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Gold-catalyzed (4 + 2)-annulations between α -alkyl alkenylgold carbenes and benzisoxazoles with reactive alkyl groups†

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This work reports new (4 + 2)-annulations of α -alkyl vinylgold carbenes with benzisoxazoles to afford 3,4dihydroquinoline derivatives with high anti-stereoselectivity. The annulations are operable with carbenes in both acyclic and cyclic forms. This reaction sequence involves an initial formation of imines from αalkylgold carbenes and benzisoxazoles, followed by a novel carbonyl-enamine reaction to yield 3,4dihydroquinoline derivatives. This system presents the first alkyl C-H reactivity of α -alkyl gold carbenes with an external substrate.

Introduction

Metal carbenes are versatile intermediates to implement a vast number of useful reactions including cyclopropanation, X-H insertion (X = C, N and O), skeletal rearrangement and annulation reactions (eqn (1)).1 Despite their widespread applications, applicable metal carbenes, derived from diazo precursors, are mainly restricted to donor/acceptor (D/A) types I (R = H, aryl and alkenyl; EWG = CN, ketones and esters)whereas highly desirable α-alkyl metal carbenes II are less efficient because of a competitive 1,2-hydrogen shift to form olefins (eqn (2)).1 This side reaction is particularly serious for gold carbenes because their $LAu = C^{+}$ carbons are highly cationic.² Few intermolecular reactions involving Ar-Pd(II) catalysts focused on α-alkyl metal carbenes of D/A types.3 The limited utility of α-alkyl carbenoids features an unsolved and challenging task in metal carbene chemistry. We seek new α -alkyl carbenoids beyond commonly used D/A carbenes II, aiming at two objectives: (i) suppression of a 1,2-H shift and (ii) an alkyl C-H reaction with an external substrate.

Interest in the reactions of benzisoxazoles is rapidly growing in gold catalysis because of their various annulation modes with gold π -alkynes. 4-6 To explore the reactivity of benzisoxazoles toward gold carbenes,7 we first tested the reactions with D/Atype benzyl α -oxogold carbene II' (R = Ph and EWG = CO_2Et), yielding an olefin product III' efficiently (eqn (3)). We envisage that D/D type carbenes such as α -alkyl alkenylgold carbenes IV

Currently used carbenes: D/A types cyclopropanation, X-H insertion. rearrangement, annulation products `FWG M = Rh(II), Au(I), Cu(I)EWG = CN. ketone, este Ag(I), Zn(II), Pd(II) Our initial tests CO₂Et III' (>70%) reactive alkyl C-H bonds: no 1.2-hydrogen shift

might be viable species to achieve new annulations with benzisoxazoles because their gold-stabilized allyl cation character IV is unfavorable for a 1,2-H shift. According to this hypothesis, this work reports novel intermolecular (4 + 2)-annulations between α-alkyl vinylgold carbenes and benzisoxazoles, thus manifesting an unprecedented C-H reactivity of α-alkyl metal carbenes.

Results and discussion

As shown in eqn (5), we further tested the reaction of acyclic alkylgold carbenes A that were generated in situ from cyclopropene derivatives 1a-1b and gold catalysts.8 With IPrAuCl/ AgSbF₆, quinoline derivatives 3a and 3b were isolated in satisfactory yields (72-75%), together with enones 1a-O and 1b-O in minor proportions (17-19%). A 1,2-hydrogen shift was

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effectively suppressed with vinylcarbenes A, supporting our

Our primary interest is to construct complicated frameworks via cascade reactions. Fig. 1 depicts several bioactive compounds (VI-1)-(VI-6) bearing a common tricyclic framework VI, which can be easily constructed from cyclopentenylgold carbene A' and benzisoxazole. Indenoquinoline (VI-1) showed antiproliferative activities against breast (MCF-7) and lung epithelial (A-549) cells.94 Species VI-2 and VI-3 served as 5HT2c agonists and CRTH₂ receptor modulators, respectively. 9b,c Compounds VI-4 and VI-5 were N-containing steroids found in higher plants.9d,e Species VI-6 is a key intermediate for the total synthesis of naturally occurring (-)-isoschizogaline^{9f} and (-)-isoschizozygamine.9g

In this new task, we optimized the annulation cascades between vinylallene 4a and benzisoxazole 2a in dichloromethane (DCM) using various gold catalysts; species 4a serves as a precursor for cyclopentenylgold carbene A' (Table 1).10

An initial test of IPrAuCl/AgSbF₆ at a 5 mol% loading afforded a new azacyclic product 5a and cyclopentadiene 4a' in 62% and 25% yields, respectively (entry 1); the latter was derived from a 1,2-H shift of gold carbenes A' that was generated from cyclizations of gold-stabilized pentadienyl cation A-I. Notably, an increased gold loading (10 mol%) enhanced the yield of desired 5a up to 85%. Other gold catalysts LAuCl/AgSbF₆ (L = P(OPh)₃, PPh₃ and P(t-Bu)₂(obiphenyl)) gave 5a in 40-82% yields with $L = P(OPh)_3$ being the most effective (entries 3-5). For various silver salts as in IPrAuCl/AgX (X = OTf and NTf₂), resulting 5a was obtained in 65% and 71% yields, respectively (entries 6-7). AgNTf₂ was entirely inactive (entry 8). IPrAuCl/AgSbF₆ in various solvents gave 5a in the following yields: DCE 70%, MeCN 20% and 1,4dioxane 0 (entries 9-11). The molecular structure of compound 5a was characterized with X-ray diffraction,11 showing an anti-configuration between the alcohol and phenyl groups.

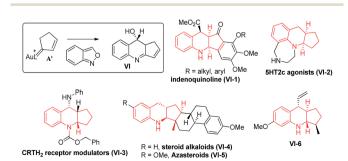


Fig. 1 Suitable alkylgold carbenes to access bioactive molecules.

Table 1 Catalytic reactions with various gold catalysts

				Yield ^b [%]		
Entry	Catalyst [mol%]	Solvent	t [h]	4a	5a	4a′
1	IPrAuCl/AgSbF ₆ (5)	DCM	12	8	62	25
2	IPrAuCl/AgSbF ₆ (10)	DCM	3	_	85	12
3	(PhO) ₃ PAuCl/AgSbF ₆ (10)	DCM	3	_	82	16
4	Ph ₃ PAuCl/AgSbF ₆ (10)	DCM	4	_	55	36
5	LAuCl/AgSbF ₆ (10) ^c	DCM	3	_	40	52
6	IPrAuCl/AgOTf (10)	DCM	4	_	65	26
7	IPrAuCl/AgNTf ₂ (10)	DCM	4	_	71	20
8	$AgSbF_6$ (10)	DCM	24	95	_	
9	IPrAuCl/AgSbF ₆ (10)	DCE	5	_	70	24
10	IPrAuCl/AgSbF ₆ (10)	MeCN	12	_	20	65
11	IPrAuCl/AgSbF ₆ (10)	Dioxane	10	_	_	90

 $^{a}\left[\mathbf{4a}
ight]=0.05$ M. b Product yields are reported after purification from silica column. ^c L = $P(t-Bu)_2(o-biphenyl)$. IPr = 1,3bis(diisopropylphenyl)imidazole-2-ylidene, DCE = 1,2-dichloroethane.

Table 2 assesses the generality of these gold-catalyzed reactions using various vinylallenes 4b-4t catalyzed with IPrAuCl/ AgSbF₆ (10 mol%) in DCM. All resulting products 5b-5t assumed anti-configurations with the alcohol and R¹ groups being mutually trans. We tested the reaction of trisubstituted

Table 2 Catalytic annulations with various alkenylallenes

^a [4] = 0.05 M. ^b Product yields are reported after purification from a silica column.

vinylallenes **4b–4f** bearing $R^1 = 4$ -MePh, 4-OMePh, 4-ClPh, 4-CF₃Ph and n-Bu, yielding desired **5b–5f** in 78–88% yields (entries 1–5). For species **4g** and **4h** bearing 3-phenyl substituents (X = OMe and Cl), their corresponding products **5g** and **5h** were obtained in 84% and 87% yields, respectively (entries 6 and 7). The reactions were extensible to other vinylallenes **4i–4k** bearing 2-naphthyl, 2-furan and 2-thiophene, further delivering desired products **5i–5k** in 82–84% yields (entries 8–10). We tested the reaction on vinylallene **4l** bearing distinct $R^1 = Me$ and $R^2 = Ph$, which yielded compound **5l** with an anticonfiguration in which the hydroxy and methyl groups are mutually *trans* (entry 11); this configuration was established by the ¹H NOE effect. Additional alkyl-substituted vinylallenes **4m**–

4p yielded desired **5m–5p** in satisfactory yields (80–85%, entries 12–15). Variations of the \mathbb{R}^2 group with an *n*-butyl group as in

species 4q gave expected product 5q in 86% yield (entry 16). We

prepared species $4\mathbf{r}$ bearing varied $R^2 = Ph$ and $R^3 = n$ -butyl, producing compound $5\mathbf{r}$ in 80% yield (entry 17). For 1,3-disubstituted vinylallenes $4\mathbf{s}$ and $4\mathbf{t}$ ($R^3 = H$), their resulting

compounds 5s and 5t were obtained in 82-83% yields (entries

18 and 19). We tested these new annulations on distinct substrates such as enynyl acetates 6a–6g, bearing varied phenyl (R=4- XC_6H_4 , X=H, Cl, Br, Me, and OMe), 2-thienyl and isopropyl substituents; these enyne acetates can be catalyzed with the same gold catalyst to yield distinct α -alkylgold carbenes A' (see Table 3). To our pleasure, new alkylgold carbenes A', generated from these enynyl acetates, were trapped efficiently with benzisox-azole 2a to afford the desired (4 + 2)-annulation products 7a–7g in satisfactory yields (61–74%), further manifesting the reaction generality (entries 1–7). For unsubstituted propargyl acetate 6h (R=H), its reaction led to a 68% recovery of initial 6h (entry 8). Even if the reaction is successful, a dehydration of compound 7h would occur to give quinoline products. The molecular

structure of compound 7a (R = Ph) was confirmed with

Table 3 Annulation reactions with enynyl acetates

 a **6** = 0.05 M. b Product yields are reported after purification from a silica column. c A 68% recovery of initial **6h** is found in entry 8.

Table 4 Catalytic annulations with various benzisoxazoles

 a ${\bf 4a}=0.05.$ b Product yields are reported after purification from a silica column.

X-ray diffraction analysis that revealed an *anti*-configuration (Table 3).¹¹

The scope of these catalytic reactions is further expanded with various applicable benzisoxazoles $2\mathbf{b}$ – $2\mathbf{j}$ substituted with the C(3), C(5) and C(6) carbons. Other C(5)-substituted benzisoxazoles $2\mathbf{b}$ – $2\mathbf{f}$ ($\mathbf{R}^1 = \mathbf{Me}$, OMe, Br, Cl, and $-\mathbf{OCO}_2\mathbf{E}\mathbf{t}$) maintained high efficiencies to deliver *anti*-configured products $\mathbf{8b}$ – $\mathbf{8f}$ in 80–90% yields (entries 1–5). High reaction efficiencies were maintained also for C(6)-substituted benzisoxazoles $2\mathbf{g}$ – $2\mathbf{i}$ that furnished products $\mathbf{8g}$ – $\mathbf{8i}$ in 86–92% yields (entries 6–8). A final applicable reaction with a C(3)-substituted benzisoxazole $2\mathbf{j}$ enabled the production of a tertiary alcohol $\mathbf{8j}$, reflecting the reaction feasibility (entry 9). 1 H NOE spectra were recorded to verify the stereochemistry of compound $\mathbf{8j}$ (Table 4).

Gold-catalyzed reactions of 3,5-dimethylisoxazole 2a' with vinylallenes 4a and 4u delivered 2-aminocyclopentadienes 9a and 9b in 72% and 64% yields, respectively (eqn (6)). 5a,13,14 The molecular structure of compound 9b was characterized with X-ray diffraction. Cyclizations of compounds 9a and 9b with a gold catalyst were unsuccessful because of the two different forms of the enol imines (eqn (6)). To rationalize the origin of the stereoselectivity, compound 5a was treated with $Zn(OTf)_2$ (20 mol%) in refluxing DCE to examine the hydroxyl epimerization that turned out to be slow. An equilibrium, anti/syn = 4:1, was attained for species 5a after reflux in DCE for 48 h (eqn (7)).

Scheme 1 shows the stereoselective functionalizations of anti-5a via NaBH₄ reductions and m-CPBA oxidations,

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Scheme 1 Chemical functionalizations.

Scheme 2 A plausible reaction mechanism.

respectively yielding compounds **5a-H** and **5a-O** as single diastereomeric products. The stereochemistries of compounds **5a-H** and **5a-O** were established with ¹H NOE spectra. Likewise, the acetate species **7a** was readily removed under basic conditions, yielding the enol form **7a'** as shown by its NMR in CD₃COCD₃ and CDCl₃. We also studied an O₃-induced oxidative cleavage of the acetate derivative **5a-OAc** to cleave the olefin group, yielding the peroxide **5a-O**₃ in 85% yield. The molecular structure of species **5a-O**₃ has been characterized by X-ray diffraction.¹¹

As depicted in Scheme 2, we postulate an initial formation of imines between alkylgold carbene **A** and benzisoxazole, yielding 2-iminoyl benzaldehyde **C**. This hypothesis is supported by our observation of 3,5-dimethylisoxazole, depicted in eqn (6). A tautomerization of imine species **C** is expected to form enamines **D** bearing an NH···O=C hydrogen bond. We believe that this enamine form, unlike other enamine-carbonyl couplings, is stabilized with the NH···O=C bond to enable a concerted process, analogous to the well-known carbonyl-ene reactions. A boat-like conformation **D** is favorable to yield *anti-*5 stereoselectively.

Conclusions

This work reports novel gold-catalyzed (4 + 2)-annulations between alkylgold carbenes and benzisoxazoles 2 to form 3,4-dihydroquinoline derivatives. Gold carbenes in cyclic and acyclic forms are both applicable. In this reaction sequence, the gold complex catalyzes an initial formation of imines between alkylgold carbenes^{13,14} and benzisoxazoles; the resulting intermediates bear an enamine moiety that is bound to an aldehyde via a hydrogen bond to induce a carbonyl-enamine reaction. Control experiments with 3,5-dimethylisoxazoles supported this postulated mechanism. This new synthetic design involving α -

alkyl metal carbenes of **D/D** types will attract growing interest because of its distinct utility.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

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Notes and references

- 1 (a) H. M. L. Davies and E. G. Antoulinakis, *Intermolecular Metal-Catalyzed Carbenoid Cyclopropanations in Organic Reactions*, ed. L. E. Overman, John Wiley & Sons, Inc., New York, NY, 2001, vol. 57, pp. 1–326; (b) H. M. L. Davies and R. E. J. Beckwith, *Chem. Rev.*, 2003, 103, 2861; (c) M. P. Doyle, R. Duffy, M. Ratnikov and L. Zhou, *Chem. Rev.*, 2010, 110, 704; (d) Q.-Q. Cheng, Y. Yu, J. Yedoyan and M. P. Doyle, *ChemCatChem*, 2018, 10, 488.
- For selected reviews for gold carbenes, see: (a) D. Qian and J. Zhang, Chem. Soc. Rev., 2015, 44, 677; (b) L. Liu and J. Zhang, Chem. Soc. Rev., 2016, 45, 506; (c) E. López, S. Gonzalez-Pelayo and L. A. López, Chem. Rec., 2017, 17, 312; (d) C. Obradors and A. M. Echavarren, Chem. Commun., 2014, 50, 16; (e) L. Zhang, Acc. Chem. Res., 2014, 47, 877; (f) D. P. Day and P. W. H. Chan, Adv. Synth. Catal., 2016, 358, 1368.
- 3 The reactions were only reported for Ar–Pd(II) species; see: (a) C. Peng, Y. Wang and J. Wang, *J. Am. Chem. Soc.*, 2008, **130**, 1566; (b) Z. Zhang, Y. Liu, M. Gong, X. Zhao, Y. Zhang and J. Wang, *Angew. Chem., Int. Ed.*, 2010, **49**, 1139.
- 4 (a) L. Li, T.-D. Tan, Y.-D. Tan, Y.-Q. Zhang, X. Liu and L.-W. Ye, *Org. Biomol. Chem.*, 2017, **15**, 8483; (b) D. B. Huple, S. Ghorpade and R.-S. Liu, *Adv. Synth. Catal.*, 2016, **358**, 1348; (c) S. S. Giri and R.-S. Liu, *Chem. Sci.*, 2018, **9**, 2991.
- 5 (a) A.-H. Zhou, Q. He, C. Shu, Y.-F. Yu, S. Liu, T. Zhao, W. Zhang, X. Lu and L.-W. Ye, Chem. Sci., 2015, 6, 1265; (b) X.-Y. Xiao, A.-H. Zhou, C. Shu, F. Pan, T. Li and L.-W. Ye, Chem.-Asian J., 2015, 10, 1854; (c) H. Jin, L. Huang, J. Xie, M. Rudolph, F. Rominger and A. S. K. Hashmi, Angew. Chem., Int. Ed., 2016, 55, 794; (d) H. Jin, B. Tian, X. Song, J. Xie, M. Rudolph, F. Rominger and A. S. K. Hashmi, Angew. Chem., Int. Ed., 2016, 55, 12688.
- 6 (a) Z. Zeng, H. Jin, J. Xie, B. Tian, M. Rudolph, F. Rominger and A. S. K. Hashmi, *Org. Lett.*, 2017, 19, 1020; (b)
 M. Chen, N. Sun, H. Chen and Y. Liu, *Chem. Commun.*, 2016, 52, 6324; (c) W. Xu, G. Wang, N. Sun and Y. Liu, *Org. Lett.*, 2017, 19, 3307.
- 7 The reactions of isoxazoles with rhodium carbenes were noted by Davies' group; distinct [3 + 3]-annulations were reported for α-alkenylrhodium esters as shown below. (*a*) J. R. Manning and H. M. L. Davies, *Tetrahedron*, 2008, **64**,

6901; (b) J. R. Manning and H. M. L. Davies, J. Am. Chem. Soc., 2008, 130, 8602.

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$$CO_2R$$
 Rh + NO CO_2R $3,3-shift$ CO_2R

- 8 (a) A. Archambeau, F. Miege, J. Crossy and C. Meyer, in Patai's Chemistry of Functional Groups, ed. Z. Rappoport, J. F. Liebman and I. Marek, John Wiley & Sons Ltd., Hoboken, NJ, 2014, pp. 631-700; (b) Y. Deng and M. P. Doyle, Isr. J. Chem., 2016, 56, 399-408; (c) F. Miege, C. Meyer and J. Cossy, Org. Lett., 2010, 12, 4144; (d) F. Miege, C. Meyer and J. Cossy, Chem.-Eur. J., 2012, 18, 7810; (e) C. Li, Y. Zeng and J. Wang, Tetrahedron Lett., 2009, 50, 2956; (f) Z.-B. Zhu and M. Shi, Chem.-Eur. J., 2008, 14, 10219; (g) S. B. Wagh, Y.-C. Hsu and R.-S. Liu, ACS Catal., 2016, 6, 7160; (h) Z. Liu, Q. Li, P. Liao and X. Bi, Chem.-Eur. J., 2017, 23, 4756; (i) W. Rao, M. J. Koh, D. Li, H. Hirao and P. W. H. Chan, J. Am. Chem. Soc., 2013, 135, 7926; (i) W. Rao and P. W. H. Chan, Chem.-Eur. I., 2014, 20, 713; (k) J. Yan, G. L. Tay, C. Neo, B. R. Lee and P. W. H. Chan, Org. Lett., 2015, 17, 4176.
- 9 (a) S. Chakrabarty, M. S. Croft, M. G. Marko and G. Moyna, *Bioorg. Med. Chem.*, 2013, 21, 1143; (b) Merck Sharp and Dohme Corp, WO 2012/174176 A1, 2012; (c) P. S. Ramamoorthy and R. E. McDevitt, *US Pat.*, US 2004/0019040, 2004; (d) J. Wöfling, É. Frank, G. Schneider, M. T. Bes and L. F. Tietze, *Synlett*, 1998, 1205; (e) L. F. Tietze and A. Modi, *Med. Res. Rev.*, 2000, 20, 304; (f) R. M. Kariba, P. J. Houghton and A. Yenesew, *J. Nat. Prod.*, 2002, 65, 566; (g) J. L. Hubbs and C. H. Heathcock, *Org. Lett.*, 1999, 1, 1315.
- 10 (a) S. Bhunia and R.-S. Liu, J. Am. Chem. Soc., 2008, 130, 16488; (b) J. H. Lee and F. D. Toste, Angew. Chem., Int. Ed., 2007, 46, 912; (c) H. Funami, H. Kasama and N. Iwasawa, Angew. Chem., Int. Ed., 2007, 46, 909; (d) G. Lemiere, V. Gandon, K. Cariou, T. Fukuyama, A. L. Dhimane,

- L. Fensterbank and M. Malacria, *Org. Lett.*, 2007, **9**, 2207; (e) V. Gandon, G. Lemiere, A. Hours, L. Fensterbank and M. Malacria, *Angew. Chem., Int. Ed.*, 2008, **47**, 7534; (f) M. R. Fructos, M. Besora, A. A. C. Braga, M. M. Díaz-Requejo, F. Maseras and P. J. Pérez, *Organometallics*, 2017, **36**, 172; (g) F.-Q. Shi, X. Li, Y. Xia, L. Zhang and Z.-X. Yu, *J. Am. Chem. Soc.*, 2007, **129**, 15503; (h) W. Rao, D. Susanti, B. J. Ayers and P. W. H. Chan, *J. Am. Chem. Soc.*, 2015, **137**, 6350; (i) W. Rao, J. W. Boyle and P. W. H. Chan, *Chem. Eur. J.*, 2016, **22**, 6532.
- 11 Crystallographic data of compounds 5a, 5a-O₃, 7a, and 9b were deposited at the Cambridge Crystallographic Data Center: 5a (CCDC 1819135); 5a-O₃ (CCDC 1819137); 7a (CCDC 1819138); and 9b (CCDC 1819136).†
- 12 L. Zhang and S. Wang, J. Am. Chem. Soc., 2006, 128, 1442.
- 13 For gold-catalyzed nitrene reactions of alkynes; see selected examples: (a) D. J. Gorin, N. R. Davies and F. D. Toste, J. Am. Chem. Soc., 2005, 127, 11260; (b) A. Wetzel and F. Gagosz, Angew. Chem., Int. Ed., 2011, 50, 7354; (c) B. Lu, Y. Luo, L. Liu, L. Ye, Y. Wang and L. Zhang, Angew. Chem., Int. Ed., 2011, 50, 8358; (d) C. Shu, Y.-H. Wang, B. Zhou, X.-L. Li, Y.-L. Li, Y.-F. Ping, X. Lu and L.-W. Ye, J. Am. Chem. Soc., 2015, 137, 9567; (e) H.-H. Hung, Y.-C. Liao and R.-S. Liu, J. Org. Chem., 2013, 78, 7970.
- 14 Benzisoxazoles serve as nitrene sources in rhodium-catalyzed C-H functionalizations, see selected examples: (a) S. Yu, G. Tang, Y. Li, X. Zhou, Y. Lan and X. Li, Angew. Chem., Int. Ed., 2016, 55, 8696; (b) M. Zou, J. Liu, C. Tang and N. Jiao, Org. Lett., 2016, 18, 3030; (c) S. Yu, Y. Li, X. Zhou, H. Wang, L. Kong and X. Li, Org. Lett., 2016, 18, 2812.
- 15 The reactions of enamines and aldehydes are generally implemented with Lewis acids or a base, see: (a) R. Matsubara, N. Kawai and S. Kobayashi, *Angew. Chem., Int. Ed.*, 2006, **45**, 3814; (b) T. Kochi, T. P. Tang and J. A. Ellman, *J. Am. Chem. Soc.*, 2003, **125**, 11276; (c) R. Matsubara, P. Vital, Y. Nakamura, H. Kiyohara and S. Kobayashi, *Tetrahedron*, 2004, **60**, 9769.