ChemComm

COMMUNICATION

Check for updates

Cite this: Chem. Commun., 2022, 58, 9377

Received 15th June 2022, Accepted 21st July 2022

DOI: 10.1039/d2cc03347j

rsc.li/chemcomm

Two boron atoms *versus* one: high-performance deep-blue multi-resonance thermally activated delayed fluorescence emitters[†]‡

Chin-Yiu Chan,§*^a Subeesh Madayanad Suresh,§^b Yi-Ting Lee,§^a Youichi Tsuchiya, ^b^a Tomas Matulaitis,^b David Hall,^{bc} Alexandra M. Z. Slawin, ^b^b Stuart Warriner,^d David Beljonne, ^b^c Yoann Olivier, ^b^e Chihaya Adachi ^{*af} and Eli Zysman-Colman ^{*b}

Two new deep-blue narrowband multi-resonant emitters, 1B-DTACrs and 2B-DTACrs, one of which shows thermally activated delayed fluorescence (TADF), based on boron, nitrogen, and oxygen doped nanographenes are reported. Devices based on 2B-DTACrs showed an EQE_{max} of 14.8% and CIE coordinates of (0.150, 0.044), which are very close to the BT.2020 requirement for blue pixels.

Thermally activated delayed fluorescence (TADF) materials can achieve 100% internal quantum efficiency (IQE) by harvesting both singlet and triplet excitons. In 2012, Adachi and coworkers first reported organic light-emitting diodes (OLEDs) that showed IQEs of 100% based on organic donor-acceptor (D–A) type TADF emitters.¹ Successively, a wide variety of D–A type TADF emitters have been developed, resulting in devices showing high external quantum efficiencies (EQEs) of over 20% for blue, green, and red emission colors.^{2–6} Nonetheless, D–A type TADF emitters usually possess broad emission bands that result in devices with poor color purity pixels. To achieve the BT.2020 standard for high-definition displays, narrowband

^a Center for Organic Photonics and Electronics Research (OPERA), Kyushu University, Motooka, Nishi, Fukuoka 819-0395, Japan.

- ^b Organic Semiconductor Centre, EaStCHEM School of Chemistry, University of St Andrews, St Andrews, KY16 9ST, UK. E-mail: eli.zysman-colman@st-andrews.ac.uk
- ^c Laboratory for Chemistry of Novel Materials, University of Mons, 7000, Mons, Belgium

electroluminescence (EL) in devices that show high EQE is highly desired.

Beyond the D-A type emitters, in 2016, Hatakeyama and coworkers first reported pure-blue boron-containing multi-resonance TADF (MR-TADF) emitters that feature narrowband emissions and high photoluminescence quantum yields (Φ_{PL} , PLQYs).⁷ The narrowband emission behaviour of MR-TADF emitters is the most suitable for high-definition display with a wide colour gamut, i.e., BT.2020 standard. Since then, the development of MR-TADF emitters has attracted much attention.8-10 To date, there have been numerous reports on high-efficiency OLEDs based on MR-TADF emitters with blue, green, and red emissions.8-13 Nonetheless, there are a limited number of reports on high-efficiency MR-TADF emitters that can fulfil the requirement of the colour gamut for the blue pixel, i.e., Commission Internationale de lÉclairage coordinates (CIE_{x,y}) = (0.131, 0.046).¹⁴ As a result, it is necessary to develop high-efficiency deep-blue MR-TADF emitters that satisfy the colour gamut requirement of the BT.2020 standard, *i.e.*, $CIE_v \leq 0.05$.

Here, we report two boron-based deep-blue emitters containing a mixed O/N donor atom system, one of which is MR-TADF. 1B-DTACrs and 2B-DTACrs were synthesized in five steps starting from 3,5-difluorobromobenzene. The key borylation reaction proceeded in 62% and 58% yield, respectively, for 1B-DTACrs and 2B-DTACrs from the common intermediate 4. Site selective borylation was achieved by controlling the reaction temperature and the amount of BBr₃ in the reaction (Fig. 1). The identity and purity of both emitters were confirmed by a combination of ¹H NMR (Fig. S13 and S17, ESI[†]), ¹³C NMR (Fig. S14 and S18, ESI[†]) spectroscopy, melting point determination, high-resolution mass spectrometry (Fig. S15 and S19, ESI[†]), single crystal X-ray diffraction (Fig. 1), and elemental analysis (Fig. S16 and S20, ESI[†]). 1B-DTACrs and 2B-DTACrs both display deep-blue narrowband emission at 440 and 447 nm and FWHMs of 30 and 26 nm, respectively, in the device. However, 1B-DTACrs is TADF-inactive while 2B-DTACrs is TADF-active. Devices based on 1B-DTACrs and 2B-DTACrs

View Article Online

 $^{{\}it E-mail: chinyiu.chan} @opera.kyushu-u.ac.jp \\$

^d School of Chemistry, University of Leeds, Woodhouse Lane, Leeds, UK

^e Laboratory for Computational Modeling of Functional Materials, Namur Institute of Structured Matter, Université de Namur, Rue de Bruxelles, 61, B-5000 Namur, Belgium

^f International Institute for Carbon Neutral Energy Research (I2CNER), Kyushu University, 744 Motooka, Nishi, Fukuoka 819-0395, Japan.

E-mail: adachi@cstf.kyushu-u.ac.jp

[†] Electronic supplementary information (ESI) available. CCDC 2171192 and 2171191. For ESI and crystallographic data in CIF or other electronic format see DOI: https://doi.org/10.1039/d2cc03347j

The research data supporting this publication can be accessed at https://doi. org/10.17630/12747024-df19-4158-906b-1b0d26dbf5b3

[§] These three authors contributed equally to this work.

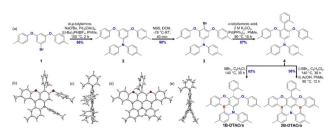


Fig. 1 (a) Synthetic route of **1B-DTACrs** and **2B-DTACrs**. ORTEP view of **1B-DTACrs** and **2B-DTACrs**; (b) plane view and (c) side view of **1B-DTACrs**, (d) plane view and (e) side view of **2B-DTACrs**. Thermal ellipsoids are displayed at 50% probability.

demonstrated maximum external quantum efficiencies (EQE_{max}) of 1.3 and 14.8%, respectively, together with corresponding CIE coordinates of (0.154, 0.049) and (0.150, 0.044), which are very close to the BT.2020 standard, *i.e.*, (0.131, 0.046).

Crystals for 1B-DTACrs were obtained by slow diffusion of hexane into a saturated solution of the emitter in THF. Crystals of 2B-DTACrs were obtained directly from a temperature gradient sublimation method (Fig. 1). The ditolylamine (DTA) unit adopts a propeller structure. In the case of 1B-DTACrs, where there is only one boron in the compound, the MR-TADF core remains nearly planar. The o-tolyl unit is approximately perpendicular to the aromatic core in both emitters. Yasuda and co-workers reported BSBS-N1,15 a structurally analogous sulfur-containing compound to 2B-DTACrs, which was described as having a highly distorted structure, whereas 2B-DTACrs possesses a nearly planar geometry at the B,Osubstituted pentacene core with an out-of-plane alignment of two aryl rings of the DTA unit.15 There are no significant intermolecular π - π interactions in the solid state for either compound, which may be a consequence of the orthogonally disposed o-tolyl unit.

The ground- and excited-state properties of 1B-DTACrs and 2B-DTACrs were modelled using DFT and spin-component scaling second order approximate coupled-cluster (SCS-CC2) methods, respectively (Fig. 2). Both compounds possess difference density distributions between the ground and excitedstate reminiscent of other MR-TADF emitters.¹⁰ Addition of the second boron atom in 2B-DTACrs resulted in a small stabilization in the HOMO and LUMO energy levels compared to those of 1B-DTACrs and a corresponding small decrease (0.12 eV) in the band gap (Fig. 2(a)). The S₁ state is stabilized from 3.38 eV in 1B-DTACrs to 3.27 eV in 2B-DTACrs, while the T_1 states are essentially isoenergetic. Thus, there is a much smaller $\Delta E_{\rm ST}$ value of 0.08 eV for 2B-DTACrs compared to 0.20 eV of **1B-DTACrs**. The oscillator strength (f) for the S₀-S1 transition is higher at 0.36 for 2B-DTACrs compared to 0.23 for 1B-DTACrs likely for the apparent largest delocalization of the S_1 excited state. Both the S_2 and T_2 states of 2B-DTACrs are also stabilized compared to those of 1B-DTACrs. The small energy difference between S_1 and T_2 (ΔE_{ST_2}) is of particular interest as many reports have highlighted that reverse intersystem crossing (RISC) occurs *via* intermediate triplet states.¹⁶ The ΔE_{ST_2} drops from 0.34 eV in **1B-DTACrs** to 0.02 eV in

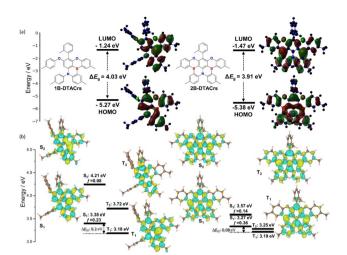


Fig. 2 (a) HOMO (bottom) and LUMO (top) electron density distributions of **1B-DTACrs** (left) and **2B-DTACrs** (right) calculated using PBE0/6-31G(d,p) (isovalue = 0.02). (b) Difference density plots for the low-lying singlet and triplet excited states for **1B-DTACrs** and **2B-DTACrs** calculated in the gas phase using SCS-CC2/pVDZ (isovalue = 0.001). Blue colour represents an area of decreased electron density, and yellow represents an increased electron density between the ground and excited states. *f* denotes the oscillator strength for the transition to the excited singlet state.

2B-DTACrs. For each compound, spin-orbital coupling (SOC) is modest between S_1 and T_1 , calculated to be 0.07 cm⁻¹ and 0.04 cm⁻¹ for **1B-DTACrs** and **2B-DTACrs**, respectively. This increases to 0.81 cm⁻¹ and 0.98 cm⁻¹ between S_1 and T_2 for **1B-DTACrs** and **2B-DTACrs**, respectively (Table S3, ESI†). These results suggest that RISC likely proceeds from T_2 to S_1 , particularly for **2B-DTACr**.¹⁷ Tables S1–S3 (ESI†) provide a full summary of the excited state energies, transitions and SOC values.

The electrochemical properties were measured using cyclic voltammetry (CV) and differential pulse voltammetry (DPV) in deaerated dichloromethane (DCM) with 0.1 M tetra-nbutylammonium hexafluorophosphate as the supporting electrolyte (Fig. S21 and Table S4, ESI†). The oxidation potentials, E^{ox} , determined from the peak of the first oxidation wave of the DPV, are 1.07 V and 1.17 V versus a saturated calomel electrode (SCE), respectively, for 1B-DTACrs and 2B-DTACrs. The corresponding reduction potentials, E^{red} , are -2.10 V and -1.89 V, respectively, versus SCE for 1B-DTACrs and 2B-DTACrs. The corresponding HOMO and LUMO levels were calculated to be -5.42 eV and -2.24 eV for 1B-DTACrs.¹⁸ The HOMO and LUMO energy levels are stabilized to -5.51 eV and -2.46 eV, respectively, for 2B-DTACrs. The electrochemical gap is thus reduced from 3.17 eV for 1B-DTACrs to 3.05 eV for 2B-DTACrs, in line with the trend predicted by the gas-phase DFT calculations.

The photophysical properties of **1B-DTACrs** and **2B-DTACrs** were firstly examined in dilute toluene solutions (10^{-5} M). The photophysical data are shown in Fig. 3, and Table 1. Both emitters showed absorption bands below 350 nm, which are assigned as locally excited (LE) π - π * transitions, while the absorption bands beyond 400 nm are assigned as short-range

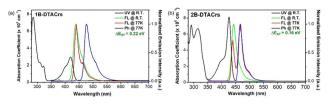


Fig. 3 Photophysical properties of (a) **1B-DTACrs** and (b) **2B-DTACrs** in toluene solution (10^{-5} M) at room temperature and 77 K (λ_{exc} = 280 nm and 320 nm for time-resolved and steady-state emission, respectively).

charge transfer (SRCT) bands. Both compounds possess narrow high-energy bands assigned to transitions to the S₁ state, with molar absorptivity coefficients, ε , of 3 and 8 × 10⁵ M⁻¹ cm⁻¹ for 1B-DTACrs and 2B-DTACrs, respectively, in line with the larger computed *f* associated with the $S_0 \rightarrow S_1$ transitions (*vide supra*). The higher calculated f value and measured ε in **2B-DTACrs** originates from the extended conjugation in the compound.¹⁹ 1B-DTACrs and 2B-DTACrs emit in the deep-blue in toluene at 438 and 443 nm, respectively. The PL spectra of 1B-DTACrs and 2B-DTACrs are narrow with full-width at half-maxima (FWHM) values of 27 and 21 nm, respectively. The narrowband emission coupled with the small Stokes-shifts reflect the rigid nature of these compounds and the small degree of reorganization in the excited state. The $\Phi_{\rm PL}$ in degassed toluene solutions of 1B-DTACrs is 99%, while 2B-DTACrs showed a rather lower $\Phi_{\rm PL}$ of 62%. When doped in a 3,3'-di(9H-carbazol-9-yl)-1,1'biphenyl (mCBP) host at 5 wt% of 1B-DTACrs or 2B-DTACrs, the $\Phi_{\rm PL}$ values remain high at 83 and 74%, respectively. Both doped films display emissions (Fig. 4) solely originated from the corresponding emitter, which indicate an efficient energy transfer from mCBP to the emitter. The transient decay profiles of both doped films reveal that 1B-DTACrs showed no TADF behaviour. By contrast, 2B-DTACrs showed TADF behaviour with a delayed lifetime (τ_d) of 13.1 µs. The TADF behaviour of 2B-DTACrs was further confirmed by temperature-dependent transient emission decay measurement (Fig. S22, ESI[†]). The $\Delta E_{\rm ST}$ values of the mCBP doped films were determined from the difference between the onsets of the fluorescence and phosphorescence spectra at 77 K (Fig. S23, ESI†). The fairly large $\Delta E_{\rm ST}$ in **1B-DTACrs-**doped film (0.21 eV) may account for the non-TADF behaviour while that of 2B-DTACrs is smaller at 0.16 eV. The rate constants of doped films of 1B-DTACrs and 2B-DTACrs were calculated, and the data are summarized in Table 1. The reverse intersystem crossing rate (k_{RISC}) of **2B-DTACrs** was found to be $1.3 \times 10^5 \text{ s}^{-1}$, which is amongst

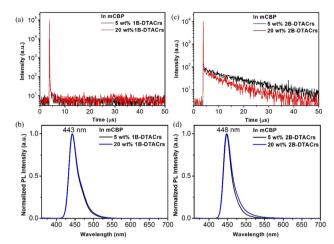


Fig. 4 Decay profiles and emission spectra of (a) and (b) **1B-DTACrs** and (c) and (d) **2B-DTACrs** at different doping concentrations in mCBP (λ_{exc} = 280 nm and 320 nm for time-resolved and steady-state emission, respectively).

the fastest k_{RISC} for deep-blue MR-TADF emitters with $\text{CIE}_y \leq 0.10$ (Table S5, ESI[†]).

next fabricated vacuum-deposited OLEDs We using 1B-DTACrs and 2B-DTACrs as the emitters. The following device configuration was used: indium-tin-oxide (ITO)-coated glass (100 nm)/HAT-CN (10 nm)/TAPC (30 nm)/mCP (10 nm)/mCP: 5 wt% of 1B-DTACrs or 2B-DTACrs (20 nm)/DPEPO (10 nm)/ TmPyPB (30 nm)/Liq (2 nm)/Al (100 nm). 1,4,5,8,9,11-Hexaazatriphenyl-enehexacarbonitrile (HAT-CN) is the holeinjection layer, 1,1-bis[(di-4-tolylamino)phenyl]cyclohexane (TAPC) is the hole-transporting layer, 1,3-bis(N-carbazolyl)benzene (mCP) is used for electron-blocking and host layers, bis[2-(diphenylphosphino)phenyl]ether oxide (DPEPO) is the hole-blocking layer, 1,3,5-tris(3-pyridyl-3-phenyl)benzene (TmPyPB) is the electrontransporting layer, and 8-hydroxyquinolinolato-lithium (Liq) and Al are the electron injection and cathode layers, respectively. Fig. 5, Fig. S24-S27 (ESI⁺) and Table 2 depict all the device characteristics. Devices based on 1B-DTACrs (Device I) and 2B-DTACrs (Device II) produced deep-blue narrowband emission at 440 and 447 nm, respectively, which are consistent to their corresponding PL spectra. The EL spectra clearly confirm the complete exciton confinement in both devices. The FWHM of Device I is 30 nm, which is slightly wider than that of Device II (FWHM = 26 nm). The corresponding $CIE_{x,y}$ are (0.154, 0.049) and (0.150, 0.044), respectively. Although the color gamuts of both devices are very close to the BT.2020 standard for the blue pixel, *i.e.*, (0.131, 0.046), the

Table 1	Photophysical	data of	1B-DTACrs	and 2B-DTACrs
---------	---------------	---------	-----------	---------------

	In toluene			5 wt% in mCBP														
Material	$\lambda_{abs} \ (nm)$		FWHM (nm)	$\Phi_{\mathrm{Air}}{}^{a}$ (%)	$ \Phi_{ m Ar}^{ b} \ (\%) $	$\Delta E_{\rm ST}$ (eV)	λ_{\max} (nm)	FWHM (nm)	$\Delta E_{\rm ST}$ (eV)	$ \Phi_{ m Ar} \ (\%) $	${\Phi_{ m p}}^c_{ m (\%)}$	$\Phi_{\mathrm{d}}{}^{d}$ (%)	$\stackrel{ au_{p}}{(ns)}$	$\stackrel{ au_d}{(\mu s)}$	$k_{ m r} \ (10^8 \ { m s}^{-1})$	$k_{ m nr} \ (10^7 \ { m s}^{-1})$	$k_{\rm ISC} \ (10^8 \ { m s}^{-1})$	$k_{ m RISC} \ (10^5 { m s}^{-1})$
1B-DTACrs 2B-DTACrs		438 443	27 21	68 20	99 64	0.22 0.16	443 448	28 24	0.21 0.16	83 74	35	 39	10 2		0.8 1.7	1.7 6.1	0.2 3.3	 1.3

^a PLQY in aerated toluene. ^b PLQY in argon-saturated toluene. ^c Prompt intensity. ^d Delayed intensity.

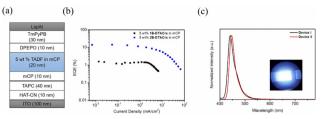


Fig. 5 Device performance of TADF OLEDs. (a) Device structure; (b) EQE versus current density; (c) EL spectra of devices at 10 cd m^{-2} (inset: photo of Device II).

 Table 2
 Device performance of deep-blue TADF OLEDs

Device	Dopant	$V_{\rm on}^{\ a}$ (V)	EQE _{max} ^b (%)		FWHM ^c (nm)	$CIE^c (x, y)$
I	1B-DTACrs		1.31	440	30	(0.154, 0.049)
II	2B-DTACrs		14.8	447	26	(0.150, 0.044)

 a Voltage at 1 cd m $^{-2}.$ b Maximum external quantum efficiency. c Value at 10 cd m $^{-2}.$

efficiencies of two devices varied significantly. Taking advantage of enhanced exciton harvesting ability of **2B-DTACrs**, Device **II** demonstrated an excellent EQE_{max} of 14.8%, which is one of the highest performing deep-blue MR-TADF OLEDs with $\text{CIE}_y \leq 0.05$ (Table S5, ESI†);¹⁴ there is, however, considerable efficiency roll-off beyond 100 cd m⁻². By contrast, Device **I** only showed a poor EQE_{max} of 1.3%, which is in part due to the fact that only singlet excitons can be utilized. Additionally, the poor carrier transporting properties in device **I**, in which a high turn-on voltage was obtained, may contribute to the lower EQE than the theoretical value (5%).

In conclusion, we presented two deep-blue p,n-doped nanographene emitters (**1B-DTACrs** and **2B-DTACrs**), in which **2B-DTACrs** with two boron atoms is TADF-active while **1B-DTACrs** with only one boron atom is TADF inactive. It is noteworthy that the emitter **2B-DTACrs** does not show redshifted emission. Devices based on **2B-DTACrs** achieved a high EQE_{max} of 14.8% and $CIE_{x,y}$ of (0.150, 0.044), which are one of highest efficiency deep-blue MR-TADF devices reported to date. Importantly, the CIE coordinates effectively meet the colour gamut requirement for the BT.2020 standard for the blue pixel.

The authors acknowledge Ms N. Nakamura and Ms K. Kusuhara for their technical assistance with this research. This work was supported financially by the JSPS Core-to-Core Program (grant number: JPJSCCA20180005) and Kyulux Inc. S. M. S. acknowledges support from the Marie Skłodowska-Curie Individual Fellowship (grant agreement No 838885 NarrowbandSSL). We would like to thank the Leverhulme Trust (RPG-2016-047) for financial support. E. Z.-C. is a Royal Society Leverhulme Trust Senior Research fellow (SRF\R1\201089). We thank the EPRSC for funding (EP/R035164/1). Computational resources have been provided by the Consortium des Équipements de Calcul Intensif (CÉCI), funded by the Fonds de la Recherche Scientifiques de Belgique (F. R. S.-FNRS) under Grant No. 2.5020.11, as well as the Tier-1 supercomputer of the Fédération Wallonie-Bruxelles, infrastructure funded by the Walloon Region under the grant agreement n1117545. Y. O. acknowledges funding by the Fonds de la Recherche Scientifique-FNRS under Grant no. F.4534.21 (MIS-IMAGINE). D. B. is a FNRS Research Director.

Conflicts of interest

There are no conflicts to declare.

Notes and references

- 1 H. Uoyama, K. Goushi, K. Shizu, H. Nomura and C. Adachi, *Nature*, 2012, **492**, 234–238.
- 2 M. Y. Wong and E. Zysman-Colman, Adv. Mater., 2017, 29, 1605444.
- 3 D. H. Ahn, S. W. Kim, H. Lee, I. J. Ko, D. Karthik, J. Y. Lee and J. H. Kwon, *Nat. Photonics*, 2019, **13**, 540–546.
- 4 H. Kaji, H. Suzuki, T. Fukushima, K. Shizu, K. Suzuki, S. Kubo, T. Komino, H. Oiwa, F. Suzuki, A. Wakamiya, Y. Murata and C. Adachi, *Nat. Commun.*, 2015, **6**, 8476.
- 5 T.-L. Wu, M.-J. Huang, C.-C. Lin, P.-Y. Huang, T.-Y. Chou, R.-W. Chen-Cheng, H.-W. Lin, R.-S. Liu and C.-H. Cheng, *Nat. Photonics*, 2018, **12**, 235–240.
- 6 Y.-L. Zhang, Q. Ran, Q. Wang, Y. Liu, C. Hänisch, S. Reineke, J. Fan and L.-S. Liao, *Adv. Mater.*, 2019, **31**, 1902368.
- 7 T. Hatakeyama, K. Shiren, K. Nakajima, S. Nomura, S. Nakatsuka, K. Kinoshita, J. Ni, Y. Ono and T. Ikuta, *Adv. Mater.*, 2016, 28, 2777–2781.
- 8 J.-M. Teng, Y.-F. Wang and C.-F. Chen, J. Mater. Chem. C, 2020, 8, 11340–11353.
- 9 H. Lee, D. Karthik, R. Lampande, J. H. Ryu and J. H. Kwon, *Front. Chem.*, 2020, **8**, 373.
- 10 S. Madayanad Suresh, D. Hall, D. Beljonne, Y. Olivier and E. Zysman-Colman, *Adv. Funct. Mater.*, 2020, **30**, 1908677.
- 11 N. Ikeda, S. Oda, R. Matsumoto, M. Yoshioka, D. Fukushima, K. Yoshiura, N. Yasuda and T. Hatakeyama, *Adv. Mater.*, 2020, 32, 2004072.
- 12 M. Yang, I. S. Park and T. Yasuda, J. Am. Chem. Soc., 2020, 142, 19468-19472.
- 13 Y. Zhang, D. Zhang, T. Huang, A. J. Gillett, Y. Liu, D. Hu, L. Cui, Z. Bin, G. Li, J. Wei and L. Duan, *Angew. Chem., Int. Ed.*, 2021, 60, 20498–20503.
- 14 I. S. Park, M. Yang, H. Shibata, N. Amanokura and T. Yasuda, *Adv. Mater.*, 2022, 34, 2107951.
- 15 M. Nagata, H. Min, E. Watanabe, H. Fukumoto, Y. Mizuhata, N. Tokitoh, T. Agou and T. Yasuda, *Angew. Chem., Int. Ed.*, 2021, 60, 20280–20285.
- 16 I. Lyskov and C. M. Marian, J. Phys. Chem. C, 2017, 121, 21145–21153.
- 17 S. M. Pratik, V. Coropceanu and J.-L. Brédas, *ACS Mater. Lett.*, 2022, 4, 440–447.
- 18 N. G. Connelly and W. E. Geiger, Chem. Rev., 1996, 96, 877-910.
- A. Pershin, D. Hall, V. Lemaur, J.-C. Sancho-Garcia, L. Muccioli, E. Zysman-Colman, D. Beljonne and Y. Olivier, *Nat. Commun.*, 2019, 10, 597.