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## Structural insights into the EthR–DNA interaction using native mass spectrometry†

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**EthR is a transcriptional repressor that increases *Mycobacterium tuberculosis* resistance to ethionamide. In this study, the EthR–DNA interaction has been investigated by native electrospray-ionization mass spectrometry for the first time. The results show that up to six subunits of EthR are able to bind to its operator.**

Tuberculosis (TB) is a contagious disease caused by *Mycobacterium tuberculosis* (Mtb) that exerts an enormous burden on human health and wellbeing worldwide. The World Health Organization (WHO) has estimated that in 2014, TB killed 1.5 million people, while another 9.6 million people were infected.<sup>1</sup> Progress against TB has been challenged by the rise of multidrug resistant (MDR) and extensively drug resistant (XDR) Mtb strains. 3.3% of new cases and 20% of previously treated cases have MDR-TB, and of those, an estimated 9.7% are XDR-TB.<sup>1</sup>

Ethionamide is a second-line drug used for the treatment of MDR-TB. Mechanistically, ethionamide is a prodrug that is activated *in vivo* by EthA, a flavin-containing monooxygenase enzyme in Mtb, to form an ethionamide-NAD adduct.<sup>2</sup> This adduct inhibits the 2-*trans*-enoyl reductase enzyme InhA, which in turn leads to the inhibition of the Mtb type II fatty acid synthase system (FAS II).<sup>3</sup> However, the potency of ethionamide is reduced by EthR, which is a transcriptional repressor of *ethA* expression.<sup>4</sup> This suggests that inhibitors of EthR activity could function as ethionamide boosters,<sup>5</sup> allowing for lower dosages of the drug to be used.

EthR belongs to the TetR/CamR repressor protein family, whose members show high sequence homology between their N-terminal DNA-binding domains, and is expected to bind to the DNA major groove *via* its helix-turn-helix (HTH) motif.<sup>6</sup> Using DNase footprinting assays, Baulard and co-workers showed that

EthR recognizes a 55 bp operator sequence within the *ethA-R* intergenic region.<sup>4</sup> Subsequent surface plasmon resonance (SPR) analysis suggested that up to eight units of EthR could bind cooperatively to a 62 bp sequence (DNA<sub>62</sub>) encompassing the operator site (Fig. S1, ESI†).<sup>4</sup> There are numerous X-ray crystal structures of the EthR dimer in complex with various small-molecule ligands,<sup>7,8</sup> but the structure of the EthR–DNA complex has not yet been solved.

Native electrospray ionization-mass spectroscopy (ESI-MS) is an ideal technique for the study of the interactions and stoichiometries of macromolecular complexes.<sup>9,10</sup> A large body of work has established that the native structure and composition of biomolecular complexes can be adequately maintained as solution species are transferred into the gas phase during the electrospray process. However, compared to the application of ESI-MS for multi-protein complexes, fewer studies on protein–DNA complexes have been reported.<sup>11–14</sup> Analysis of protein–DNA complexes containing large DNA sequences using positive-ion native MS is complicated by the heterogeneity of cation adduction and as well as difficulties with achieving a stable electrospray.<sup>12,14</sup> In this work, native MS was used to provide structural insights into the EthR–DNA interaction. Our results indicate that up to six subunits of EthR are able to bind to its operator.

Mass spectra of histidine-tagged EthR or EthR–DNA complexes were obtained by nano-electrospray ionization (nESI) from a hybrid quadrupole time-of-flight (qTOF) SYNAPT HDMS (Waters) instrument. The native MS spectrum of EthR alone confirmed the dimeric nature of EthR in solution, with the charge state distribution being centered around the 13+ state (Fig. 1a). The observed mass of dimeric EthR (50 475 ± 97 Da) was consistent with the theoretical mass of the dimer of the construct (50 456 Da). A small amount of monomeric EthR was also observed, at about 5% of the total protein content.

The native mass spectrum of DNA<sub>62</sub> alone showed that while DNA<sub>62</sub> existed primarily in its expected duplex state centered around the 10+ charge state, a fraction of the DNA was single-stranded (DNA<sub>ss62</sub>), which could be due to an excess of one of the two complementary oligonucleotides (Fig. 1b). DNA *m/z* signals

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**Fig. 3** Native MS showing recapitulation of the EthR<sub>6</sub>-DNA complex from two independent half-sites. MS spectra of (a) EthR and DNA<sub>L31</sub> in a 6 : 1 ratio, showing formation of mainly EthR<sub>2</sub>-DNA complex, and (b) EthR and DNA<sub>R31</sub> in a 6 : 1 ratio, showing formation of mainly EthR<sub>4</sub>-DNA complex, along with a small amount of EthR<sub>2</sub>-DNA and EthR<sub>3</sub>-DNA complexes. (c) When EthR was mixed with both DNA<sub>L31</sub> and DNA<sub>R31</sub>, a small amount of the putative EthR<sub>6</sub>-DNA<sub>L31</sub>-DNA<sub>R31</sub> complex was formed.

were formed, along with a small amount of the EthR<sub>4</sub>-DNA<sub>36</sub> complex (Fig. 2b). Furthermore, significant quantities of both unbound protein and DNA were observed, which is consistent with the greatly diminished binding capacity of EthR to DNA<sub>36</sub>.

To further investigate the EthR-DNA interaction, DNA<sub>62</sub> was split into two half-sites, the left-hand site (DNA<sub>L31</sub>) and the right-hand site (DNA<sub>R31</sub>) (Fig. S1, ESI†). Each site contained one copy of an imperfect direct repeat that was thought to be responsible for the EthR-DNA binding interaction.<sup>4</sup> When EthR (15 μM) was incubated with DNA<sub>L31</sub> (2.5 μM), only EthR<sub>2</sub>-DNA<sub>L31</sub> complexes were formed (Fig. 3a). On the other hand, mainly EthR<sub>4</sub>-DNA<sub>R31</sub> complexes were observed with DNA<sub>R31</sub>, along with a small amount of EthR<sub>2</sub>-DNA<sub>R31</sub> and EthR<sub>3</sub>-DNA<sub>R31</sub> species (Fig. 3b). As before, the significant quantities of both unbound protein and DNA that are observed suggests that the interactions between EthR and the individual half-sites are relatively weak.

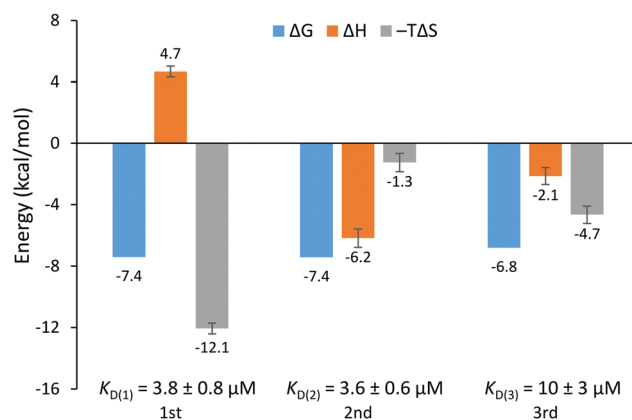
Intriguingly, when EthR was incubated with both DNA<sub>L31</sub> and DNA<sub>R31</sub> at the same time, a small amount of higher-order species could be detected that could be putatively assigned as the EthR<sub>6</sub>-DNA<sub>L31</sub>-DNA<sub>R31</sub> complex (189 872 ± 23 Da) (Fig. 3c). This complex could be formed from the association of the EthR<sub>2</sub>-DNA<sub>L31</sub> complex with the EthR<sub>4</sub>-DNA<sub>R31</sub> complex in solution. This interaction appears to be asymmetric, as homodimeric (EthR<sub>2</sub>-DNA<sub>L31</sub>)<sub>2</sub> or (EthR<sub>4</sub>-DNA<sub>R31</sub>)<sub>2</sub> species were not detected when EthR was treated with DNA<sub>L31</sub> or DNA<sub>R31</sub> separately. Additionally, DNA<sub>L31</sub> and DNA<sub>R31</sub> do not directly associate with each other in the absence of EthR (Fig. S11, ESI†), indicating that some kind of communication must exist between EthR and the DNA in order to form the putative EthR<sub>6</sub>-DNA<sub>L31</sub>-DNA<sub>R31</sub> complex.

The EthR-DNA interaction was also investigated using isothermal titration calorimetry (ITC). EthR was titrated into DNA<sub>62</sub>, generating a complex, non-sigmoidal binding isotherm

(Fig. S12, ESI†) that was analyzed using AFFINImeter software (S4SD). A stoichiometric equilibrium binding model was designed that assumed the stepwise formation of the EthR<sub>6</sub>-DNA complex from free EthR dimers and DNA<sub>62</sub>, proceeding through EthR<sub>2</sub>-DNA and EthR<sub>4</sub>-DNA intermediates (*i.e.*, EthR<sub>2</sub> + DNA<sub>62</sub> ⇌ EthR<sub>2</sub>-DNA<sub>62</sub> ⇌ EthR<sub>4</sub>-DNA<sub>62</sub> ⇌ EthR<sub>6</sub>-DNA<sub>62</sub>). Fitting of the data to this model revealed a stoichiometry of 2.9 ± 0.1 EthR dimers per duplex of DNA<sub>62</sub>, which is consistent with the native MS data. The binding affinities of the first and second dimers to DNA<sub>62</sub> were identical within experimental error ( $K_{D(1)} = 3.8 \pm 0.8 \mu\text{M}$ ,  $K_{D(2)} = 3.6 \pm 0.6 \mu\text{M}$ ). The third EthR dimer binds to DNA<sub>62</sub> with weaker affinity ( $K_{D(3)} = 10 \pm 3 \mu\text{M}$ ). The EthR-DNA<sub>62</sub> binding affinities derived from ITC in this work are about an order of magnitude weaker than those previously determined by SPR for the EthR-DNA<sub>106</sub> interaction (average  $K_D = 146 \text{ nM}$ ).<sup>4</sup>

The calculated thermodynamic parameters indicate that the binding of the first EthR dimer to DNA is entropically favorable ( $T\Delta S = +12.1 \pm 0.4 \text{ kcal mol}^{-1}$ ) but enthalpically unfavorable ( $\Delta H = +4.7 \pm 0.4 \text{ kcal mol}^{-1}$ ) (Fig. 4). The favorable increase in entropy may be due to the release of counterions or solvent molecules that were associated with the DNA.<sup>16</sup> However, the enthalpic penalty of binding stands in contrast to that observed for most major groove-binding proteins.<sup>17</sup> This suggests that significant structural rearrangement takes place in EthR and/or the DNA to accommodate binding of the first EthR dimer. Similar to what has been observed with QacR,<sup>18</sup> which binds to DNA as a dimer of dimers, this structural rearrangement could then present the DNA in a conformation that readily accepts the second dimer, thus accounting for the positive cooperativity of the EthR-DNA interaction. Indeed, binding of the second EthR dimer is driven almost entirely by enthalpy ( $\Delta H = -6.2 \pm 0.6 \text{ kcal mol}^{-1}$ ) with only a minor entropic component ( $T\Delta S = -1.3 \pm 0.6 \text{ kcal mol}^{-1}$ ). The third dimer binds *via* a combination of favorable enthalpic ( $\Delta H = -2.1 \pm 0.6 \text{ kcal mol}^{-1}$ ) and entropic ( $T\Delta S = -4.7 \pm 0.6 \text{ kcal mol}^{-1}$ ) terms.

The discrepancy in the stoichiometry of the EthR-DNA complex as determined by native MS and ITC *versus* the



**Fig. 4** Thermodynamic parameters of the EthR-DNA<sub>62</sub> interaction as determined by ITC.  $\Delta G$ ,  $\Delta H$  and  $-T\Delta S$  values are shown for the binding of the first, second and third EthR dimers with DNA. Moles of EthR are given as the dimer.



previous SPR analysis<sup>4</sup> may be a consequence of the correction that was applied to the SPR data. In that experiment, a correction factor of 0.73 was applied to the response of DNA, reflecting the different molar refractive indices of DNA *versus* proteins.<sup>19</sup> However, more recent work has shown that proteins and nucleic acids behave similarly in SPR, and so there may have been no need for this correction factor.<sup>20</sup> When this is taken into account, the previous SPR data instead suggest that each DNA<sub>62</sub> duplex binds to 6.3 EthR molecules, while each DNA<sub>37</sub> duplex binds to 3.3 EthR molecules, both numbers being consistent with the present work.

In conclusion, structural insights into the interaction between EthR and its operator have been obtained by native MS. While EthR was observed to exist as a dimer in solution as expected, the interaction of EthR with the full-length operator produced EthR<sub>6</sub>-DNA and EthR<sub>4</sub>-DNA complexes. The stoichiometry of the EthR-DNA complex was confirmed by ITC, which also revealed thermodynamic parameters that were consistent with a cooperative mode of binding. This study also highlights the capability of native MS to provide structural information on macromolecular assemblies, including where heterogeneous mixtures of complexes exist that are intractable to crystallization. Experiments are being conducted to provide structural-level detail of the EthR-DNA complex and to elucidate the precise mechanism of the EthR-DNA interaction.

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## Notes and references

- 1 World Health Organization, Global Tuberculosis Report, 2015.
- 2 T. A. Vannelli, A. Dykman and P. R. Ortiz de Montellano, *J. Biol. Chem.*, 2002, **277**, 12824.
- 3 X. Duan, X. Xiang and J. Xie, *FEMS Microbiol. Lett.*, 2014, **360**, 87.
- 4 J. Engohang-Ndong, D. Baillat, M. Aumercier, F. Bellefontaine, G. S. Besra, C. Locht and A. R. Baulard, *Mol. Microbiol.*, 2004, **51**, 175.
- 5 N. Willand, B. Dirie, X. Carette, P. Bifani, A. Singhal, M. Desroses, F. Leroux, E. Willery, V. Mathys, R. Deprez-Poulain, G. Delcroix, F. Frenois, M. Aumercier, C. Locht, V. Villeret, B. Deprez and A. R. Baulard, *Nat. Med.*, 2009, **15**, 537.
- 6 J. L. Ramos, M. Martínez-Bueno, A. J. Molina-Henares, W. Terán, K. Watanabe, X. Zhang, M. T. Gallegos, R. Brennan and R. Tobes, *Microbiol. Mol. Biol. Rev.*, 2005, **69**, 326.
- 7 F. Frénois, J. Engohang-Ndong, C. Locht, A. R. Baulard and V. Villeret, *Mol. Cell*, 2004, **16**, 301.
- 8 L. G. Dover, P. E. Corsino, I. R. Daniels, S. L. Cocklin, V. Tatituri, G. S. Besra and K. Fütterer, *J. Mol. Biol.*, 2004, **340**, 1095.
- 9 H. Hernandez and C. V. Robinson, *Nat. Protoc.*, 2007, **2**, 715.
- 10 G. R. Hilton and J. L. P. Benesch, *J. R. Soc., Interface*, 2012, **9**, 801.
- 11 A. Butterer, C. Pernstich, R. M. Smith, F. Sobott, M. D. Szczelkun and J. Tóth, *Nucleic Acids Res.*, 2014, **42**, 5139.
- 12 A. Politis, A. Y. Park, Z. Hall, B. T. Ruotolo and C. V. Robinson, *J. Mol. Biol.*, 2013, **425**, 4790.
- 13 N.-T. Nguyen-Huynh, J. Osz, C. Peluso-Iltis, N. Rochel, N. Potier and E. Leize-Wagner, *Biophys. Chem.*, 2016, **210**, 2.
- 14 V. Gabelica, C. Vreuls, P. Filée, V. Duval, B. Joris and E. D. Pauw, *Rapid Commun. Mass Spectrom.*, 2002, **16**, 1723.
- 15 F. Rosu, S. Pirotte, E. D. Pauw and V. Gabelica, *Int. J. Mass Spectrom.*, 2006, **253**, 156.
- 16 D. C. Mikles, V. Bhat, B. J. Schuchardt, C. B. McDonald and A. Farooq, *J. Mol. Recognit.*, 2014, **27**, 82.
- 17 P. L. Privalov, A. I. Dragan, C. Crane-Robinson, K. J. Breslauer, D. P. Remeta and C. A. S. A. Minetti, *J. Mol. Biol.*, 2007, **365**, 1.
- 18 M. A. Schumacher, M. C. Miller, S. Grkovic, M. H. Brown, R. A. Skurray and R. G. Brennan, *EMBO J.*, 2002, **21**, 1210.
- 19 C. Speck, C. Weigel and W. Messer, *EMBO J.*, 1999, **18**, 6169.
- 20 C. Di Primo and I. Lebars, *Anal. Biochem.*, 2007, **368**, 148.

