

Cite this: *Chem. Sci.*, 2019, 10, 5952

All publication charges for this article have been paid for by the Royal Society of Chemistry

Widely applicable background depletion step enables transaminase evolution through solid-phase screening†

Matteo Planchestainer,^a Eimear Hegarty,^a Christian M. Heckmann,^a Louise J. Gourlay^b and Francesca Paradisi^{b*ac}

Directed evolution of transaminases is a widespread technique in the development of highly sought-after biocatalysts for industrial applications. This process, however, is challenged by the limited availability of effective high-throughput protocols to evaluate mutant libraries. Here we report a rapid, reliable, and widely applicable background depletion method for solid-phase screening of transaminase variants, which was successfully applied to a transaminase from *Halomonas elongata* (HEWT), evolved through rounds of random mutagenesis towards a series of diverse prochiral ketones. This approach enabled the identification of transaminase variants in viable cells with significantly improved activity towards para-substituted acetophenones (up to 60-fold), as well as tetrahydrothiophen-3-one and related substrates. Rationalisation of the mutants was assisted by determination of the high-resolution wild-type HEWT crystal structure presented herein.

Received 20th December 2018

Accepted 26th April 2019

DOI: 10.1039/c8sc05712e

rsc.li/chemical-science

Introduction

Transaminases have attracted considerable attention for the production of compounds containing chiral amino groups, which are widespread in the pharmaceutical, agrochemical and fine chemical industry.^{1,2} Of particular interest, in terms of synthetic applications, are the amine transaminases (ATAs), (*R*)- or (*S*)-selective enzymes that are capable of aminating carbonyls (both aldehydes and ketones) in the absence of a neighbouring carboxylic group.³ Despite the enormous potential of ATAs, the widespread application of these biocatalysts in their wild type form is hindered by a number of challenges including limited substrate specificities, substrate and product inhibition, and unfavourable reaction equilibria.⁴

Protein engineering by rational design⁵⁻⁷ and directed evolution^{8,9} have played a major role in the ability to address these issues. Evolutionary strategies, in particular, are ideally suited to the optimisation of the physical properties of enzymes for large-scale manufacturing processes, as well as enabling fine-tuning of substrate and stereochemical preferences.^{10,11} The evolution of ATAs by Codexis and Merck^{12,13} pioneered

subsequent engineering efforts of many ATAs, resulting in the production of an array of highly specialised ATAs for chiral amine synthesis.^{8,9,14}

Even though a fully randomised approach to mutagenesis, targeting the whole length of the protein, has become less common, directed evolution clearly requires high-throughput screening methods to evaluate variable size libraries of engineered variants. The generation of smaller, smart libraries designed by rational targeted approaches such as CASTing,¹⁵ or saturation mutagenesis of specific residues identified *in silico*, still yield a sizable number of clones to screen (saturation at three positions affords 8000 variants). The identification of hits within live cells is advantageous for library screening because, if it is efficient, it prevents culturing of large number of clones followed by assays with cell extracts, which, in the absence of robotics, is very time-consuming.¹⁶ Recently, microfluidic screening,^{17,18} in addition to the application of desorption electrospray ionisation mass spectrometry¹⁹ on whole colonies, have emerged as powerful tools for screening large libraries, however, the equipment required in both cases is still highly specialised and expensive, therefore they are not yet suitable for widespread implementation. Solid-phase screening (SPS), on the other hand, is a simpler approach that facilitates the screening of a medium-large number of mutant variants based on colorimetric changes that are easily observed by the naked eye.²⁰ SPS was successfully applied in the evolution of monoamine oxidase (MAO-N) enzymes, using two libraries that were fully saturated at two positions each (400 variants).²¹ SPS methods applicable to ATA mutant libraries, however, are currently limited to a few strategies to screen for amino donors.

^aUniversity of Nottingham, School of Chemistry, University Park, Nottingham NG7 2RD, United Kingdom

^bStructural Biology Unit, Dep. Biosciences, Università degli Studi di Milano, Via Celoria 26, 20133 Milano, Italy

^cUniversity of Bern, Department of Chemistry and Biochemistry, Freiestrasse 3, 3012 Bern, Switzerland. E-mail: francesca.paradisi@dcb.unibe.ch

† Electronic supplementary information (ESI) available. See DOI: 10.1039/c8sc05712e



Wu and co-workers, for example, reported the use of SPS to screen libraries of approximately 100 000 ATA variants for reduced inhibition towards methoxyisopropylamine.²² Later, Bornscheuer and colleagues applied SPS to evolve an ATA towards suitable amino donors (screening *ca.* 4600 colonies), however this required co-expression of a second enzyme as a coupled reporting system, increasing the complexity of the system.^{23,24} Similarly, Turner and co-workers evolved SPS for ATA library screening, specifically to identify either (*R*)- or (*S*)-selective ATAs (screening *ca.* 1000 colonies), which also required multiple reporting enzymes, as well as flash-freezing in liquid nitrogen, to increase permeability.²⁵

Screening strategies for varying carbonyl acceptors, which are key substrates for ATAs to produce chiral amine products, appear to be much more challenging. *ortho*-Xylylenediamine²⁶ and 4-nitrophenethylamine²⁷ have been recently reported as amine donors in proof of concept studies to rapidly generate coloured by-products upon transamination. Both these strategies suffer from strong background interference in live cells due to endogenous pyruvate which is nearly always an excellent substrate for ATAs; in a mutant library screening, the number of false positives is certain to be significant without an efficient system to deplete background. Pyruvate is a primary metabolite, essential for cell growth, therefore knock-out strategies are not feasible, making it inherently difficult to minimise the background noise in SPS.

We here report the development of a simple and reliable SPS method that encapsulates (*S*)-1-phenylethylamine as a straightforward background depletion agent to minimize high levels of pyruvate activity. To highlight the power of this methodology, we successfully applied it to the screening of a fully randomised mutant library of an (*S*)-selective ATA from *Halomonas elongata* (HEWT)²⁸ against a series of prochiral aromatic and cyclic ketones. In addition, the crystal structure of the wild-type HEWT is reported, which facilitated rationalisation of the mutations introduced.

Results and discussion

Depletion of background activity and test screening

Reliable methodology to quench background activity, which leads to a high number of false positives, was designed to specifically deplete endogenous pyruvate. Unlike previously discussed methods, our alternative approach is based on pre-reacting cellular pyruvate with (*S*)-1-phenylethylamine. The reaction is catalysed by the expressed transaminase to produce alanine and acetophenone with an extremely favourable equilibrium, effectively removing pyruvate as a non-specific acceptor from the environment. The screening was initially optimised with *E. coli* BL21(DE3) cells expressing wild-type HEWT, which were spread onto a nitrocellulose membrane deposited on a LB-agar plate. This permitted easy transfer of the colonies through the different assay steps (Fig. S1†). Following expression and dialysis (Experimental in the ESI†), the membrane was left for 30 minutes on filter paper soaked with (*S*)-1-phenylethylamine and then washed with 50 mM phosphate buffer, pH 8.

Test screening of pre-treated colonies to establish timelines and reaction efficiency, were performed with both pyruvate and acetophenone and compared with a control plate not supplemented with any amino acceptor (Fig. S2†). *ortho*-Xylylenediamine was used in the first instance as an amino donor. The control plate did not give any colorimetric change up to two hours, after which time a slightly brown tinge could be observed likely due to the cell healthy metabolic system producing sufficient amounts of pyruvate detectable through the reaction. In the absence of the background depletion step, the cells almost immediately respond to the amino donor preventing any significant discrimination between exogenous amino acceptor and the endogenous pyruvate (Fig. 1), as was also observed by O'Reilly and co-workers.²⁶ Likewise, the addition of pyruvate to the membrane as the amino acceptor, causes visible darkening of the colonies within 5 minutes of incubation. Acetophenone, which in liquid-phase assays had shown no detectable activity over 3 h with wild-type HEWT, performs similarly to the control plate, with no observable colour formation during total incubation (Fig. S2†). The mild treatment, preserved cell viability, which is essential for the direct growth and plasmid extraction, bypassing the need for a replica plate. To further evaluate the effectiveness of this strategy, 4-nitrophenethylamine was used as amino donor in the assay.²⁷ Remarkably, a significant reduction in background noise was observed once more, highlighting the transferability of this methodology (should *o*-xylylenediamine not be accepted by the investigated TA). Indeed, the background was even further suppressed when compared to the *ortho*-xylylenediamine screen, with barely detectable darkening of the colonies after 24 h (Fig. S3†). Liquid nitrogen- and chloroform-induced permeabilization steps were attempted also,^{24,25} but in our case neither method was successful.

The use of amines other than (*S*)-1-phenylethylamine, such as isopropylamine and cadaverine, for the background depletion step was also investigated. Cadaverine, previously described as a “smart” amine donor,²⁹ can be used in ATA reactions *ex vivo* and may effectively convert endogenous pyruvate in whole cells. Isopropylamine is a very inexpensive amine donor, commonly used in industrial processes;¹² the generated acetone is produced in minimal quantities and expected to be harmless to colonies. While both amines also resulted in

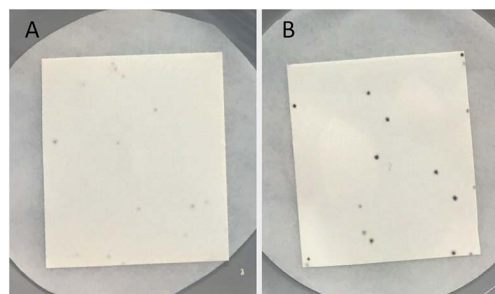


Fig. 1 *ortho*-Xylylenediamine solid screening control. *E. coli* BL21(DE3) cells expressing wild-type HEWT after 1 h of incubation with *ortho*-xylylenediamine at RT, following background depletion step (A), and in the absence of such pre-treatment (B).



a significant reduction in background (Fig. S4†), (*S*)-1-phenylethylamine appeared to be most effective, possibly due to better diffusion of this substrate. However, this demonstrates that other amines can also be used to quench endogenous pyruvate, for example with aminotransferases that do not accept 1-phenylethylamine as a substrate.

Evolution towards substituted acetophenones

To test the methodology, HEWT was subjected to directed evolution using error-prone PCR to generate a completely randomised mutant library with a high number of mutations. The enzyme has minimal activity towards the aromatic substrate *para*-nitroacetophenone (**1a**) and this was selected as the amino acceptor for the initial screening, using *ortho*-xylylenediamine as the amino donor. DMSO (10% (v/v)) was required for substrate solubility and enhanced cell-permeability. The discrimination between wild-type and an improved variant relies on the rapidity with which the enzyme converts the substrate *in vivo*. Screening of *ca.* 15 000 colonies provided two variants with enhanced catalytic activity towards *para*-nitroacetophenone. The two new variants (referred to as A1 and B1) were expressed and purified for further characterisation. They exhibited a 2-fold increase in turnover frequency with respect to the wild-type, achieved higher conversion after 2 hours, and maintained excellent enantioselectivity (Table 1). Both variants were subjected to a second round of error-prone PCR and the libraries screened as before with **1a**. Interestingly, in this case the positive colonies were isolated after a significantly shorter incubation period (2–15 minutes compared to 25 minutes for the parental colony). While the A1

template did not provide any significant hits, an improved variant was isolated from the B1 library. The new mutant, identified as B2, afforded comparable molar conversion after 2 hours and a further 1.5-fold increase in turn-over frequency (Table 1). The e.e. measured at 24 h was in all cases >99% (though the overall yields were low due to experimental conditions needed to assess the kinetic behaviour of the mutants). A third round of mutagenesis, using the B2 variant as a template, initially failed to identify better variants, as the colour change on solid screening was too rapid and did not allow discrimination between parental activity and mutants. Reducing the concentration of *para*-nitroacetophenone and *ortho*-xylylenediamine from 10 mM to 1 mM, with just 5% (v/v) DMSO prolonged the screening window to *ca.* 1 h. An additional variant, B3, was isolated from the screening and while, upon purification, it displayed comparable reaction velocity and 2 h conversions to B2, it had 2-fold higher expression levels which justifies a more rapid colour development with respect to the parental variant. Each variant was also fully characterised in terms of activity and stability at different temperatures and pHs. In all the cases, the mutants generated did not show significant alterations when compared to the WT enzyme (Fig. S5–S8†).

A second substrate, *para*-cyanoacetophenone (**2a**), was also investigated. HEWT wild-type has negligible activity towards this molecule, even lower than the nitro substitution, with a turnover frequency of $4 \times 10^{-3} \text{ s}^{-1}$. The B2 variant showed an impressive 60-fold increase in turn-over frequency ($229 \times 10^{-3} \text{ s}^{-1}$), and enhanced stereoselectivity, affording >99% of the (*S*)-enantiomer (Table 1). Again, B3 showed virtually identical catalytic properties to B2.

Table 1 HEWT evolution towards substituted acetophenones. Turnover frequencies (TOF) measured at 10 mM carbonyl concentration, 2 hour molar conversions (m.c.), and enantiomeric excess (e.e.) for wild-type HEWT and isolated variants towards *para*-nitro- (**1a**) and *para*-cyanoacetophenone (**2a**)^a

Reaction scheme: Ar-C(=O)-CH_3 (1-2a) + L-Alanine $\xrightarrow{\text{HEWT}}$ $\text{Ar-CH}_2\text{-CH}_2\text{-NH}_2$ (1-2b) + Pyruvate

Substrate	WT			A1			B1			B2			B3		
	TOF (10^{-3} s^{-1})	m.c. (%)	e.e. (%)	TOF (10^{-3} s^{-1})	m.c. (%)	e.e. (%)	TOF (10^{-3} s^{-1})	m.c. (%)	e.e. (%)	TOF (10^{-3} s^{-1})	m.c. (%)	e.e. (%)	TOF (10^{-3} s^{-1})	m.c. (%)	e.e. (%)
1a	36 ± 6	4	>99 (S)	66 ± 9	6	>99 (S)	65 ± 26	6	>99 (S)	92 ± 5	7	>99 (S)	84 ± 14	7	>99 (S)
2a	4 ± 1	9	87 (S)	50 ± 7	11	94 (S)	127 ± 5	12	>99 (S)	229 ± 7	11	>99 (S)	246 ± 7	12	>99 (S)

^a Biotransformation reactions were performed with 10 mM ketones, 500 mM L-alanine, 0.1 mM PLP, 0.1 mg mL⁻¹ (1.9 μM) enzyme in 50 mM phosphate buffer pH 8 and 10% (v/v) DMSO at 37 °C (see Experimental in the ESI, Fig. S9). All experiments were conducted in triplicate and the standard error is reported accordingly. Mutations; A1: W56C, V435A; B1: W56C, L211V, L306M; B2: W56C, L211V, L306M, V361A, Q388R, P453L; B3: W56C, L211V, A254V, L306M, V361A, Q388R, P453L.



Evolution towards small cyclic ketones

To further expand the catalytic scope of HEWT, the original mutant library was screened towards a structurally different substrate, tetrahydrothiophen-3-one (**3a**), a small cyclic ketone which if successfully aminated yields a chiral amine (**3b**) that has potential use as a building block for active pharmaceutical ingredients.³⁰

In this case, the wild-type enzyme already possesses some activity towards the target substrate (**3a**), with a turnover frequency of $206 \times 10^{-3} \text{ s}^{-1}$ and complete molar conversion (>99%) after 24 h using equimolar amounts of (*S*)-1-phenylethylamine as donor and higher catalyst concentration ($1 \text{ mg mL}^{-1} - 19 \mu\text{M}$). The optimised solid phase screening with *ortho*-xylylenediamine yielded a new variant (C1) which, upon expression and purification, displayed a 2-fold increase in turnover frequency (Table 2). On this occasion, the mutant showed a colour change after 18 minutes, compared to 30 minutes for the wild-type control. Notably, the variant matched the excellent conversion displayed by the wild-type enzyme, affording the (*S*)-enantiomer of the corresponding amine with >99% conversion and 66% e.e.



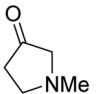
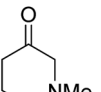
The C1 variant was subsequently screened against a panel of representative cyclic ketones (Table 2, **4a**, **5a**, **6a**). The mutant had improved TOF with both tetrahydrofuran-3-one (**4a**) and 1-methylpiperidin-3-one (**6a**) (35% and 90% increase, respectively) but it was in fact slower with 1-methylpyrrolidin-3-one (**5a**) (see Table S1†). Similarly to what was observed with the library evolved towards *para*-nitroacetophenone, an additional round of mutagenesis did not permit the identification of better variants towards tetrahydrothiophen-3-one (**3a**) as the colour change on the plates was too rapid to discriminate further enhanced variants.

This clearly demonstrates that this solid phase screen is able to identify better variants, both in terms of TOFs and expression levels, from libraries that are of desirable size for semi-rational approaches to directed evolution.¹⁰ The application of this methodology was successful even with a fully randomised approach which can only afford limited improvements of the mutants, given the size of the library, due to the high probability of unsuitable mutations.³¹

3D structural analysis of wild type HEWT and mutant variants

In order to understand the effect of the various mutations on wild-type HEWT, the crystal structure was determined at

Table 2 HEWT evolution toward small cyclic ketones. Turnover frequencies (TOF) measured at 10 mM carbonyl concentration, molar conversions (m.c.), and enantiomeric excess (e.e.) for wild-type HEWT and isolated variants towards tetrahydrothiophen-3-one (**3a**) tetrahydrofuran-3-one (**4a**), 1-methylpyrrolidin-3-one (**5a**) and 1-methyl-piperidin-3-one (**6a**). Final conversion was obtained within 2 hours of reaction^a

Substrate	WT			C1		
	TOF (10^{-3} s^{-1})	m.c. (%)	e.e. (%)	TOF (10^{-3} s^{-1})	m.c. (%)	e.e. (%)
 3a	206 ± 5	>99	66 (<i>S</i>)	460 ± 6	>99	66 (<i>S</i>)
 4a	430 ± 1	>99	70 (<i>S</i>)	580 ± 3	>99	70 (<i>S</i>)
 5a	18 ± 0.2	94	90 (<i>S</i>)	11.6 ± 0.3	94	90 (<i>S</i>)
 6a	10 ± 0.8	98	90 (<i>S</i>)	19 ± 0.8	98	90 (<i>S</i>)

^a Biotransformation reactions were performed with 10 mM ketones, 10 mM (*S*)-1-phenylethylamine, 0.1 mM PLP, 1 mg mL^{-1} ($19 \mu\text{M}$) enzyme in 50 mM phosphate buffer pH 8 and 10% (v/v) DMSO at 37 °C (see Experimental in the ESI, Fig. S10). All experiments were conducted in triplicate and the standard error is reported accordingly. Mutation in C1: D5E.



a resolution of 2 Å, as described in the methods. Atomic coordinates and structure factors have been deposited in the Protein Data Bank (www.rcsb.org) under PDB accession code 6GWI.

Two identical HEWT monomers (chains A and B; RMSD 0.23 Å) were present in the asymmetric unit (Matthews coefficient = 2.14 Å³ Da⁻¹), with an estimated solvent content of 42.5% (see Table S2†). This homodimer arrangement is biologically relevant as HEWT is a catalytically active dimer. Electron density was present for residues 2 to 451 (chain A) and 1 to 450 (chain B). The HEWT monomer comprises 3 sub-domains: (i) a principle PLP-dependent transferase-like domain (residues 82 to 318) that hosts a central mixed β-sheet, comprising 7 strands, (β9–β8–β10–β11–β12–β13–β7), with β-strand 13 being antiparallel to the rest, surrounded by 9 α-helices and two 3¹⁰ helices (η1–2) (Fig. 2); (ii) domain 2 (residues 2–79; 320–451) that contains a small N-terminal sub-domain (residues 2 to 32; α1–β1–β2–α2–β3–5) which forms the main interactions with the opposing monomer. One molecule of pyridoxal-5'-phosphate (PLP), produced by the covalent attachment to the ε-amino group of the catalytic lysine (K284), *via* an aldimine linkage, is bound to each monomer. Dimerisation involves 100 residues

contributed from each monomer, with an interface area of 5395 Å². The main dimerisation interactions include 4 salt bridges and 47 hydrogen bonds, as calculated by PDBsum available from the European Bioinformatics Institute (www.ebi.ac.uk).

Based on a secondary structure-based homology search, the DALI server identified the 4-aminobutyrate transaminase from *Pseudomonas* as the top hit (PDB entry 5KR6³²) with a RMSD of 1.3 Å over 445/460 aligned main chain residues.³³ Sequence and structure conservation between HEWT and all other structures deposited in the PDB was assessed using ENDscript 2.0 (<http://endscript.ibcp.fr>).³⁴ HEWT was compared with 124 structures with sequence conservation >30%. As expected, the highest conservation (sequence and structural) is located in the PLP-dependent transferase-like domain and active site region (Fig. 3A).

Sequencing of the selected mutants identified two amino acid changes in the A1 variant, (W56C, V435A), while the B1 variant displayed three amino acid substitutions (W56C, L211V, L306M) (notes in the ESI†). Interestingly, both mutants harbour a tryptophan to cysteine change in position 56, which is located in the active site. This residue has previously been identified as a hotspot in *in silico* rational design studies.^{7,35} In this case, removing the large indole ring and replacing it with the smaller cysteine side chain allows for easier substrate binding. Furthermore, the thiol can promote hydrogen bonding with the

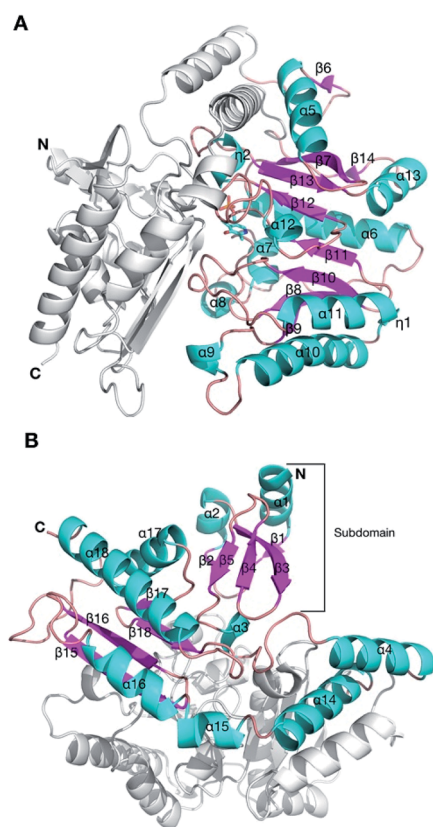


Fig. 2 The crystal structure of the HEWT WT monomer. Cartoon secondary structure representation of the (A) PLP-dependent transferase-like domain (residues 82 to 318), with PLP indicated in sticks. For simplicity, the other domain is not coloured. The N- and C-termini are labelled. (B) The second domain of HEWT, comprising a subdomain (labelled) that mediates dimerisation with the opposing monomer. In this case, the PLP-dependent domain is not shown for clarity. This figure was generated using Chimera.³⁶

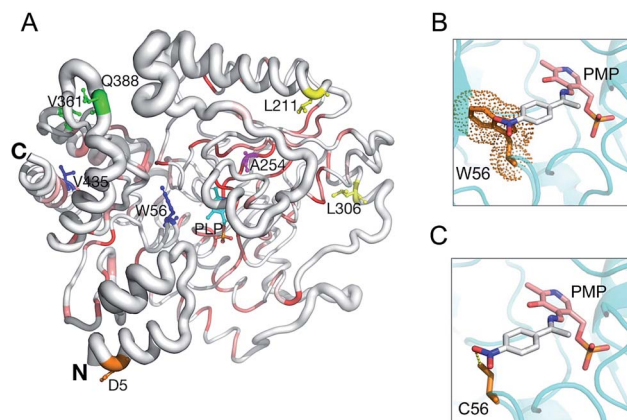


Fig. 3 HEWT 3D structure conservation and mutant residue positions. Secondary structure of the HEWT monomer shown in sausage representation, as automatically generated by ENDscript 2.0 (<http://endscript.ibcp.fr>).³⁴ Structure conservation between chain A of HEWT and 124 structure homologs present in the PDB is indicated by ribbon thickness, with regions of low conservation being thicker than highly conserved regions (thin regions). Sequence conservation is indicated by red shading; the redder the residue, the more conserved it is. Mutated residues in variants A1 (blue), B1 (yellow), B2 (green) and C1 (orange) are shown as sticks. The B3 mutant shares all B2 mutations and contains an extra A254V mutation (pink sticks). P453 from variants B2 and B3 is not present in the model. The N- and C-termini are indicated and PLP is shown in sticks. This figure was generated using Pymol 2.0.6. Modelling of the active site for the A1 and B1 mutants, where the bulky Trp (the steric encumbrance of the residue is shown as an orange cloud) (D) is substituted with a cysteine (E), allowing the aromatic ring substituent (*para*-nitroacetophenone in the aldimine intermediate complex with the pyridoxamine phosphate (PMP)) to be more easily accommodated.



aromatic substituent and stabilise *para*-nitroacetophenone (Fig. 3B and C). Accordingly, W56 is located in a highly structurally conserved region (Fig. 3A), as is V435 (Fig. 3A). The L306M mutation is peripheral and not conserved with regards to sequence, and therefore less conserved than the A1 mutations with regards to structure, and therefore is somewhat more difficult to rationalise. The L306M mutation introduces a longer side chain that may form more stabilising hydrophobic interactions with surrounding hydrophobic residues (M126, V302 and F313). From the structure, this hydrophobic cluster may in-turn stabilise regions of the protein that mediate dimerisation (Fig. S11†). In agreement with this hypothesis, thermal stability analysis of the wild-type and mutant B2 indicates that the latter is more stable (Fig. S12†). The second round of evolution introduced three additional mutations (V361A, Q388R, P453L) in the B2 variant, which are also located outside of the active site and all are exposed to the solvent; their side-chains do not make any significant stabilising intramolecular interactions. B3 introduced mostly silent mutations, apart from a single A254V mutation neighbouring the D255 residue, forming a hydrogen bond with the pyridine-type nitrogen of the co-factor PLP.

On the other hand, structural analysis of the C1 variant revealed a single mutation (D5E) positioned at the N-terminus of the protein (Fig. 3A). Using the protein structure quasi-rigid domain decomposition (PiSQRD) web server (<http://pisqrd.escience-lab.org>),³⁷ the HEWT crystal structure (chain A) was decomposed into four sub-domains that can behave as independent units during conformational fluctuations (Fig. S13†). D5 is located in domain 4 (residues 2–34 and 41–46) that adopts an α 1-helix–loop– α 2-helix structure. Based on these observations, an aspartate to glutamate mutation may alter the local protein dynamics that in-turn may cause alterations in the topology of the active site and hence substrate specificity and activity.

Conclusions

The introduction of an amine donor as a background depletion agent, which exploits the expressed ATA to quench endogenous pyruvate proved to be very effective in delivering an efficient screening platform for aminotransferase directed evolution. This high-throughput screening strategy was developed on the basis of the *ortho*-xylylenediamine assay described by O'Reilly and co-workers,²⁶ which was a fundamental tool in the evolution of HEWT towards a diverse set of aromatic and cyclic ketones. After three rounds of directed evolution, a 2.5-fold increase in activity was obtained for *para*-nitroacetophenone, while with *para*-cyanoacetophenone the identified mutant exhibited a 60-fold higher turnover frequency coupled with a substantial improvement of the enantioselectivity with respect to the wild type. Furthermore, the evolution towards tetrahydrothiophen-3-one yielded a 2-fold increase in the overall catalytic efficiency. This enzyme also showed improved catalytic properties when screened against similarly challenging cyclic ketones. The background depletion step is straightforward and reliable, and can be applied to other screening methods where endogenous pyruvate interferes with detection making the evolution of these

enzymes a much less challenging task. The use of 4-nitrophenethylamine as an alternative reporter, as well as alternative amine donors for the depletion step, proved successful. The limit of this screen is set by the power of discrimination between the mutant and the parent enzyme; if the substrate conversion is too fast, the colonies turn black rapidly, which leads to false positives. Tuning the screening conditions to slow the reaction can extend the screening window. In addition, through modulation of the parental activity, by regulating the expression levels (with the use of a promoter that allows for stricter control),³⁸ this screening method could be a very powerful tool in further rounds of evolution towards the identification of more and more active mutants.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This work was supported by the Biotechnology and Biological Sciences Research Council [grant number BB/P002536/1 and BB/M008770]. The authors thank the University of Nottingham through the Green Chemicals Beacon for support. Structural studies were supported by Università degli Studi di Milano Linea 2 funding.

References

- 1 I. Slabu, J. L. Galman, R. C. Lloyd and N. J. Turner, *ACS Catal.*, 2017, 7, 8263–8284.
- 2 M. Fuchs, J. E. Farnberger and W. Kroutil, *Eur. J. Org. Chem.*, 2015, 2015, 6965–6982.
- 3 F. Steffen-Munsberg, C. Vickers, H. Kohls, H. Land, H. Mallin, A. Nobili, L. Skalden, T. van den Bergh, H. J. Joosten, P. Berglund, M. Höhne and U. T. Bornscheuer, *Biotechnol. Adv.*, 2015, 33, 566–604.
- 4 I. Slabu, J. L. Galman, C. Iglesias, N. J. Weise, R. C. Lloyd and N. J. Turner, *Catal. Today*, 2018, 306, 96–101.
- 5 K. S. Midelfort, R. Kumar, S. Han, M. J. Karmilowicz, K. McConnell, D. K. Gehlhaar, A. Mistry, J. S. Chang, M. Anderson, A. Villalobos, J. Minshull, S. Govindarajan and J. W. Wong, *Protein Eng. Des. Sel.*, 2013, 26, 25–33.
- 6 L. Skalden, C. Peters, J. Dickerhoff, A. Nobili, H. J. Joosten, K. Weisz, M. Höhne and U. T. Bornscheuer, *ChemBioChem*, 2015, 16, 1041–1045.
- 7 K. E. Cassimjee, M. S. Humble, H. Land, V. Abedib and P. Berglund, *Org. Biomol. Chem.*, 2012, 10, 5466–5470.
- 8 H. Yun, B. Y. Hwang, J. H. Lee and B. G. Kim, *Appl. Environ. Microbiol.*, 2005, 71, 4220–4224.
- 9 I. V. Pavlidis, M. S. Weiß, M. Genz, P. Spurr, S. P. Hanlon, B. Wirz, H. Iding and U. T. Bornscheuer, *Nat. Chem.*, 2016, 1–7.
- 10 U. T. Bornscheuer, G. W. Huisman, R. J. Kazlauskas, S. Lutz, J. C. Moore and K. Robins, *Nature*, 2012, 485, 185–194.
- 11 H. Renata, Z. J. Wang and F. H. Arnold, *Angew. Chem., Int. Ed.*, 2015, 54, 3351–3367.



- 12 C. K. Savile, J. M. Janey, E. C. Mundorff, J. C. Moore, S. Tam, W. R. Jarvis, J. C. Colbeck, A. Krebber, F. J. Fleitz, J. Brands, P. N. Devine, G. W. Huisman and G. J. Hughes, *Science*, 2010, **329**, 305–310.
- 13 J. Limanto, E. R. Ashley, J. Yin, G. L. Beutner, B. T. Grau, A. M. Kassim, M. M. Kim, A. Klapars, Z. Liu, H. R. Strotman and M. D. Truppo, *Org. Lett.*, 2014, **16**, 2716–2719.
- 14 A. R. Martin, R. DiSanto, I. Plotnikov, S. Kamat, D. Shonnard and S. Pannuri, *Biochem. Eng. J.*, 2007, **37**, 246–255.
- 15 M. T. Reetz, J. D. Carballeira, J. Peyralans, H. Höbenreich, A. Maichele and A. Vogel, *Chem. – Eur. J.*, 2006, **12**, 6031–6038.
- 16 H. Xiao, Z. Bao and H. Zhao, *Ind. Eng. Chem. Res.*, 2015, **54**, 4011–4020.
- 17 R. Obexer, A. Godina, X. Garrabou, P. R. E. Mittl, D. Baker, A. D. Griffiths and D. Hilvert, *Nat. Chem.*, 2017, **9**, 50–56.
- 18 C. K. Longwell, L. Labanieh and J. R. Cochran, *Curr. Opin. Biotechnol.*, 2017, **48**, 196–202.
- 19 C. Yan, F. Parmeggiani, E. A. Jones, E. Claude, S. A. Hussain, N. J. Turner, S. L. Flitsch and P. E. Barran, *J. Am. Chem. Soc.*, 2017, **139**, 1408–1411.
- 20 F. H. Arnold and G. Georgiou, *Directed Enzyme Evolution*, Humana Press, New Jersey, 2003, vol. 230.
- 21 I. Rowles, K. J. Malone, L. L. Etchells, S. C. Willies and N. J. Turner, *ChemCatChem*, 2012, **4**, 1259–1261.
- 22 G. Matcham, M. Bhatia, W. Lang, C. Lewis, R. Nelson, A. Wang and W. Wu, *Chim. Int. J. Chem.*, 1999, **53**, 584–589.
- 23 M. S. Weiß, I. V. Pavlidis, P. Spurr, S. P. Hanlon, B. Wirz, H. Iding and U. T. Bornscheuer, *Org. Biomol. Chem.*, 2016, **14**, 10249–10254.
- 24 M. S. Weiß, I. V. Pavlidis, C. Vickers, M. Hohne and U. T. Bornscheuer, *Anal. Chem.*, 2014, **86**, 11847–11853.
- 25 S. C. Willies, J. L. Galman, I. Slabu and N. J. Turner, *Philos. Trans. R. Soc. A Math. Phys. Eng. Sci.*, 2016, **374**, 20150084.
- 26 A. P. Green, N. J. Turner and E. O'Reilly, *Angew. Chem., Int. Ed.*, 2014, **53**, 10714–10717.
- 27 D. Baud, N. Ladkau, T. S. Moody, J. M. Ward and H. C. Hailes, *Chem. Commun.*, 2015, **51**, 17225–17228.
- 28 L. Cerioli, M. Planchestainer, J. Cassidy, D. Tessaro and F. Paradisi, *J. Mol. Catal. B Enzym.*, 2015, **120**, 141–150.
- 29 A. Gomm, W. Lewis, A. P. Green and E. O'Reilly, *Chem. – Eur. J.*, 2016, **22**, 12692–12695.
- 30 X. Chen, J. Adrian, T. Cushing, H. DiMaio, L. Liang, V. Mayorga, S. Miao, M. G. Peterson, J. P. Powers, F. Spector, C. Stein, M. Wright, D. Xu, Q. Ye and J. Jaen, *Bioorganic Med. Chem. Lett.*, 2007, **17**, 2188–2192.
- 31 D. A. Drummond, B. L. Iverson, G. Georgiou and F. H. Arnold, *J. Mol. Biol.*, 2005, **350**, 806–816.
- 32 M. Wilding, T. S. Peat, S. Kalyanamoorthy, J. Newman, C. Scott and L. S. Jermiin, *Green Chem.*, 2017, **19**, 5375–5380.
- 33 L. Holm and P. Rosenström, *Nucleic Acids Res.*, 2010, **38**, 545–549.
- 34 X. Robert and P. Gouet, *Nucleic Acids Res.*, 2014, **42**, W320–W324.
- 35 A. Nobili, Y. Tao, I. V. Pavlidis, T. Van Den Bergh, H. J. Joosten, T. Tan and U. T. Bornscheuer, *ChemBioChem*, 2015, **16**, 805–810.
- 36 E. F. Pettersen, T. D. Goddard, C. C. Huang, G. S. Couch, D. M. Greenblatt, E. C. Meng and T. E. Ferrin, *J. Comput. Chem.*, 2004, **25**, 1605–1612.
- 37 T. Aleksiev, R. Potestio, F. Pontiggia, S. Cozzini and C. Micheletti, *Bioinformatics*, 2009, **25**, 2743–2744.
- 38 B. C. Joseph, S. Pichaimuthu and S. Srimeenakshi, *J. Cell Sci. Ther.*, 2015, **6**, 221.

