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Optical excitations of graphene-like materials: group III-nitrides

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By using first-principles calculations, we have studied the electronic and optical characteristics of group III-nitrides, such as BN, AlN, GaN, and InN monolayers. The optimized geometry, quasi-particle energy spectra, charge density distributions, band-decomposed charge densities, and Van Hove singularities in density of states are described in the work using physical and chemical pictures and orbital hybridizations found in B–N, Al–N, Ga–N, and In–N chemical bonds. Moreover, the dielectric functions, energy loss functions, absorption coefficients, and reflectance spectra with electron–hole interactions of optical properties are successfully achieved. More importantly, the close relations between electronic and optical properties are successfully demonstrated. The theoretical framework will be useful to research other graphene-like materials.

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1 Introduction

Nowadays, two-dimensional (2D) materials have recently gained much attention from researchers owing to their unique characteristics and practical applications.^{1–3} Since the first successful production in 2004,⁴ graphene has attracted much observation due to the distinctive characteristics of its honeycomb structure, a single-atom layer thick 2D system made up of sp²-bonding carbon atoms.⁵ Up to now, considerable efforts have been focused on developing other 2D graphene-like systems, such as transition metal dichalcogenides,^{6,7} hexagonal boron-nitride,⁸ post-transition monochalcogenides,⁹ group III–V monolayers,^{10,11} and group IV–IV monolayers.^{12,13} Among them, graphene-like structures, such as BN, AlN, GaN, and InN monolayers which belong to group III–V materials display diverse chemical and physical properties^{14–17} and provide broad applications in biomedical,¹⁸ electronic,¹⁹ composite, environmental,²⁰ optoelectronic,²¹ and “green” energy-related fields.²²

Group III–V two-dimensional materials have successfully studied both experimentation and theory measurements in terms of experimentation and theory.^{17,23–29} Taking the BN monolayer as an example, several hexagonal-BN layers have been fabricated on a large scale *via* the exfoliation method using benzyl benzoate as the milling agent.³⁰ A two-dimensional BN monolayer is produced by the liquid exfoliation of layered materials.³¹ These properties make the BN monolayer an ideal

material for a variety of applications. In terms of electronic and optical properties, the BN monolayer has a wide bandgap of around 6.86 eV and a huge optical gap of 6.0 eV,^{10,32} which makes it an excellent insulator.³³ P. Tsipas and his co-authors synthesized an AlN monolayer by using plasma-assisted molecular beam epitaxy (MBE) that grows epitaxially on Ag(111).³⁴ Besides, the AlN monolayer has been a wide gap semiconductor of 2.93 eV and 5.36 eV by using density functional theory (DFT) and many-body perturbation correction (GW), respectively.¹⁷ On the other hand, the optical absorption of the AlN monolayer with the Bethe–Salpeter equation (BSE) suggests that the first absorption peak is located at 4.02 eV and a binding energy of the first exciton of 2.05 eV is achieved. For the GaN monolayer, theoretical investigations also suggest that it can form a stable nanostructure with an indirect band gap measuring 1.95 eV³⁵ and 4.44 eV³⁶ by using DFT and GW levels, respectively. Recently, significant strides have been made in the MBE method of producing indium nitride (InN) films, resulting in notable breakthroughs.^{37–40} Also, experimental methods have proven successful in synthesizing a wide range of nanostructured InN materials.^{41–43} Furthermore, the InN monolayer has been extensively studied using theoretical methods, which have confirmed its outstanding electronic properties. These investigations have revealed that the InN monolayer possesses a considerable band gap, specifically 0.56 eV and 1.70 eV, respectively, according to DFT and GW calculations.^{44,45} This significant band gap further emphasizes the potential of the InN monolayer for a wide range of electronic applications.

Despite these graphene-like monolayers having been considered in previous studies, accurate calculations of their electronic and optical properties are still lacking, which should be addressed to examine their expectations in electronic and optoelectronic applications. Furthermore, the relationship

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between both the initial and final states of orbital hybridizations, as well as the significant excitation peaks observed in the optical spectra, which could be responsible for emphasizing specific optical characteristics, are yet to be investigated in these materials.

In this study, we investigated the 2D group-III nitrides: BN, AlN, GaN, and InN relying on data from first-principles calculations in both DFT and GW/GW + BSE standards of the electronic and optical properties. The delicate calculations and analysis include the optimal geometric, quasi-particle band structure, the atom-dominant energy spectrum in various energy ranges, the

charge density distributions, band-decomposed charge densities, Van Hove singularities in the density of states, the optical properties with the entire energy spectrum, a lot of prominent absorption structures, the electron loss functions, reflectance, and absorbance spectra. More importantly, the electronic properties and the optical responses are connected through orbital hybridizations in chemical bonds. The rich and unique electronic and optical properties of graphene-like materials can be utilized for electronic and optoelectronic applications. The developed theoretical framework is useful for other graphene-like and emerging materials.

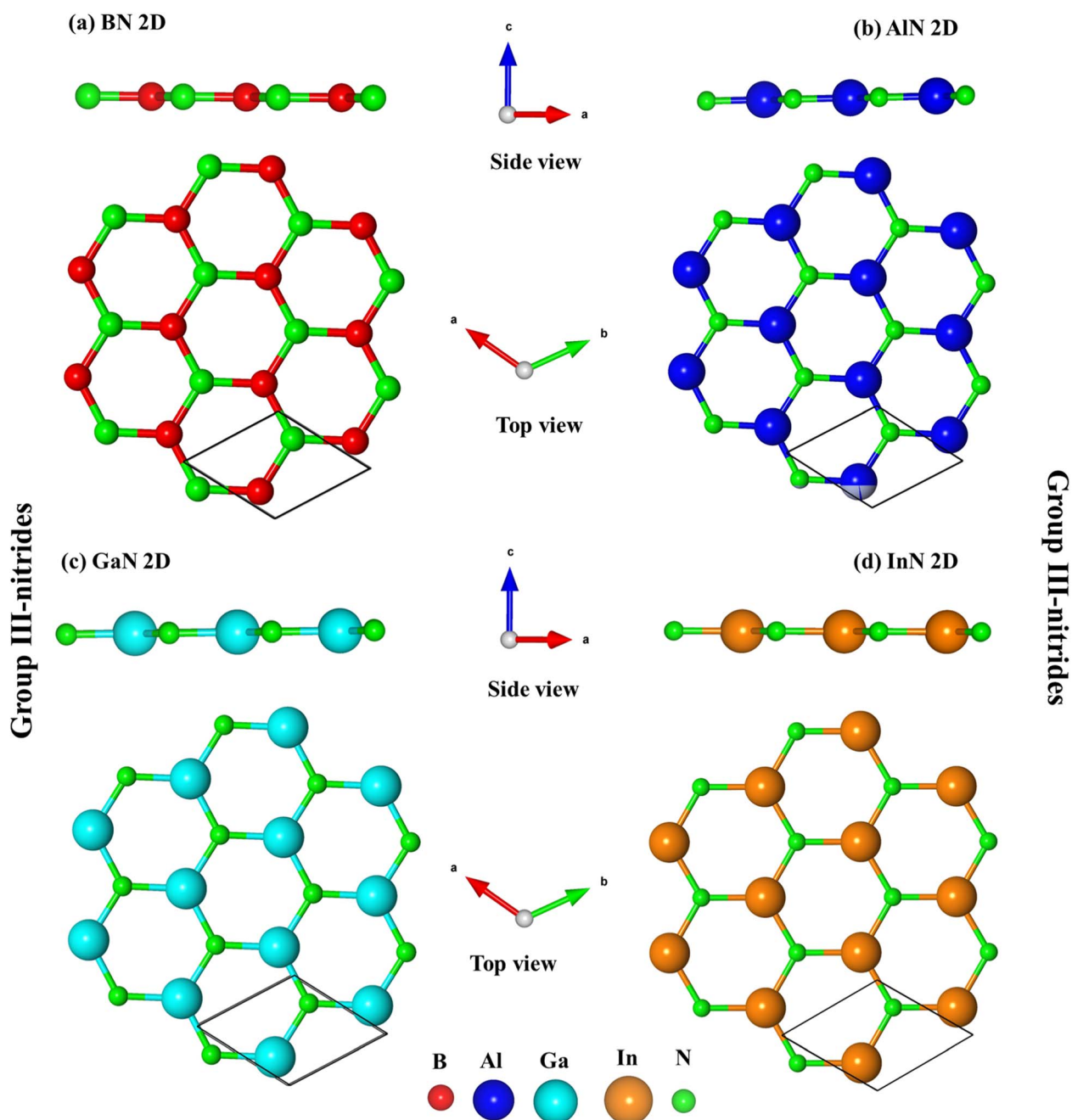


Fig. 1 (a–d) Side view and top view for BN, AlN, GaN, and InN monolayers, respectively. The solid black is a unit cell with two atoms. The red, blue, cyan, yellow, and green balls represent by B, Al, Ga, In, and N atoms.



2 Calculation methods

The density functional theory⁴⁶ via the Vienna *Ab initio* Simulation Package (VASP)⁴⁷ is utilized to investigate the optimal geometric structure and electronic and optical properties of 2D group III-nitride materials. The Perdew–Burke–Ernzerhof generalized gradient approximation was used for the exchange–correlation functions.^{48–51} The projector-augmented wave method⁵² was utilized for the characterization of the ion and valence electron interactions. The unit cells of these monolayers are built with a vacuum region larger than 16 Å to avoid spurious interactions. The cutoff energy for the expansion of the plane wave basis was set to 500 eV. The Brillouin zone was integrated with a special *k*-point mesh of 200 × 200 × 1 in the *Γ*-centered sampling technique for structural optimization.⁵³ The convergence condition of the ground state is set to be 10^{−8} eV between two consecutive simulation steps, and all atoms could fully relax during the geometric optimization until the Hellmann–Feynman force acting on each atom was smaller than 0.01 eV. The Heyd–Scuseria–Ernzerhof hybrid-function (HSE06) method⁵⁴ is added to get more band gaps. To evaluate the dynamic stability of group III–V systems, phonon energy dispersions were calculated within the harmonic approximations by using density functional perturbation theory as implemented in the PHONOPY code.⁵⁵ In these calculations, at least five neighboring interactions (corresponding to a 5 × 5 × 1 supercell) have been taken into account to get a reliable phonon band structure.

On top of DFT wave functions, the quasi-particle energies are obtained within approximation for the self-energy, the cutoff energy for the response function was set to 300 eV, and 60 × 60 × 1 *Γ*-centered point sampling was used to represent reciprocal space. The Wannier interpolation procedure performed in the WANNIER90 code⁵⁶ was used to plot the accurate GW standard. The single particle excitation spectra, which are related to the imaginary part of the dielectric functions $\varepsilon_2(\omega)$, could be described by the following Fermi's golden rule:⁵⁷

$$\varepsilon_2(\omega) = \frac{8\pi^2 e^2}{\omega^2} \sum_{\mathbf{v}\mathbf{k}} |e\langle \mathbf{v}\mathbf{k} | \mathbf{v} | \mathbf{c}\mathbf{k} \rangle|^2 \delta(\omega - (E_{\mathbf{c}\mathbf{k}} - E_{\mathbf{v}\mathbf{k}})) (f(E^{\mathbf{v}}(\mathbf{k})) - f(E^{\mathbf{c}}(\mathbf{k}))) \quad (1)$$

in which the first term $|e\langle \mathbf{v}\mathbf{k} | \mathbf{v} | \mathbf{c}\mathbf{k} \rangle|^2$ is the square of the electric moment, which is responsible for the strength of the excitation peaks, the second-term $\delta(\omega - (E_{\mathbf{c}\mathbf{k}} - E_{\mathbf{v}\mathbf{k}}))$ is the joined of the density of states, which corresponds to the available excitation transition channels, and the final term $(f(E^{\mathbf{v}}(\mathbf{k})) - f(E^{\mathbf{c}}(\mathbf{k})))$ is responsible for the excitations and de-excitations of electrons.

In addition to the single-particle excitations, the presence of stable excitons may strongly affect the optical properties. To evaluate this bound state of the electron–hole pair, we solve the standard BSE.⁵⁸

$$\varepsilon_2(\omega) = \frac{8\pi^2 e^2}{\omega^2} \sum_{\mathbf{v}\mathbf{k}} |e\langle 0 | \mathbf{v} | S \rangle|^2 \delta(\omega - \Omega_S) \quad (2)$$

Note that the *k*-point sampling, energy cutoff, and number of bands were set to the same values as in the GW level. The close

connection of the initial and the final orbital hybridizations with the prominence of optical excitations, and the effects of the electron–hole couplings on the optical properties are the main study foci of the current work. Other optical properties, such as the energy loss functions,⁵⁹ absorption coefficient,⁶⁰ reflectivity,⁶¹ refractive index, and extinction coefficient can be obtained from the dielectric function using the following equations:

$$L(\omega) = \frac{\varepsilon_2(\omega)}{\varepsilon_1^2(\omega) + \varepsilon_2^2(\omega)} \quad (3)$$

$$R(\omega) = \left| \frac{\sqrt{\varepsilon_1(\omega) + \varepsilon_2(\omega)} - 1}{\sqrt{\varepsilon_1(\omega) + \varepsilon_2(\omega)} + 1} \right|^2 \quad (4)$$

$$\alpha(\omega) = \sqrt{2\omega} \left[\sqrt{\varepsilon_1^2(\omega) + \varepsilon_2^2(\omega)} - \varepsilon_1(\omega) \right]^{1/2} \quad (5)$$

3 Results and discussion

Fig. 1a–d indicate the optimal structures of 2D BN, AlN, GaN, and InN monolayers. The optimized lattice constants are equal to 2.51 Å, 3.13 Å, 3.25 Å, and 3.64 Å, respectively (Table 1). These values are suitable for previous studies.^{34,62} Similar to that of graphene, the unit cell of these materials possesses two atoms; the top view presents a honeycomb lattice, while the side view exhibits a planar structure with the height difference between two sub-lattices equalling zero, thus, leading to impure sp² bonding configurations. The B–N, Al–N, Ga–N, and In–N bond lengths are equal to 1.45 Å, 1.81 Å, 1.88 Å, and 2.10 Å, respectively, which agree well with previous studies.^{10,63} Similar to that of graphene,⁶⁴ group III–V honeycomb lattices are constructed by the rather strong impure σ bonding, while the impure π bonds due to the impure p_z orbital hybridizations are rather weak and well separated from the σ ones. Even though they

Table 1 Lattice constants and energy gap values with various levels of theories of the BN, AlN, GaN, and InN monolayers, respectively

Materials	Lattice constants (Å)	Energy gap (eV)			
		DFT	DFT + SOC	HSE	GW
BN	2.51	4.22	4.46	5.71	6.44
	2.50 ^a	4.65 ^e		5.70 ^e	6.86 ^f 6.10 ^h
AlN	3.13	2.84	2.87	4.02	5.18
	3.12 ^b	2.91 ^e		4.04 ^e	5.57 ^f 5.80 ^h
GaN	3.25	1.89	1.90	3.19	4.32
	3.19 ^c	2.16 ^e		3.44 ^e	4.10 ^g 3.50 ^h
InN	3.64	0.35	0.35	1.51	1.41
	3.57 ^d	0.56 ^e		1.53 ^e	1.70 ^g

^a Ref. 68. ^b Ref. 34. ^c Ref. 69. ^d Ref. 10. ^e Ref. 44. ^f Ref. 10. ^g Ref. 45. ^h Ref. 70.





Fig. 2 Phonon dispersion curves are calculated for (a) BN, (b) AlN, (c) GaN, and (d) InN monolayers, respectively. Note that LA, TA, and ZA are longitudinal, transverse, and out-of-plane acoustic branches and LO, TO, and ZO are longitudinal, transverse, and out-of-plane optic modes.

have a similar structure, the electronic and optical properties of group III-V materials are predicted to be rather different from that of graphene due to the presence of impure orbital hybridizations, and the broken hexagon symmetry. Information about the optimal geometric properties of group III-V two-dimensional materials could be detected by high-precision measurements, *e.g.*, the side and top views of the graphene-

like materials can be tested by using tunneling electron microscopy (TEM)^{65,66} and scanning tunneling microscopy (STM),⁶⁷ respectively.

To verify the dynamic stability, the phonon energy dispersion of the studied materials has been calculated and shown in Fig. 2a-d. The phonon band structure of group III-nitride monolayers shows six phonon branches since their unit cells



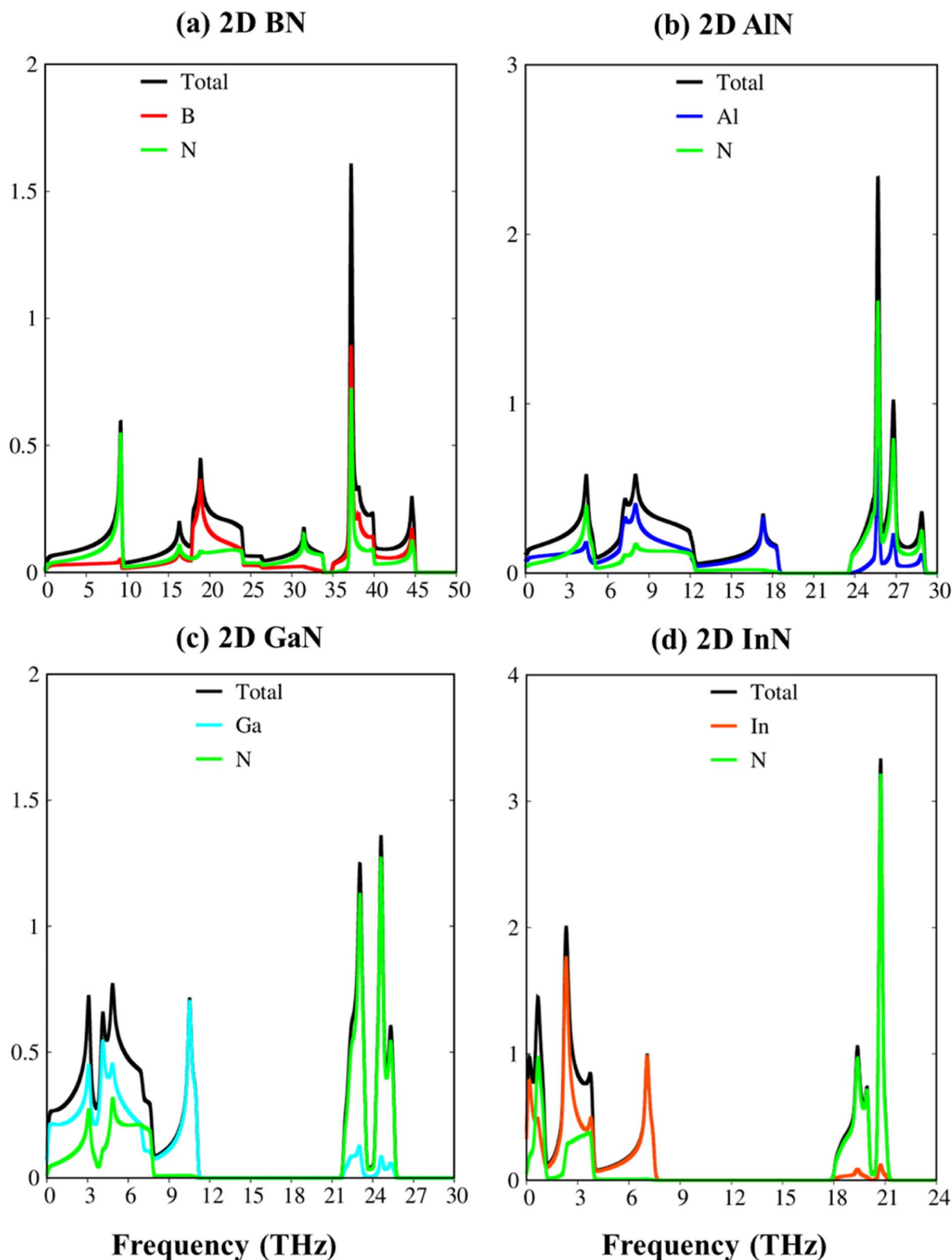


Fig. 3 The vibrational atom-dominant density of states, corresponding to the phonon bands of (a) BN, (b) AlN, (c) GaN, and (d) InN monolayers.

possess two atoms. According to the vibration of atoms along the long wavelength limit, the phonon energy sub-bands could be classified into three acoustic modes (vibrations of atoms with the

same phase) and three optic modes (vibrations of atoms with the opposite phase). The transverse acoustic (TA) and longitudinal acoustic (LA) phonon branches of BN, AlN, GaN, and InN





Fig. 4 (a–d) Band structures of BN, AlN, GaN, and InN monolayers, corresponding to DFT and GW standards, respectively, denoted by dashed and solid curves. Noted: dashed red circles denote indirect gap semiconductors. Green and purple arrows are vertically excited from occupied states to unoccupied ones in the band structure of the GW level.

monolayers present linear dispersion which satisfies the Debye model, while the out-of-plane acoustic modes (ZA) exhibit a quadratic form which is the main characteristic of two-dimensional materials.⁷¹ The crossing behavior of the in-plane

and out-of-plane phonon branches indicated the well-separated nature of the former and the latter and reflected the planar nature of the group III–V materials. Another intriguing feature is the presence of an obvious phonon gap. The presence of this large



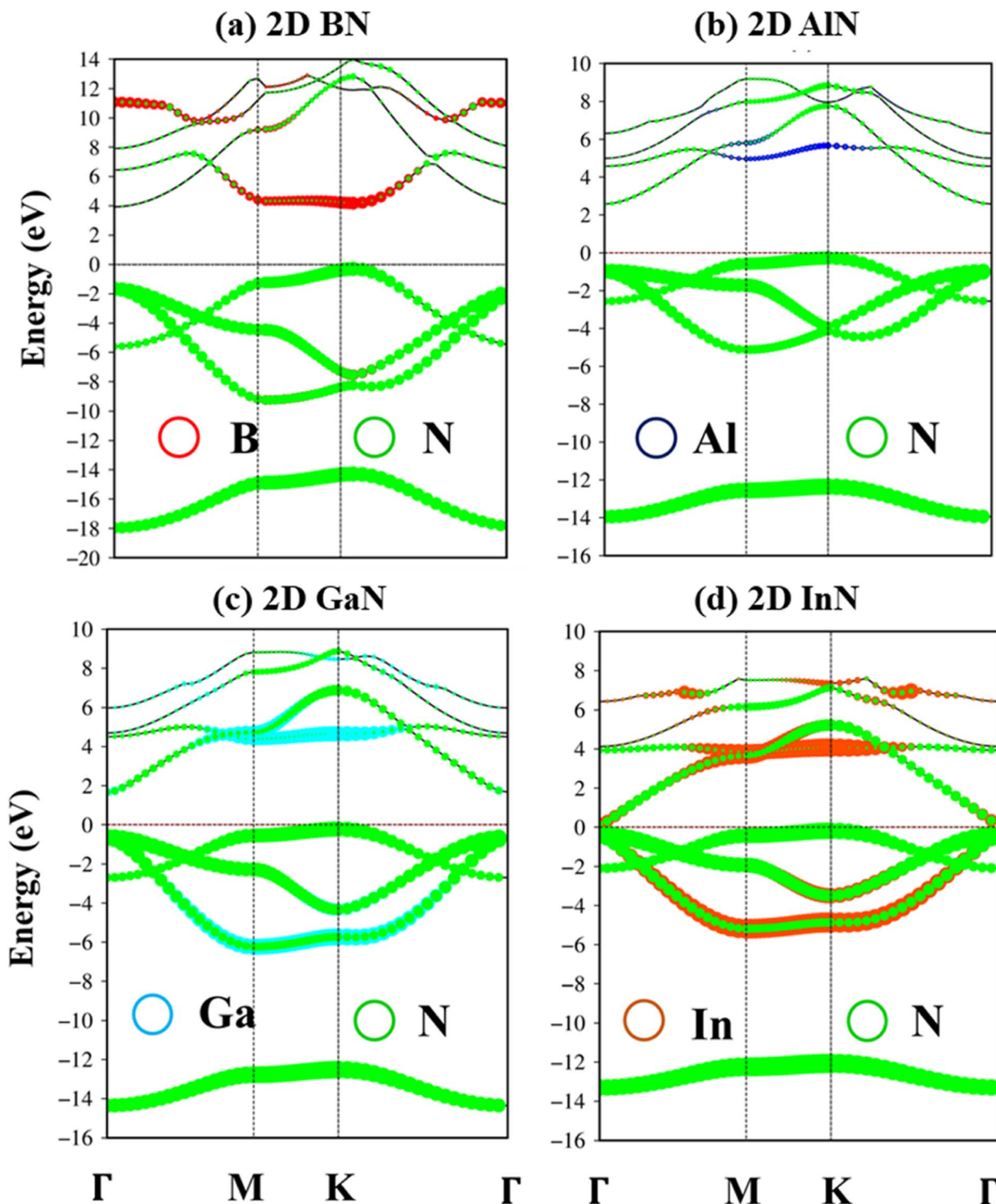


Fig. 5 (a–d) B, Al, Ga, In, and N atoms of BN, AlN, GaN, and InN two-dimensional monolayers are represented by red, blue, cyan, orange, and green balls, respectively.

acoustic-optical phonon spacing originated from the atomic mass difference in the unit cell. In general, the phonon band gap of group III–V monolayers which can describe the difference between low-frequency acoustic and high-frequency optic branches could be classified in the following order: BN (1.0 THz), AlN (5.0 THz), GaN (10.5 THz), and InN (11.5 THz) as expected.

Such a frequency gap is a typical feature of phonon transport in group-III nitride semiconductors⁷² and also denoted the absence of phonon scattering between optical (except the ZO branch) and acoustic branches. Important properties we would like to emphasize are the absence of imaginary frequencies in the phonon spectra, and thus, all studied two-dimensional materials



are dynamically stable. As shown in Fig. 3a–d, the corresponding phonon density of states reveals that B/Al/Ga/In atoms are the principal contributors to the low-frequency acoustic branches but the high-frequency optical modes are mostly dominated by the N atoms. More importantly, the vanishing frequencies between the optic mode and acoustic branch states of these materials, which are increased, are equal to phonon band gaps. Consequently, the remarkable chemical tunability and orbital interactions of N atoms will drastically alter phonon characteristics and have the potential to greatly reduce lattice thermal conductivity.

The electronic band structures of group III–V monolayers are shown in Fig. 4 and 5a–d with an effective range within the interval of -20.0 eV to 14.0 eV. There are four energy sub-bands in the occupied states which are related to the four active orbitals. The occupied state and the unoccupied one are highly asymmetric to each other around the Fermi level indicating complicated hybridizations. The presence of several critical points, the band edge states with the vanishing of the group velocity, will induce strong Van Hove singularities in the density of states and thus be responsible for the optical excitations. The absence of the linear and isotropic Dirac cone and the presence of a large quasi-particle band gap reflected the broken hexagon symmetry and the presence of the onsite energy differences. The

electronic band gap values are calculated at various levels of theories, such as DFT, DFT + spin-orbit coupling (SOC), HSE06, and GW standards, as can be seen in Table 1. These energy gap values match well with those in previous studies. Remarkably, (i) the SOC influences are found to be negligible. (ii) HSE06 calculations modify the exchange energy with portions of the exact exchange, parameterized to ensure agreement with the experiment for specific properties over a specific data set of materials. On the other hand, the GW level goes beyond the mean-field theory and can calculate exact many-body terms of electron–electron interactions.^{73,74} The order of accuracy is illustrated as $\text{GW} > \text{HSE} > \text{DFT}$. (iii) More importantly, DFT uses the Kohn–Sham (KS) equation with the single-body method, making it a powerful tool to study ground state properties of condensed matter systems. However, its predictions of a fundamental energy gap are often smaller than experimental observations due to inaccurate exchange–correlation approximations. Therefore, we have used the high-precision GW calculation method to obtain an accurate band structure for these graphene-like monolayers. The observed increase in the electronic band gap can be attributed to the electron–electron self-energy effects, as extensively discussed in numerous prior publications.^{75–78} The GW gap (called a quasi-particle gap)

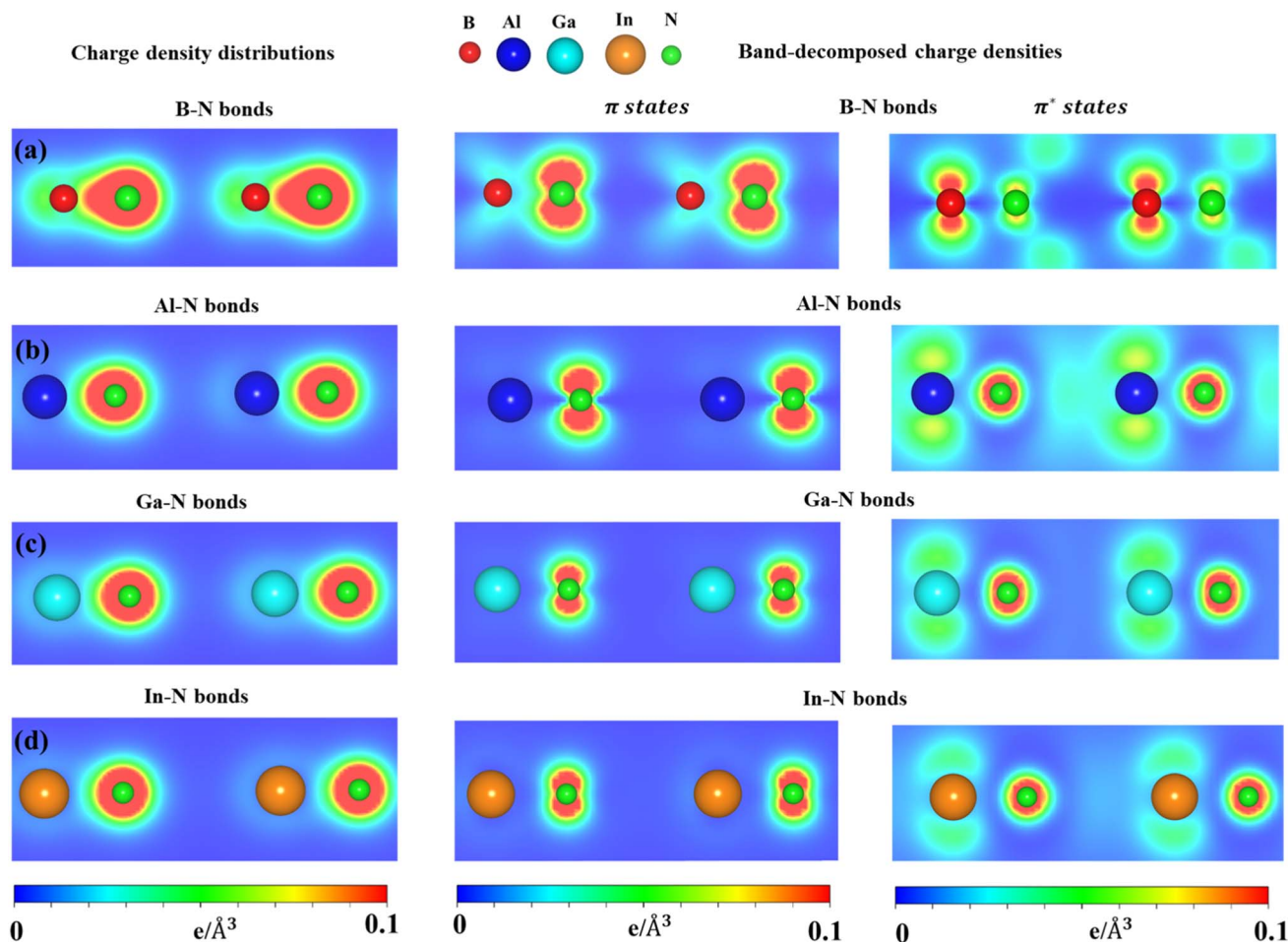


Fig. 6 (a–d) Charge density distributions and band decomposed charge densities of the BN, AlN, GaN, and InN monolayers, respectively. The red, blue, cyan, yellow, and green balls are the B, Al, Ga, In, and N atoms.



agrees closely with the electronic gap measured in scanning tunneling spectroscopy (STS) experiments. This means that the GW gap is comparable to the experimental results.

The impure π sub-bands, which are governed by $2p_z/3p_z/4p_z/5p_z$ orbitals, starting from the K , get the saddle point at M and

end at the Γ . Beyond the π sub-bands, group III-V semiconductors' electronic band structures also present three impure σ energy subbands belonging to the s , p_x , and p_y active orbitals. Here, impure σ bonds, which originated from B-2s/Al-3s/Ga-4s/In-5s and N-2s orbitals, appear at the deepest valence band due

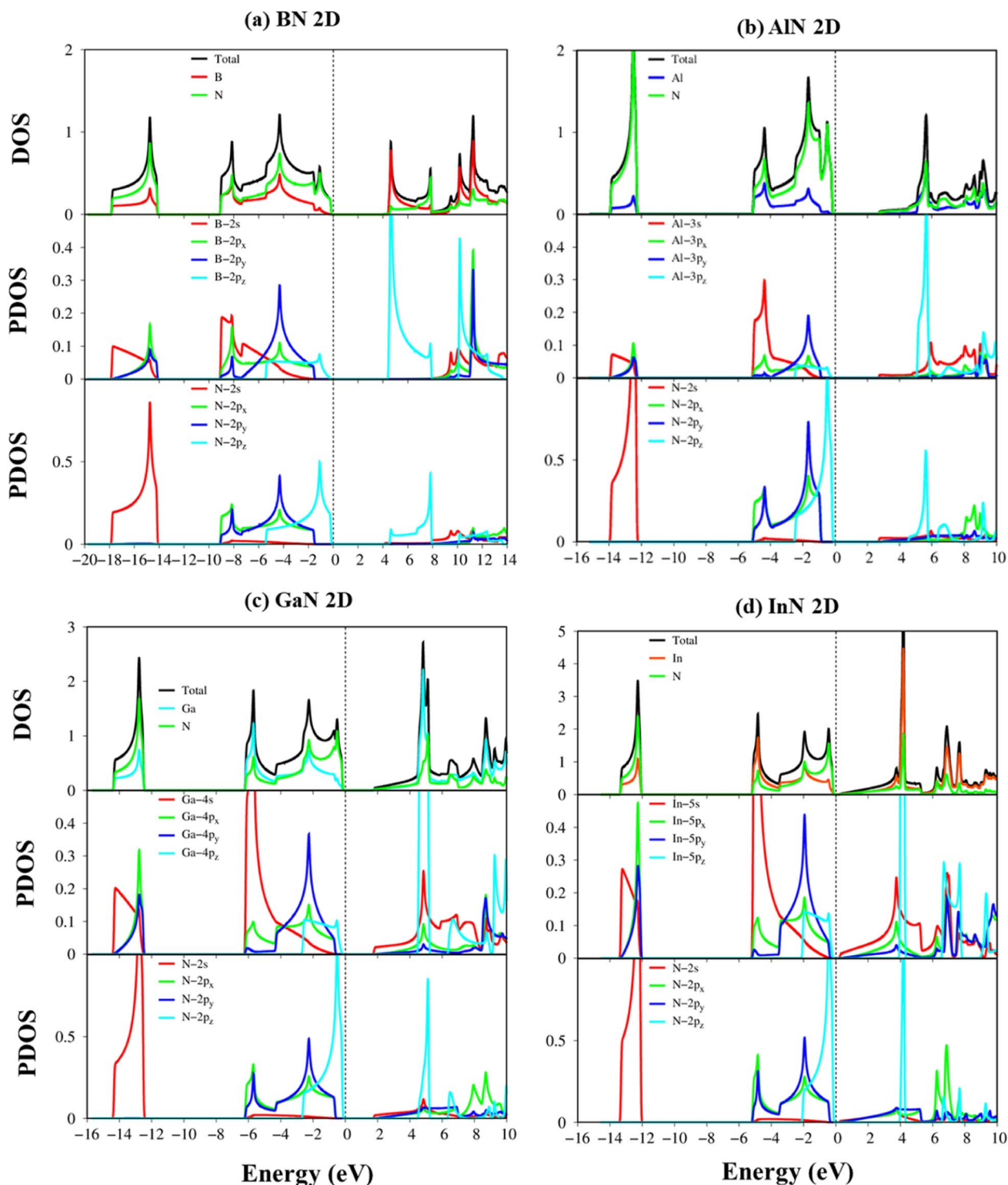


Fig. 7 (a–d) Van Hove singularities of the density of states (DOS) and project-density of states (PDOS) of the BN, AlN, GaN, and InN monolayers, respectively, and those coming from B-/Al-/Ga-/In-/N- orbitals (red, blue, cyan, yellow, and green curves), as well as B-(2s, $2p_x$, $2p_y$, and $2p_z$), Al-(3s, $3p_x$, $3p_y$, and $3p_z$), Ga-(4s, $4p_x$, $4p_y$, and $4p_z$), In-(5s, $5p_x$, $5p_y$, and $5p_z$), and N-(2s, $2p_x$, $2p_y$, and $2p_z$) orbitals (red, green, blue and cyan curves).



to the highest ionization energies. The crossing behavior of impure σ and π sub-bands can be observed, which reflects the well separated features of the former and the latter and is associated with the planar nature of the group III-V monolayers.

According to the dominance of specific atoms, the electronic band structure of group III-V materials is also classified into two distinct regions: (i) the B, Al, Ga, In, and N atoms contributed to the entire band structure under the red, blue, cyan,

yellow, and green circles (Fig. 5a-d) and nitrogen atoms are mostly in the occupied states. (ii) In contrast, the boron, aluminum, gallium, and indium atoms contributed mainly to unoccupied states of the electronic spectra. These behaviors, which connect with projected atom-/orbital hybridizations in PDOS (Fig. 7a-d discussed later), can provide sufficient information about hybridizations in the chemical bonds.

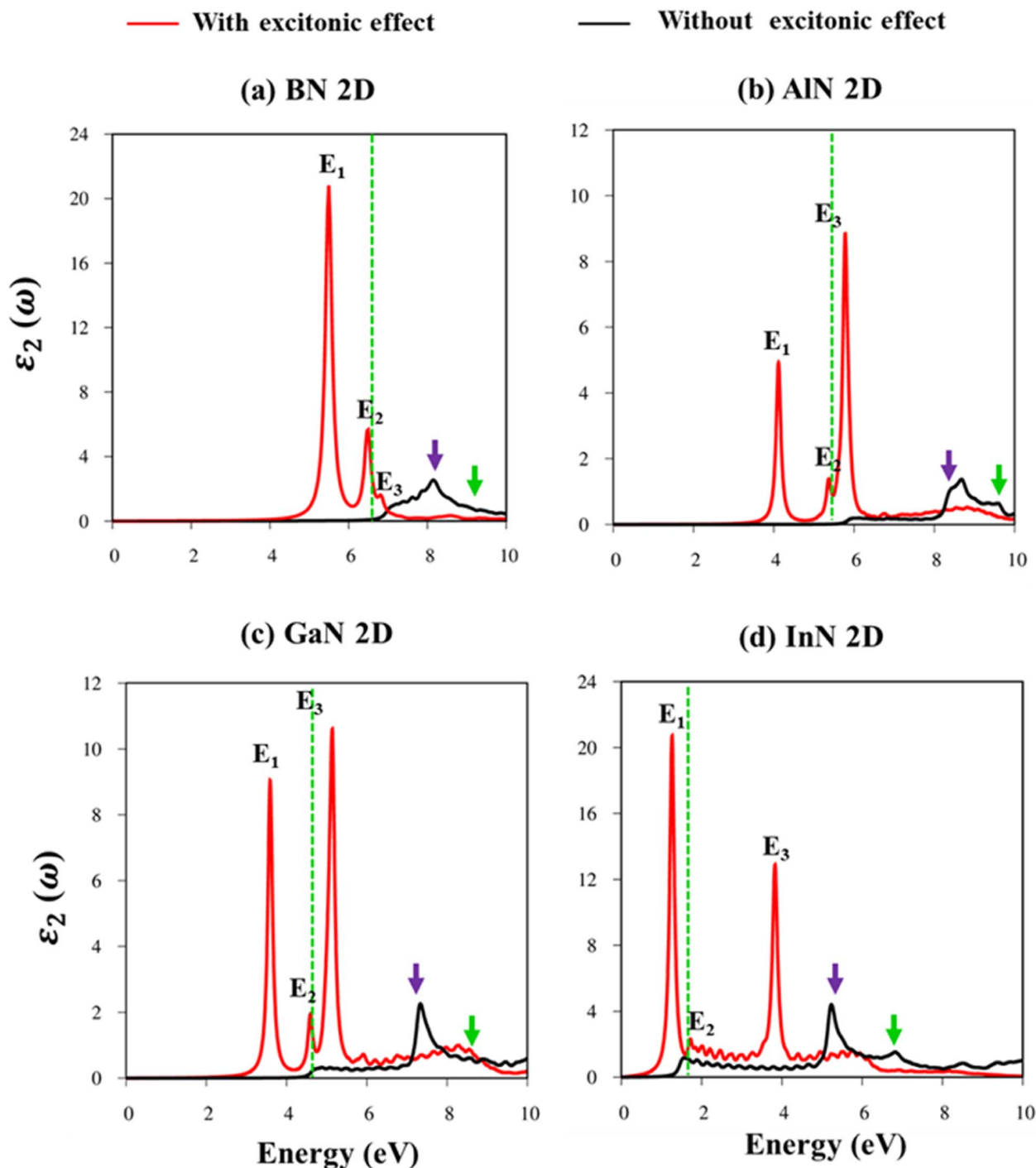


Fig. 8 (a–d) The imaginary part of the dielectric functions with and without exciton effects (red and black curves, respectively) of the BN, AlN, GaN, and InN monolayers, respectively. The dashed vertical greens correspond to the energy gap at the GW level.



Table 2 Prominent absorption structures: energy, label, colored indicators, and identified orbital hybridizations without electron–hole interactions in the B–N chemical bonds

Energy (eV)	Label	Distinct colors	Orbital hybridizations in B–N bonds
5.50	E ₁	Threshold frequency	
6.80	E ₂		
7.10	E ₃		
8.50		Purple	2p _z –2p _z
9.50		Green	2p _x , 2p _y –2p _x , and 2p _y

Table 3 Prominent absorption structures: energy, label, colored indicators, and identified orbital hybridizations without electron–hole interactions in the Al–N chemical bonds

Energy (eV)	Label	Distinct colors	Orbital hybridizations in Al–N bonds
4.10	E ₁	Threshold frequency	
5.35	E ₂		
5.80	E ₃		
8.30		Purple	3p _z –2p _z
9.30		Green	3p _x , 3p _y –2p _x , and 2p _y

The charge density distributions (Fig. 6a–d) and the highest occupied and lowest unoccupied band decomposed charge densities could provide useful information to understand the orbital hybridizations in the chemical bonds. As shown in Fig. 6, the B–N, Al–N, Ga–N, and In–N chemical bonds exhibit asymmetric charge density distribution ρ ; in particular, charge density is mostly located around the nitrogen atom which is associated with larger ionization energy. The asymmetric charge density also indicated the ionic-covalent nature of the B-/Al-/Ga-/In–N chemical bonds. The band decomposed charge density (Fig. 6) on the right panels indicated that the π bonding attends to the vacuum and is well separated from the σ bonding. Even though impure sp^2 hybridizations are present, group III–V monolayers could be classified from graphene *via* the impure π and impure σ bondings.

To fully comprehend the orbital hybridizations in the B-/Al-/Ga-/In–N chemical bonding, the orbital-projected density of states has been established and is displayed in Fig. 7a–d. The strong prominent peaks of van Hove singularities appear in the density of states in the density of states spectra. These prominent peaks mostly present the step functions or the divergence peaks, which have mostly arisen from the parabolic or the flat energy dispersions of two-dimensional systems. These prominent peaks are responsible for the strong optical excitations. The vanishing of the density of states of BN, AlN, GaN, and InN around the Fermi level is, respectively, about 4.22 eV, 2.84 eV, 1.89 eV, and 0.35 eV, which reflect the semiconducting behavior and are consistent with the electronic band structures (DFT level). Very importantly, the density of states of impure π electrons

Table 4 Prominent absorption structures: energy, label, colored indicators, and identified orbital hybridizations without electron–hole interactions in the Ga–N chemical bonds

Energy (eV)	Label	Distinct colors	Orbital hybridizations in Ga–N bonds
3.60	E ₁	Threshold frequency	
4.50	E ₂		
5.10	E ₃		
7.30		Purple	4p _z –2p _z
8.80		Green	4p _x , 4p _y –2p _x , and 2p _y

Table 5 Prominent absorption structures: energy, label, colored indicators, and identified orbital hybridizations without electron–hole interactions in the In–N chemical bonds

Energy (eV)	Label	Distinct colors	Orbital hybridizations in the In–N bonds
1.20	E ₁	Threshold frequency	
1.60	E ₂		
3.80	E ₃		
5.30		Purple	5p _z –2p _z
6.80		Green	5p _x , 5p _y –2p _x , and 2p _y

(due to the p_z orbitals) and impure σ electrons (owing to the s, p_x, and p_y orbitals) is well separated from each other, which confirms the impure sp^2 hybridizations of the planar systems.

When two-dimensional materials are perturbed by a transverse electromagnetic wave or photon beam, the electrons in the occupied states are vertically promoted to unoccupied states. All the charge carriers in two-dimensional systems will dynamically screen this external field and create induced current density and, thus, charge density fluctuations. The complex dielectric functions can provide useful information to fully comprehend optical transition mechanisms. The close connection between the initial and the final orbital characters and the prominent peaks in the optical absorbance spectra are the most concerning aspects of the current work.

Fig. 8a–d indicate the imaginary part of the dielectric functions $\epsilon_2(\omega)$ in the presence (absence) of excitonic effects, which are represented by red (black) curves. The optical absorbance spectra of group III–V materials exhibit two prominent peaks which arise from the specific Van Hove singularities; *e.g.*, the BN monolayer exhibits two prominent peaks at 8.50 eV and 9.50 eV, which are due to the excitations of $\pi \rightarrow \pi^*$, and $\sigma \rightarrow \sigma^*$ states at the K and M critical points, respectively (Fig. 8a). Such identification could be achieved by the delicate connection between the projected density of states and the prominent excitations, that is, the strong Van Hove singularities will induce large optical excitation peaks. As a consequence, the orbital hybridizations related to optical responses are indicated in Tables 2–5, while the vertical excitations that connect the occupied states to the unoccupied





Fig. 9 (a–d) The real part and imaginary part of the dielectric functions with exciton effects (red and black curves, respectively) of the BN, AlN, GaN, and InN monolayers, respectively. The dashed green circles correspond to dielectric constants.

ones in the quasi-band structures are shown in Fig. 4a–d. The optical gap (E_g^0) at the threshold frequency is about 7.10 eV, 5.80 eV, 5.10 eV, and 1.41 eV for the BN, AlN, GaN, and InN monolayers, respectively, which are consistent with the sizable-large gap of two-dimensional semiconductors. In addition to the optical gap, the extra prominent peaks (E_1 , E_2 , and E_3) exhibit significant redshifts when the electron–hole interactions (excitons) are taken into account, *e.g.*, about 1.30 eV, 1.25 eV, 0.90 eV, and 0.40 eV for the optical gap of the BN, AlN, GaN and InN monolayers, suggesting strong energy

bindings/robust excitonic effects. Such a large change in the optical spectrum indicated that the excitonic effects are significant and the electron–hole bound states may survive at high temperatures. The strong electron–hole bound states originated from the insignificant electronic screening (large electronic band gap and small $\epsilon_1(0)$ in Fig. 9a–d) and the absence of vertical electronic screening due to the quantum confinement effects. The zero points in dielectric functions, respectively 7.5 eV, 6.5 eV, 9.2 eV, and 6.8 eV for two-dimensional BN, AlN, GaN, and InN monolayers are equal to





Fig. 10 (a–d) Reflectance spectra, energy loss functions, and absorbance spectra in the presence of an excitonic effect, respectively for the BN, AlN, GaN, and InN monolayers.

plasmon mode in energy loss functions (discussed later in Fig. 10). On the other hand, the real part $\epsilon_1(\omega)$ and the imaginary part $\epsilon_2(\omega)$ of dielectric functions are connected by using the Kramers–Kronig relationship,⁷⁹ which leads to the appearance of prominent peaks simultaneously as can be seen in Fig. 9a–d. It is interesting to note that there exists an inverse relationship between the band gap energy values (E_g) and the dielectric constants ($\epsilon_1(0)$) in various monolayers⁸⁰ ranging from BN to InN. Specifically, from BN to InN monolayers, the values of band gap energy decreased while the values of dielectric constant increased. For instance, the band gap

energy values for 2D BN, AlN, GaN, and InN monolayers are 6.44 eV, 5.18 eV, 4.32 eV, and 1.41 eV, respectively, while their corresponding dielectric constant values are 1.6, 1.8, 2.0, and 4.8.

The energy loss function (ELF), being determined using the $\text{Im}\left[\frac{-1}{\epsilon(\omega)}\right]$, could provide useful information about the collective charge excitations. Each prominent peak in the ELF will denote a plasmon mode, the quantization of the collective excitation. The ELF of group III-nitride monolayers (Fig. 10) exhibits two prominent excitation peaks similar to that of graphene, *e.g.*,



4.0 eV and 7.8 eV for the BN monolayer; the former belongs to the coherent excitation of the impure π charges (B-2p_z and N-2p_z), while the latter is related to the excitation of all carriers in the low-dimensional systems. It is very important to note that the low-frequency plasmon due to the free charge is absent in BN two-dimensional semiconductors. The other optical properties such as absorbance $\alpha(\omega)$ and reflectance $R(\omega)$ are also calculated and shown in Fig. 10. The reflectivity $R(\omega)$ is weakly dependent on frequency when a photon frequency is smaller than the onset frequency, whereas the absorption coefficient $\alpha(\omega)$ disappears owing to the absence of the electronic excitation contribution. Beyond the threshold frequency, both $\alpha(\omega)$ and $R(\omega)$ are changing significantly and sensitively in the excitation model. The diverse inter-band transitions are responsible for the rapid increase in the absorption coefficient. The inverse values of the absorption coefficient are equal to 50.0 Å, 150.0 Å, 130.0 Å, and 250.0 Å for the 2D BN, AlN, GaN, and InN monolayers, respectively, indicating that the photon energy propagating in the medium is easily absorbed by the electronic excitations.

The examination of the rich and unique electronic and optical properties of group III-V materials required high-resolution equipment. For instance, the electronic band structure including various energy dispersions can be depicted by using ARPES⁸¹ measurement, and the STS⁸² examination can provide asymmetric or shoulder Van Hove singularities around the Fermi level. Being supported by Kramers–Kronig relations,⁷⁹ the results of the adsorption, reflection, or transmission measurements can be compared with our absorption, reflection coefficients, and energy loss functions. Other information, such as diversified optical excitation peaks, the slight redshift of the optical gap due to the excitonic effect, and the strong anisotropy of the optical responses can be obtained. Our analysis of orbital hybridizations that are related to optical properties is very difficult to achieve in experiments.

4 Conclusions

To summarize, this study employs first-principles calculations to analyze the electronic and optical properties of group III-nitride monolayers. Here, the studied materials exhibit rich and unique geometric, electronic, and optical properties. Similar to graphene, these systems also exhibit planar and uniform hexagonal honeycomb structures. Also, sizable electronic indirect band gaps in semiconductors and insulators of 4.22 (6.44 eV), 2.84 eV (5.18 eV), 1.89 eV (4.32), and 0.35 eV (2.43) for DFT (GW) approximations were found in BN, AlN, GaN and InN monolayers, respectively. The charge density, the band decomposed charge density, and the Van Hove singularities in the density of states are reflected in the main orbital characters of the concerned materials. In addition, these monolayers belong to impure sp² hybridizations which are due to the orbital hybridizations of impure π and impure σ orbitals.

More importantly, these graphene-like materials demonstrate distinctive and diverse optical properties, characterized by strong excitonic effects, and were able to establish a correlation between the initial and final states of orbital hybridizations

and the prominent excitation peaks observed in the optical spectra. Moreover, these monolayers exhibit low reflectance spectra, high absorption coefficients, and distinctive plasmon modes in the energy loss functions, further highlighting their unique optical properties.

In short, our study provided valuable insights into the physical and chemical characteristics. These findings are of significant interest and suggest that these materials could be useful for a broad range of applications. Moreover, the theoretical framework used in this research can be extended to analyze other graphene-like and emerging materials, thereby opening new avenues for future research.

Conflicts of interest

The authors declare no competing interests.

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