrode (Fig. 16c).

Carbon materials such as graphene and carbon nanotubes have tunable Fermi energy levels and excellent electrical conductivity, making them suitable candidates for EC thermal camouflage materials.^{161,162} Coskun Kocabas *et al.*¹³ reported multilayer graphene that enables modulation of IR emissivity by reversible insertion of non-volatile ionic liquids under the action of electricity (Fig. 16d). The emissivity of the multilayer graphene electrode could be switched between high (0.8) and low states (0.3) at bias voltage while also exhibiting a fast switching time (<1 s). To achieve the preparation of large-area EC thermal camouflage devices, a sandwich-structured thin film based on multiwalled carbon nanotubes (MWCNTs) was reported by Lin Xiao and coauthors.¹⁶² The prepared film demonstrates efficient EC modulation with a thermal emissivity in the range of about 0.15–0.7. In addition, excellent cycling stability (optical modulation retains 89% after 3500 steps) was obtained in this film. Mid-infrared ECDs based on MWCNT films could achieve large-area preparation and superior EC performance, such as large modulation depth and high stability, and are expected to be used in adaptive thermal camouflage.

During the reversible electrochemical oxidation and reduction process, the conducting polymers generally exhibit strong IR absorption bands due to the formation of polarons and bipolarons.¹⁶⁹ The formation of polarons results in absorption peaks usually in the VIS or near-infrared (NIR) region, while the absorption peaks resulting from bipolarons formation may be in the MIR range or even the far-infrared (FIR) range. The resulting modulation of emissivity in the MIR region shows attractive application potential in thermal camouflage.

PANI demonstrates excellent emissivity modulation capabilities in the MIR spectral range, making it already show significant promise in thermal camouflage. Researchers have been researching the properties of PANI to modulate in the MIR since as early as 1999. P. Topart *et al.*¹⁷⁰ reported an IR-modulated EC device using PANI film as the active layer. The assembled device demonstrated properties of excellent MIR modulation during the doping/dedoping process, which realized a reflectivity variation of about 0.45 at 12 μm for potentials from –0.2 to 0.45 V. Subsequently, Prasanna Chandrasekhar and coauthors⁹⁴ fabricated a two-electrode flexible all-solid-state device capable of 0.4 through 45 μm region modulation. The as-prepared device demonstrated high emittance variation from 0.32 to 0.79 ($\Delta\epsilon = 0.4$) in the range of 2.5 to 45 μm .

An in-depth study of the IR EC mechanism of PANI is necessary to improve the IR modulation efficiency for its application in thermal camouflage. Many groups have contributed significantly to the mechanistic study of the IR electrochromic of PANI.¹⁷¹ Yao Li and Jiupeng Zhao *et al.*⁹⁵ demonstrated that the IR emissivity modulation of PANI film was achieved by the formation and elimination of polarons and bipolarons delocalized in the PANI chains by investigating PANI porous film in different conjugated and correspondingly dedoping states. Furthermore, there are many factors that affect the modulation of IR emissivity of PANI materials and devices,⁹⁷ such as acid dopants, thickness, and morphology of PANI film, substrate, *etc.* Yao Li's group has done a lot of work in this area.^{163,172,173} Considering the continuous energy consumption required to run the drive, flexible self-driven IR EC devices with variable optical and thermal management were reported by Qianqian Zhang, Hao Wang, and coauthors (Fig. 16e).¹⁶⁴ The as-prepared self-driven IR devices showed a reflectance contrast of 21.2%, fast switching, and a high CE of 93.6 $\text{cm}^2 \text{C}^{-1}$ at 1500 nm, which also demonstrated tunable thermal management (Fig. 16e).

5. Conclusion and Perspective

During recent decades, EC technology has made impressive advances in camouflage applications due to its unique advantages. In this review, recent advances in EC camouflage technology have been reviewed and discussed in detail in terms of color camouflage and thermal camouflage. In EC color camouflage, we systematically summarized strategies for the preparation of camouflage devices from two aspects: materials design and device design. There are three main design strategies for material design: introduction of structural colors, design of molecular structures, and preparation of composite materials. Corresponding camouflage material structures were comprehensively introduced and the possible device design strategies for EC camouflage were systemically discussed. In addition, the design ideas and advantages of these strategies were also discussed in detail, which can provide essential guidance for further design of high-performance camouflage materials and devices in the future. For EC thermal camouflage, we briefly illustrated the classification of thermal camouflage materials and summarized the applications of these IR-modulated materials in camouflage.

Due to the unremitting efforts of outstanding scientists, EC camouflage technology has achieved remarkable progress in recent years. Nonetheless, there are still some critical issues that must be considered and addressed before the real-world application of EC camouflage becomes feasible. First, for color camouflage, the color switch of the EC device needs to match the camouflage color and ensure as many color switches as possible. The currently reported EC materials and devices have been validated for color camouflage applications only by image comparison, which is far from sufficient. To improve the problem, the desired camouflage effect was demonstrated by

comparison with standard camouflage colors ($L^*a^*b^*$ values and reflectance spectra). Meanwhile, some critical EC properties for practical camouflage applications, such as cycle stability, need to be further enhanced. Furthermore, the weatherabilities are also critical parameters. However, no relevant research has been reported in the field of EC camouflage. Second, the fabrication process of high-performance wearable fabric devices is scarce. Currently, there are two main strategies for preparing EC wearable fabric devices, one is to use fabric as a substrate, and the other is to prepare fiber-shaped EC devices and then further prepare them with textile technology. The EC devices using fabric as a substrate demonstrate better flexibility than ordinary EC devices; however, their poor breathability, adhesion, and implantability limit their application in camouflage clothing and fabrics. These problems could be solved by the strategy of preparing fiber-shaped devices; however, the nonuniform electric field distribution and long electron transfer pathway lead to very slow color switching of the devices. Lastly, multi-band camouflage remains an unmet key goal, as EC devices with color camouflage while maintaining high IR emissivity modulation have rarely been realized. However, with the continuous progress and development of reconnaissance technology, the development of multi-band compatible EC camouflage materials and devices is an inevitable trend. On the one hand, more attention should be paid to the exploitation and optimization of suitable EC materials with multi-band modulation capabilities, such as polyaniline (PANI). On the other hand, combining materials with different modulation ranges or designing multilayer device structures is also a viable approach when the performance of a single material cannot meet the requirements. We anticipate that the abovementioned challenges and considerations proposed in this review will provide valuable guidance for constructing high-performance EC camouflage devices for real-world applications.

Author contributions

Haichang Fu: investigation and writing the original draft. Ling Zhang: revising the manuscript. Yujie Dong: revising the manuscript. Cheng Zhang: supervision and visualization. Weijun Li: editing and revising the manuscript, conceiving the ideas, and supervision. All authors discussed the results and commented on the manuscript.

Conflicts of interest

There are no conflicts to declare.

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