ChemComm



COMMUNICATION

View Article Online



Cite this: Chem. Commun., 2018, 54 10160

Received 28th June 2018, Accepted 14th August 2018

DOI: 10.1039/c8cc05178j

rsc.li/chemcomm

Reductive cyclisations of amidines involving aminal radicals†

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Amidines bearing simple alkenes undergo aminal radical cyclisation upon treatment with Sml₂. The mild, reductive electron transfer process delivers medicinally-relevant, polycyclic quinazolinone derivatives in good to excellent yield and typically with complete diastereocontrol.

Nitrogen-containing heterocycles are ubiquitous components in the molecular architectures of natural products, materials and drug candidates.1 As a feature in biologically active alkaloids,2 the quinazolinone ring system is a significant member of the family and its presence in nature has inspired the search for synthetic quinazolinones with medicinal potential (Scheme 1A).3 Although various approaches to these polycyclic scaffolds have been described, expedient, stereoselective synthetic strategies to quinazolinones that operate under mild conditions on simple, readily accessible substrates, are of high value.

Radical cyclisations have emerged as an important tool for the efficient generation of complex polycyclic products.⁵ However, few radical cyclisation strategies have been developed for the synthesis of quinazolinone analogues.⁶ Of these, Malacria, Courillon and Fensterbank have described an elegant radical cyclisation approach using tributyltin hydride and applied the method to a synthesis of Luotonin A. 6a-f Weaver and Bowman have also reported a radical cyclisation approach to quinazolinones using tributyltin hydride. 6g Recently, Chiba reported an elegant oxidative radical rearrangement that constructs quinazolinones. 6h,i Finally, quinazolinone scaffolds have been accessed using radical processes involving $\mathrm{Ag}(\mathfrak{l})^{6j}$ and $\mathrm{Cu}(\mathfrak{l}),^{6k}$ visible-light photoredox catalysis, 61 and DTBP in a metal-free process. 6m Despite these reports, 6 there remains a need for a straightforward method that constructs polycyclic quinazolinones under mild conditions.

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† Electronic supplementary information (ESI) available: General experimental procedures, characterization details, $^1\!H$ and $^{13}\!C$ NMR spectra of compounds, X-ray crystallographic data for 2a, 2g and 2t, nOe study and analysis of coupling constants for 2u. CCDC 1846530-1846532. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c8cc05178j

A. Selected biologically-active quinazolinone natural products

B. Cyclisations of aminal radicals formed from amidines by SET

$$\begin{array}{c|c}
R^1 & O & O \\
\hline
N & 2 & O \\
R^2 & (from Sml_2)
\end{array}$$

$$\begin{array}{c|c}
R^1 & O & O \\
\hline
N & R^2 & O \\
\hline
N & R^2
\end{array}$$

■ simple alkene radical acceptors ■ high diastereocontrol & high yield ■ medicinally-relevent guinazolinone products

Scheme 1 (A) Selected biologically-active quinazolinone natural products (B) this work: the cyclisation of aminal radicals, formed from amidines by SET, provides efficient access to quinazolinones.

We recently reported the first radical reduction, cyclisations and cyclisation cascades involving radical anions generated from urea-type carbonyls by single electron transfer (SET).7 During the study, we found that the aminal radical anion intermediates could be generated and trapped by tethered alkenes to form heterocyclic products.7a Related aminal radicals were formed and trapped, in an intermolecular sense, by Beaudry in his highly effective cross-coupling of amidines with electrondeficient alkenes.⁸ Both processes are mediated by the reductive SET reagent, samarium iodide (SmI2, Kagan's reagent).9 This highly versatile, commercially available or readily prepared reagent often proves to be the only viable mediator of challenging radical cyclisations and cyclisation cascades designed to deliver high value products not easily accessible by alternative means. 10 Crucially, in Beaudry's study, only two intramolecular examples of amidinealkene coupling were described and in all examples, both interand intramolecular, alkenes bore strongly electron-withdrawing groups.8a We recognised that the intramolecular SmI2-mediated coupling of amidines with simple unactivated alkene radical acceptors could provide expedient access to important quinazolinones. Herein, we describe the first general study of aminal radical cyclisations triggered by SET reduction of amidines using

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Table 1 Optimisation of amidine-alkene radical cyclisations^a

				Yield ^b [%]	
Entry	SmI ₂ (equiv.)	NH ₄ Cl (equiv.)	H ₂ O (equiv.)	1a	2a
1	3	_	_		50
2	3	3	_	_	57
3	3	_	100	_	61
4^c	3	_	_	_	60
5	3	_	3	_	55
6	3	_	20	_	58
7^d	3	_	100	_	26
8^e	3	3	_	_	84 (81) ^f

^a Reaction conditions: 1a (0.1 mmol, in THF) under N₂, was added proton source followed by SmI2 (0.1 M in THF). The reaction was quenched after 2 h. b Yield was determined by H NMR spectroscopy using 2,3,5,6-tetrachloronitrobenzene as internal standard. c 10 equiv. t-BuOH was added. d 100 equiv. LiBr was added. e The SmI $_2$ solution was added over 1 h by syringe pump. f Isolated yield.

SmI₂ (Scheme 1B). The radical cyclisations deliver quinazolinones in good yield and typically with complete diastereocontrol.

We began our studies by optimising the cyclisation of 1a; efficiently synthesised in one step from commercial 4-hydroxyquinazoline. Pleasingly, the desired cyclisation product 2a was obtained in 50% yield upon treatment with SmI₂ (Table 1, entry 1). The fact that SmI2, in the absence of additives that increase the reducing power of the reagent, can reduce 1a highlights the reactive nature of the N-acyl amidine functional group relative to, for example, amides, 11a acids, 11b esters, 11c and nitriles. 11d Drawing on the observations of Beaudry, NH₄Cl proved to be an effective proton source in the reductive coupling, and its use gave 2a in 57% yield (entry 2). Using the more established proton sources, H2O and t-BuOH, resulted in the formation of 2a in 61% and 60% yield, respectively (entries 3 and 4). When the amount of H₂O was reduced, the yield of 3a did not improve (entries 5 and 6). The use of LiBr as an additive in combination with H₂O¹² gave 2a in a lower 26% yield (entry 7). The key to further improvement in the yield of 2a proved to lie in the rate of addition of the SET reagent; slow addition of SmI2 gave 2a in 81% isolated yield (entry 8). It is likely that slow addition prevents the over-reduction of aminal radical I that would compete with radical cyclisation. The combination of SmI2 and H2O was clearly too reducing for the amidine substrate (even with slow addition of SmI₂) and thus Beaudry's NH₄Cl additive was used in further studies.

Using the optimised conditions, we have explored the generality of the amidine-alkene radical cyclisation (Scheme 2). In all cases, the desired quinazolinone products of cyclisation were obtained with complete diastereocontrol (>95:5 dr) and in good to excellent yield. Various functional groups on the alkenyl aryl ring were found to be compatible with the reductive conditions, including methoxy (2c, 2d), bromo (2e), chloro (2h), trifluoromethyl (2f), and acetal (2g). Furthermore, the presence of medicinally-relevant heteroaromatic rings including indole (2k), benzothienyl (2l and 2m), and

Scheme 2 Scope of the amidine radical cyclisation. Reaction conditions: **1a** (0.1 mmol) and NH₄Cl (0.3 mmol) in THF (2 mL) under N_2 , was added Sml₂ (0.1 M in THF, 3 equiv.) over 1 h using a syringe pump. The reaction was quenched after another 1 h. Isolated yield.

82% >95:5 dr

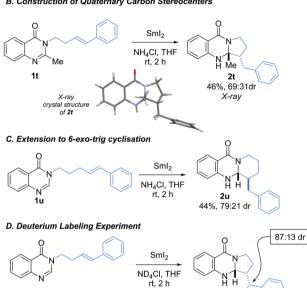
>95:5 dr

thienyl (2n) did not impede radical cyclisation. Finally, various functional groups on the benzenoid ring of the 4-quinazolinone motif, including bromo (20 and 2p), fluoro (2q and 2r) and methoxy (2s), were also tolerated in the radical cyclisation. The relative configuration of the quinazolinone products was assigned after X-ray crystallographic analysis of 2a and 2g.13

The radical cyclisation could be carried out on gram-scale with no loss of efficiency: Using the optimised conditions, 1a ChemComm Communication

A. Gram-scale Experiment Sml₂ NH₄CI, THF rt. 2 h 1a - 4 mmol. 1.1 a 2a - 0.8 g, 75%

B. Construction of Quaternary Carbon Stereocenters



E. Origin of diastereoselectivity

2a-D 80% 100% D incorporation

 $K_H/K_D = 1.18 \pm 0.1$

Scheme 3 (A) A gram-scale experiment; (B) construction of quaternary carbon stereocenters; (C) extension to 6-exo-trig cyclisation; (D) deuterium labelling experiment; (E) origin of diastereoselectivity

(4.0 mmol, 1.1 g) was converted to 2a (3.0 mmol, 0.83 g) in 75% yield (Scheme 3A). In addition, radical cyclisation of 2-methyl quinazolin-4-one 1t gave 2t containing a quaternary carbon center in 46% isolated yield as 2:1 mixture of diastereoisomers (Scheme 3B). The structure of the major diastereoisomer of 2t was confirmed by X-ray crystallographic analysis. 13 Substrate 1u underwent a challenging 6-exo-trig cyclisation to give 2u in 44% isolated yield with moderate diastereocontrol. In this case, the anti diastereoisomeric product predominated, as determined by NOE and also supported by the comparison of measured and calculated coupling constants (Scheme 3C).14 A deuterium labeling experiment was also performed using SmI₂-ND₄Cl. As expected, the labelled cyclisation product 2a-D was obtained, thus confirming that the cyclisation is terminated by protonation of a benzylic organosamarium (100% D incorporation, 83:17 dr at the labelled benzylic position). 15 The KIE measured for this reductive cyclisation suggests that proton transfer is not involved in the rate-determining step (Scheme 3D). Finally, a likely transition structure for the 5-exo-trig radical cyclisation that explains the origin of the syn-diastereoselectivity is shown in Scheme 3E. In contrast, to

the cyclisations of ketyl radical anions that often proceed to give anti-products, 16 the cyclisation of neutral aminal radical I, formed by protonation of a radical anion after SET or by protonation of the amidine prior to SET, favours cyclisation via syn-transition structure 3.

In summary, reductive amidine-alkene radical cyclisations, involving the intramolecular addition of aminal radicals to simple alkenes, deliver polycyclic quinazolinones. The radical process is mediated by single electron transfer from commercially available SmI2, operates under mild conditions on readily-available substrates, proceeds with complete diastereocontrol, and delivers a range of medicinally-relevant, quinazolinone derivatives in good to excellent yield.

We thank the EPSRC (EPSRC Established Career Fellowship to D. J. P.; EP/M005062/1), EPSRC Core Capability Grant (EP/K039547/1), the Leverhulme Trust (Research Fellowship to D. J. P.; RF-2013-286), and the University of Manchester (President's Scholarship to H. H.) for funding.

Conflicts of interest

There are no conflicts to declare.

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