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Photo induced force microscopy: chemical spectroscopy beyond the diffraction limit

 Josh A. Davies-Jones * and Philip R. Davies *

Over the last decade remarkable advances have been made in creating spectroscopic tools to interrogate surface properties using electromagnetic radiation in the near field, achieving lateral resolutions in the nanometre range. We review recent work involving one of the most promising of these spectroscopic tools, photo induced force microscopy (PiFM), which uses mechanical detection of dipole–dipole interactions enabling nanometre resolved mapping of surface properties ranging from vibrational modes of adsorbates to plasmon resonances. We discuss the origin of contrast in PiFM images, its applications as a local probe of chemical species and for mapping local electric fields in areas as diverse as zeolite films and biometric recognition. In comparison with related techniques such as photothermal atomic force microscopy-based infrared (AFM-IR) and tip enhanced Raman spectroscopy (TERS) PiFM has many advantages but perhaps its most successful application is in combination with other surface characterisation methods which provide information averaged over much greater surface areas. We discuss the most recent developments of PiFM technology which are leading to higher resolution images and are widening the range of environments in which the technique can be applied, and we consider how the field is likely to progress in the future.

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1. Introduction

Our understanding of surface chemistry has advanced at a steady pace since the experiments by Taylor and Langmuir using clean tungsten filaments in the early part of the 20th century.^{1,2} Low energy electron diffraction^{3,4} and Auger spectroscopy⁵ were developed in the 1950's & 60's to provide information on the surface structure and elemental composition whilst flash desorption experiments were developed into the sophisticated temperature programmed desorption (TPD) technique that provides fundamental information on surface kinetics.^{6,7} Photoelectron spectroscopy offered the next step forward with quantifiable identification of the elemental composition of surfaces^{8,9} but it was always recognised that elemental composition, albeit with information on the charge and hence environment of the elements concerned could not provide a complete picture of the surface chemistry and vibrational spectroscopy was developed for surfaces with techniques such as DRIFTS and ATR for non-uniform surfaces under ambient conditions and RAIRS and HREELS for single crystals in UHV. A plethora of sophisticated techniques were developed between the 70's and 90's but Binnig and Rohrer's invention of scanning tunneling microscopy in 1982¹⁰ created a step change in the understanding of surfaces by allowing ready access to topography at the Ångström scale with

few restrictions on conditions. A whole range of related proximal probe techniques have been derived since 1982 but researchers have, from the very beginning, recognised that direct chemical information on the surfaces being imaged is essential for a full interpretation of the topography. Combining local probe techniques with tools providing more direct chemical information such as X-ray photoelectron spectroscopy^{11,12} and TPD proved very successful¹³ but in many areas of research, knowing the chemistry on a local (*i.e.* nanometre scale) is critical. Ho's dramatic images of the vibrational spectrum of ethyne on a Cu(100) surface¹⁴ derived from scanning tunneling spectroscopy (STS) illustrates the power and need for such information, but that STS approach was experimentally very demanding and only feasible for single crystal surfaces in ultra-high vacuum.

A new era is now dawning for studying the chemistry at the nanometre scale with the development of a range of methods for vibrational spectroscopy that exploit near-field behaviour at nanometre scales and thus sidestep the Abbe diffraction limit. These offer the possibility of identifying the functional groups involved at the sort of local scales that phenomenon such as nanoparticle growth, microbial adhesion and cell wall changes occur. Early developments in this area included Raman spectroscopy where the massive enhancement in the signal provided by rough silver surfaces^{15,16} led eventually to the idea of using a plasmonic metal coated AFM tip to create a local field enhancement and potentially single molecule spectroscopy. Tip-enhanced Raman spectroscopy has led to several important papers and a

Cardiff Catalysis Institute, School of Chemistry, Cardiff University, Cardiff, CF10 3AT, UK. E-mail: DaviesJA21@cardiff.ac.uk, daviespr@cardiff.ac.uk



somewhat related technique of using coated nanoparticles to create local enhancements “SHINERS” has also been very interesting¹⁷ but both techniques have proved difficult to deploy; spectra appear to be highly dependent on the nature of the local probe and the stability and reproducibility of suitable probes are problematic.¹⁸ An alternative approach, using illumination of a sample through a scanning probe tip, thereby creating a highly confined area of illumination, and avoiding the diffraction limit, “scanning near field optical microscopy, SNOM or NSOM” has also been adopted successfully by many groups but again practical applications of this appealing technique have proved difficult.¹⁹

Much more recently it has been shown^{20,21} that electron energy loss spectroscopy (EELS) in transmission electron microscopy, can be obtained from impact rather than dipole scattered electrons, thereby enabling a lateral resolution that in ideal circumstances approaches atomic dimensions. This approach is undoubtedly a major step forward in our capability to fully characterise the chemistry of an interface at the best resolution one could hope for. The experiment requires the exacting conditions of the TEM experiment and it remains to be seen whether the full mid-IR vibrational spectrum of functional groups can be characterised and how well the spectra compare to the library of existing vibrational spectroscopic data but this is an exciting development that highlights the importance of highly resolved surface vibrational spectroscopy.

In this space, new approaches to obtaining infrared spectroscopy in the near field have been developed in the last decade. Two techniques, in particular, are becoming widely used. Photo-thermal infrared spectroscopy (AFM-IR)^{22–24} was the first to be developed and involves heating of the sample with a pulsed laser, the resulting rapid thermal expansion excites resonant oscillations in an AFM cantilever, the frequencies and amplitudes of the cantilever can then be related to the excitation wavelength and the local absorption. The spectra obtained from AFM-IR rely on a very different excitation mechanism to that in far field infrared absorption spectroscopy, and as with the near field Raman techniques, that has the consequence that the presence and intensity of particular vibrational modes can be difficult to relate between the two. In contrast to AFM-IR, a closely related technique that has been developed over the last decade, photo induced force microscopy (PiFM), delivers spectra that compare very closely to those from far field infrared spectroscopy (although subtle differences may arise).²⁵ PiFM therefore offers an exciting new window into local surface chemistry with all the advantages of atomic force microscopy such as rapid sample preparation, broad applicability to analysis of samples of every type and operation under ambient conditions. Since its advent in 2010²⁶ PiFM has demonstrated its applicability in many different areas but its particular strengths in characterising the local chemistry of the surface have started to be exploited in the last two years and a review of its current and likely future impact is therefore timely. In this discussion we will consider how the approach has already led to an improved understanding of interface chemistry in a wide range of research areas and try to give a perspective on how it will guide the field in the next decade.

2. The origin of spectroscopic contrast in PiFM

The concept around which the PiFM technique was originally developed is mechanical detection of the attractive force between a surface dipole and its image dipole in a scanning probe tip. Whereas AFM-IR relies on short-range repulsive forces through direct contact, PiFM operates in non-contact mode giving a near background free measurement of attractive van der Waals forces. Initial models of the PiFM contrast mechanism were based on an induced force on the tip due to the interaction between the oscillating surface dipole and the induced dipole in the tip.^{27,28} The resulting gradient force, F_g , depends upon the polarisation of the particle in the z direction and because of its $\frac{1}{Z^4}$ dependence, is highly localised that can be measured with very high spatial resolution.^{29–31}

Subsequently, experimental evidence accumulated that the cantilever response was dominated by dissipative effects rather than the dispersive response predicted by the dipole-dipole force model,^{33,34} thus indicating the involvement of a thermal process in generating PiFM contrast. Jahng *et al.*'s model³³ considered incident light causing a thermal expansion at the tip/sample junction and generating a force interaction through two processes: a “direct” contribution resulting from far-field transmission and absorption by the sample and an “indirect” contribution caused by the enhanced near field, generated by the antenna effect of the tip. The model indicated that the thermally stimulated forces would be as much as a hundred times stronger than the induced dipole force for most organic materials for mid infrared radiation and recent experimental data, utilising an innovative frequency domain force detection method that allows sub- μ s time measurements,³⁴ suggests the same is true in the visible/near infrared region where energy transfer is through electronic excitations.

More recently, Wickramasinghe *et al.*³² have published a detailed experimental study of the contrast mechanism which provides a definitive treatment of the problem and concludes that the PiFM signal is due to opto-mechanical losses. Their discussion starts by ruling out the involvement of thermal expansion; non-optical van der Waals forces are automatically excluded in PiFM but the team also rule out thermal modulation of *optical* van der Waals forces as a significant contributor to PiFM by measuring the sensitivity of the PiFM signal to thermal expansion using a mirrored lead zirconium titanate (PZT) crystal to generate similar van der Waals forces. A minimum expansion of 32 pm was determined to be necessary for detection by PiFM, slightly larger than the calculated expansion of the 60 nm polystyrene film studied in the paper and at least ten times greater than the < 3 pm expansion expected of a monolayer of self-assembled 4-methylbenzenethiol on gold surfaces which also gives a strong PiFM signal, Fig. 1. A separate experiment studying 60 nm poly(methyl methacrylate) (PMMA) on glass was used to demonstrate that the side bands monitored in PiFM (see below) are generated through amplitude rather than frequency modulation of the cantilever resonance, the excited molecule



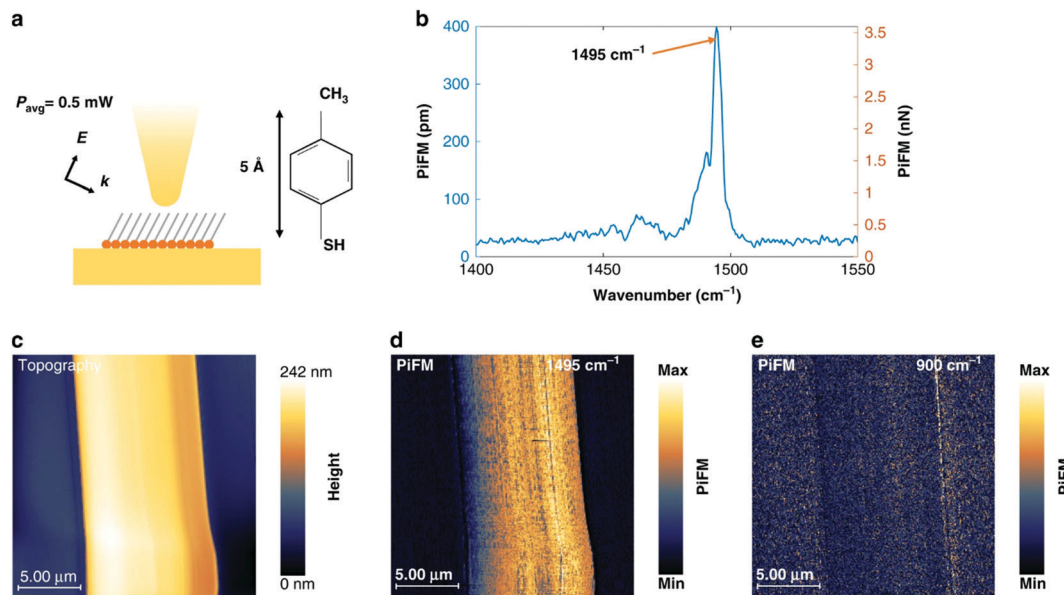


Fig. 1 Demonstration of monolayer response by PiFM. (a) 4-Methylbenzenethiol (4-MBT) adsorbed on template-stripped gold. Excitation average power is 0.5 mW focused to 20 μm -diameter spot. (b) Point spectrum of 4-MBT showing resonance at 1495 cm^{-1} . (c) Topography of gold island. (d) PiFM image at 1495 cm^{-1} . (e) PiFM image off resonance. (Reproduced from Almajhadi *et al.*³²).

changing the effective damping constant of the cantilever. This is a dissipative effect (*i.e.* mechanical energy is lost to heat) rather than a dispersive one consistent with all the experimental measurements. The team propose a model involving the “optically mediated damping” of the viscoelastic or adhesion interactions between tip and sample. One way this could happen is that the surface molecules align with the electric field of the tip and then relax back to their equilibrium position as the tip withdraws creating an oscillation with a different phase shift to the phase of the tip.

The topography of the sample and the photoinduced force gradient can be extracted simultaneously from different orthogonal mechanical resonances of the AFM cantilever. A simple implementation, “Direct mode”, has the cantilever driven at the first resonance ν_1 , and the repetition rate of the exciting laser tuned to the second resonance ν_2 , topographical and photoinduced force signals are extracted simultaneously at the two frequencies, Fig. 2. A disadvantage of this approach is that the photoinduced force is measured which includes both the scattering and the gradient force contributions, so the required information is superimposed on a significant background signal. A more sophisticated approach makes use of the ‘optical heterodyne’ method in which the second of the two driving forces is tuned to a “beat” frequency, which in the simplest case corresponds to the difference or sum of the two mechanical resonances, *i.e.* $\nu_1 - \nu_2$ (or $\nu_1 + \nu_2$).^{35,36} In this “side band” implementation, an almost background free signal is obtained from the highly localised photoinduced force *gradient* at ν_2 , because the scattering force, which varies on the micrometre scale is effectively removed. Several exciting developments of the PiFM measurement approach have been made in recent years and a more complete discussion of the force heterodyne approach is given later in Section 6.1.

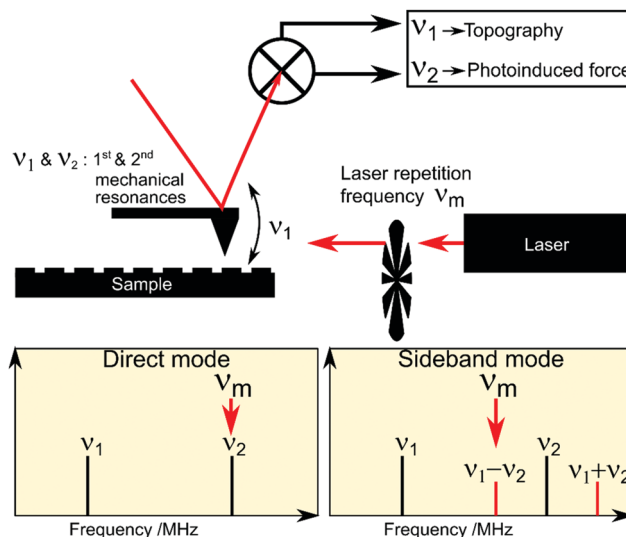


Fig. 2 Simultaneous acquisition of the surface topography and the photoinduced force from a cantilever & pulsed laser system. The topography is measured from the amplitude of the oscillation of the cantilever at its first resonating frequency, ν_1 ; the photoinduced force is measured at another mechanical resonance of the cantilever (ν_2). In the “direct” mode, the laser repetition rate (ν_m) is set to ν_2 . However, an almost background free signal can be obtained if ν_m is set to the difference between of the two mechanical resonances, *i.e.* $\nu_m = \nu_1 - \nu_2$.

2.1 Polymeric systems

PiFM’s ability to map the molecular composition of a surface is illustrated particularly well in studies of polymeric materials; typically, the PiFM images are used to show the distribution of different functional groups at the surface and with polymers,



this means that domains are easily identified. A good example is Chen *et al.*'s recent study³⁷ of the nanoscale-phase morphology of a blend of the polymer PBDB-T-2F (PM6)³⁸ with varying quantities of a small molecule, non-fullerene, acceptor (IT-4F). Non-fullerene acceptors have tuneable absorptions extending further into the near-infrared than fullerene acceptors, increasing the light absorption. Non fullerene based organic solar cells also give efficient charge transfer at much lower (virtually zero) energy offsets than fullerene-based devices and thus achieve higher open-circuit voltages. Central to the performance of the organic solar cell is the photoinduced charge transfer at the interface between the two different components and this will be dependent on several factors including domain size and phase segregation of the two components but our understanding of the relationship between charge transfer and the macroscopic non-fullerene acceptor phase properties is still lacking. PiFM's unique ability to map the chemistry across a surface provide an insight into those properties in the PBDB-T-2F/IT-4F system, Fig. 3.³⁸ Thin films of the polymer mix with differing percentages of IT-4F added to the PBDB-T-2F were examined,

Whilst the AFM images show the root-mean-square surface roughness of the samples increasing from 1.19 nm at 5% IT-4F to 1.85 nm at 50% and then decreasing to 1.21 nm at 90%, the PiFM images show the steady increase in IT-4F domain size with IT-4F concentration, Fig. 3. Chen *et al.* quantify the IT-4F domain size and show that it corresponds to the relative extent of diffusion-mediated hole transfer.³⁸

The formation of metal-organic-framework thin films is another system that demonstrates the potential of PiFM in

nanoscale chemical mapping. Weckhuysen *et al.*^{39–41} studied the layer by layer deposition of $[\text{Cu}_3(\text{btc})_2(\text{H}_2\text{O})_n]$ a surface-anchored metal-organic framework (SURMOF) known as "HKUST-1" from copper acetate ($\text{Cu}(\text{CH}_3\text{COO})_2$, CuAc_2) and 1,3,5-benzene tri-carboxylate (btc) precursors. The earliest of the three papers⁴¹ is particularly interesting with data from PiFM, photo-thermal nano-infrared spectroscopy (AFM-IR) and far field infrared reflection absorption spectroscopy (IRRAS)

There is excellent agreement in the PiFM and IRRAS spectra throughout the paper both in terms of peak positions and signal intensities but the different spectroscopies used each contribute distinctly different pieces of information to the study from their different capabilities. A broad peak at 1570 cm^{-1} in the IRRAS data from partially and fully dipped surfaces was attributed to the COO^- asymmetric stretching vibration of a Cu_2Ac_4 precursor (shown in Fig. 4c for comparison). The authors propose that these physisorbed species play a critical role in the development of the HKUST-1 SURMOF and are responsible for the rate of film growth being significantly greater than that expected from a simple layer by layer mechanism. AFM-IR spectra at individual points on the growing films, Fig. 4, reinforces this conclusion showing the presence of defect-rich SURMOF islands on both the Au/Si and Au/mica surfaces, Fig. 4c, #1 & #2, whereas areas of the Au/mica surface without SURMOF islands are covered with an amorphous layer of Cu_2Ac_4 , Fig. 4c, #3. Such localised information is only available with PiFM.

PiFM maps of the developing surfaces show the tight correlation between the AFM topography and the chemical

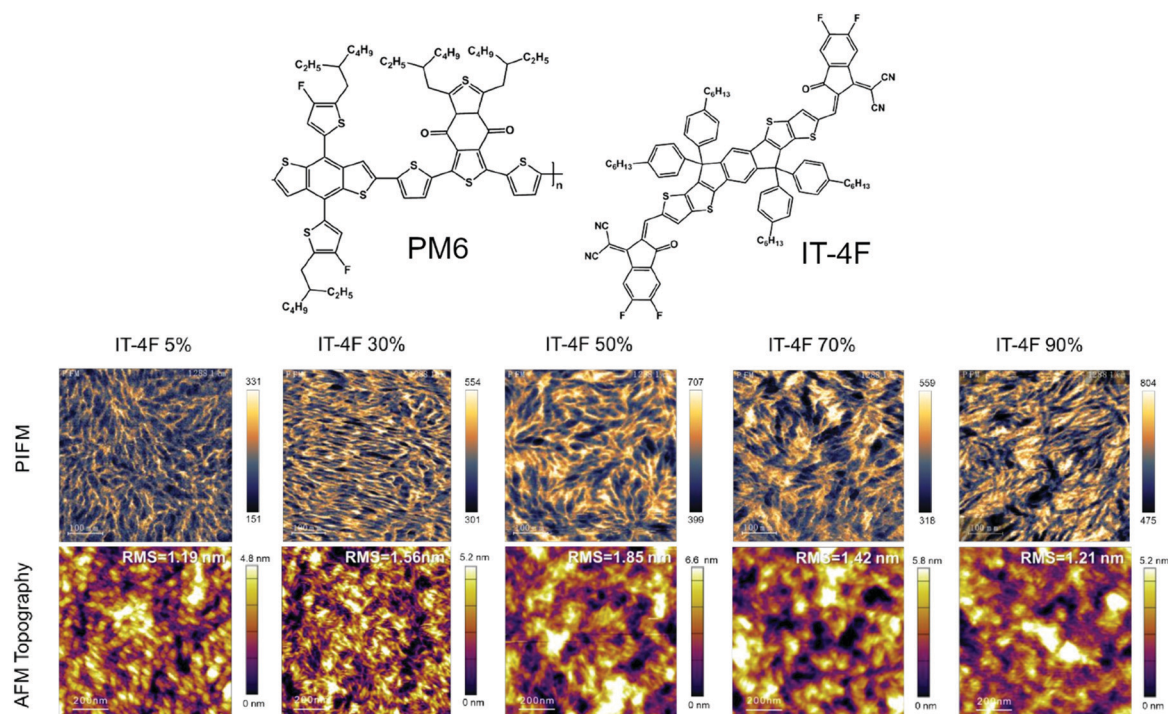


Fig. 3 Comparison of PiFM imaging of the distribution of IT-4F in PBDB-T-2F (PM6) using its IR absorption at 1288 cm^{-1} with AFM imaging of the topography, (adapted from Fig. S4 of Mandemaker *et al.*³⁸).



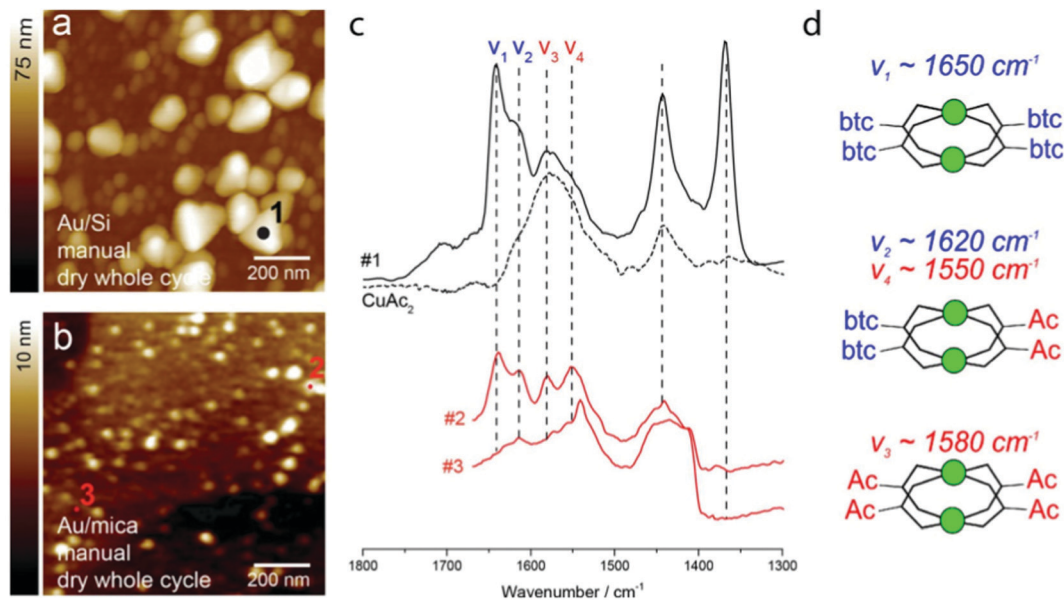


Fig. 4 AFM images and photo-thermal nanoinfrared (AFM-IR) spectra recorded on Au/Si and Au/mica HKUST-1 SURMOF thin films. (a) AFM image of a thin film on Au/Si measured after 5 deposition cycles using manual synthesis with drying each whole cycle. (b) AFM image of a thin film on Au/mica measured after 2 deposition cycles grown in the same way as (a). (c) Representative nano-IR spectra recorded at different positions for the substrates presented in (a), (b). Spectrum (#1) is taken on top of a large SURMOF grain (1 black), spectra #2 and #3 are taken at two different positions. An IRRAS spectrum of CuAc₂ (dotted line) is added for comparison. (d) Vibrational modes assigned to COO@ originating from the different Cu-paddle-wheel substitutions. Note that models are simplified for clarity by omitting C and O atoms coordinated to copper; copper atoms are depicted in green. (Reprinted from Delen *et al.*⁴¹).

composition predicted by the IRRAS and AFM-IR and the characteristic presence of btc on the high features and their absence in the lower areas, Fig. 5. In the two subsequent papers the team study the effect of temperature on the rate of SURMOF

formation³⁹ and the potential of spin coating as a deposition method for the films.⁴⁰ In both cases PiFM adds valuable information through mapping molecular composition with sub-micron lateral resolution.

A closely related paper⁴² by the same group applied PiFM to the study of ZSM-5 thin-films, prepared with a range of Al:Si ratios. The paper beautifully illustrates the power of PiFM's locally defined spectroscopy with a study of the asymmetric framework vibration of zeolites with different Si/Al ratios normally expected at 1107 cm⁻¹. The greater length of the Al-O over that of the Si-O (1.75 Å and 1.61 Å respectively), together with the lower mass of the Al atom leads to a split in this peak when Al is introduced into the lattice giving a pair of peaks at ~1109 cm⁻¹ and ~1082 cm⁻¹. The PiFM spectra, Fig. 6a-c, clearly show the two components but also the variation in their relative intensities at different positions on the zeolite film, perhaps reflecting a difference in the local concentration of aluminium. A further element to this study is very informative about the capabilities of the PiFM technique. The aluminium containing zeolite was used to catalyse the methanol-to-hydrocarbons (MTH) reaction at 623 K for 4 hours and the sample subsequently studied with both PiFM and confocal fluorescence microscopy. Whereas the latter showed a honeycomb-like fluorescent structure from the deposited coke species, PiFM showed a homogeneous distribution of hydrocarbons on the surface. The difference between the two was attributed to the greater surface sensitivity of PiFM (~30 nm).

Another example of PiFM's ability to map different polymer domains is a study of urethane-acrylic hybrid dispersions by Chen *et al.*⁴³ Urethane-acrylic hybrids are created by adding acrylic monomers during the dispersion process of the urethane

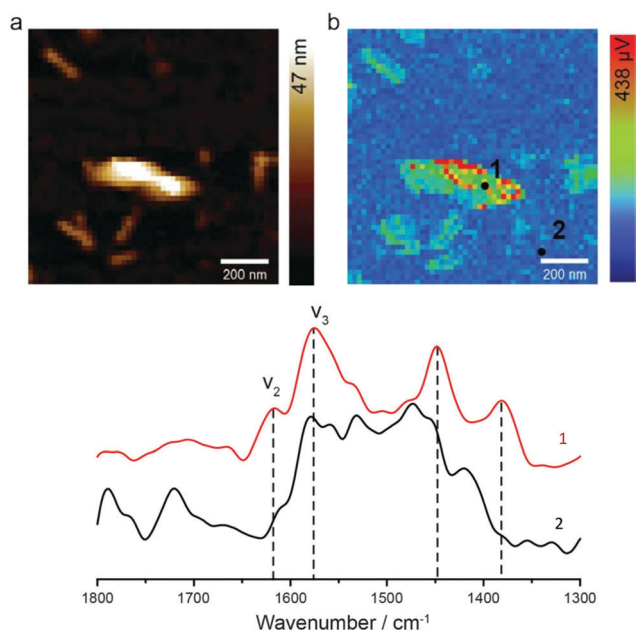


Fig. 5 PiFM maps at 1450 cm⁻¹ corresponding to the btc benzene breathing vibration, of a 1 layer HKUST-1 SURMOF thin film prepared using manual synthesis, including a drying step each half cycle. (b) Point IR spectra taken at positions 1 (red) and 2 (black), on and off the top of a high feature. (Reprinted from Delen *et al.*⁴¹).



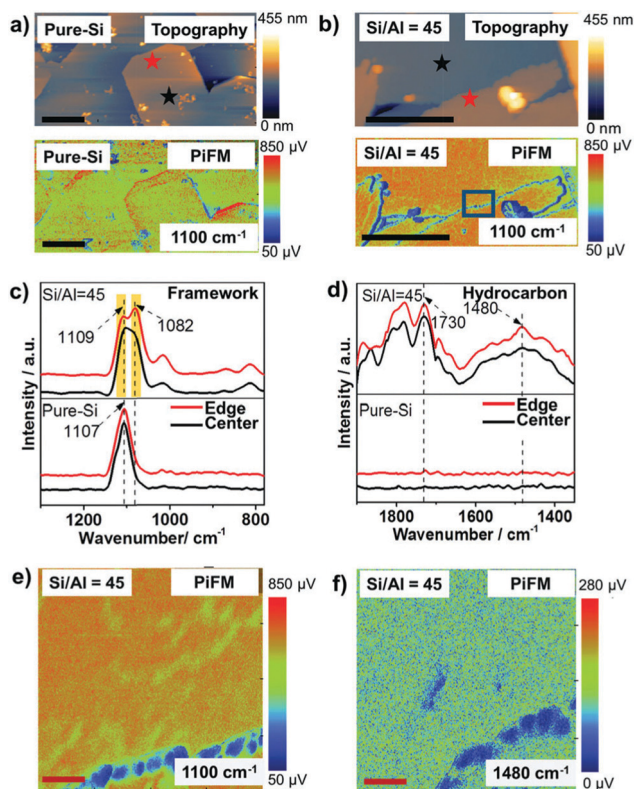


Fig. 6 (a) and (b) AFM topography (top) and photoinduced force microscopy (PiFM, bottom) image at 1100 cm^{-1} of zeolite ZSM-5 thin-films grown in a secondary growth media with ethanol as the structure directing agent and Si/Al = (a) N and (b) 45. (c) and (d) PiFM point spectra show (c) zeolite framework vibration and (d) hydrocarbon vibration distributions at different locations indicated by the stars with corresponding colours in (a) and (b) in the zeolite ZSM-5 thin-films after a 4 h methanol-to-hydrocarbons (MTH) reaction at 623 K. (e) High resolution PiFM image of the framework vibration at 1100 cm^{-1} of the zeolite thin films from the region highlighted in (b). (f) Product/coke distribution in the same area measured with PiFM at 1480 cm^{-1} representing the CQC stretch of hydrocarbons. The black and red scale bars represent 1 mm and 55 nm, respectively. Images (e) and (f) are shifted with respect to each other due to thermal drift between the tip and the sample. (Reproduced from Fu *et al.*⁴²)

prepolymer and the resulting materials show significant mechanical advantages over polymer blends. Using the characteristic acrylic and urethane absorption bands at 1730 cm^{-1} and 1705 cm^{-1} respectively, Chen *et al.* used PiFM to map the distribution of the two components and establish acrylic domain sizes of $\sim 30\text{--}40\text{ nm}$ supporting the proposal that the enhanced physical behaviour of the hybrids is due to smaller domain sizes than is found in blends. Have *et al.*⁴⁴ applied PiFM to a slightly different problem, comparing it with two commonly used light scattering techniques: nanoparticle tracking analysis and multi-angle dynamic light scattering, for the determination of nanoplastic particle size distributions and concentrations in seawater. The main drawback of PiFM's relatively slow throughput was offset by its ability to distinguish easily between nano-plastics and other nano-particles that interfere with the light scattering technique results.

3. Biological surfaces

Infrared spectroscopy has a long history of use in biological systems,^{45,46} both as an analytic tool to identify molecules present in samples but also as a means of gaining information on the structure of the molecules through characteristic changes in vibrational frequency and intensity. Such information is potentially even more powerful if it can be used to build molecularly specific maps with lateral resolutions on the order of a few nanometres. Whilst PiFM is at somewhat of a disadvantage, compared to Raman for example, in biological systems because of the difficulties caused by the presence of water damping the infrared signals, nevertheless several researchers are already exploring this field. For example, Ping *et al.*⁴⁷ have used PiFM to image genetically modified rod shaped *E. Coli* bacteria used as bio templates to produce rod-shaped silica particles. R5 proteins on the surfaces of typical *E. coli* were modified to form silafin-derived 5R5 which acts as a nucleation point for silicification at the surface of the bacterium. PiFM was used to map the distribution of 5R5 at the surface through the distinctive peak at 1200 cm^{-1} , attributed to the C–N stretching vibration of primary amines in the R5 peptide and not found in spectra of the bare *E. Coli* surfaces. The chemically sensitive mapping showed that the areas with highest protein concentrations were most prevalent along the edges of the bacteria. This correlated very well with the PiFM mapping of the Si–O–Si vibration of nucleated silica at 1100 cm^{-1} after the surface had been exposed to a silica precursor.

Two distinctive vibrations arising from the amides present in many important organic molecules such as proteins, RNA and DNA, are frequently used to characterise biological materials. These are the “amide I” band, which occurs between $\sim 1600\text{ cm}^{-1}$ and 1700 cm^{-1} and arises principally from the C=O stretch of the group, and the “amide II” band which lies between $\sim 1500\text{ cm}^{-1}$ and 1600 cm^{-1} and is due to a combination of the N–H bending and C–N stretching vibrations. In 2019, Marsh *et al.* took advantage of these distinctive bands to image artificially constructed icosahedral protein cages⁴⁸ with PiFM confirming their dimensions and shape. Russo *et al.* used the same distinctive amide bands to map the concentration of PSLG polypeptide on the surface of capped silica nanoparticles for use in a liquid crystal system, Fig. 7.⁴⁹

Ji *et al.*⁵⁰ also used the characteristic amide I and II bands in α -helical and random coiled DNA to establish PiFM as a potential analytical technique for the detection of three different biotoxins: brintoxin (ABR), ricin toxin (RT) and Clostridium perfringensexotoxin (ETX), Fig. 8. In all cases, PiFM combined with principal component analysis (PCA) was able to accurately detect the presence of the biotoxins deposited on mica surfaces. Intriguingly, Ji's work suggests that when bolstered by PCA, PiFM is capable of distinguishing single molecule protein particles. Its noticeable in Ji's data that the PiFM spectra show much narrower line widths than those of the far-field FTIR. This may be due to a number of different factors including nanoscale heterogeneities but may also be related to differing selection rules and is an aspect which should be investigated in more detail.



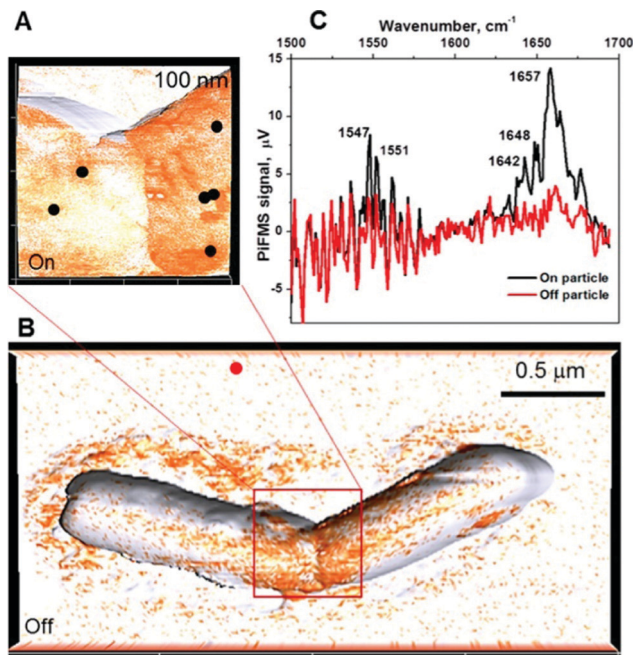


Fig. 7 Photoinduced force microscopy (PiFM) typical results: (A and B) PiFM images overlaid on 3D topography images show distribution of the PSLG polypeptide (the orange regions identify where the PSLG polypeptide is located). The black points represent the locations on the PSLG-HemiSil surface where PiFM spectra were collected, while the red point indicates the off-particle location used to collect a PiFM spectrum. (C) PiFM signal versus wavenumber for the locations indicated by the black and red dots, showing the amide I and amide II signals characteristic of the PSLG polypeptide. Exposure time was 100 ms per wavenumber. (Reprinted with permission from *Langmuir*, 2016, **32**, 13137–13148 copyright 2016 American Chemical Society.⁴⁹)

In a similar vein to Ji *et al.*, Abrego-Martinez *et al.*⁵¹ have used the characteristic amide vibrational bands of proteins in PiFM spectra to confirm the binding of the SARS-CoV-2 spike protein (S-protein) to aptamer functionalised gold surfaces as part of a proposed detection method of SARS-CoV-2 using electrochemical impedance spectroscopy to provide a fast and simple alternative to the typical PCR test. The aptamers target the receptor-binding domain in the S-proteins of SARS-CoV-2.

PiFM showed a decrease from 16.6 to 12.3 nm in surface roughness when bare gold nanoparticles functionalised with aptamers which the authors attribute to the formation of a monolayer, Fig. 9.⁵¹ When this surface was subsequently exposed to S-proteins the roughness increases to 25 nm caused by the deposition of scattered features with 15 nm height which is consistent with S-protein height detailed in previous literature.^{52,53} The PiFM spectra showed a significant increase in intensity in the 1400 cm^{-1} region attributed to the amide III vibration modes from the peptides in the S-protein confirming its presence upon the surface. There were also changes in the chemistry of the ssDNA aptamer before and after exposure to the S-proteins with the PiFM showing a shift in the amide I band from 1680 cm^{-1} to 1730 cm^{-1} due to the binding with the S-proteins. The PiFM spectra show other changes too which it would be interesting to explore, the appearance of a $\sim 1280 \text{ cm}^{-1}$ peak for example which Kitane *et al.* attributed to the PO_4^{2-} asymmetric stretching vibration of the phosphate backbone.^{54,55}

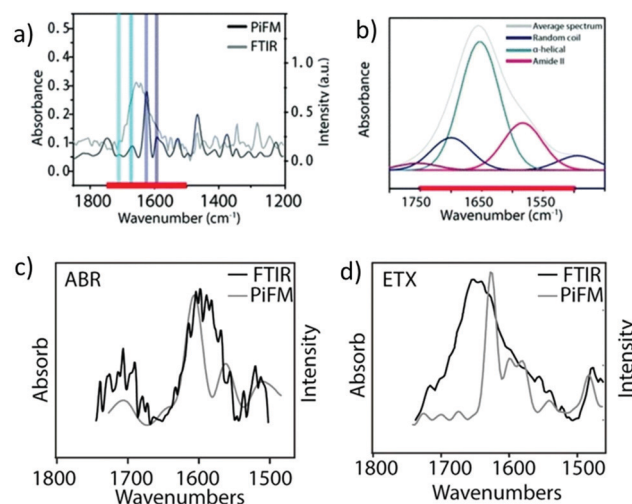


Fig. 8 Comparison between PiFM and FTIR of three biotoxins: brintoxin (ABR), ricin toxin (RT) and *Clostridium perfringens* exotoxin (ETX). (a) RT; (b) Fitting of FTIR of RT; (c) Comparison of FTIR and PiFM between 1500–1750 cm^{-1} of (c) ABR and (d) ETX. Reproduced from Ji *et al.*⁵⁰

In a slightly different area of research, Kesari *et al.*'s study of ultrathin microtome sections of Norway Spruce demonstrates the potential of PiFM for visualising the chemical composition of cell walls.⁵⁶ Using vibrational bands at 1505, 1471, 1220 and 1145 cm^{-1} , the distribution of lignin and cellulose was mapped across cell walls, Fig. 10. High lignin concentrations were found in the compound middle lamella (CML), and cell corner middle lamella (CCML) areas of the cell wall.

Much smaller features, with dimensions of between 5–20 nm were imaged by decreasing the PiFM step size from 3.9 to 1.6 and 0.6 nm. These smaller areas were uncharacteristically low in lignin for sublayers 1 and 2 but showed a strong absorption at 1051 cm^{-1} which is characteristic of cellulose. The features were therefore identified as elementary fibrils (EF's) which are dispersed throughout the cell wall. Measurements of the smallest possible cellulose elementary fibrils were found to be approximately 5 nm which agreed with literature values^{58,59} of $3.2 \times 5.3 \text{ nm}$ cross-sections, with aggregations of molecules reaching 20+ nm. Kesari *et al.*'s findings agree with those of Gusenbauer *et al.*⁶⁰ who concluded similar sized EF aggregates were present based on AFM-IR data, however single EF molecules were more easily identified with the 2–4 times better resolution of PiFM over AFM-IR.

A very different use of PiFM was demonstrated by Zhao *et al.* studying Eu^{3+} nanostructures as a potential system for biometric recognition.⁶¹ The Eu^{3+} /oleic acid complexes were placed under a protein skin and their chemical composition and morphology studied with PiFM using the amide II band at 1540 cm^{-1} , and the OEu^{3+} Phen complex carboxyl band at 1663 cm^{-1} , Fig. 11.

4. Inorganic surfaces

PiFM is well suited to the study of inorganic surfaces and its potential uses in the geological sciences has been outlined by





Fig. 9 Three-dimensional topography and PiFM intensity at $\sim 1400\text{ cm}^{-1}$ representations of the AuNPs-, Aptamer/AuNPs- and S-protein/Aptamer/AuNPs-modified SPCEs. According to the PiFM spectra and targeting selected peak shifts for best chemical contrast, the common wavenumbers at (~ 1050 , $\sim 1390\text{ cm}^{-1}$ and $1680\text{--}1730\text{ cm}^{-1}$) were selected for PiFM mapping. Reprinted J. C. Abrego-Martinez, M. Jafari, S. Chergui, C. Pavel, D. Che and M. Sij, "Aptamer-based electrochemical biosensor for rapid detection of SARS-CoV-2: nanoscale electrode-aptamer-SARS-CoV-2 imaging by photo-induced force microscopy", *Biosensors and Bioelectronics*, 2022, **195**, 113595, Copyright (2022), with permission from Elsevier.⁵¹

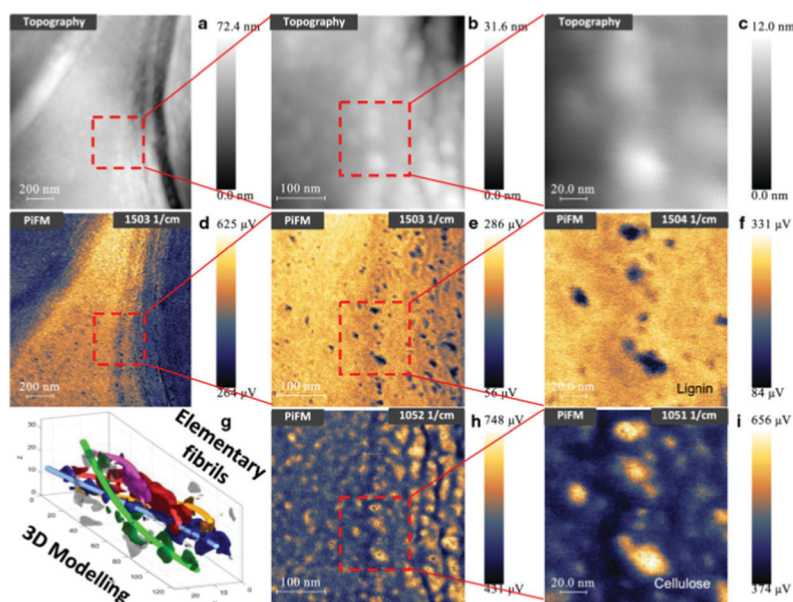


Fig. 10 Topography (a–c) and PiFM images of an ultrathin Norway spruce cross-section at ca. 1503 (d–f) and 1052 (h and i) cm^{-1} at three different magnifications (dimensions shown by the scalebars). A TEM tomography image of a helical cellulose elementary fibrils (EF) bundle with its dimensions (nm) is also shown (g) (Reza *et al.*⁵⁷). The vertical cellulose-rich patterns in the PiFM images at ca. 1050 cm^{-1} may represent tangentially organized EF bundles that are typical of the S_1 layer of Norway spruce tracheids. Reproduced from Kesari *et al.*⁵⁶

Otter *et al.*⁶² More recently Patabendigedara *et al.* used PiFM to determine the water content of different nominally anhydrous materials.⁶³ This is one of the first papers to mainly focus on the PiFM spectra rather than the topographical mapping features and shows a good comparison between FTIR and PiFM. Measurements of a polycrystalline forsterite sample were taken across grain boundaries, Fig. 12. The authors separated the high wavenumber areas of these PiFM spectra ($3000\text{--}3700\text{ cm}^{-1}$) into three distinct regions corresponding to: silicon [Si], titanium [Ti], trivalent [tiv] and magnesium [Mg] bands.

The area under each of these regions was assumed to be directly proportional to the hydroxyl content and used to calculate the water content at each point. In this way the authors were able to map and quantify the water content across the surface of

the sample. The authors concluded that the majority of water content is located in the centre of crystal grains rather than at the grain boundary as shown in the U shaped graphs in Fig. 12. While the PiFM spectra and FTIR spectra are in good agreement, there is more detail in the PiFM than the FTIR. This is similar to the results seen by Ji *et al.*⁵⁰ The authors suggest that this could be a result of several factors which warrant further investigation, but a major factor is likely to be the small area over which PiFM measurements are made in comparison to far-field FTIR, ($<10 \times 10\text{ nm}^2$ compared to $60 \times 60\text{ }\mu\text{m}^2$ spot). This means that FTIR spectra can miss additional details corresponding to nanoscale heterogeneities and the quantification work undertaken by Patabendigedara *et al.* using the PiFM data would not be possible with a traditional FTIR system.



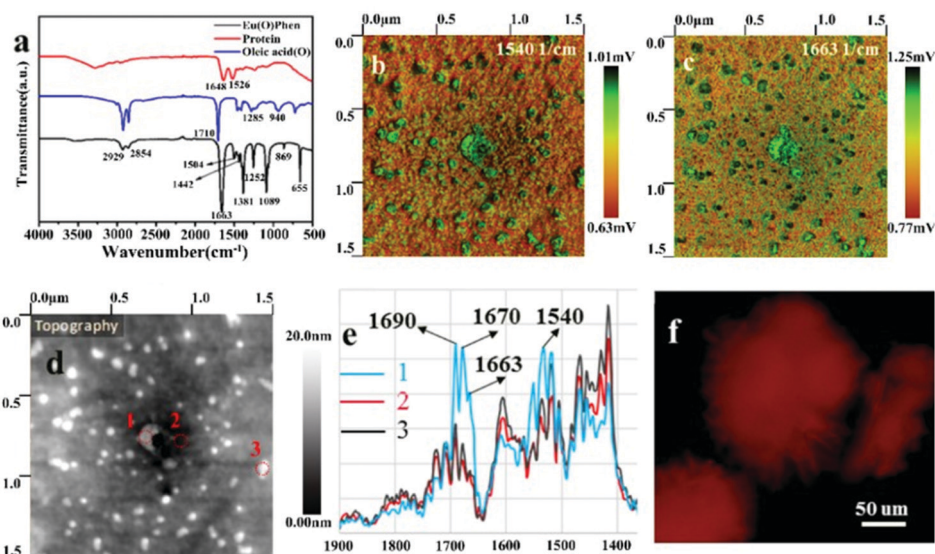


Fig. 11 The concentration of oleic acid used during nano-label synthesis was 0.05 mol L^{-1} in FTIR spectra and PiFM images. (a) FTIR spectra of protein, oleic acid, and $\text{OEu}^{3+}\text{Phen}$ complexes; PiFM images taken with laser tuned to (b) 1540 cm^{-1} for protein and (c) 1663 cm^{-1} for $\text{OEu}^{3+}\text{Phen}$ complexes in the sample; (d) the detection points of the sample, overlaid on a topographical map; (e) PiFM infrared spectra for the detection points; (f) the fluorescence microscopy image of the sample. Reproduced from Zhao *et al.*⁶¹

Nan *et al.*⁶⁴ used a combination of TEM-EDX and PiFM to analyse abiotic carbonaceous matter present in geological samples taken from the Yap Trench. Igneous peridotite rocks undergoing serpentinization on the oceanic floor are thought to play a key role as energy sources for microbial ecosystems. Furthermore, these materials have high nanoscale porosity and TEM-EDX showed that the nanopores contain large amounts of carbon. However, this technique does not give any chemical information and PiFM was combined with RAMAN to characterise the chemical composition of the nanopore. The PiFM showed that condensed carbonaceous matter (CCM) was often located in the nanopores and grain boundaries of the rocks and contained C=C stretching modes corresponding to aromatic compounds which was confirmed by the RAMAN. Analysis showed that $>70\%$ of nanopores across a $2.3 \mu\text{m}^2$ area contained CCMs.

No evidence of functional groups or thermally degraded biogenic matter was detected and since prokaryotic microorganisms would be too large to colonise the 50 nm wide nanopores, the authors conclude that the CCM is abiotic in origin. Typically, the CCM regions were also surrounded by areas rich in iron oxide leading to a suggestion that the CCMs were synthesised by a Fischer-Tropsch like reaction catalysed by the Fe minerals.

Werny *et al.*⁶⁵ analysed the morphological changes in SiO_2 supported single metallocene-based ethylene polymerization catalysts. Morphological changes such as fragmentation and sectioning are key for developing catalysts with high activity due to the build-up of polymers in the active sites of the catalyst, blocking further reaction. Fragmentation and sectioning of the catalysts results in the development of new catalytic sites which circumvents mass transport limitations and maintains catalytic activity. Two structurally homogenous SiO_2 supported catalysts

were preactivated with methylaluminumoxane and paired with either Hf or Zr metal which were selected because of their very different activities with $\text{SiO}_2/\text{MAO}/\text{Zr}$ having a $40\times$ higher productivity. Slices of each catalyst were prepared by focused ion beam and analysed using SEM and PiFM. The SEM images showed contrast between the silica body and the pore with fragmentation visible in both catalysts after an hour of reaction. The $\text{SiO}_2/\text{MAO}/\text{Hf}$ showed large areas of pristine SiO_2 whilst the $\text{SiO}_2/\text{MAO}/\text{Zr}$ catalyst had visible nano-cracks throughout the material. PiFM was used to verify the SEM data with the $\nu(\text{Si-O})$ and $\delta(\text{C-H})$ regions highlighting the core and polymer phases clearly, Fig. 13 While no pure polymeric regions could be located, average PiFM spectra showed that the polymer rich and poor regions can be identified based on the relative intensity of the $\delta(\text{C-H})$ region.

Simultaneous phase imaging of the catalysts shows sharp contrast between the polymer and SiO_2 regions because of the lower mechanical stiffness and higher viscoelasticity of the polymeric phase. This provided clear mapping of the polymer phases revealing fibre like structures weaved through the material. The combined FIB-SEM and PiFM approach uncovered significant morphological information which, when combined with real-time FTIR delivered key mechanistic insights.

Two recent papers from our own laboratory have shown the capability of PiFM to provide detailed information on the structure and formation of nanoparticles which could significantly aid the refinement of synthetic methods. Studying the development of hematite nanorods, Allender *et al.*⁶⁶ compared growth mechanisms using two different structure directing agents, in the presence of 1,2-diaminopropane the initial growth of the nanoparticles was shown to involve a goethite intermediate that subsequently underwent dissolution and recrystallisation to give the anisotropic growth of hematite. In contrast, using phosphate



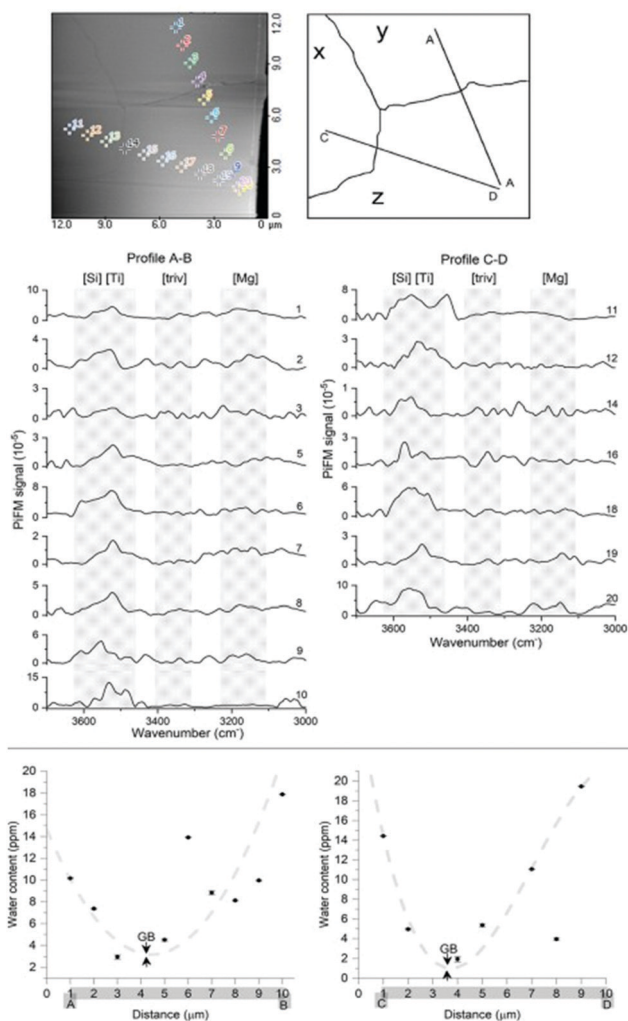


Fig. 12 Series of PiFM spectra from the polycrystalline forsterite sample. Two profiles are attained perpendicular to both two-grain boundaries. The water distribution profiles (bottom figure) were created using the estimated water content from point spectra. Patabendigedara *et al.* (Reproduced from Patabendigedara *et al.*⁶⁵).

as the directing agent akaganeite was identified on the edges and apex of the nanorods consistent with a model in which akaganeite subunits convert into hematite, Fig. 14.⁶⁷ In a separate study, Allen *et al.*⁶⁸ studied the one pot synthesis of cerium phosphate nanorods synthesising nanoparticles with aspect ratios between 3–24.8. PiFM was used to map the structure of the surface of the rods using the distinctive vibrational spectra of the hexagonal and monoclinic phosphate structures. The study showed that contrary to the results from bulk characterisation by XRD and Raman, which showed a purely hexagonal structure, the surface region of the rods contained domains of monoclinic phosphate. There was also evidence to suggest the presence of carbonate and hints of a correlation between the carbonate and monoclinic domains. Whilst the data in these studies was largely supportive of other more established techniques, the unique capabilities of PiFM to improve our understanding of nanoparticle properties and synthesis is very clearly demonstrated.

5. Surface field mapping

The examples discussed so far have focused on the vibrational spectroscopy of chemical bonds with PiFM, however the technique is also sensitive to oscillations at different frequencies, and in particular, the near infrared and visible region. It is thus able to detect surface plasmon polaritons (‘SPP’) which are oscillations in the electrons confined to the interface between a conductor and a dielectric.⁶⁹ SPP’s find use in fields as wide ranging as tip-enhanced Raman scattering and photocatalysis, and, being very sensitive to the condition of the surface, they are widely used to monitor adsorption in a variety of different sensor applications. More recently, SPP’s have been studied in connection with nano-photonics technologies because the relatively short wavelengths of the surface waves allow these very high-speed devices to be built with much smaller dimensions than is possible using “volume” oscillations.⁷⁰ PiFM has a significant advantage over other techniques used to study SPP’s because both the excitation and response measurement occur in the near field and this results in much higher signal to noise ratios than is the case when either the measurement or excitation occurs in the far-field, for example near field scanning optical microscopy (NSOM).⁷¹ Liu *et al.* exploited PiFM’s high quality signal to study graphene plasmons (GPs), a popular research topic due to their uniquely long propagation length and tunability.^{72,73} Liu measured the distribution of GPs across a monolayer of graphene on a YbF₃ spacer on top of a grating of gold, using PiFM to characterise the plasmonic nature of the grating. The results showed the GPs to be mainly concentrated on the areas over the trenches in the grating. By selecting random points along the surface of the material the authors were able to calculate the average PiFM signal and hence the amplitude of GPs in the trench and grating edge, they also calculated the propagation length of the plasmons at a given excitation wavelength, an important parameter for practical application of SSP’s.

Like the plasmon polaritons in graphene, phonon polaritons in hexagonal boron nitride (h-BN) have been extensively investigated for potential uses in nano-photonics. Two strong ‘Reststrahlen’ bands are used to study phonon polaritons: “RS1” in the region 780–830 cm⁻¹ and “RS2” between 1370–1610 cm⁻¹. While RS2 has been well studied with SNOM, RS1 remains a challenge because of a lack of the necessary continuous-wave lasers needed for the interferometric detection schemes of s-SNOM. However, Ambrosio *et al.*⁷⁴ have shown that PiFM can be used to analyse both the RS1 and RS2 bands of h-BN. Tamagnone *et al.*⁷⁵ compared the ability of PiFM and SNOM to study the h-BN polaritons and showed that PiFM was able to map all the resonant modes in a single scan with strong agreement to both calculated and SNOM distributions. PiFM produced a strong signal for mapping the amplitude of the RS1 band, shown in Fig. 15. However, for the RS2 band, whilst SNOM and PiFM showed the same wave distribution, SNOM provided clearer and stronger signals under this condition.

Working at much shorter wavelengths, Xu *et al.*⁷⁶ produced bi-functional photocatalysts capable of absorbing both the IR and UV-VIS components of solar energy. The hybrid structure consisted of vertically stacked IR-absorbing graphene



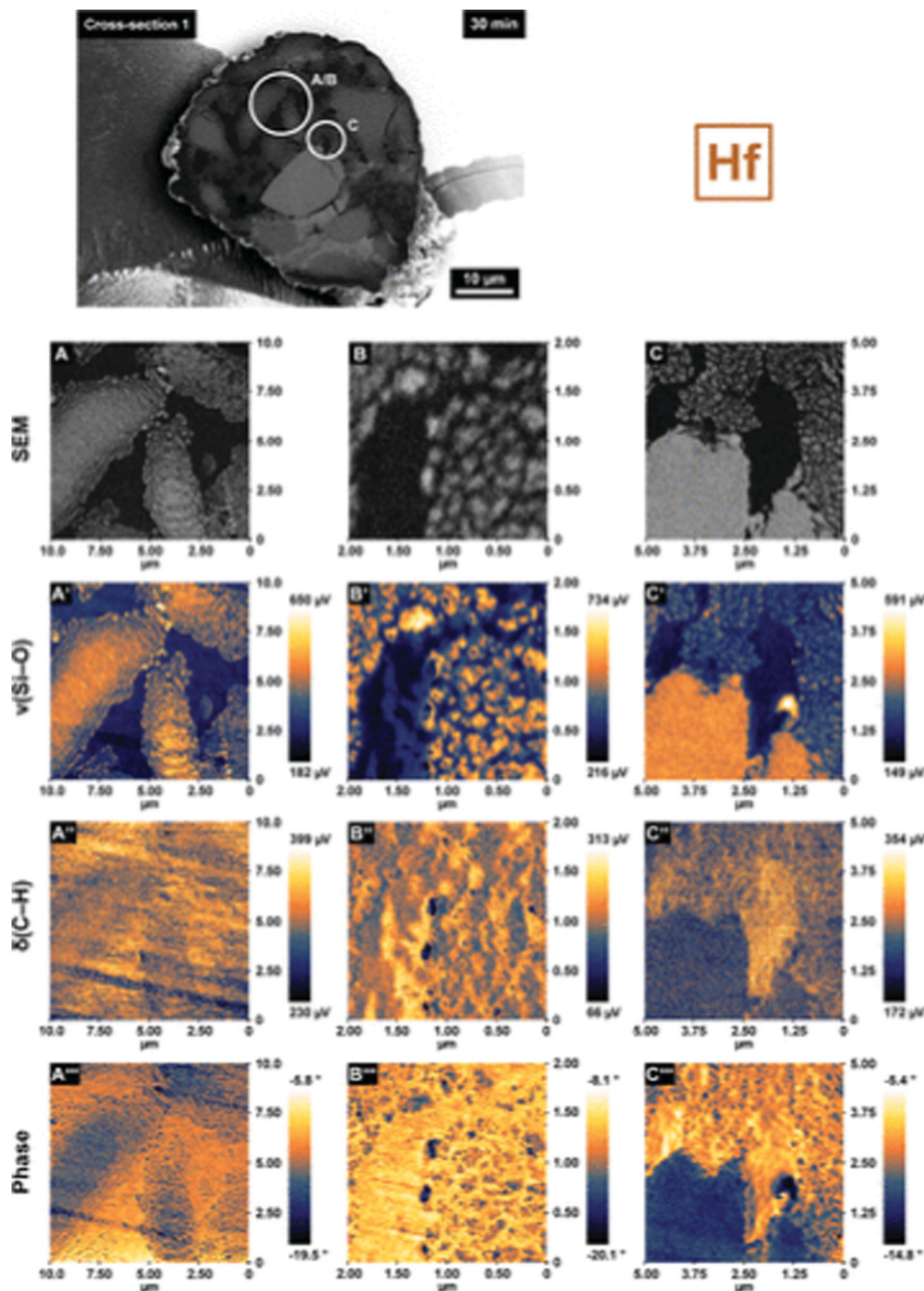


Fig. 13 Local catalyst morphology as recorded by scanning electron microscopy (SEM; A, B, C), and chemical composition as imaged by infrared photoinduced force microscopy (IR PiFM) ($\nu(\text{Si-O})$, $1050\text{--}1030\text{ cm}^{-1}$, A'–C' and $\delta(\text{C-H})$, $1472\text{--}1460\text{ cm}^{-1}$, A''–C''), and relative mechanical properties as determined by phase shift imaging (A'''–C'''). All data were collected on the cross-sections of a single 30 min prepolymerized $\text{SiO}_2/\text{MAO}/\text{Hf}$ particle. Images in panels B–B''' were recorded in the same area as those in panels A–A''' but at a higher magnification. (Reproduced from Maximilian *et al.*⁶⁵).

“nano-petals” supported by UV-VIS absorbing graphitic carbon nitride. PiFM was used to measure the propagation of GPs along the material. Topographical information highlighted sharp petal edges which are interfaces between the two structures. When combined with spectral information these topographical features showed as hot spots of plasmonic activity and acted as nano-antennas for field enhancement, Fig. 16c.

6. Method development

PiFM is a relatively new technology, and its capabilities are rapidly improving, including increasing spatial resolution, sensitivity, and conditions under which the technique can operate. Some relatively simple methods have been suggested to increase the quality of PiFM imaging, carrying out measurements



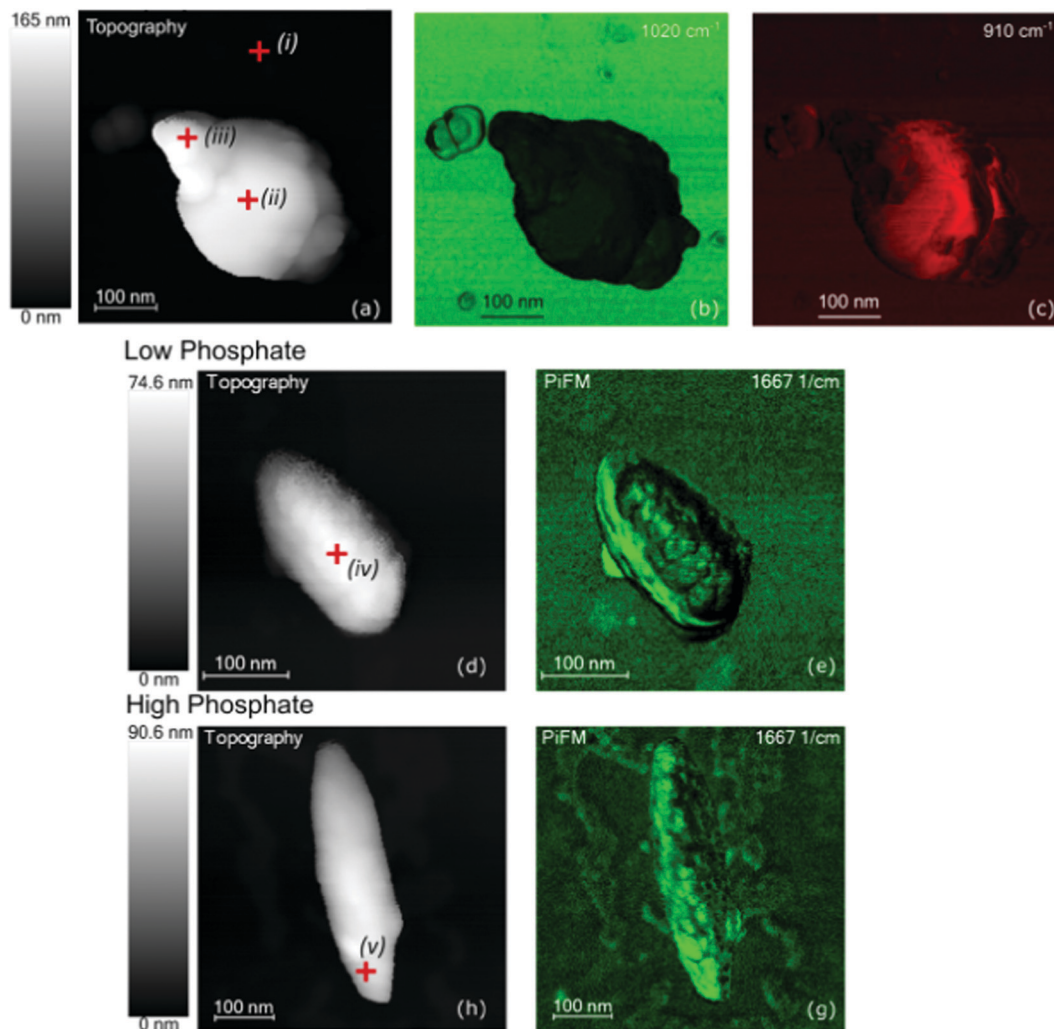


Fig. 14 Topography (left-hand side) and PiFM intensity maps of three nanoparticles synthesized under different conditions. The frequencies at which the intensity maps were recorded are indicated on the images. (a)–(c) A 1,2-diaminopropane-directed particle synthesized at 160 °C and therefore still in the process of forming the anisotropic rods. (d) and (e) A particle synthesized in the presence of 2.5×10^{-3} M phosphate solution and aged for 48 h. (h) and (g) A particle synthesized in the presence of 6×10^{-3} M phosphate solution and aged for 72 h. Reproduced from Davies *et al.*⁶⁶

under vacuum for example which, not only maintains a surface in its pristine state,⁷⁷ but also prevents surface water layers and photothermally induced acoustic waves, which, as O’Callahan *et al.* discuss can have a detrimental effect on the cantilever.⁷⁸ However, carrying out PiFM measurements in vacuum incurs other problems, including increased impact of the photothermal effect arising from the cantilever absorbing scattered light giving a vibration with a different phase to the photoinduced force.⁷⁹

6.1 Alternative heterodyne measurement configurations

One aspect in which there has been considerable interest is the sideband heterodyne detection method described in Section 2. Xu *et al.*, propose two alternative heterodyne measurement approaches for PiFM, illustrated in Fig. 17.⁸⁰ The first, referred to as the “harmonic” heterodyne configuration is designed to counter the high laser repetition rate necessary to conduct the sideband measurement described in Section 2, a typical configuration requiring a rate of ~ 1.32 MHz which is out of

the range of many light sources. Xu *et al.* show that the repetition rate can be set at an integer fraction of the frequency difference between ν_1 and ν_2 , (i.e. $\frac{1}{n}(\nu_1 - \nu_2)$, $n = 2, 3 \dots$) reducing both the laser repetition rate and the laser power. The effectiveness of the harmonic configuration is demonstrated with images of a mixed polymer film of a similar quality to the high frequency approach. A second approach, called the “sequential” heterodyne configuration, which can be combined with the harmonic mode, introduces another modulation to the infrared light source at a frequency ν_m , is demonstrated with a polarization switch and introduces additional possible modulation channels.

Yamanishi *et al.* have considered another aspect of the sideband heterodyne implementation,^{79,81} which is that in order to prevent the tip jumping to contact, the cantilever is usually driven at its first mechanical resonance frequency, ν_1 , and the photoinduced force therefore measured near to the second at ν_2 . However, the higher spring constant of the



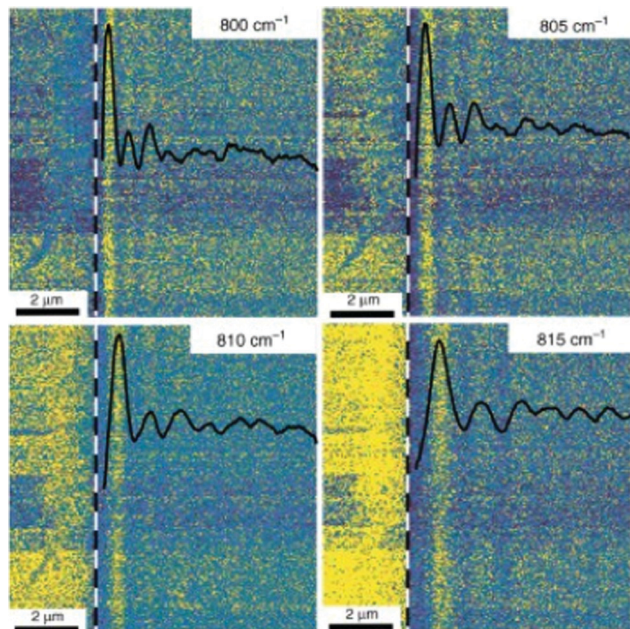


Fig. 15 PiFM imaging (and profile) of the fringes that are induced by the interference of the guided modes with the material component at the tip location for various frequencies (wavenumbers) of the illuminating light in the first RS band. The white dashed line represents the position of the flake edge. (Reproduced from Ambrosio *et al.*⁷⁴).

cantilever at ν_2 leads to a lower sensitivity. This is particularly relevant to PiFM measurements in vacuum where the higher cantilever Q factor results in a greater degree of signal degradation. Yamanishi *et al.* have proposed^{79,81} an alternative “heterodyne frequency modulation” arrangement that uses frequency rather than amplitude modulation to create the heterodyne signal. The cantilever is driven at ν_1 but, unlike the sideband configuration, the intensity of the illuminating laser is modulated at $2\nu_1 + \nu_m$. This results in a mixed signal at $\nu_1 + \nu_m$ and $3\nu_1 + \nu_m$ so that the photoinduced force gradient is measured close to ν_1 with a much-enhanced sensitivity and an estimated minimum force measurement of 419 fN.

Using heterodyne frequency modulation configuration described above, Yamanishi *et al.*⁸³ have studied dumbbell

shaped Zn–Ag–In–S (ZAIS) quantum dots recording images with a lateral resolution of ~ 0.7 nm, Fig. 18, ~ 10 times better than the standard configurations and represent the best resolution achieved with PiFM to date. Furthermore, Yamane *et al.* have suggested that because PiFM measures both the vertical and horizontal components of the gradient of the electric field⁸⁴ even higher resolutions are theoretically possible. They speculate that even atomic resolution might be achieved for PiFM measurements undertaken in cryogenic conditions using “picocavities” as a ‘super tip’.⁸⁵

6.2 Tip selection

As the calculations of Yamane *et al.* discussed above make clear,⁸⁵ PiFM measurements rely greatly on the nature of the tip used. In connection with this point, Huang *et al.*⁸⁶ investigated PiFMs ability to measure the surface field of gold nanoparticles supported on glass and calculated the optical force between a 30 nm gold nanoparticle and AFM probes made from different materials. The calculations for the Au coated AFM tips closely matched experimental results producing a signal 18.5 times stronger than that from thermal noise, Fig. 19, and showed that only Au or Ag coated tips produced signals strong enough for PiFM measurements unless measures were taken to reduce the background noise such as conducting the analysis in low temperatures and/or under vacuum.

The gold coated tips typically used in AFM measurements are produced through the sputtering Si AFM probes with gold, but this can produce larger diameter probes with low symmetry due to rogue gold particles being deposited randomly and these imperfections can lead to produce unwanted distortion of PiFM images.⁸⁷ Rajaei *et al.*⁸⁸ circumvented the need for a gold-sputtered tip while maintaining strong signals by using a gold nanoparticle adsorbed at the surface. A typical Si tipped cantilever was used to measure the surface field of gold dimers and clusters excited by linearly polarised light. This method works as rather than depending on the force being exerted from the tip to a glass slide, the dipole force is exerted by the gold nanoparticles upon the tip and resulted in signal to noise ratios of the order of 120:1 and lateral resolutions of ~ 0.6 nm. Clearly the approach in its present form is only suitable for

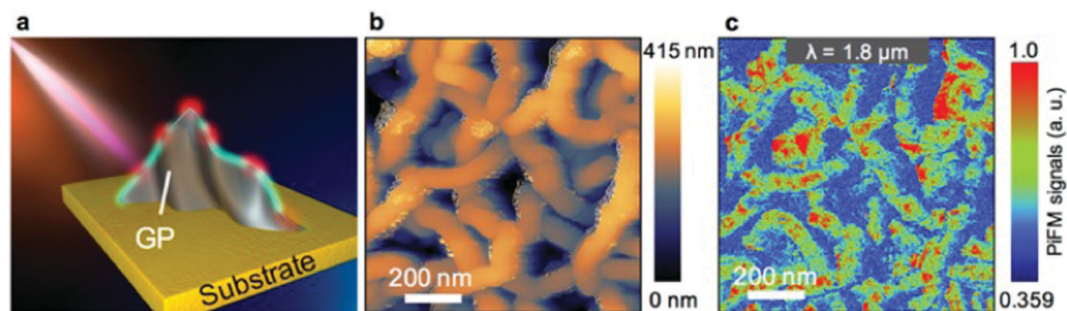


Fig. 16 PiFM near-field nano-imaging of protruding GP edge nano-antennas in the near-IR range: (a) schematic illustration of PiFM near-field nano-imaging of a single GP. The red dot along the GP edge represents the ‘hot spot’ field enhancement signal. (b) Typical PiFM topography of GP edge nanoantenna arrays. (c) Corresponding PiFM signals with apparent ‘hotspots’ along the edge nano-antennas at an incident laser wavelength λ of 1.8 μm . (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.) (Reproduced from Xu *et al.*⁷⁶).



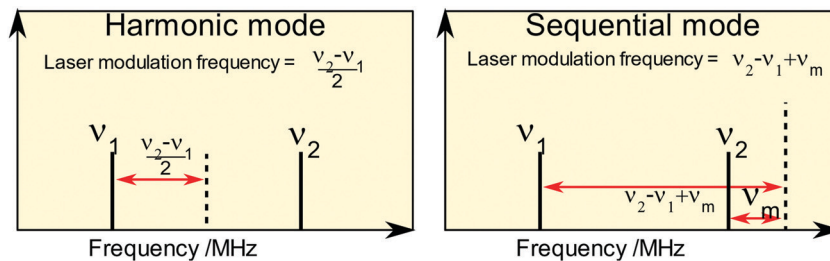


Fig. 17 Generalized heterodyne configurations proposed by Wang *et al.*⁸² Harmonic heterodyne detection is illustrated for the first integer fraction of $\nu_2 - \nu_1$, but higher order fractions are also possible. Sequential mode involves the addition of another modulation which could be in intensity or for example in frequency.

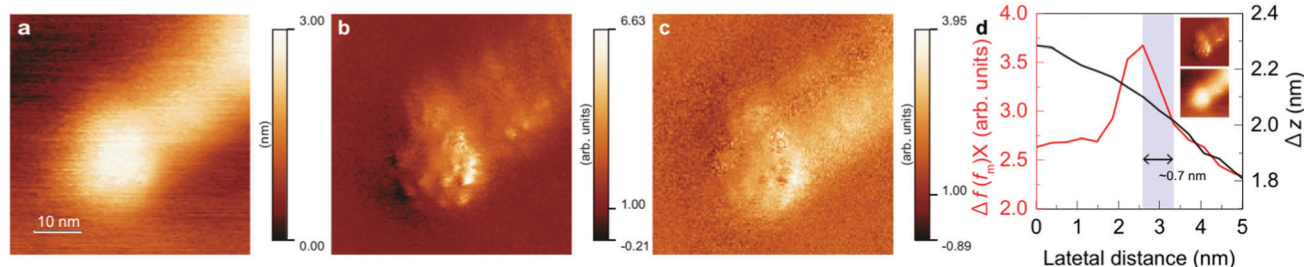


Fig. 18 PiFM imaging on one nanoellipsoid of a ZAIS QD. (a) AFM image of the ZAIS QD. (b) and (c) PiFM images of the ZAIS QD at 660 and 785 nm, respectively. (d) $\Delta f(f_m)X$ at 660 nm, and the profile of Δz in the middle of the ZAIS QD. This plot indicates that a spatial imaging resolution of ~ 0.7 nm was achieved. The positions at which the profiles were acquired are indicated in the insets. During imaging, the tip-sample distance was controlled *via* feedback control ($\Delta f = -20$ Hz) with $A = 10$ nm. (Reproduced from Yamanishi *et al.*⁸³).

imaging gold nanoparticles but it does indicate future avenues of research that might lead to improved PiFM imaging resolution and signal strength.

6.3 Photoinduced magnetic force microscopy (PiFM)

There is great interest in materials that interact with the magnetic field of light rather than its electric field because manipulation of magnetic transitions in nanostructures at optical frequencies would provide significant increases in memory capacity and read/write speeds.⁸⁹ However, the lack of

symmetry between electric and magnetic effects in nature results in magnetism in materials at optical frequencies being inherently very weak compared to electric response.⁹⁰ Furthermore, there is a major challenge in measuring such phenomena when the electric field response usually swamps the magnetic one. Azimuthally polarized laser beams provide an opportunity to sidestep this limitation with an electric field polarization that is null on the beam axis where the longitudinal magnetic field peaks.⁹¹ Zeng *et al.*^{92,93} have shown that a typical photo induced force microscope can be modified to measure the magnetic dipole under these conditions. Force maps of a sharply focused azimuthally polarized beam (APB) aimed through a glass slip at a gold coated AFM tip were recorded showing a typical doughnut-shaped morphology. The electromagnetic field components were modelled as a function of distance from the APB axis to demonstrate that the magnetic contributions predominate in the beam. A truncated Si cone was examined for this experiment as a potential nanoprobe due to its azimuth morphology.⁸⁹ Experimentally the PiFM configuration was modified with the truncated cone placed on top of the glass slip acting as a mie resonator. With this arrangement, the force map of the APB taken upon the Si cone reduced the contribution of the electric field component resulting in an area of exclusive magnetic response; the authors christened the new arrangement Photoinduced Magnetic Force Microscopy (PiFM).

Fig. 20. shows the importance of the azimuthal symmetry of the probe where (d) & (h) show that the typical sharp probe obtained no photoinduced force contrast. The symmetry of the

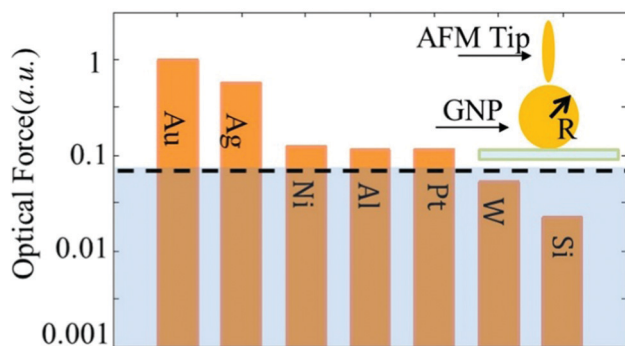


Fig. 19 Normalized optical force detected by AFM tips coated with different metals while measuring a 30 nm Au particle with incident light at 660 nm in the $|E_z|$ incident field of different permittivity of the tip calculated by COMSOL. (Reproduced from Huang *et al.*⁸⁶).



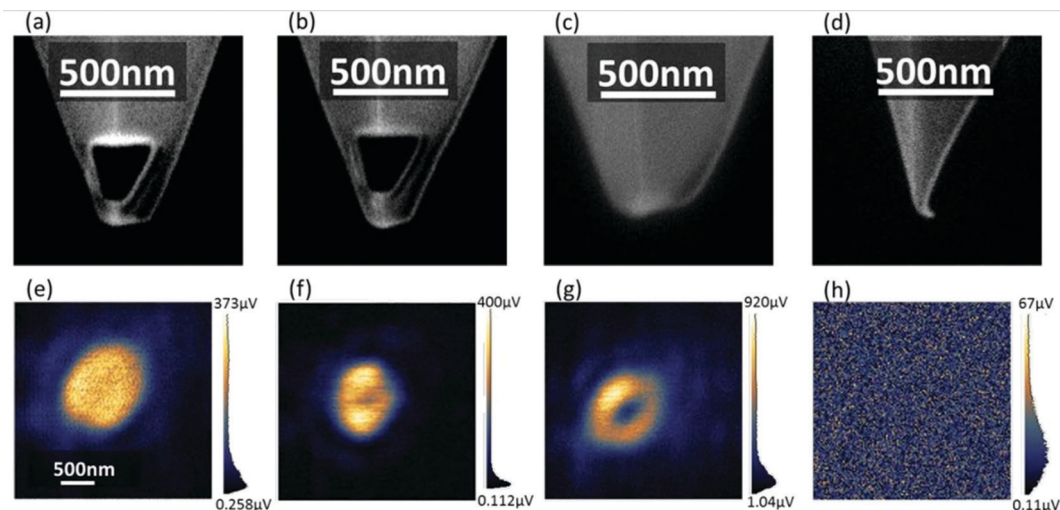


Fig. 20 (a)–(d) Focused ion beam images of the “on-state” Si truncated cone probe, “off-state” Si truncated cone probe, blunt Si probe, and sharp Si probe, respectively; (e)–(h) Corresponding measured force maps upon APB illumination from the bottom of the glass slip using the on-state Si truncated cone probe, off-state Si truncated cone probe, blunt Si probe, and sharp Si probe, respectively. The on-state probe (a) is measuring the solid-centre circular spot (e) typical of the APB magnetic field. (Reproduced from Zeng *et al.*⁹⁴).

force maps taken with tips (a) and (b) match well with those predicted for the APB magnetic field. In the case of probe (c), a doughnut-shaped force map is obtained, this is typical of an electric dipole–dipole map and does not measure a significant magnetic component. Using the truncated probe, a high signal to noise ratio of approximately 15 could be obtained. It is worth noting that the resolution of these images is low and is likely equal to the size of the nanoprobe apex at 200 nm but the system could be improved through the design of smaller probes and using rare-earth metals which could have exclusive magnetic responses to any optical beam rather than ABP. Zeng and Kamandi *et al.* note the potential of this PiFM configuration for the measurement of chiral force or PiCFM. Theoretically, if a chiral sample and the nanoprobe were illuminated with chiral light the photoinduced force would be different depending on the chirality of the light.⁹⁵ This principle has been shown to have credibility through the modelling of a chiral sphere in the presence of chiral light.

7. *In situ* and real-time measurements

NSOM⁹⁶ has been developed to take accurate measurements in the aqueous environment but requires complex setup and is strongly affected by the extent of signal absorption by the water.⁹⁷ AFM-IR and PiFM are simple alternatives to NSOM, but cantilever oscillations are mechanically damped in water leading to a significant degradation of measurements.⁹⁷ One approach proposed to overcome the weaker signals in solution is to enhance the local electric field using metal nanoantennas in the substrate.⁹⁸ In 2021, Li *et al.*⁹⁹ exploited this approach to obtain aqueous measurements with PiFM using a standard vista scope instrument with the assistance of gold nanoantennas supported upon ZnSe wafers. Initial experiments used the PIF signal arising from specific vibrational frequencies of a 1–2 nm thick polydimethylsiloxane (PDMS) coating on the AFM tip to,

qualitatively, map the electric field distribution around the antenna, demonstrating a lateral resolution of less than 10 nm even in the aqueous environment, Fig. 21. Having established the effectiveness of the approach, a self-assembled monolayer of bovine haemoglobin was imaged over the nanoantennas; the data suggests that the method has considerable promise for future studies in solution.

Connor *et al.*¹⁰⁰ used the fast signal response of PiFM to make real-time measurements of the reversible electrochemical oxidation of poly[2,5-bis(thiophenyl)-1,4-bis(2-(2-methoxyethoxy)ethoxy)ethoxy]benzene] (PB2T-TEG) polymers in vacuum. The paper shows that the polymer can switch between two different crystalline phases depending on the oxidation environment and that these phases can impact the ion-polar pair transport characteristics of the material.

Wide-angle X-ray spectroscopy (WAXS) shows that there is a change in the lattice spacing of the polymer when exposed to voltages between 0.2–0.5 V, corresponding to a change between the oxidised and neutral forms. The proposed mechanism is that under the applied bias, the glycolated side chains lose their crystallinity which allows the backbone of the polymer to untwist leading to the nucleation of dopant ions in the lattice. No clear change in the topography of the material is detected using AFM, but by following the 843 cm^{-1} P–F vibration of PF_6^- , the movement of the ion front can be monitored in real time using PiFM. Using a scan rate of 1 line per second over the same field of view, the movement of the interface between the two forms of the polymer was recorded in a series of movies. Fig. 22 shows the ion front moving across the $3 \times 3\ \mu\text{m}$ field of view over a period of 240 minutes. An important observation is that the ion front maintains a sharp edge during the motion of the ions demonstrating non-Fickian transport.

Delen *et al.*¹⁰¹ have also used PiFM in real time, studying the absorption of deuterated water on a bifunctional surface layer composed of the hydrophilic HKUST-1 surface-anchored





Fig. 21 PiFM measurements on an antenna fabricated with a mask of 5 μm PS sphere. (a) and (e): AFM images of the same region recorded in air and water, respectively. Scale bar: 1 μm . (b) and (f): PiFM images of the same region of the antennas in (a) and (e) were simultaneously recorded at 1266 cm^{-1} , respectively. (c) and (g): Signals along the red arrows of the topography (black curves) and PiFM images (blue curves). (This figure has been published in CCS Chemistry 2021; "Antenna Enhanced Infrared Photoinduced Force Imaging in Aqueous Environment with Super-Resolution and Hypersensitivity" is available online at <https://doi.org/10.31635/ccschem.021.202101341>⁹⁹).

metal-organic framework (SURMOF) described in Section 2.1, and a hydrophobic, zeolitic-imidazolate Frameworks (ZIF-8). Nanometre-sized hydrophilic and hydrophobic domains were identified in the mixed layer and the hydrophilic HKUST-1 material shown to absorb D_2O on existing defective sites, as well as generating free deuterated carboxylic acid confirming proposed mechanisms of the hydrolysis of HKUST-1 paddlewheels.

7.1 Tandem measurements

Tip based surface spectroscopy techniques have different strengths and weaknesses, that effects the combination of

multiple techniques. This is an area of much research. PiFM has been combined with two other systems in recent years, TERS,¹⁰² SNOM¹⁰³ and Kelvin probe force microscopy (KPFM).¹⁰⁴

Meyer *et al.*¹⁰² integrated a typical TERS system with a PiFM to image 2D MoS_2 monolayer flakes. Signal optimisation is a crucial aspect of any TERS experiment but since PiFM can map the electromagnetic field of various materials and light, it can be used in tandem with the TERS to assist in the focusing and alignment of the laser. The PiFM was able to agree with computational calculations of the field distribution of light in three separate cases: Overfocused, focused and under focused.



Fig. 22 Ion front movement in PB2T-TEG and P3MEEMT. (A) Example image of the samples used for ion front imaging. The grey area is oxidized in 100 mM KCl at +0.7 V (vs. Ag/AgCl) and the red area is neutral. (B) Optical transmission images of the interface between oxidized and neutral regions of the film for PB2T-TEG and P3MEEMT. The transmission images are acquired in the same spectral region as the polaron absorption ($750 \pm 25\text{ nm}$). (C) Time series of spatially correlated PiFM and topography images as the interface migrates in PB2T-TEG. The white line designates the 50% intensity threshold from the PiFM channel. (Reprinted with permission from Bischak *et al.*, "A Reversible Structural Phase Transition by Electrochemically-Driven Ion Injection into a Conjugated Polymer", *J. Am. Chem. Soc.*, 2020, **142**, 7434–7442 Copyright 2020 American Chemical Society¹⁰⁰).



systems and there are realistic expectations of that being improved with careful experimental design and choice of AFM probe. The frequency range studied to date stretches across the electromagnetic spectrum from mid infrared to the visible region, encompassing not only plasmonic oscillations but also the very interesting molecular vibrational frequencies that have been a source of chemical information for 70 years or more. Comparison between PiFM and far field FTIR shows an excellent match in peak intensities and positions allowing direct reference to the rich library of FTIR spectra in the literature, but PiFM also shows much improved spectral resolution offering the prospect of mapping of very closely related domains at surfaces with far more precision than has been possible previously and will allow much more rigorous testing of theoretical models of the surface adsorption. The theoretical understanding of PiFM has also developed, to put modelling of the phenomenon on a firm foundation capable of underpinning predictive calculations that will inspire future developments. Real time measurements with PiFM have also started to appear; so far these have reported phenomena that can be monitored on a timescale of minutes using full AFM images but since signal detection in the PiFM is of the order of 10 milliseconds,¹⁰⁶ there is the potential for much more progress in this area. Closely related to that point and an aspect that has not received a great deal of attention so far, is the question of the quantitative measurement of surface species with PiFM. This is an area that needs careful investigation in the near future and is likely to offer new challenges to the experimentalist and theoretician alike. Development of the technique to tackle more demanding environments particularly *in situ* experiments in aqueous environments is inevitably on the wish list of interface scientists and recent results have shown this is possible but before such difficult experiments are fully developed we can expect a great deal more work in gas phase environments to be undertaken. Thermal effects have also not yet been explored in any great detail and we can confidently predict there will be much to discover in the surface chemistry of systems as the temperature is changed.

In conclusion, PiFM has made great strides in just over a decade but there is much more to be done and many more opportunities to exploit.

Conflicts of interest

There are no conflicts to declare.

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